



Microscopic Dynamics of Liquid and Solid Hydrogen

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Why Hydrogen?

- Forms a quantum fluid and solid in nature
- Two nuclear spin isomers
- Large quantized rotational levels
- Can be modeled from first principles



A presentation in three parts

- Diffusion of liquid normal hydrogen
- Molecular mean square displacement of solid para hydrogen
- Applications of research studies in an introductory physics classroom.

Motivation for the study of liquid H₂ diffusion

- Large zero-point energy contribution
- Theoretical calculations are not fully in agreement with experiment.
- Neutron scattering as a new experimental technique



Neutron Scattering and Diffusive Dynamics of Low temperature Hydrogen

 $\Gamma + \frac{0.6}{5}$

0.4

0.2

2

-2

4

• Why neutron scattering?

Structure
$$S_{inc}(Q, \omega) = \frac{1}{\pi \hbar} \frac{DQ^2}{\omega^2 + (DQ)^2}$$

$$\Gamma_{fwhm} = 2\hbar DQ^2$$

Our instrument: The disk chopper spectrometer (DCS)

Diffusive time scales: $\sim 10^{-12}$ s

 $\lambda_{incident} = 6 \text{ Å}$

Resolution ~ $30\mu eV$







- We collected data over a 5 temperatures between 15.4 K and 20.4 K.
- We also performed measurements of the empty can background, darkcounts, and vanadium resolution.





Modeling Temperature Dependence and Conclusion







- Experimental results show no qualitative temperature dependence
- Theory predicts T-dependence approaching melting temp.
- What is the contribution of zeropoint vs. thermal energy to the vibration of solid H₂?

Mean Squared Displacement of solid p-H₂

- Large zero-point contributions limit bragg peak diffraction techniques normally used to measure <µ²>
- The rotational line has a contribution from <µ²> that we can utilize.

$$I(Q) \sim j_1^2(Qa) \cdot e^{-(\frac{<\mu^2>}{3})Q^2}$$









Conclusions

- Thermal contribution evident approaching melting point
- Disagreement with previous experiment
- Qualitative agreement with theoretical calculations.



What did I learn and experience?

- My first professional research experience
- Diversity of science
- "In what units?"
 - Engineer speak
 - Atmosphere of sharing



Integrating research knowledge into physics labs

Pose an inquiry question with possible hypothetical hypotheses.

Students formulate their own experimental methodology

- Choose independent variable and controls
- Choose materials and measuring devices

Experimental uncertainties

Presentation of results and inferences.





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<u>Questions?</u>

