

TECHNICAL ASSESSMENT FOR THE SNAP PROGRAM

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INTRODUCTION

Section 612 of the US Clean Air Act of 1990 (Reference 1) requires the EPA to enact regulations making it unlawful to replace any Class I or Class II substance with any substitute that may impact human health or the environment and to publish lists of both prohibited and acceptable substitutes. The risk assessments required for each substitute will be conducted under the Significant New Applications Policy (SNAP) Program. In order for the EPA to make informed decisions about the possible substitutes, the New Mexico Engineering Research Institute was tasked under contract to ICF Incorporated to identify the current halon use areas and to suggest viable, near-term substitutes for the specific Halon 1211, 1301, and 2402 applications. Here, the term "substitutes" includes both replacement agents (halon-like chemicals) as well as alternatives (non-halon-like agents such as water, foam, and CO₂). This paper describes the effort undertaken to determine all the halon uses to date and to identify proposed substitutes.

HALON APPLICATIONS

A written and telephone survey was performed to identify the known Halon 1211, 1301, and 2402 uses to date. The survey resulted in the list of uses presented in Table 1. Halon 1301 and 1211 were indicated to be in wide use in a variety of applications in fixed systems and portable use, respectively. Halon 2402 was identified as having limited use as a fire extinguishing agent in the United States, except for possible replacement in existing systems (Reference 2). It has been used extensively in other countries, particularly in Italy, in Eastern European countries, and in the Soviet Republics. The properties of Halon 2402 make it most suitable for a streaming discharge from portable systems or in localized applications. At one time, Caterpillar, Inc., had a Halon 2402 fire extinguishing system in their tractors; trucks; and large-wheel loaders for engine compartments, transmission areas, and related non-occupied area use (Reference 3). However, they no longer make this unit. It has been reported that Boeing Aircraft has used Halon 2402 in some of their

TABLE 1. HALON USE SECTORS

1.	PORTABLE EXTINGUISHERS FOR FIXED FACILITIES
1.	Residential
2.	Commercial
3.	Military and Government
2.	AEROSPACE APPLICATIONS
1.	Handheld Extinguishers for On-Board Aircraft
2.	Ground-Based Aircraft Fire Protection
3.	Aircraft Cargo Bays
4.	Dry Bay Explosion Suppression
5.	Aircraft Engine Nacelle
6.	Spacecraft Fire Protection
7.	Aircraft Manufacture
3.	MARINE APPLICATIONS
1.	Crew Quarters
2.	Machinery Spaces
3.	Control Rooms
4.	Automobile Ferry Boats
4.	CIVILIAN GROUND TRANSPORTATION
1.	Rail Transportation
2.	Vehicle Fixed Systems
3.	Vehicle Handhelds
3.	TOTAL-FLOOD ELECTRONICS PROTECTION
1.	Data Processing
2.	Telecommunications
3.	Production Control Rooms
4.	Nuclear Power Plant Control Rooms
5.	Medical Facilities
7.	OIL PRODUCTION, REFINING, TRANSFER, AND STORAGE
1.	North Slope Facilities
2.	Pipeline Pumping Stations
3.	Off-Shore Oil Production
4.	Other Fixed Facilities
3.	PROTECTION OF NON-ELECTRONIC FACILITIES
1.	Flammable Liquid Storage
2.	Explosion Suppression
3.	Libraries, Museums, Art Galleries
4.	Records Storage
5.	Bank Vaults and Depositories
6.	Warehouses
7.	Cooking and Food Processing
4.	TOTAL-FLOOD PROTECTION OF SPECIAL FACILITIES
1.	Antarctic Research Facilities
2.	Toll Booths
3.	Research Laboratories
4.	Anechoic Chambers
5.	APPLICATIONS UNIQUE TO THE MILITARY
1.	Tactical Vehicles
2.	Aircraft Fuel Cell Inerting
3.	Missile Vector Control
4.	High Security Fixed Facilities
6.	TRAINING-AND TESTING

older **700** series **aircraft**, possibly the **707,727**, and **747**; however, personnel now at Boeing have **no** knowledge of that use (Reference **4**). In the past, Frito-Lay, Inc., used a Halon **2402** system in their plant at Rosenberg, Texas. However, this system is **no** longer in use. They **are** now primarily using carbon dioxide for their frying vats (Reference **5**). Fenwal Safety Systems have **used** Halon **2402** in some explosion suppression systems in the past.

Some of those systems **are** still in existence; however, it has proven difficult or impossible to get replacement agent (Reference **6**). Halon **2402** has been used for *thrust* vector control in missiles; however, alternatives **are** under investigation (Reference **7**).

NEAR-TERM REPLACEMENTS

Near-term replacements **are** defined as halon-like agents (**halocarbons**) with significant toxicological information in existence with, at the very least, data available **on** acute toxicity, with **fire** or explosion suppression testing that indicates the material could be used **as** a replacement for **one** **or** more halon in some applications, and the compound is or is expected to be commercially available. Here, the availability is not necessarily limited to availability **as** a fire extinguishant. In several cases, the chemical is or is expected to be a CFC replacement in refrigeration and other CFC applications.

Four elements **are** needed for an agent to qualify **as** a halon replacement: an acceptable environmental impact (primarily, a low **ODP**), a low toxicity, cleanliness/volatility, and effectiveness. It is relatively easy to meet any **three** of these requirements, but a replacement that meets **all** four requirements well has yet been identified. A number of candidate "near-term" replacement agents have been examined (Table **2**), but all have trade-offs. None of the currently available halon replacement agents (**or** alternative agents) can be used as drop-in substitutes in existing systems and equipment. Here, we use the term "drop-in" to mean **no** change in equipment design or amount of agent storage required.

TABLE 2. NEAR-TERM REPLACEMENT CANDIDATES		
Candidate	Chemical Formula	Commercialization as Halon Replacement
HBFC-22B1	CHF ₂ Br	Great Lakes (FM 100)
HCFC-22	CHClF ₂	Du Pont (FE 232)
HCFC-123	CF ₃ CHCl ₂	Du Pont (FE 241)
HCFC-124	CF ₃ HCIF	
HFC-23	CHF ₃	Du Pont (FE 13)
HFC-32	CH ₂ F ₂	
HFC-125	CF ₃ CHF ₂	Du Pont (FE 25)
HFC-134a	CF ₃ CH ₂ F	
HFC-227ea	CF ₃ CHF ₂ CF ₃	Great Lakes (FM 200)
FC-3-1-10	C ₄ F ₁₀	3M (PFC 410)
FC-5-1-14	C ₆ F ₁₄	3M (PFC 614)
NAF S-III	Blend (Primarily HCFC)	N. A. Fire Guardian
NAF P	Blend (Primarily CFCs)	N. A. Fire Guardian
Halotron I	Blend (Primarily HCFC)	American Pacific Corp.

In an attempt to compare the different replacements, information on several selection parameters was compiled in Table 3 on the list of near-term replacements.

ALTERNATIVE AGENTS

In many applications in which halons have been traditionally used, other types of fire extinguishing agents could be substituted for halogenated extinguishants. Table 4 lists some of the alternative (non-halocarbon) agents available to date. In some cases, the use of these alternative agent would also require alternative engineering or logistical approaches. Generally, water, dry chemicals, foam, and combination alternatives are suitable for some Halon 1211 applications, with CO₂ extinguishers also suitable in certain instances. Water sprinklers, CO₂ flooding, and inert gases are most suitable as Halon 1301 alternatives.

TABLE 3. ASSESSMENT OF HALON REPLACEMENT CANDIDATES

Parameter	Candidates													
	22B1	22	123	124	23	32	125	134a	227ea	3-1-10	5-1-14	NAF S-III	NAF P	Halotron I
ODP, relative to CFC-11 ^a	-1.4	0.055	0.02	0.022	0	0	0	0	0	0	0	0.044 ^b	0.71	0.02
GWP, relative to CFC-11 ^c		0.36	0.018	0.096	13 ^d		0.58	0.27	0.7 ^e	18.2 ^e	6-12 ^m	0.31 ^b		0.11-0.23 ^m
Lifetime, years ^c	-7 ^f	15.3	1.6	6.6	400 ^d		28.1	15.5		>500 ^e	>500 ^e			
Availability	Now	Now	Now		Now			Now		1993	Now	Now	Now	End 1992
Ext Concentration, % ^b	4.4 ⁱ	11.6	6.3	8.2	12.4	9 ^j	9.4	10.5	5.9 ^f	5.5 ^j	4.4 ^k	8.6 ^{b,j}		~6.5 ^m
WEq ^o	1.1	2.4	1.8	2.6	2.0	1.1	2.6	2.5	2.4	3.07	2.8	1.9		1.8
SVEq ^o	1.3	3.0	2.3	2.9	4.6	1.7	3.2	3.1	2.5	3.03	3.1	2.3		2.30
Inertion Conc., % ^p	8.0	18.8		12	19.8	17.5	14.7	13.5	12	9.5	7.2			
WEq ^q	1.6	2.5		2.6	2.2	1.4	2.8	2.2	3.2	3.5	3.8			
SVEq ^q	1.6	3.2		2.8	4.8	2.2	3.4	2.7	3.4	3.5	3.4			
Toxicity, LC ₅₀ % ^s	10.8 ^l	27-30	3.2 ^u	23-36	66.3	>76	70	50	>80		>30 ^v	30 ^b		~3.2 ^m
Primary Application ^w	S	T	S	T	T	T	T	T	T	T	S	T	S	S

^a Calculated from atmospheric data provided by Dr. A. R. Ravishankara, Aeronomy Laboratory, National Oceanic and Atmospheric Administration, Boulder Colorado.

^b Provided by North American Fire Guardian, April 16, 1992.

^c Average values from Reference 8 unless otherwise noted.

^d Reference 9.

^e Reference 10.

^f Reference 11.

^g Reference 12.

^h Reference 13 unless otherwise noted.

ⁱ Reference 14 gives an extinguishment concentration of 3.9 percent for same fuel.

^j Reference 15.

^k Reference 16.

^l Specified as the "Minimum Design Concentration" in Reference 17. A higher estimated concentration is expected from data supplied to EPA SNAP program by North American Fire Guardian.

^m Estimated.

ⁿ Calculated weight of candidate required for flame extinguishment relative to that required for a given halon. $WEq = (EC_C/EC_R)X(MW_C/MW_R)$; EC = exting. conc., MW = Mole Weight, C = Candidate, R = reference halon. Relative to Halon 1211 for HBFC-22B1, HCFC-123, FC-5-1-14, and Halotron I; relative to Halon 1301 for other agents.

^o Calculated storage volume of candidate required for flame extinguishment relative to that required for a given halon. $SVEq = WEqX(LD_R/LD_C)$; LD = Liquid Density. Relative to Halon 1211 for HBFC-22B1, HCFC-123, FC-5-1-14, and Halotron I; relative to Halon 1301 for other agents.

^p Reference 18. The explosion inertion concentration of Halon is 4.3 percent.

^q Calculated weight of candidate required for explosion inertion relative to that required for Halon 1301. $WEq = (IC_C/IC_R)X(MW_C/MW_R)$.

^r Calculated storage volume of candidate required for explosion inertion relative to that required for Halon 1301. $SVEq = WEqX(LD_R/LD_C)$.

^s Unless otherwise noted, 4-hour rat values from Reference 15.

^t Reference 19.

^u Reference 20.

^v Reference 21.

^w S = streaming, T = Total Flood.

TABLE 4. ALTERNATIVE AGENTS	
WATER	COMBINATION
FOAM	1. Loaded Stream
1. Low-Expansion Foam	2. <i>other</i>
2. High- and Medium-Expansion Foam	INERT GAS
DRY CHEMICAL	1. Carbon Dioxide
1. Monoammonium Phosphate	2. Nitrogen
2. Sodium Bicarbonate	3. Argon
3. Potassium Bicarbonate	4. Helium
4. Proprietary	5. Inert Gases

HALON USE SECTORS AND SUBSTITUTES

Table 5 contains a list of specific halon applications divided into ten major groups. This table omits fire hazards for which halons have received little use (e.g., aircraft hanger fixed systems). Since for most applications, little performance difference exists between the physically acting replacement candidates (HCFCs, HFCs, and FCs), in many cases, only "streaming halocarbon," "flooding halocarbon," or just "halocarbon" has been noted. A streaming halocarbon is one that has suitable physical properties to allow satisfactory streaming in the application noted. Similarly, a flooding halocarbon is one that is more gaseous and will satisfactorily fill a space. Since the degree of streaming or total-flood capability needed depends on the application, no attempt has been made at this time to determine which candidates could be used. However, Table 3 indicates which replacements have properties making them more suitable for streaming or for total flood. In general, the substitutes are listed in order of decreasing preference based on the assessment as it now stands. *It must be recognized that substitutes are proposed rather than recommended.* In many cases, little or no testing has been performed to show the applicability of the suggested substitutes. Combinations of substitutes shown in Table 5 may provide optimal protection for specific applications.

TABLE 5. HALON USE SECTORS AND SUGGESTED SUBSTITUTES		
Application	Substitutes	
A. PORTABLE EXTINGUISHERS FOR FIXED FACILITIES		
1. Residential		Dry Chemical Carbon Dioxide Water Foam Combination
2. Commercial	a. Offices:	Carbon Dioxide Dry Chemical Water
	b. Manufacturing:	Dry Chemical Carbon Dioxide Water Streaming Halocarbon
	c. Retail Sales:	Dry Chemical Carbon Dioxide Water
3. Military and Government	a. Offices:	Carbon Dioxide Dry Chemical Water
	b. Special Facilities:	Dry Chemical Carbon Dioxide Water Foam Streaming Halocarbon
B. AEROSPACE APPLICATIONS		
1. Handheld Extinguishers for On-Board Aircraft	a. Passenger Areas	Loaded Stream Streaming Halocarbon
	a. Cockpit	Streaming Halocarbon Carbon Dioxide
2. Ground-Based Aircraft Fire Protection	a. Flightline	Streaming Halocarbon - Perfluorohexane - 123 or Halotron I Carbon Dioxide
	b. Crash/Rescue	Dry Chemical Foam
3. Aircraft Cargo Bays		Carbon Dioxide Inert Gas Flooding Halocarbon
4. Dry Bay Explosion Suppression		Halocarbon Dry Chemical
5. Aircraft Engine Nacelle		Halocarbon Dry Chemical

TABLE 5. HALON USE SECTORS AND SUGGESTED SUBSTITUTES	
Application	Substitutes
6. Spacecraft Fire Protection	Fixed Systems: Carbon Dioxide Inert Gas FC or HFC Portable: FC or HFC
7. Aircraft Manufacture	Portable Plug-In Halocarbon Carbon Dioxide
C. MARINE APPLICATIONS	
1. Crew Quarters	Fixed System: Water Portable: Dry Chemical Carbon Dioxide Water
2. Machinery Spaces	Fixed System: Carbon Dioxide ^a Foam ^a Water ^a Halocarbon Portable: Dry Chemical Carbon Dioxide Foam Halocarbon
3. Control Rooms	Fixed System: Flooding Halocarbon
4. Automobile Ferry Boats	Fixed System: Flooding Halocarbon
D. CIVILIAN GROUND TRANSPORTATION	
1. Rail Transportation	Dry Chemical Carbon Dioxide
2. Vehicle Fixed Systems	Dry Chemical Carbon Dioxide
3. Vehicle Handhelds	Dry Chemical Carbon Dioxide
E. TOTAL-FLOOD ELECTRONICS PROTECTION	
1. Nuclear Power Plant Control Rooms	Flooding Halocarbon Inert Gases
2. Telecommunications	Water Sprinklers Inert Gases
3. Production Control Rooms	Essential: Flooding Halocarbon Inert Gases Nonessential: Water Sprinklers Flooding Halocarbon Inert Gases

REFERENCES

1. *Clean Air Act Amendments of 1990*, Conference Report to accompany S.1630, 101st Congress, 2nd Session, U.S. House of Representatives, Report 101-1952, 26 October 1990.
2. Arthur Kaufman, New Jersey Fire Shield, personal communication, 31 March 1992.
3. Harold F. Shultz, Caterpillar, Inc., personal communication, 30 March 1992.
4. David Wirth and Alankar Gupta, Boeing Commercial Airplane Company, personal communication, 31 March 1992.
5. Mark Guillard, Frito-Lay, Inc., personal communication, 6 April 1992.
6. Henry Garzia, Fenwal Safety Systems, personal communication, 3 April 1992.
7. McCarson, T. D., and Nimitz, J. S., *An Assessment of Halon 2402 Emissions From Strategic Target System (STARS) Launches*, Sandia National Laboratories, Albuquerque, New Mexico, 1 February 1991. NMERI, OC/91/20.
8. Fisher, D. A., Hales, C. H., Wang, W.-C., Ko, M. K. W., and Sze, N. D., "Relative Effects on Global Warming of Halogenated Methanes and Ethanes of Social and Industrial Interest," *Scientific Assessment of Stratospheric Ozone: 1989*, World Meteorological Organization, Global Ozone Research and Monitoring Project, Report No. 20, p. 395, 1989.
9. Ellington, R.T., Meo, M., "Development of a Greenhouse Gas Emission Index," *Chemical Engineering Progress*, July 1990, pp 58-63.
10. Trowbridge, L.D., "Greenhouse Warming Potential of Candidate Gaseous Diffusion Plant Coolants," U.S. Department of Energy, DE91009986, March 1991.
11. "FM-Report," Great Lakes Chemical Corporation, West Lafayette, Indiana, August 1991.
12. Talukdar, R., Mellouiki, A., Gierczak, T., Burkholder, J. B., McKeen, S. A., and Ravishankara, A. R., *Science*, Vol. 252, pp. 693-695, 1991.
13. Moore, T. A., Moore, J. P., and Floden, J. R., "Technology Transfer for New Laboratory Apparatuses for Fire Suppression Testing of Halon Alternatives," International Conference on CFC and Halon Alternatives, Baltimore, Maryland, 27-29 November 1990.
14. "FM-100, Product Information," Great Lakes Chemical Corporation, West Lafayette, Indiana, November 1990.
15. Skaggs, S. R., Nimitz, J. S., Moore, T. A., and Tapscott, R. E., *Perfluorocarbons as Total-Flood Extinguishing Agents, Task I Report - Candidate Survey*, Air and Energy Engineering Research Laboratory, U. S. Environmental Protection Agency, Research Triangle Park, North Carolina, March 1991.

16. ***Report of the Halons Technical Options Committee***, December **1991**.
17. Information submitted by North American Fire Guardian Technology on NAF **S-III** to NFPA Technical Committee on Alternative Protection Options to Halon, October **1992**. Released with permission of Elio Guglielmi, North American Fire Guardian, 6 April **1992**.
18. Heinonen, E. W., and Skaggs, S. R., "Fire Suppression and Inertion Testing of Halon **1301** Replacement Agents," Halon Alternatives Technical Working Conference **1992**, Albuquerque, NM, **12-14** May **1992**.
19. "FM-**100**, Product Information," Great Lakes Chemical Corporation, West Lafayette, Indiana, February **1991**.
20. "FE-**232 Streaming** Agent for Portable Applications," *DuPont Airernarive Fire Extinguisflants*, Data Sheet, Du Pont de Nemours & Company, Wilmington, DE.
21. Material Safety Data Sheet, 3M, August **1991**.