

Thermodynamics and Correlation Effects on Diffusion in Al-Ni Melts

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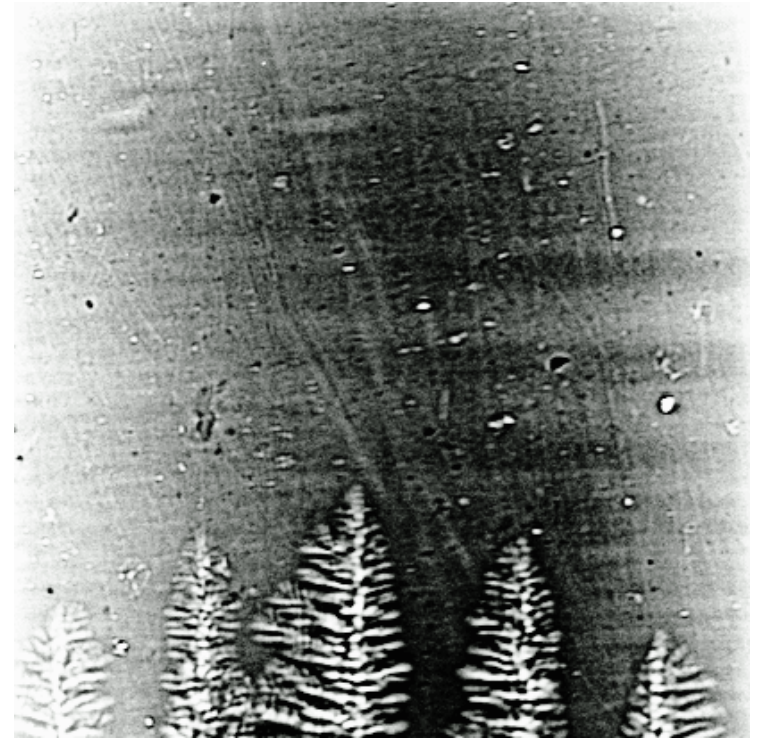
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Outline

- Motivation
- Darken's equation for liquids
- Experimental methods
 - Quasielastic Neutron Scattering (QNS)
 - Capillary techniques + X-ray Radiography (XRR)
 - Molecular Dynamic Simulation (MDS)
- Results in Al-Ni
- Conclusions

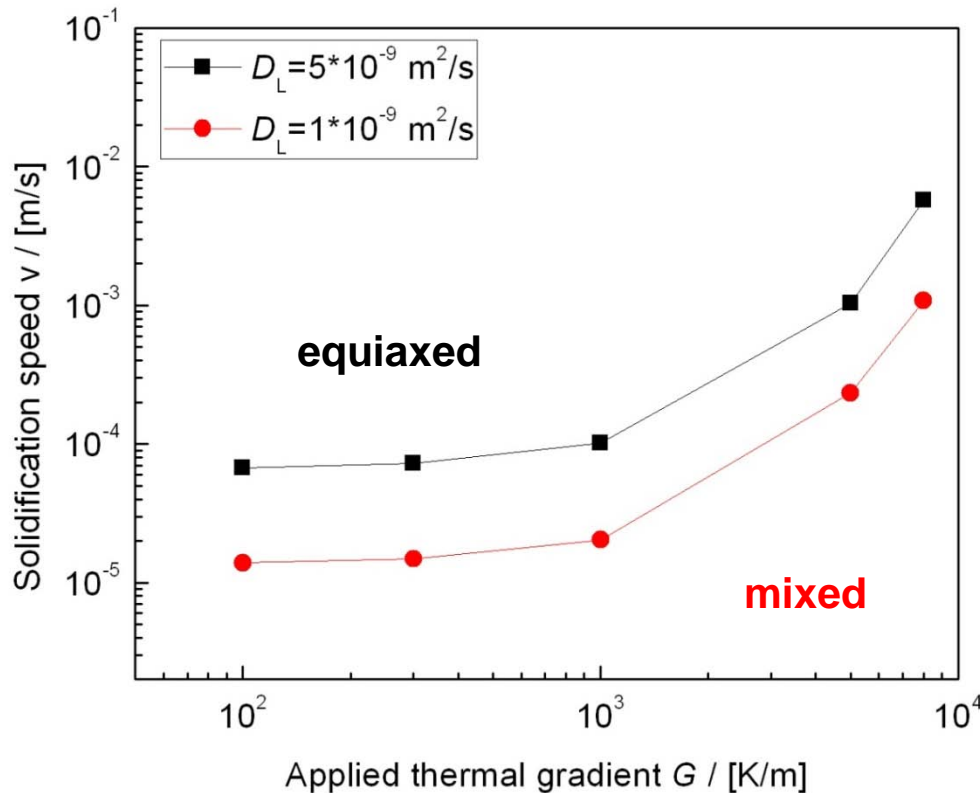
Motivation

- Materials Design from the Melt
- ⇒ Interplay of inter-diffusion and crystal growth
- ⇒ thermophysical properties
- ⇒ Influence of thermodynamics on diffusion



X-ray radiography: Al-Cu

Columnar to Equiaxed Transition (CET) in Al-Ti



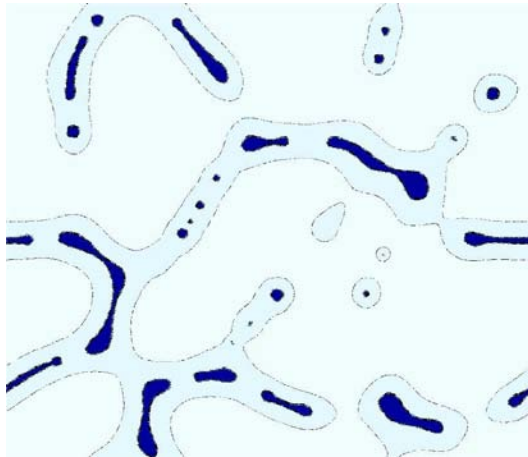
analytical Hunt model parameters:

- $N_0 = 10^9 \text{ m}^{-3}$
- $\Delta T_N = 3.8 \text{ K}$
- $c_0 = 47 \text{ at}\%$

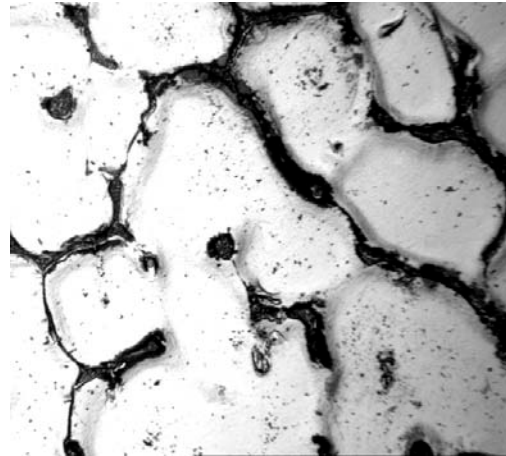
→ factor 8 difference

CET prediction depends sensitively on $D_L(c,T)$

Pseudo Front Tracking Modelling of Solidification in Al-Cu4

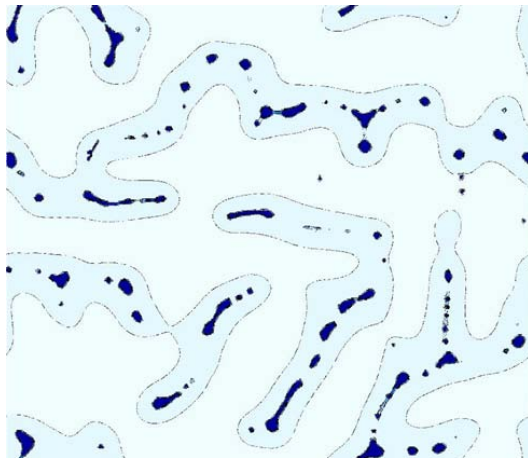


$$D_{\text{AlCu}} = 4.8 \times 10^{-9} \text{ m}^2\text{s}^{-1}$$



500 x 430 μm^2

micrograph section



$$D_{\text{AlCu}} = 1.0 \times 10^{-9} \text{ m}^2\text{s}^{-1}$$

- microstructure and phase distribution:
best agreement for exp. D_{AlCu}
- model predictions require as input
accurate liquid diffusion data

Darken's equation in Liquids?

- **solids:**

$$D_{AB} = (D_B N_A + D_A N_B) \Phi S$$

S as additional factor (“vacancy wind”)

- **liquids:**

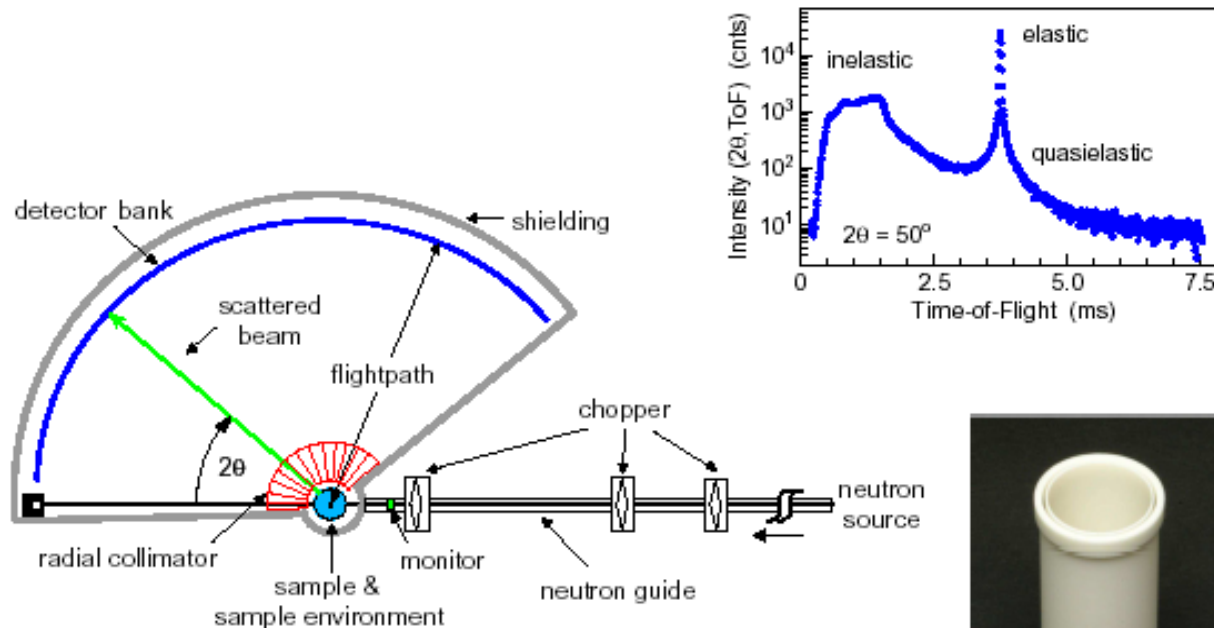
$$D_{AB} = (D_B N_A + D_A N_B + D_{cross}) \Phi = (D_B N_A + D_A N_B) \Phi S$$

$$S = 1 + \frac{D_{cross}}{N_A D_B + N_B D_A}$$

Darken, Trans. AIME (1948)
Manning, Phys. Rev. B (1961)
Kehr et al., Phys. Rev. B (1989)

Quasielastic Neutron Scattering (QNS)

- $\Delta D^*/D < 5\%$, incoherent scatterer only (Cu, Ni, Ti,...), convection free



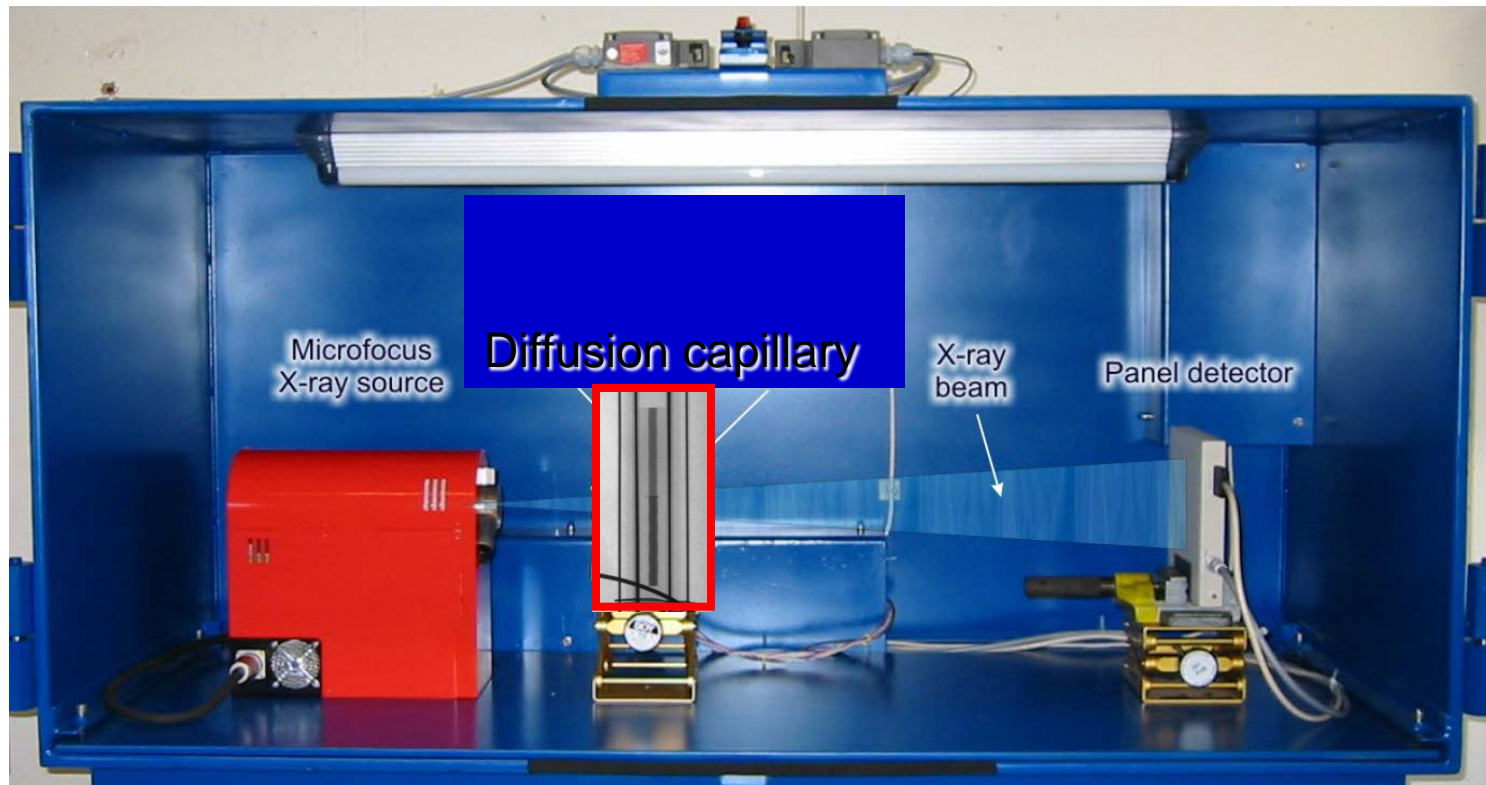
$$I(2\theta, \text{ToF}) \rightarrow \Phi(q, t) = \frac{\langle \rho_q(t)^* \rho_q(0) \rangle}{\langle |\rho_q(0)|^2 \rangle}$$



Meyer, Phys. Rev. B. (2010)

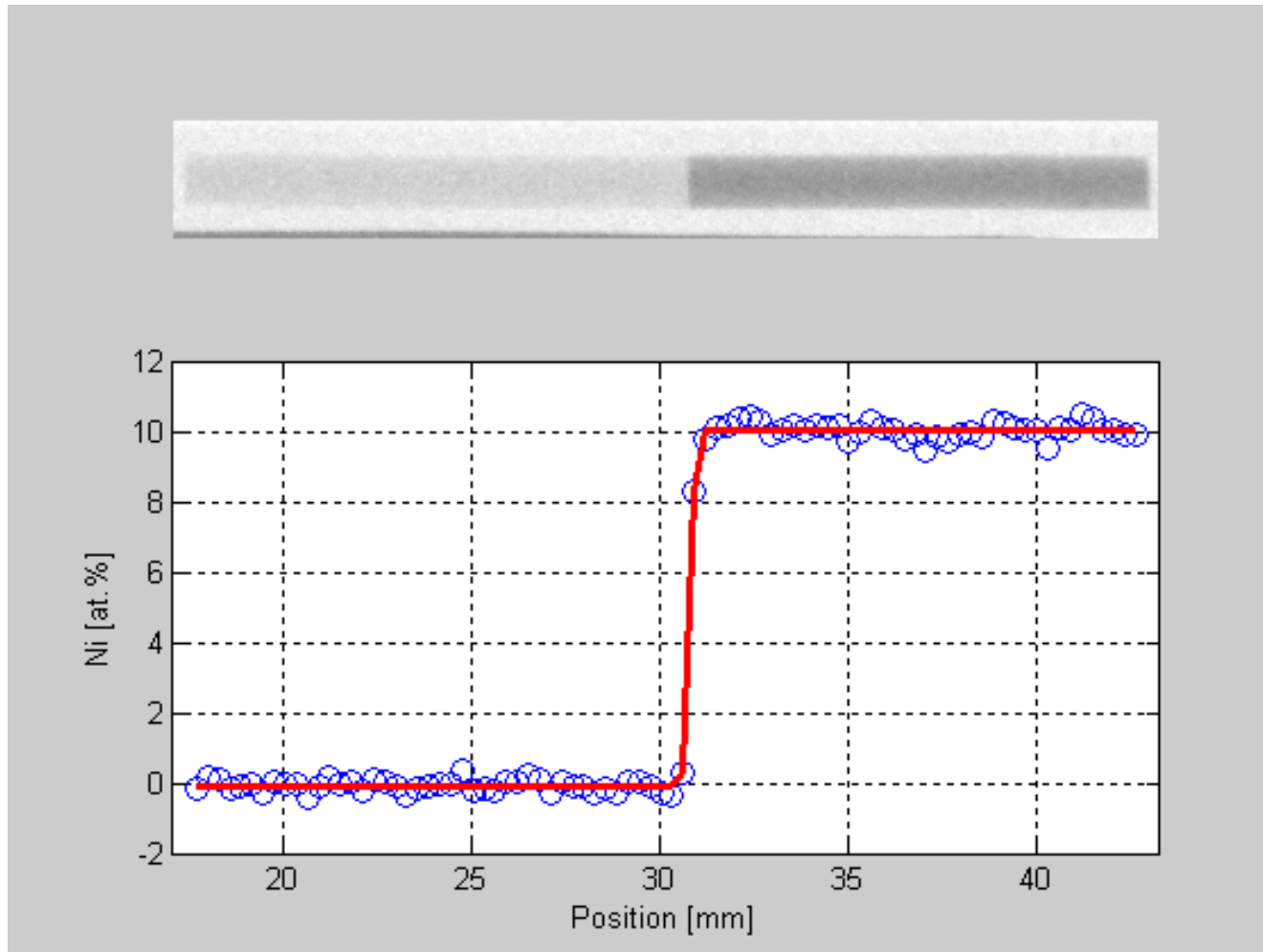
Capillary Techniques with X-ray Radiography

- $\Delta D_{AB}/D < 20\%$, contrast required



Griesche et al., Mat. Sci. For. (2010)

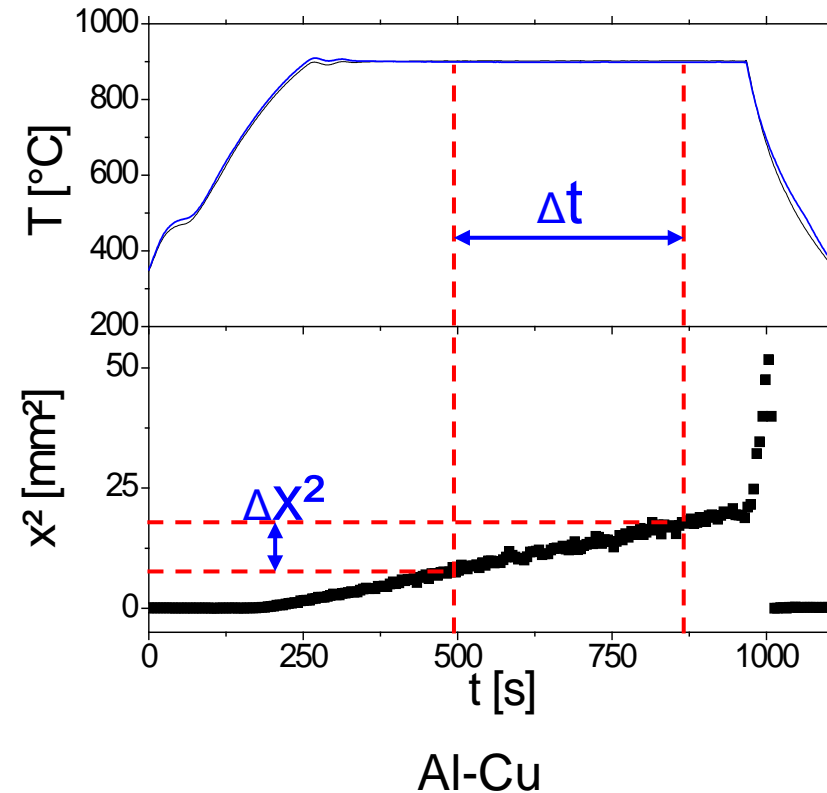
Al vs. AlNi₁₀ at.-%



Griesche et al., Rev. Sci. Instr. (2010)

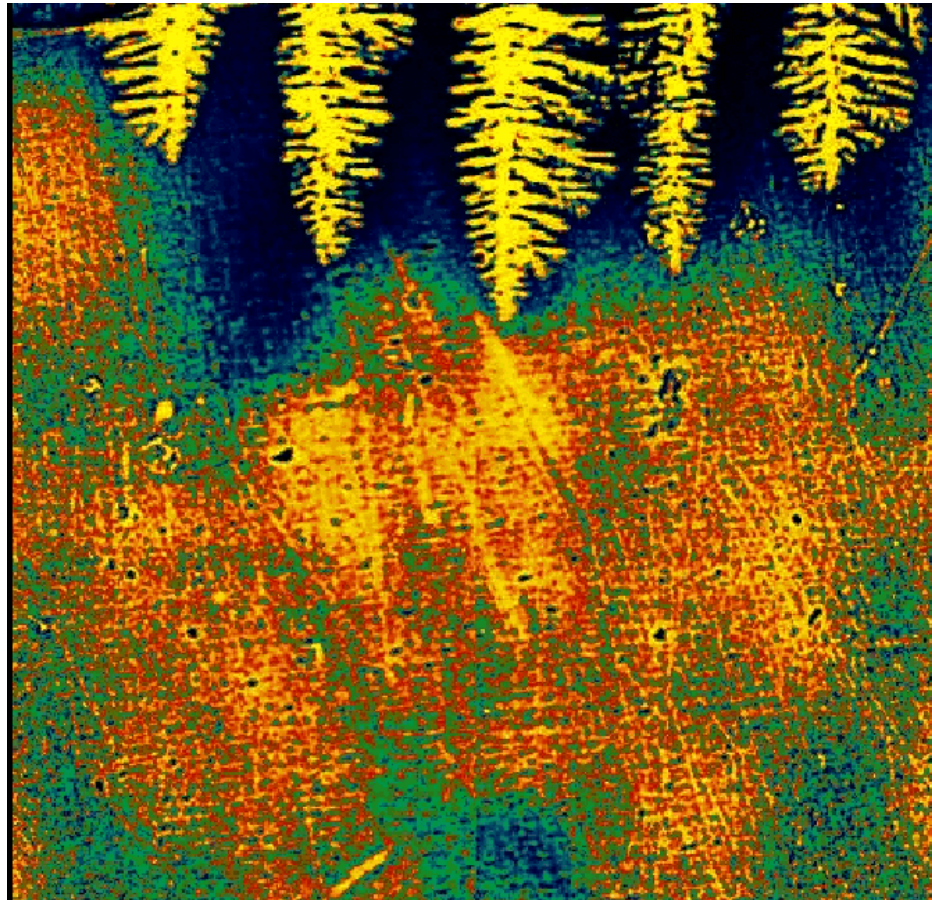
Image Analysis

- $\Delta x^2 = 2D\Delta t$
- good statistics
- deviations from pure diffusion detectable (convection)
- no black box (better accuracy)
- $D(T)$ measureable



Zhang et al., Phys. Rev. Lett. (2010)

... more movies: AlCu_{30} at.-%



Henri Nguyen-Thi, pers. comm.

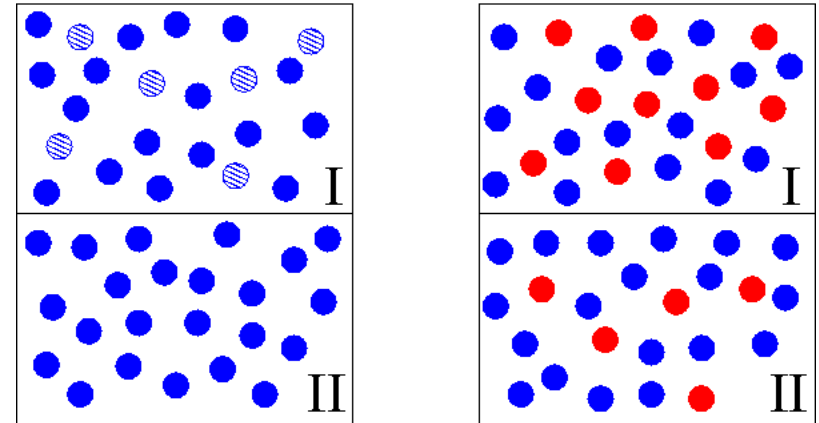
Molecular Dynamic Simulation

- D_A, D_B, D_{AB}
- D_{cross} from velocity correlation functions

$$S = 1 + \frac{D_{\text{cross}}}{N_A D_B + N_B D_A}$$

$$\Phi = \frac{Nk_B T x_A x_B}{S_{\text{cc}}(q=0)}$$

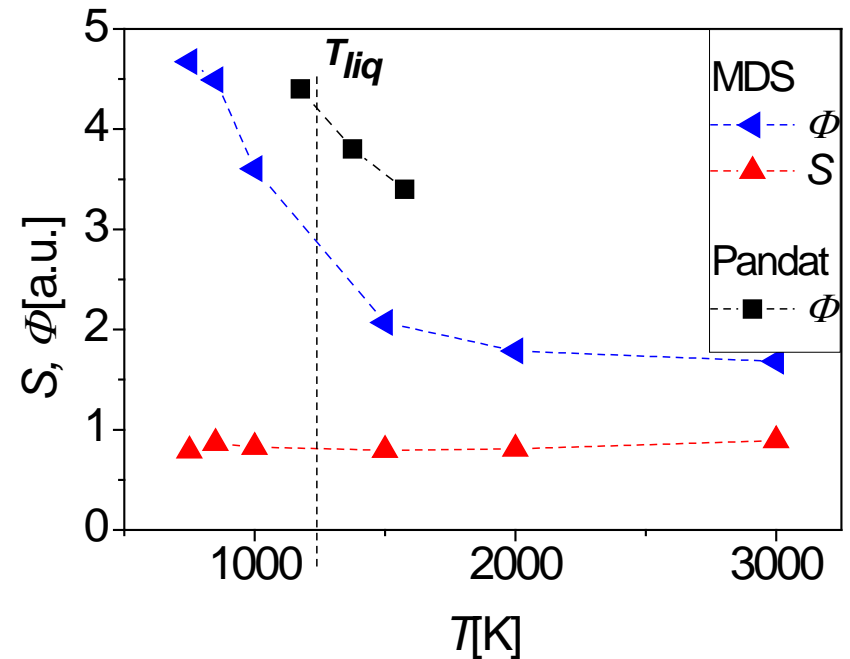
$$L = S (D_B N_A + D_A N_B)$$



$$D = \lim_{t \rightarrow \infty} C \frac{\langle (\vec{r}(t) - \vec{r}(0))^2 \rangle}{6t}$$

Results $\text{Al}_{80}\text{Ni}_{20}$

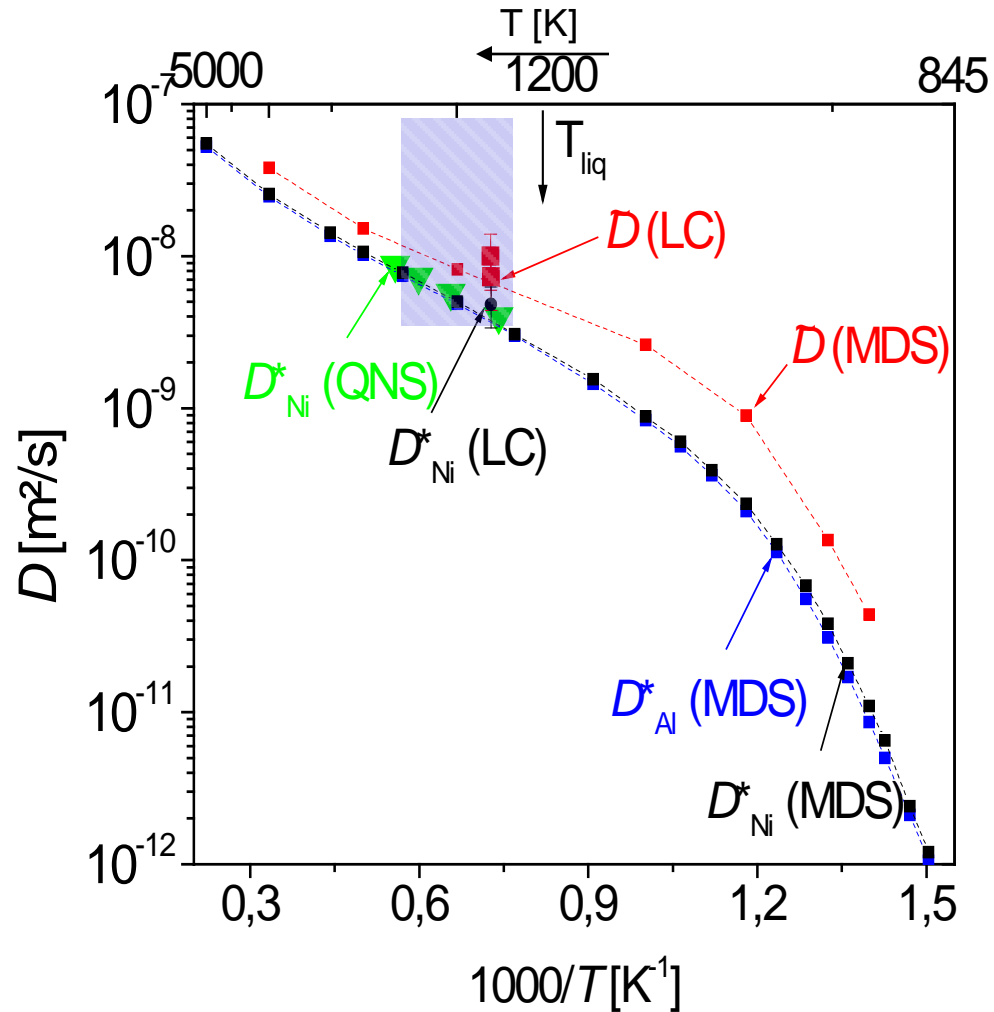
- Φ (Pandata) $\sim \Phi$ (MDS)
- strong temperature dependence of Φ in undercooled melt
- $S \approx 0,8$



R. Schmid-Fetzer, Pandat calculations (2004)
Horbach et al., Phys. Rev. B (2007)

Results $\text{Al}_{80}\text{Ni}_{20}$

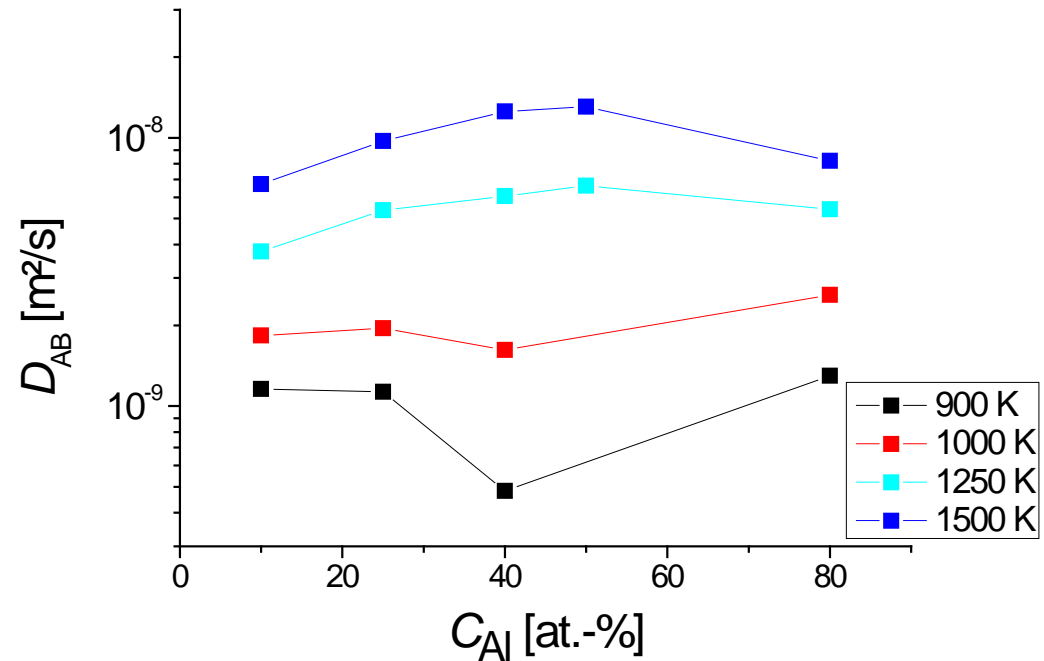
- sim. & exp. show excellent agreement
- $D_{\text{Ni}}^* \approx D_{\text{Al}}^*$
- $D_{\text{AlNi}} > D_{\text{Ni}}^*, D_{\text{Al}}^*$
- slowing down of diffusivity in undercooled region (arresting of atoms)
- change in diffusion mechanism



Horbach et al., Phys. Rev. B (2007)

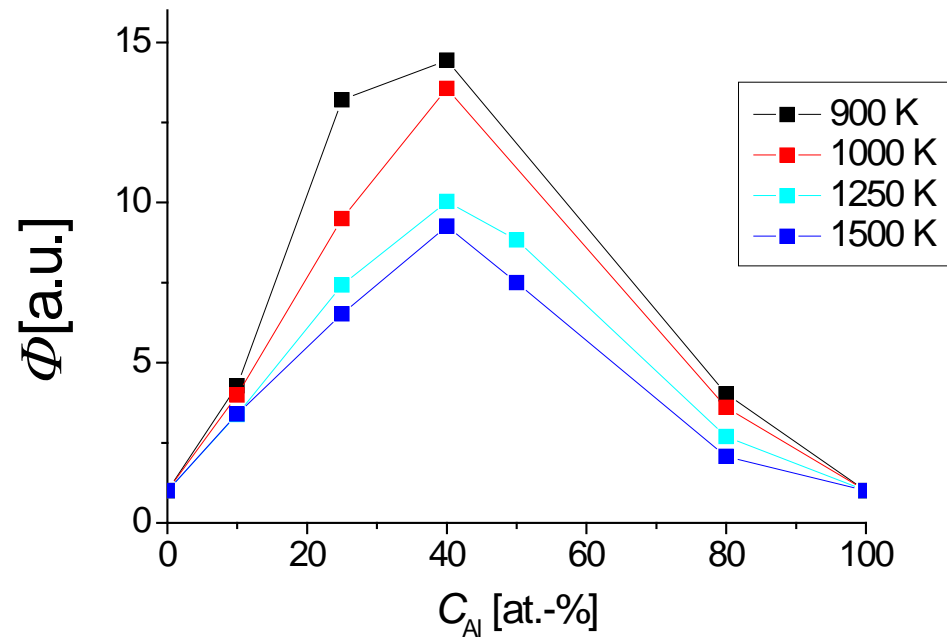
Al-Ni (MDS): D_{AB}

- D_{AB} increases with T
- Maximum of D_{AB} in equilibrium melt
- Minimum of D_{AB} in undercooled melt



Al-Ni (MDS): Φ

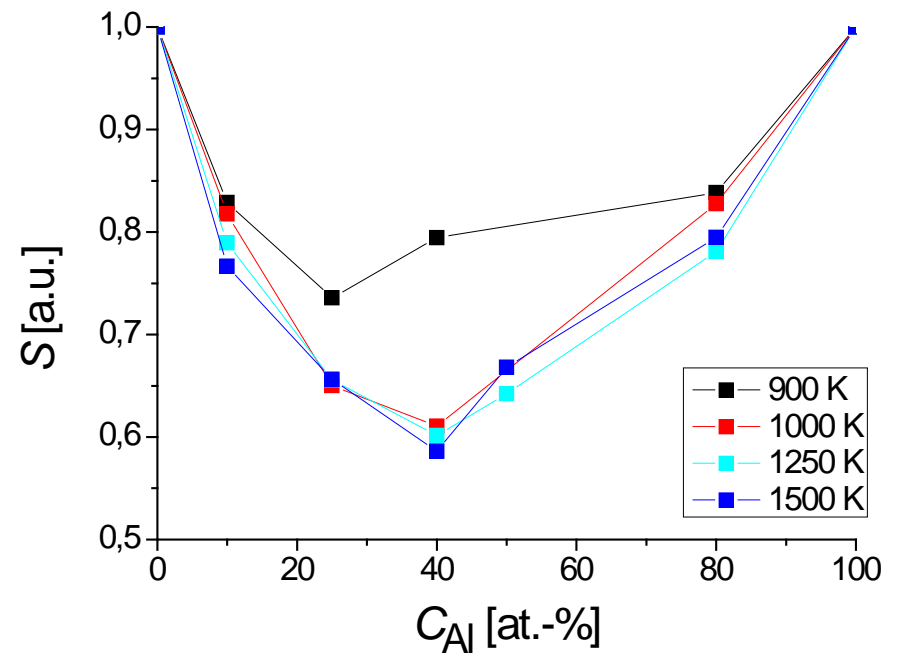
- Φ decreases with T
- always a maximum for concentrated solution
- Φ overcompensated by „cage effect “



Al-Ni (MDS): S

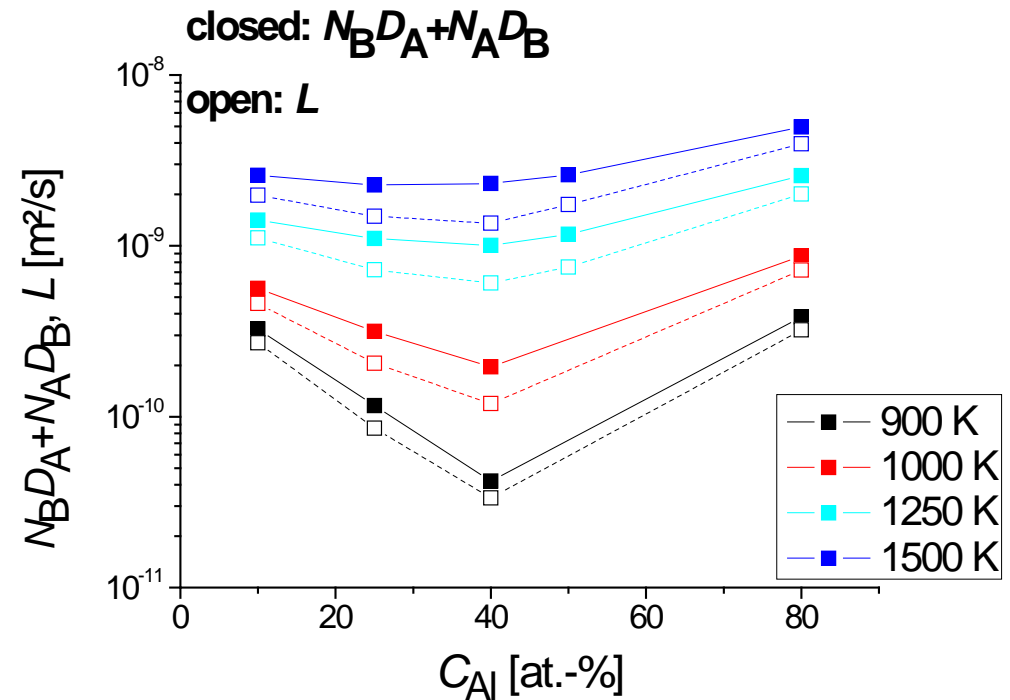
- increasing S with undercooling

$$S = 1 + \frac{D_{cross}}{N_A D_B + N_B D_A}$$



Al-Ni (MDS): $(N_B D_A + N_A D_B), L$

- $L = S (N_B D_A + N_A D_B)$
- slowing down due to cage effect



Conclusions

- coupling of experimental techniques (QNS, diffusion capillary techniques + *in situ* diagnostics) with simulation methods (MDS) helps revealing mechanisms
- slowing down of dynamics in undercooled melts due to collective effects (arresting of atoms)
- cross-correlation effects decrease with increasing undercooling

Thank you for your attention !