# A Rigorous Metrological Basis for Seawater pH

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# "The Other CO<sub>2</sub> Problem"



**R** Doney SC, et al. 2009. Annu. Rev. Mar. Sci. 1:169–92

Image: Doney et al., Annual Review of Marine Science, 2009

1. Multiple definitions of "pH" are applied to seawater

2. Only one SW pH CRM is available – no multi-point calibrations

3. Secondary measurement is precise, but requires purification of reagents and suffers from poor traceability to the primary method

#### Current state of SW pH: Multiple SW pH definitions

Multiple SW pH definitions yield a pH range >50X greater than 0.002 pH unit target uncertainty for climate change studies.

pH scale	Definition of "pH"	Ionic Strength / mol kg <sup>-1</sup>	"pH" value
NIST SRMs	$-\log_{10} a(H^{+})$	< 0.1	8.332
Free	-log <sub>10</sub> <i>c</i> (H <sup>+</sup> )	0.4 - 0.9	8.195
Total	$-\log_{10} \Sigma[c(H^+) + c(HSO_4^-)]$	0.4 – 0.9	8.087
Seawater	$-\log_{10} \Sigma[c(H^+) + c(HSO_4^-) + c(HF)]$	0.4 - 0.9	8.078

# NIST research:

Activity coefficients at higher ionic strength

 $pH_{SRM} = -\log a_H \gamma_{Cl}^{\circ} + \log \gamma_{Cl}^{\circ}$ 



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Single point calibration

#### **Certificate of Analysis**

#### pH reference material (tris buffer in synthetic seawater)

Batch 10 (Bottled on August 3, 2011)

Analysis Results The pH

$$pH = -\log_{10}\left(\frac{[H^+]}{mol \ kg-soln^{-1}}\right)$$

was determined on the total hydrogen ion scale using a hydrogen / silver – silver chloride cell (DelValls & Dickson, 1998). Two separate samples were measured in triplicate, on two different dates

The pH estimated was  $8.0924 \pm 0.0004$  (6; 2)

The cited uncertainty represents the standard deviation. Figures in parentheses are the number of analyses made (total number of analyses; number of separate bottles analyzed).

#### **Estimated Overall Uncertainty**

Although the measurements are capable of high precision, the overall uncertainty will be significantly greater. We are working to understand this value better, but presently expect it to be of the order of 0.004 in pH.



### NIST research: Development of buffers for multi-point calibrations

Multi-point pH calibration
GUM compliant U



Measurement of pH<sub>T</sub> values of Tris buffers in artificial seawater at varying mole ratios of Tris:Tris·HCI



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#### Measurement of SW pH(S): Spectrometric detection via *meta*-cresol purple



# mCP impurities and SW pH(S)



- Identify impurities to improve dye synthesis and purification process
- Molecular characterization of high purity dye: qNMR

Images: Liu, Patsavas, and Byrne. Environ. Sci. and Technol. 2011, 45 (11) pp 4862-4868.

### NIST research: qNMR – Bruker Avance II Spectrometer





DISCLAIMER: Certain commercial equipment, instruments, or materials are identified in this presentation to specify adequately the experimental procedure. Such identification does not imply recommendation or endorsement by the National Institute of Standards and Technology, nor does it imply that the materials or equipment identified are necessarily the best available for the purpose.

### NIST research: qNMR preliminary results for *m*CP



## Where do we go from here?

# Problems with candidate RMs:





### **Traceability of SW pH(S) :** From the SI via NIST to field measurements



### Traceability of SW pH(S) : Ocean acidification and public policy

#### **OCEAN ACIDIFICATION**

Community Forum: Monday, April 15, 2013

Join us to learn about ocean acidification and the strategies and actions to protect marine resources recommended by the state's Blue Ribbon Panel on Ocean Acidification

#### Local/State





Federal

To provide for ocean acidification research and monitoring, and for other purposes.



Copenhagen Policy Brief No. 1

International

Ocean Acidification: A Hidden Risk for Sustainable Development