

Ionization Potential and Appearance Potential Measurements, 1971-1981

Rhoda D. Levin and Sharon G. Lias

Ion Kinetics and Energetics Data Center
National Measurement Laboratory
National Bureau of Standards
Washington, DC 20234



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Foreword

The National Standard Reference Data System provides access to the quantitative data of physical science, critically evaluated and compiled for convenience and readily accessible through a variety of distribution channels. The System was established in 1963 by action of the President's Office of Science and Technology and the Federal Council for Science and Technology, and responsibility to administer it was assigned to the National Bureau of Standards.

NSRDS receives advice and planning assistance from a Review Committee of the National Research Council of the National Academy of Sciences-National Academy of Engineering. A number of Advisory Panels, each concerned with a single technical area, meet regularly to examine major portions of the program, assign relative priorities, and identify specific key problems in need of further attention. For selected specific topics, the Advisory Panels sponsor subpanels which make detailed studies of users' needs, the present state of knowledge, and existing data resources as a basis for recommending one or more data compilation activities. This assembly of advisory services contributes greatly to the guidance of NSRDS activities.

The System now includes a complex of data centers and other activities in academic institutions and other laboratories. Components of the NSRDS produce compilations of critically evaluated data, reviews of the state of quantitative knowledge in specialized areas, and computations of useful functions derived from standard reference data. The centers and projects also establish criteria for evaluation and compilation of data and recommend improvements in experimental techniques. They are normally associated with research in the relevant field.

The technical scope of NSRDS is indicated by the categories of projects active or being planned: nuclear properties, atomic and molecular properties, solid state properties, thermodynamic and transport properties, chemical kinetics, and colloid and surface properties.

Reliable data on the properties of matter and materials are a major foundation of scientific and technical progress. Such important activities as basic scientific research, industrial quality control, development of new materials for building and other technologies, measuring and correcting environmental pollution depend on quality reference data. In NSRDS, the Bureau's responsibility to support American science, industry, and commerce is vitally fulfilled.



ERNEST AMBLER, *Director*

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Rhoda D. Levin and Sharon G. Lias

Ion Kinetics and Energetics Data Center, National Measurement Laboratory, National Bureau of Standards, Washington, DC 20234

A compilation is presented of the ionization potential and appearance potential measurements which appeared in the refereed literature in the time period 1971-1981. The data are sorted according to the identity of the ionic species formed in the ionization process. Precursor molecules or radicals are identified by a structural formula and, in the case of compounds containing rings, by name according to the Chemical Abstracts system of nomenclature. Chemical Abstracts Registry Numbers are provided where available. A complete bibliography and author index are provided.

Key words: appearance potential; charge transfer spectrum; electron impact ionization; ionization potential; photoelectron spectroscopy; photoionization; spectroscopy.

Introduction

In 1969, the NBS Ion Energetics Data Center, under the direction of Dr. Henry M. Rosenstock, published a compilation of ionization potential and appearance potential measurements covering the literature through mid-1967 [1]. This was followed in 1977 by an update covering the literature through 1971 [2]. Both these volumes contained, wherever possible, critical evaluations of the data in which the ionization threshold measurements were paired with thermochemical data on corresponding neutral species to generate values for the heats of formation of ions in the gas phase. The current publication, which consists of a listing of ionization and appearance potential measurements which appeared in the literature between 1971 and 1981 (plus a few older measurements not included in the earlier volumes), is the first step toward a new update.

The 1977 compilation [2] is 10 years out of date at this writing, and the collection given here contains data from approximately 2000 papers which have appeared in the intervening years. The early publication of this encyclopedic list of measurements without an accompanying evaluation serves several purposes. Especially for that body of users whose interest lies in the ionization potentials themselves, or for those who require a bibliographic guide to mass spectrometric and photoelectron spectroscopic measurements, the present volume as it is will serve the need. For those users whose primary interest is in evaluated heats of formation of ions, this volume will best be used as an adjunct to the 1977 compilation, to call attention to newer measurements, until the appearance of an updated critical evaluation.

Data listed in the compilation "Ion Energetics Measurements, 1971-1973" by H. M. Rosenstock, D. Sims, S. S. Shroyer, and W. J. Webb [3] have been included as an integral part of the present book.

This compilation is restricted to processes involving positive ion formation. Data concerned with the energetics of negative ions are being compiled by Dr. John Bartmess of Indiana University, to be published separately in the Journal of Physical and Chemical Reference Data. That publication will list the heats of formation and, where available, entropies of negative ions, along with the electron affinities of corresponding neutral species and, where available, the acidities of the corresponding conjugate acids.

Literature Coverage

The literature for the period 1971-1981 was covered initially by an issue-by-issue search of the following journals: Canadian Journal of Chemistry, Canadian Journal of Physics, Chemical Communications, Chemical Physics, Chemical Physics Letters, Chemische Berichte, Faraday Transactions II, Helvetica Chimica Acta, High Temperature, International Journal of Mass Spectrometry and Ion Physics, Journal of the American Chemical Society, Journal of Chemical Physics, Journal of Electron Spectroscopy and Related Phenomena, Journal of Inorganic and Nuclear Chemistry, Journal of the Optical Society of America, Journal of Organometallic Chemistry, Journal of Physical Chemistry, Organic Mass Spectrometry, Tetrahedron, and Tetrahedron Letters. This search was supplemented by a systematic use of standard abstracting services such as Chemical Abstracts and the Mass Spectrometry Bulletin (of the Mass Spectrometry Data Centre, The University of Nottingham, U.K.). Papers listed

¹ Figures in brackets indicate literature references.

in review articles describing relevant measurements were also checked against the bibliography as a monitor of the completeness of coverage. With the exception of certain journals published in the Soviet Union to which we did not have ready access, the literature coverage is estimated to be better than 95% complete. Only data appearing in refereed journals are included. The cut-off date is approximately March 1981.

Description of the Compilation

The table of ionization and appearance potential measurements follows a format similar to that used in the earlier volumes [1,2,3]. That is, one will find data for a particular system listed under the empirical formula for the ion that is generated in the ionization process of interest. For example, in order to find the ionization potential of acetone, one finds the empirical formula of the acetone ion, $C_3H_6O^+$ (boldface), then identifies those measurements involving acetone precursor molecules by looking in the first column of the table. For the appearance potential of a fragmentation process of the acetone ion (e.g., $CH_3COCH_3 \rightarrow CH_3CO^+ + CH_3 + e^-$), one would locate the empirical formula of the product ion, $C_2H_3O^+$, and identify those measurements involving acetone precursors in the first column. The neutral precursor species are identified by a semi-structural formula, and for compounds containing rings, the compound name according to the system of nomenclature used by the Chemical Abstracts Services. The Chemical Abstracts Registry Number is given for all compounds, when available. For a very few papers which appeared late in 1980, it was necessary if the data were to be included, to use the nomenclature used by the original authors and to omit the Registry Numbers. In some cases, comments about the experimental observation are also given.

The column of the table, labelled "Other Products," contains an indication of the identity of neutral or negative ion fragment species when these are known (e.g., CH_3 in the fragmentation process: $CH_3COCH_3 \rightarrow CH_3CO^+ + CH_3 + e^-$). When the process described is just the removal of an electron, this column contains two asterisks. A word of caution is in order here—some techniques (particularly photoelectron spectroscopy) measure the energy required to remove an electron from a molecule, but do not identify the resulting ionic species. Certain molecular ions (e.g., neo- $C_5H_{12}^+$, CCl_4^+) are formed on a dissociative potential surface, and cannot be said to exist in the gas phase. Therefore, it must be stated that the listing of the empirical formula of a particular ion does not necessarily imply that the ion exists in the gas phase.

The fourth column of the table gives the measured energy required to form the listed ion from the neutral molecule or radical in the second column. All values are given in electron volts. When the original data have been reported in units other than electron volts, conversion to electron volts has been made using the following conversion factors: $1 \text{ eV} = 8065.479 \text{ cm}^{-1} = 96.48456 \text{ kJ mol}^{-1} = 23.06036 \text{ kcal mol}^{-1}$. Error limits, when cited, are those given by the original authors.

Ionization potentials given are adiabatic values unless the designation (V) appears after the value, in which case the vertical ionization potential has been given. Photoelectron spectroscopy papers often report only vertical ionization potentials. Although in many cases, these probably coincide with the adiabatic values, we have followed the policy of labelling such ionization potentials "vertical" unless the original authors specifically report the measurement of an adiabatic ionization potential. Because of the original emphasis of this compilation effort on deriving heats of formation of ions, it was initially assumed that the users of this volume would find only adiabatic values useful, and therefore, vertical values were included only when adiabatic values were not given in a particular paper. This same emphasis, as well as space considerations, has deemed that only the lowest ionization potential be included here, except for monatomic, diatomic, and triatomic species, for which transitions leading to higher electronic states are also included. Users whose interests are in vertical ionization potentials or excited states of polyatomic ions will find this volume useful as a bibliographic guide to the literature of photoelectron spectroscopy.

Where available, the ionization potentials corresponding to the formation of doubly-charged ions have been included. Data for ionization processes leading to ions having three or more positive charges are not included.

The fifth column of the table gives an indication of the experimental technique used in the measurement. The abbreviations are as follows:

Abbreviation	Technique
S	Spectroscopy
PI	Photoionization
PE	Photoelectron Spectroscopy
AUG	Auger Electron Spectroscopy
PEN	Penning Ionization
EI	Electron Impact
CTS	Charge Transfer Spectrum
OTH	Other

For detailed descriptions of these various techniques, the reader is referred to the chapter appearing at the beginning of the 1977 compilation [2].

The final column of the table lists the number of the reference in the bibliography at the end of the table. An author index is also provided.

The index lists the empirical formulas of the ionic species, ordered according to an alphabetical sorting scheme. The empirical formulas are written with the atoms given in increasing order of atomic number, with the exception of hydrogen which appears after carbon in carbon-containing ions. The alphabetization is carried out on these formulas as written. For example, the ions CHF_3^+ , $CHCl_3^+$, $CFCl_3^+$, and CCl_3I^+ would be alphabetized first according to the atom which appears immediately after the C in the empirical formula, then according to the following atom: CCl_3I^+ , $CFCl_3^+$, $CHCl_3^+$, CHF_3^+ .

As in the earlier volumes [1,2,3], the actual ordering of the ionic species in the compilation is determined by the atom in the molecule which has the highest atomic number, with the overall ordering following the periodic chart in increasing order. To find an ion whose highest atomic number atom is X, find that portion of the compilation devoted to species having X as the highest atomic number atom. In this portion, the sort will first list species containing only X (X^+ , X_2^+ , X_3^+ , etc.), then ions compounded of X and one other element, these other atoms appearing in increasing order of atomic number. Within the set of ions $A_nX_m^+$, all ions with m equal to 1 will appear first while n advances from $n=1$ to the maximum value; then m will be advanced to 2, and so on. When all $A_nX_m^+$ ions (where A has an atomic number lower than that of X) have been listed, $A_nB_pX_m^+$ ions appear (ordering of atomic numbers: $A < B < X$). The indexes are advanced in the order n, p, m . The sort then proceeds to species containing four different atoms, etc.

Acknowledgments

Mrs. Kathy Maugh, Mr. José Portal, and Dr. Pierre Ausloos have all contributed greatly to this work by participating in the abstracting of data from the literature. The authors would also like to acknowledge Dr. Henry M. Rosenstock and the former staff of the NBS Ion Energetics Data Center for the development of the computer processing

procedures used in the production of this book, and the technical assistance of Mr. Robert Thompson, Mrs. Carla Messina, and Mr. George Dines in using those techniques. We would particularly like to thank Ms. Carol Martin for her careful proofreading of the final tables, and Mr. David Stier for writing programs which simplified the final editing process and improved the format of the book. This project was supported by the Office of Standard Reference Data of the National Bureau of Standards and the U.S. Department of Energy Pollutant Characterization and Safety Research Division. That portion of the compilation originally published in reference [3] was supported in part by the National Institute of General Medical Sciences, National Institutes of Health (NIGMS). The advice and encouragement of Dr. L. H. Gevantman is gratefully acknowledged.

References

- [1] Franklin, J. L., Dillard, J. G., Rosenstock, H. M., Herron, J. T., Draxl, K., and Field, F. H., "Ionization Potentials, Appearance Potentials, and Heats of Formation of Gaseous Positive Ions," Nat. Stand. Ref. Data Ser., Nat. Bur. Stand. (U.S.), 26 (1969).
- [2] Rosenstock, H. M., Draxl, K., Steiner, B. W., and Herron, J. T., "Energetics of Gaseous Ions," J. Phys. Chem. Ref. Data 6, Suppl. 1 (1977).
- [3] Rosenstock, H. M., Sims, D., Shroyer, S. S., and Webb, W. J., "Ion Energetics Measurements. Part I. 1971-1973," Nat. Stand. Ref. Data Ser., Nat. Bur. Stand. (U.S.), 66 (1980).
- [4] Bartmess, J. E., to be published.

Index of Ions

Ac^+	552	$\text{BC}_3\text{H}_9\text{O}^+$	224
Ag^+	492	$\text{BC}_3\text{H}_9\text{O}_2^+$	224
AgEu^+	527	$\text{BC}_3\text{H}_9\text{O}_3^+$	224
AgHo^+	529	$\text{BC}_3\text{H}_9\text{S}^+$	342
AgI^+	518	$\text{BC}_3\text{H}_9\text{S}_2^+$	342
AgI_3^+	518	$\text{BC}_3\text{H}_9\text{S}_3^+$	342
Ag_2^+	493	$\text{BC}_3\text{H}_{12}\text{N}^+$	167
Ag_2I^+	518	$\text{BC}_4\text{H}_{11}\text{N}_2^+$	168
Ag_3^+	493	$\text{BC}_4\text{H}_{12}\text{N}^+$	167
Ag_3I_2^+	518	$\text{BC}_4\text{H}_{12}\text{N}_2\text{Br}^+$	470
Ag_3I_3^+	518	$\text{BC}_4\text{H}_{12}\text{N}_2\text{Cl}^+$	381
Al^+	290	$\text{BC}_4\text{H}_{12}\text{N}_2\text{F}^+$	283
AlAg^+	493	$\text{BC}_4\text{H}_{12}\text{N}_2\text{I}^+$	512
AlAu^+	544	$\text{BC}_4\text{H}_{13}\text{N}_2^+$	168
AlAu_2^+	545	$\text{BC}_5\text{H}_7\text{NBr}^+$	469
AlBr^+	475	$\text{BC}_5\text{H}_7\text{NCl}^+$	381
AlBr_3^+	475	$\text{BC}_5\text{H}_7\text{N}_2\text{O}^+$	268
AlCl^+	393	$\text{BC}_5\text{H}_8\text{N}^+$	168
AlCl_3^+	393	$\text{BC}_5\text{H}_{15}\text{N}_2^+$	168
AlCl_4Cs^+	521	$\text{BC}_6\text{H}_5\text{Cl}_2^+$	377
AlCl_4K^+	405	$\text{BC}_6\text{H}_5\text{F}_2^+$	278
AlCl_4Rb^+	482	$\text{BC}_6\text{H}_7\text{NF}_3^+$	283
AlH^+	514	$\text{BC}_6\text{H}_{10}\text{N}^+$	168
AlI^+	514	$\text{BC}_6\text{H}_{10}\text{NO}^+$	267
AlSi^+	309	$\text{BC}_6\text{H}_{12}\text{N}^+$	168
Al_2^+	291	$\text{BC}_6\text{H}_{12}\text{NO}_3^+$	268
Al^2+	291	$\text{BC}_6\text{H}_{14}\text{N}_3^+$	169
Al_2Au^+	545	$\text{BC}_6\text{H}_{18}\text{N}_3^+$	169
Al_2Br_6^+	475	$\text{BC}_8\text{H}_{11}\text{O}_2^+$	224
Al_2Cl_6^+	393	$\text{BC}_8\text{H}_{17}\text{N}_2^+$	168
Am^+	555	$\text{BC}_8\text{H}_{19}\text{N}_2^+$	168
Ar^+	402	$\text{BC}_9\text{H}_{11}\text{N}_2^+$	168
Ar^{+2}	403	$\text{BC}_9\text{H}_{13}\text{N}_2^+$	168
ArKr^+	482	$\text{BC}_9\text{H}_{16}\text{N}^+$	168
ArXe^+	520	$\text{BC}_{10}\text{H}_{11}\text{N}_2^+$	168
Ar_2^+	403	$\text{BC}_{10}\text{H}_{15}\text{N}_2^+$	169
As^+	454	$\text{BC}_{10}\text{H}_{20}\text{N}^+$	168
AsBr^+	481	$\text{BC}_{11}\text{H}_{13}\text{Co}^+$	437
AsBr_2^+	481	$\text{BC}_{12}\text{H}_{10}^+$	122
AsBr_3^+	481	$\text{BC}_{12}\text{H}_{18}\text{SBr}^+$	478
AsI_3^+	517	$\text{BC}_{12}\text{H}_{18}\text{SCI}^+$	400
AsTl^+	549	$\text{BC}_{12}\text{H}_{19}\text{S}^+$	342
As_2^+	454	$\text{BC}_{13}\text{H}_2\text{OS}^+$	360
As_4^+	454	$\text{BC}_{13}\text{H}_{21}\text{S}^+$	342
Au^+	544	$\text{BC}_{14}\text{H}_{10}^+$	122
AuEu^+	545	$\text{BC}_{16}\text{H}_{15}\text{Co}^+$	437
Au_2^+	544	$\text{BC}_{16}\text{H}_{20}\text{N}^+$	168
Au_2Eu^+	545	$\text{BC}_{18}\text{H}_{15}^+$	122
B^+	43	$\text{BC}_{18}\text{H}_{25}\text{U}^+$	554
BAu^+	544	$\text{BC}_{21}\text{H}_{15}\text{F}_4^+$	278
$\text{BCH}_3\text{Br}_2^+$	467	BCl^+	371
$\text{BCH}_3\text{Cl}_2^+$	377	BCl_2^+	371
BCH_3F_2^+	278	BCl_3^+	371
BCH_3O^+	224	BF^+	269
BCH_8N^+	167	BFCl^+	389
$\text{BC}_2\text{H}_6\text{Br}^+$	467	BFCl_2^+	389
$\text{BC}_2\text{H}_6\text{Cl}^+$	377	BF_2^+	269
$\text{BC}_2\text{H}_6\text{F}^+$	278	BF_2Cl^+	389
$\text{BC}_2\text{H}_6\text{I}^+$	511	BF_3^+	269
$\text{BC}_2\text{H}_6\text{NBr}_2^+$	470	BO^+	172
$\text{BC}_2\text{H}_6\text{NCl}_2^+$	381	BOAu^+	544
$\text{BC}_2\text{H}_6\text{NF}_2^+$	283	BOF^+	283
$\text{BC}_2\text{H}_6\text{NI}_2^+$	512	BOF_2^+	284
$\text{BC}_2\text{H}_6\text{N}^+$	167	BO_2^+	172
$\text{BC}_2\text{H}_6\text{N}^+$	167	BO_2Ba^+	521
$\text{BC}_2\text{H}_6\text{NF}_2\text{P}^+$	323	BO_2In^+	496
BC_3H_9^+	122	BO_2K^+	404

BO_2K^+	404	$\text{B}_4\text{C}_2\text{H}_{14}\text{NF}_2\text{P}^+$	324
BO_2Na^+	290	$\text{B}_4\text{C}_3\text{H}_6\text{O}_3\text{Fe}^+$	432
BO_2Na_2^+	290	B_5CH_9^+	121
BO_2Tl^+	548	$\text{B}_5\text{CH}_{11}^+$	121
BO_2Ti_2^+	548	$\text{B}_5\text{C}_2\text{H}_7^+$	121
BO_4W^+	533	$\text{B}_5\text{C}_3\text{H}_6\text{O}_3\text{Fe}^+$	432
BO_7W_2^+	534	$\text{B}_5\text{C}_5\text{H}_3\text{O}_5\text{Fe}^+$	433
$\text{BO}_{10}\text{W}_3^+$	534	$\text{B}_8\text{C}_2\text{H}_{10}^+$	122
$\text{BO}_{13}\text{W}_4^+$	534	$\text{B}_9\text{CH}_{11}\text{S}^+$	342
BP^+	310	$\text{B}_{10}\text{C}_2\text{H}_{12}^+$	122
BSCl^+	399	Ba^+	521
$\text{B}_2\text{C}_2\text{H}_6\text{S}_3^-$	342	Ba^{+2}	521
$\text{B}_2\text{C}_2\text{H}_7\text{NS}_2^+$	350	BaI^+	522
$\text{B}_2\text{C}_3\text{H}_9\text{NOS}^+$	364	Be^+	43
$\text{B}_2\text{C}_3\text{H}_9\text{NS}_2^+$	351	$\text{BeC}_5\text{H}_5\text{Br}^+$	467
$\text{B}_2\text{C}_3\text{H}_9\text{N}_3\text{Br}_2^+$	470	$\text{BeC}_5\text{H}_5\text{Cl}^+$	377
$\text{B}_2\text{C}_3\text{H}_9\text{N}_3\text{Cl}_2^+$	381	BeC_6H_5^+	121
$\text{B}_2\text{C}_3\text{H}_{10}\text{N}_2\text{S}^+$	350	BeC_6H_8^+	121
$\text{B}_2\text{C}_3\text{H}_{11}\text{N}_3^+$	169	BeC_7H_6^+	121
$\text{B}_2\text{C}_4\text{H}_{12}\text{N}_2\text{Br}_2^+$	470	BeC_8H_8^+	121
$\text{B}_2\text{C}_4\text{H}_{12}\text{N}_2\text{Cl}_2^+$	381	$\text{BeC}_{10}\text{H}_2\text{O}_4\text{F}_{12}^+$	287
$\text{B}_2\text{C}_4\text{H}_{12}\text{N}_2\text{F}_2^+$	283	$\text{BeC}_{10}\text{H}_{10}^+$	121
$\text{B}_2\text{C}_4\text{H}_{12}\text{N}_2\text{O}^+$	267	$\text{BeC}_{10}\text{H}_{14}\text{O}_4^+$	224
$\text{B}_2\text{C}_4\text{H}_{12}\text{N}_2\text{S}^+$	350	$\text{BeC}_{12}\text{H}_{10}^+$	121
$\text{B}_2\text{C}_4\text{H}_{12}\text{N}_4\text{Cl}_2^+$	381	BeCl_2^+	371
$\text{B}_2\text{C}_4\text{H}_{13}\text{N}_3^+$	169	BeF^+	269
$\text{B}_2\text{C}_5\text{H}_{15}\text{N}_3^+$	169	BeFCl^+	389
$\text{B}_2\text{C}_5\text{H}_{15}\text{N}_3\text{O}_2^+$	268	BeF_2^+	269
$\text{B}_2\text{C}_5\text{H}_{15}\text{N}_3\text{S}_2^+$	351	Bi^+	551
$\text{B}_2\text{C}_5\text{H}_{16}\text{N}_2\text{SiS}^+$	368	Bi_2^+	551
$\text{B}_2\text{C}_6\text{H}_{18}\text{N}_2^+$	168	Bi_3^+	551
$\text{B}_2\text{C}_6\text{H}_{18}\text{N}_3\text{As}^+$	455	Bi_4^+	551
$\text{B}_2\text{C}_6\text{H}_{18}\text{N}_3\text{P}^+$	316	Bk^+	555
$\text{B}_2\text{C}_6\text{H}_{18}\text{N}_4^+$	169	Br^+	462
$\text{B}_2\text{C}_6\text{H}_{18}\text{N}_4\text{O}_2^+$	268	BrAg^+	494
$\text{B}_2\text{C}_6\text{H}_{18}\text{N}_4\text{S}_2^+$	351	BrAg_2^+	494
$\text{B}_2\text{C}_7\text{H}_{21}\text{N}_3\text{Si}^+$	304	BrBa^+	522
$\text{B}_2\text{C}_7\text{H}_{21}\text{N}_3\text{Sn}^+$	500	BrCs^+	521
$\text{B}_2\text{C}_7\text{H}_{21}\text{N}_5^+$	169	BrIn^+	517
$\text{B}_2\text{C}_8\text{H}_{21}\text{N}_3^+$	169	BrRb^+	483
$\text{B}_2\text{C}_8\text{H}_{24}\text{N}_4^+$	169	BrRb_2^+	483
$\text{B}_2\text{C}_8\text{H}_{24}\text{N}_5\text{P}^+$	316	BrSr^+	483
$\text{B}_2\text{C}_8\text{H}_{24}\text{N}_6^+$	170	BrTl^+	549
$\text{B}_2\text{C}_{12}\text{H}_{16}\text{Co}^+$	437	BrW^+	539
$\text{B}_2\text{C}_{12}\text{H}_{16}\text{O}_2\text{Co}^+$	439	BrYb^+	530
$\text{B}_2\text{C}_{22}\text{H}_{20}\text{Co}^+$	437	Br_2^+	463
B_2Cl_2^+	371	Br_2Ag_3^+	494
B_2F_4^+	269	Br_2Cd^+	495
$\text{B}_2\text{O}_6\text{W}^+$	534	Br_2Nd^+	526
$\text{B}_2\text{O}_{12}\text{W}_3^+$	534	Br_2Pb^+	551
$\text{B}_3\text{C}_2\text{H}_5^+$	121	Br_2Sn^+	503
$\text{B}_3\text{C}_2\text{H}_{11}\text{NF}_2\text{P}^+$	323	Br_2Tm^+	530
$\text{B}_3\text{C}_2\text{H}_{12}\text{NF}_2\text{P}^+$	323	Br_2W^+	539
$\text{B}_3\text{C}_3\text{H}_9\text{N}_3\text{Cl}_3^+$	381	Br_2Yb^+	530
$\text{B}_3\text{C}_3\text{H}_9\text{N}_3\text{F}_3^+$	283	Br_3Ag_3^+	494
$\text{B}_3\text{C}_3\text{H}_{12}\text{N}_3^+$	169	Br_3In^+	497
$\text{B}_3\text{C}_5\text{H}_5\text{O}_3\text{Fe}^+$	433	Br_3Sb^+	506
$\text{B}_3\text{C}_5\text{H}_7\text{O}_3\text{Fe}^+$	433	Br_3Tm^+	530
$\text{B}_3\text{C}_6\text{H}_{18}\text{N}_3^+$	169	Br_3W^+	539
$\text{B}_3\text{C}_8\text{H}_{24}\text{N}_5^+$	169	Br_3W_2^+	539
$\text{B}_4\text{C}_2\text{H}_4\text{Br}_2^+$	468	Br_4Hf^+	531
$\text{B}_4\text{C}_2\text{H}_4\text{Cl}_2^+$	377	Br_4W^+	539
$\text{B}_4\text{C}_2\text{H}_4\text{I}_2^+$	512	Br_4W_2^+	539
$\text{B}_4\text{C}_2\text{H}_5\text{Br}^+$	467	Br_4Zr^+	539
$\text{B}_4\text{C}_2\text{H}_5\text{Cl}^+$	377	Br_5W^+	484
$\text{B}_4\text{C}_2\text{H}_5\text{I}^+$	511	Br_5W_2^+	539
$\text{B}_4\text{C}_2\text{H}_6^+$	121	Br_5W_2^+	539
$\text{B}_4\text{C}_2\text{H}_8^+$	121	Br_7Re_3^+	540
$\text{B}_4\text{C}_2\text{H}_{12}\text{NF}_2\text{P}^+$	323		542

C ⁺	44	CH ₂ Cl ⁺	372
C ⁺²	44	CH ₂ Cl ₂ ⁺	375
CBr ⁺	463	CH ₂ D ⁺	46
CBr ₃ ⁺	463	CH ₂ DSi ⁺	293
CBr ₄ ⁺	464	CH ₂ D ₂ Si ⁺	293
CCl ⁺	371	CH ₂ F ⁺	274
CCl ₂ ⁺	371	CH ₂ FCl ⁺	391
CCl ₂ Br ₂ ⁺	479	CH ₂ F ₃ As ⁺	456
CCl ₃ ⁺	371	CH ₂ F ₃ P ⁺	321
CDN ⁺	125	CH ₂ F ₃ S ⁺	366
CDO ⁺	174	CH ₂ I ⁺	509
CD ₂ ⁺	45	CH ₂ I ₂ ⁺	511
CD ₂ O ⁺	174	CH ₂ N ⁺	125
CD ₃ ⁺	46	CH ₂ NF ⁺	281
CD ₃ NO ₂ ⁺	246	CH ₂ NF ₂ ⁺	282
CD ₃ O ⁺	175	CH ₂ NO ⁺	226
CD ₄ O ⁺	175	CH ₂ NS ⁺	343
CF ⁺	269	CH ₂ N ₂ ⁺	146
CFBr ₃ ⁺	474	CH ₂ N ₄ ⁺	164
FCF ⁺	389	CH ₂ O ⁺	174
FCF ₂ ⁺	390	CH ₂ OAs ⁺	455
FCF ₃ ⁺	390	CH ₂ OS ⁺	352
CFP ⁺	321	CH ₂ O ₂ ⁺	204
CFSCI ⁺	401	CH ₂ PCl ₃ ⁺	397
CF ₂ ⁺	270	CH ₂ PS ⁺	368
CF ₂ Br ₂ ⁺	473	CH ₂ S ⁺	329
CF ₂ Cl ⁺	389	CH ₂ S ₂ ⁺	337
CF ₂ ClBr ⁺	479	CH ₃ ⁺	45
CF ₂ Cl ₂ ⁺	390	CH ₃ AlBr ₂ ⁺	476
CF ₂ PCl ₃ ⁺	399	CH ₃ AlI ₂ ⁺	514
CF ₂ S ⁺	365	CH ₃ B ⁺	464
CF ₃ ⁺	271	CH ₃ Cl ⁺	372
CF ₃ Br ⁺	473	CH ₃ Cl ₂ Ge ⁺	453
CF ₃ Cl ⁺	390	CH ₃ Cl ₃ Ge ⁺	453
CF ₃ PCl ₃ ⁺	399	CH ₃ Cl ₃ Ti ⁺	407
CF ₄ S ⁺	365	CH ₃ D ⁺	46
CF ₄ ⁺	271	CH ₃ DO ⁺	175
CF ₃ Br ⁺	473	CH ₃ DSi ⁺	293
CF ₃ Cl ⁺	390	CH ₃ D ₂ Si ⁺	294
CF ₃ I ⁺	514	CH ₃ F ₂ P ⁺	321
CF ₃ IHg ⁺	548	CH ₃ F ₂ Si ⁺	308
CF ₃ PCl ₂ ⁺	398	CH ₃ F ₃ Si ⁺	308
CF ₄ ⁺	271	CH ₃ Ga ⁺	447
CH ⁺	44	CH ₃ I ⁺	509
CHBr ⁺	464	CH ₃ N ⁺	126
CHBr ₂ ⁺	466	CH ₃ NB ₂ ⁺	469
CHBr ₃ ⁺	467	CH ₃ NCI ₂ ⁺	381
CHCl ₂ ⁺	375	CH ₃ NF ₄ ⁺	323
CHCl ₃ ⁺	376	CH ₃ NO ⁺	226
CHD ⁺	45	CH ₃ NOGe ⁺	451
CHDO ⁺	174	CH ₃ NOSi ⁺	306
CHD ₂ ⁺	46	CH ₃ NO ₂ ⁺	246
CHD ₂ O ⁺	175	CH ₃ NS ⁺	343
CHD ₃ ⁺	46	CH ₃ NSGe ⁺	453
CHD ₃ O ⁺	175	CH ₃ NSiS ⁺	368
CHF ⁺	274	CH ₃ N ₂ ⁺	146
CHFCI ₂ ⁺	392	CH ₃ N ₃ ⁺	162
CHF ₂ ⁺	276	CH ₃ O ⁺	175
CHF ₂ Cl ⁺	391	CH ₃ OAs ⁺	455
CHF ₃ ⁺	277	CH ₃ OPCl ₂ ⁺	397
CHI ₂ ⁺	511	CH ₃ OPSCl ₂ ⁺	402
CHI ₃ ⁺	511	CH ₃ OS ⁺	352
CHN ⁺	125	CH ₃ O ₂ ⁺	204
CHNF ₂ ⁺	282	CH ₃ O ₂ FS ⁺	367
CHNO ⁺	226	CH ₃ O ₂ F ₂ P ⁺	324
CHNS ⁺	343	CH ₃ O ₂ PBr ₂ ⁺	477
CHO ⁺	174	CH ₃ O ₂ PCl ₂ ⁺	398
CHOF ⁺	284	CH ₃ O ₂ SCl ⁺	401
CHOMn ⁺	422	CH ₃ PCl ₂ ⁺	396
CHO ₂ ⁺	204	CH ₃ PCl ₂ Se ⁺	462
CHP ⁺	310		
CHS ⁺	328		
CH ₂ ⁺	44		
CH ₂ Br ⁺	464		
CH ₂ Br ₂ ⁺	466		

CH ₃ PSBr ₂ ⁺	479	COF ₄ ⁺	284
CH ₃ PSCl ₂ ⁺	402	COF ₆ SiP ₂ Cl ₃ Co ⁺	441
CH ₃ S ⁺	329	COF ₁₂ P ₄ Fe ⁺	434
CH ₃ Si ⁺	293	COFe ⁺	430
CH ₄ ⁺	46	COMn ⁺	422
CH ₄ N ⁺	126	COMo ⁺	486
CH ₄ NBr ⁺	468	CONi ⁺	443
CH ₄ NCl ⁺	378	COS ⁺	352
CH ₄ N ₂ ⁺	146	COSe ⁺	460
CH ₄ N ₂ O ⁺	240	COSiCl ₃ Co ⁺	441
CH ₄ N ₂ S ⁺	346	COW ⁺	534
CH ₄ O ⁺	175	CO ₃ K ₂ ⁺	404
CH ₄ OAs ⁺	455	CP ⁺	310
CH ₄ OP ⁺	317	CPCl ₅ ⁺	396
CH ₄ OS ⁺	352	CP ₂ ⁺	310
CH ₄ O ²⁺	175	CRh ⁺	491
CH ₄ O ₂ P ⁺	317	CRhCe ⁺	524
CH ₄ O ₃ P ⁺	318	CRuCe ⁺	524
CH ₄ S ⁺	329	CS ⁺	327
CH ₄ S ₂ ⁺	337	CSCl ₂ ⁺	399
CH ₄ Si ⁺	293	CSCr ⁺	418
CH ₄ SiCl ₂ ⁺	395	CSFe ⁺	434
CH ₅ N ⁺	126	CSFe ₂ ⁺	434
CH ₅ NO ⁺	227	CSMn ⁺	425
CH ₅ O ₂ P ⁺	318	CSMnI ⁺	516
CH ₅ P ⁺	310	CSMo ⁺	489
CH ₅ Si ⁺	293	CSSe ⁺	461
CH ₆ N ₂ ⁺	146	CSW ⁺	538
CH ₆ OSi ⁺	305	CS ₂ ⁺	328
CH ₆ Si ⁺	294	CS ₂ ⁺	458
CH ₆ SiS ⁺	367	CS ₂ E ₂ ⁺	458
CH ₉ NSi ₂ ⁺	304	CSiCe ⁺	524
CLa ⁺	522	CSiP ⁺	325
CN ⁺	124	CSi ₂ ⁺	293
CNEu ⁺	527	CTh ⁺	552
CNF ⁺	279	CU ⁺	554
CNFP ⁺	322	C ₂ ⁺	44
CNF ₂ P ⁺	322	C ₂ Al ⁺	291
CNF ₂ PS ⁺	370	C ₂ Al ₂ ⁺	291
CNGa ⁺	448	C ₂ Ce ⁺	523
CNK ⁺	404	C ₂ Cl ₆ ⁺	371
CNK ₂ ⁺	404	C ₂ D ⁺	46
CNOBr ⁺	472	C ₂ D ₂ ⁺	47
CNOBr ₃ ⁺	472	C ₂ D ₃ ⁺	48
CNOCl ⁺	385	C ₂ D ₃ O ⁺	178
CNOClBr ₂ ⁺	479	C ₂ D ₄ ⁺	49
CNOCl ₂ Br ⁺	479	C ₂ D ₄ O ⁺	179
CNOCl ₃ ⁺	385	C ₂ Eu ⁺	527
CNOFCl ₂ ⁺	393	C ₂ FCl ₂ ⁺	390
CNOF ₂ Cl ⁺	393	C ₂ F ₂ ⁺	270
CNOF ₂ P ⁺	324	C ₂ F ₂ Cl ⁺	390
CNOF ₃ ⁺	287	C ₂ F ₂ Cl ₂ ⁺	390
CNOI ⁺	513	C ₂ F ₃ ⁺	271
CNO ₂ ⁺	226	C ₂ F ₃ Br ⁺	473
CNO ₃ F ₃ Hg ⁺	547	C ₂ F ₃ Cl ⁺	390
CNP ⁺	525	C ₂ F ₃ Cl ₃ ⁺	391
CN ₂ F ₂ ⁺	280	C ₂ F ₃ S ₂ Cl ⁺	401
CN ₃ F ₃ Hg ⁺	547	C ₂ F ₄ ⁺	272
CN ₄ ⁺	125	C ₂ F ₄ Br ₂ ⁺	474
CO ⁺	173	C ₂ F ₄ Cl ₂ ⁺	390
CO ²⁺	173	C ₂ F ₄ I ₂ ⁺	514
COBr ₂ ⁺	470	C ₂ F ₄ S ₂ ⁺	365
COCl ⁺	382	C ₂ F ₅ ⁺	272
COCl ₂ ⁺	382	C ₂ F ₅ Cl ⁺	390
COCo ⁺	438	C ₂ F ₅ I ⁺	514
COCr ⁺	410	C ₂ F ₆ ⁺	272
COF ⁺	284	C ₂ F ₆ PCl ⁺	398
COF ₂ ⁺	284	C ₂ F ₆ S ₂ Hg ⁺	547
COF ₃ SiPCl ₃ Co ⁺	441	C ₂ Fe ⁺	429

C ₂ H ⁺	46	C ₂ H ₃ NO ⁺	227
C ₂ HBr ⁺	464	C ₂ H ₃ NO ₃ ⁺	261
C ₂ HCl ⁺	372	C ₂ H ₃ NS ⁺	343
C ₂ HCl ₅ ⁺	377	C ₂ H ₃ N ₃ ⁺	162
C ₂ HD ⁺	47	C ₂ H ₃ N ₃ O ⁺	245
C ₂ HD ₂ ⁺	48	C ₂ H ₃ O ⁺	176
C ₂ HD ₃ ⁺	49	C ₂ H ₃ OBr ⁺	470
C ₂ HD ₃ O ⁺	179	C ₂ H ₃ OCl ⁺	382
C ₂ HF ⁺	274	C ₂ H ₃ OF ⁺	284
C ₂ HFCl ⁺	391	C ₂ H ₃ OF ₃ ⁺	286
C ₂ HF ₂ ⁺	276	C ₂ H ₃ OPCl ₂ ⁺	398
C ₂ HF ₂ Cl ⁺	391	C ₂ H ₃ O ₂ ⁺	204
C ₂ HF ₃ ⁺	277	C ₂ H ₃ O ₂ Br ⁺	471
C ₂ HF ₃ Cl ₂ ⁺	392	C ₂ H ₃ O ₂ Cl ⁺	384
C ₂ HF ₆ P ⁺	322	C ₂ H ₃ O ₂ I ⁺	513
C ₂ HI ⁺	509	C ₂ H ₃ P ⁺	311
C ₂ HOCl ⁺	382	C ₂ H ₃ S ⁺	329
C ₂ HOCl ₃ ⁺	385	C ₂ H ₄ ⁺	48
C ₂ HO ₂ F ₃ ⁺	286	C ₂ H ₄ Br ⁺	464
C ₂ HO ₂ Mn ⁺	422	C ₂ H ₄ Br ₂ ⁺	466
C ₂ H ₂ ⁺	47	C ₂ H ₄ Cl ⁺	372
C ₂ H ₂ Br ₂ ⁺	466	C ₂ H ₄ ClBr ⁺	479
C ₂ H ₂ Cl ⁺	372	C ₂ H ₄ Cl ₂ ⁺	375
C ₂ H ₂ Cl ₂ ⁺	375	C ₂ H ₄ DO ⁺	179
C ₂ H ₂ Cl ₄ ⁺	377	C ₂ H ₄ F ⁺	274
C ₂ H ₂ Co ₂ ⁺	437	C ₂ H ₄ FBr ⁺	474
C ₂ H ₂ D ₂ ⁺	48	C ₂ H ₄ F ₂ ⁺	276
C ₂ H ₂ D ₃ ⁺	49	C ₂ H ₄ Ga ⁺	448
C ₂ H ₂ D ₃ O ⁺	179	C ₂ H ₄ I ₂ ⁺	511
C ₂ H ₂ F ⁺	274	C ₂ H ₄ N ⁺	126
C ₂ H ₂ FBr ⁺	474	C ₂ H ₄ NO ⁺	227
C ₂ H ₂ FCl ⁺	391	C ₂ H ₄ NS ⁺	343
C ₂ H ₂ F ₂ ⁺	276	C ₂ H ₄ N ₂ ⁺	146
C ₂ H ₂ F ₂ Br ₂ ⁺	475	C ₂ H ₄ N ₂ O ₂ ⁺	254
C ₂ H ₂ F ₃ I ⁺	514	C ₂ H ₄ N ₄ ⁺	164
C ₂ H ₂ I ₂ ⁺	511	C ₂ H ₄ O ⁺	178
C ₂ H ₂ N ⁺	126	C ₂ H ₄ OAs ⁺	455
C ₂ H ₂ NBr ⁺	468	C ₂ H ₄ OPCl ₃ ⁺	398
C ₂ H ₂ NCl ⁺	378	C ₂ H ₄ OS ⁺	352
C ₂ H ₂ NF ⁺	281	C ₂ H ₄ O ₂ ⁺	205
C ₂ H ₂ NOCl ⁺	389	C ₂ H ₄ O ₂ S ⁺	357
C ₂ H ₂ N ₂ Se ⁺	460	C ₂ H ₄ O ₃ ⁺	219
C ₂ H ₂ N ₃ Br ⁺	469	C ₂ H ₄ O ₃ S ⁺	359
C ₂ H ₂ N ₃ Cl ⁺	380	C ₂ H ₄ O ₄ ⁺	222
C ₂ H ₂ N ₄ ⁺	164	C ₂ H ₄ PCl ₃ ⁺	397
C ₂ H ₂ O ⁺	175	C ₂ H ₄ PSCl ₃ ⁺	402
C ₂ H ₂ OCl ₂ ⁺	384	C ₂ H ₄ S ⁺	329
C ₂ H ₂ O ₂ ⁺	204	C ₂ H ₃ S ³ ⁺	339
C ₂ H ₂ O ₄ ⁺	222	C ₂ H ₄ Si ⁺	294
C ₂ H ₂ S ⁺	329	C ₂ H ₅ ⁺	49
C ₂ H ₂ S ₃ ⁺	339	C ₂ H ₅ Br ⁺	464
C ₂ H ₂ Se ⁺	459	C ₂ H ₅ Cl ⁺	372
C ₂ H ₃ ⁺	47	C ₂ H ₅ F ⁺	274
C ₂ H ₃ Br ⁺	464	C ₂ H ₅ I ⁺	509
C ₂ H ₃ Cl ⁺	372	C ₂ H ₅ N ⁺	126
C ₂ H ₃ Cl ₂ ⁺	377	C ₂ H ₅ NO ⁺	227
C ₂ H ₃ D ⁺	48	C ₂ H ₅ NO ₂ ⁺	246
C ₂ H ₃ DO ⁺	178	C ₂ H ₅ NS ⁺	343
C ₂ H ₃ D ₂ ⁺	49	C ₂ H ₅ O ⁺	179
C ₂ H ₃ D ₂ O ⁺	179	C ₂ H ₅ OAs ⁺	455
C ₂ H ₃ D ₃ ⁺	49	C ₂ H ₅ OBr ⁺	470
C ₂ H ₃ D ₃ O ⁺	180	C ₂ H ₅ OCl ⁺	383
C ₂ H ₃ F ⁺	274	C ₂ H ₅ OF ⁺	284
C ₂ H ₃ F ₂ ⁺	276	C ₂ H ₅ OI ⁺	512
C ₂ H ₃ F ₂ Cl ⁺	391	C ₂ H ₅ OPSCl ₂ ⁺	402
C ₂ H ₃ F ₃ ⁺	277	C ₂ H ₅ OSiCl ₃ ⁺	395
C ₂ H ₃ Ga ⁺	447	C ₂ H ₅ O ₂ ⁺	205
C ₂ H ₃ I ⁺	509	C ₂ H ₅ O ₂ As ⁺	455
C ₂ H ₃ N ⁺	126	C ₂ H ₅ O ₂ PCl ₂ ⁺	398
C ₂ H ₃ NF ⁺	281	C ₂ H ₅ P ⁺	311

C ₂ H ₅ PCl ₂ ⁺	396	C ₂ H ₅ OCl ⁺	383
C ₂ H ₅ S ⁺	330	C ₂ H ₅ OP ⁺	317
C ₂ H ₅ SCl ⁺	399	C ₂ H ₅ OPS ₂ ⁺	369
C ₂ H ₅ Se ⁺	459	C ₂ H ₅ O ₂ As ⁺	455
C ₂ H ₆ ⁺	49	C ₂ H ₆ O ₂ PS ⁺	369
C ₂ H ₆ AlBr ⁺	476	C ₂ H ₆ O ₃ P ⁺	318
C ₂ H ₆ AlCl ⁺	393	C ₂ H ₆ P ⁺	311
C ₂ H ₆ AlI ⁺	514	C ₂ H ₆ PS ⁺	369
C ₂ H ₆ Cd ⁺	495	C ₂ H ₆ Si ⁺	294
C ₂ H ₆ ClGe ⁺	453	C ₂ H ₆ Ge ⁺	449
C ₂ H ₆ ClSn ⁺	503	C ₂ H ₆ N ₂ ⁺	146
C ₂ H ₆ Cl ₂ Ge ⁺	453	C ₂ H ₆ N ₂ O ₂ P ⁺	320
C ₂ H ₆ Cl ₂ Sn ⁺	503	C ₂ H ₆ N ₂ S ⁺	346
C ₂ H ₆ FP ⁺	321	C ₂ H ₆ Si ⁺	294
C ₂ H ₆ FSi ⁺	308	C ₂ H ₆ NSi ⁺	303
C ₂ H ₆ F ₂ Ge ⁺	452	C ₂ H ₁₀ N ₂ OP ⁺	319
C ₂ H ₆ F ₂ Si ⁺	308	C ₂ La ⁺	522
C ₂ H ₆ F ₂ SiAs ⁺	456	C ₂ Lu ⁺	531
C ₂ H ₆ Ga ⁺	448	C ₂ NOF ₃ Hg ⁺	547
C ₂ H ₆ Hg ⁺	546	C ₂ NOF ₆ ⁺	287
C ₂ H ₆ N ⁺	126	C ₂ N ₂ ⁺	125
C ₂ H ₆ NBr ⁺	468	C ₂ N ²⁺	124
C ₂ H ₆ NCl ⁺	378	C ₂ N ₂ F ₆ ⁺	280
C ₂ H ₆ NF ₂ ⁺	282	C ₂ N ₂ K ₃ ⁺	404
C ₂ H ₆ NF ₂ P ⁺	322	C ₂ N ₂ O ⁺	226
C ₂ H ₆ NF ₃ Si ⁺	309	C ₂ N ₂ O ₄ Fe ⁺	433
C ₂ H ₆ NF ₄ P ⁺	323	C ₂ N ₂ S ⁺	343
C ₂ H ₆ NOPCl ₂ ⁺	398	C ₂ N ₂ S ₂ ⁺	343
C ₂ H ₆ NPCl ₂ ⁺	397	C ₂ OCl ₂ ⁺	382
C ₂ H ₆ NPSCl ₂ ⁺	402	C ₂ OCl ₃ ⁺	382
C ₂ H ₆ NSiCl ₃ ⁺	395	C ₂ OCl ₄ ⁺	382
C ₂ H ₆ N ₂ ⁺	146	C ₂ OF ₃ ⁺	284
C ₂ H ₆ N ₂ O ⁺	240	C ₂ OF ₆ ⁺	430
C ₂ H ₆ N ₂ O ₂ ⁺	254	C ₂ OSCr ⁺	418
C ₂ H ₆ N ₂ P ₂ F ₆ ⁺	323	C ₂ OSMo ⁺	489
C ₂ H ₆ N ₂ S ₂ ⁺	350	C ₂ OSW ⁺	538
C ₂ H ₆ O ⁺	179	C ₂ O ₂ Br ₂ ⁺	470
C ₂ H ₆ OAs ⁺	455	C ₂ O ₂ Cl ₂ ⁺	382
C ₂ H ₆ OPCl ⁺	397	C ₂ O ₂ Co ⁺	438
C ₂ H ₆ OPS ⁺	369	C ₂ O ₂ Cr ⁺	411
C ₂ H ₆ OPS ₂ ⁺	369	C ₂ O ₂ F ₂ ⁺	284
C ₂ H ₆ OS ⁺	352	C ₂ O ₂ SiP ₂ Cl ₃ Co ⁺	441
C ₂ H ₆ O ₂ ⁺	205	C ₂ O ₂ F ₃ P ₃ Fe ⁺	434
C ₂ H ₆ O ₂ As ⁺	455	C ₂ O ₃ Fe ⁺	430
C ₂ H ₆ O ₂ P ⁺	318	C ₂ O ₂ Mn ⁺	422
C ₂ H ₆ O ₂ PS ⁺	369	C ₂ O ₂ Mo ⁺	486
C ₂ H ₆ O ₂ PS ₂ ⁺	369	C ₂ O ₂ Ni ⁺	443
C ₂ H ₆ O ₂ S ⁺	357	C ₂ O ₂ SiCl ₃ Co ⁺	441
C ₂ H ₆ O ₃ P ⁺	318	C ₂ O ₂ W ⁺	534
C ₂ H ₆ O ₃ PS ⁺	369	C ₂ P ⁺	310
C ₂ H ₆ O ₃ S ⁺	359	C ₂ Rh ⁺	491
C ₂ H ₆ PCl ⁺	396	C ₂ RhCe ⁺	524
C ₂ H ₆ PClSe ⁺	462	C ₂ RuCe ⁺	524
C ₂ H ₆ PSBr ⁺	479	C ₂ S ₂ Cl ₄ ⁺	399
C ₂ H ₆ PSCl ⁺	402	C ₂ S ₂ Fe ₂ ⁺	434
C ₂ H ₆ S ⁺	330	C ₂ S ₂ Mn ₂ ⁺	425
C ₂ H ₆ S ₂ ⁺	337	C ₂ S ₄ ⁺	328
C ₂ H ₆ Se ⁺	459	C ₂ Sc ⁺	405
C ₂ H ₆ Si ⁺	294	C ₂ Si ⁺	293
C ₂ H ₆ SiCl ⁺	394	C ₂ Th ⁺	552
C ₂ H ₆ SiCl ₂ ⁺	395	C ₂ Ti ⁺	406
C ₂ H ₆ SiCl ₃ As ⁺	457	C ₂ U ⁺	554
C ₂ H ₆ SiPCl ₃ ⁺	399	C ₂ Y ⁺	483
C ₂ H ₆ SiI ₂ ⁺	299	C ₂ Zr ⁺	483
C ₂ H ₆ Te ⁺	506	C ₃ ⁺	44
C ₂ H ₆ Zn ⁺	446	C ₃ D ₆ ⁺	53
C ₂ H ₇ As ⁺	454	C ₃ D ₆ O ⁺	181
C ₂ H ₇ N ⁺	127	C ₃ F ⁺	270
C ₂ H ₇ NO ⁺	227	C ₃ F ₂ ⁺	270
C ₂ H ₇ NOS ⁺	360	C ₃ F ₃ ⁺	271

C ₃ F ₃ Br ⁺	473	C ₃ H ₄ O ⁺	180
C ₃ F ₃ Cl ⁺	390	C ₃ H ₄ OS ₂ ⁺	360
C ₃ F ₃ I ⁺	514	C ₃ H ₄ O ₂ ⁺	205
C ₃ F ₄ ⁺	272	C ₃ H ₄ O ₂ S ⁺	357
C ₃ F ₆ ⁺	272	C ₃ H ₄ O ₃ ⁺	219
C ₃ F ₆ S ₃ ⁺	365	C ₃ H ₄ O ₄ ⁺	222
C ₃ F ₉ P ⁺	321	C ₃ H ₄ S ₃ ⁺	340
C ₃ H ⁺	49	C ₃ H ₅ ⁺	51
C ₃ HD ₃ ⁺	53	C ₃ H ₅ Br ⁺	464
C ₃ HD ₆ ⁺	53	C ₃ H ₅ Cl ⁺	372
C ₃ HF ⁺	274	C ₃ H ₅ CHg ⁺	547
C ₃ HF ₂ ⁺	276	C ₃ H ₅ D ⁺	53
C ₃ HF ₃ ⁺	277	C ₃ H ₅ D ₃ O ⁺	181
C ₃ HN ⁺	127	C ₃ H ₅ F ⁺	275
C ₃ HNF ₆ ⁺	283	C ₃ H ₅ I ⁺	509
C ₃ HO ⁺	180	C ₃ H ₅ N ⁺	128
C ₃ HOF ₄ Cl ⁺	393	C ₃ H ₅ NO ⁺	227
C ₃ HO ₃ Mn ⁺	422	C ₃ H ₅ NOS ⁺	360
C ₃ HS ⁺	330	C ₃ H ₅ NO ₂ ⁺	247
C ₃ H ₂ ⁺	49	C ₃ H ₅ NS ⁺	343
C ₃ H ₂ D ₆ ⁺	53	C ₃ H ₅ NS ₂ ⁺	349
C ₃ H ₂ F ⁺	274	C ₃ H ₅ N ₃ O ⁺	245
C ₃ H ₂ F ₂ ⁺	276	C ₃ H ₅ O ⁺	180
C ₃ H ₂ N ⁺	127	C ₃ H ₅ OCl ⁺	383
C ₃ H ₂ NCl ⁺	378	C ₃ H ₅ OF ⁺	284
C ₃ H ₂ NO ⁺	227	C ₃ H ₅ OPCl ₂ ⁺	398
C ₃ H ₂ N ²⁺	128	C ₃ H ₅ QS ⁺	353
C ₃ H ₂ N ₂ ⁺	147	C ₃ H ₅ S ⁺	330
C ₃ H ₂ N ₂ O ₂ S ⁺	364	C ₃ H ₅ S ₂ ⁺	337
C ₃ H ₂ N ₂ O ₃ ⁺	263	C ₃ H ₆ ^{...}	52
C ₃ H ₂ O ⁺	180	C ₃ H ₆ Br ⁺	464
C ₃ H ₂ OCo ₂ ⁺	438	C ₃ H ₆ Br ₂ ⁺	466
C ₃ H ₂ OF ₃ Br ⁺	475	C ₃ H ₆ Cl ⁺	373
C ₃ H ₂ OF ₄ ⁺	286	C ₃ H ₆ D ⁺	53
C ₃ H ₂ OF ₆ ⁺	286	C ₃ H ₆ F ⁺	275
C ₃ H ₂ O ₂ S ⁺	357	C ₃ H ₆ FBr ⁺	474
C ₃ H ₂ O ₃ ⁺	219	C ₃ H ₆ F ₂ ⁺	277
C ₃ H ₂ S ₃ ⁺	340	C ₃ H ₆ N ⁺	128
C ₃ H ₃ ⁺	50	C ₃ H ₆ NF ⁺	281
C ₃ H ₃ Br ⁺	464	C ₃ H ₆ NF ₃ PCl ⁺	399
C ₃ H ₃ Cl ⁺	372	C ₃ H ₆ NO ⁺	227
C ₃ H ₃ Co ⁺	437	C ₃ H ₆ NOCl ⁺	385
C ₃ H ₃ D ₃ ⁺	53	C ₃ H ₆ NS ⁺	344
C ₃ H ₃ D ₃	274	C ₃ H ₆ NSe ⁺	459
C ₃ H ₃ F ⁺	429	C ₃ H ₆ N ₂ ⁺	147
C ₃ H ₃ Fe ⁺	509	C ₃ H ₆ N ₂ OF ₃ P ⁺	324
C ₃ H ₃ I ⁺	128	C ₃ H ₆ N ₂ OPCl ₃ ⁺	398
C ₃ H ₃ N ⁺	227	C ₃ H ₆ N ₂ O ₂ ⁺	255
C ₃ H ₃ NO ⁺	343	C ₃ H ₆ N ₂ O ₃ ⁺	263
C ₃ H ₃ NS ⁺	349	C ₃ H ₆ N ₂ S ⁺	346
C ₃ H ₃ NS ₂ ⁺	147	C ₃ H ₆ O ⁺	180
C ₃ H ₃ N ₂ ⁺	162	C ₃ H ₆ OS ⁺	353
C ₃ H ₃ N ₃ ⁺	441	C ₃ H ₆ OS ₂ ⁺	360
C ₃ H ₃ Ni ⁺	180	C ₃ H ₆ O ₂ ⁺	205
C ₃ H ₃ O ⁺	286	C ₃ H ₆ O ₂ S ⁺	357
C ₃ H ₃ OF ₃ ⁺	286	C ₃ H ₆ O ₃ ⁺	220
C ₃ H ₃ OF ₅ ⁺	286	C ₃ H ₆ O ₃ ⁺	220
C ₃ H ₃ O ₂ F ₃ ⁺	286	C ₃ H ₆ O ₃ ⁺	220
C ₃ H ₃ Ru ⁺	490	C ₃ H ₆ S ⁺	330
C ₃ H ₃ W ⁺	532	C ₃ H ₆ S ₂ ⁺	337
C ₃ H ₄ ⁺	50	C ₃ H ₆ S ₃ ⁺	340
C ₃ H ₄ D ₂ ⁺	53	C ₃ H ₆ SiCl ₂ ⁺	395
C ₃ H ₄ D ₃ ⁺	53	C ₃ H ₇ ⁺	53
C ₃ H ₄ D ₃ O ⁺	181	C ₃ H ₇ Br ⁺	464
C ₃ H ₄ F ⁺	274	C ₃ H ₇ Cl ⁺	373
C ₃ H ₄ N ₂ ⁺	147	C ₃ H ₇ F ⁺	275
C ₃ H ₄ N ₂ O ⁺	240	C ₃ H ₇ I ⁺	510
C ₃ H ₄ N ₂ O ₂ ⁺	254	C ₃ H ₇ N ⁺	128
C ₃ H ₄ N ₂ S ₂ ⁺	350	C ₃ H ₇ NO ⁺	227
C ₃ H ₄ N ₃ Br ⁺	469	C ₃ H ₇ NO ₂ ⁺	247
C ₃ H ₄ N ₃ Cl ⁺	380	C ₃ H ₇ NO ₂ S ⁺	363

C ₃ H ₇ NS ⁺	344	C ₃ NI ⁺	512
C ₃ H ₇ O ⁺	181	C ₃ NO ₄ Co ⁺	439
C ₃ H ₇ OBr ⁺	470	C ₃ N ₂ O ⁺	226
C ₃ H ₇ OCl ⁺	383	C ₃ N ₃ F ₃ ⁺	280
C ₃ H ₇ OF ⁺	284	C ₃ OF ₃ Cl ₃ ⁺	393
C ₃ H ₇ OI ⁺	512	C ₃ OF ₅ ⁺	284
C ₃ H ₇ O ₃ P ⁺	318	C ₃ OF ₅ Cl ⁺	393
C ₃ H ₇ S ⁺	330	C ₃ OF ₆ ⁺	284
C ₃ H ₇ Se ⁺	459	C ₃ O ₂ ⁺	174
C ₃ H ₈ ⁺	53	C ₃ O ₂ Fe ⁺	430
C ₃ H ₈ Hg ⁺	546	C ₃ O ₂ SCr ⁺	418
C ₃ H ₈ N ⁺	128	C ₃ O ₂ SMo ⁺	489
C ₃ H ₈ N ₂ ⁺	147	C ₃ O ₂ SW ⁺	538
C ₃ H ₈ N ₂ O ⁺	240	C ₃ O ₃ Cr ⁺	411
C ₃ H ₈ N ₂ O ₂ ⁺	255	C ₃ O ₃ F ₃ SiPCl ₂ Co ⁺	441
C ₃ H ₈ N ₂ S ⁺	346	C ₃ O ₃ F ₃ SiPCl ₃ Co ⁺	441
C ₃ H ₈ N ₂ S ₂ ⁺	350	C ₃ O ₃ F ₆ P ₂ Fe ⁺	434
C ₃ H ₈ O ⁺	181	C ₃ O ₃ F ₉ P ₃ Cr ⁺	418
C ₃ H ₈ OS ⁺	353	C ₃ O ₃ Fe ⁺	430
C ₃ H ₈ O ₄ P ⁺	319	C ₃ O ₃ Fe ⁺²	430
C ₃ H ₈ S ⁺	330	C ₃ O ₃ Mn ⁺	422
C ₃ H ₈ S ₂ ⁺	337	C ₃ O ₃ Mo ⁺	486
C ₃ H ₈ Si ⁺	294	C ₃ O ₃ Ni ⁺	443
C ₃ H ₉ Al ⁺	291	C ₃ O ₃ SiCl ₃ Co ⁺	441
C ₃ H ₉ As ⁺	454	C ₃ O ₄ W ⁺	534
C ₃ H ₉ BrPb ⁺	551	C ₃ S ₂ ⁺	328
C ₃ H ₉ BrSn ⁺	504	C ₃ Th ⁺	552
C ₃ H ₉ ClGe ⁺	453	C ₃ U ⁺	554
C ₃ H ₉ ClPb ⁺	551	C ₄ Br ₂ ⁺	463
C ₃ H ₉ ClSn ⁺	503	C ₄ Cl ₂ ⁺	371
C ₃ H ₉ FSi ⁺	308	C ₄ D ₄ ⁺	54
C ₃ H ₉ Ga ⁺	448	C ₄ D ₄ S ⁺	331
C ₃ H ₉ Ge ⁺	449	C ₄ D ₆ ⁺	55
C ₃ H ₉ N ⁺	128	C ₄ F ₂ ⁺	270
C ₃ H ₉ NO ⁺	228	C ₄ F ₃ ⁺	271
C ₃ H ₉ N ₃ F ₁₂ P ₆ Cr ⁺	418	C ₄ F ₄ ⁺	272
C ₃ H ₉ N ₃ F ₁₂ P ₆ Mo ⁺	489	C ₄ F ₆ ⁺	272
C ₃ H ₉ N ₃ F ₁₂ P ₆ W ⁺	538	C ₄ F ₆ Co ₂ ⁺	439
C ₃ H ₉ N ₃ Si ⁺	304	C ₄ F ₈ ⁺	273
C ₃ H ₉ OAs ⁺	455	C ₄ F ₁₂ As ₂ ⁺	456
C ₃ H ₉ OP ⁺	317	C ₄ F ₁₂ P ₂ ⁺	321
C ₃ H ₉ OSi ⁺	305	C ₄ F ₁₂ P ₄ ⁺	321
C ₃ H ₉ O ₂ PS ₂ ⁺	369	C ₄ HBr ⁺	464
C ₃ H ₉ O ₃ As ⁺	456	C ₄ HCl ⁺	373
C ₃ H ₉ O ₃ P ⁺	318	C ₄ HF ⁺	275
C ₃ H ₉ O ₃ PCr ⁺	416	C ₄ HI ⁺	510
C ₃ H ₉ O ₃ PS ⁺	369	C ₄ HN ₂ F ₃ ⁺	282
C ₃ H ₉ O ₃ PSe ⁺	461	C ₄ HO ₄ Co ⁺	438
C ₃ H ₉ O ₃ PW ⁺	536	C ₄ HO ₄ Mn ⁺	423
C ₃ H ₉ O ₄ P ⁺	319	C ₄ H ₂ ⁺	53
C ₃ H ₉ P ⁺	311	C ₄ H ₂ D ₄ ⁺	55
C ₃ H ₉ PS ⁺	369	C ₄ H ₂ F ₇ ⁺	277
C ₃ H ₉ Pb ⁺	550	C ₄ H ₂ F ₄ ⁺	278
C ₃ H ₉ Sb ⁺	505	C ₄ H ₂ N ₂ ⁺	147
C ₃ H ₉ Si ⁺	294	C ₄ H ₂ N ₂ F ₂ ⁺	282
C ₃ H ₉ SiBr ⁺	476	C ₄ H ₂ N ₂ S ⁺	346
C ₃ H ₉ SiCl ⁺	394	C ₄ H ₂ O ₂ ⁺	206
C ₃ H ₉ SiMn ⁺	424	C ₄ H ₂ O ₂ Cl ₂ ⁺	384
C ₃ H ₉ Sn ⁺	497	C ₄ H ₂ O ₂ Co ₂ ⁺	438
C ₃ H ₁₀ NP ⁺	315	C ₄ H ₂ O ₃ ⁺	220
C ₃ H ₁₀ N ₂ ⁺	147	C ₄ H ₃ O ₄ Fe ⁺	432
C ₃ H ₁₀ Si ⁺	295	C ₄ H ₃ SBr ₂ ⁺	478
C ₃ H ₁₀ Sn ⁺	497	C ₄ H ₃ SI ₂ ⁺	515
C ₃ H ₁₂ N ₃ OP ⁺	319	C ₄ H ₃ ⁺	54
C ₃ H ₁₆ Ge ⁺	449	C ₄ H ₃ BrTe ⁺	508
C ₃ La ⁺	522	C ₄ H ₃ ClSe ⁺	462
C ₃ NBr ⁺	468	C ₄ H ₃ ClTe ⁺	508
C ₃ NCl ⁺	378	C ₄ H ₃ I ⁺	510
C ₃ NF ⁺	279	C ₄ H ₃ N ⁺	129

C ₄ H ₃ NOS ⁺	361	C ₄ H ₆ O ₄ ⁺	222
C ₄ H ₃ NO ₂ S ⁺	363	C ₄ H ₆ S ⁺	331
C ₄ H ₃ NO ₂ Se ⁺	461	C ₄ H ₆ S ₃ ⁺	340
C ₄ H ₃ NO ₃ ⁺	261	C ₄ H ₆ SiCl ₂ ⁺	395
C ₄ H ₃ N ₂ F ⁺	281	C ₄ H ₇ ⁺	55
C ₄ H ₃ N ₂ OB ₂ ⁺	472	C ₄ H ₇ Br ⁺	465
C ₄ H ₃ N ₂ OCl ⁺	386	C ₄ H ₇ Ge ⁺	449
C ₄ H ₃ N ₂ OF ⁺	287	C ₄ H ₇ N ⁺	129
C ₄ H ₃ OBr ⁺	470	C ₄ H ₇ NO ⁺	228
C ₄ H ₃ OCHg ⁺	547	C ₄ H ₇ NO ₂ ⁺	247
C ₄ H ₃ O ₄ CoGe ⁺	454	C ₄ H ₇ NO ₃ ⁺	261
C ₄ H ₃ O ₄ SiCo ⁺	440	C ₄ H ₇ NS ₂ ⁺	349
C ₄ H ₃ S ⁺	331	C ₄ H ₇ N ₂ O ⁺	245
C ₄ H ₃ SBr ⁺	477	C ₄ H ₇ N ₂ S ⁺	348
C ₄ H ₃ SCl ⁺	399	C ₄ H ₇ O ⁺	182
C ₄ H ₃ SClHg ⁺	547	C ₄ H ₇ O ₃ P ⁺	318
C ₄ H ₃ SI ⁺	515	C ₄ H ₇ O ₄ PCl ₂ ⁺	398
C ₄ H ₃ TeI ⁺	519	C ₄ H ₇ Si ⁺	295
C ₄ H ₄ ⁺	54	C ₄ H ₇ Sn ⁺	497
C ₄ H ₄ F ₃ ⁺	277	C ₄ H ₈ ⁺	55
C ₄ H ₄ N ⁺	129	C ₄ H ₈ Br ₂ ⁺	466
C ₄ H ₄ NO ₂ Br ⁺	473	C ₄ H ₈ FBr ⁺	474
C ₄ H ₄ NO ₂ Cl ⁺	387	C ₄ H ₈ N ⁺	129
C ₄ H ₄ N ₂ ⁺	147	C ₄ H ₈ NO ⁺	228
C ₄ H ₄ N ₂ O ⁺	240	C ₄ H ₈ NOCl ⁺	385
C ₄ H ₄ N ₂ O ₂ ⁺	255	C ₄ H ₈ NO ₂ Se ⁺	461
C ₄ H ₄ N ₂ O ₃ ⁺	263	C ₄ H ₈ N ₂ ⁺	148
C ₄ H ₄ O ⁺	181	C ₄ H ₈ N ₂ O ₂ ⁺	255
C ₄ H ₄ OS ⁺	353	C ₄ H ₈ N ₂ S ⁺	346
C ₄ H ₄ O ₂ ⁺	206	C ₄ H ₈ N ₄ ⁺	164
C ₄ H ₄ O ₃ ⁺	220	C ₄ H ₈ O ⁺	182
C ₄ H ₄ O ₄ ⁺	222	C ₄ H ₈ OPCl ⁺	397
C ₄ H ₄ O ₈ Mo ₂ ⁺	487	C ₄ H ₈ OS ⁺	353
C ₄ H ₄ S ⁺	331	C ₄ H ₈ O ₂ ⁺	206
C ₄ H ₄ SSe ⁺	461	C ₄ H ₈ O ₃ S ⁺	358
C ₄ H ₄ STe ⁺	507	C ₄ H ₈ O ₄ ⁺	222
C ₄ H ₄ S ₂ ⁺	337	C ₄ H ₈ S ⁺	331
C ₄ H ₄ S ₃ ⁺	340	C ₄ H ₈ S ₂ ⁺	337
C ₄ H ₄ Se ⁺	459	C ₄ H ₈ S ₃ Sn ⁺	502
C ₄ H ₄ Te ⁺	506	C ₄ H ₈ Se ⁺	459
C ₄ H ₅ ⁺	54	C ₄ H ₈ Si ⁺	295
C ₄ H ₅ N ⁺	129	C ₄ H ₈ Te ⁺	506
C ₄ H ₅ NO ₂ ⁺	247	C ₄ H ₉ ⁺	56
C ₄ H ₅ NS ⁺	344	C ₄ H ₉ As ⁺	454
C ₄ H ₅ NS ₂ ⁺	349	C ₄ H ₉ Br ⁺	465
C ₄ H ₅ N ₃ ⁺	162	C ₄ H ₉ Cl ⁺	373
C ₄ H ₅ N ₃ O ⁺	245	C ₄ H ₉ F ₂ P ⁺	321
C ₄ H ₅ O ⁺	182	C ₄ H ₉ I ⁺	510
C ₄ H ₅ O ₂ F ₃ ⁺	286	C ₄ H ₉ N ⁺	129
C ₄ H ₅ O ₃ Cl ⁺	384	C ₄ H ₉ NO ⁺	228
C ₄ H ₆ ⁺	54	C ₄ H ₉ NOS ⁺	361
C ₄ H ₆ Co ₂ ⁺	437	C ₄ H ₉ NOSi ⁺	306
C ₄ H ₆ F ₂ Si ⁺	308	C ₄ H ₉ NO ₂ ⁺	247
C ₄ H ₆ F ₆ P ₂ ⁺	322	C ₄ H ₉ NS ⁺	344
C ₄ H ₆ Ga ⁺	448	C ₄ H ₉ NS ₂ ⁺	349
C ₄ H ₆ N ⁺	129	C ₄ H ₉ NSiS ⁺	368
C ₄ H ₆ NF ₆ P ⁺	323	C ₄ H ₉ N ₂ OF ₂ P ⁺	324
C ₄ H ₆ NOSe ⁺	461	C ₄ H ₉ O ⁺	183
C ₄ H ₆ N ₂ ⁺	148	C ₄ H ₉ OSiMn ⁺	424
C ₄ H ₆ N ₂ O ⁺	241	C ₄ H ₉ O ₂ As ⁺	456
C ₄ H ₆ N ₂ O ₂ ⁺	255	C ₄ H ₉ O ₄ PCl ₂ ⁺	398
C ₄ H ₆ N ₂ S ⁺	346	C ₄ H ₉ O ₃ P ⁺	318
C ₄ H ₆ N ₂ S ₂ ⁺	350	C ₄ H ₉ O ₄ PCr ⁺	416
C ₄ H ₆ N ₃ Br ⁺	469	C ₄ H ₉ O ₄ PW ⁺	536
C ₄ H ₆ N ₃ Cl ⁺	380	C ₄ H ₉ O ₄ PCl ⁺	396
C ₄ H ₆ N ₄ ⁺	164	C ₄ H ₉ S ⁺	331
C ₄ H ₆ O ⁺	182	C ₄ H ₉ Si ⁺	295
C ₄ H ₆ O ₂ ⁺	206	C ₄ H ₉ SiCl ⁺	394
C ₄ H ₆ O ₃ S ⁺	357	C ₄ H ₁₀ ⁺	56
C ₄ H ₆ O ₃ ⁺	220	C ₄ H ₁₀ Cd ⁺	495

C ₄ H ₁₀ F ₂ Si ⁺	308	C ₄ O ₃ SMo ⁺	489
C ₄ H ₁₀ Hg ⁺	546	C ₄ O ₃ SW ⁺	538
C ₄ H ₁₀ N ⁺	129	C ₄ O ₃ Cl ₂ Rh ₂ ⁺	492
C ₄ H ₁₀ NF ₂ P ⁺	323	C ₄ O ₄ Cr ⁺	411
C ₄ H ₁₀ NSe ⁺	459	C ₄ O ₄ F ₃ PFe ⁺	434
C ₄ H ₁₀ N ₂ ⁺	148	C ₄ O ₄ F ₆ P ₂ Cr ⁺	418
C ₄ H ₁₀ N ₂ O ⁺	241	C ₄ O ₄ Fe ⁺	430
C ₄ H ₁₀ N ₂ OS ⁺	362	C ₄ O ₄ FeBr ₂ ⁺	480
C ₄ H ₁₀ N ₂ S ₂ ⁺	350	C ₄ O ₄ FeI ₂ ⁺	516
C ₄ H ₁₀ N ₄ ⁺	164	C ₄ O ₄ Mn ⁺	422
C ₄ H ₁₀ O ⁺	183	C ₄ O ₄ Mo ⁺	486
C ₄ H ₁₀ OS ⁺	353	C ₄ O ₄ Ni ⁺	443
C ₄ H ₁₀ O ₂ ⁺	207	C ₄ O ₄ W ⁺	534
C ₄ H ₁₀ O ₂ As ⁺	456	C ₅ SBr ₃ ⁺	477
C ₄ H ₁₀ O ₂ PSCl ⁺	402	C ₅ SI ₄ ⁺	515
C ₄ H ₁₀ O ₂ S ⁺	358	C ₅ S ₈ ⁺	328
C ₄ H ₁₀ O ₂ SiCl ₂ ⁺	395	C ₅ Th ⁺	553
C ₄ H ₁₀ O ₃ ⁺	220	C ₅ Ti ⁺	406
C ₄ H ₁₀ O ₃ PCl ⁺	397	C ₅ U ⁺	554
C ₄ H ₁₀ S ⁺	331	C ₅ F ⁺	270
C ₄ H ₁₀ S ₂ ⁺	337	C ₅ F ₂ ⁺	270
C ₄ H ₁₀ S ₂ Sn ⁺	502	C ₅ F ₃ ⁺	271
C ₄ H ₁₀ Si ⁺	295	C ₅ F ₄ ⁺	272
C ₄ H ₁₀ Zn ⁺	446	C ₅ F ₅ ⁺	272
C ₄ H ₁₁ As ⁺	454	C ₅ F ₁₅ P ₅ ⁺	321
C ₄ H ₁₁ N ⁺	130	C ₅ HN ₃ ⁺	162
C ₄ H ₁₁ NO ⁺	228	C ₅ HOF ₁₁ ⁺	287
C ₄ H ₁₁ NO ₂ ⁺	247	C ₅ HO ₅ Mn ⁺	423
C ₄ H ₁₁ O ₂ As ⁺	456	C ₅ HO ₅ Re ⁺	541
C ₄ H ₁₁ O ₂ PS ₂ ⁺	370	C ₅ H ₂ ⁺	57
C ₄ H ₁₁ O ₃ P ⁺	318	C ₅ H ₂ N ₃ SCl ⁺	401
C ₄ H ₁₁ P ⁺	311	C ₅ H ₂ O ₂ F ₆ ⁺	286
C ₄ H ₁₁ SiCl ⁺	394	C ₅ H ₂ O ₃ Co ₂ ⁺	438
C ₄ H ₁₂ Al ₂ Br ₂ ⁺	476	C ₅ H ₃ ⁺	57
C ₄ H ₁₂ Al ₂ Cl ₂ ⁺	393	C ₅ H ₃ Br ⁺	465
C ₄ H ₁₂ Al ₂ I ₂ ⁺	514	C ₅ H ₃ Cl ⁺	373
C ₄ H ₁₂ As ₂ ⁺	455	C ₅ H ₃ D ₄ ⁺	60
C ₄ H ₁₂ Ge ⁺	449	C ₅ H ₃ I ⁺	510
C ₄ H ₁₂ NP ⁺	315	C ₅ H ₃ NCl ₂ ⁺	381
C ₄ H ₁₂ N ₂ ⁺	148	C ₅ H ₃ NO ⁺	228
C ₄ H ₁₂ N ₂ FP ⁺	322	C ₅ H ₃ NO ₅ Cr ⁺	414
C ₄ H ₁₂ N ₂ F ₃ P ⁺	323	C ₅ H ₃ NO ₉ W ⁺	535
C ₄ H ₁₂ N ₂ OPCl ⁺	398	C ₅ H ₃ NS ⁺	344
C ₄ H ₁₂ N ₂ OS ⁺	362	C ₅ H ₃ NS ₃ ⁺	350
C ₄ H ₁₂ N ₂ PCI ⁺	397	C ₅ H ₃ N ₂ OF ₃ ⁺	288
C ₄ H ₁₂ N ₂ PSCl ⁺	402	C ₅ H ₃ N ₃ O ⁺	245
C ₄ H ₁₂ N ₂ SiCl ₂ ⁺	395	C ₅ H ₃ OS ⁺	353
C ₄ H ₁₂ N ₄ ⁺	164	C ₅ H ₃ O ₂ ⁺	207
C ₄ H ₁₂ N ₆ ⁺	167	C ₅ H ₃ O ₅ GeRe ⁺	542
C ₄ H ₁₂ ORe ⁺	541	C ₅ H ₃ O ₅ MnGe ⁺	454
C ₄ H ₁₂ PAu ⁺	545	C ₅ H ₃ O ₅ PCr ⁺	416
C ₄ H ₁₂ P ₂ ⁺	314	C ₅ H ₃ O ₅ SiMn ⁺	424
C ₄ H ₁₂ Pb ⁺	550	C ₅ H ₃ O ₅ SiRe ⁺	541
C ₄ H ₁₂ SGe ⁺	452	C ₅ H ₄ ⁺	57
C ₄ H ₁₂ SPb ⁺	551	C ₅ H ₄ D ₃ ⁺	60
C ₄ H ₁₂ SSn ⁺	501	C ₅ H ₄ N ⁺	130
C ₄ H ₁₂ Si ⁺	295	C ₅ H ₄ NBr ⁺	468
C ₄ H ₁₂ SiS ⁺	367	C ₅ H ₄ NCl ⁺	378
C ₄ H ₁₂ Sn ⁺	497	C ₅ H ₄ NOBr ⁺	472
C ₄ H ₁₃ NSi ⁺	303	C ₅ H ₄ NOCl ⁺	385
C ₄ H ₁₃ N ₂ O ₃ P ⁺	320	C ₅ H ₄ N ₂ ⁺	149
C ₄ H ₁₄ N ₃ OP ⁺	319	C ₅ H ₄ N ₂ O ₂ ⁺	255
C ₄ I ₂ ⁺	509	C ₅ H ₄ N ₂ O ₃ ⁺	263
C ₄ La ⁺	522	C ₅ H ₄ N ₄ ⁺	164
C ₄ Lu ⁺	531	C ₅ H ₄ N ₄ O ⁺	246
C ₄ NF ₃ ⁺	280	C ₅ H ₄ N ₄ O ₂ ⁺	260
C ₄ N ₂ ⁺	125	C ₅ H ₄ N ₄ O ₃ ⁺	265
C ₄ N ₄ F ₄ ⁺	280	C ₅ H ₄ O ⁺	183
C ₄ O ₂ Cl ₂ ⁺	382	C ₅ H ₄ OS ⁺	353
C ₄ O ₃ SCR ⁺	419	C ₅ H ₄ OSe ⁺	460

C ₅ H ₄ OTe ⁺	507	C ₅ H ₇ ⁺	58
C ₅ H ₄ O ₂ ⁺	207	C ₅ H ₇ N ⁺	130
C ₅ H ₄ O ₂ S ⁺	358	C ₅ H ₇ NOS ⁺	361
C ₅ H ₄ O ₂ Se ⁺	461	C ₅ H ₇ NO ₂ ⁺	247
C ₅ H ₄ O ₂ Te ⁺	507	C ₅ H ₇ NO ₃ ⁺	261
C ₅ H ₄ O ₃ ⁺	220	C ₅ H ₇ NS ₂ ⁺	349
C ₅ H ₄ S ₂ ⁺	337	C ₅ H ₇ N ₃ ⁺	162
C ₅ H ₄ S ₃ ⁺	340	C ₅ H ₇ N ₃ O ⁺	246
C ₅ H ₅ ⁺	57	C ₅ H ₈ ⁺	60
C ₅ H ₅ As ⁺	454	C ₅ H ₈ Br ₂ ⁺	466
C ₅ H ₅ Bi ⁺	552	C ₅ H ₈ ClBr ⁺	479
C ₅ H ₅ Co ⁺	437	C ₅ H ₈ FBr ⁺	474
C ₅ H ₅ Cr ⁺	409	C ₅ H ₈ Ge ⁺	449
C ₅ H ₅ F ₂ P ⁺	321	C ₅ H ₈ N ⁺	130
C ₅ H ₅ F ₃ Si ⁺	308	C ₅ H ₈ NO ⁺	229
C ₅ H ₅ Fe ⁺	429	C ₅ H ₈ NO ₂ Br ⁺	473
C ₅ H ₅ Ge ⁺	449	C ₅ H ₈ NO ₂ Cl ⁺	387
C ₅ H ₅ In ⁺	496	C ₅ H ₈ N ₂ ⁺	149
C ₅ H ₅ La ⁺	522	C ₅ H ₈ N ₂ O ⁺	241
C ₅ H ₅ Mg ⁺	290	C ₅ H ₈ N ₂ S ⁺	346
C ₅ H ₅ Mn ⁺	421	C ₅ H ₈ O ⁺	184
C ₅ H ₅ MnI ⁺	516	C ₅ H ₈ OS ⁺	354
C ₅ H ₅ N ⁺	130	C ₅ H ₈ O ₂ ⁺	207
C ₅ H ₅ NO ⁺	228	C ₅ H ₈ O ₃ ⁺	220
C ₅ H ₅ NOCr ⁺	413	C ₅ H ₈ SCl ₂ ⁺	400
C ₅ H ₅ NONi ⁺	443	C ₅ H ₈ S ₄ ⁺	341
C ₅ H ₅ NO ₂ ⁺	247	C ₅ H ₈ Si ⁺	295
C ₅ H ₅ NS ⁺	344	C ₅ H ₉ ⁺	61
C ₅ H ₅ NSC ⁺	418	C ₅ H ₉ Br ⁺	465
C ₅ H ₅ N ₂ OB ₂ ⁺	472	C ₅ H ₉ FSi ⁺	308
C ₅ H ₅ N ₂ OCl ⁺	387	C ₅ H ₉ I ⁺	510
C ₅ H ₅ N ₂ OF ⁺	287	C ₅ H ₉ N ⁺	130
C ₅ H ₅ N ₅ ⁺	167	C ₅ H ₉ NO ⁺	229
C ₅ H ₅ N ₅ O ⁺	246	C ₅ H ₉ NOSe ⁺	461
C ₅ H ₅ Nd ⁺	525	C ₅ H ₉ NO ₂ ⁺	247
C ₅ H ₅ Ni ⁺	442	C ₅ H ₉ NO ₃ ⁺	261
C ₅ H ₅ OCHg ⁺	547	C ₅ H ₉ NS ⁺	344
C ₅ H ₅ O ₂ F ₃ ⁺	286	C ₅ H ₉ NS ₂ ⁺	349
C ₅ H ₅ P ⁺	311	C ₅ H ₉ N ₃ O ⁺	246
C ₅ H ₅ Pr ⁺	524	C ₅ H ₉ N ₃ S ⁺	348
C ₅ H ₅ Ru ⁺	490	C ₅ H ₉ O ⁺	184
C ₅ H ₅ SCl ⁺	399	C ₅ H ₉ OB ₂ ⁺	470
C ₅ H ₅ SCIHg ⁺	548	C ₅ H ₉ O ₂ ⁺	207
C ₅ H ₅ Sb ⁺	505	C ₅ H ₉ O ₂ F ₃ SiP ₃ Mn ⁺	425
C ₅ H ₅ Si ⁺	295	C ₅ H ₉ O ₂ SiMn ⁺	424
C ₅ H ₅ SiCl ₃ ⁺	395	C ₅ H ₉ O ₃ PCr ⁺	416
C ₅ H ₅ Tl ⁺	548	C ₅ H ₉ O ₃ PW ⁺	536
C ₅ H ₅ W ⁺	532	C ₅ H ₉ SiBr ⁺	476
C ₅ H ₆ ⁺	58	C ₅ H ₉ SiCl ⁺	394
C ₅ H ₆ Cl ₂ ⁺	375	C ₅ H ₉ SiI ⁺	514
C ₅ H ₆ D ⁺	60	C ₅ H ₁₀ ⁺	61
C ₅ H ₆ N ⁺	130	C ₅ H ₁₀ Br ₂ ⁺	467
C ₅ H ₆ N ₂ ⁺	149	C ₅ H ₁₀ Ge ⁺	449
C ₅ H ₆ N ₂ O ⁺	241	C ₅ H ₁₀ N ⁺	130
C ₅ H ₆ N ₂ O ₂ ⁺	255	C ₅ H ₁₀ NBr ⁺	468
C ₅ H ₆ N ₂ O ₃ ⁺	263	C ₅ H ₁₀ NCl ⁺	378
C ₅ H ₆ O ⁺	183	C ₅ H ₁₀ N ₂ ⁺	149
C ₅ H ₆ OCo ₂ ⁺	438	C ₅ H ₁₀ N ₂ O ⁺	241
C ₅ H ₆ OS ⁺	353	C ₅ H ₁₀ N ₂ S ⁺	346
C ₅ H ₆ OSe ⁺	460	C ₅ H ₁₀ N ₄ ⁺	165
C ₅ H ₆ O ₂ ⁺	207	C ₅ H ₁₀ O ⁺	184
C ₅ H ₆ O ₂ S ⁺	358	C ₅ H ₁₀ O ₂ ⁺	208
C ₅ H ₆ O ₃ ⁺	220	C ₅ H ₁₀ S ⁺	332
C ₅ H ₆ S ⁺	331	C ₅ H ₁₀ Si ⁺	295
C ₅ H ₆ SSe ⁺	462	C ₅ H ₁₁ ⁺	62
C ₅ H ₆ STE ⁺	508	C ₅ H ₁₁ As ⁺	454
C ₅ H ₆ S ₂ ⁺	338	C ₅ H ₁₁ Br ⁺	465
C ₅ H ₆ Se ⁺	459	C ₅ H ₁₁ I ⁺	510
C ₅ H ₆ Si ⁺	295	C ₅ H ₁₁ N ⁺	131
C ₅ H ₆ Te ⁺	507	C ₅ H ₁₁ NO ⁺	229

C ₅ H ₁₁ NO ₂ ⁺	247	C ₆ F ₃ Br ₃ ⁺	474
C ₅ H ₁₁ NO ₂ S ⁺	363	C ₆ F ₃ Cl ₃ ⁺	391
C ₅ H ₁₁ NO ₂ Se ⁺	461	C ₆ F ₄ ⁺	272
C ₅ H ₁₁ NS ⁺	344	C ₆ F ₄ Br ₂ ⁺	474
C ₅ H ₁₁ N ₂ OF ₂ P ⁺	324	C ₆ F ₅ ⁺	272
C ₅ H ₁₁ O ⁺	185	C ₆ F ₅ Br ⁺	473
C ₅ H ₁₁ O ₃ P ⁺	319	C ₆ F ₅ Cl ⁺	390
C ₅ H ₁₂ ⁺	62	C ₆ F ₅ I ⁺	514
C ₅ H ₁₂ Hg ⁺	546	C ₆ F ₆ ⁺	273
C ₅ H ₁₂ N ⁺	131	C ₆ F ₆ N ₂ ⁺	280
C ₅ H ₁₂ N ₂ ⁺	149	C ₆ F ₁₂ ⁺	273
C ₅ H ₁₂ N ₂ O ⁺	241	C ₆ F ₁₂ P ₂ ⁺	321
C ₅ H ₁₂ N ₂ OPF ₂ ⁺	324	C ₆ F ₁₅ P ₃ ⁺	321
C ₅ H ₁₂ N ₂ S ⁺	346	C ₆ HCl ₅ ⁺	377
C ₅ H ₁₂ O ⁺	185	C ₆ HFCl ₄ ⁺	392
C ₅ H ₁₂ OGe ⁺	451	C ₆ HF ₃ Cl ₂ ⁺	392
C ₅ H ₁₂ OS ⁺	354	C ₆ HF ₄ Br ⁺	475
C ₅ H ₁₂ OSi ⁺	305	C ₆ HF ₄ Cl ⁺	392
C ₅ H ₁₂ O ₂ Si ⁺	306	C ₆ HF ₅ ⁺	278
C ₅ H ₁₂ S ⁺	332	C ₆ HOF ₅ ⁺	286
C ₅ H ₁₂ S ₂ ⁺	338	C ₆ HS ₂ Br ₃ ⁺	478
C ₅ H ₁₂ S ₂ Sn ⁺	502	C ₆ H ₂ ⁺	62
C ₅ H ₁₂ S ₄ ⁺	341	C ₆ H ₂ Cl ₂ ⁺	375
C ₅ H ₁₂ Si ⁺	295	C ₆ H ₂ Cl ₄ ⁺	377
C ₅ H ₁₂ Sn ⁺	498	C ₆ H ₂ FCl ₃ ⁺	392
C ₅ H ₁₃ N ⁺	131	C ₆ H ₂ F ₂ Br ₂ ⁺	475
C ₅ H ₁₃ NBr ₂ ⁺	469	C ₆ H ₂ F ₂ Cl ₂ ⁺	392
C ₅ H ₁₃ NO ⁺	229	C ₆ H ₂ F ₃ Br ⁺	475
C ₅ H ₁₃ NO ₂ ⁺	248	C ₆ H ₂ F ₃ Cl ⁺	391
C ₅ H ₁₃ N ₂ OP ⁺	319	C ₆ H ₂ F ₄ ⁺	278
C ₅ H ₁₄ N ₂ ⁺	150	C ₆ H ₂ NF ₅ ⁺	283
C ₅ H ₁₄ Si ⁺	296	C ₆ H ₂ OF ₄ ⁺	286
C ₅ H ₁₄ Sn ⁺	498	C ₆ H ₂ O ₂ Cl ₄ ⁺	385
C ₅ H ₁₅ NSi ⁺	303	C ₆ H ₂ O ₄ Cl ₂ Fe ⁺	436
C ₅ H ₁₅ PS ₂ Sn ⁺	502	C ₆ H ₂ O ₄ Co ₂ ⁺	438
C ₅ H ₁₅ P ₅ ⁺	315	C ₆ H ₂ O ₄ FeBr ₂ ⁺	480
C ₅ H ₁₅ Ta ⁺	532	C ₆ H ₂ S ₂ Br ₂ ⁺	478
C ₅ H ₁₆ N ₃ OP ⁺	320	C ₆ H ₃ Br ₃ ⁺	467
C ₅ NCl ₃ ⁺	378	C ₆ H ₃ Cl ₃ ⁺	377
C ₅ NF ₅ ⁺	280	C ₆ H ₃ D ₂ ⁺	64
C ₅ NOCl ₃ ⁺	385	C ₆ H ₃ FB ₂ ⁺	475
C ₅ N ₄ ⁺	125	C ₆ H ₃ FCI ₂ ⁺	392
C ₅ OF ₆ Co ₂ ⁺	440	C ₆ H ₃ F ₂ Br ⁺	475
C ₅ O ₄ ScR ⁺	419	C ₆ H ₃ F ₂ Cl ⁺	391
C ₅ O ₄ SMo ⁺	489	C ₆ H ₃ F ₃ ⁺	277
C ₅ O ₄ SW ⁺	538	C ₆ H ₃ Ge ⁺	449
C ₅ O ₅ BrRe ⁺	542	C ₆ H ₃ NO ₄ MnBr ⁺	480
C ₅ O ₅ ClMn ⁺	428	C ₆ H ₃ N ₃ ⁺	162
C ₅ O ₅ ClRe ⁺	542	C ₆ H ₃ N ₃ O ₆ ⁺	267
C ₅ O ₅ Cr ⁺	411	C ₆ H ₃ N ₅ ⁺	167
C ₅ O ₅ F ₃ PCr ⁺	418	C ₆ H ₃ OF ₃ ⁺	286
C ₅ O ₅ F ₃ PMo ⁺	489	C ₆ H ₃ OF ₅ ⁺	367
C ₅ O ₅ F ₃ PW ⁺	538	C ₆ H ₃ OF ₃ Se ⁺	461
C ₅ O ₅ Fe ⁺	430	C ₆ H ₃ O ₂ F ₃ ⁺	286
C ₅ O ₅ IrRe ⁺	542	C ₆ H ₃ O ₅ Mn ⁺	423
C ₅ O ₅ MnBr ⁺	480	C ₆ H ₃ O ₅ Re ⁺	541
C ₅ O ₅ MnI ⁺	516	C ₆ H ₃ S ₂ Br ⁺	478
C ₅ O ₅ Mo ⁺	486	C ₆ H ₃ S ₂ I ⁺	515
C ₅ O ₅ PBr ₃ Mo ⁺	490	C ₆ H ₃ Si ⁺	296
C ₅ O ₅ PCl ₃ Cr ⁺	420	C ₆ H ₄ ⁺	62
C ₅ O ₅ PCl ₃ Mo ⁺	489	C ₆ H ₄ Br ⁺	465
C ₅ O ₅ PCl ₃ W ⁺	538	C ₆ H ₄ BrI ⁺	517
C ₅ O ₅ PCrBr ₃ ⁺	480	C ₆ H ₄ Br ₂ ⁺	467
C ₅ O ₅ SiCl ₃ Mn ⁺	428	C ₆ H ₄ Cl ⁺	373
C ₅ O ₅ W ⁺	534	C ₆ H ₄ Cl ₂ ⁺	376
C ₅ S ₁₀ ⁺	328	C ₆ H ₄ D ₂ ⁺	65
C ₆ Cl ₄ ⁺	371	C ₆ H ₄ F ⁺	275
C ₆ Cl ₆ ⁺	371	C ₆ H ₄ FBr ⁺	474
C ₆ D ₁₂ ⁺	69	C ₆ H ₄ FCl ⁺	391
C ₆ D ₁₂ O ₂ ⁺	210	C ₆ H ₄ F ₂ ⁺	277

C ₆ H ₁₈ Si ₂ ⁺	299	C ₇ H ₅ N ₂ O ₃ Cl ⁺	388
C ₆ H ₁₈ Si ₂ S ⁺	368	C ₇ H ₅ N ₃ O ₆ ⁺	267
C ₆ H ₁₈ Sn ₂ ⁺	500	C ₇ H ₅ O ⁺	187
C ₆ H ₁₈ W ⁺	532	C ₇ H ₅ OCl ⁺	383
C ₆ H ₁₉ NSi ₂ ⁺	304	C ₇ H ₅ OClCr ⁺	420
C ₆ N ₂ ⁺	125	C ₇ H ₅ OSF ₆ ⁺	435
C ₆ N ₄ ⁺	125	C ₇ H ₅ OSMn ⁺	426
C ₆ O ₂ Cl ₄ ⁺	382	C ₇ H ₅ O ₂ ⁺	210
C ₆ O ₂ F ₄ ⁺	284	C ₇ H ₅ O ₂ Br ⁺	471
C ₆ O ₂ F ₆ Co ₂ ⁺	440	C ₇ H ₅ O ₂ ClFe ⁺	436
C ₆ O ₅ CrSe ⁺	462	C ₇ H ₅ O ₂ F ⁺	285
C ₆ O ₅ F ₃ Mn ⁺	423	C ₇ H ₅ O ₂ F ₃ PMn ⁺	425
C ₆ O ₅ SCr ⁺	419	C ₇ H ₅ O ₂ FeBr ⁺	480
C ₆ O ₅ SMo ⁺	489	C ₇ H ₅ O ₂ FeI ⁺	516
C ₆ O ₅ SW ⁺	538	C ₇ H ₅ O ₂ Mn ⁺	422
C ₆ O ₆ Cr ⁺	411	C ₇ H ₅ O ₂ PCl ₃ Mn ⁺	428
C ₆ O ₆ Mo ⁺	486	C ₇ H ₅ O ₂ PMnBr ₃ ⁺	480
C ₆ O ₆ S ₂ Fe ₂ ⁺	435	C ₇ H ₅ O ₂ W ⁺	534
C ₆ O ₆ W ⁺	534	C ₇ H ₅ S ₂ Fe ₂ ⁺	435
C ₆ S ₂ Br ₄ ⁺	477	C ₇ H ₅ S ₂ Mn ⁺	426
C ₇ F ₈ ⁺	273	C ₇ H ₅ S ₂ Mn ₂ ⁺	426
C ₇ HO ₄ F ₆ Ir ⁺	543	C ₇ H ₆ ⁺	70
C ₇ HO ₄ F ₆ Rh ⁺	491	C ₇ H ₆ Cl ⁺	373
C ₇ H ₂ O ₃ Co ₂ ⁺	439	C ₇ H ₆ Cl ₂ ⁺	376
C ₇ H ₃ F ₅ ⁺	278	C ₇ H ₆ F ⁺	275
C ₇ H ₃ NO ₂ S ⁺	363	C ₇ H ₆ NO ⁺	231
C ₇ H ₃ NO ₃ ⁺	262	C ₇ H ₆ NOCl ⁺	386
C ₇ H ₃ NO ₄ Fe ⁺	433	C ₇ H ₆ NOF ⁺	287
C ₇ H ₃ NO ₅ Cr ⁺	414	C ₇ H ₆ NO ₂ ⁺	248
C ₇ H ₃ OF ₅ ⁺	286	C ₇ H ₆ N ₂ ⁺	152
C ₇ H ₃ O ₆ Fe ⁺	432	C ₇ H ₆ N ₂ O ₃ ⁺	264
C ₇ H ₄ F ₃ Br ⁺	475	C ₇ H ₆ N ₂ O ₅ ⁺	267
C ₇ H ₄ F ₃ Cl ⁺	391	C ₇ H ₆ O ⁺	188
C ₇ H ₄ F ₄ ⁺	278	C ₇ H ₆ OCr ⁺	411
C ₇ H ₄ N ⁺	133	C ₇ H ₆ O ₂ ⁺	210
C ₇ H ₄ NF ⁺	281	C ₇ H ₆ O ₃ ⁺	220
C ₇ H ₄ NI ⁺	512	C ₇ H ₆ O ₄ Co ₂ ⁺	438
C ₇ H ₄ NO ⁺	230	C ₇ H ₆ O ₄ Fe ⁺	431
C ₇ H ₄ NOCl ⁺	385	C ₇ H ₆ O ₅ ClMnSn ⁺	503
C ₇ H ₄ NO ₂ ⁺	248	C ₇ H ₆ O ₅ SCr ⁺	419
C ₇ H ₄ NO ₃ ⁺	262	C ₇ H ₆ O ₆ PCr ⁺	417
C ₇ H ₄ N ₂ O ₂ ⁺	256	C ₇ H ₆ O ₆ PW ⁺	537
C ₇ H ₄ N ₂ O ₃ ⁺	263	C ₇ H ₆ S ₂ ⁺	338
C ₇ H ₄ OB _r ⁺	471	C ₇ H ₆ Si ⁺	296
C ₇ H ₄ OCl ⁺	383	C ₇ H ₇ ⁺	70
C ₇ H ₄ OF ⁺	285	C ₇ H ₇ Br ⁺	465
C ₇ H ₄ O ₃ Fe ⁺	431	C ₇ H ₇ Cl ⁺	373
C ₇ H ₄ O ₃ PMn ⁺	424	C ₇ H ₇ ClHg ⁺	547
C ₇ H ₄ O ₄ F ₃ Rh ⁺	491	C ₇ H ₇ F ⁺	275
C ₇ H ₄ O ₅ Fe ⁺	432	C ₇ H ₇ FSiBr ⁺	476
C ₇ H ₄ O ₆ SCr ⁺	419	C ₇ H ₇ FSiCl ⁺	396
C ₇ H ₄ S ₂ Mn ⁺	426	C ₇ H ₇ I ⁺	511
C ₇ H ₅ S ₂ Mn ₂ ⁺	426	C ₇ H ₇ NO ⁺	231
C ₇ H ₅ S ₃ ⁺	340	C ₇ H ₇ NOCr ⁺	413
C ₇ H ₅ D ₂ ⁺	72	C ₇ H ₇ NOS ⁺	361
C ₇ H ₅ F ₃ ⁺	277	C ₇ H ₇ NOSMn ⁺	427
C ₇ H ₅ N ⁺	133	C ₇ H ₇ NOSMnI ⁺	516
C ₇ H ₅ NO ⁺	231	C ₇ H ₇ NO ₂ ⁺	249
C ₇ H ₅ NOS ⁺	361	C ₇ H ₇ NO ₂ FSi ⁺	309
C ₇ H ₅ NOS ₂ Mn ⁺	428	C ₇ H ₇ NO ₂ S ⁺	363
C ₇ H ₅ NOS ₂ Mn ₂ ⁺	428	C ₇ H ₇ NO ₃ ⁺	262
C ₇ H ₅ NO ₂ S ⁺	363	C ₇ H ₇ NO ₅ W ⁺	535
C ₇ H ₅ NO ₂ SCr ⁺	419	C ₇ H ₇ N ₂ FS ⁺	366
C ₇ H ₅ NO ₃ Cr ⁺	414	C ₇ H ₇ N ₂ O ⁺	242
C ₇ H ₅ NO ₄ ⁺	265	C ₇ H ₇ N ₂ OB _r ⁺	473
C ₇ H ₅ NO ₅ Cr ⁺	414	C ₇ H ₇ N ₂ OCl ⁺	387
C ₇ H ₅ NS ⁺	344	C ₇ H ₇ N ₂ OF ⁺	287
C ₇ H ₅ NS ₂ ⁺	349	C ₇ H ₇ N ₂ OI ⁺	513
C ₇ H ₅ N ₂ Cl ⁺	379	C ₇ H ₇ N ₂ S ⁺	347
C ₇ H ₅ N ₂ O ₃ ⁺	263	C ₇ H ₇ N ₂ SBr ⁺	478

C ₇ H ₇ N ₂ SCl ⁺	400	C ₇ H ₁₀ SSe ⁺	462
C ₇ H ₇ N ₂ Si ⁺	515	C ₇ H ₁₁ S ₂ ⁺	338
C ₇ H ₇ N ₃ O ₂ S ⁺	364	C ₇ H ₁₁ S ₃ ⁺	340
C ₇ H ₇ O ⁺	189	C ₇ H ₁₁ ⁺	74
C ₇ H ₇ OB _{Br} ⁺	471	C ₇ H ₁₁ N ⁺	135
C ₇ H ₇ OCl ⁺	383	C ₇ H ₁₁ NO ⁺	231
C ₇ H ₇ OF ⁺	285	C ₇ H ₁₁ NO ₂ ⁺	249
C ₇ H ₇ OI ⁺	513	C ₇ H ₁₁ N ₃ O ⁺	246
C ₇ H ₇ OMn ⁺	422	C ₇ H ₁₁ OCl ⁺	384
C ₇ H ₇ O ₂ ⁺	210	C ₇ H ₁₁ O ₂ Br ⁺	471
C ₇ H ₇ O ₃ ⁺	223	C ₇ H ₁₁ P ⁺	311
C ₇ H ₇ O ₄ Ir ⁺	543	C ₇ H ₁₂ ⁺	74
C ₇ H ₇ O ₄ Rh ⁺	491	C ₇ H ₁₂ NBr ⁺	468
C ₇ H ₇ SBr ⁺	478	C ₇ H ₁₂ NCl ⁺	379
C ₇ H ₇ SMn ⁺	425	C ₇ H ₁₂ NI ⁺	512
C ₇ H ₇ SMnI ⁺	516	C ₇ H ₁₂ NO ₂ ⁺	249
C ₇ H ₈ ⁺	72	C ₇ H ₁₂ N ₂ ⁺	152
C ₇ H ₈ ⁺²	73	C ₇ H ₁₂ N ₂ O ⁺	242
C ₇ H ₈ Cr ⁺	409	C ₇ H ₁₂ N ₂ O ₂ ⁺	256
C ₇ H ₈ FSi ⁺	308	C ₇ H ₁₂ Si ⁺	191
C ₇ H ₈ N ⁺	134	C ₇ H ₁₂ O ₂ ⁺	211
C ₇ H ₈ N ₂ ⁺	152	C ₇ H ₁₂ O ₄ ⁺	223
C ₇ H ₈ N ₂ O ⁺	242	C ₇ H ₁₂ S ⁺	333
C ₇ H ₈ N ₂ OS ⁺	362	C ₇ H ₁₂ S ₄ ⁺	342
C ₇ H ₈ N ₂ O ₂ ⁺	256	C ₇ H ₁₃ ⁺	74
C ₇ H ₈ N ₂ S ⁺	347	C ₇ H ₁₃ N ⁺	135
C ₇ H ₈ O ⁺	189	C ₇ H ₁₃ NO ⁺	232
C ₇ H ₈ OCr ⁺	411	C ₇ H ₁₃ OS ⁺	355
C ₇ H ₈ OS ⁺	355	C ₇ H ₁₃ O ₂ ⁺	211
C ₇ H ₈ OS ₂ ⁺	360	C ₇ H ₁₄ ⁺	74
C ₇ H ₈ O ₂ ⁺	210	C ₇ H ₁₄ N ₂ ⁺	153
C ₇ H ₈ O ₂ PMn ⁺	424	C ₇ H ₁₄ N ₂ O ⁺	242
C ₇ H ₈ O ₃ S ⁺	358	C ₇ H ₁₄ O ⁺	191
C ₇ H ₈ S ⁺	333	C ₇ H ₁₄ OS ⁺	355
C ₇ H ₈ S ₃ ⁺	340	C ₇ H ₁₄ O ₂ ⁺	211
C ₇ H ₈ SiCl ⁺	394	C ₇ H ₁₅ ⁺	75
C ₇ H ₈ Te ⁺	507	C ₇ H ₁₅ N ⁺	135
C ₇ H ₉ ⁺	73	C ₇ H ₁₅ NO ⁺	232
C ₇ H ₉ Br ⁺	465	C ₇ H ₁₅ NO ₃ Si ⁺	307
C ₇ H ₉ N ⁺	134	C ₇ H ₁₅ O ₂ P ⁺	318
C ₇ H ₉ NO ⁺	231	C ₇ H ₁₅ O ₄ PCr ⁺	416
C ₇ H ₉ NOS ⁺	361	C ₇ H ₁₅ O ₄ PW ⁺	536
C ₇ H ₉ NOSe ⁺	461	C ₇ H ₁₅ P ⁺	311
C ₇ H ₉ NOTe ⁺	507	C ₇ H ₁₆ Hg ⁺	546
C ₇ H ₉ NO ₂ ⁺	249	C ₇ H ₁₆ N ₂ ⁺	153
C ₇ H ₉ NS ⁺	345	C ₇ H ₁₆ N ₃ OF ₂ P ⁺	324
C ₇ H ₉ NSe ⁺	460	C ₇ H ₁₆ N ₄ ⁺	165
C ₇ H ₉ N ₃ S ⁺	348	C ₇ H ₁₆ S ₂ Sn ⁺	502
C ₇ H ₉ N ₅ ⁺	167	C ₇ H ₁₆ Sn ⁺	498
C ₇ H ₉ O ₄ CoSn ⁺	503	C ₇ H ₁₇ NO ⁺	232
C ₇ H ₉ O ₄ F ₃ SiPMn ⁺	425	C ₇ H ₁₇ OPS ⁺	369
C ₇ H ₉ O ₄ PF ₆ ⁺	433	C ₇ H ₁₈ Ge ⁺	450
C ₇ H ₉ O ₄ SiMn ⁺	424	C ₇ H ₁₈ NP ⁺	315
C ₇ H ₉ O ₇ PCr ⁺	417	C ₇ H ₁₈ N ₂ ⁺	153
C ₇ H ₉ O ₇ PW ⁺	537	C ₇ H ₁₈ N ₂ Si ⁺	303
C ₇ H ₉ Si ⁺	296	C ₇ H ₁₈ N ₃ OPCr ⁺	418
C ₇ H ₁₀ ⁺	73	C ₇ H ₁₈ N ₃ OPFe ⁺	433
C ₇ H ₁₀ F ₆ Si ⁺	308	C ₇ H ₁₈ N ₃ OPMo ⁺	488
C ₇ H ₁₀ N ⁺	135	C ₇ H ₁₈ Pb ⁺	550
C ₇ H ₁₀ NO ⁺	231	C ₇ H ₁₈ Sn ⁺	498
C ₇ H ₁₀ NO ₂ ⁺	249	C ₇ H ₁₉ NSi ⁺	303
C ₇ H ₁₀ N ₂ ⁺	152	C ₇ H ₁₉ O ₂ Si ₂ ⁺	306
C ₇ H ₁₀ N ₂ O ⁺	242	C ₇ H ₁₉ SiAs ⁺	456
C ₇ H ₁₀ N ₂ O ₂ ⁺	256	C ₇ H ₁₉ SiP ⁺	325
C ₇ H ₁₀ O ⁺	190	C ₇ H ₂₀ Si ₁₂ ⁺	299
C ₇ H ₁₀ OS ⁺	355	C ₇ H ₂₁ P ₂ ClPt ⁺	543
C ₇ H ₁₀ OSe ⁺	460	C ₇ H ₂₁ P ₂ IPt ⁺	544
C ₇ H ₁₀ O ₂ ⁺	211	C ₇ NF ₅ ⁺	280
C ₇ H ₁₀ S ⁺	333	C ₇ O ₃ F ₆ Co ₂ ⁺	440

C ₇ O ₆ F ₃ Re ⁺	541	C ₈ H ₇ D ₂ NO ₂ ⁺	250
C ₈ F ₁₀ ⁺	273	C ₈ H ₇ N ⁺	135
C ₈ F ₂₀ P ⁺	321	C ₈ H ₇ NO ⁺	232
C ₈ HO ₂ F ₁₇	287	C ₈ H ₇ NOBr ⁺	472
C ₈ H ₂ ⁺	75	C ₈ H ₇ NOBr ₂ ⁺	473
C ₈ H ₂ N ₂ F ₄ ⁺	283	C ₈ H ₇ NOCl ⁺	386
C ₈ H ₂ O ₂ Co ₂ ⁺	439	C ₈ H ₇ NOCl ₂ ⁺	388
C ₈ H ₃ F ₅ ⁺	278	C ₈ H ₇ NOF ₂ ⁺	288
C ₈ H ₃ NF ₄ ⁺	282	C ₈ H ₇ NOS ⁺	361
C ₈ H ₃ NO ₂ Sc ⁺	419	C ₈ H ₇ NO ₂ ⁺	250
C ₈ H ₃ NO ₆ Cr ⁺	415	C ₈ H ₇ NO ₂ Cr ⁺	414
C ₈ H ₄ Ge ⁺	450	C ₈ H ₇ NO ₂ S ⁺	363
C ₈ H ₄ NO ⁺	232	C ₈ H ₇ NO ₃ ⁺	262
C ₈ H ₄ N ₂ ⁺	153	C ₈ H ₇ NO ₄ ⁺	265
C ₈ H ₄ N ₂ F ₂ ⁺	282	C ₈ H ₇ NS ⁺	345
C ₈ H ₄ N ₂ O ₅ Cr ⁺	415	C ₈ H ₇ NS ₂ ⁺	349
C ₈ H ₄ O ⁺	191	C ₈ H ₇ O ⁺	191
C ₈ H ₄ O ₂ ⁺	211	C ₈ H ₇ OBr ⁺	471
C ₈ H ₄ O ₅ S ⁺	358	C ₈ H ₇ OCl ⁺	384
C ₈ H ₄ O ₅ S ₃ ⁺	360	C ₈ H ₇ OSMn ⁺	426
C ₈ H ₄ O ₃ ⁺	221	C ₈ H ₇ O ₂ ⁺	212
C ₈ H ₄ S ₃ ⁺	340	C ₈ H ₇ O ₂ Br ⁺	471
C ₈ H ₄ S ₃ Br ₂ ⁺	478	C ₈ H ₇ O ₂ Cl ⁺	384
C ₈ H ₄ Si ⁺	296	C ₈ H ₇ O ₂ I ⁺	285
C ₈ H ₅ Br ⁺	466	C ₈ H ₇ O ₂ Mn ⁺	513
C ₈ H ₅ Cl ⁺	374	C ₈ H ₇ O ⁺	422
C ₈ H ₅ I ⁺	511	C ₈ H ₈ ⁺	75
C ₈ H ₅ NO ₂ ⁺	249	C ₈ H ₈ Cl ₂ ⁺	376
C ₈ H ₅ N ₂ F ⁺	281	C ₈ H ₈ La ⁺	522
C ₈ H ₅ N ₂ OCl ⁺	387	C ₈ H ₈ NFS ⁺	366
C ₈ H ₅ N ₃ O ₃ ⁺	265	C ₈ H ₈ NO ⁺	232
C ₈ H ₅ O ₂ ClCr ⁺	420	C ₈ H ₈ NOBr ⁺	472
C ₈ H ₅ O ₂ SMn ⁺	427	C ₈ H ₈ NOCl ⁺	386
C ₈ H ₅ O ₃ ⁺	221	C ₈ H ₈ NOF ⁺	287
C ₈ H ₅ O ₃ Mn ⁺	422	C ₈ H ₈ NOI ⁺	513
C ₈ H ₅ O ₅ Re ⁺	541	C ₈ H ₈ NO ₂ ⁺	250
C ₈ H ₅ O ₃ W ⁺	534	C ₈ H ₈ NO ₂ Cl ⁺	388
C ₈ H ₆ ⁺	75	C ₈ H ₈ NS ⁺	345
C ₈ H ₆ Cl ⁺	376	C ₈ H ₈ NSBr ⁺	478
C ₈ H ₆ D ₃ O ⁺	192	C ₈ H ₈ NSCl ⁺	400
C ₈ H ₆ N ⁺	135	C ₈ H ₈ NSI ⁺	515
C ₈ H ₆ NOF ₃ ⁺	288	C ₈ H ₈ N ₂ ⁺	154
C ₈ H ₆ NSBr ⁺	478	C ₈ H ₈ N ₂ O ₃ Cl ₂ ⁺	388
C ₈ H ₆ NSCl ⁺	400	C ₈ H ₈ N ₂ O ₃ ⁺	264
C ₈ H ₆ N ₂ ⁺	154	C ₈ H ₈ N ₂ S ⁺	347
C ₈ H ₆ N ₂ O ⁺	242	C ₈ H ₈ Nd ⁺	525
C ₈ H ₆ N ₂ O ₂ ⁺	256	C ₈ H ₈ Ni ⁺	442
C ₈ H ₆ N ₂ O ₂ S ⁺	364	C ₈ H ₈ O ⁺	191
C ₈ H ₆ N ₂ O ₄ Cl ₂ ⁺	389	C ₈ H ₈ OCr ⁺	411
C ₈ H ₆ N ₂ S ₂ ⁺	350	C ₈ H ₈ O ₂ ⁺	212
C ₈ H ₆ O ⁺	191	C ₈ H ₈ O ₂ Cr ⁺	411
C ₈ H ₆ O ₂ Br ₂ ⁺	472	C ₈ H ₈ O ₂ Fe ⁺	431
C ₈ H ₆ O ₂ Cl ₂ ⁺	385	C ₈ H ₈ O ₃ ⁺	221
C ₈ H ₆ O ₂ Cr ⁺	411	C ₈ H ₈ O ₃ Fe ⁺	431
C ₈ H ₆ O ₂ F ₂ ⁺	285	C ₈ H ₈ Pr ⁺	524
C ₈ H ₆ O ₂ Hg ⁺	547	C ₈ H ₈ Ru ⁺	490
C ₈ H ₆ O ₂ I ₂ ⁺	513	C ₈ H ₈ S ⁺	333
C ₈ H ₆ O ₄ ⁺	223	C ₈ H ₈ S ₂ ⁺	338
C ₈ H ₆ O ₄ Co ₂ ⁺	439	C ₈ H ₈ W ₂ ⁺	533
C ₈ H ₆ O ₅ Fe ⁺	432	C ₈ H ₉ ⁺	76
C ₈ H ₆ O ₅ Sc ⁺	419	C ₈ H ₉ Cl ⁺	374
C ₈ H ₆ O ₆ Cr ⁺	413	C ₈ H ₉ N ⁺	136
C ₈ H ₆ O ₆ Fe ⁺	432	C ₈ H ₉ NO ⁺	233
C ₈ H ₆ S ⁺	333	C ₈ H ₉ NOS ⁺	361
C ₈ H ₆ S ₂ ⁺	338	C ₈ H ₉ NO ₂ ⁺	250
C ₈ H ₆ S ₂ Hg ⁺	547	C ₈ H ₉ NO ₂ S ⁺	363
C ₈ H ₆ S ₃ ⁺	341	C ₈ H ₉ NO ₅ Cr ⁺	414
C ₈ H ₆ Se ⁺	459	C ₈ H ₉ NO ₅ W ⁺	535
C ₈ H ₆ Te ⁺	507	C ₈ H ₉ NS ⁺	345
C ₈ H ₇ Cl ⁺	374	C ₈ H ₉ N ₂ O ⁺	243

C ₈ H ₉ N ₂ OCl ⁺	387	C ₈ H ₁₂ O ⁺	193
C ₈ H ₉ N ₂ S ⁺	347	C ₈ H ₁₂ OS ⁺	355
C ₈ H ₉ N ₂ SCl ⁺	400	C ₈ H ₁₂ O ₂ ⁺	212
C ₈ H ₉ O ⁺	192	C ₈ H ₁₂ O ₈ CrMo ⁺	489
C ₈ H ₉ OBr ⁺	471	C ₈ H ₁₂ O ₈ Cr ₂ ⁺	413
C ₈ H ₉ OCl ⁺	384	C ₈ H ₁₂ O ₈ Mo ₂ ⁺	487
C ₈ H ₉ OF ⁺	285	C ₈ H ₁₂ O ²⁺	334
C ₈ H ₉ O ₅ MnSn ⁺	503	C ₈ H ₁₂ S ₂ ⁺	339
C ₈ H ₉ O ₅ PCr ⁺	416	C ₈ H ₁₂ Si ⁺	297
C ₈ H ₉ O ₅ PMo ⁺	488	C ₈ H ₁₃ ⁺	79
C ₈ H ₉ O ₅ PW ⁺	536	C ₈ H ₁₃ N ⁺	137
C ₈ H ₉ O ₅ SiMn ⁺	424	C ₈ H ₁₃ NO ⁺	233
C ₈ H ₉ O ₅ SnRe ⁺	542	C ₈ H ₁₃ NOGe ⁺	451
C ₈ H ₉ O ₈ PCr ⁺	417	C ₈ H ₁₃ NOSi ⁺	307
C ₈ H ₉ O ₈ PMo ⁺	488	C ₈ H ₁₃ NOSn ⁺	501
C ₈ H ₉ O ₈ PW ⁺	537	C ₈ H ₁₃ NO ₂ ⁺	250
C ₈ H ₁₀ ⁺	77	C ₈ H ₁₃ Nsi ⁺	303
C ₈ H ₁₀ Cr ⁺	409	C ₈ H ₁₃ OSMn ⁺	426
C ₈ H ₁₀ FSi ⁺	308	C ₈ H ₁₃ P ⁺	312
C ₈ H ₁₀ FSiBr ⁺	476	C ₈ H ₁₄ ⁺	80
C ₈ H ₁₀ FSiCl ⁺	396	C ₈ H ₁₄ CrGe ⁺	453
C ₈ H ₁₀ N ⁺	136	C ₈ H ₁₄ N ⁺	137
C ₈ H ₁₀ NCl ⁺	379	C ₈ H ₁₄ NBr ⁺	468
C ₈ H ₁₀ NO ₂ FSi ⁺	309	C ₈ H ₁₄ NCl ⁺	379
C ₈ H ₁₀ N ₂ Cl ₂ ⁺	381	C ₈ H ₁₄ N ₂ ⁺	154
C ₈ H ₁₀ N ₂ O ⁺	243	C ₈ H ₁₄ N ₂ O ⁺	243
C ₈ H ₁₀ N ₂ OS ⁺	362	C ₈ H ₁₄ N ₂ O ₂ ⁺	257
C ₈ H ₁₀ N ₂ O ₃ ⁺	256	C ₈ H ₁₄ Ni ⁺	442
C ₈ H ₁₀ N ₂ S ⁺	347	C ₈ H ₁₄ N ₃ ⁺	194
C ₈ H ₁₀ N ₂ S ₂ ⁺	350	C ₈ H ₁₄ O ²⁺	213
C ₈ H ₁₀ N ₄ F ₅ P ₃ ⁺	323	C ₈ H ₁₄ O ₂ S ₂ ⁺	360
C ₈ H ₁₀ N ₄ O ₂ ⁺	261	C ₈ H ₁₄ O ₅ PCl ₃ ⁺	398
C ₈ H ₁₀ O ⁺	192	C ₈ H ₁₄ Pd ⁺	492
C ₈ H ₁₀ OFSi ⁺	309	C ₈ H ₁₄ Pt ⁺	543
C ₈ H ₁₀ OS ⁺	355	C ₈ H ₁₄ S ⁺	334
C ₈ H ₁₀ OSe ⁺	460	C ₈ H ₁₄ Si ⁺	297
C ₈ H ₁₀ O ₂ ⁺	212	C ₈ H ₁₅ N ⁺	138
C ₈ H ₁₀ O ₆ PW ⁺	537	C ₈ H ₁₅ NO ⁺	233
C ₈ H ₁₀ S ⁺	333	C ₈ H ₁₅ NO ₂ Cl ₂ ⁺	388
C ₈ H ₁₀ SSe ⁺	462	C ₈ H ₁₅ N ₃ ⁺	163
C ₈ H ₁₀ S ₂ ⁺	338	C ₈ H ₁₅ O ₅ PCr ⁺	417
C ₈ H ₁₀ Se ₂ ⁺	459	C ₈ H ₁₅ O ₅ PW ⁺	536
C ₈ H ₁₁ ⁺	78	C ₈ H ₁₆ ⁺	80
C ₈ H ₁₁ As ⁺	454	C ₈ H ₁₆ NO ₂ Cl ⁺	388
C ₈ H ₁₁ FSi ⁺	308	C ₈ H ₁₆ N ₂ ⁺	154
C ₈ H ₁₁ F ₃ Si ₂ ⁺	309	C ₈ H ₁₆ N ₂ O ⁺	243
C ₈ H ₁₁ F ₆ As ⁺	456	C ₈ H ₁₆ N ₂ O ₂ ⁺	257
C ₈ H ₁₁ N ⁺	136	C ₈ H ₁₆ N ₄ ⁺	165
C ₈ H ₁₁ NO ⁺	233	C ₈ H ₁₆ O ⁺	194
C ₈ H ₁₁ NOS ⁺	362	C ₈ H ₁₆ OS ⁺	355
C ₈ H ₁₁ NS ₂ ⁺	349	C ₈ H ₁₆ O ₂ ⁺	213
C ₈ H ₁₁ OSi ⁺	305	C ₈ H ₁₆ O ₄ ⁺	223
C ₈ H ₁₁ O ₂ F ₃ ⁺	286	C ₈ H ₁₇ N ⁺	138
C ₈ H ₁₁ O ₃ SMn ⁺	427	C ₈ H ₁₇ NO ⁺	233
C ₈ H ₁₁ P ⁺	311	C ₈ H ₁₇ NO ₄ Si ⁺	307
C ₈ H ₁₁ SGe ⁺	452	C ₈ H ₁₇ FP ⁺	321
C ₈ H ₁₁ SPb ⁺	551	C ₈ H ₁₇ Ge ⁺	450
C ₈ H ₁₁ SSn ⁺	501	C ₈ H ₁₇ Hg ⁺	546
C ₈ H ₁₁ Si ⁺	296	C ₈ H ₁₈ NO ⁺	234
C ₈ H ₁₁ SiS ⁺	367	C ₈ H ₁₈ N ₂ ⁺	155
C ₈ H ₁₁ SiCl ⁺	402	C ₈ H ₁₈ N ₂ O ₂ ⁺	257
C ₈ H ₁₁ Si ₂ Cl ₃ ⁺	395	C ₈ H ₁₈ N ₂ S ⁺	347
C ₈ H ₁₂ ⁺	78	C ₈ H ₁₈ N ₃ O ₂ PFe ⁺	433
C ₈ H ₁₂ N ⁺	137	C ₈ H ₁₈ N ₃ O ₂ PMo ⁺	488
C ₈ H ₁₂ NO ⁺	233	C ₈ H ₁₈ N ₃ P ⁺	316
C ₈ H ₁₂ NO ₂ ⁺	250	C ₈ H ₁₈ N ₄ ⁺	165
C ₈ H ₁₂ N ₂ ⁺	154	C ₈ H ₁₈ N ₄ S ₃ Ni ⁺	444
C ₈ H ₁₂ N ₂ O ⁺	243	C ₈ H ₁₈ O ⁺	194
C ₈ H ₁₂ N ₂ O ₂ S ₄ Fe ⁺	436	C ₈ H ₁₈ OS ⁺	355
C ₈ H ₁₂ N ₄ ⁺	165	C ₈ H ₁₈ O ₂ ⁺	213

C ₈ H ₁₀ O ₂ S ⁺	358	C ₉ H ₉ ⁺	82
C ₈ H ₁₀ PCl ⁺	396	C ₉ H ₈ Cl ₂ ⁺	376
C ₈ H ₁₀ S ⁺	334	C ₉ H ₈ DO ⁺	195
C ₈ H ₁₀ S ₂ ⁺	339	C ₉ H ₈ NO ⁺	234
C ₈ H ₁₀ Si ₂ Br ₂ ⁺	476	C ₉ H ₈ NOF ₃ ⁺	288
C ₈ H ₁₀ Si ₂ Cl ₂ ⁺	395	C ₉ H ₈ N ₂ O ⁺	243
C ₈ H ₁₀ Sn ⁺	498	C ₉ H ₈ N ₂ S ₂ ⁺	350
C ₈ H ₁₀ O ₂ PS ₃ ⁺	370	C ₉ H ₈ O ⁺	194
C ₈ H ₁₀ P ⁺	312	C ₉ H ₈ OS ⁺	356
C ₈ H ₂₀ Ge ⁺	450	C ₉ H ₈ O ₂ ⁺	213
C ₈ H ₂₀ NO ₃ P ⁺	320	C ₉ H ₈ O ₂ Cr ⁺	412
C ₈ H ₂₀ N ₂ ⁺	155	C ₉ H ₈ O ₂ SCr ⁺	419
C ₈ H ₂₀ N ₃ P ⁺	316	C ₉ H ₈ O ₃ Cr ⁺	412
C ₈ H ₂₀ N ₄ ⁺	165	C ₉ H ₈ O ₃ Fe ⁺	431
C ₈ H ₂₀ O ₄ P ₂ S ₄ Ni ⁺	444	C ₉ H ₈ O ₃ PMn ⁺	424
C ₈ H ₂₀ O ₄ P ₂ S ₄ Pd ⁺	492	C ₉ H ₈ O ₃ Ru ⁺	490
C ₈ H ₂₀ O ₄ P ₂ S ₄ Pt ⁺	543	C ₉ H ₈ O ₄ ⁺	223
C ₈ H ₂₀ O ₄ Si ⁺	306	C ₉ H ₈ O ₅ SCr ⁺	419
C ₈ H ₂₀ Si ⁺	297	C ₉ H ₉ ⁺	82
C ₈ H ₂₀ Si ₂ ⁺	299	C ₉ H ₉ Cl ⁺	374
C ₈ H ₂₀ Sn ⁺	498	C ₉ H ₉ F ₃ ⁺	278
C ₈ H ₂₁ NSi ⁺	303	C ₉ H ₉ N ⁺	138
C ₈ H ₂₁ NSi ₂ ⁺	304	C ₉ H ₉ NO ⁺	234
C ₈ H ₂₁ O ₂ Si ₂ ⁺	306	C ₉ H ₉ NOS ⁺	362
C ₈ H ₂₂ Si ⁺	299	C ₉ H ₉ NO ₃ Cr ⁺	414
C ₈ H ₂₂ Si ₂ Cd ⁺	495	C ₉ H ₉ N ₃ Cl ⁺	381
C ₈ H ₂₂ Sn ₂ ⁺	500	C ₉ H ₉ N ₃ F ₂ ⁺	282
C ₈ H ₂₃ NSi ₂ ⁺	304	C ₉ H ₉ N ₃ S ⁺	348
C ₈ H ₂₄ N ₄ Ge ⁺	451	C ₉ H ₉ O ⁺	195
C ₈ H ₂₄ N ₄ Hf ⁺	531	C ₉ H ₉ O ₅ ⁺	223
C ₈ H ₂₄ N ₄ Mo ⁺	486	C ₉ H ₁₀ ⁺	82
C ₈ H ₂₄ N ₄ Si ⁺	304	C ₉ H ₁₀ NO ⁺	234
C ₈ H ₂₄ N ₄ Si ₂ ⁺	304	C ₉ H ₁₀ NOCl ⁺	386
C ₈ H ₂₄ N ₄ Sn ⁺	500	C ₉ H ₁₀ N ₂ Br ⁺	469
C ₈ H ₂₄ N ₄ Ti ⁺	406	C ₉ H ₁₀ N ₃ Cl ⁺	380
C ₈ H ₂₄ N ₄ V ⁺	408	C ₉ H ₁₀ N ₃ F ⁺	281
C ₈ H ₂₄ N ₄ Zr ⁺	484	C ₉ H ₁₀ N ₂ I ⁺	512
C ₈ H ₂₄ P ₂ Pt ⁺	543	C ₉ H ₁₀ N ₂ S ⁺	347
C ₈ H ₂₄ Si ₃ ⁺	301	C ₉ H ₁₀ N ₃ Cl ⁺	380
C ₈ N ₂ F ₃ ⁺	280	C ₉ H ₁₀ N ₃ O ₂ ⁺	260
C ₈ N ₂ F ₆ ⁺	280	C ₉ H ₁₀ N ₃ O ₂ Cl ⁺	388
C ₈ N ₂ O ₂ Cl ₂ ⁺	385	C ₉ H ₁₀ O ⁺	195
C ₈ O ₃ Cl ₄ ⁺	382	C ₉ H ₁₀ OCr ⁺	411
C ₈ O ₄ F ₆ Co ₂ ⁺	440	C ₉ H ₁₀ O ₂ ⁺	213
C ₈ O ₈ F ₁₂ Mo ₂ ⁺	487	C ₉ H ₁₀ O ₃ ⁺	221
C ₉ H ₄ O ₃ ⁺	221	C ₉ H ₁₀ O ₃ Fe ⁺	431
C ₉ H ₅ FS ₃ ⁺	366	C ₉ H ₁₀ O ₇ PCr ⁺	417
C ₉ H ₅ NO ₄ Fe ⁺	433	C ₉ H ₁₀ O ₇ PW ⁺	537
C ₉ H ₅ O ₃ ClCr ⁺	420	C ₉ H ₁₀ S ⁺	334
C ₉ H ₅ S ₃ Cl ⁺	399	C ₉ H ₁₁ Cl ⁺	374
C ₉ H ₆ N ₂ ⁺	155	C ₉ H ₁₁ N ⁺	139
C ₉ H ₆ N ₂ S ⁺	347	C ₉ H ₁₁ NO ⁺	234
C ₉ H ₆ N ₃ F ₅ ⁺	283	C ₉ H ₁₁ NO ₂ ⁺	250
C ₉ H ₆ OS ⁺	355	C ₉ H ₁₁ NO ₃ ⁺	262
C ₉ H ₆ O ₂ ⁺	213	C ₉ H ₁₁ N ₂ ⁺	155
C ₉ H ₆ O ₂ S ⁺	358	C ₉ H ₁₁ N ₂ Br ⁺	469
C ₉ H ₆ O ₃ Cr ⁺	412	C ₉ H ₁₁ N ₃ Cl ⁺	380
C ₉ H ₆ O ₃ S ⁺	359	C ₉ H ₁₁ N ₂ F ⁺	281
C ₉ H ₆ O ₅ Co ₂ ⁺	439	C ₉ H ₁₁ N ₂ I ⁺	512
C ₉ H ₆ S ⁺	341	C ₉ H ₁₁ N ₂ O ⁺	243
C ₉ H ₇ ⁺	81	C ₉ H ₁₁ N ₂ OCl ⁺	387
C ₉ H ₇ N ⁺	138	C ₉ H ₁₁ N ₂ OF ₂ P ⁺	324
C ₉ H ₇ NO ⁺	234	C ₉ H ₁₁ N ₂ S ⁺	347
C ₉ H ₇ NO ₂ ⁺	250	C ₉ H ₁₁ N ₂ SCl ⁺	400
C ₉ H ₇ NO ₃ Cr ⁺	414	C ₉ H ₁₁ N ₃ ⁺	163
C ₉ H ₇ N ₂ O ₃ ⁺	264	C ₉ H ₁₁ N ₃ O ⁺	260
C ₉ H ₇ O ₂ SMn ⁺	427	C ₉ H ₁₁ O ₆ ⁺	224
C ₉ H ₇ O ₃ ⁺	221	C ₉ H ₁₁ S ⁺	334
C ₉ H ₇ O ₃ Mn ⁺	423	C ₉ H ₁₂ ⁺	83
C ₉ H ₇ P ⁺	312	C ₉ H ₁₂ Cr ⁺	410

C ₉ H ₁₂ N ₂ ⁺	156	C ₉ H ₁₈ ⁺	86
C ₉ H ₁₂ N ₂ O ⁺	243	C ₉ H ₁₈ Cl ⁺	379
C ₉ H ₁₂ N ₂ O ₂ ⁺	257	C ₉ H ₁₈ NO ⁺	235
C ₉ H ₁₂ N ₂ S ⁺	347	C ₉ H ₁₈ NO ₂ ⁺	251
C ₉ H ₁₂ O ⁺	195	C ₉ H ₁₈ NO ₂ Cl ⁺	388
C ₉ H ₁₂ OBr ⁺	471	C ₉ H ₁₈ N ₂ ⁺	156
C ₉ H ₁₂ O ₂ ⁺	214	C ₉ H ₁₈ N ₃ O ₃ PCr ⁺	418
C ₉ H ₁₂ O ₂ S ⁺	358	C ₉ H ₁₈ N ₃ O ₃ PFe ⁺	433
C ₉ H ₁₂ O ₃ ⁺	221	C ₉ H ₁₈ N ₃ O ₃ PMo ⁺	488
C ₉ H ₁₂ S ⁺	334	C ₉ H ₁₈ N ₃ S ₆ Fe ⁺	435
C ₉ H ₁₃ ⁺	85	C ₉ H ₁₈ O ⁺	196
C ₉ H ₁₃ As ⁺	454	C ₉ H ₁₈ O ₃ ⁺	221
C ₉ H ₁₃ ClGe ⁺	453	C ₉ H ₁₈ S ₃ ⁺	341
C ₉ H ₁₃ ClSn ⁺	503	C ₉ H ₁₉ N ⁺	140
C ₉ H ₁₃ FSi ⁺	308	C ₉ H ₂₀ Ge ⁺	450
C ₉ H ₁₃ N ⁺	139	C ₉ H ₂₀ Hg ⁺	546
C ₉ H ₁₃ NFSi ⁺	309	C ₉ H ₂₀ N ₂ ⁺	156
C ₉ H ₁₃ NO ⁺	234	C ₉ H ₂₀ Sn ⁺	498
C ₉ H ₁₃ NO ₂ ⁺	251	C ₉ H ₂₁ N ⁺	140
C ₉ H ₁₃ NO ₂ Si ⁺	307	C ₉ H ₂₁ NBr ₂ ⁺	469
C ₉ H ₁₃ NO ₃ Si ⁺	307	C ₉ H ₂₁ NSi ⁺	303
C ₉ H ₁₃ NS ⁺	345	C ₉ H ₂₁ O ₃ P ⁺	319
C ₉ H ₁₃ OFSi ⁺	309	C ₉ H ₂₁ P ⁺	312
C ₉ H ₁₃ OSiBr ⁺	476	C ₉ H ₂₂ Si ⁺	297
C ₉ H ₁₃ OSiCl ⁺	395	C ₉ H ₂₂ Sn ⁺	498
C ₉ H ₁₃ P ⁺	312	C ₉ H ₂₃ O ₂ Si ₂ ⁺	306
C ₉ H ₁₃ SiBr ⁺	476	C ₉ H ₂₄ N ₄ ⁺	165
C ₉ H ₁₃ SiCl ⁺	394	C ₉ H ₂₄ Si ₂ ⁺	299
C ₉ H ₁₄ ⁺	85	C ₉ H ₂₄ Sn ₂ ⁺	500
C ₉ H ₁₄ F ₂ Si ₂ ⁺	309	C ₉ H ₂₅ NSi ₂ ⁺	304
C ₉ H ₁₄ Ge ⁺	450	C ₉ H ₂₅ Si ₂ P ⁺	325
C ₉ H ₁₄ NO ₂ ⁺	251	C ₉ H ₂ -NSi ₃ ⁺	304
C ₉ H ₁₄ NP ⁺	315	C ₉ NF ₇ ⁺	280
C ₉ H ₁₄ NSi ⁺	303	C ₉ O ₃ F ₆ Co ₂ ⁺	440
C ₉ H ₁₄ N ₂ ⁺	156	C ₁₀ Cl ₄ ⁺	372
C ₉ H ₁₄ N ₂ Ge ⁺	451	C ₁₀ F ₈ ⁺	273
C ₉ H ₁₄ N ₂ O ⁺	243	C ₁₀ F ₁₂ S ₄ ⁺	365
C ₉ H ₁₄ N ₂ O ₃ ⁺	264	C ₁₀ H ₂ O ₄ F ₁₂ Cu ⁺	445
C ₉ H ₁₄ N ₂ S ⁺	348	C ₁₀ H ₂ O ₄ F ₁₂ Mg ⁺	290
C ₉ H ₁₄ N ₂ Si ⁺	304	C ₁₀ H ₂ O ₄ F ₁₂ Ni ⁺	444
C ₉ H ₁₄ O ⁺	195	C ₁₀ H ₂ O ₄ F ₁₂ Zn ⁺	447
C ₉ H ₁₄ OCrGe ⁺	453	C ₁₀ H ₂ O ₆ ⁺	224
C ₉ H ₁₄ OS ⁺	356	C ₁₀ H ₂ O ₆ F ₁₂ U ⁺	555
C ₉ H ₁₄ OSi ⁺	305	C ₁₀ H ₄ NO ₅ ClCr ⁺	420
C ₉ H ₁₄ O ₂ ⁺	214	C ₁₀ H ₄ NO ₅ CrBr ⁺	480
C ₉ H ₁₄ Pb ⁺	550	C ₁₀ H ₄ O ₆ Cl ₂ Co ₂ ⁺	441
C ₉ H ₁₄ SGe ⁺	452	C ₁₀ H ₅ NO ₃ Cr ⁺	414
C ₉ H ₁₄ SPb ⁺	551	C ₁₀ H ₅ NO ₅ W ⁺	535
C ₉ H ₁₄ SSn ⁺	501	C ₁₀ H ₅ NO ₆ Cr ⁺	415
C ₉ H ₁₄ Si ⁺	297	C ₁₀ H ₆ ⁺	86
C ₉ H ₁₄ SiS ⁺	367	C ₁₀ H ₆ Br ₂ ⁺	467
C ₉ H ₁₄ Si ₂ Cl ₂ ⁺	395	C ₁₀ H ₆ Cl ₄ ⁺	376
C ₉ H ₁₄ Sn ⁺	498	C ₁₀ H ₆ N ₂ ⁺	156
C ₉ H ₁₅ N ⁺	139	C ₁₀ H ₆ O ₂ ⁺	214
C ₉ H ₁₅ NO ⁺	234	C ₁₀ H ₆ O ₃ ⁺	221
C ₉ H ₁₅ NO ₂ ⁺	251	C ₁₀ H ₆ O ₄ ⁺	223
C ₉ H ₁₅ N ₂ O ₂ ⁺	257	C ₁₀ H ₆ O ₆ Co ₂ ⁺	439
C ₉ H ₁₅ O ₆ PCr ⁺	417	C ₁₀ H ₆ S ₂ ⁺	339
C ₉ H ₁₅ O ₆ PW ⁺	537	C ₁₀ H ₆ N ⁺	140
C ₉ H ₁₆ ⁺	85	C ₁₀ H ₇ NO ₂ ⁺	251
C ₉ H ₁₆ NOCI ⁺	386	C ₁₀ H ₈ O ₃ Mn ⁺	423
C ₉ H ₁₆ NO ₂ ⁺	251	C ₁₀ H ₈ ⁺	86
C ₉ H ₁₆ N ₂ ⁺	156	C ₁₀ H ₈ ⁺²	87
C ₉ H ₁₆ O ⁺	196	C ₁₀ H ₈ ⁺³	87
C ₉ H ₁₇ N ⁺	140	C ₁₀ H ₈ Cl ₂ Fe ⁺	436
C ₉ H ₁₇ NO ⁺	235	C ₁₀ H ₈ F ₆ S ₄ Ni ⁺	444
C ₉ H ₁₇ NO ₂ ⁺	251	C ₁₀ H ₈ N ₂ ⁺	156
C ₉ H ₁₇ NO ₂ Cl ₂ ⁺	388	C ₁₀ H ₈ N ₂ O ₂ ⁺	257
C ₉ H ₁₇ NO ₃ S ⁺	364	C ₁₀ H ₈ O ⁺	196
C ₉ H ₁₇ N ₂ O ₂ ⁺	257	C ₁₀ H ₈ OS ₂ ⁺	360

C ₁₀ H ₈ OS ₃ ⁺	360	C ₁₀ H ₁₂ ⁺	88
C ₁₀ H ₈ O ₂ ⁺	214	C ₁₀ H ₁₂ Mo ⁺	485
C ₁₀ H ₈ O ₂ F ₆ S ₂ Ni ⁺	444	C ₁₀ H ₁₂ NO ⁺	235
C ₁₀ H ₈ O ₃ Cr ⁺	412	C ₁₀ H ₁₂ NOCl ⁺	386
C ₁₀ H ₈ O ₃ Fe ⁺	431	C ₁₀ H ₁₂ N ₂ ⁺	156
C ₁₀ H ₈ O ₃ Mo ⁺	486	C ₁₀ H ₁₂ O ⁺	196
C ₁₀ H ₈ O ₃ W ⁺	534	C ₁₀ H ₁₂ OCr ⁺	411
C ₁₀ H ₈ O ₄ Cr ⁺	413	C ₁₀ H ₁₂ O ₂ ⁺	214
C ₁₀ H ₈ O ₄ F ₆ Co ⁺	440	C ₁₀ H ₁₂ S ₂ ⁺	339
C ₁₀ H ₈ O ₄ F ₆ Cu ⁺	445	C ₁₀ H ₁₂ S ₃ ⁺	341
C ₁₀ H ₈ O ₄ F ₆ Ni ⁺	443	C ₁₀ H ₁₂ S ₄ ⁺	342
C ₁₀ H ₈ O ₄ Fe ⁺	432	C ₁₀ H ₁₂ Se ₄ ⁺	459
C ₁₀ H ₈ S ⁺	334	C ₁₀ H ₁₂ W ⁺	533
C ₁₀ H ₈ S ₃ ⁺	341	C ₁₀ H ₁₃ Cl ⁺	374
C ₁₀ H ₉ ClFe ⁺	436	C ₁₀ H ₁₃ F ⁺	275
C ₁₀ H ₉ N ⁺	140	C ₁₀ H ₁₃ N ⁺	140
C ₁₀ H ₉ NO ⁺	235	C ₁₀ H ₁₃ NO ⁺	235
C ₁₀ H ₉ NO ₆ SFe ₂ ⁺	436	C ₁₀ H ₁₃ NO ₂ ⁺	251
C ₁₀ H ₉ NS ⁺	345	C ₁₀ H ₁₃ N ₂ ⁺	157
C ₁₀ H ₉ O ₃ Mn ⁺	423	C ₁₀ H ₁₃ N ₂ Cl ⁺	380
C ₁₀ H ₉ P ⁺	312	C ₁₀ H ₁₃ N ₂ O ⁺	243
C ₁₀ H ₁₀ ⁺	87	C ₁₀ H ₁₃ N ₂ OCl ⁺	387
C ₁₀ H ₁₀ Br ₂ Ta ⁺	532	C ₁₀ H ₁₃ N ₂ OF ₂ P ⁺	324
C ₁₀ H ₁₀ Br ₂ Zr ⁺	484	C ₁₀ H ₁₃ N ₂ S ⁺	348
C ₁₀ H ₁₀ Cl ₂ Hf ⁺	531	C ₁₀ H ₁₃ N ₂ SCl ⁺	400
C ₁₀ H ₁₀ Cl ₃ Ta ⁺	532	C ₁₀ H ₁₃ N ⁺	163
C ₁₀ H ₁₀ Cl ₃ Ti ⁺	407	C ₁₀ H ₁₃ O ₃ SMn ⁺	427
C ₁₀ H ₁₀ Cl ₃ Zr ⁺	484	C ₁₀ H ₁₃ P ⁺	312
C ₁₀ H ₁₀ Co ⁺	437	C ₁₀ H ₁₃ Ta ⁺	532
C ₁₀ H ₁₀ Cr ⁺	410	C ₁₀ H ₁₄ ⁺	89
C ₁₀ H ₁₀ F ₂ Ti ⁺	407	C ₁₀ H ₁₄ Ge ⁺	450
C ₁₀ H ₁₀ Fe ⁺	429	C ₁₀ H ₁₄ N ⁺	140
C ₁₀ H ₁₀ Hg ⁺	547	C ₁₀ H ₁₄ NOCl ⁺	386
C ₁₀ H ₁₀ La ⁺	522	C ₁₀ H ₁₄ N ₂ ⁺	157
C ₁₀ H ₁₀ Mg ⁺	290	C ₁₀ H ₁₄ N ₂ O ⁺	244
C ₁₀ H ₁₀ Mn ⁺	421	C ₁₀ H ₁₄ N ₂ S ⁺	348
C ₁₀ H ₁₀ NO ⁺	235	C ₁₀ H ₁₄ O ⁺	197
C ₁₀ H ₁₀ NOF ₃ ⁺	288	C ₁₀ H ₁₄ OBr ⁺	471
C ₁₀ H ₁₀ N ₂ ⁺	156	C ₁₀ H ₁₄ OSi ⁺	305
C ₁₀ H ₁₀ N ₂ Cl ₂ ⁺	381	C ₁₀ H ₁₄ O ₂ ⁺	214
C ₁₀ H ₁₀ N ₂ O ₃ ⁺	264	C ₁₀ H ₁₄ O ₂ CrGe ⁺	453
C ₁₀ H ₁₀ Nd ⁺	525	C ₁₀ H ₁₄ O ₂ S ₂ Co ⁺	440
C ₁₀ H ₁₀ Ni ⁺	442	C ₁₀ H ₁₄ O ₂ S ₂ Cu ⁺	445
C ₁₀ H ₁₀ O ⁺	196	C ₁₀ H ₁₄ O ₂ S ₂ Ni ⁺	444
C ₁₀ H ₁₀ O ₂ ⁺	214	C ₁₀ H ₁₄ O ₃ ⁺	221
C ₁₀ H ₁₀ O ₂ Cr ⁺	412	C ₁₀ H ₁₄ O ₄ Cl ₂ Sn ⁺	503
C ₁₀ H ₁₀ O ₂ Fe ⁺	431	C ₁₀ H ₁₄ O ₄ Co ⁺	438
C ₁₀ H ₁₀ O ₃ Fe ⁺	431	C ₁₀ H ₁₄ O ₄ Cu ⁺	445
C ₁₀ H ₁₀ O ₃ Ru ⁺	490	C ₁₀ H ₁₄ O ₄ Mg ⁺	290
C ₁₀ H ₁₀ Pb ⁺	550	C ₁₀ H ₁₄ O ₄ Ni ⁺	443
C ₁₀ H ₁₀ Pr ⁺	524	C ₁₀ H ₁₄ O ₄ Zn ⁺	447
C ₁₀ H ₁₀ Ru ⁺	490	C ₁₀ H ₁₄ O ₆ U ⁺	554
C ₁₀ H ₁₀ Si ⁺	297	C ₁₀ H ₁₄ S ⁺	335
C ₁₀ H ₁₀ Sn ⁺	498	C ₁₀ H ₁₄ S ₄ Co ⁺	440
C ₁₀ H ₁₀ TiBr ₂ ⁺	480	C ₁₀ H ₁₄ S ₄ Ni ⁺	444
C ₁₀ H ₁₀ V ⁺	407	C ₁₀ H ₁₄ Si ⁺	297
C ₁₀ H ₁₀ W ₂ ⁺	533	C ₁₀ H ₁₅ ⁺	90
C ₁₀ H ₁₀ ZrI ₂ ⁺	518	C ₁₀ H ₁₅ Br ⁺	466
C ₁₀ H ₁₁ Cl ⁺	374	C ₁₀ H ₁₅ Cl ⁺	374
C ₁₀ H ₁₁ DO ⁺	197	C ₁₀ H ₁₅ F ⁺	276
C ₁₀ H ₁₁ N ⁺	140	C ₁₀ H ₁₅ N ⁺	140
C ₁₀ H ₁₁ NO ⁺	235	C ₁₀ H ₁₅ NO ⁺	235
C ₁₀ H ₁₁ NO ₄ ⁺	265	C ₁₀ H ₁₅ NO ₂ ⁺	251
C ₁₀ H ₁₁ NO ₅ Cr ⁺	414	C ₁₀ H ₁₅ OSMn ⁺	426
C ₁₀ H ₁₁ NO ₅ W ⁺	535	C ₁₀ H ₁₅ O ₂ PCr ⁺	417
C ₁₀ H ₁₁ N ₃ ⁺	163	C ₁₀ H ₁₅ O ₇ PW ⁺	537
C ₁₀ H ₁₁ N ₃ S ⁺	349	C ₁₀ H ₁₅ P ⁺	312
C ₁₀ H ₁₁ O ⁺	196	C ₁₀ H ₁₅ SMn ⁺	425
C ₁₀ H ₁₁ O ₃ SMn ⁺	427	C ₁₀ H ₁₆ ⁺	90
C ₁₀ H ₁₁ Re ⁺	540	C ₁₀ H ₁₆ Ge ⁺	450

C ₁₀ H ₁₆ NFSi ⁺	309	C ₁₀ H ₂₄ Sn ⁺	499
C ₁₀ H ₁₆ NP ⁺	315	C ₁₀ H ₂₅ O ₂ Si ₂ ⁺	306
C ₁₀ H ₁₆ N ₂ ⁺	157	C ₁₀ H ₂₅ P ₅ ⁺	315
C ₁₀ H ₁₆ N ₂ O ₂ ⁺	257	C ₁₀ H ₂₇ NSi ₂ ⁺	304
C ₁₀ H ₁₆ N ₄ ⁺	166	C ₁₀ H ₂₇ Si ₂ As ⁺	456
C ₁₀ H ₁₆ O ⁺	197	C ₁₀ H ₂₇ Si ₂ P ⁺	325
C ₁₀ H ₁₆ OSi ⁺	305	C ₁₀ H ₂₈ N ₂ Si ₄ ⁺	304
C ₁₀ H ₁₆ O ₂ ⁺	215	C ₁₀ H ₃₀ N ₅ Nb ⁺	485
C ₁₀ H ₁₆ O ₂ S ⁺	358	C ₁₀ H ₃₀ N ₅ Ta ⁺	532
C ₁₀ H ₁₆ O ₂ Si ⁺	306	C ₁₀ H ₃₀ Si ₄ ⁺	302
C ₁₀ H ₁₆ O ₃ ⁺	222	C ₁₀ H ₃₀ Si ₅ ⁺	302
C ₁₀ H ₁₆ O ₄ Ni ⁺	443	C ₁₀ O ₆ F ₆ Co ₂ ⁺	440
C ₁₀ H ₁₆ O ₄ Pd ⁺	492	C ₁₀ O ₁₀ Mn ₂ ⁺	422
C ₁₀ H ₁₆ O ₄ Pt ⁺	543	C ₁₀ O ₁₀ Re ₂ ⁺	541
C ₁₀ H ₁₆ P ₂ ⁺	314	C ₁₁ HO ₃ F ₂ ⁺	287
C ₁₀ H ₁₆ Pb ⁺	550	C ₁₁ H ₄ N ₂ O ₅ W ⁺	535
C ₁₀ H ₁₆ S ⁺	335	C ₁₁ H ₅ N ₃ ⁺	163
C ₁₀ H ₁₆ SGe ⁺	452	C ₁₁ H ₆ O ₄ Cr ⁺	413
C ₁₀ H ₁₆ SPb ⁺	551	C ₁₁ H ₇ ⁺	92
C ₁₀ H ₁₆ SSn ⁺	501	C ₁₁ H ₇ N ⁺	141
C ₁₀ H ₁₆ S ₂ ⁺	342	C ₁₁ H ₇ NO ₃ Cr ⁺	414
C ₁₀ H ₁₆ Si ⁺	297	C ₁₁ H ₇ NO ₃ W ⁺	535
C ₁₀ H ₁₆ SiS ⁺	367	C ₁₁ H ₇ NO ₃ Cr ⁺	415
C ₁₀ H ₁₆ Sn ⁺	499	C ₁₁ H ₈ N ₂ ⁺	158
C ₁₀ H ₁₇ ⁺	91	C ₁₁ H ₈ N ₂ O ⁺	244
C ₁₀ H ₁₇ FSi ₂ ⁺	309	C ₁₁ H ₈ O ⁺	198
C ₁₀ H ₁₇ N ⁺	141	C ₁₁ H ₈ OS ⁺	356
C ₁₀ H ₁₇ NO ⁺	235	C ₁₁ H ₈ O ₂ ⁺	215
C ₁₀ H ₁₇ NO ₂ ⁺	252	C ₁₁ H ₈ O ₃ Fe ⁺	432
C ₁₀ H ₁₇ OCl ⁺	384	C ₁₁ H ₈ O ₄ Cr ⁺	413
C ₁₀ H ₁₇ P ⁺	312	C ₁₁ H ₈ O ₄ Mo ⁺	486
C ₁₀ H ₁₇ Si ₂ Cl ⁺	395	C ₁₁ H ₈ O ₄ Scr ⁺	419
C ₁₀ H ₁₈ ⁺	91	C ₁₁ H ₈ O ₅ Cr ⁺	413
C ₁₀ H ₁₈ N ₂ ⁺	157	C ₁₁ H ₉ ⁺	92
C ₁₀ H ₁₈ N ₂ O ⁺	244	C ₁₁ H ₉ F ⁺	276
C ₁₀ H ₁₈ N ₂ O ₂ ⁺	257	C ₁₁ H ₉ I ⁺	511
C ₁₀ H ₁₈ N ₂ O ₄ ⁺	266	C ₁₁ H ₉ NO ₂ ⁺	252
C ₁₀ H ₁₈ N ₃ O ₄ PCr ⁺	418	C ₁₁ H ₁₀ ⁺	92
C ₁₀ H ₁₈ N ₃ O ₄ PF ₆ ⁺	433	C ₁₁ H ₁₀ O ⁺	198
C ₁₀ H ₁₈ N ₃ O ₄ PMo ⁺	488	C ₁₁ H ₁₀ OMo ⁺	486
C ₁₀ H ₁₈ Ni ⁺	442	C ₁₁ H ₁₀ OW ₂ ⁺	535
C ₁₀ H ₁₈ O ⁺	197	C ₁₁ H ₁₀ O ₂ ⁺	215
C ₁₀ H ₁₈ O ₂ ⁺	215	C ₁₁ H ₁₀ O ₂ S ⁺	358
C ₁₀ H ₁₈ S ₆ ⁺	342	C ₁₁ H ₁₀ O ₃ Cr ⁺	412
C ₁₀ H ₁₈ Si ₂ ⁺	299	C ₁₁ H ₁₀ O ₄ PMn ⁺	424
C ₁₀ H ₁₈ Sn ⁺	499	C ₁₁ H ₁₀ S ⁺	335
C ₁₀ H ₁₉ N ⁺	141	C ₁₁ H ₁₀ SFe ₂ ⁺	435
C ₁₀ H ₁₉ NO ⁺	236	C ₁₁ H ₁₀ SMn ₂ ⁺	426
C ₁₀ H ₁₉ NO ₂ Cl ₂ ⁺	388	C ₁₁ H ₁₁ Cr ⁺	410
C ₁₀ H ₂₀ ⁺	91	C ₁₁ H ₁₁ Mn ⁺	421
C ₁₀ H ₂₀ NO ₂ Cl ⁺	388	C ₁₁ H ₁₁ N ⁺	141
C ₁₀ H ₂₀ N ₂ ⁺	157	C ₁₁ H ₁₁ NOS ⁺	362
C ₁₀ H ₂₀ N ₄ ⁺	166	C ₁₁ H ₁₁ NO ₂ ⁺	252
C ₁₀ H ₂₀ O ⁺	197	C ₁₁ H ₁₁ NO ₃ Cr ⁺	414
C ₁₀ H ₂₀ O ₅ ⁺	224	C ₁₁ H ₁₂ ⁺	93
C ₁₀ H ₂₀ S ⁺	335	C ₁₁ H ₁₂ N ₂ Cl ₂ ⁺	381
C ₁₀ H ₂₁ P ⁺	312	C ₁₁ H ₁₂ N ₂ O ₂ ⁺	257
C ₁₀ H ₂₂ Hg ⁺	547	C ₁₁ H ₁₂ O ⁺	198
C ₁₀ H ₂₂ N ₂ ⁺	157	C ₁₁ H ₁₂ O ₂ ⁺	215
C ₁₀ H ₂₂ N ₂ O ⁺	244	C ₁₁ H ₁₂ O ₂ Cr ⁺	412
C ₁₀ H ₂₂ Si ₂ ⁺	299	C ₁₁ H ₁₂ O ₃ Fe ⁺	432
C ₁₀ H ₂₃ N ⁺	141	C ₁₁ H ₁₂ O ₃ PMn ⁺	424
C ₁₀ H ₂₄ N ₂ ⁺	157	C ₁₁ H ₁₂ O ₄ Fe ⁺	432
C ₁₀ H ₂₄ N ₂ S ₃ Sn ₂ ⁺	502	C ₁₁ H ₁₂ O ₅ Fe ⁺	432
C ₁₀ H ₂₄ N ₃ P ⁺	316	C ₁₁ H ₁₃ Cl ⁺	374
C ₁₀ H ₂₄ N ₄ ⁺	166	C ₁₁ H ₁₃ DO ⁺	198
C ₁₀ H ₂₄ O ₃ Ti ⁺	406	C ₁₁ H ₁₃ N ⁺	141
C ₁₀ H ₂₄ O ₄ Ti ⁺	406	C ₁₁ H ₁₃ NO ⁺	236
C ₁₀ H ₂₄ Si ₂ ⁺	299	C ₁₁ H ₁₃ N ₃ Cl ₂ ⁺	381
C ₁₀ H ₂₄ Si ₄ ⁺	302	C ₁₁ H ₁₃ O ⁺	198

C ₁₁ H ₁₃ O ₇ ⁺	224	C ₁₁ H ₂₄ S ₂ Sn ⁺	502
C ₁₁ H ₁₄ ⁺	93	C ₁₁ H ₂₅ NS ₂ Sn ⁺	502
C ₁₁ H ₁₄ NO ⁺	236	C ₁₁ H ₂ O ₂ Si ₂ ⁺	306
C ₁₁ H ₁₄ NOCi ⁺	386	C ₁₁ H ₃ NSi ₂ P ₂ ⁺	325
C ₁₁ H ₁₄ N ₂ ⁺	158	C ₁₂ F ₈ ⁺	273
C ₁₁ H ₁₄ N ₂ O ⁺	244	C ₁₂ F ₁₀ ⁺	273
C ₁₁ H ₁₄ O ⁺	198	C ₁₂ H ₃ O ₁₂ Re ₃ ⁺	541
C ₁₁ H ₁₄ O ₂ ⁺	215	C ₁₂ H ₆ ⁺	95
C ₁₁ H ₁₄ O ₃ CrGe ⁺	453	C ₁₂ H ₆ N ₄ ⁺	166
C ₁₁ H ₁₅ Cl ⁺	374	C ₁₂ H ₆ O ₂ ⁺	216
C ₁₁ H ₁₅ NO ⁺	236	C ₁₂ H ₆ O ₃ ⁺	222
C ₁₁ H ₁₅ N ₂ O ⁺	244	C ₁₂ H ₇ NO ₂ ⁺	252
C ₁₁ H ₁₅ N ₂ OCl ⁺	387	C ₁₂ H ₇ NO ₅ Cr ⁺	414
C ₁₁ H ₁₅ N ₂ S ⁺	348	C ₁₂ H ₇ NO ₆ Cr ⁺	415
C ₁₁ H ₁₅ N ₂ SCl ⁺	400	C ₁₂ H ₇ N ₂ OCl ⁺	387
C ₁₁ H ₁₅ N ₃ ⁺	163	C ₁₂ H ₇ N ₃ ⁺	163
C ₁₁ H ₁₅ N ₃ ²⁺	167	C ₁₂ H ₇ N ₃ O ⁺	265
C ₁₁ H ₁₅ O ₃ PCr ⁺	417	C ₁₂ H ₈ ⁺	95
C ₁₁ H ₁₅ O ₃ PMo ⁺	488	C ₁₂ H ₈ Br ₂ ⁺	467
C ₁₁ H ₁₅ O ₅ PW ⁺	536	C ₁₂ H ₈ FBr ⁺	474
C ₁₁ H ₁₅ O ₈ PCr ⁺	418	C ₁₂ H ₈ F ₂ ⁺	277
C ₁₁ H ₁₅ O ₈ PMo ⁺	488	C ₁₂ H ₈ NO ⁺	236
C ₁₁ H ₁₅ O ₈ PW ⁺	537	C ₁₂ H ₈ NOBr ⁺	472
C ₁₁ H ₁₆ ⁺	94	C ₁₂ H ₈ NOCl ⁺	386
C ₁₁ H ₁₆ NO ₂ Br ⁺	473	C ₁₂ H ₈ NOF ⁺	287
C ₁₁ H ₁₆ NO ₂ F ₃ ⁺	288	C ₁₂ H ₈ NOI ⁺	513
C ₁₁ H ₁₆ NO ₃ ⁺	244	C ₁₂ H ₈ N ₂ ⁺	158
C ₁₁ H ₁₆ N ₂ S ⁺	348	C ₁₂ H ₈ N ₂ O ⁺	244
C ₁₁ H ₁₆ N ₃ ⁺	163	C ₁₂ H ₈ N ₃ O ⁺	264
C ₁₁ H ₁₆ N ₃ Cl ⁺	380	C ₁₂ H ₈ N ₃ ⁺	163
C ₁₁ H ₁₆ O ⁺	198	C ₁₂ H ₈ N ₃ Cl ⁺	380
C ₁₁ H ₁₆ OBr ⁺	471	C ₁₂ H ₈ O ⁺	199
C ₁₁ H ₁₆ O ₂ ⁺	215	C ₁₂ H ₈ OS ⁺	356
C ₁₁ H ₁₆ S ⁺	335	C ₁₂ H ₈ OSe ⁺	461
C ₁₁ H ₁₆ Si ⁺	297	C ₁₂ H ₈ OTe ⁺	507
C ₁₁ H ₁₇ ⁺	94	C ₁₂ H ₈ O ₂ ⁺	216
C ₁₁ H ₁₇ N ⁺	142	C ₁₂ H ₈ O ₃ ⁺	358
C ₁₁ H ₁₇ NO ₂ ⁺	252	C ₁₂ H ₈ O ₃ SBr ₂ ⁺	479
C ₁₁ H ₁₇ NO ₃ ⁺	262	C ₁₂ H ₈ S ⁺	335
C ₁₁ H ₁₇ N ₃ ⁺	163	C ₁₂ H ₈ SCl ₂ ⁺	400
C ₁₁ H ₁₇ O ₃ PCr ⁺	417	C ₁₂ H ₉ ⁺	96
C ₁₁ H ₁₈ ⁺	94	C ₁₂ H ₉ Br ⁺	466
C ₁₁ H ₁₈ N ₃ O ₅ PCr ⁺	418	C ₁₂ H ₉ Cl ⁺	374
C ₁₁ H ₁₈ N ₃ O ₅ PMo ⁺	488	C ₁₂ H ₉ F ⁺	276
C ₁₁ H ₁₈ N ₃ O ₅ PW ⁺	537	C ₁₂ H ₉ I ⁺	511
C ₁₁ H ₁₈ O ⁺	198	C ₁₂ H ₉ N ⁺	142
C ₁₁ H ₁₈ O ₂ ⁺	216	C ₁₂ H ₉ NO ⁺	236
C ₁₁ H ₁₈ SSn ⁺	501	C ₁₂ H ₉ NO ₅ W ⁺	535
C ₁₁ H ₁₈ SiS ⁺	367	C ₁₂ H ₉ NS ⁺	345
C ₁₁ H ₁₉ N ⁺	142	C ₁₂ H ₉ N ₂ Cl ⁺	380
C ₁₁ H ₁₉ NOSi ⁺	307	C ₁₂ H ₉ N ₃ ⁺	163
C ₁₁ H ₂₀ ⁺	95	C ₁₂ H ₉ N ₄ O ₂ ⁺	260
C ₁₁ H ₂₀ NO ⁺	236	C ₁₂ H ₉ N ₄ O ₂ F ⁺	288
C ₁₁ H ₂₀ N ₂ O ₄ ⁺	266	C ₁₂ H ₉ O ₄ Rh ⁺	491
C ₁₁ H ₂₀ O ⁺	199	C ₁₂ H ₉ SCl ⁺	399
C ₁₁ H ₂₀ OSi ₂ ⁺	306	C ₁₂ H ₁₀ ⁺	96
C ₁₁ H ₂₀ O ₃ ⁺	216	C ₁₂ H ₁₀ As ⁺	454
C ₁₁ H ₂₀ O ₅ Si ₂ ⁺	306	C ₁₂ H ₁₀ Br ⁺	552
C ₁₁ H ₂₀ O ₅ Si ₂ ²⁺	306	C ₁₂ H ₁₀ Ga ⁺	448
C ₁₁ H ₂₀ PAu ⁺	545	C ₁₂ H ₁₀ Hg ⁺	546
C ₁₁ H ₂₀ Si ₂ ⁺	299	C ₁₂ H ₁₀ NOS ₂ Mn ₂ ⁺	428
C ₁₁ H ₂₁ N ⁺	142	C ₁₂ H ₁₀ N ₂ ⁺	158
C ₁₁ H ₂₁ NSi ₂ ⁺	304	C ₁₂ H ₁₀ N ₂ O ⁺	244
C ₁₁ H ₂₁ N ₂ O ₂ ⁺	258	C ₁₂ H ₁₀ N ₂ O ₃ ⁺	258
C ₁₁ H ₂₂ ⁺	95	C ₁₂ H ₁₀ N ₂ O ₂ Mn ⁺	8
C ₁₁ H ₂₂ NO ⁺	236	C ₁₂ H ₁₀ N ₃ F ₄ P ₃ ⁺	323
C ₁₁ H ₂₂ N ₂ ⁺	158	C ₁₂ H ₁₀ N ₄ O ₂ ⁺	261
C ₁₁ H ₂₂ O ⁺	199	C ₁₂ H ₁₀ O ⁺	199
C ₁₁ H ₂₂ Si ₂ ⁺	299	C ₁₂ H ₁₀ OS ⁺	356
C ₁₁ H ₂₃ P ⁺	312		

C ₁₂ H ₁₀ OSFe ₂ ⁺	435	C ₁₂ H ₁₅ DO ⁺	200
C ₁₂ H ₁₀ O ₂ Fe ⁺	431	C ₁₂ H ₁₅ N ⁺	142
C ₁₂ H ₁₀ O ₂ S ⁺	359	C ₁₂ H ₁₅ NO ⁺	237
C ₁₂ H ₁₀ O ₂ Ti ⁺	406	C ₁₂ H ₁₅ O ₂ SMn ⁺	427
C ₁₂ H ₁₀ O ₂ W ₂ ⁺	535	C ₁₂ H ₁₅ O ₂ SMn ⁺	427
C ₁₂ H ₁₀ S ⁺	335	C ₁₂ H ₁₆	97
C ₁₂ H ₁₀ S ₂ Fe ₂ ⁺	435	C ₁₂ H ₁₆ Mo ⁺	485
C ₁₂ H ₁₀ S ₂ Mn ₂ ⁺	426	C ₁₂ H ₁₆ NO ⁺	237
C ₁₂ H ₁₀ Sb ⁺	505	C ₁₂ H ₁₆ NOCl ⁺	386
C ₁₂ H ₁₀ Si ₂ ⁺	300	C ₁₂ H ₁₆ NS ⁺	345
C ₁₂ H ₁₁ ⁺	96	C ₁₂ H ₁₆ NSCl ⁺	400
C ₁₂ H ₁₁ As ⁺	455	C ₁₂ H ₁₆ N ₂ ⁺	159
C ₁₂ H ₁₁ N ⁺	142	C ₁₂ H ₁₆ N ₂ O ₄ S ₄ Fe ⁺	436
C ₁₂ H ₁₁ NO ⁺	237	C ₁₂ H ₁₆ O ⁺	200
C ₁₂ H ₁₁ NO ₂ ⁺	252	C ₁₂ H ₁₆ OS ⁺	356
C ₁₂ H ₁₁ N ₃ ⁺	163	C ₁₂ H ₁₆ O ₂ ⁺	216
C ₁₂ H ₁₁ P ⁺	312	C ₁₂ H ₁₆ O ₃ S ⁺	359
C ₁₂ H ₁₂ ⁺	96	C ₁₂ H ₁₆ O ₃ ⁺	341
C ₁₂ H ₁₂ Cr ⁺	410	C ₁₂ H ₁₆ Si ⁺	298
C ₁₂ H ₁₂ Fe ⁺	429	C ₁₂ H ₁₆ Sn ⁺	499
C ₁₂ H ₁₂ Mo ⁺	485	C ₁₂ H ₁₆ W ⁺	533
C ₁₂ H ₁₂ N ₂ ⁺	158	C ₁₂ H ₁₇ P ⁺	312
C ₁₂ H ₁₂ N ₂ O ⁺	244	C ₁₂ H ₁₈ ⁺	98
C ₁₂ H ₁₂ N ₂ O ₂ S ⁺	364	C ₁₂ H ₁₈ Cr ⁺	410
C ₁₂ H ₁₂ N ₂ S ⁺	348	C ₁₂ H ₁₈ Ge ⁺	450
C ₁₂ H ₁₂ Nb ⁺	485	C ₁₂ H ₁₈ NO ⁺	237
C ₁₂ H ₁₂ O ⁺	200	C ₁₂ H ₁₈ N ₂ O ₂ Cu ⁺	445
C ₁₂ H ₁₂ OS ⁺	356	C ₁₂ H ₁₈ N ₂ O ₂ Ni ⁺	443
C ₁₂ H ₁₂ O ₂ ⁺	216	C ₁₂ H ₁₈ N ₂ O ₂ Pd ⁺	492
C ₁₂ H ₁₂ O ₂ S ⁺	359	C ₁₂ H ₁₈ N ₂ S ₂ Co ⁺	440
C ₁₂ H ₁₂ O ₃ Cr ⁺	412	C ₁₂ H ₁₈ N ₂ S ₂ Cu ⁺	445
C ₁₂ H ₁₂ O ₃ Mo ⁺	486	C ₁₂ H ₁₈ N ₂ S ₂ Ni ⁺	444
C ₁₂ H ₁₂ O ₃ W ⁺	534	C ₁₂ H ₁₈ N ₂ S ₂ Pd ⁺	492
C ₁₂ H ₁₂ S ₂ ⁺	339	C ₁₂ H ₁₈ O ⁺	200
C ₁₂ H ₁₂ Si ⁺	297	C ₁₂ H ₁₈ OB _{Br} ⁺	471
C ₁₂ H ₁₂ Ti ⁺	406	C ₁₂ H ₁₈ OS ⁺	356
C ₁₂ H ₁₂ V ⁺	407	C ₁₂ H ₁₈ O ₂ ⁺	216
C ₁₂ H ₁₂ Zr ⁺	484	C ₁₂ H ₁₈ S ⁺	335
C ₁₂ H ₁₃ As ⁺	455	C ₁₂ H ₁₈ Si ⁺	298
C ₁₂ H ₁₃ N ⁺	142	C ₁₂ H ₁₈ Sn ⁺	499
C ₁₂ H ₁₃ NO ⁺	237	C ₁₂ H ₁₉ NO ₂ ⁺	252
C ₁₂ H ₁₃ NO ₂ ⁺	252	C ₁₂ H ₁₉ NO ₂ S ⁺	363
C ₁₂ H ₁₃ NS ₂ ⁺	349	C ₁₂ H ₁₉ NO ₃ ⁺	262
C ₁₂ H ₁₃ N ₃ ⁺	163	C ₁₂ H ₂₀	98
C ₁₂ H ₁₃ P ⁺	312	C ₁₂ H ₂₀ NO ⁺	237
C ₁₂ H ₁₄ ⁺	97	C ₁₂ H ₂₀ N ₂ ⁺	159
C ₁₂ H ₁₄ Br ₂ Mo ⁺	490	C ₁₂ H ₂₀ N ₂ O ₂ ⁺	258
C ₁₂ H ₁₄ Cl ₂ Mo ⁺	489	C ₁₂ H ₂₀ N ₂ S ⁺	348
C ₁₂ H ₁₄ Cl ₂ Nb ⁺	485	C ₁₂ H ₂₀ N ₂ S ₂ ⁺	350
C ₁₂ H ₁₄ Co ⁺	437	C ₁₂ H ₂₀ O ⁺	200
C ₁₂ H ₁₄ Cr ⁺	410	C ₁₂ H ₂₀ O ₂ S ⁺	359
C ₁₂ H ₁₄ Fe ⁺	430	C ₁₂ H ₂₀ O ₄ Sn ⁺	501
C ₁₂ H ₁₄ Mg ⁺	290	C ₁₂ H ₂₀ O ₉ Cr ₂ ⁺	413
C ₁₂ H ₁₄ Mn ⁺	421	C ₁₂ H ₂₀ S ⁺	336
C ₁₂ H ₁₄ Mo ⁺	485	C ₁₂ H ₂₀ S ₄ ⁺	342
C ₁₂ H ₁₄ MoI ₂ ⁺	515	C ₁₂ H ₂₁ NO ⁺	237
C ₁₂ H ₁₄ NO ₂ ⁺	252	C ₁₂ H ₂₂ ⁺	99
C ₁₂ H ₁₄ N ₂ ⁺	159	C ₁₂ H ₂₂ NO ⁺	237
C ₁₂ H ₁₄ N ₂ O ₅ Cr ⁺	415	C ₁₂ H ₂₂ NO ₂ ⁺	252
C ₁₂ H ₁₄ N ₂ O ₅ Mo ⁺	487	C ₁₂ H ₂₂ N ₂ ⁺	159
C ₁₂ H ₁₄ N ₂ O ₅ W ⁺	535	C ₁₂ H ₂₂ O ⁺	200
C ₁₂ H ₁₄ Ni ⁺	442	C ₁₂ H ₂₂ OSi ₂ ⁺	306
C ₁₂ H ₁₄ O ⁺	200	C ₁₂ H ₂₂ O ₂ ⁺	216
C ₁₂ H ₁₄ O ₂ ⁺	216	C ₁₂ H ₂₂ O ₃ ⁺	222
C ₁₂ H ₁₄ O ₆ S ₂ Fe ₂ ⁺	436	C ₁₂ H ₂₂ Si ₂ ⁺	300
C ₁₂ H ₁₄ Os ⁺	542	C ₁₂ H ₂₃ N ⁺	143
C ₁₂ H ₁₄ Ru ⁺	490	C ₁₂ H ₂₃ NO ₂ Cl ₂ ⁺	388
C ₁₂ H ₁₄ V ⁺	407	C ₁₂ H ₂₄ ⁺	99
C ₁₂ H ₁₄ W ⁺	533	C ₁₂ H ₂₄ NO ⁺	237
C ₁₂ H ₁₅ Cl ⁺	375	C ₁₂ H ₂₄ O ⁺	216

C ₁₂ H ₂₁ O ₄ ⁺	223	C ₁₃ H ₁₀ NO ₂ ⁺	253
C ₁₂ H ₂₁ O ₆ ⁺	224	C ₁₃ H ₁₀ N ₂ ⁺	159
C ₁₂ H ₂₄ Si ₂ ⁺	300	C ₁₃ H ₁₀ N ₂ O ⁺	244
C ₁₂ H ₂₅ NO ⁺	238	C ₁₃ H ₁₀ N ₂ O ₂ ⁺	258
C ₁₂ H ₂₅ N ₂ ⁺	159	C ₁₃ H ₁₀ N ₂ O ₄ ⁺	266
C ₁₂ H ₂₆ N ₄ ⁺	166	C ₁₃ H ₁₀ O ⁺	201
C ₁₂ H ₂₇ N ⁺	143	C ₁₃ H ₁₀ OS ₂ Fe ₂ ⁺	435
C ₁₂ H ₂₇ NBr ₂ ⁺	469	C ₁₃ H ₁₀ O ₂ ⁺	216
C ₁₂ H ₂₇ O ₃ PS ⁺	369	C ₁₃ H ₁₀ O ₆ Fe ₂ ⁺	435
C ₁₂ H ₂₇ P ⁺	313	C ₁₃ H ₁₀ O ₆ W ₂ ⁺	535
C ₁₂ H ₂₇ PCr ⁺	416	C ₁₃ H ₁₀ S ⁺	336
C ₁₂ H ₂₈ N ₂ ⁺	159	C ₁₃ H ₁₁ ⁺	100
C ₁₂ H ₂₈ N ₄ ⁺	166	C ₁₃ H ₁₁ N ⁺	143
C ₁₂ H ₂₈ Si ₂ ⁺	300	C ₁₃ H ₁₁ NO ⁺	238
C ₁₂ H ₂₈ Sn ⁺	499	C ₁₃ H ₁₁ NO ₂ ⁺	253
C ₁₂ H ₂₉ O ₂ Si ₂ ⁺	306	C ₁₃ H ₁₁ NS ⁺	345
C ₁₂ H ₃₀ O ₆ P ₆ S ₆ Co ⁺	440	C ₁₃ H ₁₁ N ₄ O ₂ ⁺	261
C ₁₂ H ₃₀ O ₆ P ₆ S ₆ Cr ⁺	419	C ₁₃ H ₁₁ O ⁺	201
C ₁₂ H ₃₀ O ₆ P ₆ S ₆ In ⁺	496	C ₁₃ H ₁₁ OCl ⁺	384
C ₁₂ H ₃₀ O ₆ P ₆ S ₆ Rh ⁺	492	C ₁₃ H ₁₁ OSCl ⁺	401
C ₁₂ H ₃₀ Si ₃ ⁺	300	C ₁₃ H ₁₁ OSi ⁺	305
C ₁₂ H ₃₂ P ₂ S ₂ Sn ₂ ⁺	502	C ₁₃ H ₁₁ OSn ⁺	500
C ₁₂ H ₃₃ NSi ₃ ⁺	304	C ₁₃ H ₁₁ O ₂ SSn ⁺	502
C ₁₂ H ₃₆ N ₂ Si ₄ ⁺	304	C ₁₃ H ₁₁ SGe ⁺	452
C ₁₂ H ₃₆ N ₂ Si ₄ Ge ⁺	452	C ₁₃ H ₁₁ SSn ⁺	501
C ₁₂ H ₃₆ N ₂ Si ₄ Hg ⁺	547	C ₁₃ H ₁₁ SiS ⁺	368
C ₁₂ H ₃₆ N ₂ Si ₄ Pb ⁺	551	C ₁₃ H ₁₂ ⁺	100
C ₁₂ H ₃₆ N ₂ Si ₄ Sn ⁺	501	C ₁₃ H ₁₂ N ⁺	144
C ₁₂ H ₃₆ N ₂ Si ₄ Zn ⁺	447	C ₁₃ H ₁₂ NO ₂ ⁺	253
C ₁₂ H ₃₆ N ₆ Mo ₂ ⁺	486	C ₁₃ H ₁₂ N ₂ ⁺	159
C ₁₂ H ₃₆ N ₆ P ₂ Fe ⁺	433	C ₁₃ H ₁₂ N ₂ O ⁺	244
C ₁₂ H ₃₆ N ₆ P ₂ Mo ⁺	488	C ₁₃ H ₁₂ N ₄ O ₂ ⁺	258
C ₁₂ H ₃₆ N ₆ P ₂ W ⁺	536	C ₁₃ H ₁₂ O ⁺	261
C ₁₂ H ₃₆ N ₆ W ⁺	533	C ₁₃ H ₁₂ OS ⁺	201
C ₁₂ H ₃₆ Si ₅ ⁺	302	C ₁₃ H ₁₂ O ₂ ⁺	356
C ₁₂ H ₃₆ Si ₆ ⁺	302	C ₁₃ H ₁₂ S ⁺	216
C ₁₂ O ₁₂ Os ₃ ⁺	543	C ₁₃ H ₁₃ N ⁺	336
C ₁₂ O ₁₂ Ru ₃ ⁺	490	C ₁₃ H ₁₃ NO ⁺	144
C ₁₃ H ₇ N ₃ O ⁺	246	C ₁₃ H ₁₃ NO ₂ ⁺	238
C ₁₃ H ₇ N ₄ ⁺	166	C ₁₃ H ₁₃ NO ₂ S ⁺	363
C ₁₃ H ₇ O ₆ ClCr ⁺	420	C ₁₃ H ₁₃ Si ⁺	298
C ₁₃ H ₇ O ₆ FGr ⁺	415	C ₁₃ H ₁₃ Ti ⁺	406
C ₁₃ H ₈ NO ₂ ⁺	252	C ₁₃ H ₁₄ ⁺	101
C ₁₃ H ₈ N ₂ ⁺	159	C ₁₃ H ₁₄ NO ₂ ⁺	253
C ₁₃ H ₈ N ₄ ⁺	166	C ₁₃ H ₁₄ N ₂ ⁺	159
C ₁₃ H ₈ O ⁺	200	C ₁₃ H ₁₄ N ₂ O ⁺	245
C ₁₃ H ₈ OS ⁺	356	C ₁₃ H ₁₄ N ₂ O ₄ ⁺	266
C ₁₃ H ₈ O ₂ ⁺	216	C ₁₃ H ₁₄ O ₂ ⁺	217
C ₁₃ H ₈ Cr ⁺	413	C ₁₃ H ₁₄ Si ⁺	298
C ₁₃ H ₈ S ⁺	336	C ₁₃ H ₁₅ Ge ⁺	450
C ₁₃ H ₉ ⁺	99	C ₁₃ H ₁₅ NO ⁺	238
C ₁₃ H ₉ D ₂ ⁺	100	C ₁₃ H ₁₅ NO ₂ ⁺	253
C ₁₃ H ₉ N ⁺	143	C ₁₃ H ₁₅ NO ₂ S ⁺	364
C ₁₃ H ₉ NO ⁺	238	C ₁₃ H ₁₅ Nb ⁺	485
C ₁₃ H ₉ NOS ⁺	362	C ₁₃ H ₁₆ ⁺	101
C ₁₃ H ₉ NOS ⁺	252	C ₁₃ H ₁₆ N ₂ ⁺	159
C ₁₃ H ₉ NO ₂ ⁺	363	C ₁₃ H ₁₆ Si ⁺	298
C ₁₃ H ₉ NO ₂ S ⁺	263	C ₁₃ H ₁₆ Sn ⁺	499
C ₁₃ H ₉ NO ₃ ⁺	265	C ₁₃ H ₁₆ W ⁺	533
C ₁₃ H ₉ NO ₄ ⁺	201	C ₁₃ H ₁₇ Cl ⁺	375
C ₁₃ H ₉ O ⁺	384	C ₁₃ H ₁₇ N ⁺	144
C ₁₃ H ₉ OCl ⁺	313	C ₁₃ H ₁₇ NO ₂ S ⁺	364
C ₁₃ H ₉ P ⁺	100	C ₁₃ H ₁₇ N ₂ ⁺	164
C ₁₃ H ₁₀ ⁺	100	C ₁₃ H ₁₇ O ₂ PCr ⁺	417
C ₁₃ H ₁₀ D ⁺	143	C ₁₃ H ₁₇ O ₂ ⁺	224
C ₁₃ H ₁₀ N ⁺	468	C ₁₃ H ₁₈ ⁺	101
C ₁₃ H ₁₀ NBr ⁺	379	C ₁₃ H ₁₈ N ₂ O ⁺	245
C ₁₃ H ₁₀ NCl ⁺	281	C ₁₃ H ₁₈ O ⁺	201
C ₁₃ H ₁₀ NF ⁺	512	C ₁₃ H ₁₈ O ₂ Cr ⁺	411
C ₁₃ H ₁₀ NI ⁺	238	C ₁₃ H ₁₈ OSi ⁺	305

C ₁₃ H ₁₈ SiFe ⁺	433	C ₁₄ H ₁₁ N ⁺	144
C ₁₃ H ₁₉ BrMoSn ⁺	504	C ₁₄ H ₁₁ NO ⁺	239
C ₁₃ H ₁₉ ClMoSn ⁺	504	C ₁₄ H ₁₁ NO ₅ Cr ⁺	414
C ₁₃ H ₁₉ MoSnI ⁺	519	C ₁₄ H ₁₁ NS ⁺	345
C ₁₃ H ₂₀ MoSn ⁺	504	C ₁₄ H ₁₁ OS ⁺	357
C ₁₃ H ₂₀ NO ⁺	238	C ₁₄ H ₁₁ P ⁺	313
C ₁₃ H ₂₀ O ⁺	201	C ₁₄ H ₁₁ S ⁺	336
C ₁₃ H ₂₀ OSi ⁺	305	C ₁₄ H ₁₂	103
C ₁₃ H ₂₀ O ₂ ⁺	217	C ₁₄ H ₁₂ N ⁺	160
C ₁₃ H ₂₀ SnW ⁺	540	C ₁₄ H ₁₂ N ₂ O ₂ ⁺	258
C ₁₃ H ₂₁ NO ⁺	238	C ₁₄ H ₁₂ N ₂ O ₄ ⁺	266
C ₁₃ H ₂₁ SnTa ⁺	532	C ₁₄ H ₁₂ O ⁺	202
C ₁₃ H ₂₂ ⁺	101	C ₁₄ H ₁₂ O ₂ ⁺	217
C ₁₃ H ₂₂ Ge ⁺	450	C ₁₄ H ₁₂ O ₃ S ⁺	359
C ₁₃ H ₂₂ NO ⁺	239	C ₁₄ H ₁₂ O ₃ ⁺	222
C ₁₃ H ₂₂ N ₂ ⁺	160	C ₁₄ H ₁₂ S ⁺	336
C ₁₃ H ₂₂ Si ₂ ⁺	300	C ₁₄ H ₁₃	104
C ₁₃ H ₂₂ Sn ⁺	499	C ₁₄ H ₁₃ N ⁺	144
C ₁₃ H ₂₃ NO ⁺	239	C ₁₄ H ₁₃ NO ⁺	239
C ₁₃ H ₂₄ ⁺	101	C ₁₄ H ₁₃ NO ₂ ⁺	253
C ₁₃ H ₂₄ NO ⁺	239	C ₁₄ H ₁₃ NO ₄ ⁺	265
C ₁₃ H ₂₄ NO ₂ ⁺	253	C ₁₄ H ₁₃ NO ₅ Cr ⁺	414
C ₁₃ H ₂₄ N ₂ ⁺	160	C ₁₄ H ₁₃ Sn ⁺	499
C ₁₃ H ₂₄ Si ₂ ⁺	300	C ₁₄ H ₁₄	104
C ₁₃ H ₂₅ NO ⁺	239	C ₁₄ H ₁₄ N ₂ ⁺	160
C ₁₃ H ₂₅ N ₂ O ₂ ⁺	258	C ₁₄ H ₁₄ N ₂ O ₂ ⁺	258
C ₁₃ H ₂₆ ⁺	102	C ₁₄ H ₁₄ N ₂ O ₃ ⁺	264
C ₁₃ H ₂₇ OPCr ⁺	416	C ₁₄ H ₁₄ N ₂ S ₂ ⁺	350
C ₁₃ H ₂₈ Sn ₂ ⁺	500	C ₁₄ H ₁₄ N ₄ O ₂ ⁺	261
C ₁₃ H ₃₃ N ₃ Ti ⁺	406	C ₁₄ H ₁₄ O ⁺	202
C ₁₃ H ₃₆ N ₆ OP ₂ Fe ⁺	434	C ₁₄ H ₁₄ OS ⁺	357
C ₁₃ H ₃₆ N ₆ OP ₂ Mo ⁺	488	C ₁₄ H ₁₄ OSi ⁺	305
C ₁₄ F ₁₀ ⁺	273	C ₁₄ H ₁₄ OSn ⁺	500
C ₁₄ HO ₄ F ₂₉ ⁺	287	C ₁₄ H ₁₄ O ₂ ⁺	217
C ₁₄ H ₇ O ₆ F ₃ Cr ⁺	415	C ₁₄ H ₁₄ O ₂ S ⁺	359
C ₁₄ H ₈ ⁺	102	C ₁₄ H ₁₄ O ₂ SSn ⁺	502
C ₁₄ H ₈ Br ₂ ⁺	467	C ₁₄ H ₁₄ S ⁺	336
C ₁₄ H ₈ Cl ₂ ⁺	376	C ₁₄ H ₁₄ SGe ⁺	453
C ₁₄ H ₈ NO ₂ Cl ⁺	388	C ₁₄ H ₁₄ SSn ⁺	502
C ₁₄ H ₈ O ₂ ⁺	217	C ₁₄ H ₁₄ Si ⁺	298
C ₁₄ H ₈ O ₃ ⁺	222	C ₁₄ H ₁₄ SiS ⁺	368
C ₁₄ H ₈ O ₄ ⁺	223	C ₁₄ H ₁₅ N ⁺	144
C ₁₄ H ₈ O ₆ ⁺	224	C ₁₄ H ₁₅ NO ⁺	239
C ₁₄ H ₈ S ₄ ⁺	342	C ₁₄ H ₁₅ P ⁺	313
C ₁₄ H ₉ Br ⁺	466	C ₁₄ H ₁₆	104
C ₁₄ H ₉ Cl ⁺	375	C ₁₄ H ₁₆ Cr ⁺	410
C ₁₄ H ₉ F ⁺	276	C ₁₄ H ₁₆ Mo ⁺	486
C ₁₄ H ₉ N ⁺	144	C ₁₄ H ₁₆ N ₂ ⁺	160
C ₁₄ H ₉ NO ₂ ⁺	253	C ₁₄ H ₁₆ OS ₂ ⁺	360
C ₁₄ H ₉ OS ⁺	356	C ₁₄ H ₁₆ O ₂ ⁺	217
C ₁₄ H ₉ O ₂ ⁺	217	C ₁₄ H ₁₆ Ti ⁺	406
C ₁₄ H ₉ O ₂ S ⁺	359	C ₁₄ H ₁₈	105
C ₁₄ H ₁₀ ⁺	102	C ₁₄ H ₁₈ Ge ⁺	450
C ₁₄ H ₁₀ ²	103	C ₁₄ H ₁₈ N ₂ ⁺	160
C ₁₄ H ₁₀ NF ₃ ⁺	282	C ₁₄ H ₁₈ O ₂ ⁺	217
C ₁₄ H ₁₀ N ₂ O ⁺	245	C ₁₄ H ₁₈ O ₂ Cr ⁺	412
C ₁₄ H ₁₀ N ₂ O ₂ ⁺	258	C ₁₄ H ₁₈ Si ⁺	298
C ₁₄ H ₁₀ O ⁺	202	C ₁₄ H ₁₈ Sn ⁺	499
C ₁₄ H ₁₀ OS ⁺	356	C ₁₄ H ₁₉ NO ⁺	239
C ₁₄ H ₁₀ O ₂ ⁺	217	C ₁₄ H ₂₀ N ₂ O ₂ S ₄ Fe ⁺	436
C ₁₄ H ₁₀ O ₂ S ₂ Fe ₂ ⁺	435	C ₁₄ H ₂₀ O ⁺	202
C ₁₄ H ₁₀ O ₃ ⁺	222	C ₁₄ H ₂₀ O ₂ ^{..}	218
C ₁₄ H ₁₀ O ₃ S ⁺	359	C ₁₄ H ₂₀ O ₁₀ ^{..}	224
C ₁₄ H ₁₀ O ₃ Fe ₂ ^{..}	435	C ₁₄ H ₂₀ S ₃ ^{..}	341
C ₁₄ H ₁₀ O ₄ Fe ₂ ^{..}	432	C ₁₄ H ₂₁ O ₈ PCr ⁺	418
C ₁₄ H ₁₀ O ₄ W ₂ ^{..}	535	C ₁₄ H ₂₁ O ₈ PW ⁺	537
C ₁₄ H ₁₀ O ₆ Cr ⁺	413	C ₁₄ H ₂₂ ^{..}	105
C ₁₄ H ₁₀ O ₇ Cr ⁺	413	C ₁₄ H ₂₂ O ^{..}	202
C ₁₄ H ₁₀ S ⁺	336	C ₁₄ H ₂₂ O ₂ ^{..}	218
C ₁₄ H ₁₁ ⁺	103	C ₁₄ H ₂₂ Si ₂ ^{..}	300

C ₁₄ H ₂₃ NO ⁺	239	C ₁₅ H ₁₅ Pr ⁺	525
C ₁₄ H ₂₃ P ⁺	313	C ₁₅ H ₁₆ ⁺	107
C ₁₄ H ₂₄ ⁺	105	C ₁₅ H ₁₆ NO ₂ ⁺	254
C ₁₄ H ₂₄ Si ₂ ⁺	300	C ₁₅ H ₁₆ N ₂ ⁺	160
C ₁₄ H ₂₅ NO ⁺	240	C ₁₅ H ₁₆ OS ⁺	357
C ₁₄ H ₂₆ ⁺	105	C ₁₅ H ₁₆ O ₂ ⁺	218
C ₁₄ H ₂₆ Si ₂ ⁺	300	C ₁₅ H ₁₆ Sn ⁺	499
C ₁₄ H ₂₇ NO ⁺	240	C ₁₅ H ₁₈ ⁺	107
C ₁₄ H ₂₇ O ₂ PCr ⁺	416	C ₁₅ H ₁₈ NO ₂ ⁺	254
C ₁₄ H ₂₈ ⁺	105	C ₁₅ H ₁₈ N ₂ O ₂ Co ⁺	439
C ₁₄ H ₃₀ Sn ⁺	499	C ₁₅ H ₁₈ N ₂ O ₂ Rh ⁺	491
C ₁₄ H ₃₂ Si ₂ ⁺	300	C ₁₅ H ₁₈ N ₂ O ₂ ⁺	261
C ₁₄ H ₃₆ N ₂ Si ₂ Ge ⁺	452	C ₁₅ H ₁₈ O ₃ Cr ⁺	413
C ₁₄ H ₃₆ N ₂ Si ₂ Pb ⁺	550	C ₁₅ H ₁₈ O ₆ Cl ₃ Co ⁺	441
C ₁₄ H ₃₆ N ₂ Si ₂ Sn ⁺	501	C ₁₅ H ₁₈ O ₆ CoBr ₃ ⁺	480
C ₁₄ H ₃₆ N ₆ O ₂ P ₂ Fe ⁺	434	C ₁₅ H ₁₉ N ₂ O ₁₀ Rh ⁺	491
C ₁₄ H ₃₆ N ₆ O ₂ P ₂ Mo ⁺	488	C ₁₅ H ₂₀ NO ₄ Rh ⁺	491
C ₁₄ H ₃₆ N ₆ O ₂ P ₂ W ⁺	537	C ₁₅ H ₂₀ N ₂ OS ⁺	362
C ₁₄ H ₃₆ N ₃ P ₂ ⁺	325	C ₁₅ H ₂₀ N ₂ O ₂ ⁺	258
C ₁₄ H ₃₈ Si ₄ Ge ⁺	452	C ₁₅ H ₂₁ O ₆ Co ⁺	438
C ₁₄ H ₃₈ Si ₄ Hg ⁺	547	C ₁₅ H ₂₁ O ₆ Cr ⁺	413
C ₁₄ H ₃₈ Si ₄ Pb ⁺	550	C ₁₅ H ₂₁ O ₆ Fe ⁺	432
C ₁₄ H ₃₈ Si ₄ Sn ⁺	501	C ₁₅ H ₂₁ O ₆ Mn ⁺	423
C ₁₅ H ₃ O ₆ F ₁₈ Al ⁺	292	C ₁₅ H ₂₁ O ₆ Rh ⁺	491
C ₁₅ H ₃ O ₆ F ₁₈ Co ⁺	440	C ₁₅ H ₂₂ Si ₂ ⁺	300
C ₁₅ H ₃ O ₆ F ₁₈ Cr ⁺	415	C ₁₅ H ₂₃ Cl ⁺	375
C ₁₅ H ₃ O ₆ F ₁₈ Fe ⁺	433	C ₁₅ H ₂₄ ⁺	107
C ₁₅ H ₃ O ₆ F ₁₈ Ga ⁺	448	C ₁₅ H ₂₄ N ₂ O ₃ ⁺	264
C ₁₅ H ₃ O ₆ F ₁₈ Mn ⁺	423	C ₁₅ H ₂₄ O ₆ Si ₂ Cr ⁺	416
C ₁₅ H ₃ O ₆ F ₁₈ Ru ⁺	490	C ₁₅ H ₂₄ O ₆ Si ₂ Mo ⁺	487
C ₁₅ H ₃ O ₆ F ₁₈ Sc ⁺	405	C ₁₅ H ₂₄ O ₆ Si ₂ W ⁺	536
C ₁₅ H ₃ O ₆ F ₁₈ Ti ⁺	407	C ₁₅ H ₂₄ Si ₂ ⁺	300
C ₁₅ H ₄ O ₆ F ₁₈ V ⁺	408	C ₁₅ H ₂₅ P ⁺	313
C ₁₅ H ₄ O ₆ Co ₂ ⁺	439	C ₁₅ H ₂₇ O ₃ PCr ⁺	416
C ₁₅ H ₉ ⁺	105	C ₁₅ H ₂₈ ⁺	107
C ₁₅ H ₉ N ⁺	144	C ₁₅ H ₂₈ NO ₃ ⁺	263
C ₁₅ H ₁₀ Cl ₂ ⁺	376	C ₁₅ H ₂₈ NO ₃ S ⁺	364
C ₁₅ H ₁₀ N ₂ O ₃ ⁺	264	C ₁₅ H ₂₉ NO ₃ ⁺	263
C ₁₅ H ₁₀ O ⁺	202	C ₁₅ H ₂₉ N ₂ O ₂ ⁺	258
C ₁₅ H ₁₀ O ₂ ⁺	218	C ₁₅ H ₃₀ NO ⁺	240
C ₁₅ H ₁₀ O ₅ W ₂ ⁺	535	C ₁₅ H ₃₀ N ₂ O ₂ ⁺	259
C ₁₅ H ₁₀ S ⁺	336	C ₁₅ H ₃₁ NO ⁺	240
C ₁₅ H ₁₁ ⁺	105	C ₁₅ H ₃₂ Sn ⁺	499
C ₁₅ H ₁₁ N ⁺	144	C ₁₅ H ₃₅ P ₅ ⁺	315
C ₁₅ H ₁₁ NO ⁺	240	C ₁₅ H ₃₆ N ₆ O ₃ P ₂ Cr ⁺	418
C ₁₅ H ₁₁ NO ₂ ⁺	253	C ₁₅ H ₃₆ N ₆ O ₃ P ₂ Fe ⁺	434
C ₁₅ H ₁₁ NO ₃ S ⁺	364	C ₁₅ H ₃₆ N ₆ O ₃ P ₂ Mo ⁺	488
C ₁₅ H ₁₁ O ₂ S ⁺	359	C ₁₅ H ₃₆ N ₆ O ₃ P ₂ W ⁺	537
C ₁₅ H ₁₁ P ⁺	313	C ₁₆ F ₁₀ ⁺	273
C ₁₅ H ₁₂ ⁺	105	C ₁₆ H ₈ ⁺	107
C ₁₅ H ₁₂ O ⁺	202	C ₁₆ H ₈ F ₈ ⁺	278
C ₁₅ H ₁₂ O ₂ ⁺	218	C ₁₆ H ₈ N ₂ ⁺	160
C ₁₅ H ₁₂ O ₆ F ₁₈ Al ⁺	292	C ₁₆ H ₉ ⁺	107
C ₁₅ H ₁₂ O ₆ F ₉ Cr ⁺	415	C ₁₆ H ₁₀ N ₂ O ₂ ⁺	259
C ₁₅ H ₁₂ O ₆ F ₉ Fe ⁺	433	C ₁₆ H ₁₀ O ⁺	203
C ₁₅ H ₁₃ ⁺	106	C ₁₆ H ₁₀ O ₆ W ₂ ⁺	535
C ₁₅ H ₁₃ NO ⁺	240	C ₁₆ H ₁₁ ⁺	108
C ₁₅ H ₁₄ ⁺	106	C ₁₆ H ₁₁ N ₃ O ₄ ⁺	266
C ₁₅ H ₁₄ N ₂ ⁺	160	C ₁₆ H ₁₂ ⁺	108
C ₁₅ H ₁₄ N ₂ O ₂ ⁺	258	C ₁₆ H ₁₂ NCl ⁺	379
C ₁₅ H ₁₄ N ₂ O ₄ ⁺	266	C ₁₆ H ₁₂ N ₂ O ₂ ⁺	259
C ₁₅ H ₁₄ S ⁺	336	C ₁₆ H ₁₂ O ⁺	203
C ₁₅ H ₁₅ ClTh ⁺	553	C ₁₆ H ₁₂ O ₂ ⁺	218
C ₁₅ H ₁₅ ClU ⁺	555	C ₁₆ H ₁₃ ⁺	108
C ₁₅ H ₁₅ La ⁺	522	C ₁₆ H ₁₃ N ⁺	145
C ₁₅ H ₁₅ N ⁺	145	C ₁₆ H ₁₃ NO ⁺	240
C ₁₅ H ₁₅ NO ₂ ⁺	253	C ₁₆ H ₁₃ NO ₂ ⁺	254
C ₁₅ H ₁₅ N ₃ O ₂ ⁺	260	C ₁₆ H ₁₃ NS ₂ ⁺	349
C ₁₅ H ₁₅ Nd ⁺	525	C ₁₆ H ₁₄ ⁺	108
C ₁₅ H ₁₅ O ⁺	203	C ₁₆ H ₁₄ N ₂ O ₂ Co ⁺	439

C ₁₆ H ₁₄ N ₂ O ₂ Cu ⁺	445	C ₁₆ H ₄₄ Si ₄ Zr ⁺	484
C ₁₆ H ₁₄ N ₂ O ₂ Mn ⁺	423	C ₁₇ H ₈ N ₃ ⁺	164
C ₁₆ H ₁₄ N ₂ O ₂ Ni ⁺	443	C ₁₇ H ₉ NO ₄ ⁺	266
C ₁₆ H ₁₄ N ₂ S ₂ ⁺	350	C ₁₇ H ₁ O ₄ Rh ⁺	491
C ₁₆ H ₁₄ O ₂ ⁺	218	C ₁₇ H ₁₂ ⁺	109
C ₁₆ H ₁₄ O ₄ ⁺	223	C ₁₇ H ₁₂ O ⁺	203
C ₁₆ H ₁₄ S ₂ ⁺	339	C ₁₇ H ₁₂ S ₃ ⁺	341
C ₁₆ H ₁₅ N ⁺	145	C ₁₇ H ₁₃ N ⁺	145
C ₁₆ H ₁₅ NS ⁺	345	C ₁₇ H ₁₄ ⁺	109
C ₁₆ H ₁₆ ⁺	108	C ₁₇ H ₁₄ NOCl ⁺	386
C ₁₆ H ₁₆ N ₂ O ₂ ⁺	259	C ₁₇ H ₁₄ O ⁺	203
C ₁₆ H ₁₆ N ₄ Co ⁺	437	C ₁₇ H ₁₅ ⁺	109
C ₁₆ H ₁₆ N ₄ Cu ⁺	444	C ₁₇ H ₁₅ N ⁺	145
C ₁₆ H ₁₆ N ₄ Ni ⁺	442	C ₁₇ H ₁₆ O ₂ ⁺	218
C ₁₆ H ₁₆ O ⁺	203	C ₁₇ H ₁₇ D ₃ O ₂ ⁺	218
C ₁₆ H ₁₆ O ₂ ⁺	218	C ₁₇ H ₁₇ N ₂ OSCl ⁺	401
C ₁₆ H ₁₆ Th ⁺	553	C ₁₇ H ₁₈ N ₂ OS ⁺	362
C ₁₆ H ₁₆ U ⁺	554	C ₁₇ H ₁₈ O ⁺	203
C ₁₆ H ₁₇ N ⁺	145	C ₁₇ H ₁₉ O ₂ ⁺	218
C ₁₆ H ₁₈ ⁺	109	C ₁₇ H ₁₉ Si ⁺	298
C ₁₆ H ₁₈ N ₂ ⁺	160	C ₁₇ H ₁₉ N ⁺	145
C ₁₆ H ₁₈ N ₂ S ⁺	348	C ₁₇ H ₁₉ NO ₃ ⁺	263
C ₁₆ H ₁₈ N ₄ ⁺	166	C ₁₇ H ₁₉ N ₂ SCl ⁺	400
C ₁₆ H ₁₈ O ⁺	203	C ₁₇ H ₂₀ N ₂ ⁺	161
C ₁₆ H ₁₈ OS ⁺	357	C ₁₇ H ₂₀ N ₂ O ⁺	245
C ₁₆ H ₁₈ O ₂ ⁺	218	C ₁₇ H ₂₀ N ₂ S ⁺	348
C ₁₆ H ₁₈ S ⁺	336	C ₁₇ H ₂₀ O ⁺	203
C ₁₆ H ₂₀ ⁺	109	C ₁₇ H ₂₀ OS ⁺	357
C ₁₆ H ₂₀ N ₂ ⁺	160	C ₁₇ H ₂₀ O ₂ ⁺	218
C ₁₆ H ₂₀ N ₂ O ₆ Mo ₂ ⁺	487	C ₁₇ H ₂₀ Si ⁺	298
C ₁₆ H ₂₂ La ⁺	522	C ₁₇ H ₂₂ N ₂ ⁺	161
C ₁₆ H ₂₂ Nd ⁺	525	C ₁₇ H ₂₂ O ₂ ⁺	219
C ₁₆ H ₂₂ PAu ⁺	545	C ₁₇ H ₂₅ N ₂ O ₂ ⁺	259
C ₁₆ H ₂₂ P ₂ I ₂ Pt ⁺	544	C ₁₇ H ₂₅ P ₂ BrPt ⁺	544
C ₁₆ H ₂₂ Pr ⁺	525	C ₁₇ H ₂₅ P ₂ ClPt ⁺	544
C ₁₆ H ₂₂ Si ₂ ⁺	301	C ₁₇ H ₂₅ P ₂ IPt ⁺	544
C ₁₆ H ₂₄ N ₂ ⁺	160	C ₁₇ H ₂₈ Si ₃ ⁺	301
C ₁₆ H ₂₄ N ₂ O ⁺	245	C ₁₇ H ₂₉ N ⁺	145
C ₁₆ H ₂₄ O ₄ S ⁺	359	C ₁₇ H ₂₉ P ⁺	313
C ₁₆ H ₂₆ ⁺	109	C ₁₇ H ₃₂ N ₂ O ₅ ⁺	267
C ₁₆ H ₂₆ O ₄ S ⁺	360	C ₁₇ H ₃₃ N ₂ O ₂ ⁺	259
C ₁₆ H ₂₇ O ₄ PCr ⁺	416	C ₁₈ H ₈ N ₄ ⁺	166
C ₁₆ H ₂₇ O ₄ PW ⁺	536	C ₁₈ H ₁₀ ⁺	109
C ₁₆ H ₂₈ NO ₄ ⁺	265	C ₁₈ H ₁₁ NO ₂ ⁺	254
C ₁₆ H ₂₈ NO ₄ S ⁺	364	C ₁₈ H ₁₂ ⁺	109
C ₁₆ H ₂₈ N ₂ ⁺	161	C ₁₈ H ₁₂ F ₃ P ⁺	322
C ₁₆ H ₂₈ N ₄ ⁺	166	C ₁₈ H ₁₂ O ⁺	203
C ₁₆ H ₂₉ N ₂ O ₃ ⁺	264	C ₁₈ H ₁₂ PCl ₃ ⁺	397
C ₁₆ H ₂₉ N ₂ O ₄ ⁺	266	C ₁₈ H ₁₄ ⁺	110
C ₁₆ H ₃₀ N ₂ O ₃ P ₂ Cr ⁺	418	C ₁₈ H ₁₅ Al ⁺	291
C ₁₆ H ₃₀ Si ₂ ⁺	301	C ₁₈ H ₁₅ As ⁺	455
C ₁₆ H ₃₂ Si ₃ ⁺	301	C ₁₈ H ₁₅ Bi ⁺	552
C ₁₆ H ₃₄ N ₂ ⁺	161	C ₁₈ H ₁₅ ClSn ⁺	503
C ₁₆ H ₃₆ N ₆ O ₄ P ₂ Cr ⁺	418	C ₁₈ H ₁₅ Ga ⁺	448
C ₁₆ H ₃₆ N ₆ O ₄ P ₂ Mo ⁺	488	C ₁₈ H ₁₅ GeBr ⁺	481
C ₁₆ H ₃₆ N ₆ O ₄ P ₂ W ⁺	537	C ₁₈ H ₁₅ N ⁺	145
C ₁₆ H ₃₆ P ₄ ⁺	314	C ₁₈ H ₁₅ O ₃ P ⁺	319
C ₁₆ H ₃₆ Si ₇ ⁺	303	C ₁₈ H ₁₅ P ⁺	313
C ₁₆ H ₃₆ Sn ⁺	499	C ₁₈ H ₁₅ PCr ⁺	416
C ₁₆ H ₄₀ N ₄ Cr ⁺	410	C ₁₈ H ₁₅ Sb ⁺	505
C ₁₆ H ₄₀ N ₄ Hf ⁺	531	C ₁₈ H ₁₅ Si ⁺	298
C ₁₆ H ₄₀ N ₄ Mo ⁺	486	C ₁₈ H ₁₅ Sn ⁺	500
C ₁₆ H ₄₀ N ₄ Ti ⁺	406	C ₁₈ H ₁₆ ⁺	110
C ₁₆ H ₄₀ N ₄ Zr ⁺	484	C ₁₈ H ₁₆ Ge ⁺	450
C ₁₆ H ₄₄ OSi ₄ Re ⁺	542	C ₁₈ H ₁₆ NP ⁺	315
C ₁₆ H ₄₄ Si ₄ Cr ⁺	415	C ₁₈ H ₁₆ N ₂ O ₂ ⁺	259
C ₁₆ H ₄₄ Si ₄ Hf ⁺	531	C ₁₈ H ₁₆ N ₂ O ₂ S ₄ Fe ⁺	436
C ₁₆ H ₄₄ Si ₄ Pb ⁺	550	C ₁₈ H ₁₆ O ⁺	203
C ₁₆ H ₄₄ Si ₄ Sn ⁺	501	C ₁₈ H ₁₆ S ⁺	336
C ₁₆ H ₄₄ Si ₄ Ti ⁺	407	C ₁₈ H ₁₆ Si ⁺	299

C ₂₅ H ₂₂ PSMn ⁺	428	C ₃₆ H ₄₄ N ₄ Cu ⁺	445
C ₂₅ H ₂₂ SMnAs ⁺	457	C ₃₆ H ₄₄ N ₄ Fe ⁺	430
C ₂₅ H ₂₂ SMnSb ⁺	505	C ₃₆ H ₄₄ N ₄ Mg ⁺	290
C ₂₅ H ₂₃ O ₂ P ⁺	318	C ₃₆ H ₄₄ N ₄ Ni ⁺	442
C ₂₅ H ₂₃ P ⁺	314	C ₃₆ H ₄₄ N ₄ Pd ⁺	492
C ₂₅ H ₂₅ N ₂ I ⁺	512	C ₃₆ H ₄₄ N ₄ Zn ⁺	446
C ₂₅ H ₃₇ N ₃ O ₄ ⁺	267	C ₃₆ H ₄₆ N ₄ ⁺	166
C ₂₅ H ₃₇ N ₃ O ₃ ⁺	265	C ₃₆ H ₁₆ ⁺	119
C ₂₆ H ₁₄ ⁺	115	C ₃₆ H ₁₆ ⁺	119
C ₂₆ H ₁₆ ⁺	115	C ₃₆ H ₂₀ ⁺	119
C ₂₆ H ₃₂ OPSMn ⁺	428	C ₃₆ H ₂₂ ⁺	119
C ₂₆ H ₃₂ OSMnAs ⁺	457	C ₃₆ H ₅₆ ⁺	119
C ₂₆ H ₃₂ OSMnSb ⁺	505	C ₄₀ H ₂₀ ⁺	119
C ₂₆ H ₂₂ O ₂ MnAs ⁺	457	C ₄₀ H ₃₀ O ₄ P ₂ Mo ⁺	488
C ₂₆ H ₂₂ O ₂ MnSb ⁺	505	C ₄₀ H ₃₀ O ₄ P ₂ W ⁺	537
C ₂₆ H ₂₂ O ₂ PMn ⁺	424	C ₄₀ H ₅₆ ⁺	119
C ₂₆ H ₂₃ O ₃ PCr ⁺	417	C ₄₂ H ₁₈ ⁺	119
C ₂₆ H ₂₄ ⁺	116	C ₄₂ H ₂₀ ⁺	120
C ₂₆ H ₂₁ N ₂ ⁺	161	C ₄₂ H ₂₂ ⁺	120
C ₂₆ H ₂₆ Si ₂ ⁺	301	C ₄₂ H ₂₄ ⁺	120
C ₂₆ H ₃₂ Si ₃ ⁺	302	C ₄₂ H ₃₀ ⁺	120
C ₂₆ H ₃₇ N ₃ O ₄ ⁺	267	C ₄₄ H ₂₀ ⁺	120
C ₂₆ H ₄₀ N ₂ S ⁺	364	C ₄₄ H ₂₈ N ₄ ClFe ⁺	436
C ₂₆ H ₃₈ Si ₆ ⁺	302	C ₄₄ H ₂₈ N ₄ ClMn ⁺	428
C ₂₇ H ₂₃ O ₃ PCr ⁺	419	C ₄₄ H ₂₈ N ₄ Co ⁺	438
C ₂₇ H ₂₇ NFP ⁺	322	C ₄₄ H ₂₈ N ₄ Cu ⁺	445
C ₂₇ H ₃₃ P ⁺	314	C ₄₄ H ₂₈ N ₄ Fe ⁺	430
C ₂₇ H ₃₉ Si ₄ P ⁺	325	C ₄₄ H ₂₈ N ₄ Mg ⁺	290
C ₂₇ H ₄₀ N ₄ O ₃ S ⁺	364	C ₄₄ H ₂₈ N ₄ Mn ⁺	422
C ₂₈ H ₁₄ ⁺	116	C ₄₄ H ₂₈ N ₄ Ni ⁺	442
C ₂₈ H ₁₆ ⁺	116	C ₄₄ H ₂₈ N ₄ Pb ⁺	550
C ₂₈ H ₂₀ ⁺	117	C ₄₄ H ₂₈ N ₄ Zn ⁺	447
C ₂₈ H ₂₀ N ₂ O ₂ S ₄ Fe ⁺	436	C ₄₄ H ₃₀ N ₄ ⁺	166
C ₂₈ H ₂₃ O ₆ PCr ⁺	419	C ₄₅ H ₃₁ PCl ⁺	396
C ₂₈ H ₂₃ O ₆ PCr ⁺	417	C ₄₆ H ₂₀ ⁺	120
C ₂₈ H ₃₄ ⁺	117	C ₄₈ H ₂₄ ⁺	120
C ₂₉ H ₂₅ P ⁺	314	C ₅₀ H ₂₈ ⁺	120
C ₂₉ H ₃₅ N ₂ I ⁺	512	C ₅₄ H ₃₀ ⁺	120
C ₃₀ H ₁₄ ⁺	117	C ₅₅ H ₇₂ N ₄ O ₃ Mg ⁺	290
C ₃₀ H ₁₆ ⁺	117	C ₅₈ H ₃₂ ⁺	120
C ₃₀ H ₁₈ ⁺	117	Ca ⁺	405
C ₃₀ H ₂₀ S ₂ ⁺	339	Ca ⁺²	405
C ₃₀ H ₃₂ N ₄ ⁺	166	CaBr ⁺	480
C ₃₀ H ₃₉ P ⁺	314	CaI ⁺	515
C ₃₀ H ₄₅ N ₅ O ₆ ⁺	267	CaI ₂ ⁺	515
C ₃₀ H ₆₀ Si ₆ ⁺	303	Cd ⁺	495
C ₃₁ H ₃₇ N ₂ P ⁺	316	CdI ₂ ⁺	518
C ₃₂ H ₁₄ ⁺	118	Ce ⁺	523
C ₃₂ H ₁₆ ⁺	118	CeAu ⁺	545
C ₃₂ H ₁₆ N ₈ Co ⁺	438	CeIr ⁺	543
C ₃₂ H ₁₆ N ₈ Cu ⁺	445	CePt ⁺	544
C ₃₂ H ₁₆ N ₈ Fe ⁺	430	Ce ₂ ⁺	523
C ₃₂ H ₁₆ N ₈ Mn ⁺	422	Cf ⁺	555
C ₃₂ H ₁₆ N ₈ Ni ⁺	442	Cl ⁺	370
C ₃₂ H ₁₆ N ₈ Zn ⁺	447	Cl ⁺²	370
C ₃₂ H ₁₈ ⁺	118	ClAg ⁺	493
C ₃₂ H ₁₈ N ₈ ⁺	167	ClAg ₂ ⁺	493
C ₃₂ H ₂₁ NCl ⁺	379	ClAg ₃ ⁺	493
C ₃₃ H ₂₀ N ₂ Cl ⁺	380	ClAs ⁺	457
C ₃₃ H ₅₇ O ₆ Fe ⁺	432	ClBa ⁺	521
C ₃₄ H ₁₆ ⁺	118	ClBrSn ⁺	504
C ₃₄ H ₁₈ ⁺	118	ClBr ₂ Ag ₃ ⁺	494
C ₃₄ H ₂₀ ⁺	118	ClBr ₃ Sn ⁺	504
C ₃₅ H ₂₇ P ⁺	314	ClCo ⁺	440
C ₃₆ H ₁₆ ⁺	119	ClCs ⁺	520
C ₃₆ H ₁₈ ⁺	119	ClCu ⁺	446
C ₃₆ H ₂₀ ⁺	119	ClCu ₂ ⁺	446
C ₃₆ H ₂₃ NCl ⁺	379	ClGd ⁺	528
C ₃₆ H ₃₀ Si ₂ ⁺	301	ClI ⁺	515
C ₃₆ H ₄₄ N ₄ Co ⁺	437	ClIn ⁺	496

CIK ⁺	405	Cl ₄ Mo ⁺	489
CIMo ⁺	489	Cl ₄ Nb ⁺	485
CINb ⁺	485	Cl ₄ Ta ⁺	532
CINd ⁺	526	Cl ₄ Th ⁺	553
CIRb ⁺	482	Cl ₄ Ti ⁺	407
CISr ⁺	483	Cl ₄ U ⁺	555
CITl ⁺	548	Cl ₄ W ⁺	538
CIV ⁺	408	Cl ₄ Zr ⁺	484
ClW ⁺	538	Cl ₅ Cu ₅ ⁺	446
ClYb ⁺	530	Cl ₅ Mo ⁺	489
ClZr ⁺	484	Cl ₅ Nb ⁺	485
Cl ₂ ⁺	370	Cl ₅ Re ⁺	542
Cl ₂ Ag ₂ ⁺	493	Cl ₅ Ta ⁺	532
Cl ₂ Ag ₃ ⁺	493	Cl ₅ W ⁺	538
Cl ₂ As ⁺	457	Cl ₆ Ga ₂ ⁺	448
Cl ₂ Ba ⁺	522	Cl ₆ W ⁺	538
Cl ₂ BrAg ₃ ⁺	494	Cl ₆ Re ₃ ⁺	542
Cl ₂ Ca ⁺	405	Cm ⁺	555
Cl ₂ Cd ⁺	495	Co ⁺	437
Cl ₂ Co ⁺	441	Cr ⁺	409
Cl ₂ Cr ⁺	420	Cs ⁺	520
Cl ₂ Cs ₂ ⁺	521	CsAu ⁺	545
Cl ₂ CuAg ₂ ⁺	494	Cs ²⁺	520
Cl ₂ Cu ₂ ⁺	446	Cs ₂ ⁺	520
Cl ₂ Cu ₂ Ag ⁺	494	Cu ⁺	444
Cl ₂ Cu ₃ ⁺	446	CuDy ⁺	529
Cl ₂ Fe ⁺	436	CuGe ⁺	454
Cl ₂ Gd ⁺	528	CuHo ⁺	529
Cl ₂ Hg ⁺	547	CuSn ⁺	503
Cl ₂ K ₂ ⁺	405	CuTb ⁺	528
Cl ₂ Mn ⁺	428	Cu ₂ ⁺	444
Cl ₂ Mo ⁺	489	Cu ₂ Sn ⁺	503
Cl ₂ Nb ⁺	485	Cu ₃ Br ₃ ⁺	480
Cl ₂ Nd ⁺	526	Cu ₃ I ₃ ⁺	516
Cl ₂ Ni ⁺	444	Cu ₄ Br ₃ ⁺	480
Cl ₂ Pb ⁺	551	Cu ₄ Br ₄ ⁺	480
Cl ₂ Rb ₂ ⁺	482	D ⁺	42
Cl ₂ Se ⁺	462	DBr ⁺	463
Cl ₂ Se ₂ ⁺	462	DF ⁺	269
Cl ₂ Sn ⁺	503	DI ⁺	508
Cl ₂ Sr ⁺	483	DLi ⁺	42
Cl ₂ Ta ⁺	532	DO ⁺	171
Cl ₂ V ⁺	408	D ₂ ⁺	42
Cl ₂ W ⁺	538	D ₂ N ⁺	123
Cl ₂ Yb ⁺	530	D ₂ N ₂ ⁺	124
Cl ₂ Zn ⁺	447	D ₂ O ⁺	172
Cl ₂ Zr ⁺	484	D ₂ Si ⁺	292
Cl ₃ Ag ₃ ⁺	494	D ₃ N ⁺	123
Cl ₃ Ag ₄ ⁺	494	D ₃ Si ⁺	292
Cl ₃ As ⁺	457	Dy ⁺	528
Cl ₃ CuAg ₂ ⁺	494	Er ⁺	529
Cl ₃ Cu ₂ Ag ⁺	494	Eu ⁺	526
Cl ₃ Cu ₃ ⁺	446	Eu ⁺²	527
Cl ₃ Cu ₄ ⁺	446	Eu ₂ ⁺	527
Cl ₃ Ga ⁺	448	F ⁺	268
Cl ₃ Ge ⁺	453	FAg ⁺	493
Cl ₃ In ⁺	496	FAI ⁺	292
Cl ₃ Mo ⁺	489	FAs ⁺	456
Cl ₃ Nb ⁺	485	FBr ⁺	473
Cl ₃ Nd ⁺	526	FCe ⁺	523
Cl ₃ Sb ⁺	505	FCl ⁺	389
Cl ₃ Ta ⁺	532	FCr ⁺	415
Cl ₃ V ⁺	408	FCs ⁺	520
Cl ₃ W ⁺	538	FGa ⁺	448
Cl ₃ Zr ⁺	484	FGe ⁺	451
Cl ₄ Ag ₃ ⁺	494	FI ⁺	513
Cl ₄ Cu ₂ ⁺	446	FK ₂ ⁺	404
Cl ₄ Cu ₃ ⁺	446	FKr ⁺	482
Cl ₄ Ge ⁺	453	FLa ⁺	523
Cl ₄ Hf ⁺	531	FMn ⁺	423

FMo ⁺	487	F ₁ S ⁺	365
FN ₃ ⁺	290	F ₁ SW ⁺	538
FP ⁺	320	F ₁ Si ⁺	307
FPSBr ₂ ⁺	479	F ₁ U ⁺	554
FS ⁺	365	F ₁ W ⁺	536
FS ₂ ⁺	365	F ₁ Xe ⁺	519
FTI ⁺	548	F ₂ Br ⁺	473
FTI ₂ ⁺	548	F ₂ Ce ₂ ⁺	524
FV ⁺	408	F ₂ Ga ₂ ⁺	448
FW ⁺	536	F ₂ I ⁺	514
F ₂ ⁺	268	F ₂ La ₂ ⁺	523
F ₂ ³⁵ Cl ⁺	389	F ₂ Mo ⁺	487
F ₂ Al ⁺	292	F ₂ P ⁺	320
F ₂ As ⁺	456	F ₂ S ⁺	365
F ₂ Cd ⁺	495	F ₂ ScI ⁺	401
F ₂ Ce ⁺	523	F ₂ U ⁺	554
F ₂ Cr ⁺	415	F ₂ W ⁺	536
F ₂ Ga ⁺	448	F ₃ Mo ⁺	487
F ₂ Ge ⁺	451	F ₃ Re ⁺	541
F ₂ Kr ⁺	482	F ₃ S ⁺	365
F ₂ La ⁺	523	F ₃ Si ₂ ⁺	307
F ₂ Mn ⁺	423	F ₃ U ⁺	554
F ₂ Mo ⁺	487	F ₄ Xe ⁺	519
F ₂ P ⁺	320	F ₄ Re ⁺	541
F ₂ PBr ⁺	477	F ₄ U ⁺	492
F ₂ PCl ⁺	398	F ₁₂ P ₄ Cl ₂ Rh ₂ ⁺	444
F ₂ PI ⁺	515	F ₁₂ P ₄ Ni ⁺	492
F ₂ PSBr ⁺	479	F ₁₂ P ₄ Pd ⁺	543
F ₂ S ⁺	365	F ₁₂ P ₄ Pt ⁺	434
F ₂ SW ⁺	538	F ₁₂ P ₅ Fe ⁺	490
F ₂ S ₂ ⁺	365	F ₁₅ P ₅ Ru ⁺	418
F ₂ S ₂ W ⁺	538	F ₁₈ P ₆ Cr ⁺	488
F ₂ Se ⁺	461	F ₁₈ P ₆ Mo ⁺	538
F ₂ Si ⁺	307	F ₁₈ P ₆ W ⁺	447
F ₂ Sn ⁺	501	Fe ⁺	428
F ₂ Tl ₂ ⁺	548	Fe ²⁺	429
F ₂ V ⁺	408	Fe ₂ ⁺	429
F ₂ W ⁺	536	Ga ⁺	447
F ₂ Xe ⁺	519	GaBi ⁺	552
F ₂ Zn ⁺	447	GaBr ₃ ⁺	481
F ₃ As ⁺	456	Gal ⁺	517
F ₃ Bi ⁺	552	Gal ₃ ⁺	517
F ₃ Br ⁺	473	GaSb ⁺	506
F ₃ Ce ⁺	523	Ga ₂ ⁺	447
F ₃ Cl ⁺	389	Gd ⁺	527
F ₃ Cr ⁺	415	Ge ⁺	448
F ₃ Ge ⁺	451	GeAu ⁺	545
F ₃ Mn ⁺	423	GeI ₄ ⁺	517
F ₃ Mo ⁺	487	GeSe ⁺	462
F ₃ P ⁺	320	GeTe ⁺	508
F ₃ PCo ⁺	440	Ge ₂ ⁺	449
F ₃ PS ⁺	370	H ⁺	42
F ₃ S ⁺	365	HB ⁺	43
F ₃ SW ⁺	538	HBNF ⁺	279
F ₃ Sb ⁺	505	HBNF ₄ P ⁺	322
F ₃ Si ⁺	307	HBO ₄ ⁺	173
F ₃ SiBr ⁺	476	HBS ⁺	327
F ₃ SiCl ⁺	396	HBe ⁺	43
F ₃ SiPCl ₃ Co ⁺	441	HBr ⁺	463
F ₃ V ⁺	408	HCa ⁺	405
F ₃ W ⁺	536	HCl ⁺	370
F ₄ AlCs ⁺	520	HD ⁺	42
F ₄ AlK ⁺	405	HDN ⁺	123
F ₄ As ⁺	456	HDO ⁺	172
F ₄ Bi ⁺	552	HF ⁺	268
F ₄ Ge ⁺	451	HF ₃ P ⁺	320
F ₄ Ge ₂ ⁺	451	HF ₃ Si ⁺	307
F ₄ Mn ⁺	423	HF ₁₂ P ₄ Co ⁺	440
F ₄ Mo ⁺	487	HF ₁₂ P ₄ Ir ⁺	543
F ₄ P ₂	320	HF ₁₂ P ₄ Rh ⁺	491
		HF ₁₅ P ₅ Mn ⁺	425

H ₁ ⁺	508	H ₃ GeI ⁺	517
HLi ⁺	42	H ₃ N ⁺	123
HLi ₂ ⁺	43	H ₃ N ⁺²	123
HMn ⁺	421	H ₃ NF ₄ SiP ₂ ⁺	326
HN ⁺	123	H ₃ NO ⁺	226
HNBr ₂ ⁺	468	H ₃ NOSiS ⁺	368
HNCl ₂ ⁺	378	H ₃ N ₂ ⁺	124
HNF ₂ ⁺	279	H ₃ N ₃ Ge ⁺	451
HN ₄ P ₂ ⁺	322	H ₃ N ₃ Si ⁺	303
HN ₆ P ₂ ⁺	322	H ₃ O ⁺	172
HNO ⁺	226	H ₃ P ⁺	310
HNO ₅ ⁺	360	H ₃ S ⁺	327
HNO ₂ ⁺	226	H ₃ Sb ⁺	504
HNO ₃ ⁺	226	H ₃ Si ⁺	292
HN ₂ ⁺	124	H ₃ SiBr ⁺	476
HN ₃ ⁺	124	H ₃ SiCl ⁺	394
HO ⁺	171	H ₃ SiI ⁺	514
HOCl ⁺	382	H ₄ F ₃ ⁺	269
HOF ⁺	283	H ₄ Ge ⁺	449
HO ₂ ⁺	172	H ₄ N ⁺	123
HP ⁺	310	H ₄ NaCl ₄ ⁺	393
HS ⁺	326	H ₄ N ₂ ⁺	124
HSe ⁺	458	H ₄ N ₄ ⁺	124
HSi ⁺	292	H ₄ O ₂ ⁺	172
HSiCl ₃ ⁺	394	H ₄ P ₂ ⁺	310
HTe ⁺	506	H ₄ SGe ⁺	452
H ₂ ⁺	42	H ₄ Si ⁺	292
H ₂ B ⁺	43	H ₄ SiS ⁺	367
H ₂ BNF ⁺	279	H ₄ Sn ⁺	497
H ₂ BNF ₂ ⁺	279	H ₅ B ₃ ⁺	43
H ₂ Br ⁺	463	H ₅ B ₃ F ₃ P ⁺	321
H ₂ Br ₂ ⁺	463	H ₅ N ₂ F ₂ P ⁺	322
H ₂ Cl ⁺	370	H ₅ PGe ⁺	452
H ₂ Cl ₂ ⁺	370	H ₅ SiP ⁺	325
H ₂ Cl ₂ Ge ⁺	453	H ₅ Si ₂ ⁺	293
H ₂ F ⁺	269	H ₆ BN ⁺	124
H ₂ F ₂ Ge ⁺	451	H ₆ B ₃ ⁺	43
H ₂ F ₂ Si ⁺	307	H ₆ B ₃ N ₃ ⁺	124
H ₂ F ₃ SiAs ⁺	456	H ₆ Ge ₂ Se ⁺	462
H ₂ F ₃ SiP ⁺	325	H ₆ Ge ₂ Te ⁺	508
H ₂ F ₁₂ P ₄ Fe ⁺	434	H ₆ NF ₂ SiP ⁺	326
H ₂ GeBr ₂ ⁺	481	H ₆ N ₃ OP ⁺	319
H ₂ GeI ₂ ⁺	517	H ₆ OGe ₂ ⁺	451
H ₂ Li ⁺	42	H ₆ OSi ₂ ⁺	305
H ₂ N ⁺	123	H ₆ SGe ₂ ⁺	452
H ₂ NBr ⁺	468	H ₆ Si ₂ ⁺	293
H ₂ NCl ⁺	377	H ₆ Si ₂ S ⁺	367
H ₂ NF ₂ P ⁺	322	H ₆ Si ₂ Se ⁺	461
H ₂ N ₂ ⁺	124	H ₆ Si ₂ Te ⁺	507
H ₂ O ⁺	171	H ₈ B ₄ ⁺	43
H ₂ O ₂ ⁺	172	H ₈ B ₅ ⁺	43
H ₂ P ⁺	310	H ₈ B ₅ Br ⁺	463
H ₂ S ⁺	326	H ₈ B ₅ Cl ⁺	371
H ₂ S ₂ ⁺	327	H ₈ B ₅ I ⁺	509
H ₂ Se ⁺	458	H ₈ B ₉ SiBr ⁺	477
H ₃ Si ⁺	292	H ₈ Si ₃ ⁺	293
H ₃ SiBr ₂ ⁺	476	H ₁₀ B ₅ ⁺	43
H ₃ SiCl ₂ ⁺	394	H ₁₀ B ₉ S ⁺	327
H ₃ SiI ₂ ⁺	514	H ₁₀ NGe ₃ ⁺	451
H ₃ Te ⁺	506	H ₁₀ NSi ₃ ⁺	303
H ₃ As ⁺	454	H ₁₀ PGe ₃ ⁺	452
H ₃ B ⁺	43	H ₁₀ Si ₃ As ⁺	456
H ₃ BF ₃ P ⁺	320	H ₁₀ Si ₃ P ⁺	325
H ₃ B ₃ N ₃ Cl ₃ ⁺	378	H ₁₀ B ₄ ⁺	43
H ₃ B ₃ N ₃ F ₃ ⁺	279	H ₁₀ B ₆ ⁺	44
H ₃ ClGe ⁺	453	H ₁₀ Si ₄ ⁺	293
H ₃ FGe ⁺	451	H ₁₁ B ₅ ⁺	43
H ₃ FSi ⁺	307	H ₁₁ B ₅ Si ⁺	293
H ₃ F ₂ ⁺	269	H ₁₁ B ₉ S ⁺	327
H ₃ GeBr ⁺	481	H ₁₁ B ₁₁ S ⁺	327

H ₁₂ B ₃ Al ⁺	291	K ⁺	403
H ₁₂ B ₆ ⁺	44	K ₃ ⁺	403
H ₁₂ Si ₅ ⁺	293	K ₄ ⁺	403
H ₁₄ B ₁₀ ⁺	44	K ₅ ⁺	403
H ₁₆ B ₄ Hf ⁺	531	K ₇ ⁺	403
H ₁₆ B ₄ U ⁺	554	K ₈ ⁺	403
H ₁₆ B ₄ Zr ⁺	483	Kr ⁺	481
Hf ⁺	531	Kr ⁺²	481
Hg ⁺	545	KrXe ⁺	520
Hg ₂ ⁺	545	Kr ₂ ⁺	481
Hg ₃ ⁺	545	La ⁺	522
Hg ₄ ⁺	545	LaAu ⁺	545
Hg ₅ ⁺	546	LaIr ⁺	543
Hg ₆ ⁺	546	Li ⁺	42
Hg ₇ ⁺	546	LiBi ⁺	552
Hg ₈ ⁺	546	LiBr ⁺	463
Hg ₉ ⁺	546	LiCH ₃ ⁺	120
Hg ₁₀ ⁺	546	LiCl ⁺	370
Hg ₁₁ ⁺	546	Lil ⁺	509
Hg ₁₂ ⁺	546	LiK ⁺	404
Ho ⁺	529	LiNa ⁺	289
HoAu ⁺	545	LiO ⁺	172
Ho ₂ ⁺	529	LiOSi ⁺	305
I ⁺	508	LiPb ⁺	550
ICe ⁺	524	Li ₂ ⁺	42
ICe ⁺²	524	Li ₂ Br ₂ ⁺	463
ICs ⁺	521	Li ₂ C ₄ H ₀ ⁺	120
IDy ⁺	529	Li ₂ Cl ₂ ⁺	370
IEr ⁺	530	Li ₂ I ₂ ⁺	509
IEu ⁺	527	Li ₂ O ⁺	172
IGd ⁺	528	Li ₃ ⁺	42
IHo ⁺	529	Li ₃ C ⁺	120
INd ⁺	526	Li ₃ Cl ₃ ⁺	371
IPr ⁺	525	Li ₃ C ₄ H ₀ ⁺	121
ISm ⁺	526	Li ₃ C ₈ H ₁₈ ⁺	121
ITb ⁺	528	Li ₃ C ₁₂ H ₂₇ ⁺	121
ITl ⁺	549	Li ₃ C ₁₆ H ₃₆ ⁺	121
ITm ⁺	530	Lu ⁺	531
I ₂ ⁺	508	Mg ⁺	290
I ₂ ⁺²	508	MgCl ₂ ⁺	393
I ₂ Ba ⁺	522	MgI ₂ ⁺	514
I ₂ Ce ⁺	524	Mn ⁺	420
I ₂ Dy ⁺	529	MnI ⁺	516
I ₂ Er ⁺	530	MnSe ⁺	462
I ₂ Eu ⁺	527	Mn ₂ ⁺	421
I ₂ Gd ⁺	528	Mo ⁺	485
I ₂ Ho ⁺	529	Mo ₂ ⁺	485
I ₂ Nd ⁺	526	N ⁺	122
I ₂ Pb ⁺	551	N ⁺²	122
I ₂ Pr ⁺	525	NCe ⁺	523
I ₂ Sm ⁺	526	NCl ₃ ⁺	377
I ₂ Tb ⁺	528	NF ⁺	278
I ₃ Tm ⁺	530	NFS ⁺	366
I ₃ Ce ⁺	524	NF ₂ ⁺	279
I ₃ Dy ⁺	529	NF ₃ ⁺	279
I ₃ Er ⁺	530	NF ₃ S ⁺	366
I ₃ Gd ⁺	528	NF ₅ P ₂ ⁺	322
I ₃ Ho ⁺	529	NF ₆ P ₃ ⁺	322
I ₃ Nd ⁺	526	NGe ₂ ⁺	451
I ₃ Pr ⁺	525	NHf ⁺	531
I ₃ Tb ⁺	528	NO ⁺	225
I ₃ Tm ⁺	530	NO ⁺²	225
I ₄ Hf ⁺	531	NOBr ⁺	472
In ⁺	496	NOCl ⁺	385
InI ⁺	518	NOF ⁺	287
InI ₃ ⁺	519	NOF ₃ ⁺	287
Ir ⁺	543	NOS ₂ ⁺	306
K ⁺	403	NO ₂ ⁺	225
KBr ⁺	480	NO ₂ Cl ⁺	385
KI ⁺	515	NO ₂ F ⁺	287

NO ₃ Cs ⁺	520	OAlSi ⁺	309
NO ₃ Cs ₂ ⁺	520	OAl ₂ ⁺	291
NO ₃ K ⁺	404	OBa ⁺	521
NO ₃ Rb ⁺	482	OBr ⁺	470
NO ₃ Tl ⁺	548	OBrW ⁺	540
NP ⁺	315	OBr ₂ W ⁺	540
NS ⁺	343	OBr ₃ W ⁺	540
NSCl ⁺	400	OBr ₄ W ⁺	540
NS ₂ ⁺	343	OCa ⁺	405
NSiGe ⁺	452	OCe ⁺	523
NSi ₂ ⁺	303	OCl ⁺	382
NV ⁺	407	OClIV ⁺	408
N ₂ ⁺	122	OCl ₂ ⁺	382
N ₂ F ⁺	279	OCl ₂ V ⁺	409
N ₂ F ₂ ⁺	279	OCl ₃ V ⁺	409
N ₂ F ₄ ⁺	279	OCs ₂ ⁺	520
N ₂ O ⁺	225	ODy ⁺	529
N ₂ O ²⁺	225	OEr ⁺	530
N ₂ O ₃ ⁺	226	OEu ⁺	527
N ₂ O ₅ ⁺	226	OEu ₂ ⁺	527
N ₂ O ₆ Cu ⁺	445	OF ⁺	283
N ₂ S ₃ ⁺	343	OFAI ⁺	292
N ₃ Br ⁺	468	OFV ⁺	408
N ₃ Cl ⁺	377	OF ₂ ⁺	283
N ₃ F ₆ P ₃ ⁺	322	OF ₂ Al ⁺	292
N ₃ O ₆ Co ⁺	439	OF ₂ Ge ⁺	452
N ₃ O ₁₀ V ⁺	408	OF ₂ S ⁺	366
N ₃ P ₆ Cl ₆ ⁺	397	OF ₂ V ⁺	408
N ₃ S ₃ ⁺	343	OF ₂ Mo ⁺	487
N ₄ O ₁₂ Ti ⁺	406	OF ₂ P ⁺	324
N ₅ S ₄ ⁺	343	OF ₃ V ⁺	408
Na ⁺	289	OF ₂ P ₂ ⁺	324
NaAg ⁺	493	OF ₂ Xe ⁺	520
NaAu ⁺	544	OF ₃ Re ⁺	541
NaBr ⁺	475	OFe ⁺	430
NaCl ⁺	393	OGd ⁺	527
NaCl ₃ Gd ⁺	528	OGd ₂ ⁺	528
NaI ⁺	514	OGe ⁺	451
NaK ⁺	404	OHf ⁺	531
NaK ₂ ⁺	404	OHo ⁺	529
Na ₂ ⁺	289	OHo ₂ ⁺	529
Na ₂ Cl ₂ ⁺	393	OIn ₂ ⁺	496
Na ₂ K ⁺	404	OK ⁺	404
Na ₂ K ₂ ⁺	404	OK ₂ ⁺	404
Na ₃ ⁺	289	OLa ⁺	522
Na ₃ K ⁺	404	OLu ⁺	531
Na ₄ ⁺	289	OLu ₂ ⁺	531
Na ₄ K ⁺	404	ONa ⁺	290
Na ₅ ⁺	289	ONaP ⁺	324
Na ₅ K ⁺	405	OND ⁺	526
Na ₆ ⁺	289	ONp ⁺	555
Na ₇ ⁺	289	OP ⁺	316
Na ₈ ⁺	289	OPBr ₃ ⁺	477
Na ₉ ⁺	289	OPCI ⁺	397
Na ₁₀ ⁺	289	OPCI ₃ ⁺	397
Na ₁₁ ⁺	289	OPb ⁺	550
Na ₁₂ ⁺	289	OPr ⁺	525
Na ₁₃ ⁺	289	ORe ⁺	540
Na ₁₄ ⁺	289	OS ⁺	351
Nb ⁺	484	OSBr ₂ ⁺	478
Nd ⁺	525	OSBr ₃ ⁺	479
Ne ⁺	288	OSCl ₂ ⁺	401
Ne ⁺²	289	OSCl ₃ ⁺	401
Ni ⁺	441	OSU ⁺	555
No ⁺	555	OS ₂ ⁺	351
Np ⁺	555	¹⁸ OSi ⁺	352
O ⁺	170	OSm ⁺	526
O ⁺²	170	OSn ⁺	500
OAl ⁺	291	OSr ⁺	483
OAlCl ⁺	393	OTa ⁺	532

O _{Tb} ⁺	528	O ₃ W ₂ ⁺	533
O _{Tb} ₂ ⁺	528	O ₄ BaRe ⁺	542
O _{Te} ⁺	507	O ₄ CsRe ⁺	542
O _{Th} ⁺	553	O ₄ KRe ⁺	542
O _{Ti} ⁺	406	O ₄ MoCs ₂ ⁺	521
O _{Tl} ⁺	548	O ₄ NaRe ⁺	541
O _{Tl} ₂ ⁺	548	O ₄ Na ₂ Mo ⁺	487
O _{Tm} ⁺	530	O ₄ Os ⁺	542
O _U ⁺	554	O ₄ P ₂ ⁺	317
O _V ⁺	408	O ₄ RbRe ⁺	542
O _Y ⁺	483	O ₄ ReTl ⁺	549
O _{Yb} ⁺	530	O ₄ Ru ⁺	490
O _{Zr} ⁺	484	O ₅ STl ₂ ⁺	548
O ₂ ⁺	170	O ₄ W ₂ ⁺	533
O ₂ ⁺²	171	O ₅ P ₂ ⁺	317
O ₂ Al ⁺	291	O ₆ Re ₂ ⁺	541
O ₂ Al ₂ ⁺	291	O ₅ VW ⁺	539
O ₂ BrW ⁺	540	O ₅ W ₂ ⁺	533
O ₂ Br ₂ W ⁺	540	O ₆ As ₄ ⁺	455
O ₂ Ce ⁺	523	O ₆ P ₃ ⁺	317
O ₂ Ce ₂ ⁺	*	O ₆ P ₄ ⁺	317
O ₂ Cl ⁺	382	O ₆ Re ₂ ⁺	541
O ₂ Cl ₂ Cr ⁺	420	O ₆ Sb ₄ ⁺	505
O ₂ Cl ₃ Mo ⁺	489	O ₆ W ₂ ⁺	533
O ₂ Eu ₂ ⁺	527	O ₇ P ₃ ⁺	317
O ₂ FS ⁺	366	O ₇ P ₄ ⁺	317
O ₂ F ₂ S ⁺	366	O ₇ Re ₂ ⁺	541
O ₂ Fe ⁺	430	O ₈ P ₄ ⁺	317
O ₂ Gd ⁺	528	O ₈ VW ₂ ⁺	539
O ₂ Gd ₂ ⁺	528	O ₈ V ₄ ⁺	408
O ₂ Ge ₂ ⁺	451	O ₈ W ₃ ⁺	533
O ₂ Hf ⁺	531	O ₉ P ₄ ⁺	317
O ₂ Ho ₂ ⁺	529	O ₉ V ₂ W ₂ ⁺	539
O ₂ IW ⁺	540	O ₉ W ₃ ⁺	533
O ₂ I ₂ W ⁺	540	O ₁₀ P ₄ ⁺	317
O ₂ NaP ⁺	324	O ₁₀ V ₂ W ₂ ⁺	539
O ₂ P ⁺	317	O ₁₀ V ₃ W ⁺	539
O ₂ PAg ⁺	493	O ₁₀ V ₄ ⁺	408
O ₂ Pb ⁺	550	O ₁₁ VW ₃ ⁺	539
O ₂ Re ⁺	540	O ₁₁ W ₄ ⁺	533
O ₂ S ⁺	351	O ₁₂ W ₄ ⁺	533
O ₂ SCl ⁺	401	O ₁₃ V ₂ W ₁₃ ⁺	539
O ₂ SCl ₂ ⁺	401	O ₁₃ V ₃ W ₂ ⁺	539
O ₂ SFCI ⁺	401	Os ⁺	542
O ₂ Se ⁺	460	P ⁺	309
O ₂ Ta ⁺	532	PA ₈ ⁺	456
O ₂ Tb ₂ ⁺	528	PA ₈ ₃ ⁺	457
O ₂ Te ⁺	507	PBr ⁺	477
O ₂ Th ⁺	553	PBr ₂ ⁺	477
O ₂ Ti ⁺	406	PBr ₃ ⁺	477
O ₂ U ⁺	554	PCI ⁺	396
O ₂ V ⁺	408	PClBr ⁺	479
O ₂ W ⁺	533	PClBr ₂ ⁺	480
O ₂ Zr ⁺	484	PCl ₂ ⁺	396
O ₃ ⁺	171	PCl ₂ Br ⁺	479
O ₃ ClMn ⁺	428	PCl ₃ ⁺	396
O ₃ FCI ⁺	392	PCl ₅ ⁺	396
O ₃ FMn ⁺	423	PI ₃ ⁺	514
O ₃ FRe ⁺	541	PS ⁺	368
O ₃ FS ⁺	366	PSBr ₃ ⁺	479
O ₃ IRe ⁺	542	PSCl ₃ ⁺	402
O ₃ MoCs ₂ ⁺	521	PSb ⁺	505
O ₃ NaF ⁺	325	PSe ⁺	461
O ₃ PCs ⁺	520	PTe ⁺	507
O ₃ PK ⁺	405	P ₂ ⁺	309
O ₃ PRb ⁺	482	P ₂ As ₂ ⁺	457
O ₃ P ₂ ⁺	317	P ₂ Rh ⁺	491
O ₃ Re ⁺	541	P ₃ ⁺	310
O ₃ S ⁺	351	P ₃ As ⁺	457
O ₃ U ⁺	554	P ₄ ⁺	310

P ₄ S ⁺	368	Se ₂ ⁺	458
P ₄ S ₂ ⁺	368	Se ₅ ⁺	458
P ₄ S ₃ ⁺	368	Se ₆ ⁺	458
P ₄ S ₄ ⁺	368	Si ⁺	292
P ₄ S ₅ ⁺	368	SiBr ⁺	476
P ₄ S ₆ ⁺	368	SiBr ₂ ⁺	476
P ₄ S ₇ ⁺	368	SiBr ₃ ⁺	476
P ₄ S ₈ ⁺	368	SiBr ₄ ⁺	476
P ₄ S ₉ ⁺	368	SiCl ⁺	393
P ₄ S ₁₀ ⁺	368	SiCl ₂ ⁺	393
P ₄ Se ₃ ⁺	461	SiCl ₂ Co ⁺	441
Pa ⁺	553	SiCl ₃ ⁺	394
Pb ⁺	549	SiCl ₃ Co ⁺	441
Pd ⁺	492	SiCl ₄ ⁺	394
PdCe ⁺	524	SiP ⁺	325
Pm ⁺	526	SiP ₂ ⁺	325
Pr ⁺	524	Si ₂ Cl ₆ ⁺	394
Pt ⁺	543	Si ₂ P ⁺	325
PtTh ⁺	553	Sm ⁺	526
Pu ⁺	555	Sn ⁺	497
Rb ⁺	482	SnI ₄ ⁺	519
Rbf ⁺	517	SnTe ⁺	508
Rb ²⁺	482	Sr ⁺	483
Rb ₂ I ⁺	517	Sr ⁺²	483
Re ⁺	540	SrI ⁺	517
Rh ⁺	490	SrI ₂ ⁺	517
RhCe ⁺	524	Ta ⁺	532
RhLa ⁺	523	Tb ⁺	528
Rh ₂ ⁺	491	TePb ⁺	551
Ru ⁺	490	Te ₂ ⁺	506
RuTh ⁺	553	Te ₃ ⁺	506
RuU ⁺	555	Te ₄ ⁺	506
S ⁺	326	Te ₅ ⁺	506
SBr ₂ ⁺	477	Te ₆ ⁺	506
SCe ⁺	524	Th ⁺	552
SCl ⁺	399	TiBr ₄ ⁺	480
SCl ₂ ⁺	399	TiI ₄ ⁺	515
SEu ⁺	527	TiPt ⁺	544
SEu ₂ ⁺	527	TiRh ⁺	492
SGa ⁺	448	Ti ₂ Rh ⁺	492
SGa ₂ ⁺	448	Tl ⁺	548
SGd ⁺	528	TlBi ⁺	552
SGe ⁺	452	Tl ₂ ⁺	548
SSe ⁺	461	Tm ⁺	530
SSn ⁺	501	U ⁺	553
STi ⁺	407	U ⁺²	553
SV ⁺	408	V ⁺	407
SY ⁺	483	V ⁺⁵	407
S ₂ ⁺	326	W ⁺	532
S ₂ Br ₂ ⁺	477	Xe ⁺	519
S ₂ Ce ⁺	524	Xe ⁺²	519
S ₂ Cl ⁺	399	Xe ₂ ⁺	519
S ₂ Cl ₂ ⁺	399	Y ⁺	483
S ₂ Eu ⁺	527	YRh ⁺	492
S ₂ Eu ₂ ⁺	527	Yb ⁺	530
S ₃ As ₃ ⁺	457	Yb ⁺²	530
S ₃ As ₄ ⁺	457	Yb ₂ ⁺	530
S ₄ As ₄ ⁺	457	Zn ⁺	446
S ₈ ⁺	326	ZnBr ₂ ⁺	481
Sb ⁺	504	ZnI ₂ ⁺	516
Sb ₃ ⁺	519	Zr ⁺	483
Sb ₂ ⁺	504	ZrI ₄ ⁺	518
Sb ₃ ⁺	504		
Sc ⁺	504		
Sc ⁺	405		
ScRh ⁺	492		
Se ⁺	458		
SeBr ₂ ⁺	481		
SeSn ⁺	503		
SeTe ⁺	508		

Table of Ion Energetics Measurements

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
H⁺						
	H ₂ ^{+(2Σ_g⁺)}	1333-74-0	H	18.0±0.2	EI	3799
	CH ₄	74-82-8	CH ₃	21.3±0.3	EI	5205
				24.0±0.5	EI	3521
	C ₂ H ₂	74-86-2		20.6±0.3	EI	4876
	C ₂ H ₆	74-84-0		23.5±0.5	EI	4911
	H ₂ O	7732-18-5	OH	16.95±0.05	EI	5046
			OH(X ² Π)	18.7±0.05	EI	3906
	HCHO	50-00-0	HCO	17.41±0.07	PI	3554
	HF	7664-39-3	F	19.444	PI	3928
D⁺						
	CD ₄	59862-12-3	CD ₃	22.17±0.1	EI	5205
	D ₂ O	7789-20-0	OD	18.75±0.05	PE	4247
			OD(X ² Π)	18.7±0.05	EI	3906
H₂⁺						
	H ₂ (² Σ _g)	1333-74-0	**	15.42589±0.00005	S	3770
			**	15.4	PI	5479
			**	15.38186±0.00031	PE	3531
			**	15.43	PE	4248
			**	15.43	PE	5313
			**	15.5±1	EI	4894
	C ₂ H ₆	74-84-0		35.0±0.5	EI	4911
	HCHO	50-00-0	CO	15.42±0.06	PI	3554
HD⁺						
	HD	13983-20-5	**	15.44477±0.00007	S	3763
	CH ₃ CD ₃	2031-95-0		38.2±0.8	EI	5128
D₂⁺						
	D ₂	7782-39-0	**	15.4667±0.0001	S	5140
	CH ₃ CD ₃	2031-95-0		35.2±0.8	EI	5128
Li⁺						
	Li	7439-93-2	**	5.4	EI	4912
			**	5.5±0.3	EI	5254
	LiF	7789-24-4		~12	EI	3464
	LiCl	7447-41-8	Cl	10.17	PI	5509
Li₂⁺						
	Li ₂	14452-59-6	**	4.96±0.1	S	3768
			**	5.174±0.013	PI	5143
			**	4.86±0.1	EI	4568
			**	4.86±0.1	EI	5164
			**	5.0±0.3	EI	5254
			**	5.0	EI	4912
Li₃⁺						
	Li ₃	12596-47-3	**	4.35±0.2	EI	5164
HLi⁺						
	LiH	7580-67-8	**	7.9±0.3	EI	5254
			**	4.5±0.3	EI	5254
DLi⁺						
	LiD	13587-16-1	**	7.7±0.1	EI	4568
H₂Li⁺						
	LiH ₂	19709-52-5	**	6.14±0.2	EI	5254

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
HLi₂⁺	Li ₂ H	12339-13-8	**	4.5±0.3	EI	5254
Be⁺	Be	7440-41-7	**	9.2±1.0	EI	4113
HBe⁺	BeH	13597-97-2	** **	8.20±0.06 8.21±0.05	S S	4183 5223
B⁺	B	7440-42-8	** ** **	8.29808±0.00002 8.0 8.6±0.4	S EI EI	4182 4483 3468
	H ₂ NBH ₂	14720-35-5		19.2±0.05	EI	4522
HB⁺	H ₂ NBH ₂	14720-35-5	NH ₃	18.0±0.1	EI	4522
H₂B⁺	H ₂ NBH ₂	14720-35-5	NH ₂	17.2±0.2	EI	4522
H₃B⁺	BH ₃	13283-31-3	**	11-12	EI	3441
H₅B₃⁺	B ₃ H ₇	12429-70-8		11.5±0.3	EI	3652
H₆B₃⁺	B ₃ H ₇	12429-70-8	H	11.2±0.3	EI	3652
H₈B₄⁺	B ₄ H ₈	12007-71-5	**	10.9±0.3	EI	3652
H₁₀B₄⁺	B ₄ H ₁₀ (Tetraborane (10))	18283-93-7	**	10.76±0.04	PE	4454
H₈B₅⁺	B ₅ H ₉ 1-B ₅ H ₈ CH ₃ 2-B ₅ H ₈ CH ₃ 1-B ₅ H ₈ C ₂ H ₅ 2-B ₅ H ₈ C ₂ H ₅ 1-B ₅ H ₈ C ₃ H ₇ 1-B ₅ H ₈ Cl 2-B ₅ H ₈ Cl 1-B ₅ H ₈ Br 2-B ₅ H ₈ Br 1-B ₅ H ₈ I 2-B ₅ H ₈ I	19624-22-7 19495-55-7 23753-74-4 23753-61-9 23753-62-0 34692-67-6 19469-13-7 19469-14-8 23753-67-5 23753-64-2 30624-33-0 20199-87-5	H CH ₃ CH ₃ C ₂ H ₅ C ₂ H ₅ C ₃ H ₇ Cl Cl Br Br I I	11.84±0.01 10.45±0.02 10.61±0.05 10.33±0.05 10.31±0.01 10.98±0.01 11.75±0.05 12.20±0.10 11.38±0.05 11.75±0.05 10.70±0.05 10.72±0.05	EI EI EI EI EI EI EI EI EI EI EI EI	3547 3547 3547 3547 3547 3547 3547 3547 3547 3547 3547 3547
H₉B₅⁺	B ₅ H ₉	19624-22-7	** ** ** **	9.90 9.94 9.87±0.02 10.5 (V)	PE PE PE PE	3869 4446 4454 4949
H₁₁B₅	B ₅ H ₁₁ (Pentaborane(11))	18433-84-6	**	10.7 (V)	PE	4949

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\mathbf{H_{10}B_6^+}$	$\mathbf{B_6H_{10}}$ (Hexaborane(10))	2377-80-2	**	9.4 (V)	PE	4949
$\mathbf{H_{12}B_6^+}$	$\mathbf{B_6H_{12}}$ (Hexaborane(12))	12008-19-4	**	10.2 (V)	PE	4949
$\mathbf{H_{14}B_{10}^+}$	$\mathbf{B_{10}H_{14}}$ (Decaborane (14))	17702-41-9	**	9.88 ± 0.03	PE	4454
			**	10.15 (V)	PE	4265
$\mathbf{C^+}$	C	7440-44-0	**	10.5 ± 1.0	EI	3597
			**	10.9 ± 0.4	EI	4206
			**	10.9 ± 0.4	EI	5635
			**	11.2 ± 0.5	EI	4909
			**	11.2 ± 0.5	EI	5169
			**	11.4 ± 1.5	EI	3978
	$\mathbf{CH_4}$	74-82-8		<25.2	EI	3813
	$\mathbf{C_2H_2}$	74-86-2		22.5 ± 0.3	EI	4876
	$\mathbf{C_2H_4}$	74-85-1		24.4	EI	4118
	$\mathbf{CH_2=CD_2}$	6755-54-0		24.4	EI	4197
	$\mathbf{C_2H_6}$	74-84-0		29.6 ± 0.2	EI	4911
(^2P)	CO	630-08-0	$O(^2P)$	20.89	EI	5126
	$\mathbf{CO_2}$	124-38-9	O_2	25 ± 2	PI	5170
			O_2	22.7 ± 0.2	EI	4693
				24.6 ± 1.0	EI	4129
			20	27.8 \pm 0.1	EI	4693
	$\mathbf{CH_3Br}$	74-83-9	$H + H_2 + Br$	22.9 ± 0.5	EI	4533
$\mathbf{C^{+2}}$	$\mathbf{C^+}$	14067-05-1	**	31.0	EI	3489
(^3P)			**	37.3	EI	3489
$\mathbf{C_2^+}$	$\mathbf{C_2}$	12070-15-4	**	10.9 ± 0.4	EI	4206
			**	11.1 ± 0.5	EI	5169
			**	11.1 ± 1.0	EI	3597
	$\mathbf{C_2H_2}$	74-86-2		19.2 ± 0.2	EI	4876
	$\mathbf{C_2H_4}$	74-85-1		24.5	EI	4118
	$\mathbf{CH_2=CD_2}$	6755-54-0		24.5	EI	4197
	$\mathbf{C_2H_6}$	74-84-0		31.5 ± 0.2	EI	4911
$\mathbf{C_3^+}$	$\mathbf{C_3}$	12075-35-3	**	11.1 ± 0.5	EI	5169
$\mathbf{CH^+}$	$\mathbf{CH_4}$	74-82-8	$H_2 + H?$	22.4	EI	3813
	$\mathbf{C_2H_2}$	74-86-2		20.9 ± 0.2	EI	4876
	$\mathbf{C_2H_4}$	74-85-1		22.1	EI	4118
	$\mathbf{CH_2=CD_2}$	6755-54-0		21.9	EI	4197
	$\mathbf{C_2H_6}$	74-84-0		26.7 ± 0.5	EI	4911
	$\mathbf{CH_3Br}$	74-83-9	$H_2 + Br$	21.7 ± 0.3	EI	4533
$\mathbf{CH_2^+}$	$\mathbf{CH_2}$	60528-76-9	**	10.35 ± 0.15	EI	5365
	$\mathbf{CH_4}$	74-82-8	H_2	15.3	EI	3813
	$\mathbf{C_2H_2}$	74-86-2		20.5 ± 0.2	EI	4876
	$\mathbf{C_2H_4}$	74-85-1	CH_2	18.04 ± 0.04	PI	5130
				18.4	EI	4118

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
CH₂⁺						
	CH ₂ =CD ₂	6755-54-0		18.4	EI	4197
	C ₂ H ₆	74-84-0		17.3±0.15	EI	4911
	CH ₃ OH	67-56-1	H ₂ O	14.05±0.05	PI	3554
	CH ₃ CHO	75-07-0		15.08±0.09	PI	4350
	C ₂ H ₄ O (Oxirane)	75-21-8		14.66±0.09	PI	4350
	CH ₂ =CF ₂	75-38-7	CF ₂	16.99±0.02	PI	3930
			CF ₂	17.2±0.1	EI	3539
	CH ₃ Br	74-83-9	HBr	14.7±0.5	EI	4533
CHD⁺						
	CH ₂ =CD ₂	6755-54-0		20.0	EI	4197
CD₂⁺						
	CH ₂ =CD ₂	6755-54-0		18.4	EI	4197
	C ₂ D ₄	683-73-8	CD ₂	18.13±0.07	PI	5130
CH₃⁺						
	CH ₃	2229-07-4	**	9.81±0.02	PE	3717
			**	9.82±0.02 (V)	PE	4614
			**	9.837±0.005	PE	3942
			**	9.86±0.04 (V)	PE	3695
			**	9.86±0.04	PE	3700
			**	9.84±0.05	EI	4714
			**	9.84±0.02	PE	4899
			**	9.840±0.005 (V)	PE	4596
			**	9.6±0.3	EI	4533
	CH ₄	74-82-8	H	14.4	EI	3813
	C ₂ H ₄	74-85-1		19.3	EI	4118
	C ₂ H ₆	74-84-0		14.1±0.1	EI	4911
	CH ₃ C≡CH	74-99-7	C ₂ H	14.6±0.1	EI	3769
			C ₂ H	16.0	EI	3808
	C ₂ H ₅ C≡CH	107-00-6	C ₃ H ₃	15.1	EI	3808
	1-C ₄ H ₈	106-98-9	C ₃ H ₅	14.1	EI	3808
	iso-C ₄ H ₈	115-11-7	C ₃ H ₅	16.4	EI	3808
	(CH ₃) ₃ CC≡CH	917-92-0	C ₅ H ₇	14.7	EI	3808
	(CH ₃) ₃ CCH=CH ₂	558-37-2	C ₅ H ₉	15.4	EI	3808
	CH ₃ NH ₂	74-89-5	NH ₂	14.5	EI	3808
	C ₂ H ₅ NH ₂	75-04-7	CH ₂ NH ₂	15.6	EI	3808
	(CH ₃) ₂ NH	124-40-3	CH ₃ NH	14.8	EI	3808
	(CH ₃) ₃ N	75-50-3	(CH ₃) ₂ N	14.9	EI	3808
	(C ₂ H ₅) ₂ NH	109-89-7	C ₂ H ₅ NHCH ₂	15.4	EI	3808
	(C ₂ H ₅) ₃ N	121-44-8	(C ₂ H ₅) ₂ NCH ₂	16.7	EI	3808
	trans-CH ₃ N=NCH ₃	4143-41-3	CH ₃ +N ₂	11.32±0.05	PI	4342
	CH ₃ OH	67-56-1	OH	13.82±0.04	PI	3554
	CH ₃ CHO	75-07-0	CO+H	14.08±0.05	PI	4350
				14.08	PI	5270
			CO+H	14.11±0.05	PI	4177
	C ₂ H ₄ O (Oxirane)	75-21-8	CO+H	13.06±0.05	PI	4350
	CH ₃ CDO	4122-13-8		14.26	PI	5270
	(CH ₃) ₂ CO	67-64-1		15.61	PE	5066
				15.2	EI	3550
	((CH ₃) ₂ C(CN)NO) ₂	31018-29-8		14.60	EI	4809
	((CH ₃) ₂ C(NO)COCH ₃) ₂	30442-79-6		15.70	EI	4809
	(C ₆ H ₁₁ NO ₂) ₂	68777-99-1		15.50	EI	4809
	((CH ₃) ₂ C(NO)COOCH ₃) ₂	6144-15-6		14.20	EI	4809
	((CH ₃) ₂ C(NO)OOCCH ₃) ₂	68777-98-0		12.80	EI	4809
	((CH ₃) ₂ C(NO ₂)NO) ₂	5275-46-7		14.20	EI	4809
	CH ₃ (NF ₂)CH(NF ₂)CH ₃	15403-25-5		16.4±0.4	EI	3634
	(CH ₃) ₂ C(NF ₂) ₂	19309-63-8		14.7±0.2	EI	3634

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
CH₃⁺						
	(CH ₃ NF ₂) ₂ CH ₂	21298-22-6		14.6±0.3	EI	3634
	(CH ₃ O) ₂ PO	512-56-1		17.90±0.40	EI	3989
	(CH ₃) ₂ SO	67-68-5	CH ₃ SO	13.3±0.3	EI	5311
	(CH ₃ O) ₂ P(CH ₃ S)O	152-20-5		15.20±0.30	EI	3989
	(CH ₃ O) ₂ P(CH ₃ S)S	2953-29-9		14.50±0.40	EI	3989
	(CH ₃) ₂ CClNO	2421-26-3		13.75	EI	4809
	CH ₃ Br	74-83-9	Br	12.80±0.03	PI	4640
			Br	12.8±0.3	EI	4533
	(CH ₃) ₂ CBrNO	7119-91-7		11.95	EI	4809
	CH ₃ I	74-88-4	I	12.25±0.03	PI	4640
			I	12.260±0.013	PI	3524
			I	12.07±0.07	EI	3626
CH₂D⁺						
	CH ₃ CDO	4122-13-8		14.18	PI	5270
CHD₂⁺						
	CH ₂ =CD ₂	6755-54-0		19.5	EI	4197
	CD ₃ CHO	19901-15-6		14.25	PI	5270
CD₃⁺						
	CD ₃	17030-72-7	**	9.831±0.007 (V)	PE	4596
			**	9.5±0.1	EI	4714
	CD ₃ OD	811-98-3	OD	14.88	PI	5174
	CD ₃ CHO	19901-15-6		14.15	PI	5270
CH₄⁺						
	CH ₄	74-82-8	**	12.6	PI	5479
			**	12.51	PE	3645
			**	~12.51	PE	3529
			**	12.6	PE	4623
			**	12.64	PE	3716
			**	13.6 (V)	PE	5084
			**	12.8	EI	3813
			**	12.82±0.02	EI	5513
			**	12.94±0.04	EI	5503
	C ₂ H ₆	74-84-0		20.4±0.3	EI	4911
	CH ₃ CHO	75-07-0	CO	12.61±0.06	PI	4350
				12.61	PI	5270
	C ₂ H ₄ O (Oxirane)	75-21-8	CO	11.79±0.03	PI	4350
CH₃D⁺						
	CH ₃ CDO	4122-13-8		12.76	PI	5270
CHD₃⁺						
	CD ₃ CHO	19901-15-6		12.77	PI	5270
C₂H⁺						
	C ₂ H	2122-48-7	**	11.96±0.05	OTH	3931
			**	11.96±0.05	OTH	3929
	C ₂ H ₂	74-86-2	H	17.36±0.01	PI	3931
				17.45±0.1	EI	4876
	C ₂ H ₄	74-85-1		18.7	EI	4118
	CH ₂ =CD ₂	6755-54-0		18.9	EI	4197
	C ₂ H ₆	74-84-0		25.6±0.2	EI	4911
	CH≡CCN	1070-71-9	CN	18.19±0.04	PI	3929
	CHF ₂ C≡CH	18371-25-0	CHF ₂	16.19±0.02	EI	3769
C₂D⁺						
	C ₂ D ₂	1070-74-2	D	17.44±0.01	PI	3931

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₂H₂⁺						
	C ₂ H ₂	74-86-2	**	11.394±0.005	PI	4069
			**	11.398±0.005	PI	3921
			**	11.40	PE	4048
			**	11.40	PE	5313
			**	11.403±0.0003	PE	4575
			**	11.43 (V)	PE	4750
			**	11.49 (V)	PE	5084
			**	11.4±0.1	EI	4876
			**	~11.3	EI	4658
			**	11.37±0.05	EI	4714
			**	11.4±0.1	EI	5129
	C ₂ H ₄	74-85-1	H ₂	13.14±0.01	PI	5130
			**	13.55	PI	5018
				13.0±0.1	EI	4922
				13.13±0.04	EI	4922
			H ₂	13.1	EI	4118
			H ₂	13.11±0.02	EI	4320
	CH ₂ =CD ₂	6755-54-0		13.1	EI	4197
	trans-CHD=CHD	1517-53-9	D ₂	13.27±0.05	EI	4320
	C ₂ H ₆	74-84-0		14.7±0.1	EI	4911
	CH ₃ C≡CH	74-99-7	CH ₂	15.2±0.1	EI	3769
	C ₃ H ₆	115-07-1	CH ₄	12.92±0.05	PI	4350
	C ₃ H ₆	75-19-4	CH ₄	12.71±0.06	PI	4350
	(Cyclopropane)					
	((CH ₃) ₂ C(CN)NO) ₂	31018-29-8		16.50	EI	4809
	((CH ₃) ₂ C(NO)OOCCH ₃) ₂	68777-98-0		15.90	EI	4809
	C ₂ H ₃ F	75-02-5	HF	13.51±0.02	PI	3930
				13.51	PI	5352
			HF	13.30	PE	4993
	CH ₂ =CF ₂	75-38-7	2F	19.08±0.03	PI	3930
	cis-CHF=CHF	1630-77-9		18.4±0.2	PI	5241
	trans-CHF=CHF	1630-78-0		18.3±0.2	PI	5241
	C ₂ H ₃ Cl	75-01-4	HCl	12.47±0.1	PI	3930
	C ₂ H ₃ Br	593-60-2	HBr	12.5±0.2	PI	5079
C₂HD⁺						
	CH≡CD	XXXXX-XX-X	**	11.25±0.1	EI	4714
	CH ₂ =CD ₂	6755-54-0		13.1	EI	4197
	trans-CHD=CHD	1517-53-9	HD	13.16±0.03	EI	4320
C₂D₂⁺						
	C ₂ D ₂	1070-74-2	**	11.404±0.005	PI	3921
			**	11.20±0.1	EI	4714
	trans-CHD=CHD	1517-53-9	H ₂	13.14±0.06	EI	4320
	C ₂ D ₄	683-73-8	D ₂	13.24±0.01	PI	5130
	C ₂ D ₆	1632-99-1	2D ₂	14.8	PE	3919
C₂H₃⁺						
	C ₂ H ₃	2669-89-8	**	8.7±0.1	OTH	3930
	C ₂ H ₄	74-85-1	H	13.22±0.02	PI	5130
			**	13.55	PI	5018
			H	13.31±0.03	EI	4320
			H	13.52±0.04	EI	5503
			H	13.6	EI	4118
	C ₂ H ₆	74-84-0		14.6±0.1	EI	4911
	C ₃ H ₆	115-07-1	CH ₃	13.20±0.04	PI	4350
			CH ₃	13.78±0.03	EI	5244
	C ₃ H ₆	75-19-4	CH ₃	12.64±0.05	PI	4350
	(Cyclopropane)					
	CH ₂ CHCH ₂ CN	109-75-1		12.90	PI	5201
	C ₆ H ₅ NH	109-97-7		13.60	PI	5201

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₂H₃⁺						
	CH ₃ C(CH ₃)CN	126-98-7		13.20	PI	5201
	C ₃ H ₅ CN (Cyclopropanecarbonitrile)	5500-21-0		12.65	PI	5201
	CH ₃ CHO	75-07-0	OH	14.17±0.13	PI	4350
	C ₂ H ₄ O (Oxirane)	75-21-8	OH	12.92±0.08	PI	4350
	((CH ₃) ₂ C(NO)OOCCH ₃) ₂	68777-98-0		15.30	EI	4809
	((CH ₃) ₂ C(NO ₂)NO) ₂	5275-46-7		14.30	EI	4809
	C ₂ H ₃ F	75-02-5	F	13.84±0.04	PI	3930
			F	13.84	PI	5352
			F	13.85±0.1	PE	4993
	C ₂ H ₃ Cl	75-01-4	Cl	12.48±0.04	PI	3930
			Cl	12.56±0.09	EI	5503
	C ₂ H ₃ Br	593-60-2	Br	11.85±0.1	PI	5079
			Br	12.01±0.13	EI	5503
	(CH ₃) ₂ CBrNO	7119-91-7		14.45	EI	4809
C₂HD₂⁺						
	CH ₂ =CD ₂	6755-54-0	H	13.2	EI	4197
	trans-CHD=CHD	1517-53-9	H	13.56±0.10	EI	4320
C₂D₃⁺						
	C ₂ D ₄	683-73-8	D	13.41±0.02	PI	5130
	C ₂ D ₆	1632-99-1	D ₂ +D	14.8	PE	3919
C₂H₄⁺						
	C ₂ H ₄	74-85-1	**	10.50±0.02	PI	5018
			**	10.507±0.004	PI	4306
			**	10.51	PI	5479
			**	10.517±0.003	PI	5130
			**	10.5 (V)	PE	4225
			**	10.5 (V)	PE	4884
			**	10.50±0.01 (V)	PE	4939
			**	10.51	PE	3649
			**	10.51	PE	3739
			**	10.51	PE	3847
			**	10.51	PE	5408
			**	10.514±0.007	PE	4943
			**	10.515±0.003	PE	3957
			**	10.517±0.002	PE	4494
			**	10.56	PE	3533
			**	10.68 (V)	PE	5084
			**	10.5	EI	4118
			**	~10.5	EI	4671
			**	10.51±0.01	EI	4320
	C ₂ H ₆	74-84-0		12.1±0.1	EI	4911
	C ₃ H ₈	74-98-6	CH ₄	11.52	EI	5284
			CH ₄	11.55	EI	3488
			CH ₄	11.9	EI	3488
	C ₃ H ₄ (=O) (Cyclopropanone)	5009-27-8		10.2±0.1	EI	4689
	C ₄ H ₈ F ₄ (Cyclobutane, 1,1,2,2-tetrafluoro-)	374-12-9	C ₂ F ₄	13.15	EI	4553
C₂H₃D⁺						
	C ₂ H ₃ D	XXXXX-XX-X	**	10.518±0.007	PE	4943
C₂H₂D₂⁺						
	CH ₂ CD ₂	6755-54-0	**	10.529±0.007	PE	4943
			**	10.5	EI	4197
	cis-CHDCHD	2813-62-9	**	10.521±0.007	PE	4943

Table of Ion Energetics Measurements—Continued

Table of Ion Energistics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₃H₃⁺						
	CH≡CCH ₃	74-99-7	H	11.58±0.04	PI	5009
			H	11.2±0.1	PE	5009
			H	11.9±0.1	EI	3769
	CH ₂ =C=CH ₂	463-49-0		11.48±0.02	PI	5050
			H	11.595±0.01	PI	5106
	C ₃ H ₄ (Cyclopropene)	2781-85-3	H	10.59±0.04	PI	5014
			H	10.25±0.1	PE	5014
			H	10.9±0.1	EI	4689
	C ₃ H ₆	115-07-1	H ₂ +H	13.19±0.05	PI	4350
				14.21±0.09	EI	5244
	C ₃ H ₆ (Cyclopropane)	75-19-4	H ₂ +H ⁻	12.1±0.1	PI	4350
			H ₂ +H	12.86±0.1	PI	4350
	C ₂ H ₅ C≡CH	107-00-6	CH ₃	11.7	EI	3808
	C ₆ H ₆ (Benzene)	71-43-2	C ₃ H ₃	13.79	PI	4075
			C ₃ H ₃	16.90	PE	4630
				15.34±0.06	EI	4534
	CH ₂ CHCH ₂ CN	109-75-1		12.10	PI	5201
	CH ₂ C(CH ₃)CN	126-98-7		12.30	PI	5201
	C ₃ H ₅ CN (Cyclopropanecarbonitrile)	5500-21-0		11.80	PI	5201
	C ₄ H ₄ NH (1H-Pyrrole)	109-97-7		12.60	PI	5201
	(CH ₃) ₂ NCH=CHC≡CH	2206-24-8		15.2	EI	3674
	(C ₂ H ₅) ₂ NCH=CHC≡CH	1809-53-6		18.6	EI	3674
	CH ₃ COC≡CH	1423-60-5	CHO	11.55±0.10	PE	5289
	C ₄ H ₄ O (Furan)	110-00-9	CHO	12.10±0.10	PE	5289
	((CH ₃) ₂ C(CN)NO) ₂	31018-29-8		13.90	EI	4809
	((CH ₃) ₂ C(NO)OOCCH ₃) ₂	68777-98-0		14.30	EI	4809
	((CH ₃) ₂ C(NO ₂)NO) ₂	5275-46-7		14.05	EI	4809
	C ₄ H ₄ S (Thiophene)	110-02-1	CHS	13.06±0.05	PE	5283
	CH ₂ ClC≡CH	624-65-7	Cl	11.00	EI	5282
	CH ₃ C≡CCl	7747-84-4	Cl	10.98	EI	5282
	(CH ₃) ₂ CClNO	2421-26-3		14.35	EI	4809
	CH ₂ BrC≡CH	106-96-7	Br	10.88	EI	5282
	CH ₃ C≡CBr	2003-82-9	Br	10.90	EI	5282
	(CH ₃) ₂ CBrNO	7119-91-7		13.80	EI	4809
	CH ₃ C≡CI	624-66-8	I	10.70	EI	5282
	CH ₂ IC≡CH	659-86-9	I	10.50	EI	5282
C₃H₄⁺						
	CH ₃ C≡CH	74-99-7	**	10.37±0.01	PI	5009
			**	10.36 (V)	PE	4847
			**	10.364±0.005	PE	4575
			**	10.37	PE	4048
			**	10.38±0.01	PE	5009
			**	10.54 (V)	PE	5084
			**	10.5±0.1	EI	3769
	CH ₂ =C=CH ₂	463-49-0	**	10.017±0.003	S	3774
			**	9.696±0.002	PE	5050
			**	10. (V)	PE	4931
			**	10.02 (V)	PE	5105
			**	10.07 (V)	PE	4019
			**	9.691±0.004	PI	4807
	C ₃ H ₄ (Cyclopropene)	2781-85-3	**	9.67±0.01	PI	5014
			**	9.668±0.005	PE	5014

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₃H₄⁺						
	C ₃ H ₄	2781-85-3	**	9.67	PE	3727
			**	9.82 (V)	PE	4669
			**	9.86 (V)	PE	3505
			**	9.86 (V)	PE	4267
			**	9.7±0.1	EI	4689
	C ₃ H ₆	115-07-1	H ₂	11.91±0.03	PI	4350
	C ₃ H ₆	75-19-4	H ₂	11.64±0.15	PI	4350
	(Cyclopropane)					
	CH ₂ CHCH ₂ CN	109-75-1		11.50	PI	5201
	CH ₂ C(CH ₃)CN	126-98-7		11.75	PI	5201
	C ₃ H ₅ CN	5500-21-0		11.20	PI	5201
	(Cyclopropanecarbonitrile)					
	C ₃ H ₄ NH	109-97-7		12.00	PI	5201
	(1H-Pyrrole)					
	CH ₃ COC≡CH	1423-60-5	CO	10.68±0.05	PE	5289
	C ₃ H ₄ O	110-00-9	CO	11.60±0.10	PE	5289
	(Furan)					
	((CH ₃) ₂ C(CN)NO) ₂	31018-29-8		12.50	EI	4809
	(C ₆ H ₁₁ NO ₂) ₂	68777-99-1		15.55	EI	4809
	((CH ₃) ₂ C(NO)OOCCH ₃) ₂	68777-98-0		12.00	EI	4809
	((CH ₃) ₂ C(NO ₂)NO) ₂	5275-46-7		14.60	EI	4809
	(CH ₃) ₂ CCINO	2421-26-3		11.95	EI	4809
	(CH ₃) ₂ CBrNO	7119-91-7		11.80	EI	4809
C₃H₅⁺						
	CH ₂ =CHCH ₂	1981-80-2	**	8.13±0.02	PE	4722
			**	8.13±0.02	PE	4898
	C ₃ H ₆	115-07-1	H	11.78	PI	4369
			H	11.88±0.03	PI	4350
			H	11.90±0.05	EI	5244
	C ₃ H ₆	75-19-4	H ⁻	10.74±0.09	PI	4350
	(Cyclopropane)					
			H	11.44±0.05	PI	4350
			H	11.47	PI	4369
	1-C ₄ H ₈	106-98-9	CH ₃	11.8	EI	3808
	iso-C ₄ H ₈	115-11-7	CH ₃	11.8	EI	3808
	C ₄ H ₈	594-11-6	CH ₃	10.9	EI	3493
	(Cyclopropane, methyl-)					
	CH≡C(CH ₂) ₃ CH ₃	698-02-7		14.09±0.05	EI	3585
	CH ₃ C≡CCH ₂ CH ₂ CH ₃	764-35-2		13.9±0.01	EI	3585
	C ₆ H ₁₀	110-83-8		13.68±0.05	EI	3585
	(Cyclohexene)					
	C ₅ H ₈ =CH ₂	1528-30-9	C ₃ H ₅	10.2	EI	5586
	(Cyclopentane, methylene-)					
				14.05±0.05	EI	3585
	C ₅ H ₇ CH ₃	698-89-0		14.90±0.1	EI	3585
	(Cyclopentene, 1-methyl-)					
			C ₄ H ₇	13.7	EI	5586
	(CH ₃) ₂ NCH ₂ CH=CH ₂	2155-94-4	C ₂ H ₆ N	9.55	PI	5543
	((CH ₃) ₂ C(CN)NO) ₂	31018-29-8		10.85	EI	4809
	(C ₆ H ₁₁ NO ₂) ₂	68777-99-1		12.95	EI	4809
	((CH ₃) ₂ C(NO)OOCCH ₃) ₂	68777-98-0		11.80	EI	4809
	((CH ₃) ₂ C(NO ₂)NO) ₂	5275-46-7		11.40	EI	4809
	(C ₂ H ₅) ₂ S	352-93-2	CH ₃ SH+H	12.41±0.05	PI	4025
	C ₃ H ₆ S ₂	4829-04-3	S ₂ H	10.8±0.2	EI	3598
	(1,3-Dithiolane)					
	(iso-C ₃ H ₇)SOCH ₃	XXXXX-XX-X		12.4±0.1	EI	5311
	n-C ₃ H ₇ Cl	540-54-5	H ₂ +Cl	12.41	PI	5069
	iso-C ₃ H ₇ Cl	75-29-6	H ₂ +Cl	12.58	PI	5069
	(CH ₃) ₂ CCINO	2421-26-3		11.75	EI	4809
	n-C ₃ H ₇ Br	106-94-5	H ₂ +Br	11.86	PI	5069

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₄H₃⁺						
	C ₄ H ₃	XXXXX-XX-X **		8.31±0.1	EI	4714
	CH ₃ C≡CC=CCH ₃	2809-69-0	C ₂ H ₃	14.05±0.1	PI	5370
	C ₆ H ₆ (Benzene)	71-43-2	H+C ₂ H ₂	18.48±0.07	EI	4534
	(CH ₃) ₂ NCH=CHC≡CH	2206-24-8		14.4	EI	3674
	C ₄ H ₈ NCH=CHC≡CH (Pyrrolidine, 1-(1-buten-3-ynyl)-)	19352-85-3		15.2	EI	3674
	(C ₂ H ₅) ₂ NCH=CHC≡CH	1809-53-6		15.0	EI	3674
C₄H₄⁺						
	H ₂ C=C=C=CH ₂	2873-50-9	**	9.25±0.05	EI	5454
			**	9.15	PE	5034
	CH ₂ =CHC≡CH	689-97-4	**	9.58±0.02	PE	4374
			**	9.63	PE	3997
			**	9.64±0.03 (V)	PE	4538
			**	9.58±0.02	EI	5454
			**	9.9	EI	3767
	HC≡CCH ₂ CH ₂ C≡CH	628-16-0	C ₂ H ₂	10.42±0.08	PI	5454
			C ₂ H ₂	10.47±0.1	EI	5454
	H ₃ CC≡CC=CCH ₃	2809-69-0	C ₂ H ₂	11.27±0.2	PI	5454
	C ₆ H ₆ (Benzene)	71-43-2	C ₂ H ₂	13.85	PI	4075
			C ₂ H ₂	14.17±0.08	PI	5454
			C ₂ H ₂	14.85	PE	4630
			C ₂ H ₂	13.94±0.1	EI	5454
			C ₂ H ₂	14.1	EI	3488
	C ₅ H ₅ N (Pyridine)	110-86-1	HCN	11.8-12.0	PI	5028
			HCN	12.34±0.1	EI	5454
			HCN	13.41±0.05	EI	5413
	(CH ₃) ₂ NCH=CHC≡CH	2206-24-8	CH ₂ =NH+CH ₃	13.4	EI	3674
	C ₄ H ₈ NCH=CHC≡CH (Pyrrolidine, 1-(1-buten-3-ynyl)-)	19352-85-3		13.7	EI	3674
C₄D₄⁺						
	(CD ₂ =C) ₂	25294-38-6	**	9.20	PE	5034
C₄H₅⁺						
	C ₅ H ₈ (Cyclopentene)	142-29-0	CH ₃	11.83	EI	4203
	C ₅ H ₈ (Spiropentane)	157-40-4	CH ₃	10.20	EI	4203
	C ₅ H ₈ =CH ₂ (Cyclopentane, methylene-)	1528-30-9	C ₂ H ₅	9.7	EI	5586
C₄H₆⁺						
	trans-(CH ₂ =CH) ₂	106-99-0	**	9.03	PE	5084
			**	9.0691	S	5199
			**	9.03 (V)	PE	4688
			**	9.18±0.04	EI	4274
	C ₂ H ₅ C≡CH	107-00-6	**	10.178±0.005	PE	4575
	CH ₃ C≡CCH ₃	503-17-3	**	9.562±0.005	PE	4575
			**	9.59	PE	4048
			**	9.61	PE	4160
			**	9.79 (V)	PE	5084
	CH ₂ =C=CHCH ₃	590-19-2	**	9.33 (V)	PE	4019
			**	9.0 (V)	PE	4225
			**	9.03	PE	3847
	C ₄ H ₆ (Bicyclo[1.1.0]butane)	157-33-5	**	19.1±0.1 (V)	PE	4702
	C ₄ H ₆ (Cyclobutene)	822-35-5	**	9.43±0.03 (V)	PE	4828

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₄H₆⁺						
	C ₃ H ₆	822-35-5	**	9.43 (V)	PE	4267
			**	9.59 (V)	PE	4669
	C ₃ H ₄ (=CH ₂) (Cyclopropane, methylene-)	6142-73-0	**	9.57 (V)	PE	4669
	iso-C ₄ H ₈	115-11-7	H ₂	11.3±0.1	EI	5268
	C ₄ H ₈ (Cyclobutane)	287-23-0	H ₂	11.2±0.1	EI	5268
	CH=C(CH ₃) ₂ CH ₃	693-02-7	C ₂ H ₄	11.08±0.05	EI	3585
	CH ₃ C≡CCH ₂ CH ₂ CH ₃	764-35-2	C ₂ H ₄	11.02±0.05	EI	3585
	C ₆ H ₁₀ (Cyclohexene)	110-83-8	C ₂ H ₄	11.91±0.05	EI	3585
	C ₅ H ₈ =CH ₂ (Cyclopentane, methylene-)	1528-30-9	C ₂ H ₄	10.2	EI	5586
			C ₂ H ₄	12.32±0.05	EI	3585
	C ₅ H ₇ CH ₃ (Cyclopentene, 1-methyl-)	693-89-0	C ₂ H ₄	12.33±0.05	EI	3585
	C ₆ H ₁₀ =CH ₂ (Cyclohexane, methylene-)	1192-37-6	C ₃ H ₆	13.2	EI	5586
	C ₆ H ₁₁ Cl (Cyclohexane, chloro-)	542-18-7		11.07±0.03	PI	4078
C₄H₂D₄⁺						
	trans-(CD ₂ =CH) ₂	10545-58-1	**	9.0695	S	5199
C₄D₆⁺						
	trans-(CD ₂ =CD) ₂	1441-56-1	**	9.0698	S	5199
C₄H₇⁺						
	(CH ₃) ₂ C=CHCH ₂	513-35-9	CH ₃	11.33±0.12	EI	3544
	(CH ₃) ₂ CHCH=CH ₂	563-45-1	CH ₃	11.15±0.12	EI	3544
	C ₂ H ₅ C(CH ₃)=CH ₂	563-46-2	CH ₃	11.34±0.07	EI	3544
	1-C ₅ H ₁₀	109-67-1	CH ₃	11.35±0.07	EI	3544
	cis-2-C ₅ H ₁₀	627-20-3	CH ₃	11.24±0.02	EI	3544
	trans-2-C ₅ H ₁₀	646-04-8	CH ₃	11.35±0.03	EI	3544
	C ₅ H ₁₀ (Cyclopentane)	287-92-3	CH ₃	11.36±0.08	EI	3544
	C ₆ H ₁₂ (Cyclohexane)	110-82-7	C ₂ H ₅	11.21±0.04	PI	4078
	C ₆ H ₁₀ =CH ₂ (Cyclohexane, methylene-)	1192-37-6	C ₃ H ₅	13.7	EI	5586
	C ₆ H ₁₁ Cl (Cyclohexane, chloro-)	542-18-7		11.52±0.05	PI	4078
	CH ₂ =CHCH ₂ CH ₂ Br	5162-44-7	Br	10.6	EI	5633
	C ₆ H ₁₁ Br (Cyclohexane, bromo-)	108-85-0		11.54±0.02	PI	4078
C₄H₈⁺						
	1-C ₄ H ₈	106-98-9	**	9.59±0.02	PI	5018
			**	9.625±0.003	PE	3957
			**	9.72 (V)	PE	3950
			**	9.77±0.01 (V)	PE	4939
			**	10.0 (V)	PE	4225
	2-C ₄ H ₈	107-01-7	**	9.13 (V)	PE	5600
	iso-C ₄ H ₈	115-11-7	**	9.21	PE	3533
			**	9.239±0.003	PE	3957
			**	9.39 (V)	PE	4614
			**	9.41 (V)	PE	4669
			**	9.45 (V)	PE	4513
	cis-2-C ₄ H ₈	590-18-1	**	9.11±0.03 (V)	PE	4828
			**	9.11±0.02	PI	5018
			**	9.07	PE	3533

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₄H₈⁺						
	<i>cis</i> -2-C ₄ H ₈	590-18-1	**	9.124±0.005	PE	3957
			**	9.20 (V)	PE	4669
			**	9.29 (V)	PE	4084
			**	9.32±0.01 (V)	PE	4939
			**	9.36 (V)	PE	4513
			**	9.4 (V)	PE	4225
	<i>trans</i> -CH ₃ CH=CHCH ₃	624-64-6	**	9.11 (V)	PE	3649
			**	9.10±0.02	PI	5018
			**	9.09	PE	3533
			**	9.11	PE	4267
			**	9.122±0.005	PE	3957
			**	9.32 (V)	PE	4084
			**	9.37 (V)	PE	4513
			**	9.5 (V)	PE	4225
	C ₄ H ₈ (Cyclobutane)	287-23-0	**	9.92±0.05	PE	3757
			**	10.7±0.1 (V)	PE	4037
	C ₄ H ₈ (Cyclopropane, methyl-)	594-11-6	**	9.9±0.2	EI	3493
	<i>n</i> -C ₅ H ₁₂	109-66-0	CH ₄	11.00	EI	5284
	C ₆ H ₁₂ (Cyclohexane)	110-82-7	C ₂ H ₄	11.08±0.01	PI	4078
			C ₂ H ₄	11.45	EI	4319
	(CH ₃) ₂ CHC ₂ H ₄ CHO	1119-16-0	C ₂ H ₄ O	11.10	EI	5264
	C ₂ H ₅ CH(CH ₃)CH ₂ CHO	15877-57-3	CH ₃ CHO	9.86	EI	4729
			C ₂ H ₄ O	11.10	EI	5264
	<i>n</i> -C ₆ H ₁₁ CHO	66-25-1	C ₂ H ₄ O	10.70	EI	5264
	<i>n</i> -C ₆ H ₁₃ OH	111-27-3		9.89	EI	4729
	C ₆ H ₁₁ Cl (Cyclohexane, chloro-)	542-18-7		10.2±0.01	PI	4078
C₄H₉⁺						
	<i>tert</i> -C ₄ H ₉	1605-73-8	**	6.58±0.01	PE	4634
			**	6.70±0.03	PE	4899
			**	6.95±0.05 (V)	PE	4614
	<i>iso</i> -C ₄ H ₁₀	75-28-5	H	10.68±0.02	PI	5025
			H	10.68±0.03	PI	5345
	<i>neo</i> -C ₅ H ₁₂	463-82-1	CH ₃	10.35	PI	5482
	(<i>tert</i> -C ₄ H ₉) ₄ Li ₄	25395-78-2		11.±0.50	PI	5455
	<i>tert</i> -C ₄ H ₉ NO	917-95-3		8.9±0.1	EI	3654
	C ₆ H ₅ S(<i>tert</i> -C ₄ H ₉) ₂ (Benzene, [(1,1-dimethylethyl)thio]-)	3019-19-0		10.47±0.1	EI	4198
	<i>tert</i> -C ₄ H ₉ Cl	507-20-0	Cl	10.51±0.01	PI	5345
	C ₆ H ₁₁ Cl (Cyclohexane, chloro-)	542-18-7		10.56±0.02	PI	4078
	<i>tert</i> -C ₄ H ₉ SiCl ₃	18171-74-9	SiCl ₃	10.7±0.1	EI	5276
	(CH ₃) ₃ CGe(CH ₃) ₃	1184-91-4	(CH ₃) ₃ Ge	10.19±0.27	EI	3548
	<i>tert</i> -C ₄ H ₉ Br	507-19-7	Br	9.85±0.01	PI	5345
	(<i>tert</i> -C ₄ H ₉) ₂ Sn	3531-47-3	(CH ₃) ₃ Sn	10.03±0.23	EI	3548
	<i>tert</i> -C ₄ H ₉ I	558-17-8	I	8.98±0.01	PI	5345
	(<i>tert</i> -C ₄ H ₉) ₂ Pb	32997-03-8	(CH ₃) ₃ Pb	9.45±0.15	EI	3548
C₄H₁₀⁺						
	<i>n</i> -C ₄ H ₁₀	106-97-8	**	10.6±0.1	PE	4702
			**	11.2 (V)	PE	5084
			**	10.87±0.05	EI	3791
			**	10.89	EI	3538
	<i>iso</i> -C ₄ H ₁₀	75-28-5	**	11.4 (V)	PE	3710
			**	10.74±0.05	EI	3791

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$C_5H_2^+$	$CH \equiv CC \equiv CCH_3$	4911-55-1	H_2	13.7	EI	5404
$C_5H_3^+$	$CH \equiv CC \equiv CCH_3$	4911-55-1	H	11.6 ± 0.2	EI	5404
$C_5H_4^+$	$CH_3C \equiv CC \equiv CH$	4911-55-1	**	9.51	PE	4048
	$1,2,3,4-C_5H_4$	21986-03-8	**	9.51	PE	5404
	$1,2,3,4-C_5H_4$	21986-03-8	**	8.67	PE	4686
$C_5H_5^+$	C_5H_5 (Cyclopentadieny)	XXXXX-XX-X	**	8.41	EI	4545
	$C_6H_5CH_3$ (Benzene, methyl-)	108-88-3	$C_2H_2 + H$	16.4 ± 0.2	EI	4331
	$C_6H_5(CH_3)_2$ (Benzene, 1,4-dimethyl-)	106-42-3	$C_2H_2 + CH_3$	16.3 ± 0.2	EI	4331
	$C_6H_5C_2H_5$ (Benzene, ethyl-)	100-41-4	$C_2H_2 + CH_3$	16.2 ± 0.2	EI	4331
	$C_6H_5C_3H_7$ (Benzene, propyl-)	103-65-1	$C_2H_5 + C_2H_2$	15.5 ± 0.2	EI	4331
	$C_6H_5NH_2$ (Benzanine)	62-53-3	$HCN + H$	15.2 ± 0.2	EI	4331
	C_6H_5OH (Phenol)	108-95-2	$CO + H$	14.2 ± 0.2	EI	4331
	$C_6H_5(NO_2)CH_3$ (Benzene, 1-methyl-2-nitro-)	88-72-2	$HCN + CO + OH$	13.5 ± 0.2	EI	4331
	$C_6H_5(NO_2)CH_3$ (Benzene, 1-methyl-3-nitro-)	99-08-1	$C_2H_2 + NO_2$	14.8	EI	4331
	$C_6H_5(NO_2)CH_3$ (Benzene, 1-methyl-4-nitro-)	99-99-0	$C_2H_2 + NO_2$	15.2 ± 0.2	EI	4331
	$C_6H_5ClCH_3$ (Benzene, 1-chloro-2-methyl-)	95-49-8		15.67 ± 0.015	EI	3777
	$C_6H_5ClCH_3$ (Benzene, 1-chloro-3-methyl-)	108-41-8	$C_2H_2 + Cl$	15.7 ± 0.2	EI	4331
	$C_6H_5ClCH_3$ (Benzene, 1-chloro-4-methyl-)	106-43-4		15.71 ± 0.15	EI	3777
	$C_6H_5ClCH_3$ (Benzene, 1-chloro-4-methyl-)			15.66 ± 0.15	EI	3777
	$C_6H_5BrCH_3$ (Benzene, 1-bromo-2-methyl-)	95-46-5	$C_2H_2 + Cl$	15.7 ± 0.2	EI	4331
	$C_6H_5BrCH_3$ (Benzene, 1-bromo-3-methyl-)	591-17-3	$C_2H_2 + Br$	15.19 ± 0.15	EI	3777
	$C_6H_5BrCH_3$ (Benzene, 1-bromo-4-methyl-)	106-38-7	$C_2H_2 + Br$	15.2 ± 0.2	EI	4331
	$C_6H_5BrCH_3$ (Benzene, 1-bromo-4-methyl-)	106-38-7	$C_2H_2 + Br$	15.2 ± 0.2	EI	4331
	$C_6H_5ICH_3$ (Benzene, 1-iodo-2-methyl-)	615-37-2	$C_2H_2 + I$	15.23 ± 0.15	EI	3777
	$C_6H_5ICH_3$ (Benzene, 1-iodo-3-methyl-)	625-95-6		14.3 ± 0.2	EI	4331
	$C_6H_5ICH_3$ (Benzene, 1-iodo-3-methyl-)			14.34 ± 0.15	EI	3777
	$C_6H_5ICH_3$ (Benzene, 1-iodo-4-methyl-)	624-31-7	$C_2H_2 + I$	14.47 ± 0.15	EI	3777
	$C_6H_5ICH_3$ (Benzene, 1-iodo-4-methyl-)		$C_2H_2 + I$	14.5 ± 0.2	EI	4331
	$C_6H_5ICH_3$ (Benzene, 1-iodo-4-methyl-)		$C_2H_2 + I$	14.66 ± 0.15	EI	3777
	$C_6H_5ICH_3$ (Benzene, 1-iodo-4-methyl-)		$C_2H_2 + I$	14.7 ± 0.2	EI	4331

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₅H₆⁺						
	CH ₂ =C(CH ₃)C≡CH	78-80-8	**	9.23±0.01	PE	5407
			**	9.30±0.03 (V)	PE	4538
			**	10.1	EI	3767
	CH ₂ =CHC≡CCH ₃	646-05-9	**	9.00±0.01	PE	5407
			**	9.06±0.03 (V)	PE	4538
			**	9.4	EI	3767
	CH ₃ CH=CHC≡CH	2206-23-7	**	8.5	EI	3767
	CH ₂ =C=CHCH=CH ₂	10563-01-6	**	8.88 (V)	PE	4397
	cis-CH ₃ CH=CHC≡CH	1574-40-9	**	9.17±0.03 (V)	PE	4538
			**	9.11±0.01	OTH	5407
	trans-CH ₃ CH=CHC≡CH	2004-69-5	**	9.11±0.03 (V)	PE	4538
			**	9.05±0.01	OTH	5407
	C ₅ H ₆ (Bicyclo[2.1.0]pent-2-ene)	5164-35-2	**	8.6 (V)	PE	5621
	C ₅ H ₆ (Cyclopentadiene)	26912-33-4	**	8.56±0.01	EI	3535
	C ₅ H ₆ (1,3-Cyclopentadiene)	542-92-7	**	8.56 (V)	PE	4179
			**	8.6 (V)	PE	4373
			**	8.61 (V)	PE	5535
			**	9.0	EI	3476
	C ₃ H ₅ C≡CH (Cyclopropane, ethynyl-)	6746-94-7	**	9.58 (V)	PE	3997
	C ₇ H ₁₀ (Bicyclo[2.2.1]hept-2-ene)	498-66-8	C ₂ H ₄	9.22±0.01	EI	3535
	C ₇ H ₁₀ (Tricyclo[2.2.1.0 ^{2,6}]heptane)	279-19-6	C ₂ H ₄	9.44±0.01	EI	3535
	C ₆ H ₅ NH ₂ (Benzenamine)	62-53-3		12.04±<0.1	EI	3735
			HCN	12.13±0.06	EI	3784
			HCN	12.77±0.05	EI	5413
	C ₆ H ₅ OH (Phenol)	108-95-2	CO	12.45±0.1	EI	3817
	C ₆ H ₅ SH (Benzenethiol)	108-98-5	CS	12.18±0.1	EI	3817
	C ₇ H ₉ Br (Bicyclo[2.2.1]hept-2-ene, 5-bromo-, endo-)	5810-82-2	C ₂ H ₃ Br	10.0	EI	5633
	C ₇ H ₉ Br (Bicyclo[2.2.1]hept-2-ene, 5-bromo-, exo-)	5889-54-3	C ₂ H ₃ Br	10.0	EI	5633
C₅H₇⁺						
	CH ₂ =CHCHCH=CH ₂	XXXXX-XX-X	**	7.25	EI	4591
	CH≡CC(CH ₃) ₂	XXXXX-XX-X	**	7.44	EI	4591
	CH≡CCHCH ₂ CH ₃	XXXXX-XX-X	**	7.6	OTH	4591
	CH ₂ =CHC(=CH ₂)CH ₂	XXXXX-XX-X	**	7.9	OTH	4591
	C ₅ H ₇ (Cyclopentenyl)	XXXXX-XX-X	**	7.00	EI	4545
	C ₅ H ₇ (Cyclopentenyl)	XXXXX-XX-X	**	7.00	EI	4591
	CH ₂ =C(CH ₃)CH=CH ₂	78-79-5	H	8.85	EI	4591
			H	10.54	EI	4203
	CH ₂ =CHCH=CHCH ₃	504-60-9	H	10.52	EI	4203
	CH ₂ =CHCH ₂ CH=CH ₂	591-93-5	H	9.46	EI	4591
			H	10.23	EI	4203
	trans-CH ₂ =CHCH=CHCH ₃	2004-70-8	H	8.60	EI	4591
	C ₅ H ₈ (Cyclopentene)	142-29-0	H	9.00	EI	4591
			H	10.98	EI	4203
	C ₅ H ₈ (Spiropentane)	157-40-4	H	9.26	EI	4591

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C_5H_7^+						
	C_5H_8	157-40-4	H	9.53±0.03	EI	4203
	$\text{C}_5\text{H}_7\text{D}$ (Cyclopentene-1-d)	37729-44-5	D	11.03±0.03	EI	4203
	C_6H_{10}	XXXXX-XX-X	CH_3	8.45	EI	4591
	$\text{CH}_2=\text{C}(\text{C}_2\text{H}_5)\text{CH}=\text{CH}_2$	XXXXX-XX-X	CH_3	8.81	EI	4591
	C_6H_{10}	XXXXX-XX-X	CH_3	10.06±0.05	EI	5483
	$\text{CH}_2=\text{C}(\text{C}_2\text{H}_5)\text{CH}=\text{CH}_2$	XXXXX-XX-X	CH_3	10.08±0.05	EI	5483
	$\text{CH}_2=(\text{C}(\text{CH}_3))_2=\text{CH}_2$	513-81-5	CH_3	8.66	EI	4591
			CH_3	10.22±0.05	EI	5483
	$\text{CH}_2=\text{CH}(\text{CH}_2)_2\text{CH}=\text{CH}_2$	592-42-7	CH_3	9.29	EI	4591
			CH_3	9.35±0.05	EI	5483
	$\text{CH}\equiv\text{C}(\text{CH}_2)_3\text{CH}_3$	693-02-7	CH_3	10.04±0.05	EI	5483
			CH_3	10.87±0.05	EI	3585
	$\text{CH}_2=\text{C}(\text{CH}_3)\text{CH}_2\text{CH}=\text{CH}_2$	763-30-4	CH_3	9.16	EI	4591
			CH_3	9.40±0.05	EI	5483
	$\text{CH}_3\text{C}\equiv\text{CCH}_2\text{CH}_2\text{CH}_3$	764-35-2	CH_3	<9.93±0.05	EI	5483
			CH_3	10.63±0.05	EI	3585
	$\text{CH}\equiv\text{CC}(\text{CH}_3)_3$	917-92-0	CH_3	9.90±0.05	EI	5483
			CH_3	10.76±0.06	EI	4126
	$\text{CH}\equiv\text{CCH}(\text{CH}_3)\text{C}_2\text{H}_5$	922-59-8	CH_3	9.93±0.05	EI	5483
	C_6H_{10}	926-54-5	CH_3	10.23±0.05	EI	5483
	$\text{CH}_2=\text{CHCH}=\text{C}(\text{CH}_3)_2$	926-56-7	CH_3	10.18±0.05	EI	5483
	$\text{C}_2\text{H}_5\text{C}\equiv\text{CC}_2\text{H}_5$	928-49-4	CH_3	9.88±0.05	EI	5483
	$\text{CH}_2=\text{CHCH}(\text{CH}_3)\text{CH}=\text{CH}_2$	1115-08-8	CH_3	9.54±0.05	EI	5483
	C_6H_{10}	2787-45-3	CH_3	10.16±0.05	EI	5483
	$(\text{CH}_3)_2\text{C}=\text{C}=\text{CHCH}_3$	3043-33-2	CH_3	9.55±0.05	EI	5483
	C_6H_{10}	5194-51-4	CH_3	10.14±0.05	EI	5483
	$\text{CH}\equiv\text{CCH}_2\text{CH}(\text{CH}_3)_2$	7154-75-8	CH_3	10.03±0.05	EI	5483
	$\text{CH}_2=\text{C}=\text{C}(\text{CH}_3)\text{C}_2\text{H}_5$	7417-48-3	CH_3	9.44±0.05	EI	5483
	$\text{CH}_2=\text{C}=\text{CHCH}(\text{CH}_3)_2$	13643-05-5	CH_3	9.78±0.05	EI	5483
	$\text{CH}_3\text{C}\equiv\text{CCH}(\text{CH}_3)_2$	21020-27-9	CH_3	9.67±0.05	EI	5483
	$1,2-n-\text{C}_6\text{H}_{10}$	592-44-9	CH_3	9.12±0.05	EI	5483
	$2,3-n-\text{C}_6\text{H}_{10}$	592-49-4	CH_3	9.38±0.05	EI	5483
	$cis-\text{CH}_2=\text{CHCH}=\text{CHC}_2\text{H}_5$	XXXXX-XX-X	CH_3	8.54	EI	4591
	$cis-1,4-n-\text{C}_6\text{H}_{10}$	7318-67-4	CH_3	9.61±0.05	EI	5483
	$trans-1,4-n-\text{C}_6\text{H}_{10}$	7319-00-8	CH_3	9.60±0.05	EI	5483
	$trans-1,3-n-\text{C}_6\text{H}_{10}$	20237-34-7	CH_3	9.74±0.05	EI	5483
	$trans,cis-2,4-n-\text{C}_6\text{H}_{10}$	5194-50-3	CH_3	10.10±0.05	EI	5483
	$(\text{C}_3\text{H}_5)_2$ (1,1'-Bicyclopropyl)	5685-46-1	CH_3	9.34±0.05	EI	5483
	$\text{C}_4\text{H}_6(=\text{CHCH}_3)$ (Cyclobutane, ethenyl-)	2597-49-1	CH_3	9.88±0.05	EI	5483
	C_6H_{10} (Cyclohexene)	110-83-8	CH_3	8.95	EI	4591
			CH_3	11.22±0.05	EI	3585
	$\text{C}_5\text{H}_8=\text{CH}_2$ (Cyclopentane, methylene-)	1528-30-9	CH_3	8.2	EI	5586
	$\text{C}_5\text{H}_7\text{CH}_3$ (Cyclopentene, 1-methyl-)	693-89-0	CH_3	11.71±0.05	EI	3585
			CH_3	8.59	EI	4591
	$\text{C}_5\text{H}_7\text{CH}_3$ (Cyclopentene, 3-methyl-)	1120-62-3	CH_3	11.59±0.05	EI	3585
	$\text{C}_3\text{H}_5\text{C}(\text{CH}_3)=\text{CH}_2$ (Cyclopropane, (1-methylethenyl)-)	4663-22-3	CH_3	9.18±0.05	EI	5483
	$\text{C}_3\text{H}(\text{CH}_3)_3$ (Cyclopropene, 1,3,3-trimethyl-)	3664-56-0	CH_3	8.78±0.05	EI	5483
	$\text{C}_6\text{H}_{10}=\text{CH}_2$ (Cyclohexane, methylene-)	1192-37-6	C_2H_5	12.5	EI	5586
	$\text{C}_{10}\text{H}_{16}$ (4,7-Methano-1 <i>H</i> -indene, octahydro-, (3 <i>a</i> α ,4 <i>B</i> ,7 <i>B</i> ,7 <i>a</i> α -)-)	2825-82-3		10.0±0.1	PI	3918

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₅H₇⁺						
	C ₁₀ H ₁₅ CH ₃	XXXXXX-XX-X		<10.2±0.1	PI	3918
	C ₁₀ H ₁₅ CH ₃	50745-90-9		>10.2±0.1	PI	3918
	(4,7-Methano-1H-indene, octahydro-2-methyl, (2α,3αβ,4α,7α,7aβ)-)					
	C ₁₀ H ₁₅ CH ₃	50745-92-1		>10.5±0.1	PI	3918
	(4,7-Methano-1H-indene, octahydro-8-methyl-, stereoisomer)					
	C ₁₀ H ₁₅ C ₂ H ₅	32787-97-6		>10.2±0.1	PI	3918
	(4,7-Methano-1H-indene, 5-ethyloctahydro-, (3αα,4β,5α,7β,7aα)-)					
	C ₅ H ₅ N(CH ₃)	109-06-8	HCN	12.87±0.05	EI	5413
	(Pyridine,2-methyl-)					
	C ₅ H ₅ N(CH ₃)	108-99-6	HCN	12.94±0.05	EI	5413
	(Pyridine,3-methyl-)					
	C ₅ H ₅ N(CH ₃)	108-89-4	HCN	12.86±0.05	EI	5413
	(Pyridine,4-methyl-)					
	C ₆ H ₁₁ Cl	542-18-7		10.67±0.05	PI	4078
	(Cyclohexane, chloro-)					
C₅H₆D⁺						
	C ₅ H ₆ D	37729-44-5	H	10.98±0.03	EI	4203
	(Cyclopentene-1-d)					
C₅H₄D₃⁺						
	C ₅ H ₄ D ₄	14996-50-0	D	9.72±0.03	EI	4203
	(Spiropentane-1,1,2,2-d ₄)					
C₅H₃D₄⁺						
	C ₅ H ₄ D ₄	14996-50-0	H	9.55±0.03	EI	4203
	(Spiropentane-1,1,2,2-d ₄)					
C₅H₈⁺						
	CH ₂ =C=CHCHCH ₃	591-95-7	**	9.25	PE	5411
			**	9.22 (V)	PE	4748
	CH ₂ =C(CH ₃)CH=CH ₂	78-79-5	**	8.87 (V)	PE	5010
			**	8.89	PE	3847
			**	9.04 (V)	PE	3892
	CH ₂ =CHCH=CHCH ₃	504-60-9	**	8.67	PE	5411
			**	8.6	EI	5200
	(CH ₂ =CH) ₂ CH ₂	591-93-5	**	7.97 (V)	PE	5314
			**	9.62±0.02	PE	4010
			**	9.72 (V)	PE	4211
	CH ₃ CH=C=CHCH ₃	591-96-8	**	9.13 (V)	PE	4019
	(CH ₃) ₂ CHC≡CH	598-23-2	**	10.049±0.007	PE	4575
	(CH ₃) ₂ C=C=CH ₂	598-25-4	**	8.95 (V)	PE	4019
			**	8.9	EI	5200
	C ₃ H ₂ C≡CH	627-19-0	**	10.098±0.005	PE	4575
	C ₂ H ₃ C≡CCH ₃	627-21-4	**	9.439±0.005	PE	4575
			**	9.25±0.02	PE	4702
	1,3-trans-C ₅ H ₈	2004-70-8	**	8.67±0.02	PE	4702
			**	8.61	PE	3847
	cis-CH ₃ CH=CHCH=CH ₂	1574-41-0	**	8.64	PE	5202
			**	8.60 (V)	PE	5005
			**	8.60 (V)	PE	5010
	C ₅ H ₈	185-94-4	**	8.7±0.1	PE	4702
	(Bicyclo[2.1.0]pentane)					
	(JC-Mean value of Jahn-Teller components)					
	C ₄ H ₆ (=CH ₂)	1120-56-5	**	9.35 (V)	PE	4669
	(Cyclobutane, methylene-)					
	C ₅ H ₈	142-29-0	**	9.02±0.01	PI	5556
	(Cyclopentene)		**	9.01±0.03 (V)	PE	4828
			**	9.12 (V)	PE	4285
			**	9.17 (V)	PE	4517

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₅H₈⁺						
	C ₅ H ₈	142-29-0	**	9.18 (V)	PE	4267
			**	9.20 (V)	PE	4189
			**	9.20 (V)	PE	4669
			**	9.00	EI	4203
			**	9.1	EI	5200
	C ₃ H ₅ CH=CH ₂ (Cyclopropane, ethenyl-)	693-86-7	**	8.7	PE	4329
			**	9.1 (V)	PE	4034
			**	9.15 (V)	PE	4347
			**	9.2	PE	3576
	C ₅ H ₈ (Spiropentane)	157-40-4	**	9.26	EI	4203
	C ₅ H ₈ =CH ₂ (Cyclopentane, methylene-)	1528-30-9	CH ₂	9.2	EI	5586
	C ₆ H ₁₀ =CH ₂ (Cyclohexane, methylene-)	1192-37-6	C ₂ H ₄	12.2	EI	5586
	C(CH ₃)(CH ₂)C ₆ H ₈ CH ₃ (Cyclohexene, 1-methyl-4-(1-methylethenyl)-)	138-86-3	C ₅ H ₈	11.6	EI	5200
	n-C ₄ H ₉ CHO	110-62-3	H ₂ O	9.80±0.06	EI	5267
			H ₂ O	10.00	EI	5264
	C ₅ H ₉ OH (Cyclopentanol)	96-41-3	H ₂ O	9.66±0.06	EI	5267
C₅H₉⁺						
	CH ₃ CH=CHCHCH ₃	XXXXX-XX-X	**	7.07	EI	4591
	CH ₂ =C(CH ₃)CHCH ₃	XXXXX-XX-X	**	7.4	OTH	4591
	CH ₂ =C(C ₂ H ₅)CH ₂	XXXXX-XX-X	**	7.9	OTH	4591
	CH ₂ CH(CH ₃)CH=CH ₂	XXXXX-XX-X	**	8.0	OTH	4591
	CH ₂ =CHCHC ₂ H ₅	17829-37-7	**	7.30	EI	4591
	CH ₂ =CHC(CH ₃) ₂	29791-12-6	**	7.13	EI	4591
	C ₅ H ₉	3889-74-5	**	7.47	EI	4545
	(Cyclopentyl)		**	7.47	EI	4591
	((CH ₃) ₂ C) ₂	563-79-1	CH ₃	8.16	EI	4591
	(CH ₃) ₂ C=CHC ₂ H ₅	625-27-4	CH ₃	8.58	EI	4591
	n-C ₄ H ₉ CH=CH ₂	592-41-6	CH ₃	9.44	EI	4591
	n-C ₅ H ₇ C(CH ₃)=CH ₂	763-29-1	CH ₃	9.04	EI	4591
	sec-C ₄ H ₉ CH=CH ₂	760-20-3	CH ₃	9.44	EI	4591
	iso-C ₄ H ₉ CH=CH ₂	691-37-2	CH ₃	9.44	EI	4591
	iso-C ₃ H ₇ C(CH ₃)=CH ₂	27416-06-4	CH ₃	9.01	EI	4591
	tert-C ₄ H ₉ CH=CH ₂	558-37-2	CH ₃	9.44	EI	4591
	cis-CH ₃ CH=C(CH ₃)C ₂ H ₅	922-62-3	CH ₃	8.58	EI	4591
	trans-(CH ₃) ₂ CHCH=CHCH ₃	674-76-0	CH ₃	8.91	EI	4591
	trans-CH ₃ CH=CHC ₂ H ₅	4050-45-7	CH ₃	8.93	EI	4591
	trans-C ₂ H ₅ CH=CHC ₂ H ₅	13269-52-8	CH ₃	8.97	EI	4591
	C ₆ H ₁₂	110-82-7	CH ₃	11.07±0.04	PI	4078
	(Cyclohexane)		CH ₃	9.88	EI	4591
			CH ₃	11.15	EI	4319
	C ₅ H ₉ CH ₃ (Cyclopentane, methyl-)	96-37-7	CH ₃	10.42	EI	4591
	C ₂ H ₅ CH(C ₂ H ₅)=CH ₂	760-21-4	CH ₃	9.01	EI	4591
	C ₁₀ H ₁₆	2825-82-3		10.5±0.1	PI	3918
	(4,7-Methano-1H-indene, octahydro-, (3α,4β,7β,7α)-)					
	C ₆ H ₁₁ Cl (Cyclohexane, chloro-)	542-18-7		11.01±0.02	PI	4078
	CH ₂ =CH(CH ₂) ₃ Br	1119-51-3	Br	10.2	EI	5633
C₅H₁₀⁺						
	(CH ₃) ₂ C=CHCH ₂	513-35-9	**	8.83±0.11	EI	3544
			**	8.682±0.003	PE	3957

Table of Ion Energetics Measurements—Continued

Ion (state),	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₅H₁₀⁺						
	(CH ₃) ₂ C=CHCH ₂	513-35-9	**	8.72	PE	3533
	(CH ₃) ₂ CHCH=CH ₂	563-45-1	**	9.533±0.003	PE	3957
			**	9.60±0.03	EI	3544
	C ₂ H ₅ C(CH ₃)=CH ₂	563-46-2	**	9.148±0.003	PE	3957
			**	9.35±0.08	EI	3544
	1-C ₅ H ₁₀	109-67-1	**	9.42±0.02	PE	4695
			**	9.524±0.003	PE	3957
			**	9.54±0.02 (V)	PE	4010
			**	9.68±0.01 (V)	PE	4939
			**	9.82±0.06	EI	3544
	2-cis-C ₅ H ₁₀	627-20-3	**	8.94±0.02	PE	4695
			**	9.22±0.01 (V)	PE	4939
			**	9.036±0.005	PE	3957
			**	9.23±0.02	EI	3544
	trans-2-C ₅ H ₁₀	646-04-8	**	9.036±0.005	PE	3957
			**	9.23±0.01 (V)	PE	4939
			**	9.32±0.03	EI	3544
	C ₄ H ₇ CH ₃ (Cyclobutane, methyl-)	598-61-8	**	9.60	PE	4268
	C ₅ H ₁₀ (Cyclopentane)	287-92-3	**	10.55±0.03	PI	5556
			**	10.3±0.1	PE	4702
			**	10.40	PE	4056
			**	10.48	PE	4319
			**	10.5 (V)	PE	4189
			**	10.54±0.05	EI	4319
			**	10.91±0.07	EI	3544
C₅H₁₁⁺						
	1-C ₅ H ₁₁	2672-01-7	**	7.94±0.06	EI	4895
	2-C ₅ H ₁₁	2492-34-4	**	7.41	EI	4895
	tert-C ₄ H ₉ CH ₂	3744-21-6	**	7.91	EI	4895
	tert-C ₅ H ₁₁	4348-35-0	**	6.85	EI	4895
	tert-C ₅ H ₁₁ NO	34946-78-6		8.7±0.1	EI	3654
	(iso-C ₅ H ₁₁)SOCH ₃	55860-10-1	CH ₃ SO	9.1±0.3	EI	5311
C₅H₁₂⁺						
	n-C ₅ H ₁₂	109-66-0	**	10.2±0.1	PE	4702
			**	10.36	PE	4056
			**	10.59±0.05	EI	3791
	iso-C ₅ H ₁₂	78-78-4	**	10.3±0.1	PE	4702
			**	10.50±0.05	EI	3791
	neo-C ₅ H ₁₂	463-82-1	**	10.21±0.04	PE	3880
			**	10.25±0.1	PE	3677
			**	11.3 (V)	PE	3710
			**	11.3 (V)	PE	4050
C₆H₂⁺						
	HC≡CC≡CC≡CH	3161-99-7	**	9.50	PE	4048
			**	9.63 (V)	PE	5084
C₆H₄⁺						
	cis-CH=CCH=CHC≡CH	16668-67-0	**	9.10±0.02	PE	4374
	trans-CH=CCH=CHC≡CH	16668-68-1	**	9.07±0.02	PE	4374
	C ₆ H ₄ (1,3-Cyclohexadien-5-yne)	462-80-6	**	9.75±0.2	EI	3583
	C ₆ H ₆ (Benzene)	71-43-2	H ₂	12.94	PI	4075
			H ₂	14.04±0.06	EI	3784
				14.14±0.08	EI	4534
	C ₆ H ₅ CN (Benzonitrile)	100-47-0	HCN	13.38±0.03	EI	5080

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₆H₅⁺						
	C ₆ H ₅ CN	100-47-0	HCN	13.80±0.06 13.92±<0.1	EI EI	3784 3735
C₆H₅⁺						
	C ₆ H ₅ (Phenyl)	2396-01-2	**	8.1±0.1	PI	3752
	HC≡CCH ₂ CH ₂ C≡CH	628-16-0	H	10.16±0.08	PI	5454
			H	10.21±0.03	EI	3790
	H ₃ CC≡CC≡CCH ₃	2809-69-0	H	10.55±0.09	PI	5454
	HC≡CCH ₂ C≡CCH ₃	10420-91-4	H	10.21±0.1	EI	5454
	C ₆ H ₆ (Benzene)	71-43-2	H	12.94	PI	4075
			H	13.78±0.08	PI	5454
			H	13.74±0.1	EI	5454
			H	13.97±0.06	EI	3784
			H	14.05±<0.1	EI	3735
			H	14.56±0.07	EI	4534
	C ₆ H ₅ CH ₃ (Benzene, methyl-)	108-88-3	CH ₃	13.70	EI	4115
	C ₇ H ₈ (1,3,5-Cycloheptatriene)	544-25-2	CH ₃	14.17	EI	4115
	trans-CH ₂ =CHCH=CHCH=CHCH ₃	17679-93-5		12.3±0.15	PE	5432
	C ₆ H ₅ CH ₂ CH ₂ N(CH ₃) ₂ (Benzeneethanamine,dimethyl-)	29088-49-1	C ₆ H ₁₀ N	10.55	PI	5543
	C ₆ H ₅ CHO (Benzaldehyde)	100-52-7	CO+H	14.11	EI	3792
	C ₆ H ₅ COCH ₃ (Ethanone, 1-phenyl-)	98-86-2	CO+CH ₃	13.28	EI	3626
			CO+CH ₃	13.97	EI	3792
	(C ₆ H ₅) ₂ CO (Methanone, diphenyl-)	119-61-9	C ₆ H ₅ +CO	15.67	EI	3792
	C ₆ H ₅ COOH (Benzoic acid)	65-85-0		14.3±0.07	EI	5121
			CO+OH	15.08±0.2	EI	3973
			CO+OH	15.08	EI	3792
	C ₆ H ₅ COOCH ₃ (Benzoic acid methyl ester)	93-58-3	CH ₃ O+CO	13.82	EI	3626
				14.3±0.07	EI	5121
	C ₆ H ₅ COOC ₂ H ₅ (Benzoic acid, ethyl ester)	93-89-0	CH ₃ O+CO	14.74	EI	3792
	C ₆ H ₅ COOC ₃ H ₇ (Benzoic acid, 1-methylethyl ester)	939-48-0		14.5±0.03	EI	5121
	C ₆ H ₅ COOC ₃ H ₇ (Benzoic acid, propyl ester)	2315-68-6		15.0±0.10	EI	5121
	C ₆ H ₅ COOC ₄ H ₉ (Benzoic acid, butyl ester)	136-60-7		14.9±0.06	EI	5121
	C ₆ H ₅ COOG ₃ H ₉ (Benzoic acid, 2-methylpropyl ester)	120-50-3		15.0±0.03	EI	5121
	C ₆ H ₅ COOC ₅ H ₁₁ (Benzoic acid, methylbutyl ester)	XXXXX-XX-X		15.0±0.04	EI	5121
	C ₆ H ₅ NO (Benzene, nitroso-)	586-96-9		15.2±0.10	EI	5121
	C ₆ H ₅ CONH ₂ (Benzamide)	55-21-0	NH ₂ +CO	11.0±0.1	EI	3792
	C ₆ H ₅ COC ₆ H ₄ NH ₂ (Methanone, (2-aminophenyl)phenyl-)	2835-77-0		14.21	EI	4358
	C ₆ H ₅ COC ₆ H ₄ NH ₂ (Methanone, (3-aminophenyl)phenyl-)	2835-78-1		16.2±0.3	EI	4358
	C ₆ H ₅ NO ₂ (Benzene,nitro-)	98-95-3	NO ₂	9.46±0.05	PI	5437

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₆H₅⁺						
	C ₆ H ₅ NO ₂	98-95-3	NO ₂	11.93±0.1	EI	3447
	C ₆ H ₅ COCH ₂ H ₄ NO ₂ (Methanone, (2-nitrophenyl)phenyl-)	2243-79-0		14.9±0.3	EI	4358
	C ₆ H ₅ COCH ₂ H ₄ NO ₂ (Methanone, (3-nitrophenyl)phenyl-)	2243-80-3		15.6±0.3	EI	4358
	C ₆ H ₅ COCH ₂ H ₄ NO ₂ (Methanone, (4-nitrophenyl)phenyl-)	1144-74-7		15.5±0.3	EI	4358
	C ₆ H ₅ Cl (Benzene, chloro-)	108-90-7	Cl	12.47±0.06	PI	5181
			Cl	12.81	EI	3626
	C ₆ H ₅ COCl (Benzoyl chloride)	98-88-4	Cl+CO	13.81	EI	3792
	C ₆ H ₅ COCH ₂ H ₄ Cl (Methanone, (2-chlorophenyl)phenyl-)	5162-03-8		15.0±0.3	EI	4358
	C ₆ H ₅ COCH ₂ H ₄ Cl (Methanone, (3-chlorophenyl)phenyl-)	1016-78-0		15.2±0.3	EI	4358
	C ₆ H ₅ COCH ₂ H ₄ Cl (Methanone, (4-chlorophenyl)phenyl-)	134-85-0		15.2±0.3	EI	4358
	C ₆ H ₅ Br (Benzene, bromo-)	108-86-1	Br	11.82	EI	3626
	C ₆ H ₅ I (Benzene, iodo-)	591-50-4	I	11.34	EI	3626
C₆H₃D₂⁺	CD≡CCH ₂ CH ₂ C≡CD	XXXXX-XX-X	H	10.18±0.03	EI	3790
C₆H₆⁺						
	HC≡CCH ₂ CH ₂ C≡CH	628-16-0	**	9.98±0.05	PI	5454
			**	10.48 (V)	PE	4397
			**	9.87±0.03	EI	3790
			**	9.93±0.05	EI	5454
	CH ₂ =CHC≡CCH=CH ₂	821-08-9	**	8.50±0.02	PE	4374
	H ₃ CC≡CC≡CCH ₃	2809-69-0	**	8.90±0.05	PI	5454
			**	8.91	PE	4048
			**	8.92	PE	4731
			**	9.08 (V)	PE	5084
			**	9.03±0.1	EI	4714
	CH ₂ =C=(CH) ₂ =C=CH ₂	29776-96-3	**	8.53 (V)	PE	4397
	CH≡CCH ₂ CH=C=CH ₂	33142-15-3	**	9.65 (V)	PE	4397
	C ₆ H ₆ (Benzene)	71-43-2	**	9.2	PI	3586
			**	9.2 (V)	PE	3528
			**	9.22	PE	5408
			**	9.22 (V)	PE	5125
			**	9.23 (V)	PE	4884
			**	9.23 (V)	PE	4472
			**	9.24±0.02 (V)	PE	4913
			**	9.24	PE	4621
			**	9.24	PE	5197
			**	9.24 (V)	PE	3513
			**	9.24 (V)	PE	3673
			**	9.24 (V)	PE	4280
			**	9.24 (V)	PE	4701
			**	9.24 (V)	PE	5012
			**	9.24 (V)	PE	5378
			**	9.24 (V)	PE	5632
			**	9.25±0.03 (V)	PE	3713
			**	9.25±0.05 (V)	PE	4724
			**	9.25	PE	3520
			**	9.25	PE	5084
			**	9.25 (V)	PE	5600

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₆H₆⁺						
	C ₆ H ₆	71-43-2	**	9.27	PE	3658
			**	9.3 (V)	PE	5258
			**	9.20±0.1	EI	3624
			**	9.25±0.07	EI	4534
			**	9.26±0.06	EI	5503
			**	9.70	EI	4834
			**	9.25	CTS	3922
	C ₆ H ₆ (Bicyclo[2.2.0]hexa-2,5-diene)	5649-95-6		9.4 (V)	PE	4394
	C ₃ (=CH ₂) ₃ (Cyclopropane, tris(methylene)-)	3227-90-5	**	9.40 (V)	PE	4453
			**	9.0±0.1	S	4184
	C ₆ H ₆ (Tricyclo[3.1.0.0 ^{2,6}]hex-3-ene)	659-85-8	**	8.94 (V)	PE	5431
			**	8.54±0.04 (V)	PE	4716
	C ₈ H ₈ (Pentacyclo[4.2.0.0 ^{2,5} .0 ^{3,8} .0 ^{4,7}]octane)	277-10-1		8.55 (V)	PE	4400
	C ₆ H ₅ OCH ₃ (Benzene, methoxy-)	100-66-3	CH ₂ O	9.2±<0.1	EI	3735
			HCHO	11.27±0.1	EI	3446
				11.50	EI	3845
				11.55±<0.1	EI	3735
	(C ₆ H ₆)(CO) ₃ Cr (Chromium, (η^6 -benzene)tricarbonyl-)	12082-08-5		9.49±0.1	EI	3788
C₆H₆⁺²						
	C ₆ H ₆ (Benzene)	71-43-2	**	26.1	OTH	5141
C₆H₄D₂⁺	CD≡CCH ₂ CH ₂ C≡CD	XXXXX-XX-X	**	9.97±0.06	EI	3790
C₆H₇⁺						
	C ₇ H ₁₀ (Bicyclo[2.2.1]hept-2-ene)	498-66-8	CH ₃	10.46±0.01	EI	3535
	C ₇ H ₁₀ (Tricyclo[2.2.1.0 ^{2,6}]heptane)	279-19-6	CH ₃	10.17±0.01	EI	3535
	C ₁₀ H ₁₆ (Tricyclo[3.3.1.1 ^{3,7}]decane)	281-23-2		10.69	PI	4173
C₆H₈⁺						
	C ₅ H ₅ (CH ₃) (1,3-Cyclopentadiene, 2-methyl-)	3727-31-9	**	8.4 (V)	PE	4373
	C ₅ H ₅ (CH ₃) (1,3-Cyclopentadiene, 1-methyl-)	96-39-9	**	8.4 (V)	PE	4373
	CH ₃ C≡CC(CH ₃)=CH ₂	926-55-6	**	8.72±0.01	PE	5407
	C ₂ H ₅ C≡CCH=CH ₂	13721-54-5	**	8.91±0.01	PE	5407
	CH ₂ =C=CHC(CH ₃)=CH ₂	14763-81-6	**	8.54 (V)	PE	4829
	CH ₂ =C=C(CH ₃)CH=CH ₂	25054-29-9	**	8.54 (V)	PE	4829
	1,2, <i>trans</i> -4-C ₆ H ₈	20130-95-4	**	8.32 (V)	PE	4829
	2,3,5-C ₆ H ₈	33755-64-5	**	8.56 (V)	PE	4829
	cis-CH ₂ =CHCH=CHCH=CH ₂	2612-46-6	**	8.3±0.1	S	4235
			**	8.31±0.02	PE	5432
			**	8.32	PE	3847
	trans-CH ₂ =CHCH=CHCH=CH ₂	821-07-8	**	8.27	S	4235
			**	8.29	PE	3847
			**	8.30±0.02	PE	5432
	C ₆ H ₈ (Bicyclo[2.2.0]hex-2-ene)	3097-63-0	**	9.4 (V)	PE	4453
	C ₄ H ₄ (=CH ₂) ₂ (Cyclobutane, 1,2-bis(methylene)-)	14296-80-1	**	8.66±0.03 (V)	PE	4766

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₆H₈⁺						
	C ₄ H ₆ (=CH ₂) ₂	14296-80-1	**	8.77	PE	5265
	C ₄ H ₆ (=CH ₂) ₂ (Cyclobutane, 1,3-bis(methylene)-)	2045-78-5	**	9.08±0.03 (V)	PE	4766
	C ₄ H ₆ C≡CH (Cyclobutane, ethynyl-)	50786-62-4	**	10.02 (V)	PE	3997
			**	10.02 (V)	PE	5607
	C ₆ H ₈ (1,3-Cyclohexadiene)	592-57-4	**	8.25±0.02	PE	4702
			**	8.25±0.03 (V)	PE	4828
			**	8.25	PE	5411
			**	8.32 (V)	PE	5010
	C ₆ H ₈ (1,4-Cyclohexadiene)	628-41-1	**	8.80 (V)	PE	5538
			**	8.82±0.02	PE	4702
			**	8.82	PE	5411
			**	8.82 (V)	PE	4531
			**	8.82 (V)	PE	5535
	C ₅ H ₅ CH ₃ (1,3-Cyclopentadiene, methyl-)	26519-91-5	**	8.28±0.05 (V)	PE	3688
	C ₅ H ₅ CH ₃ (1,3-Cyclopentadiene, 1-methyl-)	96-39-9	**	8.40 (V)	PE	4179
	C ₅ H ₅ CH ₃ (1,3-Cyclopentadiene, 2-methyl-)	3727-31-9	**	8.45 (V)	PE	4179
	C ₅ H ₆ =CH ₂ (Cyclopentene, 3-methylene-)	930-26-7	**	8.40	PE	4347
	C ₆ H ₈ (Cyclopropane cyclopropylidene-)	27567-82-4	**	8.93 (V)	PE	4963
	C ₆ H ₈ (Tricyclo[3.1.0.0 ^{2,6}]hexane)	287-12-7	**	9.43 (V)	PE	4400
	C ₅ H ₆ =CH ₂ (Cyclopentane, methylene-)	1528-30-9	H ₂	8.7	EI	5586
	C ₆ H ₁₀ =CH ₂ (Cyclohexane, methylene-)	1192-37-6	CH ₄	11.2	EI	5586
	C ₁₀ H ₁₆ (4,7-Methano-1H-indene, octahydro-, (3aα,4β,7β,7aα)-)	2825-82-3		9.9±0.1	PI	3918
C₆H₉⁺						
	CH≡C(CH ₂) ₃ CH ₃	693-02-7	H	10.75±0.05	EI	3585
	CH ₃ C≡CCH ₂ CH ₂ CH ₃	764-35-2	H	10.81±0.05	EI	3585
	C ₆ H ₁₀ (Cyclohexene)	110-83-8	H	11.8±0.05	EI	3585
	C ₅ H ₆ =CH ₂ (Cyclopentane, methylene-)	1528-30-9	H	12.13±0.05	EI	3585
	C ₅ H ₅ CH ₃ (Cyclopentene, 1-methyl-)	693-89-0	H	11.97±0.05	EI	3585
	C ₁₀ H ₁₅ CH ₃ (4,7-Methano-1H-indene, octahydro-8-methyl, stereoisomer)	50745-92-1		9.5±0.1	PI	3918
	C ₁₀ H ₁₅ C ₂ H ₅ (4,7-Methano-1H-indene, 5-ethyloctahydro-, (3aα,4β,5α,7β,7aα)-)	32787-97-6		<10.2±0.1	PI	3918
	C ₆ H ₁₁ Cl (Cyclohexane, chloro-)	542-18-7		10.40±0.02	PI	4078
C₆H₁₀⁺						
	C ₅ H ₇ (CH ₃) (Cyclopentene, 3-methyl-)	1120-62-3	**	8.98±0.05 (V)	PE	4954
	trans-CH ₂ =CHC(CH ₃)=CHCH ₃	XXXXX-XX-X	**	8.37±0.05	EI	5483
	CH ₂ =C(C ₂ H ₅)CH=CH ₂	XXXXX-XX-X	**	8.81±0.05	EI	5483
	CH ₂ =C(CH ₃)C(CH ₃)=CH ₂	513-81-5	**	8.62	PE	3847
			**	8.72 (V)	PE	5010
			**	8.76 (V)	PE	3892
			**	8.54±0.04	EI	4274

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₆H₁₀⁺						
	CH ₃ =C(CH ₃)C(CH ₃)=CH ₂	513-81-5	**	8.66±0.05	EI	5483
	(CH ₂ =CHCH ₂) ₂	592-42-7	**	9.25 (V)	PE	5314
			**	9.29±0.05	EI	5483
			**	9.59±0.02 (V)	PE	4010
	C ₄ H ₉ C≡CH	693-02-7	**	10.067±0.005	PE	4575
			**	9.95±0.05	EI	5483
			**	10.52±0.05	EI	3585
	CH ₂ =C(CH ₃)CH ₂ CH=CH ₂	763-30-4	**	9.16±0.05	EI	5483
	C ₃ H ₇ C≡CCH ₃	764-35-2	**	9.366±0.005	PE	4575
			**	9.37±0.05	EI	5483
			**	9.97±0.05	EI	3585
	CH≡CC(CH ₃) ₃	917-92-0	**	9.80±0.05	EI	5483
			**	10.67±0.02	EI	4126
	CH ₃ CH ₂ CH(CH ₃)C≡CH	922-59-8	**	9.975±0.008	PE	4575
			**	9.79±0.05	EI	5483
	trans-CH ₂ =C(CH ₃)CH=CHCH ₃	926-54-5	**	8.47 (V)	PE	5010
			**	8.45±0.05	EI	5483
	(CH ₃) ₂ C=CHCH=CH ₂	926-56-7	**	8.29	PE	5202
			**	8.26±0.05	EI	5483
	C ₂ H ₅ C≡CC ₂ H ₅	928-49-4	**	9.323±0.005	PE	4575
			**	9.34±0.05	EI	5483
	CH ₂ =CHCH(CH ₃)CH=CH ₂	1115-08-8	**	9.40±0.05	EI	5483
	CH ₂ =C(CH ₃)CH=CHCH ₃	1118-58-7	**	8.47±0.02	PE	4702
	cis-CH ₂ =CHC(CH ₃)=CHCH ₃	2787-43-1	**	8.39±0.02	PE	4702
	trans-CH ₂ =CHC(CH ₃)=CHCH ₃	2787-45-3	**	16.6±0.1 (V)	PE	4702
			**	8.46±0.05	EI	5483
	(CH ₃) ₂ C=C=CHCH ₃	3043-33-2	**	8.69 (V)	PE	4019
			**	8.64±0.05	EI	5483
	C ₂ H ₅ C(=CH ₂)CH=CH ₂	3404-63-5	**	8.79±0.02	PE	4702
			**	8.79	PE	5411
	CH ₂ =CHC(CH ₃)=CHCH ₃	4549-74-0	**	8.39	PE	5411
	trans,trans-2,4-n-C ₆ H ₁₀	5194-51-4	**	8.26±0.05	EI	5483
	(CH ₃) ₂ CHCH ₂ C≡CH	7154-75-8	**	10.055±0.005	PE	4575
			**	9.83±0.05	EI	5483
	CH ₂ =C=C(CH ₃)C ₂ H ₅	7417-48-3	**	8.74±0.05	EI	5483
	CH ₂ =C=CHCH(CH ₃) ₂	13643-05-5	**	9.06±0.05	EI	5483
	(CH ₃) ₂ CHC≡CCH ₃	21020-27-9	**	9.346±0.007	PE	4575
			**	9.31±0.05	EI	5483
	1,2-n-C ₆ H ₁₀	592-44-9	**	9.00±0.05	EI	5483
	1,3-C ₆ H ₁₀	592-48-3	**	8.53±0.02	PE	4702
	2,3-n-C ₆ H ₁₀	592-49-4	**	8.76±0.05	EI	5483
	2,4-C ₆ H ₁₀	592-46-1	**	8.09±0.03 (V)	PE	4828
	2,4-trans,cis-C ₆ H ₁₀	5194-50-3	**	8.25±0.02	PE	4702
			**	8.26	PE	5202
			**	8.24±0.05	EI	5483
	2,4-cis,cis-C ₆ H ₁₀	6108-61-8	**	8.18±0.02	PE	4702
	(tert-C ₄ H ₉)C≡CH	917-92-0	**	9.923±0.010	PE	4575
	cis-1,4-n-C ₆ H ₁₀	7318-67-4	**	9.04±0.05	EI	5483
	trans-1,4-n-C ₆ H ₁₀	7319-00-8	**	8.98±0.05	EI	5483
	trans-1,3-n-C ₆ H ₁₀	20237-34-7	**	8.54±0.05	EI	5483
	trans,trans-CH ₃ CH=CHCH=CHCH ₃	5194-51-4	**	8.09	PE	3847
			**	8.93 (V)	PE	3892
	C ₆ H ₁₀ (Bicyclo[2.2.0]hexane)	186-04-9	**	9.6 (V)	PE	4453
	(C ₃ H ₅) ₂ (1,1'-Bicyclopropyl)	5685-46-1	**	9.6 (V)	PE	5344
			**	9.12±0.05	EI	5483
	C ₄ H ₉ CH=CH ₂ (Cyclobutane, ethenyl-)	2597-49-1	**	9.44 (V)	PE	4347
			**	9.44 (V)	PE	5607
			**	8.70±0.05	EI	5483

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₆H₁₀⁺						
	C ₆ H ₁₀ (Cyclohexene)	110-83-8	**	8.94±0.01	PI	5556
			**	9.11 (V)	PE	4249
			**	9.12 (V)	PE	4267
			**	9.12 (V)	PE	4285
			**	9.12 (V)	PE	5538
			**	9.57±0.05	EI	3585
	C ₅ H ₈ =CH ₂ (Cyclopentane, methylene-)	1528-30-9	**	8.55±0.01	PI	3585
			**	9.14 (V)	PE	4669
			**	7.2	EI	5586
			**	9.26±0.05	EI	3585
	C ₅ H ₇ CH ₃ (Cyclopentene, 1-methyl-)	693-89-0	**	8.55±0.01	PI	3585
			**	8.60±0.01	PI	5556
			**	9.12±0.05	EI	3585
	CH ₃ C ₅ H ₇ (Cyclopentene,3-methyl-)	1120-62-3	**	8.95±0.01	PI	5556
	C ₃ H ₅ C(CH ₃)=CH ₂ (Cyclopropane, (1-methylethyl)-)	4663-22-3	**	9.12	PE	4608
			**	8.66±0.05	EI	5483
	C ₃ H(CH ₃) ₃ (Cyclopropene, 1,3,3-trimethyl-)	3664-56-0	**	8.58±0.05	EI	5483
	C ₆ H ₁₀ (Spirohexane)	157-45-9	**	9.66 (V)	PE	5361
	C ₆ H ₁₀ =CH ₂ (Cyclohexane,methylene-)	1192-37-6	CH ₂	11.7	EI	5586
	C ₆ H ₁₀ (CH ₃) ₂ (Cyclohexane, 1,2-dimethyl-, <i>cis</i> -)	2207-01-4	2CH ₃	10.46±0.1	EI	3581
	C ₆ H ₁₀ (CH ₃) ₂ (Cyclohexane, 1,2-dimethyl-, <i>trans</i> -)	6876-23-9	2CH ₃	10.63±0.1	EI	3581
	C ₁₀ H ₁₅ CH ₃	XXXXX-XX-X		9.8±0.1	PI	3918
	C ₁₀ H ₁₅ CH ₃	50745-90-9		10.0±0.1	PI	3918
	(4,7-Methano-1H-indene, octahydro-2-methyl-, (2α,3αβ,4α,7α,7αβ)-)					
	(CH ₃) ₂ CHC ₆ H ₄ CHO	1119-16-0	H ₂ O	10.00	EI	5264
	C ₂ H ₅ CH(CH ₃)CH ₂ CHO	15877-57-3	H ₂ O	9.90	EI	5264
	n-C ₅ H ₁₁ CHO	66-25-1	H ₂ O	9.80	EI	5264
	C ₆ H ₁₁ OH (Cyclohexanol)	108-93-0	H ₂ O	10.2±0.2	EI	4617
			H ₂ O	10.4±0.05	EI	4548
	C ₆ H ₁₁ Cl (Cyclohexane, chloro-)	542-18-7		10.10±0.05	PI	4078
C₆H₉D⁺						
	C ₆ H ₉ D ₂ OH (Cyclohexanol, 3,5-d ₂ -)	XXXXX-XX-X HDO		11.3±0.10	EI	4548
	C ₆ H ₉ D ₂ OH (Cyclohexanol, 4,4-d ₂ -)	XXXXX-XX-X HDO		10.5±0.06	EI	4548
C₆H₈D₂⁺						
	C ₆ H ₈ D ₂ OH (Cyclohexanol, 3,5-d ₂ -)	XXXXX-XX-X H ₂ O		10.5±0.10	EI	4548
	C ₆ H ₈ D ₂ OH (Cyclohexanol, 4,4-d ₂ -)	XXXXX-XX-X H ₂ O		11.1±0.04	EI	4548
C₆H₇D₃⁺						
	C ₆ H ₇ D ₃ OH (Cyclohexan-3,3,5,5-d ₃ -ol)	XXXXX-XX-X HDO		10.7±0.2	EI	4617
C₆H₆D₄⁺						
	C ₆ H ₆ D ₄ OH (Cyclohexan-3,3,5,5-d ₄ -ol)	21273-04-1	H ₂ O	10.2±0.2	EI	4617

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₆H₁₁⁺					•	
	C ₆ H ₁₂ (Cyclohexane)	110-82-7	H	11.32±0.05	PI	4078
	C ₆ H ₁₁ Cl (Cyclohexane, chloro-)	542-18-7		10.20±0.05	PI	4078
	C ₆ H ₁₁ Br (Cyclohexane, bromo-)	108-85-0		9.85±0.05	PI	4078
C₆H₁₂⁺						
	(CH ₃) ₃ CCH=CH ₂	558-37-2	**	9.450±0.005	PE	3957
			**	9.7 (V)	PE	3940
	(CH ₃) ₄ C=C	563-78-0	**	8.41 (V)	PE	5535
			**	9.072±0.005	PE	3957
	(CH ₃) ₂ C=C(CH ₃) ₂	563-79-1	**	8.26	PE	3533
			**	8.271±0.005	PE	3957
			**	8.30 (V)	PE	5600
			**	8.42 (V)	PE	4243
			**	8.46 (V)	PE	4459
			**	10.52 (V)	PE	4747
	(CH ₃) ₂ CHCH ₂ CH=CH ₂	691-37-2	**	9.452±0.003	PE	3957
	(C ₂ H ₅) ₂ C=CH ₂	760-21-4	**	9.061±0.005	PE	3957
	C ₂ H ₅ CH ₂ C(CH ₃)=CH ₂	763-29-1	**	9.076±0.005	PE	3957
	1-C ₆ H ₁₂	592-41-6	**	9.31	PE	4033
			**	9.37±0.02	PE	4695
			**	9.478±0.003	PE	3957
			**	9.65±0.01 (V)	PE	4939
			**	9.33	EI	4033
	2-C ₆ H ₁₂	592-43-8	**	8.88±0.02	PE	4695
	3-trans-C ₆ H ₁₂	13269-52-8	**	8.83±0.02	PE	4695
			**	9.14±0.01 (V)	PE	4939
	cis-(CH ₃) ₂ CHCH=CHCH ₃	691-38-3	**	8.976±0.005	PE	3957
	cis-2-C ₆ H ₁₂	7688-21-3	**	8.969±0.005	PE	3957
			**	9.15±0.01 (V)	PE	4939
	cis-3-C ₆ H ₁₂	7642-09-3	**	8.954±0.005	PE	3957
			**	9.15±0.01 (V)	PE	4939
	trans-(CH ₃) ₂ CHCH=CHCH ₃	674-76-0	**	8.972±0.005	PE	3957
	trans-2-C ₆ H ₁₂	4050-45-7	**	8.966±0.005	PE	3957
			**	9.16±0.01 (V)	PE	4939
	trans-3-C ₆ H ₁₂	13269-52-8	**	8.965±0.005	PE	3957
	C ₆ H ₁₂ (Cyclohexane)	110-82-7	**	9.88±0.01	S	3757
			**	9.88±0.01	PI	4078
			**	9.89±0.01	PI	5556
			**	9.84	PE	4319
			**	9.87	PE	4056
			**	9.88	PE	5043
			**	10.3 (V)	PE	3997
			**	9.83±0.05	EI	4319
	CH ₃ C ₅ H ₉ (Cyclopentane,methyl-)	96-37-7	**	10.34±0.04	PI	5556
			**	18.3±0.1	PE	4702
C₆D₁₂⁺						
	C ₆ D ₁₂ (Cyclohexane-d ₁₂)	1735-17-7	**	9.91±0.01	S	3757
C₆H₁₃⁺						
	1-C ₆ H ₁₃	2679-29-0	**	7.92±0.06	EI	4895
	2-C ₆ H ₁₃	2493-44-9	**	7.38	EI	4895
	n-C ₅ H ₇ C(CH ₃) ₂	21058-26-4	**	6.82	EI	4895

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_6\text{H}_{11}^+$	$n\text{-C}_6\text{H}_{11}$	110-54-3	**	10.22	PE	4056
			**	18.7±0.1 (V)	PE	4702
	$(\text{iso-C}_3\text{H}_7)_2$ <i>tert</i> - $\text{C}_1\text{H}_3\text{CH}_2\text{CH}_3$	79-29-8 75-83-2	**	17.9±0.1 (V)	PE	4702
C_7H_6^+	C_7H_6 (Bicyclo[3.2.0]hepta-1,4,6-triene)	35295-58-0	**	8.41 (V)	PE	4779
	C_7H_6 (Bicyclo[4.1.0]hepta-1,3,5-triene)	4646-69-9	**	8.82 (V)	PE	4063
	$\text{C}_5\text{H}_4(=\text{C}=\text{CH}_2)$ (1,3-Cyclopentadiene, 5-ethenylidene-)	27041-32-3	**	8.29 (V)	PE	4779
	<i>cis</i> - $\text{C}_4\text{H}_4(\text{C}\equiv\text{CH})_2$ (Cyclopropane, <i>cis</i> -1,2-diethynyl-)	59502-33-9	**	8.90±0.02	PE	4374
	<i>trans</i> - $\text{C}_4\text{H}_4(\text{C}\equiv\text{CH})_2$ (Cyclopropane, <i>trans</i> -1,2-diethynyl-)	35295-57-9	**	9.00±0.02	PE	4374
	$\text{C}_6\text{H}_5\text{CH}_2\text{CN}$ (Benzeneacetonitrile)	140-29-4	HCN	12.19	EI	4934
	$\text{C}_6\text{H}_4(\text{CN})\text{CH}_3$ (Benzonitrile, 4-methyl-)	104-85-8	HCN	12.22	EI	4934
	$\text{C}_7\text{H}_7\text{CN}$ (2,4,6-Cycloheptatriene-1-carbonitrile)	13612-59-4	HCN	11.19	EI	4934
C_7H_7^+	C_7H_7 (2,4,6-Cycloheptatrien-1-yl) (JC-Mean value of Jahn-Teller components)	3551-27-7	**	6.28±0.02	PE	4820
			**	6.74±0.05	EI	3789
	$\text{C}_6\text{H}_5\text{CH}_2$ (Methyl, phenyl-)	2154-56-5	**	7.20±0.02	PE	4722
			**	7.20±0.02	PE	4898
			**	7.43±0.06 (V)	PE	4609
	$\text{C}_6\text{H}_5\text{CH}_3$ (Benzene,methyl-)	108-88-3	H	10.71±0.03	PI	5120
			H	10.71	EI	5293
			H	11.8	EI	4115
	C_7H_8 (1,3,5-Cycloheptatriene)	544-25-2	H	9.36±0.02	PI	5120
			H	10.73	EI	4115
	<i>trans</i> - $\text{CH}_2=\text{CHCH=CHCH=CHCH}_3$	17679-93-5		12.2±0.15	PE	5432
	$\text{C}_6\text{H}_4(\text{CH}_3)_2$ (Benzene, 1,2-dimethyl-)	95-47-6	CH_3	11.80±0.2	EI	4199
	$\text{C}_6\text{H}_4(\text{CH}_3)_2$ (Benzene, 1,3-dimethyl-)	108-38-8	CH_3	11.80±0.2	EI	4199
	$\text{C}_6\text{H}_4(\text{CH}_3)_2$ (Benzene, 1,4-dimethyl-)	106-42-3		11.5±0.3	EI	4223
	$\text{C}_6\text{H}_5\text{C}_2\text{H}_5$ (Benzene,ethyl-)	100-41-4	CH_3	11.85±0.2	EI	4199
	$\text{C}_6\text{H}_5\text{CH}(\text{CH}_3)_2$ (Benzene,1-methylethyl-)	98-82-8	C_2H_5	10.06	EI	5293
	$\text{C}_6\text{H}_5\text{CH}_2\text{CH}_2\text{CH}_3$ (Benzene,propyl-)	103-65-1	C_2H_5	9.85	EI	5293
	$\text{C}_6\text{H}_5\text{C}_4\text{H}_9$ (Benzene, butyl-)	104-51-8	C_3H_7	9.93	EI	5293
	$\text{C}_6\text{H}_5\text{CH}(\text{CH}_3)\text{C}_6\text{H}_5$ (Benzene,1-methylpropyl)	135-98-8	C_3H_7	10.00	EI	5293
	$\text{C}_6\text{H}_5\text{CH}_2\text{CH}(\text{CH}_3)_2$ (Benzene,2-methylpropyl)	538-93-2	C_3H_7	9.99	EI	5293
	$(\text{C}_6\text{H}_5)_2\text{CH}_2$ (Benzene, 1,1'-methylenebis-)	101-81-5	C_6H_5	11.5±0.1	EI	3807

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₇H₇⁺						
	C ₆ H ₅ C ₆ H ₄ CH ₃ (1,1'-Biphenyl, 4-methyl-)	644-08-6		13.7±0.3	EI	4223
	C ₆ H ₅ CH ₂ CH ₂ C ₆ H ₄ CH ₃ (Benzene, 1-methyl-2-(2-phenylethyl)-)	34403-05-9		11.2±0.3	EI	5230
	C ₆ H ₅ CH ₂ CH ₂ C ₆ H ₄ CH ₃ (Benzene, 1-methyl-3-(2-phenylethyl)-)	34403-06-0		11.1±0.2	EI	5230
	C ₆ H ₅ CH ₂ CH ₂ C ₆ H ₄ CH ₃ (Benzene, 1-methyl-4-(2-phenylethyl)-)	14310-20-4		11.3±0.4	EI	5230
	C ₆ H ₅ (CH ₂) ₃ C ₆ H ₅ (Benzene, 1,1'-(1,3-propanediyl)bis-)	1081-75-0		11.6	EI	4925
				11.6	EI	5230
	C ₆ H ₅ CH ₂ CH ₂ C ₆ H ₇ (1,3,5-Cycloheptatriene, 7-(2-phenylethyl)-)	712-32-6		9.3±0.1	EI	5230
	C ₆ H ₅ CH ₂ N(CH ₃) ₂ (Benzene-methanamine, dimethyl-)	28262-13-7	C ₂ H ₆ N	9.62	PI	5543
	C ₆ H ₅ CH ₂ CH ₂ N(CH ₃) ₂ (Benzene-ethanamine, dimethyl-)	29088-49-1	C ₃ H ₈ N	10.55	PI	5543
	C ₆ H ₅ (CH ₃) ₂ CHO (Benzaldehyde, 2,4-dimethyl-)	15764-16-6		11.2	EI	4051
	C ₆ H ₅ (CH ₃) ₂ CHO (Benzaldehyde, 2,5-dimethyl-)	5779-94-2		11.2	EI	4051
	C ₆ H ₅ (CH ₃) ₂ CHO (Benzaldehyde, 3,4-dimethyl-)	5973-71-7		11.1	EI	4051
	C ₆ H ₅ (CH ₃)COCH ₃ (Ethanone, 1-(4-methylphenyl)-)	122-00-9		13.8±0.3	EI	4223
	C ₆ H ₅ (CH ₃)COOH (Benzoic acid, 3-methyl-)	99-04-7	COOH	12.48±0.2	EI	3973
	C ₆ H ₅ (CH ₃)COOH (Benzoic acid, 4-methyl-)	99-94-5	COOH	12.55±0.2	EI	3973
	C ₆ H ₅ CH ₂ CH ₂ OCOCH ₃ (Acetic acid, 2-phenylethyl ester)	103-45-7		12.50	EI	3590
	CH ₃ C ₆ H ₄ NO ₂ (Benzene, 1-methyl-2-nitro-)	88-72-2	NO ₂	11.0±0.1	PI	5437
				13.1±0.3	EI	4223
	C ₆ H ₅ (NO ₂)CH ₃ (Benzene, 1-methyl-3-nitro-)	99-08-1	NO ₂	11.58±0.1	EI	3447
				12.1±0.3	EI	4223
	CH ₃ C ₆ H ₄ NO ₂ (Benzene, 1-methyl-4-nitro-)	99-99-0	NO ₂	11.3±0.1	PI	5437
			NO ₂	11.80±0.1	EI	3447
				12.3±0.3	EI	4223
	C ₆ H ₅ CH ₂ Cl (Benzene, chloromethyl-)	25168-05-2	Cl	10.16±0.05	PI	5515
	C ₆ H ₅ ClCH ₃ (Benzene, 1-chloro-2-methyl-)	95-49-8		11.21±0.1	EI	3777
	C ₆ H ₅ ClCH ₃ (Benzene, 1-chloro-3-methyl-)	108-41-8		11.34±0.1	EI	3777
	C ₆ H ₅ ClCH ₃ (Benzene, 1-chloro-4-methyl-)	106-43-4		11.42±0.1	EI	3777
				11.5±0.3	EI	4223
	C ₆ H ₅ BrCH ₃ (Benzene, 1-bromo-2-methyl-)	95-46-5		11.14±0.1	EI	3777
	C ₆ H ₅ BrCH ₃ (Benzene, 1-bromo-3-methyl-)	591-17-3		11.22±0.1	EI	3777
	C ₆ H ₅ BrCH ₃ (Benzene, 1-bromo-4-methyl-)	106-38-7		11.1±0.3	EI	4223
				11.22±0.1	EI	3777
	C ₆ H ₅ ICH ₃ (Benzene, 1-iodo-2-methyl-)	615-37-2		11.14±0.1	EI	3777
	C ₆ H ₅ ICH ₃ (Benzene, 1-iodo-3-methyl-)	625-95-6		11.0±0.3	EI	4223

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₇H₇⁺						
	C ₆ H ₅ ICH ₃	625-95-6		11.26±0.1	EI	3777
	C ₆ H ₅ ICH ₃ (Benzene, 1-iodo-4-methyl-)	624-31-7		11.15±0.1	EI	3777
C₇H₅D₂⁺						
	C ₆ H ₅ CD ₂ (Methyl-d ₂ , phenyl-)	2154-54-3	**	7.22±0.02	PE	4722
			**	7.22±0.02	PE	4898
C₇H₈⁺						
	C ₆ H ₅ (n-C ₃ H ₉) (Benzene, butyl-)	104-51-8	C ₃ H ₆	9.73±0.04	PI	4928
	C ₆ H ₅ (iso-C ₃ H ₉) (Benzene, (2-methylpropyl)-)	538-93-2	C ₃ H ₆	9.76±0.04	PI	4928
	C ₆ H ₅ (n-C ₅ H ₁₁) (Benzene, pentyl-)	538-68-1		9.72±0.04	PI	4928
	CH ₂ =C=C(CH ₃)CH=C=CH ₂	57212-57-4	**	8.39 (V)	PE	4397
	C ₆ H ₅ CH ₃ (Benzene, methyl-)	108-88-3	**	8.82	PI	3753
			**	8.72	PE	3955
			**	8.78±0.02	PE	3854
			**	8.80	PE	3868
			**	8.82	PE	4621
			**	8.82 (V)	PE	4280
			**	8.84	PE	5574
			**	8.85±0.015 (V)	PE	4107
			**	8.85 (V)	PE	4884
			**	8.900±0.03 (V)	PE	4340
			**	9.0±0.03 (V)	PE	3713
			**	9.00 (V)	PE	5258
			**	8.67	EI	3845
			**	8.80±0.1	EI	3788
			**	8.81	EI	4115
			**	8.82	EI	5293
			**	8.89±0.03	EI	3626
			**	8.71	CTS	3546
			**	8.91	CTS	4029
	C ₇ H ₈	121-46-0	**	8.6 (V)	PE	3724
	(Bicyclo[2.2.1]hepta-2,5-diene)					
			**	8.69 (V)	PE	3687
			**	8.69 (V)	PE	5538
			**	8.70 (V)	PE	3509
			**	8.73 (V)	PE	5010
			**	8.73 (V)	PE	5367
			**	8.69 (V)	PE	3824
	C ₇ H ₈	544-25-2	**	8.50 (V)	PE	5444
	(1,3,5-Cycloheptatriene)					
			**	8.52	EI	4115
	C ₇ H ₈	765-46-8	**	8.14	PE	3576
	(Spiro[2.4]hepta-4,6-diene)					
	C ₇ H ₈	XXXXX-XX-X	**	8.33 (V)	PE	4142
	(Tetracyclo[3.2.0.0 ^{2,7} .0 ^{1,6}]heptane)					
	C ₇ H ₈	35618-58-7	**	8.82 (V)	PE	5441
	(Tricyclo[4.1.0.0 ^{2,5}]hept-3-ene)					
	trans-CH ₂ =CHCH=CHCH=CHCH ₃	17679-93-5		12.4±0.15	PE	5432
	C ₆ H ₅ C ₆ H ₉	104-51-8	CH ₂ =CHCH ₃	10.10±0.1	EI	3629
	(Benzene, butyl-)					
	C ₆ H ₅ (CH ₂) ₃ C ₆ H ₅	1081-75-0		9.7±0.1	EI	4925
	(Benzene, 1,1'-(1,3-propanediyl)bis-)					
	C ₆ H ₅ CH ₂ CH ₂ C ₆ H ₅	712-32-6		10.0±0.1	EI	5230
	(1,3,5-Cycloheptatriene, 7-(2-phenylethyl)-)			9.0±0.1	EI	5230

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₇H₈⁺						
	C ₆ H ₅ (OCH ₃)CH ₃ (Benzene, 1-methoxy-3-methyl-)	100-84-5	CH ₂ O	11.22±0.1	EI	3446
	C ₆ H ₅ (OCH ₃)CH ₃ (Benzene, 1-methoxy-4-methyl-)	104-93-8	CH ₂ O	11.11±0.1	EI	3446
			HCHO	11.23	EI	3845
	(C ₆ H ₅ CH ₃)(CO) ₃ Cr (Chromium, tricarbonyl[(1,2,3,4,5,6- η)-methylbenzene]-)	12083-24-8		8.31±0.1	EI	3788
C₇H₈⁺²						
	C ₆ H ₅ CH ₃ (Benzene, methyl-)	108-88-3	**	24.2	OTH	5141
C₇H₉⁺						
	C ₇ H ₁₀ (Bicyclo[2.2.1]hept-2-ene)	498-66-8	H	11.0±0.01	EI	3535
	C ₇ H ₁₀ (Tricyclo[2.2.1.0 ^{2,6}]heptane)	279-19-6	H	11.3±0.01	EI	3535
	C ₁₀ H ₁₆ (Tricyclo[3.3.1.1 ^{3,7}]decane)	281-23-2		10.69	PI	4173
	C ₇ H ₉ Br (Bicyclo[2.2.1]hept-2-ene, 5-bromo-, <i>endo</i> -)	5810-82-2	Br	10.1	EI	5633
	C ₇ H ₉ Br (Bicyclo[2.2.1]hept-2-ene, 5-bromo-, <i>exo</i> -)	5889-54-3	Br	10.2	EI	5633
C₇H₁₀⁺						
	CH=CC(C ₂ H ₅)=CHCH ₃	14272-82-3	**	8.70±0.01	PE	5407
	C ₂ H ₅ C≡CC(CH ₃)=CH ₂	23056-94-2	**	8.66±0.01	PE	5407
	CH ₂ =CHC(CH ₃)=CHCH=CH ₂ -E	24587-26-6		8.28 (V)	PE	4380
	CH ₂ =C=C(CH ₃)C(CH ₃)=CH ₂	39968-66-6	**	8.10 (V)	PE	4829
	CH ₂ =C(CH ₃)CH=CHCH=CH ₂ -E	41233-72-1		8.31 (V)	PE	4380
	<i>trans</i> -CH ₂ =CHCH=CHCH=CHCH ₃	17679-93-5	**	7.96±0.02	PE	5432
			**	8.07	PE	3847
	C ₇ H ₁₀ (Bicyclo[2.2.1]hept-2-ene)	498-66-8	**	8.95	PE	5481
			**	8.95 (V)	PE	3509
			**	8.97 (V)	PE	3687
			**	8.97 (V)	PE	4249
			**	8.97 (V)	PE	4285
			**	8.97 (V)	PE	5538
			**	8.80±0.01	EI	3535
	C ₇ H ₁₀ (Bicyclo[4.1.0]hept-2-ene)	2566-57-6	**	8.69 (V)	PE	3849
	C ₇ H ₁₀ (1,3-Cycloheptadiene)	4054-38-0	**	8.31±0.03 (V)	PE	4828
	C ₆ H ₄ =CH ₂ (Cyclohexene, 4-methylene-)	13407-18-6	**	9.27 (V)	PE	4249
	C ₅ H ₆ (=CH ₂) ₂ (Cyclopentane, 1,2-bis(methylene)-)	20968-70-1	**	8.58	PE	5265
	C ₃ H ₅ C(CH ₃)=C=CH ₂ (Cyclopropane, (1-methyl-1,2-propadienyl)-)	51549-86-1	**	8.83	PE	4608
	C ₇ H ₁₀ (Spiro[2.4]hept-4-ene)	52708-23-3	**	8.48 (V)	PE	4347
	C ₇ H ₁₀ (Tricyclo[2.2.1.0 ^{2,6}]heptane)	279-19-6	**	9.40 (V)	PE	3741
			**	8.92±0.01	EI	3535
	C ₇ H ₁₀ (Tricyclo[4.1.0.0 ^{2,7}]heptane)	287-13-8	**	8.72 (V)	PE	4400
			**	8.72 (V)	PE	5441
	C ₁₀ H ₁₅ CH ₃ (4,7-Methano-1H-indene, octahydro-8-methyl-, stereoisomer)	50745-92-1		9.5±0.1	PI	3918

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₇H₁₁⁺						
	C ₁₀ H ₁₆ (4,7-Methano-1H-indene, octahydro-, (3α,4β,7β,7α)-)	2825-82-3		9.9±0.1	PI	3918
	C ₁₀ H ₁₅ CH ₃ (2-Methyl-exo-tricyclo[5.2.1.0 ^{2,6}]decane)	XXXXX-XX-X		<10.2±0.1	PI	3918
	C ₁₀ H ₁₅ CH ₃ (4,7-Methano-1H-indene, octahydro-2-methyl-, (2α,3αβ,4α,7α,7αβ)-)	50745-90-9		10.0±0.1	PI	3918
	C ₁₀ H ₁₅ C ₂ H ₅ (4,7-Methano-1H-indene, 5-ethyloctahydro-, (3αα,4β,5α,7β,7αα)-)	32787-97-6		<10.2±0.1	PI	3918
C₇H₁₂⁺						
	C ₅ H ₇ (C ₂ H ₅) (Cyclopentene, 3-ethyl-)	694-35-9	**	8.91±0.05 (V)	PE	4954
			**	8.88±0.01	PI	5556
	C ₅ H ₁₁ C≡CH	628-71-7	**	10.044±0.005	PE	4575
	CH ₂ =CHCH ₂ CH=C(CH ₃) ₂	763-88-2	**	8.70 (V)	PE	4211
	(CH ₃) ₂ C=C=C(CH ₃) ₂	1000-87-9	**	8.47 (V)	PE	4019
			**	8.53 (V)	PE	5362
	(C ₂ H ₅) ₂ C(CH ₃) ₂	1112-35-2	**	9.55 (V)	PE	3994
	C ₄ H ₉ C≡CCH ₃	1119-65-9	**	9.326±0.005	PE	4575
	(CH ₃) ₂ CH(CH ₂) ₂ C≡CH	2203-80-7	**	10.015±0.005	PE	4575
	C ₂ H ₅ C≡CC ₃ H ₇	2586-89-2	**	9.260±0.005	PE	4575
	CH ₂ =CH(CH ₂) ₃ CH=CH ₂	3070-53-9	**	9.52±0.02 (V)	PE	4010
	(CH ₃) ₂ CHCH ₂ C≡CCH ₃	53566-37-3	**	9.320±0.005	PE	4575
	(tert-C ₄ H ₉)C≡CCH ₃	999-78-0	**	9.276±0.010	PE	4575
	C ₇ H ₁₂ (Bicyclo[2.2.1]heptane)	279-23-2	**	10.15 (V)	PE	3509
			**	10.2 (V)	PE	3687
	C ₇ H ₁₂ (Bicyclo[4.1.0]heptane)	286-08-8	**	9.46 (V)	PE	3849
	C ₇ H ₁₂ (Cycloheptene(Z))	628-92-2	**	9.05±0.15	EI	5532
			**	9.12 (V)	PE	4285
	C ₆ H ₁₀ =CH ₂ (Cyclohexane, methylene-)	1192-37-6	**	9.12±0.02 (V)	PE	4338
			**	9.13 (V)	PE	4249
			**	9.7	EI	5586
	CH ₃ C ₆ H ₉ (Cyclohexene,1-methyl-)	591-49-1	**	8.67±0.02	PI	5556
	CH ₃ C ₆ H ₉ (Cyclohexene,3-methyl-)	591-48-0	**	8.89±0.01	PI	5556
	CH ₃ C ₆ H ₉ (Cyclohexene,4-methyl-)	591-47-9	**	8.91±0.01	PI	5556
	C ₂ H ₅ C ₅ H ₇ (Cyclopentene,1-ethyl-) (Cyclopentene,3-ethyl-)	2146-38-5	**	8.53±0.01	PI	5556
	C ₇ H ₁₂ (Cyclopropene,tetramethyl)	26385-95-5	**	8.52 (V)	PE	5480
C₇H₁₃⁺						
	C ₆ H ₁₀ (CH ₃) ₂ (Cyclohexane, 1,2-dimethyl-, cis-)	2207-01-4	CH ₃	10.55±0.05	EI	3581
	C ₆ H ₁₀ (CH ₃) ₂ (Cyclohexane, 1,2-dimethyl-, trans-)	6876-23-9	CH ₃	10.73±0.05	EI	3581
C₇H₁₄⁺						
	trans-(CH ₃) ₃ CCH=CHCH ₂	690-08-4	**	8.908±0.008	PE	3957
	(CH ₃) ₃ CC(CH ₃)=CH ₂	594-56-9	**	9.016±0.007	PE	3957
	(CH ₃) ₃ CCH ₂ CH=CH ₂	762-62-9	**	9.399±0.003	PE	3957
			**	9.6 (V)	PE	3940
	(CH ₃) ₂ CHCH ₂ C(CH ₃)=CH ₂	2213-32-3	**	9.025±0.005	PE	3957
	CH ₃ (CH ₂) ₃ C(CH ₃)=CH ₂	6094-02-6	**	9.039±0.005	PE	3957

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₇H₁₁⁺						
	C ₅ H ₅ C(CH ₃)=C(CH ₃) ₂	10574-37-5	**	8.213±0.005	PE	3957
	1-C ₇ H ₁₁	592-76-7	**	9.27±0.02	PE	4695
			**	9.442±0.003	PE	3957
	2-C ₇ H ₁₁	592-77-8	**	8.84±0.02	PE	4695
	3-C ₇ H ₁₁	592-78-9	**	8.77±0.02	PE	4695
	cis-(CH ₃) ₃ CCH=CHCH ₃	762-63-0	**	8.922±0.008	PE	3957
	cis-(CH ₃) ₂ CHCH ₂ CH=CHCH ₃	13151-17-2	**	8.917±0.005	PE	3957
	trans-CH ₃ CH ₂ C(CH ₃)HCH=CHCH ₃	3683-22-5	**	8.912±0.005	PE	3957
	trans-(CH ₃) ₂ CHCH ₂ CH=CHCH ₃	7385-82-2	**	8.919±0.005	PE	3957
	C ₇ H ₁₁ (Cycloheptane)	291-64-5	**	9.97	PE	4319
			**	9.88±0.05	EI	4319
	CH ₃ C ₆ H ₁₁ (Cyclohexane,methyl-)	108-87-2	**	9.76±0.03	PI	5556
	C ₂ H ₅ C ₅ H ₉ (Cyclopentane,ethyl-)	1640-89-7	**	10.12±0.02	PI	5556
C₇H₁₅⁺						
	1-C ₇ H ₁₅	3356-67-0	**	7.90±0.06	EI	4895
	2-C ₇ H ₁₅	3474-30-4	**	7.35±0.06	EI	4895
	n-C ₄ H ₉ C(CH ₃) ₂	40626-78-6	**	6.79	EI	4895
C₈H₂⁺	CH≡CC≡CC≡CC≡CH	XXXXX-XX-X	**	9.09±0.02	PE	4460
C₈H₆⁺						
	C ₆ H ₆	XXXXX-XX-X	**	8.95±0.1	EI	4714
	CH ₃ C≡CC≡CC≡CCH ₃	1072-20-4	**	8.60	PE	4048
	C ₆ H ₅ C≡CH (Benzene, ethynyl-)	536-74-3	**	8.75	PE	3938
			**	8.78 (V)	PE	4334
			**	8.78 (V)	PE	5259
			**	8.82±0.02 (V)	PE	5409
			**	8.88±0.02 (V)	PE	3854
	C ₆ H ₆ (Bicyclo[4.2.0]octa-1,3,5,7-tetraene)	4026-23-7	**	7.87±0.02 (V)	PE	4945
C₈H₈⁺						
	C ₆ H ₅ CH=CH ₂ (Benzene, ethenyl-)	100-42-5	**	8.40±0.02	PE	3854
			**	8.42	PE	3938
			**	8.48 (V)	PE	4884
			**	8.49 (V)	PE	3964
			**	8.50 (V)	PE	4347
			**	8.55 (V)	PE	3781
			**	8.55 (V)	PE	5632
			**	8.23±0.1	EI	4714
			**	8.28±0.04	EI	4097
	C ₈ H ₈	37846-63-2	**	8.50 (V)	PE	3933
	(Bicyclo[2.2.1]hepta-2,5-diene, 7-methylene-)					
	C ₈ H ₈	694-87-1	**	8.66±0.03 (V)	PE	4828
	(Bicyclo[4.2.0]octa-1,3,5-triene)					
	C ₄ (=CH ₂) ₄ (Cyclobutane, tetrakis(methylene)-)	3227-91-6	**	8.66 (V)	PE	4063
	C ₆ H ₄ (=CH ₂) ₂ (1,4-Cyclohexadiene,3,6-bis(methylene)-)	502-86-3	**	8.35	PE	4728
			**	7.87±0.05 (V)	PE	4510
	C ₈ H ₈	49852-40-6	**	8.9	PE	4180
	(1,5-Cyclooctadiyne)					
	C ₈ H ₈	629-20-9	**	8.0	PE	3999
	(1,3,5,7-Cyclooctatetraene)					

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₈H₈⁺						
	C ₈ H ₈ (Pentacyclo[3.3.0.0 ^{2,4} .0 ^{3,7} .0 ^{6,8}]octane)	20656-23-9	**	8.18	PE	4955
	C ₈ H ₈ (Pentacyclo[4.2.0.0 ^{2,5} .0 ^{3,8} .0 ^{4,7}]octane)	277-10-1	**	8.46	PE	4955
			**	8.4±<0.1	EI	3735
			**	9.6	PE	4726
	C ₈ H ₈ (Tetracyclo[3.2.0.0 ^{2,7} .0 ^{3,6}]heptane, 3-methylene-)	38898-42-9	**	8.48 (V)	PE	4142
	C ₈ H ₈ (Tricyclo[3.2.1.0 ^{2,8}]octa-2,6-diene)	XXXXX-XX-X	**	8.5 (V)	PE	4034
	C ₈ H ₈ (Tricyclo[4.2.0.0 ^{2,5}]octa-3,7-diene, <i>syn</i> -)	20380-30-7	**	9.08 (V)	PE	4045
			**	9.08 (V)	PE	4258
	C ₈ H ₈ (Tricyclo[4.2.0.0 ^{2,5}]octa-3,7-diene, <i>anti</i> -)	20380-31-8	**	8.90 (V)	PE	4258
			**	8.96 (V)	PE	4045
	C ₆ H ₅ CH ₂ CH ₂ C ₆ H ₄ CH ₃ (Benzene, 1-methyl-2-(2-phenylethyl)-)	34403-05-9		10.0±0.2	EI	5230
	C ₆ H ₅ CH ₂ CH ₂ C ₆ H ₄ CH ₃ (Benzene, 1-methyl-3-(2-phenylethyl)-)	34403-06-0		10.1±0.2	EI	5230
	C ₆ H ₅ CH ₂ CH ₂ C ₆ H ₄ CH ₃ (Benzene, 1-methyl-4-(2-phenylethyl)-)	14310-20-4		10.6±0.5	EI	5230
	C ₆ H ₅ CH ₂ CH ₂ CH ₂ C ₆ H ₅ (Benzene, 1,1'-(1,3-propanediyl)bis-)	1081-75-0		10.0±0.2	EI	5230
	C ₆ H ₅ CH ₂ CH ₂ C ₇ H ₇ (1,3,5-Cycloheptatriene, 7-(2-phenylethyl)-)	712-32-6		9.3±0.2	EI	5230
	C ₁₀ H ₁₁ OH (2-Naphthalenol, 1,2,3,4-tetrahydro-)	530-91-6	C ₂ H ₄ O	11.68±0.04	EI	4960
	C ₆ H ₅ CH ₂ CH ₂ OCOCH ₃ (Acetic acid, 2-phenylethyl ester)	103-45-7		8.90	EI	3590
C₈H₉⁺						
	C ₆ H ₅ (CH ₃) ₂ (Benzene, 1-2-dimethyl-)	95-47-6	H	12.10±0.2	EI	4199
	C ₆ H ₄ (CH ₃) ₂ (Benzene, 1,3-dimethyl-)	108-38-8	H	12.25±0.2	EI	4199
	C ₆ H ₄ (CH ₃) ₂ (Benzene, 1,4-dimethyl-)	106-42-3	H	12.10±0.2	EI	4199
	C ₆ H ₅ C ₂ H ₅ (Benzene, ethyl-)	100-41-4	H	10.60	EI	5293
	C ₆ H ₅ CH(CH ₃) ₂ (Benzene, 1-methylethyl-)	98-82-8	CH ₃	10.02	EI	5293
	C ₆ H ₅ CH ₂ CH ₂ CH ₃ (Benzene, propyl-)	103-65-1	CH ₃	9.98	EI	5293
	C ₆ H ₅ C ₄ H ₉ (Benzene, butyl-)	104-51-8	C ₂ H ₅	9.98	EI	5293
	C ₆ H ₅ CH(CH ₃)C ₂ H ₅ (Benzene, 1-methylpropyl)	135-98-8	C ₂ H ₅	9.93	EI	5293
	C ₆ H ₄ (CH ₃)C ₄ H ₉ (Benzene, 1-butyl-3-methyl-)	1595-04-6		11.43±0.1	EI	3629
	C ₆ H ₄ (CH ₃)C ₄ H ₉ (Benzene, 1-butyl-4-methyl-)	1595-05-7		11.03±0.1	EI	3629
	C ₆ H ₅ CH ₂ CH ₂ C ₆ H ₄ CH ₃ (Benzene, 1-methyl-2-(2-phenylethyl)-)	34403-05-9		10.15±0.1	EI	5230
	C ₆ H ₅ CH ₂ CH ₂ C ₆ H ₄ CH ₃ (Benzene, 1-methyl-3-(2-phenylethyl)-)	34403-06-0		10.35±0.1	EI	5230
	C ₆ H ₅ CH ₂ CH ₂ C ₆ H ₄ CH ₃ (Benzene, 1-methyl-4-(2-phenylethyl)-)	14310-20-4		10.0±0.1	EI	5230
	C ₆ H ₅ CH ₂ CH ₂ C ₆ H ₄ CH ₃ (Benzene, 1,1'-(1,3-propanediyl)bis-)	1081-75-0		10.4±0.2	EI	5230
	C ₆ H ₅ CH ₂ CH ₂ C ₇ H ₇ (1,3,5-Cycloheptatriene, 7-(2-phenylethyl)-)	712-32-6		8.95±0.05	EI	5230

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₈H₉⁺						
	CH ₃ C ₆ H ₄ CH ₂ N(CH ₃) ₂ (Benzene-methanamine, N,N, <i>ar</i> -trimethyl-)	56927-89-0	C ₂ H ₆ N	10.92	PI	5543
	C ₆ H ₄ (CH ₃)CH ₂ CH ₂ OCOCH ₃ (Phenethyl alcohol, <i>m</i> -methyl-, acetate)	33709-40-9		12.30	EI	3590
	C ₆ H ₄ (CH ₃)CH ₂ CH ₂ OCOCH ₃ (Phenethyl alcohol, <i>p</i> -methyl-, acetate)	22532-47-4		11.80	EI	3590
C₈H₁₀⁺						
	C ₇ H ₇ (CH ₃) (Tricyclo[4.1.0.0 ^{2,7}]hept-3-ene, 1-methyl-)	61772-33-6	**	8.45 (V)	PE	5441
	C ₇ H ₇ (CH ₃) (Tricyclo[4.1.0.0 ^{2,7}]hept-3-ene, 6-methyl-)	61772-31-4	**	8.64 (V)	PE	5441
	(C ₂ H ₅ C≡C) ₂	16387-70-5	**	8.78	PE	4731
	<i>trans</i> -1,3,5,7-C ₈ H ₁₀	3725-31-3	**	7.79±0.02	PE	4846
	C ₆ H ₄ (CH ₃) ₂ (Benzene, 1,2-dimethyl-)	95-47-6	**	8.45±0.02	PE	3854
			**	8.57±0.03 (V)	PE	4828
			**	8.57 (V)	PE	4063
			**	8.75±0.03 (V)	PE	3713
			**	8.55±0.1	EI	3788
			**	8.85±0.05	EI	4199
			**	8.61	CTS	3546
			**	8.70	CTS	4029
	C ₆ H ₄ (CH ₃) ₂ (Benzene, 1,3-dimethyl-)	108-38-3	**	8.50±0.02	PE	3854
			**	8.55 (V)	PE	4231
			**	8.71±0.015 (V)	PE	4107
			**	8.75±0.03 (V)	PE	3713
	C ₆ H ₄ (CH ₃) ₂ (Benzene, 1,3-dimethyl-)	108-38-8	**	8.90±0.05	EI	4199
	C ₆ H ₄ (CH ₃) ₂ (Benzene, 1,4-dimethyl-)	106-42-3	**	8.37±0.02	PE	3854
			**	8.43 (V)	PE	4231
			**	8.44	PE	5574
			**	8.6±0.03 (V)	PE	3713
			**	8.80±0.05	EI	4199
	C ₆ H ₅ C ₂ H ₅ (Benzene, ethyl-)	100-41-4	**	8.76	EI	5293
	C ₈ H ₁₀	694-91-7	**	8.93 (V)	PE	3824
	(Bicyclo[2.2.1]hept-2-ene, 5-methylene-)		**	9.01 (V)	PE	4249
	C ₈ H ₁₀	61885-53-8	**	8.11 (V)	PE	4723
	(Bicyclo[4.1.1]octa-2,4-diene)		**	9.00 (V)	PE	4249
	C ₆ H ₆ (=CH ₂) ₂ (Cyclohexene, 4,5-bis(methylene)-)	54290-41-4	**			
	CH≡CC ₆ H ₉ (Cyclohexene, 1-ethynyl-)	931-49-7	**	8.61±0.01	PE	5407
	C ₈ H ₁₀ (1,3,5-Cyclooctatriene)	1871-52-9	**	7.9	PE	3999
	C ₈ H ₁₀ (1,3,6-Cyclooctatriene)	3725-30-2	**	8.5	PE	3999
	C ₈ H ₁₀ (1-Cycloocten-5-yne)	68177-00-4	**	8.90	PE	5053
	C ₅ H ₄ =C(CH ₃) ₂ (1,3-Cyclopentadiene, 5-(1-methylethylidene)-)	2175-91-9	**	8.03 (V)	PE	4357
	C ₈ H ₁₀ (Dicyclopenta[cd,gh]pentalene, octahydro-)	765-72-0	**	9.12±0.02 (V)	PE	4338
	C ₈ H ₁₀ (Spiro[2.5]octa-4,6-diene)	53143-64-9	**	7.89 (V)	PE	5359
	C ₈ H ₁₀ (Spiro[3.4]octa-5,7-diene)	15439-15-3	**	8.20	PE	4268

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₈H₁₀⁺						
	C ₈ H ₁₀ (Tricyclo[3.2.1.0 ^{2,4}]oct-6-ene, (1 α ,2 α ,4 α ,5 α)-)	3635-94-7	**	9.05 (V)	PE	3509
	C ₈ H ₁₀ (Tricyclo[3.2.1.0 ^{2,4}]oct-6-ene, (1 α ,2 β ,4 β ,5 α)-)	3635-95-8	**	8.90 (V)	PE	3509
	C ₈ H ₁₀ (Tricyclo[3.2.1.0 ^{2,8}]oct-6-ene)	XXXXX-XX-X	**	8.5 (V)	PE	4034
	C ₈ H ₁₀ (Tricyclo[3.3.0.0 ^{2,6}]octene)	53754-35-1	**	8.63 (V)	PE	4259
	C ₈ H ₁₀ (Tricyclo[4.2.0.0 ^{2,5}]oct-3-ene, (1 α ,2 β ,5 β ,6 α)-)	39781-76-5	**	9.25 (V)	PE	4045
	C ₆ H ₅ (CH ₃)C ₄ H ₉ (Benzene, 1-butyl-3-methyl-)	1595-04-6	CH ₂ =CHCH ₃	10.33±0.1	EI	3629
	C ₆ H ₅ (CH ₃)C ₄ H ₉ (Benzene, 1-butyl-4-methyl-)	1595-05-7	CH ₂ =CHCH ₃	10.14±0.1	EI	3629
	CH ₃ C ₆ H ₄ CH ₂ N(CH ₃) ₂ (Benzene-methanamine,N,N, <i>ar</i> -trimethyl-)	56927-89-0	C ₂ H ₅ N	9.6	PI	5543
	(C ₆ H ₄ (CH ₃) ₂) ₂ Cr (Chromium, tricarbonyl[(1,2,3,4,5,6- η)-1,2-dimethylbenzene]-)	12129-29-2		8.51±0.1	EI	3788
C₈H₁₁⁺						
	C ₁₀ H ₁₅ C ₂ H ₅ (4,7-Methano-1 <i>H</i> -indene, 5-ethyloctahydro-, (3 α α ,4 β ,5 α ,7 β ,7 α)-)	32787-97-6		9.9±0.1	PI	3918
C₈H₁₂⁺						
	((CH ₃) ₂ C=C) ₂	2431-31-4	**	7.70	PE	5034
	(C ₂ H ₅) ₂ C=CHC≡CH	2750-71-2	**	8.54±0.01	PE	5407
	CH ₃ CH=CHCH=CHCH=CHCH ₃ -E,E,E	15192-80-0		7.95 (V)	PE	4380
	(CH ₃) ₂ C=CHCH=CHCH=CH ₂ -E,E	16895-46-8		7.88 (V)	PE	4380
	C ₃ H ₇ C≡CC(CH ₃)=CH ₂	17669-40-8	**	8.62±0.01	PE	5407
	C ₄ H ₉ C≡CCH=CH ₂	17679-92-4	**	8.83±0.01	PE	5407
	CH ₃ CH=C(CH ₃)CH=CHCH=CH ₂ -E,E	58434-77-8		8.01 (V)	PE	4380
	cis-CH≡CCH=CH(CH ₂) ₃ CH ₃	42091-89-4	**	8.91±0.01	PE	5407
	cis-CH ₃ C≡CC(C ₂ H ₅)=CHCH ₃	70058-02-5	**	8.28±0.01	PE	5407
	trans-CH≡CCH=CH(CH ₂) ₃ CH ₃	42104-42-7	**	8.87±0.01	PE	5407
	trans-CH ₃ C≡CC(C ₂ H ₅)=CHCH ₃	70058-03-6	**	8.23±0.01	PE	5407
	C ₈ H ₁₂ (Bicyclo[2.2.1]heptane, 2-methylene-)	497-35-8	**	9.02 (V)	PE	3824
			**	9.04 (V)	PE	4249
	C ₈ H ₁₂ (Bicyclo[2.2.1]heptane, 7-methylene-)	31463-35-1	**	9.40 (V)	PE	3933
	C ₈ H ₁₂ (Bicyclo[2.2.2]oct-2-ene)	931-64-6	**	9.03 (V)	PE	4285
			**	9.05±0.02 (V)	PE	4842
			**	9.07 (V)	PE	4249
	C ₈ H ₁₂ (Bicyclo[4.1.1]oct-3-ene)	61885-54-9	**	8.90 (V)	PE	4723
	C ₄ H ₅ (CH ₃) ₂ C≡CH (Cyclobutane,3-ethynyl-1,1-dimethyl)	66438-88-8	**	9.78 (V)	PE	5607
	C ₆ H ₈ (=CH ₂) ₂ (Cyclohexane,1,2-bis(methylene)-)	2819-48-9	**	8.90	PE	5265
			**	8.92 (V)	PE	4249
	C ₆ H ₁₀ =C=CH ₂ (Cyclohexane,ethenylidene-)	5664-20-0	**	8.69	PE	5625
	C ₆ H ₁₁ C≡CH (Cyclohexane, ethynyl-)	931-48-6	**	9.92 (V)	PE	3997
	C ₈ H ₁₂ (1,5-Cyclooctadiene-(E,Z)-)	5259-71-2	**	8.7 (V)	PE	5372
	C ₈ H ₁₂ (1,3-Cyclooctadiene)	1700-10-3	**	8.4	PE	3999
	C ₈ H ₁₂ (1,4-Cyclooctadiene)	1073-07-0	**	8.5	PE	3999

Table of Ion Energetics Measurements—Continued

on (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₈H₁₂⁺						
	C ₈ H ₁₂ (1,5-Cyclooctadiene)	111-78-4	**	8.9	PE	3999
	C ₈ H ₁₂ (Cyclooctyne)	1781-78-8	**	8.9	PE	4180
			**	9.10 (V) 8.60±0.01	PE PI	4362 5556
	CH ₂ =CHCH ₂ C ₅ H ₇ (Cyclopentene,1-(2-propenyl)-)	37689-19-3	**	8.89±0.02	PI	5556
	CH ₂ =CHCH ₂ C ₅ H ₇ (Cyclopentene,3-(2-propenyl)-)	14564-97-7	**	7.72	PI	3759
	C ₃ H ₄ CH ₃ (CH=C=CHCH ₃) (Cyclopropane, 1-(1,2-butadienyl)-2-methyl-cis-)	60166-71-4	**	8.96	PE	4608
	C ₃ H ₅ CH=CHC ₃ H ₅ (Cyclopropane, 1,1'-(1,2-ethenediyl)bis-(E))	10359-44-1	**	7.70	PI	3759
	C ₃ H ₅ CH=CHC ₃ H ₅ (Cyclopropane, 1,1'-(1,2-ethenediyl)bis-(Z))	23510-65-6	**	8.08	PI	3759
	(C ₃ H ₅) ₂ C=CH ₂ (Cyclopropane, 1,1'-ethenylidenebis-)	822-93-5	**	8.87	PE	4608
			**	8.78	PE	4608
	C ₈ H ₅ CH=C=C(CH ₃) ₂ (Cyclopropane, (3-methyl-1,2-butadienyl)-)	60166-72-5	**	8.81	PE	4608
	C ₈ H ₁₂ (Cyclopropane, 1-methyl-1-(1-methyl-1,2-propadienyl)-)	60166-69-0	**	9.02 (V)	PE	5361
	C ₈ H ₁₂ (Dispiro[2.0.2.2]octane)	21426-37-9	**	9.21 (V)	PE	5361
	C ₈ H ₁₂ (Dispiro[2.1.2.1]octane)	25399-32-0	**	8.44 (V)	PE	5359
	C ₈ H ₁₂ (Spiro[2.5]oct-4-ene)	7647-57-6	**	8.65	PE	4268
	C ₈ H ₁₂ (Spiro[3.4]oct-5-ene)	14783-50-7	**	8.89 (V) 8.20 (V)	PE	4347 5441
	C ₇ H ₉ CH ₃ (Tricyclo[4.1.0.0 ^{2,7}]heptane,1-methyl-)	32348-63-3	**	8.42 (V)	PE	5441
	C ₇ H ₉ CH ₃ (Tricyclo[4.1.0.0 ^{2,7}]heptane,2-methyl-)	40391-49-9	**	9.40 (V)	PE	3509
	C ₈ H ₁₂ (Tricyclo[3.2.1.0 ^{2,4}]octane, (1 α ,2 α ,4 α ,5 α -))	22389-16-8	**	9.1±0.1	EI	3492
			**	9.40 (V)	PE	3509
	C ₈ H ₁₂ (Tricyclo[3.2.1.0 ^{2,4}]octane, (1 α ,2 β ,4 β ,5 α -))	13377-46-3	**	9.1±0.1	EI	3492
			**	9.78 (V)	PE	4259
	C ₈ H ₁₂ (Tricyclo[3.3.0.0 ^{2,6}]octane)	28636-10-4	**	9.18 (V)	PE	4045
	C ₈ H ₁₂ (Tricyclo[4.2.0.0 ^{2,5}]octane, syn-)	13027-75-3	**	9.23 (V)	PE	4045
	C ₈ H ₁₂ (Tricyclo[4.2.0.0 ^{2,5}]octane, anti-)	50695-42-6	**	8.95 (V)	PE	3849
	C ₈ H ₁₂ (Tricyclo[5.1.0.0 ^{2,4}]octane, (1 α ,2 α ,4 α ,7 α -))	50895-58-4	**	9.39 (V)	PE	3849
	C ₁₀ H ₁₆ (4,7-Methano-1H-indene, octahydro-, (3 α ,4 β ,7 β ,7 α -))	2825-82-3	10.5±0.1	PI	3918	
	C ₁₀ H ₁₅ CH ₃	XXXXX-XX-X		10.0±0.1	PI	3918
C₉H₁₃⁺						
	C ₁₀ H ₁₅ CH ₃ (4,7-Methano-1H-indene, octahydro-2-methyl-, (2 α ,3 α β ,4 α ,7 α ,7 β -))	50745-90-9	10.1±0.1	PI	3918	
	C ₁₀ H ₁₅ CH ₃ (4,7-Methano-1H-indene, octahydro-8-methyl-, stereoisomer)	50745-92-1	9.5±0.1	PI	3918	

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₈H₁₄⁺						
	C ₅ H ₇ ((CH ₂) ₂ CH ₃) (Cyclopentene, 3-(1-methylethyl)-)	4276-45-3	**	8.85±0.05 (V)	PE	4954
			**	8.81±0.02	PI	5556
	(tert-C ₄ H ₉)CH ₂ C≡CH ₃	56617-18-6	**	9.284±0.007	PE	4575
	(CH ₃) ₂ C=CHCH=C(CH ₃) ₂	764-13-6	**	7.65	PE	3847
	(CH ₃) ₂ CHC=CCH(CH ₃) ₂	927-99-1	**	9.171±0.008	PE	4575
	C ₃ H ₇ C≡CC ₃ H ₇	1942-45-6	**	9.196±0.005	PE	4575
			**	9.20±0.02	PI	5583
	C ₅ H ₁₁ C≡CCH ₃	2809-67-8	**	9.302±0.005	PE	4575
			**	9.31±0.02	PI	5583
	CH ₂ =CH(CH ₂) ₄ CH=CH ₂	3710-30-3	**	9.52±0.02 (V)	PE	4010
	C ₂ H ₅ C≡CC ₄ H ₉	15232-76-5	**	9.222±0.005	PE	4575
			**	9.22±0.02	PI	5583
	1-C ₈ H ₁₄	629-05-0	**	9.95±0.02	PI	5583
	(tert-C ₄ H ₉)C≡CC ₂ H ₅	4911-60-8	**	9.180±0.010	PE	4575
	C ₈ H ₁₄ (Bicyclo[2.2.2]octane)	280-33-1	**	9.43	S	3757
			**	9.45±0.02	PE	3757
	C ₈ H ₁₄ (Bicyclo[4.1.1]octane)	7078-34-4	**	10.0 (V)	PE	4723
	C ₄ H ₅ (CH ₃) ₂ CH=CH ₂ (Cyclobutane, 3-ethenyl-1,1-dimethyl-)	52708-22-2	**	9.40 (V)	PE	4347
			**	9.40 (V)	PE	5607
	C ₆ H ₁₁ CH=CH ₂ (Cyclohexane, ethenyl-)	695-12-5	**	9.51	PE	4347
	C ₂ H ₅ C ₆ H ₉ (Cyclohexene, 1-ethyl-)	1453-24-3	**	8.48±0.01	PI	5556
	C ₂ H ₅ C ₆ H ₉ (Cyclohexene, 3-ethyl-)	2808-71-1	**	8.83±0.01	PI	5556
	C ₂ H ₅ C ₆ H ₉ (Cyclohexene, 4-ethyl-)	3742-42-5	**	8.88±0.01	PI	5556
	C ₈ H ₁₄ (Cyclooctene)	931-88-4	**	8.8	PE	3999
			**	9.02 (V)	PE	4285
	n-C ₃ H ₇ C ₅ H ₇ (Cyclopentene, 1-propyl)	3074-61-1	**	8.48±0.01	PI	5556
	n-C ₃ H ₇ C ₅ H ₇ (Cyclopentene, 3-propyl-)	34067-75-9	**	8.84±0.02	PI	5556
	C ₈ H ₁₄ (Spiro[2.5]octane)	185-65-9	**	9.46 (V)	PE	5359
	C ₈ H ₁₄ (Spiro[3.4]octane)	175-56-4	**	9.45	PE	4268
C₈H₁₆⁺						
	(CH ₃) ₃ CCH ₂ C(CH ₃)=CH ₂	107-39-1	**	8.909±0.005	PE	3957
	(CH ₃) ₂ CHC(CH ₃)=C(CH ₃) ₂	565-77-5	**	8.165±0.005	PE	3957
	C ₂ H ₅ CH ₂ C(CH ₃)=C(CH ₃) ₂	7145-20-2	**	8.186±0.005	PE	3957
	(C ₂ H ₅) ₂ C=CHC ₂ H ₅	16789-51-8	**	8.480±0.004	PE	3957
	(C ₂ H ₅) ₂ C=C(CH ₃) ₂	19780-67-7	**	8.170±0.003	PE	3957
	1-C ₈ H ₁₆	111-66-0	**	9.427±0.006	PI	5584
			**	9.60±0.01 (V)	PE	4939
	cis-(CH ₃) ₂ CHCH=CHCH(CH ₃) ₂	10557-44-5	**	8.846±0.005	PE	3957
	cis-C ₂ H ₅ C(CH ₃)=C(CH ₃)C ₂ H ₅	19550-87-9	**	8.172±0.003	PE	3957
	cis-2-C ₈ H ₁₆	7642-04-8	**	8.913±0.009	PI	5584
			**	9.10±0.01 (V)	PE	4939
	cis-3-C ₈ H ₁₆	14850-22-7	**	8.859±0.008	PI	5584
			**	8.849±0.005	PE	3957
			**	9.05±0.01 (V)	PE	4939
	cis-4-C ₈ H ₁₆	7642-15-1	**	8.836±0.006	PI	5584
			**	8.841±0.005	PE	3957
			**	9.03±0.01 (V)	PE	4939

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₈H₁₆⁺						
	<i>trans</i> -(CH ₃) ₂ CHCH=CHCH(CH ₃) ₂	692-70-6	**	8.838±0.005	PE	3957
	<i>trans</i> -C ₂ H ₅ C(CH ₃)=C(CH ₃)C ₂ H ₅	19550-88-0	**	8.156±0.003	PE	3957
	<i>trans</i> -2-C ₈ H ₁₆	13389-42-9	**	8.913±0.006	PI	5584
			**	9.09±0.01 (V)	PE	4939
	<i>trans</i> -3-C ₉ H ₁₆	14919-01-8	**	8.854±0.006	PI	5584
			**	9.03±0.01 (V)	PE	4939
	<i>trans</i> -4-C ₉ H ₁₆	14850-23-8	**	8.836±0.006	PI	5584
			**	8.830±0.005	PE	3957
			**	9.01±0.01 (V)	PE	4939
	C ₆ H ₁₀ (CH ₃) ₂ (Cyclohexane, 1,2-dimethyl-, <i>cis</i> -)	2207-01-4	**	9.90±0.07	EI	3581
	C ₆ H ₁₀ (CH ₃) ₂ (Cyclohexane, 1,2-dimethyl-, <i>trans</i> -)	6876-23-9	**	10.03±0.05	EI	3581
	C ₆ H ₅ C ₆ H ₁₁ (Cyclohexane, ethyl-)	1678-91-7	**	9.67±0.02	PI	5556
	C ₈ H ₁₆ (Cyclooctane)	292-64-8	**	9.7	PE	3999
			**	9.80	PE	4319
			**	10.08±0.05	EI	4319
	<i>n</i> -C ₃ H ₇ C ₅ H ₉ (Cyclopentane, propyl-)	2040-96-2	**	10.00±0.04	PI	5556
C₉H₇⁺						
	C ₆ H ₅ C≡CCH ₃ (Benzene, 1-propynyl-)	673-32-5		11.42±0.05	EI	4044
	C ₉ H ₈ (1H-Indene)	95-13-6	H	12.62±0.05	EI	4044
	C ₆ H ₈ (C ₆ H ₅) ₂ (Benzene, 1,1'-(2-cyclohexen-1-ylidene)bis-)	31158-25-5		13.6±0.4	EI	4018
	C ₆ H ₁₀ (C ₆ H ₅) ₂ (Benzene, 1,1'-cyclohexylidenebis-)	21113-55-3		13.3±0.4	EI	4018
	C ₆ H ₉ (CH ₃)(C ₆ H ₅) ₂ (Benzene, 1,1'-(4-methylcyclohexylidene)bis-)	32812-65-0		13.7±0.4	EI	4018
	C ₁₀ H ₁₃ (CH ₃)(C ₆ H ₅) ₂ (Naphthalene, 1,2,3,4,4a,5,6,7-octahydro-4a-methyl-2,2-diphenyl-)	50592-50-2		13.2±0.4	EI	4018
	C ₆ H ₅ C≡CCH=CHCH ₂ OH (2-Penten-4-yn-1-ol, 5-phenyl-, (<i>E</i>)-)	40317-08-6		11.43±0.05	EI	4044
	C ₆ H ₈ (=O)(C ₆ H ₅) ₂ (Cyclohexanone, 2,2-diphenyl-)	22612-62-0		14.1±0.4	EI	4018
	C ₆ H ₈ (=O)(C ₆ H ₅) ₂ (Cyclohexanone, 4,4-diphenyl-)	4528-68-1		13.5±0.4	EI	4018
	C ₆ H ₇ (=O)(CH ₃)(C ₆ H ₅) ₂ (Cyclohexanone, 2-methyl-5,5-diphenyl-)	50592-49-9		13.5±0.4	EI	4018
	C ₆ H ₇ (=O)(CH ₃)(C ₆ H ₅) ₂ (Cyclohexanone, 6-methyl-2,2-diphenyl-)	50592-52-4		13.7±0.4	EI	4018
	C ₆ H ₈ (OH)(CH ₃)(C ₆ H ₅) ₂ (Cyclohexanol, 1-methyl-4,4-diphenyl-)	50592-47-7		13.7±0.4	EI	4018
	C ₆ H ₆ (=O)(CH ₃) ₂ (C ₆ H ₅) ₂ (Cyclohexanone, 2,2-dimethyl-6,6-diphenyl-)	50592-53-5		13.8±0.4	EI	4018
	C ₁₀ H ₁₁ (=O)(CH ₃)(C ₆ H ₅) ₂ (2(3 <i>H</i>)-Naphthalenone,4,4a,5,6,7,8-hexahydro-4a-methyl-7,7-diphenyl-)	50786-03-3		13.0±0.4	EI	4018
	C ₆ H ₆ (=O)(CH ₃)(C ₆ H ₅) ₂ CH ₂ CH ₂ CHO (Cyclohexanepropanal, 1-methyl-2-oxo-3,3-diphenyl-)	XXXXX-XX-X		13.4±0.4	EI	4018
	C ₆ H ₆ (=O)(CH ₃)(C ₆ H ₅) ₂ CH ₂ CH ₂ COCH ₃ 4018 (Cyclohexanone, 2-methyl-2-(3-oxobutyl)-6,6-diphenyl-)	50592-55-7		14.2±0.4	EI	4018
	C ₆ H ₆ (=O)(C ₆ H ₅) ₂ CHS(CH ₂) ₃ CH ₃ (Cyclohexanone, 6-[butylthio)methylene]-2,2-diphenyl-)	50592-51-3		13.7±0.4	EI	4018
	C ₆ H ₆ (=O)CH ₃ (C ₆ H ₅) ₂ CH ₂ CH=C(CH ₃)Cl (Cyclohexanone, 2-(3-chloro-2-butenyl)-2-methyl-6,6-diphenyl-)	50592-54-6	Cl	13.7±0.4	EI	4018

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₉H₈⁺						
	C ₆ H ₅ (CH ₃)C≡CH (Benzene, 1-ethynyl-2-methyl-)	766-47-2	**	8.61±0.02 (V)	PE	5409
	C ₆ H ₅ (CH ₃)C≡CH (Benzene, 1-ethynyl-3-methyl-)	766-82-5	**	8.63±0.02 (V)	PE	5409
	C ₆ H ₅ (CH ₃)C≡CH (Benzene, 1-ethynyl-4-methyl-)	766-97-2	**	8.43 (V)	PE	4334
			**	8.48±0.02 (V)	PE	5409
	C ₆ H ₅ CH=C=CH ₂ (Benzene, 1,2-propadienyl-)	2327-99-3	**	8.29 (V)	PE	4493
	C ₆ H ₅ C≡CCH ₃ (Benzene, 1-propynyl-)	673-32-5	**	8.41±0.02 (V)	PE	5409
		.	**	8.49 (V)	PE	4334
	C ₆ H ₅ C ₃ H ₄ (1H-Indene)	95-13-6	**	8.15±0.015 (V)	PE	5522
			**	8.33±0.01	EI	3805
	C ₉ H ₈ (Spiro[4.4]nona-1,3,6,8-tetraene)	14867-83-5	**	7.99 (V)	PE	4049
			**	7.99 (V)	PE	4189
C₉H₉⁺						
	CH≡C(CH=CH) ₃ CH ₃	1743-34-6	H	10.7±0.1	EI	4336
	C ₆ H ₅ C ₃ H ₅ (Benzene, cyclopropyl-)	873-49-4	H	11.4±0.1	EI	4336
	C ₆ H ₅ (CH ₃)CH=CH ₂ (Benzene, 1-ethenyl-4-methyl-)	622-97-9	H	11.8±0.1	EI	4336
	C ₆ H ₅ C(CH ₃)=CH ₂ (Benzene, (1-methylethethyl)-)	98-83-9	H	11.8±0.1	EI	4336
	C ₆ H ₅ CH=CHCH ₃ (Benzene, 1-propenyl-)	637-50-3	H	11.8±0.1	EI	4336
	C ₆ H ₅ CH ₂ CH=CH ₂ (Benzene, 2-propenyl-)	300-57-2	H	11.6±0.1	EI	4336
	C ₉ H ₁₀ (1H-Indene, 2,3-dihydro-)	496-11-7	H	12.1±0.1	EI	4336
C₉H₁₀⁺						
	CH≡C(CH=CH) ₃ CH ₃	1743-34-6	**	7.2±0.1	EI	4336
	C ₆ H ₅ C ₃ H ₅ (Benzene, cyclopropyl-)	873-49-4	**	8.61 (V)	PE	4927
			**	8.66 (V)	PE	4815
			**	8.71 (V)	PE	4347
			**	8.3±0.1	EI	4336
	C ₆ H ₅ (CH ₃)CH=CH ₂ (Benzene, 1-ethenyl-2-methyl-)	611-15-4	**	8.20±0.02	PE	3854
			**	8.53 (V)	PE	3964
	C ₆ H ₄ (CH ₃)CH=CH ₂ (Benzene, 1-ethenyl-3-methyl-)	100-80-1	**	8.15±0.02	PE	3854
			**	8.37 (V)	PE	3964
	C ₆ H ₄ (CH ₃)CH=CH ₂ (Benzene, 1-ethenyl-4-methyl-)	622-97-9	**	8.20 (V)	PE	3964
			**	8.1±0.1	EI	4336
	C ₆ H ₅ C(CH ₃)=CH ₂ (Benzene, (1-methylethethyl)-)	98-83-9	**	8.52 (V)	PE	3964
			**	8.18±0.04	EI	4097
			**	8.3±0.1	EI	4336
	C ₆ H ₅ CH=CHCH ₃ (Benzene, 1-propenyl-, (E)-)	873-66-5	**	8.20±0.02	PE	3854
			**	8.32	PE	4289
			**	7.84±0.04	EI	4097
	C ₆ H ₅ CH=CHCH ₃ (Benzene, 1-propenyl-, (Z)-)	766-90-5	**	8.45	PE	4289
	C ₆ H ₅ CH=CHCH ₃ (Benzene, 1-propenyl-)	637-50-3	**	8.5±0.1	EI	4336

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₉H₁₀⁺						
	C ₆ H ₅ C(CH ₃)=CH ₂ (Benzene, 2-propenyl-)	300-57-2	**	8.20±0.02	PE	3854
			**	8.60	PE	3938
			**	9.16 (V)	PE	4211
			**	7.8±0.1	EI	4336
	C ₇ H ₆ (=CH ₂) ₂ (Bicyclo[2.2.1]hept-2-ene, 5,6-bis(methylene)-)	5628-77-3	**	8.48 (V)	PE	4249
	C ₉ H ₁₀ (Bicyclo[3.2.2]nona-2,6,8-triene)	16216-91-4	**	8.72 (V)	PE	3991
	C ₉ H ₁₀ (1H-Cyclobuta[cd]pentalene, 1a,3a,5a,5b-tetrahydro-)	58913-91-0	**	8.76	PE	4855
	C ₆ H ₈ =CH ₂ (Dicyclopenta[cd,gh]pentalene, octahydro-1-methylene-)	3721-64-0	**	8.43±0.02 (V)	PE	4338
	C ₉ H ₁₀ (1H-Indene, 2,3-dihydro-)	496-11-7	**	8.45±0.02 (V)	PE	3854
			**	8.46 (V)	PE	4063
			**	8.6±0.1	EI	4336
			**	8.60±0.01	EI	3805
			**	8.52	CTS	3546
			**	8.46±0.03 (V)	PE	4828
			**	8.50	PE	4952
	C ₉ H ₁₀ (1,2-Methanodicyclopenta[cd,gh]pentalene, octahydro-)	13084-56-5	**	9.06±0.02 (V)	PE	4338
			**	9.15±0.05 (V)	PE	5335
	C ₉ H ₁₀ (Pentacyclo[4.3.0.0 ^{2,5} .0 ^{3,8} .0 ^{4,7}]nonane)	452-61-9	**	8.47	PE	4955
	C ₉ H ₁₀ (Spiro[bicyclo[2.2.1]hepta-2,5-diene-7,1'-cyclopropane])	7092-57-1	**	8.73 (V)	PE	3780
	C ₉ H ₁₀ (Spiro[4.4]nona-1,3,6-triene)	766-30-3	**	8.27 (V)	PE	4189
	C ₉ H ₁₀ (Spiro[4.4]nona-1,3,7-triene)	24430-29-3	**	8.25 (V)	PE	4189
	C ₉ H ₁₀ (Tricyclo[3.2.2.0 ^{2,4}]nona-6,8-diene)	7092-05-9	**	8.65 (V)	PE	5605
	C ₉ H ₁₀ (Tricyclo[3.3.1.0 ^{2,8}]nona-3,6-diene)	14693-11-9	**	8.4 (V)	PE	4034
	C ₉ H ₁₀ (Tricyclo[4.2.1.0 ^{2,5}]nona-3,7-diene)	4932-71-2	**	8.7 (V)	PE	3853
	C ₉ H ₁₀ (Tricyclo[4.2.1.0 ^{2,5}]nona-3,7-diene, (1 α ,2 α ,5 α ,6 α -))	15564-45-1	**	9.03±0.03 (V)	PE	4281
	C ₉ H ₁₀ (Tricyclo[4.2.1.0 ^{2,5}]nona-3,7-diene, (1 α ,2 β ,5 β ,6 α -))	15564-44-0	**	8.65±0.05 (V)	PE	4040
	C ₉ H ₈ (=CH ₂) (Tricyclo[3.2.1.0 ^{2,4}]oct-6-ene,8-methylene-)	XXXXX-XX-X	**	8.85±0.05 (V)	PE	5335
	C ₆ H ₅ (CH ₂) ₃ NH ₂ (Benzenepropanamine)	2038-57-5	NH ₃	9.5±0.1	EI	5374
	C ₆ H ₅ (CH ₃) ₂ CH ₂ OCOCH ₃ (Phenethyl alcohol, <i>m</i> -methyl-, acetate)	33709-40-9		8.75	EI	3590
	C ₆ H ₅ (CH ₃) ₂ CH ₂ OCOCH ₃ (Phenethyl alcohol, <i>p</i> -methyl-, acetate)	22532-47-4		8.50	EI	3590
C₉H₁₂⁺						
	C ₆ H ₅ (iso-C ₃ H ₇) (Benzene, (1-methylethyl)-)	98-82-8	**	8.75 (V)	PE	4927
			**	8.72	EI	5293
			**	8.98 (V)	PE	4347
	(C ₂ H ₅) ₄ C	20685-34-1	**	9.52 (V)	PE	3994
	C ₆ H ₅ CH ₂ CH ₂ CH ₃ (Benzene, propyl-)	103-65-1	**	8.71	EI	5293
	C ₆ H ₃ (CH ₃) ₃ (Benzene, 1,2,3-trimethyl-)	526-73-8	**	8.6±0.03 (V)	PE	3713

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₉H₁₂⁺						
	C ₆ H ₅ (CH ₃) ₃ (Benzene, 1,2,4-trimethyl-)	95-63-6	**	8.5±0.03 (V)	PE	3713
	C ₆ H ₅ (CH ₃) ₃ (Benzene, 1,3,5-trimethyl-)	108-67-8	**	8.45±0.05 (V)	PE	4132
			**	8.45±0.05 (V)	PE	4724
			**	8.45 (V)	PE	5367
			**	8.65±0.03 (V)	PE	3713
			**	8.21±0.1	EI	3788
			**	8.46	CTS	4029
	C ₈ H ₈ (=CH ₂) ₂ (Bicyclo[2.2.1]heptane, 2,3-bis(methylene)-)	36439-78-8	**	8.41 (V)	PE	4249
	C ₆ H ₁₂ (Bicyclo[3.2.2]nona-2,6-diene)	14993-07-8	**	8.84 (V)	PE	3991
	C ₆ H ₁₂ (Bicyclo[3.2.2]nona-6,8-diene)	7164-08-1	**	9.00 (V)	PE	3991
	C ₆ H ₁₂ (Bicyclo[4.2.1]nona-2,4-diene)	6572-82-3	**	8.23 (V)	PE	4688
	C ₉ H ₁₀ =CH ₂ (Bicyclo[2.2.2]oct-2-ene, 5-methylene-)	19386-05-1	**	8.97 (V)	PE	4249
	CH≡CCH=C ₆ H ₁₀ (Cyclohexane,2-propynylidene-)	2806-45-3	**	8.49±0.01	PE	5407
	(C ₃ H ₅) ₂ C=C=CH ₂ (Cyclopropane, 1,1'-(1,2-propadienylidene)bis-)	60166-70-3	**	8.62	PE	4608
	C ₆ H ₁₂ (Spiro[4.4]nona-1,3-diene)	766-29-0	**	8.10 (V)	PE	4189
			**	8.14	PE	4268
	C ₆ H ₁₂ (Tetracyclo[3.3.1.0 ^{2,8} .0 ^{4,6}]nonane)	3105-29-1	**	8.67 (V)	PE	3741
	C ₆ H ₁₂ (Tetracyclo[6.1.0.0 ^{2,4} .0 ^{5,7}]nonane(1 α ,2 α ,4 α ,5 β ,7 β ,8 α -))	37831-90-6	**	9.0 (V)	PE	5192
	C ₆ H ₆ (CH ₃) ₂ (Tricyclo[4.1.0.0 ^{2,7}]hept-3-ene,1,3-dimethyl-)	66036-92-8	**	8.26 (V)	PE	5441
	C ₆ H ₆ (CH ₃) ₂ (Tricyclo[4.1.0.0 ^{2,7}]hept-3-ene,1,6-dimethyl-)	61772-32-5	**	8.30 (V)	PE	5441
	C ₆ H ₁₂ (Tricyclo[3.2.2.0 ^{2,4}]non-6-ene)	7092-58-2	**	8.8 (V)	PE	5605
	C ₆ H ₁₂ (Tricyclo[4.2.1.0 ^{2,5}]non-3-ene)	7078-40-2	**	9 (V)	PE	3853
	C ₆ H ₁₂ (Tricyclo[4.2.1.0 ^{2,5}]non-3-ene, (1 α ,2 β ,5 β ,6 α -))	16529-76-3	**	9.00±0.05 (V)	PE	4040
	C ₆ H ₁₂ (Tricyclo[4.2.1.0 ^{2,5}]non-7-ene)	6827-30-1	**	8.7 (V)	PE	3853
	C ₆ H ₁₂ (Tricyclo[4.2.1.0 ^{2,5}]non-7-ene, (1 α ,2 α ,5 α ,6 α -))	16529-83-2	**	8.92±0.03 (V)	PE	4281
	C ₆ H ₁₂ (Tricyclo[4.2.1.0 ^{2,5}]non-7-ene, exo-)	16529-82-1	**	8.70±0.05 (V)	PE	4040
	C ₆ H ₁₂ (Tricyclo[6.1.0.0 ^{2,4}]non-5-ene(1 α ,2 α ,4 α ,8 α -))	62211-27-2	**	8.90 (V)	PE	4964
	C ₆ H ₁₂ (Tricyclo[6.1.0.0 ^{2,4}]non-5-ene(1 α ,2 β ,4 β ,8 α -))	62279-39-4	**	8.96 (V)	PE	4964
	C ₆ H ₁₂ (Tricyclo[6.1.0.0 ^{2,4}]non-6-ene(1 α ,3 α ,5 α ,8 α -))	XXXXXX-XX-X	**	9.0 (V)	PE	4964
	C ₆ H ₁₂ (Tricyclo[6.1.0.0 ^{2,4}]non-6-ene(1 α ,3 β ,5 β ,8 α -))	62163-62-6	**	8.5 (V)	PE	4964
	C ₆ H ₁₀ (=CH ₂) (Tricyclo[3.2.1.0 ^{2,1}]octane,8-methylene-)	38310-48-4	**	9.10±0.05 (V)	PE	5335
	C ₆ H ₁₂ (Trispiro[2.0.2.0.2.0]nonane)	31561-59-8	**	9.12 (V)	PE	4963
	(C ₆ H ₅ (CH ₃) ₃) ₂ Cr (Chromium, tricarbonyl[(1,2,3,4,5,6- η)-1,3,5-trimethylbenzene]-)	12129-67-8		8.61±0.1	EI	3788

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₉H₁₃⁺						
	C(CH ₃)(CH ₂)C ₆ H ₈ CH ₃ (Cyclohexene, 1-methyl-4-(1-methylethylene)-)	138-86-3	CH ₃	8.9	EI	5200
	C ₁₀ H ₁₆ (4,7-Methano-1H-indene, octahydro-, (3α,4β,7β,7α)-)	2825-82-3	CH ₃	9.8±0.1	PI	3918
	C ₁₀ H ₁₅ CH ₃ (4,7-Methano-1H-indene, octahydro-2-methyl-, (2α,3αβ,4α,7α,7αβ)-)	XXXXX-XX-X		<10.2±0.1	PI	3918
	C ₁₀ H ₁₅ CH ₃ (4,7-Methano-1H-indene, octahydro-8-methyl-, stereoisomer)	50745-90-9		10.1±0.1	PI	3918
	C ₁₀ H ₁₅ CH ₃ (4,7-Methano-1H-indene, octahydro-8-methyl-, stereoisomer)	50745-92-1		9.5±0.1	PI	3918
	C ₁₀ H ₁₅ C ₂ H ₅ (4,7-Methano-1H-indene, 5-ethyloctahydro-, (3α,4β,5α,7β,7α)-)	32787-97-6		9.9±0.1	PI	3918
C₉H₁₄⁺						
	C ₄ H ₆ C≡CC(CH ₃)=CH ₂	17603-76-8	**	8.57±0.01	PE	5407
	CH ₃ C≡CCH=C(C ₂ H ₅) ₂	70058-01-4	**	8.12±0.01	PE	5407
	CH≡CC(iso-C ₃ H ₇)=C(CH ₃) ₂	61786-07-0	**	8.26±0.01	PE	5407
	cis-CH ₃ C≡CCH=CHC ₄ H ₉	53497-78-2	**	8.46±0.01	PE	5407
	trans-CH ₃ C≡CCH=CHC ₄ H ₉	53497-79-3	**	8.46±0.01	PE	5407
	C ₉ H ₁₄ (Bicyclo[3.2.2]non-2-ene)	40319-81-1	**	8.84 (V)	PE	3991
	C ₉ H ₁₄ (Bicyclo[3.2.2]non-6-ene)	7124-86-9	**	8.95 (V)	PE	3991
	C ₉ H ₁₄ (Bicyclo[3.3.1]non-1-ene)	17530-61-9	**	8.35 (V)	PE	4569
	C ₈ H ₁₂ =CH ₂ (Bicyclo[2.2.2]octane, 2-methylene-)	2972-20-5	**	8.87 (V)	PE	4249
	CH ₂ =CHCH ₂ C ₆ H ₉ (Cyclohexene, 1-(2-propenyl)-)	13511-13-2	**	8.49±0.01	PI	5556
	CH ₂ =CHCH ₂ C ₆ H ₉ (Cyclohexene, 3-(2-propenyl)-)	15232-95-8	**	8.83±0.02	PI	5556
	C ₉ H ₁₄ (1,2-Cyclononadiene)	1123-11-1	**	8.87 (V)	PE	4019
	C ₃ H ₂ (CH ₃) ₂ =C=C(CH ₃) ₂ (Cyclopropane, 1,1-dimethyl-2-(2-methyl-1-propenylidene)-)	28438-32-6	**	7.65	PE	5625
	C ₃ H ₂ (CH ₃) ₂ =C=C(CH ₃) ₂ (Cyclopropane, 1,2-dimethyl-3-(2-methyl-1-propenylidene)-cis-)	37817-36-0	**	7.76	PE	5625
	C ₃ H ₂ (CH ₃) ₂ =C=C(CH ₃) ₂ (Cyclopropane, 1,2-dimethyl-3-(2-methyl-1-propenylidene)-trans-)	37817-46-2	**	7.70	PE	5625
	C ₃ H ₅ C(C ₂ H ₅)=C=CHCH ₃ (Cyclopropane, (1-ethyl-1,2-butadienyl)-)	60042-77-5	**	8.60	PE	4608
	C ₉ H ₁₄ (Spiro[bicyclo[2.2.1]heptane-2,1'-cyclopropane])	173-89-7	**	9.45 (V)	PE	4433
	C ₉ H ₁₄ (Spiro[4.4]non-1-ene)	873-12-1	**	8.73	PE	4268
	C ₉ H ₁₄ (Tricyclo[3.2.2.0 ^{2,4}]nonane)	278-80-8	**	8.96 (V)	PE	4347
	C ₉ H ₁₄ (Tricyclo[4.2.1.0 ^{2,5}]nonane, (1α,2α,5α,6α)-)	16526-28-6	**	9.50 (V)	PE	3849
	C ₉ H ₁₄ (Tricyclo[4.2.1.0 ^{2,5}]nonane, exo-)	16526-27-5	**	9.5±0.05 (V)	PE	4281
	C ₉ H ₁₀ =C=C=CHCH ₃ (Cyclohexane, 1-propenylidene-)	20023-43-2	**	8.41	PE	4040
						5625
C₉H₁₆⁺						
	C ₆ H ₇ C(CH ₃) ₂ C≡CCH ₃ (CH ₃) ₂ CHC(CH ₃) ₂ C≡CCH ₃	XXXXX-XX-X	**	9.183±0.010	PE	4575
	CH ₂ =CH(CH ₂) ₅ CH=CH ₂	994-21-8	**	9.154±0.010	PE	4575
	C ₆ H ₁₃ C≡CCH ₃	4900-30-5	**	9.51±0.02 (V)	PE	4010
		19447-29-1	**	9.289±0.005	PE	4575
			**	9.32±0.02	PI	5583

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₉H₁₆⁺						
	C ₅ H ₁₁ C≡CC ₂ H ₅	20184-89-8	**	9.202±0.005	PE	4575
			**	9.20±0.02	PI	5583
	1-C ₉ H ₁₆	3452-09-3	**	9.93±0.02	PI	5583
	4-C ₉ H ₁₆	20184-91-2	**	9.17±0.03	PI	5583
	C ₇ H ₁₀ (CH ₃) ₂ (Bicyclo[2.2.1]heptane, 7,7-dimethyl-)	2034-53-9	**	8.30	PE	3687
	C ₉ H ₁₆ (Bicyclo[3.2.2]nonane)	283-19-2	**	9.6 (V)	PE	3991
	C ₉ H ₁₆ (Bicyclo[3.3.1]nonane)	280-65-9	**	9.35	PE	4735
	C ₉ H ₁₆ (Bicyclo[6.1.0]nonane)	286-60-2	**	9.4 (V)	PE	3509
	C ₉ H ₁₆ (Bicyclo[6.1.0]nonane, <i>trans</i> -)	39124-79-3	**	9.36 (V)	PE	3849
	<i>n</i> -C ₅ H ₇ C ₆ H ₉ (Cyclohexene,1-propyl-)	2539-75-5	**	8.43±0.01	PI	5556
	<i>n</i> -C ₅ H ₇ C ₆ H ₉ (Cyclohexene,3-propyl-)	3983-06-0	**	8.80±0.01	PI	5556
	C ₉ H ₁₆ (Cyclononene(Z))	933-21-1	**	8.81±0.15	EI	5532
	<i>n</i> -C ₄ H ₉ C ₅ H ₇ (Cyclopentene,1-butyl-)	2423-01-0	**	8.45±0.01	PI	5556
	<i>n</i> -C ₄ H ₉ C ₅ H ₇ (Cyclopentene,3-butyl-)	22531-00-6	**	8.83±0.02	PI	5556
	<i>is</i> -C ₄ H ₉ C ₅ H ₇ (Cyclopentene,1-(2-methylpropyl)-)	53098-47-8	**	8.44±0.01	PI	5556
C₉H₁₈⁺						
	CH ₃ (CH ₂) ₃ C(CH ₃) ₂	3074-64-4	**	8.145±0.005	PE	3957
	C ₂ H ₅ CH ₂ C(CH ₃) ₂ =C(CH ₃)C ₂ H ₅	3074-67-7	**	8.077±0.005	PE	3957
	(C ₂ H ₅) ₂ C=C(CH ₃)C ₂ H ₅	50787-13-8	**	8.128±0.005	PE	3957
	1-C ₉ H ₁₈	124-11-8	**	9.42±0.01	PI	5584
	cis-2-C ₉ H ₁₈	6434-77-1	**	8.90±0.01	PI	5584
	cis-3-C ₉ H ₁₈	20237-46-1	**	8.84±0.01	PI	5584
			**	9.01±0.01 (V)	PE	4939
	cis-4-C ₉ H ₁₈	10405-84-2	**	8.801±0.01	PI	5584
	trans-2-C ₉ H ₁₈	6434-78-2	**	8.90±0.01	PI	5584
	trans-3-C ₉ H ₁₈	20063-92-7	**	8.84±0.01	PI	5584
			**	9.01±0.01 (V)	PE	4939
	trans-4-C ₉ H ₁₈	10405-85-3	**	8.809±0.01	PI	5584
	(CH ₃) ₂ CHC ₆ H ₁₁ (Cyclohexane,(1-methylethyl)-)	696-29-7	**	9.55±0.03	PI	5556
	<i>n</i> -C ₄ H ₉ C ₅ H ₇ (Cyclopentane,butyl-)	2040-95-1	**	9.95±0.03	PI	5556
C₁₀H₆⁺						
	C ₆ H ₄ (C≡CH) ₂ (Benzene, 1,2 diethynyl-)	21792-52-9	**	8.69±0.02	PE	4374
	C ₆ H ₄ (C≡CH) ₂ (Benzene, 1,3 diethynyl-)	1785-61-1	**	8.82±0.02	PE	4374
	C ₆ H ₄ (C≡CH) ₂ (Benzene, 1,4 diethynyl-)	935-14-8	**	8.58±0.02	PE	4374
C₁₀H₈⁺						
	C ₁₀ H ₈ (Azulene)	275-51-4	**	7.42 (V)	PE	5397
			**	7.43±0.04	PE	4196
			**	7.44±0.03 (V)	PE	4828
	C ₁₀ H ₈ (Naphthalene)	91-20-3	**	8.1	PI	3586
			**	8.13	PE	3637

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₀H₈⁺						
	C ₁₀ H ₈	91-20-3	**	8.15±0.02 (V)	PE	4913
			**	8.15	PE	3668
			**	8.15	PE	3638
			**	8.15	PE	4066
			**	8.15	PE	4515
			**	8.15 (V)	PE	3781
			**	8.15 (V)	PE	4701
			**	8.15 (V)	PE	5632
			**	8.18±0.03 (V)	PE	4828
			**	8.31±0.03 (V)	PE	4341
			**	8.25±0.01	EI	3588
			**	8.12	CTS	3922
C₁₀H₈⁺²						
	C ₁₀ H ₈ (Naphthalene)	91-20-3	**	22.8	OTH	5141
C₁₀H₈⁺³						
	C ₁₀ H ₈ (Naphthalene)	91-20-3	**	41.2±1.0	OTH	5141
C₁₀H₁₀⁺						
	C ₆ H ₅ CH=CHCH=CH ₂ (Benzene, 1,3-butadienyl-, (E)-)	16939-57-4	**	7.95	PE	3892
	C ₆ H ₅ CH=C=CHCH ₃ (Benzene, 1,2-butadienyl-)	2327-98-2	**	8.15 (V)	PE	4493
	cis-(C ₆ H ₅)CH=CHCH=CH ₂ (Benzene, 1,3-butadienyl-)	1515-78-2	**	8.39	PE	5202
	C ₆ H ₅ C≡CC ₂ H ₅ (Benzene, 1-butynyl-)	622-76-4	**	8.33±0.02 (V)	PE	5409
	C ₆ H ₅ C ₄ H ₅ (Benzene, 1-cyclobuten-1-yl-)	3365-26-2	**	8.22	PE	4347
	C ₆ H ₅ (C ₂ H ₃) ₂ (Benzene, 1,4-diethenyl-)	105-06-6	**	8.11 (V)	PE	5537
	C ₆ H ₅ (CH ₃) ₂ C≡CH (Benzene, 1-ethynyl-2,4-dimethyl-)	16017-30-4	**	8.31±0.02 (V)	PE	5409
	CH ₂ =C(C ₆ H ₅)CH=CH ₂ (Benzene, (1-methylene-2-propenyl)-)	2288-18-8	**	8.57	PE	3892
			**	8.60 (V)	PE	5537
	C ₆ H ₅ C(CH ₃)=C=CH ₂ (Benzene, 1-methyl-1,2-propadienyl-)	22433-39-2	**	8.07 (V)	PE	4493
	C ₆ H ₅ (CH ₃)C≡CCH ₃ (Benzene, 1-methyl-2-(1-propynyl)-)	57497-13-9	**	8.23±0.02 (V)	PE	5409
	C ₆ H ₅ (CH ₃)C≡CCH ₃ (Benzene, 1-methyl-3-(1-propynyl)-)	XXXXX-XX-X	**	8.26±0.02 (V)	PE	5409
	C ₆ H ₅ (CH ₃)C≡CCH ₃ (Benzene, 1-methyl-4-(1-propynyl)-)	2749-93-1	**	8.13±0.02 (V)	PE	5409
	C ₉ H ₈ =CH ₂ (Bicyclo[4.2.1]nona-2,4,7-triene, 9-methylene-)	38898-39-4	**	8.25 (V)	PE	4094
	C ₉ H ₈ (=CH ₂) ₂ (Bicyclo[2.2.2]octa-2,5-diene, 7,8-bis(methylene)-)	51698-73-8	**	8.33±0.03 (V)	PE	4665
	C ₉ H ₈ =CH ₂ (1H-Cyclobuta[cd]pentale, 1a,3a,5a,5b-tetrahydro-1-methylene-)	64096-73-7	**	8.80	PE	4855
	C ₁₀ H ₁₀ (Cyclopenta[cd]pentale, 2a,4a,6a,6b-tetrahydro-)	6053-74-3	**	9.0 (V)	PE	4004
	C ₁₀ H ₁₀ (Hexacyclo[4.4.0 ^{2,4} .0 ^{3,9} .0 ^{5,7} .0 ^{8,10}]decane)	XXXXX-XX-X	**	8.5 (V)	PE	5192
	C ₉ H ₈ (=CH ₂) (1H-Indene, 2,3-dihydro-1-methylene-)	1194-56-5	**	8.00±0.02	PE	3854
	C ₉ H ₈ (=CH ₂) (1,2-Methanodicyclopropa[cd,gh]pentale, octahydro-3-methylene-)	64630-96-2	**	9.00±0.05 (V)	PE	5335

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{10}\text{H}_{14}^+$	$\text{C}_{10}\text{H}_{14}$ (Tetracyclo[7.1.0 ^{2,4} .0 ^{5,7}]decane(1 α ,2 α ,4 β ,5 α ,7 α ,9 α -))	62279-35-0	**	8.88 (V)	PE	4964
	$\text{C}_{10}\text{H}_{14}$ (Tricyclo[4.2.2.0 ^{2,5}]dec-7-ene)	37706-26-6	**	9.0 (V)	PE	5605
	$\text{C}_9\text{H}_{12}(=\text{CH}_2)$ (Tricyclo[4.2.1.0 ^{2,5}]nonane,9-methylene-)	XXXXXX-XX-X	**	9.20±0.05 (V)	PE	5335
	$\text{C}_6\text{H}_4(\text{CH}_3)_4$ (Benzene, 1,2,4,5-tetramethyl-)	95-93-2	**	8.07	PE	4952
$\text{C}_{10}\text{H}_{15}^+$	$\text{C}_{10}\text{H}_{16}$ (Tricyclo[3.3.1.1 ^{3,7}]decane)	281-23-2	H	10.6	PI	4173
	$\text{C}_{10}\text{H}_{15}\text{CH}_3$	XXXXXX-XX-X	CH_3	9.5±0.1	PI	3918
	$\text{C}_{10}\text{H}_{15}\text{CH}_3$ (4,7-Methano-1H-indene, octahydro-2-methyl-, (2 α ,3 α β ,4 α ,7 α ,7 α β -))	50745-90-9	CH_3	10.1±0.1	PI	3918
	$\text{C}_{10}\text{H}_{15}\text{CH}_3$ (4,7-Methano-1H-indene, octahydro-8-methyl-, stereoisomer)	50745-92-1	CH_3	9.6±0.1	PI	3918
	$\text{C}_{10}\text{H}_{15}\text{C}_2\text{H}_5$ (4,7-Methano-1H-indene, 5-ethyloctahydro-, (3 α ,4 β ,5 α ,7 β ,7 α α -))	32787-97-6		9.9±0.1	PI	3918
	$\text{C}_{12}\text{H}_{20}$ (Tricyclo[3.3.1.1 ^{3,7}]decane, 2-ethyl-)	14451-87-7		10.5	PI	4173
$\text{C}_{10}\text{H}_{16}^+$	$\text{C}_5\text{H}_{11}\text{C}\equiv\text{CC}(\text{CH}_3)=\text{CH}_2$	70058-00-3	**	8.57±0.01	PE	5407
	$\text{CH}_3\text{C}\equiv\text{CC}(iso-\text{C}_3\text{H}_7)=\text{C}(\text{CH}_3)_2$	70058-04-7	**	7.89±0.01	PE	5407
	$\text{C}_6\text{H}_{14}=\text{CH}_2$ (Bicyclo[4.2.1]nonane, 9-methylene-)	40916-48-1	**	9.0 (V)	PE	4094
	$\text{C}_4\text{H}_3(\text{CH}_3)_4\text{C}\equiv\text{CH}$ (Cyclobutane,2-ethynyl-1,1,3,3-tetramethyl)	66438-89-9	**	9.33 (V)	PE	5607
	$\text{C}_6\text{H}_4(\text{CH}_3)_4$ (1,4-Cyclohexadiene, 3,3,6,6-tetramethyl-)	2223-54-3	**	8.81 (V)	PE	4385
	$\text{C}(\text{CH}_3)(\text{CH}_2)\text{C}_6\text{H}_8\text{CH}_3$ (Cyclohexene, 1-methyl-4-(1-methylethylene)-)	138-86-3	**	8.3	EI	5200
	$(\text{C}_5\text{H}_5)_2\text{C}=\text{C}(\text{CH}_3)_2$ (Cyclopropane, 1,1'-(2-methyl-1-propenylidene)bis-)	27720-84-9	**	7.82	PI	3759
	$\text{C}_3\text{H}(\text{CH}_3)_3=\text{C}=\text{C}(\text{CH}_3)_2$ (Cyclopropane,trimethyl(2-methyl-1-propenylidene)-)	14803-30-6	**	7.57	PE	5625
	$\text{C}_{10}\text{H}_{16}$ (Dispiro[2.0.2.4]decane)	24029-74-1	**	9.22 (V)	PE	5359
	$\text{C}_{10}\text{H}_{16}$ (Dispiro[2.2.2.2] decane)	24518-94-3	**	9.17 (V)	PE	4385
	$\text{C}_{10}\text{H}_{16}$ (4,7-Methano-1H-indene, octahydro-)	6004-38-2	**	9.3	PI	4173
	$\text{C}_{10}\text{H}_{16}$ (4,7-Methano-1H-indene, octahydro-, (3 α ,4 β ,7 β ,7 α -))	2825-82-3	**	9.35±0.05	PI	3918
	$\text{C}_{10}\text{H}_{16}$ (Spiro[bicyclo[2.2.2]octane-2,1'-cyclopropane])	53764-10-6	**	9.32 (V)	PE	4433
	$\text{C}_{10}\text{H}_{16}$ (Tricyclo[3.3.1.1 ^{3,7}]decane)	281-23-2	**	9.30±0.01	S	3757
			**	9.25	PI	4173
			**	9.1±0.05	PE	3855
			**	9.20	PE	4735
			**	9.22	PE	3907
			**	9.23	PE	3886
			**	9.28±0.1	PE	3851
			**	9.28 (V)	PE	5043
			**	9.31±0.01	PE	3757
			**	9.55 (V)	PE	3990
			**	9.75±0.02 (V)	PE	4217
			**	9.75 (V)	PE	4000
			**	9.75 (V)	PE	5395

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₀H₁₆⁺	C ₁₀ H ₁₆ (Tricyclo[4.2.2.0 ^{2.5}]decane)	249-87-6	**	9.45 (V)	PE	5605
C₁₀H₁₇⁺	C ₆ H ₉ (CH ₃)C ₄ H ₈ (Naphthalene, decahydro-1-methyl-(1α,4aα,8aβ))	4683-95-8	CH ₃	10.13±0.007	EI	5451
	C ₆ H ₉ (CH ₃)C ₄ H ₈ (Naphthalene, decahydro-1-methyl-(1α,4aβ,8aα))	XXXXX-XX-X	CH ₃	10.14±0.010	EI	5451
	C ₆ H ₉ (CH ₃)C ₄ H ₈ (Naphthalene, decahydro-2-methyl-(2α,4aα,8aβ))	14398-71-1	CH ₃	10.34±0.006	EI	5451
	C ₆ H ₉ (CH ₃)C ₄ H ₈ (Naphthalene, decahydro-2-methyl-(2α,4aβ,8aα))	4683-94-7	CH ₃	10.33±0.009	EI	5451
C₁₀H₁₈⁺	(tert-C ₄ H ₉)C≡C(tert-C ₄ H ₉)	17530-24-4	**	9.054±0.010	PE	4575
	C ₆ H ₉ (tert-C ₄ H ₉) (Cyclohexene,3-(1,1-dimethylethyl)-)	14072-87-8	**	8.94±0.02 (V)	PE	5420
	C ₄ H ₉ C≡CC ₄ H ₉	1942-46-7	**	9.125±0.005	PE	4575
			**	9.14±0.02	PI	5583
	C ₆ H ₁₃ C≡CC ₂ H ₅	2384-85-2	**	9.190±0.005	PE	4575
			**	9.19±0.02	PI	5583
	1-C ₁₀ H ₁₈	764-93-2	**	9.91±0.02	PI	5583
	2-C ₁₀ H ₁₈	2384-70-5	**	9.30±0.02	PI	5583
	4-C ₁₀ H ₁₈	2384-86-3	**	9.17±0.02	PI	5583
	C ₄ H ₉ (CH ₃) ₃ CH=CH ₂ (Cyclobutane,2-ethenyl-1,1,3,3-tetramethyl)	66438-87-7	**	9.10 (V)	PE	5607
	C ₁₀ H ₁₈ (Cyclodecene(E))	2198-20-1	**	8.91±0.15	EI	5532
	C ₁₀ H ₁₈ (Cyclodecene(Z))	935-31-9	**	8.97±0.15	EI	5532
	C ₁₀ H ₁₈ (Cyclodecene)	3618-12-0	**	8.98 (V)	PE	4267
	n-C ₆ H ₉ C ₆ H ₉ (Cyclohexene,1-butyl-)	3282-53-9	**	8.41±0.01	PI	5556
	n-C ₆ H ₉ C ₆ H ₉ (Cyclohexene,3-butyl-)	3983-07-1	**	8.80±0.02	PI	5556
	n-C ₆ H ₉ C ₆ H ₉ (Cyclohexene,4-butyl-)	21524-26-5	**	8.85±0.02	PI	5556
	(CH ₃) ₂ CHCH ₂ C ₆ H ₉ (Cyclohexene,1-(2-methylpropyl)-)	3983-03-7	**	8.40±0.01	PI	5556
	C ₂ H ₅ CH(CH ₃)C ₆ H ₉ (Cyclohexene,3-(1-methylpropyl)-)	15232-91-4	**	8.74±0.02	PI	5556
	(CH ₃) ₂ CHCH ₂ C ₆ H ₉ (Cyclohexene,3-(2-methylpropyl)-)	4104-56-7	**	8.77±0.02	PI	5556
	n-C ₅ H ₁₁ C ₅ H ₇ (Cyclopentene,1-pentyl-)	4291-98-9	**	8.45±0.02	PI	5556
	n-C ₅ H ₁₁ C ₅ H ₇ (Cyclopentene,3-pentyl-)	37689-14-8	**	8.84±0.02	PI	5556
	C ₁₀ H ₁₈ (Naphthalene, decahydro-)	91-17-8	**	~9.35	PE	4735
	(CH ₃) ₂ CHC ₂ H ₅ C ₅ H ₇ (Cyclopentene,1-(3-methylbutyl)-)	37689-15-9	**	8.44±0.02	PI	5556
	(CH ₃) ₂ CHC ₂ H ₅ C ₅ H ₇ (Cyclopentene,3-(3-methylbutyl)-)	37689-16-0	**	8.83±0.02	PI	5556
C₁₀H₂₀⁺	CH ₃ (CH ₂) ₃ C(C ₂ H ₅)=C(CH ₃) ₂	19780-61-1	**	8.101±0.005	PE	3957
	CH ₃ (CH ₂) ₄ C(CH ₃)=C(CH ₃) ₂	19781-18-1	**	8.132±0.005	PE	3957
	(CH ₃) ₃ CCH ₂ C(CH ₃)=C(CH ₃) ₂	33175-59-6	**	8.097±0.005	PE	3957
	1-C ₁₀ H ₂₀	872-05-9	**	9.417±0.006	PI	5584
			**	9.59±0.01 (V)	PE	4939

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₀H₂₀⁺						
	(<i>tert</i> -C ₄ H ₉) ₂ C=CH ₂	5857-68-1	**	8.795±0.008	PE	3957
	<i>cis</i> -(CH ₃) ₃ CCH=CHC(CH ₃) ₃	692-47-7	**	8.695±0.010	PE	3957
			**	8.95 (V)	PE	4084
	<i>cis</i> -2-C ₁₀ H ₂₀	20348-51-0	**	8.899±0.010	PI	5584
			**	9.08±0.01 (V)	PE	4939
	<i>cis</i> -3-C ₁₀ H ₂₀	19398-86-8	**	8.832±0.009	PI	5584
			**	9.01±0.01 (V)	PE	4939
	<i>cis</i> -4-C ₁₀ H ₂₀	19398-88-0	**	8.784±0.004	PI	5584
			**	8.97±0.01 (V)	PE	4939
	<i>cis</i> -5-C ₁₀ H ₂₀	7433-78-5	**	8.773±0.006	PI	5584
			**	8.766±0.005	PE	3957
			**	8.94±0.01 (V)	PE	4939
	<i>cis</i> ((<i>iso</i> -C ₃ H ₇)(CH ₃)C) ₂	60643-93-8	**	8.27 (V)	PE	4459
	<i>trans</i> -(CH ₃) ₃ CCH=CHC(CH ₃) ₃	692-48-8	**	8.741±0.008	PE	3957
			**	8.89 (V)	PE	4084
	<i>trans</i> -2-C ₁₀ H ₂₀	20063-97-2	**	8.903±0.005	PI	5584
			**	9.06±0.01 (V)	PE	4939
	<i>trans</i> -3-C ₁₀ H ₂₀	19150-21-1	**	8.830±0.006	PI	5584
			**	9.00±0.01	PE	4939
	<i>trans</i> -4-C ₁₀ H ₂₀	19398-89-1	**	8.782±0.004	PI	5584
			**	8.97±0.01 (V)	PE	4939
	<i>trans</i> -5-C ₁₀ H ₂₀	7433-56-9	**	8.762±0.012	PI	5584
			**	8.760±0.005	PE	3957
			**	8.95±0.01 (V)	PE	4939
	<i>trans</i> ((<i>iso</i> -C ₃ H ₇)(CH ₃)C) ₂	60643-94-9	**	8.24 (V)	PE	4459
	C ₁₀ H ₂₀ (Cyclodecane)	293-96-9	**	10.00±0.05	EI	4319
	n-C ₄ H ₉ C ₆ H ₁₁ (Cyclohexane, butyl-)	1678-93-9	**	9.57±0.03	PI	5556
	C ₂ H ₅ CH(CH ₃)C ₆ H ₁₁ (Cyclohexane,(1-methylpropyl)-)	7058-01-7	**	9.51±0.03	PI	5556
	(CH ₃) ₂ CHCH ₂ C ₆ H ₁₁ (Cyclohexane,(2-methylpropyl)-)	1678-98-4	**	9.54±0.03	PI	5556
	n-C ₅ H ₁₁ C ₆ H ₉ (Cyclopentane,pentyl-)	3741-00-2	**	9.91±0.05	PI	5556
C₁₁H₇⁺						
	(C ₆ H ₅) ₂ (1,1'-Biphenyl)	92-52-4	CH ₃	14.80±0.2	EI	4199
C₁₁H₉⁺						
	C ₁₀ H ₇ CH ₃ (Naphthalene, 1-methyl-)	90-12-0	H	13.15±0.2	EI	4199
	C ₁₀ H ₇ CH ₃ (Naphthalene, 2-methyl-)	91-57-6	H	13.15±0.2	EI	4199
	C ₁₀ H ₆ (CH ₃) ₂ (Naphthalene, 1,5-dimethyl-)	571-61-9	CH ₃	12.85±0.05	EI	4199
	C ₁₀ H ₆ (CH ₃) ₂ (Naphthalene, 1,8-dimethyl-)	569-41-5	CH ₃	12.70±0.2	EI	4199
	C ₆ H ₅ C≡CCH=CHCH ₂ Cl (Benzene, (5-chloro-3-penten-1-ynyl)-, (<i>E</i>)-)	40316-56-1		8.95±0.05	EI	4044
	C ₁₀ H ₇ CH ₂ Cl (Naphthalene, 1-(chloromethyl)-)	86-52-2		11.21±0.05	EI	4044
	C ₁₀ H ₇ CH ₂ Cl (Naphthalene, 2-(chloromethyl)-)	2506-41-4		11.15±0.05	EI	4044
C₁₁H₁₀⁺						
	C ₁₀ H ₇ CH ₃ (Azulene, 1-methyl-)	769-31-3	**	7.26±0.03 (V)	PE	4828
	C ₁₀ H ₇ CH ₃ (Azulene, 4-methyl-)	17647-77-7	**	7.33±0.03 (V)	PE	4828

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₁H₁₀⁺						
	C ₁₀ H ₇ CH ₃ (Azulene, 5-methyl-)	1654-55-3	**	7.30±0.03 (V)	PE	4828
	C ₁₀ H ₇ CH ₃ (Azulene, 6-methyl-)	1654-52-0	**	7.34±0.03 (V)	PE	4828
	C ₁₁ H ₁₀ (Bicyclo[4.4.1]undeca-1,3,5,7,9-pentaene)	2443-46-1	**	7.90 (V)	PE	3953
	C ₉ H ₈ (=C=CH ₂) (1,2-Methanodicyclopropa[cd,gh]pentalene,3-ethenylideneoctahydro-)	65915-89-1	**	8.75 (V)	PE	5447
	C ₁₁ H ₁₀ (1,4-Methanonaphthalene, 1,4-dihydro-)	4453-90-1	**	8.30±0.05 (V)	PE	4830
			**	8.32±0.05 (V)	PE	4866
			**	8.34±0.05 (V)	PE	5019
			**	8.34 (V)	PE	4541
			**	8.34 (V)	PE	4835
	C ₁₀ H ₇ CH ₃ (Naphthalene, 1-methyl-)	90-12-0	**	7.95 (V)	PE	3685
			**	8.01±0.03 (V)	PE	4828
			**	7.80±0.03	EI	3588
			**	8.50±0.05	EI	4199
			**	7.98	CTS	3758
	C ₁₀ H ₇ CH ₃ (Naphthalene, 2-methyl-)	91-57-6	**	7.83	PE	4515
			**	7.93 (V)	PE	3685
			**	8.01±0.03 (V)	PE	4828
			**	8.10±0.03	EI	3588
			**	8.45±0.05	EI	4199
	(C ₆ H ₅) ₂ S (Benzene, 1,1'-thiobis-)	139-66-2	CS	12.57±0.1	EI	3817
C₁₁H₁₂⁺						
	C ₆ H ₅ (C ₂ H ₅) (Spiro[cyclopropane-1,3'-{1,2}methanodicyclopropa[cd,gh]pentalene] octahydro-)	65915-88-0	**	9.05 (V)	PE	5447
	C ₆ H ₅ C ₅ H ₇ (Benzene, 1-cyclopenten-1-yl-)	825-54-7	**	8.15 (V)	PE	4347
	C ₆ H ₅ C ₅ H ₇ (Benzene, 2-cyclopenten-1-yl-)	37689-22-8	**	~9.2±0.05 (V)	PE	4954
	C ₆ H ₅ C ₅ H ₇ (Benzene, 3-cyclopenten-1-yl-)	39599-89-8	**	8.62±0.01	PI	5556
	C ₆ H ₅ C≡CC ₃ H ₇ (Benzene, 1-pentynyl-)	4250-81-1	**	8.29±0.02 (V)	PE	5409
	C ₁₁ H ₁₂ (1H-Cyclobut[f]indene,2,4,5,6-tetrahydro-)	60582-10-7	**	8.05	PE	4952
	C ₁₁ H ₁₂ (1H-Cyclobut[e]indene, 2,5,6,7-tetrahydro-)	60582-11-8	**	8.19	PE	4952
	C ₁₁ H ₁₂ (1,4-Methanonaphthalene, 1,2,3,4-tetrahydro-)	4486-29-7	**	8.42±0.05 (V)	PE	4830
			**	8.45±0.05 (V)	PE	4866
	C ₁₀ H ₁₀ (=CH ₂) (Naphthalene, 1,2,3,4-tetrahydro-1-methylene-)	25108-63-8	**	7.90±0.02 (V)	PE	3854
	C ₁₁ H ₁₂ (Pentacycloundecene)	XXXXX-XX-X	**	8.7 (V)	PE	5578
	C ₉ H ₈ (=CH ₂) ₂ (Tricyclo[3.2.2.0 ^{2,4}]non-6-ene,8,9-bis(methylene)-(1 α ,2 α ,4 α ,5 α -))	36439-89-1	**	8.37±0.03 (V)	PE	4665
C₁₁H₁₄⁺						
	C ₆ H ₅ C ₅ H ₄ (C ₂ H ₅) (Benzene, (1-ethylcyclopropyl)-)	50462-84-5	**	8.70 (V)	PE	4815
	C ₆ H ₅ C ₅ H ₉ (Benzene, cyclopentyl-)	700-88-9	**	8.81 (V)	PE	4347
	C ₆ H ₂ (CH ₃) ₃ CH=CH ₂ (Benzene, 2-ethenyl-1,3,5-trimethyl-)	769-25-5	**	8.33 (V)	PE	3964

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₁H₁₁⁺						
	C ₆ H ₅ CH=CH(CH ₂) ₂ CH ₃ (Benzene, 1-pentenyl-)	826-18-6	**	8.4±0.07	EI	5374
	C ₁₁ H ₁₁ (5H-Benzocycloheptene, 6,7,8,9-tetrahydro-)	1075-16-7	**	8.40±0.02 (V)	PE	3854
			**	8.44 (V) 8.90 (V)	PE	4063
	C ₁₁ H ₁₁ (Bicyclo[4.2.1]non-7-ene, 2,5-bis(methylene)-)	72569-85-8			PE	5325
	C ₉ H ₈ (CH ₃) ₂ (Indan, 1,1-dimethyl)	4912-92-9	**	8.47	CTS	3546
	C ₉ H ₈ (CH ₃) ₂ (1H-Indene, 2,3-dihydro-2,2-dimethyl-)	20836-11-7	**	8.47	CTS	3546
	C ₁₁ H ₁₁ (Spiro[2,4]hepta-1,4,6-triene, 1,2-diethyl)	49542-94-1	**	7.87 (V)	PE	5480
	C ₉ H ₁₀ (=CH ₂) ₂ (Tricyclo[3.2.2.0 ^{2,1}]nonane, 6,7-bis(methylene)-(1 α ,2 β ,4 β ,5 α -))	36439-90-4	**	8.38±0.03 (V)	PE	4665
	C ₉ H ₈ =C(CH ₃) ₂ (Tricyclo[3.2.1.0 ^{2,1}]oct-6-ene, 8-(1-methylethylidene)-, endo-)	XXXXX-XX-X	**	7.9	PE	3687
	C ₆ H ₅ (CH ₂) ₅ NH ₂ (Benzepentanamine)	17734-21-3	NH ₃	9.4±0.1	EI	5374
C₁₁H₁₆⁺						
	C ₆ H ₅ CH ₂ (tert-C ₄ H ₉) (Benzene, (2,2-dimethylpropyl)-)	1007-26-7	**	8.7 (V)	PE	4172
			**	8.77 (V)	PE	4280
			**	~8.8	PE	4589
	C ₆ H ₅ (CH ₃)C ₄ H ₉ (Benzene, 1-butyl-3-methyl-)	1595-04-6	**	8.42±0.1	EI	3629
	C ₆ H ₅ (CH ₃)C ₄ H ₉ (Benzene, 1-butyl-4-methyl-)	1595-05-7	**	8.35±0.1	EI	3629
	C ₆ H(CH ₃) ₅ (Benzene, pentamethyl-)	700-12-9	**	7.9	CTS	3543
	C ₆ H ₁₀ C=C=C(CH ₃) ₂ (Bicyclo[4.1.0]heptane, 7-(2-methyl-1-propenylidene)-)	4544-26-7	**	7.60	PE	5625
	C ₆ H ₁₂ (=CH ₂) ₂ (Bicyclo[4.2.1]nonane, 2,5-bis(methylene)-)	72569-86-9	**	8.90 (V)	PE	5314
	(C ₃ H ₅) ₂ C=CHC ₃ H ₅ (Cyclopropane, 1,1',1''-(1-ethenyl-2-ylidene)tris-)	23603-63-6	**	7.48	PI	3759
	C ₁₁ H ₁₆ (Dispiro[cyclopropane-1,2'-bicyclo[2.2.1]heptane-3',1''-cyclopropane])	40827-29-0	**	8.76 (V)	PE	4433
	C ₁₁ H ₁₆ (Spiro[2,4]hepta-4,6-diene, 1,2-diethyl)	59313-59-6	**	8.20 (V)	PE	5480
	C ₁₀ H ₁₄ (=CH ₂) (Tricyclo[3.3.1.1 ^{3,7}]decane, 2-methylene-)	875-72-9		8.82	PE	3886
			**	8.86±0.02 (V)	PE	4217
	C ₈ H ₈ (=CH ₂)(CH ₃) ₂ (Tricyclo[3.2.1.0 ^{2,4}]octane, 3,3-dimethyl-)	XXXXX-XX-X	**	8.80±0.05 (V)	PE	5335
	C ₉ H ₁₀ =C(CH ₃) ₂ (Tricyclo[3.2.1.0 ^{2,4}]octane, 8-(1-methylethylidene)-, endo-)	XXXXX-XX-X	**	8.18	PE	3687
C₁₁H₁₇⁺						
	C ₁₀ H ₁₅ C ₂ H ₅ (4,7-Methano-1H-indene, 5-ethyloctahydro-, (3 α ,4 β ,5 α ,7 β ,7a α -))	32787-97-6	CH ₃	10.0±0.1	PI	3918
C₁₁H₁₈⁺						
	C ₁₁ H ₁₅ CH ₃	XXXXX-XX-X	**	9.35±0.05	PI	3918
	C ₆ H ₆ (CH ₃) ₄ (Cycloheptyne, 3,3,7,7-tetramethyl-)	33470-40-5	**	8.80 (V)	PE	4362
	C ₆ H ₁₁ C ₅ H ₇ (Cyclohexane, 2-cyclopenten-1-yl-)	2690-15-5	**	8.95±0.05 (V)	PE	4954
	C ₃ (CH ₃) ₄ =C=C(CH ₃) ₂ (Cyclopropane, tetramethyl(2-methyl-1-propenylidene)-)	13303-30-5	**	7.46	PE	5625
	C ₁₁ H ₁₈ (4,7-Ethano-1H-indene, octahydro-)	38255-97-9	**	9.15	PI	4173

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₁H₁₈⁺						
	C ₁₁ H ₁₈ (4,7-Methanoazulene, decahydro-)	51027-86-2	**	9.25	PI	4173
	C ₁₀ H ₁₅ CH ₃ (4,7-Methano-1H-indene, octahydro-2-methyl-, (2α,3αβ,4α,7α,7aβ)-)	50745-90-9	**	9.35±0.05	PI	3918
	C ₁₀ H ₁₅ CH ₃ (4,7-Methano-1H-indene, octahydro-8-methyl-, stereoisomer)	50745-92-1	**	9.35±0.05	PI	3918
	C ₆ H ₇ (CH ₃)C ₄ H ₈ (Naphthalene,1,2,3,4,4a,5,6,8a-octahydro-4a-methyl-cis-)	65698-42-2	**	8.92±0.02 (V)	PE	5420
	C ₆ H ₇ (CH ₃)C ₄ H ₈ (Naphthalene,1,2,3,4,4a,5,6,8a-octahydro-4a-methyl-trans-)	XXXXXX-XX-X	**	8.92±0.02 (V)	PE	5420
	C ₁₁ H ₁₈ (Naphthalene, 1,2,3,4,4a,5,6,8a-octahydro-4a-methyl-trans-)	68211-37-0	**	8.92±0.05 (V)	PE	4842
	C ₁₀ H ₁₅ CH ₃ (Tricyclo[3.3.1.1 ^{3,7}]decane, 1-methyl-)	768-91-2	**	9.17±0.02	PE	3886
C₁₁H₂₀⁺						
	1-C ₁₁ H ₂₀	2243-98-3	**	9.90±0.02	PI	5583
	2-C ₁₁ H ₂₀	60212-29-5	**	9.28±0.02	PI	5583
	3-C ₁₁ H ₂₀	60212-30-8	**	9.17±0.02	PI	5583
	4-C ₁₁ H ₂₀	60212-31-9	**	9.13±0.02	PI	5583
	5-C ₁₁ H ₂₀	2294-72-6	**	9.11±0.02	PI	5583
	(tert-C ₄ H ₉) ₂ C=C=CH ₂	22585-31-5	**	8.55 (V)	PE	4019
	C ₁₁ H ₂₀ (Cyclohexane, 1-(1,1-dimethylethyl)-4-methylene)	13294-73-0	**	9.09±0.05 (V)	PE	4842
			**	9.09±0.02 (V)	PE	5420
	n-C ₅ H ₁₁ C ₆ H ₉ (Cyclohexene,1-pentyl-)	15232-85-6	**	8.37±0.02	PI	5556
	n-C ₆ H ₁₃ C ₅ H ₇ (Cyclopentene,1-hexyl-)	4291-99-0	**	8.43±0.01	PI	5556
	n-C ₆ H ₁₃ C ₅ H ₇ (Cyclopentene,3-hexyl-)	37689-18-2	**	8.84±0.02	PI	5556
	C ₁₁ H ₂₀ (Cycloundecene(E))	13151-60-5	**	8.73±0.15	EI	5532
	C ₁₁ H ₂₀ (Cycloundecene(Z))	13151-61-6	**	8.65±0.15	EI	5532
	C ₆ H ₉ (CH ₃)C ₄ H ₈ (Naphthalene,decahydro-2-methyl-(2α,4aβ,8aα)	4683-94-7	**	9.31±0.006	EI	5451
	C ₆ H ₉ (CH ₃)C ₄ H ₈ (Naphthalene,decahydro-1-methyl-(1α,4aα,8aβ))	4683-95-8	**	9.27±0.009	EI	5451
	C ₆ H ₉ (CH ₃)C ₄ H ₈ (Naphthalene,decahydro-1-methyl-(1α,4aβ,8aα))	XXXXXX-XX-X	**	9.26±0.008	EI	5451
	C ₆ H ₉ (CH ₃)C ₄ H ₈ (Naphthalene,decahydro-2-methyl-(2α,4aα,8aβ))	14398-71-1	**	9.32±0.006	EI	5451
C₁₁H₂₂⁺						
	C ₂ H ₅ CH ₂ C(C ₂ H ₅)=C(C ₂ H ₅) ₂	50787-14-9	**	8.041±0.020	PE	3957
	n-C ₆ H ₁₃ C ₅ H ₉ (Cyclopentane,hexyl-)	4457-00-5	**	9.90±0.03	PI	5556
C₁₂H₆⁺						
	C ₆ H ₅ (C≡CH) ₃ (Benzene, 1,3,5-triethynyl-)	7567-63-7	**	8.86±0.02	PE	4374
	C ₁₂ H ₆ (1,5,9-Cyclododecatriene-3,7,11-triene)	6555-54-0	**	7.69 (V)	PE	4652
C₁₂H₈⁺						
	C ₁₂ H ₈ (Acenaphthylene)	208-96-8	**	8.22±0.04	PE	4196
	C ₁₂ H ₈ (Biphenylene)	259-79-0	**	7.53±0.05	PE	3684
			**	7.60±0.02 (V)	PE	3702

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₂H₈⁺						
	C ₁₂ H ₈	259-79-0	**	7.61±0.04	PE	4196
	C ₁₂ H ₈ (1,3,5,9-Cyclododecatetraene-7,11-diyne)	7003-42-1	**	7.54 (V)	PE	4652
C₁₂H₉⁺						
	(C ₆ H ₅) ₂ (1,1'-Biphenyl)	92-52-4	H	13.60±0.2	EI	4199
C₁₂H₁₀⁺						
	C ₁₂ H ₁₀ (Acenaphthylene, 1,2-dihydro-)	83-32-9	**	7.76±0.03 (V)	PE	4828
			**	7.82±0.04	PE	4196
	(C ₆ H ₅) ₂ (1,1'-Biphenyl)	92-52-4	**	7.95±0.02	PE	3702
			**	8.34 (V)	PE	5619
			**	8.39 (V)	PE	5364
			**	8.80±0.05	EI	4199
			**	8.35	CTS	3577
	C ₁₂ H ₁₀ (Cyclobuta[<i>a</i>]naphthalene, 1,2-dihydro-)	32277-35-3	**	7.84±0.03 (V)	PE	4952
	C ₁₂ H ₁₀ (Cyclobuta[<i>b</i>]naphthalene, 1,2-dihydro-)	6827-31-2	**	7.92±0.03 (V)	PE	4952
			**	7.96±0.03 (V)	PE	4828
	C ₁₂ H ₁₀ (Cyclopent[cd]azulene, 2a, 8b-dihydro-)	38310-40-6	**	7.46 (V)	PE	4008
	C ₁₂ H ₁₀ (4a, 8a-Ethenonaphthalene)	19539-78-7	**	8.1 (V)	PE	4006
C₁₂H₁₀⁺²						
	(C ₆ H ₅) ₂ (1,1'-Biphenyl)	92-52-4	**	22.1	OTH	5141
C₁₂H₁₁⁺						
	C ₁₀ H ₆ (CH ₃) ₂ (Naphthalene, 1,5-dimethyl-)	571-61-9	H	12.85±0.05	EI	4199
	C ₁₀ H ₆ (CH ₃) ₂ (Naphthalene, 1,8-dimethyl-)	569-41-5	H	13.00±0.2	EI	4199
C₁₂H₁₂⁺						
	C ₁₁ H ₆ (CH ₃) (1,4-Methanonaphthalene, 1,4-dihydro-6-methyl-)	4897-73-8	**	8.12±0.05 (V)	PE	5019
	C ₁₀ H ₆ (CH ₃) ₂ (Azulene, 4,5-dimethyl-)	56594-77-5	**	7.18±0.03 (V)	PE	4828
	C ₁₀ H ₆ (CH ₃) ₂ (Azulene, 4,6-dimethyl-)	56594-78-6	**	7.29±0.03 (V)	PE	4828
	C ₁₀ H ₆ (CH ₃) ₂ (Azulene, 4,7-dimethyl-)	46030-99-3	**	7.20±0.03 (V)	PE	4828
	C ₁₀ H ₆ (CH ₃) ₂ (Azulene, 4,8-dimethyl-)	7206-52-2	**	7.27±0.03 (V)	PE	4828
	C ₁₀ H ₆ (CH ₃) ₂ (Azulene, 5,6-dimethyl-)	10556-12-4	**	7.17±0.03 (V)	PE	4828
	C ₁₀ H ₆ (CH ₃) ₂ (Azulene, 5,7-dimethyl-)	56594-76-4	**	7.08±0.03 (V)	PE	4828
	C ₁₂ H ₁₂ (1,5,9-Cyclododecatriyne)	60323-50-4	**	9.24 (V)	PE	4781
	C ₁₂ H ₁₂ (Cyclopent[cd]azulene, 2a, 4a, 8a, 8b-tetrahydro-)	56004-38-7	**	8.50 (V)	PE	5606
	C ₁₂ H ₁₂ (4a, 8a-Ethenonaphthalene, 1,4-dihydro-)	38310-32-6	**	8.0 (V)	PE	4006
	C ₁₂ H ₁₂ (5,9-Methano-5H-benzocycloheptene, 6,9-dihydro-)	24309-43-1	**	8.42±0.05	PE	4866

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₂H₁₂⁺						
	C ₁₂ H ₁₂ (2,7-Methano-1H-cyclopropa[<i>b</i>]naphthalene, 1a,2,7,7a-tetrahydro- (1α,2β,7β,7α)-)	15577-76-1	**	8.40±0.05 (V)	PE	4866
	C ₁₂ H ₁₂ (1,2,5-Metheno-1H-cyclobuta[<i>de</i>]naphthalene, 1a,2,4a,5,7a,7b-hexahydro-)	68109-02-4	**	8.30 (V)	PE	5119
	C ₁₀ H ₆ (CH ₃) ₂ (Naphthalene, 1,3-dimethyl-)	575-41-7	**	7.86±0.03 (V)	PE	4828
	C ₁₀ H ₆ (CH ₃) ₂ (Naphthalene, 1,4-dimethyl-)	571-58-4	**	7.82±0.03 (V)	PE	4828
	C ₁₀ H ₆ (CH ₃) ₂ (Naphthalene, 1,5-dimethyl-)	571-61-9	**	7.85±0.03 (V)	PE	4828
			**	8.30±0.05	EI	4199
	C ₁₀ H ₆ (CH ₃) ₂ (Naphthalene, 1,8-dimethyl-)	569-41-5	**	7.64±0.03 (V)	PE	4828
			**	8.30±0.05	EI	4199
	C ₁₀ H ₆ (CH ₃) ₂ (Naphthalene, 2,3-dimethyl-)	581-40-8	**	7.89±0.03 (V)	PE	4828
	C ₁₀ H ₆ (CH ₃) ₂ (Naphthalene, 2,7-dimethyl-)	582-16-1	**	7.89±0.03 (V)	PE	4828
C₁₂H₁₄⁺						
	C ₆ H ₅ C ₆ H ₉ (Benzene,2-cyclohexen-1-yl-)	15232-96-9	**	7.96±0.02	PI	5556
	C ₆ H ₅ C ₆ H ₉ (Benzene,3-cyclohexen-1-yl-)	4994-16-5	**	8.57±0.01	PI	5556
	C ₆ H ₅ CH ₂ C ₅ H ₇ (Benzene,(1-methyl-2-cyclopenten-1-yl-)	XXXXXX-XX-X	**	8.47±0.02	PI	5556
	C ₁₁ H ₁₂ (=CH ₂) (5H-Benzocycloheptene, 6,7,8,9-tetrahydro-5-methylene-)	40562-09-2	**	8.45±0.02 (V)	PE	3854
	C ₇ H ₈ (=CH ₂) ₂ (=C(CH ₃) ₂) (Bicyclo[2.2.1]hept-2-ene,5,6-bis(methylene)-7-(1-methylethylidene)-)	36439-83-5	**	8.40±0.03 (V)	PE	4665
	C ₉ H ₈ =C(CH ₃) ₂ (1H-Cyclobuta[cd]pentalene,1a,3a,5a,5b-tetrahydro-1- (1-methylethylidene-))	64096-77-1	**	8.15	PE	4855
	C ₁₂ H ₁₄ (1,4:5,8-Dimethanonaphthalene, 1,4,4a,5,8a-hexahydro-,(1α,4α,4aα,5α,8α,8aα)-)	1076-13-7	**	8.08±0.03 (V)	PE	4301
	C ₁₂ H ₁₄ (1,4:5,8-Dimethanonaphthalene, 1,4,4a,5,8,8a-hexahydro-, (1α,4α,4aα,5β,8β,8aα)-)	15914-94-0	**	8.46±0.03 (V)	PE	4301
	C ₁₂ H ₁₄ (4a, 8a-Ethenonaphthalene, 1,2,3,4-tetrahydro-)	24139-33-1	**	8.0 (V)	PE	4006
	C ₁₂ H ₁₄ (4a, 8a-Ethenonaphthalene, 1,4,5,8-tetrahydro-)	20295-17-4	**	8.7 (V)	PE	4006
	C ₁₂ H ₁₄ (Hexacyclododecane)	XXXXXX-XX-X	**	9.0 (V)	PE	5578
	C ₁₂ H ₁₄ (5-Indacene, 1,2,3,5,6,7-hexahydro-)	495-52-3	**	7.94	PE	4952
	C ₁₂ H ₁₄ (as-Indacene, 1,2,3,6,7,8-hexahydro-)	1076-17-1	**	8.09	PE	4952
	C ₁₂ H ₁₄ (1,2-Methanodicyclopenta[cd,g,h]pentalene,octahydro-3-(1-methylethylidene)-)	15391-62-5	**	8.52±0.05 (V)	PE	4866
	C ₁₀ H ₁₀ (=CH ₂) ₂ (Tricyclo[4.2.2.0 ^{2,5}]dec-7-ene,9,10-bis(methylene)-(1a,2a,5a,6a)-)	57297-56-0	**	8.40±0.03 (V)	PE	4665
	C ₉ H ₈ =C(CH ₃) ₂ (Tricyclo[4.2.1.0 ^{2,5}]nona-3,7-diene, 9-(1-methylethylidene)-, (1α,2a,5a,6a)-)	27237-73-6	**	8.33±0.03 (V)	PE	4281
C₁₂H₁₆⁺						
	C ₆ H ₅ C ₃ H ₄ (iso-C ₃ H ₇) (Benzene, [1-(1-methylethyl)cyclopropyl]-)	63339-99-1	**	8.63 (V)	PE	4815
	C ₆ H ₅ CH=CHC(CH ₃) ₃ (Benzene, (3,3-dimethyl-1-butenyl)-, (E)-)	3846-66-0	**	7.80±0.04	EI	4097

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₂H₁₆⁺						
	C ₆ H ₅ CH=CHC(CH ₃) ₃ (Benzene, (3,3-dimethyl-1-butenyl)-, (Z)-)	3740-05-4	**	8.29±0.04	EI	4097
	C ₆ H ₅ C(C(CH ₃) ₃)=CH ₂ (Benzene, (2,2-dimethyl-1-methylenepropyl)-)	5676-29-9	**	8.25±0.04	EI	4097
	C ₁₂ H ₁₆ (Benzocyclooctene, 5,6,7,8,9,10-hexahydro-)	1076-69-3	**	8.42 (V)	PE	4063
	C ₆ H ₄ (CH ₂) ₆ (Bicyclo[6.2.2]dodeca-8,10,11-triene)	53011-74-8	**	8.00 (V)	PE	5339
	C ₁₂ H ₁₆ (4a,8a-Ethanonaphthalene, 1,4,5,8-tetrahydro-)	5103-78-6	**	9.00±0.05 (V)	PE	4593
	C ₁₂ H ₁₆ (4a, 8a-Ethenonaphthalene, 1,2,3,4,5,8-hexahydro-)	24139-32-0	**	8.9 (V)	PE	4006
	C ₁₂ H ₁₆ (Tetraspiro[2.0.2.0.2.0.2.0]dodecane)	24375-17-5	**	8.22 (V)	PE	4963
	C ₁₀ H ₁₂ (=CH ₂) ₂ (Tricyclo[4.2.2.0 ^{2,5}]decane, 7,8-bis(methylene)-(1α,2β,5β,6α)-)	36439-92-6	**	8.27±0.03 (V)	PE	4665
	C ₆ H ₁₀ =C(CH ₃) ₂ (Tricyclo[4.2.1.0 ^{2,5}]non-7-ene, 9-(1-methylethylidene)-, (1α,2α,5α,6α)-)	53848-19-4	**	8.39±0.03 (V)	PE	4281
C₁₂H₁₈⁺						
	(n-C ₄ H ₉ C≡C) ₂	1120-29-2	**	8.67	PE	4731
	(tert-C ₄ H ₉ C≡C) ₂	6130-98-9	**	8.61±0.02 (V)	PE	4816
	C ₆ H ₄ (CH(CH ₃) ₂) ₂ (Benzene, 1,4-bis(1-methylethyl))	100-18-5	**	8.35	PE	5574
	C ₆ (CH ₃) ₆ (Benzene, hexamethyl-)	87-85-4	**	7.9 (V)	PE	5600
			**	7.8	CTS	3543
	C ₆ (CH ₃) ₆ (Bicyclo[2.2.0]hexa-2,5-diene, 1,2,3,4,5,6-hexamethyl-)	7641-77-2		7.83 (V)	PE	4296
			**	7.92 (V)	PE	4297
	C ₆ H ₁₂ =C(CH ₃) ₂ (1H-Cyclobuta[cd]pentalene, octahydro-1-(1-methylethylidene)-)	66149-44-8	**	8.19	PE	4855
	C ₃ =(C(CH ₃) ₂) ₃ (Cyclopropane, tris (1-methylethylidene)-) (JC-Mean value of Jahn-Teller components)	2799-44-2	**	7.49	PE	4390
	C ₁₂ H ₁₈ (1,4:5,8-Dimethanonaphthalene, decahydro-, (1α,4α,4α,5α,8α,8α)-)	53862-33-2	**	9.50±0.03 (V)	PE	4301
	C ₁₂ H ₁₈ (1,4:5,8-Dimethanonaphthalene, decahydro-, (1α,4α,4α,5β,8β,8α)-)	15914-95-1	**	9.57±0.03 (V)	PE	4301
	C ₁₂ H ₁₈ (Disipiro[cyclopropane-1,2'-bicyclo[2.2.2]octane-3',1"-cyclopropane])	40827-30-3	**	8.67 (V)	PE	4433
	C ₁₂ H ₁₈ (4a, 8a-Ethenonaphthalene, 1,2,3,4,5,6,7,8-octahydro-)	38992-78-8	**	9.05 (V)	PE	4006
	C ₆ H ₁₂ =C(CH ₃) ₂ (Tricyclo[4.2.1.0 ^{2,5}]nonane, 9-(1-methylethylidene)-, (1α,2α,5α,6α)-)	53848-20-7	**	8.30±0.03 (V)	PE	4281
	(C ₆ (CH ₃) ₆ (CO) ₃ Cr (Chromium, tricarbonyl[(1,2,3,4,5,6-η)-hexamethylbenzene]-)	12088-11-8		8.55±0.1	EI	3788
C₁₂H₂₀⁺						
	C ₁₂ H ₂₀ (Acenaphthylene, dodecahydro-)	2146-36-3	**	9.05	PI	4173
	C ₆ H ₁₁ C ₆ H ₉ (Cyclohexene, 1-cyclohexyl-)	3282-54-0	**	8.30±0.01	PI	5556
	C ₆ H ₁₁ C ₆ H ₉ (Cyclohexene, 3-cyclohexyl-)	1808-09-9	**	8.68±0.01	PI	5556
	C ₆ H ₈ (CH ₃) ₄ (Cyclooctyne, 3,3,8,8-tetramethyl-)	XXXXX-XX-X	**	8.90 (V)	PE	4362
	C ₁₀ H ₁₅ C ₂ H ₅ (4,7-Methano-1H-indene, 5-ethyloctahydro-, (3α,4β,5α,7β,7aa)-)	32787-97-6	**	9.35±0.05	PI	3918
	C ₁₀ H ₁₄ (CH ₃) ₂ (Tricyclo[3.3.1.1 ^{3,7}]decane)	702-79-4	**	9.15	PE	4735

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₂H₂₀⁺	C ₁₂ H ₂₀ (Tricyclo[3.3.1.1 ^{3,7}]decane, 2-ethyl-)	14451-87-7	**	9.2	PI	4173
C₁₂H₂₂⁺	<i>trans</i> -C ₃ H ₂ (<i>tert</i> -C ₄ H ₉) ₂ =CH ₂ C ₅ H ₁₁ C≡CC ₅ H ₁₁	XXXXX-XX-X 6975-99-1	** **	8.22±0.04 9.090±0.005 9.06±0.03	EI PE PI	4575 4575 5583
	<i>trans,trans</i> -((<i>tert</i> -C ₄ H ₉)CH=CH) ₂	22430-49-5	**	8.23±0.04	EI	4274
	1-C ₁₂ H ₂₂	765-03-7	**	9.90±0.02	PI	5583
	2-C ₁₂ H ₂₂	629-49-2	**	9.29±0.02	PI	5583
	3-C ₁₂ H ₂₂	6790-27-8	**	9.17±0.02	PI	5583
	4-C ₁₂ H ₂₂	22058-01-1	**	9.14±0.03	PI	5583
	5-C ₁₂ H ₂₂	19780-12-2	**	9.09±0.03	PI	5583
	C ₁₂ H ₂₂ (Cyclododecene(E))	1486-75-5	**	8.74±0.15	EI	5532
	C ₁₂ H ₂₂ (Cyclododecene(Z))	1129-89-1	**	8.78±0.15	EI	5532
	<i>n</i> -C ₆ H ₁₃ C ₆ H ₉ (Cyclohexene, 1-hexyl-)	3964-66-7	**	8.37±0.03	PI	5556
	<i>n</i> -C ₆ H ₁₃ C ₆ H ₉ (Cyclohexene, 3-hexyl-)	15232-78-7	**	8.78±0.01	PI	5556
	<i>n</i> -C ₇ H ₁₅ C ₅ H ₇ (Cyclopentene, 1-heptyl-)	4292-00-6	**	8.41±0.03	PI	5556
C₁₂H₂₁⁺	<i>cis</i> -(CH ₃) ₃ CCH ₂ C(CH ₃)=CHC(CH ₃) ₃ C ₁₂ H ₂₄ (Cyclododecane)	27656-50-4 294-62-2	** **	8.346±0.005 10.04±0.05	PE EI	3957 4319
C₁₃H₉⁺	C ₁₄ H ₉ CH ₃ (Phenanthrene, 4-methyl-)	832-64-4	C ₂ H ₃	12.7±0.1	EI	3454
	C ₁₄ H ₈ (CH ₃) ₂ (Phenanthrene, 4,5-dimethyl-)	3674-69-9		12.4±0.1	EI	3454
	C ₆ H ₅ (C ₆ H ₅) ₂ (Benzene, 1,1'-(2-cyclohexen-1-ylidene)bis-)	31158-25-5		13.0±0.4	EI	4018
	C ₆ H ₁₀ (C ₆ H ₅) ₂ (Benzene, 1,1'-cyclohexylidenebis-)	21113-55-3		13.3±0.4	EI	4018
	C ₆ H ₇ (CH ₃)(C ₆ H ₅) ₂ (Cyclohexene, 1-methyl-4,4-diphenyl-)	50592-48-8		13.4±0.4	EI	4018
	C ₆ H ₆ (CH ₃)(C ₆ H ₅) ₂ (Benzene, 1,1'-(4-methylcyclohexylidene)bis-)	32812-65-0		13.2±0.4	EI	4018
	C ₁₀ H ₁₃ (CH ₃)(C ₆ H ₅) ₂ (Naphthalene, 1,2,3,4,4a,5,6,7-octahydro-4a-methyl-2,2-diphenyl-)	50592-50-2		13.4±0.4	EI	4018
	C ₆ H ₆ (=O)(C ₆ H ₅) ₂ (2-Cyclohexen-1-one, 4,4-diphenyl-)	4528-64-7		14.4±0.4	EI	4018
	C ₆ H ₆ (=O)(C ₆ H ₅) ₂ (Cyclohexanone, 2,2-diphenyl-)	22612-62-0		13.8±0.4	EI	4018
	C ₆ H ₆ (=O)(C ₆ H ₅) ₂ (Cyclohexanone, 4,4-diphenyl-)	4528-68-1		14.4±0.4	EI	4018
	C ₆ H ₇ (=O)(CH ₃)(C ₆ H ₅) ₂ (Cyclohexanone, 2-methyl-5,5-diphenyl-)	50592-49-9		14.0±0.4	EI	4018
	C ₆ H ₇ (=O)(CH ₃)(C ₆ H ₅) ₂ (Cyclohexanone, 6-methyl-2,2-diphenyl-)	50592-52-4		14.1±0.4	EI	4018
	C ₆ H ₈ (OH)(C ₆ H ₅) ₂ (Cyclohexanol, 1-methyl-4,4-diphenyl-)	50592-47-7		13.9±0.4	EI	4018
	C ₆ H ₆ (=O)(CH ₃) ₂ (C ₆ H ₅) ₂ (Cyclohexanone, 2,2-dimethyl-6,6-diphenyl-)	50592-53-5		13.4±0.4	EI	4018
	C ₆ H ₆ (=O)(CH ₃)(C ₆ H ₅) ₂ CH ₂ CH ₂ CHO (Cyclohexanopropanal, 1-methyl-2-oxo-3,3-diphenyl-)	XXXXX-XX-X		13.6±0.4	EI	4018
	C ₆ H ₆ (=O)(CH ₃)(C ₆ H ₅) ₂ CH ₂ CH ₂ COCH ₃ (Cyclohexanone, 2-methyl-2-(3-oxobutyl)-6,6-diphenyl-)	50592-55-7		13.6±0.4	EI	4018

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{13}\text{H}_9^+$	$\text{C}_6\text{H}_5(\text{O})(\text{C}_6\text{H}_5)=\text{CHS}(\text{CH}_2)_3\text{CH}_3$ (Cyclohexanone, 6-[(butylthio)methylene]-2,2-diphenyl-) $\text{C}_6\text{H}_5(\text{O})\text{CH}_3(\text{C}_6\text{H}_5)_2\text{CH}_2\text{CH}=\text{C}(\text{CH}_3)\text{Cl}$ (Cyclohexanone, 2-(3-chloro-2-butene)-2-methyl-6,6-diphenyl-)	50592-51-3 50592-54-6	Cl	13.7 ± 0.4 13.3 ± 0.4	EI EI	4018 4018
$\text{C}_{13}\text{H}_{10}^+$	$\text{C}_{13}\text{H}_{10}$ (9H-Fluorene)	86-73-7	** ** ** **	7.93 ± 0.02 (V) 7.89 ± 0.03 7.91 (V) 8.52	PE PI PE EI	3702 5552 5619 4228
$\text{C}_{13}\text{H}_{11}^+$	$(\text{C}_6\text{H}_5)_2\text{CH}_2$ (Benzene, 1,1'-methylenebis-) $\text{C}_6\text{H}_5\text{C}_6\text{H}_4\text{CH}_3$ (1,1'-Biphenyl, 4-methyl-) $\text{C}_{10}\text{H}_7\text{CH}=\text{CHCH}_3$ (Naphthalene, 1-(1-propenyl)-) $\text{C}_{11}\text{H}_9\text{C}_6\text{H}_4$ (Spiro[7H-benzocycloheptene-7,1'-cyclopropane]) $\text{C}_6\text{H}_5(\text{CH}_3)\text{C}_6\text{H}_4\text{CH}_3$ (1,1'-Biphenyl, 2,2'-dimethyl-) $\text{C}_6\text{H}_5(\text{CH}_3)\text{C}_6\text{H}_4\text{CH}_3$ (1,1'-Biphenyl, 3,3'-dimethyl-) $\text{C}_6\text{H}_5(\text{CH}_3)\text{C}_6\text{H}_4\text{CH}_3$ (1,1'-Biphenyl, 4,4'-dimethyl-) $(\text{C}_6\text{H}_5)_3\text{CH}$ (Benzene, 1,1',1''-methylidynetris-) $\text{C}_6\text{H}_5\text{CH}_2\text{C}_6\text{H}_4\text{OH}$ (Phenol, 4-(phenylmethyl)-) $\text{C}_6\text{H}_5\text{CH}_2\text{C}_6\text{H}_4\text{OCH}_3$ (Benzene, 1-methoxy-4-(phenylmethyl)-) $\text{C}_6\text{H}_5\text{CH}_2\text{C}_6\text{H}_4\text{NO}_2$ (Benzene, 1-nitro-4-(phenylmethyl)-)	101-81-5 644-08-6 22767-77-7 29150-13-8 605-39-0 612-75-9 613-33-2 519-73-3 101-53-1 834-14-0 1817-77-2	H H H H CH_3 CH_3 CH_3 C_6H_5 OH OCH_3 NO_2	11.2 ± 0.1 11.3 ± 0.1 12.2 ± 0.1 11.4 ± 0.1 11.75 ± 0.2 13.40 ± 0.2 12.65 10.9 11.0 ± 0.2 11.6 ± 0.1 10.5 ± 0.1	EI EI EI EI EI EI EI PI EI EI EI EI	5429 5429 5429 5429 4199 4199 4199 4055 3807 3807 3807 3807
$\text{C}_{13}\text{H}_{10}\text{D}^+$	$\text{C}_6\text{H}_5(\text{CH}_2\text{D})\text{C}_6\text{H}_4\text{CH}_2\text{D}$ (1,1'-Biphenyl, 2,2'-di(methyl-d)-) $\text{C}_6\text{H}_5(\text{CH}_2\text{D})\text{C}_6\text{H}_4\text{CH}_2\text{D}$ (1,1'-Biphenyl, 4,4'-di(methyl-d)-)	52889-80-2 52889-82-4	CH_2D	11.80 ± 0.2 12.95	EI EI	4199 4199
$\text{C}_{13}\text{H}_9\text{D}_2^+$	$\text{C}_6\text{H}_5(\text{CH}_2\text{D})\text{C}_6\text{H}_4\text{CH}_2\text{D}$ (1,1'-Biphenyl, 4,4'-di(methyl-d)-)	52889-82-4	CH_3	12.65 ± 0.2	EI	4199
$\text{C}_{13}\text{H}_{12}^+$	$(\text{C}_6\text{H}_5)_2\text{CH}_2$ (Benzene, 1,1'-methylenebis-)	101-81-5	** ** ** ** ** ** **	8.55 ± 0.03 8.67 ± 0.05 (V) 8.8 (V) 8.80 ± 0.02 (V) 8.7 ± 0.1 9.00 ± 0.05 9.4	PI PE PE PE EI EI EI	5552 4620 4211 3854 5429 3806 4228
	$\text{C}_{13}\text{H}_{12}$ (1H-Benz[f]indene, 2,3-dihydro-)	1624-26-6	**	7.85 ± 0.03 (V)	PE	4828
	$\text{C}_6\text{H}_5\text{C}_6\text{H}_4\text{CH}_3$ (1,1'-Biphenyl, 2-methyl-)	643-58-3	**	8.10 ± 0.02	PE	3702
	$\text{C}_6\text{H}_5\text{C}_6\text{H}_4\text{CH}_3$ (1,1'-Biphenyl, 3-methyl-)	643-93-6	**	7.95 ± 0.02	PE	3702

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₃H₁₂⁺						
	C ₆ H ₅ C ₆ H ₃ CH ₃ (1,1'-Biphenyl, 4-methyl-)	644-08-6	**	7.80±0.02	PE	3702
			**	8.6±0.1	EI	5429
	C ₁₃ H ₁₂ (5,10-Methanobenzocyclooctene, 5,10-dihydro-)	33627-04-2	**	8.25±0.05	PE	4866
	C ₁₃ H ₁₂ (3,8-Methanocyclobuta[b]naphthalene, 2a,3,8a-tetrahydro-(2a α ,3 α ,8 α ,8a α)-)	54483-68-4	**	8.35±0.05 (V)	PE	4866
	C ₁₃ H ₁₂ (3,8-Methanocyclobuta[b]naphthalene, 2a,3,8a-tetrahydro-((2a α ,3 β ,8 β ,8a α)-)	54483-73-7	**	8.42±0.05 (V)	PE	4866
	C ₁₀ H ₇ CH=CHCH ₃ (Naphthalene, 1-(1-propenyl)-)	22767-77-7	**	8.4±0.1	EI	5429
	C ₁₁ H ₈ C ₂ H ₄ (Spiro[7H-benzocycloheptene-7,1'-cyclopropane])	29150-13-8	**	8.0±0.1	EI	5429
C₁₃H₁₄⁺						
	C ₁₃ H ₁₄ (Azulene, 4,6,8-trimethyl-)	941-81-1	**	7.10 (V)	PE	5397
	C ₃ H ₃ (C ₆ H ₅)=C=C(CH ₃) ₂ (Benzene,[2-(2-methyl-1-propenylidene)cyclopropyl]-)	4544-23-4	**	7.73	PE	5625
	C ₁₃ H ₁₄ (1,2,4-Ethanylidenecyclobuta[cd]pentalene, octahydro-5,7-bis (methylene)-)	42607-62-5	**	8.50	PE	4036
	C ₁₃ H ₁₄ (5,10-Methanobenzocyclooctene, 5,6,7,10-tetrahydro-)	42919-37-9	**	8.66±0.05	PE	4866
	C ₁₃ H ₁₄ (5,10-Methanobenzocyclooctene, 5,6,9,10-tetrahydro-)	42919-38-0	**	8.54±0.05	PE	4866
	C ₁₃ H ₁₄ (3,8-Methanocyclobuta[b]naphthalene, 1,2,2a,3,8,8a-hexahydro-(2a α ,3 α ,8 α ,8a α)-)	67145-41-9	**	8.46±0.05 (V)	PE	4866
	C ₁₃ H ₁₄ (3,8-Methanocyclobuta[b]naphthalene, 1,2,2a,3,8,8a-hexahydro-(2a α ,3 β ,8 β ,8a α)-)	67109-90-4	**	8.60±0.05 (V)	PE	4866
C₁₃H₁₆⁺						
	C ₁₃ H ₁₆ (Bicyclo[5.4.2]trideca-7,9,11,12-tetraene)	XXXXX-XX-X	**	8.2 (V)	PE	3999
	C ₁₃ H ₁₆ (1,2,4-Ethanylidenecyclobuta[cd]pentalene, octahydro-5-methyl-7-methylene-, (1 α ,1a β ,2 α ,3a β ,4 α ,5 α ,5a β ,5b β -)	42607-64-7	**	9.10	PE	4036
	C ₁₃ H ₁₆ (5,10-Methanobenzocyclooctene, 5,6,7,8,9,10-hexahydro-)	33627-05-3	**	8.52±0.05 (V)	PE	4866
C₁₃H₁₈⁺						
	C ₆ H ₅ C ₃ H ₃ (tert-C ₄ H ₉) (Benzene, [1-(1,1-dimethylethyl)cyclopropyl]-)	63340-00-1	**	8.63 (V)	PE	4815
	C ₆ H ₄ (CH ₂) ₇ (Bicyclo[7.2.2]trideca-9,11,12-triene)	3761-63-5	**	8.21 (V)	PE	5339
C₁₃H₂₂⁺						
	C ₅ H ₁₁ (C ₄ H ₉) ₂ (1,3-Cyclopentadiene, 1,3-bis(1,1-dimethylethyl)-)	XXXXX-XX-X	**	7.79 (V)	PE	4324
	C ₁₃ H ₂₂ (1H-Phenalene, dodecahydro-)	2935-07-1	**	8.85	PI	4173
C₁₃H₂₄⁺						
	1-C ₁₃ H ₂₄	26186-02-7	**	9.90±0.02	PI	5583
	2-C ₁₃ H ₂₄	28467-75-6	**	9.28±0.02	PI	5583
	3-C ₁₃ H ₂₄	60186-78-9	**	9.14±0.03	PI	5583
	4-C ₁₃ H ₂₄	60186-79-0	**	9.07±0.03	PI	5583
	5-C ₁₃ H ₂₄	60186-80-3	**	9.09±0.03	PI	5583

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₃H₂₄⁺						
	6-C ₁₃ H ₂₄	42371-66-4	**	9.05±0.03	PI	5583
	n-C ₇ H ₁₅ C ₆ H ₉ (Cyclohexene,1-heptyl-)	15232-86-7	**	8.37±0.02	PI	5556
	C ₁₃ H ₂₄ (Cyclotridecene(E))	2484-65-3	**	8.63±0.15	EI	5532
C₁₃H₂₆⁺						
	((CH ₃) ₃ C) ₂ C=CHCH(CH ₃) ₂	50787-12-7	**	8.307±0.008	PE	3957
C₁₄H₈⁺						
	C ₁₈ H ₈ N ₄ (Dibenzo[f,h]quinoxaline-2,3-dicarbonitrile)	55408-49-6	2(CN) ₂	11.91	EI	5488
C₁₄H₁₀⁺						
	C ₁₄ H ₁₀ (Anthracene)	120-12-7	**	7.47	S	3857
			**	7.4	PI	3586
			**	7.40	PI	3877
			**	7.40	PE	3668
			**	7.40 (V)	PE	5436
			**	7.40 (V)	PE	5630
			**	7.41±0.02 (V)	PE	4913
			**	7.41±0.05	PE	3684
			**	7.41 (V)	PE	4701
			**	7.42±0.02 (V)	PE	4430
			**	7.43±0.03 (V)	PE	4887
			**	7.44±0.03 (V)	PE	4341
			**	7.47±0.01	PE	3644
			**	7.47±0.01	PE	3657
			**	7.47	PE	4364
			**	7.35	CTS	3577
			**	7.4	CTS	3543
	C ₆ H ₅ C≡CC ₆ H ₅ (Benzene, 1,1'-(1,2-ethynediyl)bis-)	501-65-5	**	7.90±0.02	PE	3854
			**	8.0±0.05	PE	3684
	C ₁₀ H ₆ C ₄ H ₄ (Cyclohepta[de]naphthalene)	208-20-8	**	7.10 (V)	PE	5597
	C ₁₄ H ₁₀ (Cyclopenta[e/f]heptalene)	209-42-7	**	6.84 (V)	PE	4572
	C ₁₀ H ₆ C ₄ H ₄ (6b,8a-Dihydrocyclobut[a]acenaphthylene)	XXXXX-XX-X	**	7.72 (V)	PE	5597
	C ₁₀ H ₆ C ₄ H ₄ (2,3-Dihydro-1,2,3-metheno-1H-phenalene)	XXXXX-XX-X	**	7.55 (V)	PE	5597
	C ₁₄ H ₁₀ (Phenanthrene)	85-01-8	**	7.85 (V)	PE	5619
			**	7.86±0.01	PE	3644
			**	7.86±0.02 (V)	PE	4913
			**	7.86 (V)	PE	4701
			**	7.87±0.02 (V)	PE	4430
			**	7.91 (V)	PE	5364
			**	7.92±0.02 (V)	PE	3702
			**	7.92±0.05	PE	3684
			**	8.03±0.01	EI	3588
			**	8.25	CTS	3577
	C ₆ H ₅ (C ₆ H ₅) ₂ (Benzene, 1,1'-(2-cyclohexen-1-ylidene)bis-)	31158-25-5		10.4±0.4	EI	4018
	C ₆ H ₅ (C ₆ H ₅) ₂ (Benzene, 1,1'-cyclohexylidenebis-)	21113-55-3		10.8±0.4	EI	4018
	C ₆ H ₅ (CH ₃)(C ₆ H ₅) ₂ (Benzene, 1,1'-(4-methylcyclohexylidene)bis-)	32812-65-0		10.2±0.4	EI	4018
	C ₁₀ H ₁₃ (CH ₃)(C ₆ H ₅) ₂ (Naphthalene, 1,2,3,4,4a,5,6,7-octahydro-4a-methyl-2,2-diphenyl-)	50592-50-2		9.3±0.4	EI	4018

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₄H₁₀⁺						
	C ₆ H ₅ (=O)(C ₆ H ₅) ₂ (Cyclohexanone, 2,2-diphenyl-)	22612-62-0		10.7±0.4	EI	4018
	C ₆ H ₈ (=O)(C ₆ H ₅) ₂ (Cyclohexanone, 4,4-diphenyl-)	4528-68-1		13.2±0.4	EI	4018
	C ₆ H ₇ (=O)(CH ₃)(C ₆ H ₅) ₂ (Cyclohexanone, 2-methyl-5,5-diphenyl-)	50592-49-9		9.6±0.4	EI	4018
	C ₆ H ₇ (=O)(CH ₃)(C ₆ H ₅) ₂ (Cyclohexanone, 6-methyl-2,2-diphenyl-)	50592-52-4		10.3±0.4	EI	4018
	C ₆ H ₈ (OH)(CH ₃)(C ₆ H ₅) ₂ (Cyclohexanol, 1-methyl-4,4-diphenyl-)	50592-47-7		10.5±0.4	EI	4018
	C ₆ H ₆ (=O)(CH ₃)(C ₆ H ₅) ₂ CH ₂ CH ₂ CHO (Cyclohexanepropanal, 1-methyl-2-oxo-3,3-diphenyl-)	XXXXX-XX-X		10.2±0.4	EI	4018
	C ₆ H ₆ (=O)(CH ₃)(C ₆ H ₅) ₂ CH ₂ CH ₂ COCH ₃ (Cyclohexanone, 2-methyl-2-(3-oxobutyl)-6,6-diphenyl-)	50592-55-7		10.0±0.4	EI	4018
	C ₁₄ H ₁₂ S (Dibenzo[b,e]thiepin,6,11-dihydro-)	1207-93-8	H ₂ S	9.76	EI	5414
	C ₁₄ H ₁₂ SO ₂ (Dibenzo[b,e]thiepin,6,11-dihydro-5,5-dioxide-)	23772-26-1	H ₂ SO ₂	10.00	EI	5414
	C ₆ H ₆ (=O)CH ₃ (C ₆ H ₅) ₂ CH ₂ CH=C(CH ₃)Cl (Cyclohexanone, 2-(3-chloro-2-butenyl)-2-methyl-6,6-diphenyl-)	50592-54-6	Cl	10.5±0.4	EI	4018
C₁₄H₁₀⁺2						
	C ₁₄ H ₁₀ (Anthracene)	120-12-7	**	21.3	OTH	5141
	C ₆ H ₅ C≡CC ₆ H ₅ (Benzene, 1,1'-(1,2-ethynediyl)bis-)	501-65-5	**	23.3	OTH	5141
C₁₄H₁₁⁺						
	C ₁₄ H ₁₂ S (Dibenzo[b,e]thiepin,6,11-dihydro-)	1207-93-8	HS	11.05	EI	5414
	C ₁₄ H ₁₂ SO ₂ (Dibenzo[b,e]thiepin,6,11-dihydro-5,5-dioxide-)	23772-26-1	HSO ₂	10.35	EI	5414
C₁₄H₁₂⁺						
	(C ₆ H ₅) ₂ CH=CH (Benzene,1,1'-(1,2-ethenediyil)bis-(E)-)	103-30-0	**	7.70±0.03	PI	5552
			**	7.70±0.02	PE	3854
			**	7.76	PE	3657
			**	7.87 (V)	PE	4464
			**	7.90±0.05 (V)	PE	4377
			**	7.91±0.05 (V)	PE	4333
	C ₆ H ₅ CH=CHC ₆ H ₅ (Benzene, 1,1'-(1,2-ethenediyil)bis-(Z))	645-49-8	**	7.80±0.02	PE	3854
			**	8.17 (V)	PE	4464
	C ₁₄ H ₁₂ (Benzene, 1,1'-(1,2-ethenediyil)bis-)	588-59-0	**	7.5	PI	3586
			**	7.93±0.03 (V)	PE	4767
			**	7.94	PE	5124
			**	10.30 (V)	PE	4856
			**	7.9	CTS	3577
	(C ₆ H ₅) ₂ C-CH ₂ (Benzene, 1,1'-ethenylidenebis-)	530-48-3	**	8.00±0.02	PE	3854
	C ₁₄ H ₁₂ (Phenanthrene, 9,10-dihydro-)	776-35-2	**	7.55±0.02	PE	3702
			**	8.19 (V)	PE	5364
	C ₆ H ₈ (C ₆ H ₅) ₂ (Benzene, 1,1'-(2-cyclohexen-1-ylidene)bis-)	31158-25-5		9.8±0.4	EI	4018
	C ₆ H ₁₀ (C ₆ H ₅) ₂ (Benzene, 1,1'-cyclohexylidenebis-)	21113-55-3		9.8±0.4	EI	4018
	C ₆ H ₇ (CH ₃)(C ₆ H ₅) ₂ (Cyclohexene, 1-methyl-4,4-diphenyl-)	50592-48-8		9.8±0.4	EI	4018

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₄H₁₂⁺						
	C ₆ H ₅ (CH ₃)(C ₆ H ₅) ₂ (Benzene, 1,1'-(4-methylcyclohexylidene)bis-)	32812-65-0		10.1±0.4	EI	4018
	C ₁₀ H ₁₃ (CH ₃)(C ₆ H ₅) ₂ (Naphthalene, 1,2,3,4,4a,5,6,7-octahydro-4a-methyl-2,2-diphenyl-)	50592-50-2		9.5±0.4	EI	4018
	C ₆ H ₅ (=O)(C ₆ H ₅) ₂ (Cyclohexanone, 2,2-diphenyl-)	22612-62-0		9.5±0.4	EI	4018
	C ₆ H ₅ (=O)(C ₆ H ₅) ₂ (Cyclohexanone, 4,4-diphenyl-)	4528-68-1		10.0±0.4	EI	4018
	C ₆ H ₇ (=O)(CH ₃)(C ₆ H ₅) ₂ (Cyclohexanone, 2-methyl-5,5-diphenyl-)	50592-49-9		10.0±0.4	EI	4018
	C ₆ H ₆ (=O)(CH ₃)(C ₆ H ₅) ₂ (Cyclohexanone, 6-methyl-2,2-diphenyl-)	50592-52-4		10.4±0.4	EI	4018
	C ₆ H ₆ (OH)(CH ₃)(C ₆ H ₅) ₂ (Cyclohexanol, 1-methyl-4,4-diphenyl-)	50592-47-7		10.1±0.4	EI	4018
	C ₆ H ₆ (=O)(CH ₃) ₂ (C ₆ H ₅) ₂ (Cyclohexanone, 2,2-dimethyl-6,6-diphenyl-)	50592-53-5		9.9±0.4	EI	4018
	C ₆ H ₆ (=O)(CH ₃)(C ₆ H ₅) ₂ CH ₂ CH ₂ CHO (Cyclohexanopropanal, 1-methyl-2-oxo-3,3-diphenyl-)	XXXXX-XX-X		10.3±0.4	EI	4018
	C ₆ H ₆ (=O)(CH ₃)(C ₆ H ₅) ₂ CH ₂ CH ₂ COCH ₃ (Cyclohexanone, 2-methyl-2-(3-oxobutyl)-6,6-diphenyl-)	50592-55-7		10.5±0.4	EI	4018
	C ₆ H ₆ (=O)(C ₆ H ₅)=CHS(CH ₂) ₃ CH ₃ (Cyclohexanone, 6-[{(butylthio)methylene]-2,2-diphenyl-})	50592-51-3		10.1±0.4	EI	4018
	C ₁₄ H ₁₂ SO ₂ (Dibenzo[b,e]thiepin,6,11-dihydro-5,5-dioxide-)	23772-26-1	SO ₂	10.20	EI	5414
	C ₆ H ₆ (=O)CH ₃ (C ₆ H ₅) ₂ CH ₂ CH=C(CH ₃)Cl (Cyclohexanone, 2-(3-chloro-2-butenyl)-2-methyl-6,6-diphenyl-)	50592-54-6	Cl	10.0±0.4	EI	4018
C₁₄H₁₃⁺						
	C ₆ H ₄ (CH ₃)C ₆ H ₄ CH ₃ (1,1'-Biphenyl, 2,2'-dimethyl-)	605-39-0	H	12.20±0.2	EI	4199
	C ₆ H ₄ (CH ₃)C ₆ H ₄ CH ₃ (1,1'-Biphenyl, 3,3'-dimethyl-)	612-75-9	H	13.00±0.2	EI	4199
	C ₆ H ₄ (CH ₃)C ₆ H ₄ CH ₃ (1,1'-Biphenyl, 4,4'-dimethyl-)	613-33-2	H	12.85	EI	4199
C₁₄H₁₄⁺						
	C ₆ H ₅ CH ₂ CH ₂ C ₆ H ₅ (Benzene, 1,1'-(1,2-ethanediyl)bis-)	103-29-7	**	9.00±0.05	EI	3806
	(C ₆ H ₇) ₂ (Bicycloheptatrienyl)	39473-62-6	**	8.62 (V)	PE	4820
	C ₁₄ H ₁₄ (Bicyclo[2.2.2]octane,2,3,5,6,7,8-hexa(methylene)-)	XXXXX-XX-X	**	8.38 (V)	PE	5315
	(C ₆ H ₄ CH ₃) ₂ (1,1'-Biphenyl, 2,2'-dimethyl-)	605-39-0	**	8.05±0.02	PE	3702
	(C ₆ H ₄ CH ₃) ₂ (1,1'-Biphenyl, 3,3'-dimethyl-)	612-75-9	**	8.80±0.05	EI	4199
	(C ₆ H ₄ CH ₃) ₂ (1,1'-Biphenyl, 4,4'-dimethyl-)	613-33-2	**	7.85±0.02	PE	3702
	C ₆ H ₄ (CH ₃)C ₆ H ₄ CH ₃ (1,1'-Biphenyl, 4,4'-dimethyl-)	613-33-2	**	8.70±0.05	EI	4199
	C ₆ H ₅ C ₆ H ₄ C ₂ H ₅ (1,1'-Biphenyl, 2-ethyl-)	1812-51-7	**	8.55±0.02 (V)	PE	3702
	C ₁₄ H ₁₄ (1,4-Methanonaphthalene, 1,4-dihydro-9-((1-methylethylidene)-))	7350-72-3	**	8.01 (V)	PE	4541
C₁₄H₁₆⁺						
	C ₁₄ H ₁₆ (Anthracene, 1,4,5,8,9-hexahydro-)	5910-28-1	**	8.16 (V)	PE	4531
	C ₃ H ₂ (C ₆ H ₅)(CH ₃)=C=C(CH ₃) ₂ (Benzene,[2-methyl-3-(2-methyl-1-propenylidene)cyclopropyl]-cis-)	33530-26-6	**	7.65	PE	5625
	C ₃ H ₂ (C ₆ H ₅)(CH ₃)=C=C(CH ₃) ₂ (Benzene,[2-methyl-3-(2-methyl-1-propenylidene)cyclopropyl]-trans-)	33530-27-7	**	7.63	PE	5625

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₄H₁₆⁺	C ₁₀ H ₇ (CH ₂) ₃ CH ₃ (Naphthalene, 1-butyl-)	1634-09-0	**	7.76	PE	3960
	C ₁₀ H ₄ (CH ₃) ₄ (Naphthalene, 2,3,6,7-tetramethyl-)	1134-40-3	**	7.60±0.03 (V)	PE	4828
C₁₄H₁₈⁺	(tert-C ₄ H ₉) ₂ (C≡C) ₃	20264-60-2	**	8.32±0.02 (V)	PE	4816
C₁₄H₂₂⁺	C ₆ H ₅ (C(CH ₃) ₃) ₂ (Benzene, 1,4-bis(1,1-dimethylethyl))	1012-72-2	**	8.30	PE	5574
C₁₄H₂₄⁺	(CH ₃) ₂ CC(C ₃ H ₇)C(C ₃ H ₇)C(CH ₂)CH ₃	54580-22-2	**	8.14 (V)	PE	4459
	C ₄ (CH ₃) ₄ (C(CH ₃) ₂) ₂ (Cyclobutane, 1,1,2,2-tetramethyl-3,4-bis(1-methylethylidene)-)	1133-23-9	**	7.49 (V)	PE	4459
C₁₄H₂₆⁺	C ₆ H ₁₃ C≡CC ₆ H ₁₃	35216-11-6	**	9.067±0.005	PE	4575
			**	9.03±0.04	PI	5583
	1-C ₁₄ H ₂₆	765-10-6	**	9.89±0.02	PI	5583
	2-C ₁₄ H ₂₆	638-60-8	**	9.26±0.03	PI	5583
	3-C ₁₄ H ₂₆	60212-32-0	**	9.17±0.02	PI	5583
	4-C ₁₄ H ₂₆	60212-33-1	**	9.11±0.03	PI	5583
	5-C ₁₄ H ₂₆	60212-34-2	**	9.10±0.03	PI	5583
	6-C ₁₄ H ₂₆	3730-08-3	**	9.09±0.02	PI	5583
	((CH ₃) ₂ C=C(iso-C ₃ H ₇) ₂	54580-23-3	**	8.22 (V)	PE	4459
	C ₁₄ H ₂₆ (Cyclotetradecene(E))	6568-33-8	**	8.70±0.15	EI	5532
			**	8.65±0.15	EI	5532
C₁₄H₂₈⁺	((CH ₃) ₃ C ₂ C=CHC(CH ₃) ₃	28923-90-2	**	8.169±0.012	PE	3957
	((iso-C ₃ H ₇) ₂ C ₂	7090-88-2	**	8.13 (V)	PE	4459
C₁₅H₉⁺	C ₁₄ H ₉ CH ₃ (Phenanthrene, 4-methyl-)	832-64-4	H ₂ +H	14.4±0.1	EI	3454
	C ₁₄ H ₈ (CH ₃) ₂ (Phenanthrene, 2,7-dimethyl-)	1576-69-8		17.6±0.1	EI	3454
	C ₁₄ H ₈ (CH ₃) ₂ (Phenanthrene, 4,5-dimethyl-)	3674-69-9		15.1±0.1	EI	3454
	C ₁₄ H ₆ (CH ₃) ₄ (Phenanthrene, 2,4,5,7-tetramethyl-)	7396-38-5	3CH ₃	14.5±0.1	EI	3454
	C ₁₄ H ₆ (CH ₃) ₄ (Phenanthrene, 3,4,5,6-tetramethyl-)	7343-06-8	3CH ₃	16.5±0.1	EI	3454
C₁₅H₁₁⁺	C ₁₄ H ₉ CH ₃ (Phenanthrene, 4-methyl-)	832-64-4	H	12.0±0.1	EI	3454
	C ₁₄ H ₈ (CH ₃) ₂ (Phenanthrene, 2,7-dimethyl-)	1576-69-8	CH ₃	13.5±0.1	EI	3454
	C ₁₄ H ₈ (CH ₃) ₂ (Phenanthrene, 4,5-dimethyl-)	3674-69-9	CH ₃	10.8±0.1	EI	3454
C₁₅H₁₂⁺	C ₁₄ H ₉ CH ₃ (Anthracene, 9-methyl-)	779-02-2	**	7.24±0.03 (V)	PE	4887
			**	7.25	PE	4171
			**	7.27 (V)	PE	5436
	C ₃ H ₂ (C ₆ H ₅) ₂ (Benzene, 1,1'-(1-cyclopropene-1,2-diyl)bis-)	24168-52-3	**	10.27 (V)	PE	4856

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₅H₁₂⁺						
	(C ₇ H ₆) ₂ C	73045-26-8	**	8.05 (V)	PE	5463
	(Bicyclo[2.2.1]hepta-2,5-diene,7,7'-methanediylidenebis-)					
	C ₁₅ H ₁₂	949-41-7	**	7.77 (V)	PE	4927
	(1H-Cyclopropa[<i>I</i>]phenanthrene, 1a,9b-dihydro-)					
	C ₁₅ H ₁₂	256-81-5	**	7.95 (V)	PE	4611
	(5H-Dibenzo [<i>a,d</i>]cycloheptene)					
	C ₁₄ H ₉ CH ₃	832-69-9	**	7.7±0.03	EI	3588
	(Phenanthrene, 1-methyl-)					
	C ₁₄ H ₉ CH ₃	2531-84-2	**	7.9±0.04	EI	3588
	(Phenanthrene, 2-methyl-)					
	C ₁₄ H ₉ CH ₃	832-71-3	**	7.68±0.01	EI	3588
	(Phenanthrene, 3-methyl-)					
	C ₁₄ H ₉ CH ₃	832-64-4	**	7.1±0.1	EI	3454
	(Phenanthrene, 4-methyl-)					
	C ₁₄ H ₉ CH ₃	883-20-5	**	7.70±0.02	EI	3588
	(Phenanthrene, 9-methyl-)					
	(C ₇ H ₆) ₂ C	73050-57-4	**	7.80 (V)	PE	5463
	(Tetracyclo[3.2.0.0 ^{2,7} .0 ^{4,6}]heptane,3,3'-methanetetrabis-)					
C₁₅H₁₃⁺						
	C ₆ H ₁₀ (C ₆ H ₅) ₂	21113-55-3		10.3±0.4	EI	4018
	(Benzene, 1,1'-cyclohexylidenebis-)					
	C ₆ H ₉ (CH ₃)(C ₆ H ₅) ₂	32812-65-0		10.6±0.4	EI	4018
	(Benzene, 1,1'-(4-methylcyclohexylidene)bis-)					
	C ₁₀ H ₁₃ (CH ₃)(C ₆ H ₅) ₂	50592-50-2		10.3±0.4	EI	4018
	(Naphthalene, 1,2,3,4,4a,5,6,7-octahydro-4a-methyl-2,2-diphenyl-)					
	C ₆ H ₈ (=O)(C ₆ H ₅) ₂	22612-62-0		9.7±0.4	EI	4018
	(Cyclohexanone, 2,2-diphenyl-)					
	C ₆ H ₈ (=O)(C ₆ H ₅) ₂	4528-68-1		10.5±0.4	EI	4018
	(Cyclohexanone, 4,4-diphenyl-)					
	C ₆ H ₇ (=O)(CH ₃)(C ₆ H ₅) ₂	50592-49-9		10.8±0.4	EI	4018
	(Cyclohexanone, 2-methyl-5,5-diphenyl-)					
	C ₆ H ₇ (=O)(CH ₃)(C ₆ H ₅) ₂	50592-52-4		10.3±0.4	EI	4018
	(Cyclohexanone, 6-methyl-2,2-diphenyl-)					
	C ₆ H ₈ (OH)(CH ₃)(C ₆ H ₅) ₂	50592-47-7		10.1±0.4	EI	4018
	(Cyclohexanol, 1-methyl-4,4-diphenyl-)					
	C ₆ H ₈ (=O)(CH ₃) ₂ (C ₆ H ₅) ₂	50592-53-5		10.3±0.4	EI	4018
	(Cyclohexanone, 2,2-dimethyl-6,6-diphenyl-)					
	C ₁₀ H ₁₁ (=O)(CH ₃)(C ₆ H ₅) ₂	50786-03-3		9.9±0.4	EI	4018
	(2(3 <i>H</i>)-Naphthalenone,4,4a,5,6,7,8-hexahydro-4a-methyl-7,7-diphenyl-)					
	C ₆ H ₆ (=O)(CH ₃)(C ₆ H ₅) ₂ CH ₂ CH ₂ CHO	XXXXXX-XX-X		10.5±0.4	EI	4018
	(Cyclohexanepropanal, 1-methyl-2-oxo-3,3-diphenyl-)					
	C ₆ H ₆ (=O)(CH ₃)(C ₆ H ₅) ₂ CH ₂ CH ₂ COCH ₃	50592-55-7		10.6±0.4	EI	4018
	(Cyclohexanone, 2-methyl-2-(3-oxobutyl)-6,6-diphenyl-)					
	C ₆ H ₆ (=O)(C ₆ H ₅)=CHS(CH ₂) ₃ CH ₃	50592-51-3		10.8±0.4	EI	4018
	(Cyclohexanone, 6-[butylthio)methylene]-2,2-diphenyl-)					
	C ₆ H ₆ (=O)CH ₃ (C ₆ H ₅) ₂ CH ₂ CH=C(CH ₃)Cl	50592-54-6	Cl	10.6±0.4	EI	4018
	(Cyclohexanone, 2-(3-chloro-2-buteno)-2-methyl-6,6-diphenyl-)					
C₁₅H₁₄⁺						
	C ₁₅ H ₁₄	1138-48-3	**	8.20	PE	5260
	(Benzene,1,1'-(1,2-cyclopropanediyl)bis-, <i>cis</i> -)					
	C ₁₅ H ₁₄	1138-47-2	**	8.05	PE	5260
	(Benzene,1,1'-(1,2-cyclopropanediyl)bis-, <i>trans</i> -)					
	(C ₆ H ₅) ₂ CH=CCH ₃	833-81-8	**	8.10±0.05 (V)	PE	4377
	(Benzene, <i>trans</i> -1,1'-(1-methyl-1,2-ethenediyl)bis-)					
	(C ₇ H ₆) ₂ CH ₂	73045-27-9	**	8.25 (V)	PE	5463
	(Dispiro[bicyclo[2.2.1]hepta-2,5-diene-7,1'-cyclopropane-2',7''-bicyclo[2.2.1]hepta[2,5]diene])					
	(C ₇ H ₆) ₂ CH ₂	73050-58-5		8.4 (V)	PE	5463
	(Dispiro[tetracyclo[3.2.0.0 ^{2,7} .0 ^{4,6}]heptane-3,1'-cyclopropane-2',3''-tetracyclo[3.2.0.0 ^{2,7} .0 ^{4,6}]heptane)					

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{15}\text{H}_{14}^+$	$\text{C}_{13}\text{H}_8(\text{CH}_3)_2$ (9 <i>H</i> -Fluorene, 9,9-dimethyl-)	4569-45-3	**	7.8 (V)	PE	4081
$\text{C}_{15}\text{H}_{16}^+$	$\text{C}_6\text{H}_5\text{CH}_2\text{CH}_2\text{C}_6\text{H}_4\text{CH}_3$ (Benzene, 1-methyl-2-(2-phenylethyl)-) $\text{C}_6\text{H}_5\text{CH}_2\text{CH}_2\text{C}_6\text{H}_4\text{CH}_3$ (Benzene, 1-methyl-3-(2-phenylethyl)-) $\text{C}_6\text{H}_5\text{CH}_2\text{CH}_2\text{C}_6\text{H}_4\text{CH}_3$ (Benzene, 1-methyl-4-(2-phenylethyl)-) $\text{C}_6\text{H}_5(\text{CH}_2)_3\text{C}_6\text{H}_5$ (Benzene, 1,1'-(1,3-propanediyl)bis-) $\text{C}_6\text{H}_5\text{C}_6\text{H}_4\text{CH}(\text{CH}_3)_2$ (1,1'-Biphenyl, 2-isopropyl-) $\text{C}_6\text{H}_5\text{C}_6\text{H}_4\text{C}_3\text{H}_7$ (1,1'-Biphenyl, 2-propyl-) $\text{C}_6\text{H}_5\text{CH}_2\text{CH}_2\text{C}_7\text{H}_7$ (1,3,5-Cycloheptatriene, 7-(2-phenylethyl)-)	34403-05-9 34403-06-0 14310-20-4 1081-75-0 19486-60-3 20282-28-4 712-32-6	** ** ** ** ** ** **	8.64 ± 0.05 8.59 ± 0.05 8.58 ± 0.05 8.60 ± 0.1 8.79 ± 0.05 8.50 ± 0.02 (V) 8.50 ± 0.02 (V) 8.06 ± 0.05	EI EI EI EI EI PE PE EI	5230 5230 5230 4925 5230 3702 3702 5230
$\text{C}_{15}\text{H}_{18}^+$	$\text{C}_{10}\text{H}_3(\text{CH}_3)_5$ (Azulene, 1,2,4,6,8-pentamethyl-)	XXXXX-XX-X	**	6.85 ± 0.03 (V)	PE	4828
$\text{C}_{15}\text{H}_{24}^+$	$\text{C}_9\text{H}_6(\text{CH}_3)_6$ (Tetracyclo[6.1.0.0 ^{2,4} .0 ^{5,7}]nonane, 3,3,6,6,9,9-hexamethyl- (1 α ,2 α ,4 α ,5 β ,7 β ,8 α -)) $\text{C}_{11}\text{H}_{12}(\text{CH}_3)_4$ (Undec-1,5,8-triene, 1,4,4,8-tetramethyl-)	51898-92-1 XXXXX-XX-X	** **	8.5 (V) 9.54 (V)	PE PE	5192 5314
$\text{C}_{15}\text{H}_{28}^+$	$\text{C}_{15}\text{H}_{28}$ (Cyclopentadecene(E)) $\text{C}_{15}\text{H}_{28}$ (Cyclopentadecene(Z))	2146-35-2 34458-54-3	** **	8.83 ± 0.15 8.80 ± 0.15	EI EI	5532 5532
$\text{C}_{16}\text{H}_8^+$	C_{16}H_8 (Dibenzo[<i>a,e</i>]cyclooctene, 5,6,11,12-tetrahydro-)	53397-65-2	**	7.76 (V)	PE	4652
$\text{C}_{16}\text{H}_{10}^+$	$\text{C}_{16}\text{H}_{10}$ (Azuleno[2,1,8- <i>i,j</i>]azulene) $\text{C}_{16}\text{H}_{10}$ (Cyclohept[<i>f,g</i>]acenaphthylene) $\text{C}_{16}\text{H}_{10}$ (Dibenzo[<i>a,e</i>]cyclooctene, 5,6-didehydro-) $\text{C}_{16}\text{H}_{10}$ (Fluoranthene) $\text{C}_{16}\text{H}_{10}$ (Pyrene)	3526-04-3 194-32-1 53397-66-3 206-44-0 129-00-0 1576-69-8 3674-69-9	** ** ** ** ** ** ** ** ** ** **	7.14 ± 0.03 (V) 7.13 ± 0.04 7.56 (V) 7.95 ± 0.04 7.41 (V) 7.41 (V) 7.42 (V) 7.45 ± 0.01 7.45 17.7 ± 0.1 > 16	PE PE PE PE PE PE PE PE PE PE CTS EI EI	4263 4196 4652 4196 3951 4701 5364 3657 3577 3454 3454

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₆H₁₁⁺						
	C ₁₄ H ₆ (CH ₃) ₄ (Phenanthrene, 2,4,5,7-tetramethyl-)	7396-38-5	2CH ₃ +H	15.6±0.1	EI	3454
	C ₁₄ H ₆ (CH ₃) ₄ (Phenanthrene, 3,4,5,6-tetramethyl-)	7343-06-8	2CH ₃ +H	14.3±0.1	EI	3454
C₁₆H₁₂⁺						
	C ₁₄ H ₈ (=CH ₂) ₂ (Anthracene, 9,10-dihydro-9,10-bis(methylene)-)	3302-51-0	**	7.95 (V)	PE	4540
	C ₁₆ H ₁₂ (Azulene,2-phenyl-)	19227-07-7	**	7.20 (V)	PE	5397
	C ₁₆ H ₁₂ (Azulene,6-phenyl-)	23781-82-0	**	7.25 (V)	PE	5397
	C ₁₆ H ₁₂ (Azuleno[2,1,8- <i>i,j</i> a]azulene, 10b, 10c-dihydro-)	38765-94-5	**	7.33±0.03 (V)	PE	4263
	C ₁₆ H ₁₂ (Cyclohept[<i>f,g</i>]acenaphthylene, 1,2-dihydro-)	518-03-6	**	6.85±0.04	PE	4196
	C ₁₆ H ₁₂ (Dibenzo[<i>a,e</i>]cyclooctene)	262-89-5	**	7.8 (V)	PE	4652
	C ₁₀ H ₈ C ₆ H ₅ (Naphthalene, 2-phenyl-)	612-94-2	**	7.75	PE	4066
	C ₁₄ H ₆ (CH ₃) ₄ (Phenanthrene, 2,4,5,7-tetramethyl-)	7396-38-5	2CH ₃	14.0±0.1	EI	3454
	C ₁₄ H ₆ (CH ₃) ₄ (Phenanthrene, 3,4,5,6-tetramethyl-)	7343-06-8	2CH ₃	13.5±0.1	EI	3454
C₁₆H₁₃⁺						
	C ₁₆ H ₁₁ (Phenanthrene, 2,7-dimethyl-)	1576-69-8	H	13.5±0.1	EI	3454
	C ₁₄ H ₈ (CH ₃) ₂ (Phenanthrene, 4,5-dimethyl-)	3674-69-9	H	12.3±0.1	EI	3454
C₁₆H₁₄⁺						
	C ₆ H ₅ (CH=CH) ₂ C ₆ H ₅ (Benzene, 1,1'(1,3-butadiene-1,4-diyl)bis-)	886-65-7	**	7.54±0.03 (V)	PE	4767
			**	7.56	PE	5124
			**	8.05	PE	5202
	C ₁₄ H ₈ (CH ₃) ₂ (Cyclopenta[e]heptalene, 3,5-dimethyl-)	20672-23-5	**	6.73 (V)	PE	4572
	C ₁₄ H ₈ (CH ₃) ₂ (Phenanthrene, 2,7-dimethyl-)	1576-69-8	**	8.0±0.1	EI	3454
	C ₁₄ H ₈ (CH ₃) ₂ (Phenanthrene, 4,5-dimethyl-)	3674-69-9	**	7.6±0.1	EI	3454
	C ₆ H ₆ (=O)(C ₆ H ₅) ₂ (2-Cyclohexen-1-one, 4,4-diphenyl-)	4528-64-7		9.3±0.4	EI	4018
	C ₆ H ₆ (=O)(C ₆ H ₅) ₂ (Cyclohexanone, 2,2-diphenyl-)	22612-62-0		9.6±0.4	EI	4018
	C ₆ H ₇ (=O)(CH ₃)(C ₆ H ₅) ₂ (Cyclohexanone, 6-methyl-2,2-diphenyl-)	50592-52-4		9.2±0.4	EI	4018
	C ₆ H ₆ (=O)(CH ₃) ₂ (C ₆ H ₅) ₂ (Cyclohexanone, 2,2-dimethyl-6,6-diphenyl-)	50592-53-5		9.4±0.4	EI	4018
	C ₆ H ₆ (=O)(CH ₃)(C ₆ H ₅) ₂ CH ₂ CH ₂ CHO (Cyclohexanepropanal, 1-methyl-2-oxo-3,3-diphenyl-)	XXXXX-XX-X		9.4±0.4	EI	4018
	C ₆ H ₆ (=O)(CH ₃)(C ₆ H ₅) ₂ CH ₂ CH ₂ COCH ₃ (Cyclohexanone, 2-methyl-2-(3-oxobutyl)-6,6-diphenyl-)	50592-55-7		9.3±0.4	EI	4018
	C ₆ H ₆ (=O)CH ₃ (C ₆ H ₅) ₂ CH ₂ CH=CHCl (Cyclohexanone, 2-(3-chloro-2-butenyl)-2-methyl-6,6-diphenyl-)	50592-54-6	Cl	9.1±0.4	EI	4018
C₁₆H₁₅⁺						
	C ₁₆ H ₁₅ (1,6-Ethenocyclopenta[cd]pentaleno[2,1,6- <i>g,h,a</i>] pentalene, 1,1a,3a,3b,5a,5b,6,6a,6b,6c-decahydro-)	66081-13-8	**	8.74 (V)	PE	4832

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₆H₁₆⁺						
	C ₁₆ H ₁₆ (Tricyclo[8.2.2.2 ^{4,7}]hexadeca-4,6,10,12,13,15-hexaene)	1633-22-3	**	8.00 (V)	PE	4510
			**	7.60	PE	4158
			**	7.8	PE	5600
			**	8.08 (V)	PE	4088
			**	8.10 (V)	PE	5575
	(C ₆ H ₄ CH ₂ CH ₂) ₂ (Tricyclo[9.3.1.1 ^{4,8}]hexadeca-1(15),4,6,8(16),11,13-hexaene)	2319-97-3	**	8.20 (V)	PE	5575
			**	8.24 (V)	PE	4088
			**	8.24 (V)	PE	4231
C₁₆H₁₈⁺						
	(tert-C ₄ H ₉) ₂ (C≡C) ₄	20264-61-3	**	8.12±0.02 (V)	PE	4816
	C ₆ H ₅ C ₆ H ₄ C ₄ H ₉ (1,1'-Biphenyl, 2-butyl-)	XXXXX-XX-X	**	8.50±0.02 (V)	PE	3702
C₁₆H₂₀⁺						
	C ₁₀ H ₂ (CH ₃) ₆ (Azulene, 2,4,5,6,7,8-hexamethyl-)	63297-21-2	**	6.84±0.03 (V)	PE	4828
	C ₁₀ H ₂ (CH ₃) ₆ (Azulene, 3,4,5,6,7,8-hexamethyl-)	XXXXX-XX-X	**	6.73±0.03 (V)	PE	4828
	C ₈ (CH ₃) ₄ =CH ₂ ₄ (Tricyclo[3.3.0.0 ^{2,6}]octane,1,2,5,6-tetramethyl- 3,4,7,8-tetrakis(methylene)-)	34106-16-6	**	7.97±0.02 (V)	PE	5562
	C ₈ (CH ₃) ₄ =CH ₂ ₄ (Tricyclo[4.2.0.0 ^{2,5}]octane,1,2,5,6-tetramethyl- 3,4,7,8-tetrakis(methylene)-(1 α ,2 β ,5 β ,6 α -))	34101-24-1	**	8.10±0.02 (V)	PE	5562
C₁₆H₂₆⁺						
	C ₆ H ₅ (CH ₂ C(CH ₃) ₃) ₂ (Benzene,1,4-bis(2,2-dimethylpropyl))	1020-87-7	**	8.25	PE	5574
C₁₇H₁₂⁺						
	C ₁₇ H ₁₂ (10b,10c-Methanoazuleno[2,1,8-ij]azulene)	38801-41-1	**	7.15±0.03 (V)	PE	4263
	C ₁₇ H ₁₂ (1,1'-Spirobi[1H-indene])	165-42-4	**	7.80 (V)	PE	4083
C₁₇H₁₄⁺						
	C ₁₇ H ₁₄ (12H-1,11-Methenobenzo[1,2:4,5]dicycloheptene, 11a,12a-dihydro-)	25835-57-8	**	7.37±0.03 (V)	PE	4263
C₁₇H₁₅⁺						
	C ₁₄ H ₆ (CH ₃) ₄ (Phenanthrene, 2,4,5,7-tetramethyl-)	7396-38-5	CH ₃	11.5±0.1	EI	3454
	C ₁₈ H ₁₈ (Phenanthrene, 3,4,5,6-tetramethyl-)	7343-06-8	CH ₃	11.5±0.1	EI	3454
C₁₈H₁₀⁺						
	C ₁₈ H ₁₀ (Naphthacene)	92-24-0	**	6.9	PI	3586
C₁₈H₁₂⁺						
	C ₁₈ H ₁₂ (Benz[a]anthracene)	56-55-3	**	7.41 (V)	PE	4701
			**	7.41±0.02 (V)	PE	4913
			**	7.42 (V)	PE	4039
			**	7.47±0.01	PE	3644
			**	7.56±0.01	PE	3657
			**	7.5	CTS	3577
	C ₁₈ H ₁₂ (Benzo[c]phenanthrene)	195-19-7	**	7.60 (V)	PE	4701

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₈H₁₂⁺	C ₁₈ H ₁₂	195-19-7	**	7.62 (V)	PE	4039
			**	7.60±0.02 (V)	PE	4913
	C ₁₈ H ₁₂ (Chrysene)	218-01-9	**	7.59±0.02 (V)	PE	4913
			**	7.59 (V)	PE	4701
			**	7.60±0.01	PE	3644
			**	7.61 (V)	PE	4039
			**	7.75	CTS	3577
	C ₁₈ H ₁₂ (Naphthacene)	92-24-0	**	6.97±0.02 (V)	PE	4913
			**	7.01	PE	3668
			**	7.01 (V)	PE	4039
			**	7.04±0.04	PE	4196
	C ₁₈ H ₁₂ (Tetracyclo[6.6.2.1 ^{3,13} .1 ^{6,10}]octadeca-1,3(17),4,6,8,10(18),11,13,15-nonaene)	27313-56-0	**	8.06 (V)	PE	3647
			**	8.06 (V)	PE	4088
	C ₁₈ H ₁₂ (Triphenylene)	217-59-4	**	7.84±0.01	PE	3657
			**	7.86 (V)	PE	4039
			**	7.88±0.02 (V)	PE	4913
			**	7.88 (V)	PE	4701
			**	7.89±0.04	PE	4196
			**	8.1	CTS	3577
C₁₈H₁₄⁺	C ₆ H ₄ =C(C ₆ H ₅) ₂	2175-90-8	**	7.96 (V)	PE	4357
	(Benzene, (2,4-cyclopentadien-1-ylidene)phenylmethyl)-)					
	(C ₆ H ₅) ₂ C ₆ H ₄	92-94-4	**	7.83	PE	4478
	(1,1':4',1"-Terphenyl)					
	C ₁₈ H ₁₄	84-15-1	**	7.99±0.01	PE	3657
	(1,1':2',1"-Terphenyl)					
	C ₁₈ H ₁₄	92-06-8	**	8.01±0.01	PE	3657
	(1,1':3',1"-Terphenyl)					
	(C ₆ H ₅) ₂ C ₆ H ₄	92-94-4	**	7.78±0.01	PE	3657
	(1,1':4',1"-Terphenyl)					
C₁₈H₁₆⁺	C ₆ H ₅ (CH=CH) ₂ C ₆ H ₅	1720-32-7	**	7.27±0.03 (V)	PE	4767
	(Benzene, 1,1'-(1,3,5-hexatriene-1,6-diyil)bis-)					
	**					
	C ₁₈ H ₁₆	28255-97-2	**	7.33	PE	5124
	(11,1-Metheno-1H-cyclohepta[b]heptalene, 11a,12,13,13a-tetrahydro-)					
	C ₁₆ H ₁₀ (CH ₃) ₂	956-84-3	**	7.40±0.05 (V)	PE	4263
	(Pyrene, 10b,10c-dihydro-10b,10c-dimethyl-, trans-)					
C₁₈H₁₈⁺	(tert-C ₄ H ₉) ₂ (C≡C) ₅	XXXXXX-XX-X	**	8.06±0.02 (V)	PE	4816
	(C ₄ H ₉) ₂ C(CH ₃) ₃	13719-97-6	**	7.13±0.03 (V)	PE	4887
	(Anthracene, 9-(1,1-dimethylethyl)-)					
	C ₁₄ H ₆ (CH ₃) ₄	17597-70-5	**	6.59 (V)	PE	4572
	(Cyclopenta[e]heptalene, 3,5,8,10-tetramethyl-)					
	(C ₆ H ₃) ₂ (CH ₂) ₆	XXXXXX-XX-X	**	8.0±0.1 (V)	PE	5600
	([2,2]J(1,2,4)Cyclophane)					
	C ₁₄ H ₆ (CH ₃) ₄	7396-38-5	**	7.8±0.1	EI	3454
	(Phenanthrene, 2,4,5,7-tetramethyl-)					
	C ₁₄ H ₆ (CH ₃) ₄	7343-06-8	**	7.5±0.1	EI	3454
	(Phenanthrene, 3,4,5,6-tetramethyl-)					
	(C ₆ H ₃) ₂ (CH ₂) ₆	27165-88-4	**	7.70±0.02 (V)	PE	5600
	(Tetracyclo[6.6.2.1 ^{3,13} .1 ^{6,10}]octadeca-1,3(17),6,8,10(18),13-hexaene)					
	**					
				8.20±0.05 (V)	PE	5600

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₈H₁₈⁺	(C ₆ H ₅) ₂ (CH ₂) ₆	27165-88-4	**	7.88 (V)	PE	4701
C₁₈H₂₀⁺	C ₆ H ₁₀ (C ₆ H ₅) ₂ (Benzene, 1,1'-cyclohexylidenebis-)	21113-55-3	**	8.9±0.2	EI	4074
	C ₁₈ H ₂₀ (Naphthacene, 1,4,5,6,7,10,11,12-octahydro-)	60700-47-2	**	8.14 (V)	PE	4531
	C ₁₆ H ₁₄ (CH ₃) ₂ (Tricyclo[9.3.1.1 ^{4,8}]hexadeca-1(15),4,6,8(16),11,13-hexaene, 5,12-dimethyl-)	55705-29-8	**	7.98 (V)	PE	4231
C₁₉H₁₆⁺	(C ₆ H ₅) ₃ CH (Benzene, 1,1',1''-methylidynetris-)	519-73-3	**	8.34±0.03	PI	4055
			**	8.34±0.04	PI	5552
			**	8.40±0.05 (V)	PE	4620
C₁₉H₂₀⁺	C ₆ H ₇ (CH ₃)(C ₆ H ₅) ₂ (Cyclohexene, 1-methyl-4,4-diphenyl-)	50592-48-8	**	8.7±0.4	EI	4018
			H ₂ O	9.2±0.4	EI	4018
	C ₆ H ₈ (OH)(CH ₃)(C ₆ H ₅) ₂ (Cyclohexanol, 1-methyl-4,4-diphenyl-)	50592-47-7				
C₁₉H₂₂⁺	C ₆ H ₉ (CH ₃)(C ₆ H ₅) ₂ (Benzene, 1,1'-(4-methylcyclohexylidene)bis-)	32812-65-0	**	8.8±0.2	EI	4074
			**	8.8±0.2	EI	4074
C₂₀H₁₂⁺	C ₂₀ H ₁₂ (Azuleno[1,2,3-cd]phenalene)	54100-60-6	**	6.58 (V)	PE	4637
			**	7.76 (V)	PE	4637
			**	7.12±0.01	PE	3644
	C ₂₀ H ₁₂ (Azuleno[5,6,7-cd]phenalene)	6580-41-2	**	7.39±0.01	PE	5364
			**	7.41 (V)	PE	3657
			**	7.43±0.04	PE	4701
	C ₂₀ H ₁₂ (Benzo[a]pyrene)	50-32-8	**	6.90±0.01	PE	4196
			**	6.97 (V)	PE	3657
			**	7.00±0.01	PE	3644
	C ₂₀ H ₁₂ (Benzo[e]pyrene)	192-97-2	**	7.1	CTS	3577
			**	6.97 (V)	PE	4712
			**	6.97 (V)	PE	4701
	C ₂₀ H ₁₂ (Perylene)	198-55-0	**	7.00±0.01	PE	3644
			**	7.00±0.01	PE	3577
			**	7.1	CTS	3577
C₂₀H₁₄⁺	C ₁₄ H ₉ C ₆ H ₅ (Anthracene, 9-phenyl-)	602-55-1	**	7.25 (V)	PE	5436
			**	7.25 (V)	PE	5630
	C ₁₄ H ₉ C ₆ H ₅ (Phenanthrene, 9-phenyl-)	844-20-2	**	7.65 (V)	PE	4262
C₂₀H₁₈⁺	(C ₆ H ₅ CHCHCHCH) ₂ (Benzene, 1,1'-(1,3,5,7-octatetraene-1,8-diyl)bis-)	22828-29-1	**	7.19	PE	5124
			**	7.60 (V)	PE	5575
	C ₆ H ₄ (CH ₂ CH ₂) ₂ C ₁₀ H ₆ (5,14:8,11-Diethanobenzocyclododecane, 6,7,12,13-tetrahydro-)	4432-72-8	**	7.56 (V)	PE	5575
	C ₆ H ₄ (CH ₂ CH ₂) ₂ C ₁₀ H ₆ (1,5-(Ethano[1,4]benzenoethano)naphthalene)	60058-13-1	**			

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₂₀H₂₀⁺	(C ₆ H ₅) ₂ (CH ₂) ₈ ([2.2.2.2](1,2,3,4)Cyclophane)	XXXXX-XX-X **		7.9±0.1 (V)	PE	5600
	(C ₆ H ₅) ₂ (CH ₂) ₈ ([2.2.2.2](1,2,3,5)Cyclophane)	XXXXX-XX-X **		7.75±0.02 (V)	PE	5600
	(C ₆ H ₅) ₂ (CH ₂) ₈ ([2.2.2.2](1,2,4,5)Cyclophane)	XXXXX-XX-X **		7.67±0.02 (V)	PE	5600
C₂₀H₂₁⁺	C ₁₂ (CH ₃) ₈ (1,3,7,9-Cyclododecatetrayne, 5,5,6,6,11,11,12,12-octamethyl-)	61414-48-0 **		8.27±0.03 (V)	PE	4938
	(C ₆ H ₅)(CH ₂) ₄ C ₆ (CH ₃) ₈ (Tricyclo[8.2.2.2 ^{4,7}]hexadeca-4,6,10,12,13,15-hexaene, 4,5,7,8-tetramethyl-)	XXXXX-XX-X **		7.47 (V)	PE	5600
	C ₁₀ H ₁₂ (CH ₃) ₈ (Tricyclo[8.2.2.2 ^{4,7}]hexadeca-4,6,10,12,13,15-hexaene, 5,11,13,15-tetramethyl-,stereoisomer)	35233-71-7 **		7.52 (V)	PE	4771
	C ₆ H ₅ CH ₃ CH ₂ C ₆ (CH ₃) ₄ CH ₃ CH ₂ (Tricyclo[8.2.2.2 ^{4,7}]hexadeca-4,6,10,12,13,15-hexaene,5,6,15,16-tetramethyl-)	65304-59-8 **		7.55 (V)	PE	5575
C₂₀H₂₈⁺	(C ₆ H ₁₁ C) ₂ (Tricyclo[3.3.1.1 ^{3,7}]decane, tricyclo[3.3.1.1 ^{3,7}]decylidene-)	30541-56-1 **		7.84 (V)	PE	4459
C₂₀H₃₀⁺	C ₆ H ₅ (tert-C ₄ H ₉) ₃ (Pentalene,1,3,5-tris(1,1-dimethylethyl)-)	50356-52-0 **		7.11 (V)	PE	5613
C₂₀H₃₆⁺	((tert-C ₄ H ₉) ₂ C=C) ₂ C ₄ (tert-C ₄ H ₉) ₄ (1,3-Cyclobutadiene,1,2,3,4-tetrakis(1,1-dimethylethyl)-)	33512-45-7 ** 66809-05-0 **		7.0 6.35 (V)	PE PE	5034 5094
	C ₄ (tert-C ₄ H ₉) ₄ (Tricyclo[1.1.0.0 ^{2,1}]butane,tetrakis(1,1-dimethylethyl)-) (JC-Mean value of Jahn-Teller components)	66809-06-1 **		7.50 (V)	PE	5094
C₂₁H₁₅⁺	C ₁₀ H ₆ (CH ₃)C ₁₀ H ₆ CH ₃ (1,1'-Binaphthalene, 2,2'-dimethyl-)	32834-84-7	CH ₃	13.25±0.2	EI	4199
	C ₁₀ H ₆ (CH ₃)C ₁₀ H ₆ CH ₃ (1,1'-Binaphthalene, 3,3'-dimethyl-)	34042-82-5	CH ₃	12.25±0.2	EI	4199
	C ₁₀ H ₆ (CH ₃)C ₁₀ H ₆ CH ₃ (1,1'-Binaphthalene, 7,7'-dimethyl-)	34003-80-0	CH ₃	12.75±0.2	EI	4199
	C ₁₀ H ₆ (CH ₃)C ₁₀ H ₆ CH ₃ (1,1'-Binaphthalene, 8,8'-dimethyl-)	32693-05-3	CH ₃	11.50±0.2	EI	4199
	C ₁₀ H ₆ (CH ₃)C ₁₀ H ₆ CH ₃ (1,1'-Binaphthalene, 2,2'-dimethyl-)	32834-84-7	CH ₃	13.25	EI	3477
	C ₁₀ H ₆ (CH ₃)C ₁₀ H ₆ CH ₃ (1,1'-Binaphthalene, 3,3'-dimethyl-)	34042-82-5	CH ₃	12.25	EI	3477
	C ₁₀ H ₆ (CH ₃)C ₁₀ H ₆ CH ₃ (1,1'-Binaphthalene, 7,7'-dimethyl-)	34003-80-0	CH ₃	12.75	EI	3477
	C ₁₀ H ₆ (CH ₃)C ₁₀ H ₆ CH ₃ (1,1'-Binaphthalene, 8,8'-dimethyl-)	32693-05-3	CH ₃	11.50	EI	3477
	C ₁₂ H ₃₀ (Cyclopropene, bis-3,3'-triphenyl-)	XXXXX-XX-X **		9.3±0.05	EI	4628
	C ₃ (C ₆ H ₅) ₂ BF ₃ (Cyclopropenium, triphenyl-,tetrafluoroborate(1-))	741-16-2	BF ₃ ,F	9.3±0.05	EI	4628
	C ₃ (C ₆ H ₅) ₂ Cl (Cyclopropenium, triphenyl-,chloride)	58090-78-1	Cl	8.51±0.05	EI	4628
	C ₃ (C ₆ H ₅) ₂ Br (Cyclopropenium, triphenyl-,bromide)	4919-51-1	Br	8.35±0.05	EI	4628
	C ₃ (C ₆ H ₅) ₂ I (Cyclopropenium, triphenyl-,iodide)	58090-79-2	I	8.6±0.05	EI	4628

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₂₁H₁₄D⁺						
	C ₁₀ H ₆ (CH ₂ D)C ₁₀ H ₆ CH ₂ D (1,1'-Binaphthalene, 2,2'-di(methyl-d)-)	52889-79-9	CH ₂ D	13.05±0.2	EI	4199
	C ₁₀ H ₆ (CH ₂ D)C ₁₀ H ₆ CH ₂ D (1,1'-Binaphthalene, 8,8'-di(methyl-d)-)	52963-27-6	CH ₂ D	11.35±0.2	EI	4199
C₂₂H₁₂⁺						
	C ₂₂ H ₁₂ (Benzol[g,h,i]perylene)	191-24-2	**	7.15 (V)	PE	4701
			**	7.15 (V)	PE	4712
			**	7.19±0.01	PE	3644
	C ₂₂ H ₁₂ (Dibenzo[d,e,f,m,n,o]chrysene)	191-26-4	**	6.92±0.04	PE	4196
C₂₂H₁₄⁺						
	C ₂₂ H ₁₄ (Benzol[b]chrysene)	214-17-5	**	7.20±0.02 (V)	PE	4913
	C ₂₂ H ₁₄ (Benzol[a]naphthacene)	226-88-0	**	6.97±0.02 (V)	PE	4913
	C ₂₂ H ₁₄ (Benzol[b]chrysene)	214-17-5	**	7.14±0.04	PE	4196
	C ₂₂ H ₁₄ (Benzol[a]naphthacene)	226-88-0	**	7.06±0.04	PE	4196
	C ₂₂ H ₁₄ (3,4-Benzotetraphene)	XXXXX-XX-X	**	7.35±0.01	PE	3657
	C ₂₂ H ₁₄ (Benzol[b]triphenylene)	215-58-7	**	7.39±0.02 (V)	PE	4913
			**	7.39 (V)	PE	4701
			**	7.44±0.04	PE	4196
	C ₂₂ H ₁₄ (Dibenz[a,h]anthracene)	53-70-3	**	7.38±0.02 (V)	PE	4913
			**	7.38±0.04	PE	4196
			**	7.38 (V)	PE	4701
	C ₂₂ H ₁₄ (Dibenz[a,y]anthracene)	224-41-9	**	7.39±0.04	PE	4196
			**	7.40±0.02 (V)	PE	4913
			**	7.40 (V)	PE	4701
	C ₂₂ H ₁₄ (Dibenzo[c,g]phenanthrene)	188-52-3	**	7.47±0.04	PE	4196
			**	7.51 (V)	PE	4488
			**	7.51 (V)	PE	4701
			**	7.51±0.02 (V)	PE	4913
	C ₂₂ H ₁₄ (Pentacene)	135-48-8	**	6.61±0.02 (V)	PE	4913
			**	6.64	PE	3668
			**	6.74±0.01	PE	3644
	C ₂₂ H ₁₄ (Pentaphene)	222-93-5	**	7.27±0.02 (V)	PE	4913
			**	7.27 (V)	PE	4701
			**	7.34±0.04	PE	4196
	C ₂₂ H ₁₄ (Picene)	213-46-7	**	7.52±0.02 (V)	PE	4913
			**	7.52 (V)	PE	4701
			**	7.54±0.04	PE	4196
C₂₂H₁₈⁺						
	C ₁₀ H ₆ (CH ₃)C ₁₀ H ₆ CH ₃ (1,1'-Binaphthalene, 2,2'-dimethyl-)	32834-84-7	**	8.20±0.05	EI	4199
	C ₁₀ H ₆ (CH ₃)C ₁₀ H ₆ CH ₃ (1,1'-Binaphthalene, 3,3'-dimethyl-)	34042-82-5	**	8.20±0.05	EI	4199
	C ₁₀ H ₆ (CH ₃)C ₁₀ H ₆ CH ₃ (1,1'-Binaphthalene, 7,7'-dimethyl-)	34003-80-0	**	8.15±0.05	EI	4199

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₂₂H₁₈⁺						
	C ₁₀ H ₆ (CH ₃)C ₁₀ H ₆ CH ₃ (1,1'-Binaphthalene, 8,8'-dimethyl-)	32693-05-3	**	8.00±0.05	EI	4199
	C ₁₀ H ₆ (CH ₃)C ₁₀ H ₆ CH ₃ (1,1'-Binaphthalene, 2,2'-dimethyl-)	32834-84-7	**	8.20	EI	3477
	C ₁₀ H ₆ (CH ₃)C ₁₀ H ₆ CH ₃ (1,1'-Binaphthalene, 3,3'-dimethyl-)	34042-82-5	**	8.00	EI	3477
	C ₁₀ H ₆ (CH ₃)C ₁₀ H ₆ CH ₃ (1,1'-Binaphthalene, 7,7'-dimethyl-)	34003-80-0	**	8.15	EI	3477
	C ₁₀ H ₆ (CH ₃)C ₁₀ H ₆ CH ₃ (1,1'-Binaphthalene, 8,8'-dimethyl-)	32693-05-3	**	8.00	EI	3477
C₂₂H₂₀⁺	(C ₆ H ₅ CHCHCHCHCH) ₂ (Benzene, 1,1'-(1,3,5,7,9-decapentaene-1,10-diy)bis-)	XXXXX-XX-X **		7.05	PE	5124
C₂₂H₂₂⁺	(C ₆ H) ₂ (CH ₂) ₁₀ [[2.2.2.2](1,2,3,4,5)Cyclophane]	XXXXX-XX-X **		7.67±0.02 (V)	PE	5600
C₂₃H₂₆⁺	C ₁₀ H ₁₃ (CH ₃)(C ₆ H ₅) ₂ (Naphthalene, 1,2,3,4,4a,5,6,7-octahydro-4a-methyl-2,2-diphenyl-)	50592-50-2	**	8.9±0.2	EI	4074
C₂₄H₁₂⁺	C ₂₄ H ₁₂ (Coronene)	191-07-1	**	7.29 (V)	PE	4701
			**	7.29 (V)	PE	4712
			**	7.34 (V)	PE	3951
			**	7.5	CTS	3577
	C ₂₄ H ₁₂ (Tribenzo[a,e,i]cyclododecene, 5,6,11,12,17,18-hexadehydro-)	5385-26-2	**	7.45 (V)	PE	4652
C₂₄H₁₄⁺	C ₂₄ H ₁₄ (Benzo[rsi]pentaphene)	189-55-9	**	7.07±0.04	PE	4196
	C ₂₄ H ₁₄ (Benzo[a]perylene)	191-85-5	**	6.71 (V)	PE	4712
	C ₂₄ H ₁₄ (Benzo[b]perylene)	197-70-6	**	6.89 (V)	PE	4712
			**	6.92±0.04	PE	4196
	C ₂₄ H ₁₄ (Dibenzo[de,qr]naphthacene)	193-09-9	**	7.41±0.04	PE	4196
	C ₂₄ H ₁₄ (Dibenzo[fg,op]naphthacene)	192-51-8	**	7.40±0.04	PE	4196
	C ₂₄ H ₁₄ (Dibenzo[a,h]pyrene)	XXXXX-XX-X **		7.39 (V)	PE	4701
C₂₄H₁₆⁺	(C ₁₀ H ₆ C ₂ H ₂) ₂ (Pentacyclo[10.4.4.4 ^{4,9} .0 ^{6,22} .0 ^{15,19}]tetracosa-2,4,6,8,10, 12,14,16,17,19,21,23-dodecaene)	43012-17-5	**	7.40 (V)	PE	5575
	(C ₁₀ H ₆ C ₂ H ₂) ₂ (Pentacyclo[11.5.3.3 ^{4,10} .0 ^{7,23} .0 ^{16,20}]tetracosa-1(19),2,4,6,8, 10(22),11,13,15,17,20,23-dodecaene)	51557-75-6	**	7.20 (V)	PE	5575
C₂₄H₂₀⁺	(C ₁₀ H ₆ CH ₂ CH ₂) ₂ (5,16;8,13-Diethenodibenzo[a,g]cyclododecene,6,7,14,15-tetrahydro-)	14724-91-5	**	7.25 (V)	PE	5575
			**	7.50 (V)	PE	5575
	(C ₁₀ H ₆ CH ₂ CH ₂) ₂ (Pentacyclo[10.4.4.4 ^{4,9} .0 ^{5,21} .0 ^{16,20}]tetracosa-1(17),4,6,8, 12,14,16(20),18,21,23-decaene)	54835-57-3	**	7.05 (V)	PE	5575

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{24}\text{H}_{20}^+$	$(\text{C}_{10}\text{H}_6\text{CH}_2\text{CH}_2)_2$ $(\text{C}_{10}\text{H}_6\text{CH}_2\text{CH}_2)_2$ (Pentacyclo[10.4.4.4 ⁹ .0 ^{6,22} .0 ^{15,19}]tetracosa-4,6,8,12,14,16, 17,19,21,23-decaene) $(\text{C}_{10}\text{H}_6\text{CH}_2\text{CH}_2)_2$ (Pentacyclo[11.5.3.3 ^{4,10} .0 ^{7,23} .0 ^{16,20}]tetracosa-1(19),4,6,8,10(22), 13,15,17,20,23-decaene) $(\text{C}_{10}\text{H}_6\text{CH}_2\text{CH}_2)_2$ (Pentacyclo[13.3.2.2 ^{6,10} .1 ^{3,18} .1 ^{9,12}]tetracosa-1,3(21),6,8,10, 12(22),15,17,19,23-decaene)	54835-57-3 73608-51-2 7130-24-7 73608-52-3	** ** ** **	7.25 (V) 7.52 (V) 7.37 (V) 6.60 (V)	PE PE PE PE	5575 5575 5575 5575
$\text{C}_{24}\text{H}_{22}^+$	$(\text{C}_6\text{H}_5\text{CHCHCHCHCHCH})_2$ (Benzene, 1,1'-(1,3,5,7,9,11-dodecahexaene-1,12-diyl)bis-) $\text{C}_{10}\text{H}_7(\text{CH}_2)_4\text{C}_{10}\text{H}_7$ (Naphthalene, 1,1'-(1,4-butanediyl)bis-)	XXXXX-XX-X	**	7.07 7.67	PE PE	5124 3960
$\text{C}_{24}\text{H}_{24}^+$	$(\text{C}_6)_2(\text{CH}_2)_{12}$ ([2.2.2.2.2](1,2,3,4,5,6)Cyclophane)	XXXXX-XX-X	**	7.55 ± 0.02 (V)	PE	5600
$\text{C}_{25}\text{H}_{16}^+$	$\text{C}_{25}\text{H}_{16}$ (9,9'-Spirobi[9H-fluorene])	159-66-0	**	7.7 (V)	PE	4081
$\text{C}_{26}\text{H}_{14}^+$	$\text{C}_{26}\text{H}_{14}$ (Dibenzo[b,ghi]perylene) $\text{C}_{26}\text{H}_{14}$ (Dibenzo[b,pqr]perylene) $\text{C}_{26}\text{H}_{14}$ (Dibenzo[cd,lm]perylene) $\text{C}_{26}\text{H}_{14}$ (Naphtho[1,2,3,4-ghi]perylene) $\text{C}_{26}\text{H}_{14}$ (Naphtho[8,1,2-bcd]perylene)	5869-30-7 190-95-4 188-96-5 190-84-1 188-89-6	** ** ** ** ** **	6.99 (V) 7.12 (V) 6.72 ± 0.02 (V) 6.77 ± 0.04 6.96 (V) 6.82 ± 0.04	PE PE PE PE PE	4712 4712 4852 4196 4712 4196
$\text{C}_{26}\text{H}_{16}^+$	$\text{C}_{26}\text{H}_{16}$ (Benzo[c]picene) $\text{C}_{26}\text{H}_{16}$ (Benzo[a]pentacene) $\text{C}_{26}\text{H}_{16}$ (Benzo[c]pentaphene) $\text{C}_{26}\text{H}_{16}$ (Benzo[h]pentaphene) $\text{C}_{26}\text{H}_{16}$ (Benzo[b]picene) $\text{C}_{26}\text{H}_{16}$ (Benzo[c]picene) $\text{C}_{26}\text{H}_{16}$ (Dibenzo[a,j]naphthacene) $\text{C}_{26}\text{H}_{16}$ (Dibenzo[b,k]chrysene) $\text{C}_{26}\text{H}_{16}$ (Dibenzo[g,p]chrysene)	217-37-8 239-98-5 222-54-8 214-91-5 217-42-5 217-37-8 227-04-3 217-54-9 191-68-4	** ** ** ** ** ** ** ** ** ** ** ** ** **	7.36 ± 0.02 (V) 6.61 ± 0.02 (V) 6.72 ± 0.04 7.14 ± 0.04 7.20 ± 0.02 (V) 7.30 ± 0.04 7.36 ± 0.02 (V) 7.17 ± 0.02 (V) 7.20 ± 0.04 7.20 (V) 6.99 ± 0.02 (V) 6.97 ± 0.04 6.98 ± 0.02 (V) 7.18 ± 0.04	PE PE PE PE PE PE PE PE PE PE PE PE PE PE	4913 4913 4196 4196 4913 4196 4913 4913 4913 4701 4913 4196 4913 4196 4913 4196

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₂₆H₁₆⁺						
	C ₂₆ H ₁₆	191-68-4	**	7.20±0.02 (V)	PE	4913
	C ₂₆ H ₁₆ (Dibenzo[<i>a,c</i>]naphthacene)	216-00-2	**	6.96±0.04	PE	4196
			**	6.98±0.02 (V)	PE	4913
	C ₂₆ H ₁₆ (Dibenzo[<i>a,j</i>]naphthacene)	227-04-3		7.02±0.04	PE	4196
	(C ₁₃ H ₈) ₂ (9H-Fluorene, 9(9H-fluoren-9-ylidene)-)	746-47-4	**	7.27±0.04	PE	4196
	C ₂₆ H ₁₆ (Hexacene)	258-31-1	**	6.36±0.02 (V)	PE	4913
			**	6.44±0.04	PE	4196
	C ₂₆ H ₁₆ (Hexaphene)	222-78-6	**	6.92±0.02 (V)	PE	4913
			**	7.02±0.04	PE	4196
	C ₂₆ H ₁₆ (Naphtho[1,2- <i>b</i>]chrysene)	220-77-9	**	7.19±0.02 (V)	PE	4913
	C ₂₆ H ₁₆ (Naphtho[2,3- <i>g</i>]chrysene)	196-64-5	**	7.15±0.02 (V)	PE	4913
			**	7.15±0.04	PE	4196
	C ₂₆ H ₁₆ (Naphtho[2,1- <i>a</i>]naphthacene)	220-82-6	**	7.22±0.04	PE	4196
			**	6.83±0.02 (V)	PE	4913
	C ₂₆ H ₁₆ (Phenanthro[3,4- <i>c</i>]phenanthrene)	187-83-7	**	7.37 (V)	PE	4488
	C ₂₆ H ₁₆ (Tribenz[<i>a,c,h</i>]anthracene)	215-26-9	**	7.35±0.04	PE	4196
			**	7.40±0.02 (V)	PE	4913
			**	7.40 (V)	PE	4701
C₂₆H₂₄⁺						
	(C ₆ H ₅ CHCHCHCHCHCHCH) ₂ (Benzene,1,1'-(1,3,5,7,9,11,13-tetradecaheptaene-1,14-diyl)bis-)	62622-57-5	**	7.2±0.2	PE	5124
C₂₈H₁₄⁺						
	C ₂₈ H ₁₄ (Benz[<i>a</i>]coronene)	190-70-5	**	7.08 (V)	PE	4701
			**	7.08 (V)	PE	4712
	C ₂₈ H ₁₄ (Benz[<i>pqr</i>]naphtho[8,1,2- <i>bcd</i>]perylene)	190-71-6	**	6.92±0.04	PE	4196
	C ₂₈ H ₁₄ (Phenanthro[1,10,9,8- <i>opqua</i>]perylene)	190-39-6	**	6.30 (V)	PE	4712
C₂₈H₁₆⁺						
	C ₂₈ H ₁₆ (Benz[<i>p</i>]naphtho[1,8,7- <i>ghi</i>]chrysene)	385-14-8	**	7.00±0.04	PE	4196
	C ₂₈ H ₁₆ (Dibenzo[<i>de,st</i>]pentacene)	14147-38-7	**	6.82 (V)	PE	4712
	C ₂₈ H ₁₆ (Dibenzo[<i>de,uv</i>]pentacene)	193-11-3	**	7.03±0.04	PE	4196
	C ₂₈ H ₁₆ (Dibenzo[<i>fg,qr</i>]pentacene)	197-74-0	**	6.86 (V)	PE	4712
	C ₂₈ H ₁₆ (Dibenzo[<i>fg,st</i>]pentacene)	192-59-6	**	7.33±0.04	PE	4196
	C ₂₈ H ₁₆ (Dibenzo[<i>fg,ij</i>]pentaphene)	197-69-3	**	6.85 (V)	PE	4712
	C ₂₈ H ₁₆ (Dibenzo[<i>a,o</i>]perylene)	190-36-3	**	6.51 (V)	PE	4712
	C ₂₈ H ₁₆ (Dibenzo[<i>a,n</i>]perylene)	191-81-1	**	6.64 (V)	PE	4712
	C ₂₈ H ₁₆ (Dibenzo[<i>a,j</i>]perylene)	191-87-7	**	6.51 (V)	PE	4712

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₂₈H₁₆⁺	C ₂₈ H ₁₆ (Naphtho[1,2,3,4- <i>rst</i>]pentaphene)	191-20-8	**	7.09±0.04	PE	4196
C₂₈H₂₀⁺	C ₂₈ H ₂₀ (Azulene,1,2,3-triphenyl-)	XXXXX-XX-X	**	6.9 (V)	PE	5397
C₂₈H₃₄⁺	(C ₆ H ₅ (CH ₃) ₃) ₃ CH (Benzene, 1,1',1"-methylidynetris[2,4,6-trimethyl-])	52719-55-8	**	7.68±0.05 (V)	PE	4620
C₃₀H₁₄⁺	C ₃₀ H ₁₄ (Dibenzo[<i>bc,ef</i>]coronene)	190-31-8	**	6.50 (V)	PE	4712
	C ₃₀ H ₁₄ (Dibenzo[<i>bc,kl</i>]coronene)	190-55-6	**	6.42±0.02 (V)	PE	4852
C₃₀H₁₆⁺	C ₃₀ H ₁₆ (Anthra[1,2,3,4- <i>ghi</i>]perylene)	190-85-2	**	6.77 (V)	PE	4712
	C ₃₀ H ₁₆ (Benzo[<i>st</i>]naphtho[2,1,8,7- <i>defg</i>]pentacene)	14258-76-5	**	7.04 (V)	PE	4712
	C ₃₀ H ₁₆ (Benzo[<i>uv</i>]naphtho[2,1,8,7- <i>defg</i>]pentacene)	5869-31-8	**	6.78 (V)	PE	4712
	C ₃₀ H ₁₆ (Benzo[<i>qr</i>]naphtho[2,1,8,7- <i>fgih</i>]pentacene)	190-87-4	**	6.97 (V)	PE	4712
	C ₃₀ H ₁₆ (Tetrabenzo[<i>de,hi,mn,qr</i>]naphthacene)	385-13-7	**	6.90±0.04	PE	4196
	C ₃₀ H ₁₆ (Tribenzo[<i>de,kl,rst</i>]pentaphene)	188-72-7	**	6.42±0.02 (V)	PE	4852
	C ₃₀ H ₁₆ (Tribenzo[<i>b,n,pqr</i>]perylene)	190-81-8	**	7.13 (V)	PE	4701
			**	7.13 (V)	PE	4712
C₃₀H₁₈⁺	C ₃₀ H ₁₈ (Benzo[<i>p</i>]hexaphene)	222-81-1	**	6.59±0.02 (V)	PE	4913
	C ₃₀ H ₁₈ (Benzo[<i>c</i>]naphtho[2,1- <i>p</i>]chrysene)	27798-46-5	**	7.19±0.02 (V)	PE	4913
	C ₃₀ H ₁₈ (Dibenzo[<i>a,c</i>]pentacene)	216-08-0	**	6.62±0.02 (V)	PE	4913
			**	6.67±0.04	PE	4196
	C ₃₀ H ₁₈ (Dibenzo[<i>a,l</i>]pentacene)	227-09-8	**	6.64±0.02 (V)	PE	4913
	C ₃₀ H ₁₈ (Dibenzo[<i>c,m</i>]pentaphene)	222-51-5	**	7.11±0.02 (V)	PE	4913
			**	7.11 (V)	PE	4701
	C ₃₀ H ₁₈ (Dibenzo[<i>b,n</i>]picene)	213-44-5	**	7.17±0.02 (V)	PE	4913
	C ₃₀ H ₁₈ (Dinaphtho[2,1- <i>c</i> :1',2'- <i>g</i>]phenanthrene)	16914-68-4	**	7.25 (V)	PE	4488
	C ₃₀ H ₁₈ (Heptaphene)	222-75-3	**	6.89±0.02 (V)	PE	4913
			**	6.98±0.04	PE	4196
	C ₃₀ H ₁₈ (Naphtho[2,3- <i>c</i>]pentaphene)	222-58-2	**	7.04±0.02 (V)	PE	4913
	C ₃₀ H ₁₈ (Tetrabenzo[<i>a,c,h,j</i>]anthracene)	215-11-2	**	7.43±0.02 (V)	PE	4913
			**	7.43 (V)	PE	4701
			**	7.45±0.04	PE	4196
	C ₃₀ H ₁₈ (Tribenz[<i>a,c,y</i>] naphthacene)	215-96-3	**	6.99±0.02 (V)	PE	4913

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{30}\text{H}_{18}^+$	$\text{C}_{30}\text{H}_{18}$ (Trinaphthylene)	196-62-3	**	7.35 ± 0.02 (V)	PE	4913
$\text{C}_{32}\text{H}_{14}^+$	$\text{C}_{32}\text{H}_{14}$ (Ovalene)	190-26-1	**	6.71 (V)	PE	4712
			**	6.86 ± 0.01	PE	3644
$\text{C}_{32}\text{H}_{16}^+$	$\text{C}_{32}\text{H}_{16}$ (Dibenzo[<i>a,g</i>]coronene)	190-66-9	**	7.04 (V)	PE	4712
			**	7.04 (V)	PE	4701
	$\text{C}_{32}\text{H}_{16}$ (Dibenzo[<i>a,j</i>]coronene)	190-72-7	**	6.92 (V)	PE	4712
$\text{C}_{32}\text{H}_{18}^+$	$\text{C}_{32}\text{H}_{18}$ (Naphtho[2,3- <i>a</i>]coronene)	190-74-9	**	6.92 (V)	PE	4701
			**	6.88 (V)	PE	4712
$\text{C}_{32}\text{H}_{18}^+$	$\text{C}_{32}\text{H}_{18}$ (Dibenzo[<i>g,wx</i>]hexacene)	192-60-9	**	7.01 ± 0.04	PE	4196
	$\text{C}_{32}\text{H}_{18}$ (Dibenzo[<i>hi,uv</i>]hexacene)	192-54-1	**	7.30 ± 0.04	PE	4196
$\text{C}_{34}\text{H}_{16}^+$	$\text{C}_{34}\text{H}_{16}$ (Benzo[<i>pqr</i>]dinaphtho[8,1,2- <i>bcd</i> :2',1',8'- <i>lmn</i>]perylene)	188-11-4	**	6.74 ± 0.02 (V)	PE	4852
	$\text{C}_{34}\text{H}_{16}$ (Dibenzo[<i>g,ij</i>]phenanthro[2,1,10,9,8,7- <i>pqrstuv</i>]pentaphene)	187-94-0	**	6.82 ± 0.02 (V)	PE	4852
			**	6.82 (V)	PE	4712
$\text{C}_{34}\text{H}_{18}^+$	$\text{C}_{34}\text{H}_{18}$ (Benzo[<i>rst</i>]phenanthro[1,10,9- <i>cde</i>]pentaphene)	190-93-2	**	6.42 ± 0.02 (V)	PE	4852
	$\text{C}_{34}\text{H}_{18}$ (Dibenzo[<i>a,rst</i>]naphtho[8,1,2- <i>cde</i>]pentaphene)	191-46-8	**	6.59 ± 0.02 (V)	PE	4852
	$\text{C}_{34}\text{H}_{18}$ (Dibenzo[<i>g,ij</i>]naphtho[1,2,3,4- <i>rst</i>]pentaphene)	313-63-3	**	6.84 (V)	PE	4712
	$\text{C}_{34}\text{H}_{18}$ (Dibenzo[<i>m,pqr</i>]naphtho[1,2,3,4- <i>tuv</i>]picene)	XXXXX-XX-X	**	6.59 ± 0.02 (V)	PE	4852
	$\text{C}_{34}\text{H}_{18}$ (Tetrabeno[<i>de,hi,op,st</i>]pentacene)	191-79-7	**	6.27 ± 0.02 (V)	PE	4852
	$\text{C}_{34}\text{H}_{18}$ (Tetrabeno[<i>de,h,kl,rst</i>]pentaphene)	188-13-6	**	6.22 ± 0.02 (V)	PE	4852
	$\text{C}_{34}\text{H}_{18}$ (Tetrabeno[<i>a,cd,j,lm</i>]perylene)	191-53-7	**	6.71 ± 0.02 (V)	PE	4852
	$\text{C}_{34}\text{H}_{18}$ (Tetrabeno[<i>c,m,pqr,tuv</i>]picene)	XXXXX-XX-X	**	6.48 ± 0.02 (V)	PE	4852
$\text{C}_{34}\text{H}_{20}^+$	$\text{C}_{34}\text{H}_{20}$ (Benz[<i>j</i>]heptaphene)	214-87-9	**	6.90 ± 0.02 (V)	PE	4913
	$\text{C}_{34}\text{H}_{20}$ (Benzo[<i>a</i>]phenanthro[9,10- <i>c</i>]naphthacene)	385-16-0	**	6.73 ± 0.02 (V)	PE	4913
	$\text{C}_{34}\text{H}_{20}$ (Naphtho[2,1- <i>c</i>]phenanthro[4,3- <i>g</i>]phenanthrene)	20495-12-9	**	7.15 (V)	PE	4488
	$\text{C}_{34}\text{H}_{20}$ (Tetrabeno[<i>b,g,k,p</i>]chrysene)	385-15-9	**	6.83 ± 0.02 (V)	PE	4913
	$\text{C}_{34}\text{H}_{20}$ (Tetrabeno[<i>a,c,j,l</i>]naphthacene)	215-95-2	**	7.00 ± 0.02 (V)	PE	4913
			**	7.00 (V)	PE	4701

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₃₆H₁₆⁺						
	C ₃₆ H ₁₆ (Dinaphtho[8,1,2-abc:2',1',8-klm]coronene)	53086-28-5	**	6.76±0.02 (V)	PE	4852
			**	6.85±0.04	PE	4196
	C ₃₆ H ₁₆ (Dinaphtho[8,1,2-abc:8',1',2'-jkl]coronene)	190-47-6	**	6.70±0.04	PE	4196
C₃₆H₁₈⁺						
	C ₃₆ H ₁₈ (Dibenzo[f _g ,i _j]phenanthro[9,10,1,2,3-pqrst]pentaphene)	188-00-1	**	7.10 (V)	PE	4701
	C ₃₆ H ₁₈ (Tribenzo[a,d,g]coronene)	313-62-2	**	6.88 (V)	PE	4712
			**	6.88 (V)	PE	4701
C₃₆H₂₀⁺						
	C ₃₆ H ₂₀ (Dibenzo[hi,wx]heptacene)	197-73-9	**	6.68 (V)	PE	4712
	C ₃₆ H ₂₀ (Dinaphtho[1,2,3-fg:1',2',3'-qr]pentacene)	36474-85-8	**	6.82 (V)	PE	4712
C₃₈H₁₆⁺						
	C ₃₈ H ₁₆ (Naphth[2,1',8',7':4,10,5]anthra[1,9,8-abcd]coronene)	41163-25-1	**	6.81±0.02 (V)	PE	4852
			**	6.90 (V)	PE	4701
C₃₈H₁₈⁺						
	C ₃₈ H ₁₈ (Benzos[rs]dinaphtho[2,1,8,7-klmn:3',2',1',8',7'-vwxyz]hexaphene)	190-90-9	**	6.38±0.02 (V)	PE	4852
	C ₃₈ H ₁₈ (Dibenzo[jk,uv]dinaphtho[2,1,8,7-defg:2',1',8',7'-opqr]pentacene)	190-89-6	**	6.50±0.02 (V)	PE	4852
C₃₈H₂₀⁺						
	C ₃₈ H ₂₀ (Benzos[wz]naphtho[2,1,8,7-hijk]heptacene)	14529-73-8	**	6.72 (V)	PE	4712
	C ₃₈ H ₂₀ (Tribenzo[f _g ,mn,xyz]heptaphene)	34814-77-2	**	6.40±0.02 (V)	PE	4852
	C ₃₈ H ₂₀ (Tribenzo[de,h,kl]naphtho[1,2,3,4-rst]pentaphene)	187-96-2	**	6.06±0.02 (V)	PE	4852
C₃₈H₂₂⁺						
	C ₃₈ H ₂₂ (Diphenanthro[3,4-c:4'3'-g]phenanthrene)	20495-14-1	**	7.07 (V)	PE	4488
	C ₃₈ H ₂₂ (Tetrabenz[a,c,l,n]pentacene)	216-07-9	**	6.65±0.02 (V)	PE	4913
C₃₈H₅₆⁺						
	(C ₆ H ₆ (CH ₃) ₃ (CHCHC(CH ₃) ₂ CH) ₂) ₂ (β-Carotene,(all-E)-1,1'-(3,7,12,16-tetramethyl-1,3,5,7,9,11,13,15,17-octadecanonaene-1,18-diy)bis[2,6,6-trimethylcyclohexene])	7235-40-7		6.4±0.2	OTH	5278
C₄₀H₂₀⁺						
	C ₄₀ H ₂₀ (Benzos[1,2,3-cd:4,5,6-c'd]diperylene)	188-73-8	**	6.11±0.02 (V)	PE	4852
C₄₀H₅₆⁺						
	C ₄₀ H ₅₆ (1,3,5,7,9,11,13,15,17-Octadecanonene,3,7,12,16-tetramethyl-1,18-cyclohex-1-ene,2,6,6-trimethyl-)	XXXXXX-XX-X	**	6.5	PE	5093
C₄₂H₁₈⁺						
	C ₄₂ H ₁₈ (Hexabenzo[bc,ef,hi,kl,no,qr]coronene)	190-24-9	**	6.87±0.02 (V)	PE	4852
			**	6.87 (V)	PE	4712

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₄₂H₁₈⁺	C ₄₂ H ₁₈	190-24-9	**	6.89 (V)	PE	4701
C₄₂H₂₀⁺	C ₄₂ H ₂₀ (Dibenzo[<i>g,mn</i>]phenanthro[2,1,10,9,8,7- <i>wxyzab,b</i>]heptaphene)	34814-80-7	**	6.72±0.02 (V)	PE	4852
C₄₂H₂₂⁺	C ₄₂ H ₂₂ (Dibenzo[<i>g,mn</i>]naphtho[1,2,3,4- <i>xyz</i>]heptaphene)	34814-82-9	**	6.18±0.02 (V)	PE	4852
	C ₄₂ H ₂₂ (Hexabenzo[<i>a,cd,fj,lm,o</i>]perylene)	190-22-7	**	6.71±0.02 (V)	PE	4852
C₄₂H₂₄⁺	C ₄₂ H ₂₄ (Anthra[2,3- <i>j</i>]heptaphene)	214-77-7	**	6.85±0.02 (V)	PE	4913
	C ₄₂ H ₂₄ (Benzog[<i>g</i>]phenanthro[3,4- <i>c:6,5-c'</i>]diphenanthrene)	57520-29-3	**	6.99 (V)	PE	4488
	C ₄₂ H ₂₄ (Dibenzo[<i>f,j</i>]phenanthro[9,10- <i>s</i>]picene)	190-23-8	**	7.52±0.02 (V)	PE	4913
C₄₂H₃₀⁺	C ₆ (C ₆ H ₅) ₆ (Benzene, hexaphenyl-)	XXXXX-XX-X	**	8.47±0.05	EI	4628
	C ₄₂ H ₃₀ (Cyclopropene, bis-3,3'-triphenyl-)	XXXXX-XX-X	**	7.72±0.05	EI	4628
C₄₄H₂₀⁺	C ₄₄ H ₂₀ (Dibenzo[<i>a,jk</i>]phenanthro[8,9,10,1,2- <i>cdefgh</i>]pyranthrene)	70346-75-7	**	6.79±0.02 (V)	PE	4852
C₄₆H₂₆⁺	C ₄₆ H ₂₆ (Bisbenzo[5,6]phenanthro[3,4- <i>c:4',3'-g</i>]phenanthrene)	57468-45-8	**	6.95 (V)	PE	4488
	C ₄₆ H ₂₆ (Tetrabenzo[<i>a,c,g,s</i>]heptaphene)	62662-49-1	**	6.88±0.02 (V)	PE	4913
C₄₈H₂₄⁺	C ₄₈ H ₂₄ (Hexabenzo[<i>a,d,g,j,m,p</i>]coronene)	1065-80-1	**	6.75 (V)	PE	4712
			**	6.78 (V)	PE	4701
C₅₀H₂₈⁺	C ₅₀ H ₂₈ (Dinaphtho[1,2- <i>g</i> :1',2'- <i>g</i>]naphtho[2,1- <i>c</i> :7,8- <i>c'</i>]diphenanthrene)	57468-46-9	**	6.93 (V)	PE	4488
C₅₄H₃₀⁺	C ₅₄ H ₃₀ (Bisnaphtho[1',2':5,6]phenanthro[3,4- <i>c:4',3'-g</i>]phenanthrene)	24386-06-9	**	6.91 (V)	PE	4488
C₅₈H₃₂⁺	C ₅₈ H ₃₂ (Diphenanthro[4,3- <i>g</i> :4',3'- <i>g</i>]naphtho[2,1- <i>c</i> :7,8- <i>c'</i>]diphenanthrene)	57483-71-3	**	6.88 (V)	PE	4488
Li₃C⁺	CLi ₃	70378-93-7	**	4.6±0.3	EI	5334
LiCH₃⁺	(<i>tert</i> -C ₄ H ₉) ₂ Li ₁	25395-78-2		11.0±0.50	PI	5455
Li₂C₄H₉⁺	(<i>tert</i> -C ₄ H ₉) ₂ Li ₁	25395-78-2		8.1±0.25	PI	5455

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
Li₄C₄H₉⁺	(<i>tert</i> -C ₄ H ₉) ₄ Li ₄	25395-78-2		8.1±0.25	PI	5455
Li₄C₈H₁₈⁺	(<i>tert</i> -C ₄ H ₉) ₄ Li ₄	25395-78-2		8.1±0.25	PI	5455
Li₄C₁₂H₂₇⁺	(<i>tert</i> -C ₄ H ₉) ₄ Li ₄	25395-78-2		6.2±0.25	PI	5455
Li₄C₁₆H₃₆⁺	(<i>tert</i> -C ₄ H ₉) ₄ Li ₄	25395-78-2	**	6.2±0.25	OTH	5455
BeC₆H₅⁺	(C ₆ H ₅) ₂ Be (Beryllium, diphenyl-)	22300-89-6	C ₆ H ₅	13.4±0.2	EI	3815
BeC₆H₈⁺	(C ₅ H ₅)(CH ₃)Be (Beryllium, (η^5 -2,4-cyclopentadien-1-yl)methyl-)	36351-95-8	**	9.43 (V)	PE	5384
BeC₇H₆⁺	(C ₅ H ₅)(C ₂ H)Be (Beryllium, (η^5 -2,4-cyclopentadien-1-yl)ethynyl-)	52140-36-0	**	9.40 (V)	PE	5384
BeC₈H₈⁺	(C ₅ H ₅)(C ₂ CH ₃)Be (Beryllium(η^5 -2,4-cyclopentadien-1-yl)propynyl-)	XXXXX-XX-X	**	8.82 (V)	PE	5384
BeC₁₀H₁₀⁺	(C ₅ H ₅) ₂ Be (Beryllium,2,4-cyclopentadien-1-yl(η^5 -2,4-cyclopentadien-1-yl)-)	37048-03-6	**	7.45 (V)	PE	5108
BeC₁₂H₁₀⁺	(C ₆ H ₅) ₂ Be (Beryllium, diphenyl-)	22300-89-6	**	9.20±0.10	EI	3815
B₅CH₉⁺	CH ₃ B ₅ (2-Carbahexaborane(9))	12385-35-2	**	10.4 (V)	PE	4949
B₅CH₁₁⁺	B ₅ H ₈ CH ₃ (Pentaborane(9), 1-methyl-) B ₅ H ₈ CH ₃ (Pentaborane(9), 2-methyl-)	19495-55-7 23753-74-4	** **	10.20 (V) 10.30 (V)	PE PE	4519 4519
B₃C₂H₅⁺	C ₂ H ₅ B ₃ (1,5-Dicarbapentaborane)	20693-66-7	** **	10.54 10.9 (V)	PE PE	4446 4949
B₄C₂H₆⁺	C ₂ H ₆ B ₄ (1,6-Dicarbahexaborane(6))	20693-67-8	** **	9.9 (V) 9.77	PE PE	4949 4446
B₄C₂H₈⁺	C ₂ H ₈ B ₄ (2,3-Dicarbahexborane(8))	18972-20-8	**	9.6 (V)	PE	4949
B₅C₂H₇⁺	C ₂ H ₇ B ₅ (2,4-Dicarbaheptaborane(7))	20693-69-0	**	10.54	PE	4446

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
B₅C₂H₇⁺	C ₂ H ₇ B ₅	20693-69-0	**	10.6 (V)	PE	4949
B₈C₂H₁₀⁺	C ₂ B ₈ H ₁₀ (1,10-Dicarbadecaborane(10))	23653-23-8	**	10.5 (V)	PE	5324
B₁₀C₂H₁₂⁺	C ₂ B ₁₀ H ₁₂ (1,12-Dicarbadodecaborane(12))	20644-12-6	**	10.6 (V)	PE	5324
	C ₂ H ₁₂ B ₁₀ (1,7-Dicarbadodecaborane)	16986-24-6	**	10.19	PE	4446
BC₃H₉⁺	(CH ₃) ₃ B	593-90-8	**	10.68 (V)	PE	4398
			**	10.69 (V)	PE	4243
			**	10.69	PE	5485
BC₁₂H₁₀⁺	(C ₆ H ₅) ₃ B (Borane, triphenyl-)	960-71-4	C ₆ H ₅	10.2	PI	4055
BC₁₄H₁₉⁺	C ₆ H ₅ BC ₈ H ₁₄ (9-Borabicyclo[3.3.1]nonane, 9-phenyl-)	23418-91-9	**	9.16 (V)	PE	4956
BC₁₈H₁₅⁺	(C ₆ H ₅) ₃ B (Borane, triphenyl-)	960-71-4	**	8.60±0.03	PI	4055
N⁺	N	17778-88-0	**	14.549	PI	4355
	N ₂	7727-37-9	**	24.3	EI	5617
			N(⁴ S ^o)	24.34	EI	5051
			N	24.4±0.25	EI	3797
	NH ₃	7664-41-7	H ₂ +H	≤22.5	EI	3811
	N ₂ O	10024-97-2	NO	20±1	PI	5170
			NO	19.494	PE	4752
N⁺²	N ₂	7727-37-9	N	60.3±2	EI	3797
N₂⁺	N ₂	7727-37-9	**	15.5812±0.002	S	3561
(² Σ _g)			**	15.5	PI	5479
(² Π _g)			**	16.7	PI	5479
(² Σ _u)			**	18.8	PI	5479
(² Σ _g)			**	15.58	PE	4248
(² Σ _g)			**	15.58 (V)	PE	5055
(² Σ _g)			**	15.60 (V)	PE	4022
(² Σ _g)			**	15.61	PE	4073
(A ² Π _u)			**	16.695±0.002	PE	3935
(² Π _g)			**	16.70	PE	4248
(² Π _g)			**	16.73	PE	4073
(² Π _u)			**	16.98 (V)	PE	4022
(² Σ _u)			**	18.75	PE	4248
(² Σ _u)			**	18.78 (V)	PE	4022
(² Σ _u)			**	18.87 (V)	PE	3714
(² Σ _u)			**	24.6 (V)	PE	3714

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
N₂⁺						
(² Π _g)	N ₂	7727-37-9	**	29.0 (V)	PE	4615
(² Σ _g ⁺)			**	35 (V)	PE	3714
(² Σ _g ⁻)			**	39.8 (V)	PE	4615
(² Σ _u)			**	28-29 (V)	PE	3714
(² Σ _u ⁺)			**	32-33 (V)	PE	3714
(² Σ _u ⁻)			**	36-37 (V)	PE	3714
(² Σ _g)			**	15.58±0.02	EI	4877
(² Σ _g ⁺)			**	61.1±0.5	EI	5346
N ₂ H ₂		3618-05-1	H ₂	14.00±0.05	EI	4896
iso-N ₂ H ₂		28647-38-3	H ₂	13.70±0.05	EI	5248
N ₂ O		10024-97-2	O	18±1	PI	5170
N₂⁺²						
(¹ Σ _g)	N ₂	7727-37-9	**	43	EI	3452
(¹ Π _u)			**	43.1±0.5	OTH	5007
N ₂ ⁺		13966-04-6		45.2±0.5	OTH	5007
				28	EI	3452
HN⁺						
(² H)	NH	XXXXX-XX-X	**	13.49±0.01 (V)	PE	5011
	NH ₃	7664-41-7	H ₂	17.2	EI	3811
H₂N⁺						
(³ B ₁)	NH ₂	15194-15-7	**	11.46±0.01	PE	5011
(¹ A ₁)			**	12.45±0.01	PE	5011
(¹ B ₁)			**	14.27±0.01 (V)	PE	5011
NH ₃		7664-41-7	**	15.768±0.004	PI	5146
			H	15.0	EI	3811
CH ₃ NH ₂		74-89-5	CH ₃	15.9	EI	3808
HDN⁺						
NH ₂ D		13587-49-0	**	15.79±0.01	PI	5146
NHD ₂		13780-28-4	**	15.90±0.01	PI	5146
D₂N⁺						
ND ₂		54842-55-6	**	11.45±0.01	PE	5011
NHD ₂		13780-28-4	**	15.79±0.01	PI	5146
ND ₃		13550-49-7	**	15.89±0.01	PI	5146
H₃N⁺						
NH ₃		7664-41-7	**	10.18±0.09	PE	4497
			**	10.15	PE	3719
			**	10.2	PE	4623
			**	10.85 (V)	PE	5540
			**	11.3 (V)	PE	4845
			**	10.2	EI	3811
			**	10.45	EI	4759
NH ₄ Cl		12125-02-9		10.10±0.05	PI	4592
H₃N⁺²						
NH ₃		7664-41-7	**	35.3±0.7	OTH	5266
D₃N⁺						
ND ₃		13550-49-7	**	10.21	PE	3719
H₄N⁺						
C ₂ H ₅ NH ₂		75-04-7	C ₂ H ₂ +H	12.72±0.02	EI	3487
(CH ₃) ₂ NH		124-40-3		14.05±0.05	EI	3487
NH ₄ Cl		12125-02-9	Cl	10.10±0.05	PI	4592

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
HN₂⁺						
	N ₂ H	36882-13-0		7.8±0.05	EI	5248
	N ₂ H ₂	3618-05-1	H	10.98±0.05	EI	4896
			H	11.33±0.05	EI	5248
	N ₂ H ₂	15626-43-4	H	10.89±0.08	EI	4903
	<i>iso</i> -N ₂ H ₂	28647-38-3	H	10.77±0.05	EI	5248
H₂N₂⁺						
	N ₂ H ₂	3618-05-1	**	9.59±0.01	PE	4587
			**	9.59	PE	4408
			**	9.59	PE	5137
			**	9.7±0.1	EI	4896
				9.80±0.05	EI	5248
	<i>iso</i> -N ₂ H ₂	28647-38-3		9.52±0.05	EI	5248
	<i>trans</i> -N ₂ H ₂	15626-42-3	**	9.65±0.08	EI	4904
	N ₂ H ₄	302-01-2	2H	10.75±0.08	EI	4904
D₂N₂⁺						
	N ₂ D ₂	14989-24-3	**	9.61	PE	4408
			**	9.61	PE	5137
	N ₂ D ₂	40712-39-8	**	9.61±0.01	PE	4587
H₃N₂⁺						
	N ₂ H ₃	13598-46-4		7.85±0.05	EI	5248
	N ₂ H ₄	302-01-2	H	10.86±0.05	EI	5248
H₄N₂⁺						
	N ₂ H ₄	302-01-2	**	8.98±0.05	PE	4521
			**	9.90 (V)	PE	4137
			**	9.91 (V)	PE	3862
			**	10.07	PE	3747
			**	10.68 (V)	PE	5381
				8.93±0.05	EI	5248
HN₃⁺						
	HN ₃ (² A')	7782-79-8	**	10.70	PE	4500
			**	10.72±0.02	PE	3670
			**	10.72 (V)	PE	5151
			**	10.74	PE	4595
	(² A')		**	12.24±0.02 (V)	PE	3670
H₄N₄⁺						
	<i>trans</i> -H ₂ NN=NNH ₂	54410-57-0	**	8.99 (V)	PE	4432
H₆BN⁺						
	(BH ₃)(NH ₃)	xxxx-xx-x	**	9.44±0.02	PE	3699
H₆B₃N₃⁺						
	B ₃ H ₆ N ₃ (Borazine)	6569-51-3	**	9.88	PE	3637
			**	10.09 (V)	PE	3673
			**	10.14±0.01	PE	3506
CN⁺						
	((CH ₃) ₂ C(CN)NO) ₂	31018-29-8		16.50	EI	4809
	((CH ₃) ₂ C(NO)OOCCCH ₃) ₂	68777-98-0		15.90	EI	4809
	PF ₂ CN	14118-40-2	PF ₂	19.8±0.3	EI	4543
C₂N²⁺						
	C ₂ H ₅ CN	107-12-0		41.2	EI	5337

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₂N₂⁺	C ₂ N ₂	460-19-5	** **	13.51 (V) 35.5±0.5	PE OTH	5525 5147
C₄N₂⁺	CNC≡CCN	1071-98-3	** **	11.84 (V) 11.84 (V)	PE PE	4765 5525
C₆N₂⁺	CNC≡CC≡CCN	16419-78-6	**	11.2	S	4254
CN_i⁺	N ₃ C≡N	764-05-6	** **	10.96 (V) 11.00±0.01	PE PE	4392 4746
C₅N₄⁺	C(CN) ₄	24331-09-7		13.94	PE	4417
C₆N₄⁺	(NC) ₂ C=C(CN) ₂	670-54-2	** ** **	11.67±0.02 11.765±0.008 11.79±0.05 (V)	PI PI PE	5505 4306 4859
CHN⁺	HCN	74-90-8	** ** ** ** ** ** ** ** **	13.60 (V) 13.607±0.002 13.61±0.01 14.00±0.01 14.011±0.003 19.06±0.01 ~19.7 31.0 (V) 13.71	PE PE PE PE PE PE EI PE EI	5055 4525 3840 3840 4525 3840 4525 4525 3737
	CH ₃ NH ₂	74-89-5		12.5±0.1 (V)	PE	5457
	CH ₃ NC	75-05-8		12.5±0.1 (V)	PE	5457
	((CH ₃) ₂ C(CN)NO) ₂	31018-29-8		14.10	EI	4809
	DCN ⁺	3017-23-0	** **	13.613±0.002 13.999±0.003	PE PE	4525 4525
	CH ₂ N ⁺	CH ₂ CHCH ₂ CN CH ₂ C(CH ₃)CN C ₃ H ₅ CN (Cyclopropanecarbonitrile) C ₃ H ₅ NH (1H-Pyrrole)	109-75-1 126-98-7 5500-21-0 109-97-7	11.90 12.05 11.50 12.40	PI PI PI PI	5201 5201 5201 5201

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
CH_3N^+ (A')	$\text{CH}_2=\text{NH}$	2053-29-4	**	~10.0	PE	4489
CH_4N^+						
	CH_3NH_2	74-89-5		10.55	EI	4878
				10.70	EI	4878
	$(\text{CH}_3)_2\text{NH}$	124-40-3		10.80	EI	4878
	HCONHCH_3	123-39-7		11.65	EI	4878
	$\text{CH}_3\text{CONHCH}_3$	79-16-3		11.50	EI	4878
	$\text{NHCH}_3\text{CONH}_2$	598-50-5		11.65	EI	4878
	$(\text{NHCH}_3)_2\text{CO}$	96-31-1		11.45	EI	4878
	$\text{N}(\text{CH}_3)_2\text{CONHCH}_3$	632-14-4		11.45	EI	4878
	$\text{C}_2\text{H}_5\text{NO}_2$	56-40-6		10.27 ± 0.05	EI	3571
	$\text{NH}(\text{CH}_3)\text{CSNH}_2$	598-52-7		11.10	EI	4878
	$(\text{NHCH}_3)_2\text{CS}$	534-13-4		11.25	EI	4878
	$\text{N}(\text{CH}_3)_2\text{CSNHCH}_3$	2489-77-2		11.60	EI	4878
CH_5N^+						
	CH_3NH_2	74-89-5	**	8.80 ± 0.02	PE	3890
			**	8.89 ± 0.1	PE	4480
			**	9.08	PE	5510
			**	9.58 (V)	PE	4884
			**	9.58 (V)	PE	5249
			**	9.64 (V)	PE	4068
			**	9.64 (V)	PE	5063
			**	9.65 (V)	PE	4087
			**	9.45	EI	4759
$\text{C}_2\text{H}_2\text{N}^+$						
	$\text{C}_3\text{H}_4\text{N}_2$ (1 <i>H</i> -Imidazole)	288-32-4	HCN	13.2	EI	3910
	$\text{C}_3\text{H}_3\text{NO}$ (Oxazole)	288-42-6	HCO	14.1	EI	5400
$\text{C}_2\text{H}_3\text{N}^+$						
	CH_3CN	75-05-8	**	12.20 ± 0.01	PE	4679
			**	12.21 (V)	PE	4884
			**	12.46 (V)	PE	5525
	CH_3NC	593-75-9	**	11.32 (V)	PE	5525
	$\text{CH}_2\text{CHCH}_2\text{CN}$	109-75-1		11.10	PI	5201
	$\text{CH}_2\text{C}(\text{CH}_3)\text{CN}$	126-98-7		11.65	PI	5201
	$\text{C}_3\text{H}_5\text{CN}$ (Cyclopropanecarbonitrile)	5500-21-0		11.00	PI	5201
	$\text{C}_4\text{H}_4\text{NH}$ (1 <i>H</i> -Pyrrole)	109-97-7		11.75	PI	5201
	$\text{C}_3\text{H}_3\text{NO}$ (Oxazole)	288-42-6	CO	11.0	EI	5400
$\text{C}_2\text{H}_4\text{N}^+$						
	$(\text{CH}_3)_2\text{NCH}=\text{CHC}\equiv\text{CH}$	2206-24-8		13.1	EI	3674
	$(\text{C}_2\text{H}_5)_2\text{NCH}=\text{CHC}\equiv\text{CH}$	1809-53-6		13.6	EI	3674
$\text{C}_2\text{H}_5\text{N}^+$						
	$\text{CH}_2=\text{NCH}_3$	1761-67-7	**	9.90 ± 0.02 (V)	PE	4776
	$\text{CH}_3\text{CH}=\text{NH}$	20729-41-3	**	10.18 ± 0.02 (V)	PE	4776
	$\text{C}_2\text{H}_5\text{N}$ (Aziridine)	151-56-4	**	9.2 ± 0.1	PE	4990
			**	9.85 ± 0.02 (V)	PE	4133
$\text{C}_2\text{H}_6\text{N}^+$						
	$\text{C}_2\text{H}_5\text{NH}_2$	75-04-7	H	9.61 ± 0.09	EI	5467
	$(\text{CH}_3)_2\text{NH}$	124-40-3	H	9.41 ± 0.06	EI	5467

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₂H₆N⁺						
	(CH ₃) ₂ NH	124-40-3		10.50	EI	4878
				10.55	EI	4878
	(CH ₃) ₃ N	75-50-3	CH ₃	10.68±0.09	EI	5467
				11.25	EI	4878
	C ₂ H ₅ NHCH ₃	624-78-2	CH ₃	8.49±0.05	EI	5467
	n-C ₃ H ₇ NH ₂	107-10-8	CH ₃	10.2±0.3	EI	5467
	iso-C ₃ H ₇ NH ₂	75-31-0	CH ₃	8.86±0.05	EI	5467
	(C ₂ H ₅) ₂ NH	109-89-7	C ₂ H ₅	11.42±0.05	EI	5467
	n-C ₄ H ₉ NH ₂	109-73-9	C ₂ H ₅	9.49±0.09	EI	5467
	(CH ₃) ₂ NCH ₂ CH=CH ₂	2155-94-4	C ₃ H ₅	8.58	PI	5543
	(CH ₃) ₂ CH(CH ₂) ₂ NH ₂	107-85-7	iso-C ₃ H ₇	9.59±0.12	EI	5467
	n-C ₅ H ₁₁ NH ₂	110-58-7	C ₃ H ₇	9.34±0.10	EI	5467
	n-C ₄ H ₉ NHCH ₃	110-68-9	iso-C ₃ H ₇	8.37±0.06	EI	5467
	(CH ₃) ₂ NCH=CHC≡CH	2206-24-8	CH=CHC≡CH	12.7	EI	3674
	(CH ₃) ₂ NC ₄ H ₉	927-62-8	C ₆ H ₉	9.75±0.10	EI	5467
	C ₂ H ₅ NHC ₄ H ₉	13360-63-9	C ₆ H ₉	8.61±0.05	EI	5467
	(tert-C ₄ H ₉)N(CH ₃) ₂	918-02-5	tert-C ₄ H ₉	10.96±0.07	EI	5467
	C ₆ H ₅ CH ₂ N(CH ₃) ₂ (Benzinemethanamine,dimethyl-)	28262-13-7	C ₆ H ₅ CH ₂	9.62	PI	5543
	C ₆ H ₅ CH ₂ CH ₂ N(CH ₃) ₂ (Benzeneethanamine,dimethyl-)	29088-49-1	C ₆ H ₅ C ₂ H ₄	8.50	PI	5543
	CH ₃ C ₆ H ₄ CH ₂ N(CH ₃) ₂ (Benzinemethanamine,N,N, <i>ar</i> -trimethyl-)	56927-89-0	C ₆ H ₉	9.49	PI	5543
	HCON(CH ₃) ₂	68-12-2		11.60	EI	4878
	C ₂ H ₅ NHCHO	627-45-2	HCO	9.7±0.15	EI	5467
	CH ₃ CON(CH ₃) ₂	127-19-5		12.15	EI	4878
	N(CH ₃) ₂ CONH ₂	1320-50-9		11.65	EI	4878
	N(CH ₃) ₂ CONHCH ₃	632-14-4		11.70	EI	4878
	((CH ₃) ₂ N) ₂ CO	632-22-4		10.10	EI	4878
	N(CH ₃) ₂ CSNHCH ₃	2489-77-2		10.85	EI	4878
	((CH ₃) ₂ N) ₂ CS	2782-91-4		10.35	EI	4878
C₂H₇N⁺						
	C ₂ H ₅ NH ₂	75-04-7	**	8.76±0.1	PE	4480
			**	9.44±0.18 (V)	PE	3987
			**	9.471 (V)	PE	4527
			**	9.50 (V)	PE	4032
			**	9.50 (V)	PE	4068
			**	9.50 (V)	PE	5249
	(CH ₃) ₂ NH	124-40-3	**	8.07	PE	3589
			**	8.15±0.1	PE	4480
			**	8.2±0.1	PE	4990
			**	8.25±0.02	PE	3890
			**	8.30	PE	5510
			**	8.85 (V)	PE	4588
			**	8.929 (V)	PE	4527
			**	8.95 (V)	PE	5540
			**	8.97 (V)	PE	5063
			**	8.83	EI	4759
C₃HN⁺						
	CH≡CCN	1070-71-9	**	11.6	S	3755
			**	11.64±0.01	PI	3929
			**	11.75 (V)	PE	5525
C₃H₂N⁺						
	CH ₂ CHCH ₂ CN	109-75-1		12.05	PI	5201
	CH ₂ C(CH ₃)CN	126-98-7		12.20	PI	5201
	C ₃ H ₅ CN (Cyclopropanecarbonitrile)	5500-21-0		11.75	PI	5201
	C ₄ H ₅ NH (1H-Pyrrole)	109-97-7		12.50	PI	5201

Table of Ion Energetics Measurements—Continued

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₄H₃N⁺						
	CH ₂ =C=CHCN	1001-56-5	**	10.35 (V)	PE	4748
	CH ₃ C≡CCN	13752-78-8	**	10.78±0.02	PE	4765
			**	10.95 (V)	PE	5525
	(CH ₃) ₂ NCH=CHC≡CH	2206-24-8	2CH ₃	15.1	EI	3674
	C ₄ H ₈ NCH=CHC≡CH	19352-85-3	C ₄ H ₈	15.3	EI	3674
	(Pyrrolidine, 1-(1-butene-3-ynyl)-)					
	(C ₂ H ₅) ₂ NCH=CHC≡CH	1809-53-6		16.5	EI	3674
C₄H₄N⁺						
	CH ₂ CHCH ₂ CN	109-75-1	H	12.30	PI	5201
	CH ₂ C(CH ₃)CN	126-98-7	H	12.55	PI	5201
	C ₃ H ₅ CN (Cyclopropanecarbonitrile)	5500-21-0	H	12.10	PI	5201
	C ₄ H ₄ NH (1H-Pyrrole)	109-97-7	H	12.85	PI	5201
C₄H₅N⁺						
	CH ₂ CHCH ₂ CN	109-75-1	**	10.22	PE	5201
	CH ₂ C(CH ₃)CN	126-98-7	**	10.34	PE	5201
			**	10.37±0.02 (V)	PE	4609
			**	10.37±0.05 (V)	PE	4859
	trans-CH ₃ CH=CHCN	627-26-9	**	10.23±0.05 (V)	PE	4859
	C ₃ H ₅ CN (Cyclopropanecarbonitrile)	5500-21-0	**	10.25	PE	5201
	C ₄ H ₄ NH (1H-Pyrrole)	109-97-7	**	8.207±0.003	PI	5430
			**	8.208±0.005	PI	5274
			**	8.20±0.01	PI	4058
			**	8.23 (V)	PE	4009
				8.21	PE	5201
			**	~8.1	EI	4656
			**	8.22±0.05	EI	4316
			**	8.40±0.05	EI	3482
C₄H₆N⁺						
	(CH ₃) ₂ CC≡N	3225-31-8	**	8.56±0.06 (V)	PE	4609
	tert-C ₄ H ₉ CN	630-18-2		12.5	EI	4809
	((CH ₃) ₂ C(CN)NO) ₂	31018-29-8		9.00	EI	4809
C₄H₇N⁺						
	C ₄ H ₇ N (1H-Pyrrole, 2,5-dihydro-)	109-96-6	**	8.61±0.05 (V)	PE	4830
C₄H₈N⁺						
	(CH ₃) ₂ NCH ₂ CH=CH ₂	2155-94-4	CH ₃	9.62	PI	5543
C₄H₉N⁺						
	CH ₃ CH=NC ₂ H ₅	1190-79-0	**	9.440 (V)	PE	4527
	C ₂ H ₃ N(CH ₃) ₂ (Aziridine, 2,2-dimethyl-)	2658-24-4	**	9.29±0.02 (V)	PE	4133
	C ₄ H ₉ N (Pyrrolidine)	123-75-1	**	8.77±0.02 (V)	PE	4133
			**	8.77±0.02 (V)	PE	4480
			**	8.77±0.05 (V)	PE	4830
			**	8.82±0.03 (V)	PE	4452
			**	8.82 (V)	PE	4742
C₄H₁₀N⁺						
	(C ₂ H ₅) ₃ N	121-44-8	C ₂ H ₅	13.14	EI	3674

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₄H₁₁N⁺						
	(C ₂ H ₅) ₂ NH	109-89-7	**	7.85±0.1	PE	4480
			**	8.630 (V)	PE	4527
			**	8.68 (V)	PE	4588
	C ₂ H ₅ N(CH ₃) ₂	598-56-1	**	7.74±0.05	PE	4192
	n-C ₄ H ₉ NH ₂	109-73-9	**	9.40 (V)	PE	4068
	sec-C ₄ H ₉ NH ₂	13952-84-6	**	8.46±0.1	PE	4480
	iso-C ₄ H ₉ NH ₂	78-81-9	**	8.50±0.1	PE	4480
	tert-C ₄ H ₉ NH ₂	75-64-9	**	8.46±0.1	PE	4480
C₅H₄N⁺						
	(CH ₃) ₂ NCH=CHC≡CH	2206-24-8	CH ₃ +H ₂	12.4	EI	3674
	C ₄ H ₈ NCH=CHC≡CH (Pyrrolidine, 1-(1-buten-3-ynyl)-)	19352-85-3		15.0	EI	3674
C₅H₅N⁺						
	C ₅ H ₅ N (Pyridine)	110-86-1	**	9.25	PI	5028
			**	9.4	PI	3586
			**	9.26	PE	4867
			**	9.263	PE	3707
			**	9.51 (V)	PE	5258
			**	9.59 (V)	PE	3513
			**	9.60±0.5 (V)	PE	3685
			**	9.66 (V)	PE	4240
			**	9.7 (V)	PE	3832
			**	~9.5	EI	4530
			**	9.66±0.03	EI	3626
			**	9.70±0.05	EI	3498
			**	9.70	EI	5292
			**	9.74±0.05	EI	5413
			**	9.85±0.1	EI	4302
C₅H₆N⁺						
	(CH ₃) ₂ NCH=CHC≡CH	2206-24-8	CH ₃	11.2	EI	3674
	C ₄ H ₈ NCH=CHC≡CH (Pyrrolidine, 1-(1-buten-3-ynyl)-)	19352-85-3	CH ₂ =CHCH ₂	11.3	EI	3674
	(C ₂ H ₅) ₂ NCH=CHC≡CH	1809-53-6		13.9	EI	3674
C₅H₇N⁺						
	C ₄ H ₄ N(CH ₃) (1H-Pyrrole, 1-methyl-)	96-54-8	**	8.4	EI	3580
			**	7.94±0.02	PI	5430
	C ₅ H ₇ N (Pyridine, 1,4-dihydro-)	3337-17-5	**	7.46 (V)	PE	4586
	C ₄ H ₄ NCH ₃ (Pyrrole, 2-methyl-)	636-41-9	**	8.01±0.05	EI	3482
C₅H₈N⁺						
	(CH ₃) ₂ NCH ₂ C≡CH	7223-38-3	H	9.29	PI	5543
C₅H₉N⁺						
	C ₄ H ₆ N(CH ₃) (1H-Pyrrole, 2,5-dihydro-1-methyl-)	554-15-4	**	8.21±0.05 (V)	PE	4830
	(CH ₃) ₂ NCH ₂ C≡CH	7223-38-3	**	8.17	PI	5543
			**	8.22±0.05	PE	4192
	n-C ₄ H ₉ N≡C	2769-64-4	**	11.1 (V)	PE	4649
	C ₅ H ₉ N (Pyridine, 1,2,3,6-tetrahydro-)	694-05-3	**	8.64±0.05 (V)	PE	4830
C₅H₁₀N⁺						
	(CH ₃) ₂ NCH ₂ CH=CH ₂	2155-94-4	H	9.56	PI	5543

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₅H₁₁N⁺						
	C ₄ H ₈ N(CH ₃) (Pyrrolidine, 1-methyl-)	120-94-5	**	8.41±0.02 (V)	PE	4480
			**	8.41±0.02 (V)	PE	4133
			**	8.41±0.05 (V)	PE	4830
	(CH ₃) ₂ NCH ₂ CH=CH ₂	2155-94-4	**	7.84	PI	5543
			**	7.84±0.05	PE	4192
	C ₂ H ₅ CH=NC ₂ H ₅	18328-91-1	**	9.45 (V)	PE	4814
	C ₂ H ₂ N(CH ₃) ₃ (Aziridine, 1,2,2-trimethyl-)	23132-47-0	**	8.68±0.02 (V)	PE	4133
	C ₅ H ₁₁ N (Piperidine)	110-89-4	**	7.85±0.1	PE	4480
			**	8.05±0.05	PE	4996
			**	8.64±0.02 (V)	PE	4133
			**	8.64±0.05 (V)	PE	4830
			**	8.65±0.10 (V)	PE	5308
			**	8.66±0.03 (V)	PE	4452
			**	8.660 (V)	PE	4527
			**	8.67 (V)	PE	5540
C₅H₁₂N⁺						
	(C ₂ H ₅) ₃ N	121-44-8	CH ₃	11.48	EI	3674
C₅H₁₃N⁺						
	(C ₂ H ₅) ₂ (CH ₃)N	616-39-7	**	7.42±0.1	PE	4480
			**	8.32 (V)	PE	4564
	tert-C ₅ H ₁₁ NH ₂	5813-64-9	**	8.46±0.1	PE	4480
	neo-C ₅ H ₁₁ NH ₂	110-58-7	**	8.54±0.1	PE	4480
C₆H₅N⁺						
	C ₅ H ₅ CN (Cyclopentadienecarbonitrile)	27659-36-5	**	9.7	EI	3476
C₆H₆N⁺						
	C ₆ H ₅ COC ₆ H ₄ NH ₂ (Methanone, (2-aminophenyl)phenyl-)	2835-77-0		15.0±0.3	EI	4358
	C ₆ H ₅ COC ₆ H ₄ NH ₂ (Methanone, (3-aminophenyl)phenyl-)	2835-78-1		14.6±0.2	EI	4358
	C ₆ H ₅ COC ₆ H ₄ NH ₂ (Methanone, (4-aminophenyl)phenyl-)	1137-41-3		15.5±0.3	EI	4358
	C ₆ H ₄ (NH ₂)COOH (Benzoic acid, 3-amino-)	99-05-8	CO+OH	14.26±0.2	EI	3973
	C ₆ H ₄ (NH ₂)COOH (Benzoic acid, 4-amino-)	150-13-0	CO+OH	14.77±0.2	EI	3973
	C ₆ H ₄ (NO ₂)NH ₂ (Benzenamine, 3-nitro-)	99-09-2	NO ₂	11.23±0.1	EI	3447
	C ₆ H ₄ (NO ₂)NH ₂ (Benzenamine, 4-nitro-)	100-01-6	NO ₂	11.53±0.1	EI	3447
	C ₆ H ₄ ClNH ₂ (Benzenamine, 2-chloro-)	95-51-2	Cl	13.10	EI	4834
	C ₆ H ₄ BrNH ₂ (Benzenamine, 2-bromo-)	615-36-1	Br	12.50	EI	4834
	C ₆ H ₄ INH ₂ (Benzenamine, 2-iodo-)	615-43-0	I	11.60	EI	4834
C₆H₇N⁺						
	C ₆ H ₅ NH ₂ (Benzenamine)	62-53-3	**	7.7	PI	3586
			**	7.70±0.01	PI	4028
			**	7.65±0.02	PE	3890
			**	7.66	PE	3988
			**	7.71±0.01	PE	4154

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₆H₇N⁺						
	C ₆ H ₅ NH ₂	62-53-3	**	7.71	PE	3955
			**	7.80	PE	4621
			**	8.03 (V)	PE	4884
			**	8.05 (V)	PE	4106
			**	8.05 (V)	PE	4893
			**	8.10 (V)	PE	4159
			**	7.61±0.1	EI	3788
			**	7.63	EI	3845
			**	7.89±0.03	EI	3626
			**	7.89	EI	3485
			**	8.09±<0.1	EI	3735
			**	8.27±0.05	EI	5413
			**	8.35	EI	4834
			**	8.05 (V)	PE	5272
	CH ₃ C ₅ H ₄ N (Pyridine,2-methyl-)	109-06-8	**	9.18 (V)	PE	5258
			**	9.20±0.05 (V)	PE	3685
			**	9.20 (V)	PE	5527
			**	9.37±0.05	EI	5413
			**	9.4±0.1	EI	4302
	CH ₃ C ₅ H ₄ N (Pyridine,3-methyl-)	108-99-6	**	9.29 (V)	PE	5258
			**	9.43±0.05	EI	5413
			**	9.4±0.1	EI	4302
	CH ₃ C ₅ H ₄ N (Pyridine,4-methyl-)	108-89-4	**	9.41 (V)	PE	5258
			**	9.46±0.05	EI	5413
			**	9.50±0.05 (V)	PE	3685
			**	9.5±0.1	EI	4302
			**	9.55±0.05	EI	3498
			**	9.55	EI	5292
	C ₆ H ₅ (NH ₂)OCH ₃ (Benzenamine, 3-methoxy-)	536-90-3	CH ₂ O	10.51±0.1	EI	3446
	C ₆ H ₅ (NH ₂)OCH ₃ (Benzenamine, 4-methoxy-)	104-94-9	HCHO	9.58	EI	3845
	C ₆ H ₅ NHCOCH ₃ (Acetamide, N-phenyl-)	103-84-4		10.60	EI	4834
			CH ₂ =C=O	10.45±0.03	EI	3483
	C ₆ H ₅ NHCONH ₂ (Urea, phenyl-)	64-10-8		10.1	EI	4834
	(C ₆ H ₅ NH ₂)(CO) ₃ Cr (Chromium, (η^6 -benzenamine)tricarbonyl-)	12108-11-1		7.96±0.1	EI	3788
C₆H₈N⁺						
	(CH ₃) ₂ NCH=CHC≡CH	2206-24-8	H	10.1	EI	3674
C₆H₉N⁺						
	(CH ₃) ₂ NCH=CHC≡CH	2206-24-8	**	7.7	EI	3674
	C ₆ H ₆ NCH ₃ (Pyridine, 1,4-dihydro-N-methyl-)	33666-44-3	**	7.39 (V)	PE	4255
			**	7.39 (V)	PE	4586
	C ₄ H ₂ NH(CH ₃) ₂ (1H-Pyrrole,2,5-dimethyl-)	625-84-3	**	7.69 (V)	PE	5387
	C ₄ H ₄ NC ₂ H ₅ (Pyrrole, 2-ethyl-)	1551-06-0	**	7.97±0.05	EI	3482
C₆H₁₁N⁺						
	C ₅ H ₅ N(CH ₃) (Pyridine, 1,2,3,6-tetrahydro-1-methyl-)	694-55-3	**	8.67±0.05 (V)	PE	4830
	(CH ₂ =CHCH ₂) ₂ NH	124-02-7	**	8.79±0.3 (V)	PE	4818
			**	8.79 (V)	PE	5469

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₆H₁₁N⁺						
	CH ₃ CH=CHCH=NC ₂ H ₅	3653-19-8	**	9.3 (V)	PE	4814
C₆H₁₃N⁺						
	C ₅ H ₁₀ N(CH ₃) (Piperidine, 1-methyl-)	626-67-5	**	7.74±0.05	PE	4996
			**	8.29±0.02 (V)	PE	4133
			**	8.29±0.02 (V)	PE	4480
			**	8.29±0.05 (V)	PE	4830
	n-C ₃ H ₇ N=CHCH ₂ CH ₃	7707-70-2	**	8.55±0.2	EI	4360
	n-C ₃ H ₇ N=C(CH ₃) ₂	22023-64-9	**	8.31±0.2	EI	4360
	(iso-C ₃ H ₇)CH=NC ₂ H ₅	1743-56-2	**	9.25 (V)	PE	4814
	iso-C ₃ H ₇ N=C(CH ₃) ₂	3332-08-9	**	8.36±0.2	EI	4360
	iso-C ₃ H ₇ N=CHCH ₂ CH ₃	28916-23-6	**	8.50±0.2	EI	4360
	C ₆ H ₁₃ N (1H-Azepine, hexahydro-)	111-49-9	**	8.41±0.02 (V)	PE	4133
	C ₆ H ₁₁ NH ₂ (Cyclohexanamine)	108-91-8	**	8.37±0.1	PE	4480
	C ₅ H ₁₀ NCH ₃ (Piperidine, 2-methyl-)	109-05-7	**	7.76±0.05	PE	4996
	C ₅ H ₁₀ NCH ₃ (Piperidine, 3-methyl-)	626-56-2	**	7.94±0.05	PE	4996
	C ₅ H ₁₀ NCH ₃ (Piperidine, 4-methyl-)	626-58-4	**	8.01±0.05	PE	4996
C₆H₁₅N⁺						
	(C ₂ H ₅) ₃ N	121-44-8	**	7.11±0.1	PE	4480
			**	7.20±0.09	PE	4497
			**	8.08 (V)	PE	4564
			**	8.19±0.05 (V)	PE	3987
	n-C ₆ H ₁₃ NH ₂	111-26-2	**	8.63±0.05	PI	5508
	(n-C ₃ H ₇) ₂ NH	142-84-7	**	7.76±0.1	PE	4480
			**	8.59±0.3 (V)	PE	4818
	(iso-C ₃ H ₇) ₂ NH	108-18-9	**	7.59±0.1	PE	4480
C₇H₄N⁺						
	C ₆ H ₄ (CN)COOH (Benzoic acid, 4-cyano-)	619-65-8	CO+OH	15.68±0.2	EI	3973
	C ₆ H ₄ (NO ₂)CN (Benzonitrile, 3-nitro-)	619-24-9	NO ₂	12.25±0.1	EI	3447
	C ₆ H ₄ (NO ₂)CN (Benzonitrile, 4-nitro-)	619-72-7	NO ₂	12.42±0.1	EI	3447
C₇H₅N⁺						
	C ₆ H ₅ N≡C (Benzene, isocyanato-)	931-54-4	**	9.50 (V)	PE	4649
	C ₆ H ₅ CN (Benzonitrile)	100-47-0	**	9.62	PE	3938
			**	9.69	PE	4621
			**	9.70 (V)	PE	4334
			**	9.70 (V)	PE	4969
			**	9.71 (V)	PE	5259
			**	9.72 (V)	PE	5272
			**	9.7	EI	3916
			**	9.77	EI	3845
			**	10.13±0.03	EI	5080
	C ₆ H ₄ (CN)OCH ₃ (Benzonitrile, 3-methoxy-)	1527-89-5	CH ₂ O	12.23±0.1	EI	3446
	C ₆ H ₄ (CN)OCH ₃ (Benzonitrile, 4-methoxy-)	874-90-8	CH ₂ O	12.30±0.1	EI	3446
			HCHO	12.39	EI	3845

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₇H₈N⁺						
	C ₆ H ₄ (NH ₂)CH ₃ (Benzenamine, 2-methyl-)	95-53-4	H	11.25±0.05	PI	4028
	C ₆ H ₄ (NH ₂)CH ₃ (Benzenamine, 4-methyl-)	106-49-0	H	11.00±0.1	PI	4028
	C ₆ H ₄ (NH ₂)C ₄ H ₉ (Benzenamine, 3-butyl-)	5369-17-5		12.13±0.1	EI	3629
	C ₆ H ₄ (NH ₂)C ₄ H ₉ (Benzenamine, 4-butyl-)	104-13-2		11.10±0.1	EI	3629
	C ₆ H ₅ CH ₂ C ₆ H ₄ NH ₂ (Benzenamine, 4-(phenylmethyl)-)	1135-12-2	C ₆ H ₅	10.6±0.1	EI	3807
	(C ₆ H ₄ NH ₂) ₂ CH ₂ (Benzenamine, 4,4'-methylenabis-)	101-77-9		10.6±0.1	EI	3807
	C ₆ H ₄ (CH ₃)NHCOCH ₃ (Acetamide, N-(2-methylphenyl)-)	120-66-1	CH ₃ CO	13.97±0.02	EI	3631
	C ₆ H ₄ (CH ₃)NHCOC ₂ H ₅ (Acetamide, N-(4-methylphenyl)-)	103-89-9	CH ₃ CO	14.21±0.02	EI	3631
	C ₆ H ₄ (NH ₂)CH ₂ CH ₂ OCOCH ₃ (Benzeneethanol, 4-amino-, acetate(ester))	33709-38-5		11.00	EI	3590
	C ₆ H ₄ (NO ₂)CH ₂ C ₆ H ₄ NH ₂ (Benzenamine, 4-[(4-nitrophenyl)methyl]-)	726-17-0		11.6±0.2	EI	3807
	C ₅ H ₅ N(CH=CH ₂)BF ₄ (Pyridinium, 1-ethenyl-tetrafluoroborate (1-))	XXXXX-XX-X		9.0±0.1	EI	5502
C₇H₉N⁺						
	C ₆ H ₄ (NH ₂)CH ₃ (Benzenamine, 2-methyl-)	95-53-4	**	7.44±0.02	PI	4028
			**	7.45±0.02	PE	3890
			**	7.52	PE	3988
			**	7.83 (V)	PE	4106
			**	7.83 (V)	PE	5272
			**	7.84 (V)	PE	4893
	C ₆ H ₄ (NH ₂)CH ₃ (Benzenamine, 3-methyl-)	108-44-1	**	7.55	PE	3988
			**	7.66 (V)	PE	5272
			**	7.66 (V)	PE	4106
			**	7.82 (V)	PE	4893
	C ₆ H ₄ (NH ₂)CH ₃ (Benzenamine, 4-methyl-)	106-49-0	**	7.24±0.02	PI	4028
			**	7.37	PE	3988
			**	7.44±0.02	PI	4028
			**	7.62 (V)	PE	4106
			**	7.81 (V)	PE	4893
			**	7.85±0.05 (V)	PE	5013
	C ₆ H ₅ NHCH ₃ (Benzenamine, N-methyl-)	100-61-8	**	7.32	PE	3988
			**	7.35±0.02	PE	3890
	C ₆ H ₅ CH ₂ NH ₂ (Benzeneethanamine)	100-46-9	**	9.10±0.01 (V)	PE	4154
	C ₅ H ₅ N(CH ₃) ₂ (Pyridine, 2,5-dimethyl-)	589-93-5	**	8.80±0.05 (V)	PE	3685
	C ₅ H ₅ N(CH ₃) ₂ (Pyridine, 2,6-dimethyl-)	108-48-5	**	8.87	PE	4867
			**	9.23±0.05	EI	3498
			**	8.90±0.05 (V)	PE	3685
			**	9.23	EI	5292
	(CH ₃) ₂ C ₅ H ₃ N (Pyridine, 3,4-dimethyl-)	583-58-4	**	9.15 (V)	PE	5527
	(CH ₃) ₂ C ₅ H ₃ N (Pyridine, 3,5-dimethyl-)	591-22-0	**	9.25 (V)	PE	5527
	C ₆ H ₄ (NH ₂)C ₄ H ₉ (Benzenamine, 3-butyl-)	5369-17-5	CH ₂ =CHCH ₃	10.10±0.1	EI	3629

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₇H₉N⁺						
	C ₆ H ₅ (NH ₂)C ₂ H ₅ (Benzenamine, 4-butyl-)	104-13-2	CH ₂ =CHCH ₃	9.37±0.1	EI	3629
	C ₆ H ₅ (CH ₃)NHCOCH ₃ (Acetamide, N-(2-methylphenyl)-)	120-66-1	CH ₂ =C=O	10.05±0.02	EI	3631
	C ₆ H ₅ (CH ₃)NHCOCH ₃ (Acetamide, N-(4-methylphenyl)-)	103-89-9	CH ₂ =C=O	10.12±0.02	EI	3631
C₇H₁₀N⁺	(C ₂ H ₅) ₂ NCH=CHC≡CH	1809-53-6	CH ₃	13.1	EI	3674
C₇H₁₁N⁺						
	C ₇ H ₁₁ N (1-Azabicyclo[2.2.2]oct-2-ene)	13929-94-7	**	8.02	PE	5185
	C ₇ H ₁₁ N (2-Azabicyclo[2.2.2]oct-5-ene)	3693-58-1	**	8.35±0.05 (V)	PE	4830
	C ₇ H ₁₀ NH (2-Azabicyclo[3.2.1]oct-6-ene)	71017-41-9	**	8.60 (V)	PE	5481
	C ₆ H ₁₁ N≡C (Cyclohexane, isocyanato-)	931-53-3	**	11.0 (V)	PE	4649
	C ₅ H ₂ N(CH ₃) ₃ (Pyrrole, 1,3,4-trimethyl-)	30144-12-8	**	7.3	EI	3580
C₇H₁₃N⁺						
	C ₅ H ₇ (N(CH ₃) ₂) (2-Cyclopenten-1-amine, N,N-dimethyl-)	13044-51-4	**	9.32±0.05 (V)	PE	4954
	(CH ₂ =CHCH ₂) ₂ (CH ₃)N	2424-01-3	**	8.41±0.3 (V)	PE	4818
	C ₇ H ₁₃ N (1-Azabicyclo[2.2.2]octane)	100-76-5	**	7.50±0.09	PE	4497
			**	8.06±0.015 (V)	PE	4286
	C ₇ H ₁₃ N (2-Azabicyclo[2.2.2]octane)	280-38-6	**	8.22±0.05 (V)	PE	4830
	C ₇ H ₁₃ N (4-Azabicyclo[2.2.2]octane)	100-76-5	**	7.50±0.1	PE	4480
	C ₅ H ₇ N(CH ₃) ₂ (1-Cyclopenten-1-amine, N,N-dimethyl-)	4840-12-4	**	7.46 (V)	PE	5185
C₇H₁₅N⁺						
	C ₆ H ₁₂ NCH ₃ (1 <i>H</i> -Azepine, hexahydro-1-methyl-)	1192-95-6	**	8.29±0.02 (V)	PE	4133
	C ₅ H ₉ N(CH ₃) ₂ (Cyclopentanamine, N,N-dimethyl-)	18636-91-4	**	8.34 (V)	PE	5185
C₈H₆N⁺						
	C ₆ H ₄ (CN)C ₄ H ₉ (Benzonitrile, 3-butyl-)	20651-74-5		12.90±0.1	EI	3629
	C ₆ H ₄ (CN)C ₄ H ₉ (Benzonitrile, 4-butyl-)	20651-73-4		12.71±0.1	EI	3629
C₈H₇N⁺						
	C ₆ H ₅ CH ₂ CN (Benzeneacetonitrile)	140-29-4	**	9.34	EI	4934
	C ₆ H ₅ CH ₂ N≡C (Benzene, (isocyanomethyl)-)	10340-91-7	**	9.47 (V)	PE	4649
	C ₆ H ₄ (CH ₃)CN (Benzonitrile,2-methyl-)	529-19-1	**	9.38 (V)	PE	5272
			**	9.40 (V)	PE	5259
	C ₆ H ₄ (CH ₃)CN (Benzonitrile,3-methyl-)	620-22-4	**	9.34 (V)	PE	5259
			**	9.40 (V)	PE	5272
	C ₆ H ₄ (CH ₃)CN (Benzonitrile,4-methyl-)	104-85-8	**	9.38 (V)	PE	5259

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₈H₁₄N⁺						
	C ₁₀ H ₁₈ N ₂ O ₂ (8-Azabicyclo[3.2.1]octan-3-ol,8-methyl-methylcarbamate(ester), <i>exo</i> -)	67139-53-1	C ₂ H ₄ NO ₂	10.2±0.3	EI	5401
	C ₁₅ H ₂₀ N ₂ O ₂ (8-Azabicyclo[3.2.1]octan-3-ol,8-methyl-phenylcarbamate(ester)- <i>exo</i> -)	29364-21-4	C ₇ H ₆ NO ₂	9.2±0.3	EI	5401
	C ₁₅ H ₂₀ N ₂ OS (Carbamothioic acid,phenyl-0-(8-methyl-8-azabicyclo[3.2.1]oct-3-yl)ester, <i>endo</i> -)	67139-54-2	C ₇ H ₆ NOS	8.4±0.3	EI	5401
	C ₁₅ H ₂₀ N ₂ OS (Carbamothioic acid,phenyl-0-(8-methyl-8-azabicyclo[3.2.1]oct-3-yl)ester, <i>exo</i> -)	67139-55-3	C ₇ H ₆ NOS	8.6±0.3	EI	5401
	C ₈ H ₁₇ NO ₃ S (8-Azabicyclo[3.2.1]octan-3-ol,8-methyl-methanesulfonate(ester), <i>endo</i> -)	35130-97-3	CH ₃ O ₃ S	9.1±0.3	EI	5401
	C ₈ H ₁₇ NO ₃ S (8-Azabicyclo[3.2.1]octan-3-ol,8-methyl-methanesulfonate(ester), <i>exo</i> -)	35136-87-9	CH ₃ O ₃ S	9.6±0.3	EI	5401
	C ₈ H ₁₄ NCl (8-Azabicyclo[3.2.1]octane,3-chloro-8-methyl- <i>endo</i> -)	13514-03-9	Cl	9.1±0.3	EI	5401
	C ₈ H ₁₄ NCl (8-Azabicyclo[3.2.1]octane,3-chloro-8-methyl- <i>exo</i> -)	2292-12-8	Cl	9.5±0.3	EI	5401
	C ₈ H ₁₄ NBr (8-Azabicyclo[3.2.1]octane,3-bromo-8-methyl- <i>endo</i> -)	27809-79-6	Br	9.1±0.3	EI	5401
	C ₈ H ₁₄ NBr (8-Azabicyclo[3.2.1]octane,3-bromo-8-methyl- <i>exo</i> -)	2292-11-7	Br	8.9±0.3	EI	5401
C₈H₁₅N⁺						
	C ₈ H ₁₂ N(CH ₃) ₃ (2-Azabicyclo[2.2.2]octane,2-methyl-)	55100-40-8	**	7.78±0.05 (V)	PE	4830
	<i>tert</i> -C ₈ H ₉ CH=NCH ₂ CH=CH ₂	68003-54-3	**	9.31 (V)	PE	4968
	<i>tert</i> -C ₈ H ₉ CH=NCH=CHCH ₃	68003-65-6	**	8.69 (V)	PE	4968
	C ₈ H ₁₃ N (9-Azabicyclo[4.2.1]nonane)	284-18-4	**	8.50 (V)	PE	4136
	C ₈ H ₁₂ NCH ₃ (1-Azabicyclo[2.2.2]octane, 4-methyl-)	45651-41-0	**	8.06±0.015 (V)	PE	4286
	C ₈ H ₁₁ N(CH ₃) ₂ (1-Cyclohexen-1-amine,N,N-dimethyl-)	13815-46-8	**	7.56 (V)	PE	5185
	((CH ₃) ₂ N)CH=C(CH ₃) ₂ (Pyrrolidine, 1-(2-methyl-1-propenyl)-)	2403-57-8	**	7.66±0.03 (V)	PE	4452
C₈H₁₇N⁺						
	C ₆ H ₁₁ N(CH ₃) ₂ (Cyclohexanamine,N,N-diethyl-)	XXXXX-XX-X	**	8.09 (V)	PE	5185
	((CH ₃) ₂ N)CH ₂ CH(CH ₃) ₂ (Pyrrolidine, 1-(2-methylpropyl)-)	39198-81-7	**	8.17±0.03 (V)	PE	4452
C₉H₇N⁺						
	C ₈ H ₇ N (Isoquinoline)	119-65-3	**	8.50	PE	3638
			**	8.50	PE	4515
			**	8.54 (V)	PE	3723
	C ₈ H ₇ N (Quinoline)	91-22-5	**	8.3	PI	3586
			**	8.62	PE	3638
			**	8.62	PE	4066
			**	8.62 (V)	PE	3723
C₉H₉N⁺						
	C ₆ H ₅ CH(N=C)CH ₃ (Benzene, (1-isocyanethyl)-(R)-)	21872-33-3	**	9.37 (V)	PE	4649
	C ₆ H ₅ C ₆ H ₂ NCH ₃ (1H-Indole,1-methyl-)	603-76-9	**	7.74±0.03	PI	5552
			**	7.48±0.015	PE	5522
			**	7.71 (V)	PE	4586
	CH ₃ C ₆ H ₃ C ₂ H ₂ NH (1H-Indole,2-methyl-)	95-20-5	**	7.44±0.015	PE	5522

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_9\text{H}_9\text{N}^+$						
	$\text{CH}_3\text{C}_6\text{H}_3\text{C}_2\text{H}_2\text{NH}$ (1H-Indole,3-methyl-)	83-34-1	**	7.54 ± 0.015	PE	5522
	$\text{CH}_3\text{C}_6\text{H}_3\text{C}_2\text{H}_2\text{NH}$ (1H-Indole,4-methyl-)	16096-32-5	**	7.60 ± 0.015	PE	5522
	$\text{C}_6\text{H}_4\text{C}_2\text{H}(\text{CH}_3)\text{NH}$ (1H-Indole,6-methyl-)	3420-02-8	**	7.54 ± 0.015	PE	5522
	$\text{C}_6\text{H}_4\text{C}_2\text{H}(\text{CH}_3)\text{NH}$ (1H-Indole,7-methyl-)	933-67-5	**	7.53 ± 0.015	PE	5522
	$\text{C}_8\text{H}_6\text{NCH}_3$ (2H-Isoindole, 2-methyl-)	33804-84-1	**	7.12 (V)	PE	4935
			**	7.22 (V)	PE	4586
$\text{C}_9\text{H}_{11}\text{N}^+$						
	$\text{C}_9\text{H}_{11}\text{N}$ (Isoquinoline, 1,2,3,4-tetrahydro-)	91-21-4	**	8.57 ± 0.05 (V)	PE	4830
	$\text{C}_9\text{H}_{11}\text{N}$ (Quinoline, 1,2,3,4-tetrahydro-)	635-46-1	**	7.00 ± 0.02	PE	3890
$\text{C}_9\text{H}_{13}\text{N}^+$						
	$\text{C}_6\text{H}_4\text{CH}_3(\text{N}(\text{CH}_3)_2)$ (Benzeneamine,N,N,3-trimethyl-)	121-72-2	**	7.24 (V)	PE	5272
	$\text{C}_6\text{H}_4\text{CH}_3(\text{N}(\text{CH}_3)_2)$ (Benzeneamine,N,N,2-trimethyl-)	609-72-3	**	7.92 (V)	PE	5272
	$\text{C}_7\text{H}_{12}\text{NC}\equiv\text{CH}$ (1-Azabicyclo[2.2.2]octane, 4-ethynyl-)	52547-86-1	**	8.30 ± 0.015 (V)	PE	4286
	$\text{C}_6\text{H}_4\text{CH}_3\text{N}(\text{CH}_3)_2$ (Benzenamine, N,N,4-trimethyl-)	99-97-8	**	6.95	PE	3988
			**	6.9 ± 0.1	PE	4401
			**	7.27 (V)	PE	5272
	$\text{C}_6\text{H}_2(\text{CH}_3)_3\text{NH}_2$ (Benzenamine, 2,4,6-trimethyl-)	88-05-1	**	7.15	PE	3988
	$\text{C}_6\text{H}_3(\text{CH}_3)_2\text{NHCH}_3$ (Benzenamine, N,2,6-trimethyl-)	767-71-5	**	7.34	PE	3988
	$\text{C}_6\text{H}_4(\text{CH}_3)\text{N}(\text{CH}_3)_2$ (Benzenamine, N,N,2-trimethyl-)	609-72-3	**	7.40 ± 0.02	PE	3890
			**	7.44	PE	3988
			**	7.92 (V)	PE	4106
	$\text{C}_6\text{H}_4(\text{CH}_3)\text{N}(\text{CH}_3)_2$ (Benzenamine, N,N,3-trimethyl-)	121-72-2	**	7.06	PE	3988
			**	7.24 (V)	PE	4106
			**	7.27 (V)	PE	4106
	$\text{C}_6\text{H}_5\text{CH}_2\text{CH}_2\text{NHCH}_3$ (Benzeneethanamine, N-methyl-)	589-08-2	**	8.66 ± 0.20 (V)	PE	4672
	$\text{C}_6\text{H}_5\text{CH}_2\text{CH}(\text{NH}_2)\text{CH}_3$ (Benzeneethanamine, α -methyl- (\pm)-)	300-62-9	**	8.99 ± 0.06 (V)	PE	4758
			**	8.91 ± 0.14 (V)	PE	4672
	$\text{C}_6\text{H}_5\text{CH}_2\text{N}(\text{CH}_3)_2$ (Benzenemethanamine,dimethyl-)	103-83-3	**	7.69	PI	5543
			**	7.69 ± 0.05	PE	4192
	$\text{C}_6\text{H}_5(\text{CH}_3)_2\text{NH}_2$ (Benzepropanamine)	2038-57-5	**	8.89 ± 0.12 (V)	PE	4672
	$\text{C}_5\text{H}_4\text{NC}(\text{CH}_3)_3$ (Pyridine, 4-(1,1-dimethylethyl)-)	3978-81-2	**	9.30 ± 0.05 (V)	PE	3685
$\text{C}_9\text{H}_{15}\text{N}^+$						
	$(\text{CH}_2=\text{CHCH}_2)_3\text{N}$	102-70-5	**	8.30 ± 0.3 (V)	PE	4818
			**	8.30 (V)	PE	5469
	$\text{C}_9\text{H}_{15}\text{N}$ (1-Azatricyclo[3.3.1.1 ^{3,7}]decane)	281-27-6	**	7.57 ± 0.02	PE	4217
	$\text{C}_9\text{H}_8\text{NC}_2\text{H}_7$ (Pyrrolidine, 1-(1-cyclopenten-1-yl)-)	7148-07-4	**	7.10 ± 0.05 (V)	PE	4654

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_9\text{H}_{17}\text{N}^+$	$(\text{CH}_3)_2\text{C}=\text{NC}_6\text{H}_{11}$ $\text{C}_8\text{H}_{14}\text{NCH}_3$ (9-Azabicyclo[3.3.1]nonane, 9-methyl-) $\text{C}_7\text{H}_{12}\text{NC}_2\text{H}_5$ (1-Azabicyclo[2.2.2]octane, 4-ethyl-) $\text{C}_6\text{H}_{11}\text{N}=\text{C}(\text{CH}_3)_2$ (Cyclohexanamine, N-(1-methylethylidene)-) $(\text{CH}_2)_5\text{NCH}=\text{C}(\text{CH}_3)_2$ (Piperidine, 1-(2-methyl-1-propenyl)-)	XXXXX-XX-X ** 491-25-8 ** 45732-65-8 ** 6407-36-9 ** 673-33-6 **		8.23 7.84 (V) 8.05 ± 0.015 (V) 8.23 7.98 ± 0.03 (V)	PE PE PE PE PE	5589 5091 4286 4043 4452
$\text{C}_9\text{H}_{19}\text{N}^+$	$((\text{CH}_2)_5\text{N})\text{CH}_2\text{CH}(\text{CH}_3)_2$ (Piperidine, 1-(2-methylpropyl)-) $\text{C}_5\text{H}_7\text{N}(\text{CH}_3)_4$ (Piperidine, 2,2,6,6-tetramethyl-)	10315-89-6 ** 768-66-1 **		8.16 ± 0.03 (V) 7.39	PE PE	4452 4278
$\text{C}_{9,21}\text{H}_{21}\text{N}^+$	$(n-\text{C}_3\text{H}_7)_3\text{N}$ <i>tert</i> - $\text{C}_5\text{H}_{11}(tert-\text{C}_4\text{H}_9)\text{NH}$	102-69-2 ** 58471-09-3 **		7.03 ± 0.09 7.03 ± 0.1 8.04 ± 0.3 (V) 7.81 ± 0.1	PE PE PE PE	4497 4480 4818 4480
$\text{C}_{10}\text{H}_7\text{N}^+$	$\text{C}_{10}\text{H}_7\text{N}$ (Pyrrolo[2,1,5- <i>cd</i>]indolizine)	209-81-4 **		7.63 (V)	PE	4812
$\text{C}_{10}\text{H}_9\text{N}^+$	$\text{C}_{10}\text{H}_7(\text{NH}_2)$ (1-Naphthalenamine) $\text{C}_9\text{H}_8\text{NCH}_3$ (Isoquinoline, 3-methyl-) $\text{C}_{10}\text{H}_7\text{NH}_2$ (2-Naphthalenamine) $\text{C}_{10}\text{H}_9\text{N}$ (Naphthalen-1,4-imine, 1,4-dihydro-)	134-32-7 ** 1125-80-0 ** 91-59-8 ** 5176-20-5 **		7.3 7.46 (V) 8.11 7.74 ± 0.02 7.10 ± 0.02 7.2 7.56 (V) 8.25 ± 0.05 (V)	PI PE PE PE PE PI PE PE	3586 4466 4515 4143 4143 3586 4466 4830
$\text{C}_{10}\text{H}_{11}\text{N}^+$	$\text{C}_6\text{H}_5\text{CH}=\text{NCH}=\text{CHCH}_3$ (z) $\text{C}_{10}\text{H}_{11}\text{N}$ (Naphthalen-1,4-imine, 1,2,3,4-tetrahydro-) $\text{C}_6\text{H}_5\text{CH}=\text{NCH}_2\text{CH}=\text{CH}_2$ (2-Propen-1-amine, N-(phenylmethylene)-(E)-)	53146-18-2 ** 5176-30-7 ** 68003-55-4 **		8.33 (V) 8.44 ± 0.05 (V) 8.87 (V)	PE PE PE	4968 4830 4968
$\text{C}_{10}\text{H}_{13}\text{N}^+$	$\text{C}_9\text{H}_{10}\text{N}(\text{CH}_3)_3$ (Isoquinoline, 1,2,3,4-tetrahydro-2-methyl-)	1612-65-3 **		8.60 ± 0.05 (V)	PE	4830
$\text{C}_{10}\text{H}_{14}\text{N}^+$	$\text{CH}_3\text{C}_6\text{H}_4\text{CH}_2\text{N}(\text{CH}_3)_2$ (Benzinemethanamine, N,N, <i>ar</i> -trimethyl-)	56927-89-0 H		9.5	PI	5543
$\text{C}_{10}\text{H}_{15}\text{N}^+$	$\text{C}_9\text{H}_{13}\text{N}=\text{CH}_2$ (1-Azatricyclo[3.3.1.1 ^{3,7}]decane, 4-methylene-) $\text{C}_6\text{H}_4(\text{NH}_2)\text{C}_4\text{H}_9$ (Benzynamine, 3-butyl-)	42949-22-4 ** 5369-17-5 **		7.78 ± 0.02 (V) 7.51 ± 0.1	PE EI	4217 3629

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₀H₁₅N⁺						
	C ₆ H ₅ (NH ₂)C ₄ H ₉ (Benzenamine, 4-butyl-)	104-13-2	**	7.61±0.1	EI	3629
	C ₆ H ₅ N(C ₂ H ₅) ₂ (Benzenamine, N,N-diethyl-)	91-66-7	**	6.95±0.02	PE	3890
	C ₆ H ₂ (CH ₃) ₃ NHCH ₃ (Benzenamine, N,2,4,6-tetramethyl-)	13021-14-2	**	7.22	PE	3988
	C ₆ H ₃ (CH ₃) ₂ N(CH ₃) ₂ (Benzenamine, N,N,2,6-tetramethyl-)	769-06-2	**	7.30±0.02	PE	3890
			**	7.42	PE	3988
	C ₆ H ₅ CH ₂ CH ₂ N(CH ₃) ₂ (Benzeneethanamine, N,N-dimethyl-)	1126-71-2	**	7.70±0.05	PE	4192
			**	8.35±0.14 (V)	PE	4672
	C ₆ H ₅ CH ₂ CH ₂ N(CH ₃) ₂ (Benzeneethanamine,dimethyl-)	29088-49-1	**	7.70	PI	5543
	CH ₃ C ₆ H ₄ CH ₂ N(CH ₃) ₂ (Benzinemethanamine,N,N, <i>ar</i> -trimethyl-)	56927-89-0	**	7.61	PI	5543
	C ₆ H ₅ CH ₂ CH(CH ₃)NHCH ₃ (Benzeneethanamine, N, α -dimethyl-)	7632-10-2	**	8.60±0.20 (V)	PE	4672
C₁₀H₁₇N⁺						
	((CH ₂) ₄ N)(C ₆ H ₉) (Pyrrolidine, 1-(1-cyclohexen-1-yl)-)	1125-99-1	**	7.10±0.03 (V)	PE	4452
			**	7.14±0.05	PE	4654
	C ₅ H ₁₀ NC ₅ H ₇ (Piperidine, 1-(1-cyclopenten-1-yl)-)	1614-92-2	**	7.4±0.05 (V)	PE	4654
C₁₀H₁₉N⁺						
	((CH ₂) ₄ N)(C ₆ H ₁₁) (Pyrrolidine, 1-cyclohexyl-)	7731-02-4	**	7.96±0.03 (V)	PE	4452
	C ₇ H ₁₂ N(iso-C ₄ H ₇) (1-Azabicyclo[2.2.2]octane, 4-(1-methylethyl)-)	45842-68-0	**	7.99±0.015 (V)	PE	4286
	C ₉ H ₁₄ NC ₂ H ₅ (9-Azabicyclo[3.3.1]nonane,9-ethyl-)	64776-29-0	**	7.76 (V)	PE	5091
	C ₁₀ H ₁₉ N (1-Azabicyclo[3.3.3]undecane)	31023-92-4	**	6.94±0.09	PE	4497
C₁₀H₂₃N⁺						
	n-C ₁₀ H ₂₁ NH ₂	2016-57-1	**	8.63±0.05	PI	5508
C₁₁H₇N⁺						
	C ₁₀ H ₇ CH (1-Naphthalenecarbonitrile)	86-53-3	**	8.61 (V)	PE	4466
	C ₁₀ H ₇ CN (2-Naphthalenecarbonitrile)	613-46-7	**	8.64 (V)	PE	4466
C₁₁H₁₁N⁺						
	C ₁₁ H ₉ (NH ₂) (1,4-Methanonaphthalene-5-amine, 1,4-dihydro-)	61346-80-3	**	7.84±0.05 (V)	PE	5019
	C ₁₁ H ₉ (NH ₂) (1,4-Methanonaphthalene-6-amine, 1,4-dihydro-)	35391-95-8	**	7.60±0.05	PE	5019
	C ₁₀ H ₈ N(CH ₃) (Naphthalen-1,4-imine,1,4-dihydro-9-methyl-)	55258-00-9	**	8.18±0.05 (V)	PE	4830
C₁₁H₁₃N⁺						
	C ₁₀ H ₁₀ N(CH ₃) (Naphthalen-1,4-imine, 1,2,3,4-tetrahydro-9-methyl-)	55257-99-3	**	8.33±0.05 (V)	PE	4830
	C ₆ H ₄ (CN)C ₄ H ₉ (Benzonitrile, 3-butyl-)	20651-74-5	**	9.77±0.1	EI	3629
	C ₆ H ₄ (CN)C ₄ H ₉ (Benzonitrile, 4-butyl-)	20651-73-4	**	10.08±0.1	EI	3629
	C ₁₁ H ₁₃ N (2H-1,4-Ethanoquinoline, 3,4-dihydro-)	4363-25-1	**	7.85±0.02	PE	3890

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₁H₁₃N⁺	C ₆ H ₅ CH=NCH=C(CH ₃) ₂ (2-Propen-1-amine, 2-methyl-N-(phenylmethylene)-(E)-)	68003-68-9	**	8.05 (V)	PE	4968
C₁₁H₁₇N⁺	C ₇ H ₈ N(tert-C ₄ H ₉) (2-Azabicyclo[3.2.1]octa-3,6-diene,2-(1,1-dimethylethyl)-) C ₆ H ₂ (CH ₃) ₃ N(CH ₃) ₂ (Benzenamine, N,N,2,4,6-pentamethyl-)	71017-51-1 13021-15-3	**	7.06 (V) 7.24	PE	5481 3988
C₁₁H₁₉N⁺	C ₅ H ₁₀ NC ₆ H ₅ (Piperidine, 1-(1-cyclohexen-1-yl)-)	2981-10-4	**	7.44±0.03 (V)	PE	4452
C₁₁H₂₁N⁺	((CH ₂) ₂ N)(C ₆ H ₁₁) (Piperidine, 1-cyclohexyl-) C ₇ H ₁₂ N(tert-C ₄ H ₉) (1-Azabicyclo[2.2.2]octane, 4-(1,1-dimethylethyl)-) C ₇ H ₁₂ N(tert-C ₄ H ₉) (2-Azabicyclo[3.2.1]octane,2-(1,1-dimethylethyl)-) C ₈ H ₁₄ NCH(CH ₃) ₂ (9-Azabicyclo[3.3.1]nonane,9-(1-methylethyl)-) C ₈ H ₁₄ NCH ₂ CH ₂ CH ₃ (9-Azabicyclo[3.3.1]nonane,9-propyl-)	3319-01-5 45980-26-5 71017-52-2 64776-33-6 73320-99-7	**	7.93±0.03 (V) 7.97±0.015 (V) 8.30 (V) 7.68 (V) 7.71 (V)	PE	4452 4286 5481 5091 5091
C₁₂H₉N⁺	C ₁₁ H ₉ (CN) (1,4-Methanonaphthalene-5-carbonitrile, 1,4-dihydro-) C ₁₁ H ₉ (CN) (1,4-Methanonaphthalene-6-carbonitrile, 1,4-dihydro-) (C ₆ H ₅) ₂ NH (9H-Carbazole) C ₁₁ H ₉ CN (1,4-Methanonaphthalene-2-carbonitrile, 1,4-dihydro-) C ₁₁ H ₉ CN (1,4-Methanonaphthalene-6-carbonitrile, 1,4-dihydro-) C ₁₂ H ₉ N (Pyrido[2,1,6-de]quinolizine)	61346-79-0 16513-60-3 86-74-8 71906-57-5 16513-60-3 519-61-9	** ** ** ** ** ** ** ** ** ** ** **	8.94±0.05 8.94±0.05 (V) 8.87±0.05 (V) 7.57±0.03 7.50 (V) 7.68 (V) 8.77±0.05 (V) 8.77 (V) 8.85±0.05 (V) 5.87 (V)	PE PE PE PI PE PE PE PE PE	5019 5235 5019 5552 5619 4159 5235 4835 5235 4812
C₁₂H₁₁N⁺	(C ₆ H ₅) ₂ NH (Benzenamine, N-phenyl-) C ₆ H ₅ C ₆ H ₄ NH ₂ ([1,1'-Biphenyl]-2-amine)	122-39-4 90-41-5	** ** ** **	7.14±0.03 7.18±0.01 7.44 (V) 7.28±0.02	PI PE PE PE	4028 4154 4159 3702
C₁₂H₁₃N⁺	C ₁₀ H ₇ N(CH ₃) ₂ (2-Naphthalenamine, N,N-dimethyl-) C ₁₀ H ₇ N(CH ₃) ₂ (1-Naphthalenamine, N,N-dimethyl-)	2436-85-3 86-56-6	** ** **	7.12 (V) 7.59 (V) 7.00±0.02	PE PE PE	4466 4466 4143
C₁₂H₁₅N⁺	C ₁₂ H ₁₅ N (1H,5H-Benz[e]quinolizine, 2,3,6,7-tetrahydro-)	479-59-4	**	6.65±0.02	PE	3890

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₂H₂₃N⁺	C ₈ H ₁₄ NC(CH ₃) ₃ (9-Azabicyclo[3.3.1]nonane,9-(1,1-dimethylethyl)-)	64776-36-9	**	7.30 (V)	PE	5091
C₁₂H₂₇N⁺	(n-C ₄ H ₉) ₃ N	102-82-9	**	6.98±0.1	PE	4480
C₁₃H₉N⁺	C ₁₃ H ₉ N (Acridine)	260-94-6	**	7.8	PI	3586
			**	7.85 (V)	PE	5436
			**	7.88±0.02 (V)	PE	4430
			**	8.13±0.02 (V)	PE	4551
	C ₁₃ H ₉ N (Benzof[<i>f</i>]quinoline)	85-02-9	**	8.14±0.02 (V)	PE	4430
	C ₁₃ H ₉ N (Benzof[<i>h</i>]quinoline)	230-27-3	**	8.04±0.02 (V)	PE	4430
	C ₁₃ H ₉ N (Phenanthridine)	229-87-8	**	8.31±0.02 (V)	PE	4430
C₁₃H₁₀N⁺	C ₆ H ₅ C(=CH ₂)C ₅ H ₄ N (Pyridine,2-(1-phenylethethyl)-)	XXXXXX-XX-X H		9.5	EI	5570
	C ₆ H ₅ C(=CH ₂)C ₅ H ₄ N (Pyridine,3-(1-phenylethethyl)-)	XXXXXX-XX-X H		9.9	EI	5570
	C ₆ H ₅ C(=CH ₂)C ₅ H ₄ N (Pyridine,4-(1-phenylethethyl)-)	54813-56-8 H		10.0	EI	5570
	C ₆ H ₄ (CH ₃)C(=CH ₂)C ₅ H ₄ N (Pyridine,2-[1-(3-methylphenyl)ethenyl]-)	XXXXXX-XX-X CH ₃		9.7	EI	5570
	C ₆ H ₄ (CH ₃)C(=CH ₂)C ₅ H ₄ N (Pyridine,2-[1-(4-methylphenyl)ethenyl]-)	XXXXXX-XX-X CH ₃		9.8	EI	5570
	C ₆ H ₄ FC(=CH ₂)C ₅ H ₄ N (Pyridine,2-[1-(2-fluorophenyl)ethenyl]-)	XXXXXX-XX-X F		9.5	EI	5570
	C ₆ H ₄ ClC(=CH ₂)C ₅ H ₄ N (Pyridine,2-[1-(2-chlorophenyl)ethenyl]-)	XXXXXX-XX-X Cl		9.2	EI	5570
	C ₆ H ₄ ClC(=CH ₂)C ₅ H ₄ N (Pyridine,2-[1-(4-chlorophenyl)ethenyl]-)	XXXXXX-XX-X Cl		9.9	EI	5570
	C ₆ H ₄ BrC(=CH ₂)C ₅ H ₄ N (Pyridine,2-[1-(2-bromophenyl)ethenyl]-)	XXXXXX-XX-X Br		9.0	EI	5570
	C ₆ H ₄ BrC(=CH ₂)C ₅ H ₄ N (Pyridine,2-[1-(4-bromophenyl)ethenyl]-)	XXXXXX-XX-X Br		9.7	EI	5570
	C ₆ H ₄ IC(=CH ₂)C ₅ H ₄ N (Pyridine,2-[1-(2-iodophenyl)ethenyl]-)	XXXXXX-XX-X I		8.8	EI	5570
C₁₃H₁₁N⁺	C ₁₃ H ₁₁ N (Acridine, 9,10-dihydro-)	92-81-9	**	7.33 (V)	PE	4159
	C ₆ H ₅ CH=NC ₆ H ₅ (Benzenamine, N-(phenylmethylene)-)	538-51-2	**	8.25 (V)	PE	4475
			**	8.27±0.05 (V)	PE	4333
	C ₁₂ H ₉ NCH ₃ (2H-Benzof[<i>j</i>]isoindole, 2-methyl-)	59788-14-6	**	6.56 (V)	PE	4935
	C ₆ H ₅ C(=CH ₂)C ₅ H ₄ N (Pyridine,2-(1-phenylethethyl)-)	XXXXXX-XX-X **		8.65	EI	5570
	C ₆ H ₅ C(=CH ₂)C ₅ H ₄ N (Pyridine,3-(1-phenylethethyl)-)	XXXXXX-XX-X **		8.73	EI	5570
	C ₆ H ₅ C(=CH ₂)C ₅ H ₄ N (Pyridine,4-(1-phenylethethyl)-)	54813-56-8 **		8.90	EI	5570
	C ₆ H ₅ CH=CHC ₆ H ₅ N (Pyridine, <i>trans</i> -2-(2-phenylethethyl)-)	538-49-8 **		7.99±0.05 (V)	PE	4377
	C ₆ H ₅ CH=CHC ₆ H ₅ N (Pyridine, <i>trans</i> -3-(2-phenylethethyl)-)	5097-91-6 **		8.10±0.05 (V)	PE	4377

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₃H₁₁N⁺	C ₆ H ₅ CH=CHC ₅ H ₄ N (Pyridine, <i>trans</i> -4-(2-phenylethenyl)-)	5097-93-8	**	8.34±0.05 (V)	PE	4377
C₁₃H₁₂N⁺	(C ₆ H ₅ NH ₂) ₂ CH ₂ (Benzenamine, 4,4'-methylenabis-)	101-77-9	NH ₂	10.7±0.1	EI	3807
C₁₃H₁₃N⁺	(C ₆ H ₅) ₂ NCH ₃ (Benzenamine,N-methyl-N-phenyl-)	552-82-9	**	6.94±0.03	PI	5552
	C ₆ H ₅ CH ₂ C ₆ H ₄ NH ₂ (Benzenamine, 4-(phenylmethyl)-)	1135-12-2	**	7.33 (V) 7.67±0.05	PE EI	4159 3806
C₁₃H₁₇N⁺	C ₇ H ₁₂ NC ₆ H ₅ (1-Azabicyclo[2.2.2]octane, 4-phenyl-)	51069-11-5	**	8.13±0.015 (V)	PE	4286
C₁₄H₉N⁺	C ₁₄ H ₉ N (Cyclopent[4,5]azepino[2,1,7- <i>cd</i>]pyrrolizine)	27884-38-4	**	7.06 (V)	PE	4812
C₁₄H₁₁N⁺	C ₁₃ H ₉ NCH ₃ (Acridine,9-methyl-)	611-64-3	**	7.68 (V)	PE	5436
	C ₆ H ₅ CH ₂ C ₆ H ₄ CN (Benzonitrile, 4-(phenylmethyl)-)	23450-31-9	**	9.25±0.05	EI	3806
	C ₁₄ H ₁₁ N (5H-Dibenzo [<i>b,f</i>]azepine)	256-96-2	**	6.78	PE	4611
C₁₄H₁₃N⁺	C ₆ H ₅ N=CHC ₆ H ₅ CH ₃ (Benzenamine,N-[3-methylphenyl)methylene-])	6906-25-8	**	8.07 (V)	PE	5486
	C ₁₄ H ₁₃ N (5H-Dibenz[b,f]azepine, 10,11-dihydro-)	494-19-9	**	7.25 (V)	PE	4159
	C ₆ H ₄ (CH ₃)C(=CH ₂)C ₅ H ₄ N (Pyridine,2-[1-(3-methylphenyl)ethenyl]-)	XXXXX-XX-X	**	8.48	EI	5570
	C ₆ H ₄ (CH ₃)C(=CH ₂)C ₅ H ₄ N (Pyridine,2-[1-(4-methylphenyl)ethenyl]-)	XXXXX-XX-X	**	8.45	EI	5570
	C ₆ H ₄ (CH ₃)CH=CHC ₅ H ₄ N (Pyridine, <i>trans</i> -3-[2-(4-methylphenyl)ethenyl]-)	6892-33-7	**	7.90±0.05 (V)	PE	4377
	C ₆ H ₅ CH=C(CH ₃)C ₅ H ₄ N (Pyridine, <i>trans</i> -4-(1-methyl-2-phenylethenyl)-)	18150-12-4	**	8.39±0.05 (V)	PE	4377
	C ₆ H ₄ (CH ₃)N=CHC ₆ H ₅ (Benzenamine,2-methyl-N-(phenylmethylene)-)	5877-55-4	**	8.06 (V)	PE	5486
C₁₄H₁₅N⁺	C ₆ H ₄ (CH ₂ CH ₂) ₂ C ₆ H ₂ NH (15-Azatricyclo[8.2.2.1 ^{4,7}]pentadeca-4,6,10,12,13-pentaene)	51053-69-1	**	7.26	PE	5575
	C ₆ H ₅ CH ₂ CH ₂ C ₆ H ₄ NH ₂ (Benzenamine, 4-(2-phenylethyl)-)	13024-49-2	**	7.55±0.05	EI	3806
C₁₅H₉N⁺	C ₁₄ H ₉ CN (9-Anthracenecarbonitrile)	1210-12-4	**	7.80±0.03 (V)	PE	4887
C₁₅H₁₁N⁺	C ₁₄ H ₉ N(CH ₃) (Cyclopenta[<i>i,j</i>]pyrido[2,1,6- <i>de</i>]quinolizine, 3-methyl-)	21533-76-6	**	6.37 (V)	PE	4812
	C ₁₅ H ₁₁ N (16-Azatricyclo[9.2.2.1 ^{4,8}]hexadeca-2,4,6,8(16),9,11,13,14-octae)	51647-34-8	**	8.03 (V)	PE	4824

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{15}\text{H}_{11}\text{N}^+$	$\text{C}_6\text{H}_5\text{NC}_6\text{H}_5$ (Quinoline, 2-phenyl-)	612-96-4	**	8.10	PE	4066
$\text{C}_{15}\text{H}_{15}\text{N}^+$	$\text{C}_{15}\text{H}_{15}\text{N}$ (16-Azatricyclo[9.2.2.1 ^{4,8}]hexadeca-4,6,8(16),11,13,14-hexaene) $\text{C}_6\text{H}_4(\text{CH}_2\text{CH}_2)_2\text{C}_5\text{H}_3\text{N}$ (5-Azatricyclo[8.2.2.2 ^{4,7}]hexadeca-4,6,10,12,13,15-hexaene) $\text{C}_6\text{H}_4(\text{CH}_3)\text{N}=\text{CHC}_6\text{H}_4\text{CH}_3$ (Benzenamine, 2-methyl-N-[(3-methylphenyl)methylene]-) $\text{C}_6\text{H}_4(\text{CH}_3)_2\text{N}=\text{CHC}_6\text{H}_5$ (Benzenamine, 2,6-dimethyl-N-(phenylmethylene)-)	42082-72-4 37877-95-5 33629-97-9 3096-95-5	** ** ** **	8.05 (V) 8.20 (V) 8.00 (V) 8.00 (V)	PE PE PE PE	4824 5575 5486 5486
$\text{C}_{16}\text{H}_{13}\text{N}^+$	$\text{C}_{15}\text{H}_{10}\text{N}(\text{CH}_3)$ (16-Azatricyclo[9.2.2.1 ^{4,8}]hexadeca-2,4,6,8(16),9,11,13,14-octaene, 6-methyl-) $\text{C}_{14}\text{H}_7\text{N}(\text{CH}_3)_2$ (Cyclopent[4,5]azepino[2,1,7- <i>cd</i>]pyrrolizine, 6,8-dimethyl-) $\text{C}_3\text{H}_3(\text{CN})(\text{C}_6\text{H}_5)_2$ (Cyclopropanecarbonitrile, 1,2-diphenyl-)	70389-17-2 65738-45-6 10224-14-3	** ** **	7.92 (V) 6.99 (V) 8.80 ± 0.08	PE PE EI	4824 4812 3575
$\text{C}_{16}\text{H}_{15}\text{N}^+$	$\text{C}_{14}\text{H}_9\text{N}(\text{CH}_3)_2$ (3H-Indole, 3,3-dimethyl-2-phenyl-)	6636-32-4	**	8.10 (V)	PE	4421
$\text{C}_{16}\text{H}_{17}\text{N}^+$	$\text{C}_{15}\text{H}_{14}\text{N}(\text{CH}_3)$ (16-Azatricyclo[9.2.2.1 ^{4,8}]hexadeca-4,6,8(16),11,13,14-hexaene, 6-methyl-) $\text{C}_6\text{H}_3(\text{CH}_3)_2\text{N}=\text{CHC}_6\text{H}_4\text{CH}_3$ (Benzamine, 2,6-dimethyl-N-[(3-methylphenyl)methylene]-) $\text{C}_{16}\text{H}_{15}\text{NH}_2$ (Tricyclo[8.2.2.2 ^{4,7}]hexadeca-4,6,10,12,13,15-hexaen-5-amine)	70389-16-1 57387-52-7 10122-95-9	** ** **	8.06 (V) 7.90 (V) 6.90	PE PE PE	4824 5486 4158
$\text{C}_{17}\text{H}_{13}\text{N}^+$	$\text{C}_{16}\text{H}_{10}\text{NCH}_3$ (2H-Dibenz[e,g]isoindole, 2-methyl-)	59788-15-7	**	7.15 (V)	PE	4935
$\text{C}_{17}\text{H}_{15}\text{N}^+$	$\text{C}_{15}\text{H}_{10}\text{N}(\text{CH}_3)_2$ (16-Azatricyclo[9.2.2.1 ^{4,8}]hexadeca-2,4,6,8(16),9,11,13,14-octaene, 12,14-dimethyl-)	64000-97-1	**	7.67 (V)	PE	4824
$\text{C}_{17}\text{H}_{19}\text{N}^+$	$\text{C}_{15}\text{H}_{13}\text{N}(\text{CH}_3)_2$ (16-Azatricyclo[9.2.2.1 ^{4,8}]hexadeca-4,6,8(16),11,13,14-hexaene, 12,14-dimethyl-)	70389-13-8	**	7.70 (V)	PE	4824
$\text{C}_{17}\text{H}_{29}\text{N}^+$	$\text{C}_5\text{H}_2\text{N}(\text{C}(\text{CH}_3)_3)_3$ (Pyridine, 2,4,6-tris(1,1-dimethylethyl)-)	20336-15-6	** **	8.6 (V) 8.6 (V)	PE PE	3685 3934
$\text{C}_{18}\text{H}_{15}\text{N}^+$	$(\text{C}_6\text{H}_5)_3\text{N}$ (Benzamine, N,N-diphenyl-)	603-34-9	** ** **	7.00 ± 0.05 (V) 6.80 ± 0.05 6.75 ± 0.01	PE PI PE	4368 4028 4154

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₈H₁₇N⁺	C ₁₀ H ₆ (CH ₂ CH ₂) ₂ C ₄ H ₂ NH (5,14-Ethenobenzocyclododecen-8,11-imine,6,7,12,13-tetrahydro-)	73650-66-5	**	7.35 (V)	PE	5575
C₁₈H₂₇N⁺	C ₉ H ₁₄ NC ₆ H ₄ C(CH ₃) ₃ (9-Azabicyclo[3.3.1]nonane,9-[4-(1,1-dimethylethyl)phenyl])	XXXXX-XX-X	**	6.94 (V)	PE	5091
C₁₉H₁₃N⁺	C ₁₃ H ₈ NC ₆ H ₅ (Acridine,9-phenyl-)	602-56-2	**	7.75 (V)	PE	5436
	C ₁₃ H ₈ NC ₆ H ₅ (Phenanthridine, 6-phenyl-)	2720-93-6	**	7.80 (V) 8.20 (V)	PE	5630 4262
C₁₉H₁₉N⁺	C ₁₅ H ₇ N(CH ₃) ₄ (16-Azatricyclo[9.2.2.1 ^{4,8}]hexadeca-2,4,6,8(16),9,11,13,14-octae-	64000-98-2	**	7.54 (V)	PE	4824
				12,13,14,15-tetramethyl-)		
C₁₉H₂₃N⁺	C ₁₅ H ₁₁ N(CH ₃) ₄ (16-Azatricyclo[9.2.2.1 ^{4,8}]hexadeca-4,6,8(16),11,13,14-hexa-	70389-15-0	**	7.57 (V)	PE	4824
				12,13,14,15-tetramethyl-)		
C₂₀H₂₃N⁺	C ₁₅ H ₁₂ =CHCH ₂ CH ₂ N(CH ₃) ₂ (1-Propanamine, 3-(10,11-dihydro-5H-dibenzo[a,d]cyclohepten-5-ylidene)-N,N-dimethyl-)	50-48-6	**	8.26±0.07	CTS	4079
CH₂N₂⁺	CH ₂ N ₂ H ₂ NC≡N CH ₂ N ₂ (3H-Diazirine)	334-88-3 420-04-2 157-22-2	** ** **	9.00 10.65 (V) 10.3	PE PE PE	4595 4294 3727
CH₃N₂⁺	CH ₃ N=NCH ₃ <i>trans</i> -CH ₃ N=NCH ₃	503-28-6 4143-41-3	CH ₃ CH ₃	9.2 9.20±0.03	EI PI	3632 4342
CH₄N₂⁺	CH ₃ N=NH	XXXXX-XX-X	**	8.8±0.1	PE	4587
CH₆N₂⁺	H ₂ NNH(CH ₃)	60-34-4	** ** ** **	9.34 (V) 8.40±0.05 9.32 (V) 9.36 (V)	PE PE PE PE	5381 4521 4137 4514
C₂H₄N₂⁺	CH ₂ =NN=CH ₂	503-27-5	**	8.95	PE	4499
C₂H₆N₂⁺	(CH ₃ N) ₂ <i>trans</i> -CH ₃ N=NCH ₃	503-28-6 4143-41-3	** ** ** **	8.30 8.95±0.05 (V) 9.0 (V) 8.45±0.05 8.20	PE PE PE PI PE	4587 4614 4467 4342 3649
C₂H₈N₂⁺	(CH ₃) ₂ NNH ₂	57-14-7	** ** **	8.05±0.05 8.82 (V) 8.85 (V)	PE PE PE	4521 5381 4514

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₂H₈N₂⁺						
	(CH ₃) ₂ NNH ₂	57-14-7	**	8.88 (V)	PE	4137
	(CH ₃ NH) ₂	540-73-8	**	9.00 (V)	PE	4137
			**	9.02 (V)	PE	5068
			**	9.02 (V)	PE	5381
			**	9.62	PE	3747
	C ₂ H ₅ NHHN ₂	624-80-6	**	8.12±0.05	PE	4521
			**	9.20 (V)	PE	4137
C₃H₂N₂⁺						
	CH ₂ (CN) ₂	109-77-3	**	12.88	PE	4067
C₃H₃N₂⁺						
	C ₃ H ₄ N ₂ (1 <i>H</i> -Imidazole)	288-32-4	H	12.8	EI	3910
C₃H₄N₂⁺						
	C ₃ H ₄ N ₂ (1 <i>H</i> -Imidazole)	288-32-4	**	8.96 (V)	PE	5092
			**	8.78 (V)	PE	4009
			**	9.12	EI	3910
	C ₃ H ₄ N ₂ (1 <i>H</i> -Pyrazole)	288-13-1	**	9.15 (V)	PE	5213
			**	9.15 (V)	PE	4009
C₃H₆N₂⁺						
	(CH ₃) ₂ NC≡N	1467-79-4		9.44 (V)	PE	4294
	(CH ₃) ₂ C=N=N	2684-60-8	**	7.88	PE	4047
	C ₃ H ₆ N ₂ (3 <i>H</i> -Diazirine, 3,3-dimethyl-)	5161-49-9	**	9.76 (V)	PE	3505
C₃H₈N₂⁺						
	(CH ₃) ₂ NN=CH ₂	2035-89-4	**	7.85	PE	3884
	CH ₃ NHN=CHCH ₃	17167-73-6	**	7.67	PE	3884
	C ₃ H ₆ NNH ₂ (1-Azetidinamine)	53779-89-8	**	8.828 (V)	PE	4156
	CH ₂ N ₂ (CH ₃) ₂ (Diaziridine, 1,2-dimethyl-)	6794-95-2	**	9.42 (V)	PE	3888
			**	9.42 (V)	PE	4277
	CH ₂ N ₂ (CH ₃) ₂ (Diaziridine, 3,3-dimethyl-)	4901-76-2	**	9.90 (V)	PE	3888
	C ₃ H ₈ N ₂ (Pyrazolidine)	504-70-1	**	7.90 (V)	PE	4085
			**	9.16 (V)	PE	4134
C₃H₁₀N₂⁺						
	(CH ₃) ₂ NNH(CH ₃)	1741-01-1	**	8.74 (V)	PE	5381
			**	8.67 (V)	PE	4137
	n-C ₃ H ₇ NHHN ₂	5039-61-2	**	9.07 (V)	PE	4137
	iso-C ₃ H ₇ NHHN ₂	2257-52-5	**	8.42±0.05	PE	4521
			**	9.05 (V)	PE	4137
C₄H₂N₂⁺						
	cis-CH(CN)=CH(CN)	928-53-0	**	11.15	PE	3778
	trans-CH(CN)=CH(CN)	764-42-1	**	11.15	PE	3778
			**	11.16±0.03	PI	5505
	C(CN) ₂ =CH ₂	922-64-5	**	11.38±0.05 (V)	PE	4859
C₄H₄N₂⁺						
	C ₄ H ₄ N ₂ (Pyrazine)	290-37-9	**	9.28±0.01	S	3773

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₅H₁₄N₂⁺						
	((CH ₃) ₂ N) ₂ CH ₂	XXXXX-XX-X **		7.74±0.05	PE	4192
	(C ₂ H ₅)(CH ₃)NN(CH ₃) ₂	50599-41-2 **		8.18	PE	5280
		**		8.18 (V)	PE	4137
	<i>n</i> -C ₄ H ₉ N(CH ₃)NH ₂	20240-62-4 **		7.82±0.05	PE	4521
	<i>iso</i> -C ₃ H ₇ NNH(CH ₃) ₂	5824-85-1 **		8.52 (V)	PE	4137
C₆H₄N₂⁺						
	C ₅ H ₄ NCN (2-Pyridinecarbonitrile)	100-70-9	**	10.12 (V)	PE	4240
			**	10.33±0.05	EI	3498
			**	10.33	EI	5292
			**	10.5±0.1	EI	4302
	C ₅ H ₄ NCN (3-Pyridinecarbonitrile)	100-54-9	**	10.10 (V)	PE	4240
			**	10.37 (V)	PE	5527
			**	10.4±0.1	EI	4302
	C ₅ H ₄ NCN (4-Pyridinecarbonitrile)	100-48-1	**	10.30 (V)	PE	4240
			**	10.7 (V)	PE	5527
			**	10.4±0.1	EI	4302
C₆H₆N₂⁺						
	C ₆ H ₄ (NH) ₂ (2,5-Cyclohexadiene,1,4-diimine)	4377-73-5	**	9.36±0.03	PI	5552
	C ₆ H ₆ N ₂ (7,8-Diazatetracyclo[3.3.0.0 ^{2,4} .0 ^{3,6}]oct-7-ene)	34122-54-8	**	8.54 (V)	PE	4135
C₆H₇N₂⁺						
	C ₆ H ₄ (NH ₂)NHCOCH ₃ (Acetamide, <i>N</i> -(2-aminophenyl)-)	34801-09-7	CH ₃ CO	13.93±0.02	EI	3631
	C ₆ H ₄ (NH ₂)NHCOCH ₃ (Acetamide, <i>N</i> -(4-aminophenyl)-)	122-80-5	CH ₃ CO	13.72±0.02	EI	3631
C₆H₈N₂⁺						
	C ₆ H ₄ (NH ₂) ₂ (1,2-Benzenediamine)	95-54-5	**	7.2	PE	4201
			**	7.69 (V)	PE	5474
			**	7.78 (V)	PE	4893
	C ₆ H ₄ (NH ₂) ₂ (1,3-Benzenediamine)	108-45-2	**	7.14	PI	4328
			**	7.44	PE	4201
			**	7.60 (V)	PE	5474
			**	7.74 (V)	PE	4893
	C ₆ H ₄ (NH ₂) ₂ (1,4-Benzenediamine)	106-50-3	**	6.89±0.03	PI	5552
			**	6.84	PE	4201
			**	7.34 (V)	PE	5474
			**	7.61 (V)	PE	4893
			**	7.16	EI	4089
	C ₆ H ₅ NHNH ₂ (Phenylhydrazine)	100-63-0	**	7.86 (V)	PE	5474
	C ₆ H ₂ N ₂ (CH ₃) ₂ (Pyrazine, 2,6-dimethyl-)	108-50-9	**	8.80	PE	3860
	C ₅ NH ₃ (CH ₃)NH ₂ (2-Pyridinamine, 6-methyl-)	1824-81-3	**	9.1	CTS	3730
	C ₅ H ₄ NNHCH ₃ (2-Pyridinamine, <i>N</i> -methyl-)	4597-87-9	**	8.26±0.05	EI	3891

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₆H₈N₂⁺						
	C ₅ NH ₃ (CH ₃)NH ₂ (3-Pyridinamine, 4-methyl-)	3430-27-1	**	9.3	CTS	3730
	C ₅ H ₄ NNHCH ₃ (3-Pyridinamine, N-methyl-)	18364-47-1	**	8.53±0.05	EI	3891
	C ₅ H ₄ NNHCH ₃ (4-Pyridinamine, N-methyl-)	1121-58-0	**	8.75±0.05	EI	3891
	C ₅ H ₄ N(NH)CH ₃ (2(1H)-Pyridinimine, 1-methyl-)	4088-63-5	**	7.91±0.05	EI	3891
	C ₅ H ₄ N(NH)CH ₃ (4(1H)-Pyridinimine, 1-methyl-)	16562-40-6	**	7.85±0.05	EI	3891
	C ₅ H ₄ N(NH)CH ₃ (Pyridinium, 3-amino-1-methyl-, hydroxides, inner salt)	38879-42-2	**	7.45±0.1	EI	3891
	C ₆ H ₄ (NH ₂)NHCOC ₃ (Acetamide, N-(2-aminophenyl-))	34801-09-7	CH ₂ =C=O	10.49±0.02	EI	3631
	C ₆ H ₄ (NH ₂)NHCOC ₃ (Acetamide, N-(4-aminophenyl-))	122-80-5	CH ₂ =C=O	10.06±0.02	EI	3631
C₆H₁₀N₂⁺						
	C ₆ H ₁₀ N ₂ (2,3-Diazabicyclo[2.2.2]oct-2-ene)	3310-62-1	**	7.79±0.04	PE	3828
C₆H₁₂N₂⁺						
	(CH ₃) ₂ C=NN=C(CH ₃) ₂	627-70-3	**	7.97	PE	4043
			**	8.6	PE	4814
	(C ₂ H ₅ CH=N) ₂	15601-98-6	**	9.0 (V)	PE	4814
	(C ₃ H ₆ N) ₂ (1,1-Biazetidine)	67092-91-5	**	8.2 (V)	PE	4780
	C ₆ H ₁₂ N ₂ (1,2-Diazabicyclo[2.2.2]octane)	329-94-2	**	8.52 (V)	PE	4134
	C ₆ H ₁₂ N ₂ (1,4-Diazabicyclo[2.2.2]octane)	280-57-9	**	7.20	PI	5045
			**	7.52±0.02 (V)	PE	4480
			**	7.52 (V)	PE	4038
			**	7.609	PE	4214
			**	7.61 (V)	PE	4141
			**	7.70 (V)	PE	5623
	C ₆ H ₁₂ N ₂ (1,5-Diazabicyclo[3.2.1]octane)	280-28-4	**	8.24 (V)	PE	5623
			**	8.89 (V)	PE	4141
	C ₆ H ₁₂ N ₂ (1,5-Diazabicyclo[3.3.0]octane)	XXXXX-XX-X	**	7.87 (V)	PE	5504
	C ₂ N ₂ (CH ₃) ₄ (1,2-Diazete, 3,4-dihydro-3,3,4,4-tetramethyl-)	54166-22-2	**	8.87 (V)	PE	4651
	C ₃ H ₆ N ₂ C ₃ H ₆ (1H,5H-Pyrazolo[1,2-a]pyrazole,tetrahydro-)	5397-67-1	**	7.87	PE	5280
			**	7.90 (V)	PE	5381
			**	7.87 (V)	PE	4134
			**	7.91 (V)	PE	3889
	C ₄ H ₆ N ₂ (CH ₃) ₂ (Pyridazine, 1,2,3,6-tetrahydro-1,2-dimethyl-)	26163-36-0	**	8.89 (V)	PE	4277
			**	8.12 (V)	PE	4134
C₆H₁₄N₂⁺						
	CH ₂ =C(N(CH ₃) ₂) ₂	815-62-3	**	7.5 (V)	PE	4291
	cis-(iso-C ₃ H ₇) ₂ N=N	23201-84-5	**	8.24 (V)	PE	4429
	trans-C ₃ H ₇ N=NC ₃ H ₇	55204-42-7	**	8.61 (V)	PE	4429
	trans-(iso-C ₃ H ₇) ₂ N=N	15464-00-3	**	8.47 (V)	PE	4429
	C ₄ H ₈ N ₂ (CH ₃) ₂ (Piperazine, 1,4-dimethyl-)	106-58-1	**	8.77 (V)	PE	4141

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₆H₁₄N₂⁺						
	C ₄ H ₈ N ₂ (CH ₃) ₂ (Pyridazine, hexahydro-1,2-dimethyl-)	26163-37-1	**	7.77 (V)	PE	3887
			**	7.78 (V)	PE	4277
			**	7.78 (V)	PE	5353
			**	7.81	PE	5280
			**	7.81 (V)	PE	4134
			**	8.57 (V)	PE	4277
	C ₄ H ₈ N ₂ (CH ₃) ₂ (Pyrimidine, hexahydro-1,3-dimethyl-)	10556-96-4	**	8.11 (V)	PE	4141
	C ₄ H ₈ NN(CH ₃) ₂ (1-Pyrrolidinamine,N,N-dimethyl-)	53779-90-1	**	7.97	PE	5280
C₆H₁₆N₂⁺						
	(CH ₃) ₂ NCH ₂ CH ₂ N(CH ₃) ₂	51-80-9	**	7.61±0.05	PE	4192
	(C ₃ H ₆ NH ₂) ₂	124-09-4	**	7.52 (V)	PE	5538
	(NH(C ₃ H ₇)) ₂	1615-83-4	**	8.62 (V)	PE	5381
	(C ₂ H ₅) ₂ NN(CH ₃) ₂	21849-74-1		8.10	PE	5280
			**	8.10 (V)	PE	4137
	((C ₂ H ₅)(CH ₃)N) ₂	23337-93-1	**	8.08	PE	5280
	(n-C ₃ H ₇) ₂ NNH ₂	4986-50-9	**	8.51	PE	4137
	(n-C ₃ H ₇)(CH ₃)NN(CH ₃) ₂	60678-65-1	**	8.14	PE	5280
	(NH(iso-C ₃ H ₇)) ₂	3711-34-0	**	8.45 (V)	PE	5381
			**	8.34 (V)	PE	4085
			**	8.59 (V)	PE	4137
	(iso-C ₃ H ₇)(CH ₃)NN(CH ₃) ₂	49840-63-3	**	8.09	PE	5280
			**	8.09 (V)	PE	4137
C₇H₆N₂⁺						
	C ₆ H ₅ CHN ₂ (Benzene, (diazomethyl)-)	766-91-6	**	7.72±0.02 (V)	PE	4674
	C ₇ H ₆ N ₂ (1H-Benzimidazole)	51-17-2	**	8.44 (V)	PE	5092
			**	8.45 (V)	PE	5396
	C ₇ H ₆ N ₂ (Imidazo[1,2-a]pyridine)	274-76-0	**	8.19 (V)	PE	4812
	C ₆ H ₄ CHN ₂ H (1H-Indazole)	271-44-3	**	8.35 (V)	PE	5396
C₇H₈N₂⁺						
	C ₇ H ₈ N ₂ (3,4-Diazatricyclo[4.2.1.0 ^{2,5}]nona-3,7-diene)	23979-29-5	**	9.05±0.05 (V)	PE	4040
	C ₇ H ₈ N ₂ (3,5,6-Methenocyclopentapyrazole,3,3a,4,5,6,6a-hexahydro-)	16104-45-3	**	8.23±0.05	PE	4449
			**	8.65 (V)	PE	4135
C₇H₁₀N₂⁺						
	C ₇ H ₁₀ N ₂ (3,4-Diazatricyclo[4.2.1.0 ^{2,5}]non-3-ene)	23979-30-8	**	8.90±0.05 (V)	PE	4040
	C ₅ H ₄ NN(CH ₃) ₂ (4-Pyridinamine,N,N-dimethyl-)	1122-58-3	**	7.82 (V)	PE	5527
			**	8.3±0.1	EI	4302
	C ₅ H ₄ NN(CH ₃) ₂ (2-Pyridinamine, N,N-dimethyl-)	5683-33-0	**	7.8±0.1	EI	4302
			**	7.7	CTS	3730
C₇H₁₂N₂⁺						
	C ₅ H ₆ N ₂ (CH ₃) ₂ (2,3-Diazabicyclo[2.2.1]hept-5-ene, 2,3-dimethyl-)	14288-15-4	**	7.63 (V)	PE	4277
			**	7.63 (V)	PE	5353
			**	7.72 (V)	PE	4134
			**	7.74 (V)	PE	3889

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₇H₁₂N₂⁺	C ₇ H ₁₂ N ₂ (6,7-Diazabicyclo[3.2.2]non-6-ene)	43195-77-3	**	7.64±0.04	PE	3828
	C ₃ N ₂ (CH ₃) ₄ (4H-Pyrazole,3,4,4,5-tetramethyl-)	19078-32-1	**	9.57 (V)	PE	5381
			**	10.12 (V)	PE	4085
C₇H₁₄N₂⁺	C ₅ H ₈ N ₂ (CH ₃) ₂ (2,3-Diazabicyclo[2.2.1]heptane, 2,3-dimethyl-)	14287-89-9	**	7.48 (V)	PE	4277
			**	7.48 (V)	PE	5353
			**	7.58 (V)	PE	3889
			**	7.66	PE	5280
	C ₅ H ₈ N ₂ (CH ₃) ₂ (2,3-Diazabicyclo[2.2.1]heptane, 2,3-dimethyl-, (2- <i>endo</i> ,3- <i>exo</i>)-)	53798-46-2	**	7.66 (V)	PE	4134
	C ₇ H ₁₄ N ₂ (1,5-Diazabicyclo[3.2.2]nonane)	283-47-6	**	7.43 (V)	PE	4141
	C ₇ H ₁₄ N ₂ (1,5-Diazabicyclo[3.3.1]nonane)	281-17-4	**	7.75 (V)	PE	4141
	C ₆ H ₁₁ N ₂ CH ₃ (1,2-Diazabicyclo[2.2.2]octane, 2-methyl-)	6523-29-1	**	8.02 (V)	PE	4134
	C ₃ H ₈ N ₂ (CH ₃) ₄ (3H-Pyrazole, 4,5-dihydro-3,3,5,5-tetramethyl-)	2721-31-5	**	8.63 (V)	PE	4429
	C ₄ H ₈ N ₂ C ₃ H ₆ (1H-Pyrazolo[1,2- <i>a</i>]pyridazine, hexahydro-)	5721-43-7	**	7.63	PE	5280
C₇H₁₆N₂⁺			**	7.63 (V)	PE	4134
	C ₄ H ₈ N ₂ (CH ₃) ₃ (Pyridazine, 1,2,3,6-tetrahydro-1,2,3-trimethyl-)	38704-94-8	**	8.08 (V)	PE	4134
			**	7.88	PE	5280
	C ₅ H ₁₀ N ₂ (CH ₃) ₂ (1H-1,2-Diazepin, hexahydro-1,2-dimethyl-)	49840-68-8	**	7.85 (V)	PE	5477
C₇H₁₈N₂⁺	C ₃ H ₈ N ₂ (CH ₃) ₄ (Imidazolidine,1,2,2,3-tetramethyl-)	33709-65-8	**	8.09	PE	5280
	C ₅ H ₁₀ NN(CH ₃) ₂ (1-Piperidinamine,N,N-dimethyl-)	49840-60-0	**	8.06	PE	5280
	C ₃ H ₆ N ₂ (C ₂ H ₅) ₂ (Pyrazolidine,1,2-diethyl-)	22825-58-7	**	8.06 (V)	PE	4134
			**	7.81 (V)	PE	3887
	C ₄ H ₇ N ₂ (CH ₃) ₃ (Pyridazine, hexahydro-1,2,3-trimethyl-)	38704-92-6	**	7.83	PE	5280
			**	7.83 (V)	PE	4134
			**	8.03 (V)	PE	4141
C₈H₄N₂⁺	(C ₂ H ₅) ₂ NN(C ₂ H ₅)(CH ₃) (<i>n</i> -C ₄ H ₉)(CH ₃)NN(CH ₃) ₂	50599-43-4	**	8.02	PE	5280
		52598-10-4	**	8.12	PE	5280
			**	8.12 (V)	PE	4137
	(<i>tert</i> -C ₄ H ₉)(CH ₃)NN(CH ₃) ₂	60678-73-1	**	7.89	PE	5280
C₈H₄N₂⁺	C ₆ H ₄ (CN) ₂ (1,2-Benzenedicarbonitrile)	91-15-6	**	10.10 (V)	PE	4969
			**	10.27 (V)	PE	5259
	C ₆ H ₄ (CN) ₂ (1,3-Benzenedicarbonitrile)	626-17-5	**	10.20 (V)	PE	5259
			**	10.60 (V)	PE	4969
	C ₆ H ₄ (CN) ₂ (1,4-Benzenedicarbonitrile)	623-26-7	**	10.1 (V)	PE	5259
			**	10.10 (V)	PE	4969

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_8\text{H}_6\text{N}_2^+$						
	$\text{C}_8\text{H}_6\text{N}_2$ (Cinnoline)	253-66-7	**	<8.8	PE	3638
			**	8.90 (V)	PE	3722
	$\text{C}_8\text{H}_6\text{N}_2$ (1,5-Naphthyridine)	254-79-5	**	9.20 (V)	PE	3722
	$\text{C}_8\text{H}_6\text{N}_2$ (1,6-Naphthyridine)	253-72-5	**	9.07 (V)	PE	3722
	$\text{C}_8\text{H}_6\text{N}_2$ (1,7-Naphthyridine)	253-69-0	**	8.99 (V)	PE	3722
	$\text{C}_8\text{H}_6\text{N}_2$ (1,8-Naphthyridine)	254-60-4	**	9.20 (V)	PE	3722
	$\text{C}_8\text{H}_6\text{N}_2$ (2,6-Naphthyridine)	253-50-9	**	8.87 (V)	PE	3722
	$\text{C}_8\text{H}_6\text{N}_2$ (2,7-Naphthyridine)	253-45-2	**	8.98 (V)	PE	3722
	$\text{C}_8\text{H}_6\text{N}_2$ (Phthalazine)	253-52-1	**	8.70 (V)	PE	3722
	$\text{C}_8\text{H}_6\text{N}_2$ (Quinazoline)	253-82-7	**	9.00	PE	3638
			**	9.08 (V)	PE	3722
	$\text{C}_8\text{H}_6\text{N}_2$ (Quinoxaline)	91-19-0	**	9.00 (V)	PE	3722
			**	9.01	PE	3638
$\text{C}_8\text{H}_8\text{N}_2^+$						
	$(\text{C}_4\text{H}_4\text{N})_2$ (1,1'-Bi-1H-pyrrole)	38602-81-2	**	8.30 (V)	PE	5387
	$\text{C}_8\text{H}_8\text{N}_2$ (9,10-Diazapentacyclo[4.4.0.0 ^{2,5} .0 ^{3,8} .0 ^{4,7}]dec-9-ene)	24046-80-8	**	7.68±0.05	PE	4449
$\text{C}_8\text{H}_{12}\text{N}_2^+$						
	$\text{C}_7\text{H}_{12}\text{NCN}$ (1-Azabicyclo[2.2.2]octane-4-carbonitrile)	26458-78-6	**	8.71±0.015 (V)	PE	4286
	$\text{C}_6\text{H}_4(\text{NH}_2)\text{N}(\text{CH}_3)_2$ (1,4-Benzenediamine, N,N-dimethyl-)	99-98-9	**	6.46	PI	4328
	$\text{C}_8\text{H}_{12}\text{N}_2$ (7,8-Diazatricyclo[4.2.2.0 ^{2,5}]dec-7-ene,(1 α ,2 β ,5 β ,6 α)-)	25863-08-5	**	7.68±0.05	PE	4449
	$\text{C}_4\text{N}_2(\text{CH}_3)_4$ (Pyrazine, tetramethyl-)	1124-11-4	**	8.6 (V)	PE	4161
$\text{C}_8\text{H}_{14}\text{N}_2^+$						
	$\text{C}_6\text{H}_8\text{N}_2(\text{CH}_3)_2$ (2,3-Diazabicyclo[2.2.2]oct-2-ene,1,4-dimethyl-)	49570-30-1	**	8.06 (V)	PE	4429
	$\text{C}_6\text{H}_8\text{N}_2(\text{CH}_3)_2$ (2,3-Diazabicyclo[2.2.2]oct-5-ene, 2,3-dimethyl-)	14287-91-3	**	7.49 (V)	PE	4134
			**	7.51 (V)	PE	4277
			**	7.51 (V)	PE	5353
			**	7.59 (V)	PE	3889
	$\text{C}_8\text{H}_{14}\text{N}_2$ (7,8-Diazabicyclo[4.2.2]dec-7-ene)	32634-64-3	**	7.38±0.04	PE	3828
	$\text{C}_8\text{H}_{14}\text{N}_2$ (1,3-Diazatricyclo[3.3.1.1 ^{3,7}]decane)	281-29-8	**	7.75 (V)	PE	4659
$\text{C}_8\text{H}_{16}\text{N}_2^+$						
	$\text{C}_5\text{H}_{10}\text{N}_2\text{C}_3\text{H}_6$ (2H-Azirin-3-amine, N,N-diethyl-2,2-dimethyl-)	28942-55-4	**	7.68 (V)	PE	4780
	$(\text{C}_4\text{H}_8\text{N})_2$ (1,1'-Bipyrrolidine)	18389-95-2	**	7.888 (V)	PE	4156
			**	7.91	PE	5280
			**	9.95 (V)	PE	5381
	$\text{C}_6\text{H}_{16}\text{N}_2(\text{CH}_3)_2$ (2,3-Diazabicyclo[2.2.2]octane, 2,3-dimethyl-)	14287-92-4	**	7.45 (V)	PE	4277

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₈H₁₆N₂⁺						
	C ₆ H ₁₀ N ₂ (CH ₃) ₂	14287-92-4	**	7.45 (V)	PE	5353
			**	7.46	PE	5280
	C ₆ H ₁₀ N ₂ (CH ₃) ₂ (2,3-Diazabicyclo[2.2.2]octane, 2,3-dimethyl-, <i>trans</i> -)	53779-85-4	**	7.46 (V)	PE	4134
	C ₈ H ₁₆ N ₂ (1 <i>H</i> ,5 <i>H</i> -Pyrazolo[1,2- <i>a</i>][1,2]diazepine, hexahydro-)	49840-69-9	**	7.58 (V)	PE	4134
			**	7.58	PE	5280
	C ₆ H ₄ N ₂ (CH ₃) ₄ (Pyridazine, 1,2,3,6-tetrahydro-1,2,4,5-tetramethyl-)	14003-02-2	**	7.92 (V)	PE	4134
	C ₆ H ₄ N ₂ (CH ₃) ₄ (Pyridazine, 3,4,5,6-tetrahydro-3,3,6,6-tetramethyl-)	19403-24-8	**	7.89 (V)	PE	4429
	C ₈ H ₁₆ N ₂ (Pyridazino[1,2- <i>a</i>]pyridazine, octahydro-)	3661-15-2	**	7.59 (V)	PE	3889
			**	7.60 (V)	PE	4134
			**	7.61	PE	5280
C₈H₁₈N₂⁺						
	<i>trans</i> -(<i>tert</i> -C ₄ H ₉ N) ₂	927-83-3	**	8.2±0.2 (V)	PE	4581
			**	8.20 (V)	PE	4429
	(<i>iso</i> -C ₄ H ₉ N) ₂	3896-19-3	**	8.20 (V)	PE	4429
	C ₆ H ₁₂ NN(CH ₃) ₂ (1 <i>H</i> -Azepin-1-amine, hexahydro-N,N-dimethyl-)	60678-76-4	**	8.09	PE	5280
	C ₂ H ₄ N ₂ (C ₈ H ₇) ₂ (1,2-Diazetidine, 1,2-bis(1-methylethyl)- <i>trans</i>)	67092-87-9	**	7.6 (V)	PE	4780
	C ₄ H ₈ N ₂ (C ₂ H ₅) ₂ (Pyridazine, 1,2-diethylhexahydro-)	60678-82-2	**	7.81	PE	5280
	C ₄ H ₆ (CH ₃) ₂ N ₂ (CH ₃) ₂ (Pyridazine, hexahydro-1,2,3,6-tetramethyl, <i>cis</i> -)	26171-64-2	**	7.76	PE	5280
			**	7.76 (V)	PE	4134
			**	7.82 (V)	PE	3887
	C ₄ H ₆ (CH ₃) ₂ N ₂ (CH ₃) ₂ (Pyridazine, hexahydro-1,2,3,6-tetramethyl, <i>trans</i> -)	38704-91-5	**	7.55	PE	5280
			**	7.78 (V)	PE	3887
			**	7.82 (V)	PE	4134
C₈H₂₀N₂⁺						
	(NH(C ₄ H ₉)) ₂	1744-71-4	**	8.65 (V)	PE	5381
	((C ₂ H ₅) ₂ N) ₂	4267-00-9	**	7.94	PE	5280
			**	7.94 (V)	PE	4137
			**	7.94 (V)	PE	5381
			**	8.10 (V)	PE	3889
	(<i>n</i> -C ₄ H ₉) ₂ NNH ₂	7422-80-2	**	7.75±0.05	PE	4521
	(<i>n</i> -C ₅ H ₇) ₂ NN(CH ₃) ₂	60678-72-0	**	7.98	PE	5280
	(NH(<i>iso</i> -C ₄ H ₉)) ₂	3711-37-3	**	8.70 (V)	PE	5381
	(<i>iso</i> -C ₄ H ₉) ₂ NNH ₂	16596-38-6	**	7.73±0.05	PE	4521
	(<i>iso</i> -C ₅ H ₇) ₂ NN(CH ₃) ₂	60678-66-2	**	7.65	PE	5280
	((<i>iso</i> -C ₅ H ₇)(CH ₃)N) ₂	60678-71-9	**	7.92	PE	5280
C₉H₆N₂⁺						
	C ₉ H ₆ N ₂ (Pyrazino[2,1,6- <i>cd</i>]pyrrolizine)	27884-36-2	**	7.65 (V)	PE	4812
C₉H₁₁N₂⁺						
	C ₆ H ₅ N=CHN(CH ₃) ₂ (Methanimidamide, <i>N,N</i> -dimethyl- <i>N'</i> -phenyl-)	1783-25-1	H	9.0±0.1	EI	4359
			H	9.0	EI	4337
	C ₆ H ₄ (F)N=CHN(CH ₃) ₂ (Methanimidamide, <i>N'</i> -(2-fluorophenyl)- <i>N,N</i> -dimethyl-)	53666-09-4	F	8.9	EI	4337
	C ₆ H ₄ (Cl)N=CHN(CH ₃) ₂ (Methanimidamide, <i>N'</i> -(2-chlorophenyl)- <i>N,N</i> -dimethyl-)	2103-49-3	Cl	8.6±0.1	EI	4359
			Cl	8.6	EI	4337

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_9\text{H}_{11}\text{N}_2^+$	$\text{C}_6\text{H}_4(\text{Br})\text{N}=\text{CHN}(\text{CH}_3)_2$ (Methanimidamide, N' -(2-bromophenyl)- N,N -dimethyl-)	53746-69-3	Br	8.4	EI	4337
	$\text{C}_6\text{H}_4(\text{I})\text{N}=\text{CHN}(\text{CH}_3)_2$ (Methanimidamide, N' -(2-iodophenyl)- N,N -dimethyl-)	53666-10-7	I	8.4	EI	4337
$\text{C}_9\text{H}_{12}\text{N}_2^+$	$\text{C}_6\text{H}_5\text{N}=\text{CHN}(\text{CH}_3)_2$ (Methanimidamide, N,N -dimethyl- N' -phenyl-)	1783-25-1	**	7.3 ± 0.1	EI	4359
			**	7.3	EI	4337
$\text{C}_9\text{H}_{14}\text{N}_2^+$	$\text{C}_9\text{H}_{14}\text{N}_2$ (3,4-Diazatricyclo[4.2.1.0 ^{2,5}]non-7-ene, 3,4-dimethyl-(1 α ,2 β ,5 β ,6 α -))	67144-64-3	**	7.68 (V)	PE	4780
	$\text{C}_8\text{H}_{12}\text{N}_2(=\text{CH}_2)$ (1,3-Diazatricyclo[3.3.1.1 ^{3,7}]decane, 6-methylene-)	51500-09-5	**	7.53 (V)	PE	4659
$\text{C}_9\text{H}_{16}\text{N}_2^+$	$\text{C}_9\text{H}_{16}\text{N}_2$ (3,4-Diazatricyclo[4.2.1.0 ^{2,5}]nonane, 3,4-dimethyl-(1 α ,2 β ,5 β ,6 α -))	67144-63-2	**	7.64 (V)	PE	4780
	$\text{C}_9\text{H}_{16}\text{N}_2$ (1,4-Methanopyrazino[1,2- a]pyridazine, octahydro-)	72282-74-7	**	7.19 (V)	PE	5133
$\text{C}_9\text{H}_{18}\text{N}_2^+$	$\text{C}_5\text{H}_{10}\text{N}(\text{C}_4\text{H}_8\text{N})$ (Piperidine, 1-(1-pyrrolidinyl)-)	49840-66-6	**	7.951 (V)	PE	4156
			**	7.95	PE	5280
$\text{C}_9\text{H}_{20}\text{N}_2^+$	$\text{C}_3\text{H}_6\text{N}_2(\text{C}_3\text{H}_7)_2$ (1-Azetidinamine, N,N -dipropyl-)	67092-89-1	**	7.5 (V)	PE	4780
	$\text{C}_3\text{H}_6\text{N}_2(\text{C}_3\text{H}_7)_2$ (Pyrazolidine, 1,2-bis(1-methylethyl)-)	38704-87-9	**	7.81 (V)	PE	4134
			**	7.89 (V)	PE	3889
			**	7.81	PE	5280
	$\text{C}_3\text{H}_4\text{N}_2(\text{CH}_3)_2(\text{C}_2\text{H}_5)_2$ (Pyrazolidine, 4,4-diethyl-1,2-dimethyl-, <i>trans</i> -)	53779-87-6	**	7.59 (V)	PE	4134
$\text{C}_{10}\text{H}_6\text{N}_2^+$	$\text{C}_{12}\text{H}_6\text{N}_4$ (2,3-Pyrazinedicarbonitrile,5-phenyl-)	52109-66-7	(CN) ₂	13.20	EI	5488
$\text{C}_{10}\text{H}_8\text{N}_2^+$	$(\text{C}_5\text{H}_4\text{N})_2$ (2,2'-Bipyridine)	366-18-7	**	8.35 ± 0.02	PE	3702
	$(\text{C}_5\text{H}_4\text{N})_2$ (4,4'-Bipyridine)	553-26-4	**	9.10 ± 0.02	PE	3702
$\text{C}_{10}\text{H}_{10}\text{N}_2^+$	$\text{C}_{10}\text{H}_6(\text{NH}_2)_2$ (1,5-Naphthalenediamine)	2243-62-1	**	6.74 ± 0.02	PE	4143
	$\text{C}_{10}\text{H}_6(\text{NH}_2)_2$ (1,8-Naphthalenediamine)	479-27-6		6.65 ± 0.02	PE	4143
$\text{C}_{10}\text{H}_{12}\text{N}_2^+$	$\text{C}_4\text{H}_4\text{NNC}_4\text{H}_2(\text{CH}_3)_2$ (1,1'-Bi-1H-pyrrole,2,5-dimethyl)	24046-14-8	**	7.77 (V)	PE	5387
	$\text{C}_6\text{H}_5\text{CH}_2\text{C}_3\text{H}_5\text{N}_2$ (1H-Imidazole,4,5-dihydro-2-(phenylmethyl)-)	59-98-3	**	8.50 (V)	PE	5096
	$\text{C}_8\text{H}_9\text{NCH}_2\text{CH}_2\text{NH}_2$ (1H-Indole-3-ethanamine)	61-54-1	**	7.69 ± 0.08 (V)	PE	4672

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{10}\text{H}_{13}\text{N}_2^+$	$\text{C}_6\text{H}_3(\text{Cl})(\text{CH}_3)\text{N}=\text{CHN}(\text{CH}_3)_2$ (Methanimidamide, <i>N'</i> (2-chloro-4-methylphenyl)- <i>N,N</i> -dimethyl-)	53666-35-6	Cl	8.6 ± 0.1	EI	4359
	$\text{C}_6\text{H}_3(\text{Cl})(\text{CH}_3)\text{N}=\text{CHN}(\text{CH}_3)_2$ (Methanimidamide, <i>N'</i> (2-chloro-5-methylphenyl)- <i>N,N</i> -dimethyl-)	53666-41-4	Cl	8.5 ± 0.1	EI	4359
$\text{C}_{10}\text{H}_{14}\text{N}_2^+$	$\text{C}_6\text{H}_5\text{N}=\text{N}(\text{tert-C}_4\text{H}_9)$ (Diazene, (1,1-dimethylethyl)phenyl-)	1775-83-3	**	8.35 ± 0.2 (V)	PE	4581
	$\text{C}_6(\text{NH}_2)_2(\text{CH}_3)_4$ (1,4-Benzenediamine, 2,3,5,6-tetramethyl-)	3102-87-2	**	8.63 ± 0.03	PI	5552
$\text{C}_{10}\text{H}_{16}\text{N}_2^+$	$\text{C}_6\text{H}_4(\text{N}(\text{CH}_3)_2)_2$ (1,4-Benzenediamine, N,N,N',N'-tetramethyl-)	100-22-1	**	6.1 ± 0.1	PE	4401
			**	6.20 ± 0.05	PI	3729
			**	6.7	CTS	3543
			**	6.75 (V)	PE	5382
	$\text{C}_6(\text{NH}_2)_2(\text{CH}_3)_4$ (1,4-Benzenediamine, 2,3,5,6-tetramethyl-)	3102-87-2	**	6.43	PI	4328
	$\text{C}_{10}\text{H}_{16}\text{N}_2$ (1,4-Ethanopyridazino[1,2- <i>a</i>]pyridazine, 1,4,6,7,8,9-hexahydro-)	72282-73-6	**	7.07 (V)	PE	5133
$\text{C}_{10}\text{H}_{18}\text{N}_2^+$	$\text{C}_6\text{H}_6\text{N}_2(\text{CH}_3)_4$ (2,3-Diazabicyclo[2.2.2]oct-5-ene, 1,2,3,4-tetramethyl-)	53779-88-7	**	7.43 (V)	PE	4134
	$\text{C}_{10}\text{H}_{18}\text{N}_2$ (1,4-Ethanopyridazino[1,2- <i>a</i>]pyridazine, octahydro-)	72282-72-5	**	7.06 (V)	PE	5133
	$\text{C}_3\text{N}_2(\text{CH}_3)_4=\text{C}(\text{CH}_3)_2$ (3H-Pyrazole, 4,5-dihydro-3,3,5,5-tetramethyl-4-(1-methylethylidene)-)	55204-47-2	**	8.58 (V)	PE	4429
$\text{C}_{10}\text{H}_{20}\text{N}_2^+$	$\text{C}_6\text{H}_8\text{N}_2(\text{tert-C}_4\text{H}_9)(\text{CH}_3)$ (2,3-Diazabicyclo[2.2.1]heptane, 2-(1,1-dimethylethyl)-3-methyl-)	42842-99-9	**	7.34	PE	5280
			**	7.33 (V)	PE	4134
	$\text{C}_6\text{H}_{14}\text{N}_2(\text{CH}_3)_2$ (9-Azabicyclo[3.3.1]nonan-9-amine, N,N-dimethyl-)	60678-79-7	**	7.53 (V)	PE	5091
	$\text{C}_6\text{H}_{12}\text{NNC}_4\text{H}_8$ (1H-Azepine, hexahydro-1-(pyrrolidinyl)-)	60678-75-3	**	7.60	PE	5280
	$(\text{C}_5\text{H}_{10}\text{N})_2$ (1,1'-Bipiperidine)	6130-94-5	**	7.89	PE	5280
			**	7.892 (V)	PE	4156
			**	8.05 (V)	PE	4085
	$\text{C}_6\text{H}_8(\text{CH}_3)_2\text{N}_2(\text{CH}_3)_2$ (2,3-Diazabicyclo[2.2.2]octane, 1,2,3,4-tetramethyl-)	59498-94-1	**	7.43	PE	5280
	$\text{C}_6\text{H}_8\text{N}_2(\text{CH}_3)_4$ (2,3-Diazabicyclo[2.2.2]octane, 1,2,3,4-tetramethyl-, <i>trans</i> -)	53779-86-5	**	7.43 (V)	PE	4134
	$(\text{C}_3\text{H}_4(\text{CH}_3)_2)_2\text{N}_2$ (1H,5H-Pyrazolo[1,2- <i>a</i>]pyrazole, tetrahydro-2,2,6,6-tetramethyl-)	2940-98-9	**	7.53	PE	5280
			**	7.53 (V)	PE	4134
$\text{C}_{10}\text{H}_{22}\text{N}_2^+$	$\text{C}_4\text{H}_4(\text{CH}_3)_4\text{N}_2(\text{CH}_3)_2$ (Pyridazine, hexahydro-1,2,3,3,6,6-hexamethyl-)	60678-80-0	**	7.46	PE	5280
$\text{C}_{10}\text{H}_{24}\text{N}_2^+$	$(\text{iso-C}_3\text{H}_7)_2\text{NN}(\text{iso-C}_3\text{H}_7)(\text{CH}_3)$	XXXXX-XX-X	**	7.60	PE	5280
	$(n-\text{C}_3\text{H}_7)_2\text{NN}(\text{C}_2\text{H}_5)_2$	52598-09-1	**	7.87	PE	5280
			**	7.87 (V)	PE	4137
	$(n-\text{C}_4\text{H}_9)_2\text{NN}(\text{CH}_3)_2$	60678-67-3	**	7.96	PE	5280
	$(\text{iso-C}_3\text{H}_7)_2\text{NN}(\text{C}_2\text{H}_5)_2$	XXXXX-XX-X	**	8.126 (V)	PE	4156

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₀H₂₄N₂⁺	<i>iso</i> -C ₃ H ₇ N(CH ₃)N(<i>iso</i> -C ₃ H ₇) ₂ ((<i>tert</i> -C ₄ H ₉)(CH ₃)N) ₂	49840-64-4 52291-46-0	** ** ** ** ** **	7.59 (V) 7.895 (V) 7.67 (V) 7.67 (V) 7.67 (V) 7.920 (V)	PE PE PE PE PE PE	4156 4156 5280 4137 5381 4156
C₁₁H₈N₂⁺	C ₁₁ H ₈ N ₂ (1H-Perimidine) C ₁₁ H ₈ N ₂ (9H-Pyrido[3,4- <i>b</i>]indole)	204-02-4 244-63-3	**	6.80 7.99±0.06 (V)	CTS PE	4035 4758
C₁₁H₁₄N₂⁺	C ₆ H ₄ (CH ₃)CH ₂ C ₃ H ₅ N ₂ (1H-Imidazole, 4,5-dihydro-2-[(2-methylphenyl)methyl]-) C ₈ H ₆ NCH ₂ CH ₂ NHCH ₃ (1H-Indole-3-ethanamine, N-methyl-) C ₈ H ₅ N(CH ₃)CH ₂ CH ₂ NH ₂ (1H-Indole-3-ethanamine, 5-methyl-) C ₈ H ₆ NCH ₂ N(CH ₃) ₂ (1H-Indole-3-methanamine, N,N-dimethyl-)	3038-50-4 61-49-4 1821-47-2 87-52-5	** ** ** **	8.60 (V) 7.60±0.08 (V) 7.64±0.05 (V) 7.69±0.16 (V)	PE PE PE PE	5096 4672 4672 4672
C₁₁H₂₂N₂⁺	C ₁₁ H ₂₂ N ₂ (8-Azabicyclo[3.2.1]octan-3-amine, 8-methyl-N-propyl- <i>endo</i> -) C ₁₁ H ₂₂ N ₂ (8-Azabicyclo[3.2.1]octan-3-amine, 8-methyl-N-propyl- <i>exo</i> -) C ₆ H ₁₂ NNC ₅ H ₁₀ (1H-Azepine, hexahydro-1-(1-piperidinyl)-)	67216-34-6 67139-56-4 60778-60-1	** ** **	8.0±0.15 8.1±0.15 7.87	EI EI PE	5401 5401 5280
C₁₂H₈N₂⁺	C ₁₂ H ₈ N ₂ (Benzoc[<i>c</i>]cinnoline) C ₁₂ H ₈ N ₂ (1,10-Phenanthroline) C ₁₂ H ₈ N ₂ (4,7-Phenanthroline) C ₁₂ H ₈ N ₂ (Phenazine)	230-17-1 66-71-7 230-07-9 92-82-0	** ** ** **	~8.69±0.02 (V) 8.51±0.02 (V) 8.35±0.02 (V) 8.33±0.02 (V)	PE PE PE PE	4430 4430 4430 4430
C₁₂H₁₀N₂⁺	C ₆ H ₅ N=NC ₆ H ₅ (Diazene, diphenyl-) <i>trans</i> -C ₆ H ₅ N=NC ₆ H ₅ (Diazene, diphenyl-, (<i>trans</i>)-) <i>trans</i> -(C ₅ H ₄ N) ₂ CH=CH (Pyridine, 2,2'-(1,2-ethenediyl)bis-(E)-) <i>trans</i> -(C ₅ H ₄ N) ₂ CH=CH (Pyridine, 4,4'-(1,2-ethenediyl)bis-(E)-) <i>trans</i> -(C ₅ H ₄ N) ₂ CH=CH (Pyridine, 2-[2-(3-pyridinyl)ethenyl]-(E)-) <i>trans</i> -(C ₅ H ₄ N) ₂ CH=CH (Pyridine, 2-[2-(4-pyridinyl)ethenyl]-(E)-) C ₁₁ H ₈ N ₂ CH ₃ (9H-Pyrido[3,4- <i>b</i>]indole, 1-methyl-)	103-33-3 17082-12-1 13341-40-7 13362-78-2 13362-75-9 14802-41-6 486-84-0	** ** ** ** ** ** **	8.5 (V) 8.46 (V) 8.5±0.05 (V) 8.83±0.03 (V) 8.33±0.03 (V) 8.50±0.03 (V) 7.83±0.06 (V)	PE PE PE PE PE PE PE	4467 4475 5320 4805 4805 4805 4758
C₁₂H₁₂N₂⁺	C ₆ H ₄ (NH ₂)C ₆ H ₄ NH ₂ ([1,1'-Biphenyl]-4,4'-diamine)	92-87-5	**	6.88	PI	4328

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₂H₁₂N₂⁺	(C ₆ H ₅ NH) ₂ (Hydrazine,1,2-diphenyl-)	122-66-7	** **	7.78±0.05 7.78 (V)	PE PE	5322 5381
C₁₂H₁₄N₂⁺	(C ₄ H ₂ NHCH ₂ CH ₂) ₂ (13,14-Diazatricyclo[8.2.1.1 ^{4,7}]tetradeca-4,6,10,12-tetraene)	73650-67-6	**	7.45 (V)	PE	5575
C₁₂H₁₆N₂⁺	(C ₄ H ₂ N(CH ₃) ₂) ₂ (1,1'-Bi-1H-pyrrole,2,2',5,5'-tetramethyl-) C ₁₂ H ₁₆ N ₂ (Benzencarboximidamide, N,N-dimethyl-N'-1-propenyl) C ₈ H ₆ NCH ₂ CH ₂ N(CH ₃) ₂ (1H-Indole-3-ethanamine, N,N-dimethyl-)	10507-71-8 68003-59-8 61-50-7	** ** **	7.73 (V) 7.20 (V) 7.57±0.05 (V)	PE PE PE	5387 4968 4672
C₁₂H₂₀N₂⁺	C ₆ H ₁₀ NN(C ₆ H ₁₀) (Cyclohexanone, cyclohexylidenehydrazone) (C ₆ H ₁₀ N) ₂	4278-87-9 XXXXX-XX-X	** **	7.84 7.84	PE PE	4043 5589
C₁₂H₂₂N₂⁺	C ₈ H ₁₄ N ₂ C ₄ H ₈ (Pyridazino[1,2- <i>b</i>]phthalazine,dodecahydro-, <i>trans</i> -)	60678-83-3	**	7.51	PE	5280
C₁₂H₂₅N₂⁺	C ₁₂ H ₂₅ N ₂ ⁺ (Hexyl,1,1,5-trimethyl-5-[(1-methylethyl)azo]-)	73322-99-3	**	7.39 (V)	PE	5091
C₁₂H₂₈N₂⁺	(<i>n</i> -C ₄ H ₉) ₂ NN(C ₂ H ₅) ₂ ((<i>n</i> -C ₅ H ₇) ₂ N) ₂	60678-68-4 60678-69-5	** **	7.77 7.74	PE PE	5280 5280
C₁₃H₈N₂⁺	C ₁₁ H ₈ (CN) ₂ (1,4-Methanonaphthalene-2,5-dicarbonitrile, 1,4-dihydro-) C ₁₁ H ₈ (CN) ₂ (1,4-Methanonaphthalene-2,6-dicarbonitrile, 1,4-dihydro-) C ₁₁ H ₈ (CN) ₂ (1,4-Methanonaphthalene-2,7-dicarbonitrile, 1,4-dihydro-) C ₁₁ H ₈ (CN) ₂ (1,4-Methanonaphthalene-2,8-dicarbonitrile, 1,4-dihydro-)	71925-32-1 71925-30-9 71925-31-0 71925-33-2	** ** ** **	9.31±0.05 (V) 9.30±0.05 (V) 9.27±0.05 (V) 9.27±0.05 (V)	PE PE PE PE	5235 5235 5235 5235
C₁₃H₁₀N₂⁺	C ₁₃ H ₁₀ N ₂ (1H-Phenalen-9-amine, 1-iminio-)	67618-27-3	**	7.27 ±0.1 (V)	PE	4951
C₁₃H₁₂N₂⁺	C ₆ H ₅ NNC ₆ H ₄ CH ₃ (Diazene,(4-methylphenyl)phenyl-(E)-)	6720-39-4	**	~8.3 (V)	PE	5320
C₁₃H₁₄N₂⁺	(C ₆ H ₄ NH ₂) ₂ CH ₂ (Benzenamine, 4-4'-methylenebis-)	101-77-9	** **	7.20 7.75±0.05	PI EI	4328 3806
C₁₃H₁₆N₂⁺	C ₁₀ H ₁₁ C ₃ H ₅ N ₂ (1H-Imidazole,4,5-dihydro-2-(1,2,3,4,-tetrahydro-1-naphthalenyl)-)	84-22-0	**	8.33 (V)	PE	5096

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$C_{13}H_{22}N_2^+$	$C_{13}H_{22}N_2$ (5,8-Ethano-1H-pyrazolo[1,2-a]pyridazine,2,2-diethyl-2,3,5,8-tetrahydro-) $C_{13}H_{22}N_2$ (Spiro[cyclohexane-1,3'-{3H-2,6]methanoimidazo[1,5-a]pyridine})	72282-76-9	**	7.04 (V)	PE	5133
$C_{13}H_{24}N_2^+$	$C_6H_{10}N_2C_3H_4(C_2H_5)_2$ (5,8-Ethano-1H-pyrazolo[1,2-a]pyridazine,2,2-diethylhexahydro-)	23211-28-1	**	6.92	PE	5280
				6.93 (V)	PE	4134
$C_{14}H_{12}N_2^+$	$C_{13}H_9N_2(CH_3)$ (1H-Cyclopenta[gh]perimidine, 6,7-dihydro-1-methyl-)	18969-93-2	**	6.53	CTS	4035
$C_{14}H_{14}N_2^+$	$(C_5H_3N)_2(CH_2)_4$ (15,16-Diazatricyclo[9.3.1.1 ^{4,8}]hexadeca-1(15),4,6,8(16),11,13-hexaene) $C_{14}H_{14}N_2$ (1,4-Ethanonaphtho[1,8-ef]-1,4-diazepine, 2,3-dihydro-) $C_{10}H_7CH_2C_3H_5N_2$ (1H-Imidazole,4,5-dihydro-2-(1-naphthalenylmethyl)-)	6574-83-0	**	8.35	PE	4386
				7.56 (V)	PE	4419
				8.46 (V)	PE	5096
$C_{14}H_{16}N_2^+$	$C_6H_4(NH_2)CH_2CH_2C_6H_4NH_2$ (Benzeneamine, 4,4'-(1,2-ethanediyl)bis-) $(C_4H_2N)_2(C_3H_6)_2$ (10b,10c-Diazadicyclopenta[ef,kl]heptalene,3,4,5,8,9,10-hexahydro-) $(C_6H_5N(CH_3))_2$ (Hydrazine,1,2-dimethyl-1,2-diphenyl-)	621-95-4	**	7.45±0.05	EI	3806
				7.72 (V)	PE	5387
				7.30±0.05	PE	5322
$C_{14}H_{18}N_2^+$	$C_{10}H_6(N(CH_3)_2)_2$ (1,5-Naphthalenediamine, N,N,N',N'-tetramethyl-) $C_{10}H_6(N(CH_3)_2)_2$ (1,8-Naphthalenediamine, N,N,N',N'-tetramethyl-) $C_{14}H_{18}N_2$ (1,4-Ethanopyridazino[1,2-b]phthalazine,1,2,3,4,6,11-hexahydro-)	10075-69-1	**	6.70±0.02	PE	4143
				6.45±0.02	PE	4143
				7.21 (V)	PE	5133
$C_{15}H_{14}N_2^+$	$C_{13}H_7(=NCH_3)NHCH_3$ (Phenalene,9-methylamino-1-methylimino-)	XXXXX-XX-X	**	6.98±0.04 (V)	PE	5595
$C_{15}H_{16}N_2^+$	$C_3H_5N_2(C_6H_5)_2$ (Pyrazolidine,1,2-diphenyl-)	63378-86-9	**	7.50±0.05	PE	5322
$C_{16}H_8N_2^+$	$C_{18}H_8N_4$ (Dibenzo[f,h]quinoxaline-2,3-dicarbonitrile)	55408-49-6	(CN) ₂	12.30	EI	5488
$C_{16}H_{18}N_2^+$	$C_{16}H_{18}N_2$ (2H-1,5-Propano-1H-naphtho[1,8-bc]-1,5-diazocine,3,4-dihydro-) $C_4H_8N_2(C_6H_5)_2$ (Pyridazine,hexahydro-1,2-diphenyl-)	59950-40-2	**	6.90 (V)	PE	4419
				7.30±0.05	PE	5322
$C_{16}H_{20}N_2^+$	$C_6H_4(N(CH_3)_2)C_6H_4N(CH_3)_2$ ([1,1'-Biphenyl]-4,4'-diamine, N,N,N',N'-tetramethyl-)	366-29-0	**	6.40	PI	4328
$C_{16}H_{24}N_2^+$	$C_{16}H_{24}N_2$ (1H-Imidazole,2-[[4-(1,1-dimethylethyl)-2,6-dimethylphenyl]methyl]-4,5-dihydro-)	526-36-3	**	8.49 (V)	PE	5096

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{16}\text{H}_{28}\text{N}_2^+$	$(\text{C}_6\text{H}_{14}\text{N})_2$ (9,9'-Bi-9-azabicyclo[3.3.1]nonane)	62796-83-2	**	6.94 (V)	PE	5091
$\text{C}_{16}\text{H}_{34}\text{N}_2^+$	<i>trans</i> -((CH ₃) ₃ CCH ₂ C(CH ₃) ₂) ₂ N=N	55204-43-8	**	8.00 (V)	PE	
$\text{C}_{17}\text{H}_{20}\text{N}_2^+$	$\text{C}_5\text{H}_{10}\text{N}_2(\text{C}_6\text{H}_5)_2$ (1H-1,2-Diazepine, hexahydro-1,2-diphenyl-)	63378-89-2	**	7.30 ± 0.05	PE	5322
$\text{C}_{17}\text{H}_{22}\text{N}_2^+$	$(\text{C}_6\text{H}_4\text{N}(\text{CH}_3)_2)_2\text{CH}_2$ (Benzenamine, 4,4'-methylenebis(N,N-dimethyl-))	101-61-1	**	6.72	PI	4328
			**	7.1	CTS	3543
$\text{C}_{18}\text{H}_{18}\text{N}_2^+$	$\text{C}_6\text{H}_5\text{C}_3\text{H}_3(\text{CN})\text{C}_6\text{H}_4\text{N}(\text{CH}_3)_2$ (Cyclopropanecarbonitrile, 2-(<i>p</i> -(dimethylamino)phenyl)-1-phenyl-)	6114-58-5	**	6.90 ± 0.10	EI	3575
$\text{C}_{18}\text{H}_{20}\text{N}_2^+$	$\text{C}_6\text{H}_{10}\text{N}_2(\text{C}_6\text{H}_5)_2$ (2,3-Diazabicyclo[2.2.2]octane,2,3-diphenyl-)	63378-90-5	**	7.15 ± 0.05	PE	5322
$\text{C}_{18}\text{H}_{24}\text{N}_2^+$	$(\text{C}_6\text{H}_5\text{N}(\text{i}-\text{C}_3\text{H}_7))_2$ (Hydrazine,1,2-bis(1-methylethyl)-1,2-diphenyl-) $(\text{C}_6\text{H}_5\text{N}(\text{C}_3\text{H}_7))_2$ (Hydrazine,1,2-diphenyl-1,2-dipropyl-)	63378-85-8	**	7.20 ± 0.05	PE	5322
			**	7.24 ± 0.05	PE	5322
$\text{C}_{19}\text{H}_{20}\text{N}_2^+$	$\text{C}_6\text{H}_4(\text{CH}_3)\text{C}_3\text{H}_3(\text{CN})\text{C}_6\text{H}_4\text{N}(\text{CH}_3)_2$ (Cyclopropanecarbonitrile, 2-(<i>p</i> -(dimethylamino)phenyl)-1- <i>p</i> -tolyl-)	32589-51-8	**	6.80 ± 0.07	EI	3575
$\text{C}_{19}\text{H}_{24}\text{N}_2^+$	$\text{C}_{14}\text{H}_{12}\text{N}(\text{CH}_2)_3\text{N}(\text{CH}_3)_2$ (<i>S</i> -Dibenz[b,f]azepine-5-propanamine, 10,11-dihydro- <i>N,N</i> -dimethyl-)	50-49-7	**	8.21 ± 0.07	CTS	4079
$\text{C}_{20}\text{H}_{18}\text{N}_2^+$	$\text{C}_6\text{H}_4(\text{CH}_2\text{NC}_6\text{H}_5)_2$ (Phthalazine,1,2,3,4-tetrahydro-2,3-diphenyl-)	16460-56-3	**	7.32 ± 0.05	PE	5322
$\text{C}_{20}\text{H}_{22}\text{N}_2^+$	$\text{C}_8\text{H}_{12}\text{N}_2(\text{C}_6\text{H}_5)_2$ (1,3-Diazatricyclo[3.3.1.1 ^{3,7}]decane, 5,7-diphenyl-)	38705-08-7	**	7.54 ± 0.03 (V)	PE	4163
$\text{C}_{20}\text{H}_{24}\text{N}_2^+$	$\text{C}_6\text{H}_{10}(\text{CH}_2\text{NC}_6\text{H}_5)_2$ (Phthalazine,decahydro-2,3-diphenyl, <i>trans</i> -)	63378-88-1	**	7.01 ± 0.05	PE	5322
$\text{C}_{20}\text{H}_{34}\text{N}_2^+$	$\text{C}_4(\text{CH}_3)_4(=\text{NC}_6\text{H}_{11})_2$ (Cyclohexanamine, N,N'-(2,2,4,4-tetramethyl-1,3-cyclobutanediyliidene)bis-)	6119-44-4	**	8.33 (V)	PE	5499
$\text{C}_{24}\text{H}_{16}\text{N}_2^+$	$\text{C}_{24}\text{H}_{16}\text{N}_2$ (25,26-Diazapentacyclo[19.3.1.1 ^{9,13} .0 ^{4,16} .0 ^{6,18}]hexacosa-1(25),2,4,6(18),7,9,11,13(26),14,16,19,21,23-tridecaene)	64031-65-8	**	6.97 (V)	PE	4824
$\text{C}_{26}\text{H}_{24}\text{N}_2^+$	$(\text{C}_6\text{H}_5\text{N}(\text{CH}_2\text{C}_6\text{H}_5)_2$ (Hydrazine,1,2-diphenyl-1,2-bis(phenylmethyl)-)	29334-75-6	**	7.59 ± 0.05	PE	5322

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
CH₃N₃⁺	CH ₃ N ₃	624-90-8	**	9.81±0.02	PE	3670
C₂H₃N₃⁺	C ₂ H ₃ N ₃ (1 <i>H</i> -1,2,4-Triazole)	288-88-0	**	10.6 (V)	PE	5228
			**	10.0 (V)	PE	4009
	C ₂ H ₃ N ₃ (1 <i>H</i> -1,2,3-Triazole)	288-36-8	**	10.06 (V)	PE	4009
C₃H₃N₃⁺	C ₂ H ₃ N ₃ (1 <i>H</i> -1,2,4-Triazole)	288-88-0	**	10.0 (V)	PE	4009
	C ₃ H ₃ N ₃ (1,2,4-Triazine)	290-38-0	**	9.61 (V)	PE	4707
	C ₃ H ₃ N ₃ (1,3,5-Triazine)	290-87-9	**	9.98	PE	3679
C₄H₅N₃⁺			**	10.01±0.01	PE	3720
	C ₃ H ₂ N ₃ (CH ₃) (1,2,4-Triazine, 3-methyl-)	24108-33-6	**	9.26 (V)	PE	4707
	C ₃ H ₂ N ₃ (CH ₃) (1,2,4-Triazine, 5-methyl-)	21134-95-2	**	9.31 (V)	PE	4707
C₅HN₃⁺	C ₃ H ₂ N ₃ (CH ₃) (1,2,4-Triazine, 6-methyl-)	21134-96-3	**	9.35 (V)	PE	4707
	C(CN) ₂ =CHCN	997-76-2	**	~11.55	PE	4859
	C ₁₂ H ₆ N ₄ (2,3-Pyrazinedicarbonitrile,5-phenyl-)	52109-66-7	C ₆ H ₅ CN	11.27	EI	5488
C₅H₇N₃⁺	C ₃ HN ₃ (CH ₃) ₂ (1,2,4-Triazine, 3,5-dimethyl-)	24108-34-7	**	9.02 (V)	PE	4707
	C ₃ HN ₃ (CH ₃) ₂ (1,2,4-Triazine, 5,6-dimethyl-)	21134-90-7	**	9.15 (V)	PE	4707
	C ₁₃ H ₈ N ₄ (2,3-Pyrazinedicarbonitrile,5-methyl-6-phenyl-)	52109-67-8	C ₆ H ₅ CN	11.92	EI	5488
C₆H₅N₃⁺	C ₆ H ₅ N ₃ (Benzene, azido-)	622-37-7	**	8.72±0.02 (V)	PE	4674
	C ₆ H ₅ N ₃ (1 <i>H</i> -Benzotriazole)	95-14-7	**	9.20±0.05	EI	4316
	C ₄ H ₃ N ₂ C ₂ H ₂ N (Imidazo[1,2- <i>b</i>]pyridazine)	766-55-2	**	8.33 (V)	PE	5396
	C ₃ N ₃ (CH ₃) ₃ (1,2,4-Triazine, 3,5,6-trimethyl-)	24108-36-9	**	8.84 (V)	PE	4707
C₆H₁₅N₃⁺	(CH ₂ =NCH ₃) ₃	108-74-7	**	8.33±0.05 (V)	PE	4776
	(CH ₃ CH=NH) ₃	638-14-2	**	8.45±0.05 (V)	PE	4776
	C ₃ H ₆ N ₃ (CH ₃) ₃ (1,2,4-Triazine, hexahydro-1,2,4-trimethyl-)	66175-25-5	**	8.10 (V)	PE	5215

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_8\text{H}_{15}\text{N}_3^+$	$\text{C}_7\text{H}_{12}\text{N}_3(\text{CH}_3)$ (1,3,5-Triazatricyclo[3.3.1.1.3,7]decane, 7-methyl-)	38705-10-1	**	8.08 (V)	PE	4141
$\text{C}_9\text{H}_{11}\text{N}_3^+$	$\text{C}_6\text{H}_5\text{NC}_3\text{H}_4\text{N}_2\text{H}_2$ (Imidazolidine,2-(phenylimino)-)	XXXXX-XX-X	**	7.85 (V)	PE	5545
$\text{C}_{10}\text{H}_{11}\text{N}_3^+$	$\text{C}_6\text{H}_4\text{C}_4\text{H}_7\text{N}_3$ (Imidazo[2,1- <i>b</i>]quinazoline,1,2,3,5-tetrahydro-)	32725-29-4	**	7.46 (V)	PE	5545
$\text{C}_{10}\text{H}_{13}\text{N}_3^+$	$\text{C}_6\text{H}_4(\text{CH}_3)\text{NC}_3\text{H}_4\text{N}_2\text{H}_2$ (Imidazolidine,2-(2-methylphenylimino)-)	XXXXX-XX-X	**	7.75 (V)	PE	5545
$\text{C}_{11}\text{H}_5\text{N}_3^+$	$\text{C}_{12}\text{H}_6\text{N}_4$ (2,3-Pyrazinedicarbonitrile,5-phenyl-)	52109-66-7	HCN	11.61	EI	5488
	$\text{C}_{13}\text{H}_8\text{N}_4$ (2,3-Pyrazinedicarbonitrile,5-methyl-6-phenyl-)	52109-67-8	CH_3CN	11.48	EI	5488
$\text{C}_{11}\text{H}_{15}\text{N}_3^+$	$\text{C}_6\text{H}_3(\text{CH}_3)_2\text{NC}_3\text{H}_4\text{N}_2\text{H}_2$ (Imidazolidine,2-(2,6-dimethylphenylimino)-)	XXXXX-XX-X	**	7.63 (V)	PE	5545
$\text{C}_{11}\text{H}_{16}\text{N}_3^+$	$\text{C}_6\text{H}_4(\text{N}(\text{CH}_3)_2)\text{N}=\text{CHN}(\text{CH}_3)_2$ (Methanimidamide, <i>N</i> '-[3-(dimethylamino)phenyl]- <i>N,N</i> -dimethyl-)	53666-32-3	H	8.8 ± 0.1	EI	4359
	$\text{C}_6\text{H}_4(\text{N}(\text{CH}_3)_2)\text{N}=\text{CHN}(\text{CH}_3)_2$ (Methanimidamide, <i>N</i> '-[4-(dimethylamino)phenyl]- <i>N,N</i> -dimethyl-)	53666-31-2	H	9.0 ± 0.1	EI	4359
	$\text{C}_6\text{H}_3(\text{Cl})(\text{N}(\text{CH}_3)_2)\text{N}=\text{CHN}(\text{CH}_3)_2$ (Methanimidamide, <i>N</i> '-[2-chloro-4-(dimethylamino)phenyl]- <i>N,N</i> -dimethyl-)	53666-30-1	Cl	9.1 ± 0.1	EI	4359
	$\text{C}_6\text{H}_3(\text{Cl})(\text{N}(\text{CH}_3)_2)\text{N}=\text{CHN}(\text{CH}_3)_2$ (Methanimidamide, <i>N</i> '-[2-chloro-5-(dimethylamino)phenyl]- <i>N,N</i> -dimethyl-)	53666-39-0	Cl	8.9 ± 0.1	EI	4359
	$\text{C}_6\text{H}_4(\text{N}(\text{CH}_3)_2)\text{N}=\text{CHN}(\text{CH}_3)_2$ (Methanimidamide, <i>N</i> '-[3-(dimethylamino)phenyl]- <i>N,N</i> -dimethyl-)	53666-32-3	**	6.3 ± 0.1	EI	4359
	$\text{C}_6\text{H}_4(\text{N}(\text{CH}_3)_2)\text{N}=\text{CHN}(\text{CH}_3)_2$ (Methanimidamide, <i>N</i> '-[4-(dimethylamino)phenyl]- <i>N,N</i> -dimethyl-)	53666-31-2	**	6.1 ± 0.1	EI	4359
$\text{C}_{12}\text{H}_7\text{N}_3^+$	$\text{C}_{13}\text{H}_8\text{N}_4$ (2,3-Pyrazinedicarbonitrile,5-methyl-6-phenyl-)	52109-67-8	HCN	14.82	EI	5488
$\text{C}_{12}\text{H}_8\text{N}_3^+$	$\text{C}_{13}\text{H}_8\text{N}_4$ (2,3-Pyrazinedicarbonitrile,5-methyl-6-phenyl-)	52109-67-8	CN	15.10	EI	5488
$\text{C}_{12}\text{H}_9\text{N}_3^+$	$\text{C}_4\text{H}_3\text{N}_2\text{C}_2\text{HNC}_6\text{H}_5$ (Imidazo[1,2- <i>b</i>]pyridazine,2-phenyl-)	1844-54-8	**	7.73 (V)	PE	5396
$\text{C}_{12}\text{H}_{11}\text{N}_3^+$	$\text{C}_6\text{H}_5\text{NNC}_6\text{H}_4\text{NH}_2$ (Benzenamine,4-(phenylazo)-(E)-)	25548-34-9	**	7.67 ± 0.05 (V)	PE	5320
	$\text{C}_{11}\text{H}_6\text{N}_2(\text{NH}_2)\text{CH}_3$ (1 <i>H</i> -Perimidin-2-amine, 1-methyl-)	20551-10-4	**	6.41	CTS	4035
$\text{C}_{12}\text{H}_{13}\text{N}_3^+$	$(\text{C}_6\text{H}_4\text{NH}_2)_2\text{NH}$ (1,4-Benzenediamine, <i>N</i> -(4-aminophenyl)-)	537-65-5	**	6.20	PI	4328

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₈H₁₇N₃⁺	C ₁₀ H ₁₁ NC ₃ H ₄ N ₂ H ₂ (Imidazolidine,2-[5,6,7,8-tetrahydronaphthal-1-yl]imino-)	XXXXX-XX-X **		7.62 (V)	PE	5545
C₁₇H₈N₃⁺	C ₁₈ H ₈ N ₄ (Dibenzo[<i>f,h</i>]quinoxaline-2,3-dicarbonitrile)	55408-49-6	CN	13.10	EI	5488
CH₂N₄⁺	CH ₂ N ₄ (1 <i>H</i> -Tetrazole)	288-94-8	**	11.3 (V)	PE	4009
C₂H₂N₄⁺	C ₂ H ₂ N ₄ (1,2,4,5-Tetrazine)	290-96-0	**	9.14	PE	3679
			**	9.24	PE	3740
C₂H₄N₄⁺	C ₂ H ₄ N ₄ (1 <i>H</i> -1,2,4-Triazole-5-amine)	XXXXX-XX-X		8.3	EI	5487
	C ₄ H ₈ N ₄ (1 <i>H</i> -1,2,4-Triazol-5-amine,1-ethyl-)	58661-94-2	C ₂ H ₄	10.0	EI	5487
	C ₄ H ₈ N ₄ (4 <i>H</i> -1,2,4-Triazol-3-amine,4-ethyl-)	42786-06-1	C ₂ H ₄	10.2	EI	5487
	C ₄ H ₈ N ₄ (1 <i>H</i> -1,2,4-Triazole-3-amine,1-ethyl-)	42786-04-9	C ₂ H ₄	10.2	EI	5487
	C ₅ H ₁₀ N ₄ (4 <i>H</i> -1,2,4-Triazol-3-amine,4-propyl-)	58661-97-5		10.4	EI	5487
	C ₅ H ₁₀ N ₄ (1 <i>H</i> -1,2,4-Triazole-3-amine,1-propyl-)	58661-95-3	C ₃ H ₆	10.2	EI	5487
	C ₅ H ₁₀ N ₄ (1 <i>H</i> -1,2,4-Triazole-5-amine,1-propyl-)	58661-96-4	C ₃ H ₆	9.9	EI	5487
C₄H₆N₄⁺	C ₂ N ₄ (CH ₃) ₂ (1,2,4,5-Tetrazine, 3,6-dimethyl-)	1558-23-2	**	9.08 (V)	PE	3679
C₄H₈N₄⁺	C ₄ H ₈ N ₄ (1 <i>H</i> -1,2,4-Triazol-5-amine,1-ethyl-)	58661-94-2		8.5	EI	5487
	C ₄ H ₈ N ₄ (4 <i>H</i> -1,2,4-Triazol-3-amine,4-ethyl-)	42786-06-1		8.3	EI	5487
	C ₄ H ₈ N ₄ (1 <i>H</i> -1,2,4-Triazole-3-amine,1-ethyl-)	42786-04-9		8.2	EI	5487
C₄H₁₀N₄⁺	C ₂ N ₂ (CH ₃) ₂ =N ₂ (1,2,3,4-Tetrazine,1,4,5,6-tetrahydro-1,4-dimethyl-)	39247-66-0	**	8.03 (V)	PE	5604
C₄H₁₂N₄⁺	((CH ₃) ₂ N ₂) ₂ (2-Tetrazene,1,1,4,4-tetramethyl-)	6130-87-6	**	7.7 (V)	PE	5604
C₅H₄N₄⁺	C ₅ H ₄ N ₄ (1 <i>H</i> -Purine)	120-73-0	**	9.52±0.03 (V)	PE	4445
	C ₅ H ₄ NN ₃ (Tetrazolo[1,5- <i>a</i>]Pyridine)	274-87-3	**	8.85 (V)	PE	5396
	C ₅ H ₄ N ₄ (1 <i>H</i> -1,2,3-Triazolo[4,5- <i>c</i>]pyridine)	273-05-2	**	9.10±0.05	EI	4316
	C ₅ H ₄ N ₄ ([1,2,4]Triazolo[1,5- <i>a</i>]pyrazine)	399-66-6	**	9.6 (V)	PE	5492

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_5\text{H}_4\text{N}_4^+$	$\text{C}_5\text{H}_4\text{N}_4$ (1 <i>H</i> -1,2,3-Triazolo[4,5- <i>b</i>]pyridine)	273-34-7	**	9.20±0.05	EI	4316
$\text{C}_5\text{H}_{10}\text{N}_4^+$	$\text{C}_5\text{H}_{10}\text{N}_4$ (4 <i>H</i> -1,2,4-Triazol-3-amine,4-propyl-)	58661-97-5		8.3	EI	5487
	$\text{C}_5\text{H}_{10}\text{N}_4$ (1 <i>H</i> -1,2,4-Triazole-3-amine,1-propyl-)	58661-95-3		8.1	EI	5487
	$\text{C}_5\text{H}_{10}\text{N}_4$ (1 <i>H</i> -1,2,4-Triazole-5-amine,1-propyl-)	58661-96-4		8.3	EI	5487
$\text{C}_6\text{H}_6\text{N}_4^+$	$\text{C}_5\text{H}_3\text{N}_4\text{CH}_3$ (1 <i>H</i> -Purine, 6-methyl-)	2004-03-7	**	9.3 (V)	PE	5492
	$\text{C}_5\text{H}_3\text{N}_4\text{CH}_3$ (7 <i>H</i> -Purine, 7-methyl-)	18346-04-8	**	9.4 (V)	PE	5492
	$\text{C}_5\text{H}_3\text{N}_4\text{CH}_3$ (9 <i>H</i> -Purine, 9-methyl-)	20427-22-9	**	9.4 (V)	PE	5492
$\text{C}_6\text{H}_{12}\text{N}_4^+$	$\text{C}_6\text{H}_{12}\text{N}_4$ (1,3,5,7-Tetraazatricyclo[3.3.1.1. ^{3,7}]decane)	100-97-0	**	8.53 (V)	PE	4141
$\text{C}_6\text{H}_{16}\text{N}_4^+$	$\text{C}_2\text{H}_4\text{N}_4(\text{CH}_3)_4$ (1,2,4,5-Tetrazacyclohexane, 1,2,4,5-tetramethyl-)	XXXXXX-XX-X	**	7.90 (V)	PE	5504
	$\text{C}_2\text{H}_4\text{N}_4(\text{CH}_3)_4$ (1,2,4,5-Tetrazine, hexahydro-1,2,4,5-tetramethyl-)	20717-38-8	**	7.90 (V)	PE	4277
			**	7.90 (V)	PE	5215
			**	7.90 (V)	PE	5353
			**	9.00 (V)	PE	4277
$\text{C}_7\text{H}_{16}\text{N}_4^+$	$\text{C}_2\text{H}_4\text{N}_4(\text{CH}_3)_2(\text{C}_3\text{H}_6)$ (6 <i>H</i> -Pyrazolo[1,2- <i>a</i>][1,2,4,5]tetrazine,hexahydro-2,3-dimethyl-)	70517-50-9	**	7.76 (V)	PE	5489
$\text{C}_8\text{H}_{12}\text{N}_4^+$	<i>trans</i> -(NCC(CH ₃) ₂) ₂ N=N	34241-39-9	**	9.62 (V)	PE	4429
$\text{C}_8\text{H}_{16}\text{N}_4^+$	$\text{C}_2\text{H}_4\text{N}_4(\text{CH}_3)_2(\text{C}_4\text{H}_6)$ (Pyridazino[1,2- <i>a</i>][1,2,4,5]tetrazine,1,2,3,4,6,9-hexahydro-2,3-dimethyl-)	53233-92-4	**	7.77 (V)	PE	5489
	$\text{C}_2\text{H}_4\text{N}_4(\text{C}_3\text{H}_6)_2$ (1 <i>H</i> ,5 <i>H</i> ,7 <i>H</i> ,11 <i>H</i> -Dipyrazolo[1,2- <i>a</i> :1',2'- <i>d</i>][1,2,4,5]tetrazine,tetrahydro-)	37882-92-1	**	7.55 (V)	PE	5489
	$\text{C}_8\text{H}_{16}\text{N}_4$ (1,3,6,8-Tetraazatricyclo[4.4.1.1 ^{3,8}]dodecane)	51-46-7	**	7.389	PE	4214
$\text{C}_9\text{H}_{18}\text{N}_4^+$	$\text{C}_2\text{H}_4\text{N}_4(\text{CH}_3)_2(\text{C}_4\text{H}_6)$ (Pyridazino[1,2- <i>a</i>][1,2,4,5]tetrazine,octahydro-2,3-dimethyl-)	61012-98-4	**	7.69 (V)	PE	5489
$\text{C}_8\text{H}_{20}\text{N}_4^+$	($\text{N}_2(\text{C}_2\text{H}_5)_2$) ₂ (2-Tetrazene,1,1,4,4-tetraethyl-)	13304-29-5	**	7.1 (V)	PE	5604
	$\text{C}_2\text{H}_2\text{N}_4(\text{CH}_3)_6$ (1,2,4,5-Tetrazine, hexahydro-1,2,3,4,5,6-hexamethyl- <i>trans</i> -)	71899-35-9	**	7.63 (V)	PE	5215
$\text{C}_9\text{H}_{24}\text{N}_4^+$	$\text{C}(\text{N}(\text{CH}_3)_2)_4$	10524-51-3	**	7.19 (V)	PE	4588

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{10}\text{H}_{16}\text{N}_4^+$	$\text{C}_2\text{H}_4\text{N}_4(\text{C}_6\text{H}_6)_2$ (6H,13H-Dipyridazino[1,2-a:1',2'-d][1,2,4,5-tetrazine,1,4,8,11-tetrahydro-])	37882-93-2	** **	7.51 (V) 7.73 (V)	PE PE	5489 5215
$\text{C}_{10}\text{H}_{20}\text{N}_4^+$	$\text{C}_2\text{H}_4\text{N}_4(\text{C}_6\text{H}_6)_2$ (6H,13H-Dipyridazino[1,2-a:1',2'-d][1,2,4,5]tetrazine,octahydro-) $(\text{C}_3\text{H}_4\text{N}_2(\text{CH}_3)_2)_2$ (Imidazolidine,2-(1,3-dimethyl-2-imidazolidinylidene)-1,3-dimethyl-)	5767-20-4 1911-01-9	** ** **	7.46 (V) <5.41 6.06 (V)	PE PI PE	5489 5277 3512
$\text{C}_{10}\text{H}_{24}\text{N}_4^+$	$((\text{CH}_3)_2\text{N})_4\text{C}_2$	996-70-3	** **	<5.36 5.95 (V)	PI PE	5277 3512
$\text{C}_{12}\text{H}_6\text{N}_4^+$	$\text{C}_{12}\text{H}_6\text{N}_4$ (2,3-Pyrazinedicarbonitrile,5-phenyl-)	52109-66-7	**	8.68	EI	5488
$\text{C}_{12}\text{H}_{26}\text{N}_4^+$	$((\text{CH}_3)_2\text{N})_2\text{C}=\text{CH}_2$	10596-53-9	**	5.60 ± 0.10	PI	5278
$\text{C}_{12}\text{H}_{28}\text{N}_4^+$	$((\text{CH}_3)_2\text{CH}_2\text{NN})_2$ (2-Tetrazene,1,1,4,4-tetrakis(1-methylethyl))	13304-31-9	**	6.9 (V)	PE	5604
$\text{C}_{13}\text{H}_7\text{N}_4^+$	$\text{C}_{13}\text{H}_6\text{N}_4$ (2,3-Pyrazinedicarbonitrile,5-methyl-6-phenyl-)	52109-67-8	H	9.35	EI	5488
$\text{C}_{13}\text{H}_6\text{N}_4^+$	$\text{C}_{13}\text{H}_6\text{N}_4$ (2,3-Pyrazinedicarbonitrile,5-methyl-6-phenyl-)	52109-67-8		8.65	EI	5488
$\text{C}_{16}\text{H}_{18}\text{N}_4^+$	$\text{C}_{16}\text{H}_{18}\text{N}_4$ (Aniline, 2,2'-[1,2-ethanediylbis(nitrilomethylidyne)]bis-)	XXXXX-XX-X	**	7.83 ± 0.04	EI	4668
$\text{C}_{16}\text{H}_{28}\text{N}_4^+$	$(\text{C}_8\text{H}_{14}\text{N}_2)_2$ (9-Azabicyclo[3.3.1]nonane,9,9'-azobis-)	67282-66-0	**	7.07 (V)	PE	5091
$\text{C}_{18}\text{H}_8\text{N}_4^+$	$\text{C}_{18}\text{H}_8\text{N}_4$ (Dibenzo[f,h]quinoxaline-2,3-dicarbonitrile)	55408-49-6	**	8.20	EI	5488
$\text{C}_{18}\text{H}_{24}\text{N}_4^+$	$\text{C}_{18}\text{H}_{24}\text{N}_4$ (1,2,4,5-Tetrazine, hexahydro-1,4-dimethyl-2,5-bis(phenylmethyl)-)	61012-91-7	**	7.71 (V)	PE	5215
$\text{C}_{30}\text{H}_{32}\text{N}_4^+$	$\text{C}_2\text{H}_4\text{N}_4(\text{CH}_2\text{C}_6\text{H}_5)_4$ (1,2,4,5-Tetrazine, hexahydro-1,3,5,6-tetrakis(4-methylphenyl)-)	38422-60-5	**	7.44 (V)	PE	5215
$\text{C}_{36}\text{H}_{46}\text{N}_4^+$	$\text{C}_{20}\text{H}_{10}\text{N}_4(\text{C}_6\text{H}_5)_8$ (21H, 23H-Porphine,2,3,7,8,12,13,17,18-octaethyl-)	2683-82-1	** **	6.25 (V) 6.39 ± 0.03 (V)	PE PE	4557 5476
$\text{C}_{44}\text{H}_{30}\text{N}_4^+$	$\text{C}_{20}\text{H}_{10}\text{N}_4(\text{C}_6\text{H}_5)_4$ (21H, 23H-Porphine,5,10,15,20-tetraphenyl-)	917-23-7	**	6.39 (V)	PE	4557

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₄₄H₃₀N₄⁺	C ₂₀ H ₁₀ N ₄ (C ₆ H ₅) ₄	917-23-7	**	6.32±0.2	OTH	4962
C₅H₅N₅⁺	C ₅ H ₃ N ₄ (NH ₂) (1H-Purin-6-amine)	73-24-5	**	8.44±0.03 (V)	PE	4445
			**	8.3±0.1	EI	5555
			**	8.48 (V)	PE	4644
			**	8.48 (V)	PE	5492
C₆H₃N₅⁺	C ₅ H ₃ N(CN)N ₃ (Tetrazolo[1,5- <i>a</i>]pyridine-8-carbonitrile)	40306-97-6	**	9.22 (V)	PE	5396
C₆H₇N₅⁺	C ₅ H ₃ N ₄ (NHCH ₃) (1H-Purin-6-amine, N-methyl-)	443-72-1	**	8.15 (V)	PE	5492
			**	8.15 (V)	PE	4644
			**	8.39 (V)	PE	4644
	C ₅ H ₂ N ₄ (CH ₃)NH ₂ (7H-Purin-6-amine, 7-methyl-)	935-69-3	**	8.64 (V)	PE	5492
	C ₅ H ₂ N ₄ (NH ₂)CH ₃ (9H-Purin-6-amine, 9-methyl-)	700-00-5	**	8.39 (V)	PE	5492
	C ₅ H ₂ N ₄ (NH ₂)CH ₃ (1H-Purin-6-amine, 9-methyl-)	XXXXX-XX-X	**	7.9±0.1	EI	5555
C₇H₉N₅⁺	C ₅ H ₂ N ₄ (CH ₃)NHCH ₃ (9H-Purin-6-amine, N,9-dimethyl-)	2009-52-1	**	7.95 (V)	PE	5492
	C ₅ H ₃ N ₄ N(CH ₃) ₂ (1H-Purin-6-amine, N,N-dimethyl-)	938-55-6	**	7.78 (V)	PE	5492
C₁₁H₁₈N₅⁺	C ₁₁ H ₁₃ N ₄ NH ₂ (9H-Purin-6-amine, 9-cyclohexyl-)	4235-94-3	**	9.1	CTS	3915
C₄H₁₂N₆⁺	C ₂ H ₄ N ₂ H ₂ N ₂ C ₂ H ₄ N ₂ H ₂ ([1,2,4,5]Tetrazino[1,2- <i>a</i>][1,2,4,5]tetrazine, octahydro-)	1743-13-1	**	11.05 (V)	PE	5381
C₃₂H₁₈N₆⁺	C ₃₂ H ₁₈ N ₆ (29H,31H-Phthalocyanine)	574-93-6	**	7.36±0.10	EI	3829
BCH₈N⁺	(CH ₃ NH ₂)(BH ₃)	1722-33-4	**	9.66±0.01	PE	3699
BC₂H₈N⁺	(CH ₃) ₂ NBH ₂	1838-13-7	**	9.51	PE	3584
BC₂H₉N⁺	((CH ₃) ₂ NH)(BH ₂)	74-94-2	**	9.39±0.01	PE	3699
BC₃H₁₂N⁺	((CH ₃) ₃ N)(BH ₃)	75-22-9	**	9.28±0.2	PE	3699
BC₄H₁₂N⁺	(CH ₃) ₂ NB(CH ₃) ₂	1113-30-0	**	8.92	PE	3584
			**	8.92 (V)	PE	4243
			**	9.02 (V)	PE	5581

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
BC₅H₈N⁺	C ₅ H ₅ N·BH ₃ (Pyridine, compound with borane (1:1))	110-51-0	**	9.72 (V)	PE	4536
BC₆H₁₀N⁺	C ₅ H ₄ N(CH ₃)·BH ₃ (Pyridine, 4-methyl-, compound with borane (1:1))	3999-39-1	**	9.50 (V)	PE	4536
BC₆H₁₂N⁺	(C ₃ H ₆) ₂ BN (1H,5H-[1,2]Azaborolo[1,2-a][1,2]azaborole,tetrahydro-)	16153-13-2	** **	8.80 (V) 8.06	PE PE	5609 3584
BC₉H₁₆N⁺	C ₅ H ₄ N(<i>tert</i> -C ₄ H ₉)·BH ₃ (Pyridine, 4-(1,1-dimethylethyl)-, compound with borane (1:1))	56898-51-2	**	9.45 (V)	PE	4536
BC₁₀H₂₀N⁺	(CH ₃) ₂ BNC ₈ H ₁₄ (1-Azabicyclo[3.3.1]nonane,1-dimethylboryl-) (CH ₃) ₂ NBC ₈ H ₁₄ (Methanamine,N-methyl-N-(9-boratabicyclo[3.3.1]non-9-yl))	XXXXX-XX-X	** **	8.53 (V) 8.73 (V)	PE PE	5581 5581
BC₁₆H₂₈N⁺	C ₈ H ₁₄ BNC ₈ H ₁₄ (1-Azabicyclo[3.3.1]nonane,1-(9-boratabicyclo[3.3.1]non-9-yl)-)	XXXXX-XX-X	**	8.31 (V)	PE	5581
BC₄H₁₁N₂⁺	C ₂ H ₅ BN ₂ (CH ₃) ₂ (1,3,2-Diazaborolidine, 1,3-dimethyl-)	38151-26-7	**	7.55 (V)	PE	4298
BC₄H₁₃N₂⁺	((CH ₃) ₂ N) ₂ BH	2386-98-3	**	7.76	PE	3584
BC₅H₁₅N₂⁺	((CH ₃) ₂ N) ₂ B(CH ₃)	6914-63-2	**	7.63	PE	3584
B₂C₆H₁₈N⁺	((CH ₃) ₂ BNCH ₃) ₂ (Boranediamine,N-(dimethylboryl)-N,N',N',1-tetramethyl-)	73263-55-5	**	9.02 (V)	PE	5628
BC₈H₁₇N₂⁺	C ₂ H ₅ BN ₂ (CH ₃) ₂ C(CH ₃) ₃ (1H-1,3,2-Diazaborole, 2-(1,1-dimethylethyl)-2,3-dihydro-1,3-dimethyl-)	53088-51-0	**	7.25 (V)	PE	4298
BC₈H₁₉N₂⁺	C ₂ H ₄ BN ₂ (CH ₃) ₂ C(CH ₃) ₃ (1,3,2-Diazaborolidine, 2-(1,1-dimethylethyl)-1,3-dimethyl-)	53088-52-1	**	7.46 (V)	PE	4298
BC₉H₁₁N₂⁺	C ₆ H ₅ C ₂ H ₅ BN ₂ CH ₃ (1H-1,3,2-Diazaborole, 2,3-dihydro-1-methyl-2-phenyl-)	53088-50-9	**	7.53 (V)	PE	4298
BC₉H₁₃N₂⁺	C ₆ H ₅ C ₂ H ₅ BN ₂ CH ₃ (1,3,2-Diazaborolidine, 1-methyl-2-phenyl-)	6076-64-8	**	7.91 (V)	PE	4298
BC₁₀H₁₃N₂⁺	C ₆ H ₅ C ₂ H ₂ BN ₂ (CH ₃) ₂ (1H-1,3,2-Diazaborole, 2,3-dihydro-1,3-dimethyl-2-phenyl-)	41422-89-3	**	7.34 (V)	PE	4298

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
BC₁₀H₁₅N₂⁺	C ₆ H ₅ C ₂ H ₄ BN ₂ (CH ₃) ₂ (1,3,2-Diazaborolidine, 1,3-dimethyl-2-phenyl-)	5709-94-4	**	7.48 (V)	PE	4298
B₂C₃H₁₁N₃⁺	N ₃ B ₂ H ₂ (CH ₃) ₃ (1,2,4,3,5-Triazadiborolidine, 1,2,4-trimethyl-)	53246-11-0	**	7.78 (V)	PE	4526
	N ₃ B ₂ H ₂ (CH ₃) ₃ (1,2,4,3,5-Triazadiborolidine, 1,3,5-trimethyl-)	40392-35-6	**	7.76 (V)	PE	4526
B₃C₃H₁₂N₃⁺	C ₃ H ₁₂ B ₃ N ₃ (Borazine, 1,3,5-trimethyl-)	1004-35-9	**	8.99 (V)	PE	3943
			**	9.28±0.02	PE	3506
	C ₃ H ₁₂ B ₃ N ₃ (Borazine, 2,4,6-trimethyl-)	5314-85-2	**	9.50 (V)	PE	3943
			**	9.64±0.03	PE	3506
B₂C₄H₁₃N₃⁺	N ₃ B ₂ H(CH ₃) ₄ (1,2,4,3,5-Triazadiborolidine, 1,2,3,5-tetramethyl-)	31732-40-8	**	7.51 (V)	PE	4526
	N ₃ B ₂ H(CH ₃) ₄ (1,2,4,3,5-Triazadiborolidine, 1,3,4,5-tetramethyl-)	40392-34-5	**	7.73 (V)	PE	4526
	N ₃ B ₂ (CH ₃) ₅ (1,2,4,3,5-Triazadiborolidine, 1,2,3,4,5-pentamethyl-)	31732-41-9	**	7.47 (V)	PE	4526
BC₆H₁₄N₃⁺	C ₆ H ₁₄ BN ₃ ([1,3,2]Diazaborino[1,2- <i>a</i>][1,3,2]diazaborine, octahydro-)	1730-15-0	**	7.90	PE	3584
BC₆H₁₈N₃⁺	B(N(CH ₃) ₂) ₃	4375-83-1	**	7.60 (V)	PE	3704
B₃C₆H₁₈N₃⁺	C ₆ H ₁₈ B ₃ N ₃ (Borazine, hexamethyl-)	877-07-6	**	8.53 (V)	PE	3943
B₂C₈H₂₁N₃⁺	N ₃ B ₂ (CH ₃) ₄ C(CH ₃) ₃ (1,2,4,3,5-Triazadiborolidine, 4-(1,1-dimethylethyl)-1,2,3,5-tetramethyl-)	57877-83-5	**	7.45 (V)	PE	4526
B₂C₆H₁₈N₄⁺	B ₂ N ₄ (CH ₃) ₆ (1,2,4,5,3,6-Tetrazadiborine, hexahydro-1,2,3,4,5,6-hexamethyl-)	7318-93-6	**	6.83 (V)	PE	4299
B₂C₈H₂₄N₄⁺	((CH ₃) ₂ N)BB(N(CH ₃) ₂) ₂	1630-79-1	** **	7.3 (V) 7.58	PE	3512 3584
B₂C₇H₂₁N₅⁺	N ₃ B ₂ (CH ₃) ₃ (N(CH ₃) ₂) ₂ (1,2,4,3,5-Triazadiborolidine-3,5-diamine, N,N,N',N'-heptamethyl-)	53246-08-5	**	7.05 (V)	PE	4526
B₃C₈H₂₄N₅⁺	C ₈ H ₂₄ B ₃ N ₅ (Boranediamine, N,N,N',N'-tetramethyl-1-(2,3,4,5-tetramethyl-1,2,4,3,5-triazadiborolidin-1-yl)-)	53324-00-8	**	~7.29 (V)	PE	4526
	C ₈ H ₂₄ B ₃ N ₅ (Boranediamine, N,N,N',N'-tetramethyl-1-(1,2,3,5-tetramethyl-1,2,4,3,5-triazadiborolidin-4-yl)-)	53323-99-2	**	~7.2 (V)	PE	4526
	(RX N ₃ B ₂ (CH ₃) ₄ B(N(CH ₃) ₂) ₂					

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
B₂C₈H₂₄N₆⁺	B ₂ N ₄ (CH ₃) ₄ (N(CH ₃) ₂) ₂ (1,2,4,5,3,6-Tetrazadiborine-3,6-diamine, tetrahydro- <i>N,N,N',N'</i> ,1,2,4,5-octamethyl-)	54154-16-4	**	7.09 (V)	PE	4299
O⁺						
(¹ P ^e)	O	17778-80-2	**	14.040	S	5209
			**	13.618	PI	5000
(² P)			**	18.63	PE	3701
			**	14.0±0.5	EI	4436
			**	14.2±1	EI	4687
(⁴ S ^o)	O ₂	7782-44-7	O(³ P)	18.69±0.04	EI	4318
(⁴ S ^o)			O(¹ D)	20.52±0.05	EI	4318
(² D ^o)			O(³ P)	22.09±0.1	EI	4318
(³ P)	O ₃	10028-15-6	O ₂	15.21±0.1	PI	5004
	H ₂ O	7732-18-5	H ₂	19.0	EI	3967
			2H	26.8	EI	3967
(⁴ S)	CO	630-08-0	C(⁴ S)	23.44	EI	5126
	CO ₂	124-38-9		19.393±0.008	PI	4349
			CO	19±1	PI	5170
			CO	19.067	PE	5064
(⁴ S _u)			CO	19.071	PE	4886
			CO	19.05±0.05	EI	4693
				22.6±1.0	EI	4129
	NO	10102-43-9	N	20.1±0.3	EI	3945
	N ₂ O	10024-97-2	N ₂	15±1	PI	5170
	HOF	14034-79-8	HF	14.34	PI	3932
(⁴ S _u)	COS	463-58-1	CS ⁻	19.45±0.08	EI	4905
O⁺²						
O ^{+(²P)}		14581-93-2	**	30	EI	3489
			**	32	EI	3489
(¹ D)			**	38	EI	3489
(⁵ S)			**	42	EI	3489
CO	630-08-0	C(¹ D)		61	EI	3489
CO ⁺	12144-04-6	C(¹ D)		47	EI	3489
O₂⁺						
(² Π _{3/2g})	O ₂	7782-44-7	**	12.127	PE	4675
(² Π _g)			**	12.07±0.01	PI	4020
(² Π _{1/2})			**	12.071±0.001	PE	4491
(² Π _{g'})			**	12.071	PE	5064
(² Π _{g'})			**	12.076±0.002	PE	4770
(² Π _{3/2g})			**	12.077	PE	3834
(² Π _g)			**	12.08	PE	4073
(² Π _{1/2g})			**	12.102	PE	3834
(² Π _g)			**	12.33±0.01 (V)	PE	4415
(⁴ Π _u)			**	16.101	PE	5064
(⁴ πu)			**	16.105	PE	3664
(² Π _u)			**	16.5	PE	3698
(² Π _u)			**	17.15	PE	5064
(² Π _u)			**	~17.45	PE	3534
(² Φ _u)			**	17.5	PE	3698
(⁴ Σ _g)			**	18.171	PE	5064
(² Δ _g)			**	18.803±0.006	PE	4288
(² Δ _g)			**	18.81	PE	3534
(² Φ _u)			**	19.1±0.01	PE	5142
(² Δ _g)			**	19.9±0.01	PE	5142
(² Σ _g)			**	20.296	PE	5064
(² Π _u)			**	22.8±0.1	PE	3975
(⁴ Σ _g)			**	24.6	PE	3975
(⁴ Σ _g)			**	39.7 (V)	PE	4629
(² Σ _g)			**	40.33 (V)	PE	4629

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
O₂⁺						
	O ₂	7782-44-7	**	12.0±0.5	EI	4436
			**	12.2±0.2	EI	4131
			**	60.5±0.8	EI	5346
	O ₃	10028-15-6	O	13.125±0.004	PI	5004
O₂⁺²						
(³ P _g , ³ S _u) O ₂		7782-44-7	**	43.0±0.5	OTH	5007
(³ P _u)			**	48.0±1.0	OTH	5007
O₃⁺						
(² A ₁)	O ₃	10028-15-6	**	12.519±0.004	PI	5004
(² A ₁)			**	12.3±0.1?	PE	4539
(² A ₁)			**	12.44±0.01	PE	4239
(² A ₁)			**	12.53±0.1	PE	4170
(² A ₁)			**	12.56	PE	4169
(² A ₂)			**	13.02 (V)	PE	4169
(² B ₂)			**	13.02 (V)	PE	4239
(² A ₂)			**	13.03±0.02	PE	4170
(² B ₂)			**	13.57±0.01	PE	4170
(² B ₂)			**	13.57 (V)	PE	4169
(² A ₂)			**	13.57 (V)	PE	4239
(² A ₂)			**	15.57 (V)	PE	4239
(² B ₂)			**	16.54 (V)	PE	4239
(² A ₁)			**	17.45 (V)	PE	4239
(² B ₁ , ² B ₂ , ² A ₁ , ² B ₂)			**	19.99 (V)	PE	4239
(² B ₁ , ² B ₂)			**	20.3±0.1 (V)	PE	4170
(² A ₁)			**	-24.5 (V)	PE	4239
HO⁺						
(³ S ⁻)	OH	3352-57-6	**	13.01 (V)	PE	4773
(¹ Δ)			**	15.20 (V)	PE	4773
			**	13.5±1.0	EI	4054
			**	12.88	OTH	3932
H ₂ O		7732-18-5	H	18.115±0.008	PI	5146
			H	18.08±0.05	EI	5046
			H	18.2	EI	3967
HCOOH		64-18-6	HCO	17.97±0.06	PI	4177
HOF		14034-79-8	F	15.07	PI	3932
OD⁺						
D ₂ O		7789-20-0	D	18.219±0.008	PI	5146
			D	18.19±0.03	PE	4247
H₂O⁺						
H ₂ O		7732-18-5	**	12.612	S	5101
(² B ₁)			**	12.619±0.006	S	3983
(² B ₁)			**	12.6	PI	5479
(² A ₁)			**	13.8	PI	5479
(² B ₂)			**	17.2	PI	5479
			**	11.8 (V)	PE	4845
(² B ₁)			**	12.6	PE	4623
(² B ₂)			**	12.60±0.02 (V)	PE	4970
(² B ₁)			**	12.61 (V)	PE	4537
(² B ₁)			**	12.61 (V)	PE	4850
(² B ₁)			**	12.615±0.001	PE	4351
(² B ₁)			**	12.615±0.001	PE	5506
(² B ₁)			**	12.616	PE	5064
(² B ₁)			**	12.619	PE	3941
(² B ₁)			**	12.62	PE	3719
(² B ₁)			**	12.624	PE	3530
(² B ₁)			**	12.624	PE	4602

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
H₂O⁺						
	H ₂ O	7732-18-5	**	12.627	PE	5626
(² A ₁)			**	13.78	PE	3719
(² A ₁)			**	13.8	PE	4623
(² A ₁)			**	13.930±0.010	PE	3530
(² A ₁)			**	14.75±0.03 (V)	PE	4970
(² A ₁)			**	14.8	PE	3941
(² B ₂)			**	17.02	PE	3719
(² B ₂)			**	17.2	PE	4623
(² B ₂)			**	17.378±0.008	PE	4351
(² B ₂)			**	17.390	PE	3530
(² B ₂)			**	18.54	PE	3941
(² B ₁)			**	18.74±0.04 (V)	PE	4970
(² A ₁)			**	32.2	PE	4623
(² A ₁)			**	32.2 (V)	PE	3719
(² A ₁)			**	32.61±0.05 (V)	PE	4970
(² B ₁)			**	12.63±0.03	EI	5046
(² B ₁)			**	12.7	EI	3967
	H ₂ ¹⁸ O	14314-42-2	**	12.615±0.001	PE	5506
HDO⁺						
	HDO	14940-63-7	**	12.630	PE	5626
D₂O⁺						
(² B ₁)	D ₂ O	7789-20-0	**	12.636±0.006	S	3983
(² B ₁)			**	12.637	S	5101
(² B ₁)			**	12.633±0.001	PE	4351
			**	12.633±0.001	PE	5506
(² B ₁)			**	12.633	PE	3530
(² B ₁)			**	12.637	PE	4602
			**	12.639	PE	5626
(² A ₁)			**	13.930±0.010	PE	3530
(² B ₂)			**	17.412±0.008	PE	4351
(² B ₁)	D ₂ O		**	12.65±0.03	EI	5046
H₃O⁺						
	(H ₂ O) ₂	25655-83-8	OH	11.73±0.03	PI	5015
	C ₂ H ₅ OH	64-17-5		14.30±0.02	EI	3487
HO₂⁺						
	HO ₂	3170-83-0	**	11.67±0.15	EI	4920
H₂O₂⁺						
	H ₂ O ₂	7722-84-1	**	10.54	PE	4577
			**	11.69 (V)	PE	4168
H₄O₂⁺						
	(H ₂ O) ₂	25655-83-8	**	<11.21±0.09	PI	5015
LiO⁺						
	LiO	12142-77-7	**	8.45±0.20	EI	3909
Li₂O⁺						
	Li ₂ O	12057-24-8	**	6.19±0.20	EI	3909
BO⁺						
	BO	12505-77-0	**	13.2±0.2	EI	4483
	BO	13840-87-4	**	13.0±0.5	EI	3473
BO₂⁺						
	BO ₂	13840-88-5	**	14.0±1.0	EI	4054

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
HBO₂⁺	BHO ₂	13460-50-9	**	13.5±1.0	EI	4054
CO⁺						
(² Σ ⁺)	CO	630-08-0	**	11.3969	S	5167
(² Π _{1/2})			**	14.014	S	3760
(² Σ ⁺)			**	16.550	S	3760
(² Σ _{2p})			**	19.672	S	3760
(² Π _{2p})			**	14.0	PI	5479
(² Σ _{2s})			**	16.5	PI	5479
(² Σ ⁺)			**	19.7	PI	5479
(² Σ _{2p})			**	14.01	PE	4073
(² Σ ⁺)			**	14.01 (V)	PE	4022
(² Π)			**	14.01 (V)	PE	5055
(² Π)			**	16.55	PE	4073
(² Σ ⁺)			**	16.91 (V)	PE	4022
(² Σ ⁺)			**	19.69 (V)	PE	3714
(² Σ _{2s})			**	19.72 (V)	PE	4022
(² Σ ⁺)			**	39.0	PE	3975
(² Σ ⁺)			**	39.7 (V)	PE	4615
(² Σ ⁺)			**	14.07±0.05	EI	4958
(² Σ ⁺)	CO ₂	124-38-9	O(³ S)	29.0	PI	4095
			0	19±2	PI	5170
			0	19.466	PE	4886
			0	19.466	PE	5064
(² Σ ⁺)			0	21.433	PE	4886
(² Π)			0	21.976	PE	4886
			0	19.42±0.075	EI	4693
				20.9±1.0	EI	4129
COS		463-58-1	S?	15.6	EI	3779
CO⁺²						
CO₂⁺	CO	630-08-0	**	41.8±0.5	EI	4958
(² Π _g)	CO ₂	124-38-9	**	13.77	PI	4932
(X ² Π _{3/2g})			**	13.773±0.002	PI	3925
(² Π _{3/2g})			**	13.774±0.003	PI	4349
(X ² Π _{3/2g})			**	13.776±0.008	PI	4069
(² Π _{1/2g})			**	13.788±0.003	PI	4349
(² Σ _g ⁺)			**	19.391±0.001	PI	4886
			**	36.2	PI	5127
			**	13±1	PI	5170
			**	35±3	PI	5170
(² Π _{3/2g})			**	13.776±0.002	PE	4910
(² Π _{1/2g})			**	13.797±0.002	PE	4910
(² Π _{3/2u} , ² Π _{1/2u})			**	17.316±0.003	PE	4910
(² Σ _u ⁺)			**	18.076±0.002	PE	4910
(² Σ _g ⁺)			**	19.395±0.003	PE	4910
(² Π _g)			**	13.773 (V)	PE	4886
			**	13.776±0.002	PE	5256
(² Π _g)			**	13.777±0.002	PE	5132
(² Π _g)			**	13.78	PE	4073
(² Π _g)			**	13.78 (V)	PE	4850
(² Π _g)			**	13.788	PE	5064
(² Π _g)			**	13.79 (V)	PE	5055
(² Π _g)			**	13.80±0.01	PE	3965
(² Π _g)			**	17.31 (V)	PE	4886
(² Π _g)			**	17.311±0.002	PE	5132
(² Π _g)			**	17.34±0.01	PE	3965
(² Σ _u)			**	18.068±0.002	PE	5132
(² Σ _g ⁺)			**	18.07 (V)	PE	4886

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
CO₂⁺						
(² S _u ⁺)	CO ₂	124-38-9	**	18.08±0.01	PE	3965
(² S _g)			**	19.386±0.002	PE	5132
(² S _g ⁺)			**	19.39±0.01	PE	3965
(² S _u)			**	37	PE	4095
(² S _g)			**	38.4	PE	4095
			**	13.79±0.05	EI	5240
			**	13.83±0.05	EI	4693
			**	13.89±0.03	EI	4877
			**	13.92±0.2	EI	5588
			**	37.2±0.5	EI	5240
C₃O₂⁺						
	C ₃ O ₂	504-64-3	**	10.605	PE	3728
CHO⁺						
(² A')	HCO	17030-74-9	**	8.55±0.01	PE	5008
	HCHO	50-00-0	H	11.89±0.03	PI	3554
	CH ₃ OH	67-56-1	H ₂ +H	13.06±0.10	PI	3554
	CD ₃ OH	1849-29-2	D ₂ +D	13.8±0.6	EI	5173
	CH ₃ CHO	75-07-0	CH ₃	11.79±0.03	PI	4177
	CH ₃ CDO	4122-13-8		12.67	PI	5270
	CD ₃ CHO	19901-15-6		11.98	PI	5270
			CH ₃	12.03±0.03	PI	4350
				12.03	PI	5270
	C ₂ H ₄ O (Oxirane)	75-21-8	CH ₃	11.54±0.03	PI	4350
	(CH ₃) ₂ O	115-10-6		13.96±0.2	EI	4071
	CH ₃ OCD ₃	13725-27-4		13.97±0.2	EI	4071
	C ₂ H ₅ OCD ₃	16995-14-5		13.13±0.2	EI	4071
	C ₄ H ₄ O (Furan)	110-00-9	C ₃ H ₃	13.2±0.1	PE	5289
	HCOOH	64-18-6	OH	12.79±0.03	PI	4177
			OH	13.0±0.1	PI	5135
	HNCO	420-05-3	N	15.52	EI	4507
	HCONH ₂	75-12-7		13.70	EI	4878
	HCONNHCH ₃	123-39-7		12.40	EI	4878
	HCON(CH ₃) ₂	68-12-2		14.50	EI	4878
CDO⁺						
	DCO	15233-68-8	**	8.56±0.01	PE	5008
	CD ₃ OH	1849-29-2	D ₂ +H	13.53±0.5	EI	5173
	CD ₃ OD	811-98-3		14.88	PI	5174
	CH ₃ CDO	4122-13-8		11.95	PI	5270
	CD ₃ CHO	19901-15-6		12.65	PI	5270
	CH ₃ OCD ₃	13725-27-4		13.87±0.2	EI	4071
	C ₂ H ₅ OCD ₃	16995-14-5		13.57±0.2	EI	4071
CH₂O⁺						
	H ₂ CO	50-00-0	**	10.874±0.002	S	5071
			**	10.88±0.02	PI	3554
			**	10.90±0.03	PI	3765
			**	10.1 (V)	PE	4467
			**	10.885±0.005	PE	5519
	CH ₃ OH	67-56-1	H ₂	12.05±0.12	PI	3554
CHDO⁺						
	CD ₃ OH	1849-29-2	D ₂	12.78±0.3	EI	5173
CD₂O⁺						
	D ₂ CO	XXXXX-XX-X	**	10.901±0.006	S	5071
	CD ₃ OH	1849-29-2	HD	12.28±0.4	EI	5173

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
CH₃O⁺						
	CH ₃ OH	67-56-1	H	11.55±0.03	PI	3554
			H	11.69	EI	4915
			H	11.76±0.11	EI	5503
	(CH ₃) ₂ O	115-10-6	CH ₃	<11.8	EI	4915
			CH ₃	12.42±0.1	EI	4071
	C ₂ H ₅ OH	13725-27-4	CH ₃	11.30	EI	4915
	C ₂ H ₅ OCH ₃	540-67-0	C ₂ H ₅	<11.7	EI	4915
				12.86±0.1	EI	4071
	n-C ₃ H ₇ OH	71-23-8	C ₂ H ₅	11.16±0.03	EI	3626
	(C ₂ H ₅) ₂ O	60-29-7		11.92	EI	5072
	C ₂ H ₅ CH(OH)CH ₃	78-92-2		12.40	EI	5072
	((CH ₃) ₂ C(NO)OOCCH ₃) ₂	68777-98-0		14.20	EI	4809
CHD₂O⁺						
	CD ₃ OH	1849-29-2	D	11.30±0.3	EI	5173
	C ₂ H ₅ OCD ₃	16995-14-5		12.86±0.05	EI	4071
CD₃O⁺						
	CD ₃ OH	1849-29-2	H	11.40±0.5	EI	5173
	CD ₃ OD	811-98-3	D	12.71	PI	5174
CH₄O⁺						
	CH ₃ OH	67-56-1	**	10.83±0.03	PI	3554
			**	10.85±0.01	PI	4957
			**	10.846±0.002	PE	4770
			**	10.86 (V)	PE	4850
			**	10.94 (V)	PE	4068
			**	10.95	PE	4087
			**	10.95 (V)	PE	4032
			**	10.95 (V)	PE	4884
			**	10.95 (V)	PE	5249
			**	10.96 (V)	PE	3941
			**	10.97±0.03 (V)	PE	4484
			**	10.90±0.03	EI	4877
			**	10.90±0.12	EI	5503
CH₄O²⁺						
	CH ₃ OH	67-56-1	**	33.2±0.5	OTH	5147
CH₃DO⁺						
	CH ₃ OD	4206-31-9	**	10.861±0.002	PE	4770
CHD₃O⁺						
	CD ₃ OH	1849-29-2	**	10.84±0.1	EI	5173
CD₄O⁺						
	CD ₃ OD	811-98-3	**	11.00	PI	5174
			**	10.885±0.002	PE	4770
C₂H₂O⁺						
	CH ₂ =C=O	463-51-4	**	9.614±0.008	PI	5458
			**	9.60 (V)	PE	5610
			**	9.61±0.02	PE	5458
			**	9.63±0.02	PE	5211
	CH ₃ CHO	75-07-0	H ₂ ?	13.06±0.09	PI	4350
	C ₂ H ₄ O (Oxirane)	75-21-8	H ₂ ?	13.07±0.05	PI	4350
	C ₃ H ₄ (=O) (Cyclopropanone)	5009-27-8		9.9±0.1	EI	4689
	C ₄ H ₄ O (Furan)	110-00-9	C ₂ H ₂	11.80±0.10	PE	5289

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₂H₂O⁺						
	C ₄ H ₆ O (Cyclobutanone)	1191-95-3	C ₂ H ₄	10.53±0.15	EI	3794
	C ₃ H ₃ NO (Oxazole)	288-42-6	HCN	12.15±0.6	EI	5400
C₂H₄O⁺						
	CH ₃ CHO	75-07-0	H	10.82±0.03	PI	4177
			H	10.90±0.03	PI	4350
			H	10.90	PI	5270
	C ₂ H ₄ O (Oxirane)	75-21-8	H	11.62±0.05	PI	4350
	CH ₃ CDO	4122-13-8	D	10.92	PI	5270
	(CH ₃) ₂ CO	67-64-1	CH ₃	10.52±0.02	PI	5412
			CH ₃	12.22	PE	5066
			CH ₃	10.28±0.05	EI	3626
				10.30	EI	4535
			CH ₃	11.3	EI	3550
	CH ₃ COCH=CH	1423-60-5	C ₂ H	12.10±0.10	PE	5289
	C ₂ H ₅ COCH ₃	78-93-3		10.69	EI	4535
	iso-C ₃ H ₇ COCH ₃	563-80-4		10.68	EI	4535
	tert-C ₄ H ₉ COCH ₃	75-97-8		~11.3	EI	4535
	CH ₃ COOCH ₃	79-20-9		10.94	EI	5070
	C ₆ H ₅ OOCCH ₃ (Acetic acid, phenyl ester)	122-79-2	cyclo-C ₆ H ₅ O	12.78±0.2	EI	3484
			C ₆ H ₅ O	12.83±0.03	EI	3483
	C ₆ H ₄ (CH ₃)OOCCH ₃ (Acetic acid, 3-methylphenyl ester)	122-46-3	C ₆ H ₄ (CH ₃)O	13.83±0.2	EI	3484
	C ₆ H ₄ (CH ₃)OOCCH ₃ (Acetic acid, 4-methylphenyl ester)	140-39-6		13.97±0.2	EI	3484
	C ₆ H ₅ CH ₂ CH ₂ OOCCH ₃ (Acetic acid, 2-phenylethyl ester)	103-45-7		11.70	EI	3590
	C ₆ H ₄ (CH ₃)CH ₂ CH ₂ OOCCH ₃ (Phenethyl alcohol, <i>m</i> -methyl-, acetate)	33709-40-9		11.90	EI	3590
	C ₆ H ₄ (CH ₃)CH ₂ CH ₂ OOCCH ₃ (Phenethyl alcohol, <i>p</i> -methyl-, acetate)	22532-47-4		11.90	EI	3590
	C ₆ H ₄ (OCH ₃)OOCCH ₃ (Phenol, 3-methoxy-, acetate)	5451-83-2	C ₆ H ₄ (OCH ₃)O	13.92±0.2	EI	3484
	C ₆ H ₄ (OCH ₃)OOCCH ₃ (Phenol, 4-methoxy-, acetate)	1200-06-2	C ₆ H ₄ (OCH ₃)O	14.57±0.2	EI	3484
	C ₆ H ₄ (OCH ₃)CH ₂ CH ₂ OOCCH ₃ (Phenethyl alcohol, <i>m</i> -methoxy-, acetate)	33709-39-6		11.80	EI	3590
	C ₆ H ₄ (OCH ₃)CH ₂ CH ₂ OOCCH ₃ (Phenethyl alcohol, <i>p</i> -methoxy-, acetate)	22532-51-0		12.20	EI	3590
	C ₆ H ₄ (COOH)OOCCH ₃ (Benzoinic acid, 4-(acetyloxy)-)	2345-34-8	C ₆ H ₄ (COOH)O	12.46±0.2	EI	3484
	CH ₃ CONH ₂	60-35-5		11.70	EI	4878
	CH ₃ CONHCH ₃	79-16-3		12.40	EI	4878
	CH ₃ CON(CH ₃) ₂	127-19-5		12.55	EI	4878
	C ₅ H ₆ NCOCH ₃ (Pyridine, 1-acetyl-1,2,3,4-tetrahydro-)	19615-27-1		13.5	EI	4046
	C ₅ H ₁₀ NCOCH ₃ (Piperidine, 1-acetyl-)	618-42-8		15.1	EI	4046
	C ₆ H ₅ NHCOCH ₃ (Acetamide, <i>N</i> -phenyl-)	103-84-4		13.22±0.03	EI	3483
	C ₆ H ₄ (NH ₂)CH ₂ CH ₂ OOCCH ₃ (Benzeneethanol, 4-amino-, acetate(ester))	33709-38-5		12.30	EI	3590
	C ₆ H ₄ (NO ₂)OOCCH ₃ (Acetic acid, 3-nitrophenyl ester)	1523-06-4		10.94±0.2	EI	3484
	C ₆ H ₄ (NO ₂)OOCCH ₃ (Acetic acid, 4-nitrophenyl ester)	830-03-5		10.85±0.2	EI	3484
	((CH ₃) ₂ C(NO)COCH ₃) ₂	30442-79-6		11.60	EI	4809

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₂H₃O⁺						
	(C ₆ H ₁₁ NO ₂) ₂	68777-99-1		10.40	EI	4809
	((CH ₃) ₂ C(NO)OOCCH ₃) ₂	68777-98-0		10.20	EI	4809
	C ₆ H ₄ FOOCCH ₃	29650-44-0	C ₆ H ₄ FO	12.23±0.03	EI	3483
	(Phenol, 2-fluoro-, acetate)					
	C ₆ H ₄ FOOCCH ₃	405-51-6	C ₆ H ₄ FO	12.72±0.03	EI	3483
	(Phenol, 4-fluoro-, acetate)					
	C ₆ H ₃ F ₂ OOCCH ₃	36914-77-9		12.00±0.03	EI	3480
	(Phenol, 2,4-difluoro-, acetate)					
	C ₆ H ₃ F ₂ OOCCH ₃	36914-78-0		12.24±0.03	EI	3480
	(Phenol, 2,6-difluoro-, acetate)					
	CH ₃ COCF ₃	421-50-1		11.45	EI	3550
	C ₆ H ₄ FNHCOCH ₃	399-31-5		13.59±0.03	EI	3483
	(Acetamide, N-(2-fluorophenyl)-)					
	C ₆ H ₄ FNHCOCH ₃	351-83-7		13.42±0.03	EI	3483
	(Acetamide, N-(4-fluorophenyl)-)					
	C ₆ H ₃ F ₂ NHCOC ₃	399-36-0		13.18±0.03	EI	3480
	(Acetamide, N-(2,4-difluorophenyl)-)					
	C ₆ H ₃ F ₂ NHCOC ₃	3869-29-5		13.80±0.03	EI	3480
	(Acetamide, N-(2,6-difluorophenyl)-)					
	C ₆ H ₄ ClOOCCH ₃	4525-75-1		12.55±0.03	EI	3483
	(Acetic acid, 2-chlorophenyl ester)					
	C ₆ H ₄ ClOOCCH ₃	13031-39-5		12.36±0.2	EI	3484
	(Acetic acid, 3-chlorophenyl ester)					
	C ₆ H ₄ ClOOCCH ₃	876-27-7		12.39±0.03	EI	3483
	(Acetic acid, 4-chlorophenyl ester)					
				12.73±0.2	EI	3484
	C ₆ H ₄ ClCH ₂ CH ₂ OOCCH ₃	33709-41-0		11.60	EI	3590
	(Phenethyl alcohol, m-chloro-, acetate)					
	C ₆ H ₃ Cl ₂ OOCCH ₃	6341-97-5		12.11±0.03	EI	3480
	(Phenol, 2,4-dichloro-, acetate)					
	C ₆ H ₃ Cl ₂ OOCCH ₃	28165-71-1		12.09±0.03	EI	3480
	(Phenol, 2,6-dichloro-, acetate)					
	C ₆ H ₄ CINHCOCH ₃	533-17-5		13.91±0.03	EI	3483
	(Acetamide, N-(2-chlorophenyl)-)					
	C ₆ H ₄ CINHCOCH ₃	539-03-7		13.00±0.03	EI	3483
	(Acetamide, N-(4-chlorophenyl)-)					
	C ₆ H ₃ Cl ₂ NHCOC ₃	6975-29-7		13.08±0.03	EI	3480
	(Acetamide, N-(2,4-dichlorophenyl)-)					
	C ₆ H ₃ Cl ₂ NHCOC ₃	17700-54-8		13.40±0.03	EI	3480
	(Acetamide, N-(2,6-dichlorophenyl)-)					
	C ₆ H ₄ BrCOOCH ₃	1829-37-4		12.24±0.03	EI	3483
	(Phenol, 2-bromo-, acetate)					
	C ₆ H ₄ BrOOCCH ₃	35065-86-2		12.36±0.2	EI	3484
	(Phenol, 3-bromo-, acetate)					
	C ₆ H ₄ BrOOCCH ₃	1927-95-3		12.87±0.2	EI	3484
	(Phenol, 4-bromo-, acetate)					
				13.06±0.03	EI	3483
	C ₆ H ₃ Br ₂ OOCCH ₃	36914-79-1		12.01±0.03	EI	3480
	(Phenol, 2,4-dibromo-, acetate)					
	C ₆ H ₃ Br ₂ OOCCH ₃	28165-72-2		12.36±0.03	EI	3480
	(Phenol, 2,6-dibromo-, acetate)					
	C ₆ H ₄ BrNHCOCH ₃	614-76-6		14.68±0.03	EI	3483
	(Acetamide, N-(2-bromophenyl)-)					
	C ₆ H ₄ BrNHCOCH ₃	103-88-8		13.96±0.03	EI	3483
	(Acetamide, N-(4-bromophenyl)-)					
	C ₆ H ₃ Br ₂ NHCOC ₃	23373-04-8		13.10±0.03	EI	3480
	(Acetamide, N-(2,4-dibromophenyl)-)					
	C ₆ H ₃ Br ₂ NHCOC ₃	33098-80-5		13.21±0.03	EI	3480
	(Acetamide, N-(2,6-dibromophenyl)-)					
	C ₆ H ₄ IOOCCH ₃	32865-61-5	C ₆ H ₄ IO	12.47±0.03	EI	3483
	(Phenol, 2-iodo-, acetate)					

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₂H₃O⁺						
	C ₆ H ₄ IOOCCH ₃ (Phenol, 4-iodo-, acetate)	33527-94-5	C ₆ H ₄ IO	12.74±0.03	EI	3483
	C ₆ H ₃ I ₂ OOCCH ₃ (Phenol, 2,4-diiodo-, acetate)	36914-80-4		12.15±0.03	EI	3480
	C ₆ H ₃ I ₂ OOCCH ₃ (Phenol, 2,6-diiodo-, acetate)	28165-73-3		12.02±0.03	EI	3480
	C ₆ H ₄ INHCOCH ₃ (Acetamide, N-(2-iodophenyl)-)	19591-17-4		13.56±0.03	EI	3483
	C ₆ H ₄ INHCOCH ₃ (Acetamide, N-(4-iodophenyl)-)	622-50-4		13.16±0.03	EI	3483
C₂D₃O⁺						
	CD ₃ CHO	19901-15-6	H	10.91	PI	5270
	(CD ₃) ₂ CO	666-52-4	CD ₃	10.56±0.02	PI	5412
C₂H₄O⁺						
	C ₄ H ₇ (OH) (Cyclobutanol)	2919-23-5	C ₂ H ₄	9.87	EI	4729
	C ₂ H ₃ O(CH ₂ OH) (Oxiranemethanol)	556-52-5	CH ₂ O	10.30	EI	4729
	CH ₃ CHO	75-07-0	**	10.19	S	5273
			**	10.20±0.02	PI	4177
			**	10.20±0.03	PI	3765
			**	10.22±0.01	PI	4350
			**	10.22	PI	5270
			**	10.2298±0.0007	PI	4306
			**	10.20	PE	4471
			**	10.20	PE	4520
			**	10.21	PE	4224
			**	10.227±0.005	PE	5519
			**	10.23 (V)	PE	4850
			**	10.24±0.02	PE	4220
			**	10.26 (V)	PE	4513
			**	10.9 (V)	PE	4467
			**	10.23	EI	4729
	CH ₂ =CHOH	557-75-5	**	9.0±0.15	OTH	4729
	C ₂ H ₄ O (Oxirane)	75-21-8	**	10.558±0.1	PI	4868
			**	10.56±0.01	PI	4350
			**	10.4±0.1	PE	4990
			**	10.560	PE	4868
			**	10.568 (V)	PE	4527
			**	10.57	EI	4729
	C ₂ H ₅ OH	64-17-5	H ₂	~10.45	EI	4729
	iso-C ₂ H ₇ OH	67-63-0	CH ₄	10.23±0.02	PI	5512
	CH ₂ =CHOC ₂ H ₅	109-92-2	C ₂ H ₄	10.19	EI	4729
	n-C ₃ H ₇ CHO	123-72-8	C ₂ H ₄	10.52	EI	4729
	n-C ₄ H ₉ CHO	110-62-3	C ₃ H ₆	11.40	EI	5264
	iso-C ₄ H ₉ CHO	26140-47-6	C ₃ H ₆	10.57	EI	4729
	(CH ₃) ₂ CHC ₂ H ₄ CHO	1119-16-0	C ₄ H ₈	11.40	EI	5264
	C ₂ H ₅ CH(CH ₃)CH ₂ CHO	15877-57-3	2-C ₄ H ₈	10.88	EI	4729
			C ₄ H ₈	11.40	EI	5264
	n-C ₅ H ₁₁ CHO	66-25-1	C ₄ H ₈	11.60	EI	5264
	n-C ₆ H ₁₃ OH	111-27-3		~10.7	EI	4729
	C ₃ H ₆ O ₂ (1,3-dioxolane)	646-06-0	CH ₂ O	10.87	EI	4729
C₂H₃DO⁺						
	CH ₃ CDO	4122-13-8	**	10.21	PI	5270

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_2\text{HD}_3\text{O}^+$	CD_3CHO	19901-15-6	**	10.19	PI	5270
$\text{C}_2\text{D}_4\text{O}^+$	$\text{C}_2\text{D}_4\text{O}$ (Oxirane- d_4)	6552-57-4	**	10.571	PE	4868
$\text{C}_2\text{H}_5\text{O}^+$						
	CH_3OCH_2	23653-97-6	**	6.94	EI	4915
	$\text{C}_2\text{H}_5\text{OH}$	64-17-5	H	10.75 ± 0.03	EI	5467
			H	10.67	EI	4915
	$(\text{CH}_3)_2\text{O}$	115-10-6	H	10.99	EI	4915
			H	10.70 ± 0.13	EI	5503
			H	11.23 ± 0.04	EI	5467
			H	11.55 ± 0.15	EI	4071
	$\text{C}_2\text{H}_5\text{OCH}_3$	540-67-0	CH_3	10.47	EI	4915
			CH_3	10.91 ± 0.1	EI	4071
	$n\text{-C}_3\text{H}_7\text{OH}$	71-23-8	CH_3	11.35 ± 0.04	EI	5467
	$iso\text{-C}_3\text{H}_7\text{OH}$	67-63-0	CH_3	10.40 ± 0.03	PI	5512
			CH_3	10.26	EI	4915
	$(\text{C}_2\text{H}_5)_2\text{O}$	60-29-7	C_2H_5	11.85	EI	4915
				11.83	EI	4603
	$\text{C}_2\text{H}_5\text{CH}(\text{OH})\text{CH}_3$	78-92-2	C_2H_5	10.22	EI	4915
	$\text{CH}_3\text{CD}_2\text{OC}_2\text{H}_5$	XXXXX-XX-X		11.71	EI	4603
	$\text{CH}_3\text{CH}(\text{OH})\text{CH}_2\text{OH}$	57-55-6	CH_2OH	10.25	EI	4915
	$\text{CH}_3\text{OCH}_2\text{CH}_2\text{OH}$	109-86-4	CH_2OH	10.36	EI	4915
	$\text{CH}_3\text{OCH}_2\text{CH}_2\text{OCH}_3$	110-71-4	CH_3OCH_2	10.27	EI	4915
	$\text{C}_2\text{H}_5\text{ONO}$	79-24-3	NO	10.62 ± 0.07	EI	5467
	$((\text{CH}_3)_2\text{C}(\text{NO})\text{OOCCH}_3)_2$	68777-98-0		12.75	EI	4809
	$\text{CH}_2\text{BrCH}_2\text{OH}$	540-51-2	Br	10.47 ± 0.05	EI	5467
$\text{C}_2\text{H}_4\text{DO}^+$	$\text{CH}_3\text{CD}_2\text{OC}_2\text{H}_5$	XXXXX-XX-X		11.91	EI	4603
$\text{C}_2\text{H}_3\text{D}_2\text{O}^+$						
	CH_3OCD_3	13725-27-4	D	11.53 ± 0.1	EI	4071
	$\text{CH}_3\text{CD}_2\text{OC}_2\text{H}_5$	XXXXX-XX-X		13.1	EI	4603
$\text{C}_2\text{H}_2\text{D}_3\text{O}^+$						
	CH_3OCD_3	13725-27-4	H	11.15 ± 0.1	EI	4071
	$\text{C}_2\text{H}_5\text{OCD}_3$	16995-14-5	CH_3	10.41 ± 0.06	EI	5503
			CH_3	11.01 ± 0.1	EI	4071
$\text{C}_2\text{H}_6\text{O}^+$						
	$\text{C}_2\text{H}_5\text{OH}$	64-17-5	**	10.59 (V)	PE	5514
			**	10.61 (V)	PE	4850
			**	10.62 (V)	PE	3941
			**	10.64 (V)	PE	4068
			**	10.64 (V)	PE	5249
			**	10.65 ± 0.03 (V)	PE	4484
			**	10.65 (V)	PE	5088
	$(\text{CH}_3)_2\text{O}$	115-10-6	**	9.8±0.1	PE	4990
			**	9.98 (V)	PE	4850
			**	10.0 ± 0.2 (V)	PE	4774
			**	10.03 (V)	PE	4884
			**	10.04 (V)	PE	3656
			**	10.04 (V)	PE	3844
			**	10.052 (V)	PE	4527
			**	11.94 (V)	PE	5249
			**	10.12 ± 0.2	EI	4071

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₂H₃D₃O⁺						
	CH ₃ OCD ₃	13725-27-4	**	10.00±0.1	EI	4071
C₃HO⁺						
	CH ₃ COC≡CH	1423-60-5	CH ₃	11.00±0.10	PE	5289
C₃H₂O⁺						
	C ₃ H ₂ (=O) (2-Cyclopropen-1-one)	2961-80-0	**	9.47	PE	4270
			**	10.0±0.3	EI	4689
C₃H₃O⁺						
	CH ₂ =CHCOCH ₃ (C ₆ H ₁₁ NO ₂) ₂	78-94-4 68777-99-1	CH ₃	10.44±0.05 13.25	EI EI	5445 4809
C₃H₄O⁺						
	CH ₂ =CHCHO	107-02-8	**	10.13	PE	3864
			**	10.15 (V)	PE	4195
	CH≡CCH ₂ OH	107-19-7	**	10.45 (V)	PE	4847
	CH ₃ CH=C=O	6004-44-0	**	8.95 (V)	PE	5610
	C ₃ H ₄ (=O) (Cyclopropanone)	5009-27-8	**	9.1±0.1	EI	4689
	CH ₂ =CHCHO (2-Propenal)	107-02-8	**	10.11	PE	5360
C₃H₅O⁺						
	C ₂ H ₅ COCH ₃ (C ₂ H ₅) ₂ CO (C ₆ H ₁₁ NO ₂) ₂	123-72-8 96-22-0 68777-99-1		10.22 10.10 10.75	EI EI EI	4535 4535 4809
C₃H₆O⁺						
	(CH ₃) ₂ CO	67-64-1	**	9.705	S	5006
			**	9.71	S	5273
			**	9.694±0.006	PI	5412
			**	9.700±0.001	PI	4306
			**	9.71±0.03	PI	3765
			**	9.5 (V)	PE	4467
			**	9.68 (V)	PE	4850
			**	9.70 (V)	PE	4513
			**	9.709±0.005	PE	5519
			**	9.709 (V)	PE	4527
			**	9.71±0.01	PE	4535
			**	9.71±0.02 (V)	PE	4524
			**	9.71	PE	4224
			**	9.71 (V)	PE	4233
			**	9.72	PE	3649
			**	9.72 (V)	PE	4285
			**	9.72 (V)	PE	5538
			**	9.75±0.025	PE	3626
			**	9.71±0.03	EI	4535
			**	9.74	EI	3485
	CH ₂ =CHCH ₂ OH	107-18-6	**	9.63	PE	3864
			**	10.22 (V)	PE	3863
	CH ₂ =CHOCH ₃	107-25-5	**	8.95	PE	3863
			**	8.96	PE	4246
			**	9.05 (V)	PE	4291
	C ₂ H ₅ CHO	123-38-6	**	9.85 (V)	PE	4513
			**	9.953±0.005	PE	5519
			**	9.96 (V)	PE	4850
			**	9.99	PE	4224
	C ₃ H ₆ O (Oxetane)	503-30-0	**	9.63	PE	3980
				9.679 (V)	PE	4527

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₃H₆O⁺						
	C ₂ H ₃ OCH ₃ (Oxirane, methyl-)	75-56-9	**	10.26 (V)	PE	4747
	n-C ₃ H ₇ COCH ₃	107-87-9	C ₂ H ₄	10.08	EI	5039
	n-C ₄ H ₉ CHO	110-62-3	C ₂ H ₄	9.82	EI	5039
			C ₂ H ₄	10.00	EI	5264
	C ₅ H ₉ OH (Cyclopentanol)	96-41-3	C ₂ H ₄	9.98	EI	5039
	(CH ₃) ₂ CHC ₂ H ₄ CHO	1119-16-0	C ₃ H ₆	11.00	EI	5264
	C ₂ H ₅ CH(CH ₃)CH ₂ CHO	15877-57-3	C ₃ H ₆	10.10	EI	5264
	n-C ₅ H ₁₁ CHO	66-25-1	C ₃ H ₆	9.72	EI	5039
			C ₃ H ₆	10.20	EI	5264
	n-C ₃ H ₇ CH(CH ₃)CHO	123-15-9	C ₃ H ₆	9.80	EI	5039
	n-C ₄ H ₉ COCH ₃	591-78-6	C ₃ H ₆	10.04	EI	5039
	sec-C ₅ H ₁₁ CHO	123-15-9	C ₃ H ₆	10.30	EI	5264
	iso-C ₄ H ₉ COCH ₃	108-10-1	C ₃ H ₆	9.98	EI	5039
C₃D₆O⁺						
	(CD ₃) ₂ CO	666-52-4	**	9.695±0.006	PI	5412
			**	9.68	PE	3649
C₃H₇O⁺						
	CH ₃ CHOCH ₃	20615-69-4	**	<6.50	EI	4915
	C ₂ H ₅ OCH ₃	540-67-0	H	10.32	EI	4915
			H	10.32±0.1	EI	4071
	n-C ₃ H ₇ OH	71-23-8	H	10.2	EI	3916
			H	10.48±0.03	EI	3626
	iso-C ₃ H ₇ OH	67-63-0	H	10.3±0.5	PI	5512
			H	<10.48	EI	4915
	(C ₂ H ₅) ₂ O	60-29-7	CH ₃	10.26	EI	4915
	iso-C ₃ H ₇ OCH ₃	598-53-8	CH ₃	9.82	EI	4915
	tert-C ₄ H ₉ OH	75-65-0	CH ₃	9.86	EI	4915
				10.1±0.2	EI	4124
	n-C ₃ H ₇ (CH ₃)OH	71-23-8	CH ₃	10.18	EI	4915
	tert-C ₅ H ₁₁ OH	75-85-4	C ₂ H ₅	9.80	EI	4915
	C ₂ H ₅ OCH ₂ CH ₂ OH	110-80-5	CH ₂ OH	10.26	EI	4915
	CH ₃ OCH(CH ₃)CH ₂ OH	1589-47-5	CH ₂ OH	9.68	EI	4915
C₃H₄D₅O⁺						
	C ₂ H ₅ OCD ₃	16995-14-5	H	10.22±0.1	EI	4071
C₃H₈O⁺						
	C ₂ H ₅ OCH ₃	540-67-0	**	9.72 (V)	PE	5088
			**	9.62±0.1	EI	4071
	n-C ₃ H ₇ OH	71-23-8	**	10.15±0.025	PE	3626
			**	10.49 (V)	PE	4068
			**	10.51 (V)	PE	3941
			**	10.51 (V)	PE	4850
			**	10.52±0.03 (V)	PE	4484
			**	10.0	EI	3916
			**	10.16±0.03	EI	3626
	iso-C ₃ H ₇ OH	67-63-0	**	10.10±0.02	PI	5512
			**	10.36 (V)	PE	4068
			**	10.42 (V)	PE	3941
			**	10.44 (V)	PE	4850
			**	10.49±0.03 (V)	PE	4484
C₃H₅D₅O⁺						
	C ₂ H ₅ OCD ₃	16995-14-5	**	9.64±0.1	EI	4071
C₄H₄O⁺						
	CH ₃ COC=CH	1423-60-5	**	10.19	PE	5289

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_6\text{H}_6\text{O}^+$						
	$\text{C}_6\text{H}_5\text{OC}_2\text{H}_5$	103-73-1	C_2H_4	11.3	EI	3479
	$\text{C}_7\text{H}_6\text{O}_2$ (2,4,6-Cycloheptatrien-1-one, 2-hydroxy-)	533-75-5	CO	10.8	EI	3479
	$\text{C}_6\text{H}_4(\text{OH})\text{OCH}_3$ (Phenol, 4-methoxy-)	150-76-5	HCHO	10.30	EI	3845
	$\text{C}_6\text{H}_5\text{OOCCH}_3$ (Acetic acid, phenyl ester)	122-79-2	$\text{CH}_2=\text{C=O}$	9.57 ± 0.03	EI	3483
			$\text{CH}_2=\text{C=O}$	9.89 ± 0.2	EI	3484
	$\text{C}_6\text{H}_5\text{OCH}_2\text{CH}_2\text{F}$ (Benzene, 2-fluoroethoxy-)	405-97-0	$\text{C}_2\text{H}_3\text{F}$	11.18	EI	5083
	$\text{C}_6\text{H}_5\text{OCH}_2\text{CH}_2\text{Cl}$ (Benzene, 2-chloroethoxy-)	622-86-6	$\text{C}_2\text{H}_3\text{Cl}$	10.80	EI	5083
	$\text{C}_6\text{H}_5\text{OCH}_2\text{CH}_2\text{Br}$ (Benzene, 2-bromoethoxy-)	589-10-6	$\text{C}_2\text{H}_3\text{Br}$	9.71	EI	5083
$\text{C}_6\text{H}_8\text{O}^+$						
	$\text{CH}_3(\text{CH}=\text{CH})_2\text{CHO}$	142-83-6	**	9.22 ± 0.03 (V)	PE	4767
	$\text{C}_6\text{H}_8\text{O}$ (2-Cyclohexen-1-one)	930-68-7	**	9.20 (V)	PE	4195
			**	9.23 ± 0.05	PE	5086
			**	9.37 (V)	PE	4285
	$\text{C}_6\text{H}_8\text{O}$ (3-Cyclohexen-1-one)	4096-34-8	**	9.42 (V)	PE	4285
	$\text{C}_4\text{H}_3\text{OC}_2\text{H}_5$ (Furan, 2-ethyl-)	3208-16-0	**	8.45 ± 0.05	EI	3482
	$\text{C}_6\text{H}_8\text{O}$ (7-Oxabicyclo[2.2.1]hept-2-ene)	6705-50-6	**	9.44 ± 0.02 (V)	PE	3843
$\text{C}_6\text{H}_{10}\text{O}^+$						
	$\text{C}_5\text{H}_7(\text{OCH}_3)$ (Cyclopentene, 3-methoxy-)	39819-74-4	**	9.45 ± 0.05 (V)	PE	4954
	$\text{C}_5\text{H}_7(\text{OCH}_3)$ (Cyclopentene, 4-methoxy-)	40955-64-4	**	9.12 ± 0.03 (V)	PE	4468
	$\text{C}_5\text{H}_7\text{O}(\text{CH}_3)$ (2H-Pyran, 3,4-dihydro-6-methyl-)	16015-11-5	**	8.40 (V)	PE	4569
	$\pi\text{-C}_3\text{H}_7\text{CH}=\text{CHCHO}$	505-57-7	**	9.65	PE	5360
	$\text{CH}\equiv\text{CC}(\text{CH}_3)(\text{OH})\text{C}_2\text{H}_5$	77-75-8	**	10.03 (V)	PE	4847
	$\text{CH}_2=\text{CHCH}_2\text{CH}_2\text{COCH}_3$	109-49-9	**	9.50 (V)	PE	4195
	$(\text{C}_2\text{H}_5)_2\text{C}=\text{C=O}$	24264-08-2	**	8.24	EI	4660
	$\text{CH}_3\text{CH}=\text{C}(\text{C}_2\text{H}_5)\text{CHO}$ (2-Butenal, 2-ethyl-)	19780-25-7	**	9.53	PE	5360
	$\text{C}_6\text{H}_{10}\text{O}$ (Cyclohexanone)	108-94-1	**	9.14 ± 0.03	PI	3765
			**	9.14 ± 0.02 (V)	PE	3517
			**	9.18	PE	5085
			**	9.18 (V)	PE	5043
			**	9.28 (V)	PE	4285
			**	9.5 ± 0.2	EI	4074
	$\text{C}_3\text{H}_4(\text{CH}_3)\text{COCH}_3$ (Ethanone, 1-(1-methylcyclopropyl)-)	1567-75-5	**	9.3 (V)	PE	5528
	$\text{C}_3\text{H}_4(\text{CH}_3)\text{COCH}_3$ (Ethanone, 1-(2-methylcyclopropyl)-)	930-56-3	**	9.38 (V)	PE	5528
	$\text{CH}_3\text{CH}=\text{CHC}(=\text{O})\text{C}_2\text{H}_5$	2497-21-4	**	9.32	PE	5360
	$\text{C}_6\text{H}_{10}\text{O}$ (7-Oxabicyclo[2.2.1]heptane)	279-49-2	**	9.57 ± 0.02 (V)	PE	3843
	$\text{CH}_3\text{CH}_2\text{CH}=\text{C}(\text{CH}_3)\text{CHO}$	623-36-9	**	9.54	PE	5360
	$\text{iso-C}_3\text{H}_7\text{COCH}=\text{CH}_2$	1606-47-9	**	9.39	PE	5360
	$\text{CH}_3\text{CH}=\text{C}(\text{CH}_3)\text{C}(=\text{O})\text{CH}_3$	565-62-8	**	9.35	PE	5360
	$(\text{CH}_3)_2\text{C}=\text{CHC}(=\text{O})\text{CH}_3$	141-79-7	**	9.11	PE	5360
	$\text{C}_3\text{HN}(\text{=O})_2(\text{C}_2\text{H}_5)_2$ (2,4-Azetidinedione, 3,3-diethyl-)	42282-85-9		9.60	EI	4660

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_6\text{H}_{10}\text{O}^+$	$\text{C}_3\text{N}(=\text{O})_2(\text{C}_2\text{H}_5)_2\text{C}_6\text{H}_5$ (2,4-Azetidinedione, 3,3-diethyl-1-phenyl-)	15745-94-5		9.61	EI	4660
$\text{C}_6\text{H}_{11}\text{O}^+$	$(\text{C}_6\text{H}_{11}\text{NO}_2)_2$	68777-99-1		9.65	EI	4809
$\text{C}_6\text{H}_{12}\text{O}^+$						
	$\text{C}_5\text{H}_9(\text{OCH}_3)$ (Cyclopentane, methoxy-)	5614-37-9	**	9.40 ± 0.03 (V)	PE	4468
	$\text{tert-C}_4\text{H}_9\text{COCH}_3$	75-97-8	**	8.88 ± 0.04	PE	3851
			**	9.11 ± 0.01	PE	4535
			**	9.117 ± 0.005	PE	5519
			**	9.21 (V)	PE	4224
			**	9.17 ± 0.06	EI	4535
			**	9.24	PE	4395
			**	9.18 ± 0.03	PI	3765
	$(\text{CH}_3)_2\text{CHC}_2\text{H}_4\text{CHO}$	1119-16-0	**	9.80	EI	5264
	$\text{C}_2\text{H}_5\text{CH}(\text{CH}_3)\text{CH}_2\text{CHO}$	15877-57-3	**	9.90	EI	5264
	$n-\text{C}_5\text{H}_{11}\text{CHO}$	66-25-1	**	9.62 ± 0.02	PE	4695
			**	9.722 ± 0.005	PE	5519
			**	9.80	EI	5264
	$n-\text{C}_3\text{H}_7\text{COC}_2\text{H}_5$	589-38-8	**	9.12 ± 0.02	PE	4695
	$n-\text{C}_4\text{H}_9\text{COCH}_3$	591-78-6	**	9.44 ± 0.03	PI	3765
			**	9.24 ± 0.02	PE	4695
			**	9.331 ± 0.005	PE	5519
			**	9.38 (V)	PE	4850
			**	9.2	EI	3916
	$iso-\text{C}_4\text{H}_9\text{COCH}_3$	108-10-1	**	9.42	PE	4224
			**	9.296 ± 0.005	PE	5519
	$sec-\text{C}_5\text{H}_{11}\text{CHO}$	123-15-9	**	9.70	EI	5264
	$sec-\text{C}_4\text{H}_9\text{COCH}_3$	565-61-7	**	9.209 ± 0.005	PE	5519
	$iso-\text{C}_3\text{H}_7\text{COC}_2\text{H}_5$	565-69-5	**	9.098 ± 0.005	PE	5519
	$neo-\text{C}_5\text{H}_{11}\text{CHO}$	2987-16-8	**	9.610 ± 0.005	PE	5519
	$\text{C}_6\text{H}_{11}\text{OH}$ (Cyclohexanol)	108-93-0	**	10.0 ± 0.2	EI	4617
$\text{C}_6\text{H}_{14}\text{O}^+$						
	$tert-\text{C}_4\text{H}_9\text{OC}_2\text{H}_5$	637-92-3	**	9.39 ± 0.015 (V)	PE	4434
	$(n-\text{C}_3\text{H}_7)_2\text{O}$	111-43-3	**	9.49 (V)	PE	4850
$\text{C}_7\text{H}_5\text{O}^+$						
	$\text{C}_6\text{H}_5\text{CHO}$ (Benzaldehyde)	100-52-7	H	11.26	EI	3792
	$\text{C}_6\text{H}_5\text{COCH}_3$ (Acetophenone)	98-86-2	CH_3	10.50 ± 0.01	EI	5059
			CH_3	9.6	EI	3916
			CH_3	10.38	EI	3792
	$(\text{C}_6\text{H}_5)_2\text{CO}$ (Methanone, diphenyl-)	119-61-9		11.35 ± 0.1	EI	4335
			C_6H_5	11.72	EI	3792
			C_6H_5	12.00 ± 0.1	EI	5493
	$\text{C}_6\text{H}_5\text{COOH}$ (Benzoic acid)	65-85-0	OH	11.5 ± 0.07	EI	5121
			OH	12.11 ± 0.2	EI	3973
			OH	12.11	EI	3792
	$\text{C}_6\text{H}_5\text{COOCH}_3$ (Benzoic acid, methyl ester)	93-58-3	OCH_3	10.8 ± 0.05	EI	5121
			OCH_3	11.40	EI	3792
	$\text{C}_6\text{H}_5\text{COOC}_2\text{H}_5$ (Benzoic acid, ethyl ester)	93-89-0	OC_2H_5	10.8 ± 0.07	EI	5121
	$\text{C}_6\text{H}_5\text{COOC}_3\text{H}_7$ (Benzoic acid, 1-methylethyl ester)	939-48-0	OC_3H_7	11.2 ± 0.10	EI	5121

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_7\text{H}_5\text{O}^+$						
	$\text{C}_6\text{H}_5\text{COOC}_3\text{H}_7$ (Benzoic acid, propyl ester)	2315-68-6	OC_3H_7	11.2 ± 0.05	EI	5121
	$\text{C}_6\text{H}_5\text{COC}_4\text{H}_3\text{O}$ (Methanone, 2-furanylphenyl-)	2689-59-0	$\text{C}_4\text{H}_3\text{O}$	12.3 ± 0.1	EI	5493
	$\text{C}_6\text{H}_5\text{COOC}_4\text{H}_9$ (Benzoic acid, butyl ester)	136-60-7	OC_4H_9	11.2 ± 0.10	EI	5121
	$\text{C}_6\text{H}_5\text{COOC}_4\text{H}_9$ (Benzoic acid, 2-methylpropyl ester)	120-50-3	OC_4H_9	11.3 ± 0.10	EI	5121
	$\text{C}_6\text{H}_5\text{COOC}_5\text{H}_{11}$ (Benzoic acid, methylbutyl ester)	XXXXXX-XX-X	OC_5H_{11}	11.2 ± 0.10	EI	5121
	$\text{C}_6\text{H}_5\text{COOC}_6\text{H}_5$ (Benzoic acid, phenyl ester)	93-99-2		10.0	EI	5631
	$\text{C}_6\text{H}_5\text{COOC}_6\text{H}_4\text{OCH}_3$ (Phenol, 4-methoxy-, benzoate)	1523-19-9		10.6	EI	5631
	$\text{C}_6\text{H}_5\text{CONH}_2$ (Benzamide)	55-21-0	NH_2	11.09	EI	3792
	$\text{C}_6\text{H}_5\text{COC}_5\text{H}_4\text{N}$ (Methanone, phenyl-2-pyridinyl-)	91-02-1	$\text{C}_5\text{H}_4\text{N}$	11.7 ± 0.1	EI	5493
	$\text{C}_6\text{H}_5\text{COC}_5\text{H}_4\text{N}$ (Methanone, phenyl-3-pyridinyl-)	5424-19-1	$\text{C}_5\text{H}_4\text{N}$	11.7 ± 0.1	EI	5493
	$\text{C}_6\text{H}_5\text{COC}_5\text{H}_4\text{N}$ (Methanone, phenyl-4-pyridinyl-)	14548-46-0	$\text{C}_5\text{H}_4\text{N}$	10.8 ± 0.1	EI	5493
	$\text{C}_6\text{H}_5\text{COC}_4\text{H}_3\text{NCH}_3$ (Methanone, (1-methyl-1H-pyrrol-2-yl)phenyl-)	37496-06-3		13.1 ± 0.1	EI	5493
	$\text{C}_5\text{H}_8\text{NCOC}_6\text{H}_5$ (Pyridine, 1-benzoyl-1,2,3,4-tetrahydro-)	50838-24-9		12.4	EI	4046
	$\text{C}_5\text{H}_{10}\text{NCOC}_6\text{H}_5$ (Piperidine, 1-benzoyl-)	776-75-0		14.4	EI	4046
	$\text{C}_6\text{H}_5\text{COC}_6\text{H}_4\text{NH}_2$ (Methanone, (2-aminophenyl)phenyl-)	2835-77-0		12.6 ± 0.2	EI	4358
	$\text{C}_6\text{H}_5\text{COC}_6\text{H}_4\text{NH}_2$ (Methanone, (3-aminophenyl)phenyl-)	2835-78-1		12.3 ± 0.2	EI	4358
	$\text{C}_6\text{H}_5\text{COC}_6\text{H}_4\text{NH}_2$ (Methanone, (4-aminophenyl)phenyl-)	1137-41-3		12.5 ± 0.2	EI	4358
	$\text{C}_6\text{H}_5\text{COC}_4\text{H}_3\text{N}_2$ (Methanone, phenylpyrazinyl-)	3430-09-9	$\text{C}_4\text{H}_3\text{N}_2$	10.8 ± 0.1	EI	5493
	$\text{C}_6\text{H}_5\text{COC}_4\text{H}_3\text{N}_2$ (Methanone, phenyl-4-pyrimidinyl-)	68027-80-5	$\text{C}_3\text{H}_3\text{N}_2$	10.7 ± 0.1	EI	5493
	$\text{C}_6\text{H}_5\text{COC}_6\text{H}_4\text{NO}_2$ (Methanone, (2-nitrophenyl)phenyl-)	2243-79-0		11.05 ± 0.1	EI	4358
	$\text{C}_6\text{H}_5\text{COC}_6\text{H}_4\text{NO}_2$ (Methanone, (3-nitrophenyl)phenyl-)	2243-80-3		11.15 ± 0.1	EI	4358
	$\text{C}_6\text{H}_5\text{COC}_6\text{H}_4\text{NO}_2$ (Methanone, (4-nitrophenyl)phenyl-)	1144-74-7		11.4 ± 0.1	EI	4358
	$\text{C}_6\text{H}_5\text{COOC}_6\text{H}_4\text{NO}_2$ (Benzoic acid, 4-nitrophenyl ester)	959-22-8		10.2	EI	5631
	$\text{C}_6\text{H}_5\text{COC}_4\text{H}_3\text{S}$ (Methanone, phenyl-2-thienyl-)	135-00-2	$\text{C}_4\text{H}_3\text{S}$	12.0 ± 0.1	EI	5493
	$\text{C}_6\text{H}_5\text{COCl}$ (Benzoyl chloride)	98-88-4	Cl	10.31	EI	3792
	$\text{C}_6\text{H}_5\text{COC}_6\text{H}_4\text{Cl}$ (Methanone, (2-chlorophenyl)phenyl-)	5162-03-8		11.1 ± 0.1	EI	4358
	$\text{C}_6\text{H}_5\text{COC}_6\text{H}_4\text{Cl}$ (Methanone, (3-chlorophenyl)phenyl-)	1016-78-0		11.3 ± 0.1	EI	4358
	$\text{C}_6\text{H}_5\text{COC}_6\text{H}_4\text{Cl}$ (Methanone, (4-chlorophenyl)phenyl-)	134-85-0		11.5 ± 0.1	EI	4358
$\text{C}_7\text{H}_6\text{O}^+$						
	$\text{C}_6\text{H}_5\text{CHO}$ (Benzaldehyde)	100-52-7	**	9.50 ± 0.02	PI	4031
			**	9.50 ± 0.02	PI	4057

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₇H₆O⁺						
	C ₆ H ₅ CHO	100-52-7	**	9.6	PI	3586
			**	9.40	PE	3938
			**	9.49	PE	4621
			**	9.54 (V)	PE	4850
			**	10.0 (V)	PE	4467
			**	9.74	EI	3792
	C ₇ H ₆ O (2,4,6-Cycloheptatriene-1-one)	539-80-0	**	8.89±0.03 (V)	PE	4391
			**	8.82 (V)	PE	5444
			**	8.90±0.02 (V)	PE	4140
	C ₆ H ₄ (=O)(=CH ₂) (2,4-Cyclohexadien-1-one, 6-methylene-)	27890-67-1	**	8.80 (V)	PE	4744
	C ₆ H ₅ CH ₂ C ₆ H ₄ OH (Phenol, 4-(phenylmethyl)-)	101-53-1	C ₆ H ₅	11.1±0.2	EI	3807
C₇H₇O⁺						
	C ₆ H ₄ (OCH ₃)CH ₃ (Benzene, 1-methoxy-3-methyl-)	100-84-5	CH ₃	11.60±0.1	EI	3446
	C ₆ H ₄ (OCH ₃)CH ₃ (Benzene, 1-methoxy-4-methyl-)	104-93-8	CH ₃	11.45±0.1	EI	3446
	C ₆ H ₄ (OH)C ₆ H ₉ (Phenol, 3-butyl-)	4074-43-5		12.79±0.1	EI	3629
	C ₆ H ₄ (OH)C ₆ H ₉ (Phenol, 4-butyl-)	1638-22-8		11.45±0.1	EI	3629
	C ₆ H ₄ (CH ₃)OOCCH ₃ (Acetic acid, 2-methylphenyl ester)	533-18-6	CH ₃ CO	13.16±0.02	EI	3631
	C ₆ H ₄ (CH ₃)OOCCH ₃ (Acetic acid, 4-methylphenyl ester)	140-39-6	CH ₃ CO	13.47±0.02	EI	3631
	C ₆ H ₄ (OCH ₃)COOH (Benzoic acid, 3-methoxy-)	586-38-9	COOH	13.07±0.2	EI	3973
	C ₆ H ₄ (OCH ₃)COOH (Benzoic acid, 4-methoxy-)	100-09-4	COOH	12.80±0.2	EI	3973
	C ₆ H ₄ (NO ₂)CH ₃ (Benzene, 1-methyl-3-nitro-)	99-08-1	NO	9.98±0.1	EI	3447
	CH ₃ C ₆ H ₄ NO ₂ (Benzene, 1-methyl-4-nitro-)	99-99-0	NO	10.91±0.05	PI	5437
			NO	10.34±0.1	EI	3447
	C ₆ H ₄ (NO ₂)OCH ₃ (Benzene, 1-methoxy-3-nitro-)	555-03-3	NO ₂	11.44±0.1	EI	3447
	C ₆ H ₄ (NO ₂)OCH ₃ (Benzene, 1-methoxy-4-nitro-)	100-17-4	NO ₂	11.63±0.1	EI	3447
C₇H₈O⁺						
	C ₆ H ₄ CH ₃ (OH) (Phenol, 2-methyl-)	95-48-7	**	8.48 (V)	PE	5272
			**	8.50 (V)	PE	4891
			**	8.24±0.02	PE	3890
	C ₆ H ₄ CH ₃ (OH) (Phenol, 3-methyl-)	108-39-4	**	8.52 (V)	PE	5272
			**	8.41 (V)	PE	4891
	C ₆ H ₄ CH ₃ (OH) (Phenol, 4-methyl-)	106-44-5	**	8.38 (V)	PE	5272
			**	8.35 (V)	PE	4891
			**	8.34	EI	4089
	C ₆ H ₅ CH ₂ OH (Benzene methanol)	100-51-6	**	9.11 (V)	PE	4850
			**	9.23 (V)	PE	4744
			**	9.00±0.1	EI	3788
	C ₆ H ₅ OCH ₃ (Benzene, methoxy-)	100-66-3	**	8.20±0.02	PE	3890
			**	8.24	PE	4621

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$C_7H_8O^+$	$C_6H_5OCH_3$	100-66-3	**	8.25 (V)	PE	4850
			**	8.39 (V)	PE	5272
			**	8.42 (V)	PE	3781
			**	8.42 (V)	PE	4884
			**	8.45 (V)	PE	5310
			**	8.46 (V)	PE	4327
			**	8.20	EI	3845
			**	8.20	EI	3845
			**	8.25 ± 0.1	EI	3788
			**	8.39 ± 0.1	EI	3446
			**	8.6	EI	3479
			**	8.6	EI	3916
			**	$8.76 \pm <0.1$	EI	3735
			**	8.18	CTS	3758
			**	8.37	CTS	4029
	C_7H_8O (Bicyclo[2.2.1]hept-2-en-7-one)	694-71-3	**	9.25 (V)	PE	4285
	C_7H_8O (Bicyclo[2.2.1]hept-5-en-2-one)	694-98-4	**	8.86 (V)	PE	4285
	C_7H_8O (2-Oxabicyclo[3.2.1]octa-3,6-diene)	4729-06-0	**	8.04-8.24 (V)	PE	5481
	$C_6H_4(OH)C_4H_9$ (Phenol, 3-butyl-)	4074-43-5	CH ₂ =CHCH ₃	11.07 ± 0.1	EI	3629
	$C_6H_4(OH)C_4H_9$ (Phenol, 4-butyl-)	1638-22-8	CH ₂ =CHCH ₃	10.32 ± 0.1	EI	3629
	$C_6H_4(OCH_3)_2$ (Benzene, 1,3-dimethoxy-)	151-10-0	CH ₂ O	10.98 ± 0.1	EI	3446
	$C_6H_4(OCH_3)_2$ (Benzene, 1,4-dimethoxy-)	150-78-7	HCHO	11.00	EI	3845
	$C_6H_4(CH_3)OOCC_3$ (Acetic acid, 2-methylphenyl ester)	533-18-6	CH ₂ =C=O	9.44 ± 0.02	EI	3631
	$C_6H_4(CH_3)OOCC_3$ (Acetic acid, 3-methylphenyl ester)	122-46-3	CH ₂ =C=O	10.03 ± 0.2	EI	3484
	$C_6H_4(CH_3)OOCC_3$ (Acetic acid, 4-methylphenyl ester)	140-39-6	CH ₂ =C=O	9.26 ± 0.02	EI	3631
	$C_6H_5OOCOCH_3$ (Carbonic acid, methyl phenyl ester)	13509-27-8	CH ₂ =C=O	9.75 ± 0.2	EI	3484
	$(C_6H_5CH_2OH)(CO_3)Cr$ (Chromium, [(1,2,3,4,5,6- η)-benzenemethanol]tricarbonyl-)	12116-45-9	CO ₂	10.3	EI	3479
	$(C_6H_5OCH_3)(CO_3)Cr$ (Chromium, tricarbonyl[(1,2,3,4,5,6- η)-methoxybenzene]-)	12116-44-8		9.40 ± 0.1	EI	3788
$C_7H_{10}O^+$	$C_7H_9(OH)$	13118-70-2	**	9.41 (V)	PE	4511
	(Bicyclo[2.2.1]hept-2-en-7-ol- <i>syn</i> -)					
	$C_7H_9(OH)$	694-70-2	**	9.19 (V)	PE	4511
	(Bicyclo[2.2.1]hept-2-en-7-ol- <i>anti</i> -)					
	$C_7H_{10}O$	497-38-1	**	9.14 (V)	PE	4285
	(Bicyclo[2.2.1]heptane-2-one)					
	$C_7H_{10}O$	10218-02-7	**	9.06 (V)	PE	4285
	(Bicyclo[2.2.1]heptan-7-one)					
	$C_7H_{10}O$	1121-66-0	**	9.25 (V)	PE	4285
	(2-Cyclohepten-1-one)					
	$C_7H_{10}O$	1121-64-8	**	9.14 (V)	PE	4285
	(3-Cyclohepten-1-one)					
	$(C_3H_5)_2CO$	1121-37-5	**	9.28 (V)	PE	4233
	(Methanone, dicyclopropyl-)					
	$C_7H_{10}O$	59171-38-9	**	8.01-8.18 (V)	PE	5481
	(2-Oxabicyclo[3.2.1]oct-3-ene)					

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₇H₁₂O⁺						
	C ₆ H ₉ (OCH ₃) (Cyclohexene, 4-methoxy-)	15766-93-5	**	9.01±0.03 (V)	PE	4468
	C ₆ H ₉ O(CH ₃) (Cyclohexanone, 2-methyl-)	583-60-8	**	9.05	PE	5085
			**	9.5±0.2	EI	4074
	C ₆ H ₉ O(CH ₃) (Cyclohexanone, 4-methyl-)	589-92-4	**	9.16	PE	5085
	CH ₂ =C(CH ₃)CH ₂ CH ₂ COCH ₃	3240-09-3	**	9.40 (V)	PE	4195
	C ₇ H ₁₂ O (Cycloheptanone)	502-42-1	**	9.14 (V)	PE	4285
			**	9.17±0.02 (V)	PE	3517
C₇H₁₄O⁺						
	C ₆ H ₁₁ (OCH ₃) (Cyclohexane, methoxy-)	931-56-6	**	9.22±0.03 (V)	PE	4468
	C ₂ H ₅ C(CH ₃) ₂ COCH ₃	20669-04-9	**	9.019±0.005	PE	5519
	n-C ₄ H ₉ COC ₂ H ₅	106-35-4	**	9.02±0.02	PE	4695
	n-C ₅ H ₁₁ COCH ₃	110-43-0	**	9.18±0.02	PE	4695
			**	9.298±0.005	PE	5519
			**	9.36 (V)	PE	4850
	n-C ₆ H ₁₃ CHO	111-71-7	**	9.65±0.02	PE	4695
	(n-C ₃ H ₇) ₂ CO	123-19-3	**	9.12±0.03	PI	3765
			**	9.04±0.02	PE	4695
			**	9.10±0.01	PE	5519
	iso-C ₅ H ₁₁ COCH ₃	110-12-3	**	9.284±0.005	PE	5519
	(iso-C ₃ H ₇) ₂ CO	565-80-0	**	8.94±0.01	PE	4535
			**	8.947±0.005	PE	5519
			**	8.99±0.04	EI	4535
	neo-C ₅ H ₁₁ COCH ₃	590-50-1	**	9.226±0.005	PE	5519
	C ₆ H ₁₀ (OH)CH ₃ (Cyclohexanol, 1-methyl-)	590-67-0	**	9.8±0.2	EI	4074
C₈H₄O⁺						
	C ₆ H ₈ O(CH ₃) ₂ (Cyclohexanone, 4,4-dimethyl-)	4255-62-3	**	9.12	PE	5085
C₈H₆O⁺						
	C ₆ H ₅ CH=C=O	3496-32-0	**	8.17 (V)	PE	5610
	C ₆ H ₆ O (Benzene, ethynyoxy-)	4279-76-9		8.7	EI	5290
	C ₆ H ₄ C ₂ H ₂ O (Benzofuran)	271-89-6	**	8.37±0.015 (V)	PE	5522
			**	8.8	EI	5290
			**	8.85±0.05	EI	4316
	C ₈ H ₆ O (Phenol, 2-ethynyl-)	5101-44-0		8.5	EI	5290
	C ₉ H ₆ O ₂ (2H-1-Benzopyran-2-one)	91-64-5	CO	10.8	EI	5290
C₈H₇O⁺						
	CH ₃ C ₆ H ₄ COCH ₃ (Ethanone, 1-(4-methylphenyl))	122-00-9	CH ₃	10.52±0.05	EI	5059
	C ₆ H ₄ (CH ₃)COOH (Benzoic acid, 3-methyl-)	99-04-7	OH	12.38±0.2	EI	3973
	C ₆ H ₄ (CH ₃)COOH (Benzoic acid, 4-methyl-)	99-94-5	OH	12.07±0.2	EI	3973
	C ₆ H ₅ COCOC ₆ H ₄ CH ₃ (Ethanedione, (4-methylphenyl)phenyl-)	2431-00-7	C ₆ H ₅ CO	9.84±0.10	EI	3823
C₉H₈O⁺						
	C ₇ H ₅ OCH ₃ 2,4,6-Cycloheptatriene-1-one, 2-methyl-)	29639-50-0	**	8.61±0.03 (V)	PE	4391

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_8\text{H}_8\text{O}^+$						
	$\text{C}_6\text{H}_5\text{COCH}_3$ (Benzaldehyde, methyl-)	1334-78-7	**	8.9 (V)	PE	4467
	$\text{C}_6\text{H}_5\text{CH}_2\text{CHO}$ (Benzeneacetaldehyde)	122-78-1	**	8.80	PE	3938
	$\text{C}_6\text{H}_8\text{O}$ (Benzofuran, 2,3-dihydro-)	496-16-2	**	8.02	PE	4573
	$\text{C}_6\text{H}_5\text{COCH}_3$ (Ethanone, 1-phenyl-)	98-86-2	**	9.29 ± 0.2	PI	4031
			**	9.29 ± 0.2	PI	4057
			**	9.6	PI	3586
			**	9.1 ± 0.1	PE	4401
			**	9.35 (V)	PE	4850
			**	9.37 (V)	PE	5272
			**	9.45 (V)	PE	4804
			**	9.1	EI	3916
			**	9.50	EI	3792
	$\text{C}_6\text{H}_4\text{O}(=\text{CH}_2)_2$ (7-Oxabicyclo[2.2.1]hept-2-ene, 5,6-bis(methylene)-)	56582-02-6	**	8.87 ± 0.03 (V)	PE	4665
	$\text{C}_8\text{H}_8\text{O}$ (9-Oxabicyclo[4.2.1]nona-2,4,7-triene)	7140-63-8	**	8.56 (V)	PE	4688
	$\text{C}_6\text{H}_5\text{C}_2\text{H}_3\text{O}$ (Oxirane, phenyl-)	96-09-3	**	9.04 (V)	PE	4927
			**	9.07 (V)	PE	4747
			**	9.23 (V)	PE	5364
	$\text{C}_{10}\text{H}_{11}\text{OH}$ (1-Naphthalenol, 1,2,3,4-tetrahydro-)	529-33-9	C_2H_4	10.42 ± 0.03	EI	4960
$\text{C}_8\text{H}_9\text{O}^+$						
	$\text{C}_6\text{H}_4(\text{OCH}_3)\text{C}_4\text{H}_9$ (Benzene, 1-butyl-3-methoxy-)	20893-43-0		12.04 ± 0.1	EI	3629
	$\text{C}_6\text{H}_4(\text{OCH}_3)\text{C}_4\text{H}_9$ (Benzene, 1-butyl-4-methoxy-)	18272-84-9		10.79 ± 0.1	EI	3629
	$\text{C}_6\text{H}_5\text{CH}_2\text{C}_6\text{H}_4\text{OCH}_3$ (Benzene, 1-methoxy-4-(phenylmethyl)-)	834-14-0	C_6H_5	11.9 ± 0.1	EI	3807
	$\text{C}_{16}\text{H}_{18}\text{O}$ (Benzene, 1-methoxy-4-(3-phenylpropyl)-)	40715-68-2		10.7 ± 0.1	EI	4925
	$\text{C}_6\text{H}_4(\text{OCH}_3)\text{CH}_2\text{CH}_2\text{OCOCH}_3$ (Phenethyl alcohol, <i>m</i> -methoxy-, acetate)	33709-39-6		12.10	EI	3590
	$\text{C}_6\text{H}_4(\text{OCH}_3)\text{CH}_2\text{CH}_2\text{OCOCH}_3$ (Phenethyl alcohol, <i>p</i> -methoxy-, acetate)	22532-51-0		11.50	EI	3590
$\text{C}_8\text{H}_6\text{D}_3\text{O}^+$						
	$\text{C}_{17}\text{H}_{17}\text{D}_3\text{O}_2$ (Benzene, 1-methoxy-3-[3-(4-methoxy- <i>d</i> ₃ -phenyl)propyl]-)	67081-97-4		11.1 ± 0.1	EI	4925
$\text{C}_8\text{H}_{10}\text{O}^+$						
	$\text{CH}_3(\text{CH}=\text{CH})_3\text{CHO}$	17609-31-3	**	8.42 ± 0.03 (V)	PE	4767
	$\text{C}_6\text{H}_5\text{OC}_2\text{H}_5$ (Benzene, ethoxy-)	103-73-1	**	8.36 (V)	PE	5310
			**	8.41 (V)	PE	4327
			**	8.6	EI	3479
	$\text{C}_6\text{H}_5\text{CH}_2\text{OCH}_3$ (Benzene, (methoxymethyl)-)	538-86-3	**	9.07 (V)	PE	4927
			**	9.12 (V)	PE	3781
	$\text{C}_6\text{H}_4(\text{CH}_3)\text{OCH}_3$ (Benzene, 1-methoxy-2-methyl-)	578-58-5	**	7.90	PE	4573
			**	8.03 ± 0.02	PE	3890
			**	8.24 (V)	PE	5272
			**	8.24 (V)	PE	5310
	$\text{C}_6\text{H}_4(\text{CH}_3)\text{OCH}_3$ (Benzene, 1-methoxy-3-methyl)	100-84-5	**	8.28 (V)	PE	5272

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₈H₁₀O⁺						
	C ₆ H ₄ (CH ₃)OCH ₃	100-84-5	**	8.35±0.1	EI	3446
	C ₆ H ₄ (OCH ₃)CH ₃ (Benzene, 1-methoxy-4-methyl-)	104-93-8	**	8.14±0.01 (V)	PE	4389
			**	8.16 (V)	PE	4327
			**	8.17 (V)	PE	4211
			**	8.18 (V)	PE	5272
			**	7.85	EI	3845
			**	8.33±0.1	EI	3446
			**	7.91	CTS	3758
	C ₈ H ₁₀ O (Bicyclo[2.2.2]oct-5-en-2-one)	2220-40-8	**	8.73 (V)	PE	4285
	C ₆ H ₆ O(=CH ₂) ₂ (7-Oxabicyclo[2.2.1]heptane,2,3-bis(methylene)-)	53011-95-3	**	8.79±0.03 (V)	PE	4665
	C ₈ H ₁₀ O (9-Oxabicyclo[4.2.1]nona-2,4-diene)	19740-75-1	**	8.55 (V)	PE	4688
	C ₆ H ₃ (CH ₃) ₂ OH (Phenol, 2,4-dimethyl-)	105-67-9	**	8.18 (V)	PE	5272
	C ₆ H ₃ (CH ₃) ₂ OH (Phenol, 2,6-dimethyl-)	576-26-1	**	8.05±0.02	PE	3890
			**	8.26 (V)	PE	5272
			**	8.34 (V)	PE	4327
	C ₈ H ₁₀ O (Tricyclo[3.2.1.0 ^{2,4}]octan-8-one, (1 α ,2 α ,4 α ,5 α -))	14224-86-3	**	8.8±0.1	EI	3492
	C ₈ H ₁₀ O (Tricyclo[3.2.1.0 ^{2,4}]octan-8-one, exo-)	7076-83-7	**	9.2±0.1	EI	3492
	C ₆ H ₄ (OCH ₃)C ₄ H ₉ (Benzene, 1-butyl-3-methoxy-)	20893-43-0	CH ₂ =CHCH ₃	10.52±0.1	EI	3629
	C ₆ H ₄ (OCH ₃)C ₄ H ₉ (Benzene, 1-butyl-4-methoxy-)	18272-84-9	CH ₂ =CHCH ₃	10.38±0.1	EI	3629
	C ₁₀ H ₁₈ O (Benzene, 1-methoxy-3-(3-phenylpropyl-))	67081-95-2		9.7±0.1	EI	4925
	C ₁₂ H ₁₇ D ₃ O ₂ (Benzene, 1-methoxy-3-[3-(4-methoxy-d ₃ -phenyl)propyl-])	67081-97-4		9.8±0.1	EI	4925
	C ₆ H ₅ OOCOC ₂ H ₅ (Carbonic acid, ethyl phenyl ester)	3878-46-4	CO ₂	10.0	EI	3479
C₈H₁₂O⁺						
	C ₇ H ₉ (OCH ₃) (Bicyclo[2.2.1]hept-2-ene, 5-methoxy-endo-)	17190-92-0	**	8.69±0.03 (V)	PE	4468
	C ₇ H ₉ (OCH ₃) (Bicyclo[2.2.1]hept-2-ene, 5-methoxy-exo-)	17190-87-3	**	8.68±0.03 (V)	PE	4468
	C ₇ H ₉ (OCH ₃) (Bicyclo[2.2.1]hept-2-ene, 7-methoxy-syn-)	36197-25-8	**	8.84±0.03 (V)	PE	4468
			**	8.95 (V)	PE	4511
	C ₇ H ₉ (OCH ₃) (Bicyclo[2.2.1]hept-2-ene, 7-methoxy-anti-)	13041-10-6	**	9.02±0.03 (V)	PE	4468
			**	9.11 (V)	PE	4511
	C ₄ H ₃ O(<i>tert</i> -C ₄ H ₉) (Furan, 2-(1,1-dimethylethyl)-)	7040-43-9	**	8.32	CTS	4382
	C ₄ H ₃ O(<i>tert</i> -C ₄ H ₉) (Furan, 3-(1,1-dimethylethyl)-)	7040-42-8	**	8.58	CTS	4382
	C ₈ H ₁₂ O (Bicyclo[2.2.2]octan-2-one)	2716-23-6	**	9.10 (V)	PE	4285
	C ₈ H ₁₁ OH (Bicyclo[2.2.2]oct-2-en-1-ol)	68211-36-9	**	9.21±0.05 (V)	PE	4842
	anti-C ₈ H ₁₁ OH (Bicyclo[2.2.2]oct-5-en-2-ol-(1 α ,2 α ,4 α -))	6688-07-9	**	9.14±0.02 (V)	PE	4703
	syn-C ₈ H ₁₁ OH (Bicyclo[2.2.2]oct-5-en-2-ol-(1 α ,2 β ,4 α -))	19245-72-8	**	9.25±0.02 (V)	PE	4703
	C ₆ H ₁₀ (OH)C≡CH (Cyclohexanol, 1-ethynyl-)	78-27-3	**	10.6 (V)	PE	4847

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$C_8H_{12}O^+$	$C_8H_{12}O$ (2-Cycloocten-1-one) $C_8H_{12}O$ (3-Cycloocten-1-one) $C_8H_{12}O$ (9-Oxabicyclo[3.3.1]non-1-ene) $C_8H_{12}O$ (9-Oxabicyclo[4.2.1]non-7-ene) $C_8H_{11}OH$ (Tricyclo[3.2.1.0 ^{2,4}]octan-8-ol, <i>endo-syn</i> -) $C_8H_{11}OH$ (Tricyclo[3.2.1.0 ^{2,4}]octan-8-ol, <i>endo-anti</i> -) $C_8H_{11}OH$ (Tricyclo[3.2.1.0 ^{2,4}]octan-8-ol, <i>exo-syn</i> -) $C_8H_{11}OH$	1728-25-2	**	9.18 (V)	PE	4285
		4734-90-1	**	9.12 (V)	PE	4285
		40164-27-0	**	8.60 (V)	PE	4569
		20642-83-5	**	8.89 (V)	PE	4688
		7076-81-5	**	8.8±0.1	EI	3492
		16384-97-7	**	9.1±0.1	EI	3492
		7076-80-4	**	9.1±0.1	EI	3492
			**	9.3±0.1	EI	3492
$C_8H_{14}O^+$	$C_7H_{11}(OCH_3)$ (Bicyclo[2.2.1]heptane, 2-methoxy- <i>endo</i> -) $C_7H_{11}(OCH_3)$ (Bicyclo[2.2.1]heptane, 7-methoxy-) (<i>iso</i> -C ₃ H ₇) ₂ C=C=O $C_8H_{13}OH$ (Bicyclo[2.2.2]octan-1-ol) $C_8H_{14}(=O)$ (Cyclooctanone) $n-C_3H_7CH=C(CH_3)C(=O)CH_3$ (3-Hepten-2-one, 3-methyl-) $C_8H_{14}O$ (9-Oxabicyclo[3.3.1]nonane) $C_8H_{14}O$ (9-Oxabicyclo[4.2.1]nonane) $C_3HN(=O)_2(iso-C_3H_7)_2$ (2,4-Azetidinedione, 3,3-bis(1-methylethyl)-) $C_{10}H_{17}NO_2$ (2,4-Azetidinedione, 1-methyl-3,3-bis(1-methylethyl)-) $C_{11}H_{16}NO_2F_3$ (2,4-Azetidinedione, 3,3-bis(1-methylethyl)-1-(2,2,2-trifluoroethyl)-)	10395-55-3	**	9.17±0.03 (V)	PE	4468
		36197-12-3	**	9.27±0.03 (V)	PE	4468
		XXXXX-XX-X	**	8.09	EI	4660
		20534-58-1	**	9.65±0.05 (V)	PE	4842
			**	9.08 (V)	PE	5043
			**	9.09±0.02 (V)	PE	3517
		39899-08-6	**	9.22	PE	5360
		281-05-0	**	9.05 (V)	PE	4569
		284-20-8	**	9.12 (V)	PE	4688
		17197-62-5	HN=C=O	9.49	EI	4660
		38951-66-5	CH ₃ N=C=O	9.39	EI	4660
		56519-50-7	**	9.55	EI	4660
$C_8H_{16}O^+$	<i>tert</i> -C ₄ H ₉ CO(iso-C ₃ H ₇) <i>n</i> -C ₆ H ₁₃ COCH ₃ <i>n</i> -C ₄ H ₉ COCH ₂ CH ₂ CH ₃	5857-36-3	**	8.797±0.005	PE	5519
		111-13-7	**	9.40±0.03	PI	3765
			**	9.38 (V)	PE	4850
		589-63-9	**	9.10±0.05	PI	3765
$C_8H_{18}O^+$	$(n-C_4H_9)_2O$ (<i>tert</i> -C ₄ H ₉) ₂ O (1,1'Oxybis (1,1-Dimethylethane))	142-96-1	**	9.40 (V)	PE	4850
			**	9.51±0.015 (V)	PE	4434
		XXXXX-XX-X	**	8.81	PE	4577
$C_9H_8O^+$	CH≡CCH(OH)C ₆ H ₅ (Benzenemethanol, α -ethynyl-) $C_7H_4(=O)(=CH_2)_2$ (Bicyclo[2.2.1]hept-2-en-7-one, 5,6-bis(methylene)-) $C_9H_8=O$ (Bicyclo[4.2.1]nona-2,4,7-trien-9-one) $C_9H_8(=O)$ (1H-Inden-1-one, 2,3-dihydro-)	4187-87-5	**	10.69 (V)	PE	4847
		57297-57-1	**	8.57±0.03 (V)	PE	4665
		34733-74-9	**	8.28 (V)	PE	4363
		83-33-0	**	9.31	EI	4863

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₉H₉O⁺						
	C ₆ H ₂ (CH ₃) ₂ (CH ₂ D)CHO (Benzaldehyde, 2,4-dimethyl-5-(methyl-d-)-)	38479-87-7	CH ₂ D	12.3±0.1	EI	4041
	C ₆ H ₂ (CH ₃) ₂ (CH ₂ D)CHO (Benzaldehyde, 2,5-dimethyl-4-(methyl-d-)-)	38479-86-6	CH ₂ D	11.4±0.1	EI	4041
C₉H₈DO⁺						
	C ₆ H ₂ (CH ₃) ₂ (CH ₂ D)CHO (Benzaldehyde, 2,4-dimethyl-5-(methyl-d-)-)	38479-87-7	CH ₃	11.5±0.1	EI	4041
	C ₆ H ₂ (CH ₃) ₂ (CH ₂ D)CHO (Benzaldehyde, 2,5-dimethyl-4-(methyl-d-)-)	38479-86-6	CH ₃	11.4±0.1	EI	4041
C₉H₁₀O⁺						
	C ₉ H ₁₀ O (2H-1-Benzopyran, 3,4-dihydro-)	493-08-3	**	7.93	PE	4573
	C ₇ H ₆ (=O)(=CH ₂) ₂ (Bicyclo[2.2.1]heptan-7-one,2,3-bis(methylene)-)	38680-06-7	**	8.64±0.03 (V)	PE	4665
	C ₉ H ₁₀ O (2-Cyclopropen-1-one,2,3-dicyclopropyl-)	42152-37-4	**	8.55 (V)	PE	5390
	C ₆ H ₄ (CH ₃)COCH ₃ (Ethanone,1-(2-methylphenyl-))	577-16-2	**	9.15 (V)	PE	5272
	C ₆ H ₄ (CH ₃)COCH ₃ (Ethanone,1-(3-methylphenyl-))	585-74-0	**	9.14 (V)	PE	5272
	C ₆ H ₄ (CH ₃)COCH ₃ (Ethanone,1-(4-methylphenyl-))	122-00-9	**	9.12 (V)	PE	5272
	C ₆ H ₄ (OCH ₃)CH ₂ CH ₂ OOCCH ₃ (Phenethyl alcohol, <i>m</i> -methoxy-, acetate)	33709-39-6		8.40	EI	3590
	C ₆ H ₄ (OCH ₃)CH ₂ CH ₂ OOCCH ₃ (Phenethyl alcohol, <i>p</i> -methoxy-, acetate)	22532-51-0		8.25	EI	3590
C₉H₁₂O⁺						
	C ₆ H ₅ O(iso-C ₃ H ₇) (Benzene, (1-methylethoxy)-)	2741-16-4	**	8.42 (V)	PE	5310
			**	8.32 (V)	PE	4327
	C ₆ H ₄ (CH ₃)OC ₂ H ₅ (Benzene,1-ethoxy-2-methyl-)	614-71-1	**	8.21 (V)	PE	5310
	C ₆ H ₄ (CH ₃)OC ₂ H ₅ (Benzene, 1-ethoxy-4-methyl-)	622-60-6	**	8.13 (V)	PE	4327
	C ₆ H ₅ (CH ₃) ₂ OCH ₃ (Benzene,1-methoxy-2,4-dimethyl-)	6738-23-4	**	7.95 (V)	PE	5272
	C ₆ H ₅ (CH ₃) ₂ OCH ₃ (Benzene, 2-methoxy-1,3-dimethyl-)	1004-66-6	**	8.10±0.02	PE	3890
			**	8.51 (V)	PE	5272
			**	8.53 (V)	PE	4327
	syn-C ₉ H ₁₁ OH (Bicyclo[4.2.1]nona-2,4-dien-9-ol <i>syn</i> -)	64725-61-7	**	8.62±0.02 (V)	PE	4703
	anti-C ₉ H ₁₁ OH (Bicyclo[4.2.1]nona-2,4-dien-9-ol <i>anti</i> -)	64725-60-6	**	8.43±0.02 (V)	PE	4703
	C ₉ H ₁₂ (=O) (Bicyclo[4.2.1]non-7-en-9-one)	42948-91-4	**	9.10 (V)	PE	4363
	C ₉ H ₁₂ (=O) (Tricyclo[3.2.1.1 ^{3,6}]nonan-2-one)	XXXXX-XX-X	**	8.67 (V)	PE	5043
	C ₉ H ₁₂ (=O) (Tricyclo[3.2.1.1 ^{3,6}]nonan-7-one)	XXXXX-XX-X	**	8.81 (V)	PE	5043
	C ₁₀ H ₁₂ O ₂ (2,5-Cyclohexadiene-1,4-dione, 2,3,5,6-tetramethyl-)	527-17-3	CO	10.1±0.05	PI	3523
C₉H₁₄O⁺						
	C ₉ H ₁₄ (=O) (Bicyclo[4.2.1]nonan-9-one)	14252-11-0	**	8.90 (V)	PE	4363
	C ₉ H ₁₄ (=O) (Bicyclo[4.3.0]nonan-7-one)	XXXXX-XX-X	**	9.08±0.08	EI	5038

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_9\text{H}_{14}\text{O}^+$	$\text{C}_9\text{H}_{14}(=\text{O})$ (Bicyclo[4.3.0]nonan-8-one)		XXXXX-XX-X **	9.14 ± 0.08	EI	5038
	<i>syn</i> - $\text{C}_9\text{H}_{13}\text{OH}$ (Bicyclo[4.2.1]non-3-en-9-ol <i>syn</i> -)	64725-59-3	**	9.14 ± 0.02 (V)	PE	4703
	<i>anti</i> - $\text{C}_9\text{H}_{13}\text{OH}$ (Bicyclo[4.2.1]non-3-en-9-ol <i>anti</i> -)	64725-58-2	**	9.11 ± 0.02 (V)	PE	4703
	$\text{C}_8\text{H}_{11}\text{OCH}_3$ (Bicyclo[2.2.2]oct-2-ene, 1-methoxy-)	25489-02-5	**	9.17 ± 0.05 (V)	PE	4842
	$\text{C}_8\text{H}_{11}\text{OCH}_3$ (Bicyclo[2.2.2]oct-2-ene, 5-methoxy-)	56206-38-3	**	8.77 ± 0.03 (V)	PE	4468
	$\text{C}_{10}\text{H}_{15}(=\text{O})\text{CH}_3$ (Bicyclo[4.4.0]decen-3-one, 2-methyl-)		XXXXX-XX-X **	10.50 ± 0.08	EI	5038
	$\text{C}_{10}\text{H}_{15}(=\text{O})\text{CH}_3$ (Bicyclo[4.4.0]decen-3-one, 4-methyl-)		XXXXX-XX-X **	10.65 ± 0.08	EI	5038
$\text{C}_9\text{H}_{16}\text{O}^+$	$\text{C}_8\text{H}_{13}\text{OCH}_3$ (Bicyclo[2.2.2]octane, 1-methoxy-)	7697-14-5	**	9.17 ± 0.05 (V)	PE	4842
	$\text{C}_8\text{H}_{13}\text{OCH}_3$ (Bicyclo[2.2.2]octane, 2-methoxy-)	56206-39-4	**	9.07 ± 0.03	PE	4468
$\text{C}_9\text{H}_{18}\text{O}^+$	$n\text{-C}_4\text{H}_9\text{COCH}_3$	821-55-6	**	9.38 (V)	PE	4850
	$(iso\text{-C}_4\text{H}_9)_2\text{CO}$	108-83-8	**	9.04 ± 0.03	PI	3765
			**	8.98 ± 0.01	PE	5519
	$(tert\text{-C}_4\text{H}_9)_2\text{CO}$	815-24-7	**	8.67 ± 0.01	PE	4535
			**	8.67 ± 0.02	PE	5519
			**	8.79 ± 0.05	EI	4535
			**	8.65 ± 0.03	PI	3765
$\text{C}_{10}\text{H}_8\text{O}^+$	$\text{C}_{10}\text{H}_8\text{O}$ (2-Cyclopropen-1-one, 2-methyl-3-phenyl-)	26307-30-2	**	8.64 (V)	PE	5390
	$\text{C}_{10}\text{H}_7\text{OH}$ (1-Naphthalenol)	90-15-3	**	7.76 ± 0.03	PI	5552
			**	7.78 (V)	PE	4466
	$\text{C}_{10}\text{H}_7\text{OH}$ (2-Naphthalenol)	135-19-3	**	7.85 ± 0.05	PI	5552
			**	7.90 (V)	PE	4466
$\text{C}_{10}\text{H}_{10}\text{O}^+$	$\text{C}(\text{C}_6\text{H}_5)(\text{C}_2\text{H}_5)=\text{C=O}$ (1-Buten-1-one, 2-phenyl-)	XXXXX-XX-X **		7.94	EI	4660
	$\text{C}_2\text{HN}(\text{=O})_2\text{C}_2\text{H}_5(\text{C}_6\text{H}_5)$ (2,4-Azetidinedione, 3-ethyl-3-phenyl-)	42282-82-6	**	8.97	EI	4660
	$\text{C}_{12}\text{H}_{13}\text{NO}_2$ (2,4-Azetidinedione, 3-ethyl-1-methyl-3-phenyl-)	56519-51-8	**	8.83	EI	4660
$\text{C}_{10}\text{H}_{11}\text{O}^+$	$\text{C}_{10}\text{H}_{11}\text{OH}$ (1-Naphthalenol, 1,2,3,4-tetrahydro-)	529-33-9	H	9.67 ± 0.11	EI	4960
	$\text{C}_{10}\text{H}_{11}\text{OH}$ (2-Naphthalenol, 1,2,3,4-tetrahydro-)	530-91-6	H	11.6	EI	4960
$\text{C}_{10}\text{H}_{12}\text{O}^+$	$\text{C}_6\text{H}_4(\text{OCH}_3)\text{C}_3\text{H}_5$ (Benzene, 1-cyclopropyl-4-methoxy-)	4030-17-5	**	8.05 (V)	PE	4815
	$\text{C}_6\text{H}_4(\text{OCH}_3)\text{CH}_2\text{CH}=\text{CH}_2$ (Benzene, 1-methoxy-4-(2-propenyl)-)	140-67-0	**	8.20 (V)	PE	4211
	$\text{C}_{10}\text{H}_{11}\text{OH}$ (1-Naphthalenol, 1,2,3,4-tetrahydro-)	529-33-9	**	8.70 ± 0.01	EI	4960

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₀H₁₂O⁺						
	C ₁₀ H ₁₁ OH (2-Naphthalenol, 1,2,3,4-tetrahydro-)	530-91-6	**	8.67±0.02	EI	4960
C₁₀H₁₁DO⁺						
	C ₆ H ₂ (CH ₃) ₂ (CH ₂ D)CHO (Benzaldehyde, 2,4-dimethyl-5-(methyl-d)-)	38479-87-7	**	8.7±0.1	EI	4041
	C ₆ H ₂ (CH ₃) ₂ (CH ₂ D)CHO (Benzaldehyde, 2,5-dimethyl-4-(methyl-d)-)	38479-86-6	**	8.7±0.1	EI	4041
C₁₀H₁₄O⁺						
	C ₆ H ₅ O(<i>tert</i> -C ₄ H ₉) (Benzene, (1,1-dimethylethoxy-))	6669-13-2	**	8.66 (V)	PE	4327
			**	8.71±0.015 (V)	PE	
			**	8.77 (V)	PE	5310
	C ₆ H ₄ (CH ₃)O(iso-C ₃ H ₇) (Benzene, 1-methyl-2-(1-methylethoxy-))	33426-60-7	**	8.24 (V)	PE	5310
	C ₆ H ₃ (CH ₃) ₂ OC ₂ H ₅ (Benzene, 2-ethoxy-1,3-dimethyl-)	26620-08-6	**	8.49 (V)	PE	4327
	C ₆ H ₂ (CH ₃) ₃ OCH ₃ (Benzene, 2-methoxy-1,3,5-trimethyl-)	4028-66-4	**	8.28 (V)	PE	5310
	C ₆ H ₄ (CH ₃)OCH(CH ₃) ₂ (Benzene, 1-methyl-4-(1-methylethoxy-))	22921-10-4	**	8.09 (V)	PE	4327
	<i>syn</i> -C ₉ H ₁₁ OCH ₃ (Bicyclo[4.2.1]nona-2,4-diene, 9-methoxy- <i>syn</i> -)	64725-62-8	**	8.28±0.02 (V)	PE	4703
	C ₆ H ₄ (OH)C ₄ H ₉ (Phenol, 3-butyl-)	4074-43-5	**	8.92±0.1	EI	3629
	C ₆ H ₄ (OH)C ₄ H ₉ (Phenol, 4-butyl-)	1638-22-8	**	8.67±0.1	EI	3629
	C ₆ H ₄ (OH)C ₄ H ₉ (Phenol, 2-(1,1-dimethylethyl-))	88-18-6	**	8.10±0.02	PE	3890
	C ₁₀ H ₁₄ (=O) (Tricyclo[3.3.1.1 ^{3,7}]decanone)	700-58-3	**	8.67 (V)	PE	5043
			**	8.59	PE	3886
			**	8.80±0.02 (V)	PE	4217
	C ₁₀ H ₁₄ (=O) (Tricyclo[4.2.1.1 ^{2,6}]decan-8-one)	XXXXX-XX-X	**	8.57 (V)	PE	5043
	C ₁₀ H ₁₄ (=O) (Tricyclo[4.2.1.1 ^{3,6}]decan-8-one)	XXXXX-XX-X	**	8.96 (V)	PE	5043
C₁₀H₁₆O⁺						
	C ₁₀ H ₁₆ O (Bicyclo[2.2.1]heptan-2-one, 1,7,7-trimethyl-)	76-22-2	**	8.76±0.03	PI	3765
	C ₁₀ H ₁₅ OH (Tricyclo[3.3.1.1 ^{3,7}]decan-1-ol)	768-95-6	**	9.09±0.05	PE	3886
	C ₁₀ H ₁₅ OH (Tricyclo[3.3.1.1 ^{3,7}]decan-2-ol)	700-57-2	**	9.09±0.07	PE	3886
C₁₀H₁₈O⁺						
	C ₆ H ₈ (<i>tert</i> -C ₄ H ₉)(OH) (2-Cyclohexen-1-ol,4-(1,1-dimethylethyl)- <i>cis</i> -)	35376-39-7	**	9.33±0.02 (V)	PE	5420
	C ₆ H ₈ (<i>tert</i> -C ₄ H ₉)(OH) (2-Cyclohexen-1-ol,4-(1,1-dimethylethyl)- <i>trans</i> -)	35376-40-0	**	9.18±0.02 (V)	PE	5420
	C ₆ H ₈ O(<i>tert</i> -C ₄ H ₉) (Cyclohexanone,4- <i>tert</i> -butyl-)	98-53-3	**	9.04	PE	5085
C₁₀H₂₀O⁺						
	C ₆ H ₁₀ (<i>tert</i> -C ₄ H ₉)(OH) (Cyclohexanol,4-(1,1-dimethylethyl)- <i>cis</i> -)	937-05-3	**	9.82±0.02 (V)	PE	5420
	C ₆ H ₁₀ (<i>tert</i> -C ₄ H ₉)(OH) (Cyclohexanol,4-(1,1-dimethylethyl)- <i>trans</i> -)	21862-63-5	**	9.91±0.02 (V)	PE	5420

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{11}\text{H}_9\text{O}^+$	$\text{C}_{11}\text{H}_8\text{O}$ (7-H-Benzocyclohepten-7-one)	4443-91-8	**	8.61 ± 0.03 (V)	PE	4391
	$\text{C}_{10}\text{H}_7\text{CHO}$ (1-Naphthalenecarboxaldehyde)	66-77-3	**	8.43 ± 0.03	PI	5552
$\text{C}_{11}\text{H}_{10}\text{O}^+$	$\text{C}_{10}\text{H}_7\text{OCH}_3$ (Naphthalene, 1-methoxy-)	2216-69-5	**	7.72 (V)	PE	3781
	$\text{C}_{10}\text{H}_7\text{OCH}_3$ (Naphthalene, 2-methoxy-)	93-04-9	**	7.87 (V)	PE	3781
$\text{C}_{11}\text{H}_{12}\text{O}^+$	<i>syn</i> - $\text{C}_{11}\text{H}_{11}\text{OH}$ (1,4-Methanonaphthalen-9-ol, 1,2,3,4-tetrahydrostereoisomer)	1198-20-5	**	8.80 ± 0.02 (V)	PE	4703
			**	8.62 ± 0.02 (V)	PE	4703
	$\text{C}_{20}\text{H}_{26}\text{O}_2$ (D-Homoestra-1,3,5(10)-trien-17a-one, 3-methoxy-)	1232-89-9		11.46 ± 0.05	EI	3571
	$\text{C}_{20}\text{H}_{26}\text{O}_2$ (D-Homoestra-1,3,5(10)-trien-17a-one, 3-methoxy-, (8a)-)	1232-88-8		11.20 ± 0.05	EI	3571
$\text{C}_{11}\text{H}_{13}\text{O}^+$	$\text{C}_6(\text{CH}_3)_4(\text{CH}_2\text{D})\text{CHO}$ (Benzaldehyde, 2,3,5,6-tetramethyl-4-(methyl-d)-)	43022-36-2	CH_2D	11.2 ± 0.1	EI	4041
$\text{C}_{11}\text{H}_{14}\text{O}^+$	$\text{C}_6\text{H}_4(\text{OCH}_3)\text{C}_3\text{H}_4(\text{CH}_3)$ (Benzene, 1-methoxy-4-(1-methylcyclopropyl)-)	63340-01-2	**	8.09 (V)	PE	4815
	$\text{C}_{11}\text{H}_{14}\text{O}$ (2-Cyclopropen-1-one,2,3-bis(1-methylcyclopropyl)-)	58287-34-6	**	8.44 (V)	PE	5390
	$\text{C}_6\text{H}_5\text{CO}(\text{CH}_2)_3\text{CH}_3$ (1-Pantanone, 1-phenyl-)	1009-14-9	**	9.3 (V)	PE	4804
	$\text{tert-C}_4\text{H}_9\text{COC}_6\text{H}_5$ (1-Propanone, 2,2 dimethyl-1-phenyl-)	938-16-9	**	8.70	PE	4395
			**	9.02 (V)	PE	4804
$\text{C}_{11}\text{H}_{13}\text{DO}^+$	$\text{C}_6(\text{CH}_3)_4(\text{CH}_2\text{D})\text{CHO}$ (Benzaldehyde, 2,3,5,6-tetramethyl-4-(methyl-d)-)	43022-36-2	CH_3	11.2 ± 0.1	EI	4041
$\text{C}_{11}\text{H}_{16}\text{O}^+$	$\text{C}_6\text{H}_4(\text{CH}_3)\text{O}(\text{tert-C}_4\text{H}_9)$ (Benzene, 1-(1,1-dimethylethoxy)-2-methyl-)	15359-96-3	**	8.45 (V)	PE	5310
	$\text{C}_6\text{H}_4(\text{OCH}_3)\text{C}_4\text{H}_9$ (Benzene, 1-butyl-3-methoxy-)	20893-43-0	**	8.17 ± 0.1	EI	3629
	$\text{C}_6\text{H}_4(\text{OCH}_3)\text{C}_4\text{H}_9$ (Benzene, 1-butyl-4-methoxy-)	18272-84-9	**	8.24 ± 0.1	EI	3629
	$\text{C}_6\text{H}_4(\text{CH}_3)\text{OC}(\text{CH}_3)_3$ (Benzene, 1-(1,1-dimethylethoxy)-4-methyl-)	15359-98-5	**	8.23 (V)	PE	4327
	$\text{C}_6\text{H}_3(\text{CH}_3)_2\text{OCH}(\text{CH}_3)_2$ (Benzene, 1,3-dimethyl-2-(1-methylethoxy)-)	54350-31-1	**	8.49 (V)	PE	4327
	$\text{C}_6\text{H}_2(\text{CH}_3)_3\text{OC}_2\text{H}_5$ (Benzene, 2-ethoxy-1,3,5-trimethyl-)	61248-63-3	**	8.28 (V)	PE	5310
	$\text{C}_{10}\text{H}_{13}(=\text{O})\text{CH}_3$ (2(3 <i>H</i>)-Naphthalenone, 4,4a,5,6,7,8-hexahydro-4a-methyl-)	826-56-2	**	9.6 ± 0.2	EI	4074
$\text{C}_{11}\text{H}_{18}\text{O}^+$	$\text{C}_{10}\text{H}_{15}(=\text{O})\text{CH}_3$ (Bicyclo[4.4.0]decan-3-one, 2-methyl-)	XXXXXX-XX-X	**	9.32 ± 0.08	EI	5038
	$\text{C}_{10}\text{H}_{15}(=\text{O})\text{CH}_3$ (Bicyclo[4.4.0]decan-3-one, 4-methyl-)	XXXXXX-XX-X	**	9.41 ± 0.08	EI	5038
	$\text{C}_9\text{H}_{13}(=\text{O})(\text{C}_2\text{H}_5)$ (Bicyclo[4.3.0]nonan-7-one, 1-ethyl-)	XXXXXX-XX-X	**	9.40 ± 0.08	EI	5038

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₁H₁₈O⁺						
	C ₉ H ₁₃ (=O)(C ₂ H ₅) (Bicyclo[4.3.0]nonan-8-one, 7-ethyl-)	XXXXX-XX-X **		9.45±0.08	EI	5038
	C ₃ (C ₄ H ₉) ₂ =O (2-Cyclopropen-1-one, 2,3-bis(1,1-dimethylethyl)-)	19985-79-6 **		8.23 (V)	PE	4361
			**	8.36 (V)	PE	5390
	C ₆ H ₆ (CH ₃)(OH)C ₄ H ₈ (4a(2H)-Naphthalenol, 1,3,4,7,8,8a-hexahydro-8a-methyl-cis-)	68211-44-9 **		9.26±0.02 (V)	PE	5420
	C ₆ H ₆ (CH ₃)(OH)C ₄ H ₈ (4a(2H)-Naphthalenol, 1,3,4,7,8,8a-hexahydro-8a-methyl-trans-)	XXXXX-XX-X **		9.35±0.02 (V)	PE	5420
	C ₁₁ H ₁₇ OH (4a(2H)-Naphthalenol, 1,3,4,7,8,8a-hexahydro-8a-methyl-trans-)	67497-82-9 **		9.35±0.05 (V)	PE	4842
C₁₁H₂₀O⁺						
	C ₆ H ₈ (tert-C ₄ H ₉)(OCH ₃) (Cyclohexene, 3-(1,1-dimethylethyl)-6-methoxy-cis-)	71555-63-0 **		9.29±0.03 (V)	PE	5420
	C ₆ H ₈ (tert-C ₄ H ₉)(OCH ₃) (Cyclohexene, 3-(1,1-dimethylethyl)-6-methoxy-trans-)	71555-64-1 **		8.97±0.02 (V)	PE	5420
	C ₁₁ H ₁₉ OH (Cyclohexanol, 5-(1,1-dimethylethyl)-2-methylene-cis-)	19245-69-3 **		9.18±0.05 (V)	PE	4842
			**	9.18±0.02 (V)	PE	5420
	C ₁₁ H ₁₉ OH (Cyclohexanol, 5-(1,1-dimethylethyl)-2-methylene-trans-)	19245-70-6 **		9.37±0.05 (V)	PE	4842
			**	9.37±0.02 (V)	PE	5420
	C ₃ H ₂ (C ₄ H ₉) ₂ =O (Cyclopropanone, 2,3-bis(1,1-dimethylethyl)-, trans-)	14743-58-9 **		8.45 (V)	PE	4361
	tert-C ₄ H ₉ CO(C ₂ H ₅)=C(CH ₃)CH ₃ (4-Hexen-3-one, 4-ethyl-2,2,5-trimethyl-)	68165-37-7 **		8.74	PE	5360
	C ₆ H ₈ (CH ₃)(OH)C ₄ H ₈ (4a(2H)-Naphthalenol, octahydro-8a-methyl-cis-)	5173-74-0 **		9.45±0.02 (V)	PE	5420
	C ₁₁ H ₁₉ OH (4a(2H)-Naphthalenol, octahydro-8a-methyl-trans-)	5173-73-9 **		9.41±0.05 (V)	PE	4842
C₁₁H₂₂O⁺						
	C ₆ H ₁₀ (tert-C ₄ H ₉)(OCH ₃) (Cyclohexane, 1-(1,1-dimethylethyl)-4-methoxy-cis-)	15875-99-7 **		9.36±0.02 (V)	PE	5420
	C ₆ H ₁₀ (tert-C ₄ H ₉)(OCH ₃) (Cyclohexane, 1-(1,1-dimethylethyl)-4-methoxy-trans-)	15876-31-0 **		9.32±0.02 (V)	PE	5420
C₁₂H₈O⁺						
	(C ₆ H ₄) ₂ O (Dibenzofuran)	132-64-9 **		8.09 (V)	PE	5619
			**	8.77	EI	4228
C₁₂H₁₀O⁺						
	(C ₆ H ₅) ₂ O (Benzene, 1,1'-oxybis-)	101-84-8 **		8.09±0.03	PI	5552
			**	8.0	PE	4228
	C ₁₁ H ₇ OCH ₃ (7-H-Benzocyclohepten-7-one, 6-methyl-)	4900-73-6 **		8.46±0.03 (V)	PE	4391
	C ₆ H ₅ C ₆ H ₄ OH ([1,1'-Biphenyl]-2-ol)	90-43-7 **		7.80±0.02	PE	3702
	C ₆ H ₅ C ₆ H ₄ OH ([1,1'-Biphenyl]-4-ol)	92-69-3 **		7.78±0.03	PI	5552
	C ₁₀ H ₇ COCH ₃ (Ethanone, 1-(1-naphthalenyl)-)	941-98-0 **		8.23 (V)	PE	4466
	C ₁₀ H ₇ COCH ₃ (Ethanone, 1-(2-naphthalenyl)-)	93-08-3 **		8.31 (V)	PE	4466
	C ₁₀ H ₇ C ₂ H ₃ O (Oxirane, 2-naphthalenyl-)	20861-99-8 **		8.21 (V)	PE	5364

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{12}\text{H}_{12}\text{O}^+$	$\text{C}_{11}\text{H}_9(\text{OCH}_3)$ (1,4-Methanonaphthalene, 1,4-dihydro-5-methoxy-)	53308-23-9	**	8.10 ± 0.05 (V)	PE	5019
			**	8.10 (V)	PE	4835
	$\text{C}_{11}\text{H}_9(\text{OCH}_3)$ (1,4-Methanonaphthalene, 1,4-dihydro-6-methoxy-)	4897-71-6	**	7.87 ± 0.05	PE	5019
			**	7.87 (V)	PE	4835
$\text{C}_{12}\text{H}_{14}\text{O}^+$	<i>syn</i> - $\text{C}_{11}\text{H}_{11}\text{OCH}_3$ 1,4-Methanonaphthalene, 1,2,3,4-tetrahydro-9-methoxy-stereoisomer)	64725-57-1	**	8.46 ± 0.02 (V)	PE	4703
	$\text{C}_{12}\text{H}_{14}\text{O}$ (4a,8a-Ethanonaphthalene-9-one, 1,4,5,8-tetrahydro-)	60964-67-2	**	8.85 ± 0.05 (V)	PE	4593
	<i>anti</i> - $\text{C}_{11}\text{H}_{11}\text{OCH}_3$ (1,4-Methanonaphthalene, 1,2,3,4-tetrahydro-9-methoxy-stereoisomer)	64725-57-1	**	8.61 ± 0.02 (V)	PE	4703
	<i>anti</i> - $\text{C}_{11}\text{H}_{10}(\text{OH})\text{CH}_3$ (1,4-Methanonaphthalen-9-ol, 1,2,3,4-tetrahydro-6-methyl-(1a,4a,6R*)-)	1201-10-1	**	8.31 ± 0.02 (V)	PE	4703
	<i>syn</i> - $\text{C}_{11}\text{H}_{10}(\text{OH})\text{CH}_3$ (1,4-Methanonaphthalen-9-ol, 1,2,3,4-tetrahydro-6-methyl-(1a,4a,9S*)-)	16306-87-7	**	8.41 ± 0.02 (V)	PE	4703
$\text{C}_{12}\text{H}_{16}\text{O}^+$	$\text{C}_6\text{H}_4(\text{OCH}_3)\text{C}_3\text{H}_4(\text{C}_2\text{H}_5)$ (Benzene, 1-(1-ethylcyclopropyl)-4-methoxy-)	63340-02-3	**	8.11 (V)	PE	4815
	$\text{C}_6\text{H}_4(\text{C}(\text{CH}_3)_3)\text{COCH}_3$ (Ethanone, 1-[4-(1,1-dimethylethyl)phenyl]-)	943-27-1	**	9.01 ± 0.05 (V)	PE	5097
	$\text{C}_6\text{H}_4(\text{CH}_3)\text{CO}(\text{CH}_2)_3\text{CH}_3$ (1-Pentanone, 1-(4-methylphenyl)-)	1671-77-8	**	9.02 (V)	PE	4804
$\text{C}_{12}\text{H}_{15}\text{DO}^+$	$\text{C}_6(\text{CH}_3)_4(\text{CH}_2\text{D})\text{CHO}$ (Benzaldehyde, 2,3,5,6-tetramethyl-4-(methyl-d)-)	43022-36-2	**	8.3 ± 0.1	EI	4041
$\text{C}_{12}\text{H}_{18}\text{O}^+$	$\text{C}_6\text{H}_2(\text{CH}_3)_3\text{O(iso-C}_3\text{H}_7)$ (Benzene, 1,3,5-trimethyl-2-(1-methylethoxy-))	13605-05-5	**	8.15 (V)	PE	5310
	$\text{C}_6\text{H}_3(\text{CH}_3)_2\text{OC}(\text{CH}_3)_3$ (Benzene, 2-(1,1-dimethylethoxy)-1,3-dimethyl-)	54350-32-2	**	8.47 (V)	PE	4327
	$\text{C}_{10}\text{H}_{15}\text{COCH}_3$ (Ethanone, 1-tricyclo[3.3.1.1 ^{3,7}]dec-1-yl-)	1660-04-4	**	8.82 ± 0.05	PE	3851
$\text{C}_{12}\text{H}_{20}\text{O}^+$	$\text{C}_6\text{H}_6(\text{CH}_3)(\text{OCH}_3)\text{C}_4\text{H}_8$ (Naphthalene, 1,2,3,4,4a,5,6,8a-octahydro-8a-methoxy-4a-methyl- <i>cis</i> -)	71546-87-7	**	9.34 ± 0.02 (V)	PE	5420
	$\text{C}_6\text{H}_6(\text{CH}_3)(\text{OCH}_3)\text{C}_4\text{H}_8$ (Naphthalene, 1,2,3,4,4a,5,6,8a-octahydro-8a-methoxy-4a-methyl- <i>trans</i> -)	68211-38-1	**	9.00 ± 0.05 (V)	PE	4842
			**	9.35 ± 0.02 (V)	PE	5420
$\text{C}_{12}\text{H}_{22}\text{O}^+$	$\text{C}_{11}\text{H}_{19}\text{OCH}_3$ (Cyclohexane, 1-(1,1-dimethylethyl)-3-methoxy-4-methylene- <i>cis</i> -)	68211-39-2	**	8.97 ± 0.05 (V)	PE	4842
			**	8.97 ± 0.02 (V)	PE	5420
	$\text{C}_{11}\text{H}_{19}\text{OCH}_3$ (Cyclohexane, 1-(1,1-dimethylethyl)-3-methoxy-4-methylene- <i>trans</i> -)	68211-40-5	**	9.30 ± 0.05 (V)	PE	4842
			**	9.30 ± 0.02 (V)	PE	5420
	$\text{C}_6\text{H}_8(\text{CH}_3)(\text{OCH}_3)\text{C}_4\text{H}_8$ (Naphthalene, decahydro-4a-methoxy-8a-methyl- <i>cis</i> -)	17987-54-1	**	9.08 ± 0.02 (V)	PE	5420
	$\text{C}_{11}\text{H}_{19}\text{OCH}_3$ (Naphthalene, decahydro-4a-methoxy-8a-methyl- <i>trans</i> -)	17987-53-0	**	9.10 ± 0.05 (V)	PE	4842
$\text{C}_{13}\text{H}_8\text{O}^+$	$(\text{C}_6\text{H}_5)_2\text{CO}$ (9H-Fluoren-9-one)	486-25-9	**	8.36 ± 0.03	PI	5552

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{13}\text{H}_8\text{O}^+$						
	$(\text{C}_6\text{H}_4)_2\text{CO}$	486-25-9	**	8.36 ± 0.02	PI	3523
	$\text{C}_{13}\text{H}_8(=\text{O})$ (1H-Phenalen-1-one)	548-39-0	**	8.20 ± 0.04 (V)	PE	5193
	$\text{C}_6\text{H}_5\text{COC}_6\text{H}_4\text{NH}_2$ (Methanone, (2-aminophenyl)phenyl-)	2835-77-0		11.2 ± 0.2	EI	4358
	$\text{C}_6\text{H}_5\text{COC}_6\text{H}_4\text{NH}_2$ (Methanone, (3-aminophenyl)phenyl-)	2835-78-1		12.3 ± 0.3	EI	4358
	$\text{C}_6\text{H}_5\text{COC}_6\text{H}_4\text{NH}_2$ (Methanone, (4-aminophenyl)phenyl-)	1137-41-3		11.5 ± 0.2	EI	4358
$\text{C}_{13}\text{H}_9\text{O}^+$						
	$\text{C}_6\text{H}_5\text{COC}_6\text{H}_4\text{Cl}$ (Methanone, (2-chlorophenyl)phenyl-)	5162-03-8		10.9 ± 0.2	EI	4358
	$\text{C}_6\text{H}_5\text{COC}_6\text{H}_4\text{Cl}$ (Methanone, (3-chlorophenyl)phenyl-)	1016-78-0		11.0 ± 0.1	EI	4358
	$\text{C}_6\text{H}_5\text{COC}_6\text{H}_4\text{Cl}$ (Methanone, (4-chlorophenyl)phenyl-)	134-85-0		11.1 ± 0.2	EI	4358
$\text{C}_{13}\text{H}_{10}\text{O}^+$						
	$\text{C}_6\text{H}_5\text{C}_6\text{H}_4\text{CHO}$ ([1,1'-Biphenyl]-4-carboxaldehyde)	3218-36-8	**	8.47 ± 0.03	PI	5552
	$(\text{C}_6\text{H}_5)_2\text{CO}$ (Methanone, diphenyl-)	119-61-9	**	9.14 ± 0.03	PI	4031
			**	9.14 ± 0.03	PI	4057
			**	9.4	PI	3586
			**	9.05 ± 0.05 (V)	PE	4844
			**	9.4 ± 0.1	EI	5493
			**	9.45 ± 0.1	EI	4335
			**	9.45 ± 0.1	EI	4358
			**	9.46	EI	3792
	$(\text{C}_6\text{H}_4)_2\text{CH}_2\text{OC}(=\text{O})$ (Dibenz[b,e]oxepin-11(6H)-one)	4504-87-4	CO	11.5	EI	5340
$\text{C}_{13}\text{H}_{11}\text{O}^+$						
	$\text{C}_6\text{H}_5\text{CH}_2\text{C}_6\text{H}_4\text{OCH}_3$ (Benzene, 1-methoxy-4-(phenylmethyl)-)	834-14-0	CH_3	11.9 ± 0.1	EI	3807
$\text{C}_{13}\text{H}_{12}\text{O}^+$						
	$\text{C}_{11}\text{H}_9(\text{COCH}_3)$ (Ethanone, 1-(1,4-dihydro-1,4-methanonaphthalen-5-yl)-)	61346-78-9	**	8.49 ± 0.05 (V)	PE	5019
	$\text{C}_{11}\text{H}_9(\text{COCH}_3)$ (Ethanone, 1-(1,4-dihydro-1,4-methanonaphthalen-6-yl)-)	63509-77-3	**	8.57 ± 0.05 (V)	PE	5019
	$\text{C}_6\text{H}_5\text{CH}_2\text{OC}_6\text{H}_5$ (Benzene, phenoxymethyl-)	946-80-5	**	8.31	CT	5336
	$\text{C}_{11}\text{H}_6\text{O}(\text{CH}_3)_2$ (7H-Benzocyclohepten-7-one, 6,8-dimethyl-)	2484-16-4	**	8.29 ± 0.03 (V)	PE	4391
	$\text{C}_{11}\text{H}_6\text{O}(\text{CH}_3)_2$ (7H-Benzocyclohept-7-one, 2,3-dimethyl-)	55027-90-2	**	8.25 ± 0.03 (V)	PE	4391
	$\text{C}_6\text{H}_5\text{CH}_2\text{C}_6\text{H}_4\text{OH}$ (Phenol, 4-(phenylmethyl)-)	101-53-1	**	8.45 ± 0.05	EI	3806
$\text{C}_{13}\text{H}_{18}\text{O}^+$						
	$\text{C}_{13}\text{H}_{18}\text{O}$ (Benzene, 1-methoxy-4-[1-(1-methylethyl)cyclopropyl]-)	63340-03-4	**	8.10 (V)	PE	4815
$\text{C}_{13}\text{H}_{20}\text{O}^+$						
	$\text{C}_6\text{H}_2(\text{CH}_3)_3\text{O}(\text{tert-C}_4\text{H}_9)$ (Benzene, 2-(1,1-dimethylethoxy)-1,3,5-trimethyl-)	61248-61-1	**	8.27 (V)	PE	5310
	$\text{C}_5\text{H}_2(\text{O})(\text{C}_4\text{H}_9)_2$ (2,4-Cyclopentadien-1-one, 2,5-bis(1,1-dimethylethyl)-)	36319-88-7	**	8.50 (V)	PE	4293
	$\text{C}_5\text{H}_2(\text{O})(\text{C}_4\text{H}_9)_2$ (2,4-Cyclopentadien-1-one, 3,4-bis(1,1-dimethylethyl)-)	28786-71-2	**	8.60 (V)	PE	4293

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₄H₁₀O⁺						
	C ₁₄ H ₁₀ O (9(10H)-Anthracenone)	90-44-8	**	8.83±0.03	PI	5552
			**	8.83±0.03	PI	3523
	C ₁₄ H ₁₀ O (Dibenz [b,f]oxepin)	257-05-6	**	7.45	PE	4611
	(C ₆ H ₅) ₂ C=C=O (Ethenone, diphenyl-)	525-06-4	**	7.85	EI	4660
	C ₁₄ H ₁₀ O (Phenanthro[9,10- <i>b</i>]oxirene, 1 <i>a</i> ,9 <i>b</i> -dihydro-)	585-08-0	**	8.19 (V)	PE	5364
			**	8.24 (V) 8.46	OTH EI	4927 4660
	C ₉ N=(O) ₂ (C ₆ H ₅) ₃ (2,4-Azetidinedione, 1,3,3-triphenyl-)	15745-93-4	**			
	(C ₆ H ₅) ₂ CH ₂ SC(=O) (Dibenzo[<i>b,e</i>]thiepin-11(6H)-one)	1531-77-7	S	9.45	EI	5340
	C ₁₄ H ₁₀ SO ₃ (Dibenzo[<i>b,e</i>]thiepin-11(6H)-one-5,5-dioxide)	33301-21-2	SO ₂	10.00	EI	5414
C₁₄H₁₂O⁺						
	C ₁₄ H ₁₂ O (Oxirane, <i>cis</i> -2,3-diphenyl-)	1689-71-0	**	8.68	PE	5260
	C ₁₄ H ₁₂ O (Oxirane, <i>trans</i> -2,3-diphenyl-)	1439-07-2	**	8.60	PE	5260
C₁₄H₁₄O⁺						
	C ₆ H ₅ CH ₂ C ₆ H ₄ OCH ₃ (Benzene, 1-methoxy-4-(phenylmethyl)-)	834-14-0	**	8.20±0.05	EI	3806
	C ₆ H ₅ CH ₂ OC ₆ H ₄ CH ₃ (Benzene, 1-methyl-4-(phenylmethoxy)-)	834-25-3	**	7.91	CTS	5336
	C ₆ H ₄ (CH ₂ CH ₂) ₂ C ₆ H ₂ O (15-Oxatricyclo[8.2.2.1 ^{4,7}]pentadeca-4,6,10,12,13-pentaene)	5040-51-7	**	7.78 (V)	PE	5575
C₁₄H₂₀O⁺						
	C ₁₄ H ₂₀ O (Benzene, 1-[1-(1,1-dimethylethyl)cyclopropyl]-4-methoxy-)	63340-04-5	**	8.05 (V)	PE	4815
C₁₄H₂₂O⁺						
	C ₆ H ₃ (C ₄ H ₉) ₂ OH (Phenol, 2,6-bis(1,1-dimethylethyl-))	128-39-2	**	7.70±0.02	PE	3890
	C ₆ H ₃ (C ₄ H ₉) ₂ OH (Phenol, 3,5-bis(1,1-dimethylethyl-))	1138-52-9	**	7.90±0.02	PE	3890
C₁₅H₁₀O⁺						
	C ₁₄ H ₉ CHO (9-Anthracenecarboxaldehyde)	642-31-9	**	7.69±0.03	PI	5552
			**	7.67±0.03 (V) 8.47 (V)	PE PE	4887 5390
	C ₁₅ H ₁₀ O (2-Cyclopropen-1-one,2,3-diphenyl-)	886-38-4	**			
			**	10.56 (V) 8.5±0.1 (V)	PE PE	4856 4391
	C ₁₅ H ₁₀ O (5H-Dibenzo[a,c]cyclohepten-5-one)	4444-43-3	**			
	C ₁₅ H ₁₀ O (5H-Dibenzo[a,d]cyclohepten-5-one)	2222-33-5	**	8.06±0.03 (V)	PE	4391
C₁₅H₁₂O⁺						
	C ₁₄ H ₉ OCH ₃ (Anthracene, 9-methoxy-)	2395-96-2	**	7.21±0.03 (V)	PE	4887
	<i>trans</i> -C ₆ H ₅ CH=CHC ₆ H ₄ CHO (Benzaldehyde,4-(2-phenylethynyl)-)	32555-96-7	**	7.92±0.04	PI	5552

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{15}\text{H}_{15}\text{O}^+$	$\text{C}_{20}\text{H}_{22}\text{O}_2$ (<i>D</i> -Homoestra-1,3,5(10),6,8-pentaen-17 α -one, 3-methoxy-) $\text{C}_{20}\text{H}_{22}\text{O}_2$ (<i>D</i> -Homoestra-1,3,5(10),6,8-pentaen-17 α -one, 3-methoxy-, (14 β)-)	1232-90-2 1232-91-3		11.46±0.05 10.84±0.09	EI EI	3571 3571
$\text{C}_{16}\text{H}_{10}\text{O}^+$	$\text{C}_{16}\text{H}_{10}\text{O}$ (4,6-Ethenodibenz[<i>b,f</i>]oxepine, (<i>Z,Z</i>)-)	42073-03-0	**	7.95 (V)	PE	4088
$\text{C}_{16}\text{H}_{12}\text{O}^+$	$\text{C}_{14}\text{H}_9\text{C}_2\text{H}_3\text{O}$ (Oxirane,9-anthracenyl-)	61695-73-6	**	7.41 (V)	PE	5364
$\text{C}_{16}\text{H}_{16}\text{O}^+$	$\text{C}_{16}\text{H}_{16}\text{O}$ (6,12-Methano-7 <i>H</i> -benzocycloundecen-14-one,8,9,10,11-tetrahydro-) $\text{C}_{20}\text{H}_{22}\text{O}_2$ (<i>D</i> -Homoestra-1,3,5(10),6,8-pentaen-17 α -one, 3-methoxy-) $\text{C}_{20}\text{H}_{22}\text{O}_2$ (<i>D</i> -Homoestra-1,3,5(10),6,8-pentaen-17 α -one, 3-methoxy-, (14 β)-)	25401-39-2 1232-90-2 1232-91-3	** ** **	8.31±0.03 (V) 10.79±0.07 10.44±0.11	PE EI EI	4391 3571 3571
$\text{C}_{16}\text{H}_{18}\text{O}^+$	$\text{C}_{16}\text{H}_{18}\text{O}$ (Benzene, 1-methoxy-3-(3-phenylpropyl)-) $\text{C}_{16}\text{H}_{18}\text{O}$ (Benzene, 1-methoxy-4-(3-phenylpropyl)-)	67081-95-2 40715-68-2	** **	8.15±0.05 8.18±0.05	EI EI	4925 4925
$\text{C}_{17}\text{H}_{12}\text{O}^+$	$\text{C}_{17}\text{H}_{12}\text{O}$ (Methanone, phenyl-1-azulenyl-)	XXXXX-XX-X	**	7.55 (V)	PE	5397
$\text{C}_{17}\text{H}_{14}\text{O}^+$	$\text{C}_{15}\text{H}_8\text{O}(\text{CH}_3)_2$ (8 <i>H</i> -Cyclohepta[<i>b</i>]naphthalen-8-one,7,9-dimethyl-) $\text{C}_6\text{H}_5\text{CH}_2\text{OC}_{10}\text{H}_7$ (Naphthalene,1-(phenylmethoxy)-) $\text{C}_6\text{H}_5\text{CH}_2\text{OC}_{10}\text{H}_7$ (Naphthalene,2-(phenylmethoxy)-)	39787-00-3 607-58-9 613-62-7	** ** **	7.83±0.03 (V) 7.63 7.82	PE CTS CTS	4391 5336 5336
$\text{C}_{17}\text{H}_{18}\text{O}^+$	$\text{C}_{17}\text{H}_{18}\text{O}$ (6,13-Methanobenzocyclododecene-15-one,7,8,9,10,11,12-hexahydro-)	55027-91-3	**	8.2±0.1 (V)	PE	4391
$\text{C}_{17}\text{H}_{20}\text{O}^+$	$\text{C}_{11}\text{H}_6\text{O}[(\text{CH}_3)_2\text{CH}]_2$ (7- <i>H</i> -Benzocyclohepten-7-one,6,8-bis(1-methylethyl)-)	55027-89-9	**	8.15±0.03 (V)	PE	4391
$\text{C}_{18}\text{H}_{12}\text{O}^+$	$\text{C}_{16}\text{H}_9\text{C}_2\text{H}_3\text{O}$ (Oxirane,1-pyrenyl-)	61695-74-7	**	7.43 (V)	PE	5364
$\text{C}_{18}\text{H}_{16}\text{O}^+$	$\text{C}_{10}\text{H}_6(\text{CH}_2\text{CH}_2)_2\text{C}_4\text{H}_2\text{O}$ (8,11-Epoxy-5,14-ethenobenzocyclododecene,6,7,12,13-tetrahydro-)	24178-85-6	**	7.46 (V)	PE	5575
$\text{C}_{18}\text{H}_{18}\text{O}^+$	$\text{C}_6\text{H}_8(=\text{O})(\text{C}_6\text{H}_5)_2$ (Cyclohexanone, 4,4-diphenyl-)	4528-68-1	**	8.8±0.2	EI	4074
$\text{C}_{18}\text{H}_{20}\text{O}^+$	$\text{C}_{18}\text{H}_{20}\text{O}$ (6,14-Methanobenzocyclotridecene-16-one,8,9,10,11,12,12-hexahydro-)	25401-40-5	**	8.13±0.03 (V)	PE	4391

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{19}\text{H}_{18}\text{O}^+$	$\text{C}_{15}\text{H}_8\text{O}(\text{CH}_2\text{CH}_3)_2$ (8H-Cyclohepta[b]naphthalen-8-one, 7,9-diethyl-)	55027-92-4	**	7.83 ± 0.03 (V)	PE	4391
$\text{C}_{19}\text{H}_{20}\text{O}^+$	$\text{C}_6\text{H}_7(=\text{O})(\text{CH}_3)(\text{C}_6\text{H}_5)_2$ (Cyclohexanone, 2-methyl-5,5-diphenyl-)	50592-49-9	**	8.8 ± 0.2	EI	4074
$\text{C}_{19}\text{H}_{22}\text{O}^+$	$\text{C}_6\text{H}_8(\text{OH})(\text{CH}_3)(\text{C}_6\text{H}_5)_2$ (Cyclohexanol, 1-methyl-4,4-diphenyl-)	50592-47-7	**	9.2 ± 0.2	EI	4074
$\text{C}_{20}\text{H}_{24}\text{O}^+$	$\text{C}_{20}\text{H}_{24}\text{O}$ (6,16-Methanobenzocyclopentadecen-18-one, 8,9,10,11,12,13,14,15-octahydro-)	25401-41-6	**	8.10 ± 0.03 (V)	PE	4391
$\text{C}_{21}\text{H}_{22}\text{O}^+$	$\text{C}_{15}\text{H}_8\text{O}(\text{CH}(\text{CH}_3)_2)_2$ (8H-Cyclohepta[b]naphthalen-8-one, 7,9-bis(1-methylethyl)-)	55027-93-5	**	7.76 ± 0.03 (V)	PE	4391
$\text{C}_{22}\text{H}_{18}\text{O}^+$	$\text{C}_{14}\text{H}_8(\text{CH}_2\text{CH}_2)_2\text{C}_4\text{H}_2\text{O}$ (9,10-(Ethano[2,5]furanoethano)anthracene)	34721-69-2	**	6.87 (V)	PE	5575
$\text{C}_{23}\text{H}_{24}\text{O}^+$	$\text{C}_{10}\text{H}_{11}(=\text{O})(\text{CH}_3)(\text{C}_6\text{H}_5)_2$ (2(3H-Naphthalenone, 4,4a,5,6,7,8-hexahydro-4a-methyl-7,7-diphenyl-)	50786-03-3	**	8.9 ± 0.2	EI	4074
$\text{C}_{23}\text{H}_{30}\text{O}^+$	$\text{C}_{23}\text{H}_{30}\text{O}$ (6,19-Methanobenzocyclooctadecen-21-one, 7,8,9,10,11,12,13,14,15,16,17,18,-dodecahydro-)	25401-43-8	**	8.15 ± 0.03 (V)	PE	4391
CHO_2^+	HCOOH	64-18-6		12.26	PI	4959
			H	12.29 ± 0.03	PI	4177
			H	12.36 ± 0.1	PI	5135
	$((\text{CH}_3)_2\text{C}(\text{NO})\text{OOCCH}_3)_2$	68777-98-0		12.75	EI	4809
CH_2O_2^+	HCOOH	64-18-6	**	11.329 ± 0.002	S	5465
			**	11.05 ± 0.03	PI	3765
			**	11.16 ± 0.03	PI	4177
			**	11.314 ± 0.002	PI	4306
			**	10.7 (V)	PE	4467
			**	11.3	PE	3883
			**	11.33	PE	3874
			**	11.34 (V)	PE	4850
			**	11.35 ± 0.03	PE	3734
			**	11.51 (V)	PE	4513
CH_3O_2^+	HCOOC_2H_5	109-94-4	$\text{CH}_2 = \text{CH}$	10.9 ± 0.05	EI	4831
	$\text{HCOOCH}_2\text{CH}_2\text{CH}_3$	110-74-7	$\text{CH}_2 = \text{CHCH}_2$	10.45 ± 0.05	EI	4831
	$\text{HCOOCH}(\text{CH}_3)_2$	625-55-8	$\text{CH}_2 = \text{CHCH}_3$	10.38 ± 0.05	EI	4831
$\text{C}_2\text{H}_2\text{O}_2^+$	$(\text{CHO})_2$	107-22-2	**	10.52 (V)	PE	5517
	$((\text{CH}_3)_2\text{C}(\text{NO})\text{OOCCH}_3)_2$	68777-98-0		10.30	EI	4809
$\text{C}_2\text{H}_3\text{O}_2^+$	$((\text{CH}_3)_2\text{C}(\text{NO})\text{COOCH}_3)_2$	6144-15-6		11.05	EI	4809
	$((\text{CH}_3)_2\text{C}(\text{NO})\text{OOCCH}_3)_2$	68777-98-0		10.35	EI	4809

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_2\text{H}_4\text{O}_2^+$						
	CH_3COOH	64-19-7	**	10.38±0.03	PI	3765
			**	10.644±0.002	PI	4306
			**	10.66±0.05	PI	4959
			**	10.664±0.003	PI	5161
			**	10.63 (V)	PE	4850
			**	10.65	PE	3874
			**	10.69±0.03	PE	3734
			**	10.70	PE	3718
			**	10.8 (V)	PE	4426
			**	10.84 (V)	PE	5251
			**	10.87 (V)	PE	4513
			**	11.5 (V)	PE	4467
				10.66±0.05	EI	5263
			**	10.66	EI	5039
	HCOOCH_3	107-31-3	**	10.3 (V)	PE	4467
			**	10.85±0.05	PE	4831
			**	10.85	PE	3718
			**	10.85 (V)	PE	4850
	$\text{CH}_3\text{CH}_2\text{CH}_2\text{COOH}$	107-92-6	C_2H_4	10.60±0.05	EI	5263
			C_2H_4	10.60	EI	5039
	$\text{CH}_3(\text{CH}_2)_3\text{COOH}$	109-52-4	C_3H_6	10.56±0.05	EI	5263
			C_3H_6	10.56	EI	5039
	$\text{CH}_3(\text{CH}_2)_4\text{COOH}$	142-62-1	C_4H_8	10.52±0.05	EI	5263
			1-C ₄ H ₈	10.52	EI	5039
	$\text{CH}_3(\text{CH}_2)_5\text{COOH}$	111-14-8	C_5H_{10}	10.54±0.05	EI	5263
			1-C ₅ H ₁₀	10.54	EI	5039
	$((\text{CH}_3)_2\text{C}(\text{NO})\text{OOCCH}_3)_2$	68777-98-0		11.35	EI	4809
$\text{C}_2\text{H}_5\text{O}_2^+$						
	$\text{CH}_3\text{COOC}_2\text{H}_5$	141-78-6	$\text{CH}_2=\text{CH}$	10.6±0.1	EI	4831
	$\text{CH}_3\text{COOCH}(\text{CH}_3)_2$	108-21-4	$\text{CH}_2=\text{CHCH}_2$	9.96±0.05	EI	4831
	$\text{CH}_3\text{COOCH}_2\text{CH}_2\text{CH}_3$	109-60-4	$\text{CH}_2=\text{CHCH}_2$	9.94±0.05	EI	4831
$\text{C}_2\text{H}_6\text{O}_2^+$	$(\text{CH}_3\text{O})_2$	690-02-8	**	9.71 (V)	PE	5068
$\text{C}_3\text{H}_4\text{O}_2^+$	$\text{CH}_2=\text{CHCOOH}$	79-10-7	**	10.60	PE	3864
$\text{C}_3\text{H}_6\text{O}_2^+$						
	$\text{C}_2\text{H}_5\text{COOH}$	79-09-4	**	10.525±0.003	PI	5161
			**	10.44±0.03	PE	3734
			**	10.51 (V)	PE	4850
			**	10.54	PE	3874
			**	10.72 (V)	PE	4513
			**	10.41	EI	5039
	$\text{CH}_3\text{COOCH}_3$	79-20-9	**	10.25±0.05	PE	4831
			**	10.25 (V)	PE	4850
			**	10.33	PE	3718
			**	10.59 (V)	PE	3937
			**	11.0 (V)	PE	4467
	HCOOC_2H_5	109-94-4	**	10.61±0.05	PE	4831
			**	10.61 (V)	PE	4850
			**	10.62	PE	3718
	$\text{C}_3\text{H}_6\text{O}_2$ (1,2-Dioxolane)	4362-13-4	**	9.86 (V)	PE	5212
	$\text{C}_3\text{H}_6\text{O}_2$ (1,3-Dioxolane)	646-06-0	**	10.1 (V)	PE	3733
	$n\text{-C}_3\text{H}_7\text{COOCH}_3$	623-42-7		10.17±0.05	EI	5070
			C_2H_4	10.18	EI	5039
	$sec\text{-C}_4\text{H}_9\text{COOH}$	116-53-0	C_2H_4	10.27	EI	5039
	$\text{CH}_3\text{CH}(\text{CH}_3)\text{CH}(\text{CH}_3)\text{COOH}$	14287-61-7	C_3H_6	10.20	EI	5039
	$n\text{-C}_4\text{H}_9\text{COOCH}_3$	624-24-8	C_3H_6	10.06	EI	5039
	$iso\text{-C}_4\text{H}_9\text{COOCH}_3$	556-24-1	C_3H_6	10.16	EI	5039

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₄H₂O₂⁺	C ₄ H ₂ (=O) ₂ (3-Cyclobutene-1,2-dione)	32936-74-6	**	9.79 (V)	PE	4808
	C ₆ H ₄ O ₂ (2,5-Cyclohexadiene-1,4-dione)	106-51-4	C ₂ H ₂	11.2±0.05	PI	3523
C₄H₄O₂⁺	C ₄ H ₄ (=O) ₂ (1,2-Cyclobutanedione)	33689-28-0	**	9.61 (V)	PE	4808
	C ₄ H ₄ O ₂ (1,4-Dioxin)	290-67-5	**	7.75±0.02	PE	4740
	C ₄ H ₄ O(=O) (2(3H)-Furanone)	20825-71-2	**	10.70 (V)	PE	3826
C₄H₆O₂⁺	CH ₂ =CHCOOCH ₃	96-33-3	**	10.72 (V)	PE	3937
	CH ₃ CO ₂ CH=CH ₂	108-05-4	**	9.85±0.05 (V)	PE	4859
	(CH ₃ CO) ₂	431-03-8	**	9.47 (V)	PE	5538
			**	9.55 (V)	PE	3936
			**	9.55 (V)	PE	4520
			**	9.57 (V)	PE	4233
			**	9.72 (V)	PE	5517
	CH ₂ =CHCH ₂ COOH	625-38-7	**	10.02	PE	5086
	CH ₂ =C(CH ₃)COOH	3724-65-0	**	10.15	PE	5086
	cis-CH ₃ CH=CHCOOH	503-64-0	**	10.08	PE	5086
	trans-CH ₃ CH=CHCOOH	107-93-7	**	10.08	PE	5086
	C ₃ H ₅ COOH	1759-53-1	**	10.64	PE	5086
	(Cyclopropanecarboxylic acid)					
	C ₄ H ₆ O ₂ (1,2-Dioxin-3,6-dihydro-)	18715-02-1	**	9.66	PE	5318
	C ₄ H ₆ O ₂ (1,4-Dioxin, 2,3-dihydro-)	543-75-9	**	8.07±0.02	PE	4740
	C ₄ H ₆ O(=O) (2(3H)-Furanone, dihydro-)	96-48-0	**	10.26 (V)	PE	4742
			**	10.26 (V)	PE	3826
C₄H₈O₂⁺	HCOOCH ₂ CH ₂ CH ₃	110-74-7	**	10.50±0.05	PE	4831
			**	10.50	PE	4850
			**	10.62	PE	3718
	CH ₃ COOC ₂ H ₅	141-78-6	**	9.90±0.05	PE	4831
			**	9.90 (V)	PE	4850
			**	10.24	PE	3718
			**	10.16	EI	5039
	C ₂ H ₅ COOCH ₃	554-12-1	**	10.30 (V)	PE	4850
			**	10.15	EI	5039
	HCOOCH(CH ₃) ₂	625-55-8	**	10.44±0.05	PE	4831
			**	10.44 (V)	PE	4850
	n-C ₃ H ₇ COOH	107-92-6	**	10.22 (V)	PE	3937
			**	10.38 (V)	PE	4850
			**	10.46	PE	3874
			**	10.24	EI	5039
	iso-C ₃ H ₇ COOH	79-31-2	**	10.33±0.03	PE	3734
			**	10.33	PE	3874
			**	10.12	EI	5039
			**	10.329±0.005	PI	5161
			**	10.30 (V)	PE	3937
	C ₄ H ₈ O ₂ (1,2-Dioxane)	5703-46-8	**	10.0 (V)	PE	5212
	C ₄ H ₈ O ₂ (1,3-Dioxane)	505-22-6	**	10.1 (V)	PE	3733
			**	10.12 (V)	PE	4082

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
	$\text{C}_4\text{H}_8\text{O}_2$ (1,4-Dioxane)	123-91-1	**	9.41 (V)	PE	4082
			**	9.43 (V)	PE	3733
	$(\text{C}_2\text{H}_5)_2\text{CHCOOH}$	88-09-5	C_6H_4	10.14	EI	5039
	$\text{C}_2\text{H}_5\text{C}(\text{CH}_3)_2\text{COOH}$	595-37-9	C_6H_4	10.02	EI	5039
	$n-\text{C}_3\text{H}_7\text{COOC}_2\text{H}_5$	105-54-4	C_6H_4	10.06	EI	5039
	$sec\text{-C}_4\text{H}_9\text{COOCH}_3$	868-57-5	C_6H_4	9.81	EI	5039
	$n-\text{C}_3\text{H}_7\text{C}(\text{CH}_3)_2\text{COOH}$	1185-39-3	C_6H_6	9.96	EI	5039
	$n-\text{C}_5\text{H}_{11}\text{COOC}_2\text{H}_5$	123-66-0	$1-\text{C}_4\text{H}_8$	9.96	EI	5039
$\text{C}_4\text{H}_{10}\text{O}_2^+$	<i>tert</i> - $\text{C}_4\text{H}_9\text{OOH}$	75-91-2	**	10.24 (V)	PE	4251
$\text{C}_5\text{H}_3\text{O}_2^+$	$\text{C}_6\text{H}_5\text{COC}_4\text{H}_3\text{O}$ (Methanone, 2-furanylphenyl-)	2689-59-0	C_6H_5	12.4 ± 0.1	EI	5493
$\text{C}_5\text{H}_4\text{O}_2^+$	$\text{C}_5\text{H}_4\text{O}_2$ (4-Cyclopentene-1,3-dione)	930-60-9	**	10.25 (V)	PE	3826
	$\text{C}_4\text{H}_3\text{OCHO}$ (2-Furancarboxaldehyde)	98-01-1	**	9.50 ± 0.05	EI	3482
	$\text{C}_5\text{H}_4\text{O}(=\text{O})$ (4H-Pyran-4-one)	108-97-4	**	9.35 ± 0.05 (V)	PE	5002
$\text{C}_5\text{H}_6\text{O}_2^+$	$\text{CH}_2=\text{C}=\text{CHCOOCH}_3$	18913-35-4	**	10.02 (V)	PE	4748
	$\text{C}_5\text{H}_6(=\text{O})_2$ (1,3-Cyclopentanedione)	3859-41-4	**	9.46 ± 0.05	PE	3848
			**	9.53 (V)	PE	5020
	$\text{C}_5\text{H}_5(=\text{O})\text{OH}$ (2-Cyclopenten-1-one, 3-hydroxy-)	5870-62-2	**	9.22 ± 0.05 (V)	PE	3848
	$\text{C}_3\text{H}_2\text{O}_2(=\text{CH}_2)_2$ (1,3-Dioxolane, 4,5-bis(methylene)-)	4362-68-9	**	8.62	PE	5265
	$\text{C}_4\text{H}_3\text{O}(=\text{O})\text{CH}_3$ (2(3H)-Furanone, 5-methyl-)	591-12-8	**	9.62 ± 0.05	EI	4666
	$\text{C}_4\text{H}_3\text{O}(=\text{O})\text{CH}_3$ (2(5H)-Furanone, 5-methyl-)	591-11-7	**	10.12 ± 0.05	EI	4666
$\text{C}_5\text{H}_8\text{O}_2^+$	$\text{CH}_2=\text{C}(\text{CH}_3)\text{COOCH}_3$	80-62-6	**	10.28 (V)	PE	3937
	$\text{CH}_3\text{COCH}_2\text{COCH}_3$	123-54-6	**	8.85 ± 0.05	PE	3848
			**	9.00 (V)	PE	4195
			**	9.15 (V)	PE	5100
			**	9.18 ± 0.07 (V)	PE	3682
	$(\text{CH}_3)_2\text{C}=\text{CHCOOH}$	541-47-9	**	9.63	PE	5086
	$\text{CH}_3\text{CO}_2\text{C}(\text{CH}_3)=\text{CH}_2$	591-87-7	**	9.74 ± 0.05 (V)	PE	4859
	$\text{C}_2\text{H}_5\text{CH}=\text{CHCOOH}$	626-98-2	**	10.14	PE	5086
	$\text{HCOC}(\text{CH}_3)_2\text{CHO}$	1185-34-8	**	9.8 (V)	PE	4195
	$\text{CH}_3\text{CH}=\text{CHCH}_2\text{COOH}$	1617-32-9	**	9.41	PE	5086
	$\text{CH}_2=\text{C}(\text{C}_2\text{H}_5)\text{COOH}$	3586-58-1	**	10.06	PE	5086
	$\text{CH}_3\text{CH}=\text{C}(\text{CH}_3)\text{COOH}$	13201-46-2	**	9.50	PE	5086
	$\text{CH}_2=\text{C}(\text{CH}_3)\text{CH}_2\text{COOH}$	53774-20-2	**	9.52	PE	5086
	$\text{C}_4\text{H}_7\text{COOH}$ (Cyclobutanecarboxylic acid)	3721-95-7	**	10.35	PE	5086
	$\text{C}_5\text{H}_8\text{O}_2$ (2,3-Dioxabicyclo[2.2.1]heptane)	279-35-6	**	8.96 (V)	PE	5563
			**	8.99 (V)	PE	5212
$\text{C}_5\text{H}_9\text{O}_2^+$	$((\text{CH}_3)_2\text{CO})_2$	XXXXX-XX-X	CH_3	10.08 ± 0.05	PI	5412
	$((\text{CH}_3)_2\text{C}(\text{NO})\text{OOCCH}_3)_2$	68777-98-0		9.45	EI	4809

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$C_5H_{10}O_2^+$	$CH_3COOCH(CH_3)_2$	108-21-4	**	9.95 ± 0.05	PE	4831
			**	10.08	PE	3718
	$CH_3COOCH_2CH_2CH_3$ $HCOO(CH_2)_3CH_3$	109-60-4	**	9.92 (V)	PE	4850
		592-84-7	**	10.52 ± 0.05	PE	4831
			**	10.52 (V)	PE	4850
			**	10.54	PE	3718
	$n-C_4H_9COOH$	109-52-4	**	10.53 (V)	PE	3874
	$n-C_3H_7COOCH_3$	623-42-7	**	10.15 (V)	PE	4850
	$iso-C_4H_9COOH$	503-74-2	**	10.51 (V)	PE	3874
	$tert-C_4H_9COOH$	75-98-9	**	10.3 (V)	PE	4426
	$C_5H_{10}O_2$ (1,2-Dioxepane)	505-63-5	**	9.75 (V)	PE	5212
	$C_5H_{10}O_2(CH_3)_2$ (1,3-Dioxolane, 2,2-dimethyl-)	2916-31-6	**	9.71 (V)	PE	3733
$C_6H_4O_2^+$	$C_6H_4O_2$ (2,5-Cyclohexadiene-1,4-dione)	106-51-4	**	9.7	PI	3586
			**	9.96 ± 0.01	PI	3523
			**	9.96 ± 0.01	PI	5505
			**	9.99 ± 0.05 (V)	PE	5558
			**	10.01	PE	4463
			**	10.03 (V)	PE	3936
			**	10.11	PE	5082
	$C_6H_4(=O)_2$ (3,5-Cyclohexadiene-1,2-dione)	583-63-1	**	9.6 (V)	PE	4616
			**	9.60 (V)	PE	4808
$C_6H_5O_2^+$	$C_6H_4(OH)OCH_3$ (Phenol, 4-methoxy-)	150-76-5	CH ₃	11.10 ± 0.1	EI	3446
	$C_6H_4(OH)OOCCH_3$ (1,2-Benzenediol monoacetate)	2848-25-1	CH ₃ CO	12.54 ± 0.02	EI	3631
	$C_6H_4(OH)OOCCH_3$ (1,4-Benzenediol monoacetate)	3233-32-7	CH ₃ CO	13.83 ± 0.02	EI	3631
	$C_6H_4(NO_2)OH$ (Phenol, 4-nitro-)	100-02-7	NO	9.90 ± 0.1	EI	3447

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₆H₆O₂⁺						
	C ₆ H ₄ (OH) ₂ (1,2-Benzenediol)	120-80-9	**	8.56 (V)	PE	4891
	C ₆ H ₄ (OH) ₂ (1,3-Benzenediol)	108-46-3	**	8.63 (V)	PE	4891
	C ₆ H ₆ O ₂ (1,4-Benzenediol)	123-31-9	**	7.95±0.03	PI	3523
			**	7.95±0.05	PI	5552
			**	8.44 (V)	PE	4891
	C ₄ (=O) ₂ (CH ₃) ₂ (3-Cyclobutene-1,2-dione, 3,4-dimethyl-)	1121-15-9	**	9.06 (V)	PE	4808
			**	9.10 (V)	PE	4861
	C ₄ H ₃ OCOCH ₃ (Ethanone, 1-(2-furanyl)-)	1192-62-7	**	9.27±0.05	EI	3482
	C ₆ H ₄ (OH)OOCH ₃ (1,2-Benzenediol monoacetate)	2848-25-1	CH ₂ =C=O	9.30±0.02	EI	3631
	C ₆ H ₄ (OH)OOCCH ₃ (Benzeneacetic acid, 4-hydroxy-)	3233-32-7	CH ₂ =C=O	9.28±0.02	EI	3631
C₆H₈O₂⁺						
	C ₆ H ₈ (=O) ₂ (1,3-Cyclohexanedione)	504-02-9	**	9.52±0.05	PE	3848
			**	9.60 (V)	PE	5020
	C ₆ H ₈ (=O) ₂ (1,4-Cyclohexanedione)	637-88-7	**	9.65 (V)	PE	3936
			**	~9.85 (V)	PE	5090
	C ₅ H ₅ (=O) ₂ CH ₃ (1,3-Cyclopentanedione, 2-methyl-)	765-69-5	**	9.40±0.1 (V)	PE	3848
	C ₅ H ₄ (=O)(OH)CH ₃ (2-Cyclopenten-1-one, 3-hydroxy-2-methyl-)	5870-63-3	**	8.84±0.05	PE	3848
	C ₆ H ₈ O ₂ (2,3-Dioxabicyclo[2.2.2]oct-5-ene)	6671-70-1	**	8.76 (V)	PE	5563
	C ₄ H ₄ O ₂ (=CH ₂) ₂ (1,4-Dioxane, 2,3-bis(methylene)-)	70517-24-7	**	8.38	PE	5265
	C ₄ H ₂ O(=O)(CH ₃) ₂ (3(2H)-Furanone, 2,5-dimethyl-)	14400-67-0	**	9.23±0.05	EI	4673
	C ₃ H ₅ COCOCH ₃ (1,2-Propanedione, 1-cyclopropyl-)	15940-89-3	**	9.33 (V)	PE	4233
C₆H₁₀O₂⁺						
	CH ₃ COC(CH ₃)HCOCH ₃	815-57-6	**	8.55 (V)	PE	4195
	trans-CH ₃ CH=CHCOOC ₂ H ₅	623-70-1	**	10.11 (V)	PE	3937
	C ₆ H ₆ (=O)OH (Cyclohexanone, 2-hydroxy-)	533-60-8	**	9.70 (V)	PE	4509
	C ₆ H ₁₀ O ₂ (2,3-Dioxabicyclo[2.2.2]octane)	280-53-5	**	8.82 (V)	PE	5212
			**	8.83 (V)	PE	5563
C₆H₁₁O₂⁺						
	C ₄ H ₆ O ₂ (CH ₃) ₂ (1,3-Dioxane, 4,6-dimethyl-, cis-)	3390-18-9	H	9.693±0.005	EI	3481
	C ₄ H ₆ O ₂ (CH ₃) ₂ (1,3-Dioxane, 4,6-dimethyl-, trans-)	1121-87-5	H	9.540±0.003	EI	3481
	C ₄ H ₅ O ₂ (CH ₃) ₃ (1,3-Dioxane, 2,4,6-trimethyl-, (2 α ,4 α ,6 α -))	19145-91-6	CH ₃	9.593±0.006	EI	3481
	C ₄ H ₅ O ₂ (CH ₃) ₃ (1,3-Dioxane, 2,4,6-trimethyl-, (2 α ,4 α ,6 β -))	36402-73-0	CH ₃	9.448±0.002	EI	3481

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₆H₁₂O₂⁺	((CH ₃) ₂ CO) ₂ CH ₃ COOCH(CH ₃)C ₂ H ₅ C ₂ H ₅ COOCH ₂ CH ₂ CH ₃ CH ₃ COO(CH ₂) ₃ CH ₃ CH ₂ =C(OC ₂ H ₅) ₂ <i>tert</i> -C ₄ H ₉ COOCH ₃ <i>cis</i> -C ₅ H ₈ (OH)OCH ₃ (Cyclopentanol, 2-methoxy-, <i>cis</i> -) <i>trans</i> -C ₅ H ₈ (OH)OCH ₃ (Cyclopentanol, 2-methoxy-, <i>trans</i> -) C ₂ O ₄ (CH ₃) ₄ (1,2-Dioxetane, 3,3,4,4-tetramethyl-) C ₆ H ₁₂ O ₂ (1,2-Dioxocane)	XXXXXX-XX-X ** 105-46-4 ** 106-36-5 ** 123-86-4 ** ** ** 2678-54-8 ** 598-98-1 ** 13051-91-7 ** 7429-45-0 ** 35856-82-7 ** 6572-89-0 **		9.26±0.03 9.97±0.05 9.96 (V) 9.92±0.05 10.02±0.05 10.17 8.3 (V) 9.90±0.04 9.80 (V) 9.60 (V) 8.53 9.29 (V)	PI PE PE PE PE PE PE PE PE PE PE	5412 4831 4850 4831 4831 3718 4291 3851 4450 4450 4577 5212
C₆D₁₂O₂⁺	((CD ₃) ₂ CO) ₂	XXXXXX-XX-X **		9.25±0.03	PI	5412
C₆H₁₄O₂⁺	(<i>iso</i> -C ₃ H ₇ O) ₂	16642-57-2 **		9.16 (V)	PE	5212
C₇H₅O₂⁺	C ₆ H ₄ (OH)COOH (Benzoic acid, 3-hydroxy-) C ₆ H ₄ (OH)COOH (Benzoic acid, 4-hydroxy-) C ₆ H ₄ (COOH) ₂ (1,3-Benzenedicarboxylic acid) C ₆ H ₄ (COOH) ₂ (1,4-Benzenedicarboxylic acid)	99-06-9 99-96-7 121-91-5 100-21-0	OH OH COOH COOH	12.51±0.2 12.00±0.2 12.42±0.2 12.56±0.2	EI EI EI EI	3973 3973 3973 3973
C₇H₆O₂⁺	C ₆ H ₄ (O ₂ CH ₂) (1,3-Benzodioxole) C ₆ H ₅ COOH (Benzoic acid) C ₇ H ₆ O ₂ (Bicyclo[2.2.1]hept-5-ene-2,3-dione) C ₇ H ₆ O ₂ (2,5-Cyclohexadiene-1,4-dione, 2-methyl-) C ₆ H ₃ (=O) ₂ (CH ₃) (3,5-Cyclohexadiene-1,2-dione, 4-methyl-)	274-09-9 65-85-0 17994-26-2 553-97-9 3131-54-2	** ** ** ** **	8.21 (V) 9.75±0.2 9.75 8.73±0.05 (V) 9.78±0.02 9.78 9.40 (V)	PE EI EI PE PI PE PE	5567 3973 3792 4851 3523 4463 4808
C₇H₇O₂⁺	C ₆ H ₄ (OCH ₃) ₂ (Benzene, 1,3-dimethoxy-) C ₆ H ₄ (OCH ₃) ₂ (Benzene, 1,4-dimethoxy-) C ₆ H ₄ (NO ₂)OCH ₃ (Benzene, 1-methoxy-3-nitro-) C ₆ H ₄ (NO ₂)OCH ₃ (Benzene, 1-methoxy-4-nitro-)	151-10-0 150-78-7 555-03-3 100-17-4	CH ₃ CH ₃ NO NO	11.17±0.1 10.98±0.1 9.39±0.1 10.03±0.1	EI EI EI EI	3446 3446 3447 3447
C₇H₈O₂⁺	C ₆ H ₄ (OH)CH ₂ OH (Benzemethanol, 2-hydroxy-) C ₇ H ₈ O ₂ (Bicyclo[2.2.1]heptane-2,3-dione)	90-01-7 6236-71-1	** **	8.58 (V) 9.00±0.05 (V)	PE PE	4744 4851

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_7\text{H}_8\text{O}_2^+$	$\text{C}_6\text{H}_4(\text{OH})\text{OCH}_3$ (Phenol, 4-methoxy-)	150-76-5	**	7.50	EI	3845
			**	8.02 ± 0.1	EI	3446
	$\text{C}_6\text{H}_4(\text{OCH}_3)\text{OOCCH}_3$ (Phenol, 3-methoxy-, acetate)	5451-83-2	$\text{CH}_2=\text{C=O}$	9.56 ± 0.2	EI	3484
$\text{C}_7\text{H}_{10}\text{O}_2^+$	$\text{C}_6\text{H}_4(\text{OCH}_3)\text{OOCCH}_3$ (Phenol, 4-methoxy-, acetate)	1200-06-2	$\text{CH}_2=\text{C=O}$	9.48 ± 0.2	EI	3484
$\text{C}_7\text{H}_{10}\text{O}_2^+$	$\text{C}_5\text{H}_7(\text{OOCCH}_3)_2$ (2-Cyclopenten-1-ol)	20657-21-0	**	9.61 ± 0.05 (V)	PE	4954
	$\text{C}_6\text{H}_7(=\text{O})_2\text{CH}_3$ (1,3-Cyclohexanedione, 2-methyl-)	1193-55-1	**	9.37 ± 0.05	PE	3848
	$\text{C}_5\text{H}_4(=\text{O})_2(\text{CH}_3)_2$ (1,3-Cyclopentanenedione, 2,2-dimethyl-)	3883-58-7	**	9.08 ± 0.05	PE	3848
			**	9.22 (V)	PE	4742
			**	9.22 (V)	PE	4810
	$\text{C}_5\text{H}_5(=\text{O})_2\text{C}_2\text{H}_5$ (1,3-Cyclopentanenedione, 2-ethyl-)	823-36-9	**	9.35 ± 0.1 (V)	PE	3848
	$\text{C}_5\text{H}_4(=\text{O})(\text{OH})\text{C}_2\text{H}_5$ (2-Cyclopenten-1-one, 2-ethyl-3-hydroxy-)	5857-25-0	**	8.79 ± 0.05	PE	3848
	$\text{C}_3\text{O}_2(=\text{CH}_2)_2(\text{CH}_3)_2$ (1,3-Dioxolane, 2,2-dimethyl-4,5-bis(methylene)-)	70517-23-6	**	8.30	PE	5265
	$\text{C}_4\text{HO}(\text{CH}_3)_2\text{OCH}_3$ (Furan, 3-methoxy-2,5-dimethyl-)	57556-12-4	**	7.86 ± 0.05	EI	4673
	$\text{C}_4\text{HO}(=\text{O})(\text{CH}_3)_3$ (2(3H)-Furanone, 3,3,5-trimethyl-)	35983-73-4	**	9.00 ± 0.05	EI	4666
	$\text{C}_4\text{HO}(=\text{O})(\text{CH}_3)_3$ (3(2H)-Furanone, 2,2,5-trimethyl-)	1559-45-1	**	9.04 ± 0.05	EI	4673
	$\text{C}_7\text{H}_{10}\text{O}_2$ (Spiro[2,3-dioxabicyclo[2.2.1]heptane,7,1'-cyclopropane]	XXXXX-XX-X	**	8.87 (V)	PE	5563
$\text{C}_7\text{H}_{12}\text{O}_2^+$	$\text{CH}_3\text{COC}(\text{CH}_3)_2\text{COCH}_3$	3142-58-3	**	9.30 (V)	PE	4195
	$\text{C}_6\text{H}_9(=\text{O})\text{OCH}_3$ (Cyclohexanone, 2-methoxy-)	17429-00-4	**	9.06 (V)	PE	4509
	$\text{C}_7\text{H}_{12}\text{O}_2$ (6,7-Dioxabicyclo[3.2.2]nonane)	283-35-2	**	8.97 (V)	PE	5212
$\text{C}_7\text{H}_{13}\text{O}_2^+$	$\text{C}_4\text{H}_4\text{O}_2(\text{CH}_3)_4$ (1,3-Dioxane, 2,2,4,6-tetramethyl-, cis-)	17227-17-7	CH_3	9.332 ± 0.006	EI	3481
	$\text{C}_4\text{H}_4\text{O}_2(\text{CH}_3)_4$ (1,3-Dioxane, 2,2,4,6-tetramethyl-, trans-)	20268-00-2	CH_3	9.128 ± 0.008	EI	3481
$\text{C}_7\text{H}_{14}\text{O}_2^+$	$\text{C}_2\text{H}_5\text{COOCH}_2\text{CH}(\text{CH}_3)_2$	540-42-1	**	9.94 (V)	PE	4850
	$cis\text{-C}_6\text{H}_{10}(\text{OH})\text{OCH}_3$ (Cyclohexanol, 2-methoxy-, cis-)	7429-41-6	**	9.68 (V)	PE	4450
	$trans\text{-C}_6\text{H}_{10}(\text{OH})\text{OCH}_3$ (Cyclohexanol, 2-methoxy-, trans-)	7429-40-5	**	9.69 (V)	PE	4450
	$cis\text{-C}_5\text{H}_8(\text{OCH}_3)_2$ (Cyclopentane, 1,2-dimethoxy-, cis-)	61011-51-6	**	9.29 (V)	PE	4450
	$trans\text{-C}_5\text{H}_8(\text{OCH}_3)_2$ (Cyclopentane, 1,2-dimethoxy-, trans-)	29887-56-7	**	9.39 (V)	PE	4450
	$\text{C}_3\text{H}_2\text{O}_2(\text{CH}_3)_4$ (1,2-Dioxolane, 3,3,5,5-tetramethyl-)	22431-90-9	**	9.25 (V)	PE	4251
			**	9.26 (V)	PE	4577
	$\text{C}_8\text{H}_4\text{O}_2^+$	6383-11-5	**	9.23 (V)	PE	4861
	$\text{C}_8\text{H}_4(=\text{O})_2$ (Bicyclo[4.2.0]octa-1,3,5-triene-7,8-dione)					

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₈H₇O₂⁺						
	CH ₃ OCH ₂ COCH ₃ (Ethanone, 1-(4-methoxyphenyl)-)	100-06-1	CH ₃	10.69±0.04	EI	5059
	C ₆ H ₄ (OCH ₃)COOH (Benzoic acid, 3-methoxy-)	586-38-9	OH	12.51±0.2	EI	3973
	C ₆ H ₄ (OCH ₃)COOH (Benzoic acid, 4-methoxy-)	100-09-4	OH	12.53±0.2	EI	3973
C₈H₈O₂⁺						
	C ₆ H ₅ OC(=O)CH ₃ (Acetic acid, phenyl ester)	122-79-2	**	8.6±0.05	PE	5608
			**	8.75±0.03	EI	3483
			**	8.84±0.2	EI	3484
	C ₆ H ₄ (CHO)OCH ₃ (Benzaldehyde, 4-methoxy-)	123-11-5	**	8.43	PE	4621
	C ₆ H ₄ (CH ₃)COOH (Benzoic acid, 3-methyl-)	99-04-7	**	9.43±0.2	EI	3973
	C ₆ H ₄ (CH ₃)COOH (Benzoic acid, 4-methyl-)	99-94-5	**	9.23±0.2	EI	3973
	C ₆ H ₅ COOCH ₃ (Benzoic acid, methyl ester)	93-58-3	**	9.28	PE	4621
			**	9.34 (V)	PE	4850
			**	9.40±0.025	PE	3626
			**	9.35±0.03	EI	3626
			**	9.35±0.1	EI	3788
			**	9.49	EI	3792
	C ₈ H ₈ O ₂	60526-48-9	**	8.50±0.05 (V)	PE	4851
	(Bicyclo[2.2.1]hept-5-ene-2,3-dione, 5-methyl-)					
	C ₆ H ₂ O ₂ (CH ₃) ₂	137-18-8	**	9.58	PE	4463
	(2,5-Cyclohexadiene-1,4-dione, 2,5-dimethyl-)		**	9.60±0.05 (V)	PE	5558
	(C ₆ H ₅ COOCH ₃)(CO) ₃ Cr	12125-87-0		9.31±0.1	EI	3788
	(Chromium, tricarbonyl[(1,2,3,4,5,6-η)-methylbenzoate]-)					
C₈H₁₀O₂⁺						
	C ₆ H ₄ (OCH ₃) ₂ (Benzene, 1,2-dimethoxy-)	91-16-7	**	7.8 (V)	PE	4758
			**	8.17 (V)	PE	5567
	C ₆ H ₄ (OCH ₃) ₂ (Benzene, 1,3-dimethoxy-)	151-10-0	**	8.14 (V)	PE	5567
			**	8.18 (V)	PE	4758
			**	8.17±0.1	EI	3446
	C ₆ H ₄ (OCH ₃) ₂ (Benzene, 1,4-dimethoxy-)	150-78-7	**	7.54	PE	4621
			**	7.83±0.015 (V)	PE	4434
			**	7.90 (V)	PE	3781
			**	7.90 (V)	PE	4758
			**	7.90 (V)	PE	5403
			**	7.96 (V)	PE	5567
			**	7.45	EI	3845
			**	7.88±0.1	EI	3446
	C ₈ H ₁₀ (=O) ₂	XXXXXX-XX-X	**	9.28 (V)	PE	5020
	(Bicyclo[3.2.1]octane-2,4-dione)					
	C ₈ H ₁₀ (=O) ₂	XXXXXX-XX-X	**	9.78 (V)	PE	5090
	(cis-Bicyclo[3.3.0]octane-3,7-dione)					
	C ₃ H ₅ COCOC ₃ H ₅	XXXXXX-XX-X	**	9.09 (V)	PE	4233
	(Ethanedione, dicyclopropyl-)					
C₈H₁₂O₂⁺						
	C ₈ H ₁₂ O ₂	5941-48-0	**	8.85 (V)	PE	5010
	(trans,trans-CH ₃ CH=CHCH=CHCOOC ₂ H ₅)					
	C ₄ (=O) ₂ (CH ₃) ₄	933-52-8	**	8.80 (V)	PE	3936
	(1,3-Cyclobutanedione, 2,2,4,4-tetramethyl-)					

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₈H₁₂O₂⁺						
	C ₄ (=O) ₂ (CH ₃) ₄	933-52-8	**	8.80 (V)	PE	5090
	C ₆ H ₆ (=O) ₂ (CH ₃) ₂ (1,3-Cyclohexanedione, 5,5-dimethyl-)	126-81-8	**	9.28±0.05	PE	3848
	C ₆ H ₇ (=O)OC ₂ H ₅ (2-Cyclohexen-1-one, 3-ethoxy-)	5323-87-5	**	8.69±0.05	PE	3848
	C ₈ H ₁₂ O ₂ (7,8-Dioxabicyclo[4.2.2]dec-9-ene)	52148-56-8	**	9.00 (V)	PE	5563
	C ₈ H ₁₂ O ₂ (2,3-Dioxabicyclo[2.2.1]heptane, 7-[methylene(dimethyl)]-)	XXXXXX-XX-X	**	8.62 (V)	PE	5563
	C ₄ H ₃ O(=O)(tert-C ₄ H ₉) (2(3H)-Furanone, 5-(1,1-dimethylethyl)-)	19918-17-3	**	9.03±0.05	EI	4666
C₈H₁₄O₂⁺						
	C ₈ H ₁₄ O ₂	142-30-3	**	9.70 (V)	PE	4847
	(C=CH ₂)OC ₂ H ₅) ₂	55370-32-6	**	8.14	PE	5265
	C ₈ H ₁₄ O ₂ (7,8-Dioxabicyclo[4.2.2]decane)	52965-57-8	**	9.05 (V)	PE	5212
			**	9.06 (V)	PE	5563
	C ₈ H ₁₄ O ₂ (9,10-Dioxabicyclo[3.3.2]decane)	XXXXXX-XX-X	**	9.14 (V)	PE	5563
	C ₄ H ₂ O(O)(CH ₃) ₄ (3(2H)-Furanone, dihydro-2,2,5,5-tetramethyl-)	5455-94-7	**	9.29±0.03 (V)	PE	4292
C₈H₁₆O₂⁺						
	cis-C ₆ H ₁₀ (OCH ₃) ₂ (Cyclohexane, 1,2-dimethoxy-, cis-)	30363-80-5	**	9.24 (V)	PE	4450
	trans-C ₆ H ₁₀ (OCH ₃) ₂ (Cyclohexane, 1,2-dimethoxy-, trans-)	29887-60-3	**	9.31 (V)	PE	4450
	C ₄ H ₆ O ₂ (CH ₃) ₄ (1,2-Dioxane, 3,3,6,6-tetramethyl-)	22431-89-6	**	9.35 (V)	PE	4577
			**	9.55 (V)	PE	4251
C₈H₁₈O₂⁺						
	(tert-C ₄ H ₉ O) ₂	110-05-4	**	8.78 (V)	PE	4251
			**	8.78 (V)	PE	5212
C₉H₆O₂⁺						
	C ₉ H ₆ (=O) ₂ (1H-Indene-1,2(3H)dione)	16214-27-0	**	9.04±0.05 (V)	PE	4708
	C ₉ H ₆ (=O) ₂ (1H-Indene-1,3(2H)dione)	606-23-5	**	9.43±0.05 (V)	PE	4708
C₉H₈O₂⁺						
	C ₉ H ₈ O ₂ (Spiro[bicyclo[2.2.1]hept-5-ene-7,1'-cyclopropan]-2,3-dione)	60526-40-1	**	8.50±0.05 (V)	PE	4851
C₉H₁₀O₂⁺						
	C ₆ H ₄ (OCH ₃)(COCH ₃) (Ethanone, 1-(4-methoxyphenyl)-)	100-06-1	**	8.2±0.1	PE	4401
			**	8.65 (V)	PE	4804
	C ₆ H ₄ (CH ₃)OOCCCH ₃ (Acetic acid, 2-methylphenyl ester)	533-18-6	**	8.38±0.02	EI	3631
	C ₆ H ₄ (CH ₃)OOCCCH ₃ (Acetic acid, 3-methylphenyl ester)	122-46-3	**	8.98±0.2	EI	3484
	C ₆ H ₄ (CH ₃)OOCCCH ₃ (Acetic acid, 4-methylphenyl ester)	140-39-6	**	7.84±0.02	EI	3631
			**	8.61±0.2	EI	3484
	C ₉ H ₁₀ O ₂ (Bicyclo[2.2.1]hept-5-ene-2,3-dione, 7,7-dimethyl-)	60526-42-3	**	8.50±0.05	PE	4851
	C ₉ H ₁₀ O ₂ (Spiro[bicyclo[2.2.1]heptane-7,1'-cyclopropane]-2,3-dione)	70705-73-6	**	8.75±0.05 (V)	PE	4851

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_9\text{H}_{12}\text{O}_2^+$	$\text{C}_6\text{H}_4(\text{OCH}_3)(\text{OC}_2\text{H}_5)$ (Benzene, 1-ethoxy-4-methoxy-) $\text{C}_6\text{H}_3(\text{OCH}_3)_2\text{CH}_3$ (Benzene, 1,2-dimethoxy-4-methyl-) $\text{C}_6\text{H}_5\text{O}(\text{CH}_3)_2\text{OCH}_3$ (Benzene,(2-methoxyethoxy)-) $\text{C}_9\text{H}_{12}(=\text{O})_2$ (Bicyclo[3.2.2]nonane-2,4-dione)	5076-72-2 494-99-5 41532-81-4 XXXXX-XX-X	** ** ** **	7.72 ± 0.015 (V) 7.95 (V) 8.41 ± 0.05 9.15 (V)	PE PE EI PE	4434 4672 5484 5020
$\text{C}_9\text{H}_{14}\text{O}_2^+$	$\text{C}_6\text{H}_7(=\text{O})_2\text{CH}(\text{CH}_3)_2$ (1,3-Cyclohexanedione, 2-(1-methylethyl)-) $\text{C}_6\text{H}_5(=\text{O})_2(\text{CH}_3)_3$ (1,3-Cyclohexanedione, 2,5,5-trimethyl-) $\text{C}_5\text{H}_2(=\text{O})_2(\text{CH}_3)_4$ (1,3-Cyclopentanedione, 4,4,5,5-tetramethyl-)	3401-01-2 1125-11-7 XXXXX-XX-X	** ** **	9.09 ± 0.05 9.10 ± 0.05 9.18 (V)	PE PE PE	3848 3848 5020
$\text{C}_{10}\text{H}_6\text{O}_2^+$	$\text{C}_{10}\text{H}_6\text{O}_2$ (1,4-Naphthalenedione)	130-15-4	** **	9.56 ± 0.01 9.49	PI PE	3523 5082
$\text{C}_{10}\text{H}_8\text{O}_2^+$	$\text{C}_{10}\text{H}_6(\text{OH})_2$ (1,4-Naphthalenediol)	571-60-8	**	7.62 ± 0.03	PI	5552
$\text{C}_{10}\text{H}_{10}\text{O}_2^+$	$\text{C}_{10}\text{H}_{10}\text{O}_2$ (Bicyclo[2.2.1]hept-5-ene-2,3-dione, 7-(1-methylethylidene)-)	60526-38-7	**	8.30 ± 0.05 (V)	PE	4851
$\text{C}_{10}\text{H}_{12}\text{O}_2^+$	$\text{C}_{10}\text{H}_{12}\text{O}_2$ (2,5-Cyclohexadiene-1,4-dione, 2,3,5,6-tetramethyl-) $\text{C}_{10}\text{H}_{12}\text{O}_2$ (Tricyclo[3.3.1.1 ^{3,7}]decane-2,6-dione) $\text{C}_{10}\text{H}_{12}(=\text{O})_2$ (JC-Mean value of Jahn-Teller components) (Tricyclo[4.2.1.1 ^{2,6}]decane-7,8-dione)	527-17-3 39751-07-0 XXXXX-XX-X	** ** ** ** **	9.16 ± 0.03 9.16 ± 0.03 9.25 ± 0.05 (V) 9.06 9.07 (V)	PI PI PE PE PE	3523 5505 5558 3886 5043
$\text{C}_{10}\text{H}_{14}\text{O}_2^+$	$\text{C}_6\text{H}_4(\text{OCH}_3)(\text{OCH}(\text{CH}_3)_2)$ (Benzene, 1-methoxy-4-(1-methylethoxy)-) $\text{C}_6\text{H}_4(\text{OCH}_3)(\text{OC}_3\text{H}_7)$ (Benzene, 1-methoxy-4-propoxy-) $\text{C}_6(\text{CH}_3)_4(\text{OH})_2$ (1,4-Benzenediol,2,3,5,6-tetramethyl-) $\text{C}_6\text{H}_5\text{O}(\text{CH}_2)_3\text{OCH}_3$ (Benzene,(3-methoxypropoxy)-) $\text{C}_7\text{H}_5(\text{CH}_3)_3\text{O}_2$ (Bicyclo[2.2.1]heptane-2,3-dione,1,7,7-trimethyl-) $\text{C}_8\text{H}_{11}\text{OOCCH}_3$ (Tricyclo[3.2.1.0 ^{2,4}]octan-8-ol, acetate, <i>endo-syn</i> -) $\text{C}_8\text{H}_{11}\text{OOCCH}_3$ (Tricyclo[3.2.1.0 ^{2,4}]octan-8-ol, acetate, <i>endo-anti</i> -) $\text{C}_8\text{H}_{11}\text{OOCCH}_3$ (Tricyclo[3.2.1.0 ^{2,4}]octan-8-ol, acetate, <i>exo-syn</i> -) $\text{C}_8\text{H}_{11}\text{OOCCH}_3$ (Tricyclo[3.2.1.0 ^{2,4}]octan-8-ol, acetate, <i>exo-anti</i> -)	20744-02-9 20743-94-6 527-18-14 61372-56-3 465-29-2 32426-26-9 32350-51-9 32350-52-0 32350-50-8	** ** ** ** ** ** ** ** **	7.83 ± 0.015 (V) 7.80 ± 0.015 (V) 7.48 ± 0.05 8.42 ± 0.05 8.71 (V) 8.80 (V) 8.6 \pm 0.1 9.0 \pm 0.1 8.9 \pm 0.1 9.3 \pm 0.1	PE PE PI EI PE PE EI EI EI	4434 4434 5552 5484 5517 3936 3492 3492 3492 3492

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₀H₁₄O₂⁺	C ₈ H ₈ (OCH ₃) ₂ (Tricyclo[3.2.1.0 ^{2,4}]oct-6-ene, 8,8-dimethoxy-, (1 α ,2 α ,4 α ,5 α -))	14224-84-1	**	8.6±0.1	EI	3492
C₁₀H₁₆O₂⁺	C ₈ H ₁₀ (OCH ₃) ₂ (Bicyclo[2.2.2]oct-2-ene, 1,4-dimethoxy-) C ₆ H ₇ (=O) ₂ C(CH ₃) ₃ (1,3-Cyclohexanedione, 2-(1,1-dimethylethyl)-) C ₆ H ₄ (=O) ₂ (CH ₃) ₄ (1,3-Cyclohexanedione, 2,2,5,5-tetramethyl-) C ₆ H ₄ (=O) ₂ (CH ₃) ₄ (1,3-Cyclohexanedione, 4,4,6,6-tetramethyl-) C ₆ H ₆ O ₂ CH ₃ CH(CH ₃) ₂ (2,3-Dioxabicyclo[2.2.2]oct-5-ene, 1-methyl-4-(1-methylethyl)-)	59880-82-9 XXXXX-XX-X 702-50-1 60681-10-9 512-85-6 14224-85-2 7076-82-6	** ** ** ** ** ** **	9.24 (V) 9.05±0.1 9.04±0.05 9.29 (V) 8.07 8.7±0.1 8.9±0.1	PE PE PE PE PE PE EI	4619 3848 3848 5020 4577 4619 3492 3492
C₁₀H₁₈O₂⁺	C ₈ H ₁₂ (OCH ₃) ₂ (Bicyclo[2.2.2]octane, 1,4-dimethoxy-) C ₆ H ₈ O ₂ CH ₃ CH(CH ₃) ₂ (2,3-Dioxabicyclo[2.2.2]octane, 1-methyl-4-(1-methylethyl)-)	59880-84-1 5718-73-0 58-27-5	** ** **	9.14 (V) 8.09 8.50 (V)	PE PE PE	4619 4577 4619
C₁₁H₈O₂⁺	C ₆ H ₅ COC ₄ H ₃ O (Methanone, 2-furanylphenyl-) C ₁₀ H ₅ (=O) ₂ (CH ₃) (1,4-Naphthalenedione, 2-methyl-)	2689-59-0 58-27-5	** **	9.1±0.1 9.51 (V)	EI PE	5493 5093
C₁₁H₁₀O₂⁺	C ₆ H ₄ C ₃ (CH ₃) ₂ O ₂ (1H-Indene-1,2(3H)-dione, 3,3-dimethyl-)	20651-88-1	**	8.7 (V)	PE	5517
C₁₁H₁₂O₂⁺	C ₁₁ H ₁₂ O ₂ (Spiro[bicyclo[2.2.1]hept-5-ene-7,1'-cyclopentane]-2,3-dione)	60526-44-5	**	8.45±0.05 (V)	PE	4851
C₁₁H₁₄O₂⁺	C ₆ H ₄ (OCH ₃)(OCH ₂ C ₃ H ₅) (Benzene, 1-(cyclopropylmethoxy)-4-methoxy-)	54929-10-1	**	7.78±0.015 (V)	PE	4434
C₁₁H₁₆O₂⁺	C ₆ H ₄ (OCH ₃)(OC ₄ H ₉) (Benzene, 1-butoxy-4-methoxy-) C ₆ H ₄ (OCH ₃)(OC(CH ₃) ₃) (Benzene, 1-(1,1-dimethylethoxy)-4-methoxy-) C ₆ H ₄ (OCH ₃)(OC ₄ H ₉) (Benzene, 1-methoxy-4-(1-methylpropoxy-)) C ₆ H ₄ (OCH ₃)(OC ₄ H ₉) (Benzene, 1-methoxy-4-(2-methylpropoxy-)) C ₆ H ₅ O(CH ₂) ₄ OCH ₃ (Benzene, 4-methoxybutoxy-) C ₆ H ₇ (=O) ₂ (CH ₃) ₃ (Bicyclo[3.2.1]octane-2,4-dione, 1,8,8-trimethyl-) C ₁₀ H ₁₅ COOH (Tricyclo[3.3.1.1 ^{3,7}]decane-1-carboxylic acid)	20743-95-7 15360-00-6 51241-49-7 54929-09-8 20636-14-0 3278-94-2 828-51-3	** ** ** ** ** ** **	7.74±0.015 (V) 8.00±0.015 (V) 7.83±0.015 (V) 7.79±0.015 (V) 8.45±0.05 8.73 (V) 9.34	PE PE PE PE EI PE PE	4434 4434 4434 4434 5484 5020 3886

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₁H₁₈O₂⁺	C ₇ H ₆ (=O) ₂ (CH ₃) ₄ (1,2-Cycloheptanedione,3,3,7,7-tetramethyl-) C ₇ H ₈ O ₂ (CH ₃)C ₃ H ₇ (2,4-Dioxabicyclo[3.2.2]non-6-ene,1-methyl-5-(1-methylethyl)-)	68347-39-7 59880-80-7	** **	8.70 (V) 9.31 (V)	PE PE	5090 4619
C₁₁H₂₀O₂⁺	(CH ₃) ₃ CCOCH ₂ COC(CH ₃) ₃ C ₇ H ₁₀ O ₂ (CH ₃)C ₃ H ₇ (2,4-Dioxabicyclo[3.2.2]nonane, 1-methyl-5-(1-methylethyl)-)	1118-71-4 59880-83-0	** **	8.86±0.07 (V) 9.29 (V)	PE PE	3682 4619
C₁₂H₆O₂⁺	C ₁₂ H ₆ (=O) ₂ (1,2-Acenaphthalenedione)	82-86-0	**	8.77±0.05 (V)	PE	5095
C₁₂H₈O₂⁺	C ₁₂ H ₈ O ₂ (Dibenzo[b,e][1,4]dioxin)	262-12-4	**	7.78±0.05 (V)	PE	4743
C₁₂H₁₂O₂⁺	(C ₄ H ₂ OCH ₂ CH ₂) ₂ (13,14-Dioxatricyclo[8.2.1.1 ^{4,7}]tetradeca-4,6,10,12-tetraene) C ₁₂ H ₁₂ O ₂ (4a,8a-Ethanonaphthalene-9,10-dione, 1,4,5,8-tetrahydro-)	73650-68-7 21377-44-6	** **	7.60 8.70±0.05 (V)	PE PE	5575 4593
C₁₂H₁₄O₂⁺	C ₁₂ H ₁₄ O ₂ (4a,8a-Ethanonaphthalene-9,10-dione, 1,2,3,4,5,8-hexahydro-)	21377-45-7	**	8.60±0.05 (V)	PE	4593
C₁₂H₁₆O₂⁺	C ₁₂ H ₁₆ O ₂ (4a,8a-Ethanonaphthalene-9,10-dione, octahydro-)	21377-46-8	**	8.65±0.05 (V)	PE	4593
C₁₂H₁₈O₂⁺	C ₆ H ₅ O(CH ₂) ₅ OCH ₃ (Benzene,[(5-methoxypentyl)oxy]-) C ₁₀ H ₁₅ COOCH ₃ (Tricyclo[3.3.1.1 ^{3,7}]decane-1-carboxylic acid methyl ester)	61372-57-4 711-01-3	** **	8.51±0.05 9.38±0.03	EI PE	5484 3851
C₁₂H₂₂O₂⁺	C ₆ H ₆ CH ₃ (OCH ₃) ₂ C ₃ H ₇ (Cyclohexene, 3,6-dimethoxy-3-methyl-6-(1-methylethyl)- <i>cis</i> -)	59880-81-8	**	9.21 (V)	PE	4619
C₁₂H₂₄O₂⁺	C ₆ H ₈ CH ₃ (OCH ₃) ₂ C ₃ H ₇ (Cyclohexane, 1,4-dimethoxy-1-methyl-4-(1-methylethyl)- <i>cis</i> -)	59922-36-0	**	9.26 (V)	PE	4619
C₁₃H₈O₂⁺	C ₁₃ H ₇ (=O)OH (1H-Phenalen-1-one,9-hydroxy-) C ₁₄ H ₈ O ₂ (9H-Xanthen-9-one)	7465-58-9 90-47-1	** **	8.12±0.04 (V) 8.42±0.03	PE PI	5193 3523
C₁₃H₁₀O₂⁺	C ₆ H ₅ COOC ₆ H ₅ (Benzoinic acid, phenyl ester) C ₆ H ₅ COC ₆ H ₄ OH (Methanone, (4-hydroxyphenyl)phenyl-)	93-99-2 1137-42-4	** **	9.0 8.80±0.05 (V)	EI PE	5631 4844
C₁₃H₁₂O₂⁺	C ₆ H ₅ CH ₂ OC ₆ H ₄ OH (Phenol,4-(phenylmethoxy)-)	103-16-2	**	7.83	CTS	5336

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₃H₁₄O₂⁺	C ₁₁ H ₈ (OCH ₃) ₂ (1,4-Methanonaphthalene, 1,4-dihydro-5,8-dimethoxy-)	947-58-0	**	7.77±0.05 (V)	PE	5019
C₁₃H₂₀O₂⁺	C ₆ H ₅ O(CH ₂) ₆ OCH ₃ (Benzene,[(6-methoxyhexyl)oxy]-)	61372-58-5	**	8.48±0.05	EI	5484
C₁₄H₈O₂⁺	C ₁₄ H ₈ O ₂ (1,4-Anthracenedione)	635-12-1	**	8.45±0.02	PI	3523
	C ₁₄ H ₈ O ₂ (9,10-Anthracenedione)	84-65-1	**	9.25±0.03	PI	3523
			**	9.3	PI	3586
			**	9.25	PE	5082
			**	9.40±0.08	EI	3571
	C ₁₄ H ₈ O ₂ (9,10-Phenanthrenedione)	84-11-7	**	8.64±0.03	PI	3523
C₁₄H₉O₂⁺	(C ₆ H ₄) ₂ CH ₂ OC(=O) (Dibenz[b,e]oxepin-11(6H)-one)	4504-87-4	H	10.8	EI	5340
C₁₄H₁₀O₂⁺	(C ₆ H ₄) ₂ CH ₂ OC(=O) (Dibenz[b,e]oxepin-11(6H)-one)	4504-87-4	**	9.63	EI	5340
	(C ₆ H ₅ CO) ₂ (Ethanodione, diphenyl-)	134-81-6	**	8.9±0.05 (V)	PE	4844
			**	9.1 (V)	PE	5517
			**	8.86±0.15	EI	3823
	C ₁₃ H ₇ (=O)OCH ₃ (1H-Phenalen-1-one,9-methoxy-)	35897-82-6	**	8.14±0.04 (V)	PE	5193
C₁₄H₁₂O₂⁺	C ₁₄ H ₁₂ O ₂ (Azulene,1,3-diacetyl-)	10487-55-5	**	7.95 (V)	PE	5397
	C ₁₄ H ₁₂ O ₂ (3,6-Ethanodicyclopenta[cd,gh]pentale-7,8-dione 2a,3,3a,5a,6,6a,6b,6c-octahydro-)	68217-17-4	**	8.85 (V)	PE	4849
	C ₁₄ H ₁₀ (OH) ₂ (9,10-Phenanthrenediol,9,10-dihydro-trans-)	572-41-8	**	8.13 (V)	PE	5364
C₁₄H₁₄O₂⁺	C ₁₁ H ₉ (COOC ₂ H ₅) (1,4-Methanonaphthalene-6-carboxylic acid ethyl ester, 1,4-dihydro-)	56136-20-0	**	8.51±0.05 (V)	PE	5019
	C ₆ H ₅ O(CH ₂) ₂ OC ₆ H ₅ (Benzene,1,1'-[1,2-ethanediylbis(oxy)]bis-)	104-66-5	**	8.39±0.05	EI	5484
	C ₆ H ₅ CH ₂ OC ₆ H ₄ OCH ₃ (Benzene,1-methoxy-4-(phenylmethoxy-))	6630-18-8	**	7.76	CTS	5336
	C ₁₄ H ₁₄ O ₂ (3,6-Ethanodicyclopenta[cd,gh]pentale-7,8-dione, 1,2,2a,3,3a,5a,6,6a 6b,6c-decahydro-)	68217-18-5	**	8.80 (V)	PE	4849
C₁₄H₁₆O₂⁺	C ₁₄ H ₁₆ O ₂ (3,6-Ethanodicyclopenta[cd,gh]pentale-7,8-dione, dodecahydro-)	68217-19-6	**	8.82 (V)	PE	4849
C₁₄H₁₈O₂⁺	C ₆ H ₅ OCH ₃ (OCH(C ₃ H ₅) ₂) (Benzene, 1-(dicyclopropylmethoxy)-4-methoxy-)	54929-11-2	**	7.80±0.015 (V)	PE	4434

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₄H₂₀O₂⁺	C ₆ H ₂ (=O) ₂ (tert-C ₄ H ₉) ₂ (3,5-Cyclohexadiene-1,2-dione, 3,5-bis(1,1-dimethylethyl)-) C ₆ H ₂ (=O) ₂ (tert-C ₄ H ₉) ₂ (3,5-Cyclohexadiene-1,2-dione, 3,6-bis(1,1-dimethylethyl)-)	3383-21-9 34105-76-5	** **	8.81 (V) 8.71 (V)	PE PE	4808 4808
C₁₄H₂₂O₂⁺	C ₆ H ₆ (OCH ₃)(OC ₇ H ₁₅) (Benzene, 1-(heptyloxy)-4-methoxy-)	20743-97-9	**	7.78±0.015 (V)	PE	4434
C₁₅H₁₆O₂⁺	C ₁₅ H ₁₆ O ₂	XXXXX-XX-X	**	9.0 (V)	PE	5599
C₁₅H₁₂O₂⁺	C ₁₅ H ₁₂ O ₂ C ₆ H ₅ COCOC ₆ H ₄ CH ₃ (Ethanedione, (4-methylphenyl)phenyl-) C ₁₃ H ₇ (=O)OC ₂ H ₅ (1H-Phenalen-1-one,9-ethoxy-) C ₆ H ₅ COCH ₂ COC ₆ H ₅ (1,3-Propanedione, 1,3-diphenyl-)	XXXXX-XX-X 2431-00-7 68217-42-5 120-46-7	** ** ** **	9.22 (V) 9.05±0.10 8.06±0.04 (V) 8.45±0.05 (V)	PE EI PE PE	5599 3823 5193 4844
C₁₅H₁₆O₂⁺	C ₆ H ₅ O(CH ₂) ₃ OC ₆ H ₅ (Benzene,1,1'-[1,3-propanediylbis(oxy)]bis-)	726-44-3	**	8.46±0.05	EI	5484
C₁₆H₁₂O₂⁺	C ₁₆ H ₁₂ O ₂	XXXXX-XX-X	**	7.8 (V)	PE	5599
C₁₆H₁₄O₂⁺	C ₆ H ₅ COCH ₂ CH ₂ COC ₆ H ₅ (1,4-Butanedione, 1,4-diphenyl-)	495-71-6	**	9.2±0.05 (V)	PE	4844
C₁₆H₁₆O₂⁺	C ₁₆ H ₁₆ O ₂ (2,4-Ethanobiscyclopropane[4,5]cyclopenta[1,2,3-cd:1',2',3'-gh] pentale-5,6-dione, tetradecahydro-(1a α ,1b β ,2 α ,2a β ,2b α , 3a α ,3b β ,3c β ,3d β ,4 α ,4a β ,4b α -)	68217-20-9	**	8.6 (V)	PE	4849
C₁₆H₁₈O₂⁺	C ₆ H ₅ O(CH ₂) ₄ OC ₆ H ₅ (Benzene,1,1'-[1,4-butanediylbis(oxy)]bis-)	3459-88-9	**	8.41±0.05	EI	5484
C₁₇H₁₆O₂⁺	C ₁₇ H ₁₆ O ₂ C ₁₃ H ₇ (=O)OC ₄ H ₉ (1H-Phenalen-1-one,9-butoxy-) C ₆ H ₅ COC(CH ₃) ₂ COC ₆ H ₅ (1,3-Propanedione, 2,2-dimethyl-1,3-diphenyl-)	XXXXX-XX-X 69454-53-1 41169-42-0	** ** **	7.55 (V) 8.03±0.04 (V) 9.0±0.05 (V)	PE PE PE	5599 5193 4844
C₁₇H₁₈O₂⁺	C ₁₇ H ₁₈ O ₂ C ₁₇ H ₁₆ O ₂ (Azulene,1,3-diacetyl-4,6,8-trimethyl-)	XXXXX-XX-X 841-71-4	** **	7.5 (V) 7.5 (V)	PE PE	5599 5397
C₁₇H₂₀O₂⁺	C ₆ H ₅ O (Benzene,1,1'-[1,5-pentanediylibis(oxy)]bis-)	40339-96-6	**	8.4	EI	5484
C₁₇H₁₇D₃O₂⁺	C ₁₇ H ₁₇ D ₃ O ₂ (Benzene, 1-methoxy-3-[3-(4-methoxy-d ₃ -phenyl)propyl]-)	67081-97-4	**	7.90±0.1	EI	4925

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{17}\text{H}_{22}\text{O}_2^+$	$\text{C}_{10}\text{H}_{15}(\text{OCH}_3)(\text{OC}_6\text{H}_4)$ (Tricyclo[3.3.1.1 ^{3,7}]decane, 1-(4-methoxyphenoxy)-)	49764-17-2	**	7.82±0.015 (V)	PE	4434
$\text{C}_{18}\text{H}_{18}\text{O}_2^+$	$\text{C}_{18}\text{H}_{18}\text{O}_2$	XXXXXX-XX-X	**	7.5 (V)	PE	5599
$\text{C}_{18}\text{H}_{22}\text{O}_2^+$	$\text{C}_6\text{H}_5\text{O}(\text{CH}_2)_6\text{OC}_6\text{H}_5$ (Benzene, 1,1'-(1,6-hexanediylibis(oxyl)bis-)	10125-18-5	**	8.47±0.05	EI	5484
$\text{C}_{19}\text{H}_{20}\text{O}_2^+$	$\text{C}_{19}\text{H}_{20}\text{O}_2$	XXXXXX-XX-X	**	7.4 (V)	PE	5599
$\text{C}_{20}\text{H}_{14}\text{O}_2^+$	$\text{C}_{20}\text{H}_{12}(\text{OH})_2$ (Benzo[a]pyrene, 7,8-diol, 7,8-dihydro-, <i>trans</i> -)	57404-88-3	**	7.21 (V)	PE	5364
$\text{C}_{20}\text{H}_{22}\text{O}_2^+$	$\text{C}_{20}\text{H}_{22}\text{O}_2$ (<i>D</i> -Homoestra-1,3,5(10),6,8-pentaen-17a-one, 3-methoxy-) $\text{C}_{20}\text{H}_{22}\text{O}_2$ (<i>D</i> -Homoestra-1,3,5(10),6,8-pentaen-17a-one, 3-methoxy-, (14 β)-)	1232-90-2 1232-91-3	** **	7.56±0.07 7.82±0.07	EI EI	3571 3571
$\text{C}_{20}\text{H}_{26}\text{O}_2^+$	$\text{C}_{20}\text{H}_{26}\text{O}_2$ (<i>D</i> -Homoestra-1,3,5(10)-trien-17a-one, 3-methoxy-) $\text{C}_{20}\text{H}_{26}\text{O}_2$ (<i>D</i> -Homoestra-1,3,5(10)-trien-17a-one, 3-methoxy-, (8 α)-)	1232-89-9 1232-88-8	** **	8.22±0.06 8.17±0.08	EI EI	3571 3571
$\text{C}_{22}\text{H}_{12}\text{O}_2^+$	$\text{C}_{22}\text{H}_{12}\text{O}_2$ (6,13-Pentacenedione)	3029-32-1	**	8.07±0.05	PI	3523
$\text{C}_{23}\text{H}_{40}\text{O}_2^+$	$\text{C}_6\text{H}_4(\text{OCH}_3)(\text{OC}_{16}\text{H}_{33})$ (Benzene, 1-hexadecyloxy)-4-methoxy-)	20743-99-1	**	7.72±0.015 (V)	PE	4434
$\text{C}_{24}\text{H}_{16}\text{O}_2^+$	$\text{C}_{24}\text{H}_{16}\text{O}_2$ (Azulene, 1,3-dibenzoyl-)	XXXXXX-XX-X	**	7.7 (V)	PE	5397
$\text{C}_2\text{H}_4\text{O}_3^+$	$\text{C}_2\text{H}_4\text{O}_3$ (1,2,4-Trioxolane)	289-14-5	**	10.67±0.03 (V)	PE	4980
$\text{C}_3\text{H}_2\text{O}_3^+$	$\text{C}_3\text{H}_2\text{O}_2(=O)$ (1,3-Dioxol-2-one)	872-36-6	** **	10.08 (V) 11.91 (V)	PE PE	4549 3826
$\text{C}_3\text{H}_4\text{O}_3^+$	CH_3COOOH $\text{C}_3\text{H}_4\text{O}_2(=O)$ (1,3-Dioxolan-2-one)	127-17-3 96-49-1	** ** ** ** **	10.42 (V) 10.40 10.40 10.70 11.1 (V) 11.47 (V)	PE PE PE PE PE	4520 4471 4648 4219 4549 3826

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_5\text{H}_6\text{O}_3^+$	$\text{CH}_3\text{OCOOCH}_3$	616-38-6	**	11.00 (V)	PE	4471
			**	11.00 (V)	PE	4648
			**	11.2 (V)	PE	4549
$\text{C}_5\text{H}_6\text{O}_3^+$	$\text{C}_3\text{H}_4\text{O}_3$ (1,3,5-Trioxane)	110-88-3	**	-10.8 (V)	PE	3733
$\text{C}_4\text{H}_2\text{O}_3^+$	$\text{C}_4\text{H}_2\text{O}(=\text{O})_2$ (2,5-Furandione)	108-31-6	**	11.1 (V)	PE	4269
			**	11.11 ± 0.05 (V)	PE	4708
			**	11.45 (V)	PE	3826
$\text{C}_4\text{H}_4\text{O}_3^+$	$\text{C}_4\text{H}_4\text{O}(=\text{O})_2$ (2,5-Furandione, dihydro-)	108-30-5	**	10.8 (V)	PE	4269
			**	10.84 (V)	PE	4742
			**	10.84 (V)	PE	4810
$\text{C}_4\text{H}_6\text{O}_3^+$	$\text{CH}_3\text{COCOOCH}_3$ $\text{C}_3\text{H}_3\text{O}_2(=\text{O})\text{CH}_3$ (1,3-Dioxolan-2-one, 4-methyl-)	600-22-6 108-32-7	**	9.88 (V)	PE	4520
				10.52	PE	4219
$\text{C}_4\text{H}_{10}\text{O}_3^+$	$\text{CH}(\text{OCH}_3)_3$	149-73-5	**	10.24 ± 0.07 (V)	PE	4721
$\text{C}_5\text{H}_4\text{O}_3^+$	$\text{C}_3\text{O}_2(=\text{CH}_2)_2\text{O}$ (1,3-Dioxolan-2-one,4,5-bis(methylene)-) $\text{C}_4\text{H}_3\text{OCOOH}$ (2-Furancarboxylic acid)	62458-20-2 488-93-7	**	9.30	PE	5265
			**	9.16 ± 0.05 (V)	PE	4626
$\text{C}_5\text{H}_6\text{O}_3^+$	$\text{CH}_3(\text{C}=\text{O})_3\text{CH}_3$ $\text{C}_3\text{O}_2(=\text{O})(\text{CH}_3)_2$ (1,3-Dioxol-2-one, 4,5-dimethyl-) $\text{C}_5\text{H}_6\text{O}(=\text{O})_2$ (2H-Pyran-2,6(3H)-dione,dihydro)	921-11-9 37830-90-3 108-55-4	**	9.52 (V)	PE	5347
			**	9.10 (V)	PE	4549
			**	11.17 (V)	PE	5090
$\text{C}_5\text{H}_8\text{O}_3^+$	$\text{C}_5\text{H}_8\text{O}_3$ (6,7,8-Trioxabicyclo[3.2.1]octane)	280-21-7	**	9.63 ± 0.03 (V)	PE	4980
$\text{C}_6\text{H}_4\text{O}_3^+$	$\text{C}_6\text{H}_4\text{O}_3$ (7-oxabicyclo[2.2.1]hept-5-ene-2,3-dione)	55058-68-9	**	8.95 ± 0.05 (V)	PE	4851
$\text{C}_6\text{H}_6\text{O}_3^+$	$\text{C}_4\text{H}_3\text{OCOOCH}_3$ (2-Furancarboxylic acid, methyl ester)	611-13-2	**	9.00 ± 0.05 (V)	PE	4626
			**	9.32 ± 0.05	EI	3482
$\text{C}_6\text{H}_{10}\text{O}_3^+$	$\text{C}_6\text{H}_{10}\text{O}_3$ (7,8,9-Trioxabicyclo[4.2.1]nonane)	284-22-0	**	9.61 ± 0.03 (V)	PE	4980
$\text{C}_7\text{H}_6\text{O}_3^+$	$\text{C}_6\text{H}_4(\text{OH})\text{COOH}$ (Benzoic acid, 3-hydroxy-) $\text{C}_6\text{H}_4(\text{OH})\text{COOH}$ (Benzoic acid, 4-hydroxy-)	99-06-9 99-96-7	**	9.20 \pm 0.2	EI	3973
			**	9.22 \pm 0.2	EI	3973

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_7\text{H}_6\text{O}_3^+$	$\text{C}_6\text{H}_4(\text{COOH})\text{OOCCH}_3$ (Benzoic acid, 4-(acetoxy)-)	2345-34-8	$\text{CH}_2=\text{C=O}$	10.08 ± 0.2	EI	3484
$\text{C}_8\text{H}_4\text{O}_3^+$	$\text{C}_8\text{H}_4(\text{O=O})_2$ (2,3-Benzofurandione)	4732-72-3	**	9.65 ± 0.05 (V)	PE	4708
	$\text{C}_8\text{H}_4(\text{O=O})_2$ (1,3-Isobenzofurandione)	85-44-9	**	10.25 ± 0.05 (V)	PE	4708
$\text{C}_8\text{H}_5\text{O}_3^+$	$\text{C}_6\text{H}_4(\text{COOH})_2$ (1,3-Benzenedicarboxylic acid)	121-91-5	OH	12.17 ± 0.2	EI	3973
	$\text{C}_6\text{H}_4(\text{COOH})_2$ (1,4-Benzenedicarboxylic acid)	100-21-0	OH	12.14 ± 0.2	EI	3973
$\text{C}_8\text{H}_8\text{O}_3^+$	$\text{C}_6\text{H}_4(\text{OH}))\text{CCH}_3$ (1,2-Benzenediol monoacetate)	2848-25-1	**	8.16 ± 0.02	EI	3631
	$\text{C}_6\text{H}_4(\text{OH})\text{OOCCH}_3$ (1,4-Benzenediol monoacetate)	3233-32-7	**	8.12 ± 0.02	EI	3631
	$\text{C}_6\text{H}_4(\text{OCH}_3)\text{COOH}$ (Benzoic acid, 3-methoxy-)	586-38-9	**	9.06 ± 0.2	EI	3973
	$\text{C}_6\text{H}_4(\text{OCH}_3)\text{COOH}$ (Benzoic acid, 4-methoxy-)	100-09-4	**	9.04 ± 0.2	EI	3973
	$\text{C}_8\text{H}_8\text{O}_3$ (Bicyclo[3.2.1]octane-2,3,4-trione)	25352-00-5	**	9.49 (V)	PE	4387
$\text{C}_9\text{H}_4\text{O}_3^+$	$\text{C}_9\text{H}_4\text{O}_3$ (1H-Indene-1,2,3-trione)	938-24-9	**	9.1 (V)	PE	4387
$\text{C}_9\text{H}_7\text{O}_3^+$	$\text{C}_6\text{H}_4(\text{COOCH}_3)\text{COSC}_6\text{H}_4\text{CH}_3$ (Benzoic acid, 2-[(4-methylphenyl)thio]carbonyl-methyl ester)	42797-32-0		10.98 ± 0.2	EI	4062
	$\text{C}_8\text{H}_4\text{O}(\text{O})(\text{OCH}_3)\text{SC}_6\text{H}_4\text{CH}_3$ (1(3H)-Isobenzofuranone, 3-methoxy-3-[(4-methylphenyl)thio]-)	51053-89-5		10.7 ± 0.2	EI	4062
$\text{C}_9\text{H}_{10}\text{O}_3^+$	$\text{C}_6\text{H}_4(\text{OCH}_3)\text{CO}_2\text{CH}_3$ (Benzoic acid, 4-methoxy-,methyl ester)	121-98-2	**	8.24	PE	4621
	$\text{C}_9\text{H}_{10}\text{O}_3$ (Bicyclo[3.2.2]nonane-2,3,4-trione)	57744-40-8	**	9.14 (V)	PE	4387
	$\text{C}_6\text{H}_4(\text{OCH}_3)\text{OOCCH}_3$ (Phenol, 3-methoxy-, acetate)	5451-83-2	**	8.29 ± 0.2	EI	3484
	$\text{C}_6\text{H}_4(\text{OCH}_3)\text{OOCCH}_3$ (Phenol, 4-methoxy-, acetate)	1200-06-2	**	7.92 ± 0.2	EI	3484
$\text{C}_9\text{H}_{12}\text{O}_3^+$	$\text{C}_5\text{O}_3(\text{CH}_3)_4$ (1,2,3 Cyclopentanetrione, 4,4,5,5-tetramethyl-)	1889-98-1	**	9.00 (V)	PE	4387
$\text{C}_{10}\text{H}_{18}\text{O}_3^+$	$((\text{CH}_3)_2\text{CO})_3$	XXXXX-XX-X	**	9.10 ± 0.03	PI	5412
$\text{C}_{10}\text{H}_6\text{O}_3^+$	$\text{C}_{10}\text{H}_5\text{O}_2(\text{OH})$ (1,4-Naphthalenedione, 5-hydroxy-)	481-39-0	**	8.70 ± 0.02	PI	3523
$\text{C}_{10}\text{H}_{14}\text{O}_3^+$	$\text{C}_6\text{H}_2\text{O}_3(\text{CH}_3)_4$ (1,2,3 Cyclohexanetrione, 4,4,6,6-tetramethyl-)	57744-39-5	**	9.10 (V)	PE	4387

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{10}\text{H}_{16}\text{O}_3^+$	$\text{CH}(\text{OCH}_2\text{CH}=\text{CH}_2)_3$ $\text{C}_6\text{H}_4\text{O}(=\text{O})_2(\text{CH}_3)_4$ (1,2-Cycloheptanedione-5-oxa,3,3,7,7-tetramethyl-)	16754-50-0 XXXXX-XX-X	** **	9.80 ± 0.07 (V) 8.90 (V)	PE PE	4721 5090
$\text{C}_{12}\text{H}_6\text{O}_3^+$	$\text{C}_{12}\text{H}_6\text{O}(=\text{O})_2$ (1H,3H-Naphtho[1,8- <i>cd</i>]pyran-1,3-dione)	81-84-5	**	8.92 ± 0.05 (V)	PE	5095
$\text{C}_{12}\text{H}_{22}\text{O}_3^+$	$\text{C}_4\text{H}_8\text{O}_3(\text{tert-}\text{C}_4\text{H}_9)_2$ (2,3,7-Trioxabicyclo[2.2.1]heptane, 1,4-bis(1,1-dimethylethyl)-)	XXXXX-XX-X	**	9.00 (V)	PE	5563
$\text{C}_{14}\text{H}_8\text{O}_3^+$	$\text{C}_{14}\text{H}_7\text{O}_2(\text{OH})$ (9,10-Anthracenedione, 1-hydroxy-) $\text{C}_{14}\text{H}_7\text{O}_2(\text{OH})$ (9,10-Anthracenedione, 2-hydroxy-)	129-43-1 605-32-3	** **	8.43 ± 0.05 8.70 ± 0.03	PI PI	3523 3523
$\text{C}_{14}\text{H}_{10}\text{O}_3^+$	$\text{C}_6\text{H}_5\text{COCOC}_6\text{H}_4\text{OH}$ (Ethanedione, (4-hydroxyphenyl)phenyl-)	38469-73-7	**	8.9 ± 0.05 (V)	PE	4844
$\text{C}_{14}\text{H}_{12}\text{O}_3^+$	$\text{C}_6\text{H}_5\text{COOC}_6\text{H}_4\text{OCH}_3$ (Phenol, 4-methoxy-, benzoate)	1523-19-9	**	8.6	EI	5631
$\text{C}_{18}\text{H}_{18}\text{O}_3^+$	$\text{C}_{18}\text{H}_{18}\text{O}_3^+$ (1,3-Propanedione, 1-(4-methoxyphenyl)-2,2-dimethyl-3-phenyl-)	71591-81-6	**	8.6 ± 0.05 (V)	PE	4844
$\text{C}_{20}\text{H}_{14}\text{O}_3^+$	$\text{C}_{20}\text{H}_{12}(\text{OH})_2\text{O}$ (Benz[10,11]chrysene[3,4- <i>b</i>]oxirene-7,8-diol,7,8,8a,9a-tetrahydro- (7 <i>α</i> ,8 <i>β</i> ,8 <i>aα</i> ,9 <i>aα</i> -)	60268-85-1	**	7.13 (V)	PE	5364
$\text{C}_2\text{H}_2\text{O}_4^+$	HOCOCOOH	144-62-7	** ** **	11.20 (V) 11.20 (V) 11.20 (V)	PE PE PE	4487 4648 5517
$\text{C}_2\text{H}_4\text{O}_4^+$	$(\text{HCOOH})_2$	14523-98-9	**	11.3 (V)	PE	3734
$\text{C}_3\text{H}_4\text{O}_4^+$	$\text{CH}_2(\text{COOH})_2$	141-82-2	**	11.05 (V)	PE	5243
$\text{C}_4\text{H}_4\text{O}_4^+$	<i>trans</i> -HO ₂ CCH=CHCO ₂ H	110-17-8	**	10.9 (V)	PE	4464
$\text{C}_4\text{H}_6\text{O}_4^+$	$\text{CHCH}_3(\text{COOH})_2$ $\text{CH}_3\text{OCOCOOCH}_3$	516-05-2 553-90-2	** **	10.80 (V) 10.30 (V)	PE PE	5243 4648
$\text{C}_4\text{H}_8\text{O}_4^+$	$(\text{CH}_3\text{COOH})_2$	6993-75-5	**	10.6 (V)	PE	3734
$\text{C}_6\text{H}_6\text{O}_4^+$	$\text{CH}_3\text{OOC}\equiv\text{CCOOCH}_3$ $\text{C}_4(=\text{O})_2(\text{CH}_3\text{O})_2$ (3-Cyclobutene-1,2-dione, 3,4-dimethoxy-)	762-42-5 5222-73-1	** **	10.9 (V) 9.20 (V)	PE PE	3937 4861

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_6\text{H}_8\text{O}_4^+$						
	cis-(CH ₃ OC(O)CH) ₂	624-48-6	**	10.3 (V)	PE	4464
			**	10.47 (V)	PE	3937
	trans-(CH ₃ OC(O)CH) ₂	624-49-7	**	10.5 (V)	PE	4464
			**	10.70 (V)	PE	3937
$\text{C}_6\text{H}_{10}\text{O}_4^+$						
	C ₂ H ₅ OCOCOOCC ₂ H ₅	95-92-1	**	10.19 (V)	PE	4648
$\text{C}_6\text{H}_{12}\text{O}_4^+$						
	(CH ₃ CH ₂ COOH) ₂	XXXXX-XX-X	**	10.4 (V)	PE	3734
$\text{C}_7\text{H}_7\text{O}_4^+$						
	C ₁₄ H ₂₀ O ₁₀ (α -D-Galactopyranosiduronic acid, methyl methyl ester, triacetate)	35785-35-4		10.81	EI	5227
$\text{C}_7\text{H}_{12}\text{O}_4^+$						
	C(C ₂ H ₅) ₂ (COOH) ₂	510-20-3	**	10.40 (V)	PE	5243
$\text{C}_8\text{H}_6\text{O}_4^+$						
	C ₆ H ₄ (COOH) ₂ (1,3-Benzenedicarboxylic acid)	121-91-5	**	9.98±0.2	EI	3973
	C ₆ H ₄ (COOH) ₂ (1,4-Benzenedicarboxylic acid)	100-21-0	**	9.86±0.2	EI	3973
$\text{C}_8\text{H}_{16}\text{O}_4^+$						
	C ₈ H ₁₆ O ₄ (1,4,7,10-Tetraoxacyclododecane)	294-93-9	**	9.3 (V)	PE	5104
$\text{C}_9\text{H}_8\text{O}_4^+$						
	C ₆ H ₄ (COOH)OOCCH ₃ (Benzoic acid, 4-(acetoxy)-)	2345-34-8	**	9.11±0.2	EI	3484
$\text{C}_{10}\text{H}_6\text{O}_4^+$						
	C ₁₀ H ₄ O ₂ (OH) ₂ (1,4-Naphthalenedione, 5,8-dihydroxy-)	475-38-7	**	8.20±0.02	PI	3523
$\text{C}_{12}\text{H}_{24}\text{O}_4^+$						
	((CH ₃) ₂ CO) ₄	XXXXX-XX-X	**	9.02±0.03	PI	5412
$\text{C}_{14}\text{H}_8\text{O}_4^+$						
	C ₁₄ H ₆ O ₂ (OH) ₂ (9,10-Anthracenedione, 1,4-dihydroxy-)	81-64-1	**	7.94±0.03	PI	3523
	C ₁₄ H ₆ O ₂ (OH) ₂ (9,10-Anthracenedione, 1,5-dihydroxy-)	117-12-4	**	8.53±0.03	PI	3523
	C ₁₄ H ₆ O ₂ (OH) ₂ (9,10-Anthracenedione, 2,6-dihydroxy-)	84-60-6	**	8.65±0.05	PI	3523
$\text{C}_{16}\text{H}_{14}\text{O}_4^+$						
	C ₆ H ₄ (COOCH ₃)C ₆ H ₄ COOCH ₃ ([1,1'-Biphenyl]-2,2'-dicarboxylic acid dimethyl ester)	5807-64-7	**	8.90±0.05	EI	4199
	C ₆ H ₄ (COOCH ₃)C ₆ H ₄ COOCH ₃ ([1,1'-Biphenyl]-4-4'-dicarboxylic acid dimethyl ester)	792-74-5	**	9.15±0.05	EI	4199
	(C ₆ H ₅ CH ₂ OC=O) ₂ (Ethanedioic acid bis(phenylmethyl)ester)	7579-36-4	**	9.1 (V)	PE	4609
$\text{C}_{22}\text{H}_{10}\text{O}_4^+$						
	C ₂₂ H ₁₀ O ₄ (5,7,12,14-Pentacenetrone)	23912-79-0	**	9.22±0.05	PI	3523
$\text{C}_9\text{H}_9\text{O}_5^+$						
	C ₁₄ H ₂₀ O ₁₀ (α -D-Galactopyranosiduronic acid, methyl methyl ester, triacetate)	35785-35-4		10.50	EI	5227

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{10}\text{H}_{20}\text{O}_5^+$	$\text{C}_{10}\text{H}_{20}\text{O}_5$ (1,4,7,10,13-Pentaoxacyclopentadecane)	33100-27-5	**	9.58 (V)	PE	5104
$\text{C}_9\text{H}_{11}\text{O}_6^+$	$\text{C}_{14}\text{H}_{20}\text{O}_{10}$ (α -D-Galactopyranosiduronic acid, methyl methyl ester, triacetate)	35785-35-4		10.46	EI	5227
$\text{C}_{10}\text{H}_{20}\text{O}_6^+$	$\text{C}_{10}\text{H}_{20}\text{O}_2(=O)_4$ (1 <i>H</i> ,3 <i>H</i> -Benz[1,2-c:4,5-c']difuran-1,3,5,7-tetron)	89-32-7	**	12.19±0.02	PI	4174
$\text{C}_{12}\text{H}_{24}\text{O}_6^+$	$\text{C}_{12}\text{H}_{24}\text{O}_6$ (1,4,7,10,13,16-Hexaoxacyclooctadecane)	17455-13-9	**	9.70 (V)	PE	5104
$\text{C}_{14}\text{H}_8\text{O}_6^+$	$\text{C}_{14}\text{H}_4\text{O}_2(\text{OH})_4$ (Anthraquinone, 1,4,5,8-tetrahydroxy-)	81-60-7	**	7.83±0.02	PI	3523
$\text{C}_{20}\text{H}_{24}\text{O}_6^+$	$\text{C}_{20}\text{H}_{24}\text{O}_6$ (Dibenzo[<i>b,k</i>][1,4,7,10,13,16]hexaoxacyclooctadecin,6,7,9,10,17,18,20,21-octahydro-)	14187-32-7	**	7.70 (V)	PE	5104
$\text{C}_{20}\text{H}_{36}\text{O}_6^+$	$\text{C}_{20}\text{H}_{36}\text{O}_6$ (Dibenzo[<i>b,k</i>][1,4,7,10,13,16]hexaoxacyclooctadecin,eicosahydro-)	16069-36-6	**	9.45 (V)	PE	5104
$\text{C}_{11}\text{H}_{13}\text{O}_7^+$	$\text{C}_{14}\text{H}_{20}\text{O}_{10}$ (α -D-Galactopyranosiduronic acid, methyl methyl ester, triacetate)	35785-35-4	$\text{OCH}_3, \text{CH}_3\text{COOH}$	10.27	EI	5227
$\text{C}_{13}\text{H}_{17}\text{O}_9^+$	$\text{C}_{14}\text{H}_{20}\text{O}_{10}$ (α -D-Galactopyranosiduronic acid, methyl methyl ester, triacetate)	35785-35-4	OCH_3	10.10	EI	5227
$\text{C}_{14}\text{H}_{20}\text{O}_{10}^+$	$\text{C}_{14}\text{H}_{20}\text{O}_{10}$ (α -D-Galactopyranosiduronic acid, methyl methyl ester, triacetate)	35785-35-4	**	9.96	EI	5227
$\text{BeC}_{10}\text{H}_{14}\text{O}_4^+$	$(\text{CH}_3\text{COCHCOCH}_3)_2\text{Be}$ (Beryllium, bis(2,4-pentanedionato- <i>O,O'</i>)-(T-4)-)	10210-64-7	**	8.41±0.07 (V)	PE	3682
BCH_3O^+	$(\text{BH}_3)(\text{CO})$	13205-44-2	**	11.14±0.02	PE	3699
$\text{BC}_3\text{H}_9\text{O}^+$	$(\text{CH}_3)_2\text{BOCH}_3$	4443-43-0	**	10.32 (V)	PE	4065
$\text{BC}_3\text{H}_9\text{O}_2^+$	$(\text{CH}_3\text{O})_2\text{BCH}_3$	7318-81-2	**	10.40 (V)	PE	4065
$\text{BC}_6\text{H}_{11}\text{O}_2^+$	$\text{C}_6\text{H}_5\text{B}(\text{OCH}_3)_2$ (Boronic acid, phenyl-dimethyl ester)	13471-35-7	**	9.25±0.05 (V)	PE	4956
$\text{BC}_3\text{H}_9\text{O}_3^+$	$\text{B}(\text{OCH}_3)_3$	121-43-7	**	10.40 (V)	PE	4065

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
NO⁺						
(¹ S ⁺)	NO	10102-43-9	**	9.26436±0.00006	S	5144
(³ P, ¹ P)			**	21.72	S	4176
(³ P)			**	21.721±0.006	S	3761
(¹ S ⁺)			**	9.26 (V)	PE	4843
(¹ S ⁺)			**	9.262±0.003	PE	3516
(¹ S ⁺)			**	9.27	PE	4073
(³ S ⁺)			**	15.667±0.003	PE	3516
(³ P)			**	16.562±0.003	PE	3516
(³ A)			**	16.863±0.003	PE	3516
(³ S ⁻)			**	17.586±0.003	PE	3516
(¹ S ⁻)			**	17.811±0.003	PE	3516
(³ P)			**	18.319±0.003	PE	3516
(³ P)			**	21.722±0.010	PE	3516
(³ P)			**	21.722±0.010	PE	3516
(¹ S ⁺)			**	22.727±0.10	PE	3516
(¹ S ⁺)			**	9.27±0.05	EI	3453
N ₂ O		10024-97-2	N	15.01	PI	4356
			N(² D°)	16.53±0.01	PI	4356
			N(² P°)	17.73±0.01	PI	4356
			N	16±1	PI	5170
CH ₃ NO ₂		75-52-5		11.75±0.01	PI	3524
CH ₃ ONO		624-91-9	CH ₃ O	10.917±0.008	PI	3524
((CH ₃) ₂ C(CN)NO) ₂		31018-29-8		10.20	EI	4809
((CH ₃) ₂ C(NO)COCH ₃) ₂		30442-79-6		10.50	EI	4809
(C ₆ H ₁₁ NO ₂) ₂		68777-99-1		12.20	EI	4809
((CH ₃) ₂ C(NO)COOCH ₃) ₂		6144-15-6		9.90	EI	4809
((CH ₃) ₂ C(NO)OOCCH ₃) ₂		68777-98-0		10.80	EI	4809
((CH ₃) ₂ C(NO ₂)NO) ₂		5275-46-7		10.50	EI	4809
CF ₃ NO		XXXXX-XX-X	CF ₃	12.4±0.1	EI	5220
CINO		XXXXX-XX-X	Cl	11.0±0.02	EI	5220
(CH ₃) ₂ CCINO		2421-26-3	CH ₃ CCl	12.75	EI	4809
(CH ₃) ₂ CBrNO		7119-91-7		11.10	EI	4809
NO⁺²						
(² S ⁺ , ² P)	NO	10102-43-9	**	39.3±0.5	OTH	5007
(² S ⁺)			**	42.4±1.0	OTH	5007
(² S ⁺ , ² P)			**	47.2±0.5	OTH	5007
N₂O⁺						
(² P _{3/2})	N ₂ O	10024-97-2	**	12.88±0.005	PI	4356
(² P _{1/2})			**	12.89±0.005	PI	4356
(² S ⁺)			**	16.37±0.01	PI	4356
(² P)			**	12±1	PI	5170
(² P)			**	12.886±0.002	PE	4752
(² P)			**	12.89 (V)	PE	5055
(² S ⁺)			**	12.90	PE	3998
(² S ⁺)			**	16.388±0.001	PE	4752
(² S ⁺)			**	16.40	PE	3998
(² S ⁺)			**	20.105±0.002	PE	4752
(² S ⁺)			**	12.91±0.03	EI	4877
(CH ₃) ₂ CBrNO		7119-91-7		13.15	EI	4809
N₂O²⁺						
N ₂ O		10024-97-2	**	37.3±0.5	OTH	5147
NO₂⁺						
NO ₂		10102-44-0	**	<9.62±0.01	PI	3927
			**	10.4±0.3	EI	5176
			**	35.0±0.5	EI	5176

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
N_2O_4^+	N_2O_4	10544-72-6	** ** ** ** ** **	10.8 ± 0.2 11.4 ± 0.1 (V) 11.4 ± 0.1 (V) 11.45 ± 0.1 (V) 11.6 (V) $11-12$ (V)	PE PE PE PE PE PE	4700 4709 5262 5383 4561 4631
N_2O_5^+	N_2O_5	10102-03-1	**	12.3 (V)	PE	4561
HNO^+	HNO $((\text{CH}_3)_2\text{C}(\text{NO})\text{OOCCH}_3)_2$ $(\text{CH}_3)_2\text{CBrNO}$	14332-28-6 68777-98-0 7119-91-7	**	8.6 (V) 14.20 12.75	PE EI EI	4467 4809 4809
H_3NO^+	NH_2OH	7803-49-8	**	10.59 (V) 10.64 (V)	PE PE	4768 5288
$\text{C}_6\text{H}_7\text{NO}^+$	$\text{C}_5\text{H}_4\text{N}(\text{O})\text{CH}_3$ (Pyridine, 2-methyl-, 1-oxide)	931-19-1	**	8.21 ± 0.02 (V)	PE	4275
HNO_2^+	HNO_2	7782-77-6	**	11.3 (V)	PE	4467
HNO_3^+	HNO_3	7697-37-2	** ** **	11.95 ± 0.01 11.96 12.2 (V)	PE PE PE	4477 4404 4561
$\text{C}_2\text{N}_2\text{O}^+$	NCNCO	22430-66-6	**	11.49 ± 0.02	PE	4746
$\text{C}_3\text{N}_2\text{O}^+$	$(\text{CN})_2\text{CO}$	1115-12-4	**	12.56 (V)	PE	3726
CNO_2^+	$((\text{CH}_3)_2\text{C}(\text{NO}_2)\text{NO})_2$	5275-46-7		10.15	EI	4809
$\text{C}_6\text{H}_5\text{NO}_3^+$	$\text{C}_6\text{H}_4(\text{OH})\text{NO}_2$ (Phenol, 4-nitro-)	100-02-7	**	7.38	EI	4089
CHNO^+	HNCO HCNO	75-13-8 506-85-4	** **	11.62 ± 0.02 10.83	PE PE	3670 4595
CH_2NO^+	HCONH ₂ CH_3CONH_2 $(\text{NH}_2)_2\text{CO}$ $\text{NHCH}_3\text{CONH}_2$ $\text{N}(\text{CH}_3)_2\text{CONH}_2$	75-12-7 60-35-5 57-13-6 598-50-5 1320-50-9		12.00 11.60 12.90 13.25 13.70	EI EI EI EI EI	4878 4878 4878 4878 4878
CH_3NO^+	HCONH ₂ $\text{CH}_2=\text{NOH}$ CH_3NO	75-12-7 75-17-2 865-40-7	** ** **	10.16 ± 0.03 10.50 ± 0.05 10.62 (V) 8.68 ± 0.1 (V) 9.76 ± 0.05 (V)	PI EI PE PE PE	3765 4759 4650 4465 5298

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
CH₃NO⁺						
	CH ₃ NO	865-40-7	** **	9.8 9.8 (V)	PE PE	4379 4467
CH₅NO⁺						
	H ₂ NOCH ₃	67-62-9	**	10.25 (V) 10.28 (V)	PE PE	5288 4768
	CH ₃ NHOH	593-77-1		9.82 (V)	PE	5288
C₂H₃NO⁺						
	CH ₃ NCO	624-83-9	**	10.67±0.02	PE	3670
C₂H₄NO⁺						
	HCONHCH ₃	123-39-7		11.20	EI	4878
	CH ₃ CONHCH ₃	79-16-3		11.80	EI	4878
	(NHCH ₃) ₂ CO	96-31-1		11.90	EI	4878
	N(CH ₃) ₂ CONHCH ₃	632-14-4		12.40	EI	4878
	((CH ₃) ₂ C(NO)OOCCH ₃) ₂	68777-98-0		10.30	EI	4809
	((CH ₃) ₂ C(NO ₂)NO) ₂	5275-46-7		10.15	EI	4809
	(CH ₃) ₂ CBrNO	7119-91-7		10.10	EI	4809
C₂H₅NO⁺						
	CH ₃ CONH ₂	60-35-5	** ** ** ** ** **	9.65±0.03 9.62 9.62 9.80 10.15±0.05	PI PE PE PE EI	3765 4471 4520 3718 4759
	CH ₃ CH=NOH	107-29-9	**	10.20 (V)	PE	4650
	HCONHCH ₃	123-39-7	**	10.00±0.05	EI	4759
	C ₂ H ₅ NO	925-91-7	**	10.1±0.2	EI	4099
C₂H₇NO⁺						
	NH ₂ CH ₂ CH ₂ OH	141-43-5	**	9.87±0.06 (V)	PE	3987
	CH ₃ NHOCH ₃	1117-97-1		9.48 (V)	PE	5288
	(CH ₃) ₂ NOH	5725-96-2		9.18 (V)	PE	5288
C₃H₂NO⁺						
	C ₃ H ₃ NO (Oxazole)	288-42-6	H	12.7	EI	5400
C₃H₃NO⁺						
	CH ₂ =CHNCO	3555-94-0	**	9.80±0.1 (V)	PE	5541
	C ₃ H ₃ NO (Isoxazole)	288-14-2	**	10.20 (V)	PE	5213
	C ₃ H ₃ NO (Oxazole)	288-42-6	**	9.6	EI	5400
C₃H₅NO⁺						
	C ₂ H ₅ NCO	109-90-0	**	10.32±0.05 (V)	PE	5026
	(CH ₃) ₂ CBrNO	7119-91-7		10.60	EI	4809
C₃H₆NO⁺						
	HCON(CH ₃) ₂	68-12-2		11.35	EI	4878
	CH ₃ CON(CH ₃) ₂	127-19-5		11.60	EI	4878
	N(CH ₃) ₂ CONHCH ₃	632-14-4		12.40	EI	4878
	((CH ₃) ₂ N) ₂ CO	632-22-4		11.75	EI	4878
	((CH ₃) ₂ C(NO)OOCCH ₃) ₂	68777-98-0		10.25	EI	4809
	(CH ₃) ₂ CCINO	2421-26-3		11.80	EI	4809
	(CH ₃) ₂ CBrNO	7119-91-7		10.35	EI	4809
C₃H₇NO⁺						
	HCON(CH ₃) ₂	68-12-2	**	9.45±0.05	EI	4759

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$C_3H_7NO^+$	$CH_3CONHCH_3$ $(CH_3)_2C=NOH$ C_3H_6ONH (Isoxazolidine) $((CH_3)_2C(NO)OOCCH_3)_2$	79-16-3 127-06-0 504-72-3 68777-98-0	**	9.85 (V)	PE	3718
			**	9.70 ± 0.05	EI	4759
			**	9.67 (V)	PE	4650
			**	9.57 (V)	PE	5301
				10.70	EI	4809
$C_3H_9NO^+$	$CH_3OCH_2CH_2NH_2$ $NH_2(CH_2)_3OH$ $(CH_3)_3NO$ $(CH_3)_2NOCH_3$	109-85-3 156-87-6 1184-78-7 5669-39-6	**	9.45 ± 0.09 (V)	PE	3987
			**	9.77 ± 0.20 (V)	PE	3987
			**	8.27 (V)	PE	4537
			**	8.375 ± 0.035 (V)	PE	5529
				8.81 (V)	PE	5288
$C_4H_7NO^+$	$C_4H_7N(=O)$ (2-Pyrrolidinone)	616-45-5	**	9.53 (V)	PE	4742
$C_4H_8NO^+$	$(C_6H_{11}NO_2)_2$	68777-99-1		9.40	EI	4809
$C_4H_9NO^+$	$C_3H_6ON(CH_3)$ (Isoxazolidine, 2-methyl-) $CH_3CON(CH_3)_2$ $n-C_3H_7CH=NOH$ <i>tert</i> - C_4H_9NO C_4H_9NO (Morpholine)	22445-44-9 127-19-5 110-69-0 917-95-3 110-91-8 36652-42-3	**	8.60 (V)	PE	5301
			**	9.43 (V)	PE	3718
			**	9.20 ± 0.05	EI	4759
			**	9.93 (V)	PE	4650
			**	7.99 ± 0.1 (V)	PE	4465
			**	8.95 (V)	PE	4719
			**	9.05 ± 0.05 (V)	PE	5298
			**	8.88 ± 0.05	PE	4654
			**	8.88 ± 0.05 (V)	PE	4819
			**	8.91 ± 0.03 (V)	PE	4452
				9.00 (V)	PE	5301
$C_4H_{11}NO^+$	$(CH_3)_2NC_2H_4OH$	108-01-0	**	8.82 (V)	PE	4537
			**	8.85 ± 0.04 (V)	PE	3987
	$CH_3O(CH_2)_3NH_2$	5332-73-0	**	9.37 ± 0.12 (V)	PE	3987
$C_5H_3NO^+$	C_4H_3OCN (2-Furancarbonitrile)	617-90-3	**	9.47 ± 0.05 (V)	PE	4626
			**	9.77 ± 0.05	EI	3482
$C_5H_5NO^+$	$C_5H_4N(OH)$ (2-Pyridinol) $C_5H_4N(OH)$ (3-Pyridinol) $C_5H_4N(OH)$ (4-Pyridinol) C_5H_5NO (Pyridine, 1-oxide)	109-10-4 109-00-2 626-64-2 694-59-7	**	9.11 ± 0.03 (V)	PE	4711
			**	9.28 ± 0.02	EI	3636
			**	9.15 ± 0.03 (V)	PE	4711
			**	9.5 ± 0.1	EI	4302
			**	9.55 ± 0.02	EI	3636
			**	9.55 ± 0.05	EI	3635
			**	9.8 ± 0.03	PE	4711
			**	9.6 ± 0.1	EI	4302
			**	9.89 ± 0.02	EI	3636
			**	8.38 ± 0.02	PE	4470

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₅H₅NO⁺						
	C ₅ H ₅ NO	694-59-7	**	8.38±0.02 (V)	PE	4275
			**	8.46 (V)	PE	4222
	C ₅ H ₄ NOH (2-Pyridinol)	72762-00-6	**	8.62 (V)	PE	5191
	C ₅ H ₄ NH(=O) (2(1H)-Pyridinone)	142-08-5	**	8.62±0.03 (V)	PE	4711
	C ₅ H ₄ NH(=O) (2(1H)-Pyridinone)	142-08-5	**	9.0±0.1	EI	4302
	C ₄ H ₄ NCHO (1-H-Pyrrole-2-carboxaldehyde)	1003-29-8	**	8.93±0.05	EI	3482
C₅H₈NO⁺						
	(CH ₃) ₂ NCOCH=CHCH ₃	23135-18-4	CH ₃	11.0±0.1	EI	3996
C₅H₉NO⁺						
	n-C ₄ H ₉ NCO	111-36-4	**	10.14±0.05 (V)	PE	5026
	tert-C ₄ H ₉ CNO	1609-86-5	**	9.57 (V)	PE	4674
	tert-C ₄ H ₉ C=NO	27143-81-3	**	9.55±0.05 (V)	PE	4719
	C ₃ H ₅ CHN(CH ₃)O (Methanaminium,N-(cyclopropylmethylene)-N-hydroxy-hydroxide,inner salt)	65194-05-0	**	8.30	PE	5099
	C ₄ H ₆ N(=O)CH ₃ (3-Pyrrolidinone, 1-methyl-)	68165-06-0	**	8.83 (V)	PE	4742
C₅H₁₁NO⁺						
	C ₄ H ₈ ON(CH ₃) (2H-1,2-Oxazine,tetrahydro-2-methyl-)	22445-43-8	**	8.66 (V)	PE	5301
	n-C ₃ H ₇ CHNO(CH ₃) (Oxaziridine, 2-methyl-3-propyl-)	58751-77-2	**	9.40±0.05	EI	4677
	CH ₃ COCH ₂ N(CH ₃) ₂	15364-56-4	**	7.71±0.05	PE	4192
	n-C ₃ H ₇ CONHCH ₃	17794-44-4	**	9.68±0.05	EI	4677
	n-C ₃ H ₇ CH=NOCH ₃	31376-98-4	**	9.33±0.05	EI	4677
	n-C ₃ H ₇ CH=N(O)CH ₃	44603-43-2	**	8.57±0.05	EI	4677
	tert-C ₄ H ₉ N(=CH ₂)O	41012-82-2	**	8.64	PE	5099
			**	8.64 (V)	PE	4719
	cis-C ₅ H ₈ (OH)NH ₂ (Cyclopentanol, cis-2-amino-)	57070-95-8	**	8.61	PE	4399
	trans-C ₅ H ₈ (OH)NH ₂ (Cyclopentanol, trans-2-amino-)	59260-76-3	**	8.30	PE	4399
C₅H₁₃NO⁺						
	(CH ₃) ₂ N(CH ₂) ₃ OH	3179-63-3	**	8.74±0.04 (V)	PE	3987
C₆H₅NO⁺						
	C ₆ H ₅ NO (Benzene, nitroso-)	586-96-9	**	8.09	PE	3938
			**	8.9 (V)	PE	4467
			**	8.90±0.1 (V)	PE	4465
			**	9.84±0.1 (V)	PE	4401
C₆H₆NO⁺						
	C ₆ H ₄ (NH ₂)OCH ₃ (Benzenamine, 3-methoxy-)	536-90-3	CH ₃	11.07±0.1	EI	3446
	C ₆ H ₄ (NH ₂)OCH ₃ (Benzenamine, 4-methoxy-)	104-94-9	CH ₃	10.43±0.1	EI	3446
	C ₆ H ₅ COC ₄ H ₉ NCH ₃ (Methanone,(1-methyl-1H-pyrrol-2-yl)phenyl-)	37496-06-3	C ₆ H ₅	12.2±0.1	EI	5493
	C ₆ H ₄ (OH)NHCOCH ₃ (Acetamide, N-(2-hydroxyphenyl)-)	614-80-2	CH ₃ CO	13.46±0.02	EI	3631
	C ₆ H ₄ (OH)NHCOCH ₃ (Acetamide, N-(4-hydroxyphenyl)-)	103-90-2	CH ₃ CO	13.52±0.02	EI	3631
	C ₆ H ₄ (NO ₂)NH ₂ (Benzenamine, 3-nitro-)	99-09-2	NO	9.12±0.1	EI	3447

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₆H₆NO⁺	C ₆ H ₄ (NO ₂)NH ₂ (Benzenamine, 4-nitro-)	100-01-6	NO	9.56±0.1	EI	3447
C₆H₇NO⁺	C ₅ H ₄ N(OCH ₃) (Pyridine, 2-methoxy-)	1628-89-3	**	8.82±0.03 (V)	PE	4711
			**	8.9±0.1	EI	4302
			**	8.96±0.02	EI	3636
	C ₅ H ₄ N(OCH ₃) (Pyridine, 3-methoxy-)	7295-76-3	**	9.34±0.02	EI	3636
			**	9.34±0.05	EI	3635
	C ₅ H ₄ N(OCH ₃) (Pyridine, 4-methoxy-)	620-08-6	**	9.58±0.02	EI	3636
	C ₅ H ₄ N(=O)CH ₃ (2(1 <i>H</i>)-Pyridinone, 1-methyl-)	694-85-9	**	8.58±0.02	EI	3636
			**	8.41±0.03 (V)	PE	4711
	C ₅ H ₄ N(=O)CH ₃ (4(1 <i>H</i>)-Pyridinone, 1-methyl-)	695-19-2	**	8.48±0.02	EI	3636
	C ₅ H ₃ N(CH ₃)OH (2-Pyridinol, 6-methyl-)	73229-70-6	**	8.20±0.03 (V)	PE	4711
			**	8.33 (V)	PE	5191
	C ₄ H ₄ NCOCH ₃ (Ethanone, 1-(1 <i>H</i> -pyrrol-2-yl)-)	1072-83-9	**	8.72±0.05	EI	3482
	C ₅ H ₄ N(O)CH ₃ (Pyridine, 3-methyl-, 1-oxide)	1003-73-2	**	8.20±0.02 (V)	PE	4275
	C ₅ H ₄ N(O)CH ₃ (Pyridine, 4-methyl-, 1-oxide)	1003-67-4	**	8.12±0.02 (V)	PE	4275
			**	8.17 (V)	PE	4222
	C ₅ H ₄ N(O)CH ₃ (Pyridinium, 3-hydroxy-1-methyl-, hydroxide, inner salt)	25065-00-3	**	7.90±0.02	EI	3636
			**	7.90±0.05	EI	3635
	C ₅ H ₃ N(OH)CH ₃ (3-Pyridinol, 6-methyl-)	1121-78-4	**	9.15±0.05	EI	3635
	C ₅ H ₃ NH(CH ₃) ⁼ O (2(1 <i>H</i>)-Pyridinone, 6-methyl-)	3279-76-3	**	8.19±0.03	OTH	5596
	C ₆ H ₄ (OH)NCOCH ₃ (Acetamide, <i>N</i> -(2-hydroxyphenyl)-)	614-80-2	CH ₂ =C=O	9.41±0.02	EI	3631
	C ₆ H ₄ (OH)NCOCH ₃ (Acetamide, <i>N</i> -(4-hydroxyphenyl)-)	103-90-2	CH ₂ =C=O	9.82±0.02	EI	3631
C₆H₁₁NO⁺	(CH ₃) ₂ NCOCH=CHCH ₃	23135-18-4	**	9.0±0.1	EI	3996
C₆H₁₂NO⁺	C ₁₁ H ₂₀ N ₂ O ₄ (L-Alanine, N-(N-acetyl-L-valyl)-methyl ester)	55728-13-7		9.2±0.1	PI	5279
C₆H₁₃NO⁺	cis-C ₆ H ₁₀ (OH)NH ₂ (Cyclohexanol, 2-amino-, cis-)	931-15-7	**	9.59 (V)	PE	4450
	trans-C ₆ H ₁₀ (OH)NH ₂ (Cyclohexanol, 2-amino-, trans-)	6982-39-4	**	9.49 (V)	PE	4450
	C ₁₁ H ₂₀ N ₂ O ₄ (L-Alanine, N-(N-acetyl-L-valyl)-methyl ester)	55728-13-7		8.8±0.1	PI	5279
C₆H₁₅NO⁺	(C ₂ H ₅) ₂ NCH ₂ CH ₂ OH	100-37-8	**	8.58±0.03 (V)	PE	3987
C₇H₄NO⁺	C ₆ H ₄ (CN)OCH ₃ (Benzonitrile, 3-methoxy-)	1527-89-5	CH ₃	12.75±0.1	EI	3446

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₇H₄NO⁺						
	C ₆ H ₅ (CN)OCH ₃ (Benzonitrile, 4-methoxy-)	874-90-8	CH ₃	12.65±0.1	EI	3446
	C ₆ H ₅ (NO ₂)CN (Benzonitrile, 3-nitro-)	619-24-9	NO	10.45±0.1	EI	3447
	C ₆ H ₅ (NO ₂)CN (Benzonitrile, 4-nitro-)	619-72-7	NO	10.80±0.1	EI	3447
C₇H₅NO⁺						
	C ₆ H ₅ N=C=O (Benzene, isocyanato-)	103-71-9	**	9.00 (V)	PE	4495
			**	9.2	EI	4660
	C ₆ H ₅ CNO (Benzonitrile, N-oxide)	873-67-6	**	8.96±0.02 (V)	PE	4674
C₇H₆NO⁺						
	C ₆ H ₅ COC ₆ H ₄ NH ₂ (Methanone, (2-aminophenyl)phenyl-)	2835-77-0		11.9±0.1	EI	4358
				11.9±0.1	EI	4335
	C ₆ H ₅ COC ₆ H ₄ NH ₂ (Methanone, (3-aminophenyl)phenyl-)	2835-78-1		12.0±0.1	EI	4335
				12.0±0.1	EI	4358
	C ₆ H ₅ COC ₆ H ₄ NH ₂ (Methanone, (4-aminophenyl)phenyl-)	1137-41-3		11.25±0.1	EI	4335
				11.25±0.1	EI	4358
	CH ₃ C ₆ H ₄ NO ₂ (Benzene, 1-methyl-2-nitro-)	88-72-2	OH	9.69±0.05	PI	5437
	C ₆ H ₄ (NH ₂)COOH (Benzoic acid, 3-amino-)	99-05-8	OH	12.18±0.2	EI	3973
	C ₆ H ₄ (NH ₂)COOH (Benzoic acid, 4-amino-)	150-13-0	OH	12.12±0.2	EI	3973
	C ₆ H ₄ (Cl)NHCHO (Formamide, N-(2-chlorophenyl)-)	2596-93-2	Cl	9.3±0.1	EI	4359
C₇H₇NO⁺						
	C ₆ H ₅ (NO)(CH ₃) (Benzene, 1-methyl-4-nitroso-)	623-11-0	**	8.79±0.1 (V)	PE	4465
	C ₆ H ₅ CONH ₂ (Benzamide)	55-21-0	**	9.45 (V)	PE	4918
			**	9.60	EI	3792
C₇H₉NO⁺						
	C ₅ H ₄ N(OC ₂ H ₅) (Pyridine, 4-ethoxy-)	33399-46-1	**	9.25±0.03 (V)	PE	4711
	C ₆ H ₄ (NH ₂)OCH ₃ (Benzenamine, 3-methoxy-)	536-90-3	**	7.76±0.1	EI	3446
	C ₆ H ₄ (NH ₂)OCH ₃ (Benzenamine, 4-methoxy-)	104-94-9	**	7.44	PI	4328
			**	7.08	PE	4621
			**	7.58±0.01 (V)	PE	4389
			**	7.58 (V)	PE	5403
			**	6.92	EI	3845
			**	7.60±0.1	EI	3446
			**	9.39	EI	4089
C₇H₁₀NO⁺						
	C ₅ H ₈ NCOCH=CHCH ₃ (Pyrrolidine, 1-(1-oxo-2-butenyl)-)	51944-65-1	CH ₃	11.2±0.1	EI	3996
C₇H₁₁NO⁺						
	C ₅ H ₈ NCOCH ₃ (Pyridine, 1-acetyl-1,2,3,4-tetrahydro-)	19615-27-1	**	8.8	EI	4046

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_7\text{H}_{13}\text{NO}^+$	$\text{C}_7\text{H}_{12}\text{NOH}$ (1-Azabicyclo[2.2.2]octan-4-ol) $\text{C}_9\text{H}_{10}\text{N}(\text{O})\text{CH}_3$ (2H-Azepin-2-one, hexahydro-1-methyl-) $\text{C}_9\text{H}_{10}(=\text{NOCH}_3)$ (Cyclohexanone, O-methyloxime) $\text{C}_9\text{H}_{10}(=\text{N}(\text{O})\text{CH}_3)$ (Methanamine, N-cyclohexylidene-N-oxide) $\text{C}_9\text{H}_{10}\text{NOCH}_3$ (1-Oxa-2-azaspiro[2.5]octane, 2-methyl-) $\text{C}_9\text{H}_{10}\text{NCOCH}_3$ (Piperidine, 1-acetyl-)	26458-74-2 2556-73-2 13858-85-0 58751-78-3 3400-13-3 618-42-8	**	8.48 ± 0.015 (V) 9.00 ± 0.05 9.01 ± 0.05 7.97 ± 0.05 8.93 ± 0.05 9.1	PE EI EI EI EI EI	4286 4677 4677 4677 4677 4046
$\text{C}_7\text{H}_{15}\text{NO}^+$	$\text{C}_5\text{H}_6\text{ON}(\text{CH}_3)_3$ (Isoxazolidine, 2-(1,1-dimethylethyl)-) <i>cis</i> - $\text{C}_5\text{H}_8(\text{OH})\text{N}(\text{CH}_3)_2$ (Cyclopentanol, <i>cis</i> -2-(dimethylamino)-) <i>trans</i> - $\text{C}_5\text{H}_8(\text{OH})\text{N}(\text{CH}_3)_2$ (Cyclopentanol, <i>trans</i> -2-(dimethylamino)-)	67137-81-9 57070-96-9 18760-79-7	**	8.25 7.80 7.45	PE PE PE	5301 4399 4399
$\text{C}_7\text{H}_{17}\text{NO}^+$	$(\text{C}_2\text{H}_5)_2\text{N}(\text{CH}_2)_3\text{OH}$	622-93-5	**	8.56 ± 0.05 (V)	PE	3987
$\text{C}_8\text{H}_4\text{NO}^+$	$\text{C}_6\text{H}_4(\text{CN})\text{COOH}$ (Benzoic acid, 4-cyano-)	619-65-8	OH	12.68 ± 0.2	EI	3973
$\text{C}_8\text{H}_7\text{NO}^+$	$\text{C}_6\text{H}_4(\text{NCO})\text{CH}_3$ (Benzene, 1-isocyanato-2-methyl-) $\text{C}_6\text{H}_4(\text{NCO})\text{CH}_3$ (Benzene, 1-isocyanato-3-methyl-) $\text{C}_6\text{H}_4(\text{NCO})\text{CH}_3$ (Benzene, 1-isocyanato-4-methyl-) $\text{C}_6\text{H}_4(\text{CN})\text{OCH}_3$ (Benzonitrile, 3-methoxy-) $\text{C}_6\text{H}_4(\text{CN})\text{OCH}_3$ (Benzonitrile, 4-methoxy-) $\text{C}_6\text{H}_4\text{C}_2\text{H}_2\text{NH}(\text{=O})$ (2H-Indol-2-one, 1,3-dihydro-)	614-68-6 621-29-4 622-58-2 1527-89-5 874-90-8 59-48-3	** ** ** ** ** **	8.7 ± 0.1 (V) 8.7 ± 0.1 (V) 8.83 (V) 8.6 ± 0.1 (V) 9.11 ± 0.1 8.74 8.97 ± 0.1 8.36 (V)	PE PE PE PE EI EI EI PE	5026 5026 4495 5026 3446 3845 3446 5406
$\text{C}_8\text{H}_8\text{NO}^+$	$\text{C}_6\text{H}_5\text{NHCOCH}_3$ (Acetamide, N-phenyl-) $\text{C}_6\text{H}_5\text{ClNHCOCH}_3$ (Acetamide, N-(2-chlorophenyl)-) $\text{C}_6\text{H}_5(\text{Cl})(\text{CH}_3)\text{NHCHO}$ (Formamide, N-(2-chloro-4-methylphenyl)-) $\text{C}_6\text{H}_5(\text{Cl})(\text{CH}_3)\text{NHCHO}$ (Formamide, N-(2-chloro-5-methylphenyl)-) $\text{C}_6\text{H}_4\text{BrNHCOCH}_3$ (Acetamide, N-(2-bromophenyl)-) $\text{C}_6\text{H}_4\text{INHCOCH}_3$ (Acetamide, N-(2-iodophenyl)-)	103-84-4 533-17-5 18931-77-6 18931-82-3 614-76-6 19591-17-4	H Cl Cl Cl Br I	11.00 9.40 9.40 8.86 ± 0.03 9.1 ± 0.1 9.1 ± 0.1 9.40 9.08 ± 0.03 9.30 8.57 ± 0.03	EI EI EI EI EI EI EI EI EI	4834 4834 4834 3483 4359 4359 4834 3483 4834

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₈H₉NO⁺						
	C ₆ H ₅ (CH ₃)(CONH ₂) (Benzamide, 3-methyl-)	618-47-3	**	9.11 (V)	PE	4918
	C ₆ H ₅ (CH ₃)(CONH ₂) (Benzamide, 4-methyl-)	619-55-6	**	9.14 (V)	PE	4918
	C ₆ H ₅ NH ₂ (COCH ₃) (Ethanone, 1-(4-aminophenyl)-)	99-92-3	**	7.8±0.1	PE	4401
	C ₆ H ₅ CHNO(CH ₃) (Oxaziridine, 2-methyl-3-phenyl-)	3400-12-2	**	8.36±0.05	EI	4677
	C ₆ H ₅ CH=N(O)CH ₃	XXXXX-XX-X	**	8.01 (V)	PE	5590
	C ₆ H ₅ NHC(=O)CH ₃ (Acetamide, N-phenyl)	103-84-4	**	8.30±0.10	PE	5608
			**	8.46±0.05 (V)	PE	5013
			**	8.46 (V)	PE	5406
			**	8.60	EI	4834
			**	8.18±0.03	EI	3483
	C ₆ H ₅ CH=NOCH ₃ (Benzaldehyde, O-methyloxime)	3376-32-7	**	8.76±0.05	EI	4677
	C ₆ H ₅ CONHCH ₃ (Benzamide, N-methyl-)	613-93-4	**	9.33±0.05	EI	4677
	C ₆ H ₅ CH=N(CH ₃)O (Methanamine, N-(phenylmethylene)-N-oxide)	3376-23-6	**	7.89 (V)	PE	4719
			**	8.01±0.02 (V)	PE	4674
			**	8.01	PE	5099
			**	8.08±0.05	EI	4677
	C ₇ H ₇ N(CH ₃)O (Methanaminium,N-2,4,6-cycloheptatrien-1-ylidene-N-hydroxy-hydroxide, inner salt)	65194-06-1	**	7.28	PE	5099
	C ₆ H ₅ (OH)CHN(O)CH ₃ (Phenol,4-amino-N-oxide)	16089-67-1	**	7.76±0.02 (V)	PE	4674
C₈H₁₁NO⁺						
	C ₆ H ₅ (OH)(CH ₂ NHCH ₃) (Benzene-methanamine,2-hydroxy-N-methyl-)	XXXXX-XX-X	**	8.18 (V)	PE	5134
	C ₆ H ₅ (OH)CH ₂ CH ₂ NH ₂ (Phenol, 4-(2-aminoethyl)-)	51-67-2	**	8.41±0.12 (V)	PE	4672
C₈H₁₂NO⁺						
	C ₅ H ₁₀ NCOCH=CHCH ₃ (Piperidine, 1-(1-oxo-2-butenyl)-)	3626-69-5	**	11.1±0.1	EI	3996
C₈H₁₃NO⁺						
	C ₄ H ₉ NCOCH=CHCH ₃ (Pyrrolidine, 1-(1-oxo-2-butenyl)-)	51944-65-1	**	9.0±0.1	EI	3996
C₈H₁₅NO⁺						
	C ₇ H ₁₂ NCH ₂ OH (1-Azabicyclo[2.2.2]octane-4-methanol)	26608-58-2	**	8.17±0.015 (V)	PE	4286
	C ₈ H ₁₅ NO (8-Azabicyclo[3.2.1]octan-3-ol,8-methyl- <i>endo</i> -)	120-29-6	**	8.1±0.15	EI	5401
	C ₈ H ₁₅ NO (8-Azabicyclo[3.2.1]octan-3-ol,8-methyl- <i>exo</i> -)	135-97-7	**	7.9±0.15	EI	5401
	((CH ₂) ₄ ON)CH=C(CH ₃) ₂ (Morpholine, 4-(2-methyl-1-propenyl)-)	2403-55-6	**	8.20±0.03 (V)	PE	4452
C₈H₁₇NO⁺						
	C ₄ H ₉ ON(C(CH ₃) ₃) (2H-1,2-Oxazine,2-(1,1-dimethylethyl)tetrahydro-)	54722-72-4	**	8.27 (V)	PE	5301
	CH ₃ CH(CH ₃)CON(C ₂ H ₅) ₂	33931-44-1	**	8.80 (V)	PE	4672
	cis-C ₆ H ₁₀ (OH)N(CH ₃) ₂ (Cyclohexanol, 2-(dimethylamino)-, <i>cis</i> -)	20431-82-7	**	8.64 (V)	PE	4450
	trans-C ₆ H ₁₀ (OH)N(CH ₃) ₂ (Cyclohexanol, 2-(dimethylamino)-, <i>trans</i> -)	15910-74-4	**	8.36 (V)	PE	4450

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_8\text{H}_{17}\text{NO}^+$	$((\text{CH}_2)_4\text{ON})\text{CH}_2\text{CH}(\text{CH}_3)_2$ (Morpholine, 4-(2-methylpropyl)-)	10315-98-7	**	8.46 ± 0.03 (V)	PE	4452
$\text{C}_8\text{H}_{18}\text{NO}^+$	$(\text{tert}-\text{C}_4\text{H}_9)_2\text{NO}$	2406-25-9	**	6.77	PE	3712
$\text{C}_9\text{H}_7\text{NO}^+$	$\text{C}_9\text{H}_7\text{NO}$ (Isoquinoline, 2-oxide) $\text{C}_9\text{H}_7\text{NO}$ (Quinoline, 1-oxide)	1532-72-5 1613-37-2	**	7.98 ± 0.02 (V) 8.00 ± 0.02 (V)	PE	4551 4551
$\text{C}_9\text{H}_8\text{NO}^+$	$\text{C}_6\text{H}_5\text{NHCOCH}=\text{CHCH}_3$ (2-Butenamide, <i>N</i> -phenyl-)	1733-40-0	CH_3	12.1 ± 0.3	EI	3996
$\text{C}_9\text{H}_9\text{NO}^+$	$\text{C}_6\text{H}_5(\text{CH}_3)_2\text{CNO}$ (Benzonitrile, 2,6-dimethyl-N-oxide) $\text{C}_6\text{H}_5\text{NO}$ (Isoquinoline, 3,4-dihydro-2-oxide) $\text{C}_6\text{H}_4\text{C}_3\text{H}_5\text{NO}$ (Isoquinolinium,3,4-dihydro-2-hydroxy-hydroxide,inner salt)	19111-74-1 24423-87-8 65194-03-8	**	8.62 ± 0.02 (V) 7.81 (V) 7.81	PE	4674 4719 5099
$\text{C}_9\text{H}_{10}\text{NO}^+$	$\text{C}_6\text{H}_4\text{ClNHCOCH}_3$ (Propanamide, <i>N</i> -(2-chlorophenyl)-)	2760-32-9	Cl	9.45	EI	4834
$\text{C}_{10}\text{H}_{11}\text{NO}^+$	$\text{C}_6\text{H}_5\text{N}(\text{CH}_3)\text{C}(=\text{O})\text{CH}_3$ (Acetamide, <i>N</i> -methyl- <i>N</i> -phenyl-) $\text{C}_6\text{H}_4(\text{CH}_3)\text{NHCOCH}_3$ (Acetamide, <i>N</i> -(2-methylphenyl)-) $\text{C}_6\text{H}_4(\text{CH}_3)\text{NHCOCH}_3$ (Acetamide, <i>N</i> -(4-methylphenyl)-) $\text{C}_6\text{H}_4(\text{CHO})\text{N}(\text{CH}_3)_2$ (Benzaldehyde, 4-(dimethylamino)-)	579-10-2 120-66-1 103-89-9 100-10-7	** ** ** ** **	8.81 (V) 8.34 (V) 8.03 ± 0.02 7.75 ± 0.02 7.36 ± 0.02 7.3 ± 0.1	PE PE EI EI PI PE	5406 5406 3631 3631 4028 4401
$\text{C}_{10}\text{H}_{13}\text{NO}^+$	$\text{C}_6\text{H}_4(\text{OCH}_3)(\text{CH}_2\text{NHCH}_3)$ (Methanamine, <i>N</i> -[2-methoxyphenyl methylene]-) $\text{C}_5\text{H}_4\text{N}(\text{O})(\text{tert}-\text{C}_4\text{H}_9)$ (Pyridine, 4-(1,1-dimethylethyl)-, 1-oxide) $\text{C}_9\text{H}_{13}\text{N}=\text{O}$ (1-Azatricyclo[3.3.1.1 ^{3,7}]decan-4-one) $\text{C}_6\text{H}_4(\text{OCH}_3)\text{N}(\text{CH}_3)_2$ (Benzanamine, 4-methoxy- <i>N,N</i> -dimethyl-) $\text{C}_6\text{H}_4(\text{OCH}_3)\text{N}(\text{CH}_3)_2$ (Benzanamine, 2-methoxy- <i>N,N</i> -dimethyl-) $\text{C}_6\text{H}_4(\text{O})\text{N}(\text{CH}_3)_3$ (Benzenaminine, 2-hydroxy- <i>N,N,N</i> -trimethyl-, hydroxide, inner salt) $\text{C}_6\text{H}_4(\text{OCH}_3)\text{CH}_2\text{CH}_2\text{NH}_2$ (Benzeneethanamine, 4-methoxy-) $\text{C}_5\text{H}_8\text{NCOCH}=\text{CHCH}_3$ (Pyridine, 1,2,3,4-tetrahydro-1-(1-oxo-2-butenyl)-, (E))	1125-90-2 23569-17-7 42949-24-6 701-56-4 700-75-4 31061-58-2 55-81-2 50838-23-8	** ** ** ** ** ** ** **	8.22 (V) 8.00 (V) 8.21 ± 0.02 (V) 6.7 ± 0.1 7.18 ± 0.01 (V) 7.59 ± 0.02 ~6.8 8.16 ± 0.08 (V) 8.6	PE PE EI PE PE EI EI PE EI	5134 4222 4217 4401 4389 3630 3630 4672 4046
$\text{C}_{10}\text{H}_{15}\text{NO}^+$	$\text{C}_4\text{H}_9\text{NOC}_5\text{H}_7$ (Morpholine, 4-(1-cyclopenten-1-yl)-)	936-52-7	**	7.60 ± 0.05 (V)	PE	4654

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₉H₁₅NO⁺						
	C ₉ H ₉ NOC ₅ H ₇	936-52-7	**	7.60±0.05 (V)	PE	4819
	C ₉ H ₁₀ NCOCH=CHCH ₃ (Piperidine, 1-(1-oxo-2-butenyl)-, (E))	50838-22-7		8.9	EI	4046
	C ₉ H ₁₀ NCOCH=CHCH ₃ (Piperidine, 1-(1-oxo-2-butenyl)-)	3626-69-5	CH ₃	8.9±0.1	EI	3996
C₉H₁₇NO⁺						
	C ₉ H ₁₄ NOCH ₃ (9-Azabicyclo[3.3.1]nonane,9-methoxy-)	73321-04-7	**	7.79 (V)	PE	5091
	C ₉ H ₁₇ NO (8-Azabicyclo[3.2.1]octane,3-methoxy-8-methyl- <i>endo</i> -)	XXXXX-XX-X	**	7.8±0.15	EI	5401
	C ₉ H ₁₇ NO (8-Azabicyclo[3.2.1]octane,3-methoxy-8-methyl- <i>exo</i> -)	16487-33-5	**	7.9±0.15	EI	5401
	C ₉ H ₁₇ NO (Bicyclo[2.2.1]heptan-2-ol,3-(dimethylamino)-(2- <i>exo</i> ,3- <i>endo</i>)-)	57128-85-5	**	8.35 (V)	PE	5377
	C ₉ H ₁₇ NO (Bicyclo[2.2.1]heptan-2-ol,3-(dimethylamino)-(<i>endo,endo</i>)-)	57070-90-3	**	8.60 (V)	PE	5377
	C ₉ H ₅ N(O)(CH ₃) ₄ (4-Piperidinone, 2,2,6,6-tetramethyl-)	826-36-8	**	7.74	PE	4278
			**	8.30±0.05	EI	3494
C₉H₁₈NO⁺						
	C ₉ H ₆ N(CH ₃) ₄ O (1-Piperidinyloxy, 2,2,6,6-tetramethyl-)	2564-83-2	**	6.73	PE	3712
C₁₀H₉NO⁺						
	C ₉ H ₆ N ₃ (=O)(CH ₃) (Indeno[1,2- <i>d</i>]triazol-8(3H)-one, 3a,8a-dihydro-3-methyl-)	55507-30-7	N ₂	8.8±0.2	EI	4863
C₁₀H₁₀NO⁺						
	C ₉ H ₅ CH ₂ NHCOCH=CHCH ₃ (2-Butenamide, <i>N</i> -(phenylmethyl)-)	51944-67-3	CH ₃	10.7±0.1	EI	3996
C₁₀H₁₁NO⁺						
	C ₉ H ₂ (CH ₃) ₃ (C≡NO) (Benzonitrile, 2,4,6-trimethyl-N-oxide)	2904-57-6	**	8.34 (V)	PE	4719
	C ₉ H ₅ NHCOCH=CHCH ₃ (2-Butenamide, <i>N</i> -phenyl-)	1733-40-0	**	8.35±0.02 (V)	PE	4674
			**	8.7±0.1	EI	3996
C₁₀H₁₂NO⁺						
	C ₉ H ₄ ClNHCOCH ₂ CH ₂ CH ₃ (Butanamide, <i>N</i> -(2-chlorophenyl)-)	33694-15-4	Cl	9.45	EI	4834
C₁₀H₁₃NO⁺						
	C ₉ H ₃ (CH ₃) ₂ NHCOCH ₃ (Acetamide, <i>N</i> -(2,6-dimethylphenyl)-)	2198-53-0	**	8.70±0.05 (V)	PE	5013
	C ₉ H ₄ (CH ₃)N(CH ₃) ₂ C(=O)CH ₃ (Acetamide, <i>N</i> -methyl- <i>N</i> -(2-methylphenyl)-)	29823-47-0	**	8.82 (V)	PE	5406
	C ₉ H ₄ (N(CH ₃) ₂)COCH ₃ (Ethanone, 1-[4-(dimethylamino)phenyl]-)	2124-31-4	**	7.57±0.05 (V)	PE	5097
C₁₀H₁₅NO⁺						
	C ₁₀ H ₁₅ NO (Benzeneethanamine, 4-methoxy- <i>α</i> -methyl-(<i>±</i>)-)	23239-32-9	**	8.16±0.06 (V)	PE	4758
C₁₀H₁₇NO⁺						
	C ₄ H ₈ NOC ₆ H ₉ (Morpholine, 4-(1-cyclohexen-1-yl)-)	670-80-4	**	7.67±0.05	PE	4452
			**	7.67±0.05	PE	4654
			**	7.67±0.05 (V)	PE	4819

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₀H₁₉NO⁺	((CH ₂) ₃ ON)(C ₆ H ₁₁) (Morpholine, 4-cyclohexyl-)	6425-41-8	**	8.18±0.03 (V)	PE	4452
	C ₁₀ H ₁₉ NO (Bicyclo[2.2.1]heptan-2-amine,3-methoxy-N,N-dimethyl-(2- <i>endo</i> ,3- <i>exo</i>)-)	67425-06-3	**	8.13 (V)	PE	5377
	C ₁₀ H ₁₉ NO (Bicyclo[2.2.1]heptan-2-amine,3-methoxy-N,N-dimethyl-(<i>endo,endo</i>)-)	67398-96-3	**	8.06 (V)	PE	5377
C₁₁H₁₃NO⁺	C ₆ H ₅ CH ₂ NHCOCH=CHCH ₃ (2-Butenamide, <i>N</i> -(phenylmethyl)-)	51944-67-3	**	8.6±0.1	EI	3996
C₁₁H₁₄NO⁺	C ₂₇ H ₄₀ N ₄ O ₈ S (L-Cysteine,S-(2-methoxy-2-oxoethyl)-N-[N-[N-(1-oxopropyl)-L-phenylalanyl]-L-leucyl]glycyl]-methyl ester)	35146-63-5		9.0±0.1	PI	5279
	C ₆ H ₄ CINHCOC(CH ₃) ₃ (Propanamide, N-(2-chlorophenyl)-2,2-dimethyl-)	62662-74-2	Cl	9.45	EI	4834
C₁₁H₁₅NO⁺	C ₆ H ₃ (CH ₃) ₂ N(CH ₃)C(=O)CH ₃ (Acetamide, N-(2,6-dimethylphenyl)-N-methyl-)	18835-47-7	**	8.8 (V)	PE	5406
	C ₆ H ₂ (CH ₃) ₃ CHN(CH ₃)O (Methanamine, N-[2,4,6-trimethylphenyl)methylene]-N-oxide)	41106-03-0	**	8.08	PE	5099
	C ₆ H ₅ CH=N(<i>tert</i> -C ₄ H ₉)O (2-Propanamine, 2-methyl-N-(phenylmethylene)-N-oxide)	3376-24-7	**	8.08 (V) 7.69 (V)	PE	4719
C₁₁H₂₀NO⁺	C ₁₃ H ₂₅ NO (4-Quinolinol,4-ethyldecahydro-1,2-dimethyl-(2α,4α,4aβ,8aβ)	20422-70-2	C ₂ H ₅	9.92	EI	5452
	C ₁₃ H ₂₅ NO (4-Quinolinol,4-ethyldecahydro-1,2-dimethyl-(2α,4α,4aβ,8aα))	20422-68-8	C ₂ H ₅	9.81±0.02	EI	5598
	C ₁₃ H ₂₅ NO (4-Quinolinol,4-ethyldecahydro-1,2-dimethyl-(2α,4β,4aα,8aβ))	20422-72-4	C ₂ H ₅	9.94	EI	5452
C₁₁H₂₂NO⁺	C ₃₀ H ₄₅ N ₅ O ₆ (L-Alanine, N-[N-[N-[N-(1-oxodecyl)glycyl]-L-tryptophyl]-L-alanyl]-methyl ester)	19716-78-0		9.4±0.1	PI	5279
C₁₂H₈NO⁺	C ₁₂ H ₈ NOH (Methanone, (2-nitrophenyl)-2-pyridinyl-)	27693-37-4		7.94	EI	5459
	C ₆ H ₅ CO ₂ C ₆ H ₄ N (Methanone, phenyl-2-pyridinyl-)	91-02-1		9.97	EI	5459
	C ₆ H ₄ (CH ₃)COC ₆ H ₄ N (Methanone, (2-methylphenyl)-2-pyridinyl-)	54523-78-3		9.71	EI	5459
	C ₆ H ₄ FCOC ₆ H ₄ N (Methanone, (2-fluorophenyl)-2-pyridinyl-)	XXXXX-XX-X		10.15	EI	5459
	C ₆ H ₄ ClCOC ₆ H ₄ N (Methanone, (2-chlorophenyl)-2-pyridinyl-)	1694-57-1		9.59	EI	5459
	C ₆ H ₄ BrCOC ₆ H ₄ N (Methanone, (2-bromophenyl)-2-pyridinyl-)	XXXXX-XX-X		9.37	EI	5459
	C ₆ H ₄ IICOC ₆ H ₄ N (Methanone, (2-iodophenyl)-2-pyridinyl-)	XXXXX-XX-X		9.06	EI	5459
C₁₂H₉NO⁺	C ₁₂ H ₈ NOH (Methanone, (2-nitrophenyl)-2-pyridinyl-)	27693-37-4	**	7.29	EI	5459
	C ₆ H ₅ CO ₂ C ₆ H ₄ N (Methanone, phenyl-2-pyridinyl-)	91-02-1	**	9.06	EI	5459
	C ₆ H ₅ CO ₂ C ₆ H ₄ N (Methanone, phenyl-3-pyridinyl-)	5424-19-1	**	9.1±0.1 9.6±0.1	EI	5493
					EI	5493

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{12}\text{H}_9\text{NO}^+$	$\text{C}_6\text{H}_5\text{COC}_5\text{H}_4\text{N}$ (Methanone,phenyl-4-pyridinyl-)	14548-46-0	**	9.6 ± 0.1	EI	5493
$\text{C}_{12}\text{H}_{11}\text{NO}^+$	$\text{C}_6\text{H}_5\text{COC}_5\text{H}_3\text{NCH}_3$ (Methanone,(1-methyl-1H-pyrrol-2-yl)phenyl-)	37496-06-3	**	8.7 ± 0.1	EI	5493
$\text{C}_{12}\text{H}_{13}\text{NO}^+$	$\text{C}_6\text{H}_4(\text{CN})\text{CO}(\text{CH}_3)_3\text{CH}_3$ (Benzonitrile, 4-(1-oxypropenyl)-) $\text{C}_5\text{H}_2\text{O}(\text{CH}_2\text{CH}_3)_2\text{C}_4\text{H}_2\text{NH}$ (13-Oxa-14-azatricyclo[8.2.1.1 ^{4,7}]tetradeca-4,6,10,12-tetraene) $\text{C}_5\text{H}_8\text{NCOC}_6\text{H}_5$ (Pyridine, 1-benzoyl-1,2,3,4-tetrahydro-)	30611-20-2 73650-94-9 50838-24-9	**	9.57 (V) 7.22 8.4	PE EI	4804 5575 4046
$\text{C}_{12}\text{H}_{15}\text{NO}^+$	$\text{C}_5\text{H}_{10}\text{NCOC}_6\text{H}_5$ (Piperidine, 1-benzoyl-)	776-75-0	**	8.8	EI	4046
$\text{C}_{12}\text{H}_{16}\text{NO}^+$	$\text{C}_6\text{H}_4\text{ClNHCOCH}_2\text{C}(\text{CH}_3)_3$ (Butanamide, N-(2-chlorophenyl)-3,3-dimethyl-)	XXXXX-XX-X Cl		9.40	EI	4834
$\text{C}_{12}\text{H}_{18}\text{NO}^+$	$\text{C}_{13}\text{H}_{21}\text{NO}$ (4-Quinolinol,4-ethynyldecahydro-1,2-dimethyl-(2 α ,4 α ,4 $\alpha\beta$,8 $\alpha\beta$)) $\text{C}_{13}\text{H}_{21}\text{NO}$ (4-Quinolinol,4-ethynyldecahydro-1,2-dimethyl-(2 α ,4 α ,4 $\alpha\beta$,8 $\alpha\alpha$)) $\text{C}_{13}\text{H}_{21}\text{NO}$ (4-Quinolinol,4-ethynyldecahydro-1,2-dimethyl-(2 α ,4 β ,4 $\alpha\alpha$,8 $\alpha\beta$))	16067-80-4 16067-45-1 14788-65-9	CH ₃	9.35 9.15 9.33	EI	5598 5452 5452
$\text{C}_{12}\text{H}_{20}\text{NO}^+$	$\text{C}_{13}\text{N}_{23}\text{NO}$ (4-Quinolinol,4-ethenyldecahydro-1,2-dimethyl-(2 α ,4 α ,4 $\alpha\alpha$,8 $\alpha\beta$)) $\text{C}_{13}\text{H}_{23}\text{NO}$ (4-Quinolinol-4-ethenyldecahydro-1,2-dimethyl-(2 α ,4 α ,4 $\alpha\beta$,8 $\alpha\alpha$)) $\text{C}_{13}\text{H}_{23}\text{NO}$ (4-Quinolinol-4-ethenyldecahydro-1,2-dimethyl-(2 α ,4 β ,4 $\alpha\alpha$,8 $\alpha\beta$))	20431-93-0 20431-91-8 20431-95-2	CH ₃	9.17 8.98 9.14	EI	5598 5452 5452
$\text{C}_{12}\text{H}_{21}\text{NO}^+$	$\text{C}_4(=\text{O})(\text{CH}_3)_4(=\text{NC}_4\text{H}_6)$ (Cyclobutanone, 3-(butylimino)-2,2,4,4-tetramethyl-)	23458-49-3	**	8.63 (V)	PE	5499
$\text{C}_{12}\text{H}_{22}\text{NO}^+$	$\text{C}_{13}\text{H}_{25}\text{NO}$ (4-Quinolinol,4-ethyldecahydro-1,2-dimethyl-(2 α ,4 α ,4 $\alpha\beta$,8 $\alpha\alpha$)) $\text{C}_{13}\text{H}_{25}\text{NO}$ (4-Quinolinol,4-ethyldecahydro-1,2-dimethyl-(2 α ,4 α ,4 $\alpha\alpha$,8 $\alpha\beta$)) $\text{C}_{13}\text{H}_{25}\text{NO}$ (4-Quinolinol,4-ethyldecahydro-1,2-dimethyl-(2 α ,4 β ,4 $\alpha\alpha$,8 $\alpha\beta$)) $\text{C}_{14}\text{H}_{27}\text{NO}$ (4-Quinolinol,1,4-diethyldecahydro-2-methyl-(2 α ,4 α ,4 $\alpha\alpha$,8 $\alpha\beta$)) $\text{C}_{14}\text{H}_{27}\text{NO}$ (4-Quinolinol,1,4-diethyldecahydro-2-methyl-(2 α ,4 α ,4 $\alpha\beta$,8 $\alpha\alpha$)) $\text{C}_{14}\text{H}_{27}\text{NO}$ (4-Quinolinol,1,4-diethyldecahydro-2-methyl-(2 α ,4 β ,4 $\alpha\alpha$,8 $\alpha\beta$))	20422-68-8 20422-70-2 20422-72-4 38463-60-4 38463-62-6 38463-61-5	CH ₃ CH ₃ CH ₃ C_2H_5 C_2H_5 C_2H_5	9.00 9.04 9.15 9.85 9.80 ± 0.02 9.90	EI EI EI EI EI EI	5452 5598 5452 5452 5598 5452
$\text{C}_{12}\text{H}_{24}\text{NO}^+$	$\text{C}_{17}\text{H}_{32}\text{N}_2\text{O}_5$ (L-Serine,N-[N-(1-oxodecyl)-L-alanyl]-methyl ester) $\text{C}_{20}\text{H}_{37}\text{N}_3\text{O}_5$ (L-Alanine,N-[N-[N-(1-oxodecyl)-L-alanyl]-L-alanyl]-methyl ester)	55728-15-9 55728-11-5		9.5 ± 0.1 9.0 ± 0.1	PI PI	5279 5279

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₂H₂₄NO⁺	C ₂₃ H ₄₃ N ₃ O ₅ (L-Alanine,N-[N-[N-(1-oxodecyl)-L-alanyl]-L-leucyl]-methyl ester)	55728-12-6		9.0±0.1	PI	5279
C₁₂H₂₅NO⁺	C ₁₇ H ₄₂ N ₂ O ₅ (L-Serine,N-[N-(1-oxodecyl)-L-alanyl]-methyl ester)	55728-15-9		9.4±0.1	PI	5279
C₁₃H₉NO⁺	C ₁₃ H ₉ NO (Acridine 10-oxide)	10399-73-2	**	7.45±0.02 (V)	PE	4551
C₁₃H₁₀NO⁺	C ₆ H ₅ COC ₆ H ₄ NH ₂ (Methanone, (2-aminophenyl)phenyl-)	2835-77-0	H	10.6±0.1	EI	4358
C₁₃H₁₁NO⁺	C ₁₁ H ₈ (CN)(OCH ₃) (1,4-Methanonaphthalene-2-carbonitrile, 1,4-dihydro-5-methoxy-) C ₁₁ H ₈ (CN)(OCH ₃) (1,4-Methanonaphthalene-2-carbonitrile, 1,4-dihydro-8-methoxy-) C ₁₁ H ₉ (CN)(OCH ₃) (1,4-Methanonaphthalene-2-carbonitrile, 1,4-dihydro-6-methoxy-) C ₆ H ₅ CH=N(O)C ₆ H ₅ (Benzemethanimine, α -phenyl-N-oxide) C ₆ H ₅ COC ₆ H ₄ NH ₂ (Methanone, (2-aminophenyl)phenyl-)	71906-50-8 71906-48-4 71906-42-8 59862-61-2 2835-77-0 2835-78-1 1137-41-3 54523-78-3	** ** ** ** ** ** ** **	8.46 (V) 8.44 (V) 8.22 (V) 7.75 (V) 8.25±0.1 8.45±0.1 8.45±0.1 8.4±0.1	PE PE PE PE EI EI EI EI	4835 4835 4835 5590 4358 4335 4335 4358 4358 4335 5459
C₁₃H₁₃NO⁺	C ₆ H ₅ CH ₂ OC ₆ H ₄ NH ₂ (Benzamine,4-(phenylmethoxy)-)	6373-46-2	**	7.58	CTS	5336
C₁₃H₁₅NO⁺	C ₉ H ₆ N(=O)(n-C ₄ H ₉) (Indeno[1,2- <i>b</i>]azirin-6(1H)-one, 1-butyl-1a,6a-dihydro-) C ₉ H ₆ N ₃ (=O)(n-C ₄ H ₉) (Indeno[1,2- <i>d</i>]triazol-8(3H)-one, 3-butyl-3a,8a-dihydro-)	56359-28-5 55507-31-8	** N ₂	7.90±0.1 7.8±0.2	EI EI	4863 4863
C₁₃H₂₀NO⁺	C ₁₄ H ₂₃ NO (4-Quinolinol,1-ethyl-4-ethynyldecahydro-2-methyl-(2 α ,4 β ,4a α ,8a β)) C ₁₄ H ₂₃ NO (4-Quinolinol,1-ethyl-4-ethynyldecahydro-2-methyl-(2 α ,4 α ,4a α ,8a β)) C ₁₄ H ₂₃ NO (4-Quinolinol,1-ethyl-4-ethynyldecahydro-2-methyl-(2 α ,4 α ,4a β ,8a α))	38463-55-7 38463-54-6 38463-56-8	CH ₃ CH ₃ CH ₃	9.29 9.30 9.15	EI EI EI	5598 5598 5598
C₁₃H₂₁NO⁺	C ₅ H ₅ N(CH ₃) ₂ (OH)(C≡CH)C ₄ H ₈ (4-Quinolinol,4-ethynyldecahydro-1,2-dimethyl-(2 α ,4 α ,4a β ,8a α)) C ₁₃ H ₂₁ NO (4-Quinolinol,4-ethynyldecahydro-1,2-dimethyl-(2 α ,4 β ,4a α ,8a β)) C ₁₃ H ₂₁ NO (4-Quinolinol,4-ethynyldecahydro-1,2-dimethyl-(2 α ,4 α ,4a β ,8a β))	16067-45-1 14788-65-9 16067-80-4	** ** **	7.27±0.02 7.41±0.02 7.40±0.02	EI EI EI	5598 5598 5598

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₃H₂₂NO⁺						
	C ₁₄ H ₂₅ NO (4-Quinolinol,4-ethenyl-1-ethyldecahydro-2-methyl-(2α,4α,4aα,8aβ))	38463-57-9	CH ₃	9.23	EI	5598
	C ₁₄ H ₂₅ NO (4-Quinolinol,4-ethenyl-1-ethyldecahydro-2-methyl-(2α,4β,4aα,8aβ))	38463-58-0	CH ₃	9.20	EI	5598
	C ₁₄ H ₂₅ NO (4-Quinolinol,4-ethenyl-1-ethyldecahydro-2-methyl-(2α,4α,4aβ,8aα))	38463-59-1	CH ₃	9.06	EI	5598
C₁₃H₂₃NO⁺						
	C ₁₃ H ₂₃ NO (4-Quinolinol,4-ethenyldecahydro-1,2-dimethyl-(2α,4α,4aα,8aβ))	20431-93-0	**	7.43±0.02	EI	5598
	C ₅ H ₅ N(CH ₃) ₂ (OH)(CH=CH ₂)C ₄ H ₈ (4-Quinolinol,4-ethenyldecahydro-1,2-dimethyl-(2α,4α,4aβ,8aα))	20431-91-8	**	7.26±0.02	EI	5598
	C ₁₃ H ₂₃ NO (4-Quinolinol,4-ethenyldecahydro-1,2-dimethyl-(2α,4β,4aα,8aβ))	20431-95-2	**	7.39±0.02	EI	5598
C₁₃H₂₄NO⁺						
	C ₁₄ H ₂₇ NO (4-Quinolinol,1,4-diethyldecahydro-2-methyl-(2α,4α,4aα,8aβ))	38463-60-4	CH ₃	9.09	EI	5598
	C ₁₄ H ₂₇ NO (4-Quinolinol,1,4-diethyldecahydro-2-methyl-(2α,4α,4aβ,8aα))	38463-62-6	CH ₃	9.05	EI	5598
	C ₁₄ H ₂₇ NO (4-Quinolinol,1,4-diethyldecahydro-2-methyl-(2α,4β,4aα,8aβ))	38463-61-5	CH ₃	9.18	EI	5598
C₁₃H₂₅NO⁺						
	C ₁₃ H ₂₅ NO (4-Quinolinol,4-ethyldecahydro-1,2-dimethyl-(2α,4α,4aα,8aβ))	20422-70-2	**	7.30±0.02	EI	5598
	C ₅ H ₅ N(CH ₃) ₂ (OH)(C ₂ H ₅)C ₄ H ₈ (4-Quinolinol,4-ethyldecahydro-1,2-dimethyl-(2α,4α,4aβ,8aα))	20422-68-8	**	7.19±0.02	EI	5598
	C ₁₃ H ₂₅ NO (4-Quinolinol,4-ethyldecahydro-1,2-dimethyl-(2α,4β,4aα,8aβ))	20422-72-4	**	7.32±0.02	EI	5598
C₁₄H₁₁NO⁺						
	C ₁₃ H ₇ (=O)NHCH ₃ (Phenalen-1-one,9-methylamino-)	XXXXXX-XX-X	**	7.41±0.04 (V)	PE	5595
C₁₄H₁₃NO⁺						
	C ₆ H ₄ (OCH ₃)C(=CH ₂)C ₅ H ₄ N (Pyridine,2-[1-(3-methoxyphenyl)ethenyl]-)	XXXXXX-XX-X	**	8.27	EI	5570
	C ₆ H ₄ (OCH ₃)C(=CH ₂)C ₅ H ₄ N (Pyridine,2-[1-(4-methoxyphenyl)ethenyl]-)	XXXXXX-XX-X	**	8.15	EI	5570
	C ₆ H ₄ (OCH ₃)CH=CHC ₅ H ₄ N (Pyridine, trans-3-[2-(4-methoxyphenyl)ethenyl]-)	5847-73-4	**	7.72±0.05 (V)	PE	4377
C₁₄H₁₅NO⁺						
	C ₆ H ₅ COC ₆ H ₄ N(CH ₃) ₂ (Benzanine,N,N-dimethyl-4-(phenylmethanone)-)	XXXXXX-XX-X	**	7.50±0.05	PI	5552
C₁₄H₁₉NO⁺						
	C ₁₄ H ₁₉ NO (8-Azabicyclo[3.2.1]octane,3-phenoxy- <i>endo</i> -)	XXXXXX-XX-X	**	8.1±0.15	EI	5401
	C ₁₄ H ₁₉ NO (8-Azabicyclo[3.2.1]octane,3-phenoxy- <i>exo</i> -)	16487-31-3	**	8.2±0.15	EI	5401
C₁₄H₂₃NO⁺						
	C ₄ (=O)(CH ₃) ₄ (=NC ₆ H ₁₁) (Cyclobutanone, 3-(cyclohexylimino)-2,2,4,4-tetramethyl-)	54133-31-2	**	9.23 (V)	PE	5499
	C ₁₄ H ₂₃ NO (4-Quinolinol,1-ethyl-4-ethynyldecahydro-2-methyl-(2α,4α,4aα,8aβ))	38463-54-6	**	7.33±0.02	EI	5598
	C ₁₄ H ₂₃ NO (4-Quinolinol,1-ethyl-4-ethynyldecahydro-2-methyl-(2α,4α,4aβ,8aα))	38463-56-8	**	7.16±0.02	EI	5598
	C ₁₄ H ₂₃ NO (4-Quinolinol,1-ethyl-4-ethynyldecahydro-2-methyl-(2α,4β,4aα,8aβ))	38463-55-7	**	7.28±0.02	EI	5598

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₄H₂₅NO⁺						
	C ₁₄ H ₂₅ NO (4-Quinolinol,4-ethenyl-1-ethyldecahydro-2-methyl-(2α,4α,4aα,8aβ))	38463-57-9	**	7.32±0.02	EI	5598
	C ₁₄ H ₂₅ NO (4-Quinolinol,4-ethenyl-1-ethyldecahydro-2-methyl-(2α,4α,4aβ,8aα))	38463-59-1	**	7.15±0.02	EI	5598
	C ₁₄ H ₂₅ NO (4-Quinolinol,4-ethenyl-1-ethyldecahydro-2-methyl-(2α,4β,4aα,8aβ))	38463-58-0	**	7.30±0.02	EI	5598
C₁₄H₂₇NO⁺						
	C ₁₄ H ₂₇ NO (4-Quinolinol,1,4-diethyldecahydro-2-methyl-(2α,4α,4aα,8aβ))	38463-60-4	**	7.24±0.02	EI	5598
	C ₁₄ H ₂₇ NO (4-Quinolinol,1,4-diethyldecahydro-2-methyl-(2α,4α,4aβ,8aα))	38463-62-6	**	7.09±0.02	EI	5598
	C ₁₄ H ₂₇ NO (4-Quinolinol,1,4-diethyldecahydro-2-methyl-(2α,4β,4aα,8aβ))	38463-61-5	**	7.23±0.02	EI	5598
C₁₅H₁₁NO⁺						
	C ₉ H ₆ N(O)(C ₆ H ₅) (Isoquinolinium, 4-hydroxy-2-phenyl-hydroxide, inner salt)	56359-29-6	**	7.10±0.05	EI	4863
	C ₉ H ₆ N(=O)(C ₆ H ₅) (Indeno[1,2- <i>b</i>]azirin-6(1H)-one, 1a,6a-dihydro-1-phenyl-)	42299-62-7	**	8.13±0.05	EI	4863
	C ₉ H ₆ N ₃ (=O)(C ₆ H ₅) (Indeno[1,2- <i>d</i>]triazol-8(3H)-one, 3a,8a-dihydro-3-phenyl-)	55507-27-2	N ₂	8.1±0.1	EI	4863
C₁₅H₁₃NO⁺						
	C ₁₃ H ₇ (=O)N(CH ₃) ₂ (Phenalen-1-one,9-dimethylamino-)	XXXXXX-XX-X	**	7.36±0.04 (V)	PE	5595
C₁₅H₃₀NO⁺						
	C ₁₉ H ₃₀ N ₂ O ₄ (Glycine,N-[N-(1-oxodecyl)-L-leucyl]-methyl ester)	55728-14-8		9.2±0.1	PI	5279
C₁₅H₃₁NO⁺						
	C ₁₉ H ₃₀ N ₂ O ₄ (Glycine,N-[N-(1-oxodecyl)-L-leucyl]-methyl ester)	55728-14-8		9.1±0.1	PI	5279
C₁₆H₁₃NO⁺						
	C ₉ H ₆ N ₃ (=O)(CH ₂ C ₆ H ₅) (Indeno[1,2- <i>d</i>]triazol-8(3H)-one, 3a,8a-dihydro-3-(phenylmethyl)-)	55527-79-2	N ₂	8.1±0.1	EI	4863
CH₄N₂O⁺						
	(NH ₂) ₂ CO	57-13-6	**	9.7	PE	4221
			**	10.15 (V)	PE	4471
			**	10.28 (V)	PE	4599
			**	10.33 (V)	PE	4469
C₂H₆N₂O⁺						
	(CH ₃) ₂ NNO	62-75-9	**	8.69	PE	4647
			**	9.05 (V)	PE	4451
			**	9.09 (V)	PE	4576
	CH ₃ NHCONH ₂	598-50-5	**	9.66 (V)	PE	4599
	CH ₃ NN(O)CH ₃	54168-20-6	**	~10.07±0.03 (V)	PE	4691
C₃H₄N₂O⁺						
	CH ₃ C(=O)CHN ₂	2684-62-0	**	9.21±0.05 (V)	PE	5326
C₄H₈N₂O⁺						
	(CH ₃ NH) ₂ CO	96-31-1	**	9.23 (V)	PE	4599
	(CH ₃) ₂ NCONH ₂	598-94-7	**	8.96 (V)	PE	4599
C₄H₄N₂O⁺						
	C ₄ H ₄ N ₂ O (Pyrazine, 1-oxide)	2423-65-6	**	9.17±0.02 (V)	PE	4470

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_4\text{H}_4\text{N}_2\text{O}^+$	$\text{C}_4\text{H}_4\text{N}_2\text{O}$ (Pyridazine, 1-oxide) $\text{C}_4\text{H}_4\text{N}_2\text{O}$ (Pyrimidine, 1-oxide) $\text{C}_4\text{H}_4\text{N}_2(=\text{O})$ (2(1H)-Pyrimidinone)	1457-42-7 17043-94-6 557-01-7	**	8.89 ± 0.02 8.80 ± 0.02 10.06 ± 0.05	PE PE EI	4470 4470 5159
$\text{C}_4\text{H}_6\text{N}_2\text{O}^+$	$\text{CH}_3\text{C}(=\text{O})\text{C}(\text{CH}_3)\text{N}_2$ $\text{C}(\text{CH}_3)_2(\text{CN})\text{NO}$	14088-58-5 44513-62-4	**	8.76 ± 0.05 (V) 9.77 ± 0.1 (V)	PE PE	5326 4465
$\text{C}_4\text{H}_{10}\text{N}_2\text{O}^+$	$(\text{CH}_3)_2\text{NCONHCH}_3$ $(\text{CH}_3\text{CH}_2)_2\text{NNO}$ (Ethanamine, N-ethyl-N-nitroso-)	632-14-4 55-18-5	**	8.80 (V) 8.76 (V)	PE PE	4599 4576
$\text{C}_5\text{H}_6\text{N}_2\text{O}^+$	$\text{C}_5\text{H}_6\text{N}_2\text{O}$ (1H-Imidazole, 1-acetyl-) $\text{C}_5\text{H}_4\text{N}(\text{O})\text{NH}_2$ (2-Pyridinamine 1-oxide) $\text{C}_5\text{H}_4\text{N}(\text{O})\text{NH}_2$ (3-Pyridinamine 1-oxide) $\text{C}_5\text{H}_4\text{N}(\text{O})\text{NH}_2$ (4-Pyridinamine 1-oxide) $\text{C}_4\text{H}_3\text{N}_2\text{OCH}_3$ (Pyrimidine, 2-methoxy-) $\text{C}_4\text{H}_3\text{N}_2(=\text{O})\text{CH}_3$ (2(1H)-Pyrimidinone, 1-methyl-)	2466-76-4 14150-95-9 1657-32-5 3535-75-9 931-63-5 3739-81-9	**	9.38 (V) 8.04 \pm 0.05 8.21 \pm 0.05 7.67 \pm 0.05 9.66 \pm 0.05 9.31 \pm 0.05	PE EI EI EI EI EI	5092 4117 4117 4117 5159 5159
$\text{C}_5\text{H}_8\text{N}_2\text{O}^+$	$\text{C}_5\text{H}_8\text{N}_2\text{O}$ (2,3-Diazabicyclo[2.2.1]hept-2-ene, 2-oxide)	22509-00-8	**	9.48 ± 0.03 (V)	PE	4691
$\text{C}_5\text{H}_{10}\text{N}_2\text{O}^+$	$(\text{CH}_3)_2\text{NN}=\text{CHCOCH}_3$ $(\text{CH}_3)_2\text{NN}=\text{CHCH}_2\text{CHO}$ $\text{C}_4\text{H}_8\text{NCONH}_2$ (1-Pyrrolidinecarboxamide)	XXXXX-XX-X XXXXX-XX-X 4736-71-4	** ** **	8.06 (V) 8.08 (V) 8.92 (V)	PE PE PE	5548 5548 4803
$\text{C}_5\text{H}_{12}\text{N}_2\text{O}^+$	$((\text{CH}_3)_2\text{N})_2\text{CO}$	632-22-4	** **	8.64 (V) 8.67 (V)	PE PE	4599 4469
$\text{C}_6\text{H}_4\text{N}_2\text{O}^+$	$\text{C}_6\text{H}_4\text{N}_2\text{O}$ (Benzofuran)	273-09-6	**	9.37	PE	4017
	$\text{C}_6\text{H}_4\text{N}_2\text{O}$ (1,2,3-Benzoxadiazole)	273-59-6	**	9.45 (V)	PE	5131
	$\text{C}_6\text{H}_4(\text{O})\text{NN}$ (2,4-Cyclohexadien-1-one, 6-diazo-)	4024-72-0	N ₂	9.5 \pm 0.01	EI	4317
	$\text{C}_6\text{H}_4(\text{O})\text{NN}$ (2,5-Cyclohexadien-1-one, 4-diazo-)	932-97-8	**	8.28 \pm 0.05	EI	4317
	$\text{C}_5\text{H}_4\text{N}(\text{O})\text{CN}$ (2-Pyridinecarbonitrile, 1-oxide)	2402-98-4	**	8.96 \pm 0.02 (V)	PE	4275
	$\text{C}_5\text{H}_4\text{N}(\text{O})\text{CN}$ (3-Pyridinecarbonitrile, 1-oxide)	14906-64-0	**	8.93 \pm 0.02 (V)	PE	4275
	$\text{C}_5\text{H}_4\text{N}(\text{O})\text{CN}$ (4-Pyridinecarbonitrile, 1-oxide)	14906-59-3	**	8.95 \pm 0.02 (V)	PE	4275
$\text{C}_6\text{H}_6\text{N}_2\text{O}^+$	$\text{C}_5\text{H}_4\text{NCONH}_2$ (3-Pyridinecarboxamide)	98-92-0	**	9.18	PE	5093

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₆H₈N₂O⁺						
	C ₅ H ₄ N(O)NHCH ₃ (2-Pyridinamine, <i>N</i> -methyl-, 1-oxide)	54818-70-1	**	7.67±0.05	EI	4117
	C ₅ H ₄ N(O)NHCH ₃ (3-Pyridinamine, <i>N</i> -methyl-, 1-oxide)	54818-71-2	**	7.97±0.05	EI	4117
	C ₅ H ₄ N(O)NHCH ₃ (4-Pyridinamine, <i>N</i> -methyl-, 1-oxide)	1122-92-5	**	7.45±0.05	EI	4117
	C ₅ H ₄ N(=NH)OCH ₃ (2(1H)-Pyridinimine, 1-methoxy-)	54818-76-7	**	7.46±0.05	EI	4117
C₆H₁₀N₂O⁺						
	C ₆ H ₁₀ N ₂ O (2,3-Diazabicyclo[2.2.2]oct-2-ene 2-oxide)	25926-96-9	**	9.30±0.03 (V)	PE	4691
C₆H₁₄N₂O⁺						
	C ₆ H ₁₄ N ₂ O ((CH ₃) ₂ CH) ₂ NNO (2-Propanamine, N-(1-methylethyl)-N-nitroso-)	35216-94-5 601-77-4	** **	~9.60±0.03 (V) 8.58 (V)	PE PE	4691 4576
C₇H₇N₂O⁺						
	C ₆ H ₅ NHCONH ₂ (Urea, phenyl-)	64-10-8		9.50	EI	4834
	C ₆ H ₄ CINHCONH ₂ (Urea, (2-chlorophenyl)-)	114-38-5	Cl	9.35	EI	4834
	C ₆ H ₄ BrNHCONH ₂ (Urea, (2-bromophenyl)-)	13114-90-4	Br	9.35	EI	4834
	C ₆ H ₄ INHCONH ₂ (Urea, (2-iodophenyl)-)	13114-93-7	I	9.15	EI	4834
C₇H₈N₂O⁺						
	C ₆ H ₅ (CH ₃)NNO (Benzenamine, <i>N</i> -methyl- <i>N</i> -nitroso-)	614-00-6	**	9.01 (V)	PE	4576
	C ₆ H ₅ NHCONH ₂ (Urea, phenyl-)	64-10-8	**	8.55	EI	4834
C₇H₁₀N₂O⁺						
	C ₅ H ₄ N(O)N(CH ₃) ₂ (2-Pyridinamine, <i>N,N</i> -dimethyl-, 1-oxide)	3618-79-9	**	7.62±0.05	EI	4117
	C ₅ H ₄ N(O)N(CH ₃) ₂ (3-Pyridinamine, <i>N,N</i> -dimethyl-, 1-oxide)	36100-40-0	**	7.85±0.05	EI	4117
	C ₅ H ₄ N(O)N(CH ₃) ₂ (4-Pyridinamine, <i>N,N</i> -dimethyl-, 1-oxide)	1005-31-8	**	7.21±0.05 (V)	PE	4275
			**	7.32±0.05	EI	4117
C₇H₁₂N₂O⁺						
	C ₇ H ₁₂ NNO (2-Azabicyclo[2.2.2]octane, 2-nitroso)	21744-12-7	**	8.72 (V)	PE	4576
	C ₇ H ₁₂ N ₂ O (6,7-Diazabicyclo[3.2.2]non-6-ene 6-oxide)	26081-83-4	**	9.21±0.03 (V)	PE	4691
	C ₃ N ₂ =O(CH ₃) ₄ (4H-Pyrazole-4-one,3,5-dihydro-3,3,5,5-tetramethyl-)	30467-62-0	**	8.61 (V)	PE	4429
C₇H₁₄N₂O⁺						
	C ₆ H ₁₄ N ₂ O (1H-Pyridazino[1,2- <i>c</i>][1,3,4]oxadiazine, hexahydro-)	73569-74-1	**	8.04 (V)	PE	5215
C₈H₆N₂O⁺						
	C ₆ H ₄ (CN)(CONH ₂) (Benzamide, 4-cyano-)	3034-34-2	**	9.99 (V)	PE	4918
	C ₆ H ₅ C(=O)CHN ₂ (Ethanone,2-diazo-1-phenyl-)	3282-32-4	**	8.93±0.05 (V)	PE	5326
	C ₆ H ₆ N ₂ O (Quinoxaline, 1-oxide)	6935-29-1	**	8.62±0.02 (V)	PE	4551

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₈H₉N₂O⁺	C ₆ H ₄ ClNHCONHCH ₃ (Urea, N-(2-chlorophenyl)-N'-methyl-)	15500-96-6	Cl	9.35	EI	4834
C₈H₁₀N₂O⁺	C ₆ H ₄ NO(N(CH ₃) ₂) (Benzenamine, N,N-dimethyl-4-nitroso-)	138-89-6	**	7.2±0.1	PE	4401
			**	7.78±0.1 (V)	PE	4465
	C ₆ H ₄ (NH ₂)NHCOCH ₃ (Acetamide, N-(2-aminophenyl)-)	34801-09-7	**	7.39±0.02	EI	3631
	C ₆ H ₄ (NH ₂)NHCOCH ₃ (Acetamide, N-(4-aminophenyl)-)	122-80-5	**	7.12±0.02	EI	3631
	C ₆ H ₅ NHCONHCH ₃ (Urea, N-methyl-N'-phenyl-)	1007-36-9	**	8.50±0.05	EI	4834
C₈H₁₂N₂O⁺	C ₈ H ₁₂ N ₂ (=O) (1,3-Diazatricyclo[3.3.1.1 ^{3,7}]decan-6-one)	20397-57-3	**	8.25 (V)	PE	4659
C₈H₁₄N₂O⁺	(CH ₃)CH=NN(C ₂ H ₅)CH=CHCOCH ₃ (CH ₃) ₂ C=NN(CH ₃)CH=CHCOCH ₃ C ₈ H ₁₄ N ₂ O (7,8-Diazabicyclo[4.2.2]dec-7-ene 7-oxide)	XXXXX-XX-X	**	7.79 (V)	PE	5548
		63262-98-6	**	7.78 (V)	PE	5548
		25926-97-0	**	9.13±0.03 (V)	PE	4691
C₈H₁₆N₂O⁺	C ₄ H ₄ N ₂ (O)(CH ₃) ₄ (Pyridazine, 3,4,5,6-tetrahydro-3,3,6,6-tetramethyl-1-oxide)	54143-34-9	**	~9.13±0.03 (V)	PE	4691
C₉H₈N₂O⁺	C ₆ H ₄ (CN)CHN(O)CH ₃ (Benzonitrile, 4-[(methylimino)methyl]-N ¹ -oxide)	16089-70-6	**	8.35±0.02 (V)	PE	4674
	CH ₃ C ₆ H ₄ C(=O)CHN ₂ (Ethanone,2-diazo-1-(4-methylphenyl)-)	17263-64-8	**	8.80±0.05 (V)	PE	5326
	C ₆ H ₅ C(=O)C(CH ₃) ₂ N ₂ (1-Propanone,2-diazo-1-phenyl-)	14088-57-4	**	8.52±0.05 (V)	PE	5326
C₉H₁₁N₂O⁺	C ₆ H ₃ (Cl)(N(CH ₃) ₂)NHCHO (Formamide, N-[2-chloro-5-(dimethylamino)phenyl]-)	53666-46-9	Cl	9.7±0.1	EI	4359
	C ₆ H ₄ ClNHCONHC ₂ H ₅ (Urea, N-(2-chlorophenyl)-N'-ethyl-)	62635-53-4	Cl	9.30	EI	4834
C₉H₁₂N₂O⁺	C ₆ H ₅ NHCONHC ₂ H ₅ (Urea, N-ethyl-N'-phenyl-)	621-04-5	**	8.25±0.05	EI	4834
C₉H₁₄N₂O⁺	C ₆ H ₁₄ N ₂ O (1-Pyrrolidinecarboxamide, N-1,3-butadienyl-(E)-)	61759-62-4	**	7.90 (V)	PE	4803
C₁₀H₁₃N₂O⁺	C ₆ H ₄ (OCH ₃)N=CHN(CH ₃) ₂ (Methanimidamide, N'-(3-methoxyphenyl)-N,N-dimethyl-)	1202-42-2	H	9.2±0.1	EI	4359
	C ₆ H ₄ (OCH ₃)N=CHN(CH ₃) ₂ (Methanimidamide, N'-(4-methoxyphenyl)-N,N-dimethyl-)	1202-62-6	H	9.3±0.1	EI	4359
	C ₆ H ₃ (Cl)(OCH ₃)N=CHN(CH ₃) ₂ (Methanimidamide, N'-(2-chloro-4-methoxyphenyl)-N,N-dimethyl-)	53666-34-5	Cl	8.9±0.1	EI	4359
	C ₆ H ₃ (Cl)(OCH ₃)N=CHN(CH ₃) ₂ (Methanimidamide, N'-(2-chloro-5-methoxyphenyl)-N,N-dimethyl-)	53666-40-3	Cl	8.7±0.1	EI	4359
	C ₆ H ₄ ClNHCONHCH(CH ₃) ₂ (Urea, N-(2-chlorophenyl)-N'-(1-methylethyl)-)	62635-47-6	Cl	9.20	EI	4834

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₅H₇N₃O⁺						
	C ₄ H ₂ N ₂ (CH ₃) (=O)NH ₂ (2(1H)-Pyrimidinone,4-amino-1-methyl-)	1122-47-0	**	8.65 (V)	PE	5594
			**	9.5±0.1	EI	5555
	C ₄ HN ₂ H(CH ₃) (=O)NH ₂ (2(1H)-Pyrimidinone,4-amino-5-methyl-)	554-01-8	**	8.78 (V)	PE	5594
	C ₄ HN ₂ H(CH ₃) (=O)NH ₂ (2(1H)-Pyrimidinone,4-amino-6-methyl-)	6220-50-4	**	8.73 (V)	PE	5594
	C ₄ HN ₂ H(CH ₃) (=O)NH (2(1H)-Pyrimidinone,4-imino-3-methyl-)	XXXXXX-XX-X	**	8.72 (V)	PE	5594
C₅H₉N₃O⁺						
	C ₂ N ₃ (=O)(CH ₃) ₃ (3H-1,2,4-Triazol-3-one,2,4-dihydro-2,4,5-trimethyl-)	57626-52-5	**	8.39 (V)	PE	4439
C₆H₉N₃O⁺						
	C ₄ H ₂ N ₂ (=O)(NH(CH ₃)) (CH ₃) (2(1H)-Pyrimidinone,4-amino-1,N-dimethyl-)	XXXXXX-XX-X	**	9.25±0.1	EI	5555
	C ₄ HN ₂ (CH ₃) ₂ (=O)NH ₂ (2(1H)-Pyrimidinone,4-amino-1,5-dimethyl-)	17634-60-5	**	8.50 (V)	PE	5594
	C ₄ HN ₂ (CH ₃) ₂ (=O)NH ₂ (2(1H)-Pyrimidinone,4-amino-1,6-dimethyl-)	66943-92-8	**	8.41 (V)	PE	5594
	C ₄ H ₂ N ₂ (CH ₃) (=O)NHCH ₃ (2(1H)-Pyrimidinone,1-methyl-4-(methylamino)-)	6220-49-1	**	8.58 (V)	PE	5594
C₇H₁₁N₃O⁺						
	C ₄ H ₂ N ₂ (=O)(N(CH ₃) ₂)(CH ₃) (2(1H)-Pyrimidinone,4-(dimethylamino)-1-methyl-)	2228-27-5	**	8.7±0.1	EI	5555
C₁₃H₇N₃O⁺						
	C ₁₂ H ₇ N ₂ OCN (2-Phenazinecarbonitrile-10-oxide)	59019-84-0	**	8.44 (V)	PE	4590
C₂₀H₂₅N₃O⁺						
	C ₂₀ H ₂₅ N ₃ O		**	7.25±0.10 (V)	PE	4672
C₅H₄N₄O⁺						
	C ₅ H ₄ N ₄ (=O) (6H-Purin-6-one,1,7-dihydro-)	68-94-0	**	8.55±0.03 (V)	PE	4445
C₅H₅N₅O⁺						
	C ₅ H ₃ N ₄ (=O)(NH ₂) (6H-Purin-6-one,2-amino-1,7-dihydro-)	73-40-5	**	7.85	PE	5093
			**	8.70	PE	5093
			**	8.24±0.03 (V)	PE	4445
			**	8.0±0.2	EI	5555
CH₃NO₂⁺						
	CH ₃ NO ₂	75-52-5	**	11.040±0.017	PI	3524
			**	11.07±0.01	PE	3721
			**	11.29 (V)	PE	5272
			**	11.31±0.015 (V)	PE	4107
			**	11.31 (V)	PE	4884
			**	11.8 (V)	PE	4467
	CH ₃ ONO	624-91-9	**	10.475±0.007	PI	3524
			**	11.0	PE	4379
CD₃NO₂⁺						
	CD ₃ NO ₂	13031-32-8	**	11.08±0.01	PE	3721
C₂H₅NO₂⁺						
	CH ₂ (NH ₂)COOH	56-40-6	**	8.8	PE	4221

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_2\text{H}_5\text{NO}_2^+$	$\text{CH}_2(\text{NH}_2)\text{COOH}$	56-40-6	**	9.21 ± 0.05	EI	3571
$\text{C}_3\text{H}_5\text{NO}_2^+$	$\text{CH}_3\text{COCONH}_2$	631-66-3	**	9.71 (V)	PE	4520
	$\text{C}_3\text{H}_5\text{NO}(\text{=O})$ (2-Oxazolidinone)	497-25-6	**	10.21 (V)	PE	4742
$\text{C}_3\text{H}_7\text{NO}_2^+$	$\text{NH}_2\text{COOC}_2\text{H}_5$	51-79-6	**	10.62 (V)	PE	4803
	$\text{CH}_3\text{CH}(\text{NH}_2)\text{COOH}$	56-41-7	**	8.8	PE	4221
			**	8.88	PE	4641
$\text{C}_4\text{H}_5\text{NO}_2^+$	$\text{C}_4\text{H}_5\text{N}(\text{=O})_2$ (2,5-Pyrrolidinedione)	123-56-8	**	10.01 (V)	PE	4742
			**	10.01 (V)	PE	4810
$\text{C}_4\text{H}_7\text{NO}_2^+$	$\text{CH}_2=\text{CHCH}_2\text{CH}_2\text{ONO}$	67428-02-8	**	10.02 ± 0.02 (V)	PE	4722
			**	10.02 (V)	PE	4898
	$\text{C}_3\text{H}_4\text{NO}(\text{=O})(\text{CH}_3)$ (2-Oxazolidinone, 4-methyl-)	16112-59-7	**	9.95 (V)	PE	4742
	$\text{C}_3\text{H}_4\text{NO}(\text{=O})\text{CH}_3$ (2-Oxazolidinone, 5-methyl-)	1072-70-4	**	9.99 (V)	PE	4742
$\text{C}_4\text{H}_9\text{NO}_2^+$	$\text{C}_2\text{H}_5\text{CH}(\text{NH}_2)\text{COOH}$	80-60-4	**	8.70	PE	4641
	$\text{CH}_2(\text{NH}_2)\text{COOC}_2\text{H}_5$	459-73-4	**	8.8	PE	4221
$\text{C}_4\text{H}_{11}\text{NO}_2^+$	$(\text{CH}_3)_2\text{N}(\text{O})(\text{C}_2\text{H}_4\text{OH})$	10489-99-3	**	8.86 (V)	PE	4537
$\text{C}_5\text{H}_5\text{NO}_2^+$	$\text{CH}_2=\text{C}(\text{CN})\text{CO}_2\text{CH}_3$	137-05-3	**	10.98 ± 0.05 (V)	PE	4859
	$\text{CH}_3\text{CO}_2\text{C}(\text{CN})=\text{CH}_2$	3061-65-2	**	10.76 ± 0.05 (V)	PE	4859
	$\text{C}_5\text{H}_4\text{N}(\text{O})\text{OH}$ (Pyridinium, 1,2-dihydroxy-, 1-hydroxide, inner salt)	XXXXX-XX-X	**	8.90 ± 0.05	EI	4178
	$\text{C}_5\text{H}_4\text{N}(\text{O})\text{OH}$ (Pyridinium, 1,3-dihydroxy-, 1-hydroxide, inner salt)	XXXXX-XX-X	**	8.60 ± 0.05	EI	4178
	$\text{C}_5\text{H}_4\text{N}(\text{O})\text{OH}$ (Pyridinium, 1,4-dihydroxy-, 1-hydroxide, inner salt)	XXXXX-XX-X	**	8.18 ± 0.05	EI	4178
$\text{C}_5\text{H}_7\text{NO}_2^+$	$\text{C}_5\text{H}_6\text{NH}(\text{=O})_2$ (2,6-Piperidinedione)	1121-89-7	**	9.87 (V)	PE	5614
	$\text{C}_4\text{H}_4\text{N}(\text{=O})_2(\text{CH}_3)$ (2,5-Pyrrolidinedione, 1-methyl-)	1121-07-9	**	10.71 (V)	PE	5090
$\text{C}_5\text{H}_9\text{NO}_2^+$	$\text{CH}_3\text{COC}(\text{CH}_3)_2\text{NO}$	6931-05-1	**	8.48 ± 0.1 (V)	PE	4465
	$\text{C}_3\text{H}_3\text{NO}(\text{=O})(\text{CH}_3)_2$ (2-Oxazolidinone, 4,4-dimethyl-)	26654-39-7	**	9.80 (V)	PE	4742
	$\text{C}_3\text{H}_3\text{NO}(\text{=O})(\text{CH}_3)_2$ (2-Oxazolidinone, 4,5-dimethyl-)	58628-98-1	**	9.84 (V)	PE	4742
	$\text{C}_3\text{H}_3\text{NO}(\text{=O})(\text{CH}_3)_2$ (2-Oxazolidinone, 5,5-dimethyl-)	1121-83-1	**	9.88 (V)	PE	4742
$\text{C}_5\text{H}_{11}\text{NO}_2^+$	$(\text{CH}_3)_2\text{NCH}_2\text{COOCH}_3$	7148-06-3	**	7.96 ± 0.05	PE	4192
	$n\text{-C}_3\text{H}_7\text{CH}(\text{NH}_2)\text{COOH}$	6600-40-4	**	8.53	PE	4641
	$iso\text{-C}_3\text{H}_7\text{CH}(\text{NH}_2)\text{COOH}$	72-18-4	**	8.71	PE	4641

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_5\text{H}_{13}\text{NO}_2^+$	$(\text{CH}_3)_2\text{N}(\text{O})(\text{C}_2\text{H}_4\text{OCH}_3)$	55695-37-9	**	8.37 (V)	PE	4537
$\text{C}_6\text{H}_4\text{NO}_2^+$	$\text{C}_6\text{H}_4(\text{NO}_2)_2$ (Benzene, 1,3-dinitro-)	99-65-0	NO_2	12.34 ± 0.1	EI	3447
	$\text{C}_6\text{H}_4(\text{NO}_2)_2$ (Benzene, 1,4-dinitro-)	100-25-4	NO_2	12.50 ± 0.1	EI	3447
$\text{C}_6\text{H}_5\text{NO}_2^+$	$\text{C}_6\text{H}_5\text{NO}_2$ (Benzene, nitro-)	98-95-3	**	9.85 ± 0.03	PI	5505
			**	9.87 ± 0.05	PI	5437
			**	9.88 ± 0.015 (V)	PE	4107
			**	9.92 (V)	PE	4892
			**	9.93	PE	4621
			**	9.93 (V)	PE	4884
			**	9.93 (V)	PE	5272
			**	9.94 ± 0.025	PE	3626
			**	9.99 ± 0.01	PE	3721
			**	9.99	PE	3856
			**	10.8 (V)	PE	4467
			**	9.6	EI	3916
			**	9.65 ± 0.1	EI	3447
			**	9.90	EI	3485
	$\text{C}_5\text{H}_4\text{NCOOH}$ (3-Pyridinecarboxylic acid)	59-67-6	**	9.38	PE	5093
$\text{C}_6\text{H}_7\text{NO}_2^+$	$\text{C}_5\text{H}_4\text{N}(\text{O})\text{OCH}_3$ (Pyridine, 4-methoxy-, 1-oxide)	1122-96-9	**	7.74 ± 0.05 (V)	PE	4275
			**	7.89 (V)	PE	4222
	$\text{C}_5\text{H}_4\text{N}(\text{O})\text{OCH}_3$ (Pyridinium, 1-hydroxy-2-methoxy-, hydroxide, inner salt)	XXXXX-XX-X	**	8.21 ± 0.05	EI	4178
	$\text{C}_5\text{H}_4\text{N}(\text{O})\text{OCH}_3$ (Pyridinium, 1-hydroxy-3-methoxy-, hydroxide, inner salt)	XXXXX-XX-X	**	8.40 ± 0.05	EI	4178
	$\text{C}_5\text{H}_4\text{N}(\text{O})\text{OCH}_3$ (Pyridinium, 1-hydroxy-4-methoxy-, hydroxide, inner salt)	XXXXX-XX-X	**	7.98 ± 0.05	EI	4178
	$\text{C}_5\text{H}_4\text{N}(\text{O})\text{OCH}_3$ (Pyridinium, 3-hydroxy-1-methoxy-, hydroxide, inner salt)	XXXXX-XX-X	**	8.3	EI	4178
	$\text{C}_5\text{H}_4\text{N}(=\text{O})\text{OCH}_3$ (2(1 <i>H</i>)-Pyridinone, 1-methoxy-)	40775-55-1	**	8.32 ± 0.05	EI	4178
	$\text{C}_5\text{H}_4\text{N}(=\text{O})\text{OCH}_3$ (4(1 <i>H</i>)-Pyridinone, 1-methoxy-)	XXXXX-XX-X	**	8.49 ± 0.05	EI	4178
	$\text{C}_4\text{H}_4\text{NCOOCH}_3$ (1 <i>H</i> -Pyrrole-2-carboxylic acid, methyl ester)	1193-62-0	**	8.65 ± 0.05	EI	3482
$\text{C}_6\text{H}_{11}\text{NO}_2^+$	$\text{CH}_3\text{COCH}_2\text{C}(\text{CH}_3)_2\text{NO}$	60027-50-1	**	7.96 ± 0.1 (V)	PE	4465
$\text{C}_6\text{H}_{13}\text{NO}_2^+$	$n\text{-C}_4\text{H}_9\text{CH}(\text{NH}_2)\text{COOH}$	327-57-1	**	8.52	PE	4641
	$sec\text{-C}_4\text{H}_9\text{CH}(\text{NH}_2)\text{COOH}$	73-32-5	**	8.66	PE	4641
	$iso\text{-C}_4\text{H}_9\text{CH}(\text{NH}_2)\text{COOH}$	61-90-5	**	8.51	PE	4641
$\text{C}_7\text{H}_4\text{NO}_2^+$	$\text{C}_6\text{H}_5\text{COC}_6\text{H}_4\text{NO}_2$ (Methanone, (2-nitrophenyl)phenyl-)	2243-79-0		10.3 ± 0.1	EI	4358
$\text{C}_7\text{H}_6\text{NO}_2^+$	$\text{C}_6\text{H}_4(\text{NO}_2)_2\text{C}_4\text{H}_9$ (Benzene, 1-butyl-3-nitro-)	20651-76-7		13.08 ± 0.1	EI	3629

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₇H₆NO₂⁺	C ₆ H ₅ (NO ₂)C ₂ H ₅ (Benzene, 1-butyl-4-nitro-)	20651-75-6		12.54±0.1	EI	3629
C₇H₇NO₂⁺	C ₆ H ₅ (NO)(OCH ₃) (Benzene, 1-methoxy-4-nitroso-)	1516-21-8	**	8.46±0.1 (V)	PE	4465
			**	9.50 (V)	PE	4892
	CH ₃ C ₆ H ₄ NO ₂ (Benzene, 1-methyl-2-nitro-)	88-72-2	**	9.43±0.05	PI	5437
			**	9.63 (V)	PE	5272
			**	9.69±0.015 (V)	PE	4107
			**	9.50	PE	4892
	C ₆ H ₅ (CH ₃)NO ₂ (Benzene, 1-methyl-3-nitro-)	99-08-1	**	9.48 (V)	PE	5272
			**	9.49±0.015 (V)	PE	4107
			**	9.48±0.1	EI	3447
			**	9.50	PE	4892
	C ₆ H ₅ (CH ₃)NO ₂ (Benzene, 1-methyl-4-nitro-)	99-99-0	**	9.52 (V)	PE	5272
			**	9.54±0.015 (V)	PE	4107
			**	9.50±0.1	EI	3447
			**	9.56	EI	4089
			**	9.1±0.1	PE	4401
	C ₆ H ₅ (NH ₂)COOH (Benzoic acid, 3-amino-)	99-05-8	**	8.41±0.2	EI	3973
	C ₆ H ₅ (NH ₂)COOH (Benzoic acid, 4-amino-)	150-13-0	**	8.36±0.2	EI	3973
	C ₆ H ₅ OOCNH ₂ (Carbamic acid phenyl ester)	622-46-8	**	9.14 (V)	PE	4803
	C ₅ H ₄ NCOOCH ₃ (3-Pyridinecarboxylic acid, methyl ester)	93-60-7	**	9.25	PE	5093
			**	9.85±0.1	EI	4302
	C ₆ H ₅ (NO ₂)C ₂ H ₅ (Benzene, 1-butyl-3-nitro-)	20651-76-7	CH ₂ =CHCH ₃	11.52±0.1	EI	3629
	C ₆ H ₅ (NO ₂)C ₂ H ₅ (Benzene, 1-butyl-4-nitro-)	20651-75-6	CH ₂ =CHCH ₃	11.44±0.1	EI	3629
C₇H₉NO₂⁺	C ₄ H ₃ OCON(CH ₃) ₂ (2-Furancarboxamide, N,N-dimethyl-)	13156-75-7	**	8.86±0.05 (V)	PE	4626
C₇H₁₀NO₂⁺	C ₇ H ₁₀ NO(=O) (8-Azabicyclo[3.2.1]oct-8-yloxy, 3-oxo-)	38390-62-4	**	7.4±0.1	OTH	5379
	C ₄ H ₈ NO(COCH=CHCH ₃) (Morpholine, 4-(1-oxo-2-butenyl)-)	51944-66-2	**	11.1±0.1	EI	3996
C₇H₁₁NO₂⁺	C ₇ H ₁₁ NO ₂ C ₃ HN(=O) ₂ (C ₂ H ₅) ₂ (2,4-Azetidinedione, 3,3-diethyl-)	61759-61-3 42282-85-9	** **	8.21 (V) 9.57	PE EI	4803 4660
C₇H₁₂NO₂⁺	C ₁₁ H ₂₀ N ₂ O ₄ (L-Alanine, N-(N-acetyl-L-valyl)-methyl ester)	55728-13-7		9.3±0.1	PI	5279
C₈H₅NO₂⁺	C ₆ H ₅ (CN)COOH (Benzoic acid, 4-cyano-)	619-65-8	**	10.27±0.2	EI	3973
	C ₈ H ₅ N(=O) ₂ (1H-Indole-2,3-dione)	91-56-5	**	8.98±0.05 (V)	PE	4708

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_8\text{H}_5\text{NO}_2^+$	$\text{C}_8\text{H}_5\text{N}(\text{O})_2$ (1H-Isoindole-1,3(2H)-dione)	85-41-6	**	9.78 ± 0.05 (V)	PE	4708
			**	9.90 (V)	PE	5614
$\text{C}_8\text{H}_7\text{NO}_2^+$	$\text{C}_6\text{H}_5(\text{OCH}_3)(\text{C}\equiv\text{NO})$ (Benzonitrile, 4-methoxy-N-oxide)	15500-73-9	**	8.42 (V)	PE	4719
$\text{C}_8\text{H}_8\text{NO}_2^+$	$\text{C}_6\text{H}_5(\text{Cl})(\text{OCH}_3)\text{NHCHO}$ (Formamide, <i>N</i> -(2-chloro-4-methoxyphenyl)-)	53666-45-8	Cl	9.4 ± 0.1	EI	4359
	$\text{C}_6\text{H}_5(\text{Cl})(\text{OCH}_3)\text{NHCHO}$ (Formamide, <i>N</i> -(2-chloro-5-methoxyphenyl)-)	53666-47-0	Cl	9.0 ± 0.1	EI	4359
$\text{C}_8\text{H}_9\text{NO}_2^+$	$\text{C}_6\text{H}_5(\text{OCH}_3)(\text{CONH}_2)$ (Benzamide, 3-methoxy-)	5813-86-5	**	8.60 (V)	PE	4918
	$\text{C}_6\text{H}_5(\text{OCH}_3)(\text{CONH}_2)$ (Benzamide, 4-methoxy-)	3424-93-9	**	8.62 (V)	PE	4918
	$\text{C}_6\text{H}_5(\text{OH})\text{CH}=\text{N}(\text{O})\text{CH}_3$	XXXXX-XX-X	**	7.76 (V)	PE	5590
	$\text{C}_6\text{H}_5(\text{OH})\text{NHCOC}_3$ (Acetamide, <i>N</i> -(2-hydroxyphenyl)-)	614-80-2	**	7.01 ± 0.02	EI	3631
	$\text{C}_6\text{H}_5(\text{OH})\text{NHCOC}_3$ (Acetamide, <i>N</i> -(4-hydroxyphenyl)-)	103-90-2	**	7.57 ± 0.02	EI	3631
	$\text{C}_6\text{H}_5(\text{CH}_3)_2\text{NO}_2$ (Benzene, 1,3-dimethyl-2-nitro-)	81-20-9	**	9.17 ± 0.015	PE	4107
			**	9.17 (V)	PE	5272
	$\text{C}_6\text{H}_5(\text{CH}_3)_2\text{NO}_2$ (Benzene, 2,4-dimethyl-1-nitro-)	89-87-2	**	9.36 (V)	PE	5272
			**	9.38 ± 0.015 (V)	PE	4107
	$\text{C}_5\text{H}_4\text{NCH}_2\text{COOCH}_3$ (2-Pyridineacetic acid methyl ester)	1658-42-0	**	9.40 ± 0.02	EI	3627
	$\text{C}_5\text{H}_4\text{NCH}_2\text{COOCH}_3$ (3-Pyridineacetic acid methyl ester)	39998-25-9	**	9.52 ± 0.02	EI	3627
	$\text{C}_5\text{H}_4\text{NCH}_2\text{COOCH}_3$ (4-Pyridineacetic acid methyl ester)	29800-89-3	**	9.62 ± 0.02	EI	3627
	$\text{C}_5\text{H}_4\text{NCOOC}_2\text{H}_5$ (4-Pyridinecarboxylic acid ethyl ester)	1570-45-2	**	9.75 ± 0.1	EI	4302
$\text{C}_8\text{H}_7\text{D}_2\text{NO}_2^+$	$\text{C}_6\text{H}_5\text{CD}_2\text{CH}_2\text{ONO}$ (Nitrous acid 2-phenylethyl-2,2- <i>d</i> ₂ ester)	67428-03-9	**	9.13 ± 0.02 (V)	PE	4722
			**	9.13 (V)	PE	4898
$\text{C}_9\text{H}_{12}\text{NO}_2^+$	$\text{C}_8\text{H}_{12}\text{NO}(\text{O})$ (9-Azabicyclo[3.3.1]non-9-yloxy,3-oxo-)	7123-92-4	**	7.4 ± 0.1	OTH	5379
$\text{C}_8\text{H}_{13}\text{NO}_2^+$	$\text{C}_4\text{H}_9\text{NO}(\text{COCH}=\text{CHCH}_3)$ (Morpholine, 4-(1-oxo-2-butenyl)-)	51944-66-2	**	8.8 ± 0.1	EI	3996
$\text{C}_9\text{H}_7\text{NO}_2^+$	$\text{C}_8\text{H}_7\text{N}(\text{O})\text{CH}_3$ (1H-Isoindole-1,3(2H)-dione, 2-methyl-)	550-44-7	**	9.55 ± 0.05 (V)	PE	4854
$\text{C}_9\text{H}_{11}\text{NO}_2^+$	$\text{C}_6\text{H}_5(\text{OCH}_3)\text{CH}=\text{N}(\text{O})\text{CH}_3$	XXXXX-XX-X	**	7.60 (V)	PE	5590
	$\text{C}_5\text{H}_5\text{N}(\text{CH}_3)=\text{CHCOOCH}_3$ (Acetic acid, (1-methyl-2(1 <i>H</i>)-pyridinylidene)-, methyl ester)	39998-21-5	**	7.02 ± 0.02	EI	3627
	$\text{C}_5\text{H}_5\text{N}(\text{CH}_3)=\text{CHCOOCH}_3$ (Acetic acid, (1-methyl-4(1 <i>H</i>)-pyridinylidene)-, methyl ester)	39998-22-6	**	6.82 ± 0.02	EI	3627

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_9\text{H}_{11}\text{NO}_2^+$	$\text{C}_7\text{H}_9\text{NCO}_2\text{CH}_3$ (2-Azabicyclo[3.2.1]octa-3,6-diene-2-carboxylic acid methyl ester)	56125-98-0	**	8.20 (V)	PE	5481
	$\text{C}_6\text{H}_2(\text{CH}_3)_3\text{NO}_2$ (Benzene, 1,3,5-trimethyl-2-nitro-)	603-71-4	**	9.01 (V)	PE	5272
	$\text{C}_6\text{H}_5\text{CH}_2\text{CH}(\text{NH}_2)\text{COOH}$ (DL-Phenylalanine)	150-30-1	**	<8.4	PI	3766
$\text{C}_9\text{H}_{13}\text{NO}_2^+$	$\text{C}_7\text{H}_{10}\text{NCO}_2\text{CH}_3$ (2-Azabicyclo[3.2.1]oct-3-ene-2-carboxylic acid methyl ester)	56125-94-1	**	8.03 (V)	PE	5481
	$\text{C}_7\text{H}_{10}\text{NCO}_2\text{CH}_3$ (2-Azabicyclo[3.2.1]oct-6-ene-2-carboxylic acid methyl ester)	56125-95-2	**	8.60 (V)	PE	5481
	$\text{C}_6\text{H}_{13}\text{NO}_2$ (1,2-Benzenediol, 4-(2-aminopropyl)-)	555-64-6	**	8.18±0.06 (V)	PE	4758
	$\text{C}_5\text{H}_5\text{N}(\text{CH}_3)\text{CH}_2\text{COOCH}_3$ (3-Pyridineacetic acid, 1,4-dihydro-1-methyl-, methyl ester)	39998-23-7	**	6.94±0.02	EI	3627
$\text{C}_9\text{H}_{14}\text{NO}_2^+$	$\text{C}_7\text{H}_8\text{NO}(\text{=O})(\text{CH}_3)_2$ (8-Azabicyclo[3.2.1]oct-8-yloxy,1,5-dimethyl-3-oxo-)	34061-60-4	**	7.4±0.1	OTH	5379
$\text{C}_9\text{H}_{15}\text{NO}_2^+$	$\text{C}_7\text{H}_{12}\text{NCO}_2\text{CH}_3$ (2-Azabicyclo[3.2.1]octane-2-carboxylic acid methyl ester)	71017-44-2	**	8.70 (V)	PE	5481
	$\text{C}_7\text{H}_{12}\text{NOCOCH}_3$ (1-Azabicyclo[2.2.2]octane-4-ol acetate(ester))	26458-76-4	**	8.42±0.015 (V)	PE	4286
	$\text{C}_5\text{HN}(\text{=O})_2(iso-\text{C}_3\text{H}_5)_2$ (2,4-Azetidinedione, 3,3-bis(1-methylethyl)-)	17197-62-5	**	9.42	EI	4660
$\text{C}_9\text{H}_{16}\text{NO}_2^+$	$\text{C}_5\text{H}_4\text{N}(\text{O})(\text{=O})(\text{CH}_3)_4$ (1-Piperidinyloxy, 2,2,6,6-tetramethyl-4-oxo-)	2896-70-0	**	7.40±0.05	EI	3494
			**	7.4±0.1	OTH	5379
$\text{C}_9\text{H}_{17}\text{NO}_2^+$	$\text{C}_5\text{H}_4\text{N}(\text{=O})(\text{OH})(\text{CH}_3)_4$ (4-Piperidinone, 1-hydroxy-2,2,6,6-tetramethyl-)	3637-11-4	**	8.51±0.05	EI	3494
	$trans-(\text{C}_2\text{H}_5)_2\text{NCH}=\text{CHCOO}_2\text{H}_5$	13894-28-5	**	7.63 (V)	PE	388
$\text{C}_9\text{H}_{18}\text{NO}_2^+$	$\text{C}_5\text{H}_5\text{NO}(\text{CH}_3)_4\text{OH}$ (1-Piperidinyloxy, 4-hydroxy-2,2,6,6-tetramethyl-)	2226-96-2	**	7.4±0.1	OTH	5379
$\text{C}_{10}\text{H}_7\text{NO}_2^+$	$\text{C}_{10}\text{H}_7\text{NO}_2$ (Naphthalene, 1-nitro-)	86-57-7	**	8.60±0.01	PI	5505
	$\text{C}_{10}\text{H}_7\text{NO}_2$ (Naphthalene, 2-nitro-)	581-89-5	**	8.67±0.01	PI	5505
$\text{C}_{10}\text{H}_{13}\text{NO}_2^+$	$\text{C}_6\text{H}_4(\text{NO}_2)\text{C}_4\text{H}_9$ (Benzene, 1-butyl-3-nitro-)	20651-76-7	**	9.94±0.1	EI	3629
	$\text{C}_6\text{H}_4(\text{NO}_2)\text{C}_4\text{H}_9$ (Benzene, 1-butyl-4-nitro-)	20651-75-6	**	10.07±0.1	EI	3629
	$\text{C}_6\text{H}_5\text{O}_2\text{CH}_2\text{CH}(\text{NH}_2)\text{CH}_3$ (1,3-Benzodioxole, 5-ethanamine- α -methyl-(\pm)-)	51497-09-7	**	8.01±0.06 (V)	PE	4758
$\text{C}_{10}\text{H}_{15}\text{NO}_2^+$	$\text{C}_6\text{H}_5(\text{OCH}_3)_2\text{CH}_2\text{CH}_2\text{NH}_2$ (Benzeneethanamine, 3,4-dimethoxy-)	120-20-7	**	8.03±0.16 (V)	PE	4672

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₀H₁₇NO₂⁺	C ₁₀ H ₁₇ NO ₂ (8-Azabicyclo[3.2.1]octan-3-ol,8-methyl acetate(ester), <i>endo</i> -)	3423-27-6	**	8.0±0.15	EI	5401
	C ₁₀ H ₁₇ NO ₂ (8-Azabicyclo[3.2.1]octan-3-ol,8-methyl-acetate(ester), <i>exo</i> -)	3423-26-5	**	8.1±0.15	EI	5401
	C ₁₀ H ₁₇ NO ₂ (2,4-Azetidinedione, 1-methyl-3,3-bis(1-methylethyl)-)	38951-66-5	**	9.27	EI	4660
C₁₁H₉NO₂⁺	C ₁₁ H ₉ (NO ₂) (1,4-Methanonaphthalene, 1,4-dihydro-5-nitro-)	58673-43-1	**	8.87±0.05 (V)	PE	5019
	C ₁₁ H ₉ (NO ₂) (1,4-Methanonaphthalene, 1,4-dihydro-6-nitro-)	XXXXX-XX-X	**	8.96±0.05 (V)	PE	5019
C₁₁H₁₁NO₂⁺	C ₃ HN(=O) ₂ C ₆ H ₅ (C ₆ H ₅) (2,4-Azetidinedione, 3-ethyl-3-phenyl-)	42282-82-6	**	8.90	EI	4660
	C ₁₁ H ₁₁ NO ₂ (Carbamic acid, 1,3-butadienyl-phenyl ester, (E)-)	61759-55-5	**	8.30 (V)	PE	4803
C₁₁H₁₇NO₂⁺	C ₁₁ H ₁₇ NO ₂ (Benzeneethanamine, 2,5-dimethoxy- α -methyl-(\pm)-)	13641-74-2	**	7.70±0.06 (V)	PE	4758
	C ₁₁ H ₁₇ NO ₂ (Benzeneethanamine, 2,4-dimethoxy- α -methyl-(\pm)-)	52850-81-4	**	7.91±0.06 (V)	PE	4758
	C ₁₁ H ₁₇ NO ₂ (Benzeneethanamine, 3,4-dimethoxy- α -methyl-)	120-26-3	**	8.18±0.06 (V)	PE	4758
				8.03±0.06 (V)	PE	4758
C₁₂H₇NO₂⁺	C ₁₂ H ₇ N(=O) ₂ (1H-Benz[de]isoquinoline-1,3(2H)-dione)	81-83-4	**	8.68±0.05 (V)	PE	5095
C₁₂H₁₁NO₂⁺	C ₂₅ H ₃₇ N ₃ O ₄ (L-Tryptophan,N-[N-(1-oxodecyl)-L-alanyl]-methyl ester)	55728-17-1		8.9±0.1	PI	5279
C₁₂H₁₃NO₂⁺	C ₁₂ H ₁₃ NO ₂ (2,4-Azetidinedione, 3-ethyl-1-methyl-3-phenyl-)	56519-51-8	**	8.82	EI	4660
C₁₂H₁₄NO₂⁺	C ₂₇ H ₄₉ N ₄ O ₆ S (L-Cysteine,S-(2-methoxy-2-oxoethyl)-N-[N-[N-(1-oxopropyl)-L-phenylalanyl]-L-leucyl]glycyl]-methyl ester)	35146-63-5		9.1±0.1	PI	5279
C₁₂H₁₉NO₂⁺	C ₁₂ H ₁₉ NO ₂ (Benzeneethanamine, 2,5-dimethoxy- α ,4-dimethyl-(\pm)-)	26011-50-7	**	7.62±0.06 (V)	PE	4758
C₁₂H₂₂NO₂⁺	C ₃₀ H ₄₅ N ₃ O ₆ (L-Alanine,N-[N-[N-(1-oxodecyl)glycyl]-L-tryptophyl]-L-alanyl]-methyl ester)	19716-78-0		9.5±0.1	PI	5279
C₁₃H₈NO₂⁺	C ₆ H ₅ COC ₆ H ₄ NO ₂ (Methanone, (3-nitrophenyl)phenyl-)	2243-80-3		11.2±0.1	EI	4358
	C ₆ H ₅ COC ₆ H ₄ NO ₂ (Methanone, (4-nitrophenyl)phenyl-)	1144-74-7		11.5±0.1	EI	4358
C₁₃H₉NO₂⁺	C ₁₃ H ₉ N(=O) ₂ (1H-Benz[de]isoquinoline-1,3(2H)-dione,2-methyl-)	2382-08-3	**	8.57±0.05 (V)	PE	5095

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{13}\text{H}_{10}\text{NO}_2^+$	$(\text{C}_6\text{H}_4\text{NO}_2)_2\text{CH}_2$ (Benzene, 1,1'-methylenebis[4-nitro-])	1817-74-9	NO_2	11.1 ± 0.1	EI	3807
$\text{C}_{13}\text{H}_{11}\text{NO}_2^+$	$\text{C}_6\text{H}_5\text{CH}_2\text{C}_6\text{H}_4\text{NO}_2$ (Benzene, 1-nitro-4-(phenylmethyl)-)	1817-77-2	**	9.35 ± 0.05	EI	3806
$\text{C}_{13}\text{H}_{12}\text{NO}_2^+$	$\text{C}_6\text{H}_4(\text{COC}_5\text{H}_8\text{N})_2$ (Pyridine, 1,1'-(1,2-phenylenedicarbonyl)bis[1,2,3,4-tetrahydro-])	52881-76-2		10.5	EI	4346
	$\text{C}_6\text{H}_4(\text{COC}_5\text{H}_8\text{N})_2$ (Pyridine, 1,1'-(1,4-phenylenedicarbonyl)bis[1,2,3,4-tetrahydro-])	52881-77-3		11.6	EI	4346
$\text{C}_{13}\text{H}_{14}\text{NO}_2^+$	$\text{C}_5\text{H}_{10}\text{NCOC}_6\text{H}_4\text{COC}_5\text{H}_8\text{N}$ (Pyridine, 1,2,3,4-tetrahydro-1-[4-(1-piperidinylcarbonyl)benzoyl]-)	52881-78-4		11.4	EI	4346
	$\text{C}_6\text{H}_4(\text{COC}_5\text{H}_{10}\text{N})_2$ (Piperidine, 1,1'-(1,2-phenylenedicarbonyl)bis-)	38256-33-6		11.8	EI	4346
	$\text{C}_6\text{H}_4(\text{COC}_5\text{H}_{10}\text{N})_2$ (Piperidine, 1,1'-(1,4-phenylenedicarbonyl)bis-)	15088-30-9		12.7	EI	4346
$\text{C}_{13}\text{H}_{15}\text{NO}_2^+$	$\text{C}_3\text{N}(=\text{O})_2(\text{C}_6\text{H}_5)_2\text{C}_6\text{H}_5$ (2,4-Azetidinedione, 3,3-diethyl-1-phenyl-)	15745-94-5	**	8.71	EI	4660
$\text{C}_{13}\text{H}_{24}\text{NO}_2^+$	$\text{C}_{25}\text{H}_{37}\text{N}_3\text{O}_4$ (L-Tryptophan, N-[N-(1-oxodecyl)-L-alanyl]-methyl ester)	55728-17-1		9.5 ± 0.1	PI	5279
	$\text{C}_{20}\text{H}_{34}\text{N}_4\text{O}_4$ (L-Histidine, N-[N-(1-oxodecyl)-L-alanyl]-methyl ester)	31944-64-6		8.8 ± 0.1	PI	5279
	$\text{C}_{17}\text{H}_{32}\text{N}_2\text{O}_5$ (L-Serine, N-[N-(1-oxodecyl)-L-alanyl]-methyl ester)	55728-15-9		9.6 ± 0.1	PI	5279
	$\text{C}_{20}\text{H}_{37}\text{N}_3\text{O}_5$ (L-Alanine, N-[N-[N-(1-oxodecyl)-L-alanyl]-L-alanyl]-methyl ester)	55728-11-5		9.5 ± 0.1	PI	5279
	$\text{C}_{23}\text{H}_{43}\text{N}_3\text{O}_5$ (L-Alanine, N-[N-[N-(1-oxodecyl)-L-alanyl]-L-leucyl]-methyl ester)	55728-12-6		9.6 ± 0.1	PI	5279
$\text{C}_{14}\text{H}_9\text{NO}_2^+$	$\text{C}_{14}\text{H}_9\text{NO}_2$ (Anthracene, 9-nitro-)	602-60-8	**	7.88 ± 0.03 (V)	PE	4887
	$\text{C}_6\text{H}_4\text{C}_3\text{O}_2\text{NC}_5\text{H}_5$ (Pyridinium,2,3-dihydro-1,3-dioxo-1H-indene-2-ylide)	1283-93-8	**	7.6	CTS	5592
$\text{C}_{14}\text{H}_{13}\text{NO}_2^+$	$\text{C}_6\text{H}_5\text{CH}_2\text{CH}_2\text{C}_6\text{H}_4\text{NO}_2$ (Benzene, 1-nitro-4-(2-phenylethyl)-)	14310-29-3	**	9.17 ± 0.05	EI	3806
$\text{C}_{15}\text{H}_{11}\text{NO}_2^+$	$\text{C}_6\text{H}_4\text{C}_3\text{O}_2\text{C}_5\text{H}_4\text{NCH}_3$ (Pyridinium,3-(1,3-dihydro-1,3-dioxo-2H-inden-2-yl)-1-methyl-hydroxide, inner salt)	59804-88-5	**	7.20	CTS	5592
	$\text{C}_6\text{H}_4\text{C}_3\text{O}_2\text{C}_5\text{H}_4\text{NCH}_3$ (Pyridinium,1,3-dihydro-1,3-dioxo-2H-inden-2-ylide)	59804-81-8	**	7.35	CTS	5592
	$\text{C}_6\text{H}_4\text{C}_3\text{O}_2\text{NC}_5\text{H}_4\text{CH}_3$ (Pyridinium,3-methyl-1,3-dihydro-1,3-dioxo-2H-inden-2-ylide)	59804-82-9	**	7.55	CTS	5592
				7.55	CTS	5592
$\text{C}_{15}\text{H}_{15}\text{NO}_2^+$	$\text{C}_6\text{H}_5\text{CH}_2\text{OC}_6\text{H}_4\text{NHCOCH}_3$ (Acetamide,N-[4-(phenylmethoxy)phenyl]-)	41927-14-4	**	7.88	CTS	5336

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₅H₁₆NO₂⁺	C ₆ H ₄ (CH ₂ COC ₅ H ₈ N) ₂ (Pyridine, 1,1'-[1,2-phenylenebis(1-oxo-2,1-ethanediyl)]bis[1,2,3,4-tetrahydro-])	52881-80-8		10.8	EI	4346
C₁₅H₁₈NO₂⁺	C ₆ H ₄ (CH ₂ COC ₅ H ₁₀ N) ₂ (Piperidine, 1,1'-[1,2-phenylenebis(1-oxo-2,1-ethanediyl)]bis-)	52881-79-5		12.1	EI	4346
C₁₆H₁₃NO₂⁺	C ₆ H ₆ N(O)(C ₆ H ₄ OCH ₃) (Isoquinolinium, 4-hydroxy-2-(4-methoxyphenyl)-hydroxide, inner salt)	56359-30-9	**	6.93±0.05	EI	4863
	C ₆ H ₆ N(=O)(C ₆ H ₄ OCH ₃) (Indeno[1,2- <i>b</i>]azirin-6(1H)-one, 1a,6a-dihydro-1-(4-methoxyphenyl)-)	55507-32-9	**	7.68±0.05	EI	4863
	C ₆ H ₆ N ₃ (=O)(C ₆ H ₄ OCH ₃) (Indeno[1,2- <i>d</i>]triazol-8(3H)-one, 3a,8a-dihydro-3-(4-methoxyphenyl)-)	55507-28-3	N ₂	7.8±0.1	EI	4863
C₁₈H₁₁NO₂⁺	C ₆ H ₄ C ₃ O ₂ NC ₉ H ₇ (Isoquinolinium, 1,3-dihydro-1,3-dioxo-2H-inden-2-ylide)	27609-07-0	**	7.5	CTS	5592
	C ₆ H ₄ C ₃ O ₂ NC ₉ H ₇ (Quinolinium, 1,3-dihydro-1,3-dioxo-2H-inden-2-ylide)	59804-80-7	**	7.45	CTS	5592
C₁₉H₁₆NO₂⁺	(C ₆ H ₄ COCl ₅ H ₈ N) ₂ (Pyridine, 1,1'-(1,1'-biphenyl]-2,2'-diyldicarbonyl)bis[1,2,3,4-tetrahydro-])	52882-85-6		10.9	EI	4346
	(C ₆ H ₄ COCl ₅ H ₈ N) ₂ (Pyridine, 1,1'-(1,1'-biphenyl]-4-4'diyldicarbonyl)bis[1,2,3,4-tetrahydro-])	52882-88-9		11.1	EI	4346
C₁₉H₁₈NO₂⁺	C ₆ H ₄ (COCl ₅ H ₈ N)C ₆ H ₄ COCl ₅ H ₁₀ N (Pyridine, 1,2,3,4-tetrahydro-1-[[2'-(1-piperidinylcarbonyl)[1,1'-biphenyl]-2-carbonyl]-])	52882-86-7		10.4	EI	4346
	(C ₆ H ₄ COCl ₅ H ₁₀ N) ₂ (Piperidine, 1,1'-(1,1'-biphenyl]-2,2-diyldicarbonyl)bis-)	52882-84-5		12.0	EI	4346
	(C ₆ H ₄ COCl ₅ H ₁₀ N) ₂ (Piperidine, 1,1'-(1,1'-biphenyl]-4-4'-diyldicarbonyl)bis-)	52882-87-8		12.3	EI	4346
C₂₀H₁₅NO₂⁺	C ₃ (C ₆ H ₅) ₂ O ₂ NC ₅ H ₅ (Pyridinium, 1-benzoyl-2-oxo-2-phenylethylide)	17281-65-1	**	8.14	CTS	5591
C₂₁H₁₅NO₂⁺	C ₃ N(=O) ₂ (C ₆ H ₅) ₃ (2,4-Azetidinedione, 1,3,3-triphenyl-)	15745-93-4	**	8.37	EI	4660
C₂₄H₁₇NO₂⁺	C ₃ (C ₆ H ₅) ₂ O ₂ NC ₉ H ₇ (Quinolinium, 1-benzoyl-2-oxo-2-phenylethylide)	XXXXX-XX-X	**	7.92	CTS	5591
C₂H₄N₂O₂⁺	NH ₂ COCONH ₂	471-46-5	**	9.41	PE	4487
			**	9.80 (V)	PE	4462
			**	9.80 (V)	PE	5517
	NH ₂ CONHCHO	1190-24-5	**	10.58 (V)	PE	4599
C₂H₆N₂O₂⁺	(CH ₃) ₂ NNO ₂ <i>trans</i> -(CH ₃ NO) ₂	4164-28-7	**	9.53	PE	4647
		XXXXX-XX-X	**	8.68 (V)	PE	4465
C₃H₄N₂O₂⁺	C ₂ HN ₂ O ₂ CH ₃ (Sydnone, 3-methyl-)	6939-12-4	**	9.0	CTS	4348

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_3\text{H}_6\text{N}_2\text{O}_2^+$	$\text{CH}_3\text{CONHCONH}_2$	591-07-1	**	10.3 (V)	PE	4599
$\text{C}_3\text{H}_8\text{N}_2\text{O}_2^+$	$(\text{CH}_3)_2\text{NCH}_2\text{NO}_2$	53915-73-4	**	9.17 (V)	PE	4192
$\text{C}_4\text{H}_4\text{N}_2\text{O}_2^+$						
	$\text{C}_4\text{H}_4\text{N}_2\text{O}_2$ (Pyrazine, 1,4-dioxide)	2423-84-9	**	8.33 ± 0.02 (V)	PE	4470
	$\text{C}_4\text{H}_4\text{N}_2\text{O}_2$ (Pyridazine, 1,2-dioxide)	19194-87-7	**	8.51 ± 0.02 (V)	PE	4470
	$\text{C}_4\text{H}_4\text{N}_2(=\text{O})_2$ (2,4(1H,3H)-Pyrimidinedione)	66-22-8	**	9.45 (V)	PE	4754
			**	9.50 ± 0.03 (V)	PE	4445
			**	9.59 (V)	PE	5472
			**	9.60 (V)	PE	4599
			**	9.68 (V)	PE	5577
			**	9.35 ± 0.1	EI	5555
			**	9.53 ± 0.02	EI	3571
	$\text{C}_4\text{H}_4\text{NNO}_2$ (Pyrrole, 2-nitro-)	5919-26-6	**	9.30 ± 0.05	EI	3482
$\text{C}_4\text{H}_6\text{N}_2\text{O}_2^+$						
	$\text{C}_2\text{N}_2\text{O}(=\text{O})(\text{CH}_3)_2$ (1,3,4-Oxadiazol-2(5H)-one, 5,5-dimethyl-)	28873-61-2	**	10.20 (V)	PE	4929
	$\text{C}_4\text{H}_6\text{N}_2\text{O}_2$ (2,4(1H,3H)-Pyrimidinedione, dihydro-)	504-07-4	**	10.00 (V)	PE	4599
			**	10.0 ± 0.1	EI	5555
$\text{C}_4\text{H}_8\text{N}_2\text{O}_2^+$	$\text{CH}_3\text{NHCOCONHCH}_3$	615-35-0	**	9.33	PE	4462
$\text{C}_5\text{H}_4\text{N}_2\text{O}_2^+$						
	$\text{C}_5\text{H}_4\text{NNO}_2$ (Pyridine, 2-nitro-)	15009-91-3	**	10.1 ± 0.1	EI	4302
	$\text{C}_5\text{H}_4\text{NNO}_2$ (Pyridine, 3-nitro-)	2530-26-9	**	10.3 ± 0.1	EI	4302
	$\text{C}_5\text{H}_4\text{NNO}_2$ (Pyridine, 4-nitro-)	1122-61-8	**	10.4	PE	4536
			**	10.2 ± 0.1	EI	4302
$\text{C}_5\text{H}_6\text{N}_2\text{O}_2^+$						
	$\text{C}_4\text{H}_3\text{N}_2(=\text{O})_2\text{CH}_3$ (2,4(1H,3H)-Pyrimidinedione, 1-methyl-)	615-77-0	**	9.0 ± 0.1	EI	5555
	$\text{C}_4\text{H}_3\text{N}_2(=\text{O})_2\text{CH}_3$ (2,4(1H,3H)-Pyrimidinedione, 5-methyl-)	65-71-4	**	9.02 (V)	PE	4754
			**	9.14 ± 0.03 (V)	PE	4445
			**	9.20 (V)	PE	4599
			**	8.95 ± 0.1	EI	5555
$\text{C}_6\text{H}_6\text{N}_2\text{O}_2^+$						
	$\text{C}_6\text{H}_4\text{NH}_2(\text{NO}_2)$ (Benzeneamine, 2-nitro-)	88-74-4	**	8.27 ± 0.01	PI	5552
			**	8.43 (V)	PE	3856
	$\text{C}_6\text{H}_4\text{NH}_2(\text{NO}_2)$ (Benzeneamine, 3-nitro-)	99-09-2	**	8.31 ± 0.02	PI	5552
			**	8.60 (V)	PE	3856
	$\text{C}_6\text{H}_4\text{NH}_2(\text{NO}_2)$ (Benzeneamine, 4-nitro-)	100-01-6	**	8.34 ± 0.01	PI	5552
			**	8.60 (V)	PE	3856
			**	8.43	EI	4089
			**	8.62 ± 0.1	EI	3447

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₁H₂₁N₂O₂⁺	C ₅ H ₅ N(O)(CH ₃) ₄ NHCOCH ₃ (1-Piperidinyloxy, 4-(acetylamino)-2,2,6,6-tetramethyl-)	14691-89-5	**	7.40±0.05	EI	3494
C₁₂H₁₀N₂O₂⁺	(NO ₂)C ₆ H ₄ C ₆ H ₄ NH ₂ ([1,1'-Biphenyl]-4-amine-4'-nitro-) C ₁₀ H ₄ N ₂ (=O) ₂ (CH ₃) ₂ (Cyclobuta[b]quinoxaline-1,2-dione, 3,8-dihydro-3,8-dimethyl-)	1211-40-1 33527-99-0	**	7.46±0.03 7.13 (V)	PI PE	5552 4861
C₁₂H₂₀N₂O₂⁺	C ₁₂ H ₂₀ O ₂ N ₂ (2-Pentanone, 4,4'-(1,2-ethanediyldinitrilo)bis-)	6310-76-5	**	7.71 (V)	PE	3822
C₁₃H₁₀N₂O₂⁺	C ₁₂ H ₇ N ₂ O(OCH ₃) (Phenazine, 2-methoxy-10-oxide) C ₁₂ H ₇ N ₂ O(OCH ₃) (Phenazine, 2-methoxy-5-oxide) NO ₂ C ₆ H ₄ N=CHC ₆ H ₅ (Benzenamine, 4-nitro-N-(phenylmethylene-)) C ₆ H ₅ NNC ₆ H ₄ COOH (Benzoic acid, 4-(phenylazo)-(E)-) C ₆ H ₅ (NO ₂)CH=CHC ₆ H ₅ N (Pyridine, trans-3-[2-(4-nitrophenyl)ethenyl-])	2876-29-1 3224-54-2 69173-79-1 37790-20-8 5847-74-5	**	7.74 (V) 7.84 (V) 8.76 (V) ~8.75 (V) 8.58±0.05 (V)	PE PE PE PE PE	4590 4590 5486 5320 4377
C₁₃H₁₂N₂O₂⁺	C ₆ H ₄ (NO ₂)CH ₂ C ₆ H ₄ NH ₂ (Benzenamine, 4-[(4-nitrophenyl)methyl-])	726-17-0	**	7.87±0.05	EI	3806
C₁₃H₂₅N₂O₂⁺	C ₂₀ H ₃₄ N ₄ O ₄ (L-Histidine, N-[N-(1-oxodecyl)-L-alanyl]-methyl ester)	31944-64-6		9.1±0.1	PI	5279
C₁₄H₁₀N₂O₂⁺	(C ₆ H ₅) ₂ C ₂ N ₂ O ₂ (Sydnone, 3,4-diphenyl-)	3815-83-6		7.8	CTS	4348
C₁₄H₁₂N₂O₂⁺	NO ₂ C ₆ H ₄ N=CHC ₆ H ₄ CH ₃ (Benzenamine, N-[(3-methylphenyl)methylene]-4-nitro-) NO ₂ C ₆ H ₃ (CH ₃)N=CHC ₆ H ₅ (Benzenamine, 2-methyl-4-nitro-N-(phenylmethylene)-)	XXXXX-XX-X XXXXX-XX-X	**	8.58 (V) 8.66 (V)	PE PE	5486 5486
C₁₄H₁₄N₂O₂⁺	C ₆ H ₄ (NH ₂)CH ₂ CH ₂ C ₆ H ₄ NO ₂ (Benzenamine, 4-[2-(4-nitrophenyl)ethyl-])	7357-96-2	**	7.78±0.05	EI	3806
C₁₅H₁₄N₂O₂⁺	NO ₂ C ₆ H ₂ (CH ₃) ₂ N=CHC ₆ H ₅ (Benzenamine, 2,6-dimethyl-4-nitro-N-(phenylmethylene)-) NO ₂ C ₆ H ₃ (CH ₃)N=CHC ₆ H ₄ CH ₃ (Benzenamine, 2-methyl-4-nitro-N-[(3-methylphenyl)methylene]-)	XXXXX-XX-X XXXXX-XX-X	**	8.51 (V) ~8.49 (V)	PE PE	5486 5486
C₁₅H₂₀N₂O₂⁺	C ₁₅ H ₂₀ N ₂ O ₂ (8-Azabicyclo[3.2.1]octan-3-ol,8-methyl-phenylcarbamate(ester), <i>exo</i> -)	29364-21-4	**	8.0±0.15	EI	5401
C₁₅H₂₉N₂O₂⁺	C ₂₀ H ₃₇ N ₃ O ₅ (L-Alanine, N-[N-[N-(1-oxodecyl)-L-alanyl]-L-alanyl]-methyl ester)	55728-11-5		8.9±0.1	PI	5279

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₅H₃₀N₂O₂⁺	C ₂₀ H ₃₇ N ₃ O ₅ (L-Alanine,N-[N-[N-(1-oxodecyl)-L-alanyl]-L-alanyl]-methyl ester)	55728-11-5		8.7±0.1	PI	5279
C₁₆H₁₀N₂O₂⁺	C ₁₆ H ₁₀ N ₂ O ₂ ([Δ ^{2,2'} -Biindoline]-3,3'-dione)	12626-73-2	**	7.17	PI	3586
C₁₆H₁₂N₂O₂⁺	C ₆ H ₄ (NO ₂)C ₃ H ₃ (CN)C ₆ H ₅ (Cyclopropanecarbonitrile, 1-(p-nitrophenyl)-2-phenyl-)	10432-22-1	**	9.05±0.10	EI	3575
C₁₆H₁₆N₂O₂⁺	NO ₂ C ₆ H ₂ (CH ₃) ₂ N=CHC ₆ H ₄ CH ₃ (Benzenamine,2,6-dimethyl-4-nitro-N-[3-methylphenyl)methylene]-) C ₁₆ H ₁₆ N ₂ O ₂ 94-93-9 (Phenol, 2,2'-[1,2-ethanediylbis(nitrilmethylidyne)]bis-)	XXXXX-XX-X	**	8.43 (V)	PE	5486
C₁₇H₂₅N₂O₂⁺	C ₂₇ H ₄₀ N ₄ O ₆ S (L-Cysteine,S-(2-methoxy-2-oxoethyl)-N-[N-[N-(1-oxopropyl)-L-phenylalanyl]-L-leucyl]glycyl]-methyl ester)	35146-63-5		8.8±0.1	PI	5279
C₁₇H₃₃N₂O₂⁺	C ₁₉ H ₃₆ N ₂ O ₄ (Glycine,N-[N-(1-oxodecyl)-L-leucyl]-methyl ester)	55728-14-8		9.6±0.1	PI	5279
C₁₈H₁₆N₂O₂⁺	C ₆ H ₄ (NH ₂)OC ₆ H ₄ OC ₆ H ₄ NH ₂ (Benzenamine, 4,4'-[1,4-phenylenebis(oxy)]bis-)	3491-12-1	**	6.60	PI	4328
C₁₈H₂₀N₂O₂⁺	C ₆ H ₄ (COC ₅ H ₈ N) ₂ (Pyridine, 1,1'-(1,2-phenyleneddicarbonyl)bis[1,2,3,4-tetrahydro-]) C ₆ H ₄ (COC ₅ H ₈ N) ₂ 52881-77-3 (Pyridine, 1,1'-(1,4-phenyleneddicarbonyl)bis[1,2,3,4-tetrahydro-])	52881-76-2	**	8.7	EI	4346
C₁₈H₂₂N₂O₂⁺	C ₅ H ₁₀ NCOC ₆ H ₄ COC ₅ H ₈ N (Pyridine, 1,2,3,4-tetrahydro-1-[4-(1-piperidinylcarbonyl)benzoyl]-)	52881-78-4	**	8.7	EI	4346
C₁₈H₂₄N₂O₂⁺	C ₆ H ₄ (COC ₅ H ₁₀ N) ₂ (Piperidine, 1,1'-(1,2-phenyleneddicarbonyl)bis-) C ₆ H ₄ (COC ₅ H ₁₀ N) ₂ 15088-30-9 (Piperidine, 1,1'-(1,4-phenyleneddicarbonyl)bis-)	38256-33-6	**	8.9	EI	4346
C₁₈H₃₅N₂O₂⁺	C ₂₃ H ₄₃ N ₃ O ₅ (L-Alanine,N-[N-[N-(1-oxodecyl)-L-alanyl]-L-leucyl]-methyl ester)	55728-12-6		8.5±0.1	PI	5279
C₂₀H₂₄N₂O₂⁺	C ₆ H ₄ (CH ₂ COC ₅ H ₈ N) ₂ (Pyridine, 1,1'-(1,2-phenylenebis(1-oxo-2,1-ethanediyl))bis[1,2,3,4-tetrahydro-])	52881-80-8	**	8.6	EI	4346
C₂₀H₂₈N₂O₂⁺	C ₆ H ₄ (CH ₂ COC ₅ H ₁₀ N) ₂ (Piperidine, 1,1'-(1,2-phenylenebis(1-oxo-2,1-ethanediyl))bis-)	52881-79-5	**	8.8	EI	4346
C₂₁H₁₄N₂O₂⁺	C ₃ (C ₆ H ₅) ₂ O ₂ NC ₅ H ₄ CN (Pyridinium,4-cyano-1-benzoyl-2-oxo-2-phenylethylide)	59805-16-2	**	7.94	CTS	5591

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₂₁H₂₆N₂O₂⁺	C ₂₁ H ₂₆ N ₂ O ₂ (Phenol, 2,2'-[1,7-heptanediylibis(nitrilomethylidyne)]bis-)	52279-42-2	**	8.26±0.06	EI	4213
C₂₄H₂₄N₂O₂⁺	(C ₆ H ₄ COC ₅ H ₈ N) ₂ (Pyridine, 1,1'-(1,1'-biphenyl)-2,2'-diyldicarbonyl)bis[1,2,3,4-tetrahydro-] (C ₆ H ₄ COC ₅ H ₈ N) ₂ (Pyridine, 1,1'-(1,1'-biphenyl)-4-4'diyldicarbonyl)bis[1,2,3,4-tetrahydro-])	52882-85-6 52882-88-9	**	8.4 8.3	EI EI	4346 4346
C₂₄H₂₆N₂O₂⁺	C ₆ H ₄ (COC ₅ H ₈ N)C ₆ H ₄ COC ₅ H ₁₀ N (Pyridine, 1,2,3,4-tetrahydro-1-[2'-(1-piperidinylcarbonyl)[1,1'-biphenyl]-2-carbonyl-])	52882-86-7	**	8.4	EI	4346
C₂₄H₂₈N₂O₂⁺	(C ₆ H ₄ COC ₅ H ₁₀ N) ₂ (Piperidine, 1,1'-(1,1'-biphenyl)-2,2-diyldicarbonyl)bis-) (C ₆ H ₄ COC ₅ H ₁₀ N) ₂ (Piperidine, 1,1'-(1,1'-biphenyl)-4-4'diyldicarbonyl)bis-)	52882-84-5 52882-87-8	**	8.5 8.4	EI EI	4346 4346
C₂₅H₁₈N₂O₂⁺	C ₅ (C ₆ H ₅) ₂ O ₂ (NC ₅ H ₄) ₂ (4,4'-Bipyridinium,1-benzoyl-2-oxo-2-phenylethylide)	59805-17-3	**	7.66	CTS	5591
C₉H₁₀N₃O₂⁺	C ₆ H ₄ (NO ₂)N=CHN(CH ₃) ₂ (Methanimidamide, N,N-dimethyl-N'-(3-nitrophenyl)-) C ₆ H ₄ (NO ₂)N=CHN(CH ₃) ₂ (Methanimidamide, N,N-dimethyl-N'-(4-nitrophenyl)-) C ₆ H ₃ (Cl)(NO ₂)N=CHN(CH ₃) ₂ (Methanimidamide, N'-(2-chloro-4-nitrophenyl)-N,N-dimethyl-) C ₆ H ₃ (Cl)(NO ₂)N=CHN(CH ₃) ₂ (Methanimidamide, N'-(2-chloro-5-nitrophenyl)-N,N-dimethyl-)	2103-47-1 1205-59-0 53666-38-9 53666-43-6	H H Cl Cl	9.5±0.1 9.5±0.1 9.3±0.1 9.1±0.1	EI EI EI EI	4359 4359 4359 4359
C₉H₁₁N₃O₂⁺	C ₆ H ₄ (NO ₂)N=CHN(CH ₃) ₂ (Methanimidamide, N,N-dimethyl-N'-(3-nitrophenyl)-) C ₆ H ₄ (NO ₂)N=CHN(CH ₃) ₂ (Methanimidamide, N,N-dimethyl-N'-(4-nitrophenyl)-)	2103-47-1 1205-59-0	** **	7.8±0.1 7.9±0.1	EI EI	4359 4359
C₁₂H₉N₃O₂⁺	C ₆ H ₅ NNC ₆ H ₄ NO ₂ (Diazene,(nitrophenyl)phenyl-(E)-)	37790-23-1	**	9.05±0.05 (V)	PE	5320
C₁₅H₁₅N₃O₂⁺	C ₁₁ H ₃ N ₃ (=O) ₂ (CH ₃) ₄ (Benzo[g]pyrido[2,3 d]pyrimidin-2,4-dione, 3,7,8,10-tetramethyl-)	XXXXXX-XX-X	**	7.94 (V)	PE	4992
C₁₈H₁₇N₃O₂⁺	C ₆ H ₄ (NO ₂)C ₃ H ₃ (CN)C ₆ H ₄ N(CH ₃) ₂ (Cyclopropanecarbonitrile, 2-(p-(dimethylamino)phenyl)-1-(p-nitrophenyl)-)	28752-34-3	**	8.30±0.07	EI	3575
C₂₀H₂₅N₃O₂⁺	C ₂₀ H ₂₅ N ₃ O ₂ (Phenol, 2,2'-[iminobis(3,1-propanediylnitrilomethylidyne)]bis-)	52279-45-5	**	8.31±0.07	EI	4213
C₂₂H₃₂N₃O₂⁺	C ₃₀ H ₄₅ N ₅ O ₆ (L-Alanine,N-[N-[N-(1-oxodecyl)glycyl]-L-tryptophyl]-L-alanyl]-methyl ester)	19716-78-0		9.6±0.1	PI	5279
C₅H₄N₄O₂⁺	C ₅ H ₄ N ₄ (=O) ₂ (1H-Purine-2,6-dione,3,7-dihydro-)	69-89-6	** **	8.55 8.89±0.03 (V)	PE PE	5093 4445

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₆H₄N₄O₂⁺	C ₆ H ₄ N ₄ (=O) ₂ (2,4-(1H,3H)-Pteridinedione)	487-21-8	**	9.20 (V)	PE	5577
C₈H₁₀N₄O₂⁺	C ₅ HN ₄ (=O) ₂ (CH ₃) ₃ (1H-Purine-2,6-dione,3,7-dihydro-1,3,7-trimethyl-)	58-08-2	**	7.95	PE	5093
C₁₂H₁₀N₄O₂⁺	C ₁₀ H ₄ N ₄ (=O) ₂ (CH ₃) ₂ (Benzog[<i>g</i>]pteridine-2,4(1H,3H)-dione, 1,3-dimethyl-) C ₁₀ H ₄ N ₄ (=O) ₂ (CH ₃) ₂ (Benzog[<i>g</i>]pteridine-2,4(3H,10H)-dione, 3,10-dimethyl-)	2962-90-5 4074-59-3	**	8.63 (V) 8.47 (V)	PE	4992 4992
C₁₃H₁₁N₄O₂⁺	C ₆ H ₂ (CH ₃) ₂ C ₄ N ₄ (CH ₃) ₂ (=O) ₂ (Methyl-isooalloxazine)	XXXXXX-XX-X	**	8.72 (V)	PE	5577
C₁₃H₁₂N₄O₂⁺	C ₁₀ H ₃ N ₄ (=O) ₂ (CH ₃) ₃ (Benzog[<i>g</i>]pteridine-2,4(3H,10H)-dione, 3,6,10-trimethyl-) C ₁₀ H ₃ N ₄ (=O) ₂ (CH ₃) ₃ (Benzog[<i>g</i>]pteridine-2,4(3H,10H)-dione, 3,9,10-trimethyl-)	XXXXXX-XX-X	**	8.16 (V)	PE	4992
C₁₄H₁₄N₄O₂⁺	C ₁₀ H ₂ N ₄ (=O) ₂ (CH ₃) ₄ (Benzog[<i>g</i>]pteridine-2,4(3H,10H)-dione, 3,7,8,10-tetramethyl-)	18636-32-3	**	8.22 (V)	PE	4992
C₁₅H₁₈N₄O₂⁺	C ₁₀ H ₃ N ₄ (=O) ₂ (CH ₃) ₅ (Benzog[<i>g</i>]pteridine-2,4(1H,3H)-dione,5,10-dihydro-1,3,7,8,10-pentamethyl-)	14453-97-5	**	7.00 (V)	PE	4992
C₂H₃NO₃⁺	NH ₂ COCOOH	471-47-6	**	10.51 (V)	PE	4487
C₄H₃NO₃⁺	C ₄ H ₃ ONO ₂ (Furan, 2-nitro-)	609-39-2	**	9.75±0.05 (V)	PE	4626
			**	10.04±0.05	EI	3482
C₄H₇NO₃⁺	C ₂ H ₅ O(CO) ₂ NH ₂	XXXXXX-XX-X	**	9.85 (V)	PE	5549
C₅H₇NO₃⁺	CH ₃ CONHC(=CH ₂)COOH	XXXXXX-XX-X	**	9.24 (V)	PE	4983
C₅H₉NO₃⁺	CH ₃ COOC(CH ₃) ₂ NO	17746-46-2	**	8.28±0.1 (V)	PE	4465
C₆H₅NO₃⁺	C ₆ H ₄ (OH)(NO ₂) (Phenol, 2-nitro-) C ₆ H ₄ (OH)(NO ₂) (Phenol, 3-nitro-) C ₆ H ₄ (OH)(NO ₂) (Phenol, 4-nitro-) C ₆ H ₄ (NO ₂)OOCCH ₃ (Acetic acid, 3-nitrophenyl ester) C ₆ H ₄ (NO ₂)OOCCH ₃ (Acetic acid, 4-nitrophenyl ester)	88-75-5 554-84-7 100-02-7 1523-06-4 830-03-5	** ** ** ** CH ₂ =C=O	9.29 (V) 9.33 (V) 9.38 (V) 8.84±0.1 10.85±0.2 10.76±0.2	PE PE PE EI EI EI	4473 4473 4473 3447 3484 3484

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_6\text{H}_{11}\text{NO}_3^+$	$\text{C}_2\text{H}_5\text{O}(\text{CO})_2\text{N}(\text{CH}_3)_2$	XXXXX-XX-X **		9.31 (V)	PE	5549
$\text{C}_6\text{H}_{15}\text{NO}_3^+$	$\text{N}(\text{CH}_2\text{CH}_2\text{OH})_3$	102-71-6	**	~8.7 (V)	PE	4413
$\text{C}_7\text{H}_3\text{NO}_3^+$	$\text{C}_6\text{H}_3\text{NO}(=\text{O})_2$ (Furo[3,4,-b]pyridine-5,7-dione)	699-98-9	**	10.5 ± 0.1 (V)	PE	4889
$\text{C}_7\text{H}_4\text{NO}_3^+$	$\text{C}_6\text{H}_5\text{COC}_6\text{H}_4\text{NO}_2$ (Methanone, (2-nitrophenyl)phenyl-)	2243-79-0		12.0 ± 0.2	EI	4335
	$\text{C}_6\text{H}_5\text{COC}_6\text{H}_4\text{NO}_2$ (Methanone, (3-nitrophenyl)phenyl-)	2243-80-3		12.0 ± 0.2	EI	4358
	$\text{C}_6\text{H}_5\text{COC}_6\text{H}_4\text{NO}_2$ (Methanone, (4-nitrophenyl)phenyl-)	1144-74-7		12.3 ± 0.2	EI	4358
	$\text{C}_6\text{H}_4(\text{NO}_2)\text{COOH}$ (Benzoic acid, 3-nitro-)	121-92-6	OH	12.35 ± 0.2	EI	4335
	$\text{C}_6\text{H}_4(\text{NO}_2)\text{COOH}$ (Benzoic acid, 4-nitro-)	62-23-7	OH	11.58 ± 0.2	EI	3973
$\text{C}_7\text{H}_7\text{NO}_3^+$	$\text{C}_6\text{H}_4(\text{NO}_2)(\text{OCH}_3)$ (Benzene, 1-methoxy-2-nitro-)	91-23-6	**	9.04 (V)	PE	4473
	$\text{C}_6\text{H}_4(\text{NO}_2)(\text{OCH}_3)$ (Benzene, 1-methoxy-3-nitro-)	555-03-3	**	9.01 (V)	PE	4473
	$\text{C}_6\text{H}_4(\text{NO}_2)(\text{OCH}_3)$ (Benzene, 1-methoxy-4-nitro-)	100-17-4	**	9.09 ± 0.1	EI	3447
			**	8.6 ± 0.1	PE	4401
			**	8.79	PE	4621
			**	9.04 ± 0.1	EI	3447
			**	9.07 (V)	PE	4473
			**	9.08 ± 0.01 (V)	PE	4389
$\text{C}_8\text{H}_7\text{NO}_3^+$	$\text{C}_6\text{H}_4\text{NO}_2(\text{COCH}_3)$ (Ethanone, 1-(4-nitrophenyl)-)	100-19-6	**	10.15 ± 0.1 (V)	PE	4401
$\text{C}_9\text{H}_{11}\text{NO}_3^+$	$\text{C}_6\text{H}_4(\text{OH})\text{CH}_2\text{CH}(\text{NH}_2)\text{COOH}$ (DL-Tyrosine)	556-03-6	**	<8.4	EI	3766
$\text{C}_{11}\text{H}_{17}\text{NO}_3^+$	$\text{C}_6\text{H}_2(\text{OCH}_3)_3\text{CH}_2\text{CH}_2\text{NH}_2$ (Benzeneethanamine, 3,4,5-trimethoxy-)	54-04-6	**	8.18 ± 0.24 (V)	PE	4672
$\text{C}_{12}\text{H}_{19}\text{NO}_3^+$	$\text{C}_{12}\text{H}_{19}\text{NO}_3$ (Benzeneethanamine, 3,4,5-trimethoxy-N-methyl-)	4838-96-4		8.44 ± 0.40 (V)	PE	4672
	$\text{C}_{12}\text{H}_{19}\text{NO}_3$ (Benzeneethanamine, 2,3,4-trimethoxy- α -methyl-(\pm))	22199-12-8	**	8.09 ± 0.06 (V)	PE	4758
	$\text{C}_{12}\text{H}_{19}\text{NO}_3$ (Benzeneethanamine, 2,4,5-trimethoxy- α -methyl-(\pm))	22199-15-1	**	7.66 ± 0.06 (V)	PE	4758
	$\text{C}_{12}\text{H}_{19}\text{NO}_3$ (Benzeneethanamine, 2,4,6-trimethoxy- α -methyl-(\pm))	22199-16-2	**	7.76 ± 0.06 (V)	PE	4758
	$\text{C}_{12}\text{H}_{19}\text{NO}_3$ (Benzeneethanamine, 3,4,5-trimethoxy- α -methyl-(\pm))	22199-17-3	**	8.16 ± 0.06 (V)	PE	4758

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₃H₉NO₃⁺	C ₆ H ₅ COC ₆ H ₄ NO ₂ (Methanone, (2-nitrophenyl)phenyl-)	2243-79-0	**	9.6±0.1	EI	4358
			**	9.6±0.1	EI	4335
	C ₆ H ₅ COC ₆ H ₄ NO ₂ (Methanone, (3-nitrophenyl)phenyl-)	2243-80-3		9.8±0.1	EI	4335
			**	9.8±0.1	EI	4358
	C ₆ H ₅ COC ₆ H ₄ NO ₂ (Methanone, (4-nitrophenyl)phenyl-)	1144-74-7	**	9.95±0.1	EI	4335
			**	9.95±0.1	EI	4358
C₁₅H₂₈NO₃⁺	C ₂₀ H ₃₆ N ₂ O ₆ (L-Alanine,N-[N-(1-oxodecyl)-L- α -glutamyl]-dimethyl ester)	55728-16-0		9.6±0.1	PI	5279
C₁₅H₂₉NO₃⁺	C ₂₀ H ₃₆ N ₂ O ₆ (L-Alanine,N-[N-(1-oxodecyl)-L- α -glutamyl]-dimethyl ester)	55728-16-0		9.4±0.1	PI	5279
C₁₇H₁₉NO₃⁺	C ₁₆ H ₁₄ NO(OH ₂)CH ₃ (Morphinan-3,6-diol,7,8-didehydro-4,5-epoxy-17-methyl-(5 α ,6 α -)-)	57-27-2	**	8.3 (V)	PE	4646
C₁₉H₂₁NO₃⁺	C ₁₆ H ₁₄ NO(OH ₂)CH ₂ CH=CH ₂ (Morphinan-3,6-diol,7,8-didehydro-4,5-epoxy-17-(2-propenyl)-(5 α ,6 α -)-)	62-67-9	**	8.15 (V)	PE	4646
C₂₀H₁₃NO₃⁺	C ₆ H ₅ OC ₆ H ₃ C ₃ O ₂ NC ₅ H ₅	XXXXX-XX-X	**	7.55	CTS	5592
C₃H₂N₂O₃⁺	C ₃ H ₂ N ₂ O ₃ (Imidazolidinetrione)	120-89-8	**	10.67	PE	4471
C₃H₆N₂O₃⁺	C(CH ₃) ₂ (NO ₂)NO	5275-46-7	**	9.92±0.1 (V)	PE	4465
C₄H₄N₂O₃⁺	C ₃ HN ₂ O ₃ (CH ₃) (Imidazolidinetrione, methyl-)	3659-97-0	**	10.52	PE	4471
	C ₄ H ₄ N ₂ (=O) ₃ (2,4,6(1H,3H,5H)-Pyrimidinetrione)	67-52-7	**	10.20	PE	5093
C₅H₄N₂O₃⁺	C ₃ H ₄ N(O)NO ₂ (Pyridine, 4-nitro-, 1-oxide)	1124-33-0	**	9.03±0.02 (V)	PE	4275
C₅H₆N₂O₃⁺	C ₃ N ₂ O ₃ (CH ₃) ₂ (Imidazolidinetrione, dimethyl-)	5176-82-9	**	10.19	PE	4471
C₆H₁₀N₂O₃⁺	C ₆ H ₁₀ (NO)(NO ₂) (Cyclohexane, 1-nitro-1-nitroso-)	14296-14-1	**	9.55 (V)	PE	4465
C₇H₄N₂O₃⁺	C ₆ H ₄ (NO ₂)(C≡NO) (Benzonitrile, 4-nitro-N-oxide)	2574-03-0	**	~9.5 (V)	PE	4719
C₇H₅N₂O₃⁺	C ₆ H ₃ (Cl)(NO ₂)NHCHO (Formamide, N-(2-chloro-4-nitrophenyl)-)	16135-32-3	Cl	10.2±0.1	EI	4359

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_7\text{H}_5\text{N}_2\text{O}_3^+$	$\text{C}_6\text{H}_3(\text{Cl})(\text{NO}_2)\text{NHCHO}$ (Formamide, <i>N</i> -(2-chloro-5-nitrophenyl)-)	53666-48-1	Cl	9.9 ± 0.1	EI	4359
$\text{C}_7\text{H}_6\text{N}_2\text{O}_3^+$	$\text{C}_6\text{H}_4(\text{NO}_2)(\text{CONH}_2)$ (Benzamide, 3-nitro-)	645-09-0	**	10.28 (V)	PE	4918
	$\text{C}_6\text{H}_4(\text{NO}_2)(\text{CONH}_2)$ (Benzamide, 4-nitro-)	619-80-7	**	10.33 (V)	PE	4918
$\text{C}_8\text{H}_8\text{N}_2\text{O}_3^+$	$\text{C}_6\text{H}_4(\text{NO}_2)\text{NHCOCH}_3$ (Acetamide, <i>N</i> -(2-nitrophenyl)-)	552-32-9	**	8.85	EI	4834
$\text{C}_9\text{H}_7\text{N}_2\text{O}_3^+$	$\text{C}_6\text{H}_4(\text{NO}_2)\text{NHCOCH}=\text{CHCH}_3$ (2-Butenamide, <i>N</i> -(4-nitrophenyl)-)	51944-68-4	CH_3	13.6 ± 0.3	EI	3996
$\text{C}_{10}\text{H}_{14}\text{N}_2\text{O}_3^+$	$\text{C}_3\text{N}_2\text{O}_3(\text{C}_6\text{H}_7)_2$ (Imidazolidinetrione, dipropyl-)	21036-96-4	**	9.90	PE	4471
$\text{C}_{10}\text{H}_{10}\text{N}_2\text{O}_3^+$	$\text{C}_6\text{H}_4(\text{NO}_2)\text{NHCOCH}=\text{CHCH}_3$ (2-Butenamide, <i>N</i> -(4-nitrophenyl)-)	51944-68-4	**	9.1 ± 0.1	EI	3996
$\text{C}_{12}\text{H}_8\text{N}_2\text{O}_3^+$	$\text{C}_6\text{H}_4(\text{NO}_2)\text{COC}_5\text{H}_4\text{N}$ (Methanone, (2-nitrophenyl)-2-pyridinyl-)	27693-37-4	**	9.71	EI	5459
$\text{C}_{14}\text{H}_{14}\text{N}_2\text{O}_3^+$	$\text{C}_{14}\text{H}_{14}\text{N}_2\text{O}_3$	XXXXX-XX-X	**	8.00 (V)	PE	5590
$\text{C}_{15}\text{H}_{10}\text{N}_2\text{O}_3^+$	$\text{C}_6\text{H}_5\text{N}(\text{O})(\text{C}_6\text{H}_4\text{NO}_2)$ (Isoquinolinium, 4-hydroxy-2-(4-nitrophenyl)-hydroxide, inner salt)	56359-31-0	**	7.29 ± 0.05	EI	4863
	$\text{C}_6\text{H}_5\text{N}(\text{O})=\text{O}(\text{C}_6\text{H}_4\text{NO}_2)$ (Indeno[1,2- <i>b</i>]azirin-6(1H)-one, 1a,6a-dihydro-1-(4-nitrophenyl)-)	55507-33-0	**	8.71 ± 0.05	EI	4863
	$\text{C}_6\text{H}_5\text{N}_3(\text{O})=\text{O}(\text{C}_6\text{H}_4\text{NO}_2)$ (Indeno[1,2- <i>d</i>]triazol-8(3H)-one, 3a,8a-dihydro-3-(4-nitrophenyl)-)	55507-29-4	N_2	8.8 ± 0.1	EI	4863
	$\text{C}_{15}\text{H}_{24}\text{N}_2\text{O}_3^+$	28834-17-5	**	9.26 (V)	PE	4465
$\text{C}_{16}\text{H}_{29}\text{N}_2\text{O}_3^+$	$\text{C}_{15}\text{H}_{24}(\text{NO})(\text{NO}_2)$ (Bicyclo[7.2.0]undecane, 6,10,10-trimethyl-2-methylene-5-nitro-6-nitroso-					
	$\text{C}_{20}\text{H}_{37}\text{N}_3\text{O}_5$ (L-Alanine, N-[N-[N-(1-oxodecyl)-L-alanyl]-L-alanyl]-methyl ester)	55728-11-5		9.8 ± 0.1	PI	5279
$\text{C}_{18}\text{H}_{25}\text{N}_2\text{O}_3^+$	$\text{C}_{27}\text{H}_{40}\text{N}_4\text{O}_8\text{S}$ (L-Cysteine, S-(2-methoxy-2-oxoethyl)-N-[N-[N-(1-oxopropyl)-L-phenylalanyl]-L-leucyl]glycyl]-methyl ester)	35146-63-5		9.1 ± 0.1	PI	5279
$\text{C}_{19}\text{H}_{35}\text{N}_2\text{O}_3^+$	$\text{C}_{23}\text{H}_{43}\text{N}_3\text{O}_5$ (L-Alanine, N-[N-[N-(1-oxodecyl)-L-alanyl]-L-leucyl]-methyl ester)	55728-12-6		8.8 ± 0.1	PI	5279
$\text{C}_{20}\text{H}_{24}\text{N}_2\text{O}_3^+$	$\text{C}_{20}\text{H}_{24}\text{N}_2\text{O}_3$ (Phenol, 2,2'-[oxybis(3,1-propanediyl)nitrilomethylidyne)]bis-)	52279-43-3	**	8.40 ± 0.10	EI	4213

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_8\text{H}_5\text{N}_3\text{O}_3^+$	$\text{O}_2\text{NC}_6\text{H}_4\text{C}(=\text{O})\text{CHN}_2$ (Ethanone, 2-diazo-(4-nitrophenyl)-)	4203-31-0	**	9.41 ± 0.05 (V)	PE	5326
$\text{C}_{12}\text{H}_7\text{N}_3\text{O}_3^+$	$\text{C}_{12}\text{H}_7\text{N}_2\text{ONO}_2$ (Phenazine, 2-nitro-10-oxide)	2876-33-7	**	8.46 (V)	PE	4590
$\text{C}_{18}\text{H}_{34}\text{N}_3\text{O}_3^+$	$\text{C}_{20}\text{H}_{37}\text{N}_3\text{O}_5$ (L-Alanine, N-[N-[N-(1-oxodecyl)-L-alanyl]-L-alanyl]-methyl ester)	55728-11-5		8.8 ± 0.1	PI	5279
$\text{C}_{19}\text{H}_{28}\text{N}_3\text{O}_3^+$	$\text{C}_{27}\text{H}_{40}\text{N}_4\text{O}_8\text{S}$ (L-Cysteine, S-(2-methoxy-2-oxoethyl)-N-[N-[N-(1-oxopropyl)-L-phenylalanyl]-L-leucyl]-glycyl]-methyl ester)	35146-63-5		9.0 ± 0.1	PI	5279
$\text{C}_{23}\text{H}_{32}\text{N}_3\text{O}_3^+$	$\text{C}_{30}\text{H}_{45}\text{N}_5\text{O}_6$ (L-Alanine, N-[N-[N-(1-oxodecyl)glycyl]-L-tryptophyl]-L-alanyl]-methyl ester)	19716-78-0		9.5 ± 0.1	PI	5279
$\text{C}_5\text{H}_4\text{N}_4\text{O}_3^+$	$\text{C}_5\text{H}_4\text{N}_4(=\text{O})_3$ (1H-Purine-2,6,8(3H)-trione, 7,9-dihydro-)	69-93-2	**	8.15	PE	5093
$\text{C}_{25}\text{H}_{37}\text{N}_4\text{O}_3^+$	$\text{C}_{30}\text{H}_{45}\text{N}_5\text{O}_6$ (L-Alanine, N-[N-[N-(1-oxodecyl)glycyl]-L-tryptophyl]-L-alanyl]-methyl ester)	19716-78-0		9.5 ± 0.1	PI	5279
$\text{C}_7\text{H}_5\text{NO}_4^+$	$\text{C}_6\text{H}_4(\text{NO}_2)\text{COOH}$ (Benzoic acid, 3-nitro-)	121-92-6	**	10.31 ± 0.2	EI	3973
	$\text{C}_6\text{H}_4(\text{NO}_2)\text{COOH}$ (Benzoic acid, 4-nitro-)	62-23-7	**	10.18 ± 0.2	EI	3973
$\text{C}_8\text{H}_7\text{NO}_4^+$	$\text{C}_6\text{H}_4(\text{NO}_2)\text{OOCCH}_3$ (Acetic acid, 3-nitrophenyl ester)	1523-06-4	**	9.43 ± 0.2	EI	3484
	$\text{C}_6\text{H}_4(\text{NO}_2)\text{OOCCH}_3$ (Acetic acid, 4-nitrophenyl ester)	830-03-5	**	9.48 ± 0.2	EI	3484
$\text{C}_{10}\text{H}_{11}\text{NO}_4^+$	$\text{C}_6\text{H}_2(\text{OCH}_3)_3(\text{C} \equiv \text{NO})$ (Benzonitrile, 2,4,6-trimethoxy-N-oxide)	2904-59-8	**	7.95 (V)	PE	4719
	$\text{C}_3(\text{OCH}_3)_2\text{O}_2\text{NC}_5\text{H}_5$ (Pyridinium, 2-methoxy-1-(methoxycarbonyl)-2-oxoethylide)	1291-37-8	**	7.83	CTS	5591
$\text{C}_{13}\text{H}_9\text{NO}_4^+$	$\text{C}_6\text{H}_5\text{COOC}_6\text{H}_4\text{NO}_2$ (Benzoic acid, 4-nitrophenyl ester)	959-22-8	**	9.3	EI	5631
$\text{C}_{14}\text{H}_{13}\text{NO}_4^+$	$\text{C}_3(\text{OCH}_3)_2\text{O}_2\text{NC}_9\text{H}_7$ (Isoquinolium, 2-methoxy-1-(methoxycarbonyl)-2-oxoethylide)	17870-65-4	**	7.67	CTS	5591
	$\text{C}_3(\text{OCH}_3)_2\text{O}_2\text{NC}_9\text{H}_7$ (Quinolinium, 1-(2-methoxy-1-(methoxycarbonyl)-2-oxoethyl)-hydroxide, inner salt)	17870-64-3	**	7.67	CTS	5591
$\text{C}_{16}\text{H}_{28}\text{NO}_4^+$	$\text{C}_{20}\text{H}_{36}\text{N}_2\text{O}_6$ (L-Alanine, N-[N-(1-oxodecyl)-L- α -glutamyl]-dimethyl ester)	55728-16-0		9.7 ± 0.1	PI	5279

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₇H₉NO₄⁺	C ₁₇ H ₉ NO ₄ (Naphthalo[2,3- <i>f</i>]quinoline-7,12-dione, 5,6-dihydroxy-)	568-02-5	**	7.35	PI	3586
C₆H₄N₂O₄⁺	C ₆ H ₄ (NO ₂) ₂ (Benzene, 1,2-dinitro-)	528-29-0	**	10.71 (V)	PE	4892
	C ₆ H ₄ (NO ₂) ₂ (Benzene, 1,3-dinitro-)	99-65-0	**	10.43±0.02	PI	5505
			**	10.40 (V)	PE	4892
			**	10.62±0.1	EI	3447
	C ₆ H ₄ (NO ₂) ₂ (Benzene, 1,4-dinitro-)	100-25-4	**	10.50±0.02	PI	5552
			**	10.3±0.1	PE	4401
			**	10.65 (V)	PE	4892
			**	10.63±0.1	EI	3447
C₁₀H₁₈N₂O₄⁺	trans-(CH ₃ C=OOC(CH ₃) ₂) ₂ N=N	55204-45-0	**	8.74 (V)	PE	4429
C₁₁H₂₀N₂O₄⁺	C ₁₁ H ₂₀ N ₂ O ₄ (L-Alanine,N-(N-acetyl-L-valyl)-methyl ester)	55728-13-7	**	8.6±0.1	PI	5279
C₁₃H₁₆N₂O₄⁺	(C ₆ H ₄ NO ₂) ₂ CH ₂ (Benzene, 1,1'-methylenebis[4-nitro-])	1817-74-9	**	9.98±0.05	EI	3806
C₁₃H₁₄N₂O₄⁺	C ₅ H ₅ NH(=O) ₂ NC ₈ H ₈ (=O) ₂ (1H-Isoindole-1,3(2H)-dione,2-(2,6-dioxo-3-piperidinyl)-4,5,6,7-tetrahydro-)	60242-08-2	**	9.50 (V)	PE	5614
C₁₄H₁₂N₂O₄⁺	C ₆ H ₄ (NO ₂)CH ₂ CH ₂ C ₆ H ₄ NO ₂ (Benzene, 1,1'-(1,2-ethanediyl)bis[4-nitro-])	736-30-1	**	9.77±0.05	EI	3806
C₁₅H₁₄N₂O₄⁺	C ₃ (OCH ₃) ₂ O ₂ (NC ₅ H ₄) ₂ (4,4'-Bipyridinium,2-methoxy-1-(methoxycarbonyl)-2-oxoethylidide)	59805-15-1	**	7.50	CTS	5591
C₁₆H₂₉N₂O₄⁺	C ₂₀ H ₃₄ N ₂ O ₄ (L-Histidine,N-[N-(1-oxodecyl)-L-alanyl]-methyl ester)	31944-64-6		9.6±0.1	PI	5279
C₁₈H₃₀N₂O₄⁺	C ₄ (N(C ₂ H ₅) ₂) ₂ (COOC ₂ H ₅) ₂ (1,3-Cyclobutadiene-1,3-dicarboxylic acid, 2,4-bis(diethylamino)-, diethyl ester)	20913-35-3	**	7.55 (V)	PE	3885
C₁₉H₃₆N₂O₄⁺	C ₁₉ H ₃₆ N ₂ O ₄ (Glycine,N-[N-(1-oxodecyl)-L-leucyl]-methyl ester)	55728-14-8	**	8.9±0.1	PI	5279
C₆H₅N₃O₄⁺	C ₆ H ₅ NH ₂ (NO ₂) ₂ (Benzeneamine,2,5-dinitro-)	619-18-1	**	8.89±0.01	PI	5552
C₁₆H₁₁N₃O₄⁺	C ₃ H ₃ (CN)((C ₆ H ₄)NO ₂) ₂ (Cyclopropanecarbonitrile, 1,2-bis(<i>p</i> -nitrophenyl)-)	28752-28-5	**	9.30±0.05	EI	3575

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₂₀H₂₈N₃O₄⁺	C ₂₇ H ₄₀ N ₄ O ₆ S (L-Cysteine,S-(2-methoxy-2-oxoethyl)-N-[N-[N-[N-(1-oxopropyl)-L-phenylalanyl]-L-leucyl]glycyl]-methyl ester)	35146-63-5		9.2±0.1	PI	5279
C₂₅H₃₇N₃O₄⁺	C ₂₅ H ₃₇ N ₃ O ₄ (L-Tryptophan,N-[N-(1-oxodecyl)-L-alanyl]-methyl ester)	55728-17-1	**	7.5±0.1	PI	5279
C₂₀H₃₁N₄O₄⁺	C ₂₀ H ₃₁ N ₄ O ₄ (L-Histidine,N-[N-(1-oxodecyl)-L-alanyl]-methyl ester)	31944-64-6	**	8.7±0.1	PI	5279
C₂₆H₃₇N₄O₄⁺	C ₃₀ H ₄₃ N ₅ O ₆ (L-Alanine,N-[N-[N-(1-oxodecyl)glycyl]-L-tryptophyl]-L-alanyl]-methyl ester)	19716-78-0		9.5±0.1	PI	5279
C₆H₄N₂O₅⁺	C ₆ H ₃ (NO ₂) ₂ OH (Phenol, 2,4-dinitro-)	51-28-5	**	9.57	PE	5093
C₇H₆N₂O₅⁺	C ₆ H ₃ (NO ₂) ₂ OCH ₃ (Benzene, 1-methoxy-2,4-dinitro-)	119-27-7	**	9.30	PE	5093
C₁₇H₃₂N₂O₅⁺	C ₁₇ H ₃₂ N ₂ O ₅ (L-Serine,N-[N-(1-oxodecyl)-L-alanyl]-methyl ester)	55728-15-9		9.1±0.1	PI	5279
C₂₀H₃₇N₃O₅⁺	C ₂₀ H ₃₇ N ₃ O ₅ (L-Alanine,N-[N-[N-(1-oxodecyl)-L-alanyl]-L-alanyl]-methyl ester)	55728-11-5	**	8.6±0.1	PI	5279
C₂₃H₄₃N₃O₅⁺	C ₂₃ H ₄₃ N ₃ O ₅ (L-Alanine,N-[N-[N-(1-oxodecyl)-L-alanyl]-L-leucyl]-methyl ester)	55728-12-6	**	8.4±0.1	PI	5279
C₂₀H₃₆N₂O₆⁺	C ₂₀ H ₃₆ N ₂ O ₆ (L-Alanine,N-[N-(1-oxodecyl)-L-α-glutamyl]-dimethyl ester)	55728-16-0	**	9.1±0.1	PI	5279
C₆H₃N₃O₆⁺	C ₆ H ₃ (NO ₂) ₃ (Benzene,1,3,5-trinitro-)	99-35-4	**	10.96±0.02	PI	5505
C₇H₅N₃O₆⁺	C ₆ H ₂ (NO ₂) ₃ CH ₃ (Benzene,2-methyl-1,3,5-trinitro-)	118-96-7	**	10.59±0.04	PI	5552
C₃₀H₄₅N₅O₆⁺	C ₃₀ H ₄₅ N ₅ O ₆ (L-Alanine,N-[N-[N-(1-oxodecyl)glycyl]-L-tryptophyl]-L-alanyl]-methyl ester)	19716-78-0	**	7.7±0.1	PI	5279
BC₆H₁₀NO⁺	C ₅ H ₄ N(OCH ₃)·BH ₃ (Pyridine, 4-methoxy-, compound with borane(1:1))	56898-50-1	**	9.30 (V)	PE	4536
B₂C₄H₁₂N₂O⁺						
	N ₂ B ₂ O(CH ₃) ₄ (1,2,4,3,5-Oxadiazadiborolidine, 2,3,4,5-tetramethyl-)	57877-89-1	**	8.39 (V)	PE	4526
	N ₂ B ₂ O(CH ₃) ₄ (1,3,4,2,5-Oxadiazadiborolidine, 2,3,4,5-tetramethyl-)	40392-38-9	**	7.88 (V)	PE	4526

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
CF⁺						
	CFCl ₃	75-69-4	Cl ₂ +Cl 3Cl	15.7 18.35	PI PI	5196 5196
C₃F⁺						
	C ₆ F ₆ (Bicyclo[2.2.0]hexa-2,5-diene, 1,2,3,4,5,6-hexafluoro-)	6733-01-3		22.4±0.5	EI	4961
C₅F⁺						
	CF ₃ C≡CC≡CCF ₃	10524-09-1		23.7±0.5	EI	4961
CF₂⁺						
(² A ₁)	CF ₂	2154-59-8	**	11.4±0.3	EI	4544
			**	11.54±0.1	EI	4554
			**	11.42±0.01	PE	4239
(² B ₂)			**	16.40 (V)	PE	4239
(² A ₂)			**	17.4(V)	PE	4239
(² A ₁)			**	19.2 (V)	PE	4239
(² B ₁)			**	20.83 (V)	PE	4239
(² B ₂)			**	22.2 (V)	PE	4239
(² A ₁)			**	24.0 (V)	PE	4239
			**	11.54±0.10	EI	3818
			**	9.74	OTH	3930
C ₂ F ₄		116-14-3	CF ₂	15.2±0.1	EI	3539
CF ₃ NO		XXXXX-XX-X	FNO	16.7±0.2	EI	5220
CF ₃ Cl		75-72-9	F ⁻ +Cl	15.90±0.3	PI	5399
			F+Cl ⁻	16.00±0.1	PI	5399
			F+Cl	18.84	PI	4757
			F+Cl	18.85±0.05	PI	5399
			F+Cl	18.85	PI	5196
CF ₂ Cl ₂		75-71-8	Cl ₂	14.90±0.3	PI	5399
			Cl ⁻ +Cl	16.98	PI	4757
			2Cl	17.22	PI	5196
C₂F₂⁺						
C ₂ F ₂		689-99-6	**	11.18	PE	4681
			**	11.18	PE	5313
C₃F₂⁺						
	CF ₃ C≡CC≡CCF ₃	10524-09-1	F+C ₃ F ₃	14.9±0.4	EI	4961
	C ₆ F ₆ (Benzene, hexafluoro-)	392-56-3		15.8±0.1	EI	4961
	C ₆ F ₆ (Bicyclo[2.2.0]hexa-2,5-diene, 1,2,3,4,5,6-hexafluoro-)	6733-01-3		13.5±0.5	EI	4961
C₄F₂⁺						
(CF≡C) ₂		64788-23-4	**	10.05	PE	5313
			**	10.35 (V)	PE	4681
CF ₃ C≡CC≡CCF ₃		10524-09-1		18.9±0.5	EI	4961
C ₆ F ₆ (Benzene, hexafluoro-)		392-56-3		19.8±0.5	EI	4961
C ₆ F ₆ (Bicyclo[2.2.0]hexa-2,5-diene, 1,2,3,4,5,6-hexafluoro-)		6733-01-3		16.0±0.5	EI	4961
C₅F₂⁺						
	CF ₃ C≡CC≡CCF ₃	10524-09-1		21.2±0.5	EI	4961
	C ₆ F ₆ (Benzene, hexafluoro-)	392-56-3		20.7±0.1	EI	4961
			F ₂ +CF ₂	24.8±0.4	EI	4961
			F+CF ₃	24.8±0.4	EI	4961
C ₆ F ₆ (Bicyclo[2.2.0]hexa-2,5-diene, 1,2,3,4,5,6-hexafluoro-)		6733-01-3		17.8±0.5	EI	4961

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₅F₂⁺						
	C ₆ F ₆	6733-01-3	F ₂ +CF ₂ F+CF ₃	25.±0.4 25.±0.4	EI EI	4961 4961
CF₃⁺						
	CF ₃	2264-21-3	**	9.5	OTH	5554
	CF ₄	75-73-0	F	14.7±0.3	PI	5175
	CF ₃ C≡CC≡CCF ₃	10524-09-1		17.6±0.5	EI	4961
	C ₆ F ₆ (Benzene, hexafluoro-)	392-56-3		15.3±0.5	EI	4961
			CF+C ₄ F ₂	21.3±0.4	EI	4961
			CF+C ₃ +CF ₂	25.2±0.4	EI	4961
	C ₆ F ₆ (Bicyclo[2.2.0]hexa-2,5-diene, 1,2,3,4,5,6-hexafluoro-)	6733-01-3		16.5±0.5	EI	4961
			CF+C ₄ F ₂	19.4±0.4	EI	4961
			CF+C ₃ +CF ₂	22.7±0.4	EI	4961
	CH ₃ CF ₃	420-46-2	CH ₃	13.94±0.1	EI	3478
	(CF ₂ =CH) ₂	407-70-5	C ₃ H ₂ F	13.9±0.1	EI	5554
	(CF ₃) ₂ CO	684-16-2		13.8	EI	3550
	CH ₃ COCF ₃	421-50-1		14.6	EI	3550
	CF ₃ NO	XXXXX-XX-X	NO	12.6±0.1	EI	5220
	CF ₃ Cl	75-72-9	Cl	12.55	PI	5196
			Cl	12.65	PI	4757
	C ₂ F ₅ I	354-64-3	CF ₂ ,I	13.73±0.1	EI	4862
C₂F₃⁺						
	C ₂ F ₃ Cl	79-38-9	Cl	15.4±0.1	EI	4070
C₃F₃⁺						
	CF ₃ C≡CC≡CCF ₃	10524-09-1		15.0±0.2	EI	4961
	C ₆ F ₆ (Benzene, hexafluoro-)	392-56-3	C ₃ F ₃	16.5±0.4	EI	4961
			CF+C ₂ F ₂	17.1±0.2	EI	4961
				21±0.4	EI	4961
	C ₆ F ₆ (Bicyclo[2.2.0]hexa-2,5-diene, 1,2,3,4,5,6-hexafluoro-)	6733-01-3		15.0±0.2	EI	4961
			C ₃ F ₃	15.6±0.4	EI	4961
			CF+C ₂ F ₂	19.6±0.4	EI	4961
C₄F₃⁺						
	C ₆ F ₆ (Bicyclo[2.2.0]hexa-2,5-diene, 1,2,3,4,5,6-hexafluoro-)	6733-01-3		16.8±0.2	EI	4961
C₅F₃⁺						
	CF ₃ C≡CC≡CCF ₃	10524-09-1		14.8±0.2	EI	4961
			CF ₃	15.±0.4	EI	4961
			F+CF ₂	18.4±0.4	EI	4961
			F+CF ₂	18.5±0.4	EI	4961
			CF+F ₂	23.6±0.4	EI	4961
	C ₆ F ₆ (Benzene, hexafluoro-)	392-56-3		15.8±0.1	EI	4961
			CF ₃	16.1±0.4	EI	4961
			F+CF ₂	18.8±0.4	EI	4961
			F+CF ₂	18.8±0.4	EI	4961
	C ₆ F ₆ (Bicyclo[2.2.0]hexa-2,5-diene, 1,2,3,4,5,6-hexafluoro-)	6733-01-3		13.8±0.1	EI	4961
			CF ₃	15.±0.4	EI	4961
			F+CF ₂	17.5±0.4	EI	4961
			F+CF ₂	17.7±0.4	EI	4961
CF₄⁺						
	CF ₄	75-73-0	**	16.25±0.04 (V)	PE	3880

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C_2F_4^+	C_2F_4	116-14-3	** ** ** ** **	10.10 10.14 10.32 10.52 (V) 10.56 ± 0.02 (V)	PE PE PE PE PE	3649 5408 3589 4084 5017
	$\text{C}_4\text{H}_4\text{F}_4$ (Cyclobutane, 1,1,2,2-tetrafluoro-)	374-12-9	C_2H_4	12.60	EI	4553
C_3F_4^+	1,2- C_3F_4	461-68-7	**	11.24 (V)	PE	5105
C_4F_4^+	$\text{CF}_2=\text{C}=\text{C}=\text{CF}_2$	2252-95-1	**	9.30 (V)	PE	4738
C_5F_4^+	$\text{CF}_3\text{C}\equiv\text{CC}\equiv\text{CF}$ $\text{CF}_3\text{C}\equiv\text{CC}\equiv\text{CCF}_3$	64788-24-5 10524-09-1	** CF_2	10.85 (V) 14.2 ± 0.2 15.5 ± 0.4 16.3 ± 0.4	PE EI EI EI	4681 4961 4961 4961
	C_6F_6 (Benzene, hexafluoro-)	392-56-3	CF_2	16.4 ± 0.2 13.8 ± 0.2 14.8 ± 0.4	EI EI EI	4961 4961 4961
C_6F_4^+	$\text{CF}_3\text{C}\equiv\text{CC}\equiv\text{CCF}_3$	10524-09-1		20.2 ± 0.2	EI	4961
C_2F_5^+	$\text{C}_2\text{F}_5\text{I}$	354-64-3	I	11.71 ± 0.1	EI	4862
C_5F_5^+	C_6F_6 (Benzene, hexafluoro-)	392-56-3		16.4 ± 0.2	EI	4961
	C_6F_6 (Bicyclo[2.2.0]hexa-2,5-diene, 1,2,3,4,5,6-hexafluoro-)	6733-01-3		18.5 ± 0.2	EI	4961
C_6F_5^+	$\text{CF}_3\text{C}\equiv\text{CC}\equiv\text{CCF}_3$	10524-09-1	F	14.8 ± 0.2 15.1 ± 0.4	EI EI	4961 4961
	C_6F_6 (Benzene, hexafluoro-)	392-56-3	F	16.86 ± 0.05	EI	4127
	C_6F_6 (Bicyclo[2.2.0]hexa-2,5-diene, 1,2,3,4,5,6-hexafluoro-)	6733-01-3		17.2 ± 0.2 14.6 ± 0.2	EI EI	4961 4961
	$\text{C}_6\text{F}_5\text{Cl}$ (Benzene, chloropentafluoro-)	344-07-0	Cl	15.85 ± 0.05	EI	4127
	$\text{C}_6\text{F}_5\text{Br}$ (Benzene, bromopentafluoro-)	344-04-7	Br	14.93 ± 0.05	EI	4127
	$\text{C}_6\text{F}_5\text{I}$ (Benzene, pentafluoroiodo-)	827-15-6	I	13.21 ± 0.05	EI	4127
C_2F_6^+	C_2F_6	76-16-4	**	14.6 (V)	PE	4321
C_3F_6^+	$\text{CF}_3\text{CF}=\text{CF}_2$	116-15-4	** **	10.62 10.62	PE PE	3589 4165
C_4F_6^+	$\text{CF}_3\text{C}\equiv\text{CCF}_3$	692-50-2	**	12.31	PE	3589

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₄F₆⁺	CF ₃ C≡CCF ₃	692-50-2	**	12.35±0.01	PE	4633
C₆F₆⁺	CF ₃ C≡CC≡CCF ₃	10524-09-1	**	10.99±0.01	PE	4633
			**	11.5±0.1	EI	4961
	C ₆ F ₆ (Benzene, hexafluoro-)	392-56-3	**	9.90±0.01	S	3559
			**	9.90±0.05	PE	4821
			**	9.90 (V)	PE	3873
			**	9.91	PE	5408
			**	9.93	PE	3637
			**	10.09 (V)	PE	4884
			**	10.14 (V)	PE	5252
			**	10.2±0.1	EI	4961
			**	10.09 (V)	PE	4472
	C ₆ F ₆ (Bicyclo[2.2.0]hexa-2,5-diene,1,2,3,4,5,6-hexafluoro-)	6733-01-3	**	10.08±0.05	PE	4414
			**	10.4 (V)	PE	4453
			**	10.4±0.1	EI	4961
C₄F₈⁺	cis-2-C ₄ F ₈	1516-65-0	**	11.46 (V)	PE	4084
	trans-2-C ₄ F ₈	1516-64-9	**	11.55 (V)	PE	3649
			**	11.55 (V)	PE	4084
C₇F₈⁺	C ₆ F ₅ CF ₃ (Benzene,pentafluoro(trifluoromethyl)-)	434-64-0	**	9.9	PE	5521
			**	9.9 (V)	PE	5461
C₁₀F₈⁺	C ₁₀ F ₈ (Naphthalene, octafluoro-)	313-72-4	**	8.85	PE	3637
			**	8.90±0.05	PE	4821
C₁₂F₈⁺	C ₁₂ F ₈ (Acenaphthylene, octafluoro-)	1554-93-4	**	9.1±0.1 (V)	PE	4821
C₈F₁₀⁺	C ₆ F ₄ (CF ₃) ₂ (Benzene,1,2,4,5-tetrafluoro-3,6-bis(trifluoromethyl)-)	651-89-8	**	9.9	PE	5521
			**	9.9 (V)	PE	5461
C₁₂F₁₀⁺	(C ₆ F ₅) ₂ (1,1'-Biphenyl, decafluoro-)	434-90-2	**	9.40±0.02	PE	3702
C₁₄F₁₀⁺	C ₁₄ F ₁₀ (Anthracene, decafluoro-)	1580-19-4	**	8.28±0.05	PE	4821
	C ₁₄ F ₁₀ (Phenanthrene, decafluoro-)	1580-20-7	**	8.75±0.05	PE	4821
C₁₆F₁₀⁺	C ₁₆ F ₁₀ (Pyrene, decafluoro-)	1493-68-1	**	8.36±0.05	PE	4821
C₆F₁₂⁺	(CF ₃) ₂ C=C(CF ₃) ₂	360-57-6	**	12.61 (V)	PE	4084

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
CHF⁺						
	<i>cis</i> -CHF=CHF	1630-77-9	CHF	18.1±0.2	PI	5241
	<i>trans</i> -CHF=CHF	1630-78-0	CHF	18.1±0.2	PI	5241
CH₂F⁺						
	CH ₂ F	3744-29-4	**	8.90	EI	3732
			**	9.16±0.02	OTH	3930
	CH ₂ F ₂	75-10-5	F	14.06	EI	3732
	CH ₂ =CF ₂	75-38-7	CF	14.84±0.02	PI	3930
	<i>cis</i> -CHF=CHF	1630-77-9	CF	14.3±0.1	PI	5241
	<i>trans</i> -CHF=CHF	1630-78-0	CF	14.3±0.1	PI	5241
C₂HF⁺						
	C ₂ HF	2713-09-9	**	11.26	PE	5313
	C ₂ H ₃ F	75-02-5	H ₂	13.72±0.02	PI	3930
				13.72	PI	5352
			H ₂	13.70±0.1	PE	4993
	CH ₂ =CF ₂	75-38-7	HF	14.18±0.03	PI	3930
	<i>trans</i> -CHF=CHF	1630-78-0	HF	13.7±0.1	PI	5241
C₂H₂F⁺						
	C ₂ H ₃ F	75-02-5	H	13.56±0.04	PI	3930
			H	13.56	PI	5352
			H	13.55	PE	4993
	CH ₂ =CF ₂	75-38-7	F	14.37±0.02	PI	3930
	<i>cis</i> -CHF=CHF	1630-77-9	F	13.9±0.1	PI	5241
	<i>trans</i> -CHF=CHF	1630-78-0	F	13.9±0.1	PI	5241
	CH ₂ =FCl	2317-91-1	Cl	13.7±0.1	EI	4070
C₂H₃F⁺						
	C ₂ H ₃ F	75-02-5	**	10.35±0.01	PI	3930
			**	10.363±0.015	PI	5616
			**	10.3	PE	4993
			**	10.36	PE	5408
			**	10.37	PE	5352
			**	10.56±0.02 (V)	PE	5017
	(CH ₃) ₂ CHF	420-26-8	CH ₄	11.53±0.03	PI	5003
C₂H₄F⁺						
	CH ₃ CHF	29526-61-2	**	7.93	PI	5003
	C ₂ H ₃ F	353-36-6	H	12.04±0.03	PI	5003
	(CH ₃) ₂ CHF	420-26-8	CH ₃	11.75±0.03	PI	5003
	CH ₃ CHF ₂	75-37-6	F	14.80±0.1	EI	3478
C₂H₅F⁺						
	C ₂ H ₅ F	353-36-6	**	12.43 (V)	PE	3984
			**	12.43 (V)	PE	4321
			**	12.43 (V)	PE	5088
C₃HF⁺						
	CHF ₂ C≡CH	18371-25-0	HF	12.6±0.15	EI	3769
C₃H₂F⁺						
	CHF ₂ C≡CH	18371-25-0	F	14.2±0.2	EI	3769
	(CF ₂ =CH) ₂	407-70-5	CF ₃	12.4±0.1	EI	5554
C₃H₃F⁺						
	CH ₂ FC≡CH	2805-22-3	**	10.95 (V)	PE	4684
C₃H₄F⁺						
	C ₄ H ₄ F ₄ (Cyclobutane, 1,1,2,2-tetrafluoro-)	374-12-9	CF ₃	12.85	EI	4553

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₃H₅F⁺						
	CH ₂ =CHCH ₂ F	818-92-8	**	10.11	PE	3863
			**	10.38 (V)	PE	4260
			**	10.56 (V)	PE	4091
C₃H₆F⁺						
	(CH ₃) ₂ CF	14665-81-7	**	7.14	PI	5003
	(CH ₃) ₂ CHF	420-26-8	H	11.23±0.03	PI	5003
C₃H₇F⁺						
	(CH ₃) ₂ CHF	420-26-8	**	11.08±0.02	PI	5003
	n-C ₃ H ₇ F	460-13-9	**	11.96 (V)	PE	3984
C₄HF⁺						
	CF≡CC≡CH	XXXXX-XX-X	**	10.10	PE	5313
C₆H₄F⁺						
	C ₆ H ₄ (F)COOH (Benzoic acid, 3-fluoro-)	455-38-9	CO+OH	15.25±0.2	EI	3973
	C ₆ H ₄ (F)COOH (Benzoic acid, 4-fluoro-)	456-22-4	CO+OH	15.33±0.2	EI	3973
	C ₆ H ₄ FNO ₂ (Benzene, 1-fluoro-3-nitro-)	402-67-5	NO ₂	12.22±0.1	EI	3447
	C ₆ H ₄ FNO ₂ (Benzene, 1-fluoro-4-nitro-)	350-46-9	NO ₂	12.37±0.1	EI	3447
C₆H₅F⁺						
	C ₆ H ₅ F (Benzene, fluoro-)	462-06-6	**	9.20	S	3559
			**	9.11	PE	3955
			**	9.17	PE	4621
			**	9.19 (V)	PE	3873
			**	9.22	PE	5408
			**	9.22 (V)	PE	5125
			**	9.35±0.03 (V)	PE	3713
			**	9.37 (V)	PE	4884
			**	9.75	EI	4834
	C ₆ H ₅ FOCH ₃ (Benzene, 1-fluoro-3-methoxy-)	456-49-5	CH ₂ O	11.76±0.1	EI	3446
	C ₆ H ₅ FOCH ₃ (Benzene, 1-fluoro-4-methoxy-)	459-60-9	CH ₂ O	11.55±0.1	EI	3446
C₇H₆F⁺						
	C ₆ H ₅ FC ₄ H ₉ (Benzene, 1-butyl-3-fluoro-)	20651-66-5		11.69±0.1	EI	3629
	C ₆ H ₅ FC ₄ H ₉ (Benzene, 1-butyl-4-fluoro-)	20651-65-4		11.25±0.1	EI	3629
C₇H₇F⁺						
	C ₆ H ₅ CH ₂ F (Benzene, (fluoromethyl)-)	350-50-5	**	9.55 (V)	PE	3992
	C ₆ H ₅ FC ₄ H ₉ (Benzene, 1-butyl-3-fluoro-)	20651-66-5	CH ₂ =CHCH ₃	10.21±0.1	EI	3629
	C ₆ H ₅ FC ₄ H ₉ (Benzene, 1-butyl-4-fluoro-)	20651-65-4	CH ₂ =CHCH ₃	10.29±0.1	EI	3629
C₁₀H₁₃F⁺						
	C ₆ H ₅ FC ₄ H ₉ (Benzene, 1-butyl-3-fluoro-)	20651-66-5	**	9.19±0.1	EI	3629
	C ₆ H ₅ FC ₄ H ₉ (Benzene, 1-butyl-4-fluoro-)	20651-65-4	**	9.15±0.1	EI	3629

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₀H₁₅F⁺	C ₁₀ H ₁₅ F (Tricyclo[3.3.1.1 ^{3,7}]decane, 2-fluoro-)	16668-83-0	**	9.46	PE	3886
C₁₁H₉F⁺	C ₁₁ H ₉ F (1,4-Methanonaphthalene, 5-fluoro-1,4-dihydro-)	61346-81-4	**	8.66±0.05 (V)	PE	5019
	C ₁₁ H ₉ F (1,4-Methanonaphthalene, 6-fluoro-1,4-dihydro-)	58653-71-7	**	8.62±0.05 (V)	PE	5019
C₁₂H₉F⁺	C ₆ H ₅ C ₆ H ₄ F (1,1'-Biphenyl, 2-fluoro-)	321-60-8	**	8.20±0.02	PE	3702
	C ₆ H ₅ C ₆ H ₄ F (1,1'-Biphenyl, 4-fluoro-)	324-74-3	**	8.00±0.02	PE	3702
C₁₄H₉F⁺	C ₁₄ H ₉ F (Anthracene, 9-fluoro-)	529-85-1	**	7.46±0.03 (V)	PE	4887
CHF₂⁺	CHF ₂	2670-13-5	**	<8.90	EI	3732
			**	9.45	OTH	5554
	CH ₂ F ₂	75-10-5	H	13.11	EI	3732
	CHF ₂ C≡CH	18371-25-0	C ₃ H	13.8±0.1	EI	3769
	(CF ₂ =CH) ₂	407-70-5	C ₃ F ₂ H	14.3±0.1	EI	5554
C₂HF₂⁺	CH ₂ =CF ₂	75-38-7	H	15.80±0.04	PI	3930
	cis-CHF=CHF	1630-77-9	H	14.9±0.2	PI	5241
	trans-CHF=CHF	1630-78-0	H	15.4±0.2	PI	5241
C₂H₂F₂⁺	CH ₂ =CF ₂	75-38-7	**	10.29±0.01	PI	3930
			**	10.29	PE	5408
			**	10.69±0.02 (V)	PE	5017
	cis-CHF=CHF	1630-77-9	**	10.23	PE	5408
			**	10.43 (V)	PE	3649
			**	10.44±0.02 (V)	PE	5017
	trans-CHF=CHF	1630-78-0	**	10.21	PE	5408
			**	10.38±0.02 (V)	PE	5017
			**	10.38 (V)	PE	3649
	(CH ₃) ₂ CF ₂	420-45-1	CH ₄	11.57±0.03	PI	5003
	C ₄ H ₄ F ₄	374-12-9	C ₂ H ₂ F ₂	12.15	EI	4553
	(Cyclobutane, 1,1,2,2-tetrafluoro-)					
C₂H₃F₂⁺	CH ₃ CF ₂	40640-67-3	**	7.92	PI	5003
	CH ₃ CHF ₂	75-37-6	H	12.18±0.03	PI	5003
	(CH ₃) ₂ CF ₂	420-45-1	CH ₃	11.81±0.03	PI	5003
	CH ₃ CF ₃	420-46-2	F	15.14±0.1	EI	3478
C₂H₄F₂⁺	CH ₃ CHF ₂	75-37-6	**	12.8 (V)	PE	4321
C₃HF₂⁺	CHF ₂ C≡CH	18371-25-0	H	12.9±0.1	EI	3769
	(CF ₂ =CH) ₂	407-70-5	CF ₂ H	14.0±0.1	EI	5554
C₃H₂F₂⁺	CF ₂ =C=CH ₂	430-64-8	**	9.79±0.03	PE	4833
	CHF ₂ C≡CH	18371-25-0	**	11.6±0.1	EI	3769
	(1,2-Propadiene, 1,1-difluoro-)					

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_3\text{H}_2\text{F}_2^+$	$(\text{CF}_2=\text{CH})_2$	407-70-5	CF_2	14.4 ± 0.2	EI	5554
$\text{C}_3\text{H}_6\text{F}_2^+$	$(\text{CH}_3)_2\text{CF}_2$	420-45-1	**	11.42 ± 0.02	PI	5003
$\text{C}_6\text{H}_4\text{F}_2^+$	$\text{C}_6\text{H}_4\text{F}_2$ (Benzene, 1,2-difluoro-)	367-11-3	**	9.30	S	4271
			**	9.30 (V)	PE	3873
			**	9.6 \pm 0.03 (V)	PE	3713
	$\text{C}_6\text{H}_4\text{F}_2$ (Benzene, 1,3-difluoro-)	372-18-9	**	9.35	S	4271
			**	9.32 (V)	PE	3873
			**	9.6 \pm 0.03 (V)	PE	3713
	$\text{C}_6\text{H}_4\text{F}_2$ (Benzene, 1,4-difluoro-)	540-36-3	**	9.18	S	4271
			**	9.15 (V)	PE	3873
			**	9.29	PE	5408
			**	9.4 \pm 0.03 (V)	PE	3713
$\text{C}_{12}\text{H}_8\text{F}_2^+$	$(\text{C}_6\text{H}_4\text{F})_2$ (1,1'-Biphenyl, 2,2'-difluoro-)	388-82-9	**	8.35 ± 0.02	PE	3702
	$(\text{C}_6\text{H}_4\text{F})_2$ (1,1'-Biphenyl, 3,3'-difluoro-)	396-64-5	**	8.35 ± 0.02	PE	3702
	$(\text{C}_6\text{H}_4\text{F})_2$ (1,1'-Biphenyl, 4,4'-difluoro-)	398-23-2	**	8.00 ± 0.02	PE	3702
CHF_3^+	CF_3H	75-46-7	**	14.8 ± 0.05 (V)	PE	5419
C_2HF_3^+	C_2HF_3	359-11-5	**	10.14	PE	5408
			**	10.53 (V)	PE	3649
			**	10.54 \pm 0.02 (V)	PE	5017
$\text{C}_2\text{H}_3\text{F}_3^+$	CH_3CF_3	420-46-2	**	13.26 ± 0.1	EI	3478
			**	13.8 (V)	PE	4321
C_3HF_3^+	$\text{CF}_3\text{C}=\text{CH}$	661-54-1	**	11.83	PE	3589
			**	11.96 ± 0.02	PE	4765
$\text{C}_4\text{H}_2\text{F}_3^+$	$(\text{CF}_2=\text{CH})_2$	407-70-5	F	15.2 ± 0.1	EI	5554
$\text{C}_4\text{H}_4\text{F}_3^+$	$\text{C}_4\text{H}_4\text{F}_4$ (Cyclobutane, 1,1,2,2-tetrafluoro-)	374-12-9	F ⁻	13.5 ± 1	EI	4553
$\text{C}_6\text{H}_3\text{F}_3^+$	$\text{C}_6\text{H}_3\text{F}_3$ (Benzene, 1,3,5-trifluoro-)	372-38-3	**	9.26 (V)	PE	3873
			**	9.64	PE	3764
			**	9.64	PE	5408
$\text{C}_7\text{H}_5\text{F}_3^+$	$\text{C}_6\text{H}_5\text{CF}_3$ (Benzene, (trifluoromethyl)-)	98-08-8	**	9.68	PE	4621

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_9\text{H}_9\text{F}_3^+$	$\text{C}_6\text{F}_3(\text{CH}_3)_3$ (Benzene, 1,3,5-trifluoro-2,4,6-trimethyl-)	363-64-4	**	8.76 ± 0.02	PE	5521
			**	8.76 (V)	PE	5461
$\text{C}_4\text{H}_2\text{F}_4^+$	$(\text{CF}_2 = \text{CH})_2$	407-70-5	**	10.6 ± 0.1	EI	5554
$\text{C}_6\text{H}_2\text{F}_4^+$	$\text{C}_6\text{H}_2\text{F}_4$ (Benzene, 1,2,3,4-tetrafluoro-)	551-62-2	**	9.56 (V)	PE	3873
			**	9.60	PE	5408
	$\text{C}_6\text{H}_2\text{F}_4$ (Benzene, 1,2,3,5-tetrafluoro-)	2367-82-0	**	9.56 (V)	PE	3873
			**	9.36 (V)	PE	3873
$\text{C}_7\text{H}_4\text{F}_4^+$	$\text{C}_6\text{HF}_4\text{CH}_3$ (Benzene, 1,2,4,5-tetrafluoro-3-methyl-)	5230-78-4	**	9.16 ± 0.02	PE	5521
			**	9.16 (V)	PE	5461
			**	9.82	S	3559
C_6HF_5^+	C_6HF_5 (Benzene, pentafluoro-)	363-72-4	**	9.64 (V)	PE	3873
			**	9.73	PE	5408
			**	9.90 (V)	PE	5252
			**	9.4	PE	5521
$\text{C}_7\text{H}_3\text{F}_5^+$	$\text{C}_6\text{F}_5\text{CH}_3$ (Benzene, pentafluoromethyl-)	771-56-2	**	9.4	PE	5461
			**	9.4 (V)	PE	5252
			**	9.81 (V)	PE	3854
			**	9.18 \pm 0.02	PE	3854
$\text{C}_{16}\text{H}_8\text{F}_8^+$	$\text{C}_{10}\text{H}_8\text{F}_8$ (Tricyclo[8.2.2.2 ^{4,7}]hexadeca-4,6,10,12,13,15-hexaene, 2,2,3,3,8,8,9,9,-octafluoro-)	3345-29-7	**	8.90	PE	4158
			**	13.16 (V)	PE	5485
			**	11.22 (V)	PE	4243
$\text{BC}_2\text{H}_6\text{F}^+$	$(\text{CH}_3)_2\text{BF}$	353-46-8	**	11.25 (V)	PE	5485
			**	9.61 (V)	PE	4956
$\text{BC}_6\text{H}_5\text{F}_2^+$	$\text{C}_6\text{H}_5\text{BF}_2$ (Borane, difluorophenyl-)	368-98-9	**	11.86 \pm 0.2	EI	4628
NF^+	NF_2	3744-07-8	F ⁻	11.86 \pm 0.2	EI	3785

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
NF⁺						
	NF ₂	3744-07-8	F	15.46±0.2	EI	3785
	N ₂ F ₄	10036-47-2	NF ₂ +F	16.6	EI	3785
	(CH ₃) ₂ C(NF ₂) ₂	19309-63-8		13.9±0.3	EI	3634
	(CH ₂ NF ₂) ₂ CH ₂	21298-22-6		13.0±0.3	EI	3634
N₂F⁺						
	N ₂ F ₄	10036-47-2	F ₂ +F 3F	14.2±0.3 16.7±0.3	EI EI	3785 3785
NF₂⁺						
(¹ A ₁)	NF ₂	3744-07-8	**	12.1±0.1 (V)	PE	3671
(¹ A ₁)			**	12.1	PE	3693
(³ B ₁)			**	14.6±0.1 (V)	PE	3671
(³ B ₁)			**	14.6	PE	3693
(¹ B ₁ , ³ B ₂ , ³ A ₂ , ¹ A ₂)			**	16.4	PE	3693
(¹ B ₂)			**	17.6	PE	3693
			**	11.76±0.1	EI	3785
	N ₂ F ₄	10036-47-2	F ⁻ +NF NF ₂	12.40±0.1 12.70±0.1	EI EI	3785 3785
	(CH ₃) ₂ C(NF ₂) ₂	19309-63-8		13.9±0.4	EI	3634
	(CH ₂ NF ₂) ₂ CH ₂	21298-22-6		14.8±0.4	EI	3634
N₂F₂⁺						
	trans-N ₂ F ₂	13776-62-0	**	12.8	PE	3649
	N ₂ F ₄	10036-47-2	2F	16.0±0.1	EI	3785
NF₃⁺						
	NF ₃	7783-54-2	**	12.97±0.04	PE	3641
			**	13.18±0.1	EI	3578
N₂F₄⁺						
	N ₂ F ₄	10036-47-2	**	12.00±0.1	EI	3785
HNF₂⁺						
	HNF ₂	10405-27-3	**	11.53±0.08	PE	5253
HBNF⁺						
	H ₂ NBF ₂	50673-31-9	HF	14.0±0.2	EI	4522
H₂BNF⁺						
	H ₂ NBF ₂	50673-31-9	F	16.1±0.4	EI	4522
H₂BNF₂⁺						
	H ₂ NBF ₂	50673-31-9	**	12.4±0.4	EI	4522
H₃B₃N₃F₃⁺						
	B ₃ H ₃ N ₃ F ₃ (Borazine, 2,4,6-trifluoro-)	13779-24-3	**	10.46	PE	3637
			**	10.66 (V)	PE	3673
			**	10.66 (V)	PE	3943
CNF⁺						
	FCN	1495-50-7	**	13.34±0.02	PE	4676
			**	14.48±0.02	PE	4676
			**	19.3±0.1 (V)	PE	4676
			**	22.6±0.1 (V)	PE	4676
C₃NF⁺						
	CF≡CCN	32038-83-8	**	11.51±0.02	PE	4765

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
CHNF₂⁺	(CH ₃) ₂ C(NF ₂) ₂ (CH ₂ NF ₂) ₂ CH ₂	19309-63-8 21298-22-6		13.2±0.3 13.7±0.3	EI EI	3634 3634
CH₂NF₂⁺	CH ₂ (NF ₂)CH(NF ₂)CH ₃ (CH ₂ NF ₂) ₂ CH ₂	15403-25-5 21298-22-6		13.1±0.2 13.6±0.3	EI EI	3634 3634
C₂H₆NF₂⁺	CH ₂ (NF ₂)CH(NF ₂)CH ₃ (CH ₃) ₂ C(NF ₂) ₂ (CH ₂ NF ₂) ₂ CH ₂	15403-25-5 19309-63-8 21298-22-6		10.8±0.2 11.1±0.3 11.8±0.3	EI EI EI	3634 3634 3634
C₆H₅NF₂⁺	C ₆ H ₃ F ₂ NHCOCH ₃ (Acetamide, N-(2,4-difluorophenyl)-) C ₆ H ₃ F ₂ NHCOCH ₃ (Acetamide, N-(2,6-difluorophenyl)-)	399-36-0 3896-29-5	CH ₂ =C=O	9.70±0.03 9.52±0.03	EI EI	3480 3480
C₄H₂N₂F₂⁺	C ₄ H ₂ F ₂ N ₂ (Pyrazine, 2,3-difluoro-) C ₄ H ₂ F ₂ N ₂ (Pyrazine, 2,6-difluoro-) C ₄ H ₂ F ₂ N ₂ (Pyridazine, 3,6-difluoro-) C ₄ H ₂ F ₂ N ₂ (Pyrimidine, 2,4-difluoro-) C ₄ H ₂ F ₂ N ₂ (Pyrimidine, 4,6-difluoro-)	52751-15-2 33873-09-5 33097-39-1 2802-61-1 2802-62-2	**	10.35 (V) 10.30 (V) 10.17 10.65 (V) 10.95 (V)	PE PE PE PE PE	5530 5530 5530 5530 5530
C₈H₄N₂F₂⁺	C ₈ H ₄ N ₂ F ₂ (1,8-Naphthyridine, 2,7-difluoro-) C ₈ H ₄ N ₂ F ₂ (Quinazoline, 2,4-difluoro-) C ₈ H ₄ N ₂ (F) ₂ (Quinoxaline, 2,3-difluoro-)	56595-11-0 56595-10-9 7066-36-6	**	9.26 (V) 9.30 (V) 9.30 (V)	PE PE PE	4523 4523 3959
C₉H₉N₃F₂⁺	C ₆ H ₃ F ₂ NC ₃ H ₄ N ₂ H ₂ (Imidazolidine, 2-(2,6-difluorophenylimino)-)	XXXXX-XX-X	**	8.12 (V)	PE	5545
C₆H₄NF₃⁺	C ₆ H ₄ N(CF ₃) (Pyridine, 4-(trifluoromethyl)-)	XXXXX-XX-X	**	10.1 (V)	PE	4536
C₁₄H₁₀NF₃⁺	C ₆ H ₄ CF ₃ C(=CH ₂)C ₆ H ₄ N (Pyridine, 2-[1-[3-trifluoromethyl]phenyl]ethenyl]-) C ₆ H ₄ CF ₃ C(=CH ₂)C ₆ H ₄ N (Pyridine, 2-[1-[4-trifluoromethyl]phenyl]ethenyl]-)	XXXXX-XX-X	**	9.02 8.97	EI EI	5570 5570
C₄HN₂F₃⁺	C ₄ HF ₃ N ₂ (Pyrimidine, 2,4,6-trifluoro-)	696-82-2	**	10.93 (V)	PE	5530
C₈H₃NF₄⁺	C ₆ F ₄ C ₂ H ₂ NH (1H-Indole, 4,5,6,7-tetrafluoro-)	16264-67-8	**	8.30±0.015 (V)	PE	5522

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_6\text{H}_2\text{N}_2\text{F}_4^+$	$\text{C}_6\text{H}_2\text{N}_2\text{F}_4$ (2,7-Naphthyridine, 1,3,6,8-tetrafluoro-) $\text{C}_6\text{H}_2\text{N}_2(\text{F})_4$ (Quinoxaline, 5,6,7,8-tetrafluoro-)	56595-13-2 33319-19-6	**	9.55 (V) 9.50 (V)	PE	4523 3959
$\text{C}_6\text{H}_2\text{NF}_5^+$	$\text{C}_6\text{F}_5\text{NH}_2$ (Benzenamine, 2,3,4,5,6-pentafluoro-)	771-60-8	**	8.40 ± 0.02	PE	3890
$\text{C}_9\text{H}_6\text{N}_3\text{F}_5^+$	$\text{C}_6\text{F}_5\text{NC}_3\text{H}_4\text{N}_2\text{H}_2$ (Imidazolidine, 2-(pentafluorophenylimino)-)	XXXXX-XX-X	**	8.60 (V)	PE	5545
C_3HNF_6^+	$(\text{CF}_3)_2\text{C}=\text{NH}$	1645-75-6	**	11.8 (V)	PE	4814
$\text{C}_6\text{H}_7\text{NF}_6^+$	$(\text{CH}_3)_2\text{NC}(\text{CF}_3)=\text{C}(\text{CF}_3)\text{H}$	35186-00-6	**	8.22	PE	3589
$\text{BC}_4\text{H}_{12}\text{N}_2\text{F}^+$	$((\text{CH}_3)_2\text{N})_2\text{BF}_2$	383-90-4	**	8.04	PE	3584
$\text{BC}_2\text{H}_6\text{NF}_2^+$	$(\text{CH}_3)_2\text{NBF}_2$	359-18-2	**	9.71	PE	3584
$\text{B}_2\text{C}_4\text{H}_{12}\text{N}_2\text{F}_2^+$	$(\text{FCH}_3)\text{BNCH}_3$	73775-17-1	**	9.34 (V)	PE	5628
$\text{BC}_6\text{H}_7\text{NF}_3^+$	$\text{C}_5\text{H}_4\text{N}(\text{CF}_3)\text{-BH}_3$ (Pyridine, 4-trifluoromethyl)-, compound with borane (1:1))	56898-54-5	**	10.04 (V)	PE	4536
$\text{B}_3\text{C}_3\text{H}_6\text{N}_3\text{F}_3^+$	$\text{C}_3\text{H}_9\text{B}_3\text{N}_3\text{F}_3$ (Borazine, 2,4,6-trifluoro-1,3,5-trimethyl-)	13722-15-1	**	9.48 (V)	PE	3943
OF^+	FO	12061-70-0	**	12.77	PE	5425
			**	12.79 ± 0.1	OTH	3920
OF_2		7783-41-7	F	≤ 14.438	PI	3920
	OF ₂	7783-41-7	**	13.11 ± 0.01	PI	3920
OF_2^+			**	13.11	PE	3649
			**	13.26 (V)	PE	3694
			**	15.74	PE	3649
			**	16.17 (V)	PE	3694
			**	16.44 (V)	PE	3649
			**	16.47 (V)	PE	3694
			**	17.9	PE	3649
			**	20.7 (V)	PE	3649
HOF^+	HOF	14034-79-8	**	12.71 ± 0.01	PI	3932
			**	12.69 ± 0.03	PE	3831
			**	14.50 ± 0.03	PE	3831
			**	15.9 ± 0.05	PE	3831
BOF^+	BOF	23361-56-0	**	14 ± 1	EI	4054

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
BOF₂⁺	BOF ₂	12006-82-5	**	17±1	EI	4054
COF⁺	CF ₂ O	353-50-4	F	14.85±0.2	PI	5041
COF₂⁺	CF ₂ O	353-50-4	** ** ** ** ** ** ** **	13.02 13.04 13.6 (V) 14.09 19.15 19.8 (V) 21.1 (V) ~22.7	PE PE PE PE PE PE PE PE	3649 3726 5041 3649 3649 3649 3649 3649
C₂O₂F₂⁺	(COF) ₂	359-40-0	**	12.20±0.02	PE	4696
C₂OF₅⁺	(CF ₃) ₂ CO	684-16-2		11.65	EI	3550
COF₄⁺	CF ₃ OF	373-91-1	**	13.6 (V)	PE	3941
C₆O₂F₄⁺	C ₆ F ₄ O ₂ (2,5-Cyclohexadiene,1,4-dione,2,3,5,6-tetrafluoro-)	527-21-9	**	10.96±0.05 (V)	PE	5558
C₃OF₅⁺	(CF ₃) ₂ CO	684-16-2		16	EI	3550
C₃OF₆⁺	(CF ₃) ₂ CO	684-16-2	** **	11.44 12.09±0.02 (V)	PE PE	3649 4524
CHOF⁺ (²A')	HFCO	1493-02-3	**	12.37±0.02	PE	4496
C₂H₅OF⁺	CH ₃ CFO	557-99-3	**	11.51±0.02	PE	4220
C₂H₅OF⁺	CH ₂ FCH ₂ OH	371-62-0	**	10.98 (V)	PE	5088
C₃H₅OF⁺	CH ₃ COCH ₂ F C ₂ H ₃ OCH ₂ F (Oxirane, (fluoromethyl)-)	430-51-3 503-09-3	** **	10.20±0.02 (V) 10.78 (V)	PE PE	4524 4747
C₃H₇OF⁺	CH ₂ FCH ₂ OCH ₃	627-43-0	**	10.18 (V)	PE	5088
C₆H₄OF⁺	C ₆ H ₄ FOCH ₃ (Benzene, 1-fluoro-3-methoxy-) C ₆ H ₄ FOCH ₃ (Benzene, 1-fluoro-4-methoxy-) C ₆ H ₄ FNO ₂ (Benzene, 1-fluoro-3-nitro-) C ₆ H ₄ FNO ₂ (Benzene, 1-fluoro-4-nitro-)	456-49-5 459-60-9 402-67-5 350-46-9	CH ₃ CH ₃ NO NO	12.53±0.1 11.99±0.1 10.25±0.1 10.64±0.1	EI EI EI EI	3446 3446 3447 3447

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₆H₅OF⁺						
	C ₆ H ₅ (F)(OH) (Phenol, 2-fluoro-)	367-12-4	**	8.95 (V)	PE	4891
			**	8.97±0.02 (V)	PE	5184
	C ₆ H ₅ (F)(OH) (Phenol, 3-fluoro-)	372-20-3	**	8.99 (V)	PE	4891
			**	9.05±0.02 (V)	PE	5184
	C ₆ H ₅ (F)(OH) (Phenol, 4-fluoro-)	371-41-5	**	8.77 (V)	PE	4891
			**	8.79±0.02 (V)	PE	5184
	C ₆ H ₅ FOOCCH ₃ (Phenol, 2-fluoro-, acetate)	29650-44-0	CH ₂ =C=O	9.17±0.03	EI	3483
	C ₆ H ₅ FOOCCH ₃ (Phenol, 4-fluoro-, acetate)	405-51-6	CH ₂ =C=O	9.55±0.03	EI	3483
C₇H₄OF⁺						
	FC ₆ H ₄ COCH ₃ (Ethanone, 1-(4-fluorophenyl))	403-42-9	CH ₃	10.39±0.03	EI	5059
	C ₆ H ₄ (F)COOH (Benzoic acid, 3-fluoro-)	455-38-9	OH	12.50±0.2	EI	3973
	C ₆ H ₄ (F)COOH (Benzoic acid, 4-fluoro-)	456-22-4	OH	12.33±0.2	EI	3973
C₇H₇OF⁺						
	C ₆ H ₄ FOCH ₃ (Benzene, 1-fluoro-3-methoxy-)	456-49-5	**	8.70±0.1	EI	3446
	C ₆ H ₄ FOCH ₃ (Benzene, 1-fluoro-4-methoxy-)	459-60-9	**	8.58±0.1	EI	3446
C₈H₉OF⁺						
	C ₆ H ₅ OCH ₂ CH ₂ F (Benzene, 2-fluoroethoxy-)	405-97-0	**	8.63	EI	5083
C₇H₅O₂F⁺						
	C ₆ H ₄ (F)COOH (Benzoic acid, 3-fluoro-)	455-38-9	**	9.91±0.2	EI	3973
	C ₆ H ₄ (F)COOH (Benzoic acid, 4-fluoro-)	456-22-4	**	9.91±0.2	EI	3973
C₈H₇O₂F⁺						
	C ₆ H ₅ FOOCCH ₃ (Phenol, 2-fluoro-, acetate)	29650-44-0	**	8.78±0.03	EI	3483
	C ₆ H ₅ FOOCCH ₃ (Phenol, 4-fluoro-, acetate)	405-51-6	**	8.27±0.03	EI	3483
C₆H₄OF₂⁺						
	C ₆ H ₅ (OH)F ₂ (Phenol, 2,4-difluoro-)	367-27-1	**	8.98±0.02 (V)	PE	5184
	C ₆ H ₅ (OH)F ₂ (Phenol, 2,5-difluoro-)	XXXXX-XX-X	**	9.10±0.02 (V)	PE	5184
	C ₆ H ₅ (OH)F ₂ (Phenol, 3,5-difluoro-)	XXXXX-XX-X	**	9.04±0.02 (V)	PE	5184
	C ₆ H ₅ F ₂ OOCCH ₃ (Phenol, 2,4-difluoro-, acetate)	36914-77-9	CH ₂ =C=O	9.63±0.03	EI	3480
	C ₆ H ₅ F ₂ OOCCH ₃ (Phenol, 2,6-difluoro-, acetate)	36914-78-0	CH ₂ =C=O	9.69±0.03	EI	3480
C₈H₆O₂F₂⁺						
	C ₆ H ₅ F ₂ OOCCH ₃ (Phenol, 2,4-difluoro-, acetate)	36914-77-9	**	8.60±0.03	EI	3480
	C ₆ H ₅ F ₂ OOCCH ₃ (Phenol, 2,6-difluoro-, acetate)	36914-78-0	**	8.88±0.03	EI	3480

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_2\text{H}_3\text{OF}_3^+$	$\text{CF}_3\text{CH}_2\text{OH}$	75-89-8	**	11.7 (V)	PE	3941
$\text{C}_3\text{H}_3\text{OF}_3^+$	CH_3COCF_3	421-50-1	**	11.00 ± 0.02 (V)	PE	4524
$\text{C}_6\text{H}_3\text{OF}_3^+$	$\text{C}_6\text{H}_2(\text{OH})\text{F}_3$ (Phenol,2,3,4-trifluoro-) $\text{C}_6\text{H}_2(\text{OH})\text{F}_3$ (Phenol,2,4,5-trifluoro-)	XXXXX-XX-X	**	9.19 ± 0.02 (V)	PE	5184
$\text{C}_2\text{HO}_2\text{F}_3^+$	CF_3COOH	76-05-1	**	11.46	PE	3718
			**	12.00 ± 0.03 (V)	PE	3734
			**	12.00 (V)	PE	3874
			**	12.06 (V)	PE	5251
			**	12.08 ± 0.05 (V)	PE	4986
$\text{C}_3\text{H}_3\text{O}_2\text{F}_3^+$	$\text{HCOOCH}_2\text{CF}_3$	32042-38-9	**	11.31	PE	3718
$\text{C}_4\text{H}_5\text{O}_2\text{F}_3^+$	$\text{CF}_3\text{COOC}_2\text{H}_5$ $\text{CH}_3\text{COOCH}_2\text{CF}_3$	383-63-1	**	11.6 (V)	PE	3718
		406-95-1	**	10.84	PE	3718
$\text{C}_5\text{H}_5\text{O}_2\text{F}_3^+$	$\text{CF}_3\text{COCH}_2\text{COCH}_3$	367-57-7	**	9.92 ± 0.07 (V)	PE	3682
$\text{C}_6\text{H}_3\text{O}_2\text{F}_3^+$	$\text{C}_3\text{H}_3\text{OCOCF}_3$ (Ethanone, 2,2,2-trifluoro-1-(2-furanyl)-)	18207-47-1	**	9.77 ± 0.05	EI	3482
$\text{C}_8\text{H}_{11}\text{O}_2\text{F}_3^+$	$(\text{CH}_3)_3\text{CCOCH}_2\text{COCF}_3$	22767-90-4	**	9.87 ± 0.07 (V)	PE	3682
$\text{C}_3\text{H}_2\text{OF}_4^+$	$(\text{CHF}_2)_2\text{CO}$	360-52-1	**	11.33 ± 0.02 (V)	PE	4524
$\text{C}_6\text{H}_2\text{OF}_4^+$	$\text{C}_6\text{H}(\text{OH})\text{F}_4$ (Phenol,2,3,5,6-tetrafluoro-)	769-39-1	**	9.40 ± 0.02 (V)	PE	5184
$\text{C}_3\text{H}_3\text{OF}_5^+$	$\text{C}_2\text{F}_5\text{CH}_2\text{OH}$	422-05-9	**	11.68 (V)	PE	3941
C_6HOF_5^+	$\text{C}_6\text{F}_5\text{OH}$ (Phenol,pentafluoro-)	771-61-9	**	9.37 ± 0.02 (V)	PE	5184
			**	9.20 ± 0.02	PE	3890
$\text{C}_7\text{H}_3\text{OF}_5^+$	$\text{C}_6\text{F}_5\text{OCH}_3$ (Benzene, pentafluoromethoxy-)	389-40-2	**	9.10 ± 0.02	PE	3890
$\text{C}_3\text{H}_2\text{OF}_6^+$	$\text{CF}_3\text{CH}(\text{OH})\text{CF}_3$	920-66-1	**	12.23 (V)	PE	3941
$\text{C}_5\text{H}_2\text{O}_2\text{F}_6^+$	$\text{CF}_3\text{COCH}_2\text{COCF}_3$	1522-22-1	**	10.74 ± 0.07 (V)	PE	3682

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₅HO₅F₁₁⁺	CF ₃ CF ₂ CF ₂ OCHFCF ₃	3330-15-2	**	12.6	PE	4424
C₈HO₂F₁₇⁺	F(CF(CF ₃)CF ₂ O) ₂ CHFCF ₃	3330-14-1	**	12.78 (V)	PE	4424
C₁₁HO₃F₂₃⁺	F(CF(CF ₃)CF ₂ O) ₃ CHFCF ₃	3330-16-3	**	12.96 (V)	PE	4424
C₁₄HO₄F₂₉⁺	F(CF(CF ₃)CF ₂ O) ₄ CHFCF ₃	26738-51-2	**	13.47 (V)	PE	4424
BeC₁₀H₂O₄F₁₂⁺	(CF ₃ COCHCOCF ₃) ₂ Be (Beryllium, bis(1,1,1,5,5,5-hexafluoro-2,4-pentanedionato-O,O')-, (T-4)-)	19648-82-9	**	10.39±0.07 (V)	PE	3682
NOF⁺ (² A')	ONF	7789-25-5	**	12.63±0.03	PE	4420
			**	12.66	PE	4404
NO₂F⁺	NO ₂ F	10022-50-1	**	13.09	PE	4404
NOF₃⁺	NOF ₃	13847-65-9	**	13.36±0.01	PE	3641
CNOF₃⁺	CF ₃ NO	XXXXX-XX-X	**	10.5±0.1	EI	5220
	CF ₃ NO	334-99-6		11.06±0.05 (V)	PE	5298
C₂NOF₆⁺	(CF ₃) ₂ NO	2154-71-4	**	10.7±0.1 (V)	PE	3671
C₇H₆NOF⁺	C ₆ H ₄ (F)(CONH ₂) (Benzamide, 4-fluoro-)	824-75-9	**	9.50 (V)	PE	4918
C₈H₈NOF⁺	C ₆ H ₄ FNHCOCH ₃ (Acetamide, N-(2-fluorophenyl)-)	399-31-5	**	8.65	EI	4834
			**	8.27±0.03	EI	3483
	C ₆ H ₄ FNHCOCH ₃ (Acetamide, N-(4-fluorophenyl)-)	351-83-7	**	8.20±0.03	EI	3483
C₁₂H₈NOF⁺	C ₆ H ₄ FCOC ₅ H ₄ N (Methanone, (2-fluorophenyl)-2-pyridinyl-)	XXXXX-XX-X	**	9.11	EI	5459
C₄H₃N₂OF⁺	C ₄ H ₃ N ₂ F(=O) (2(1H)-Pyrimidinone, 5-fluoro-)	2022-78-8	**	10.08±0.05	EI	5159
C₅H₅N₂OF⁺	C ₄ H ₂ N ₂ FOCH ₃ (Pyrimidine, 5-fluoro-2-methoxy-)	17148-49-1	**	9.65±0.05	EI	5159
	C ₄ H ₂ N ₂ F(=O)CH ₃ (2(1H)-Pyrimidinone, 5-fluoro-1-methyl-)	63331-05-5	**	9.21±0.05	EI	5159
C₇H₇N₂OF⁺	C ₆ H ₄ FNHCONH ₂ (Urea, (2-fluorophenyl)-)	656-31-5	**	8.50	EI	4834

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₆H₄NO₂F⁺						
	C ₆ H ₄ (F)(NO ₂) (Benzene, 1-fluoro-2-nitro-)	1493-27-2	**	9.86 (V)	PE	4892
	C ₆ H ₄ (F)(NO ₂) (Benzene, 1-fluoro-3-nitro-)	402-67-5	**	9.88	PE	4892
			**	9.93±0.1	EI	3447
	C ₆ H ₄ (F)(NO ₂) (Benzene, 1-fluoro-4-nitro-)	350-46-9	**	9.90	PE	4892
			**	10.00±0.1	EI	3447
C₁₂H₉N₄O₂F⁺						
	C ₁₀ H ₃ N ₄ (F)(=O) ₂ (CH ₃) ₂ (Benzo[g]pteridine-2,4(3H,10H)-dione, 3,10-dimethyl-7-fluoro-)	XXXXXX-XX-X	**	8.51 (V)	PE	4992
C₈H₇NOF₂⁺						
	C ₆ H ₃ F ₂ NHCOCH ₃ (Acetamide, N-(2,4-difluorophenyl)-)	399-36-0	**	8.21±0.03	EI	3480
	C ₆ H ₃ F ₂ NHCOCH ₃ (Acetamide, N-(2,6-difluorophenyl)-)	3896-29-5	**	8.52±0.03	EI	3480
C₆H₄NOF₃⁺						
	C ₄ H ₄ NCOCF ₃ (Ethanone, 2,2,2-trifluoro-1-(1 <i>H</i> -pyrrol-2-yl)-)	2557-70-2	**	9.18±0.05	EI	3482
	C ₅ H ₄ N(O)CF ₃ (Pyridine, 4-(trifluoromethyl)-1-oxide-)	XXXXXX-XX-X	**	8.90 (V)	PE	4536
C₈H₆NOF₃⁺						
	C ₆ H ₅ NHCO(CF ₃) (Acetamide, 2,2,2-trifluoro-N-phenyl-)	404-24-0	**	8.93±0.05 (V)	PE	5013
C₉H₈NOF₃⁺						
	C ₆ H ₅ (CH ₃)NHCO(CF ₃) (Acetamide, 2,2,2-trifluoro-N-(2-methylphenyl)-)	2727-68-6	**	8.84±0.05 (V)	PE	5013
	C ₆ H ₅ (CH ₃) ₂ NHCO(CF ₃) (Acetamide, 2,2,2-trifluoro-N-(3-methylphenyl)-)	2727-69-7	**	8.73±0.05 (V)	PE	5013
	C ₆ H ₅ (CH ₃) ₂ NHCO(CF ₃) (Acetamide, 2,2,2-trifluoro-N-(4-methylphenyl)-)	350-96-9	**	8.61±0.05 (V)	PE	5013
C₁₀H₁₀NOF₃⁺						
	C ₆ H ₅ (CH ₃) ₂ NHCO(CF ₃) (Acetamide, N-(2,3-dimethylphenyl)-2,2,2-trifluoro-)	14719-31-4	**	8.62±0.05 (V)	PE	5013
	C ₆ H ₅ (CH ₃) ₂ NHCO(CF ₃) (Acetamide, N-(3,4-dimethylphenyl)-2,2,2-trifluoro-)	XXXXXX-XX-X	**	8.51±0.05 (V)	PE	5013
	C ₆ H ₅ (CH ₃) ₂ NHCO(CF ₃) (Acetamide, N-(2,4-dimethylphenyl)-2,2,2-trifluoro-)	14618-47-4	**	8.56±0.05 (V)	PE	5013
	C ₆ H ₅ (CH ₃) ₂ NHCO(CF ₃) (Acetamide, N-(3,5-dimethylphenyl)-2,2,2-trifluoro-)	14818-53-2	**	8.59±0.05 (V)	PE	5013
	C ₆ H ₅ (CH ₃) ₂ NHCO(CF ₃) (Acetamide, N-(2,5-dimethylphenyl)-2,2,2-trifluoro-)	14618-48-5	**	8.70±0.05 (V)	PE	5013
	C ₆ H ₅ (CH ₃) ₂ NHCO(CF ₃) (Acetamide, N-(2,6-dimethylphenyl)-2,2,2-trifluoro-)	7497-27-0	**	8.99±0.05 (V)	PE	5013
C₅H₃N₂OF₃⁺						
	C ₅ H ₃ N ₂ OF ₃ (1 <i>H</i> -Imidazole, 1-(trifluoroacetyl)-)	1546-79-8	**	9.91 (V)	PE	5092
C₁₁H₁₆NO₂F₃⁺						
	C ₁₁ H ₁₆ NO ₂ F ₃ (2,4-Azetidinedione, 3,3-bis(1-methylethyl)-1-(2,2,2-trifluoroethyl)-)	56519-50-7	**	9.50	EI	4660
Ne⁺ (² P _{3/2})	Ne	7440-01-9	**	21.56471±0.00001 S		3754

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
Ne^+ (^2P)	Ne	7440-01-9	**	21.59 (V)	PE	4970
			**	22.0	PE	4623
			**	48.49 ± 0.01 (V)	PE	4970
			**	49.0	PE	4623
Ne^{+2}	Ne	7440-01-9	**	62.8 ± 0.2	EI	4503
Na^+	Na	7440-23-5	**	5.2	EI	4912
			**	5.3 ± 0.2	EI	3609
			**	5.55 ± 0.2	EI	5588
			**	5.6 ± 0.3	EI	4518
	NaBO_2 NaF	XXXXX-XX-X	BO_2	9.66 ± 0.15	EI	4663
			F	9.98 ± 0.15	EI	4663
				~12	EI	3464
	Na_2	25681-79-2	**	4.866 ± 0.014	PI	4914
			**	4.9	EI	4912
Na_3^+	Na_3	37279-42-8	**	3.97 ± 0.05	PI	4914
Na_4^+	Na_4	39297-86-4	**	4.27 ± 0.05	PI	4914
Na_5^+	Na_5	39297-87-5	**	4.05 ± 0.05	PI	4914
Na_6^+	Na_6	39297-88-6	**	4.12 ± 0.05	PI	4914
Na_7^+	Na_7	39297-89-7	**	4.04 ± 0.05	PI	4914
Na_8^+	Na_8	39297-90-0	**	4.10 ± 0.05	PI	4914
Na_9^+	Na_9	66457-73-6	**	4.0 ± 0.01	PI	4914
Na_{10}^+	Na_{10}	XXXXX-XX-X	**	3.9 ± 0.1	PI	4914
Na_{11}^+	Na_{11}	66457-74-7	**	3.8 ± 0.1	PI	4914
Na_{12}^+	Na_{12}	XXXXX-XX-X	**	3.6 ± 0.1	PI	4914
Na_{13}^+	Na_{13}	66457-75-8	**	3.6 ± 0.1	PI	4914
Na_{14}^+	Na_{14}	66457-76-9	**	3.5 ± 0.1	PI	4914
LiNa^+	NaLi	12333-49-2	**	4.94 ± 0.10	EI	4912

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
ONa⁺						
	NaO	12401-86-4	**	12.9	EI	4518
BO₂Na⁺						
	NaBO ₂	XXXXXX-XX-X	**	9.18±0.10	EI	4663
BO₂Na₂⁺						
	(NaBO ₂) ₂	XXXXXX-XX-X BO ₂		10.15±0.12	EI	4663
	Na ₂ BO ₂ F	XXXXXX-XX-X F ⁻		6.18±0.10	EI	4663
	Na ₂ BO ₂ F	XXXXXX-XX-X F		10.15±0.12	EI	4663
FNa₂⁺						
	Na ₂ F ₂	12285-64-2	F ⁻	5.86±0.10	EI	4663
			F	10.00±0.10	EI	4663
	Na ₂ BO ₂ F	XXXXXX-XX-X BO ₂ ⁻		5.86±0.10	EI	4663
	Na ₂ BO ₂ F	XXXXXX-XX-X BO ₂		10.00±0.10	EI	4663
Mg⁺						
	Mg	7439-95-4	**	7.63±0.08	EI	4114
			**	7.72±0.05	EI	5342
	(C ₅ H ₅) ₂ Mg	1284-72-6		13.9±0.5	EI	3793
	(Magnesium, bis(η^5 -2,4-cyclopentadien-1-yl)-)					
C₅H₅Mg⁺						
	(C ₅ H ₅) ₂ Mg	1284-72-6		11.0±0.2	EI	3793
	(Magnesium, bis(η^5 -2,4-cyclopentadien-1-yl)-)					
C₁₀H₁₀Mg⁺						
	(C ₅ H ₅) ₂ Mg	1284-72-6	**	8.11 (V)	PE	3688
	(Magnesium, bis(η^5 -2,4-cyclopentadien-1-yl)-)		**	8.0±0.1	EI	3793
C₁₂H₁₄Mg⁺						
	(C ₅ H ₄ CH ₃) ₂ Mg	40672-08-0	**	7.78 (V)	PE	3688
	(Magnesocene, 1,1'-dimethyl-)					
C₃₆H₄₄N₄Mg⁺						
	((C ₂ H ₅) ₂ C ₄ NCH) ₄ Mg	20910-35-4	**	6.19±0.03 (V)	PE	5476
	(Magnesium,[2,3,7,8,12,13,17,18-octaethyl-21H,23H-porphinato(2-)-N ²¹ ,N ²² ,N ²³ ,N ²⁴]-[SP-4-1]-)					
C₄₄H₂₈N₄Mg⁺						
	C ₂₀ H ₈ N ₄ (C ₆ H ₅) ₄ Mg	14640-21-2	**	5.91±0.2	OTH	4962
	(Magnesium, [5,10,15,20-tetraphenyl-21H,23H-porphinato(2-)-N ²¹ ,N ²² ,N ²³ ,N ²⁴]-[SP-4-1]-)		**	6.48 (V)	PE	4557
C₁₀H₁₄O₄Mg⁺						
	(CH ₃ COCHCOCH ₃) ₂ Mg	14024-56-7	**	8.42 (V)	PE	4384
	(Magnesium, bis(2,4-pentanedionato-O,O')-(T-4)-)					
C₅₅H₇₂N₄O₅Mg⁺						
	C ₃₄ H ₃₃ N ₄ O ₃ MgCOOC ₂₀ H ₃₉	42617-16-3	**	6.1±0.2	OTH	5278
	(Chlorophyll a)					
C₁₀H₂O₄F₁₂Mg⁺						
	(CF ₃ COCHCOCF ₃) ₂ Mg	19648-85-2	**	10.28 (V)	PE	4384
	(Magnesium, bis(1,1,1,5,5,5 hexafluoro-2,4-pentanedionato-O,O')-(T-4)-)					
Al⁺						
	Al	7429-90-5	**	6.0±0.3	PE	4860

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
Al⁺						
	Al	7429-90-5	** ** ** ** **	6.0±0.2 6.0±0.3 6.0±1 6.0 6.6±0.6	EI EI EI EI EI	5171 5067 4687 4872 3440
Al²⁺	Al ⁺	14903-36-7	**	18.82873±0.0001	S	5081
Al₂⁺						
	Al ₂	32752-94-6	** **	5.4±1.0 5.4±1.0	EI EI	4005 4014
	Al ₂ O	12004-36-3		15.2±0.5	EI	4005
H₁₂B₃Al⁺						
	Al(BH ₄) ₃	13771-22-7	** **	12.9±0.1 (V) 12.9 (V)	PE PE	4825 4888
C₂Al⁺						
	AlC ₂	37297-57-7	**	9.3±1.0	EI	4014
C₂Al₂⁺						
	Al ₂ C ₂	12122-01-9	**	8.0±0.5	EI	4014
C₃H₉Al⁺						
	(CH ₃) ₃ Al	75-24-1	**	9.76 (V)	PE	4398
C₁₈H₁₅Al⁺						
	(C ₆ H ₅) ₃ Al (Aluminum, triphenyl-)	841-76-9	**	8.53±0.03	PI	4055
OAl⁺						
	AlO	14457-64-8	** ** ** ** ** ** ** ** **	9.5±0.2 9.5±1 9.5 9.53±0.15 9.9±0.5 10.3±1 9±1 10±1	EI EI EI EI EI EI EI EI	5171 3617 4872 3816 4678 4687 3463 3620
	Al ₂ O	12004-36-3		15.1±0.3	EI	4005
O₂Al⁺						
	AlO ₂	11092-32-3	** ** **	10.5±1.0 10±1 10±1	EI EI EI	5171 3463 3617
OAl₂⁺						
	Al ₂ O	12004-36-3	** ** ** ** ** ** ** **	7.7±0.2 7.7±0.5 8.0±0.5 8.1±1 8.20±0.15 8.5±0.2 8.5±1 9±1	EI EI EI EI EI EI EI	4005 3985 4678 4687 3816 5171 3617 3620
O₂Al₂⁺						
	Al ₂ O ₂	12252-63-0	** ** **	9.9±0.5 10.0±1 10±1	EI EI EI	5171 4687 3617

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
FAl⁺	AlF	13595-82-9	** **	9.86±0.05 9	S EI	4229 3606
F₂Al⁺	AlF ₂	13569-23-8	**	10	EI	3606
OFAl⁺	AlOF	13596-12-8	** **	10.5±1 11	EI EI	3462 3606
OF₂Al⁺	AlOF ₂	38344-66-0	**	13±1	EI	3606
C₁₅H₁₂O₆F₉Al⁺	(CF ₃ COCHCOCH ₃) ₃ Al (Aluminum, tris(1,1,1-trifluoro-2,4-pentanedionato-O,O')-)	14354-59-7	**	9.22±0.07 (V)	PE	3682
C₁₅H₃O₆F₁₈Al⁺	(CF ₃ COCHCOOCF ₃) ₃ Al (Aluminum, tris(1,1,1,5,5,5-hexafluoro-2,4-pentanedionato-O,O')-, (OC-6-11)-)	15306-18-0	**	10.33±0.07 (V)	PE	3682
Si⁺	Si	7440-21-3	** ** ** **	8.15172±0.00003 8.1±0.5 8.2±0.5 8.5±0.5	S EI EI EI	4582 3969 4200 3610
	SiH ₄	7803-62-5		13.3	EI	3813
	CH ₃ SiH ₃	992-94-9		12.1	EI	4625
	CH ₃ SiD ₃	1066-43-9		11.8	EI	4625
HSi⁺ (¹ Σ ⁺)	SiH	13774-94-2	**	7.91	OTH	3564
	SiH ₄	7803-62-5		14.7	EI	3813
	CH ₃ SiH ₃	992-94-9		14.8	EI	4625
H₂Si⁺	SiH ₄	7803-62-5	H ₂ H ₂ 2H?	11.8 11.9±0.1 16.2	EI EI EI	3813 5276 3813
	Si ₂ H ₆	1590-87-0	SiH ₄	11.95±0.1	EI	5276
	CH ₃ SiH ₃	992-94-9	CH ₄	11.5±0.1 11.7	EI EI	5276 4625
D₂Si⁺	CH ₃ SiD ₃	1066-43-9		11.6	EI	4625
H₃Si⁺	SiH ₄	7803-62-5	H	12.2	EI	3813
			H	12.3±0.1	EI	5276
	Si ₂ H ₆	1590-87-0	SiH ₃	11.75±0.1	EI	5276
	CH ₃ SiH ₃	992-94-9		12.5	EI	4625
D₃Si⁺	CH ₃ SiD ₃	1066-43-9		12.4	EI	4625
H₄Si⁺	SiH ₄	7803-62-5	** ** **	12.3 (V) 11.60 11.7	PE PE PE	4972 3716 5276

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
H₅Si₂⁺	Si ₂ H ₆	1590-87-0	H	11.4±0.1	EI	5276
H₆Si₂⁺	Si ₂ H ₆	1590-87-0	** ** **	10.0 10.53 (V) 10.53 (V)	PE PE PE	5276 4160 4558
H₆Si₃⁺	Si ₃ H ₈	7783-26-8	**	9.87 (V)	PE	4558
H₁₀Si₄⁺	n-Si ₄ H ₁₀	7783-29-1	**	9.62 (V)	PE	4558
H₁₂Si₅⁺	n-Si ₅ H ₁₂	14868-53-2	**	9.36 (V)	PE	4558
H₁₁B₅Si⁺	B ₅ H ₉ (SiH ₃) (Pentaborane(9), 2,3-μ-silyl-)	22044-27-5	**	10.17 (V)	PE	4519
H₁₁B₅Si⁺	B ₅ H ₉ (SiH ₃) (Pentaborane(9), 1-silyl-) B ₅ H ₉ (SiH ₃) (Pentaborane(9), 2-silyl-)	28556-29-8 22142-52-5	** **	10.40 (V) 10.42 (V)	PE PE	4519 4519
C₂Si⁺	SiC ₂	12071-27-1	** **	10.1±0.5 10.3±0.5	EI EI	4005 3969
CSi₂⁺	Si ₂ C Si ₂ C	XXXXX-XX-X XXXXX-XX-X	** **	9.0±0.5 9.5±0.5	EI EI	4005 3969
CH₃Si⁺	CH ₃ SiH ₃ CH ₃ SiD ₃ CH ₂ =CHSi(CH ₃) ₃	992-94-9 1066-43-9 754-05-2		12.8 12.1 15	EI EI EI	4625 4625 3809
CH₂DSi⁺	CH ₃ SiD ₃	1066-43-9		11.4	EI	4625
CH₃Si⁺	CH ₃ SiH ₃ (CH ₃) ₂ SiH ₂	992-94-9 1111-74-6	H ₂ CH ₄	11.3 11.4±0.1 11.1±0.1	EI EI EI	4625 5276 5276
CH₃DSi⁺	CH ₃ SiD ₃	1066-43-9	2D	11.5	EI	4625
CH₂D₂Si⁺	CH ₃ SiD ₃	1066-43-9	H,D	11.4	EI	4625
CH₃Si⁺	CH ₃ SiH ₃ (CH ₃) ₂ SiH ₂ CH ₂ =CHSi(CH ₃) ₃ ((CH ₃) ₂ H ₂ Si) ₂	992-94-9 1111-74-6 754-05-2 870-26-8	H H CH ₃ CH ₃ SiH ₂	11.8±0.1 11.8 11.5±0.1 15 11.4±0.1	EI EI EI EI EI	5276 4625 5276 3809 5276

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
CH₃D₂Si⁺	CH ₃ SiD ₃	1066-43-9	D	11.8	EI	4625
CH₆Si⁺	CH ₃ SiH ₃	992-94-9	** **	10.7 11.6 (V)	PE PE	5276 4972
C₂H₄Si⁺	CH=CSiH ₃	1066-27-9	**	10.73 (V)	PE	4160
C₂H₆Si⁺	CH ₂ =CHSiH ₃ (CH ₃) ₂ SiH ₂ (CH ₃) ₃ SiH (CH ₃) ₂ HSiSiH ₂ CH ₃ ((CH ₃) ₂ HSi) ₂	7291-09-0 1111-74-6 993-07-7 814-74-4 814-98-2	** ** H ₂ CH ₄ CH ₃ SiH ₃ (CH ₃) ₂ SiH ₂	10.37 (V) 10.4 (V) 10.7±0.1 10.5±0.1 10.75±0.1 10.7±0.1	PE PE EI EI EI EI	3950 3940 5276 5276 5276 5276
C₂H₇Si⁺	(CH ₃) ₂ SiH ₂ (CH ₃) ₃ SiH CH ₂ =CHSi(CH ₃) ₃ ((CH ₃) ₂ HSi) ₂ (CH ₃) ₂ HSiSiH ₂ CH ₃ ((CH ₃) ₂ HSi) ₂	1111-74-6 993-07-7 754-05-2 870-26-8 814-74-4 814-98-2	H CH ₃ 13 SiH ₃ CH ₃ SiH ₂ (CH ₃) ₂ SiH	11.1±0.1 10.9±0.1 13 10.3±0.1 10.75±0.1 10.8±0.1	EI EI EI EI EI EI	5276 5276 3809 5276 5276 5276
C₂H₈Si⁺	(CH ₃) ₂ SiH ₂	1111-74-6	** **	10.3 11.2 (V)	PE PE	5276 4972
C₃H₈Si⁺	(CH ₃) ₂ Si=CH ₂ CH ₂ =CHCH ₂ SiH ₃ C ₃ H ₈ Si (Silacyclobutane) CH ₂ =CHSi(CH ₃) ₃ C ₃ H ₈ Si(CH ₃) ₂ (Silacyclobutane, 1,1-dimethyl-)	4112-23-6 18191-59-8 287-29-6 754-05-2 2295-12-7	** ** ** C ₂ H ₄ C ₂ H ₄	7.5±0.3 9.49 (V) 10.05 (V) 10 9.61	OTH PE PE EI PI	5287 3950 4077 3809 5287
C₃H₉Si⁺	(CH ₃) ₃ SiH (CH ₃) ₄ Si CH ₂ =CHSi(CH ₃) ₃ (CH ₃) ₃ SiC ₂ H ₅ (CH ₃) ₂ HSiSiH ₂ CH ₃ (CH ₃) ₃ SiSiH ₃ ((CH ₃) ₂ HSi) ₂ (CH ₃) ₃ SiSiH(CH ₃) ₂ (CH ₃) ₆ Si ₂ (CH ₃) ₆ Si ₂ C ₆ H ₅ Si ₂ (CH ₃) ₅ (Disilane, pentamethylphenyl-) (C ₆ H ₅) ₂ SiCH ₃ Si(CH ₃) ₃ (Disilane, 1,1,1,2-tetramethyl-2,2-diphenyl-) (C ₆ H ₅) ₂ Si(CH ₃) ₂ (Disilane, 1,1,2,2-tetramethyl-1,2-diphenyl-) (C ₆ H ₅) ₂ SiSi(CH ₃) ₃ (Disilane, 1,1,1-trimethyl-2,2,2-triphenyl-)	993-07-7 75-76-3 754-05-2 3439-38-1 814-74-4 18365-32-7 814-98-2 812-15-7 1450-14-2 1450-14-2 1130-17-2 1450-16-4 1145-98-8 1450-18-6	H CH ₃ CH ₃ CH ₃ CH ₃ C ₂ H ₃ C ₂ H ₅ SiH ₃ SiH ₃ CH ₃ SiH ₂ (CH ₃) ₂ SiH (CH ₃) ₂ SiH (CH ₃) ₃ Si (CH ₃) ₃ Si (CH ₃) ₃ Si(CH ₃) ₂ 9.9 10.22±0.18 10.08±0.09 10.59±0.03 11.04±0.03 10.83±0.09	10.5±0.1 10.03±0.04 10.25±0.1 10.53±0.20 10.63±0.05 11 10.0±0.1 9.8±0.1 9.7±0.1 10.1±0.1 10.0 10.2±0.1 9.9 10.22±0.18 10.08±0.09 10.59±0.03 11.04±0.03 10.83±0.09	EI PI EI EI EI EI EI EI EI EI EI EI EI EI EI EI EI EI EI EI EI	5276 4907 5276 3548 4126 3809 5276 5276 5276 5276 5276 5276 5276 5276 3548 3549 5276 3548 3549 3549 3549 3549 3549 3549 3549

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_3\text{H}_9\text{Si}^+$						
	$(\text{CH}_3)_2\text{NCH}_2\text{Si}(\text{CH}_3)_3$	18182-40-6	$\text{C}_3\text{H}_8\text{N}$	9.76	PI	5543
	$(\text{CH}_3)_3\text{SiOSi}(\text{CH}_3)_3$	107-46-0		15.4 ± 0.2	EI	3444
	$(\text{CH}_3)_3\text{SiOSi}(\text{CH}_3)_2\text{OSi}(\text{CH}_3)_3$	107-51-7		15.8 ± 0.2	EI	3444
	$(\text{CH}_3)_3\text{SiOSi}(\text{CH}_3)(\text{C}_2\text{H}_5)\text{OSi}(\text{CH}_3)_3$	5356-85-4		15.4 ± 0.2	EI	3444
	$(\text{CH}_3)_3\text{SiOSi}(\text{CH}_3)(\text{C}_2\text{H}_5)_2\text{OSi}(\text{CH}_3)_3$	17861-60-8		15.3 ± 0.2	EI	3444
	$\text{C}_6\text{H}_5\text{SSi}(\text{CH}_3)_3$	4551-15-9		10.18 ± 0.1	EI	4198
	(Silane, trimethyl(phenylthio)-)					
	$(\text{CH}_3)_3\text{SiCl}$	30687-62-8	Cl	11.6 ± 0.1	EI	5276
	$((\text{CH}_3)_3\text{Si})(\text{CO})_3\text{Mn}$	26500-16-3		9.81 ± 0.11	EI	5321
	$(\text{CH}_3)_3\text{SiGe}(\text{CH}_3)_3$	31608-80-7	$(\text{CH}_3)_3\text{Ge}$	10.19 ± 0.12	EI	3548
	$((\text{CH}_3)_3\text{Si})(\text{CH}_3)_3\text{Sn}$	16393-88-7	$(\text{CH}_3)_3\text{Sn}$	10.18 ± 0.26	EI	3548
$\text{C}_3\text{H}_{10}\text{Si}^+$						
	$(\text{CH}_3)_3\text{SiH}$	993-07-7	**	9.9	PE	5276
			**	10.8 (V)	PE	4972
$\text{C}_4\text{H}_7\text{Si}^+$						
	$(\text{CH}_3)_3\text{SiC}\equiv\text{CH}$	1066-54-2	CH_3	10.79 ± 0.04	EI	4126
$\text{C}_4\text{H}_8\text{Si}^+$						
	$\text{C}_4\text{H}_8\text{Si}$ (Silacyclopent-3-ene)	XXXXX-XX-X **		9.21 (V)	PE	4517
$\text{C}_4\text{H}_9\text{Si}^+$						
	$\text{CH}_2=\text{CHSi}(\text{CH}_3)_3$	754-05-2	CH_3	.9	EI	3809
$\text{C}_4\text{H}_{10}\text{Si}^+$						
	$(\text{C}_2\text{H}_5)_3\text{SiH}_2$	542-91-6	H_2	10.0 ± 0.1	EI	5276
	$(\text{C}_2\text{H}_5)_3\text{SiH}$	617-86-7	C_2H_6	9.75 ± 0.1	EI	5276
$\text{C}_4\text{H}_{12}\text{Si}^+$						
	$(\text{CH}_3)_4\text{Si}$	75-76-3	**	9.80 ± 0.03	PI	4907
			**	9.42 ± 0.1	PE	3677
			**	9.6	PE	5276
			**	9.79 ± 0.04	PE	3880
			**	10.4 (V)	PE	4972
			**	10.57 (V)	PE	5368
$(^2\text{A}_1)$			**	15.62 (V)	PE	3503
			**	9.85 ± 0.16	EI	3548
			**	9.99 ± 0.03	EI	4126
	$(\text{C}_2\text{H}_5)_2\text{SiH}_2$	542-91-6	**	9.8	PE	5276
			**	10.3 (V)	PE	4972
$\text{C}_5\text{H}_5\text{Si}^+$						
	$(\text{CH}_3)_2\text{Si}(\text{C}\equiv\text{CH})_2$	1675-60-1	CH_3	12.05 ± 0.05	EI	4126
$\text{C}_5\text{H}_6\text{Si}^+$						
	$\text{C}_5\text{H}_5\text{SiH}$ (Silabenzene)	289-77-0	**	8.0 (V)	PE	5107
$\text{C}_5\text{H}_8\text{Si}^+$						
	$\text{C}_5\text{H}_5(\text{SiH}_3)$ (Silane, 2,4-cyclopentadien-1-yl-)	33618-25-6	**	8.7 (V)	PE	4373
			**	8.7 (V)	PE	4179
$\text{C}_5\text{H}_{10}\text{Si}^+$						
	$(\text{CH}_3)_3\text{SiC}\equiv\text{CH}$	1066-54-2	**	9.9 ± 0.1	PE	4002
			**	10.40 ± 0.02	EI	4126
$\text{C}_5\text{H}_{12}\text{Si}^+$						
	$(\text{CH}_3)_3\text{SiCH}=\text{CH}_2$	754-05-2	**	9.8 (V)	PE	3908

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_5\text{H}_{12}\text{Si}^+$	$(\text{CH}_3)_3\text{SiCH}=\text{CH}_2$	754-05-2	**	9.8 (V)	PE	3940
			**	9.2	EI	3809
	$\text{C}_3\text{H}_9\text{Si}(\text{CH}_3)_2$ (Silacyclobutane, 1,1-dimethyl-)	2295-12-7	**	9.40 (V)	PE	4077
$\text{C}_5\text{H}_{14}\text{Si}^+$	$(\text{CH}_3)_3\text{SiC}_2\text{H}_5$	3439-38-1	**	9.6	PE	5276
$\text{C}_6\text{H}_3\text{Si}^+$	$\text{CH}_3\text{Si}(\text{C}\equiv\text{CH})_3$	1849-39-4	CH_3	12.81 ± 0.07	EI	4126
$\text{C}_6\text{H}_8\text{Si}^+$	$(\text{CH}_3)_2\text{Si}(\text{C}\equiv\text{CH})_2$	1675-60-1	**	10.85 ± 0.10	EI	4126
	$\text{C}_5\text{H}_5\text{SiCH}_3$ (Silabenzene, 1-methyl-)	63878-65-9	**	7.7 (V)	PE	5216
	$\text{C}_6\text{H}_5\text{SiH}_3$ (Silane, phenyl-)	694-53-1	**	9.09	PE	3868
			**	9.25	PE	3922
$\text{C}_6\text{H}_{12}\text{Si}^+$	$(\text{C}_2\text{H}_3)_2\text{Si}(\text{CH}_3)_2$	10519-87-6	**	9.8 (V)	PE	3994
			**	9.8 (V)	PE	5089
	$\text{C}_6\text{H}_{12}\text{Si}$ (Silacyclopent-2-ene, 1,1-dimethyl-)	18187-50-3	**	9.27 ± 0.03 (V)	PE	5389
	$\text{C}_4\text{H}_9\text{Si}(\text{CH}_3)_2$ (Silacyclopent-3-ene, 1,1-dimethyl-)	16054-12-9	**	9.0 (V)	PE	5550
			**	9.1 ± 0.03 (V)	PE	5389
$\text{C}_6\text{H}_{14}\text{Si}^+$	$(\text{CH}_3)_3\text{SiCH}_2\text{CH}=\text{CH}_2$	762-72-1	**	9.0 (V)	PE	3908
			**	9.0 (V)	PE	3940
	$\text{C}_5\text{H}_5\text{Si}(\text{CH}_3)_3$ (Silacyclobutane, 1,1,2-trimethyl-)	30681-90-4	**	9.20 (V)	PE	4077
	$\text{C}_4\text{H}_9\text{Si}(\text{CH}_3)_2$ (Silacyclopentane, 1,1-dimethyl-)	1072-54-4	**	9.75 (V)	PE	4077
$\text{C}_6\text{H}_{15}\text{Si}^+$	$(\text{C}_2\text{H}_5)_3\text{SiH}$	617-86-7	H	10.4 ± 0.1	EI	5276
	$(\text{C}_2\text{H}_5)_4\text{Si}$	631-36-7	C_2H_5	10.0 ± 0.1	EI	5276
$\text{C}_6\text{H}_{16}\text{Si}^+$	$(\text{C}_2\text{H}_5)_3\text{SiH}$	617-86-7	**	9.5	PE	5276
			**	9.9 (V)	PE	4985
			**	10.0 (V)	PE	4972
$\text{C}_7\text{H}_6\text{Si}^+$	$\text{CH}_3\text{Si}(\text{C}\equiv\text{CH})_3$	1849-39-4	**	11.06 ± 0.03	EI	4126
$\text{C}_7\text{H}_9\text{Si}^+$	$\text{C}_6\text{H}_5\text{SiH}(\text{CH}_3)_2$ (Silane, dimethylphenyl-)	766-77-8	CH_3	8.72	EI	4125
$\text{C}_8\text{H}_4\text{Si}^+$	$\text{Si}(\text{C}\equiv\text{CH})_4$	1849-38-3	**	11.34	EI	4126
$\text{C}_8\text{H}_{11}\text{Si}^+$	$\text{C}_6\text{H}_5\text{Si}(\text{CH}_3)_2\text{H}$ (Silane, dimethylphenyl-)	766-77-8	H	10.43 ± 0.04	EI	3549
	$\text{C}_6\text{H}_5(\text{CH}_3)\text{SiH}(\text{CH}_3)_2$ (Silane, dimethyl(4-methylphenyl)-)	1432-39-9	CH_3	8.34	EI	4125
	$\text{C}_6\text{H}_5\text{Si}(\text{CH}_3)_3$ (Silane, trimethylphenyl-)	768-32-1	CH_3	10.26 ± 0.03	EI	3549

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₈H₁₁Si⁺	C ₆ H ₅ Si ₂ (CH ₃) ₅ (Disilane, pentamethylphenyl-) (C ₆ H ₅) ₂ SiCH ₃ Si(CH ₃) ₃ (Disilane, 1,1,1,2-tetramethyl-2,2-diphenyl-) (C ₆ H ₅ (CH ₃) ₂ Si) ₂ (Disilane, 1,1,2,2-tetramethyl-1,2-diphenyl-) (C ₆ H ₅) ₃ SiSi(CH ₃) ₃ (Disilane, 1,1,1-trimethyl-2,2,2-triphenyl-)	1130-17-2 1450-16-4 1145-98-8 1450-18-6	Si(CH ₃) ₃ C ₆ H ₅ Si(CH ₃) ₂ C ₆ H ₅ Si(CH ₃) ₂ (C ₆ H ₅) ₂ SiCH ₃	9.86±0.06 9.75±0.04 9.87±0.08 10.13±0.03	EI EI EI EI	3549 3549 3549 3549
C₈H₁₂Si⁺	(C ₂ H ₅) ₄ Si C ₆ H ₅ Si(CH ₃) ₂ H (Silane, dimethylphenyl-)	1112-55-6 766-77-8	** **	9.7 (V) 8.92±0.15	PE EI	3994 3549
C₈H₁₄Si⁺	C ₅ H ₅ (Si(CH ₃) ₃) (Silane, 2,4-cyclopentadien-1-yltrimethyl-)	3559-74-8	**	8.30 (V)	PE	5535
C₈H₂₀Si⁺	(C ₂ H ₅) ₄ Si	631-36-7	** **	8.9 9.8 (V)	PE PE	5276 4985
C₉H₁₄Si⁺	C ₆ H ₅ Si(CH ₃) ₃ (Silane, trimethylphenyl-)	768-32-1	** ** ** **	9.0 (V) 9.05 (V) 8.81±0.15 8.79	PE PE EI CTS	5380 4280 3549 3922
C₉H₂₂Si⁺	(iso-C ₃ H ₇) ₃ SiH	6485-79-6	**	9.5 (V)	PE	4985
C₁₀H₁₀Si⁺	C ₁₀ H ₈ SiH ₃ (Silane, 1-naphthalenyl-)	38274-75-8	**	8.12	CTS	3922
C₁₀H₁₄Si⁺	C ₈ H ₈ Si(CH ₃) ₂ (1-Silaindan, 1,1-dimethyl-) C ₈ H ₈ Si(CH ₃) ₂ (1H-2-Silaindene, 2,3-dihydro-2,2-dimethyl-)	17158-48-4 2764-87-6	** **	8.54 8.41	CTS CTS	3546 3546
C₁₀H₁₆Si⁺	C ₆ H ₅ CH ₂ Si(CH ₃) ₃ (Silane, trimethyl(phenylmethyl))	770-09-2	** ** ** ** **	8.35 8.4 8.42 (V) 8.27 8.37	PE PE PE CTS CTS	5574 4589 4280 3922 3546
C₁₁H₁₆Si⁺	C ₆ H ₅ CH=CHSi(CH ₃) ₃ (Silane, trimethyl(2-phenylethenyl)-, (E)-) C ₆ H ₅ CH=CHSi(CH ₃) ₃ (Silane, trimethyl(2-phenylethenyl)-, (Z)-) C ₆ H ₅ C(Si(CH ₃) ₃)=CH ₂ (Silane, trimethyl(1-phenylethenyl)-)	19372-00-0 19319-11-0 1923-01-9	** ** **	7.89±0.04 8.19±0.04 8.23±0.04	EI EI EI	4097 4097 4097
C₁₂H₁₂Si⁺	(C ₆ H ₅) ₂ SiH ₂ (Silane, diphenyl-)	775-12-2	**	9.23±0.05 (V)	PE	4620

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{12}\text{H}_{16}\text{Si}^+$	$\text{C}_9\text{H}_7\text{Si}(\text{CH}_3)_3$ (Silane, 1 <i>H</i> -inden-1-yltrimethyl-))	18053-75-3	**	7.65±0.01	EI	3805
$\text{C}_{12}\text{H}_{18}\text{Si}^+$	$\text{C}_9\text{H}_7\text{Si}(\text{CH}_3)_3$ (Silane, (2,3-dihydro-1 <i>H</i> -inden-1-yl)trimethyl-))	18036-88-9	**	7.87±0.01	EI	3805
			**	8.13	CTS	3546
	$\text{C}_6\text{H}_5\text{CH}=\text{CHCH}_2\text{Si}(\text{CH}_3)_3$ (Silane, trimethyl(3-phenyl-2-propenyl)-, (E)-)	40595-34-4	**	7.61±0.04	EI	4097
	$\text{C}_6\text{H}_5\text{CH}=\text{CHCH}_2\text{Si}(\text{CH}_3)_3$ (Silane, trimethyl(3-phenyl-2-propenyl)-, (Z)-)	40595-35-5	**	7.77±0.04	EI	4097
$\text{C}_{13}\text{H}_{13}\text{Si}^+$	$(\text{C}_6\text{H}_5)_2\text{Si}(\text{CH}_3)\text{H}$ (Silane, methyldiphenyl-)	776-76-1	H	10.97±0.12	EI	3549
	$(\text{C}_6\text{H}_5)_2\text{SiCH}_3\text{Si}(\text{CH}_3)_3$ (Disilane, 1,1,1,2-tetramethyl-2,2-diphenyl-)	1450-16-4	$(\text{CH}_3)_3\text{Si}$	9.63±0.02	EI	3549
	$(\text{C}_6\text{H}_5(\text{CH}_3)_2\text{Si})_2$ (Disilane, 1,1,2,2-tetramethyl-1,2-diphenyl-)	1145-98-8	$(\text{CH}_3)_3\text{Si}$	9.60±0.02	EI	3549
	$((\text{C}_6\text{H}_5)_2\text{CH}_3\text{Si})_2$ (Disilane, 1,2-dimethyl-1,1,2,2-tetraphenyl-)	1172-76-5	$(\text{C}_6\text{H}_5)_2\text{SiCH}_3$	9.51±0.05	EI	3549
$\text{C}_{13}\text{H}_{14}\text{Si}^+$	$(\text{C}_6\text{H}_5)_2\text{Si}(\text{CH}_3)\text{H}$ (Silane, methyldiphenyl-)	776-76-1	**	8.75±0.15	EI	3549
$\text{C}_{13}\text{H}_{16}\text{Si}^+$	$\text{C}_{10}\text{H}_7\text{Si}(\text{CH}_3)_3$ (Silane, trimethyl-1-naphthalenyl-)	18052-80-7	**	8.03	CTS	3758
$\text{C}_{14}\text{H}_{14}\text{Si}^+$	$\text{C}_{12}\text{H}_8\text{Si}(\text{CH}_3)_2$ (5,5-Dibenzosilole, 5,5-dimethyl-)	13688-68-1	**	7.9 (V)	PE	4081
$\text{C}_{14}\text{H}_{18}\text{Si}^+$	$\text{C}_{10}\text{H}_7\text{CH}_2\text{Si}(\text{CH}_3)_3$ (Silane, trimethyl(1-naphthalenylmethyl)-)	18410-58-7	**	7.83	CTS	3758
			**	7.83	CTS	3922
$\text{C}_{17}\text{H}_{18}\text{Si}^+$	$\text{C}_9\text{H}_7\text{Si}(\text{CH}_3)_2\text{C}_6\text{H}_5$ (Silane, 1 <i>H</i> -inden-1-ylidemethylphenyl-)	27490-90-0	**	7.69±0.04	EI	3805
$\text{C}_{17}\text{H}_{20}\text{Si}^+$	$\text{C}_9\text{H}_7\text{Si}(\text{CH}_3)_2\text{C}_6\text{H}_5$ (Silane, (2,3-dihydro-1 <i>H</i> -inden-1-yl)dimethylphenyl-)	41273-54-5	**	7.94±0.01	EI	3805
$\text{C}_{18}\text{H}_{15}\text{Si}^+$	$(\text{C}_6\text{H}_5)_3\text{SiH}$ (Silane, triphenyl-)	789-25-3	H	9.58±0.08	EI	3549
	$(\text{C}_6\text{H}_5)_4\text{Si}$ (Silane, tetraphenyl-)	1048-08-4	C_6H_5	9.7	PI	4055
	$(\text{C}_6\text{H}_5)_3\text{SiSi}(\text{CH}_3)_3$ (Disilane, 1,1,1-trimethyl-2,2,2-triphenyl-)	1450-18-6	C_6H_5	9.93±0.08	EI	3549
			$(\text{CH}_3)_3\text{Si}$	9.35±0.03	EI	3549
	$((\text{C}_6\text{H}_5)_2\text{CH}_3\text{Si})_2$ (Disilane, 1,2-dimethyl-1,1,2,2-tetraphenyl-)	1172-76-5	$\text{C}_6\text{H}_5\text{Si}(\text{CH}_3)_2$	9.35±0.03	EI	3549
	$((\text{C}_6\text{H}_5)_3\text{Si})_2$ (Disilane, hexaphenyl-)	1450-23-3	$(\text{C}_6\text{H}_5)_3\text{Si}$	9.61±0.09	EI	3549

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{18}\text{H}_{16}\text{Si}^+$	$(\text{C}_6\text{H}_5)_3\text{SiH}$ (Silane, triphenyl-)	789-25-3	**	9.13 ± 0.05 (V)	PE	4620
			**	8.80 ± 0.15	EI	3549
$\text{C}_{22}\text{H}_{20}\text{Si}^+$	$\text{C}_{10}\text{H}_7\text{Si}(\text{CH}_3)_2\text{C}_{10}\text{H}_7$ (Silane, dimethyl-di-1-naphthalenyl-)	18753-19-0	**	8.03	CTS	3758
$\text{C}_{24}\text{H}_{16}\text{Si}^+$	$\text{C}_{24}\text{H}_{16}\text{Si}$ (5,5'-Spirobi[5H-dibenzosilole])	159-68-2	**	7.85 (V)	PE	4081
$\text{C}_{24}\text{H}_{20}\text{Si}^+$	$(\text{C}_6\text{H}_5)_4\text{Si}$ (Silane, tetraphenyl-)	1048-08-4	**	8.50 ± 0.03	PI	4055
			**	8.65 ± 0.15	EI	3549
$\text{C}_2\text{H}_6\text{Si}_2^+$	$\text{SiH}_3\text{C}\equiv\text{CSiH}_3$	XXXXX-XX-X	**	10.46 (V)	PE	4160
$\text{C}_6\text{H}_{18}\text{Si}_2^+$	$(\text{CH}_3)_6\text{Si}_2$	1450-14-2	**	8.0	PE	5276
			**	8.69 (V)	PE	3504
			**	8.35 ± 0.12	EI	3548
			**	8.46 ± 0.15	EI	3549
$\text{C}_7\text{H}_{20}\text{Si}_2^+$	$((\text{CH}_3)_3\text{Si})_2\text{CH}_2$	2117-28-4	**	9.5 (V)	PE	4457
$\text{C}_8\text{H}_{20}\text{Si}_2^+$	$\text{C}_8\text{H}_{20}\text{Si}_2$	18178-59-1	**	9.19 (V)	PE	4715
$\text{C}_8\text{H}_{22}\text{Si}_2^+$	$(\text{CH}_3)_3\text{SiCH}_2\text{CH}_2\text{Si}(\text{CH}_3)_3$	6231-76-1	**	8.78 (V)	PE	4457
$\text{C}_9\text{H}_{24}\text{Si}_2^+$	$(\text{CH}_3)_3\text{Si}(\text{CH}_2)_3\text{Si}(\text{CH}_3)_3$	2295-05-8	**	9.41 (V)	PE	4457
$\text{C}_{10}\text{H}_{18}\text{Si}_2^+$	$((\text{CH}_3)_3\text{SiCC})_2$	4526-07-2	**	8.85 (V)	PE	5332
$\text{C}_{10}\text{H}_{22}\text{Si}_2^+$	$\text{C}_{10}\text{H}_{22}\text{Si}_2$ $\text{CH}_2=\text{C}(\text{Si}((\text{CH}_3)_3)\text{C}(\text{Si}((\text{CH}_3)_3)=\text{CH}_2$ $\text{C}_{10}\text{H}_{22}\text{Si}_2$ <i>trans,trans</i> - $((\text{CH}_3)_3\text{SiCH}=\text{CH})_2$	18081-31-7 22472-36-2 22500-95-4 22430-47-3	** ** ** **	8.45 ± 0.04 8.65 ± 0.04 8.45 ± 0.04 8.43 ± 0.04	EI EI EI EI	4274 4274 4274 4274
$\text{C}_{10}\text{H}_{24}\text{Si}_2^+$	$\text{C}_{10}\text{H}_{24}\text{Si}_2$	XXXXX-XX-X	**	8.30 (V)	PE	5535
$\text{C}_{11}\text{H}_{20}\text{Si}_2^+$	$\text{C}_6\text{H}_5\text{Si}_2(\text{CH}_3)_5$ (Disilane, pentamethylphenyl-)	1130-17-2	** ** **	8.35 (V) 8.35 ± 0.15 8.37	PE EI CTS	3946 3549 3946
	$\text{C}_6\text{H}_4(\text{SiH}(\text{CH}_3)_2)\text{Si}(\text{CH}_3)_3$ (Silane,[4-(dimethylsilyl)phenyl]trimethyl-)	27856-24-2	**	8.4 ± 0.2	EI	4121
$\text{C}_{11}\text{H}_{22}\text{Si}_2^+$	$\text{C}_5\text{H}_4(\text{Si}((\text{CH}_3)_3)_2$ (Silane,2,4-cyclopentadien-1-ylidenebis(trimethyl-))	33630-76-1	**	8.05 (V)	PE	5535

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{12}\text{H}_{10}\text{Si}_2^+$	$\text{C}_8\text{H}_9\text{Si}(\text{CH}_3)_3\text{Si}(\text{CH}_3)_3$ (2-Silaindan, 2-methyl-2-(trimethylsilyl)-)	27490-20-6	**	8.37	CTS	3546
$\text{C}_{12}\text{H}_{22}\text{Si}_2^+$	$\text{C}_6\text{H}_5\text{CH}_2\text{Si}_2(\text{CH}_3)_5$ (Disilane, pentamethyl(phenylmethyl)-) $\text{C}_6\text{H}_5(\text{Si}(\text{CH}_3)_3)_2$ (Silane, 1,4-phenylenebis(trimethyl-))	3098-82-6 13183-70-5	**	8.27 8.98 (V)	CTS PE	3546 5380
$\text{C}_{12}\text{H}_{24}\text{Si}_2^+$	$\text{C}_6\text{H}_5(\text{Si}(\text{CH}_3)_3)_2$ (Silane, 2,5-cyclohexadiene-1,4-diylbis(trimethyl-trans-))	54380-47-1	**	7.70 (V)	PE	5535
$\text{C}_{12}\text{H}_{28}\text{Si}_2^+$	$\text{C}_{12}\text{H}_{28}\text{Si}_2$ (Silane, 2,3-dimethyl-2-butene-1,4-diyl)bis(trimethyl-trans-))	XXXXXX-XX-X	**	7.70 (V)	PE	5535
$\text{C}_{12}\text{H}_{30}\text{Si}_2^+$	(tert- $\text{C}_4\text{H}_9\text{Si}(\text{CH}_3)_3)_2$	63262-93-1	**	8.52 (V)	PE	4683
$\text{C}_{13}\text{H}_{22}\text{Si}_2^+$	$\text{C}_6\text{H}_5\text{CH}=\text{CHSi}_2(\text{CH}_3)_5$ (Disilane, pentamethyl(2-phenylethenyl)-, (E)-)	40595-36-6	**	7.73±0.04	EI	4097
$\text{C}_{13}\text{H}_{24}\text{Si}_2^+$	$\text{C}_6\text{H}_5\text{CH}(\text{Si}(\text{CH}_3)_3)_2$ (Silane,(phenylmethylene)bis(trimethyl-))	14595-77-8	** **	8.10 8.10 (V)	PE PE	5574 5012
$\text{C}_{14}\text{H}_{22}\text{Si}_2^+$	$\text{C}_{14}\text{H}_{22}\text{Si}_2$ (2,6-Disila-s-indacene,1,2,3,5,6,7-hexahydro-2,2,6,6-tetramethyl-)	69020-20-2	**	7.80 (V)	PE	5629
$\text{C}_{14}\text{H}_{24}\text{Si}_2^+$	$\text{C}_9\text{H}_9\text{Si}_2(\text{CH}_3)_5$ (Disilane, 1-indanylpentamethyl-) $\text{C}_6\text{H}_5\text{CH}=\text{C}(\text{Si}(\text{CH}_3)_3)_2$ (Silane, (phenylethylenylidene)bis(trimethyl-))	27490-23-9 18415-23-1	** **	8.07 8.12±0.04	CTS EI	3546 4097
$\text{C}_{14}\text{H}_{26}\text{Si}_2^+$	$\text{C}_6\text{H}_4(\text{CH}_2\text{Si}(\text{CH}_3)_3)_2$ (Silane, [1,2-phenylenebis(methylene)bis(trimethyl-)])	18412-14-1	** **	8.05 (V) 8.05 (V)	PE PE	5012 5629
	$\text{C}_{14}\text{H}_{26}\text{Si}_2$ (Silane,[1,3-phenylenebis(methylene)]bis(trimethyl-))	18412-15-2	**	8.05 (V)	PE	5629
	$\text{C}_6\text{H}_4(\text{CH}_2\text{Si}(\text{CH}_3)_3)_2$ (Silane,[1,4-phenylenebis(methylene)]bis(trimethyl-))	17557-09-4	** **	8.10 (V) 7.75	PE PE	5012 5574
	$\text{C}_{14}\text{H}_{32}\text{Si}_2^+$	XXXXXX-XX-X	**	7.75 (V)	PE	5012
$\text{C}_{14}\text{H}_{32}\text{Si}_2^+$	$\text{C}_{14}\text{H}_{32}\text{Si}_2$	XXXXXX-XX-X	**	7.90 (V)	PE	5535
$\text{C}_{15}\text{H}_{22}\text{Si}_2^+$	$\text{C}_{10}\text{H}_7\text{Si}_2(\text{CH}_3)_5$ (Disilane, pentamethyl-1-naphthalenyl-)	38446-40-1	**	7.95	CTS	3758
$\text{C}_{15}\text{H}_{24}\text{Si}_2^+$	$\text{C}_6\text{H}_5(\text{Si}(\text{CH}_3)_3)_2$ (Silane, 1 <i>H</i> -indene-1,2-diylbis(trimethyl-))	26205-36-7	**	7.54±0.01	EI	3805

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{16}\text{H}_{22}\text{Si}_2^+$	$(\text{C}_6\text{H}_5)_2\text{SiCH}_3\text{Si}(\text{CH}_3)_3$ (Disilane, 1,1,1,2-tetramethyl-2,2-diphenyl-) $(\text{C}_6\text{H}_5(\text{CH}_3)_2\text{Si})_2$ (Disilane, 1,1,2,2-tetramethyl-1,2-diphenyl-)	1450-16-4 1145-98-8	**	8.38 ± 0.15 8.11 ± 0.15	EI EI	3549 3549
$\text{C}_{16}\text{H}_{30}\text{Si}_2^+$	$\text{C}_{16}\text{H}_{30}\text{Si}_2$ (Silane,[2,5-dimethyl-1,4-phenylene]bis(methylene)]bis(trimethyl-) $\text{C}_{16}\text{H}_{30}\text{Si}_2$ (Silane,[4,6-dimethyl-1,3-phenylene]bis(methylene)]bis(trimethyl-)	69020-19-5 62347-03-9	**	7.70 (V) 7.95 (V)	PE PE	5629 5629
$\text{C}_{18}\text{H}_{34}\text{Si}_2^+$	$\text{C}_{18}\text{H}_{34}\text{Si}_2$ (Silane,[2,3,5,6-tetramethyl-1,4-phenylene]bis(methylene)]bis(trimethyl-)	69020-17-3	**	7.25 (V)	PE	5629
$\text{C}_{20}\text{H}_{30}\text{Si}_2^+$	$(\text{C}_6\text{H}_4(\text{CH}_2\text{Si}(\text{CH}_3)_3)_2$ (Silane, [1,1'-biphenyl]-4,4'-diylbis(methylene)] bis(trimethyl-)	61342-05-0	**	7.60 (V)	PE	5012
$\text{C}_{20}\text{H}_{38}\text{Si}_2^+$	$\text{C}_6\text{H}_4(\text{CH}_2\text{Si}(\text{C}_2\text{H}_5)_3)_2$ (Silane,[1,4-phenylene]bis(methylene)]bis[triethyl-])	18724-34-0	**	7.75	PE	5574
$\text{C}_{21}\text{H}_{24}\text{Si}_2^+$	$(\text{C}_6\text{H}_5)_3\text{SiSi}(\text{CH}_3)_3$ (Disilane, 1,1,1-trimethyl-2,2,2-triphenyl-)	1450-18-6	**	8.30 ± 0.15	EI	3549
$\text{C}_{24}\text{H}_{26}\text{Si}_2^+$	$\text{C}_{10}\text{H}_7(\text{Si}(\text{CH}_3)_2)_2\text{C}_{10}\text{H}_7$ (Disilane, 1,1,2,2-tetramethyl-1,2-di-1-naphthalenyl-)	38446-41-2	**	7.91	CTS	3758
$\text{C}_{26}\text{H}_{26}\text{Si}_2^+$	$((\text{C}_6\text{H}_5)_2\text{CH}_3\text{Si})_2$ (Disilane, 1,2-dimethyl-1,1,2,2-tetraphenyl-)	1172-76-5	**	8.05 ± 0.15	EI	3549
$\text{C}_{36}\text{H}_{30}\text{Si}_2^+$	$((\text{C}_6\text{H}_5)_3\text{Si})_2$ (Disilane, hexaphenyl-)	1450-23-3	**	8.16 ± 0.15	EI	3549
$\text{C}_8\text{H}_{24}\text{Si}_3^+$	$\text{Si}_3(\text{CH}_3)_8$	3704-44-7	**	8.19 (V)	PE	3504
$\text{C}_{16}\text{H}_{32}\text{Si}_3^+$	$\text{C}_6\text{H}_5\text{C}(\text{Si}(\text{CH}_3)_3)_3$ (Silane,(phenylmethylidene)tris(trimethyl-))	14595-76-7	** **	8.10 8.10 (V)	PE PE	5574 5012
$\text{C}_{17}\text{H}_{28}\text{Si}_3^+$	$\text{C}_{10}\text{H}_7\text{Si}_3(\text{CH}_3)_7$ (Trisilane, 1,1,1,2,2,3,3-heptamethyl-3-(1-naphthalenyl-)) $\text{C}_{10}\text{H}_7\text{Si}(\text{Si}(\text{CH}_3)_3)_2\text{CH}_3$ (Trisilane, 1,1,1,2,3,3,3-heptamethyl-2-)(-naphthalenyl-))	38446-42-3 38446-43-4	** **	7.93 7.85	CTS CTS	3758 3758
$\text{C}_{18}\text{H}_{36}\text{Si}_3^+$	$\text{C}_6\text{H}_3(\text{CH}_2\text{Si}(\text{CH}_3)_3)_3$ (Silane, [1,3,5-benzenetriyltris(methylene)]tris(trimethyl-))	59305-32-7		7.85 (V)	PE	5012
$\text{C}_{21}\text{H}_{42}\text{Si}_3^+$	$\text{C}_{21}\text{H}_{42}\text{Si}_3$ (Silane,[2,4,6-trimethyl-1,3,5-benzenetriyl]tris(methylene)]tris(trimethyl-))	69020-18-4	**	7.40 (V)	PE	5629

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{26}\text{H}_{32}\text{Si}_3^+$	$\text{C}_{10}\text{H}_7(\text{Si}(\text{CH}_3)_2)_3\text{C}_{10}\text{H}_7$ (Trisilane, 1,1,2,2,3,3-hexamethyl-1,3-di-1-naphthalenyl-)	38580-43-7	**	7.92	CTS	3758
$\text{C}_6\text{H}_{16}\text{Si}_4^+$	$\text{C}_6\text{H}_{16}\text{Si}_4$ (1,3,5,7-Tetrasilatricyclo[3.3.1.1 ^{3,7}]decane)	281-44-7	**	9.0 ± 0.05	PE	3855
			**	9.7 (V)	PE	4000
$\text{C}_{10}\text{H}_{24}\text{Si}_4^+$	$\text{C}_6\text{H}_{12}\text{Si}_4(\text{CH}_3)_4$ (1,3,5,7-Tetrasilatricyclo[3.3.1.1 ^{3,7}]decane, 1,3,5,7-tetramethyl-)	17995-33-4	**	8.45 ± 0.05	PE	3855
$\text{C}_{10}\text{H}_{30}\text{Si}_4^+$	$n\text{-Si}_4(\text{CH}_3)_{10}$	865-76-9	**	7.98 (V)	PE	3504
$\text{C}_{18}\text{H}_{38}\text{Si}_4^+$	$\text{C}_{18}\text{H}_{38}\text{Si}_4$ (Silane, 1,2,4,5-benzenetetracyltetrakis(trimethyl-))	17156-61-5	**	8.30 (V)	PE	5319
$\text{C}_{18}\text{H}_{40}\text{Si}_4^+$	$\text{C}_6\text{H}_4(\text{Si}(\text{CH}_3)_3)_4$ (Silane, 2,5-cyclohexadiene-1,4-diyltetrakis(trimethyl-))	XXXXX-XX-X	**	7.00 (V)	PE	5535
$\text{C}_{18}\text{H}_{44}\text{Si}_4^+$	$((\text{CH}_3)_3\text{SiCH}_2)_4\text{C}=\text{C}$	XXXXX-XX-X	**	7.15 (V)	PE	5535
$\text{C}_{20}\text{H}_{42}\text{Si}_4^+$	$\text{C}_6\text{H}_4(\text{CH}(\text{Si}(\text{CH}_3)_3)_2)_2$ (Silane, 1,4-phenylenedimethylidene)tetrakis(trimethyl-)	17557-10-7	**	7.40	PE	5574
			**	7.40 (V)	PE	5012
$\text{C}_{20}\text{H}_{48}\text{Si}_4^+$	$(tert\text{-C}_4\text{H}_9\text{SiCH}_3)_4$		**	7.42 (V)	PE	4683
$\text{C}_{22}\text{H}_{46}\text{Si}_4^+$	$\text{C}_6\text{H}_2(\text{CH}_2\text{Si}(\text{CH}_3)_3)_4$ (Silane, [1,2,4,5-benzenetetracyltetrakis(methylene)]tetrakis(trimethyl-))	64131-86-8		7.10 (V)	PE	5012
			**	7.10 (V)	PE	5629
$\text{C}_{22}\text{H}_{48}\text{Si}_4^+$	$\text{C}_{10}\text{H}_{12}(\text{Si}(\text{CH}_3)_3)_4$ (Silane, [1,2,3,4,5,6,7,8-octahydro-1,4,5,8-naphthalenetetrayl]tetrakis(trimethyl-))	XXXXX-XX-X	**	6.98 (V)	PE	5535
$\text{C}_{10}\text{H}_{30}\text{Si}_5^+$	$\text{Si}_5(\text{CH}_3)_{10}$ (Cyclopentasilane, decamethyl-)	13452-92-1	**	7.94 (V)	PE	3504
$\text{C}_{12}\text{H}_{36}\text{Si}_5^+$	$\text{Si}(\text{Si}(\text{CH}_3)_3)_4$	4098-98-0	**	8.24 (V)	PE	3504
$\text{C}_{12}\text{H}_{36}\text{Si}_6^+$	$\text{Si}_6(\text{CH}_3)_{12}$ (Cyclohexasilane, dodecamethyl-)	4098-30-0	**	7.79 (V)	PE	3504
$\text{C}_{22}\text{H}_{54}\text{Si}_6^+$	$((\text{CH}_3)_3\text{Si})_3\text{CC}_2$	20932-80-3	**	7.60 (V)	PE	5332
$\text{C}_{26}\text{H}_{58}\text{Si}_6^+$	$\text{C}_6\text{H}_4(\text{C}(\text{Si}(\text{CH}_3)_3)_3)_2$ (Silane, 1,4-phenylenedimethanetetrayl)hexakis	17557-11-8	**	7.45	PE	5574

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₂₆H₅₈Si₆⁺	C ₆ H ₅ (C(Si(CH ₃) ₃) ₃) ₂	17557-11-8	**	7.45 (V)	PE	5012
C₃₀H₆₆Si₆⁺	C ₆ (CH ₂ Si(CH ₃) ₃) ₆ (Silane, [1,2,3,4,5,6-benzenehexylhexakis(methylene)]hexakis(trimethyl-))	64131-87-9	**	7.40 (V)	PE	5012
			**	7.40 (V)	PE	5629
C₁₆H₃₆Si₇⁺	C ₁₀ H ₁₈ Si ₇ (CH ₃) ₆ (2H-1,5;8,12-Dimethano-3,6a,10-metheno-1,3,5,6a,8,10,12-heptasilaoctalene, dodecahydro-1,3,5,8,10,12-hexamethyl-)	26393-20-4	**	7.9±0.05	PE	3855
NSi₂⁺	Si ₂ N	12293-67-3	**	9.5±0.5	EI	3810
			**	9.3±0.5	EI	4200
H₃N₃Si⁺	SiH ₃ N ₃	13847-60-4	**	10.33±0.02 (V)	PE	3670
H₉NSi₃⁺	(SiH ₃) ₃ N	13862-16-3	**	9.7±0.1 (V)	PE	3661
C₂H₉NSi⁺	(CH ₃) ₂ NSiH ₃	2875-98-1	**	8.5±0.1 (V)	PE	3661
C₄H₁₃NSi⁺	NH ₂ (CH ₂ Si(CH ₃) ₃)	18166-02-4	**	9.07 (V)	PE	5102
C₅H₁₅NSi⁺	NH(CH ₃)(CH ₂ Si(CH ₃) ₃)	18135-05-2	**	8.55 (V)	PE	5102
C₆H₁₇NSi⁺	C ₆ H ₁₇ NSi (CH ₃) ₂ NCH ₂ Si(CH ₃) ₃	13014-85-2	**	8.46 (V)	PE	5102
		18182-40-6	**	7.61	PI	5543
		**	**	7.63±0.05	PE	4192
		**	**	8.20 (V)	PE	5102
C₇H₁₉NSi⁺	C ₇ H ₁₉ NSi (tert-C ₄ H ₉ NHSi(CH ₃) ₃)	5577-67-3	**	8.41±0.05 (V)	PE	4725
C₈H₁₃NSi⁺	C ₅ H ₄ NS(CH ₃) ₃ (Pyridine, 2-(trimethylsilyl)-) C ₅ H ₄ NS(CH ₃) ₃ (Pyridine, 4-(trimethylsilyl)-)	13737-04-7	**	8.90±0.05 (V)	PE	3685
		18301-46-7	**	9.30±0.05 (V)	PE	3685
C₈H₂₁NSi⁺	C ₈ H ₂₁ NSi	10545-36-5	**	7.93 (V)	PE	5102
C₉H₁₄NSi⁺	C ₆ H ₄ (N(CH ₃) ₂)SiH(CH ₃) ₂ (Benzenamine, 4-(dimethylsilyl)-N,N-dimethyl-)	2516-75-8	CH ₃	7.08	EI	4125
C₉H₂₁NSi⁺	C ₅ H ₁₀ NCH ₂ Si(CH ₃) ₃ (Piperidine, 1-[(trimethylsilyl)methyl]-)	17877-17-7	**	8.18 (V)	PE	5102
C₇H₁₈N₂Si⁺	tert-C ₄ H ₉ N=NSi(CH ₃) ₃	25811-66-9	**	7.6±0.2 (V)	PE	4581

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_5\text{H}_{12}\text{O}_2\text{Si}^+$	$\text{C}_3\text{H}_5\text{Si}(\text{OCH}_3)_2$ (Silacyclobutane, 1,1-dimethoxy-)	33446-84-3	**	10.15 (V)	PE	4077
$\text{C}_{10}\text{H}_{16}\text{O}_2\text{Si}^+$	$\text{CH}_3\text{OC}_6\text{H}_4\text{Si}(\text{CH}_3)_2\text{OCH}_3$ (Silane, methoxy(4-methoxyphenyl)dimethyl-)	62244-48-8	**	8.62	EI	5421
$\text{C}_8\text{H}_{20}\text{O}_4\text{Si}^+$	$(\text{C}_2\text{H}_5\text{O})_4\text{Si}$	78-10-4	**	9.77 (V)	PE	3503
$\text{C}_6\text{H}_{18}\text{OSi}_2^+$	$((\text{CH}_3)_3\text{Si})_2\text{O}$	107-46-0	**	9.88 (V)	PE	4181
$\text{C}_{11}\text{H}_{20}\text{OSi}_2^+$	$\text{C}_6\text{H}_4(\text{SiH}(\text{CH}_3)_2)\text{Si}(\text{CH}_3)_2\text{OCH}_3$ (Silane, [3-(dimethylsilyl)phenyl]methoxydimethyl-)	XXXXXX-XX-X	**	8.5 ± 0.2	EI	4121
	$\text{C}_6\text{H}_4(\text{SiH}(\text{CH}_3)_2)\text{Si}(\text{CH}_3)_2\text{OCH}_3$ (Silane, [4-(dimethylsilyl)phenyl]methoxydimethyl-)	33546-26-8	**	8.6 ± 0.2	EI	4121
$\text{C}_{12}\text{H}_{22}\text{OSi}_2^+$	$\text{C}_6\text{H}_4(\text{OCH}_3)\text{Si}_2(\text{CH}_3)_5$ (Disilane, (4-methoxphenyl)pentamethyl-)	4199-03-5	**	7.85	CTS	3758
$\text{C}_7\text{H}_{19}\text{O}_2\text{Si}_2^+$	$(\text{CH}_3)_3\text{SiO}(\text{CH}_2)_2\text{OSi}(\text{CH}_3)_3$	7381-30-8	CH_3	9.5 ± 0.1	EI	4300
$\text{C}_8\text{H}_{21}\text{O}_2\text{Si}_2^+$	$(\text{CH}_3)_3\text{SiO}(\text{CH}_2)_3\text{OSi}(\text{CH}_3)_3$	17887-80-8	CH_3	9.4 ± 0.1	EI	4300
$\text{C}_9\text{H}_{23}\text{O}_2\text{Si}_2^+$	$(\text{CH}_3)_3\text{SiO}(\text{CH}_2)_4\text{OSi}(\text{CH}_3)_3$	18001-91-7	CH_3	9.3 ± 0.1	EI	4300
$\text{C}_{10}\text{H}_{25}\text{O}_2\text{Si}_2^+$	$(\text{CH}_3)_3\text{SiO}(\text{CH}_2)_5\text{OSi}(\text{CH}_3)_3$	54494-06-3	CH_3	9.3 ± 0.1	EI	4300
$\text{C}_{11}\text{H}_{20}\text{O}_2\text{Si}_2^+$	$\text{C}_6\text{H}_4(\text{SiH}(\text{CH}_3)_2)\text{Si}(\text{OCH}_3)_2\text{CH}_3$ (Silane, [3-(dimethylsilyl)phenyl]dimethoxymethyl-)	XXXXXX-XX-X	**	8.8 ± 0.2	EI	4121
	$\text{C}_6\text{H}_4(\text{SiH}(\text{CH}_3)_2)\text{Si}(\text{OCH}_3)_2\text{CH}_3$ (Silane, [4-(dimethylsilyl)phenyl]dimethoxymethyl-)	34239-01-5	**	8.5 ± 0.2	EI	4121
$\text{C}_{11}\text{H}_{27}\text{O}_2\text{Si}_2^+$	$(\text{CH}_3)_3\text{SiO}(\text{CH}_2)_6\text{OSi}(\text{CH}_3)_3$	6222-22-6	CH_3	9.3 ± 0.1	EI	4300
$\text{C}_{12}\text{H}_{29}\text{O}_2\text{Si}_2^+$	$(\text{CH}_3)_3\text{SiO}(\text{CH}_2)_7\text{OSi}(\text{CH}_3)_3$	54494-07-4	CH_3	9.4 ± 0.1	EI	4300
$\text{C}_{11}\text{H}_{20}\text{O}_3\text{Si}_2^+$	$\text{C}_6\text{H}_4(\text{SiH}(\text{CH}_3)_2)\text{Si}(\text{OCH}_3)_3$ (Silane, [3-(dimethylsilyl)phenyl]trimethoxy-)	XXXXXX-XX-X	**	9.0 ± 0.2	EI	4121
NOSi_2^+	Si_2NO	12033-47-5	**	10.8 ± 0.5	EI	3810
CH_3NOSi^+	SiH_3NCO	13730-13-7	**	11.10 ± 0.02 (V)	PE	3670
$\text{C}_4\text{H}_9\text{NOSi}^+$	$(\text{CH}_3)_3\text{SiNCO}$	1118-02-1	**	10.3 ± 0.1 (V)	PE	3670

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₈H₁₃NOSi⁺	C ₅ H ₄ N(O)Si(CH ₃) ₃ (Pyridine, 4-(trimethylsilyl)-, 1-oxide)	28867-06-3	**	8.19 (V)	PE	4222
C₁₁H₁₉NOSi⁺	(CH ₃) ₂ NC ₆ H ₄ Si(CH ₃) ₂ OCH ₃ (Benzenamine, 4-(methoxydimethylsilyl)-N,N-dimethyl-)	62244-49-9	**	7.45	EI	5421
C₉H₁₃NO₂Si⁺	NO ₂ C ₆ H ₄ Si(CH ₃) ₃ (Silane, trimethyl(4-nitrophenyl)-)	4405-33-8	**	9.80 (V)	PE	5380
C₆H₁₃NO₃Si⁺	N(CH ₂ CH ₂ O) ₃ SiH (2,8,9-Trioxa-5-aza-silabicyclo[3.3.3]undecane)	283-60-3	**	~10.1	PE	4413
C₇H₁₅NO₃Si⁺	N(CH ₂ CH ₂ O) ₃ SiCH ₃ (2,8,9-Trioxa-5-aza-1-silabicyclo[3.3.3]undecane, 1-methyl-)	2288-13-3	**	8.7 (V)	PE	4413
C₉H₁₃NO₃Si⁺	NO ₂ C ₆ H ₄ Si(CH ₃) ₂ OCH ₃ (Silane, methoxydimethyl(4-nitrophenyl)-)	62244-50-2	**	9.44	EI	5421
C₈H₁₇NO₄Si⁺	N(CH ₂ CH ₂ O) ₃ SiOC ₂ H ₅ (2,8,9-Trioxa-5-aza-1-silabicyclo[3.3.3]undecane, 1-ethoxy-)	3463-21-6	**	10.6 (V)	PE	4413
F₂Si⁺ (² A ₁) SiF ₂ (² A ₁) (² A ₂ , ² B ₂) (² B ₁ , ² A ₁)		13966-66-0	**	10.78±0.05	PE	4138
			**	11.08 (V)	PE	4322
			**	15.57 (V)	PE	4322
			**	17.08 (V)	PE	4322
F₃Si⁺	CH ₃ SiF ₃	373-74-0	CH ₃	13.33±0.05	PI	4907
F₄Si⁺ SiF ₄		7783-61-1	**	15.19	PI	4907
			**	16.45 (V)	PE	4322
			**	16.46±0.04 (V)	PE	3880
			**	15.4±1	EI	4894
F₆Si₂⁺	Si ₂ F ₆	13830-68-7	**	13.20±0.02 (V)	PE	4026
H₃FSi⁺ SiH ₃ F		13537-33-2	**	12.58 (V)	PE	3511
			**	12.6±0.1 (V)	PE	3510
			**	16.1±0.1 (V)	PE	3502
			**	13.0±1	EI	4894
H₂F₂Si⁺ SiH ₂ F ₂		13824-36-7	**	12.85 (V)	PE	3511
			**	12.85 (V)	PE	3694
			**	12.9±0.1 (V)	PE	3510
			**	11.0±1	EI	4894
HF₃Si⁺ SiHF ₃		13465-71-9	**	14.48±0.02 (V)	PE	4026
			**	14.48±0.05 (V)	PE	5419
			**	11.0±1	EI	4894

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_2\text{H}_6\text{FSi}^+$	$(\text{CH}_3)_3\text{SiF}$	420-56-4	CH_3	10.70 ± 0.04	PI	4907
$\text{C}_3\text{H}_9\text{FSi}^+$	$(\text{CH}_3)_3\text{SiF}$	420-56-4	**	10.31 ± 0.04 11.0 (V)	PI	4907
			**		PE	4972
$\text{C}_5\text{H}_9\text{FSi}^+$	$(\text{CH}_3)_3\text{SiC}\equiv\text{CF}$	38346-22-4	**	9.8 ± 0.1	PE	4002
$\text{C}_6\text{H}_{15}\text{FSi}^+$	$(\text{C}_2\text{H}_5)_3\text{SiF}$	358-43-0	**	10.1 (V)	PE	4972
$\text{C}_7\text{H}_8\text{FSi}^+$	$\text{C}_6\text{H}_5\text{Si}(\text{CH}_3)_2\text{F}$ (Silane, fluorodimethylphenyl-)	454-57-9	CH_3	10.83	EI	5366
$\text{C}_8\text{H}_{10}\text{FSi}^+$	$\text{CH}_3\text{C}_6\text{H}_4\text{Si}(\text{CH}_3)_2\text{F}$ (Silane, fluorodimethyl(3-methylphenyl)-)	33664-04-9	CH_3	10.92	EI	5366
	$\text{CH}_3\text{C}_6\text{H}_4\text{Si}(\text{CH}_3)_2\text{F}$ (Silane, fluorodimethyl(4-methylphenyl)-)	33664-05-0	CH_3	10.82	EI	5366
$\text{C}_8\text{H}_{11}\text{FSi}^+$	$\text{C}_6\text{H}_5\text{Si}(\text{CH}_3)_2\text{F}$ (Silane, fluorodimethylphenyl-)	454-57-9	**	9.17	EI	5421
$\text{C}_9\text{H}_{13}\text{FSi}^+$	$\text{CH}_3\text{C}_6\text{H}_4\text{Si}(\text{CH}_3)_2\text{F}$ (Silane, fluorodimethyl(3-methylphenyl)-)	33664-04-9	**	8.86	EI	5421
	$\text{CH}_3\text{C}_6\text{H}_4\text{Si}(\text{CH}_3)_2\text{F}$ (Silane, fluorodimethyl(4-methylphenyl)-)	33664-05-0	**	8.86	EI	5421
	$\text{FC}_6\text{H}_4\text{Si}(\text{CH}_3)_3$ (Silane,(4-fluorophenyl)trimethyl-)	455-17-4	**	9.0 (V)	PE	5380
$\text{CH}_3\text{F}_2\text{Si}^+$	$(\text{CH}_3)_2\text{SiF}_2$	353-66-2	CH_3	11.70 ± 0.03	PI	4907
$\text{C}_2\text{H}_6\text{F}_2\text{Si}^+$	$(\text{CH}_3)_2\text{SiF}_2$	353-66-2	**	11.03 ± 0.03 11.5 (V)	PI	4907
			**		PE	4972
$\text{C}_4\text{H}_6\text{F}_2\text{Si}^+$	$\text{C}_4\text{H}_6\text{SiF}_2$ (Silacyclopent-3-ene, 1,1-difluoro-)	XXXXX-XX-X	**	9.62 (V)	PE	4517
$\text{C}_4\text{H}_{10}\text{F}_2\text{Si}^+$	$(\text{C}_2\text{H}_5)_2\text{SiF}_2$	358-06-5	**	10.5 (V)	PE	4972
$\text{CH}_3\text{F}_3\text{Si}^+$	CH_3SiF_3	373-74-0	**	12.48 \pm 0.04 13.2 (V) 13.24 \pm 0.02 (V)	PI	4907
			**		PE	4972
			**		PE	4026
$\text{C}_5\text{H}_5\text{F}_3\text{Si}^+$	$\text{C}_5\text{H}_5(\text{SiF}_3)$ (Silane, 2,4-cyclopentadien-1-yl trifluoro-)	55765-70-3	**	9.1 (V)	PE	4373
$\text{C}_7\text{H}_{10}\text{F}_6\text{Si}^+$	$cis-(\text{CH}_3)_3\text{SiC}(\text{CF}_3)=\text{C}(\text{CF}_3)\text{H}$	35186-03-9	**	9.86	PE	3589

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{10}\text{H}_{17}\text{FSi}_2^+$	$\text{C}_6\text{H}_4(\text{SiF}(\text{CH}_3)_2)\text{SiH}(\text{CH}_3)_2$ (Silane, [4-(dimethylsilyl)phenyl]fluorodimethyl-)	33546-29-1	**	8.5 ± 0.2	EI	4121
$\text{C}_9\text{H}_{14}\text{F}_2\text{Si}_2^+$	$\text{C}_6\text{H}_4(\text{SiF}_2\text{CH}_3)\text{SiH}(\text{CH}_3)_2$ (Silane, 1,1-difluoro[4-(dimethylsilyl)phenyl]methyl-)	XXXXX-XX-X	**	8.7 ± 0.2	EI	4121
$\text{C}_8\text{H}_{11}\text{F}_3\text{Si}_2^+$	$\text{C}_6\text{H}_4(\text{SiF}_3)\text{SiH}(\text{CH}_3)_2$ (Silane, trifluoro[4-(dimethylsilyl)phenyl]-)	XXXXX-XX-X	**	9.2 ± 0.2	EI	4121
$\text{C}_6\text{H}_{12}\text{F}_4\text{Si}_4^+$	$\text{C}_6\text{H}_{12}\text{Si}_4\text{F}_4$ (1,3,5,7-Tetrasilatricyclo[3.3.1.1 ^{3,7}]decane, 1,3,5,7-tetrafluoro-)	33664-21-0	**	9.8 ± 0.05	PE	3855
$\text{C}_9\text{H}_{13}\text{NFSi}^+$	$(\text{CH}_3)_2\text{NC}_6\text{H}_4\text{Si}(\text{CH}_3)_2\text{F}$ (Benzenamine, 4-(fluorodimethylsilyl)-N,N-dimethyl-)	62244-56-8	CH_3	11.59	EI	5366
$\text{C}_{10}\text{H}_{16}\text{NFSi}^+$	$(\text{CH}_3)_2\text{NC}_6\text{H}_4\text{Si}(\text{CH}_3)_2\text{F}$ (Benzenamine, 4-(fluorodimethylsilyl)-N,N-dimethyl-)	62244-56-8	**	7.55	EI	5421
$\text{C}_2\text{H}_6\text{NF}_3\text{Si}^+$	$\text{F}_3\text{SiN}(\text{CH}_3)_2$	812-14-6	**	9.60 ± 0.05 (V)	PE	5419
$\text{C}_9\text{H}_{10}\text{OFSi}^+$	$\text{CH}_3\text{OC}_6\text{H}_4\text{Si}(\text{CH}_3)_2\text{F}$ (Silane, fluoro(4-methoxyphenyl)dimethyl-)	62244-55-7	CH_3	11.03	EI	5366
$\text{C}_9\text{H}_{13}\text{OFSi}^+$	$\text{CH}_3\text{OC}_6\text{H}_4\text{Si}(\text{CH}_3)_2\text{F}$ (Silane, fluoro(4-methoxyphenyl)dimethyl-)	62244-55-7	**	8.42	EI	5421
$\text{C}_7\text{H}_7\text{NO}_2\text{FSi}^+$	$\text{NO}_2\text{C}_6\text{H}_4\text{Si}(\text{CH}_3)_2\text{F}$ (Silane, fluorodimethyl(4-nitrophenyl)-)	62244-57-9	CH_3	10.71	EI	5366
$\text{C}_8\text{H}_{10}\text{NO}_2\text{FSi}^+$	$\text{NO}_2\text{C}_6\text{H}_4\text{Si}(\text{CH}_3)_2\text{F}$ (Silane, fluorodimethyl(4-nitrophenyl)-)	62244-57-9	**	9.77	EI	5421
AlSi^+	SiAl	12042-55-6	**	6.5 ± 1.0	EI	4005
OAlSi^+	SiAlO	37361-47-0	** **	6.3 ± 1.0 8.0 ± 1	EI EI	4005 3985
P^+						
	P_2	12185-09-0		15.9	EI	3472
	PH_3	7803-51-2	$\text{H}_2 + \text{H}$	16.3	EI	3811
	PCl_3	7719-12-2	$\text{Cl}_2 + \text{Cl}$	18.5 ± 0.7	EI	3556
	PBr_3	7789-60-8	$\text{Br}_2 + \text{Br}$	16.7 ± 0.7	EI	3556
	LaPO_4	XXXXX-XX-X		13.0 ± 0.6	EI	5603
P_2^+	P_2	12185-09-0	** ** ** **	10.7 ± 0.1 10.60 10.62 ± 0.01 (V) 10.81 ± 0.01 (V)	S PE PE PE	3567 3695 4597 4597

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)		Method	Ref.
P₂⁺ <i>(^Σ_v)</i>	P ₂ <i>(^Σ_u)</i>	12185-09-0	**	10.84 (V)	PE	3695	
			**	15.52±0.01 (V)	PE	4597	
			**	9.7±0.5	EI	3458	
			**	9.7	EI	4001	
			**	10.3±0.5	EI	4120	
			**	11.2	EI	3472	
			**	11.4±0.5	EI	4098	
			**	11.8±0.5	EI	3555	
			P ₄	12.85±0.01	PI	4936	
			P ₂	12.85±0.03	PI	4924	
P₃⁺	P ₃ P ₄ <i>(^Σ_v)</i>	55030-78-9 12185-10-3	**	7.85±0.2	PI	4924	
				12.54±0.01	PI	4936	
			P	12.54±0.03	PI	4924	
P₄⁺	P ₄	12185-10-3	**	9.25	PI	4924	
			**	9.34	PI	4936	
			**	9.10±0.05	PE	3683	
			**	9.2	PE	3643	
			**	10.0±0.5	EI	4098	
			**	10.8±0.3	EI	3555	
HP⁺	PH ₃	7803-51-2	H ₂	12.9	EI	3811	
H₂P⁺	PH ₃	7803-51-2	H	13.4	EI	3811	
H₃P⁺	PH ₃	7803-51-2	**	9.96±0.01	PE	3703	
			**	9.96	PE	3719	
			**	9.96	PE	5516	
			**	10.59±0.05 (V)	PE	5419	
			**	10.0	EI	3811	
H₄P₂⁺	P ₂ H ₄	13445-50-6	**	9.69 (V)	PE	4584	
BP⁺	BP	20205-91-8	**	<13±2	EI	3619	
CP⁺	PC	12326-85-1	**	10.5±0.5	EI	3458	
C₂P⁺	C ₂ P	12602-39-0	**	10.9±0.5	EI	3458	
CP₂⁺	CP ₂	12601-93-3	**	9.4±0.5	EI	3458	
CHP⁺ <i>(^Π)</i>	HCP	6829-52-3	**	10.79±0.01	PE	3840	
			**	12.86±0.01	PE	3840	
CH₅P⁺	CH ₃ PH ₂	593-54-4	**	9.12±0.07	PE	4152	
			**	9.12	PE	5516	
			**	9.6±0.1 (V)	PE	3661	
			**	9.70 (V)	PE	4474	

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_2\text{H}_3\text{P}^+$	$\text{CH}_3\text{C}\equiv\text{P}$	34627-31-1	**	9.89 ± 0.01 (V)	PE	5033
$\text{C}_2\text{H}_5\text{P}^+$	$\text{C}_2\text{H}_5\text{P}$ (Phosphirane)	6569-82-0	**	9.4 ± 0.1	PE	4990
$\text{C}_2\text{H}_7\text{P}^+$	$(\text{CH}_3)_2\text{PH}$	676-59-5	** ** ** ** **	8.47 ± 0.07 8.47 8.5 ± 0.1 9.10 (V) 9.13 (V)	PE PE PE PE PE	4152 5516 4990 4474 4185
$\text{C}_3\text{H}_9\text{P}^+$	$(\text{CH}_3)_3\text{P}$	594-09-2	** ** ** ** ** ** ** ** ** ** ** ** ** ** ** **	8.01 ± 0.07 8.11 ± 0.1 8.11 8.6 ± 0.1 (V) 8.6 (V) 8.60 (V) 8.60 (V) 8.60 (V) 8.65 (V) 8.79	PE PE PE PE PE PE PE PE PE PE PE PE PE PE PE	4152 5042 5516 3661 5378 4226 4579 5368 4474 5602
$\text{C}_4\text{H}_{11}\text{P}^+$	$(\text{C}_2\text{H}_5)_2\text{PH}$ $(\text{CH}_3)_3\text{P}=\text{CH}_2$ $(\text{CH}_3)_3\text{P}=\text{CH}_2$ <i>tert</i> - $\text{C}_4\text{H}_9\text{PH}_2$	627-49-6 14580-91-7 29218-61-9 2501-94-2	** ** ** ** ** ** ** **	8.69 6.81 (V) 6.81 (V) 6.87 (V) 6.81 (V) 9.30 (V)	PE PE PE PE PE PE	3589 4579 5442 4181 5368 4474
$\text{C}_5\text{H}_5\text{P}^+$	$\text{C}_5\text{H}_5\text{P}$ (Phosphorin)	289-68-9	**	9.2 (V)	PE	3832
$\text{C}_6\text{H}_7\text{P}^+$	$\text{C}_6\text{H}_5\text{PH}_2$ (Phosphine, phenyl-)	638-21-1	**	8.47 ± 0.01	PE	4154
$\text{C}_6\text{H}_9\text{P}^+$	$(\text{C}_2\text{H}_3)_3\text{P}$	3746-01-8	**	7.52 (V)	PE	5526
$\text{C}_6\text{H}_{13}\text{P}^+$	$(\text{CH}_3)_3\text{P}=\text{CHCH}=\text{CH}_2$	30417-65-3	**	6.20 (V)	PE	4579
$\text{C}_6\text{H}_{15}\text{P}^+$	$(\text{C}_2\text{H}_5)_3\text{P}$	554-70-1	** **	7.44 (V) 8.52	PE PE	5526 5602
$\text{C}_7\text{H}_{11}\text{P}^+$	$\text{C}_4\text{H}_9(\text{CH}_3)_2\text{P}(\text{CH}_3)$ (1H-Phosphole, 1,3,4-trimethyl-)	37739-99-4	**	8.25 (V)	PE	5618
$\text{C}_7\text{H}_{15}\text{P}^+$	$(\text{CH}_3)_3\text{P}=\text{CHC}(\text{CH}_3)=\text{CH}_2$ $(\text{CH}_3)_3\text{P}=\text{CHCH}=\text{CHCH}_3$	29218-65-3 61169-15-1	** **	6.20 (V) 6.02 (V)	PE PE	4579 4579
$\text{C}_8\text{H}_{11}\text{P}^+$	$(\text{C}_6\text{H}_5)(\text{CH}_3)_2\text{P}$ (Phosphine, dimethyl phenyl)	672-66-2	**	7.58 ± 0.05	PI	5278

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₂₁H₂₁P⁺	(CH ₃ C ₆ H ₄) ₃ P (Phosphine, tris(2-methylphenyl)-) (CH ₃ C ₆ H ₄) ₃ P (Phosphine, tris(3-methylphenyl)-) (CH ₃ C ₆ H ₄) ₃ P (Phosphine, tris(4-methylphenyl)-) (C ₆ H ₅) ₃ P=C(CH ₃) ₂ (Phosphorane, (1-methylethylidene)triphenyl-)	6163-58-2 6224-63-1 1038-95-9 16666-80-1	** ** ** **	7.64 (V) 7.68 (V) 7.6 (V) 6.04 (V)	PE PE PE PE	5438 5438 5438 4579
C₂₂H₂₁P⁺	(C ₆ H ₅) ₃ P=CHCH=CHCH ₃ (Phosphorane, 2-butenylenetriphenyl-(E)-)	56374-57-3	**	5.95 (V)	PE	4579
C₂₃H₁₇P⁺	C ₅ H ₂ P(C ₆ H ₅) ₃ (Phosphorin, 2,4,6-triphenyl)	13497-36-4	**	7.80 (V)	PE	5271
C₂₃H₁₉P⁺	(C ₆ H ₅) ₃ P=C ₅ H ₄ (Phosphorane, 2,4-cyclopentadien-1-ylidenetriphenyl-)	2224-32-0	**	6.91 (V)	PE	4579
C₂₅H₂₁P⁺	(C ₆ H ₅) ₃ P=CHC ₆ H ₅ (Phosphorane, triphenyl(phenylmethylene)-)	16721-45-2	**	6.01 (V)	PE	4579
C₂₅H₂₃P⁺	C ₅ H ₂ P(C ₆ H ₅) ₃ (CH ₃) ₂ (Phosphorin, 1,1-dihydro-1,1-dimethyl-2,4,6-triphenyl-)	25959-36-8	**	5.90 (V)	PE	5271
C₂₇H₃₃P⁺	(iso-C ₃ H ₇ C ₆ H ₄) ₃ P (Phosphine, tris[4-(1-methylethyl)phenyl]-)	29949-82-4	**	7.53 (V)	PE	5438
C₂₉H₂₅P⁺	C ₉ H ₆ P(C ₆ H ₅)(CH ₂ C ₆ H ₅) ₂ (Phosphinoline, 1,1-dihydro-2-phenyl-1,1-bis(phenylmethyl)-)	39767-95-8	**	6.00	PE	4066
C₃₀H₃₉P⁺	(tert-C ₄ H ₉ C ₆ H ₄) ₃ P (Phosphine, tris[4-(1,1-dimethylethyl)phenyl]-)	54409-77-7	**	7.52 (V)	PE	5438
C₃₅H₂₇P⁺	C ₅ H ₂ P(C ₆ H ₅) ₅ (Phosphorin, 1,1-dihydro-1,1,2,4,6-pentaphenyl-)	22605-15-8	**	5.90 (V)	PE	5271
C₄H₁₂P₂⁺	((CH ₃) ₂ P) ₂ -trans ((CH ₃) ₂ P) ₂ -gauche	3676-91-3 3676-91-3	** ** **	7.88 (V) 7.88 (V) 8.79 (V)	PE PE PE	4191 4185 4185
C₁₀H₁₆P₂⁺	C ₆ H ₄ (P(CH ₃) ₂) ₂ (Phosphine, 1,4-phenylenebis(dimethyl-))	10498-57-4	**	8.2 (V)	PE	5382
C₁₆H₃₆P₄⁺	(tert-C ₄ H ₉ P) ₄ (Tetrrophosphetane, tetrabutyl-)	13969-03-4	**	7.39 (V)	PE	4942
C₂₄H₄₄P₄⁺	(C ₆ H ₁₁ P) ₄ (Tetrrophosphetane, tetracyclohexyl-)	3040-71-9	**	7.28 (V)	PE	4942

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₅H₁₅P₅⁺	(CH ₃ P) ₅ (Pentaphospholane e, pentamethyl-)	1073-98-9	**	7.58 (V)	PE	4942
C₁₀H₂₅P₅⁺	(C ₂ H ₅ P) ₅ (Pentaphospholane e, pentaethyl-)	4141-67-7	**	7.41 (V)	PE	4942
C₁₅H₃₅P₅⁺	(n-C ₃ H ₇ P) ₅ (Pentaphospholane e, pentapropyl-)	55019-74-4	**	7.26 (V)	PE	4942
NP⁺	NP	17739-47-8	**	11.85	PE	4498
(² S _g ⁺)			**	11.88±0.01	PE	4685
(² S _u ⁺)			**	12.30±0.01	PE	4685
(² I _u)			**	12.34	PE	4498
(² I _g _μ)			**	15.74±0.01	PE	4685
C₃H₁₀NP⁺	(CH ₃) ₃ P=NH	15107-02-5	**	8.19 (V)	PE	4181
			**	8.29 (V)	PE	5442
C₄H₁₂NP⁺	(CH ₃) ₃ PNCH ₃	42437-75-2	**	7.67 (V)	PE	5442
C₇H₁₈NP⁺	(CH ₃) ₃ PN(tert-C ₄ H ₉)	71328-66-0	**	7.56 (V)	PE	5442
C₉H₁₄NP⁺	(CH ₃) ₃ PNC ₆ H ₅ (Benzenamine,N-trimethylphosphoranylidene)-)	57114-54-2	**	7.05 (V)	PE	5442
C₁₀H₁₆NP⁺	C ₆ H ₄ (N(CH ₃) ₂)(P(CH ₃) ₂) (Benzenamine,4-(dimethylphosphino)-N,N-dimethyl-)	1199-66-2	**	7.30 (V)	PE	5382
C₁₈H₁₆NP⁺	(C ₆ H ₅) ₃ PNH (Phosphine imide,P,P,P-triphenyl-)	2240-47-3	**	7.95 (V)	PE	5442
C₁₉H₁₈NP⁺	(C ₆ H ₅) ₃ PNCH ₃ (Methanamine,N-(triphenylphosphoranylidene)-)	17986-01-5	**	7.54 (V)	PE	5442
C₂₀H₂₀NP⁺	(C ₆ H ₅) ₃ PNC ₂ H ₅ (Ethanamine,N-triphenylphosphoranylidene)-)	47182-04-7	**	7.43 (V)	PE	5442
C₂₀H₃₂NP⁺	(CH ₃) ₂ NC ₆ H ₄ P(C ₆ H ₁₁) ₂ (Benzenamine,4-(dicyclohexylphosphino)-N,N-dimethyl-)	40438-64-0	**	7.25 (V)	PE	5417
C₂₁H₂₂NP⁺	(C ₆ H ₅) ₃ PN(iso-C ₃ H ₇) (2-Propanamine,N-(triphenylphosphoranylidene)-)	40168-14-7	**	7.38 (V)	PE	5442
C₂₂H₂₄NP⁺	(C ₆ H ₅) ₃ PN(tert-C ₄ H ₉) (2-Propanamine,2-methyl-N-(triphenylphosphoranylidene)-)	13989-64-5	**	7.35 (V)	PE	5442

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{21}\text{H}_{20}\text{NP}^+$	$(\text{C}_6\text{H}_5)_3\text{PNC}_6\text{H}_5$ (Benzenamine,N-(triphenylphosphoranylidene)-)	2325-27-1	**	6.95 (V)	PE	5442
$\text{C}_{21}\text{H}_{26}\text{NP}^+$	$(\text{C}_6\text{H}_5)_3\text{PNC}_6\text{H}_{11}$ (Cyclohexanamine,N-(triphenylphosphoranylidene)-)	66949-28-8	**	7.37 (V)	PE	5442
$\text{C}_{31}\text{H}_{37}\text{N}_2\text{P}^+$	$\text{C}_5\text{H}_2\text{P}(\text{C}_6\text{H}_5)_3(\text{N}(\text{C}_2\text{H}_5)_2)_2$ (Phosphorin,1,1-bis(diethylamino)-1,1-dihydro-2,4,6-triphenyl-)	36231-67-1	**	5.95 (V)	PE	5271
$\text{C}_6\text{H}_{16}\text{N}_3\text{P}^+$	$\text{C}_2\text{H}_4\text{N}_2\text{P}(\text{CH}_3)_2\text{N}(\text{CH}_3)_2$ (1,3,2-Diazaphospholidin-2-amine, N,N,1,3-tetramethyl-)	6069-38-1	**	7.61 (V)	PE	5477
$\text{C}_6\text{H}_{18}\text{N}_3\text{P}^+$	$((\text{CH}_3)_2\text{N})_3\text{P}$	1608-26-0	** ** **	7.30 (V) 7.61 (V) 10.01	PE PE PE	4474 3825 5602
	$((\text{CH}_3)_2\text{N})_3\text{P}(\text{CO})_4\text{Mo}$	27342-90-1		10.1 ± 0.05	EI	3952
$\text{C}_8\text{H}_{18}\text{N}_3\text{P}^+$	$((\text{CH}_3)_2\text{N})_3\text{P}(\text{CO})_4\text{Mo}$	27342-90-1		10.1 ± 0.05	EI	3952
$\text{C}_8\text{H}_{20}\text{N}_3\text{P}^+$	$\text{C}_2\text{H}_4\text{N}_2\text{P}(\text{CH}_3)_2\text{N}(\text{C}_2\text{H}_5)_2$ (1,3,2-Diazaphospholidin-2-amine, N,N-diethyl-1,3-dimethyl-)	65173-82-2	**	7.50 (V)	PE	5477
$\text{C}_{10}\text{H}_{24}\text{N}_3\text{P}^+$	$\text{C}_2\text{H}_4\text{N}_2\text{P}(\text{CH}_3)_2\text{N}(\text{iso-C}_3\text{H}_7)_2$ (1,3,2-Diazaphospholidin-2-amine,1,3-dimethyl- N,N-bis(1-methylethyl)-)	65173-83-3	**	7.40 (V)	PE	5477
$\text{C}_{24}\text{H}_{30}\text{N}_3\text{P}^+$	$((\text{CH}_3)_2\text{NC}_6\text{H}_4)_3\text{P}$ (Benzenamine,4,4',4"-phosphinidynetris[N,N-dimethyl-])	1104-21-8	**	6.9-7.0 (V)	PE	5438
$\text{B}_2\text{C}_6\text{H}_{18}\text{N}_3\text{P}^+$	$\text{N}_3\text{B}_2(\text{CH}_3)_4\text{P}(\text{CH}_3)_2$ (1,2,4,3,5-Triazadiborolidine, 1-(dimethylphosphino)-2,3,4,5-tetramethyl-) $\text{N}_3\text{B}_2(\text{CH}_3)_4\text{P}(\text{CH}_3)_2$ (1,2,4,3,5-Triazadiborolidine, 4-(dimethylphosphino)-1,2,3,5-tetramethyl-)	53246-20-1 53246-15-4	** **	7.64 (V) 7.70 (V)	PE PE	4526 4526
$\text{B}_2\text{C}_6\text{H}_{24}\text{N}_3\text{P}^+$	$\text{C}_8\text{H}_{24}\text{B}_2\text{N}_5\text{P}$ (Phosphorous diamide, N,N,N',N'-tetramethyl-P-(1,2,3,5-tetramethyl-1,2,4,3,5-triazaborolid in-4-yl)-) $(\text{RX})_3\text{B}_2(\text{CH}_3)_4\text{P}(\text{N}(\text{CH}_3)_2)_2$	53246-16-5	**	7.57 (V)	PE	4526
OP^+	PO	14452-66-5	** ** ** ** ** ** **	8.231 8.373 8.38 8.5 ± 1 9.1 ± 0.5 9.5 ± 0.5 10.7	S S S EI EI EI EI	3762 5136 3560 3819 4678 4098 4518
	P_2O_3 $(\text{CH}_3\text{O})_3\text{PO}$ LaPO_4	1314-24-5 512-56-1 XXXXX-XX-X	$\text{O} + \text{CH}_3\text{O} + 2\text{H}$	13.5 ± 1.0 18.90 ± 0.50 11.5 ± 0.5	EI EI EI	4098 3989 5603

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
O₂P⁺						
	PO ₂	12164-97-5	**	10.5±0.1	EI	4518
			**	10.5±1	EI	3819
			**	11.5±0.5	EI	4098
	P ₂ O ₃	1314-24-5		15.4±1.0	EI	4098
	LaPO ₄	XXXXX-XX-X		10.4±0.5	EI	5603
O₃P₂⁺						
	P ₂ O ₃	1314-24-5	**	10.4±0.5	EI	4098
O₄P₂⁺						
	P ₂ O ₄	XXXXX-XX-X	**	10.8±1.0	EI	4098
O₅P₂⁺						
	P ₂ O ₅	1314-56-3	**	12.0±1.0	EI	4098
O₆P₃⁺						
	P ₃ O ₆	XXXXX-XX-X	**	12.3±1.0	EI	4098
O₇P₃⁺						
	P ₄ O ₉	XXXXX-XX-X		15.0±1.0	EI	4098
O₆P₄⁺						
	P ₄ O ₆	10248-58-5	**	10.55 (V)	PE	5343
	(2,4,6,8,9,10-Hexaoxa-1,3,5,7-tetraphosphatricyclo[3.3.1.1 ^{3,7}]decane)					
O₇P₄⁺						
	P ₄ O ₇	12065-80-4	**	11.4±0.5	EI	4098
O₈P₄⁺						
	P ₄ O ₈	12037-06-8	**	11.9±0.5	EI	4098
O₉P₄⁺						
	P ₄ O ₉	XXXXX-XX-X	**	12.4±0.5	EI	4098
O₁₀P₄⁺						
	P ₄ O ₁₀	16752-60-6	**	13.0±0.5	EI	4098
	(2,4,6,8,9,10-Hexaoxa-1,3,5,7-tetraphosphatricyclo[3.3.1.1 ^{3,7}]decane,1,3,5,7-tetraoxide)					
			**	13.0±0.5	EI	4098
CH₄OP⁺						
	(CH ₃ O) ₂ P(CH ₃ S)S	2953-29-9		13.40±0.30	EI	3989
C₂H₇OP⁺						
	(CH ₃) ₂ P(O)H	7211-39-4	**	10.32 (V)	PE	5523
C₃H₉OP⁺						
	(CH ₃) ₃ PO	676-96-0	**	9.88 (V)	PE	5442
			**	9.89 (V)	PE	5368
			**	9.9	PE	5529
C₁₉H₂₉OP⁺						
	CH ₃ OC ₆ H ₄ P(C ₆ H ₁₁) ₂	40438-63-9	**	7.88 (V)	PE	5417
	(Phosphine,dicyclohexyl(4-methoxyphenyl)-)					
CH₄O₂P⁺						
	(CH ₃ O) ₂ PO	512-56-1	2HCHO+H	14.90±0.20	EI	3989
	(CH ₃ O) ₂ P(CH ₃ S)O	152-20-5	CH ₃ S+HCHO	12.25±0.20	EI	3989
	(CH ₃ O) ₂ P(CH ₃ S)S	2953-29-9	CH ₃ S+HCHS	12.75±0.20	EI	3989
	(CH ₃ S) ₂ P(CH ₃ O)O	22608-53-3	CH ₃ S+HCHS	11.90±0.10	EI	3989

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
CH₅O₂P⁺						
	(CH ₃ O) ₃ PO	512-56-1	2HCHO	12.91±0.10	EI	3989
	(CH ₃ O) ₂ P(CH ₃ S)O	152-20-5	HCHS+HCHO	12.35±0.20	EI	3989
C₂H₆O₂P⁺						
	(CH ₃ O) ₂ P(CH ₃ S)S	2953-29-9		10.40±0.10	EI	3989
C₆H₁₁PO₂⁺						
	PO(CH=CH ₂) ₂ (OC ₂ H ₅)	30594-15-1	**	10.23 (V)	PE	5021
C₇H₁₅PO₂⁺						
	C ₇ H ₁₅ PO ₂	71431-36-2	**	10.04 (V)	PE	5021
C₁₉H₃₅O₂P⁺						
	C ₅ H ₂ P(OCH ₃) ₂ (C ₄ H ₉) ₃ (Phosphorin, 2,4,6-tris(1,1-dimethylethyl)-1,1-dihydro-1,1-dimethoxy-)	37912-85-9	**	6.7 (V)	PE	4053
C₂₁H₃₁O₂P⁺						
	C ₂ H ₅ COOC ₆ H ₄ P(C ₆ H ₁₁) ₂ (Benzoic acid, 4-(dicyclohexylphosphino)-ethyl ester)	40438-59-3	**	8.12 (V)	PE	5417
C₂₅H₂₃O₂P⁺						
	C ₅ H ₂ P(C ₆ H ₅) ₃ (OCH ₃) ₂ (Phosphorin, 1,1-dihydro-1,1-dimethoxy-2,4,6-triphenyl-)	20995-67-9	**	6.60 (V)	PE	5271
CH₄O₃P⁺						
	(CH ₃ O) ₃ PO	512-56-1	HCHO+CH ₃	13.90±0.20	EI	3989
	(CH ₃ O) ₂ P(CH ₃ S)O	152-20-5	HCHS+CH ₃	13.20±0.20	EI	3989
C₂H₆O₃P⁺						
	(CH ₃ O) ₂ PO	31682-64-1	**	11.0 (V)	PE	5190
	(CH ₃ O) ₃ PO	512-56-1	HCHO+H	14.1±0.20	EI	3989
	(CH ₃ O) ₂ P(CH ₃ S)O	152-20-5	CH ₃ S	11.90±0.10	EI	3989
C₂H₇O₃P⁺						
	HPO(OCH ₃) ₂	868-85-9	**	10.53	PE	5032
	(CH ₃ O) ₃ PO	512-56-1	HCHO	11.62±0.10	EI	3989
	(CH ₃ O) ₂ P(CH ₃ S)O	152-20-5	HCHS	11.00±0.10	EI	3989
C₃H₇O₃P⁺						
	C ₂ H ₄ O ₂ P(OCH ₃) (1,3,2-Dioxaphospholane, 2-methoxy-)	XXXXX-XX-X	**	9.06±0.1	PE	5042
C₃H₉O₃P⁺						
	P(OCH ₃) ₃	121-45-9	**	8.50	PE	5516
			**	9.0 (V)	PE	5190
			**	9.21	PE	5602
			**	9.22 (V)	PE	4705
	CH ₃ PO(OCH ₃) ₂	756-79-6	**	10.00	PE	5032
C₄H₇O₃P⁺						
	C ₄ H ₇ O ₃ P (2,6,7-Trioxa-1-phosphabicyclo[2.2.2]octane)	280-45-5	**	9.42±0.1	PE	5042
C₄H₉O₃P⁺						
	C ₃ H ₆ O ₃ P(OCH ₃) (1,3,2-Dioxaphosphorinane, 2-methoxy-)	XXXXX-XX-X	**	8.74±0.1	PE	5042
	PO(OCH ₃) ₂ (CH=CH ₂)	4645-32-3	**	10.94 (V)	PE	5021
C₄H₁₁O₃P⁺						
	HPO(OC ₂ H ₅) ₂	762-04-9	**	10.31	PE	5032

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₅H₁₁O₃P⁺	PO(OCH ₃) ₂ (CH ₂ CH = CH ₂)	757-54-0	**	9.96 (V)	PE	5021
C₆H₁₃O₃P⁺	C ₃ H ₄ O ₂ P(CH ₃) ₂ (OCH ₃) (1,3,2-Dioxaphosphorinane, 2-methoxy-4,6-dimethyl-(2α,4α,6α)-)	7735-82-2	**	8.34±0.1	PE	5042
	C ₃ H ₄ O ₂ P(CH ₃) ₂ (OCH ₃) (1,3,2-Dioxaphosphorinane, 2-methoxy-4,6-dimethyl-(2β,4α,6α)-)	41821-91-4	**	8.69±0.1	PE	5042
	PO(OC ₂ H ₅) ₂ (CH = CH ₂)	682-30-4	**	10.6 (V)	PE	5021
			**	10.6 (V)	PE	5328
C₆H₁₅O₃P⁺	(C ₂ H ₅ O) ₃ P	122-52-1	** ** **	8.8 (V) 8.92 (V) 9.15	PE PE PE	5190 4705 5602
C₉H₂₁O₃P⁺	(iso-C ₃ H ₇ O) ₃ P	116-17-6	**	8.76 (V)	PE	5139
C₁₈H₁₅O₃P⁺	(C ₆ H ₅ O) ₃ P (Phosphorous acid triphenyl ester)	101-02-0	**	8.80 (V)	PE	5139
C₂₁H₂₁O₃P⁺	(CH ₃ OC ₆ H ₄) ₃ P (Phosphine, tris(2-methoxyphenyl)-)	4731-65-1	**	7.37 (V)	PE	5438
	(CH ₃ OC ₆ H ₄) ₃ P (Phosphine, tris(3-methoxyphenyl)-)	29949-84-6	**	7.72 (V)	PE	5438
	(CH ₃ OC ₆ H ₄) ₃ P (Phosphine, tris(4-methoxyphenyl)-)	855-38-9	**	7.48 (V)	PE	5438
C₉H₈O₄P⁺	(CH ₃ O) ₃ PO	512-56-1	H	12.73±0.20	EI	3989
C₉H₉O₄P⁺	(CH ₃ O) ₃ PO	512-56-1	** ** ** ** **	9.99 10.8 (V) 10.81 (V) 10.82 (V) 10.70±0.10	PE PE PE PE EI	5516 5190 5624 4705 3989
C₆H₁₅O₄P⁺	(C ₂ H ₅ O) ₃ PO	78-40-0	** ** **	9.79 10.4 (V) 10.54 (V)	PE PE PE	5516 5190 5624
H₆N₃OP⁺	(NH ₂) ₃ PO	13597-72-3	**	10.00±0.05	EI	4759
C₅H₁₃N₂OP⁺	C ₂ H ₄ N ₂ P(CH ₃) ₂ OCH ₃ (1,3,2-Diazapholidine, 2-methoxy-1,3-dimethyl-)	7137-86-2	**	8.12 (V)	PE	5477
C₂H₁₀N₃OP⁺	((CH ₃) ₂ N)(NH ₂) ₂ PO	19316-37-1	**	8.85±0.05	EI	4759
C₃H₁₂N₃OP⁺	(CH ₃ NH) ₃ PO	6326-72-3	**	9.10±0.05	EI	4759
C₄H₁₁N₃OP⁺	((CH ₃) ₂ N) ₂ (NH ₂)PO ((CH ₃) ₂ N)(CH ₃ NH) ₂ PO	3732-86-3 16853-36-4	** **	8.60±0.05 8.75±0.05	EI EI	4759 4759

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₅H₁₆N₃OP⁺	((CH ₃) ₂ N) ₂ (CH ₃ NH)PO	10159-46-3	**	8.55±0.05	EI	4759
C₆H₁₈N₃OP⁺	OP(N(CH ₃) ₂) ₃	630-31-9	**	7.82 (V)	PE	5624
	((CH ₃) ₂ N) ₃ PO	680-31-9	**	8.7 (V)	PE	5190
			**	8.35±0.05	EI	4759
C₁₈H₂₆NO₂P⁺	NO ₂ C ₆ H ₄ P(C ₆ H ₁₁) ₂ (Phosphine,dicyclohexyl(4-nitrophenyl)-)	40438-56-0	**	8.39 (V)	PE	5417
C₂H₈N₂O₂P⁺	OP(OC ₂ H ₅)NHNH ₂	XXXXX-XX-X	**	7.95	PE	5627
C₈H₂₀NO₃P⁺	(N(C ₂ H ₅) ₂)PO(OC ₂ H ₅) ₂	3167-69-9	**	8.69	PE	5032
C₄H₁₃N₂O₃P⁺	OP(OC ₂ H ₅) ₂ NHNH ₂	56183-69-8	**	10.90 (V)	PE	5627
FP⁺	PF ₃	7783-55-3	2F	21.0±0.3	EI	4543
	PF ₂ CN	14118-40-2	F+CN	19.1±0.2	EI	4543
F₂P⁺	PF ₃	7783-55-3	F	13.5±0.1	EI	4305
			F	15.4±0.2	EI	4543
	P ₂ F ₄	13824-74-3	PF ₂	10.9±0.1	EI	4305
	PF ₂ H	14984-74-8	H	11.7±0.1	EI	4305
	PF ₂ CN	14118-40-2	CN	13.4±0.2	EI	4543
	PF ₂ I	13819-11-9	I	10.8±0.1	EI	4305
F₃P⁺	PF ₃	7783-55-3	**	11.5±0.1	PI	4543
			**	11.56	PE	5453
			**	11.57±0.01	PE	3703
			**	11.66±0.01	PE	3641
			**	12.20 (V)	PE	5602
			**	12.23±0.02 (V)	PE	3662
			**	12.28 (V)	PE	5539
			**	11.4±0.2	EI	4543
			**	11.6±0.1	EI	4305
				11.65	EI	5462
			**	11.72±0.1	EI	3578
F₅P⁺	PF ₅	7647-19-0	**	15.54 (V)	PE	3872
			**	15.6 (V)	PE	3669
F₄P₂⁺	P ₂ F ₄	13824-74-3	**	9.64 (V)	PE	3662
			**	9.3±0.1	EI	4305
HF₂P⁺	PF ₂ H	14984-74-8	**	11.0±0.1 (V)	PE	3662
			**	10.5±0.1	EI	4305
H₃BF₃P⁺	(PF ₃)(BH ₃)	14931-39-6	**	11.02±0.03	PE	3699

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
H₅B₃F₃P⁺	B ₃ H ₇ PF ₃	11126-95-7		10.8±0.3	EI	3652
CFP⁺ (² Π)	FC=P	65756-42-5	**	10.57±0.01	PE	4836
C₃F₉P⁺	(CF ₃) ₃ P	432-04-2	** ** **	11.70 (V) 11.70 (V) 11.70 (V)	PE PE PE	4191 4371 4261
C₄F₁₂P₂⁺	((CF ₃) ₂ P) ₂ - <i>trans</i> ((CF ₃) ₂ P) ₂ - <i>gauche</i>	2714-60-5	** ** **	10.71 (V) 10.71 (V) 11.57 (V)	PE PE PE	4191 4185 4185
C₆F₁₂P₂⁺	C ₂ P ₂ (CF ₃) ₄ (1,2-Diphosphete, 1,2-dihydro-1,2,3,4-tetrakis(trifluoromethyl)-)	2375-86-2	**	10.97 (V)	PE	4191
C₆F₁₅P₃⁺	(C ₂ F ₅ P) ₃ (Tetraphosphirane, tris(pentafluoroethyl)-)	29634-17-1	**	10.39 (V)	PE	4942
C₄F₁₂P₄⁺	P ₄ (CF ₃) ₄ (Tetraphosphetane, tetrakis(trifluoromethyl)-)	393-02-2	** **	10.18 (V) 10.18 (V)	PE PE	4191 4942
C₈F₂₀P₄⁺	(C ₂ F ₅ P) ₄ (Tetraphosphetane, tetrakis(pentafluoroethyl)-)	35449-91-3	**	9.99 (V)	PE	4942
C₅F₁₅P₅⁺	P ₅ (CF ₃) ₅ (Pentaphospholane, pentakis(trifluoromethyl)-)	745-23-3	** **	9.71 (V) 9.79 (V)	PE PE	4191 4942
C₂H₆FP⁺	(CH ₃) ₂ PF	507-15-3	**	9.35 (V)	PE	4474
C₈H₁₈FP⁺	(<i>tert</i> -C ₄ H ₉) ₂ PF	29146-24-5	**	8.50 (V)	PE	4474
CH₃F₂P⁺	CH ₃ PF ₂	753-59-3	**	10.35 (V)	PE	4474
C₄H₉F₂P⁺	<i>tert</i> -C ₄ H ₉ PF ₂	29149-32-4	**	9.65 (V)	PE	4474
C₅H₅F₂P⁺	C ₅ H ₅ (PF ₂) ₂ (Phosphonous difluoride, 2,4-cyclopentadien-1-yl-)	36917-22-3	**	9.2 (V)	PE	4373
C₂₃H₁₇F₂P⁺	C ₅ H ₂ P(C ₆ H ₅) ₃ F ₂ (Phosphorin, 1,1-difluoro-1,1-dihydro-2,4,6-triphenyl-)	40425-79-4	**	7.15 (V)	PE	5271
CH₂F₃P⁺	H ₂ PCF ₃	420-52-0	** **	11.15±0.05 (V) 11.18 (V)	PE PE	5419 4371

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₈H₁₂F₃P⁺			*			
	(FC ₆ H ₄) ₃ P (Phosphine, tris(3-fluorophenyl)-)	23039-94-3	**	8.32 (V)	PE	5438
	(FC ₆ H ₄) ₃ P (Phosphine, tris(4-fluorophenyl)-)	18437-78-0	**	8.12 (V)	PE	5438
C₂HF₆P⁺						
	(CF ₃) ₂ PH	460-96-8	**	11.50 (V)	PE	4371
			**	11.51 (V)	PE	4185
C₂₁H₁₂F₉P⁺						
	(CF ₃ C ₆ H ₄) ₃ P (Phosphine, tris[2-(2-trifluoromethyl)phenyl]-)	25688-42-0	**	8.30 (V)	PE	5438
	(CF ₃ C ₆ H ₄) ₃ P (Phosphine, tris(4-trifluoromethyl)phenyl)-)	13406-29-6	**	8.65 (V)	PE	5438
C₄H₆F₆P₂⁺						
	(CH ₃) ₂ PP(CF ₃) ₂	666-62-6	**	9.37 (V)	PE	4191
NF₅P₂⁺						
	PF ₂ (NPF ₃)	34118-39-3	**	11.2 (V)	PE	5398
NF₆P₃⁺						
	(F ₂ P) ₃ N	56564-56-8	**	11.2±0.1 (V)	PE	4378
N₃F₆P₃⁺						
	N ₃ P ₃ F ₆	XXXXX-XX-X	**	11.4	PE	5295
H₂NF₂P⁺						
	F ₂ PNH ₂	25757-74-8	**	10.9±0.1 (V)	PE	4378
			**	10.9 (V)	PE	3662
H₃N₂F₂P⁺						
	PHF ₂ (NH ₂) ₂	60448-09-1	**	10.7 (V)	PE	4622
HNF₄P₂⁺						
	(F ₂ P) ₂ NH	34326-59-5	**	11.3±0.1 (V)	PE	4378
HNF₆P₂⁺						
	NH(PF ₂)(PF ₃)	71481-55-5	**	11.6 (V)	PE	5398
HBNF₄P⁺						
	BF ₂ [NH(PF ₃)]	60073-67-8	**	11.5±0.1 (V)	PE	4504
CNFP⁺						
	PF ₂ CN	14118-40-2	F	15.7±0.2	EI	4543
CNF₂P⁺						
	PF ₂ CN	14118-40-2	**	11.9±0.1 (V)	PE	3662
			**	11.7±0.2	EI	4543
C₂₇H₂₇NFP⁺						
	C ₆ H ₂ P(C ₆ H ₅) ₃ (N(C ₂ H ₅) ₂)F (Phosphorin, 1-(diethylamino)-1-fluoro-1,1-dihydro-2,4,6-triphenyl-)	40425-24-9	**	6.50 (V)	PE	5271
C₄H₁₂N₂FP⁺						
	((CH ₃) ₂ N) ₂ PF	1735-82-6	**	8.18 (V)	PE	3825
C₂H₆NF₂P⁺						
	(CH ₃) ₂ NPF ₂	814-97-1	**	9.58 (V)	PE	3825
			**	9.6 (V)	PE	3662
			**	9.60 (V)	PE	4474

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₂H₆NF₂P⁺	(CH ₃) ₂ NPF ₂	814-97-1	**	10.2±0.3	EI	3652
C₄H₁₀NF₂P⁺	(C ₂ H ₅) ₂ NPF ₂	363-84-8	**	9.45 (V)	PE	4474
C₆H₁₈N₃F₂P⁺	((CH ₃) ₂ N) ₃ PF ₂	7549-83-9	**	8.04 (V)	PE	3825
C₄H₁₂N₂F₃P⁺	((CH ₃) ₂ N) ₂ PF ₃	1735-83-7	**	8.84 (V)	PE	3825
C₂H₆NF₄P⁺	(CH ₃) ₂ NPF ₄	2353-98-2	**	10.35 (V)	PE	3825
C₄H₆NF₆P⁺	(CH ₃) ₂ NP(CF ₃) ₂	432-01-9	**	9.56 (V)	PE	4261
CH₃NF₄P₂⁺	CH ₃ N(PF ₂) ₂	17648-18-9	**	10.95 (V)	PE	5376
C₂H₆N₂P₂F₆⁺	(CH ₃ NPF ₃) ₂ (1,3,2,4-Diazadiphosphetidine, 2,2,2,4,4,4-hexafluoro- 2,2,4,4-tetrahydro-1,3-dimethyl-)	3880-04-4	**	9.80	EI	5462
C₂₄H₂₀N₃F₂P₃⁺	(C ₆ H ₅) ₄ P ₃ N ₃ F ₂ (1,3,5,2,4,6-Triazatriphosphorine,2,4-difluoro-2,2,4,4,6,6-hexahydro- 2,4,6,6-tetraphenyl-)	73502-98-4	**	8.59	PE	5443
C₁₂H₁₀N₃F₄P₃⁺	(C ₆ H ₅) ₂ P ₃ N ₃ F ₄ (1,3,5,2,4,6-Triazatriphosphorine,2,2,4,6-tetrafluoro-2,2,4,4,6,6- hexahydro-4,4-diphenyl-) cis-(C ₆ H ₅) ₂ P ₃ N ₃ F ₄ (1,3,5,2,4,6-Triazatriphosphorine,2,2,4,6-tetrafluoro-2,2,4,4,6,6- hexahydro-4,6-diphenyl-)	XXXXX-XX-X	**	9.64 (V)	PE	5443
C₆H₅N₃F₅P₃⁺	C ₆ H ₅ P ₃ N ₃ F ₅ (1,3,5,2,4,6-Triazatriphosphorine,2,2,4,4,6-pentafluoro-2,2,4,4,6,6- hexahydro-6-phenyl-)	2713-48-6	**	10.07 (V)	PE	5443
C₈H₁₀N₄F₅P₃⁺	(C ₆ H ₄ N(CH ₃) ₂)P ₃ N ₃ F ₅ (1,3,5,2,4,6-Triazatriphosphorine,2-[4-(dimethylamino) phenyl]-2,4,4,6,6-pentafluoro-2,2,4,4,6,6-hexahydro-)	53968-86-8	**	7.88 (V)	PE	5443
BC₂H₉NF₂P⁺	(CH ₃) ₂ NF ₂ PB ₃ H ₃ ?	2851-73-2	**	12.2±0.3	EI	3652
B₃C₂H₁₁NF₂P⁺	(CH ₃) ₂ NF ₂ PB ₃ H ₇	11126-93-5		10.4±0.3	EI	3652
B₃C₂H₁₂NF₂P⁺	(CH ₃) ₂ NF ₂ PB ₃ H ₇	11126-93-5	H	10.5±0.3	EI	3652
B₄C₂H₁₂NF₂P⁺	(CH ₃) ₂ NF ₂ PB ₄ H ₈	12602-24-3		10.0±0.3	EI	3652

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
B₄C₂H₁₄NF₂P⁺	(CH ₃) ₂ NF ₂ PB ₄ H ₈	12602-24-3	**	9.6±0.3	EI	3652
OF₃P⁺	POF ₃	13478-20-1	**	12.77±0.04	PE	3641
OF₄P₂⁺	PF ₂ OPF ₂	13812-07-2	**	11.2 (V)	PE	3662
CH₃O₂F₂P⁺	PF ₂ O(OCH ₃)	22382-13-4	**	12.64 (V)	PE	4699
C₆H₆O₃F₉P⁺	(CF ₃ CH ₂ O) ₃ P	370-69-4	**	10.37 (V)	PE	4705
CNOF₂P⁺	PF ₂ NCO	461-59-6	**	11.05±0.02 (V)	PE	3662
C₅H₁₂N₂OF₂P⁺	CN ₂ P(=O)F(CH ₃) ₄ (1,3,2-Diazaphosphetidin-4-one, 2-fluoro-2,2-dihydro- 1,2,2,3-tetramethyl-)	32707-18-9	**	8.70±0.1	EI	5462
C₄H₉N₂OF₂P⁺	CN ₂ P(=O)F ₂ (CH ₃) ₃ (1,3,2-Diazaphosphetidin-4-one, 2,2-difluoro-2,2-dihydro- 1,2,3-trimethyl-)	31053-08-4	**	9.00±0.1	EI	5462
C₅H₁₁N₂OF₂P⁺	CN ₂ P(=O)F ₂ (CH ₃) ₂ C ₂ H ₅ (1,3,2-Diazaphosphetidin-4-one, 2-ethyl-2,2-difluoro- 2,2-dihydro-1,3-dimethyl-)	31053-09-5	**	8.90±0.1	EI	5462
C₉H₁₁N₂OF₂P⁺	C ₆ H ₅ CN ₂ P(=O)F ₂ (CH ₃) ₂ (1,3,2-Diazaphosphetidin-4-one, 2,2-difluoro-2,2- dihydro-1,2-dimethyl-3-phenyl-)	31053-06-2	**	8.15±0.1	EI	5462
	C ₆ H ₅ F ₂ CN ₂ P(=O)(CH ₃) ₂ (1,3,2-Diazaphosphetidin-4-one, 2,2-difluoro-2,2-dihydro- 1,3-dimethyl-2-phenyl-)	32707-15-6	**	8.80±0.1	EI	5462
C₁₀H₁₃N₂OF₂P⁺	C ₁₀ H ₁₃ N ₂ OF ₂ P (1,3,2-Diazaphosphetidin-4-one, 2-ethyl-2,2-difluoro- 2,2-dihydro-1-methyl-3-phenyl-)	31053-07-3	**	8.00±0.1	EI	5462
C₇H₁₆N₃OF₂P⁺	C ₇ H ₁₆ N ₃ OF ₂ P (1,3,2-Diazaphosphetidin-4-one, 2-(diethylamino)-2,2- difluoro-2,2-dihydro-1,3-dimethyl-)	32707-17-8	**	8.85±0.1	EI	5462
C₃H₆N₂OF₃P⁺	CN ₂ P(=O)F ₃ (CH ₃) ₂ (1,3,2-Diazaphosphetidin-4-one, 2,2,2-trifluoro- 2,2-dihydro-1,3-dimethyl-)	32707-12-3	**	9.60±0.1	EI	5462
ONaP⁺	NaPO	56730-08-6	**	7.7±0.5	EI	4518
O₂NaP⁺	NaPO ₂ NaPO ₂	XXXXX-XX-X	**	5.3±0.5	EI	4518
		XXXXX-XX-X	**	8.6	EI	4098

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
O₃NaP⁺	NaPO ₃ NaPO ₃	XXXXX-XX-X ** XXXXX-XX-X **		10.16±0.04 (V) 5.0±0.8	PE EI	4840 4518
SiP⁺	PSi	12137-64-3	**	9.1±0.5	EI	4102
Si₂P⁺	PSi ₂	37347-46-9	**	8.4±0.5	EI	4102
SiP₂⁺	P ₂ Si	12137-68-7	**	9.0±0.5	EI	4102
H₅SiP⁺	SiH ₃ PH ₂	14616-47-8	**	9.9±0.1 (V)	PE	3661
H₉Si₃P⁺	(SiH ₃) ₃ P	15110-33-5	**	9.3±0.1 (V)	PE	3661
CSiP⁺	CSiP	37342-74-8	**	8.9±0.5	EI	4102
C₇H₁₉SiP⁺	(CH ₃) ₃ P=CHSi(CH ₃) ₃	3272-86-4	** ** **	6.80 6.81 (V) 6.81 (V)	PE PE PE	3782 4181 5368
C₂₂H₂₅SiP⁺	(CH ₃) ₃ SiCH=P(C ₆ H ₅) ₃ (Phosphorane, triphenyl[(trimethylsilyl)methylene]-)	3739-97-7	**	6.71 (V)	PE	4579
C₉H₂₅Si₂P⁺	(CH ₃) ₃ SiSi(CH ₃) ₂ CH=P(CH ₃) ₃	29947-67-9	** **	6.85 (V) 6.87	PE PE	4181 3782
C₁₀H₂₇Si₂P⁺	((CH ₃) ₃ Si) ₂ C=P(CH ₃) ₃	3607-03-2	** **	6.92 (V) 6.92 (V)	PE PE	4181 5368
C₂₇H₃₉Si₃P⁺	((CH ₃) ₃ SiC ₆ H ₅) ₃ P (Phosphine, tris[4-trimethylsilyl]phenyl)-)	18848-96-9	**	7.67 (V)	PE	5438
C₁₄H₃₆Si₃P⁺	C ₂ Si ₃ (CH ₃) ₆ =P(CH ₃) ₃ (Phosphorane, (1,1,2,2,4,4-hexamethyl-1,2,4-trisilacyclopentane-3,5-diylidene)bis(trimethyl-)	51685-13-3	**	6.11 (V)	PE	4181
C₆H₁₈NSiP⁺	(CH ₃) ₃ SiN=P(CH ₃) ₃	6063-72-5	** **	8.30 (V) 8.30 (V)	PE PE	4181 5442
C₂₁H₂₄NSiP⁺	(C ₆ H ₅) ₃ PNSi(CH ₃) ₃ (Silanamine, 1,1,1-trimethyl-N-(triphenylphosphoranylidene)-)	13892-06-3	**	8.05 (V)	PE	5442
C₁₁H₃₁NSi₂P₂⁺	C ₁₁ H ₃₁ NSi ₂ P ₂ ⁺	39980-56-8	**	6.18 (V)	PE	4181
H₂F₃SiP⁺	H ₂ PSiF ₃	51518-19-5	**	11.06±0.05 (V)	PE	5419

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
H₆NF₂SiP⁺	PF ₂ [N(SiH ₃) ₂]	71579-71-0	**	10.8 (V)	PE	4988
H₃NF₄P₂Si⁺	N(PF ₂) ₂ (SiH ₃)	71579-72-1	**	11.2 (V)	PE	4988
S⁺						
	S	7704-34-9	** ** ** ** ** ** ** ** ** ** S ₂	10.36 10.3±0.3 10.4±0.3 10.4±0.3 10.5±0.3 10.5±0.3 10.5±0.3 10.5 ~11±0.5 13.5±0.5	S EI EI EI EI EI EI EI EI EI	4864 3449 4486 4874 3616 4580 4864 4544 3448 5229
	H ₂ S	7783-06-4	H ₂	13.5	EI	3967
	CS ₂	75-15-0		14.80±0.02	PI	4936
	(⁴ S _{1/2})		CS	14.80±0.02	PI	5435
	(⁴ S _{1/2})		CS	14.88±0.05	EI	4905
			CS(X ¹ Σ ⁺)	13.35	EI	4897
				13.40±0.08	EI	5242
			CS	15±1	EI	3812
			CS	17±1	EI	3812
	(⁴ S _{1/2})	SO ₂	SO	16.334	PE	5388
	COS	463-58-1	CO	13.52±0.05	EI	4905
			CO	13.7	EI	3779
	SCl ₂	10545-99-0		13.0±0.2	EI	4287
S₂⁺						
	S ₂	23550-45-0	** ** ** ** ** ** ** ** ** ** CS ₂	9.42±0.10 9.8±0.3 9.8±0.5 9.30 9.38±0.01 9.55 (V) 9.56 (V) 11.28 13.06 9.36±0.02 9.42±0.1 9.8±0.3 10.1±0.3 9.38±0.03	EI EI EI PE PE PE PE PE PE EI EI EI EI EI OTR PI PI	3616 4874 3615 5475 4370 4550 5475 5475 4920 4554 4486 5229 5435 5435 4936
	(² Π _{3/2,k})					
	(² Π _{1/2,k})					
	(² Π _{k,3/2})					
	(² Π _u)					
	(² Σ _k)					
	C ₃ H ₆ S ₂ (1,3-Dithiolane)	4829-04-3	CH ₂ =CHCH ₃	10.7±0.1	EI	3598
	S ₂ F ₂	13709-35-8		17.6±0.4	EI	3738
S₈⁺						
	S ₈	10544-50-0	** **	9.23 (V) 9.40 (V)	PE PE	3846 4411
HS⁺						
	H ₂ S	7783-06-4	H H	14.4 14.7±0.2	EI EI	3967 4610
H₂S⁺						
(² B ₁)	H ₂ S	7783-06-4	** **	10.466±0.002 12.777±0.005	S S	5060 5060

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
H₂S⁺						
(² B ₂)	H ₂ S	7783-06-4	**	14.643	S	5060
(² B ₁)			**	10.5	PI	5479
(² A ₁)			**	12.8	PI	5479
(² B ₂)			**	14.8	PI	5479
(² B ₁)			**	10.43	PE	4073
			**	10.43 (V)	PE	4276
(² B ₁)			**	10.47	PE	3719
(² A ₁)			**	12.752	PE	3515
(² A ₁)			**	12.78	PE	3719
(² B ₂)			**	14.78	PE	3719
(² A ₁)			**	22.0±0.2 (V)	PE	5269
(² A ₁)			**	22.2 (V)	PE	3719
(² A ₁)			**	23.3±0.2 (V)	PE	5269
			**	10.45	EI	3967
			**	10.56±0.05	EI	4610
H₃S⁺						
	C ₂ H ₅ SH	75-08-1		12.41±0.02	EI	3487
	(CH ₃) ₂ S	75-18-3		14.14±0.02	EI	3487
H₂S₂⁺						
	H ₂ S ₂	13465-07-1	**	10.01 (V)	PE	4276
HBS⁺						
(² Π)	HBS	14457-85-3	**	11.11±0.03	PE	3982
			**	11.12	PE	3871
(² Σ ⁺)			**	13.54±0.03	PE	3982
(² Σ ⁺)			**	15.83±0.1	PE	3982
H₉B₉S⁺						
	SB ₉ H ₉ (1-Thiadecaborane(9))	41646-56-4	**	10.3 (V)	PE	5324
H₁₁B₉S⁺						
	6-SB ₉ H ₁₁ (6-Thiadecaborane(11))	12447-77-7	**	9.8 (V)	PE	5324
H₁₁B₁₁S⁺						
	SB ₁₁ H ₁₁ (1-Thiadodecaborane(11))	56464-75-6	**	11.1 (V)	PE	5324
CS⁺						
	CS	2944-05-0	**	11.33±0.01	PI	4936
(² Σ _g ⁺)			**	11.33±0.01	PE	3691
			**	11.33±0.02	PE	3696
(² Σ _g ⁺)			**	11.33±0.02	PE	5208
(² Σ)			**	11.34±0.02	PE	3690
(² π)			**	12.78±0.02	PE	3690
(² π _u)			**	12.79±0.01	PE	3691
(² Π _u)			**	12.79±0.02	PE	5208
(² Σ)			**	15.83±0.02	PE	3690
(² Σ _u ⁺)			**	15.84±0.01	PE	3691
(² Σ _u ⁺)			**	18.00±0.01	PE	3691
(² Σ)			**	18.03±0.02	PE	3690
			**	11.0±0.03	EI	4920
			**	11.39±0.1	EI	4554
			**	11.39±0.10	EI	3616
	CS ₂	75-15-0	S ⁻	13.64±0.02	PI	4936
				15.75±0.02	PI	4936
(² Σ ⁺)			S ⁻	15.75±0.02	PI	5435
			S ⁻	13.90±0.1	EI	4905

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
CS⁺						
(² Σ^+)	CS ₂	75-15-0	S	15.94±0.07 14.10±0.08	EI EI	4905 5242
(² Σ^+ , ² Π)			S(² P _u)	14.5	EI	4897
(² Σ^+)			S(³ P _u)	14.7	EI	4897
			S	16.3±1	EI	3812
(² Σ^+)	COS	463-58-1	O O?	18.7±0.5 16.7	EI EI	4905 3779
CS₂⁺						
(² $\Pi_{g^{3/2}}$)	CS ₂	75-15-0	**	10.0685±0.0020	S	5439
(² $\Pi_{g^{1/2}}$)			**	10.1230±0.0020	S	5439
(² $\Pi_{1/2u}$)			**	12.586	S	3573
(² Π_u)			**	12.713	S	5048
(² $\Pi_{3/2}$)			**	10.070±0.006	PI	5299
(² $\Pi_{q^{3/2}}$)			**	10.074±0.005	PI	4936
(² $\Pi_{q^{1/2}}$)			**	10.076±0.005	PI	5435
(² Π_p)			**	10.077	PI	4994
(² $\Pi_{1/2}$)			**	10.125	PI	5299
(² $\Pi_{q^{1/2}}$)			**	10.131±0.005	PI	4936
(² $\Pi_{x^{1/2}}$)			**	10.132±0.005	PI	5435
(² Π_u)			**	12.696	PI	4994
(² Σ_u^+)			**	14.479	PI	4994
(² Σ_u^+)			**	14.48±0.02	PI	5435
(² Σ_v^+)			**	14.480±0.005	PI	4936
(² Σ_v^+)			**	16.184±0.005	PI	4936
(² Σ_v^+)			**	16.19±0.02	PI	5435
(² Σ_g^+)			**	16.192	PI	4994
(² Σ_u^+)			**	16.53±0.02	PI	4936
(² Π_p)			**	10.06±0.01	PE	3965
			**	10.06	PE	3697
(² $\Pi_{3/2}$)			**	10.06	PE	4073
(X ² $\Pi_{g^{3/2}}$)			**	10.074±0.002	PE	4979
			**	10.079±0.003	PE	5256
			**	10.10 (V)	PE	5055
(² Π_u)			**	12.67±0.01	PE	3965
(² Σ_u^+)			**	14.47±0.01	PE	3965
(² Σ_g^+)			**	16.18±0.01	PE	3965
			**	10.06±0.025	EI	5027
			**	10.05±0.08	EI	5242
			**	10.07±0.1	EI	4554
			**	10.07±0.10	EI	3616
(² Π_u)			**	12.620	OTH	5029
C₂S₄⁺						
(CS ₂) ₂		XXXXXX-XX-X **		9.36±0.02	PI	5439
(CS ₂) ₂		XXXXXX-XX-X **		~9.63	PI	5299
C₃S₆⁺						
(CS ₂) ₃		XXXXXX-XX-X **		9.22±0.02	PI	5439
C₄S₈⁺						
(CS ₂) ₄		XXXXXX-XX-X **		9.10±0.02	PI	5439
C₅S₁₀⁺						
(CS ₂) ₅		XXXXXX-XX-X **		9.04±0.02	PI	5439
CHS⁺						
C ₄ H ₄ S (Thiophene)		110-02-1	C ₃ H ₃	13.19±0.04	PE	5283
C ₄ H ₆ S ₂ (1,3-Dithiolane)		4829-04-3	CHS+CH ₄ ?	13±0.4	EI	3598

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
CHS⁺						
	(CH ₃) ₂ SO	67-68-5	H ₂ O + CH ₃	11.55 ± 0.2	EI	5311
	C ₃ H ₆ OS (1,3-Oxathiolane)	2094-97-5		12.9 ± 0.2	EI	3598
CH₂S⁺						
	HCHS	865-36-1	**	9.0 (V)	PE	4467
			**	9.33 (V)	PE	4323
			**	9.338 ± 0.010	PE	3697
			**	9.38 (V)	PE	4680
	CH ₃ SH	74-93-1	H ₂	10.8 ± 0.1	PI	4025
	(CH ₃) ₂ S	75-18-3	CH ₄	10.46 ± 0.08	PI	4025
	(C ₂ H ₅) ₂ S	352-93-2	C ₂ H ₄ + CH ₄	11.75 ± 0.03	PI	4025
	C ₃ H ₆ S ₂ (1,3-Dithiolane)	4829-04-3		11 ± 0.4	EI	3598
	C ₃ H ₆ OS (1,3-Oxathiolane)	2094-97-5		12.5 ± 0.2	EI	3598
CH₃S⁺						
	CH ₃ SH	74-93-1	H	11.37 ± 0.05	PI	4025
	(CH ₃) ₂ S	75-18-3	CH ₃	10.79 ± 0.04	PI	4025
	(C ₂ H ₅) ₂ S	352-93-2	C ₂ H ₄ + CH ₃	12.00 ± 0.05	PI	4025
	(CH ₃ CH ₂) ₂ CHSH	616-31-9	C ₂ H ₅ + C ₂ H ₄	12.1	EI	5316
	CH ₃ SCH(CH ₃)C ₂ H ₅	10359-64-5	C ₂ H ₅ + C ₂ H ₄	12.9	EI	5316
	C ₃ H ₆ S ₂ (1,3-Dithiolane)	4829-04-3		11.4 ± 0.4	EI	3598
	C ₂ H ₅ SOCH ₃	1669-98-3	C ₂ H ₄ + OH	12.23 ± 0.32	EI	5311
	(CH ₃ O) ₂ P(CH ₃ S)O	152-20-5		13.1 ± 0.30	EI	3989
	(CH ₃ S) ₂ P(CH ₃ O)O	22608-53-3		12.60 ± 0.20	EI	3989
CH₄S⁺						
	CH ₃ SH	74-93-1	**	9.44 ± 0.01	PI	4025
			**	9.415	PE	3697
			**	9.42	PE	3678
(² A'')			**	9.44	PE	4032
			**	9.44	PE	4087
			**	9.44 (V)	PE	3656
			**	9.44 (V)	PE	5632
C₂H₂S⁺						
	CH ₂ =C=S	18282-77-4	**	8.89 (V)	PE	4698
	C ₃ H ₄ S (Thiophene)	110-02-1	C ₂ H ₂	12.1 ± 0.1	PE	5283
C₂H₃S⁺						
	C ₃ H ₆ S ₂ (1,3-Dithiolane)	4829-04-3	CH ₃ S	10.8 ± 0.4	EI	3598
	C ₃ H ₆ OS (1,3-Oxathiolane)	2094-97-5	CH ₂ O + H	12.3 ± 0.1	EI	3598
C₂H₄S⁺						
	CH ₃ CHS	6851-93-0	**	8.98 ± 0.02 (V)	PE	4212
			**	9.3 (V)	PE	4467
	C ₂ H ₄ S (Thiirane)	420-12-2	**	9.051 ± 0.006	S	3882
			**	8.9 ± 0.1	PE	4990
			**	9.00	PE	3861
			**	9.05 (V)	PE	3837
	(C ₂ H ₅) ₂ S	352-93-2	C ₂ H ₆	9.89 ± 0.3	PI	4025
	C ₃ H ₆ S ₂ (1,3-Dithiolane)	4829-04-3	CH ₂ S	11.2 ± 0.3	EI	3598
	C ₃ H ₆ OS (1,3-Oxathiolane)	2094-97-5	CH ₂ O	10.5 ± 0.1	EI	3598

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₂H₅S⁺						
	(CH ₃) ₂ S	75-18-3	H	10.93±0.02	PI	4025
	(CH ₃) ₂ CHSH	75-33-2	CH ₃	11.0±0.15	EI	5058
	(C ₂ H ₅) ₂ S	352-93-2	C ₂ H ₅	10.23±0.03	PI	4025
	HSCH ₂ CH ₂ SH	26914-40-9	SH	10.4±0.15	EI	5058
	C ₃ H ₆ S ₂ (1,3-Dithiolane)	4829-04-3	CHS	11.4±0.3	EI	3598
	(CH ₃ S) ₂ CH ₂	1618-26-4	SCH ₃	10.1±0.15	EI	5058
	C ₃ H ₆ OS (1,3-Oxathiolane)	2094-97-5	CHO	10.4±0.1	EI	3598
	BrCH ₂ CH ₂ SH	24276-77-5	Br	10.1±0.15	EI	5058
C₂H₆S⁺						
	C ₂ H ₆ SH	75-08-1	**	9.29	PE	4032
	(CH ₃) ₂ S	75-18-3	**	8.687	S	4238
		**		8.706±0.010	S	3970
		**		8.69±0.01	PI	4025
		**		7.59 (V)	PE	5526
		**		8.5±0.1	PE	4990
		**		8.57±0.04	PE	3842
		**		8.65 (V)	PE	3678
		**		8.67	PE	3867
		**		8.67 (V)	PE	4276
		**		8.67 (V)	PE	5632
		**		8.7	PE	4104
		**		8.71 (V)	PE	3656
		**		8.71 (V)	PE	4884
		**		8.71 (V)	PE	5538
	(C ₂ H ₅) ₂ S	352-93-2	C ₂ H ₄	9.90±0.03	PI	4025
C₃HS⁺						
	C ₄ H ₆ S (Thiophene)	110-02-1	CH ₃	12.95±0.05	PE	5283
C₃H₅S⁺						
	C ₃ H ₆ S ₂ (1,3-Dithiolane)	4829-04-3	SH	10.5±0.1	EI	3598
C₃H₆S⁺						
	CH ₂ =CHCH ₂ SH	870-23-5	**	9.25	PE	3864
		**		9.25 (V)	PE	5427
	CH ₂ =CHSCH ₃	1822-74-8	**	8.44 (V)	PE	4246
		**		8.45 (V)	PE	4291
		**		8.45 (V)	PE	4638
		**		8.45 (V)	PE	5632
	(CH ₃) ₂ CS	4756-05-2	**	8.6 (V)	PE	4467
		**		8.60±0.05 (V)	PE	4212
	C ₃ H ₆ S (Thietane)	287-27-4	**	8.65±0.01	PI	5531
	C ₂ H ₅ SCH ₃ (Thiirane, methyl-)	1072-43-1	**	8.88 (V)	PE	4747
C₃H₇S⁺						
	(CH ₃) ₃ CSH	75-66-1	CH ₃	11.4±0.15	EI	5316
	(C ₂ H ₅) ₂ S	352-93-2	CH ₃	10.16±0.05	PI	4025
		CH ₃		10.7±0.15	EI	5316
	(CH ₃ CH ₂) ₂ CHSH	616-31-9	C ₂ H ₅	10.6±0.15	EI	5316
	CH ₃ SCH(CH ₃)C ₂ H ₅	10359-64-5	C ₂ H ₅	10.3±0.15	EI	5316
	BrCH ₂ CH ₂ CH ₂ SH	XXXXX-XX-X	Br	9.5±0.15	EI	5316
C₃H₈S⁺						
	C ₂ H ₅ SCH ₃	624-89-5	**	8.46	CTS	4272

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$C_3H_8S^+$	<i>n</i> - C_3H_7SH <i>iso</i> - C_3H_7SH	107-03-9	**	9.19	PE	4032
		75-33-2	**	9.14	PE	4032
$C_4H_8S^+$	C_4H_8S (Thiophene)	110-02-1	H	12.93 ± 0.07	PE	5283
$C_4H_8S^+$	C_4H_8S (Thiophene)	110-02-1	**	8.874 ± 0.005	S	3731
			**	8.86 ± 0.01	PI	4058
			**	8.85 (V)	PE	4690
			**	8.87 ± 0.01	PE	5283
			**	8.87 (V)	PE	3858
			**	8.90	PE	4017
			**	8.90 (V)	PE	5405
			**	~ 8.8	EI	4656
			**	8.80 ± 0.05	EI	4316
			**	9.05	CTS	3787
			**	9.12 ± 0.05	EI	3482
$C_4D_8S^+$	C_4D_8S (Thiophene- <i>d</i> ₄)	2036-39-7	**	8.874 ± 0.005	S	3731
$C_4H_6S^+$	$(CH_2=CH)S$	627-51-0	**	8.25 ± 0.01	PI	5531
			**	7.61 (V)	PE	5526
	$C_2H_3SCH=CH_2$ (Thiirane, ethenyl-)	5954-75-6	**	8.89 (V)	PE	4747
	C_4H_6S (Thiophene, 2,5-dihydro-)	1708-32-3	**	8.54 (V)	PE	3995
$C_4H_8S^+$	$CH_2=CHSC_2H_5$ $CH_3SCH_2CH=CH_2$	627-50-9	**	8.21 ± 0.01	PI	5531
		10152-76-8	**	8.6	PE	4104
			**	8.65 (V)	PE	4211
	C_4H_8S (Thiophene, tetrahydro-)	110-01-0	**	8.40 (V)	PE	3995
			**	8.42 (V)	PE	4145
			**	8.62 ± 0.05	EI	3498
			**	8.62	EI	5292
$C_4H_9S^+$	$(C_2H_5)_2S$	352-93-2	H	10.2 ± 0.1	PI	4025
$C_4H_{10}S^+$	$(C_2H_5)_2S$	352-93-2	**	8.42 ± 0.01	PI	4025
			**	7.45 (V)	PE	5526
			**	8.44 (V)	PE	4276
			**	8.44 (V)	PE	5632
			**	8.41	CTS	4272
	<i>n</i> - C_4H_9SH	109-79-5	**	9.15	PE	4032
	<i>sec</i> - C_4H_9SH	513-53-1	**	9.10	PE	4032
	<i>iso</i> - C_4H_9SH	513-44-0	**	9.12	PE	4032
	<i>tert</i> - C_4H_9SH	75-66-1	**	9.03	PE	4032
$C_5H_6S^+$	$C_4H_3SCH_3$ (Thiophene, 2-methyl-)	554-14-3	**	8.59 (V)	PE	5323

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₅H₆S⁺						
	C ₄ H ₃ SCH ₃	554-14-3	**	8.63±0.05	EI	3482
			**	8.61	CTS	3787
	C ₄ H ₃ SCH ₃ (Thiophene, 3-methyl-)	616-44-4	**	8.72	EI	3787
			**	8.84	CTS	3787
			**	8.70 (V)	PE	5323
	C ₅ H ₆ S (4H-Thiopyran)	289-70-3	**	8.0±0.1 (V)	PE	4841
C₅H₁₀S⁺						
	CH ₂ =CHS(iso-C ₃ H ₇)	18888-46-5	**	8.15±0.01	PI	5531
	CH ₂ =CHCH ₂ SC ₂ H ₅	5296-62-8	**	8.51±0.01	PI	5531
	CH ₂ =CHSC ₃ H ₇	16330-21-5	**	8.16±0.01	PI	5531
	C ₅ H ₁₀ S (2H-Thiopyran, tetrahydro-)	1613-51-0	**	8.39	PE	4246
			**	8.45 (V)	PE	3733
C₅H₁₂S⁺						
	C ₂ H ₅ S(iso-C ₃ H ₇)	5145-99-3	**	8.35±0.01	PI	5531
	(CH ₃) ₂ CSCH ₃	6163-64-0	**	8.38±0.05	PE	4153
	n-C ₃ H ₇ SC ₂ H ₅	4110-50-3	**	8.37	CTS	4272
C₆H₄S⁺						
	cis-C ₂ H ₂ S(C=CH) ₂ (Thiirane, cis-2,3-diethynyl-)	50555-56-1	**	8.80	PE	4374
	trans-C ₂ H ₂ S(C=CH) ₂ (Thiirane, trans-2,3-diethynyl-)	50555-55-0	**	8.85	PE	4374
C₆H₆S⁺						
	C ₆ H ₅ SH (Benzenthiol)	108-98-5	**	8.28	PE	3678
			**	8.39	PE	4621
			**	8.47 (V)	PE	4327
			**	8.95±0.1	EI	3817
			**	8.36	CTS	4272
C₆H₈S⁺						
	C ₄ H ₂ S(CH ₃) ₂ (Thiophene, 2,5-dimethyl-)	638-02-8	**	8.10	EI	3787
			**	8.18	CTS	3787
	C ₄ H ₃ SC ₂ H ₅ (Thiophene, 2-ethyl-)	872-55-9	**	8.67±0.05	EI	3482
			**	8.57	CTS	3787
C₆H₁₀S⁺						
	HC≡CS(iso-C ₄ H ₉)	50351-47-8	**	8.62±0.01	PI	5531
	(CH ₂ =CHCH ₂) ₂ S	592-88-1	**	8.52±0.01	PI	5531
	C ₆ H ₁₀ S (7-Thiabicyclo[2.2.1]heptane)	279-59-4	**	8.28±0.04	PE	3842
	C ₅ H ₇ SCH ₃ (2H-Thiopyran, 3,4-dihydro-6-methyl-)	13042-79-0	**	7.95 (V)	PE	4569
C₆H₁₂S⁺						
	CH ₂ =CHS(tert-C ₄ H ₉)	14094-13-4	**	8.07±0.01	PI	5531
	CH ₂ =CHSC ₄ H ₉	4789-70-2	**	8.15±0.01	PI	5531
C₆H₁₄S⁺						
	(n-C ₃ H ₇) ₂ S	111-47-7	**	8.34 (V)	PE	4276
			**	8.34 (V)	PE	5632
	(iso-C ₃ H ₇) ₂ S	625-80-9	**	8.25±0.01	PI	5531
			**	8.26 (V)	PE	4276

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₆H₁₄S⁺	(<i>iso</i> -C ₃ H ₇) ₂ S	625-80-9	**	8.26 (V)	PE	5632
C₇H₈S⁺	C ₆ H ₅ CH ₂ SH (Benzene(methanethiol))	100-53-8	**	8.85 (V)	PE	3678
	C ₆ H ₅ SCH ₃ (Benzene,(methylthio)-)	100-68-5	**	7.92±0.02	PI	5552
			**	7.96±0.01	PI	5531
			**	8.02 (V)	PE	4479
			**	8.04 (V)	PE	4884
			**	8.07 (V)	PE	3781
			**	8.07 (V)	PE	4327
			**	8.07 (V)	PE	5632
			**	8.60 (V)	PE	4327
			**	8.08	CTS	4272
			**	7.93	PE	4621
	C ₆ H ₄ (SH)CH ₃ (Benzenethiol, 2-methyl-)	137-06-4	**	8.31 (V)	PE	4327
	C ₆ H ₄ (SH)CH ₃ (Benzenethiol, 3-methyl-)	108-40-7	**	8.44 (V)	PE	4327
	C ₆ H ₄ (SH)CH ₃ (Benzenethiol, 4-methyl-)	106-45-6	**	8.33 (V)	PE	4327
	C ₇ H ₈ S (2-Thiabicyclo[3.2.1]octa-3,6-diene)	39066-37-0	**	8.03-8.12 (V)	PE	5481
C₇H₁₀S⁺	C ₇ H ₁₀ S (2-Thiabicyclo[3.2.1]oct-3-ene)	71017-55-5	**	7.92 (V)	PE	5481
C₇H₁₂S⁺	CH ₃ C≡CS(<i>iso</i> -C ₄ H ₉)	56444-80-5	**	8.15±0.01	PI	5531
	C ₇ H ₁₂ S (2-Thiabicyclo[3.2.1]octane)	279-81-2	**	8.43-8.52 (V)	PE	5481
	C ₅ H ₆ S(CH ₃) ₂ (2 <i>H</i> -Thiopyran, 3,4-dihydro-4,4-dimethyl-)	53520-28-8	**	8.06 (V)	PE	4246
C₈H₆S⁺	C ₆ H ₄ C ₂ H ₂ S (Benzothiophene)	11095-43-5	**	8.13±0.015 (V)	PE	5522
	C ₈ H ₆ S (Benzo[b]thiophene)	95-15-8	**	8.20	PE	4017
			**	8.73±0.05	EI	4316
	C ₈ H ₆ S (Benzo[c]thiophene)	270-82-6	**	7.75	PE	4017
C₈H₈S⁺	C ₆ H ₅ SCH=CH ₂ (Benzene,(ethenylthio)-)	1822-73-7	**	7.96±0.01	PI	5531
	C ₆ H ₅ CSCH ₃ (Ethanethione, 1-phenyl-)	16696-68-7	**	8.1 (V)	PE	4467
	C ₈ H ₈ S (9-Thiabicyclo[4.2.1]nona-2,4,7-triene)	35783-97-2	**	8.39 (V)	PE	4326
C₈H₁₀S⁺	C ₆ H ₅ SC ₂ H ₅ (Benzene,(ethylthio)-)	622-38-8	**	7.88±0.02	PI	5531
			**	8.0 (V)	PE	4327
			**	8.53 (V)	PE	4327
	C ₆ H ₄ (CH ₃)SCH ₃ (Benzene, 1-methyl-3-(methylthio)-)	4886-77-5	**	8.00 (V)	PE	4327
			**	8.50 (V)	PE	4327

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_8\text{H}_{10}\text{S}^+$	$\text{C}_6\text{H}_5(\text{CH}_3)\text{SCH}_3$ (Benzene, 1-methyl-4-(methylthio)-)	623-13-2	**	7.87 (V)	PE	4327
			**	7.9±0.05 (V)	PE	4389
			**	8.50 (V)	PE	4327
$\text{C}_8\text{H}_{12}\text{S}^+$	$\text{C}_6\text{H}_5\text{CH}_2\text{SCH}_3$ (Benzene, [(methylthio)methyl]-)	766-92-7	**	9.01 (V)	PE	3781
	$\text{C}_8\text{H}_{10}\text{S}$ (9-Thiabicyclo[4.2.1]nona-2,4-diene)	50669-04-0	**	8.26 (V)	PE	4326
$\text{C}_8\text{H}_{12}\text{S}^+$	$\text{C}_4\text{H}_5\text{S}(\text{tert-C}_4\text{H}_9)$ (Thiophene, 2-(1,1-dimethylethyl)-)	1689-78-7	**	8.48	CTS	4382
			**	8.54±0.05	EI	3482
	$\text{C}_4\text{H}_5\text{S}(\text{tert-C}_4\text{H}_9)$ (Thiophene, 3-(1,1-dimethylethyl)-)	1689-79-8	**	8.57	CTS	4382
	$\text{C}_8\text{H}_{12}\text{S}$ (9-Thiabicyclo[3.3.1]-non-1-ene)	50436-33-4	**	8.35 (V)	PE	4569
	$\text{C}_8\text{H}_{12}\text{S}$ (9-Thiabicyclo[4.2.1]non-7-ene)	13350-64-6	**	8.20 (V)	PE	4326
	$\text{C}_5\text{S}(\text{CH}_3)_4$ (Thiophene, tetramethyl-)	14503-51-6	**	7.93	CTS	4382
$\text{C}_8\text{H}_{14}\text{S}^+$	$\text{C}_8\text{H}_{14}\text{S}$ (9-Thiabicyclo[3.3.1]nonane)	281-15-2	**	8.20 (V)	PE	4569
	$\text{C}_8\text{H}_{14}\text{S}$ (9-Thiabicyclo[4.2.1]nonane)	6522-54-9	**	8.16 (V)	PE	4326
$\text{C}_8\text{H}_{18}\text{S}^+$	$(n\text{-C}_4\text{H}_9)_2\text{S}$	544-40-1	**	8.22 (V)	PE	4276
	$(iso\text{-C}_4\text{H}_9)_2\text{S}$	592-65-4	**	8.32	CTS	4272
	$(tert\text{-C}_4\text{H}_9)_2\text{S}$	107-47-1	**	8.07 (V)	PE	4276
			**	8.07 (V)	PE	5632
			**	8.07 (V)	PE	5632
			**	8.18±0.05 (V)	PE	4153
			**	8.19±0.1	EI	4198
	$\text{C}_6\text{H}_5\text{CH=CHSCH}_3$ (Benzene, [2-(methylthio)ethenyl]-(Z)-)	35822-50-5	**	7.75 (V)	PE	3781
	$\text{C}_6\text{H}_5\text{SCH}_2\text{CH=CH}_2$ (Benzene,(2-propenylthio)-)	5296-64-0	**	8.75 (V) 7.91 ± 0.01	PE PI	5632 5531
$\text{C}_9\text{H}_{11}\text{S}^+$	$\text{C}_6\text{H}_5\text{S}(\text{tert-C}_4\text{H}_9)$ (Benzene, [(1,1-dimethylethyl)thio]-)	3019-19-0	CH_3	12.1 ± 0.1	EI	4198
$\text{C}_9\text{H}_{12}\text{S}^+$	$\text{C}_6\text{H}_5\text{S}(n\text{-C}_5\text{H}_7)$ (Benzene,(propylthio)-)	874-79-3	**	7.81±0.03	PI	5552
	$\text{C}_6\text{H}_5(\text{CH}_3)\text{SC}_2\text{H}_5$ (Benzene, 1-(ethylthio)-3-methyl-)	34786-24-8	**	7.92 (V)	PE	4327
	$\text{C}_6\text{H}_4(\text{CH}_3)\text{SC}_2\text{H}_5$ (Benzene, 1-(ethylthio)-4-methyl-)	622-63-9	**	8.42 (V) 7.9 (V)	PE	4327
	$\text{C}_6\text{H}_5\text{SCH}(\text{CH}_3)_2$ (Benzene, [(1-methylethyl)thio]-)	3019-20-3	**	8.45 (V) 8.46 (V)	PE	4327
$\text{C}_{10}\text{H}_8\text{S}^+$	$\text{C}_6\text{H}_5\text{C}_4\text{H}_3\text{S}$ (Thiophene, 2-phenyl)	825-55-8	**	8.06	CTS	4382

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₀H₁₄S⁺						
	C ₆ H ₅ S(<i>tert</i> -C ₄ H ₉) (Benzene, [(1,1-dimethylethyl)thio]-)	3019-19-0	**	8.38±0.05	PE	4589
			**	8.17±0.1	EI	4198
			**	8.40 (V)	PE	4327
	C ₆ H ₄ (CH ₃)SCH(CH ₃) ₂ (Benzene, 1-methyl-3-[(1-methylethyl)thio]-)	14905-80-7	**	8.38 (V)	PE	4327
	C ₆ H ₄ (CH ₃)SCH(CH ₃) ₂ (Benzene, 1-methyl-4-[(1-methylethyl)thio]-)	14905-81-8	**	8.38 (V)	PE	4327
C₁₀H₁₆S⁺						
	C ₇ H ₇ (=S)(CH ₃) ₃ (Bicyclo[2.2.1]heptane-2-thione, 1,3,3-trimethyl-)	875-06-9	**	8.41 (V)	PE	4323
	C ₆ H ₄ S(CH ₃) ₄ (Thiepin, 4,5-didehydro-2,3,6,7-tetrahydro-3,3,6,6-tetramethyl-)	26825-18-3	**	8.19 (V)	PE	4362
	C ₁₀ H ₁₅ SH (Tricyclo[3.3.1.1 ^{3,7}]decane-1-thiol)	34301-54-7	**	8.78 (V)	PE	5395
C₁₀H₂₀S⁺						
	C ₆ H ₈ S(CH ₃) ₄ (1-Thiacycloheptane, 3,3,6,6-tetramethyl-)	XXXXXX-XX-X	**	8.15 (V)	PE	4362
C₁₁H₁₀S⁺						
	C ₁₀ H ₇ SCH ₃ (Naphthalene, 1-(methylthio)-)	10075-72-6	**	7.67 (V)	PE	3781
	C ₁₀ H ₇ SCH ₃ (Naphthalene, 2-(methylthio)-)	7433-79-6	**	7.71 (V)	PE	3781
			**	7.71 (V)	PE	5632
C₁₁H₁₆S⁺						
	C ₆ H ₄ (SCH ₃)(<i>tert</i> -C ₄ H ₉) (Benzene, 1-(1,1-dimethylethyl)-4-(methylthio)-)	7252-86-0	**	7.83±0.05 (V)	PE	4627
	C ₆ H ₄ (CH ₃)SC(CH ₃) ₃ (Benzene, 1-[(1,1-dimethylethyl)thio]-3-methyl-)	34786-26-0	**	8.35 (V)	PE	4327
	C ₆ H ₄ (CH ₃)SC(CH ₃) ₃ (Benzene, 1-[(1,1-dimethylethyl)thio]-4-methyl-)	7439-10-3	**	8.31 (V)	PE	4327
C₁₂H₈S⁺						
	C ₁₂ H ₈ S (Dibenzothiophene)	132-65-0	**	7.90±0.03	PI	5552
			**	7.93 (V)	PE	5619
			**	8.01 (V)	PE	3852
			**	8.34	EI	3787
			**	8.44	EI	4228
			**	8.23	CTS	3787
C₁₂H₁₀S⁺						
	(C ₆ H ₅) ₂ S (Benzene, 1,1'-thiobis-)	139-66-2	**	7.81±0.03	PI	5552
			**	7.92±0.01	PI	5531
			**	7.8	PE	4228
			**	7.86 (V)	PE	4667
			**	7.88±0.05	EI	3498
			**	7.88	EI	5292
			**	8.45±0.1	EI	3817
			**	8.04	CTS	4272
	C ₄ H ₃ SCH=CHC ₆ H ₅ (Thiophene, 2-(2-phenylethenyl)-)	3783-65-1	**	7.55	EI	3787
			**	7.78	CTS	3787
C₁₂H₁₈S⁺						
	C ₆ H ₆ (C ₂ H ₄ S)C ₄ H ₈ (4α,8α-(Methanothiomethano)naphthalene, 1,2,3,4,5,8-hexahydro-)	17853-64-4	**	8.07 (V)	PE	5194

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₂H₂₀S⁺	C ₄ H ₂ S(C ₄ H ₉) ₂ (Thiophene, 2,5-bis(1,1-dimethylethyl)-)	1689-77-6	**	7.85 (V)	PE	4324
C₁₃H₈S⁺	C ₁₄ H ₁₂ S (Dibenzo[<i>b,e</i>]thiepin,6,11-dihydro-)	1207-93-8	CH ₃	10.80	EI	5414
C₁₃H₁₀S⁺	(C ₆ H ₅) ₂ CH ₂ SC(=O) (Dibenzo[<i>b,e</i>]thiepin-11(6H)-one)	1531-77-7	CO	9.75	EI	5340
C₁₃H₁₂S⁺	C ₆ H ₄ (CH ₃)SC ₆ H ₅ (Benzene, 1-methyl-2-(phenylthio)-) C ₆ H ₄ (CH ₃)SC ₆ H ₅ (Benzene, 1-methyl-3-(phenylthio)-) C ₆ H ₄ (CH ₃)SC ₆ H ₅ (Benzene, 1-methyl-4-(phenylthio)-) C ₆ H ₅ CH ₂ SC ₆ H ₅ (Benzene,[(phenylmethyl)thio]-)	13963-35-4 13865-48-0 3699-01-2 831-91-4	**	8.01 7.99 7.95 7.87±0.02	CTS CTS CTS PI	4272 4272 4272 5552
C₁₄H₁₀S⁺	C ₁₄ H ₁₀ S (Dibenzo[<i>b,f</i>] thiepin)	257-13-6	**	7.96 (V)	PE	4611
C₁₄H₁₁S⁺	C ₁₄ H ₁₂ S (Dibenzo[<i>b,e</i>]thiepin,6,11-dihydro-) C ₁₃ H ₈ S(CH ₃) ₂ (9H-Thioxanthene, 9,9-dimethyl-)	1207-93-8 19019-10-4	H CH ₃	11.40 8.3±0.1	EI EI	5414 4664
C₁₄H₁₂S⁺	C ₁₄ H ₁₂ S (Dibenzo[<i>b,e</i>]thiepin,6,11-dihydro-)	1207-93-8	**	8.77	EI	5414
C₁₄H₁₄S⁺	C ₆ H ₃ (CH ₃) ₂ SC ₆ H ₅ (Benzene, 1,2-dimethyl-4-(phenylthio)-) C ₆ H ₄ (CH ₃)SC ₆ H ₄ CH ₃ (Benzene, 1,1'-thiobis[2-methyl-]) C ₆ H ₄ (CH ₃)SC ₆ H ₄ CH ₃ (Benzene, 1,1'-thiobis[4-methyl-]) (C ₆ H ₅ CH ₂) ₂ S (Benzene,1,1'[thiobis(methylene)bis-])	2828-65-1 4537-05-7 620-94-0 538-74-9	**	7.89 7.94 7.83 8.05±0.02	CTS CTS CTS PI	4272 4272 4272 5552
C₁₅H₁₀S⁺	C ₃ (=S)(C ₆ H ₅) ₂ (2-Cyclopropen-1-thione, 2,3-diphenyl-)	2570-01-6	**	11.25 (V)	PE	4856
C₁₅H₁₄S⁺	C ₁₃ H ₈ S(CH ₃) ₂ (9H-Thioxanthene, 9,9-dimethyl-)	19019-10-4	**	7.7±0.1	EI	4664
C₁₆H₁₈S⁺	C ₆ H ₃ (CH ₃) ₂ SC ₆ H ₃ (CH ₃) ₂ (Benzene, 1,1'-thiobis[2,6-dimethyl-])	52805-90-0	**	8.36	CTS	4272
C₁₈H₁₆S⁺	C ₁₀ H ₆ (CH ₂ CH ₂) ₂ C ₄ H ₂ S (8,11-Epithio-5,14-ethenobenzocyclododecene,6,7,12,13-tetrahydro-)	53539-29-0	**	7.50 (V)	PE	5575

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
CH_2S_2^+	$\text{C}_3\text{H}_6\text{S}_2$ (1,3-Dithiolane)	4829-04-3	C_2H_4	10.8 ± 0.2	EI	3598
CH_4S_2^+	HSCH_2SH	6725-64-0	**	9.9 (V)	PE	4405
$\text{C}_2\text{H}_6\text{S}_2^+$	$(\text{CH}_3\text{S})_2$	624-92-0	** ** ** ** ** ** ** ** ** **	8.3 8.82 (V) 8.96 (V) 8.97 (V) 8.97 (V) 8.97 (V) 8.98 (V) 9. (V)	PE PE PE PE PE PE PE PE	4188 3697 5068 4276 5538 5632 4218 4410
$\text{C}_3\text{H}_5\text{S}_2^+$	$\text{C}_3\text{H}_6\text{S}_2$ (1,3-Dithiolane)	4829-04-3	H	11.2 ± 0.2	EI	3598
$\text{C}_3\text{H}_6\text{S}_2^+$	$\text{CH}_3\text{C}=\text{SSCH}_3$ $\text{C}_3\text{H}_6\text{S}_2$ (1,3-Dithiolane)	2168-84-5 4829-04-3	** ** ** **	8.50 (V) 8.75 (V) 8.77 (V) 9.0 ± 0.05	PE PE PE EI	4427 4418 4756 3598
$\text{C}_3\text{H}_8\text{S}_2^+$	$\text{CH}_3\text{SCH}_2\text{SCH}_3$	1618-26-4	** **	8.65 (V) 8.67 (V)	PE PE	5632 4405
$\text{C}_4\text{H}_4\text{S}_2^+$	$\text{C}_4\text{H}_6\text{S}_2$ (1,4-Dithiin)	290-79-9	**	8.1 ± 0.1 (V)	PE	4841
$\text{C}_4\text{H}_8\text{S}_2^+$	$\text{CH}_2=\text{C}(\text{SCH}_3)_2$ <i>cis</i> - $\text{CH}_3\text{SCH}=\text{CHSCH}_3$ <i>trans</i> - $\text{CH}_3\text{SCH}=\text{CHSCH}_3$ $\text{C}_4\text{H}_8\text{S}_2$ (1,2-Dithiane) $\text{C}_4\text{H}_8\text{S}_2$ (1,3-Dithiane) $\text{C}_4\text{H}_8\text{S}_2$ (1,4-Dithiane)	51102-74-0 764-44-3 764-45-4 505-20-4 505-23-7 505-29-3	** ** ** ** ** ** ** ** ** ** ** ** ** ** ** **	8.2 (V) 7.80 (V) 7.85 (V) 7.96 (V) 8.36 (V) 8.36 (V) 8.33 (V) 8.33 (V) 8.46 (V) 8.58 (V)	PE PE PE PE PE PE PE PE PE PE PE PE PE PE PE PE	4291 4291 4291 5632 4276 5632 4756 5632 3733 5632 5632 3733 5632 3733
$\text{C}_4\text{H}_{10}\text{S}_2^+$	$(\text{C}_2\text{H}_5\text{S})_2$ $\text{CH}_3\text{SCH}_2\text{CH}_2\text{SCH}_3$	110-81-6 6628-18-8	** ** ** ** **	8.70 (V) 8.70 (V) 8.77 (V) 8.85 (V) 8.64 (V)	PE PE PE PE PE	4276 5632 4410 4218 5632
$\text{C}_5\text{H}_4\text{S}_2^+$	$\text{C}_5\text{H}_4\text{S}(=\text{S})$ (4 <i>H</i> -Thiopyran-4-thione)	1120-94-1	**	7.96 ± 0.05 (V)	PE	5002

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₅H₆S₂⁺	C ₄ H ₃ SSCH ₃ (Thiophene, 2-(methylthio)-)	5780-36-9	**	8.63±0.05 (V)	PE	4626
	C ₄ H ₂ S(SH)CH ₃ (2-Thiophenethiol, 5-methyl-)	3970-28-3	**	8.10±0.05 8.48±0.05	EI EI	3482 4706
C₅H₁₂S₂⁺	C ₂ H ₅ SCH ₂ SC ₂ H ₅	4396-19-4	** **	8.66 (V) 8.22±0.02	PE PI	4756 5531
C₆H₄S₂⁺	C ₆ H ₄ S ₂ (Thieno[2,3- <i>b</i>]thiophene)	250-84-0	** ** **	8.32 8.32 (V) 8.45 (V)	PE PE PE	4017 5405 5478
	C ₆ H ₄ S ₂ (Thieno[3,2- <i>b</i>]thiophene)	251-41-2	** **	8.10 8.10 (V)	PE PE	4017 5405
	C ₆ H ₄ S ₂ (Thieno[3,2- <i>b</i>]thiophene)	251-41-1	**	8.14 (V)	PE PE	3852
C₆H₈S₂⁺	C ₄ H ₂ S(CH ₃)SCH ₃ (Thiophene, 2-methyl-5-(methylthiol)-)	40990-29-2	**	8.13±0.05	EI	4706
	C ₄ HS(SH)(CH ₃) ₂ (3-Thiophenethiol, 2,5-dimethyl-)	29874-05-3	**	8.22±0.05	EI	4706
C₆H₁₀S₂⁺	cis,cis-CH ₃ SCH=CHCH=CHSCH ₃	35822-49-2	**	7.48 (V)	PE	5632
C₆H₁₄S₂⁺	(n-C ₃ H ₇ S) ₂	629-19-6	** **	8.62 (V) 8.62 (V)	PE PE	4276 5632
	(iso-C ₃ H ₇ S) ₂	4253-89-8	** ** **	8.54 (V) 8.54 (V) 8.51 (V)	PE PE PE	4276 5632 4410
C₇H₆S₂⁺	C ₇ H ₆ S ₂ (Thieno[2,3- <i>b</i>]thiophene,2-methyl-)	13393-75-4	**	8.12 (V)	PE	5478
	C ₇ H ₆ S ₂ (Thieno[2,3- <i>b</i>]thiophene,3-methyl-)	1723-34-8	**	8.04 (V)	PE	5478
C₇H₁₀S₂⁺	C ₄ HS(CH ₃) ₂ SCH ₃ (Thiophene, 2,5-dimethyl-3-(methylthiol)-)	63359-64-8	**	7.96±0.05	EI	4706
C₈H₆S₂⁺	(C ₄ H ₃ S) ₂ (3,3'-Bithiophene)	3172-56-3	**	8.2 (V)	PE	5422
C₈H₈S₂⁺	C ₈ H ₈ S ₂ (1,6-Dithiecin, 3,4,8,9-tetrahydro-2,5,7,10-tetrahydro-)	53690-50-9	**	8.73±0.02 (V)	PE	4180
C₈H₁₀S₂⁺	C ₆ H ₄ (SCH ₃) ₂ (Benzene,1,2-bis(methylthio)-)	2388-68-3	**	8.0 (V)	PE	5403
	C ₆ H ₄ (SCH ₃) ₂ (Benzene,1,3-bis(methylthio)-)	2388-69-4	**	8.0 (V)	PE	5403
	C ₆ H ₄ (SCH ₃) ₂ (Benzene, 1,4-bis(methylthio)-)	699-20-7	**	7.93 (V)	PE	3781

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₈H₁₀S₂⁺	C ₆ H ₄ (SCH ₃) ₂	699-20-7	**	7.93 (V)	PE	5403
C₈H₁₂S₂⁺	C ₄ (=S) ₂ (CH ₃) ₄ (1,3-Cyclobutanedithione, 2,2,4,4-tetramethyl-)	10181-56-3	**	8.35 (V)	PE	5499
C₈H₁₈S₂⁺	n-C ₄ H ₉ SSn-C ₄ H ₉ (tert-C ₄ H ₉ S) ₂	629-45-8 110-06-5 ** ** ** **	** ** 8.15 (V) 8.17 (V) 8.17 (V) 8.20 (V)	8.51 (V)	PE PE PE PE PE	4410 4410 4276 5632 4218
C₁₀H₆S₂⁺	C ₁₀ H ₆ S ₂ (Naphtho[1,8-cd]-1,2-dithole)	209-22-3	**	7.15 (V)	PE	4782
C₁₀H₁₂S₂⁺	C ₁₀ H ₁₂ S ₂ (1,3-Benzodithiole-2-ethyl-2-methyl-)	58657-45-7	**	7.85 (V)	PE	5410
C₁₂H₁₂S₂⁺	(C ₄ H ₂ SCH ₂ CH ₂) ₂ (13,14-Dithiatricyclo[8.2.1.1 ^{4,7}]tetradeca-4,6,10,12-tetraene) C ₁₀ H ₆ (SCH ₃) ₂ (Naphthalene, 1,4-bis(methylthio)-) C ₁₀ H ₆ (SCH ₃) ₂ (Naphthalene, 1,5-bis(methylthio)-) C ₁₀ H ₆ (SCH ₃) ₂ (Naphthalene, 1,8-bis(methylthio)-) C ₁₀ H ₆ (SCH ₃) ₂ (Naphthalene, 2,6-bis(methylthio)-)	73650-69-8 10075-73-7 ** 10075-74-8 7343-31-9 10075-77-1 **	** ** 7.58 (V) 7.58 (V) 7.66 (V) 7.55 (V) 7.59 (V) 7.59 (V)	7.95	PE PE PE PE PE PE PE	5575 5204 5612 5204 5204 5612
C₁₆H₁₄S₂⁺	C ₁₄ H ₈ S ₂ (CH ₃) ₂ (Anthracene-9,10-bis(methylthio)-)	10075-83-9	**	7.44 (V)	PE	5612
C₂₀H₃₀S₂⁺	(C ₁₀ H ₁₅ S) ₂ (Disulfide,bis(tricyclo[3.3.1.1 ^{3,7}]dec-1-yl))	34895-45-9	**	7.86 (V)	PE	5395
C₂₀H₃₂S₂⁺	C ₁₂ H ₈ S ₂ (CH ₃) ₈ (Cyclobuta[1,2-d:3,4-d']bisthiepin, 1,2,4,5,6,7,9,10-octahydro-1,1,5,5,6,6,10,10-octamethyl-)	40219-42-9		6.89 (V)	PE	4304
C₃₀H₂₀S₂⁺	C ₆ S ₂ (C ₆ H ₅) ₄ (Thieno[3,4-c]thiophene-2,5-5 ^{IV} , 1,3,4,6-tetraphenyl-)	36516-81-1	**	6.19 (V)	PE	4838
C₂H₂S₃⁺	C ₂ H ₂ S ₂ (=S) (1,3-Dithietane-2-thione)	18555-26-5	**	8.83 (V)	PE	4549
C₂H₄S₃⁺	C ₂ H ₄ S ₃ (1,2,4-Trithiolane)	289-16-7	** **	8.72±0.2 (V) 8.72 (V)	PE PE	5415 4410

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_3\text{H}_2\text{S}_3^+$	$\text{C}_3\text{H}_2\text{S}_3$ (1,3-Dithiole-2-thione)	930-35-8	**	8.26 (V)	PE	5410
			**	8.3 (V)	PE	4549
	$(\text{C}_3\text{H}_2\text{S}_2)=\text{S}$ (3H-1,2-Dithiole-3-thione)	534-25-8	**	8.42 (V)	PE	4403
$\text{C}_3\text{H}_4\text{S}_3^+$	$\text{C}_3\text{H}_4\text{S}_2\text{S}$ (1,3-Dithiolane-2-thione)	822-38-8	**	8.40 (V)	PE	4407
			**	8.40 (V)	PE	4323
$\text{C}_3\text{H}_6\text{S}_3^+$	$(\text{CH}_3\text{S})_2\text{CS}$	2314-48-9	**	8.5 (V)	PE	4323
	$\text{C}_3\text{H}_6\text{S}_3$ (1,3,5-Trithiane)	291-21-4	**	8.76 (V)	PE	3733
			**	8.83 ± 0.05 (V)	PE	4212
$\text{C}_4\text{H}_5\text{S}_3^+$	$(\text{C}_3\text{HS}_2)=\text{S}(\text{CH}_3)$ (3H-1,2-Dithiole-3-thione, 4-methyl-)	3354-41-4	**	8.23 (V)	PE	4403
	$(\text{C}_3\text{HS}_2)=\text{S}(\text{CH}_3)$ (3H-1,2-Dithiole-3-thione, 5-methyl-)	3354-40-3	**	8.25 (V)	PE	4403
	$\text{C}_4\text{H}_6\text{S}_2=\text{S}$ (1,3-Dithiane-2-thione)	1748-15-8	**	8.40 (V)	PE	4323
$\text{C}_5\text{H}_4\text{S}_3^+$	$\text{C}_5\text{H}_4\text{S}_3$ ([1,2]Dithiolo[1,5- <i>b</i>][1,2]dithiole-7- <i>S</i> ^V)	252-09-5	**	8.11 (V)	PE	3569
$\text{C}_6\text{H}_6\text{S}_3^+$	$\text{C}_5\text{H}_4\text{S}_3\text{CH}_3$ ([1,2]Dithiolo[1,5- <i>b</i>][1,2]dithiole-7- <i>S</i> ^V , 2-methyl-)	20718-55-2	**	7.83 (V)	PE	3569
$\text{C}_6\text{H}_{12}\text{S}_3^+$	$\text{C}_3\text{H}_6\text{S}_3(\text{CH}_3)_3$ (1,3,5-Trithiane, 2,4,6-trimethyl-)	2765-04-0	**	8.39 ± 0.05 (V)	PE	4212
	$\text{C}_6\text{H}_{12}\text{S}_3$ (1,2,4-Trithiolane,3,3,5,5-tetramethyl-)	38348-31-1	**	8.12 ± 0.2 (V)	PE	5415
			**	8.12 (V)	PE	4410
$\text{C}_7\text{H}_4\text{S}_3^+$	$\text{C}_7\text{H}_4\text{S}_3$ (1,3-Benzodithiole-2-thione)	934-36-1	**	8.14 (V)	PE	5410
	$(\text{C}_7\text{HS}_2)=\text{S}$ (3H-1,2-Benzodithiole-3-thione)	3354-42-5	**	8.10 (V)	PE	4403
$\text{C}_7\text{H}_8\text{S}_3^+$	$\text{C}_5\text{H}_2\text{S}_3(\text{CH}_3)_2$ ([1,2]Dithiolo[1,5- <i>b</i>][1,2]dithiole-7- <i>S</i> ^V , 2,5-dimethyl-)	2080-35-5	**	7.73 (V)	PE	3569
	$\text{C}_5\text{H}_2\text{S}_3(\text{CH}_3)_2$ ([1,2]Dithiolo[1,5- <i>b</i>][1,2]dithiole-7- <i>S</i> ^V , 3,4-dimethyl-)	29977-00-2	**	7.63 (V)	PE	3569
$\text{C}_7\text{H}_{10}\text{S}_3^+$	$(\text{C}_3\text{HS}_2)=\text{S}(\text{tert-C}_4\text{H}_9)$ (3H-1,2-Dithiole-3-thione, 5-(1,1-dimethylethyl)-)	29507-64-0	**	8.15 (V)	PE	4403
$\text{C}_8\text{H}_4\text{S}_3^+$	$\text{C}_4\text{H}_4\text{S}_3$ (Dithieno[2,3- <i>b</i> :3',2'- <i>d</i>]thiophene)	236-63-5	**	7.86 (V)	PE	5405

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₈H₄S₃⁺	C ₈ H ₄ S ₃ (Dithieno[3,2- <i>b</i> :2',3'- <i>d</i>]thiophene)	3593-75-7	**	7.8 (V)	PE	5405
	C ₈ H ₄ S ₃ (Dithieno[3,4- <i>b</i> :3',4'- <i>d</i>]thiophene)	13090-49-8	**	7.88 (V)	PE	5405
C₈H₆S₃⁺	(C ₄ H ₃ S) ₂ S (Thiophene,2,2'-thiobis-)	3988-99-6	**	8.40 (V)	PE	5356
	(C ₄ H ₃ S) ₂ S (Thiophene,3,3'-thiobis-)	3807-38-3	**	8.06 (V)	PE	5356
C₉H₆S₃⁺	(C ₃ HS ₂)=S(C ₆ H ₅) (3H-1,2-Dithiole-3-thione, 5-phenyl-)	3445-76-9	**	8.11 (V)	PE	4403
C₉H₁₈S₃⁺	C ₃ S ₃ (CH ₃) ₆ (1,3,5-Trithiane, 2,2,4,4,6,6-hexamethyl-)	828-26-2	**	7.95±0.05 (V)	PE	4212
C₁₀H₈S₃⁺	(C ₃ HS ₂)=S(C ₆ H ₄ CH ₃) (3H-1,2-Dithiole-3-thione, 5-(4-methylphenyl)-)	6921-83-1	**	8.10 (V)	PE	4403
	(C ₄ H ₂ S) ₂ C ₂ H ₄ S (4H,6H-Dithieno[3,4- <i>c</i> :3',4'- <i>e</i>]thiepin)	42850-82-8	**	8.4 (V)	PE	5422
	(C ₄ H ₂ S) ₂ C ₂ H ₄ S (Dithieno[2,3- <i>c</i> :3',2'- <i>e</i>]thiepin,4,6-dihydro-)	63286-55-5	**	8.15 (V)	PE	5422
C₁₀H₁₂S₃⁺	C ₈ H ₆ S ₃ (CH ₃) ₂ (3H-[1,2]Dithiolo[4,5,1- <i>hi</i>][1,2]benzodithiole-8-S ^{IV} , 4,5-dihydro-2,6-dimethyl-)	35437-21-9	**	7.34 (V)	PE	3569
C₁₂H₁₆S₃⁺	C ₈ H ₆ S ₃ (C ₂ H ₅) ₂ (3H-[1,2]Dithiolo[4,5,1- <i>hi</i>][1,2]benzodithiole-8-S ^{IV} , 2,6-diethyl-4,5-dihydro-)	35505-46-5	**	7.33 (V)	PE	3569
C₁₄H₂₀S₃⁺	C ₈ H ₆ S ₃ (C ₃ H ₇) ₂ (3H-[1,2]Dithiolo[4,5,1- <i>hi</i>][1,2]benzodithiole-8-S ^{IV} , 4,5-dihydro-2,6-bis(1-methylethyl)-)	35505-47-6	**	7.19 (V)	PE	3569
C₁₇H₁₂S₃⁺	C ₈ H ₂ S ₃ (C ₆ H ₅) ₂ ([1,2]Dithiolo[1,5- <i>b</i>][1,2]dithiole-7-S ^{IV} , 3,4-diphenyl-)	25730-47-6	**	7.57 (V)	PE	3569
C₅H₈S₄⁺	C ₅ H ₈ S ₄ (1,4,6,9-Tetrathiaspiro[4.4]nonane)	13145-46-5	**	8.26 (V)	PE	4756
			**	8.35 (V)	PE	4418
C₅H₁₂S₄⁺	C(SCH ₃) ₄	6156-25-8	**	8.29 (V)	PE	4756
C₆H₄S₄⁺	(C ₃ H ₂ S ₂) ₂ (1,3-Dithiole-2-(1,3-dithiole-2-ylidene)-)	31366-25-3	**	7.00	CTS	5622
			**	6.83 (V)	PE	3981
			**	6.83 (V)	PE	4481
			**	6.92±0.03 (V)	PE	4155
C₆H₈S₄⁺	(C ₃ H ₄ S ₂) ₂ (1,3-Dithiolane, 2-(1,3-dithiolan-2-ylidene)-)	24719-68-4	**	7.05±0.03 (V)	PE	4155

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_6\text{H}_8\text{S}_4^+$	$(\text{C}_3\text{H}_4\text{S}_2)_2$	24719-68-4	**	7.17 (V)	PE	4481
$\text{C}_6\text{H}_{10}\text{S}_4^+$	$\text{C}_6\text{H}_{10}\text{S}_4$ (2,2-Bi-1,3-dithiolane)	6784-47-0	**	8.6-9.0 (V)	PE	4481
$\text{C}_6\text{H}_{12}\text{S}_4^+$	$(\text{CH}_3\text{S})_2\text{C}=\text{C}(\text{SCH}_3)_2$ $\text{C}_2\text{S}_4(\text{CH}_3)_4$ (1,2,4,5-Tetrathiane, 3,3,6,6-tetramethyl-)	13046-50-9 4475-72-3	** **	7.75 (V) 8.23 ± 0.02 (V)	PE PE	4291 4402
$\text{C}_7\text{H}_{12}\text{S}_4^+$	$\text{C}_7\text{H}_{12}\text{S}_4$ (1,5,7,11-Tetrathiaspiro[5.5]undecane)	180-97-2	**	8.09 (V)	PE	4756
$\text{C}_{10}\text{H}_{12}\text{S}_4^+$	$\text{C}_6\text{S}_4(\text{CH}_3)_4$ (1,3-Dithiole, 2-(4,5-dimethyl-1,3-dithiol-2-ylidene)-4,5-dimethyl-)	50708-37-7	**	6.40 (V)	PE	4481
$\text{C}_{10}\text{H}_{16}\text{S}_4^+$	$\text{C}_{10}\text{H}_{16}\text{S}_4$ (6,7,13,14-Tetrathiadispiro[4.2.4.2]tetradecane)	184-05-4	**	8.17 ± 0.02 (V)	PE	4402
$\text{C}_{12}\text{H}_{20}\text{S}_4^+$	$\text{C}_{12}\text{H}_{20}\text{S}_4$ (7,8,15,16-Tetrathiadispiro[5.2.5.2]hexadecane)	183-85-7	**	7.98 ± 0.02 (V)	PE	4402
$\text{C}_{14}\text{H}_8\text{S}_4^+$	$(\text{C}_6\text{H}_4\text{S}_2\text{C})_2$ (1,3-Benzodithiole, 2-(1,3-benzodithiol-2-ylidene)-)	24648-13-2	**	6.81 (V)	PE	4461
$\text{C}_{10}\text{H}_{18}\text{S}_6^+$	$\text{C}_4\text{H}_8\text{S}_2$ (1,4-Dithiane)	505-29-3	**	8.46 (V)	PE	5632
$\text{B}_9\text{CH}_{11}\text{S}^+$	$\text{SB}_9\text{H}_8(\text{CH}_3)$ (1-Thiadecaborane(9),10-methyl-)	64173-76-8	**	10.0 (V)	PE	5324
$\text{BC}_3\text{H}_9\text{S}^+$	$(\text{CH}_3)_2\text{BSCH}_3$	19163-05-4	**	9.40 (V)	PE	4065
$\text{BC}_{12}\text{H}_{19}\text{S}^+$	$\text{C}_6\text{H}_5\text{SB}(n\text{-C}_3\text{H}_7)_2$ (Borinic acid, dipropylthio-phenyl ester)	4443-46-3	**	8.77 ± 0.05 (V)	PE	4848
$\text{BC}_{13}\text{H}_{21}\text{S}^+$	$\text{C}_6\text{H}_4(\text{CH}_3)\text{SB}(n\text{-C}_3\text{H}_7)_2$ (Borinic acid, dipropylthio-3-methylphenyl ester) $\text{C}_6\text{H}_4(\text{CH}_3)\text{SB}(n\text{-C}_3\text{H}_7)_2$ (Borinic acid, dipropylthio-4-methylphenyl ester)	64503-47-5 64503-46-4	** **	8.59 ± 0.05 (V) 8.48 ± 0.05 (V)	PE PE	4848 4848
$\text{BC}_3\text{H}_9\text{S}_2^+$	$(\text{CH}_3\text{S})_2\text{BCH}_3$	19163-08-7	**	8.74 (V)	PE	4065
$\text{B}_2\text{C}_2\text{H}_6\text{S}_3^+$	$\text{B}_2\text{S}_3(\text{CH}_3)_2$ (1,2,4,3,5-Trithiadiborolane, 3,5-dimethyl-)	25592-09-0	**	9.04 (V)	PE	4526
$\text{BC}_3\text{H}_9\text{S}_3^+$	$\text{B}(\text{SCH}_3)_3$	997-49-9	**	8.74 (V)	PE	4065

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
NS⁺ <i>(²Π)</i>	NS	51801-08-2	**	8.87±0.01 (V)	PE	4657
	(SN) ₂	XXXXX-XX-X		13.0±0.6	EI	4870
	(SN) ₄	XXXXX-XX-X		14.1±0.6	EI	4870
NS₂⁺	(SN) ₂	XXXXX-XX-X		12.2±0.6	EI	4870
N₂S₂⁺	(SN) ₂	XXXXX-XX-X	**	11.5±0.6	EI	4870
	S ₂ N ₂	25474-92-4	**	10.41	PE	4718
	S ₂ N ₂ (Sulfur nitride)	45346-74-5	**	10.51 (V)	PE	5355
	(SN) ₄	XXXXX-XX-X		13.7±0.6	EI	4870
N₃S₃⁺	(SN) ₄	XXXXX-XX-X		12.3±0.6	EI	4870
N₄S₄⁺	(SN) ₄	XXXXX-XX-X	**	10.4±0.6	EI	4870
	S ₄ N ₄	28950-34-7	**	9.36 (V)	PE	5355
	(Nitrogen sulfide)					
C₂N₂S⁺	S(C≡N) ₂	627-52-1	**	11.32 (V)	PE	4476
C₂N₂S₂⁺	(SCN) ₂	505-14-6	**	11.05±0.02	PE	5363
CHNS⁺	HNCS	3129-90-6	**	9.94±0.02 (V)	PE	3670
CH₂NS⁺	NH(CH ₃)CSNH ₂	598-52-7		12.45	EI	4878
CH₃NS⁺	HCSNH ₂	115-08-2	**	8.69	PE	4469
C₂H₃NS⁺	CH ₃ NCS	556-61-6	**	9.37±0.02 (V)	PE	3670
	CH ₃ SCN	556-64-9	**	9.96±0.05 (V)	PE	5026
C₂H₄NS⁺	(NHCH ₃) ₂ CS	534-13-4		12.20	EI	4878
	N(CH ₃) ₂ CSNHCH ₃	2489-77-2		11.25	EI	4878
C₂H₅NS⁺	CH ₃ CSNH ₂	62-55-5	**	8.33 (V)	PE	4323
			**	8.36	PE	4469
C₃H₃NS⁺	C ₃ H ₃ NS (Isothiazole)	288-16-4	**	9.55	PE	3587
			**	9.62 (V)	PE	5213
			**	9.80	EI	3587
	C ₃ H ₃ NS (Thiazole)	288-47-1	**	9.50 (V)	PE	5213
C₃H₅NS⁺	C ₂ H ₅ NCS	542-85-8	**	9.12±0.05 (V)	PE	5026
	C ₂ H ₅ SCN	542-90-5	**	9.77±0.05 (V)	PE	5026

Table of Ion Energetics Measurements—Continued

Ion - (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_3\text{H}_6\text{NS}^+$	$(\text{CH}_3)_2\text{N}_2\text{CS}$	2782-91-4		11.20	EI	4878
$\text{C}_3\text{H}_7\text{NS}^+$	$\text{HCSN}(\text{CH}_3)_2$	758-16-7	** **	8.16 8.2 (V)	PE PE	4469 4323
$\text{C}_4\text{H}_5\text{NS}^+$	$\text{C}_3\text{H}_2\text{NS}(\text{CH}_3)$ (Isothiazole, 3-methyl-) $\text{C}_3\text{H}_2\text{NS}(\text{CH}_3)$ (Isothiazole, 4-methyl-) $\text{C}_3\text{H}_2\text{NS}(\text{CH}_3)$ (Isothiazole, 5-methyl-)	693-92-5 693-90-3 693-97-0	** ** **	9.60 9.25 9.65	EI PE EI	3587 3587 3587
$\text{C}_4\text{H}_9\text{NS}^+$	$\text{CH}_3\text{CSN}(\text{CH}_3)_2$	631-67-4	**	7.86	PE	4469
$\text{C}_5\text{H}_3\text{NS}^+$	$\text{C}_4\text{H}_3\text{SCN}$ (2-Thiophenecarbonitrile)	1003-31-2	** **	9.83 ± 0.05 10.00	EI CTS	3482 4382
$\text{C}_5\text{H}_5\text{NS}^+$	$\text{C}_5\text{H}_4\text{N}(\text{SH})$ (2-Pyridinethiol) $\text{C}_5\text{H}_4\text{N}(\text{SH})$ (3-Pyridinethiol) $\text{C}_5\text{H}_4\text{N}(\text{SH})$ (4-Pyridinethiol) $\text{C}_5\text{H}_4\text{NH}(\text{=S})$ (2(1H)-Pyridinethione)	73018-10-7 16133-26-9 4556-23-4 2637-34-5	** ** ** ** ** **	8.79 ± 0.03 (V) 8.92 ± 0.02 8.89 ± 0.03 (V) 9.41 ± 0.02 9.25 ± 0.03 (V) 9.50 ± 0.02 7.80 ± 0.03 (V)	PE EI PE EI PE EI PE	4711 3636 4711 3636 4711 3636 4711
$\text{C}_5\text{H}_9\text{NS}^+$	$n\text{-C}_4\text{H}_9\text{NCS}$ $n\text{-C}_4\text{H}_9\text{SCN}$	592-82-5 628-83-1	** **	9.02 ± 0.05 (V) 9.64 ± 0.05 (V)	PE PE	5026 5026
$\text{C}_5\text{H}_{11}\text{NS}^+$	$\text{CH}_2=\text{C}(\text{SCH}_3)\text{N}(\text{CH}_3)_2$	24854-14-6	**	7.8 (V)	PE	4291
$\text{C}_6\text{H}_7\text{NS}^+$	$\text{C}_5\text{H}_4\text{N}(\text{SCH}_3)$ (Pyridine, 2-(methylthio)-) $\text{C}_5\text{H}_4\text{N}(\text{SCH}_3)$ (Pyridine, 3-(methylthio)-) $\text{C}_5\text{H}_4\text{N}(\text{SCH}_3)$ (Pyridine, 4-(methylthio)-) $\text{C}_5\text{H}_4\text{N}(\text{=S})\text{CH}_3$ (2(1H)-Pyridinethione, 1-methyl-) $\text{C}_5\text{H}_4\text{N}(\text{=S})\text{CH}_3$ (4(1H)-Pyridinethione, 1-methyl-)	18438-38-5 18794-33-7 22581-72-2 2044-27-1 6887-59-8	** ** ** ** ** ** ** ** **	8.24 ± 0.03 (V) 8.47 ± 0.02 8.41 ± 0.03 (V) 8.93 ± 0.02 8.73 ± 0.03 (V) 9.00 ± 0.02 7.69 ± 0.03 (V) 7.84 ± 0.02 7.6 ± 0.03 (V)	PE EI PE EI PE EI PE EI PE	4711 3636 4711 3636 4711 3636 4711 3636 4711
$\text{C}_7\text{H}_5\text{NS}^+$	$\text{C}_6\text{H}_5\text{NCS}$ (Benzene, isothiocyanato-)	103-72-0	**	8.53 (V)	PE	4495

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₇H₅NS⁺	C ₇ H ₅ NS (Benzothiazole)	95-16-9	**	8.85 (V)	PE	4437
C₇H₉NS⁺	C ₆ H ₄ (SCH ₃)NH ₂ (Benzenamine, 4-(methylthio)-)	104-96-1	**	7.6 (V)	PE	5403
			**	7.60±0.01 (V)	PE	4389
C₈H₇NS⁺	C ₇ H ₄ NS(CH ₃) (Benzothiazole, 2-methyl-)	120-75-2	**	8.65 (V)	PE	4437
C₈H₈NS⁺	C ₆ H ₅ NHCSC ₃ (Ethanethioamide, N-phenyl-)	637-53-6	H	9.60	EI	4834
	C ₆ H ₄ FNHCSC ₃ (Ethanethioamide, N-(2-fluorophenyl)-)	39184-82-2	F	9.50	EI	4834
	C ₆ H ₄ ClNHCSC ₃ (Ethanethioamide, N-(2-chlorophenyl)-)	39184-83-3	Cl	8.65	EI	4834
			Cl	8.65	EI	4834
	C ₆ H ₄ BrNHCSC ₃ (Ethanethioamide, N-(2-bromophenyl)-)	62635-46-5	Br	8.50	EI	4834
	C ₆ H ₄ IINHCSC ₃ (Ethanethioamide, N-(2-iodophenyl)-)	39184-84-4	I	8.55	EI	4834
C₈H₉NS⁺	C ₆ H ₅ NHCSC ₃ (Ethanethioamide, N-phenyl-)	637-53-6	**	8.20	EI	4834
C₉H₁₃NS⁺	C ₆ H ₄ (SCH ₃)N(CH ₃) ₂ (Benzenamine, N,N-dimethyl-4-(methylthio)-)	2388-51-4	**	7.29±0.01 (V)	PE	4389
C₁₀H₉NS⁺	C ₆ H ₅ CH ₂ (C ₃ H ₂ NS) (Isothiazole, 4-(phenylmethyl)-)	36412-26-7	**	9.05	PE	3587
			**	9.35	EI	3587
C₁₂H₉NS⁺	C ₁₂ H ₉ NS (10H-Phenothiazine)	92-84-2	**	7.26±0.08 (V)	PE	4667
			**	6.74±0.07	CTS	4079
			**	6.87	CTS	4035
C₁₂H₁₆NS⁺	C ₆ H ₄ ClNHCSC ₃ C(CH ₃) ₃ (Butanethioamide, N-(2-chlorophenyl)-3,3-dimethyl-)	62635-54-5		8.65	EI	4834
C₁₃H₁₁NS⁺	C ₁₂ H ₉ NS(CH ₃) (10H-Phenothiazine, 10-methyl-)	1207-72-3	**	7.15±0.07 (V)	PE	4667
			**	6.73±0.07	CTS	4079
C₁₄H₁₁NS⁺	C ₁₃ H ₇ (=S)NHCH ₃ (Phenalene, 1-thione-9-methylamino-)	XXXXXX-XX-X	**	7.21±0.04 (V)	PE	5595
C₁₆H₁₅NS⁺	C ₁₃ H ₇ (=S)NH(iso-C ₃ H ₇) (Phenalene, 1-thione-9-(methylethyl)amino-)	XXXXXX-XX-X	**	7.17±0.04 (V)	PE	5595

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{CH}_4\text{N}_2\text{S}^+$	$(\text{NH}_2)_2\text{CS}$	62-56-6	**	7.9	PE	4221
			**	8.41 ± 0.03 (V)	PE	4253
			**	8.41 (V)	PE	4323
			**	8.50 (V)	PE	4469
			**	8.50	EI	4834
$\text{C}_2\text{H}_8\text{N}_2\text{S}^+$	$(\text{CH}_3)_2\text{S}(=\text{NH})_2$	13904-95-5	**	8.87 (V)	PE	4827
			**	8.87 (V)	PE	5207
$\text{C}_3\text{H}_6\text{N}_2\text{S}^+$	$\text{C}_3\text{H}_6\text{N}_2=\text{S}$ (2-Imidazolidinethione)	96-45-7	**	8.15 ± 0.03 (V)	PE	4253
	$\text{C}_2\text{H}_5\text{N}_2\text{SCH}_3$ (1,2,5-Thia(S^{IV})diazole, 3,4-dihydro-3-methyl-)	24692-43-1	**	8.92 (V)	PE	4024
	$(\text{CH}_3\text{NH})_2\text{CS}$	534-13-4	**	8.08 ± 0.03 (V)	PE	4253
$\text{C}_4\text{H}_2\text{N}_2\text{S}^+$	$\text{C}_3\text{H}_2\text{NS}(\text{CN})$ (4-Iothiazolecarbonitrile)	3912-37-6	**	10.55	EI	3587
$\text{C}_4\text{H}_6\text{N}_2\text{S}^+$	$\text{C}_3\text{H}_5\text{N}_2(=\text{S})\text{CH}_3$ (2 <i>H</i> -Imidazole-2-thione, 1,3-dihydro-1-methyl-)	60-56-0	**	7.41 ± 0.03 (V)	PE	4253
$\text{C}_4\text{H}_6\text{N}_2\text{S}^+$	$\text{C}_2\text{H}_5\text{N}_2\text{S}(\text{CH}_3)_2$ (1,2,5-Thia(S^{IV})diazole, 3,4-dihydro-3,3-dimethyl-)	24692-45-3	**	9.62 (V)	PE	4024
$\text{C}_5\text{H}_8\text{N}_2\text{S}^+$	$\text{C}_3\text{H}_2\text{N}_2(=\text{S})(\text{CH}_3)_2$ (2 <i>H</i> -Imidazole-2-thione, 1,3-dihydro-1,3-dimethyl-)	6596-81-2	**	7.27 ± 0.03 (V)	PE	4253
	$\text{C}_3\text{H}_2\text{N}_2(=\text{S})(\text{CH}_3)_2$ (3 <i>H</i> -Pyrazole-3-thione, 1,2-dihydro-1,2-dimethyl-)	55833-07-3	**	7.55 (V)	PE	5309
$\text{C}_5\text{H}_{10}\text{N}_2\text{S}^+$	$\text{C}_3\text{H}_5\text{N}_2(=\text{S})(\text{CH}_3)_2$ (2-Imidazolidinethione, 1,3-dimethyl-)	13461-16-0	**	7.95 ± 0.03 (V)	PE	4253
$\text{C}_5\text{H}_{12}\text{N}_2\text{S}^+$	$((\text{CH}_3)_2\text{N})_2\text{CS}$	2782-91-4	**	7.82 ± 0.03	PE	4253
			**	7.82 (V)	PE	4323
			**	7.84 (V)	PE	4469
	$\text{C}_3\text{H}_6\text{N}_2\text{S}(\text{CH}_3)_2$ (2 <i>H</i> -1,3,4-Thiadiazine, tetrahydro-3,4-dimethyl-)	66175-24-4	**	8.18 (V)	PE	5215
			**			
$\text{C}_6\text{H}_4\text{N}_2\text{S}^+$	$\text{C}_6\text{H}_4\text{N}_2\text{S}$ (1,2,3-Benzothiadiazole)	273-77-8	**	9.15 (V)	PE	3852
			**	9.50 ± 0.05	EI	4316
	$\text{C}_6\text{H}_4\text{N}_2\text{S}$ (2,1,3-Benzothiadiazole)	273-13-2	**	8.98	PE	4017
			**	9.00 (V)	PE	3852
$\text{C}_6\text{H}_{10}\text{N}_2\text{S}^+$	$\text{C}_3\text{H}_5\text{N}_2(=\text{S})(\text{CH}_3)_3$ (3 <i>H</i> -Pyrazole-3-thione, 1,2-dihydro-1,2,4-trimethyl-)	66187-19-7	**	7.60 (V)	PE	5309

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_6\text{H}_{12}\text{N}_2\text{S}^+$	$\text{C}_4\text{H}_6\text{N}_2(=\text{S})(\text{CH}_3)_2$ (2(1 <i>H</i>)-Pyrimidinethione, tetrahydro-1,3-dimethyl-)	16597-35-6	**	7.58 (V)	PE	4323
$\text{C}_7\text{H}_7\text{N}_2\text{S}^+$	$\text{C}_6\text{H}_5\text{NHCSNH}_2$ (Thiourea, phenyl-) $\text{C}_6\text{H}_4(\text{CH}_3)\text{NHCSNH}_2$ (Thiourea, (2-methylphenyl)-) $\text{C}_6\text{H}_4(\text{OCH}_3)\text{NHCSNH}_2$ (Thiourea, (2-methoxyphenyl)-) $\text{C}_6\text{H}_4(\text{NO}_2)\text{NHCSNH}_2$ (Thiourea, (2-nitrophenyl)-) $\text{C}_6\text{H}_4\text{FNHCSNH}_2$ (Thiourea, (2-fluorophenyl)-) $\text{C}_6\text{H}_4\text{ClNHCSNH}_2$ (Thiourea, (2-chlorophenyl)-) $\text{C}_6\text{H}_4\text{BrNHCSNH}_2$ (Thiourea, (2-bromophenyl)-) $\text{C}_6\text{H}_4\text{INHCSNH}_2$ (Thiourea, (2-iodophenyl)-)	103-85-5 614-78-8 1516-37-6 51039-84-0 656-32-6 5344-82-1 5391-30-0 62635-52-3	H CH_3 ** NO_2 F Cl Br I	9.65 9.60 8.95 8.60 9.60 8.50 8.35 8.55	EI	4834 4834 4834 4834 4834 4834 4834 4834
$\text{C}_7\text{H}_8\text{N}_2\text{S}^+$	$\text{C}_6\text{H}_5\text{NHCSNH}_2$ (Thiourea, phenyl-)	103-85-5	**	8.10	EI	4834
$\text{C}_8\text{H}_8\text{N}_2\text{S}^+$	$\text{C}_7\text{H}_3\text{NS}(\text{NH}_2)\text{CH}_3$ (6-Benzothiazolamine, 2-methyl-)	2941-62-0	**	7.70 (V)	PE	4437
$\text{C}_8\text{H}_9\text{N}_2\text{S}^+$	$\text{C}_6\text{H}_5\text{N}(\text{CH}_3)\text{CSNH}_2$ (Thiourea, N-methyl-N-phenyl-) $\text{C}_6\text{H}_4\text{ClNHCSNHCH}_3$ (Thiourea, N-(2-chlorophenyl)-N'-methyl-)	4104-75-0 30954-73-5	Cl	9.70 8.35	EI EI	4834 4834
$\text{C}_8\text{H}_{10}\text{N}_2\text{S}^+$	$\text{C}_6\text{H}_5\text{N}(\text{CH}_3)\text{CSNH}_2$ (Thiourea, N-methyl-N-phenyl-) $\text{C}_6\text{H}_4(\text{CH}_3)\text{NHCSNH}_2$ (Thiourea, (2-methylphenyl)-)	2724-69-8 614-78-8	** ** **	8.00 8.05 ± 0.05 8.20	EI EI EI	4834 4834 4834
$\text{C}_8\text{H}_{18}\text{N}_2\text{S}^+$	$((\text{CH}_3)_3\text{CN})_2\text{S}$	2056-74-8	**	8.65 (V)	PE	4024
$\text{C}_9\text{H}_6\text{N}_2\text{S}^+$	$\text{C}_7\text{H}_3\text{NS}(\text{CN})\text{CH}_3$ (6-Benzothiazolecarbonitrile, 2-methyl-)	42474-60-2	**	9.15 (V)	PE	4437
$\text{C}_9\text{H}_{10}\text{N}_2\text{S}^+$	$\text{C}_7\text{H}_4\text{N}_2(\text{S})(\text{CH}_3)_2$ (2H-Benzimidazole-2-thione, 1,3-dihydro-1,3-dimethyl-)	3418-46-0	**	7.46	PE	4555
$\text{C}_{10}\text{H}_{11}\text{N}_2\text{S}^+$	$\text{C}_6\text{H}_4\text{ClNHCSNH}_2\text{C}_2\text{H}_5$ (Thiourea, N-(2-chlorophenyl)-N'-ethyl-)	19384-08-8	Cl	8.35	EI	4834
$\text{C}_{10}\text{H}_{12}\text{N}_2\text{S}^+$	$\text{C}_6\text{H}_5\text{NHCSNHC}_2\text{H}_5$ (Thiourea, N-ethyl-N'-phenyl-)	2741-06-2	**	7.95 ± 0.05	EI	4834

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₉H₁₄N₂S⁺	C ₅ H ₂ N ₂ S(CH ₃) ₄ (Isothiazolo[5,1- <i>e</i>]isothiazole-7-S ^{IV} ,1,6-dihydro-1,2,5,6-tetramethyl-)	52353-57-8	**	6.44 (V)	PE	4406
C₁₀H₁₃N₂S⁺	C ₆ H ₄ CINHCSNHCH(CH ₃) ₂ (Thiourea, N-(2-chlorophenyl)-N'-(1-methylethyl)-)	62635-49-8	Cl	8.25	EI	4834
C₁₀H₁₄N₂S⁺	C ₆ H ₅ NHCSNHCH(CH ₃) ₂ (Thiourea, N-(1-methylethyl)-N'-phenyl-)	15093-36-4	**	7.90±0.05	EI	4834
C₁₁H₁₅N₂S⁺	C ₆ H ₄ CINHCSNH(C(CH ₃) ₃ (Thiourea, N-(2-chlorophenyl)-N'-(1,1-dimethylethyl)-)	62635-50-1	Cl	8.10	EI	4834
C₁₁H₁₆N₂S⁺	C ₆ H ₅ NHCSNH(C(CH ₃) ₃ (Thiourea, N-(1,1-dimethylethyl)-N'-phenyl-)	14327-04-9	**	7.85±0.05	EI	4834
C₁₂H₁₂N₂S⁺	(C ₆ H ₄ NH ₂) ₂ S (Benzenamine, 4,4'-thiobis-)	139-65-1	**	6.75	PI	4328
C₁₂H₂₀N₂S⁺	C ₁₂ H ₂₀ N ₂ S (7-Thia-14,15-diazadispiro[5.1.5.2]pentadec-14-ene)	28037-21-0	**	8.57 (V)	PE	4429
C₁₆H₁₈N₂S⁺	C ₁₂ H ₈ NSCH ₂ CH ₂ N(CH ₃) ₂ (10 <i>H</i> -Phenothiazine-10-ethanamine, <i>N,N</i> -dimethyl-)	522-24-7	**	8.25±0.07	CTS	4079
C₁₇H₂₀N₂S⁺	C ₁₂ H ₈ NS(CH ₂) ₃ N(CH ₃) ₂ (10 <i>H</i> -Phenothiazine-10-propanamine, <i>N,N</i> -dimethyl-)	58-40-2	**	7.20±0.06 (V)	PE	4667
			**	8.22±0.07	CTS	4079
C₁₈H₂₂N₂S⁺	C ₁₂ H ₈ NSCH ₂ CH ₂ N(C ₂ H ₅) ₂ (10 <i>H</i> -Phenothiazine-10-ethanamine, <i>N,N</i> -diethyl-)	60-91-3	**	7.85±0.07	CTS	4079
C₄H₇N₃S⁺	C ₂ HN ₃ (=S)(CH ₃) ₂ (4 <i>H</i> -1,2,3-Triazole-4-thione,2,3-dihydro-2,3-dimethyl-)	64808-28-2	**	7.97 (V)	PE	5309
	C ₂ HN ₃ (S)(CH ₃) ₂ (1 <i>H</i> -1,2,3-Triazolium,4-mercaptop-1,3-dimethyl-hydroxide,inner salt)	34618-67-2	**	7.25 (V)	PE	5309
C₅H₉N₃S⁺	C ₂ N ₃ (=S)(CH ₃) ₃ (3 <i>H</i> -1,2,4-Triazole-3-thione, 2,4-dihydro-2,4,5-trimethyl-)	37526-42-4	**	7.63 (V)	PE	4439
	C ₂ N ₃ (S)(CH ₃) ₃ (4 <i>H</i> -1,2,3-Triazole-4-thione,2,3-dihydro-2,3,5-trimethyl-)	64808-27-1	**	7.95 (V)	PE	5309
	C ₂ N ₃ (S)(CH ₃) ₃ (1 <i>H</i> -1,2,3-Triazolium,4-mercaptop-1,3,5-trimethyl-hydroxide,inner salt)	66187-20-0	**	7.02 (V)	PE	5309
C₇H₉N₃S⁺	C ₆ H ₄ (NH ₂)NHCSNH ₂ (Thiourea, (2-aminophenyl)-)	3394-09-0	**	8.10	EI	4834
C₉H₉N₃S⁺	C ₆ H ₅ C ₂ HN ₃ (=S)CH ₃ (3 <i>H</i> -1,2,4-Triazole-3-thione,2,4-dihydro-4-methyl-5-phenyl-)	38942-51-7	**	7.78 (V)	PE	4439

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{10}\text{H}_{11}\text{N}_3\text{S}^+$	$\text{C}_6\text{H}_5\text{C}_2\text{N}_3(=\text{S})(\text{CH}_3)_2$ (3H-1,2,4-Triazole-3-thione, 2,4-dihydro-2,4-dimethyl-5-phenyl-)	7112-00-7	**	7.59 (V)	PE	4439
$\text{C}_{20}\text{H}_{25}\text{N}_3\text{S}^+$	$\text{C}_{12}\text{H}_8\text{NS}(\text{CH}_2)_3\text{C}_4\text{H}_8\text{N}_2\text{CH}_3$ (10H-Phenothiazine, 10-[3-(4-methyl-1-piperazinyl)propyl]-)	84-97-9	**	6.87 ± 0.07	CTS	4079
$\text{C}_3\text{H}_3\text{NS}_2^+$	$\text{C}_3\text{H}_3\text{NS=S}$ (2(3H)-Thiazolethione)	5685-05-2	**	7.74 ± 0.03 (V)	PE	4253
$\text{C}_3\text{H}_5\text{NS}_2^+$	$\text{C}_3\text{H}_5\text{NS=S}$ (2-Thiazolidinethione)	96-53-7	**	8.25 ± 0.03 (V)	PE	4253
$\text{C}_4\text{H}_5\text{NS}_2^+$	$\text{C}_3\text{H}_2\text{NS}(=\text{S})\text{CH}_3$ (2(3H)-Thiazolethione, 3-methyl-)	5685-07-4	**	7.68 ± 0.03 (V)	PE	4253
$\text{C}_4\text{H}_7\text{NS}_2^+$	$\text{C}_3\text{H}_4\text{NS}(=\text{S})\text{CH}_3$ (2-Thiazolidinethione, 3-methyl-)	1908-87-8	**	8.04 ± 0.03	PE	4253
$\text{C}_4\text{H}_9\text{NS}_2^+$	$(\text{CH}_3)_2\text{NCSSCH}_3$	3735-92-0	** **	8.01 ± 0.03 8.01 (V)	PE PE	4253 4323
$\text{C}_5\text{H}_7\text{NS}_2^+$	$\text{C}_3\text{HNS}(=\text{S})(\text{CH}_3)_2$ (2(3H)-Thiazolethione, 3,4-dimethyl-) $\text{C}_3\text{HNS}(=\text{S})(\text{CH}_3)_2$ (2(3H)-Thiazolethione, 4,5-dimethyl-)	5316-79-0 5351-51-9	** **	7.55 ± 0.03 (V) 7.56 ± 0.03 (V)	PE PE	4253 4253
$\text{C}_5\text{H}_9\text{NS}_2^+$	$\text{C}_3\text{H}_3\text{NS}(=\text{S})(\text{CH}_3)_2$ (2-Thiazolidinethione, 4,4-dimethyl-)	1908-88-9	**	8.18 ± 0.03 (V)	PE	4253
$\text{C}_6\text{H}_9\text{NS}_2^+$	$\text{C}_3\text{NS}(=\text{S})(\text{CH}_3)_3$ (2(3H)-Thiazolethione, 3,4,5-trimethyl-)	21364-38-5	**	7.45 ± 0.03 (V)	PE	4253
$\text{C}_7\text{H}_5\text{NS}_2^+$	$\text{C}_7\text{H}_5\text{NS(S)}$ (2(3H)-Benzothiazolethione)	149-30-4	**	7.99	PE	4555
$\text{C}_8\text{H}_7\text{NS}_2^+$	$\text{C}_7\text{H}_4\text{NS(S)}(\text{CH}_3)$ (2(3H)-Benzothiazolethione, 3-methyl-)	2254-94-6	**	7.81	PE	4555
$\text{C}_8\text{H}_{11}\text{NS}_2^+$	$\text{C}_5\text{H}_2\text{NS}_2(\text{CH}_3)_3$ (Methanamine, N-[1-methyl-2-(5-methyl-3H-1,2-dithiol-3-ylidene)ethylidene]-)	57254-27-0	**	7.17 (V)	PE	4406
$\text{C}_{12}\text{H}_{13}\text{NS}_2^+$	$(\text{C}_4\text{H}_2\text{S})_2\text{C}_2\text{H}_4\text{NC}_2\text{H}_5$ (4H-Dithieno[2,3-c:3',2'-e]azepine, 5-ethyl-5,6-dihydro-)	64504-69-4	**	7.9 (V)	PE	5422
$\text{C}_{16}\text{H}_{13}\text{NS}_2^+$	$(\text{C}_4\text{H}_2\text{S})_2\text{C}_2\text{H}_4\text{NC}_6\text{H}_5$ (4H-Dithieno[2,3-c:3',2'-e]azepine, 5,6-dihydro-5-phenyl-) $(\text{C}_4\text{H}_2\text{S})_2\text{C}_2\text{H}_4\text{NC}_6\text{H}_5$ (4H-Dithieno[3,4-c:3',4'-e]azepine, 5,6-dihydro-5-phenyl-)	40306-87-4 64504-70-7	** **	7.9 (V) 7.5 (V)	PE PE	5422 5422

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₂₃H₁₅NS₂⁺	C ₅ S ₂ N(C ₆ H ₅) ₃ (Thieno[3,4- <i>c</i>]isothiazole-5-S ^{IV} ,3,4,6-triphenyl-)	61164-97-4	**	6.9 (V)	PE	5341
C₂H₆N₂S₂⁺	NH ₂ NHC(=S)SCH ₃	5397-03-5	**	8.81	PE	5285
C₃H₄N₂S₂⁺	C ₂ HN ₂ S(=S)CH ₃ (1,3,4-Thiadiazole-2(3H)-thione, 5-methyl-)	29490-19-5	**	8.33 (V)	PE	4439
C₃H₈N₂S₂⁺	NH ₂ N(CH ₃)C(=S)SCH ₃	20184-94-5	**	8.39	PE	5285
C₄H₆N₂S₂⁺	C ₂ N ₂ S(=S)(CH ₃) ₂ (1,3,4-Thiadiazole-2(3H)-thione,3,5-dimethyl-)	7111-96-8	**	7.97 (V)	PE	4439
C₄H₁₀N₂S₂⁺	N(CH ₃) ₂ NHC(=S)SCH ₃	25554-63-6	**	8.37	PE	5285
C₆H₆N₂S₂⁺	C ₆ H ₅ C ₂ HN ₂ S(=S) (1,3,4-Thiadiazole-2(3H)-thione,5-phenyl-)	5585-19-3	**	8.13 (V)	PE	4439
C₈H₁₀N₂S₂⁺	NH(C ₆ H ₅)NHC(=S)SCH ₃ (Hydrazinecarbodithioic acid,2-phenyl-methyl ester)	50878-38-1	**	8.47	PE	5285
C₉H₈N₂S₂⁺	C ₆ H ₅ C ₂ N ₂ S(=S)CH ₃ (1,3,4-Thiadiazole-2(3H)-thione,3-methyl-5-phenyl-)	5770-97-8	**	7.87 (V)	PE	4439
C₁₂H₂₀N₂S₂⁺	H ₂ (CH ₃ C(=S)CH ₂ C(CH ₃)NCH ₂) ₂	40006-83-5	**	7.60 (V)	PE	5446
C₁₄H₁₄N₂S₂⁺	N(C ₆ H ₅) ₂ NHC(=S)SCH ₃ (Hydrazinecarbodithioic acid,2,2-diphenyl-methyl ester)	50878-39-2	**	7.47	PE	5285
C₁₆H₁₄N₂S₂⁺	(C ₆ H ₄ N(CH ₃)CS) ₂ (Benzothiazole,2,3-dihydro-3-methyl-2-(3-methyl-2(3H)-benzothiazolylidene)-)	2786-70-1	**	5.75±0.2	OTH	5278
C₂₁H₂₆N₂S₂⁺	C ₂₁ H ₂₆ N ₂ S ₂	50-52-2	**	7.00±0.08 (V)	PE	4667
C₅H₃NS₃⁺	(C ₅ S ₂) ₂ =S(CN)(CH ₃) (3H-1,2-Dithiole-4-carbonitrile, 5-methyl-3-thioxo-)	24045-79-2	**	8.70 (V)	PE	4403
B₂C₃H₁₀N₂S₂⁺	N ₂ B ₂ SH(CH ₃) ₃ (1,3,4,2,5-Thiadiazadiborolidine, 2,3,5-trimethyl-)	57877-85-7	**	8.32 (V)	PE	4526
B₂C₄H₁₂N₂S₂⁺	N ₂ B ₂ S(CH ₃) ₄ (1,3,4,2,5-Thiadiazadiborolidine, 2,3,4,5-tetramethyl-)	40392-37-8	**	8.00 (V)	PE	4526
B₂C₂H₇NS₂⁺	NB ₂ S ₂ H(CH ₃) ₂ (1,2,4,3,5-Dithiazadiborolidine, 3,5-dimethyl-)	57877-87-9	**	8.69 (V)	PE	4526

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
B₂C₃H₉NS₂⁺	NB ₂ S ₂ (CH ₃) ₃ (1,2,4,3,5-Dithiazadiborolidine, 3,4,5-trimethyl-)	57877-88-0	**	8.58 (V)	PE	4526
B₂C₅H₁₅N₃S₂⁺	N ₃ B ₂ (CH ₃) ₃ (SCH ₃) ₂ (1,2,4,3,5-Triazadiborolidine, 1,2,4-trimethyl-3,5-bis(methylthio)-)	40392-36-7	**	7.74 (V)	PE	4526
B₂C₆H₁₈N₄S₂⁺	B ₂ N ₄ (CH ₃) ₄ (SCH ₃) ₂ (1,2,4,5,3,6-Tetrazadiborine, hexahydro-1,2,4,5-tetramethyl-3,6-bis(methylthio)-)	54154-14-2	**	7.39 (V)	PE	4299
OS⁺						
(² P _{3/2})	SO(³ Σ ⁻)	13827-32-2	**	10.29±0.01	PE	4230
			**	10.31	PE	4186
(² P ₁)			**	10.32	PE	3701
(² P _{1/2})				10.33±0.01	PE	4230
(⁴ P ₁)			**	11.3	PE	3701
(⁴ P ₃)			**	13.50±0.05	PE	4230
(² P ₃)			**	~14.4	PE	4230
(⁴ Σ _g)			**	14.94±0.01	PE	4230
(⁴ Σ _u)			**	14.96	PE	3701
(² Σ _g)			**	16.44±0.01	PE	4230
(² Π ₁)			**	~19.6	PE	4230
(² Π ₃)			**	10.20±0.03	EI	4920
(² Π ₁)			**	10.28±0.02	EI	3816
SO ₂		7446-09-5	O	15.930±0.005	PE	5388
S ₂ O		20901-21-7	**	13.745±0.006	PI	4762
COS		463-58-1	C	19.8	EI	3779
O₂S⁺						
(² A ₁)	SO ₂	7446-09-5	**	12.3	PE	3865
(² A ₁)			**	12.31	PE	4092
(² A ₁)			**	12.50 (V)	PE	3879
(² A ₁)			**	12.54 (V)	PE	4024
(² A ₂)			**	13.01 (V)	PE	4092
(² A ₂)			**	13.24 (V)	PE	3879
(² A ₂)			**	13.25 (V)	PE	4024
(² B ₁)			**	13.30 (V)	PE	4092
(² B ₂)			**	13.47 (V)	PE	3879
(² B ₂)			**	13.56 (V)	PE	4024
(² B ₁)			**	15.99	PE	3879
(² B ₂)			**	15.992±0.003	PE	3865
(² A ₁)			**	16.324±0.004	PE	3865
(² A ₁)			**	16.33	PE	3879
(² B ₁)			**	16.498±0.004	PE	3865
(² B ₁)			**	16.57 (V)	PE	4092
(² B ₁)			**	20.06±0.05	PE	3865
O₃S⁺						
SO ₃		7446-11-9	**	12.73±0.05	PE	4388
			**	12.81±0.03	PE	4485
(² A ₂)			**	12.82±0.01	PE	4516
			**	12.82±0.03	PE	4149
			**	13.75±0.03	PE	4485
			**	~14.5 (V)	PE	4485
			**	14.83±0.03	PE	4485
			**	17.86±0.03	PE	4485
OS₂⁺						
S ₂ O		20901-21-7	**	10.58±0.01	PI	4762
(² A')			**	10.52	PE	4092
(² A')			**	10.52 (V)	PE	4244

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
OS₂⁺						
(² A')	S ₂ O	20901-21-7	**	10.53±0.02	PE	3841
(² A')			**	10.62	PE	3692
(² A'')			**	11.22	PE	4092
(² A')			**	11.22 (V)	PE	4244
(² A')			**	11.25±0.02	PE	3841
(² A'')			**	11.31±0.02	PE	3841
(² A')			**	11.32	PE	3692
(² A'')			**	11.34	PE	4092
(² A'')			**	11.34 (V)	PE	4244
(² A'')			**	11.37	PE	3692
(² A')			**	14.3±0.02	PE	3841
(² A'')			**	14.3	PE	3692
(² A'')			**	14.62 (V)	PE	4092
(² A'')			**	14.62 (V)	PE	4244
(² A'')			**	14.82 (V)	PE	4244
(² A'')			**	14.84 (V)	PE	4092
(² A'')			**	14.9±0.02	PE	3841
(² A'')			**	15.5±0.02	PE	3841
(² A')			**	15.5	PE	3692
(² A')			**	15.80 (V)	PE	4092
(² A')			**	15.80 (V)	PE	4244
(² A')			**	18.5 (V)	PE	4244
(² A')			**	18.50 (V)	PE	4092
COS⁺						
(² Π)	COS	463-58-1	**	11.190	PI	4994
(² Π)			**	15.075	PI	4994
(² Σ ⁺)			**	16.043	PI	4994
(² Σ ⁺)			**	17.955	PI	4994
			**	11.177±0.002	PE	5256
(² Π)			**	11.18±0.01	PE	3965
			**	11.19 (V)	PE	5055
(² Π _{3/2})			**	11.22	PE	4073
(² Π)			**	15.09±0.01	PE	3965
(² Σ ⁺)			**	16.05±0.01	PE	3965
(² Σ ⁺)			**	17.96±0.01	PE	3965
(² Π)			**	11.19±0.05	EI	5027
			**	11.3	EI	3779
CH₂OS⁺						
	C ₃ H ₆ OS (1,3-Oxathiolane)	2094-97-5	C ₂ H ₄	10.4±0.3	EI	3598
CH₃OS⁺						
	(CH ₃) ₂ SO	67-68-5	CH ₃	10.91±0.16	EI	5311
	(C ₂ H ₅) ₂ SO	70-29-1	C ₂ H ₄ +CH ₃	12.04±0.08	EI	5311
CH₄OS⁺						
	C ₂ H ₅ SOCH ₃	1669-98-3	C ₂ H ₄	10.00±0.11	EI	5311
	(iso-C ₃ H ₇)SOCH ₃	XXXXX-XX-X	C ₃ H ₆	9.28±0.03	EI	5311
C₂H₄OS⁺						
	CH ₃ COSH	507-09-5	**	10.06 (V)	PE	4769
	C ₂ H ₄ SO (Thiirane, 1-oxide)	7117-41-1	**	9.66 (V)	PE	3646
			**	9.66 (V)	PE	4295
C₂H₆OS⁺						
	(CH ₃) ₂ SO	67-68-5	**	9.01 (V)	PE	3646
			**	9.01 (V)	PE	4295
			**	9.11 (V)	PE	3705

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₂H₆OS⁺						
	(CH ₃) ₂ SO	67-68-5	** ** **	9.08±0.09 9.20 9.20±0.05	EI EI EI	5311 5292 3498
	(C ₂ H ₅) ₂ SO	70-29-1	C ₂ H ₄	9.86±0.08	EI	5311
C₃H₅OS⁺						
	C ₃ H ₆ OS (1,3-Oxathiolane)	2094-97-5	H	10.8±0.1	EI	3598
C₃H₆OS⁺						
	CH ₃ C=OSCH ₃	1534-08-3	**	9.65 (V)	PE	4427
	C ₂ H ₃ S(CH ₃)O	10258-86-3	**	9.02 (V)	PE	4295
	C ₃ H ₆ OS (1,3-Oxathiolane)	2094-97-5	**	9±0.05	EI	3598
C₃H₈OS⁺						
	C ₂ H ₅ SOCH ₃	1669-98-3	**	8.89±0.08	EI	5311
C₄H₄OS⁺						
	C ₄ H ₄ S(=O) (2(5H)-Thiophenone)	3354-32-3	**	9.78±0.05	EI	4666
C₄H₈OS⁺						
	CH ₃ COSC ₂ H ₅	625-60-5	**	9.44 (V)	PE	4769
	CH ₃ C=SOC ₂ H ₅	926-67-0	**	8.82 (V)	PE	4427
	C ₄ H ₈ OS (1,4-Oxathiane)	15980-15-1	**	8.67 (V)	PE	3733
	C ₂ H ₃ SCH ₂ OCH ₃ (Thiirane, methoxymethyl-)	19858-14-1	**	8.77 (V)	PE	4747
	C ₄ H ₈ SO (Thiophene, tetrahydro-1-oxide)	1600-44-8	** ** **	8.77 (V) 9.07±0.05 9.07	PE EI EI	3646 4295 3498 5292
C₄H₁₀OS⁺						
	(CH ₃ CH ₂) ₂ SO	70-29-1	** ** **	8.76 (V) 8.76 (V) 8.75±0.03	PE PE EI	3646 4295 5311
	(iso-C ₃ H ₇)SOCH ₃	XXXXX-XX-X	**	8.71±0.04	EI	5311
	(iso-C ₃ H ₇) ₂ SO	2211-89-4	C ₃ H ₆	9.22±0.18	EI	5311
C₅H₃OS⁺						
	C ₆ H ₅ COC ₄ H ₃ S (Methanone, phenyl-2-thienyl-)	135-00-2	C ₆ H ₅	11.8±0.1	EI	5493
C₅H₄OS⁺						
	C ₅ H ₄ O(=S) (4H-Pyran-4-thione)	1120-93-0	**	8.10±0.05 (V)	PE	5002
	C ₄ H ₃ SCHO (2-Thiophene carboxaldehyde)	98-03-3	**	9.37±0.05 (V)	PE	4626
	C ₅ H ₄ S(=O) (4H-Thiopyran-4-one)	1003-41-4	**	9.55±0.05 8.97±0.05 (V)	EI PE	3482 5002
C₅H₆OS⁺						
	C ₄ H ₃ S(OCH ₃) (Thiophene, 2-methoxy-)	16839-97-7	** ** **	8.14±0.05 8.30±0.05 8.08	EI EI CTS	4666 3482 4382
	C ₄ H ₃ OSCH ₃ (Furan, 2-(methylthio)-)	13129-38-9	**	8.58±0.05 (V)	PE	4626

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₅H₆OS⁺						
	C ₄ H ₂ O(SH)CH ₃ (2-Furanthiol, 5-methyl-)	60965-60-8	**	8.45±0.05	EI	4706
	C ₄ H ₃ S(=O)(CH ₃) (2(5H)-Thiophenone, 3-methyl-)	33687-85-3	**	9.60±0.05	EI	4666
	C ₄ H ₃ S(=O)(CH ₃) (2(5H)-Thiophenone, 5-methyl-)	7210-64-2	**	9.16±0.05	EI	4666
C₅H₈OS⁺						
	C ₅ H ₆ S(=O) (4H-Thiopyran-4-one, tetrahydro-)	1072-72-6	**	8.90±0.05	PE	5002
C₅H₁₂OS⁺						
	C ₂ H ₅ S(CH ₂) ₂ OCH ₃	56444-81-6	**	8.33±0.02	PI	5531
C₆H₆OS⁺						
	C ₄ H ₃ SCOCH ₃ (Ethanone, 1-(2-thienyl)-)	88-15-3	**	9.20±0.05	EI	3482
	C ₄ H ₃ SCOCH ₃ (Ethanone, 1-(3-thienyl)-)	1468-83-3	**	9.32±0.05	EI	3482
C₆H₈OS⁺						
	C ₄ H ₂ S(CH ₃)(OCH ₃) (Thiophene, 2-methoxy-3-methyl-)	33687-87-5	**	8.05±0.05	EI	4666
	C ₄ H ₂ S(CH ₃)(OCH ₃) (Thiophene, 2-methoxy-5-methyl-)	31053-55-1	**	8.01±0.05	EI	4666
	C ₄ H ₂ O(CH ₃)SCH ₃ (Furan, 2-methyl-5-(methylthiol)-)	13678-59-6	**	8.15±0.05	EI	4706
	C ₄ HO(SH)(CH ₃) ₂ (3-Furanthiol, 2,5-dimethyl-)	55764-23-3	**	8.23±0.05	EI	4706
	C ₄ H ₂ S(=O)(CH ₃) ₂ (2(3H)-Thiophenone, 3,3-dimethyl-)	33687-82-0	**	8.77±0.05	EI	4666
	C ₄ H ₂ S(=O)(CH ₃) ₂ (2(5H)-Thiophenone, 3,4-dimethyl-)	33922-75-7	**	9.44±0.05	EI	4666
	C ₄ H ₂ S(=O)(CH ₃) ₂ (2(5H)-Thiophenone, 3,5-dimethyl-)	33687-84-2	**	9.35±0.05	EI	4666
	C ₄ H ₂ S(=O)(CH ₃) ₂ (2(5H)-Thiophenone, 4,5-dimethyl-)	35983-76-7	**	9.13±0.05	EI	4666
	C ₄ H ₂ S(=O)(CH ₃) ₂ (3(2H)-Thiophenone, 2,5-dimethyl-)	3760-59-6	**	8.55±0.05	EI	4673
	C ₄ H ₅ S(=O)(C ₂ H ₅) (2(5H)-Thiophenone, 5-ethyl-)	56761-30-9	**	9.08±0.05	EI	4666
C₆H₁₁OS⁺						
	C ₄ H ₅ OS(CH ₃) ₃ (1,3-Oxathiane, 2,4,6-trimethyl-, (2 α ,4 α ,6 α -))	22521-88-6	CH ₃	8.54±0.01	EI	3803
	C ₄ H ₅ OS(CH ₃) ₃ (1,3-Oxathiane, 2,4,6-trimethyl-, (2 α ,4 α ,6 β -))	22425-91-8	CH ₃	8.67	EI	3803
	C ₄ H ₅ OS(CH ₃) ₃ (1,3-Oxathiane, 2,4,6-trimethyl-, (2 α ,4 β ,6 α -))	22425-90-7	CH ₃	8.64	EI	3803
C₆H₁₂OS⁺						
	C ₄ H ₆ OS(CH ₃) ₂ (1,3-Oxathiane, 4,6-dimethyl-, <i>cis</i> -)	22452-25-1	**	8.75	EI	3803
	C ₄ H ₆ OS(CH ₃) ₂ (1,3-Oxathiane, 4,6-dimethyl-, <i>trans</i> -)	22452-26-2	**	8.67±0.01	EI	3803
C₆H₁₄OS⁺						
	(<i>n</i> -C ₃ H ₇) ₂ SO	4253-91-2	**	8.60 (V)	PE	4295
	(<i>iso</i> -C ₃ H ₇) ₂ SO	2211-89-4	**	8.46 (V)	PE	4295
			**	8.46 (V)	PE	3646
			**	8.54±0.08	EI	5311

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_6\text{H}_{14}\text{OS}^+$	$(\text{iso-C}_5\text{H}_{11})\text{SOCH}_3$	55860-10-1	**	8.55 ± 0.05	EI	5311
$\text{C}_7\text{H}_8\text{OS}^+$	$\text{C}_6\text{H}_5\text{S}(\text{CH}_3)\text{O}$ (Benzene, (methylsulfinyl)-)	1193-82-4	**	8.79 (V)	PE	4295
$\text{C}_7\text{H}_{10}\text{OS}^+$	$\text{C}_4\text{HS}(\text{CH}_3)_2(\text{OCH}_3)$ (Thiophene, 2-methoxy-3,5-dimethyl-) $\text{C}_4\text{HO}(\text{CH}_3)_2\text{SCH}_3$ (Furan, 2,5-dimethyl-3-(methylthiol)-) $\text{C}_4\text{HS}(\text{CH}_3)_2\text{OCH}_3$ (Thiophene, 3-methoxy-2,5-dimethyl-) $\text{C}_4\text{H}_2\text{S}(=\text{O})(\text{CH}_3)\text{C}_2\text{H}_5$ (3(2H)-Thiophenone, 2-ethyl-5-methyl-) $\text{C}_4\text{H}_2\text{S}(=\text{O})(\text{CH}_3)\text{C}_2\text{H}_5$ (3(2H)-Thiophenone, 5-ethyl-2-methyl-) $\text{C}_4\text{HS}(=\text{O})(\text{CH}_3)_3$ (2(3H)-Thiophenone, 3,3,5-trimethyl-) $\text{C}_4\text{HS}(=\text{O})(\text{CH}_3)_3$ (3(2H)-Thiophenone, 2,2,5-trimethyl-)	57556-17-9 63359-68-7 57556-08-8 57556-06-6 57556-03-3 33687-83-1 57556-09-9	**	7.78 ± 0.05 7.91 ± 0.05 7.89 ± 0.05 8.22 ± 0.05 8.34 ± 0.05 8.53 ± 0.05 8.49 ± 0.05	EI	4666 4706 4673 4673 4673 4666 4673
$\text{C}_7\text{H}_{13}\text{OS}^+$	$\text{C}_4\text{H}_4\text{OS}(\text{CH}_3)_4$ (1,3-Oxathiane, 2,2,4,6-tetramethyl-, cis-) $\text{C}_4\text{H}_4\text{OS}(\text{CH}_3)_4$ (1,3-Oxathiane, 2,2,4,6-tetramethyl, trans-)	34560-79-7 34560-78-6	CH_3	8.63 ± 0.01 8.54 ± 0.01	EI	3803 3803
$\text{C}_7\text{H}_{14}\text{OS}^+$	$\text{C}_4\text{H}_5\text{OS}(\text{CH}_3)_3$ (1,3-Oxathiane, 2,4,6-trimethyl-, (2 α ,4 α ,6 α -)) $\text{C}_4\text{H}_5\text{OS}(\text{CH}_3)_3$ (1,3-Oxathiane, 2,4,6-trimethyl-, (2 α ,4 α ,6 β -)) $\text{C}_4\text{H}_5\text{OS}(\text{CH}_3)_3$ (1,3-Oxathiane, 2,4,6-trimethyl-, (2 α ,4 β ,6 α -))	22521-88-6 22425-91-8 22425-90-7	**	8.55 8.54 8.58	EI	3803 3803 3803
$\text{C}_8\text{H}_{10}\text{OS}^+$	$\text{C}_6\text{H}_4(\text{SCH}_3)(\text{OCH}_3)$ (Benzene, 1-methoxy-2-(methylthio)-) $\text{C}_6\text{H}_4(\text{SCH}_3)(\text{OCH}_3)$ (Benzene, 1-methoxy-4-(methylthio)-)	2388-73-0 1879-16-9	** **	8.05 (V) 7.80 (V) 7.80 ± 0.01 (V)	PE	5403 5403 4389
$\text{C}_8\text{H}_{12}\text{OS}^+$	$\text{C}_4\text{H}_3\text{S}(=\text{O})(\text{tert-C}_4\text{H}_9)$ (2(5H)-Thiophenone, 3-(2,2-dimethylethyl)-) $\text{C}_8\text{H}_{12}\text{OS}$	XXXXX-XX-X 57556-04-4	** **	9.25 ± 0.05 8.21 ± 0.05	EI	4666 4673
$\text{C}_8\text{H}_{16}\text{OS}^+$	$\text{C}_4\text{H}_4\text{OS}(\text{CH}_3)_4$ (1,3-Oxathiane, 2,2,4,6-tetramethyl-, cis-) $\text{C}_4\text{H}_4\text{OS}(\text{CH}_3)_4$ (1,3-Oxathiane, 2,2,4,6-tetramethyl, trans-)	34560-79-7 34560-78-6	** **	8.48 ± 0.02 8.45 ± 0.01	EI	3803 3803
$\text{C}_8\text{H}_{18}\text{OS}^+$	$((\text{CH}_3)_3\text{C})_2\text{SO}$	2211-92-9	** **	8.18 (V) 8.18 (V)	PE PE	3646 4295
$\text{C}_9\text{H}_6\text{OS}^+$	$\text{C}_6\text{H}_4\text{C}_3\text{H}_2\text{S}(=\text{O})$ (4H-1-Benzothiopyran-4-one)	491-39-4	**	8.68 (V)	PE	5491

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₉H₈OS⁺	C ₆ H ₄ C ₃ H ₄ S(=O) (4H-1-Benzothiopyran-4-one,2,3-dihydro-)	3528-17-4	**	8.53 (V)	PE	5491
C₉H₁₄OS⁺	C ₆ H ₁₄ OS (Thiophene, 3-(1,1-dimethylethyl)-2-methoxy-) C ₆ H ₁₄ OS (2(3H)-Thiophenone, 3-(1,1-dimethylethyl)-3-methyl-) C ₆ H ₁₄ OS (2(5H)-Thiophenone, 3-(1,1-dimethylethyl)-5-methyl-) C ₆ H ₁₄ OS (3(2H)-Thiophenone, 2-(1,1-dimethylethyl)-5-methyl-) C ₆ H ₁₄ OS (3(2H)-Thiophenone, 5-(1,1-dimethylethyl)-2-methyl-)	57556-16-8 57556-18-0 57556-15-7 57556-07-7 57556-05-5	**	7.67±0.05 8.38±0.05 9.07±0.05 8.09±0.05 8.10±0.05	EI	4666 4666 4666 4673 4673
C₁₁H₈OS⁺	C ₆ H ₅ COC ₄ H ₃ S (Methanone, phenyl-2-thienyl-)	135-00-2	**	9.2±0.1	EI	5493
C₁₂H₈OS⁺	C ₁₂ H ₈ SO (Dibenzothiophene, 5-oxide) C ₁₂ H ₈ OS (Phenoxathiin)	1013-23-6 262-20-4	**	8.43 (V) 7.72±0.05 (V)	PE	4295 4743
C₁₂H₁₀OS⁺	(C ₆ H ₅) ₂ SO	945-51-7	** ** **	9.02±0.05 8.58 (V) 9.02	EI PE EI	3498 4295 5292
C₁₂H₁₂OS⁺	C ₆ H ₄ (C ₂ H ₄ SO)C ₄ H ₄ (4α,8α-(Methanothiomethano)naphthalene-10-oxide)	64776-55-2	**	8.44 (V)	PE	5194
C₁₂H₁₆OS⁺	C ₆ H ₄ (C ₂ H ₄ SO)C ₄ H ₈ (4α,8α-(Methanothiomethano)naphthalene,1,2,3,4-tetrahydro-10-oxide) C ₆ H ₆ (C ₂ H ₄ SO)C ₄ H ₆ (4α,8α-(Methanothiomethano)naphthalene,1,4,5,8-tetrahydro-10-oxide)	71656-72-9 17853-53-1	** **	8.71 (V) 8.52 (V)	PE	5194 5194
C₁₂H₁₈OS⁺	C ₆ H ₆ (C ₂ H ₄ SO)C ₄ H ₈ (4α,8α-(Methanothiomethano)naphthalene,1,2,3,4,5,8-hexahydro-10-oxide)	71600-20-9	**	8.50 (V)	PE	5194
C₁₃H₈OS⁺	C ₁₃ H ₇ (=O)SH (Phenalen-1-one,9-mercaptopo-)	XXXXX-XX-X	**	7.76±0.04 (V)	PE	5595
C₁₃H₁₂OS⁺	C ₆ H ₄ (OCH ₃)SC ₆ H ₅ (Benzene, 1-methoxy-3-(phenylthio-)) C ₆ H ₄ (OCH ₃)SC ₆ H ₅ (Benzene, 1-methoxy-4-(phenylthio-))	30723-54-7 5633-57-8	** **	8.02 7.89	CTS	4272 4272
C₁₄H₉OS⁺	(C ₆ H ₅) ₂ CH ₂ SC(=O) (Dibenzo[<i>b,e</i>]thiepin-11(6H)-one)	1531-77-7	H	10.4	EI	5340
C₁₄H₁₀OS⁺	(C ₆ H ₅) ₂ CH ₂ SC(=O) (Dibenzo[<i>b,e</i>]thiepin-11(6H)-one)	1531-77-7	**	9.21	EI	5340

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{14}\text{H}_{10}\text{OS}^+$	$\text{C}_2\text{S}(=\text{O})(\text{C}_6\text{H}_5)_2$ (Thiirene, diphenyl-1,1-oxide)	31247-21-9	**	10.86 (V)	PE	4856
$\text{C}_{14}\text{H}_{11}\text{OS}^+$	$\text{C}_{14}\text{H}_{12}\text{SO}_2$ (Dibenzo[<i>b,e</i>]thiepin, 6,11-dihydro-5,5-dioxide-)	23772-26-1	OH	10.30	EI	5414
$\text{C}_{14}\text{H}_{14}\text{OS}^+$	$\text{C}_6\text{H}_5\text{O}(\text{CH}_2)_2\text{SC}_6\text{H}_5$ (Benzene,[2-phenoxyethyl]thio)-)	17414-04-9	**	8.20±0.05	EI	5484
$\text{C}_{15}\text{H}_{16}\text{OS}^+$	$\text{C}_6\text{H}_5\text{O}(\text{CH}_2)_3\text{SC}_6\text{H}_5$ (Benzene,[3-phenoxypropyl]thio)-)	59950-10-6	**	8.21±0.05	EI	5484
$\text{C}_{16}\text{H}_{18}\text{OS}^+$	$\text{C}_6\text{H}_5\text{O}(\text{CH}_2)_4\text{SC}_6\text{H}_5$ (Benzene,[4-phenoxybutyl]thio)-)	59950-11-7	**	8.25±0.05	EI	5484
$\text{C}_{17}\text{H}_{20}\text{OS}^+$	$\text{C}_6\text{H}_5\text{O}(\text{CH}_2)_5\text{SC}_6\text{H}_5$ (Benzene,[5-phenoxypentyl]thio)-)	59950-12-8	**	8.27±0.05	EI	5484
$\text{C}_{18}\text{H}_{22}\text{OS}^+$	$\text{C}_6\text{H}_5\text{O}(\text{CH}_2)_6\text{SC}_6\text{H}_5$ (Benzene,[6-phenoxyhexyl]thio)-)	59950-13-9	**	8.26±0.05	EI	5484
$\text{C}_2\text{H}_4\text{O}_2\text{S}^+$	$\text{C}_2\text{H}_4\text{SO}_2$ (Thiirane 1,1-dioxide)	1782-89-4	**	10.20 (V)	PE	4827
$\text{C}_2\text{H}_6\text{O}_2\text{S}^+$	$(\text{CH}_3)_2\text{SO}_2$	67-71-0	** ** ** **	10.65 (V) 10.65 (V) 10.80 (V) 10.97 (V)	PE PE PE PE	4827 5207 3993 3705
$\text{C}_3\text{H}_2\text{O}_2\text{S}^+$	$\text{C}_3\text{H}_2\text{O}_2(=\text{S})$ (1,3-Dioxole-2-thione)	37635-87-3	**	9.05 (V)	PE	4549
$\text{C}_3\text{H}_4\text{O}_2\text{S}^+$	$\text{C}_2\text{HS}(\text{O}_2)(\text{CH}_3)$ (Thiirene, methyl-1,1-dioxide-)	14491-01-1	**	10.40 (V)	PE	4508
	$\text{C}_3\text{H}_4\text{O}_2(=\text{S})$ (1,3-Dioxolane-2-thione)	20628-59-5	**	8.88 (V)	PE	4549
$\text{C}_3\text{H}_6\text{O}_2\text{S}^+$	$(\text{CH}_3\text{O})_2\text{CS}$ $(\text{CH}_3)(\text{CH}_2=\text{CH})\text{SO}_2$	1115-13-5 3680-02-2	** ** **	8.99 (V) 10.65 (V) 10.82 (V)	PE PE PE	4323 4827 3993
$\text{C}_4\text{H}_6\text{O}_2\text{S}^+$	$(\text{CH}_2=\text{CH})_2\text{SO}_2$	77-77-0	** **	10.56 (V) 10.62 (V)	PE PE	4827 3993
	$\text{C}_2\text{S}(\text{O}_2)(\text{CH}_3)_2$ (Thiirene, dimethyl-1,1-dioxide-)	30646-57-2	**	9.89 (V)	PE	4508
	$\text{C}_4\text{H}_6\text{SO}_2$ (Thiophene, 2,5-dihydro-1,1-dioxide)	77-79-2	** **	10.44 (V) 10.44 (V)	PE PE	4827 5207

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₄H₈O₂S⁺						
	C ₂ H ₂ S(O ₂)(CH ₃) ₂ (Thiirane, 2,3-dimethyl-1,1-dioxide, <i>cis</i> -)	54697-52-8	**	9.82 (V)	PE	4508
	C ₄ H ₈ SO ₂ (Thiophene, tetrahydro-1,1-dioxide)	126-33-0	**	9.91±0.07	PI	5040
			**	10.24 (V)	PE	4324
C₄H₁₀O₂S⁺						
	(C ₂ H ₃) ₂ SO ₂	597-35-3	**	9.96±0.03	PI	5040
C₅H₄O₂S⁺						
	C ₅ H ₄ SO ₂ ([1,2]Oxathiolo[2,3- <i>b</i>][1,2]oxathiole-7-S ^{IV})	40159-76-0	**	8.58 (V)	PE	4406
	C ₄ H ₃ SCOOH (2-Thiophenecarboxylic acid)	527-72-0	**	9.14±0.05 (V)	PE	4626
			**	9.35	EI	3804
C₅H₆O₂S⁺						
	C ₃ O ₂ (=S)(CH ₃) ₂ (1,3-Dioxole-2-thione, 4,5-dimethyl-)	37528-00-0	**	8.4 (V)	PE	4549
C₆H₆O₂S⁺						
	C ₆ H ₃ SCOOCH ₃ (2-Thiophenecarboxylic acid methyl ester)	5380-42-7	**	8.98±0.05 (V)	PE	4626
			**	9.22±0.05	EI	3482
C₇H₈O₂S⁺						
	(C ₆ H ₅)(CH ₃)SO ₂ (Benzene, (methylsulfonyl)-)	3112-85-4	**	9.74 (V)	PE	4827
C₈H₄O₂S⁺						
	C ₈ H ₄ S(=O) ₂ (Benzo[<i>b</i>]thiophene-2,3-dione)	493-57-2	**	9.14±0.05 (V)	PE	4708
	C ₈ H ₄ S(=O) ₂ (Benzo[<i>c</i>]thiophene-1,3-dione)	5698-59-9	**	9.85±0.05 (V)	PE	4708
C₈H₁₀O₂S⁺						
	(<i>iso</i> -C ₄ H ₉) ₂ SO ₂	10495-45-1	**	9.54±0.05	PI	5040
C₉H₆O₂S⁺						
	C ₆ H ₄ C ₃ H ₂ S(=O) ₂ 5491 (4H-1-Benzothiopyran-4-one-1-oxide)	37682-92-1	**	9.24 (V)	PE	5491
C₉H₁₂O₂S⁺						
	(C ₆ H ₅)(n-C ₃ H ₇)SO ₂ (Benzene,(propylsulfonyl)-)	13596-75-3	**	9.21±0.03	PI	5040
C₁₀H₁₆O₂S⁺						
	C ₆ H ₄ S(=O) ₂ (CH ₃) ₄ (1,2-Cycloheptanedione-7-thia,3,3,7,7-tetramethyl-)	XXXXXX-XX-X	**	8.75 (V)	PE	5090
C₁₁H₁₀O₂S⁺						
	C ₆ H ₄ C ₃ O ₂ S(CH ₃) ₂ (Sulfonium,dimethyl-2,3-dihydro-1,3-dioxo-1H-inden-2-ylide)	5508-42-9	**	8.05	CTS	5592
C₁₂H₈O₂S⁺						
	C ₁₂ H ₈ SO ₂ (Dibenzothiophene 5,5-dioxide)	1016-05-3	**	8.90 (V)	PE	4827
			**	9.28	EI	4228

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{12}\text{H}_{10}\text{O}_2\text{S}^+$	$(\text{C}_6\text{H}_5)_2\text{SO}_2$ (Benzene, 1,1'-sulfonylbis-)	127-63-9	**	9.16±0.03	PI	5040
			**	9.37 (V)	PE	4827
			**	9.7	EI	4228
$\text{C}_{12}\text{H}_{12}\text{O}_2\text{S}^+$	$\text{C}_6\text{H}_4(\text{C}_2\text{H}_4\text{SO}_2)\text{C}_4\text{H}_4$ (4 α ,8 α -(Methanothiomethano)naphthalene-10,10-dioxide)	23695-63-8	**	8.7 (V)	PE	5194
$\text{C}_{12}\text{H}_{16}\text{O}_2\text{S}^+$	$\text{C}_6\text{H}_6(\text{C}_2\text{H}_4\text{SO}_2)\text{C}_4\text{H}_6$ (4 α ,8 α -(Methanothiomethano)naphthalene,1,4,5,8-tetrahydro-10,10-dioxide)	17853-54-2	**	9.2 (V)	PE	5194
$\text{C}_{12}\text{H}_{20}\text{O}_2\text{S}^+$	$\text{C}_4\text{H}_2\text{S}(\text{C}_4\text{H}_9)_2\text{O}_2$ (Thiophene, 2,5-bis(1,1-dimethylethyl)- 1,1-dioxide)	6407-02-9	**	8.64 (V)	PE	4324
$\text{C}_{14}\text{H}_9\text{O}_2\text{S}^+$	$\text{C}_6\text{H}_4(\text{COSC}_6\text{H}_5)_2$ (1,2-Benzenedicarbothioic acid S,S-diphenyl ester)	42797-33-1	$\text{C}_6\text{H}_5\text{S}$	10.3±0.2	EI	4062
	$\text{C}_6\text{H}_4\text{O}(=\text{O})(\text{SC}_6\text{H}_5)_2$ (1(3 H)-Isobenzofuranone, 3,3-bis(phenylthio)-)	4792-31-8	$\text{C}_6\text{H}_5\text{S}$	10.3±0.2	EI	4062
	$\text{C}_{14}\text{H}_{12}\text{O}_2\text{S}^+$	23772-26-1	**	9.85	EI	5414
$\text{C}_{14}\text{H}_{14}\text{O}_2\text{S}^+$	$(\text{C}_6\text{H}_5\text{CH}_3)_2\text{SO}_2$ (Benzene, 1,1'-sulfonylbis[4-methyl-])	599-66-6	**	8.66±0.04	PI	5040
$\text{C}_{15}\text{H}_{11}\text{O}_2\text{S}^+$	$\text{C}_6\text{H}_4(\text{COSC}_6\text{H}_4\text{CH}_3)_2$ (1,2-Benzenedicarbothioic acid S,S-bis(4-methylphenyl)ester)	42797-34-2	$\text{C}_6\text{H}_4(\text{S})\text{CH}_3$	10.1±0.2	EI	4062
	$\text{C}_6\text{H}_4\text{O}(=\text{O})(\text{SC}_6\text{H}_4\text{CH}_3)_2$ (1(3 H)-Isobenzofuranone, 3,3-bis[(4-methylphenyl)thio]-)	42797-36-4	$\text{C}_6\text{H}_4(\text{S})\text{CH}_3$	9.9±0.2	EI	4062
	$\text{C}_2\text{H}_4\text{O}_3\text{S}^+$	3741-38-6	** ** ** **	10.30 10.93 (V) 10.93 (V) 10.30±0.05	EI PE PE EI	5292 3646 4295 3498
$\text{C}_2\text{H}_6\text{O}_3\text{S}^+$	$(\text{CH}_3\text{O})_2\text{SO}$	616-42-2	** **	10.25 (V) 10.25 (V)	PE PE	3646 4295
$\text{C}_6\text{H}_4\text{O}_3\text{S}^+$	$\text{C}_6\text{H}_4\text{O}_3\text{S}$ (1,3,2-Benzodioxathiole-2-oxide)	6255-58-9	**	9.1 (V)	PE	4616
$\text{C}_9\text{H}_6\text{O}_3\text{S}^+$	$\text{C}_6\text{H}_4\text{C}_2\text{H}_2\text{S}(=\text{O})_3$ (4H-1-Benzothiopyran-4-one-1,1-dioxide)	22810-27-1	**	9.93 (V)	PE	5491
$\text{C}_{14}\text{H}_{10}\text{O}_3\text{S}^+$	$\text{C}_{14}\text{H}_{10}\text{SO}_3$ (Dibenzo[<i>b,e</i>]thiepin-11(6H)-one-5,5-dioxide)	33301-21-2	**	9.70	EI	5414
$\text{C}_{16}\text{H}_{24}\text{O}_4\text{S}^+$	$\text{C}_6\text{O}_3\text{S}(\text{O})(\text{CH}_3)_4$ (Spirofuran-3(2H),2'-furo[3,4- <i>d</i>][1,3]oxathiol]-4(5H)-one, 4',6'-dihydro-2,2,4',4',5,5,6',6'-octamethyl-)	54196-16-6	**	7.48±0.03 (V)	PE	4292

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{16}\text{H}_{26}\text{O}_4\text{S}^+$	$\text{C}_8\text{HO}_3\text{S}(\text{OH})(\text{CH}_3)_8$ (Spiro[furan-3(2 <i>H</i>),2'-furo[3,4- <i>d</i>][1,3]oxathiol]-4-ol, 4,4',5,6'-tetrahydro-2,2,4',4',5,5, 6',6',-octamethyl-)	54739-35-4	**	7.47±0.03 (V)	PE	4292
$\text{C}_3\text{H}_4\text{OS}_2^+$	$\text{C}_3\text{H}_4\text{S}_2\text{O}$ (1,3-Dithiolan-2-one)	2080-58-2	**	9.50 (V)	PE	4407
			**	9.58 (V)	PE	4549
$\text{C}_3\text{H}_6\text{OS}_2^+$	$\text{CH}_3\text{SCSOCH}_3$	19708-81-7	**	8.71 (V)	PE	4323
$\text{C}_7\text{H}_8\text{OS}_2^+$	$\text{C}_5\text{H}_8\text{S}_2\text{O}(\text{CH}_3)_2$ (2-Propanone, 1-(5-methyl-3 <i>H</i> -1,2-dithiol-3-ylidene)-)	1005-55-6	**	7.68 (V)	PE	4406
$\text{C}_{10}\text{H}_8\text{OS}_2^+$	$(\text{C}_4\text{H}_2\text{S})_2\text{C}_2\text{H}_4\text{O}$ (4 <i>H</i> ,6 <i>H</i> -Dithieno[3,4- <i>c</i> :3',4'- <i>e</i>]oxepin) $(\text{C}_4\text{H}_2\text{S})_2\text{C}_2\text{H}_4\text{O}$ (Dithieno[2,3- <i>c</i> :3',2'- <i>e</i>]oxepin,4,6-dihydro-)	23062-34-2 63286-53-3	**	8.3 (V) 8.15 (V)	PE PE	5422 5422
$\text{C}_{14}\text{H}_{16}\text{OS}_2^+$	$(\text{C}_4\text{H}_2\text{S})_2\text{C}_2(\text{CH}_3)_4\text{O}$ (Dithieno[2,3- <i>c</i> :3',2'- <i>e</i>]oxepin,4,6-dihydro-4,4,6,6-tetramethyl-) $(\text{C}_4\text{H}_2\text{S})_2\text{C}_2(\text{CH}_3)_4\text{O}$ (4 <i>H</i> ,6 <i>H</i> -Dithieno[3,4- <i>c</i> :3',4'- <i>e</i>]oxepin,4,4,6,6-tetramethyl-)	64504-71-8 64504-72-9	**	7.8 (V) 8.0 (V)	PE PE	5422 5422
$\text{C}_6\text{H}_6\text{O}_2\text{S}_2^+$	$\text{C}_4(=\text{O})_2(\text{CH}_3\text{S})_2$ (3-Cyclobutene-1,2-dione, 3,4-bis(methylthio)-)	54131-97-4	**	8.18 (V)	PE	4861
$\text{C}_8\text{H}_{14}\text{O}_2\text{S}_2^+$	$\text{C}_3\text{H}_5\text{S}_2(\text{CH}_3)_4\text{COOH}$ (1,2-Dithiolane-3-pentanoic acid)	62-46-4	**	8.02 (V)	PE	4410
$\text{C}_{10}\text{H}_8\text{OS}_3^+$	$(\text{C}_3\text{HS}_2)=\text{S}(\text{C}_6\text{H}_4\text{OCH}_3)$ (3 <i>H</i> -1,2-Dithiole-3-thione, 5-(4-methoxyphenyl)-)	532-11-6	**	8.11 (V)	PE	4403
$\text{C}_8\text{H}_4\text{O}_2\text{S}_3^+$	$\text{C}_8\text{H}_4\text{S}_2(\text{SO}_2)$ (Dithieno[2,3- <i>b</i> :3',2'- <i>d</i>]thiophene,7,7-dioxide-) $\text{C}_8\text{H}_4\text{S}_2(\text{SO}_2)$ (Dithieno[3,2- <i>b</i> :2',3'- <i>d</i>]thiophene,4,4-dioxide-) $\text{C}_8\text{H}_4\text{S}_2(\text{SO}_2)$ (Dithieno[3,4- <i>b</i> :3',4'- <i>d</i>]thiophene,4,4-dioxide-)	28504-86-1 3807-53-2 28504-85-0	** ** **	8.5 (V) 8.4 (V) 8.7 (V)	PE PE PE	5405 5405 5405
$\text{BC}_{13}\text{H}_{21}\text{OS}^+$	$\text{C}_{13}\text{H}_{21}\text{BOS}$ (Borinic acid, dipropylthio-4-methoxyphenyl ester)	64503-45-3	**	8.17±0.05 (V)	PE	4848
HNOS^+	$\text{HN}=\text{S}=\text{O}$	13817-04-4	**	11.60 (V)	PE	5386
$\text{C}_2\text{H}_7\text{NOS}^+$	$(\text{CH}_3)_2\text{S}(\text{NH})\text{O}$	1520-31-6	** **	9.5 (V) 9.50 (V)	PE PE	5207 4827
$\text{C}_3\text{H}_5\text{NOS}^+$	$\text{C}_3\text{H}_5\text{NO}=\text{S}$ (2-Oxazolidinethione)	5840-81-3	**	8.37±0.03 (V)	PE	4253

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₄H₃NOS⁺	C ₃ H ₂ NS(CHO) (5-Isothiazolecarboxaldehyde)	5242-57-9	**	10.25	EI	3587
C₄H₉NOS⁺	(CH ₃) ₃ CNSO	38662-39-	**	10.54 (V)	PE	4024
C₅H₇NOS⁺	C ₃ HNO(=S)(CH ₃) ₂ (2(3H)-Oxazolethione, 4,5-dimethyl-)	6670-14-0	**	7.74±0.03 (V)	PE	4253
C₆H₇NOS⁺	C ₅ H ₂ NH(=S)(OH)CH ₃ (2(1H)-Pyridinethione, 3-hydroxy-6-methyl-) C ₅ H ₃ N(OH)SCH ₃ (3-Pyridinol, 2-(methylthio)-)	22989-67-9 32637-37-9	** **	8.04±0.05 8.53±0.05	EI EI	3635 3977
C₆H₉NOS⁺	C ₃ NO(S)(CH ₃) ₃ (2(3H)-Oxazolethione, 3,4,5-trimethyl-)	25444-93-3	** **	7.51 7.54±0.03 (V)	PE PE	4555 4253
C₆H₁₁NOS⁺	C ₆ H ₁₁ NSO (Cyclohexanamine, N-sulfinyl-)	30980-11-1	**	10.0 (V)	PE	4024
C₇H₅NOS⁺	C ₇ H ₅ NO(S) (2(3H)-Benzoxazolethione) C ₇ H ₅ NS(O) (Thiazolo[3,2- <i>a</i>]pyridinium, 3-hydroxy-, hydroxide, inner salt)	2382-96-9 42715-25-3	** **	8.14 6.92±0.05	PE EI	4555 3977
C₇H₇NOS⁺	C ₅ H ₃ N(O)SC ₂ H ₄ (Thiazolo[3,2- <i>a</i>]pyridinium,2,3-dihydro-8-hydroxy-hydroxide, inner salt) C ₅ H ₃ N(=O)SC ₂ H ₄ (5H-Thiazolo[3,2- <i>a</i>]pyridin-5-one,2,3-dihydro-)	23003-45-4 66201-75-0	** **	7.12±0.05 7.91±0.05	EI EI	5416 5416
C₇H₉NOS⁺	C ₅ H ₂ N(OH)(CH ₃)SCH ₃ (3-Pyridinol, 6-methyl-2-(methylthio)-) C ₄ H ₃ SCON(CH ₃) ₂ (2-Thiophenecarboxamide, N,N-dimethyl-)	23003-25-0 30717-57-8	** **	8.24±0.05 8.84±0.05 (V)	EI PE	3635 4626
C₈H₇NOS⁺	C ₇ H ₄ NO(S)(CH ₃) (2(3H)-Benzoxazolethione, 3-methyl-) C ₅ H ₃ N(O)SC ₂ H(CH ₃) (Thiazolo[3,2- <i>a</i>]pyridinium,8-hydroxy-3-methyl- hydroxide, inner salt) C ₇ H ₄ NS(O)CH ₃ (Thiazolo[3,2- <i>a</i>]pyridinium, 3-hydroxy-2-methyl-, hydroxide, inner salt) C ₇ H ₄ NS(O)CH ₃ (Thiazolo[3,2- <i>a</i>]pyridinium, 8-hydroxy-5-methyl-, hydroxide, inner salt) C ₅ H ₃ N(=O)SC ₂ H(CH ₃) (5H-Thiazolo[3,2- <i>a</i>]pyridin-5-one,3-methyl-)	13673-63-7 30276-99-4 35143-56-7 30277-17-9 71310-14-0	** ** ** ** **	7.94 7.12±0.05 6.82±0.05 7.03±0.05 7.44	PE EI EI EI EI	4555 5416 3977 3635 5416
C₈H₉NOS⁺	C ₇ H ₆ NOS(CH ₃) (1,4-Oxathiino[3,2- <i>b</i>]pyridine, 2,3-dihydro-6-methyl-) C ₅ H ₂ O(=O)(CH ₃)NC ₂ H ₄ S (Cyclopent[2,3]azirino[2,1- <i>b</i>]thiazol-7(4aH)-one, 2,3-dihydro-4a-methyl-)	35688-70-1 71310-16-2	** **	8.03±0.05 7.93	EI EI	3635 5416

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₈H₉NOS⁺	C ₅ H ₂ N(=S)(OH)(CH ₃)C ₂ H ₃ (2(1 <i>H</i>)-Pyridinethione, 1-ethenyl-3-hydroxy-6-methyl-) C ₅ H ₂ N(O)(CH ₃)SC ₂ H ₄ (Thiazolo[3,2- <i>a</i>]pyridinium,2,3-dihydro-8-hydroxy-5-methyl- hydroxide, inner salt) C ₅ H ₂ N(=O)(CH ₃)SC ₂ H ₄ (5H-Thiazolo[3,2- <i>a</i>]pyridin-5-one,2,3-dihydro-8-methyl-)	35688-69-8 23003-43-2 71310-13-9	** ** **	7.73±0.05 6.95±0.05 7.35±0.05 7.69±0.05	EI EI EI EI	3635 5416 3635 5416
C₈H₁₁NOS⁺	C ₅ H ₂ N(=S)(OH)(CH ₃)C ₂ H ₅ (2(1 <i>H</i>)-Pyridinethione, 1-ethyl-3-hydroxy-6-methyl-)	24207-15-6	**	7.75±0.05	EI	3635
C₉H₉NOS⁺	C ₅ H ₂ N(O)(CH ₃)SC ₂ H(CH ₃) (Thiazolo[3,2- <i>a</i>]pyridinium,8-hydroxy-3,5-dimethyl- hydroxide, inner salt) C ₅ H ₂ N(=O)(CH ₃)SC ₂ H(CH ₃) (5H-Thiazolo[3,2- <i>a</i>]pyridin-5-one,3,8-dimethyl-)	30277-00-0 71310-15-1	** **	6.84±0.05 7.32	EI EI	5416 5416
C₁₁H₁₁NOS⁺	C ₁₁ H ₁₁ NOS (Carbamothioic acid, 1,3-butadienyl-S-phenyl ester, (E)-)	61759-58-8	**	~8.18 (V)	PE	4803
C₁₃H₉NOS⁺	C ₇ H ₄ NS(O)C ₆ H ₅ (Thiazolo[3,2- <i>a</i>]pyridinium, 3-hydroxy-2-phenyl-, hydroxide, inner salt)	32044-03-4	**	6.70±0.05	EI	3977
C₄H₁₀N₂OS⁺	C ₂ H ₄ N ₂ S(O)(CH ₃) ₂ (1,2,5-Thiadiazolidine, 2,5-dimethyl-, 1-oxide)	15108-72-2	**	8.2 (V)	PE	4295
C₄H₁₂N₂OS⁺	((CH ₃) ₂ N) ₂ SO	3768-60-3	** **	8.53 (V) 8.53 (V)	PE PE	3646 4295
C₇H₈N₂OS⁺	C ₆ H ₄ (OH)NHCSNH ₂ (Thiourea, (2-hydroxyphenyl)-)	1520-26-9	**	8.20	EI	4834
C₈H₁₀N₂OS⁺	C ₆ H ₄ (OCH ₃)NHCSNH ₂ (Thiourea, (2-methoxyphenyl)-)	1516-37-6	**	7.80	EI	4834
C₁₅H₂₀N₂OS⁺	C ₁₅ H ₂₀ N ₂ OS (Carbamothioic acid,phenyl-0-(8-methyl-8-azabicyclo[3.2.1]oct-3-yl)ester, <i>endo</i> -) C ₁₅ H ₂₀ N ₂ OS (Carbamothioic acid,phenyl-0-(8-methyl-8-azabicyclo[3.2.1]oct-3-yl)ester, <i>exo</i> -)	67139-54-2 67139-55-3	** **	8±0.3 8±0.3	EI EI	5401 5401
C₁₇H₁₈N₂OS⁺	C ₁₂ H ₈ NSCOCH ₂ CH ₂ N(CH ₃) ₂ (10 <i>H</i> -Phenothiazine, 10-[3-(dimethylamino)-1-oxopropyl]-)	3576-44-1	**	8.26±0.07	CTS	4079
C₁₈H₂₂N₂OS⁺	C ₁₈ H ₂₂ N ₂ OS (10 <i>H</i> -Phenothiazine-10-ethanamine, 2-methoxy-N,N, <i>α</i> -trimethyl-)	7624-74-0	**	8.18±0.07	CTS	4079
C₁₉H₂₂N₂OS⁺	C ₁₂ H ₈ NSCOCH ₂ CH ₂ N(C ₂ H ₅) ₂ (10 <i>H</i> -Phenothiazine, 10-[3-(diethylamino)-1-oxopropyl]-)	3576-47-4	**	7.85±0.07	CTS	4079

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₂₀H₂₄N₂OS⁺	C ₁₂ H ₈ NSCO(CH ₂) ₃ N(C ₂ H ₅) ₂ (10 <i>H</i> -Phenothiazine, 10-[4-(diethylamino)-1-oxobutyl]-)	51307-45-0	**	7.88±0.07	CTS	4079
C₆H₅N₃OS⁺	C ₅ H ₂ N ₃ SOCH ₃ (1[2,3]Thiadiazolo[5,4- <i>b</i>]pyridine, 5-methoxy-)	54459-90-4	**	9.01±0.05	EI	4316
C₁₉H₂₃N₃OS⁺	C ₁₂ H ₇ NS(CH ₃)NHCOC ₂ H ₅ N(C ₂ H ₅) ₂ (Acetamide, 2-(diethylamino)- <i>N</i> -(10-methyl-10 <i>H</i> -phenothiazin-3-yl)-)	1952-62-1	**	7.13±0.07	CTS	4079
C₂₂H₂₇N₃OS⁺	C ₂₂ H ₂₇ N ₃ OS (Ethanone, 1-[10-[3-(4-methyl-1-piperazinyl)propyl]-10 <i>H</i> -phenothiazin-2-yl]-)	1053-74-3	**	9.05±0.07	CTS	4079
C₂₃H₂₉N₃OS⁺	C ₂₃ H ₂₉ N ₃ OS (1-Propanone, 1-[10-[3-(4-methyl-1-piperazinyl)propyl]-10 <i>H</i> -phenothiazin-2-yl]-)	20686-45-7	**	9.08±0.07	CTS	4079
C₃H₇NO₂S⁺	SHCH ₂ CH(NH ₂)COOH	3374-22-9	**	~9	PI	3766
C₄H₃NO₂S⁺	C ₄ H ₃ SNO ₂ (Thiophene, 2-nitro-)	609-40-5	**	9.73±0.05 (V)	PE	4626
			**	9.77±0.05	EI	3482
C₅H₁₁NO₂S⁺	CH ₃ SCH ₂ CH ₂ CH(NH ₂)COOH	59-51-8	**	~9	PI	3766
C₇H₃NO₂S⁺	C ₇ H ₃ NS(=O) ₂ (Thieno[3,4- <i>b</i>]pyridine-5,7-dione)	69094-37-7	**	10.05±0.05 (V)	PE	4889
C₇H₅NO₂S⁺	C ₇ H ₄ NS(O)OH (Thiazolo[3,2- <i>a</i>]pyridinium, 3,8-dihydroxy-, hydroxide, inner salt)	35143-55-6	**	8.70±0.05	EI	3977
C₇H₇NO₂S⁺	C ₆ H ₄ (NO ₂)SCH ₃ (Benzene, 1-(methylthiol)-4-nitro-)	701-57-5	**	8.59±0.01 (V)	PE	4389
C₈H₇NO₂S⁺	C ₇ H ₃ NS(O)(OH)CH ₃ (Thiazolo[3,2- <i>a</i>]pyridinium, 3,8-dihydroxy-2-methyl-, hydroxide, inner salt)	35191-20-9	**	8.60±0.05	EI	3977
C₈H₉NO₂S⁺	C ₅ H ₃ N(SCH ₃)OCOCH ₃ (3-Pyridinol, 2-(methylthio)- acetate (ester))	42715-30-0	**	7.91±0.05	EI	3977
C₁₂H₁₉NO₂S⁺	C ₁₂ H ₁₉ NO ₂ S (Benzeneethanamine, 2,5-dimethoxy- α -methyl-4-(methylthio)-(±)-)	69519-59-1	**	7.64±0.06 (V)	PE	4758
C₁₃H₉NO₂S⁺	C ₇ H ₃ NS(O)(OH)C ₆ H ₅ (Thiazolo[3,2- <i>a</i>]pyridinium, 3,8-dihydroxy-2-phenyl-, hydroxide, inner salt)	35143-57-8	**	8.42±0.05	EI	3977
C₁₃H₁₃NO₂S⁺	C ₇ H ₈ SO ₂ C ₆ H ₅ (2-Azabicyclo[3.2.1]octa-3,6-diene,2-(phenylsulfonyl)-)	2063-88-9	**	8.11 (V)	PE	5481

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₃H₁₅NO₂S⁺						
	C ₇ H ₁₀ NSO ₂ C ₆ H ₅ (2-Azabicyclo[3.2.1]oct-3-ene,2-(phenylsulfonyl)-)	2063-89-0	**	8.18 (V)	PE	5481
	C ₇ H ₁₀ NSO ₂ C ₆ H ₅ (2-Azabicyclo[3.2.1]oct-6-ene,2-(phenylsulfonyl)-)	71017-42-0	**	8.79	PE	5481
C₁₃H₁₇NO₂S⁺						
	C ₇ H ₁₂ NSO ₂ C ₆ H ₅ (2-Azabicyclo[3.2.1]octane,2-(phenylsulfonyl)-)	5503-65-1	**	8.72 (V)	PE	5481
C₃H₂N₂O₂S⁺						
	C ₃ H ₂ NS(NO ₂) (Isothiazole, 4-nitro-)	931-07-7	**	10.45	PE	3587
C₈H₆N₂O₂S⁺						
	C ₇ H ₃ NS(NO ₂)CH ₃ (Benzothiazole, 2-methyl-6-nitro-)	2941-63-1	**	9.15 (V)	PE	4437
C₁₂H₁₂N₂O₂S⁺						
	(C ₆ H ₄ NH ₂) ₂ SO ₂ (Benzenamine, 4,4'-sulfonylbis-)	80-08-0	**	7.25±0.05	PI	5040
			**	7.25	PI	4328
C₂₀H₂₄N₂O₂S⁺						
	C ₂₀ H ₂₄ N ₂ O ₂ S (Phenol, 2,2'-[thiobis(3,1-propanediyl nitrilomethylidyne)]bis-)	52279-44-4	**	8.51±0.10	EI	4213
C₇H₇N₃O₂S⁺						
	C ₆ H ₄ (NO ₂)NHCSNH ₂ (Thiourea, (2-nitrophenyl)-)	51039-84-0	**	8.30	EI	4834
C₉H₁₇NO₃S⁺						
	C ₉ H ₁₇ NO ₃ S (8-Azabicyclo[3.2.1]octan-3-ol,8-methyl-methanesulfonate(ester), <i>exo</i> -)	35136-87-9	**	7.7±0.15	EI	5401
C₁₅H₁₁NO₃S⁺						
	C ₇ H ₃ NOS(OCOCH ₃)C ₆ H ₅ (Thiazolo[3,2- <i>a</i>]pyridinium, 8-(acetoxy)-3-hydroxy-2-phenyl-, hydroxide, inner salt)	32002-92-9	**	6.27±0.05	EI	3977
C₁₅H₂₈NO₃S⁺						
	C ₂₀ H ₄₀ N ₂ O ₇ S (L-Tyrosine,N-[S-(2-methoxy-2-oxoethyl)-N-(1-oxodecyl)-L-cysteinyl]-methyl ester)	32886-16-1		8.7±0.1	PI	5279
C₁₆H₂₈NO₄S⁺						
	C ₂₀ H ₄₀ N ₂ O ₇ S (L-Tyrosine,N-[S-(2-methoxy-2-oxoethyl)-N-(1-oxodecyl)-L-cysteinyl]-methyl ester)	32886-16-1		9.0±0.1	PI	5279
C₂₆H₄₀N₂O₇S⁺						
	C ₂₀ H ₄₀ N ₂ O ₇ S (L-Tyrosine,N-[S-(2-methoxy-2-oxoethyl)-N-(1-oxodecyl)-L-cysteinyl]-methyl ester)	32886-16-1	**	8.3±0.1	PI	5279
C₂₇H₄₀N₄O₈S⁺						
	C ₂₂ H ₄₀ N ₄ O ₈ S (L-Cysteine,S-(2-methoxy-2-oxoethyl)-N-[N-[N-(1-oxopropyl)-L-phenylalanyl]-L-leucyl]glycyl]-methyl ester)	35146-63-5	**	8.3±0.1	PI	5279
C₂₂H₃₀N₄O₂S₂⁺						
	C ₂₂ H ₃₀ N ₄ O ₂ S ₂ (10H-Phenothiazine-2-sulfonamide, <i>N,N</i> -dimethyl-10[3-(4-methyl-1-piperazinyl)propyl]-)	316-81-4	**	6.81±0.07	CTS	4079
B₂C₃H₉NOS⁺						
	NB ₂ SO(CH ₃) ₃ (1,3,5,2,4-Oxathiazadiborolidine, 2,4,5-trimethyl-)	57877-90-4	**	9.00 (V)	PE	4526

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
FS⁺						
	SF	16068-96-5	** ** ** **	10.0±0.3 10.0 10.09±0.10 10.2±0.3	EI EI EI EI	4580 4544 3818 4864
	SF ₆	2551-62-4		30.5±0.5 37.6±3.0	EI EI	3818 4645
F₂S⁺						
	SF ₂	13814-25-0	** **	10.08 10.29±0.10	PE EI	5073 3818
	SF ₄	7783-60-0		17.4±0.5	EI	3818
	SF ₆	2551-62-4		27.0±0.3 27.5±0.5	EI EI	4645 3818
	S ₂ F ₂	13709-35-8		16.2±0.4	EI	3738
F₃S⁺						
	SF ₄	7783-60-0	F	12.63±0.10	EI	3818
	SF ₆	2551-62-4		19.4±0.5 19.6±0.5 20.0±0.5	PI EI EI	4917 4645 3818
F₄S⁺						
	SF ₄	7783-60-0	** **	12.03±0.05 12.08±0.10	EI EI	3578 3818
	SF ₆	2551-62-4		19.1±0.5 2F 18.44±0.10 19.6±1.0	PI EI EI	4917 3818 4645
F₅S⁺						
	SF ₆	2551-62-4		15.3±0.2	PI	4917
			F	15.50±0.10	EI	3818
			F	16.2±0.2	EI	4645
F₆S⁺						
	SF ₆	2551-62-4	**	15.7	PE	5232
FS₂⁺						
	S ₂ F ₂	13709-35-8		14.0±0.4	EI	3738
F₂S₂⁺						
	S ₂ F ₂	13709-35-8	** ** **	10.68 (V) 10.84 (V) 11.6±0.4	PE PE EI	4332 4332 3738
CF₂S⁺						
	F ₂ CS	420-32-6	** ** ** **	10.45±0.01 10.52 10.64 (V) 10.53±0.10	PE PE PE EI	3708 4080 3746 3818
C₂F₄S₂⁺						
	S=C(F)SCF ₃	371-73-3	**	10.12 (V)	PE	4345
C₃F₆S₃⁺						
	S=C(SCF ₃) ₂	461-08-5	**	9.25 (V)	PE	4345
C₁₀F₁₂S₄⁺						
	C ₆ S ₄ (CF ₃) ₄ (1,3-Dithiole, 2-(4,5-di-trifluoromethyl-1,3-dithiol-2-ylidene)-4,5-di-trifluoromethyl-)	26393-26-0	**	7.95 (V)	PE	4481

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{CH}_2\text{F}_3\text{S}^+$	$\text{CH}_2=\text{SF}_4$	66793-25-7	**	10.65 (V)	PE	4984
$\text{C}_9\text{H}_5\text{FS}_3^+$	$(\text{C}_3\text{HS}_2)=\text{S}(\text{C}_6\text{H}_4\text{F})$ (3H-1,2-Dithiole-3-thione, 5-(4-fluorophenyl)-)	54290-50-5	**	8.14 (V)	PE	4403
NFS^+						
(${}^2\text{A}'$)	NSF	18820-63-8	**	11.49 ± 0.02	PE	3665
(${}^2\text{A}'$)			**	11.54 ± 0.01	PE	3666
(${}^2\text{A}'$)			**	11.82 (V)	PE	3660
(${}^2\text{A}'$)			**	13.382 ± 0.004	PE	3666
(${}^2\text{A}'$)			**	13.39 ± 0.02	PE	3665
(${}^2\text{A}'$)			**	13.50 (V)	PE	3660
(${}^2\text{A}''$)			**	13.775 ± 0.005	PE	3666
(${}^2\text{A}''$)			**	13.78 ± 0.02	PE	3665
(${}^2\text{A}''$)			**	13.87 (V)	PE	3660
(${}^2\text{A}'$)			**	14.93 ± 0.01	PE	3666
(${}^2\text{A}'$)			**	15.35 ± 0.02	PE	3665
(${}^2\text{A}'$)			**	15.61 (V)	PE	3660
(${}^2\text{A}''$)			**	16.56 ± 0.03 (V)	PE	3666
(${}^2\text{A}'$)			**	17.24 ± 0.08 (V)	PE	3666
(${}^2\text{A}'$)			**	21.1 ± 0.1 (V)	PE	3666
NF_3S^+	NSF ₃	15930-75-3	**	12.50 (V)	PE	3660
$\text{C}_8\text{H}_8\text{NFS}^+$	$\text{C}_6\text{H}_4\text{FNHCSCl}_3$ (Ethanethioamide, N-(2-fluorophenyl)-)	39184-82-2	**	8.30	EI	4834
$\text{C}_7\text{H}_7\text{N}_2\text{FS}^+$	$\text{C}_6\text{H}_4\text{FNHCSNH}_2$ (Thiourea, (2-fluorophenyl)-)	656-32-6	**	8.15	EI	4834
$\text{C}_{21}\text{H}_{24}\text{N}_3\text{F}_3\text{S}^+$	$\text{C}_{12}\text{H}_7\text{NS}(\text{CF}_3)(\text{CH}_2)_3\text{C}_4\text{H}_8\text{N}_2\text{CH}_3$ (10H-Phenothiazine, 10-[3-(4-methyl-1-piperazinyl)propyl]-2-(trifluoromethyl)-)	117-89-5	**	7.10 ± 0.07	CTS	4079
			**	7.31 ± 0.08 (V)	PE	4667
O_2FS^+	SO_2F_2 SO_2FCl	2699-79-8 13637-84-8	**	14.8 ± 0.5 13.0 ± 0.5	EI EI	4921 4921
O_3FS^+	SO_3F	21549-02-0	**	12.85 ± 0.1 (V)	PE	3671
OF_2S^+	SOF_2	7783-42-8	**	12.19	PE	3705
			**	12.25	PE	3879
			**	12.58 (V)	PE	3646
			**	12.58 (V)	PE	4295
			**	12.6 (V)	PE	3694
			**	12.58 ± 0.10	EI	3818
$\text{O}_2\text{F}_2\text{S}^+$	SO_2F_2	2699-79-8	**	~ 13.0	PE	3879
			**	13.04 ± 0.01	PE	3675
			**	13.43 (V)	PE	3705
			**	13.55 (V)	PE	3694
			**	13.75 (V)	PE	4827

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
O₂F₂S⁺	SO ₂ F ₂	2699-79-8	**	13.75 (V)	PE	5207
CH₃O₂FS⁺	(CH ₃)SO ₂ (F)	558-25-8	** ** **	12.53 (V) 12.53 (V) 12.61 (V)	PE PE PE	4827 5207 3705
C₆H₃OF₃S⁺	C ₄ H ₃ SCOCF ₃ (Ethanone, 2,2,2-trifluoro-1-(2-thienyl)-) C ₄ H ₃ SCOCF ₃ (Ethanone, 2,2,2-trifluoro-1-(3-thienyl)-)	651-70-7 30933-31-4	**	9.70±0.05 9.63±0.05	EI EI	3482 3482
C₂₀H₂₁N₂OF₃S⁺	C ₁₂ H ₇ NS(CF ₃)COCH ₂ CH ₂ N(C ₂ H ₅) ₂ (10H-Phenothiazine, 10-[3-(diethylamino)-1-oxopropyl]-2-(trifluoromethyl)-)	30223-48-4	**	7.89±0.07	CTS	4079
C₂₂H₂₆N₃OF₃S⁺	C ₂₂ H ₂₆ N ₃ OF ₃ S (1-Piperazineethanol, 4-[3-[2-(trifluoromethyl)-10H-phenothiazin-10-yl]propyl]-)	69-23-8	**	8.64±0.07	CTS	4079
C₂₀H₁₉N₂O₂F₃S⁺	C ₁₂ H ₇ NS(CF ₃)COCH ₂ CH ₂ C ₄ H ₈ NO (10H-Phenothiazine, 10-[3-(4-morpholinyl)-1-oxopropyl]-2-(trifluoromethyl)-)	33414-29-8	**	8.54±0.07	CTS	4079
C₂₂H₂₄N₃O₂F₃S⁺	C ₂₂ H ₂₄ N ₃ O ₂ F ₃ S (10H-Phenothiazine, 10-[3-[4-(2-hydroxyethyl)-1-piperazinyl]-1-oxopropyl]-2-(trifluoromethyl)-)	33414-36-7	**	8.71±0.07	CTS	4079
H₄SiS⁺	SiH ₃ SH	14044-97-4	**	9.97 (V)	PE	3656
H₆Si₂S⁺	(SiH ₃) ₂ S	16544-95-9	** **	9.59 (V) 9.70 (V)	PE PE	3867 3656
CH₆SiS⁺	(CH ₃ S) ₂	16643-15-5	**	9.10 (V)	PE	3867
C₄H₁₂SiS⁺	(CH ₃) ₃ SiSCH ₃	3908-55-2	**	8.69±0.05 (V)	PE	4153
C₈H₁₁SiS⁺	C ₆ H ₅ SSi(CH ₃) ₃ (Silane, trimethyl(phenylthio)-)	4551-15-9	CH ₃	9.93±0.1	EI	4198
C₉H₁₄SiS⁺	C ₆ H ₅ SSi(CH ₃) ₃ (Silane, trimethyl(phenylthio)-)	4551-15-9	** **	8.67±0.05 8.28±0.1	PE EI	4589 4198
C₁₀H₁₆SiS⁺	C ₆ H ₄ (SCH ₃)Si(CH ₃) ₃ (Silane, trimethyl[4-(methylthio)phenyl]-) C ₆ H ₅ SCH ₂ Si(CH ₃) ₃ (Silane, trimethyl [(phenylthio)methyl]-)	22515-25-9 17873-08-4	** **	7.93±0.05 (V) 7.81±0.05 (V)	PE PE	4627 4627
C₁₁H₁₈SiS⁺	C ₁₁ H ₁₈ SiS (Silane, trimethyl[[4-(methylthio)phenyl]methyl]-)	59163-55-2	**	7.72±0.05 (V)	PE	4627

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{13}\text{H}_{11}\text{SiS}^+$	$\text{C}_{12}\text{H}_8\text{SiS}(\text{CH}_3)_2$ (10H-Phenothiasilin, 10,10-dimethyl-)	61431-08-1	CH_3	8.5 ± 0.1	EI	4664
$\text{C}_{14}\text{H}_{11}\text{SiS}^+$	$\text{C}_{12}\text{H}_8\text{SiS}(\text{CH}_3)_2$ (10H-Phenothiasilin, 10,10-dimethyl-)	61431-08-1	**	7.8 ± 0.1	EI	4664
$\text{C}_6\text{H}_{18}\text{Si}_2\text{S}^+$	$((\text{CH}_3)_2\text{Si})_2\text{S}$	3385-94-2	** **	8.74 ± 0.05 (V) 8.70 ± 0.1	PE EI	4153 4198
CH_3NSiS^+	SiH_3NCS	14311-54-7	**	9.54 ± 0.02 (V)	PE	3670
$\text{C}_4\text{H}_9\text{NSiS}^+$	$(\text{CH}_3)_3\text{SiNCS}$	2290-65-5	**	9.3 ± 0.1 (V)	PE	3670
$\text{B}_2\text{C}_5\text{H}_{16}\text{N}_2\text{SiS}^+$	$\text{N}_2\text{B}_2\text{SH}(\text{CH}_3)_2\text{Si}(\text{CH}_3)_3$ (1,3,4,2,5-Thiadiazadiborolidine, 2,5-dimethyl-3-(trimethylsilyl)-)	57877-86-8	**	8.25 (V)	PE	4526
H_3NOSiS^+	SiH_3NSO	57251-86-2	**	10.55 (V)	PE	4409
PS^+	SP	12281-36-6	**	9.0	EI	4001
P_4S^+	P_4S		**	10.6 ± 0.5	EI	3615
P_4S_2^+	P_4S_2	12165-70-7	**	10.6 ± 0.5	EI	3615
P_4S_3^+	P_4S_3	1314-85-8	** **	9.01 (V) 9.7 ± 0.5	PE EI	4704 3615
P_4S_4^+	P_4S_4	XXXXX-XX-X	**	10.1 ± 0.5	EI	3615
P_4S_5^+	P_4S_5	12137-70-1	**	10.4 ± 0.5	EI	3615
P_4S_6^+	P_4S_6	XXXXX-XX-X	**	10.0 ± 0.5	EI	3615
P_4S_7^+	P_4S_7	12037-82-0	**	10.1 ± 0.5	EI	3615
P_4S_8^+	P_4S_8	37295-14-0	**	9.8 ± 0.5	EI	3615
P_4S_9^+	P_4S_9	25070-46-6	**	9.8 ± 0.5	EI	3615
$\text{P}_4\text{S}_{10}^+$	P_4S_{10}	12066-62-5	**	9.6 ± 0.5	EI	3615
CH_2PS^+	$(\text{CH}_3\text{O})_2\text{P}(\text{CH}_3)\text{S}$	2953-29-9	H + HCHO + HS	14.05 ± 0.30	EI	3989

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₂H₇PS⁺	(CH ₃) ₂ P(S)H	6591-05-5	**	8.78 (V)	PE	5523
C₃H₉PS⁺	(CH ₃) ₃ PS	2404-55-9	** **	8.48±0.035 (V) 8.48 (V)	PE PE	5529 5442
C₆H₁₃PS₂⁺	C ₂ H ₅ S ₂ P(C ₂ H ₅) ₂	5745-32-4	**	8.68 (V)	PE	5569
C₆H₁₈N₃PS⁺	PS(N(CH ₃) ₂) ₃	3732-82-9	** ** **	7.66±0.003 7.66±0.02 8.05 (V)	PE PE PE	4086 4279 5627
C₂H₆OPS⁺	(CH ₃ O) ₂ P(CH ₃ S)S	2953-29-9	HCHO + HS	11.70±0.20	EI	3989
C₇H₁₇OPS⁺	(C ₃ H ₇ O)(C ₂ H ₅) ₂ PS	54867-58-2	** **	8.08±0.04 8.53 (V)	PE PE	4279 5627
C₂H₆O₂PS⁺	(CH ₃ O) ₂ P(CH ₃ S)O (CH ₃ O) ₂ P(CH ₃ S)S (CH ₃ S) ₂ P(CH ₃ O)O	152-20-5 2953-29-9 22608-53-3	CH ₃ O CH ₃ S CH ₃ S	11.82±0.20 10.10±0.10 10.50±0.10	EI EI EI	3989 3989 3989
C₂H₇O₂PS⁺	(CH ₃ O) ₂ P(CH ₃ S)O (CH ₃ O) ₂ P(CH ₃ S)S (CH ₃ S) ₂ P(CH ₃ O)O	152-20-5 2953-29-9 22608-53-3	HCHO HCHS HCHS	10.51±0.10 10.35±0.10 10.10±0.10	EI EI EI	3989 3989 3989
C₂H₆O₃PS⁺	(CH ₃ O) ₂ P(CH ₃ S)O	152-20-5	CH ₃	10.03±0.10	EI	3989
C₃H₉O₃PS⁺	(CH ₃ O) ₃ PS (CH ₃ O) ₂ P(CH ₃ S)O	152-18-1 152-20-5	** **	9.16 (V) 9.55±0.10	PE EI	4705 3989
C₆H₁₃O₃PS⁺	(C ₂ H ₅ O) ₃ PS SP(OC ₂ H ₅) ₃	126-68-1 1186-09-0	** **	8.49±0.02 8.96 (V) 8.96 (V)	PE PE PE	4279 5514 5627
C₁₂H₂₇O₃PS⁺	SP(OC ₄ H ₉) ₃	12408-16-1	**	8.02	PE	5627
C₂H₆OPS₂⁺	(CH ₃ O) ₂ P(CH ₃ S)S (CH ₃ S) ₂ P(CH ₃ O)O	2953-29-9 22608-53-3	CH ₃ O CH ₃ O	10.20±0.30 10.15±0.10	EI EI	3989 3989
C₂H₇OPS₂⁺	(CH ₃ O) ₂ P(CH ₃ S)S (CH ₃ S) ₂ P(CH ₃ O)O	2953-29-9 22608-53-3	HCHO HCHO	10.00±0.10 9.90±0.20	EI EI	3989 3989
C₂H₆O₂PS₂⁺	(CH ₃ O) ₂ P(CH ₃ S)S (CH ₃ S) ₂ P(CH ₃ O)O	2953-29-9 22608-53-3	CH ₃ CH ₃	9.65±0.20 9.47±0.10	EI EI	3989 3989
C₃H₉O₂PS₂⁺	(CH ₃ O) ₂ P(CH ₃ S)S (CH ₃ S) ₂ P(CH ₃ O)O	2953-29-9 22608-53-3	** **	9.0±0.10 9.20±0.10	EI EI	3989 3989

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₄H₁₁O₂PS₂⁺	HS ₂ P(OC ₂ H ₅) ₂	298-06-6	**	9.1 (V)	PE	4636
C₈H₁₉O₂PS₃⁺	(C ₂ H ₅ O) ₂ P(=S)SCH ₂ CH ₂ SC ₂ H ₅	XXXXX-XX-X	**	9.0 (V)	PE	5190
F₃PS⁺	F ₃ PS	2404-52-6	**	11.05±0.035 (V)	PE	5529
CNF₂PS⁺	PF ₂ NCS	461-60-9	**	10.2±0.1 (V)	PE	3662
Cl⁺						
(³ P ₂)	Cl(² P _{3/2} ^o)	22537-15-1	**	12.97±0.02	PE	5087
(³ P ₂)			**	12.97	PE	5214
(³ P ₁)			**	13.06±0.02	PE	5087
(³ P ₁)			**	13.06	PE	5214
(³ P ₀)			**	13.1	PE	5214
(¹ D ₂)			**	14.41±0.02	PE	5087
(¹ D ₂)			**	14.412	S	5209
(¹ D ₂)			**	14.42	PE	5214
(¹ S ₀)			**	16.42	PE	5214
CH ₂ Cl ₂		75-09-2	CH ₂ Cl	17.4±0.1	EI	3442
			CH ₂ Cl	17.4	EI	3490
COCl ₂		75-44-5	Cl, CO	16.5±0.2	PI	5041
(CH ₃) ₂ CClNO		2421-26-3		22.70	EI	4809
CF ₃ Cl		75-72-9	CF ₃	19.66±0.1	PI	5399
CF ₂ Cl ₂		75-71-8	CFCl ₂ ⁻	16.40±0.2	PI	5399
CFCl ₃		75-69-4	F ⁻ +CCl ₂	13.7±0.5	PI	5399
			F ⁻ +CCl ₂	15.20±0.1	PI	5399
			Cl ⁻ +CFCl	15.6±0.1	PI	5399
Ag ₃ Cl ₃		12444-97-2		~15.5	EI	3605
Cl⁺²						
Cl ⁺		14835-24-6	**	23.8137±0.0002	S	3756
			**	23.8138±0.0002	S	4175
Cl₂⁺						
(² Π _p)	Cl ₂	7782-50-5	**	11.49	PE	3507
(² Π _u)			**	14.43 (V)	PE	3507
(² Σ ⁺)			**	16.10 (V)	PE	3507
CF ₂ Cl ₂		75-71-8	CF ₂	15.40±0.1	PI	5399
HCl⁺						
HCl		7647-01-0	**	12.72±0.03	PI	5307
(CH ₃) ₂ CClNO		2421-26-3		13.35	EI	4809
H₂Cl⁺						
(HCl) ₂		XXXXX-XX-X	Cl	12.32±0.03	PI	5307
H₂Cl₂⁺						
(HCl) ₂		XXXXX-XX-X	**	11.91±0.05	PI	5307
LiCl⁺						
LiCl		7447-41-8	**	9.57	PI	5509
			**	10.01±0.02 (V)	PE	4950
Li₂Cl₂⁺						
(LiCl) ₂		12345-57-2	**	10.20	PI	5509
			**	~10.70 (V)	PE	4950

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
Li₃Cl₃⁺	(LiCl) ₃	59217-69-5	**	10.17	PI	5509
BeCl₂⁺	BeCl ₂	7787-47-5	**	12.5±1.0	EI	4113
BCl⁺	BCl	20583-55-5	**	12±1	EI	3465
BCl₂⁺	BCl ₂	13842-52-9	**	12±1.0	EI	3465
BCl₃⁺	BCl ₃	10294-34-5	**	11.62 (V)	PE	3704
B₂Cl₄⁺	B ₂ Cl ₄	13701-67-2	**	<10.42±0.02	PE	3709
H₈B₅Cl⁺	B ₅ H ₈ Cl (Pentaborane(9), 1-chloro-)	19469-13-7	**	10.03 (V)	PE	4519
	B ₅ H ₈ Cl (Pentaborane(9), 2-chloro-)	19469-14-8	**	10.24 (V)	PE	4519
CCl⁺	C ₂ F ₃ Cl	79-38-9	CF ₃	16.9±0.1	EI	4070
	CF ₂ Cl ₂	75-71-8	FCl+F ⁻	14.80±0.2	PI	5399
	CFCl=CFCl	598-88-9	CF ₂ Cl	16.4±0.2	EI	4070
	CFCl ₃	75-69-4	2Cl+F	20.00±0.2	PI	5399
				20.5	PI	5196
CCl₂⁺	CFCl=CFCl	598-88-9	CF ₂	13.8±0.1	EI	4070
	CFCl ₃	75-69-4	FCl	17.0	PI	5196
			Cl+F	17.12±0.04	PI	4757
C₄Cl₂⁺	CCl≡CC≡CCl	51104-87-1	**	9.34±0.02	PE	4162
CCl₃⁺	CCl ₃	3170-80-7	**	8.28	EI	3732
	CCl ₄	56-23-5	Cl	11.28±0.03	PI	4308
			Cl	11.37	EI	3732
	(CCl ₃) ₂ CO	116-16-5		11.75	EI	3550
	CFCl ₃	75-69-4	F	13.25±0.04	PI	4757
			F	13.50	PI	5196
C₆Cl₄⁺	C ₆ Cl ₄ (1,3-Cyclohexadien-5-yne, 1,2,3,4-tetrachloro-)	13280-72-3	**	10.66±0.2	EI	3583
	C ₆ O ₃ Cl ₄ (1,3-Isobenzofurandione, 4,5,6,7-tetrachloro-)	117-08-8		14.31±0.2	EI	3583
	C ₆ Cl ₅ I (Benzene, pentachloroiodo-)	16478-18-5		14.51±0.2	EI	3583
	C ₆ Cl ₄ I ₂ (Tetrachloro-1,2-diiodobenzene)	XXXXX-XX-X		12.85±0.2	EI	3583
	C ₂ Cl ₆	67-72-1	**	11.22 (V)	PE	4547
C₆Cl₆⁺	C ₆ Cl ₆ (Benzene, hexachloro-)	118-74-1	**	9.20 (V)	PE	3873

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₇H₇Cl⁺						
	C ₆ H ₅ CH ₂ Cl	25168-05-2	**	9.14±0.01	PI	5557
	C ₆ H ₅ CH ₂ Cl (Benzene, (chloromethyl)-)	100-44-7	**	9.30 (V)	PE	3992
	C ₆ H ₅ ClCH ₃ (Benzene, 1-chloro-2-methyl-)	95-49-8	**	8.72±0.1	EI	3777
	C ₆ H ₅ ClCH ₃ (Benzene, 1-chloro-3-methyl-)	108-41-8	**	8.67±0.1	EI	3777
	C ₆ H ₅ ClCH ₃ (Benzene, 1-chloro-4-methyl-)	106-43-4	**	8.78±0.1	EI	3777
C₈H₅Cl⁺						
	C ₆ H ₅ C≡CCl (Benzene, (chloroethynyl)-)	1483-82-5	**	8.70 (V)	PE	4334
	C ₆ H ₅ (Cl)C≡CH (Benzene, 1-chloro-4-ethynyl-)	873-73-4	**	8.75 (V)	PE	4334
C₈H₇Cl⁺						
	C ₆ H ₅ ClCH ₂ CH ₂ OCOCH ₃ (Phenethyl alcohol, <i>m</i> -chloro-, acetate)	33709-41-0		8.90	EI	3590
C₈H₉Cl⁺						
	CH ₃ C ₆ H ₄ CH ₂ Cl (Benzene, 1-(chloromethyl)-3-methyl-)	620-19-9	**	8.82±0.03	PI	5557
	CH ₃ C ₆ H ₄ CH ₂ Cl (Benzene, 1-(chloromethyl)-4-methyl-)	104-82-5	**	8.79±0.03	PI	5557
C₉H₉Cl⁺						
	C ₆ H ₅ (Cl)C ₃ H ₅ (Benzene, 1-chloro-4-cyclopropyl-)	1798-84-1	**	8.64 (V)	PE	4815
C₉H₁₁Cl⁺						
	(CH ₃) ₂ C ₆ H ₃ CH ₂ Cl (Benzene, 1-(chloromethyl)-3,5-dimethyl-)	2745-54-2	**	8.63±0.03	PI	5557
C₁₀H₁₁Cl⁺						
	C ₆ H ₅ (Cl)C ₃ H ₄ (CH ₃) (Benzene, 1-chloro-4-(1-methylcyclopropyl)-)	63340-05-6	**	8.67 (V)	PE	4815
C₁₀H₁₃Cl⁺						
	C ₆ H ₅ Cl(tert-C ₄ H ₉) (Benzene, 1-chloro-4-(1,1-dimethylethyl)-)	3972-56-3	**	8.82 (V)	PE	4438
C₁₀H₁₅Cl⁺						
	C ₁₀ H ₁₅ Cl (Tricyclo[3.3.1.1 ^{3,7}]decane, 1-chloro-)	935-56-8	**	9.30	PE	3886
C₁₁H₁₃Cl⁺						
	C ₆ H ₅ (Cl)C ₃ H ₄ (C ₂ H ₅) (Benzene, 1-chloro-4-(1-ethylcyclopropyl)-)	63340-06-7	**	8.64 (V)	PE	4815
C₁₁H₁₅Cl⁺						
	(tert-C ₄ H ₉)C ₆ H ₄ CH ₂ Cl (Benzene, 1-(chloromethyl)-3-dimethylethyl-)	38580-79-9	**	8.71±0.03	PI	5557
	(tert-C ₄ H ₉)C ₆ H ₄ CH ₂ Cl (Benzene, 1-(chloromethyl)-4-dimethylethyl-)	19692-45-6	**	8.60±0.03	PI	5557
C₁₂H₉Cl⁺						
	C ₆ H ₅ C ₆ H ₄ Cl (1,1'-Biphenyl, 2-chloro-)	2051-60-7	**	8.20±0.02	PE	3702
	C ₆ H ₅ C ₆ H ₄ Cl (1,1'-Biphenyl, 4-chloro-)	2051-62-9	**	8.10±0.02	PE	3702

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₂H₁₅Cl⁺	C ₁₂ H ₁₅ Cl (Benzene, 1-chloro-4-[1-(1-methylethyl)cyclopropyl]-)	63340-07-8	**	8.64 (V)	PE	4815
C₁₃H₁₇Cl⁺	C ₁₃ H ₁₇ Cl (Benzene, 1-chloro-4-[1-(1,1-dimethylethyl)cyclopropyl]-)	63340-08-9	**	8.64 (V)	PE	4815
C₁₄H₉Cl⁺	C ₁₄ H ₉ Cl (Anthracene, 9-chloro-)	716-53-0	**	7.45±0.03 (V)	PE	4887
C₁₅H₂₃Cl⁺	(<i>tert</i> -C ₄ H ₉) ₂ C ₆ H ₃ CH ₂ Cl (Benzene, 1-(chloromethyl)-3,5-bis(1,1-dimethylethyl)-)	51625-14-0	**	8.29±0.03	PI	5557
C₂₁H₁₅Cl⁺	C ₃ (C ₆ H ₅) ₂ Cl (Cyclopropenylium, triphenyl-,chloride)	58090-78-1	**	7.75±0.05	EI	4628
CHCl₂⁺	CHCl ₂	3474-12-2	**	8.45	EI	3732
	CHCl ₃	67-66-3	Cl	11.49±0.02	PI	4308
			Cl	11.52	EI	3732
	CHCl ₂ CH ₂ Cl	79-00-5	CH ₂ Cl	11.80	EI	3732
CH₂Cl₂⁺	CH ₂ Cl ₂	75-09-2	**	11.32±0.01	PI	4308
			**	11.28	EI	3732
C₂H₂Cl₂⁺	CH ₂ =CCl ₂	75-35-4	**	10.00 (V)	PE	4303
			**	9.99±0.02 (V)	PE	4880
	cis-CHCl=CHCl	156-59-2	**	9.80 (V)	PE	4303
	trans-CHCl=CHCl	156-60-5	**	9.72 (V)	PE	3648
			**	9.80 (V)	PE	4303
			**	11.92 (V)	PE	4022
C₂H₄Cl₂⁺	CH ₃ CHCl ₂	75-34-3	**	11.06	PI	5501
			**	11.02	PE	5501
			**	11.23±0.02 (V)	PE	4547
	CH ₂ ClCH ₂ Cl	107-06-2	**	11.05	PI	5501
			**	11.04	PE	5501
			**	11.13±0.10 (V)	PE	4732
			**	11.22±0.02 (V)	PE	4367
			**	11.39±0.03 (V)	PE	4144
			**	11.40±0.10 (V)	PE	4732
C₅H₆Cl₂⁺	C ₅ H ₆ Cl ₂ (Cyclopentene, 4,4-dichloro-)	XXXXX-XX-X	**	9.78 (V)	PE	4517
C₆H₂Cl₂⁺	C ₆ H ₂ Cl ₂ (1,3-Cyclohexadien-5-yne, 1,2-dichloro-)	24634-92-2	**	9.66±0.2	EI	3583
	C ₆ H ₂ Cl ₂ (1,3-Cyclohexadien-5-yne, 1,3-dichloro-)	24634-94-4	**	9.97±0.2	EI	3583
	C ₆ H ₂ Cl ₂ (1,3-Cyclohexadien-5-yne, 1,4-dichloro-)	XXXXX-XX-X	**	9.11±0.2	EI	3583
	C ₆ H ₂ Cl ₂ (1,3-Cyclohexadien-5-yne, 2,3-dichloro-)	24634-93-3	**	9.58±0.2	EI	3583

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_6\text{H}_2\text{Cl}_2^+$	$\text{C}_6\text{H}_2\text{O}_3\text{Cl}_2$ (1,3-Isobenzofurandione, 4,7-dichloro-) $\text{C}_6\text{H}_2\text{O}_3\text{Cl}_2$ (1,3-Isobenzofurandione, 5,6-dichloro-) $\text{C}_6\text{H}_2\text{Cl}_2\text{I}_2$ (3,4-Dichloro-1,2-diiodobenzene) $\text{C}_6\text{H}_2\text{Cl}_2\text{I}_2$ (3,5-Dichloro-1,2-diiodobenzene) $\text{C}_6\text{H}_2\text{Cl}_2\text{I}_2$ (4,5-Dichloro-1,2-diiodobenzene)	4466-59-5 942-06-3 XXXXX-XX-X XXXXX-XX-X XXXXX-XX-X		13.60±0.2 14.06±0.2 14.11±0.2 14.43±0.2 14.11±0.2	EI	3583 3583 3583 3583 3583
$\text{C}_6\text{H}_4\text{Cl}_2^+$	$\text{C}_6\text{H}_4\text{Cl}_2$ (Benzene, 1,2-dichloro-) $\text{C}_6\text{H}_4\text{Cl}_2$ (Benzene, 1,3-dichloro-) $\text{C}_6\text{H}_4\text{Cl}_2$ (Benzene, 1,4-dichloro-)	95-50-1 541-73-1 106-46-7	** ** ** ** **	9.06±0.02 9.08 (V) 9.12±0.02 9.15 (V) 8.98±0.02 8.988 (V) 9.00 (V)	PE PE PE PE PE	5138 3873 5138 3873 5138 5257 3873
$\text{C}_7\text{H}_6\text{Cl}_2^+$	$\text{C}_6\text{H}_3\text{Cl}_2\text{CH}_3$ (Benzene, 1,3-dichloro-2-methyl-) $\text{C}_6\text{H}_3\text{Cl}_2\text{CH}_3$ (Benzene, 1,3-dichloro-5-methyl-) $\text{C}_6\text{H}_3\text{Cl}_2\text{CH}_3$ (Benzene, 1,4-dichloro-2-methyl-)	118-69-4 25186-47-4 19398-61-9	** ** ** **	8.73 (V) 9.99±0.02 9.99 (V) 8.75±0.02	PE PE PE PE	5461 5521 5461 5521
$\text{C}_8\text{H}_6\text{Cl}_2^+$	$\text{C}_6\text{H}_3(\text{Cl})_2\text{CH}=\text{CH}_2$ (Benzene, 1,3-dichloro-2-ethenyl-)	28469-92-3	**	8.70±0.02	PE	3854
$\text{C}_8\text{H}_8\text{Cl}_2^+$	$\text{C}_6\text{H}_5\text{CH}_2\text{CHCl}_2$ (Benzene, (2,2-dichloroethyl)-)	4412-39-9	**	9.27 (V)	PE	4927
$\text{C}_9\text{H}_8\text{Cl}_2^+$	$\text{C}_6\text{H}_5\text{C}_3\text{H}_3\text{Cl}_2$ (Benzene, (2,2-dichlorocyclopropyl)-)	2415-80-7	**	8.97 (V)	PE	4927
$\text{C}_{10}\text{H}_6\text{Cl}_2^+$	$\text{C}_{10}\text{H}_6\text{Cl}_2$ (Azulene, 1,3-dichloro-)	14658-94-7	**	7.45 (V)	PE	5397
$\text{C}_{14}\text{H}_8\text{Cl}_2^+$	$\text{C}_{14}\text{H}_8\text{Cl}_2$ (Anthracene, 9,10-dichloro-)	605-48-1	**	7.58	PE	4364
$\text{C}_{15}\text{H}_{10}\text{Cl}_2^+$	$\text{C}_{15}\text{H}_{10}\text{Cl}_2$ (1H-Cyclopropa[<i>l</i>]phenanthrene, 1,1-dichloro-1 <i>a</i> ,9 <i>b</i> -dihydro-)	37608-29-0	**	8.06 (V)	PE	4927
CHCl_3^+	CHCl_3	67-66-3	** ** **	11.37±0.02 11.48 (V) 11.41	PI PE EI	4308 4146 3732

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_2\text{H}_3\text{Cl}_3^+$	CH_3CCl_3	71-55-6	**	11.25 (V)	PE	4547
$\text{C}_6\text{H}_3\text{Cl}_3^+$	$\text{C}_6\text{H}_3\text{Cl}_3$ (Benzene, 1,2,3-trichloro-)	87-61-6	**	9.22 (V)	PE	3873
	$\text{C}_6\text{H}_3\text{Cl}_3$ (Benzene, 1,3,5-trichloro-)	108-70-3	**	9.34 ± 0.02	PE	5138
			**	9.36 (V)	PE	3873
$\text{C}_2\text{H}_2\text{Cl}_4^+$	$\text{CH}_2\text{ClCCl}_3$	630-20-6	**	11.45 (V)	PE	4547
$\text{C}_6\text{H}_2\text{Cl}_4^+$	$\text{C}_6\text{H}_2\text{Cl}_4$ (Benzene, 1,2,3,4-tetrachloro-)	634-66-2	**	9.11 (V)	PE	3873
	$\text{C}_6\text{H}_2\text{Cl}_4$ (Benzene, 1,2,3,5-tetrachloro-)	634-90-2	**	9.16 (V)	PE	3873
	$\text{C}_6\text{H}_2\text{Cl}_4$ (Benzene, 1,2,4,5-tetrachloro-)	95-94-3	**	9.06 (V)	PE	3873
			**	9.20 ± 0.05 (V)	PE	5558
C_2HCl_5^+	$\text{CHCl}_2\text{CCl}_3$	76-01-7	**	11.28 (V)	PE	4547
C_6HCl_5^+	C_6HCl_5 (Benzene, pentachloro-)	608-93-5	**	9.11 (V)	PE	3873
$\text{BeC}_5\text{H}_5\text{Cl}^+$	$(\text{C}_5\text{H}_5)\text{BeCl}$ (Beryllium, chloro(η^5 -2,4-cyclopentadien-1-yl)-)	36346-97-1	**	9.60 (V)	PE	5384
$\text{BC}_2\text{H}_6\text{Cl}^+$	$(\text{CH}_3)_2\text{BCl}$ (Borane, chlorodimethyl)	1803-36-7	**	10.78 (V)	PE	5485
$\text{B}_4\text{C}_2\text{H}_5\text{Cl}^+$	$\text{C}_2\text{B}_4\text{H}_5\text{Cl}$ (1,6-Dicarbahexaborane(6),2-chloro-)	33616-59-0	**	9.53 (V)	PE	5553
$\text{BCH}_3\text{Cl}_2^+$	CH_3BCl_2 (Borane, dichloromethyl)	7318-78-7	**	11.51	PE	5485
$\text{B}_4\text{C}_2\text{H}_4\text{Cl}_2^+$	$\text{C}_2\text{B}_4\text{H}_4\text{Cl}_2$ (1,6-Dicarbahexaborane(6),2,4-dichloro-)	XXXXX-XX-X	**	9.38 (V)	PE	5553
$\text{BC}_6\text{H}_5\text{Cl}_2^+$	$\text{C}_6\text{H}_5\text{BCl}_2$ (Borane, dichlorophenyl-)	873-51-8	**	9.52 (V)	PE	4956
N_3Cl^+	ClN_3	13973-88-1	**	10.20 ± 0.01	PE	5001
NCl_3^+	NCl_3	10025-85-1	**	10.12 ± 0.1	PE	4737
H_2NCl^+	NH_2Cl	10599-90-3	**	9.85 ± 0.02	PE	4763
			**	10.60 (V)	PE	5544

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
HNCl₂⁺	NHCl ₂	3400-09-7	** **	9.98±0.05 10.52 (V)	PE PE	4737 5544
H₃B₃N₃Cl₃⁺	B ₃ H ₃ N ₃ Cl ₃ (Borazine, 2,4,6-trichloro-)	933-18-6	** **	10.55 (V) 10.55 (V)	PE PE	3673 3943
C₃NCl⁺	CCl=CCN	2003-31-8	**	10.95±0.02	PE	4765
C₅NCl₅⁺	C ₅ N(Cl) ₅ (Pyridine, pentachloro-)	2176-62-7	**	9.44 (V)	PE	4275
CH₄NCl⁺	CH ₃ NHCl	6154-14-9	** **	9.19±0.02 9.70±0.10 (V)	PE PE	4737 4741
C₂H₂NCl⁺	CH ₂ ClCN	107-14-2	** **	11.95±0.01 11.98 (V)	PE PE	4679 4684
C₂H₆NCl⁺	(CH ₃) ₂ NCl	1585-74-6	** **	8.67±0.02 9.25 (V)	PE PE	4737 5304
C₃H₂NCl⁺	CH ₂ =C(Cl)CN	920-37-6	**	10.58±0.05 (V)	PE	4859
C₅H₄NCl⁺	ClC ₅ H ₄ N (Pyridine, 2-chloro-)	109-09-1	**	9.54 (V)	PE	5258
			**	9.9±0.1	EI	4302
	ClC ₅ H ₄ N (Pyridine, 3-chloro-)	626-60-8	**	9.58 (V)	PE	5258
			**	9.75±0.1	EI	4302
	ClC ₅ H ₄ N (Pyridine, 4-chloro-)	626-61-9	** **	10.2 (V) 9.86 (V)	PI PE	5566 5258
			**	10.0±0.1	EI	4302
C₅H₁₀NCl⁺	C ₅ H ₁₀ NCl (Piperidine, 1-chloro-)	2156-71-0	**	9.00±0.10 (V)	PE	5308
C₆H₆NCl⁺	C ₆ H ₅ CINH ₂ (Benzeneamine, 2-chloro-)	95-51-2	**	8.50	EI	4834
	C ₆ H ₅ CINHCOC ₂ H ₅ (Acetamide, N-(2-chlorophenyl)-)	533-17-5		11.05	EI	4834
	C ₆ H ₅ CINHCOC ₂ H ₅ (Acetamide, N-(4-chlorophenyl)-)	539-03-7	CH ₂ =C=O CH ₂ =C=O	10.76±0.03 10.11±0.03	EI EI	3483 3483
	C ₆ H ₅ CINHCOC ₂ H ₅ (Propanamide, N-(2-chlorophenyl)-)	2760-32-9		10.75	EI	4834
	C ₆ H ₅ CINHCOC ₂ H ₅ (Butanamide, N-(2-chlorophenyl)-)	33694-15-4		10.70	EI	4834
	C ₆ H ₅ CINHCOC(CH ₃) ₃ (Propanamide, N-(2-chlorophenyl)-2,2-dimethyl)-)	62662-74-2		10.70	EI	4834
	C ₆ H ₅ CINHCOC(CH ₃) ₃ (Butanamide, N-(2-chlorophenyl)-3,3-dimethyl-)	XXXXX-XX-X		10.45	EI	4834

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₆H₆NCl⁺						
	C ₆ H ₄ CINHCONH ₂ (Urea, (2-chlorophenyl)-)	114-38-5		9.15	EI	4834
	C ₆ H ₄ CINHCONHCH ₃ (Urea, N-(2-chlorophenyl)-N'-methyl-)	15500-96-6		10.20	EI	4834
	C ₆ H ₄ CINHCONHC ₂ H ₅ (Urea, N-(2-chlorophenyl)-N'-ethyl-)	62635-53-4		10.05	EI	4834
	C ₆ H ₄ CINHCONHCH(CH ₃) ₂ (Urea, N-(2-chlorophenyl)-N'-(1-methylethyl)-)	62635-47-6		9.80	EI	4834
	C ₆ H ₄ CINHCONHC(CH ₃) ₃ (Urea, N-(2-chlorophenyl)-N'-(1,1-dimethylethyl)-)	62635-48-7		9.70	EI	4834
	C ₆ H ₄ CINHCSCH ₃ (Ethanethioamide, N-(2-chlorophenyl)-)	39184-83-3		11.00	EI	4834
	C ₆ H ₄ CINHCSCH ₂ C(CH ₃) ₃ (Butanethioamide, N-(2-chlorophenyl)-3,3-dimethyl-)	62635-54-5		11.00	EI	4834
C₇H₁₂NCl⁺						
	C ₇ H ₁₂ NCl (1-Azabicyclo[2.2.2]octane, 4-chloro-)	5960-95-2	**	8.55±0.015 (V)	PE	4286
C₈H₁₀NCl⁺						
	C ₆ H ₄ CIN(CH ₃) ₂ (Benzenamine, 4-chloro-N,N-dimethyl-)	698-69-1	**	7.2±0.1	PE	4401
C₈H₁₄NCl⁺						
	C ₈ H ₁₄ NCl (9-Azabicyclo[3.3.1]nonane, 9-chloro-)	73322-95-9	**	8.55 (V)	PE	5091
	C ₈ H ₁₄ NCl (8-Azabicyclo[3.2.1]octane, 3-chloro-8-methyl- <i>endo</i> -)	13514-03-9	**	8.1±0.15	EI	5401
	C ₈ H ₁₄ NCl (8-Azabicyclo[3.2.1]octane, 3-chloro-8-methyl- <i>exo</i> -)	2292-12-8	**	8.3±0.15	EI	5401
C₉H₁₈NCl⁺						
	C ₅ H ₆ N(CH ₃) ₄ Cl (Piperidine, 1-chloro-2,2,6,6-tetramethyl-)	32579-76-3	**	7.64	PE	4278
C₁₃H₁₀NCl⁺						
	C ₆ H ₄ ClC(=CH ₂)C ₅ H ₄ N (Pyridine, 2-[1-(2-chlorophenyl)ethenyl]-)	XXXXX-XX-X	**	8.55	EI	5570
	C ₆ H ₄ ClC(=CH ₂)C ₅ H ₄ N (Pyridine, 2-[1-(4-chlorophenyl)ethenyl]-)	XXXXX-XX-X	**	8.58	EI	5570
C₁₆H₁₂NCl⁺						
	C ₆ H ₄ (Cl)C ₃ H ₃ (CN)C ₆ H ₅ (Cyclopropanecarbonitrile, 1-(<i>p</i> -chlorophenyl)-2-phenyl-)	32589-55-2	**	8.18±0.10	EI	3575
C₃₂H₂₁NCl⁺						
	C ₅ (C ₆ H ₅) ₂ (NC ₅ H ₅)(C ₁₀ H ₆)Cl (Cyclopenta-1,3-diene, 1,4-diphenyl-5-pyridinium-2,3-(naphtha-1,8-diy),chloride)	XXXXX-XX-X	**	6.75	CTS	5593
C₃₆H₂₃NCl⁺						
	C ₅ (C ₆ H ₅) ₂ (NC ₆ H ₇)(C ₁₀ H ₆)Cl (Cyclopenta-1,3-diene, 1,4-diphenyl-5-quinolinium-2,3-(naphtha-1,8-diy),chloride)	XXXXXX-XX-X	**	6.70	CTS	5593
	C ₅ (C ₆ H ₅) ₂ (NC ₆ H ₇)(C ₁₀ H ₆)Cl (Cyclopenta-1,3-diene, 1,4-diphenyl-5-isoquinolinium-2,3-(naphtha-1,8-diy),chloride)	XXXXX-XX-X	**	6.72	CTS	5593
C₆H₁₁N₂Cl⁺						
	C ₃ H ₂ N ₂ Cl(CH ₃) ₃ (3H-Pyrazole, 3-chloro-4,5-dihydro-3,5,5-trimethyl-)	55204-46-1	**	9.04 (V)	PE	4429
C₇H₅N₂Cl⁺						
	C ₆ H ₄ CN ₂ HCl (1H-Indazole, 3-chloro-)	29110-74-5	**	8.41 (V)	PE	5396

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_9\text{H}_{10}\text{N}_2\text{Cl}^+$	$\text{C}_6\text{H}_5(\text{Cl})\text{N}=\text{CHN}(\text{CH}_3)_2$ (Methanimidamide, N' -(2-chlorophenyl)- N,N -dimethyl-)	2103-49-3	H	8.9	EI	4337
$\text{C}_9\text{H}_{11}\text{N}_2\text{Cl}^+$	$\text{C}_6\text{H}_5(\text{Cl})\text{N}=\text{CHN}(\text{CH}_3)_2$ (Methanimidamide, N' -(2-chlorophenyl)- N,N -dimethyl-)	2103-49-3	**	7.3±0.1	EI	4359
			**	7.3	EI	4337
$\text{C}_{10}\text{H}_{13}\text{N}_2\text{Cl}^+$	$\text{C}_6\text{H}_5(\text{Cl})(\text{CH}_3)\text{N}=\text{CHN}(\text{CH}_3)_2$ (Methanimidamide, N' -(2-chloro-4-methylphenyl)- N,N -dimethyl-)	53666-35-6	**	7.1±0.1	EI	4359
	$\text{C}_6\text{H}_5(\text{Cl})(\text{CH}_3)\text{N}=\text{CHN}(\text{CH}_3)_2$ (Methanimidamide, N' -(2-chloro-5-methylphenyl)- N,N -dimethyl-)	53666-41-4	**	7.1±0.1	EI	4359
$\text{C}_{12}\text{H}_9\text{N}_2\text{Cl}^+$	$\text{C}_6\text{H}_5\text{NNC}_6\text{H}_4\text{Cl}$ (Diazine,(4-chlorophenyl)phenyl-(E)-)	6141-95-3	**	8.55±0.05 (V)	PE	5320
$\text{C}_{33}\text{H}_{20}\text{N}_2\text{Cl}^+$	$\text{C}_5(\text{C}_6\text{H}_5)_2(\text{NC}_5\text{H}_4\text{CN})(\text{C}_{10}\text{H}_6)\text{Cl}$ (Cyclopenta-1,3-diene,1,4-diphenyl-5-(4-cyanopyridinium)-2,3-(naphtha-1,8-diyl),chloride)	XXXXX-XX-X	**	6.72	CTS	5593
$\text{C}_2\text{H}_2\text{N}_3\text{Cl}^+$	$\text{C}_2\text{H}_2\text{N}_3\text{Cl}$ (1H-1,2,4-Triazole,3-chloro-)	6818-99-1	**	10.1 (V)	PE	5228
$\text{C}_3\text{H}_4\text{N}_3\text{Cl}^+$	$\text{C}_2\text{N}_3\text{Cl}(\text{CH}_3)$ (1H-1,2,4-Triazole,3-chloro-5-methyl-)	15285-15-1	**	9.6 (V)	PE	5228
	$\text{C}_2\text{H}_2\text{N}_3\text{Cl}(\text{CH}_3)$ (1H-1,2,4-Triazole,3-chloro-1-methyl-)	56616-92-3	**	9.7 (V)	PE	5228
	$\text{C}_2\text{H}_2\text{N}_3\text{Cl}(\text{CH}_3)$ (1H-1,2,4-Triazole,5-chloro-1-methyl-)	56616-99-0	**	9.75 (V)	PE	5228
	$\text{C}_2\text{H}_2\text{N}_3\text{Cl}(\text{CH}_3)$ (4H-1,2,4-Triazole,3-chloro-4-methyl-)	56616-87-6	**	9.8 (V)	PE	5228
$\text{C}_4\text{H}_6\text{N}_3\text{Cl}^+$	$\text{C}_2\text{N}_3\text{Cl}(\text{CH}_3)_2$ (1H-1,2,4-Triazole,3-chloro-1,5-dimethyl-)	56616-94-5	**	9.4 (V)	PE	5228
	$\text{C}_2\text{N}_3\text{Cl}(\text{CH}_3)_2$ (1H-1,2,4-Triazole,5-chloro-1,3-dimethyl-)	56616-97-8	**	9.35 (V)	PE	5228
	$\text{C}_2\text{N}_3\text{Cl}(\text{CH}_3)_2$ (4H-1,2,4-Triazole,3-chloro-4,5-dimethyl-)	56616-85-4	**	9.3 (V)	PE	5228
$\text{C}_6\text{H}_4\text{N}_3\text{Cl}^+$	$\text{C}_4\text{H}_2\text{N}_2\text{ClC}_2\text{H}_2\text{N}$ (Imidazo[1,2- <i>b</i>]pyridazine,6-chloro-)	6775-78-6	**	8.55 (V)	PE	5396
$\text{C}_9\text{H}_{10}\text{N}_3\text{Cl}^+$	$\text{C}_6\text{H}_5\text{ClNC}_3\text{H}_4\text{N}_2\text{H}_2$ (Imidazolidine,2-(2-chlorophenylimino)-)	XXXXX-XX-X	**	7.96 (V)	PE	5545
$\text{C}_{11}\text{H}_{10}\text{N}_3\text{Cl}^+$	$\text{C}_6\text{H}_5(\text{Cl})(\text{N}(\text{CH}_3)_2)\text{N}=\text{CHN}(\text{CH}_3)_2$ (Methanimidamide, N' -[2-chloro-4-(dimethylamino)phenyl]- N,N -dimethyl-)	53666-30-1	**	6.4±0.1	EI	4359
	$\text{C}_6\text{H}_5(\text{Cl})(\text{N}(\text{CH}_3)_2)\text{N}=\text{CHN}(\text{CH}_3)_2$ (Methanimidamide, N' -[2-chloro-5-(dimethylamino)phenyl]- N,N -dimethyl-)	53666-39-0	**	6.4±0.1	EI	4359
$\text{C}_{12}\text{H}_8\text{N}_3\text{Cl}^+$	$\text{C}_4\text{H}_2\text{N}_2\text{ClC}_2\text{H}_2\text{N}$ (Imidazo[1,2- <i>b</i>]pyridazine,6-chloro-2-phenyl-)	1844-53-7	**	8.09 (V)	PE	5396

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{CH}_3\text{NCl}_2^+$	CH_3NCl_2	7651-91-4	** **	9.35 ± 0.02 10.06 ± 0.10 (V)	PE PE	4737 4741
$\text{C}_5\text{H}_3\text{NCl}_2^+$	$\text{Cl}_2\text{C}_5\text{H}_3\text{N}$ (Pyridine, 3,5-dichloro-)	2457-47-8	**	9.88 (V)	PE	5527
$\text{C}_6\text{H}_5\text{NCl}_2^+$	$\text{C}_6\text{H}_5(\text{Cl})_2\text{NH}_2$ (Benzenamine, 2,6-dichloro-) $\text{C}_6\text{H}_5\text{Cl}_2\text{NHCOCH}_3$ (Acetamide, N-(2,4-dichlorophenyl)-) $\text{C}_6\text{H}_5\text{Cl}_2\text{NHCOCH}_3$ (Acetamide, N-(2,6-dichlorophenyl)-)	608-31-1 6975-29-7 17700-54-8	** $\text{CH}_2=\text{C=O}$ $\text{CH}_2=\text{C=O}$	7.60 ± 0.02 10.09 ± 0.03 9.98 ± 0.03	PE EI EI	3890 3480 3480
$\text{C}_8\text{H}_{10}\text{N}_2\text{Cl}_2^+$	$\text{C}_6\text{Cl}_2(\text{CH}_3)_2(\text{NH}_2)_2$ (1,4-Benzenediamine, 2,5-dichloro-3,6-dimethyl-)	40200-66-6	**	6.86 ± 0.03	PI	5552
$\text{C}_{10}\text{H}_{10}\text{N}_2\text{Cl}_2^+$	$\text{C}_6\text{H}_5(\text{Cl})_2\text{CH}_2\text{C}_5\text{H}_5\text{N}_2$ (1H-Imidazole, 2-[(2,6-dichlorophenyl)methyl]-4,5-dihydro-)	52115-81-8	**	8.42 (V)	PE	5096
$\text{C}_{11}\text{H}_{12}\text{N}_2\text{Cl}_2^+$	$\text{C}_{11}\text{H}_{12}\text{N}_2\text{Cl}_2$ (1H-Imidazole, 2-[(2,6-dichlorophenyl)methyl]-4,5-dihydro-1-methyl-)	65248-67-1	**	8.21 (V)	PE	5096
$\text{C}_9\text{H}_9\text{N}_3\text{Cl}_2^+$	$\text{C}_6\text{H}_5\text{Cl}_2\text{NC}_3\text{H}_4\text{N}_2\text{H}_2$ (Imidazolidine, 2-(2,6-dichlorophenylimino)-)	XXXXX-XX-X	**	8.01 (V)	PE	5545
$\text{C}_{11}\text{H}_{13}\text{N}_3\text{Cl}_2^+$	$\text{C}_6\text{H}_5\text{Cl}_2\text{NC}_3\text{H}_4\text{N}_2(\text{CH}_3)_2$ (Imidazolidine, 2-(2,6-dichlorophenylimino)-1,3-dimethyl-)	XXXXX-XX-X	**	7.84 (V)	PE	5545
$\text{BC}_5\text{H}_7\text{NCl}_2^+$	$\text{C}_5\text{H}_4\text{N}(\text{Cl})\cdot\text{BH}_3$ (Pyridine, 4-chloro-, compound with borane (1:1))	56898-52-3	**	9.71 (V)	PE	4536
$\text{BC}_4\text{H}_{12}\text{N}_2\text{Cl}_2^+$	$\text{B}(\text{N}(\text{CH}_3)_2)_2\text{Cl}$	6562-41-0	** **	8.15 (V) 8.08	PE PE	3704 3584
$\text{BC}_2\text{H}_6\text{NCl}_2^+$	$(\text{CH}_3)_2\text{NBCl}_2$	1113-31-1	** **	9.56 9.68 (V)	PE PE	3584 3704
$\text{B}_2\text{C}_4\text{H}_{12}\text{N}_2\text{Cl}_2^+$	$(\text{ClCH}_3\text{BNCH}_3)_2$	73775-16-3	**	9.48 (V)	PE	5628
$\text{B}_2\text{C}_3\text{H}_9\text{N}_3\text{Cl}_2^+$	$\text{N}_3\text{B}_2\text{Cl}_2(\text{CH}_3)_3$ (1,2,4,3,5-Triazadiborolidine, 3,5-dichloro-1,2,4-trimethyl-)	53246-09-6	**	8.22 (V)	PE	4526
$\text{B}_2\text{C}_4\text{H}_{12}\text{N}_4\text{Cl}_2^+$	$\text{B}_2\text{N}_4(\text{CH}_3)_4\text{Cl}_2$ (1,2,4,5,3,6-Tetrazadiborine, 3,6-dichlorohexahydro-1,2,4,5-tetramethyl-)	54196-15-5	**	7.61 (V)	PE	4299
$\text{B}_3\text{C}_3\text{H}_9\text{N}_3\text{Cl}_3^+$	$(\text{CH}_3)_3\text{B}_3\text{N}_3\text{Cl}_3$ (Borazine, 2,4,6-trichloro-1,3,5-trimethyl-)	703-86-6	**	9.45 (V)	PE	3943

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
OCl⁺ (³ S ⁻)	ClO	14989-30-1	**	11.01±0.01 (V)	PE	4944
O₂Cl⁺ (² A ₁) (¹ B ₂)	ClO ₂	10049-04-4	** ** **	10.36±0.02 10.5±0.1 (V) 15.5±0.1 (V)	PE PE PE	3499 3671 3671
OCl₂⁺ (² B ₁) (² B ₂) (² A ₁) (² A ₂) (² B ₁)	Cl ₂ O	7791-21-1	** ** ** ** ** ** **	11.02 (V) 12.37 (V) 12.65 (V) 12.79 (V) 15.9 (V) 15.90 (V) 16.6 (V)	PE PE PE PE PE PE PE	3694 3694 3694 3694 4763 3694 4763
HOCl⁺ (² A'') (² A') (² A'') (² A')	HOCl	7790-92-3	** ** ** **	11.12±0.01 12.09±0.01 14.6±0.1 (V) 15.6±0.1 (V)	PE PE PE PE	4763 4763 4763 4763
COCl⁺ (¹ S ⁺)	COCl ₂	75-44-5	Cl	11.2±0.2	PI	5041
COCl₂⁺	COCl ₂	75-44-5	** ** **	~11.2 11.55±0.02 11.84 (V)	PE PE PE	3726 3667 5041
C₂OCl₂⁺	Cl ₂ C=C=O	4591-28-0	**	9.07±0.02 (V)	PE	5030
C₂O₂Cl₂⁺	(COCl) ₂	79-37-8	** **	10.91±0.05 11.26 (V)	PE PE	4696 5549
C₄O₂Cl₂⁺	C ₄ (Cl) ₂ (=O) ₂ (3-Cyclobutene-1,2-dione, 3,4-dichloro-)	2892-63-9	**	9.89 (V)	PE	4861
C₂OCl₃⁺	(CCl ₃) ₂ CO	116-16-5		12.0	EI	3550
C₂OCl₄⁺	CCl ₃ COCl	76-02-8	**	11.31 (V)	PE	4547
C₆O₂Cl₄⁺	C ₆ O ₄ Cl ₂ (2,5-Cyclohexadiene, 1,4-dione, 2,3,5,6-tetrachloro-)	118-75-2	**	9.90±0.05 (V)	PE	5558
C₈O₃Cl₄⁺	C ₈ O ₃ Cl ₄ (1,3-Isobenzofurandione, 4,5,6,7-tetrachloro-)	117-08-8	**	10.77±0.2	EI	3583
C₂HOCl⁺	CHCl=C=O	29804-89-5	**	9.35 (V)	PE	5610
C₂H₃OCl⁺	CH ₃ COCl	75-36-5	** **	10.85±0.05 11.03 (V)	PE PE	4220 4513

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₂H₃OCl⁺						
	CH ₃ COCl	75-36-5	**	11.03 (V)	PE	4547
	CH ₂ ClCHO	107-20-0	**	10.61 (V)	PE	4513
			**	10.61 (V)	PE	4547
C₂H₅OCl⁺						
	CH ₂ ClCH ₂ OH	107-07-3	**	10.90 (V)	PE	5088
C₂H₇OCl⁺						
	(CH ₃) ₂ O·HCl	XXXXX-XX-X	**	10.6±0.2 (V)	PE	4774
C₃H₅OCl⁺						
	CH ₃ COCH ₂ Cl	78-95-5	**	9.91±0.03	PI	3765
			**	9.93±0.02 (V)	PE	4524
	C ₂ H ₃ OCH ₂ Cl (Oxirane, (chloromethyl)-)	106-89-8	**	10.60 (V)	PE	4747
C₃H₇OCl⁺						
	CH ₂ ClCH ₂ OCH ₃	627-42-9	**	10.05 (V)	PE	5088
C₆H₄OCl⁺						
	C ₆ H ₄ ClOCH ₃ (Benzene, 1-chloro-3-methoxy-)	2845-89-8	CH ₃	11.89±0.1	EI	3446
	C ₆ H ₄ ClOCH ₃ (Benzene, 1-chloro-4-methoxy-)	623-12-1	CH ₃	11.84±0.1	EI	3446
	C ₆ H ₄ ClNO ₂ (Benzene, 1-chloro-3-nitro-)	121-73-3	NO	10.31±0.1	EI	3447
	C ₆ H ₄ ClNO ₂ (Benzene, 1-chloro-4-nitro-)	100-00-5	NO	10.61±0.1	EI	3447
C₆H₅OCl⁺						
	C ₆ H ₄ ClOOCH ₃ (Acetic acid, 2-chlorophenyl ester)	4525-75-1	CH ₂ =C=O	9.19±0.03	EI	3483
	C ₆ H ₄ ClOOCH ₃ (Acetic acid, 3-chlorophenyl ester)	13031-39-5	CH ₂ =C=O	10.11±0.2	EI	3484
	C ₆ H ₄ ClOOCH ₃ (Acetic acid, 4-chlorophenyl ester)	876-27-7	CH ₂ =C=O	9.60±0.03	EI	3483
			CH ₂ =C=O	10.17±0.2	EI	3484
C₇H₄OCl⁺						
	ClC ₆ H ₄ COCH ₃ (Ethanone, 1-(4-chlorophenyl))	99-91-2	CH ₃	10.34±0.03	EI	5059
	C ₆ H ₅ COCl ₂ (Methanone, (2-chlorophenyl)phenyl-)	5162-03-8		11.4±0.1	EI	4335
				11.4±0.1	EI	4358
	C ₆ H ₅ COCl ₂ (Methanone, (3-chlorophenyl)phenyl-)	1016-78-0	**	11.75±0.1	EI	4335
				11.75±0.1	EI	4358
	C ₆ H ₅ COCl ₂ (Methanone, (4-chlorophenyl)phenyl-)	134-85-0		11.65±0.1	EI	4335
			**	11.65±0.2	EI	4358
				11.65±0.2	EI	4335
C₇H₅OCl⁺						
	C ₆ H ₅ COCl (Benzoyl chloride)	98-88-4	**	9.85	EI	3792
C₇H₇OCl⁺						
	C ₆ H ₅ ClOCH ₃ (Benzene, 1-chloro-3-methoxy-)	2845-89-8	**	8.72±0.1	EI	3446
	C ₆ H ₅ ClOCH ₃ (Benzene, 1-chloro-4-methoxy-)	623-12-1	**	8.18	EI	3845
			**	8.52±0.1	EI	3446

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₇H₁₁OCl⁺	C ₆ H ₈ ClO(CH ₃) (Cyclohexanone, 2-chloro-2-methyl-)	10409-46-8	**	9.41	PE	5085
C₈H₇OCl⁺	C ₆ H ₄ (Cl)COCH ₃ (Ethanone, 1-(4-chlorophenyl)-)	99-91-2	**	9.60±0.05 (V)	PE	5097
C₈H₉OCl⁺	C ₆ H ₅ OCH ₂ CH ₂ Cl (Benzene, 2-chloroethoxy-)	622-86-6	**	8.50	EI	5083
C₁₀H₁₇OCl⁺	C ₆ H ₈ ClO(<i>tert</i> -C ₄ H ₉) (Cyclohexanone, cis-2-chloro-4-(1,1-dimethylethyl)-)	16508-33-1	**	9.48	PE	5085
C₁₃H₉OCl⁺	C ₆ H ₅ COC ₆ H ₄ Cl (Methanone, (2-chlorophenyl)phenyl-)	5162-03-8	**	9.55±0.1	EI	4335
			**	9.55±0.1	EI	4358
	C ₆ H ₅ COC ₆ H ₄ Cl (Methanone, (3-chlorophenyl)phenyl-)	1016-78-0	**	9.55±0.1	EI	4335
			**	9.55±0.1	EI	4358
C₁₃H₁₁OCl⁺	C ₆ H ₅ COC ₆ H ₄ Cl (Methanone, (4-chlorophenyl)phenyl-)	134-85-0	**	9.6±0.1	EI	4335
			**	9.6±0.1	EI	4358
C₁₃H₁₁OCl⁺	C ₆ H ₅ CH ₂ OC ₆ H ₄ Cl (Benzene, 1-chloro-4-(phenylmethoxy)-)	7700-27-8	**	8.34	CTS	5336
C₂H₃O₂Cl⁺	CH ₂ ClCOOH	79-11-8	**	10.99 (V)	PE	3874
C₈H₇O₂Cl⁺	C ₆ H ₄ ClOOCH ₃ (Acetic acid, 2-chlorophenyl ester)	4525-75-1	**	8.67±0.03	EI	3483
	C ₆ H ₄ ClOOCH ₃ (Acetic acid, 3-chlorophenyl ester)	13031-39-5	**	8.83±0.2	EI	3484
	C ₆ H ₄ ClOOCH ₃ (Acetic acid, 4-chlorophenyl ester)	876-27-7	**	8.42±0.03	EI	3483
			**	8.79±0.2	EI	3484
C₄H₅O₃Cl⁺	C ₂ H ₅ O(CO) ₂ Cl	XXXXX-XX-X	**	10.77 (V)	PE	5549
C₂H₂OCl₂⁺	CHCl ₂ CHO	79-02-7	**	10.83 (V)	PE	4547
	CH ₂ ClCOCl	79-04-9	**	10.30 (V)	PE	4547
C₆H₄OCl₂⁺	C ₆ H ₃ (Cl) ₂ OH (Phenol, 2,6-dichloro-)	87-65-0	**	8.65±0.02	PE	3890
	C ₆ H ₃ Cl ₂ OOCCH ₃ (Phenol, 2,4-dichloro-, acetate)	6341-97-5	CH ₂ =C=O	9.37±0.03	EI	3480
	C ₆ H ₃ Cl ₂ OOCCH ₃ (Phenol, 2,6-dichloro-, acetate)	28165-71-1	CH ₂ =C=O	9.88±0.03	EI	3480
C₄H₂O₂Cl₂⁺	C ₄ H ₂ Cl ₂ O ₂ (2,5-Cyclohexadiene-1,4-dione, 2,5-dichloro-)	XXXXX-XX-X	**	10.24±0.03	PI	5505

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₈H₆O₂Cl₂⁺	C ₆ H ₅ Cl ₂ OOCCH ₃ (Phenol, 2,4-dichloro-, acetate) C ₆ H ₅ Cl ₂ OOCCH ₃ (Phenol, 2,6-dichloro-, acetate)	6341-97-5 28165-71-1	** **	8.16±0.03 8.68±0.03	EI EI	3480 3480
C₂HOCl₃⁺	CCl ₃ CHO CHCl ₂ COCl	75-87-6 79-36-7	** **	10.88 (V) 11.27 (V)	PE PE	4547 4547
C₆H₂O₂Cl₄⁺	C ₆ Cl ₄ (OH) ₂ (1,4-Benzenediol, 2,3,5,6-tetrachloro-)	87-87-6	**	8.30±0.05	PI	5552
NOCl⁺	(² A', ² A'') NOCl (² A', ² A'') (² A', ² A'', ² A')	2696-92-6	** ** **	10.87±0.01 10.90±0.5 10.94	PE PE PE	4422 4420 4404
NO₂Cl⁺	ClNO ₂	13444-90-1	**	11.84	PE	4404
CNOC_l⁺	CINCO	13858-09-8	**	10.72±0.01	PE	5001
C₈N₂O₂Cl₂⁺	C ₆ Cl ₂ O ₂ (CN) ₂ (1,4-Cyclohexadiene, 1,2-dicarbonitrile, 4,5-dichloro-3,6-dioxo-)	84-58-2	**	10.58±0.05 (V)	PE	5558
CNOC_l₃⁺	CCl ₃ NO	3711-49-7		10.30±0.05 (V)	PE	5298
C₅NOCl₅⁺	C ₅ N(O(Cl)) ₅ (Pyridine, pentachloro-, 1-oxide)	17573-93-2	**	8.72±0.02 (V)	PE	4275
C₈H₆NOCl⁺	C(CH ₃) ₂ (Cl)NO	2421-26-3	**	9.13±0.1 (V)	PE	4465
C₈H₈NOCl⁺	C ₂ H ₅ C(CH ₃)(Cl)NO	681-01-6	**	9.29±0.1 (V)	PE	4465
C₅H₄NOCl⁺	C ₅ H ₄ N(O)Cl (Pyridine, 4-chloro-, 1-oxide)	1121-76-2	**	8.42±0.02 (V)	PE	4275
C₆H₄NOCl⁺	C ₆ H ₄ (Cl)(NO) (Benzene, 1-chloro-4-nitroso-)	932-98-9	**	9.02±0.1 (V)	PE	4465
C₆H₁₀NOCl⁺	C ₆ H ₁₀ (Cl)(NO) (Cyclohexane, 1-chloro-1-nitroso-) trans-C ₆ H ₁₀ (Cl)(NO) (Cyclohexane, trans-1-chloro-2-nitroso-)	695-64-7 1809-72-9	** **	9.28 (V) 9.13 (V)	PE PE	4465 4465
C₇H₄NOCl⁺	C ₆ H ₄ (Cl)(C≡NO) (Benzonitrile, 4-chloro-N-oxide) C ₆ H ₄ (Cl)NCO (Benzene, 1-chloro-3-isocyanato-) C ₆ H ₄ (Cl)NCO (Benzene, 1-chloro-4-isocyanato-)	15500-74-0 2909-38-8 104-12-1	** ** **	8.65 (V) 9.0±0.1 (V) 8.8±0.1 (V)	PE PE PE	4719 5026 5026

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₇H₆NOCl⁺						
	C ₆ H ₄ (Cl)(CONH ₂) (Benzamide, 3-chloro-)	618-48-4	**	9.34 (V)	PE	4918
	C ₆ H ₄ (Cl)(CONH ₂) (Benzamide, 4-chloro-)	619-56-7	**	9.35 (V)	PE	4918
	C ₆ H ₄ (Cl)NHCHO (Formamide, N-(2-chlorophenyl)-)	2596-93-2	**	8.4±0.1	EI	4359
C₈H₇NOCl⁺						
	C ₆ H ₃ Cl ₂ NHCOCH ₃ (Acetamide, N-(2,4-dichlorophenyl)-)	6975-29-7		8.81±0.03	EI	3480
	C ₆ H ₃ Cl ₂ NHCOCH ₃ (Acetamide, N-(2,6-dichlorophenyl)-)	17700-54-8		8.79±0.03	EI	3480
C₈H₈NOCl⁺						
	C ₆ H ₄ ClNHCOCH ₃ (Acetamide, N-(2-chlorophenyl)-)	533-17-5	**	8.55	EI	4834
			**	8.07±0.03	EI	3483
	C ₆ H ₄ ClNHCOCH ₃ (Acetamide, N-(4-chlorophenyl)-)	539-03-7	**	8.07±0.03	EI	3483
	C ₆ H ₄ (Cl)(CH ₃)NHCHO (Formamide, N-(2-chloro-4-methylphenyl)-)	18931-77-6	**	8.1±0.1	EI	4359
	C ₆ H ₄ (Cl)(CH ₃)NHCHO (Formamide, N-(2-chloro-5-methylphenyl)-)	18931-82-3	**	8.2±0.1	EI	4359
C₉H₁₀NOCl⁺						
	C ₆ H ₄ ClNHCO ₂ H ₅ (Propanamide, N-(2-chlorophenyl)-)	2760-32-9	**	8.45±0.05	EI	4834
C₉H₁₆NOCl⁺						
	C ₅ H ₄ N(O(CH ₃) ₃)Cl (4-Piperidinone, 1-chloro-2,2,6,6-tetramethyl-)	38951-83-6	**	8.01	PE	4278
C₁₀H₁₂NOCl⁺						
	C ₆ H ₄ ClNHCOCH ₂ CH ₂ CH ₃ (Butanamide, N-(2-chlorophenyl)-)	33694-15-4	**	8.50±0.05	EI	4834
C₁₀H₁₄NOCl⁺						
	C ₁₀ H ₁₄ (Cl)(NO) (Tricyclo[3.3.1.1 ^{3,7}]decane, 2-chloro-2-nitroso-)	33673-34-6	**	9.02 (V)	PE	4465
C₁₁H₁₄NOCl⁺						
	C ₁₁ H ₁₄ NOCl (Butanamide, N-(2-chlorophenyl)-3-methyl-)	62635-51-2	**	8.50±0.05	EI	4834
C₁₂H₈NOCl⁺						
	C ₆ H ₄ ClCOC ₅ H ₄ N (Methanone, (2-chlorophenyl)-2-pyridinyl-)	1694-57-1	**	8.98	EI	5459
C₁₂H₁₆NOCl⁺						
	C ₆ H ₄ ClNHCOCH ₂ C(CH ₃) ₃ (Butanamide, N-(2-chlorophenyl)-3,3-dimethyl-)	XXXXX-XX-X	**	8.40±0.05	EI	4834
C₁₇H₁₄NOCl⁺						
	C ₆ H ₄ (Cl)C ₃ H ₃ (CN)C ₆ H ₄ (OCH ₃) (Cyclopropanecarbonitrile, 1-(p-chlorophenyl)-2-(p-methoxyphenyl)-)	32589-54-1	**	7.70±0.05	EI	3575
C₄H₃N₂OCl⁺						
	C ₄ H ₃ N ₂ Cl(=O) (2(1H)-Pyrimidinone, 5-chloro-)	54326-16-8	**	9.78±0.05	EI	5159

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₅H₅N₂OCl⁺						
	C ₅ H ₃ N(O)(Cl)NH ₂ (2-Pyridinamine, 5-chloro-, 1-oxide)	52132-34-0	**	7.98±0.05	EI	4117
	C ₄ H ₂ N ₂ ClOCH ₃ (Pyrimidine, 5-chloro-2-methoxy-)	38373-44-3	**	9.36±0.05	EI	5159
	C ₄ H ₂ N ₂ Cl(=O)CH ₃ (2(1H)-Pyrimidinone, 5-chloro-1-methyl-)	63331-06-6	**	9.03±0.05	EI	5159
C₆H₇N₂OCl⁺						
	C ₅ H ₃ N(O)(Cl)NHCH ₃ (2-Pyridinamine, 5-chloro-N-methyl-, 1-oxide)	54818-75-6	**	7.61±0.05	EI	4117
	C ₅ H ₃ N(Cl)(=NH)OCH ₃ (2(1H)-Pyridinimine, 5-chloro-1-methoxy-)	54818-77-8	**	7.40±0.05	EI	4117
C₇H₇N₂OCl⁺						
	C ₆ H ₄ CINHCONH ₂ (Urea, (2-chlorophenyl)-)	114-38-5	**	8.45	EI	4834
C₈H₅N₂OCl⁺						
	ClC ₆ H ₄ C(=O)CHN ₂ (Ethanone, 1-(2-chlorophenyl)-2-diazo-)	XXXXX-XX-X	**	8.60±0.05 (V)	PE	5326
	ClC ₆ H ₄ C(=O)CHN ₂ (Ethanone, 1-(4-chlorophenyl)-2-diazo-)	3282-33-5	**	9.02±0.05 (V)	PE	5326
C₈H₉N₂OCl⁺						
	C ₆ H ₄ CINHCONHCH ₃ (Urea, N-(2-chlorophenyl)-N'-methyl-)	15500-96-6	**	8.35±0.05	EI	4834
C₉H₁₁N₂OCl⁺						
	C ₆ H ₃ (Cl)(N(CH ₃) ₂)NHCHO (Formamide, N-[2-chloro-5-(dimethylamino)phenyl]-)	53666-46-9	**	6.7±8.1	EI	4359
	C ₆ H ₄ CINHCONHC ₂ H ₅ (Urea, N-(2-chlorophenyl)-N'-ethyl-)	62635-53-4	**	8.25±0.05	EI	4834
C₁₀H₁₃N₂OCl⁺						
	C ₆ H ₃ (Cl)(OCH ₃)N=CHN(CH ₃) ₂ (Methanimidamide, N'-(2-chloro-4-methoxyphenyl)-N,N-dimethyl-)	53666-34-5	**	7.0±0.1	EI	4359
	C ₆ H ₃ (Cl)(OCH ₃)N=CHN(CH ₃) ₂ (Methanimidamide, N'-(2-chloro-5-methoxyphenyl)-N,N-dimethyl-)	53666-40-3	**	7.1±0.1	EI	4359
	C ₆ H ₄ CINHCONHCH(CH ₃) ₂ (Urea, N-(2-chlorophenyl)-N'-(1-methylethyl)-)	62635-47-6	**	8.15±0.05	EI	4834
C₁₁H₁₅N₂OCl⁺						
	C ₆ H ₄ CINHCONHC(CH ₃) ₃ (Urea, N-(2-chlorophenyl)-N'-(1,1-dimethylethyl)-)	62635-48-7	**	8.05±0.05	EI	4834
C₁₂H₇N₂OCl⁺						
	C ₁₂ H ₇ N ₂ OCl (Phenazine, 2-chloro-10-oxide)	1019-15-4	**	8.16 (V)	PE	4590
	C ₁₂ H ₇ N ₂ OCl (Phenazine, 2-chloro-5-oxide)	1211-09-2	**	8.20 (V)	PE	4590
C₄H₄NO₂Cl⁺						
	C ₄ H ₄ N(=O) ₂ (Cl) (2,5-Pyrrolidinedione, 1-chloro-)	128-09-6	**	10.29 (V)	PE	4742
			**	10.29 (V)	PE	4810
C₅H₈NO₂Cl⁺						
	C ₃ H ₂ NO(=O)(Cl)(CH ₃) ₂ (2-Oxazolidinone, 3-chloro-4,4-dimethyl-)	58629-01-9	**	9.68 (V)	PE	4742

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₂F₂Cl⁺						
	C ₂ F ₂ Cl	79-38-9	F	15.9±0.2	EI	4070
	CFCl=CFCl	598-88-9	Cl	14.8±0.1	EI	4070
CF₃Cl⁺						
	CF ₃ Cl	75-72-9	**	12.39	PI	4757
			**	12.45	PI	5196
			**	13.0 (V)	PE	3914
			**	13.08±0.01 (V)	PE	4916
			**	13.08±0.02 (V)	PE	4026
C₂F₃Cl⁺						
	C ₂ F ₃ Cl	79-38-9	**	9.76	S	3776
			**	9.82	PE	3589
			**	10.26 (V)	PE	4303
			**	10.6±0.1	EI	4070
C₃F₃Cl⁺						
	CF ₃ C≡CCl	673-93-8	**	11.14±0.02	PE	4765
C₂F₅Cl⁺						
	CF ₃ CF ₂ Cl	76-15-3	**	12.96 (V)	PE	4366
C₆F₅Cl⁺						
	C ₆ F ₅ Cl (Benzene, chloropentafluoro-)	344-07-0	**	9.72±0.02	PE	5305
			**	9.94 (V)	PE	5252
CFCl₂⁺						
	CF ₂ Cl ₂	75-71-8	F ⁻	12.07±0.05	PI	5399
			F	13.30±0.05	PI	5399
			F	13.81	PI	4757
			F	14.15	PI	5796
	CFCl=CFCl	598-88-9	CF	14.3±0.1	EI	4070
	CFCl ₃	75-69-4	Cl	11.57±0.04	PI	4757
			Cl	11.65	PI	5196
C₂FCl₂⁺						
	CFCl=CFCl	598-88-9	F	15.7±0.1	EI	4070
CF₂Cl₂⁺						
	CF ₂ Cl ₂	75-71-8	**	11.75±0.04	PI	4757
			**	11.75	PI	5196
			**	12.24±0.01 (V)	PE	4916
			**	12.3 (V)	PE	3914
			**	12.27±0.02 (V)	EI	4880
C₂F₂Cl₂⁺						
	CF ₂ =CCl ₂	79-35-6	**	9.62	PE	3589
			**	9.82±0.02 (V)	EI	4880
	CFCl=CFCl	598-88-9	**	10.2±0.1	EI	4070
C₂F₄Cl₂⁺						
	(CF ₂ Cl) ₂	76-14-2	**	12.47 (V)	PE	4613
			**	12.85 (V)	PE	4366
CFCl₃⁺						
	CFCl ₃	75-69-4	**	11.77±0.01	PE	4365
			**	11.85 (V)	PE	5196
			**	11.9 (V)	PE	3914
			**	11.76±0.01 (V)	PE	4916

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_2\text{F}_3\text{Cl}_3^+$	$\text{CFCl}_2\text{CF}_2\text{Cl}$	76-13-1	**	12.05 (V)	PE	4366
$\text{C}_6\text{F}_3\text{Cl}_3^+$	$\text{C}_6\text{F}_3\text{Cl}_3$ (Benzene, 1,3,5-trichloro-2,4,6-trifluoro-)	319-88-0	**	9.48 ± 0.02	PE	5305
CH_2FCl^+	CH_2FCl	593-70-4	**	11.74	PE	3914
C_2HFCl^+	$\text{CH}_2=\text{CFCl}$	2317-91-1	H	16.2 ± 0.2	EI	4070
$\text{C}_2\text{H}_2\text{FCl}^+$	$\text{CH}_2=\text{CFCl}$	2317-91-1	**	9.97	S	3776
			**	10.7 ± 0.2	EI	3539
			**	10.7 ± 0.2	EI	4070
$\text{C}_6\text{H}_4\text{FCl}^+$	$\text{C}_6\text{H}_4\text{FCl}$ (Benzene, 1-chloro-2-fluoro-)	348-51-6	**	9.16 (V)	PE	4567
			**	9.18 ± 0.02	PE	5305
	$\text{C}_6\text{H}_4\text{FCl}$ (Benzene, 1-chloro-3-fluoro-)	625-98-9	**	9.22 ± 0.02	PE	5305
	$\text{C}_6\text{H}_4\text{FCl}$ (Benzene, 1-chloro-4-fluoro-)	352-33-0	**	9.25 (V)	PE	4567
			**	9.05 (V)	PE	4567
			**	9.08 ± 0.02	PE	5305
CHF_2Cl^+	CHF_2Cl	75-45-6	**	12.6 (V)	PE	3914
$\text{C}_2\text{HF}_2\text{Cl}^+$	$\text{CF}_2=\text{CHCl}$	359-10-4	**	9.76	S	3776
$\text{C}_2\text{H}_3\text{F}_2\text{Cl}^+$	$\text{CH}_3\text{CF}_2\text{Cl}$	75-68-3	**	12.50 (V)	PE	4366
$\text{C}_6\text{H}_3\text{F}_2\text{Cl}^+$	$\text{C}_6\text{H}_3\text{F}_2\text{Cl}$ (Benzene, 1-chloro-2,4-difluoro-)	1435-44-5	**	9.17 ± 0.02	PE	5305
	$\text{C}_6\text{H}_3\text{F}_2\text{Cl}$ (Benzene, 1-chloro-3,5-difluoro-)	1435-43-4	**	9.40 ± 0.02	PE	5305
	$\text{C}_6\text{H}_3\text{F}_2\text{Cl}$ (Benzene, 2-chloro-1,3-difluoro-)	38361-37-4	**	9.37 ± 0.02	PE	5305
	$\text{C}_6\text{H}_3\text{F}_2\text{Cl}$ (Benzene, 2-chloro-1,4-difluoro-)	2367-91-1	**	9.19 ± 0.02	PE	5305
	$\text{C}_6\text{H}_3\text{F}_2\text{Cl}$ (Benzene, 4-chloro-1,2-difluoro-)	696-02-6	**	9.24 ± 0.02	PE	5305
$\text{C}_6\text{H}_2\text{F}_3\text{Cl}^+$	$\text{C}_6\text{H}_2\text{F}_3\text{Cl}$ (Benzene, 1-chloro-2,3,4-trifluoro-)	36556-42-0	**	9.42 ± 0.02	PE	5305
	$\text{C}_6\text{H}_2\text{F}_3\text{Cl}$ (Benzene, 1-chloro-2,4,5-trifluoro-)	XXXXX-XX-X	**	9.27 ± 0.02	PE	5305
	$\text{C}_6\text{H}_2\text{F}_3\text{Cl}$ (Benzene, 2-chloro-1,3,4-trifluoro-)	39153-73-6	**	9.39 ± 0.02	PE	5305
$\text{C}_7\text{H}_4\text{F}_3\text{Cl}^+$	$\text{C}_6\text{H}_4\text{ClCF}_3$ (Benzene, 1-chloro-2-trifluoromethyl-)	88-16-4	**	9.47 (V)	PE	4567
	$\text{C}_6\text{H}_4\text{ClCF}_3$ (Benzene, 1-chloro-3-trifluoromethyl-)	98-15-7	**	9.50 (V)	PE	4567

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_7\text{H}_4\text{F}_3\text{Cl}^+$	$\text{C}_6\text{H}_4\text{ClCF}_3$ (Benzene, 1-chloro-4-trifluoromethyl-)	98-56-6	**	9.56 (V)	PE	4567
$\text{C}_6\text{HF}_3\text{Cl}^+$	$\text{C}_6\text{HF}_3\text{Cl}$ (Benzene, 3-chloro-1,2,4,5-tetrafluoro-)	1835-61-6	**	9.58±0.02	PE	5305
CHFCl_2^+	CHFCl_2	75-43-4	**	12.0 (V)	PE	3914
$\text{C}_6\text{H}_3\text{FCl}_2^+$	$\text{C}_6\text{H}_3\text{FCl}_2$ (Benzene, 1,2-dichloro-3-fluoro-) $\text{C}_6\text{H}_3\text{FCl}_2$ (Benzene, 1,2-dichloro-4-fluoro-) $\text{C}_6\text{H}_3\text{FCl}_2$ (Benzene, 1,3-dichloro-2-fluoro-) $\text{C}_6\text{H}_3\text{FCl}_2$ (Benzene, 1,3-dichloro-5-fluoro-) $\text{C}_6\text{H}_3\text{FCl}_2$ (Benzene, 1,4-dichloro-2-fluoro-) $\text{C}_6\text{H}_3\text{FCl}_2$ (Benzene, 2,4-dichloro-1-fluoro-)	36556-50-0 1435-49-0 2268-05-5 1435-46-7 348-59-4 1435-48-9	** ** ** ** ** **	9.29±0.02 9.16±0.02 9.32±0.02 9.39±0.02 9.09±0.02 9.12±0.02	PE	5305 5305 5305 5305 5305 5305
$\text{C}_6\text{H}_2\text{F}_2\text{Cl}_2^+$	$\text{C}_6\text{H}_2\text{F}_2\text{Cl}_2$ (Benzene, 1,2-dichloro-3,4-difluoro-) $\text{C}_6\text{H}_2\text{F}_2\text{Cl}_2$ (Benzene, 1,3-dichloro-2,4-difluoro-) $\text{C}_6\text{H}_2\text{F}_2\text{Cl}_2$ (Benzene, 1,3-dichloro-2,5-difluoro-) $\text{C}_6\text{H}_2\text{F}_2\text{Cl}_2$ (Benzene, 1,4-dichloro-2,5-difluoro-) $\text{C}_6\text{H}_2\text{F}_2\text{Cl}_2$ (Benzene, 2,3-dichloro-1,4-difluoro-)	36556-39-5 36556-37-3 2367-80-8 XXXXX-XX-X 36556-54-4	** ** ** ** **	9.33±0.02 9.27±0.02 9.32±0.02 9.17±0.02 9.32±0.02	PE	5305 5305 5305 5305 5305
$\text{C}_2\text{HF}_3\text{Cl}_2^+$	$\text{CF}_2\text{ClCHFCl}$	354-23-4	**	12.00 (V)	PE	4366
$\text{C}_6\text{HF}_3\text{Cl}_2^+$	$\text{C}_6\text{HF}_3\text{Cl}_2$ (Benzene, 2,4-dichloro-1,3,5-trifluoro-)	2368-53-8	**	9.37±0.02	PE	5305
$\text{C}_6\text{H}_2\text{FCl}_3^+$	$\text{C}_6\text{H}_2\text{FCl}_3$ (Benzene, 1,2,3-trichloro-4-fluoro-) $\text{C}_6\text{H}_2\text{FCl}_3$ (Benzene, 1,2,4-trichloro-5-fluoro-) $\text{C}_6\text{H}_2\text{FCl}_3$ (Benzene, 1,3,5-trichloro-2-fluoro-) $\text{C}_6\text{H}_2\text{FCl}_3$ (Benzene, 2,3,5-trichloro-1-fluoro-)	36556-36-2 XXXXX-XX-X 36556-33-9 3107-20-8	** ** ** **	9.20±0.02 9.16±0.02 9.23±0.02 9.24±0.02	PE	5305 5305 5305 5305
$\text{C}_6\text{HFCl}_4^+$	C_6HFCl_4 (Benzene, 1,2,3,4-tetrachloro-5-fluoro-) C_6HFCl_4 (Benzene, 1,2,4,5-tetrachloro-3-fluoro-)	2691-93-2 319-97-1	** **	9.20±0.02 9.19±0.02	PE	5305 5305
O_3FCl^+	ClO_3F	7616-94-6	**	12.945±0.005	PE	3675

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_3\text{OF}_5\text{Cl}^+$	$\text{CCl}_2\text{COCF}_3$	79-53-8	**	11.71 ± 0.02 (V)	PE	4524
$\text{C}_3\text{OF}_3\text{Cl}_3^+$	$\text{CCl}_2\text{FCOCClF}_2$	79-52-7	**	11.21 ± 0.02 (V)	PE	4524
	CF_3COCl_3	758-42-9	**	11.24 ± 0.02 (V)	PE	4524
$\text{C}_3\text{HOF}_4\text{Cl}^+$	$\text{CCl}_2\text{COCHF}_2$	920-64-9	**	11.33 ± 0.02 (V)	PE	4524
CNOF_2Cl^+	CF_2ClNO	421-13-6		10.81 ± 0.05 (V)	PE	5298
CNOCl_2^+	CFCl_2NO	1495-28-9		10.58 ± 0.05 (V)	PE	5298
NaCl^+ $(^2\text{P}_{3/2})$	NaCl	7647-14-5	**	8.93 ± 0.1	PE	4344
			**	8.93 ± 0.1	PE	5035
			**	9.0 (V)	PE	4307
			**	9.80 ± 0.04 (V)	PE	5035
Na_2Cl_2^+	$(\text{NaCl})_2$	12258-98-9	**	10.30 (V)	PE	4344
			**	10.30 (V)	PE	5035
MgCl_2^+	MgCl_2	7786-30-3	**	10.5 (V)	PE	4761
AlCl^+	AlCl	13595-81-8	**	9.4	PE	4860
AlCl_3^+	AlCl_3	7446-70-0	**	12.01 (V)	PE	4398
			**	12.01 (V)	PE	4256
Al_2Cl_6^+	$(\text{AlCl}_3)_2$	13845-12-0	**	12.18 (V)	PE	4559
			**	12.18 (V)	PE	4256
$\text{C}_2\text{H}_6\text{AlCl}^+$	$(\text{CH}_3)_2\text{ClAl}$	1184-58-3	**	10.25 (V)	PE	4398
$\text{C}_4\text{H}_{12}\text{Al}_2\text{Cl}_2^+$	$((\text{CH}_3)_2\text{ClAl})_2$	12073-96-0	**	10.09 (V)	PE	4559
$\text{H}_4\text{NAlCl}_4^+$	NH_4AlCl_4	7784-14-7	**	10.56 ± 0.06 (V)	PE	5238
OAlCl^+	AlOCl	13596-11-7	**	12 ± 1	EI	3462
SiCl^+	SiCl_2 SiCl_3 $\text{Cl}_3\text{SiCo}(\text{Co})_2(\text{PF}_3)_2$ $\text{Cl}_3\text{SiCo}(\text{CO})_3\text{PF}_3$	13569-32-9 35880-05-8 37769-29-2 37769-28-1	Cl	12.50 ± 0.10	EI	5188
				19.20 ± 0.10	EI	5188
				16.4 ± 0.5	EI	3653
				16.2 ± 0.5	EI	3653
SiCl_2^+	SiCl_2 SiCl_4	13569-32-9 35880-05-8	**	10.93 ± 0.10	EI	5188
				17.64 ± 0.10	EI	5188

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
SiCl₃⁺						
	SiCl ₄	15056-28-7	Cl	12.6±0.1	EI	5276
	Si ₂ Cl ₆	13465-77-5	SiCl ₃	11.4±0.1	EI	5276
	SiHCl ₃	10025-78-2	H	11.9±0.1	EI	5276
	C ₆ H ₅ SSiCl ₃ (Silane, trichloro(phenylthio)-)	7579-91-1		11.43±0.1	EI	4198
SiCl₄⁺						
	SiCl ₄	10026-04-7	**	12.06 (V)	PE	3514
			**	11.8	PE	5276
			**	11.44±0.10	EI	5188
Si₂Cl₆⁺						
	Si ₂ Cl ₆	13465-77-5	**	10.4	PE	5276
			**	11.0±0.3	EI	5188
H₃SiCl⁺						
	SiH ₃ Cl	13465-78-6	**	11.61±0.02 (V)	PE	3510
			**	11.61±0.05 (V)	PE	3502
			**	11.65 (V)	PE	3511
H₂SiCl₂⁺						
	SiH ₂ Cl ₂	4109-96-0	**	11.64±0.02 (V)	PE	3510
			**	11.70 (V)	PE	3511
			**	11.70 (V)	PE	3694
HSiCl₃⁺						
	SiHCl ₃	10025-78-2	**	11.94 (V)	PE	3511
			**	11.94 (V)	PE	4146
C₂H₆SiCl⁺						
	C ₆ H ₅ SSi(CH ₃) ₂ Cl (Silane, chlorodimethyl(phenylthio)-)	52548-11-5		10.79±0.1	EI	4198
C₃H₉SiCl⁺						
	(CH ₃) ₃ SiCl	75-77-4	**	10.76 (V)	PE	3503
			**	10.84 (V)	PE	4683
			**	10.0	PE	5276
C₄H₉SiCl⁺						
	C ₃ H ₆ Si(Cl)CH ₃ (Silacyclobutane, 1-chloro-1-methyl-)	2351-34-0	**	9.95 (V)	PE	4077
C₄H₁₁SiCl⁺						
	(CH ₃) ₃ SiCH ₂ Cl	2344-80-1	**	10.17±0.1 (V)	PE	3830
C₅H₉SiCl⁺						
	(CH ₃) ₃ SiC≡CCl	7652-06-4	**	9.4±0.1	PE	4002
C₆H₁₅SiCl⁺						
	(tert-C ₄ H ₉ Si(CH ₃) ₂ Cl)	18162-48-6	**	9.77 (V)	PE	4683
C₇H₉SiCl⁺						
	C ₆ H ₄ (Cl)SiH(CH ₃) ₂ (Silane, (3-chlorophenyl)dimethyl-)	2083-13-8	CH ₃	8.90	EI	4125
	C ₆ H ₄ (Cl)SiH(CH ₃) ₂ (Silane, (4-chlorophenyl)dimethyl-)	1432-31-1	CH ₃	8.84	EI	4125
C₉H₁₃SiCl⁺						
	ClC ₆ H ₄ Si(CH ₃) ₃ (Silane,(4-chlorophenyl)trimethyl-)	10557-71-8	**	9.01 (V)	PE	5380
			**	9.03 (V)	PE	4438

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{10}\text{H}_{17}\text{Si}_2\text{Cl}^+$	$\text{C}_6\text{H}_4(\text{SiCl}(\text{CH}_3)_2)\text{SiH}(\text{CH}_3)_2$ (Silane, chloro[3-(dimethylsilyl)phenyl]dimethyl-) $\text{C}_6\text{H}_4(\text{SiCl}(\text{CH}_3)_2)\text{SiH}(\text{CH}_3)_2$ (Silane, chloro[4-(dimethylsilyl)phenyl]dimethyl-)	34259-70-6 17873-29-9	**	8.5 ± 0.2 8.6 ± 0.2	EI	4121 4121
$\text{CH}_3\text{SiCl}_2^+$	$\text{CH}_3\text{SiHCl}_2$	20156-50-7	**	11.47	S	5183
$\text{C}_2\text{H}_6\text{SiCl}_2^+$	$(\text{CH}_3)_2\text{SiCl}_2$	75-78-5	**	10.99 (V)	PE	3503
$\text{C}_3\text{H}_6\text{SiCl}_2^+$	$\text{C}_3\text{H}_6\text{SiCl}_2$ (Silacyclobutane, 1,1-dichloro-)	2351-33-9	**	10.50 (V)	PE	4077
$\text{C}_4\text{H}_6\text{SiCl}_2^+$	$\text{C}_4\text{H}_6\text{SiCl}_2$ (Silacyclopent-3-ene, 1,1-dichloro-)	XXXXX-XX-X	**	9.63 (V)	PE	4517
$\text{C}_8\text{H}_{18}\text{Si}_2\text{Cl}_2^+$	$\text{C}_8\text{H}_{18}\text{Si}_2\text{Cl}_2$	65411-94-1	**	8.96 (V)	PE	4715
$\text{C}_9\text{H}_{14}\text{Si}_2\text{Cl}_2^+$	$\text{C}_6\text{H}_4(\text{SiCl}_2\text{CH}_3)_2\text{SiH}(\text{CH}_3)_2$ (Silane, 1,1-dichloro[3-(dimethylsilyl)phenyl]methyl-)	34259-71-7	**	8.6 ± 0.2	EI	4121
$\text{C}_5\text{H}_5\text{SiCl}_3^+$	$\text{C}_5\text{H}_5(\text{SiCl}_3)$ (Silane, 5-trichloro-2,4-cyclopentadien-1-yl-)	13688-63-6	**	9.0 (V)	PE	4373
$\text{C}_8\text{H}_{11}\text{Si}_2\text{Cl}_3^+$	$\text{C}_6\text{H}_4(\text{SiCl}_3)\text{SiH}(\text{CH}_3)_2$ (Silane, trichloro[3-(dimethylsilyl)phenyl]-) $\text{C}_6\text{H}_4(\text{SiCl}_3)\text{SiH}(\text{CH}_3)_2$ (Silane, trichloro[4-(dimethylsilyl)phenyl]-)	34259-72-8 XXXXX-XX-X	** **	9.1 ± 0.2 9.3 ± 0.2	EI EI	4121 4121
$\text{C}_6\text{H}_{12}\text{Si}_4\text{Cl}_4^+$	$\text{C}_6\text{H}_{12}\text{Si}_4\text{Cl}_4$ (1,3,5,7-Tetrasilatricyclo[3.3.1.1 ^{3,7}]decane, 1,3,5,7-tetrachloro-)	18222-89-4	**	9.4 ± 0.05	PE	3855
$\text{C}_4\text{H}_{12}\text{N}_2\text{SiCl}_2^+$	$((\text{CH}_3)_2\text{N})_2\text{SiCl}_2$	13328-30-8	**	8.81 (V)	PE	3503
$\text{C}_2\text{H}_6\text{NSiCl}_3^+$	$((\text{CH}_3)_2\text{N})\text{SiCl}_3$	13307-04-5	**	9.30 (V)	PE	3503
$\text{C}_9\text{H}_{13}\text{OSiCl}_3^+$	$\text{ClC}_6\text{H}_4\text{Si}(\text{CH}_3)_2\text{OCH}_3$ (Silane,(3-chlorophenyl)methoxydimethyl-) $\text{ClC}_6\text{H}_4\text{Si}(\text{CH}_3)_2\text{OCH}_3$ (Silane,(4-chlorophenyl)methoxydimethyl-)	62244-45-5 62244-44-4	** **	9.20 9.20	EI EI	5421 5421
$\text{C}_6\text{H}_{15}\text{O}_3\text{SiCl}^+$	$(\text{C}_2\text{H}_5\text{O})_3\text{SiCl}$	4667-99-6	**	10.52 (V)	PE	3503
$\text{C}_4\text{H}_{10}\text{O}_2\text{SiCl}_2^+$	$(\text{C}_2\text{H}_5\text{O})_2\text{SiCl}_2$	4667-38-3	**	10.78 (V)	PE	3503
$\text{C}_2\text{H}_5\text{OSiCl}_3^+$	$(\text{C}_2\text{H}_5\text{O})\text{SiCl}_3$	1825-82-7	**	11.30 (V)	PE	3503

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
F₃SiCl⁺	SiF ₃ Cl	14049-36-6	**	13.44±0.02 (V)	PE	4026
C₇H₇FSiCl⁺	ClC ₆ H ₄ Si(CH ₃) ₂ F (Silane,(3-chlorophenyl)fluorodimethyl-) ClC ₆ H ₄ Si(CH ₃) ₂ F (Silane,(4-chlorophenyl)fluorodimethyl-)	62244-52-4 62244-51-3	CH ₃	11.13 11.00	EI	5366 5366
C₈H₁₀FSiCl⁺	ClC ₆ H ₄ Si(CH ₃) ₂ F (Silane,(3-chlorophenyl)fluorodimethyl-) ClC ₆ H ₄ Si(CH ₃) ₂ F (Silane,(4-chlorophenyl)fluorodimethyl-)	62244-52-4 62244-51-3	**	9.09 9.15	EI	5421 5421
PCl⁺	PCl ₃	7719-12-2		16.0±0.2	EI	3556
PCl₂⁺	PCl ₃ PCl ₂ Br	7719-12-2 13536-48-6	Cl Br	11.9±0.1 11.3±0.1	EI EI	3556 3556
PCl₃⁺	PCl ₃	7719-12-2	** ** ** ** ** **	10.5 (V) 10.51 (V) 10.52±0.03 (V) 10.52 (V) 10.54 (V) 10.5±0.1	PE PE PE PE PE EI	5190 4023 3669 4146 5539 3556
PCl₅⁺	PCl ₅	10026-13-8	** **	10.7 (V) 10.88 (V)	PE PE	5190 3669
CPCl₅⁺	CCl ₃ PCl ₂	3582-11-4	**	10.25 (V)	PE	4474
C₂H₆PCl⁺	(CH ₃) ₂ PCl	811-62-1	**	9.20 (V)	PE	4474
C₈H₁₈PCl⁺	(tert-C ₄ H ₉) ₂ PCl	13716-10-4	**	8.45 (V)	PE	4474
C₁₈H₂₆PCl⁺	ClC ₆ H ₄ P(C ₆ H ₁₁) ₂ (Phosphine, (4-chlorophenyl)dicyclohexyl-)	40438-62-8	**	8.14 (V)	PE	5417
C₄₅H₃₁PCl⁺	C ₁₅ H ₃₁ PCl (Cyclopenta-1,3-diene,1,4-diphenyl-5-triphenylphosphinylium- 2,3-(naphtha-1,8-diyl),chloride)	XXXXX-XX-X	**	6.82	CTS	5593
CH₃PCl₂⁺	CH ₃ PCl ₂	676-83-5	**	9.85 (V)	PE	4474
C₂H₅PCl₂⁺	C ₂ H ₅ PCl ₂	1498-40-4	**	9.70±0.05 (V)	PE	5033
C₄H₉PCl₂⁺	tert-C ₄ H ₉ PCl ₂	25979-07-1	**	9.30 (V)	PE	4474

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_6\text{H}_5\text{PCl}_2^+$	$\text{C}_6\text{H}_5\text{PCl}_2$ (Phosphorous dichloride,phenyl-)	644-97-3	**	9.7 (V)	PE	5190
			**	9.10±0.01	PE	4154
$\text{C}_{23}\text{H}_{17}\text{PCl}_2^+$	$\text{C}_5\text{H}_2\text{P}(\text{C}_6\text{H}_5)_3\text{Cl}_2$ (Phosphorin,1,1-dichloro-1,1-dihydro-2,4,6-triphenyl-)	40425-71-6	**	7.05 (V)	PE	5271
$\text{CH}_2\text{PCl}_3^+$	$(\text{CH}_2\text{Cl})\text{PCl}_2$	2155-78-4	**	9.58	PE	5627
$\text{C}_2\text{H}_4\text{PCl}_3^+$	$(\text{CH}_2\text{Cl})_2\text{PCl}$	22402-95-5	**	9.38	PE	5627
$\text{C}_{18}\text{H}_{12}\text{PCl}_3^+$	$(\text{ClC}_6\text{H}_4)_3\text{P}$ (Phosphine,tris(4-chlorophenyl)-)	1159-54-2	**	8.18 (V)	PE	5438
$\text{N}_3\text{P}_3\text{Cl}_6^+$	$\text{N}_3\text{P}_3\text{Cl}_6$	940-71-6	**	10.43	PE	5295
$\text{C}_4\text{H}_{12}\text{N}_2\text{PCl}_2^+$	$((\text{CH}_3)_2\text{N})_2\text{PCl}$	8348-44-5	**	8.25 (V)	PE	4474
$\text{C}_2\text{H}_6\text{NPCl}_2^+$	$(\text{CH}_3)_2\text{NPCl}_2$	683-85-2	**	9.45 (V)	PE	4261
			**	9.50 (V)	PE	4474
OPCl^+						
(^2E)	POCl	21295-50-1	**	11.85 (V)	PE	4023
$(^2\text{A}_2)$			**	12.35 (V)	PE	4023
$(^2\text{E}_{3/2})$			**	12.93 (V)	PE	4023
$(^2\text{E}_{1/2})$			**	12.98 (V)	PE	4023
$(^2\text{A}_1)$			**	13.48 (V)	PE	4023
(^2E)			**	13.85 (V)	PE	4023
$(^2\text{A}_1)$			**	15.37 (V)	PE	4023
(^2E)			**	16.53 (V)	PE	4023
OPCl_3^+						
	POCl ₃	10025-87-3	**	11.36±0.02	PE	3835
			**	11.49 (V)	PE	5624
			**	11.58±0.05	PE	3641
			**	11.89±0.03 (V)	PE	3669
			**	12.0 (V)	PE	5190
	POCl ₃	63736-95-8	**	11.89±0.02 (V)	PE	4730
$\text{C}_2\text{H}_6\text{OPCl}^+$	$(\text{CH}_3)_2\text{P}(\text{O})\text{Cl}$	1111-92-8	**	10.77 (V)	PE	5523
$\text{C}_4\text{H}_8\text{OPCl}^+$	$\text{POCl}(\text{C}_2\text{H}_5)(\text{CH}=\text{CH}_2)$	61752-99-6	**	10.62 (V)	PE	5021
$\text{C}_4\text{H}_{10}\text{O}_3\text{PCl}^+$	$\text{OPCl}(\text{OC}_2\text{H}_5)_2$	814-49-3	**	10.29	PE	5627
$\text{C}_6\text{H}_{14}\text{O}_3\text{PCl}^+$	$\text{OPCl}(\text{OC}_3\text{H}_7)_2$	2510-89-6	**	10.89	PE	5627
$\text{CH}_3\text{OPCl}_2^+$	CH_3OPCl_2	676-97-1	**	10.92	PE	5627
			**	11.4 (V)	PE	5190

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
CH₃OPCl₂⁺	CH ₃ OPCl ₂	676-97-1	** ** **	11.43 (V) 11.45 (V) 11.49 (V)	PE PE PE	5523 5021 5328
C₂H₃OPCl₂⁺	POCl ₂ (CH=CH ₂)	1438-74-0	** ** **	10.81 11.24 (V) 11.24 (V)	PE PE PE	5032 5021 5328
C₃H₅OPCl₂⁺	POCl ₂ (CH ₂ CH=CH ₂) POCl ₂ (C(CH ₃)=CH ₂)	1498-47-1 3944-27-2	** **	10.54 (V) 10.86 (V)	PE PE	5021 5021
C₆H₅OPCl₂⁺	C ₆ H ₅ (POCl ₂) (Phosphonic dichloride, phenyl-)	824-72-6	**	9.95 (V)	PE	5021
CH₃O₂PCl₂⁺	PCl ₂ O(OCH ₃)	677-24-7	**	11.50 (V)	PE	4699
C₂H₅O₂PCl₂⁺	PCl ₂ O(OC ₂ H ₅)	1498-51-7	** **	11.42 (V) 11.46 (V)	PE PE	4699 5624
C₄H₉O₂PCl₂⁺	(CH ₂ Cl) ₂ (OC ₂ H ₅)PO	13274-84-5	**	10.19	PE	5627
C₆H₅O₂PCl₂⁺	OPCl ₂ OC ₆ H ₅ (Phosphorodichloridic acid, phenyl ester)	770-12-7	**	9.1	PE	5627
C₄H₉O₄PCl₂⁺	(CH ₃ O) ₂ P(=O)OCHCCl ₂	62-73-7	**	9.4 (V)	PE	5190
C₂H₄OPCl₃⁺	(CH ₂ Cl) ₂ PClO	13482-64-9	**	10.46	PE	5627
C₈H₁₄O₅PCl₃⁺	(CH ₃ O) ₂ P(=O)CH(CCl ₃)OCOC ₃ H ₇	XXXXX-XX-X	**	10.3 (V)	PE	5190
C₄H₁₂N₂OPCl₂⁺	OPCl(N(CH ₃) ₂) ₂	1605-65-8	**	8.61	PE	5627
C₂H₆NOPCl₂⁺	(CH ₃) ₂ NPCl ₂ O	677-43-0	**	9.31 (V)	PE	5624
C₃H₆N₂OPCl₃⁺	CN ₂ P(=O)Cl ₃ (CH ₃) ₂ (1,3,2-Diazaphosphetidin-4-one, 2,2,2-trichloro-2,2-dihydro-1,3-dimethyl-)	3576-20-3	**	9.20±0.1	EI	5462
F₂PCl⁺	PF ₂ Cl	14335-40-1	**	12.8±0.1 (V)	PE	3662
C₂F₆PCl⁺	(CF ₃) ₂ PCl	650-52-2	** **	11.13 (V) 11.13 (V)	PE PE	4371 4261
CF₃PCl₂⁺	CF ₃ PCl ₂	421-58-9	** **	10.70 (V) 10.70 (V)	PE PE	4371 4261

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
CF₂PCl₃⁺	CCl ₃ PF ₂	1112-03-4	**	10.65 (V)	PE	4474
C₃H₆NF₃PCl⁺	(CH ₃) ₂ NP(Cl)CF ₃	3135-63-5	**	9.56 (V)	PE	4261
C₂H₆SiPCl₃⁺	Cl ₃ SiP(CH ₃) ₂	XXXXX-XX-X	**	9.1±0.05 (V)	PE	5419
SCI⁺	SCl ₂	10545-99-0	Cl	12.2±0.1	EI	4287
S₂Cl⁺	S ₂ Cl ₂	10025-67-9	Cl	12.2±0.2	EI	4287
SCI₂⁺	SCl ₂	10545-99-0	**	9.49	PE	4188
(² B ₁)			**	9.67 (V)	PE	4150
(² A ₁ , ² B ₂)			**	12.19 (V)	PE	4150
(² A ₂)			**	12.45 (V)	PE	4150
			**	9.7±0.1	EI	4287
S₂Cl₂⁺	S ₂ Cl ₂	10025-67-9	**	9.4	PE	4188
			**	11.3±0.2	EI	4287
BSCl⁺	ClB=S	55753-38-3	**	10.51±0.1	PE	4857
CSCl₂⁺	CCl ₂ S	463-71-8	**	9.61±0.02	PE	3667
			**	9.68	PE	4080
			**	9.80 (V)	PE	3746
C₂S₂Cl₄⁺	C ₂ S ₂ Cl ₄ (1,3-Dithietane,2,2,4,4-tetrachloro)	20464-23-7	**	9.69 (V)	PE	5572
C₂H₅SCl⁺	CH ₃ SCH ₂ Cl	2373-51-5	**	7.74 (V)	PE	5526
C₄H₃SCl⁺	C ₄ H ₃ SCl (Thiophene, 2-chloro-)	96-43-5	**	8.89±0.05 (V)	PE	4626
			**	9.06±0.05	EI	3482
			**	8.83	CTS	3787
C₅H₅SCl⁺	C ₄ H ₃ SCH ₂ Cl (Thiophene, 2-(chloromethyl)-)	765-50-4	**	8.89±0.05 (V)	PE	4626
C₁₂H₉SCl⁺	C ₆ H ₄ (Cl)SC ₆ H ₅ (Benzene, 1-chloro-3-(phenylthio)-)	38700-88-8	**	8.16	CTS	4272
	C ₆ H ₄ (Cl)SC ₆ H ₅ (Benzene, 1-chloro-4-(phenylthio)-)	13343-26-5	**	8.07	CTS	4272
C₉H₅S₃Cl⁺	(C ₃ HS ₂)=S(C ₆ H ₄ Cl) (3/2)g	5761-16-0	**	8.15 (V)	PE	4403
	Ar ₂		**	15.675±0.02 (V)	PE	4885

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_5\text{H}_8\text{SCl}_2^+$	$\text{CCl}_2=\text{CHS(iso-C}_3\text{H}_7)$	19284-67-4	**	8.14 ± 0.01	PI	5531
$\text{C}_{12}\text{H}_8\text{SCl}_2^+$	$\text{C}_6\text{H}_4(\text{Cl})\text{SC}_6\text{H}_4\text{Cl}$ (Benzene, 1,1'-thiobis[4-chloro-])	5181-10-2	**	8.13	CTS	4272
$\text{BC}_{12}\text{H}_{18}\text{SCl}^+$	$\text{C}_6\text{H}_4(\text{Cl})\text{SB}(n\text{-C}_3\text{H}_7)_2$ (Borinic acid, dipropylthio-3-chlorophenyl ester)	64541-68-0	**	8.87 ± 0.05 (V)	PE	4848
	$\text{C}_6\text{H}_4(\text{Cl})\text{SB}(n\text{-C}_3\text{H}_7)_2$ (Borinic acid, dipropylthio-4-chlorophenyl ester)	64503-48-6	**	8.73 ± 0.05 (V)	PE	4848
NSCl^+	NSCl	17178-58-4	**	10.61 ± 0.01	PE	4604
(${}^2\text{A}'$)			**	10.96 (V)	PE	3660
			**	11.38 ± 0.02	PE	4604
(${}^2\text{A}', {}^2\text{A}''$)			**	11.80 (V)	PE	3660
(${}^2\text{A}'$)			**	13.73 ± 0.02	PE	4604
(${}^2\text{A}'$)			**	13.77 (V)	PE	3660
(${}^2\text{A}'$)			**	14.28 ± 0.02	PE	4604
(${}^2\text{A}'$)			**	14.46 (V)	PE	3660
(${}^2\text{A}'$)			**	16.5 ± 0.01	PE	4604
$\text{C}_6\text{H}_6\text{NSCl}^+$	$\text{C}_7\text{H}_5\text{NS}(\text{Cl})\text{CH}_3$ (Benzothiazole, 6-chloro-2-methyl-)	4146-24-1	**	8.50 (V)	PE	4437
$\text{C}_8\text{H}_8\text{NSCl}^+$	$\text{C}_6\text{H}_4\text{ClNHCSCH}_3$ (Ethanethioamide, N-(2-chlorophenyl)-)	39184-83-3	**	8.10	EI	4834
$\text{C}_{12}\text{H}_{16}\text{NSCl}^+$	$\text{C}_6\text{H}_4\text{ClNHCSCH}_2\text{C}(\text{CH}_3)_3$ (Butanethioamide, N-(2-chlorophenyl)-3,3-dimethyl-)	62635-54-5	**	8.00 ± 0.05	EI	4834
$\text{C}_7\text{H}_7\text{N}_2\text{SCI}^+$	$\text{C}_6\text{H}_4\text{ClNHCSNH}_2$ (Thiourea, (2-chlorophenyl)-)	5344-82-1	**	8.05	EI	4834
$\text{C}_8\text{H}_9\text{N}_2\text{SCI}^+$	$\text{C}_6\text{H}_4\text{ClNHCSNHCH}_3$ (Thiourea, N-(2-chlorophenyl)-N'-methyl-)	30954-73-5	**	7.85 ± 0.05	EI	4834
$\text{C}_9\text{H}_{11}\text{N}_2\text{SCI}^+$	$\text{C}_6\text{H}_4\text{ClNHCSNHCH}_2\text{H}_5$ (Thiourea, N-(2-chlorophenyl)-N'-ethyl-)	19384-08-8	**	7.85 ± 0.05	EI	4834
$\text{C}_{10}\text{H}_{13}\text{N}_2\text{SCI}^+$	$\text{C}_6\text{H}_4\text{ClNHCSNHCH}(\text{CH}_3)_2$ (Thiourea, N-(2-chlorophenyl)-N'-(1-methylethyl)-)	62635-49-8	**	7.80 ± 0.05	EI	4834
$\text{C}_{11}\text{H}_{15}\text{N}_2\text{SCI}^+$	$\text{C}_6\text{H}_4\text{ClNHCSNH}(\text{CH}_3)_3$ (Thiourea, N-(2-chlorophenyl)-N'-(1,1-dimethylethyl)-)	62635-50-1	**	7.75 ± 0.05	EI	4834
$\text{C}_{17}\text{H}_{19}\text{N}_2\text{SCI}^+$	$\text{C}_{17}\text{H}_{19}\text{N}_2\text{SCI}$ (10H-Phenothiazine-10-propanamine, 2-chloro-N,N-dimethyl-)	50-53-3	**	7.16 ± 0.08 (V)	PE	4667
	$\text{C}_{17}\text{H}_{19}\text{N}_2\text{SCI}$ (10H-Phenothiazine-10-propanamine, 2-chloro-N,N-dimethyl-)	50-53-3	**	8.25 ± 0.07	CTS	4079

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₅H₂N₃SCl⁺	C ₅ H ₂ N ₃ SCl (1,2,3)Thiadiazolo[5,4- <i>b</i>]pyridine, 5-chloro-)	54459-89-1	**	9.57±0.05	EI	4316
C₂₀H₂₄N₃SCl⁺	C ₁₂ H ₇ NS(Cl)(CH ₂) ₃ C ₄ H ₈ N ₂ CH ₃ (10 <i>H</i> -Phenothiazine, 2-chloro-10-[3-(4-methyl-1-piperazinyl)propyl]-)	58-38-8	**	7.03±0.07	CTS	4079
O₂SCl⁺	SO ₂ Cl ₂	7791-25-5	**	11.8±0.5	EI	4921
OSCl₂⁺	Cl ₂ SO	7719-09-7	** ** ** ** ** **	11.07 (V) 11.12 (V) 11.13 (V) 11.3 (V) 11.3 (V)	PE PE PE PE PE	4295 3705 3646 3694 3879
O₂SCl₂⁺	SO ₂ Cl ₂	7791-25-5	** ** ** ** ** **	12.05 12.4 (V) 12.41 (V) 12.41 (V) 12.42 (V) 11.4±0.5	PE PE PE PE PE EI	3879 3694 4827 5207 3705 4921
OSCl₃⁺	SOCl ₃	XXXXX-XX-X	**	9.63±0.02	PE	3835
C₁₃H₁₁OSCl⁺	C ₆ H ₄ (OCH ₃)SC ₆ H ₄ Cl (Benzene, 1-chloro-4-[(4-methoxyphenyl)thio]-)	20912-69-0	**	7.91	CTS	4272
CH₃O₂SCl⁺	(CH ₃)SO ₂ (Cl)	124-63-0	** ** **	11.6 (V) 11.6 (V) 11.74 (V)	PE PE PE	4827 5207 3705
C₁₇H₁₇N₂OSCl⁺	C ₁₂ H ₇ NS(Cl)COCH ₂ CH ₂ N(CH ₃) ₂ (10 <i>H</i> -Phenothiazine, 2-chloro-10-[3-(dimethylamino)-1-oxopropyl]-)	3576-45-2	**	8.24±0.07	CTS	4079
C₁₉H₂₁N₂OSCl⁺	C ₁₂ H ₇ NS(Cl)COCH ₂ CH ₂ N(C ₂ H ₅) ₂ (10 <i>H</i> -Phenothiazine, 2-chloro-10-[3-(diethylamino)-1-oxopropyl]-)	800-22-6	**	7.87±0.07	CTS	4079
C₂₁H₂₆N₃OSCl⁺	C ₂₁ H ₂₆ N ₃ OSCl (1-Piperazineethanol, 4-[3-(2-chloro-10 <i>H</i> -phenothiazin-10-yl)propyl]-)	58-39-9	**	8.63±0.07	CTS	4079
F₃SCl⁺	SF ₃ Cl	13780-57-9	**	12.335±0.005	PE	3655
CFSCl⁺	FClCS	1495-18-7	**	10.20 (V)	PE	3746
C₂F₃S₂Cl⁺	S=C(Cl)SCF ₃	1540-66-5	**	9.57 (V)	PE	4345
O₂SFCl⁺	SO ₂ FCl	13637-84-8	** **	12.61 (V) 12.3±0.5	PE EI	3705 4921

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_8\text{H}_{11}\text{SiSCl}^+$	$\text{C}_6\text{H}_5\text{SSi}(\text{CH}_3)_2\text{Cl}$ (Silane, chlorodimethyl(phenylthio)-)	52548-11-5	**	8.76 ± 0.1	EI	4198
$\text{C}_6\text{H}_5\text{SiSCl}_3^+$	$\text{C}_6\text{H}_5\text{SSiCl}_3$ (Silane, trichloro(phenylthio)-)	7579-91-1	**	9.03 ± 0.1	EI	4198
PSCl_3^+	PSCl_3	3982-91-0	** ** ** ** ** **	9.71 ± 0.003 9.71 ± 0.03 $10.11 (\text{V})$ $10.13 \pm 0.03 (\text{V})$ $10.15 (\text{V})$ $10.15 (\text{V})$	PE PE PE PE PE PE	4086 4279 4023 3669 5514 5627
$\text{C}_2\text{H}_6\text{PSCl}^+$	$(\text{CH}_3)_2\text{P}(\text{S})\text{Cl}$	993-12-4	**	9.12 (V)	PE	5523
$\text{CH}_3\text{PSCl}_2^+$	$\text{CH}_3\text{P}(\text{S})\text{Cl}_2$	676-98-2	**	9.73 (V)	PE	5523
$\text{C}_6\text{H}_5\text{PSCl}_2^+$	$\text{C}_6\text{H}_5\text{P}(\text{Cl})_2\text{S}$ (Phosphonothioic dichloride, phenyl-)	3497-00-5	** ** **	9.02 ± 0.03 9.02 9.47 (V)	PE PE PE	4279 5514 5627
$\text{C}_2\text{H}_4\text{PSCl}_3^+$	$(\text{CH}_2\text{Cl})_2\text{PSCl}$	20459-66-9	**	9.16	PE	5627
$\text{C}_4\text{H}_{12}\text{N}_2\text{PSCl}^+$	$((\text{CH}_3)_2\text{N})_2\text{ClPS}$	3732-81-8	** ** **	8.23 ± 0.003 8.23 ± 0.02 8.75 (V)	PE PE PE	4086 4279 5627
$\text{C}_2\text{H}_6\text{NPSCl}_2^+$	$\text{PSCl}_2\text{N}(\text{CH}_3)_2$	1498-65-3	** ** **	8.97 ± 0.003 8.97 ± 0.04 9.35 (V)	PE PE PE	4086 4279 5627
$\text{C}_4\text{H}_{10}\text{O}_2\text{PSCl}^+$	$(\text{C}_2\text{H}_5\text{O})_2\text{ClPS}$	2524-04-1	** ** **	8.83 ± 0.02 9.41 (V) 9.41 (V)	PE PE PE	4279 5514 5627
$\text{CH}_3\text{OPSCl}_2^+$	$\text{PCl}_2\text{S}(\text{OCH}_3)$ $\text{Cl}_2\text{P}(\text{O})\text{SCH}_3$	2523-94-6 18281-76-0	** **	9.85 (V) 10.20 (V)	PE PE	4699 5328
$\text{C}_2\text{H}_5\text{OPSCl}_2^+$	$\text{SPCl}_2(\text{OC}_2\text{H}_5)$	1498-64-2	** ** **	9.81 (V) 9.32 ± 0.03 9.81 (V)	PE PE PE	5627 4279 5514
Ar^+	Ar	7440-37-1	** ** ** ** **	15.75973 ± 0.00001 S 15.753 ± 0.002 15.930 ± 0.002 15.713 ± 0.003 15.7	PEN PE PE EI	3923 3525 3525 3541 5022

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
Ar⁺²						
	Ar	7440-37-1	**	~43	EI	3445
			**	43.5±0.2	EI	4503
			**	43.7±0.5	EI	3625
	Ar ⁺	XXXXX-XX-X	**	~17	EI	5022
Ar₂⁺						
	Ar ₂	12595-59-4	**	14.44	PI	5195
			**	14.54±0.02	PI	4923
(1/2)u			**	15.55±0.025 (V)	PE	4885
(3/2)g			**	15.675±0.02 (V)	PE	4885
(1/2)u			**	15.87±0.015 (V)	PE	4885
(1/2)g			**	15.99±0.03 (V)	PE	4885
			**	15.2±0.2	EI	5350
			**	15.675±0.02 (V)	PE	4885
K⁺						
	K	7440-09-7	**	4.34	PE	4642
			**	4.1±0.3	EI	4873
			**	4.4	EI	4912
(² P _{3/2})	KI	7681-11-0	I ⁻	25.14±0.04 (V)	PE	5035
(² P _{1/2})				25.50±0.04 (s)	PE	5035
	KF	7789-23-3	F	9.54±0.20	EI	4663
	K ₂	25681-80-5	K	4.85	PI	4914
(² P _{3/2})	KCl	7447-40-7	Cl ⁻	24.98±0.04 (V)	PE	5035
(² P _{1/2})				25.22±0.04 (s)	PE	5035
(² P _{3/2})	KBr	7758-02-3	Br ⁻	25.04±0.04 (V)	PE	5035
(² P _{1/2})				25.36±0.04 (s)	PE	5035
	KBO ₂	XXXXX-XX-X	BO ₂	9.47±0.20	EI	4663
	NaK	12056-29-0	Na	4.96	PI	4914
K₂⁺						
	K ₂	25681-80-5	**	4.059±0.001	PI	4914
(² Sigma ⁺)			**	4.06073±0.00016	PI	1395
			**	3.9	EI	4912
K₃⁺						
	K ₃	37279-39-3	**	3.3±0.1	PI	4914
K₄⁺						
	K ₄	39297-76-2	**	3.6±0.1	PI	4914
K₅⁺						
	K ₅	39297-77-3	**	3.3±0.1	PI	4914
K₇⁺						
	K ₇	39297-79-5	**	3.3±0.1	PI	4914
K₈⁺						
	K ₈	39297-80-8	**	3.4±0.1	PI	4914

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
LiK⁺	KLi	12030-83-0	**	4.69±0.10	EI	4912
CNK⁺	KCN	151-50-8	**	9.3±0.3	EI	4875
CNK₂⁺	(KCN) ₂	XXXXX-XX-X		10.3±0.3	EI	4875
C₂N₂K₃⁺	(KCN) ₃	XXXXX-XX-X		10. ±1	EI	4875
OK⁺	KO	12401-70-6	** **	7.1±0.2 8.±1	EI EI	4745 4745
OK₂⁺	K ₂ O	12136-45-7	** ** **	7.5±0.1 7.5±0.2 10.7±0.3	EI EI EI	4745 4745 4873
BO₂K⁺	KBO ₂	XXXXX-XX-X	**	8.62±0.14	EI	4663
BO₂K₂⁺	(KBO ₂) ₂ K ₂ BO ₂ F K ₂ BO ₂ F	XXXXX-XX-X XXXXX-XX-X XXXXX-XX-X	BO ₂ F ⁻ F	9.97±0.18 5.91±0.10 9.97±0.18	EI EI EI	4663 4663 4663
CO₃K₂⁺	K ₂ CO ₃	XXXXX-XX-X	**	7.4±0.3	EI	4873
NO₃K⁺	KNO ₃	XXXXX-XX-X	**	8.96±0.03 (V)	PE	5354
FK₂⁺	K ₂ F ₂ K ₂ BO ₂ F K ₂ BO ₂ F	12285-62-0 XXXXX-XX-X XXXXX-XX-X	F ⁻ BO ₂ BO ₂	5.48±0.12 9.44±0.15 5.48±0.12 9.44±0.15	EI EI EI EI	4663 4663 4663 4663
NaK⁺	NaK	12056-29-0	** **	4.52±0.05 4.57±0.20	PI EI	4914 4912
NaK₂⁺	NaK ₂	12532-69-3	**	3.6±0.1	PI	4914
Na₂K⁺	Na ₂ K	12286-02-1	**	3.7±0.1	PI	4914
Na₂K₂⁺	Na ₂ K ₂	66459-14-1	**	4.0±0.1	PI	4914
Na₃K⁺	Na ₃ K	66419-70-3	**	4.1±0.05	PI	4914
Na₄K⁺	Na ₄ K	66419-71-4	**	4.0±0.1	PI	4914

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
Na₅K⁺	Na ₅ K	66419-72-5	**	4.1±0.1	PI	4914
F₄AlK⁺	KAlF ₄	14484-69-6	**	13.02±0.05 (V)	PE	5238
O₃PK⁺	KPO ₃	XXXXX-XX-X	**	9.44±0.03 (V)	PE	4840
ClK⁺ (³ P _{3/2})	KCl	7447-40-7	**	8.44±0.1	PE	4344
			**	8.44±0.1	PE	5035
			**	8.7 (V)	PE	4307
Cl₂K₂⁺	(KCl) ₂	12258-97-8	**	9.60 (V)	PE	4344
			**	9.60 (V)	PE	5035
AlCl₄K⁺	KAlCl ₄	13821-13-1	**	10.96±0.05 (V)	PE	5238
Ca⁺ (² S _{1/2})	Ca	7440-70-2	**	6.11321±0.00002 S		4583
			**	6.0	PE	4860
			**	6.0±0.3	EI	5067
			**	6.06±0.05	EI	4114
			**	6.08±0.06	EI	5342
			**	~6.1	EI	3486
Ca⁺²	Ca	7440-70-2	**	18	EI	3486
HCa⁺	CaH	14452-75-6	**	5.86±0.09	S	4216
OCa⁺	CaO	1305-78-8	**	6.5±1	EI	4881
Cl₂Ca⁺	CaCl ₂	10043-52-4	**	10.2 (V)	PE	4761
Sc⁺	Sc	7440-20-2	**	6.7±0.5	EI	5349
C₂Sc⁺	ScC ₂	XXXXX-XX-X	**	7.6±0.5	EI	5349
	ScC ₂	12175-91-6	**	7.7±0.2	EI	3470
C₁₅H₃O₆F₁₀Sc⁺ (CF ₃ COCHCOCF ₃) ₃ Sc (Scandium, tris(1,1,1,5,5-hexafluoro-2,4-pentanedionato-O,O')-, (OC-6-11)-)		18990-42-6	**	10.13±0.07 (V)	PE	3682
Ti⁺	Ti	7440-32-6	**	6.6±0.5	EI	3449
			**	6.7	EI	4872
			**	6.78±0.02	EI	5342
			**	6.8±0.1	EI	4114
			**	7.3±0.6	EI	4206
			**	7.3±0.6	EI	5635
			**	7.4±0.5	EI	3594
	TiO	12137-20-1	O	14.5±0.7	EI	3594
			O	14.51±0.36	EI	4103

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₂Ti⁺						
	TiC ₂	12071-32-8	** ** **	8.2±0.6 8.2±0.6 8.7±0.5	EI EI EI	4206 5635 4112
C₄Ti⁺						
	TiC ₄	12547-96-5	**	9.0±1.0	EI	4112
C₁₂H₁₂Ti⁺						
	C ₅ H ₇ Ti(C ₅ H ₅) (Titanium, (η^7 -cycloheptatrienylum)(η^5 -2,4-cyclopentadien-1-yl)-)	51203-49-7	**	6.83±0.05 (V)	PE	4132
	(C ₆ H ₅) ₂ Ti (Titanium, bis(η^6 -benzene)-)	52462-43-8	**	5.5–6.0 (V)	PE	4393
C₁₃H₁₃Ti⁺						
	C ₈ H ₈ Ti(C ₅ H ₅) (Titanium, (η^6 -1,3,5,7-cyclooctatetraene)(η^5 -2,4-cyclopentadien-1-yl)-)	11065-40-0	**	5.67±0.05 (V)	PE	4132
C₁₄H₁₆Ti⁺						
	(C ₆ H ₅ CH ₃) ₂ Ti (Titanium, bis[(1,2,3,4,5,6- η)-methylbenzene]-)	55527-82-7	**	5.4–7.2 (V)	PE	4393
C₂₀H₄₄Ti⁺						
	((CH ₃) ₃ CCH ₂) ₄ Ti	36945-13-8	**	8.33±0.1 (V)	PE	4242
C₁₃H₃₃N₃Ti⁺						
	(N(C ₂ H ₅) ₂) ₃ (CH ₃)Ti	25483-56-1	**	7.6 (V)	PE	4734
C₈H₂₄N₄Ti⁺						
	(N(CH ₃) ₂) ₄ Ti	XXXXXX-XX-X	**	7.13 (V)	PE	4588
C₁₆H₄₀N₄Ti⁺						
	(N(C ₂ H ₅) ₂) ₄ Ti	XXXXXX-XX-X	**	6.83 (V)	PE	4588
OTi⁺						
	TiO	12137-20-1	** ** ** ** ** ** **	6.4±0.1 6.7±0.1 6.7 6.8±0.5 6.8±0.5 7.22±0.35 7.3±0.5	EI EI EI EI EI EI EI	4114 5471 4872 3449 4678 4103 3594
O₂Ti⁺						
	TiO ₂	13463-67-7	** ** ** **	8.5±0.5 9.54±0.1 10.2±0.2 11.56±0.14	EI EI EI EI	3594 5471 4114 4103
C₁₂H₁₀O₂Ti⁺						
	(C ₅ H ₅) ₂ (CO) ₂ Ti (Titanium,dicarbonylbis(η^5 -2,4-cyclopentadien-1-yl)-)	12129-51-0	**	6.35 (V)	PE	5217
C₁₀H₂₄O₃Ti⁺						
	(n-C ₃ H ₇ O) ₃ (CH ₃)Ti	64516-16-1	**	9.4 (V)	PE	4734
C₁₀H₂₄O₄Ti⁺						
	(n-C ₃ H ₇ O) ₃ (CH ₃ O)Ti	64516-17-2	**	9.1 (V)	PE	4734
N₄O₁₂Ti⁺						
	(NO ₃) ₄ Ti	12372-56-4	**	12.35±0.11 (V)	PE	4999

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₀H₁₀F₂Ti⁺	(η-C ₅ H ₅) ₂ TiF ₂ (Titanium,bis(η ⁵ -2,4-cyclopentadien-1-yl)difuoro-)	309-89-7	**	8.1±0.1 (V)	PE	4987
C₁₅H₃O₆F₁₈Ti⁺	(CF ₃ COCHCOCF ₃) ₃ Ti (Titanium, tris(1,1,1,5,5-hexafluoro-2,4-pentanedionato-O,O')-, (OC-6-11)-)	22854-59-7	** **	7.94±0.07 (V) 7.98 (V)	PE PE	3682 3681
C₁₆H₄₄Si₄Ti⁺	((CH ₃) ₃ SiCH ₂) ₄ Ti	33948-28-6	**	8.58±0.1 (V)	PE	4242
STi⁺	TiS	12039-07-5	**	7.1±0.3	EI	3449
Cl₄Ti⁺	TiCl ₄	7550-45-0	** **	11.70 (V) 11.76 (V)	PE PE	5148 4694
C₁₀H₁₀Cl₂Ti⁺	(η-C ₅ H ₅) ₂ TiCl ₂ (Titanium dichlorobis(η ⁵ -2,4-cyclopentadien-1-yl)-) (η-C ₅ H ₅) ₂ TiCl ₂ Titanium, dichlorobis (η ⁵ -2,4-cyclopentadien-1-yl)-)	1271-19-8 1271-19-8	** **	8.5±0.1 (V) 8.46±0.05 (V)	PE PE	4987 4375
CH₃Cl₃Ti⁺	Ti(CH ₃)Cl ₃	2747-38-8	**	10.8 (V)	PE	4734
V⁺	V	7440-62-2	** **	~7.5 7±1	EI EI	4202 3801
	VOF ₃	13709-31-4	3F+O	31.26±0.19	EI	4546
	VOCl ₃	7727-18-6	3Cl+O	26.83±0.39	EI	4546
V⁺⁵	V ⁺⁴	22541-76-0	**	65.2812±0.0006	S	4264
C₁₀H₁₀V⁺	(C ₅ H ₅) ₂ V (Vanadocene)	1277-47-0	** **	6.78 (V) 6.81 (V)	PE PE	5507 5394
C₁₂H₁₂V⁺	C ₇ H ₇ V(C ₅ H ₅) (Vanadium, (η ⁷ -cycloheptatrienylum)(η ⁵ -2,4-cyclopentadien-1-yl)-)	12636-68-9	**	6.42±0.05 (V)	PE	4132
C₁₂H₁₄V⁺	(C ₅ H ₄ CH ₃) ₂ V (Vanadocene,1,1'-dimethyl-)	12146-93-9	**	6.60 (V)	PE	5507
C₁₈H₂₄V⁺	(C ₆ H ₃ (CH ₃) ₃) ₂ V (Vanadium, bis[(1,2,3,4,5,6-η)-1,3,5-trimethylbenzene]-)	1272-71-5	**	5.61±0.05 (V)	PE	4132
C₂₀H₃₀V⁺	(C ₅ (CH ₃) ₃) ₂ V (Vanadocene,decamethyl-)	XXXXX-XX-X	**	5.87 (V)	PE	5394
NV⁺	VN	24646-85-3	**	8±1	EI	3801

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₈H₂₄N₄V⁺	(N(CH ₃) ₂) ₄ V	XXXXX-XX-X **		6.2 (V)	PE	4588
OV⁺						
	VO	12035-98-2	**	8±1	EI	3620
	VOF ₃	13709-31-4	3F	24.41±0.10	EI	4546
	VOCl ₃	7727-18-6	3Cl	19.77±0.09	EI	4546
O₂V⁺						
	VO ₂	12036-21-4	**	12.7±0.2	EI	4131
			**	10±2	EI	3620
O₈V₄⁺						
	V ₄ O ₈	12503-87-6	**	13±1	EI	3620
O₁₀V₄⁺						
	V ₄ O ₁₀	12503-98-9	**	11.8±0.3	EI	4131
			**	12±1	EI	3620
N₃O₁₀V⁺						
	(NO ₃) ₃ VO	16017-37-1	**	12.33±0.04 (V)	PE	4999
FV⁺						
	VOF ₃	13709-31-4	2F+O	25.75±0.26	EI	4546
F₂V⁺						
	VOF ₃	13709-31-4	F+O	20.83±0.19	EI	4546
F₃V⁺						
	VOF ₃	13709-31-4	O	16.76±0.05	EI	4546
OFV⁺						
	VOF ₃	13709-31-4	2F	19.92±0.06	EI	4546
OF₂V⁺						
	VOF ₃	13709-31-4	F	15.31±0.06	EI	4546
OF₃V⁺						
	VOF ₃	13709-31-4	**	13.88±0.05	EI	4546
C₁₅H₃O₆F₁₈V⁺						
	(CF ₃ COCHCOCF ₃) ₃ V (Vanadium, tris(1,1,1,5,5-hexafluoro-2,4-pentanedionato-O,O')-, (OC-6-11)-)	15695-77-9 **		8.68±0.07 (V)	PE	3682
			**	8.68 (V)	PE	3681
SV⁺						
	VS	12166-27-7	**	~9	EI	4202
ClIV⁺						
	VOCl ₃	7727-18-6	2Cl+O	22.16±0.07	EI	4546
Cl₂V⁺						
	VOCl ₃	7727-18-6	Cl+O	18.98±0.20	EI	4546
Cl₃V⁺						
	VOCl ₃	7727-18-6	O	16.48±0.28	EI	4546
OCIV⁺						
	VOCl ₃	7727-18-6	Cl ₂ 2Cl	14.05±0.06 16.31±0.05	EI EI	4546 4546

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
OCl₂V⁺	VOCl ₃	7727-18-6	Cl	13.25±0.05	EI	4546
OCl₃V⁺	VOCl ₃	7727-18-6	** **	11.84 (V) 11.90±0.05	PE EI	5148 4546
Cr⁺						
(⁶ S)	Cr	7440-47-3	**	6.76	PE	4858
(⁶ D)			**	8.29	PE	4858
			**	6.76 (V)	OTH	5286
(CO) ₆ Cr		13007-92-6	6CO	15.36±0.03	EI	5291
(C ₆ H ₆)(CO) ₅ Cr (Chromium, (η^6 -benzene)tricarbonyl-)		12082-08-5	C ₆ H ₆ +3CO	12.2±0.2	EI	3786
(C ₆ H ₅ CH ₃)(CO) ₅ Cr (Chromium, tricarbonyl[(1,2,3,4,5,6- η)-methylbenzene]-)		12083-24-8	C ₆ H ₅ CH ₃ +3CO	13.42±0.1	EI	3788
(C ₆ H ₄ (CH ₃) ₂)(CO) ₅ Cr (Chromium, tricarbonyl[(1,2,3,4,5,6- η)-1,2-dimethylbenzene]-)		12129-29-2	C ₆ H ₅ CH ₃ +3CO	13.5±0.2	EI	5210
(C ₆ H ₃ (CH ₃) ₃)(CO) ₅ Cr (Chromium, tricarbonyl[(1,2,3,4,5,6- η)-1,3,5-trimethylbenzene]-)		12129-67-8		13.90±0.1	EI	3788
(C ₆ (CH ₃) ₆)(CO) ₅ Cr (Chromium, tricarbonyl[(1,2,3,4,5,6- η)-hexamethylbenzene]-)		12088-11-8		13.00±0.1	EI	3788
(C ₆ H ₅ CH ₂ OH)(CO) ₅ Cr (Chromium, [(1,2,3,4,5,6- η)-benzenemethanol]tricarbonyl-)		12116-45-9		14.01±0.1	EI	3788
(C ₆ H ₅ OCH ₃)(CO) ₅ Cr (Chromium, tricarbonyl[(1,2,3,4,5,6- η)-methoxybenzene]-)		12116-44-8		12.65±0.1	EI	3788
(C ₆ H ₅ COOCH ₃)(CO) ₅ Cr (Chromium, tricarbonyl[(1,2,3,4,5,6- η)-methylbenzoate]-)		12125-87-0		14.00±0.1	EI	3788
(C ₆ H ₅ NH ₂)(CO) ₅ Cr (Chromium, (η^6 -benzylamine)tricarbonyl-)		12108-11-1	C ₆ H ₅ NH ₂ +3CO	13.17±0.1	EI	3788
((CH ₃) ₂ N) ₃ P(CO) ₅ Cr		XXXXX-XX-X		22.3±0.05	EI	3952
((CH ₃) ₂ N) ₃ P ₂ (CO) ₄ Cr		19976-85-3		22.2±0.05	EI	3952
CS(CO) ₅ Cr		50358-90-2	5CO+CS	16.16±0.07	EI	5291
(C ₆ H ₅ Cl)(CO) ₅ Cr (Chromium, tricarbonyl(η^6 -chlorobenzene)-)		12082-03-0	C ₆ H ₅ Cl+3CO	14.10±0.1	EI	3788
C₅H₅Cr⁺						
(C ₅ H ₅)(CO) ₂ (NO)Cr (Chromium,dicarbonyl(η^5 -2,4-cyclopentadien-1-yl)nitrosyl-)		36312-04-6	2CO+NO	12.79±0.1	EI	5348
(C ₅ H ₅)(CO) ₂ (NS)Cr (Chromium,dicarbonyl(η^5 -2,4-cyclopentadien-1-yl)thionitrosyl-)		66539-91-1	2CO+NS	13.45±0.1	EI	5348
C₆H₆Cr⁺						
(C ₆ H ₆)(CO) ₅ Cr (Chromium, (η^6 -benzene)tricarbonyl-)		12082-08-5	3CO	9.0±0.2	EI	3786
			3CO	10.34±0.1	EI	3788
C₇H₈Cr⁺						
(C ₇ H ₈)(CO) ₅ Cr (Chromium, tricarbonyl[(1,2,3,4,5,6- η)-1,3,5-cycloheptatriene]-)		12125-72-3	3CO	10.2±0.2	EI	5210
(C ₆ H ₅ CH ₃)(CO) ₅ Cr (Chromium, tricarbonyl[(1,2,3,4,5,6- η)-methylbenzene]-)		12083-24-8	3CO	10.04±0.1	EI	3788
			3CO	10.1±0.2	EI	5210
C₈H₁₀Cr⁺						
(C ₆ H ₄ (CH ₃) ₂)(CO) ₅ Cr (Chromium, tricarbonyl[(1,2,3,4,5,6- η)-1,2-dimethylbenzene]-)		12129-29-2	3CO	9.60±0.1	EI	3788

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₉H₁₂Cr⁺	(C ₆ H ₅ (CH ₃) ₃)(CO) ₃ Cr (Chromium, tricarbonyl[(1,2,3,4,5,6- η)-1,3,5-trimethylbenzene]-)	12129-67-8	3CO	10.35±0.1	EI	3788
C₁₀H₁₀Cr⁺	(C ₅ H ₅) ₂ Cr (Chromocene)	1271-24-5	**	5.50	PE	3725
			**	5.70 (V)	PE	5394
			**	5.71 (V)	PE	5507
C₁₁H₁₁Cr⁺	C ₆ H ₆ Cr(C ₅ H ₅) (Chromium, (η^6 -benzene)(η^5 -2,4-cyclopentadien-1-yl)-)	12093-16-2	**	7.15±0.05 (V)	PE	4132
			**	6.20±0.1 (V)	PE	3686
C₁₂H₁₂Cr⁺	C ₇ H ₇ Cr(C ₅ H ₅) (Chromium, (η^7 -cycloheptatrienylium)(η^5 -2,4-cyclopentadien-1-yl)-)	12093-81-1	**	5.59±0.05 (V)	PE	4132
	(C ₆ H ₆) ₂ Cr (Chromium, bis(benzene)-)	1271-54-1	**	5.4±0.1 (V)	PE	3686
			**	5.45±0.02 (V)	PE	4447
C₁₂H₁₄Cr⁺	(C ₅ H ₄ CH ₃) ₂ Cr (Chromocene, 1,1'-dimethyl-)	12146-92-8	**	5.53 (V)	PE	5507
C₁₂H₁₈Cr⁺	(C ₆ (CH ₃) ₆)(CO) ₃ Cr (Chromium, tricarbonyl[(1,2,3,4,5,6- η)-hexamethylbenzene]-)	12088-11-8	3CO	9.82±0.1	EI	3788
C₁₄H₁₆Cr⁺	(C ₆ H ₅ CH ₃) ₂ Cr (Chromium, bis(η^6 -methyl benzene)-)	12087-58-0	**	5.24±0.1 (V)	PE	3686
C₁₈H₂₄Cr⁺	(C ₆ H ₅ (CH ₃) ₃) ₂ Cr (Chromium, bis[(1,2,3,4,5,6- η)-1,3,5-trimethylbenzene]-)	1274-07-3	**	5.01±0.05 (V)	PE	4132
C₂₀H₃₀Cr⁺	(C ₅ (CH ₃) ₅) ₂ Cr (Chromocene, decamethyl-)	XXXXXX-XX-X	**	4.93 (V)	PE	5394
C₂₀H₄₄Cr⁺	((CH ₃) ₃ CCH ₂) ₄ Cr	37007-84-4	**	7.25±0.1 (V)	PE	3830
C₂₄H₃₆Cr⁺	(C ₆ (CH ₃) ₆) ₂ Cr (Chromium, bis[(1,2,3,4,5,6- η)-hexamethylbenzene]-)	12243-39-9	**	4.68 (V)	PE	5286
C₆H₇NCr⁺	(C ₆ H ₅ NH ₂)(CO) ₃ Cr (Chromium, (η^6 -benzenamine)tricarbonyl-)	12108-11-1	3CO	9.96±0.1	EI	3788
C₁₈H₄₂N₃Cr⁺	(N(iso-C ₃ H ₇) ₂) ₃ Cr	XXXXXX-XX-X	**	6.3 (V)	PE	5036
C₁₆H₄₀N₄Cr⁺	(N(C ₂ H ₅) ₂) ₄ Cr	XXXXXX-XX-X	**	5.9 (V)	PE	5036
COCr⁺	(CO) ₆ Cr CS(CO) ₅ Cr	13007-92-6 50358-90-2	5CO 4CO + CS	14.03±0.04 14.94±0.08	EI EI	5291 5291

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₂O₂Cr⁺						
	(CO) ₆ Cr	13007-92-6	4CO	12.51±0.04	EI	5291
	CS(CO) ₅ Cr	50358-90-2	3CO+CS	13.52±0.08	EI	5291
C₃O₃Cr⁺						
	(CO) ₆ Cr	13007-92-6	3CO	11.35±0.03	EI	5291
	CS(CO) ₅ Cr	50358-90-2	2CO+CS	12.06±0.05	EI	5291
C₄O₄Cr⁺						
	(CO) ₆ Cr	13007-92-6	2CO	10.45±0.03	EI	5291
	CS(CO) ₅ Cr	50358-90-2	CO+CS	11.12±0.05	EI	5291
C₅O₅Cr⁺						
	(CO) ₆ Cr	13007-92-6	CO	9.85±0.03	EI	5291
	CS(CO) ₅ Cr	50358-90-2	CS	10.58±0.07	EI	5291
C₆O₆Cr⁺						
	(CO) ₆ Cr	13007-92-6	**	8.40±0.02 (V)	PE	3979
			**	8.40 (V)	PE	4456
			**	8.40 (V)	PE	5333
			**	8.41 (V)	PE	4692
			**	8.19±0.1	EI	3582
			**	8.20	EI	5453
			**	8.30±0.05	EI	4600
			**	8.42±0.03	EI	5291
C₇H₆OCr⁺						
	(C ₆ H ₆)(CO) ₃ Cr	12082-08-5	2CO	7.9±0.2	EI	3786
	(Chromium, (η^6 -benzene)tricarbonyl-)		2CO	8.09±0.1	EI	3788
C₇H₈OCr⁺						
	(C ₆ H ₅ CH ₂ OH)(CO) ₃ Cr	12116-45-9	3CO	10.35±0.1	EI	3788
	(Chromium, [(1,2,3,4,5,6- η)-benzenemethanol]tricarbonyl-)					
	(C ₆ H ₅ OCH ₃)(CO) ₃ Cr	12116-44-8	3CO	9.90±0.1	EI	3788
	(Chromium, tricarbonyl[(1,2,3,4,5,6- η)-methoxybenzene]-)					
C₈H₈OCr⁺						
	(C ₆ H ₅ CH ₃)(CO) ₃ Cr	12083-24-8	2CO	8.11±0.1	EI	3788
	(Chromium, tricarbonyl[(1,2,3,4,5,6- η)-methylbenzene]-)					
C₉H₁₀OCr⁺						
	(C ₆ H ₅ (CH ₃) ₂)(CO) ₃ Cr	12129-29-2	2CO	7.85±0.1	EI	3788
	(Chromium, tricarbonyl[(1,2,3,4,5,6- η)-1,2-dimethylbenzene]-)					
C₁₀H₁₂OCr⁺						
	(C ₆ H ₅ (CH ₃) ₃)(CO) ₃ Cr	12129-67-8	2CO	8.00±0.1	EI	3788
	(Chromium, tricarbonyl[(1,2,3,4,5,6- η)-1,3,5-trimethylbenzene]-)					
C₁₃H₁₈OCr⁺						
	(C ₆ (CH ₃) ₆)(CO) ₃ Cr	12088-11-8	2CO	7.70±0.1	EI	3788
	(Chromium, tricarbonyl[(1,2,3,4,5,6- η)-hexamethylbenzene]-)					
C₈H₆O₂Cr⁺						
	(C ₆ H ₆)(CO) ₃ Cr	12082-08-5	CO	7.25±0.1	EI	3788
	(Chromium, (η^6 -benzene)tricarbonyl-)		CO	7.4±0.2	EI	3786
C₈H₈O₂Cr⁺						
	(C ₆ H ₆)(CO) ₃ Cr	12116-45-9	2CO	8.19±0.1	EI	3788
	(Chromium, [(1,2,3,4,5,6- η)-benzenemethanol]tricarbonyl-)					
	(C ₆ H ₅ OCH ₃)(CO) ₃ Cr	12116-44-8	2CO	7.90±0.1	EI	3788
	(Chromium, tricarbonyl[(1,2,3,4,5,6- η)-methoxybenzene]-)					

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_8\text{H}_8\text{O}_2\text{Cr}^+$	$(\text{C}_6\text{H}_5\text{COOCH}_3)(\text{CO})_3\text{Cr}$ (Chromium, tricarbonyl[(1,2,3,4,5,6- η)-methylbenzoate]-)	12125-87-0	3CO	9.1	EI	5448
				10.00 ± 0.1	EI	3788
	$\text{C}_{11}\text{H}_8\text{O}_4\text{SCr}$ (Chromium,(carbonothioyl)dicarbonyl[(1,2,3,4,5,6- η)-methylbenzoate]-)	52140-27-9	2CO + CS	11.3	EI	5448
$\text{C}_9\text{H}_8\text{O}_2\text{Cr}^+$	$(\text{C}_6\text{H}_5\text{CH}_3)(\text{CO})_3\text{Cr}$ (Chromium, tricarbonyl[(1,2,3,4,5,6- η)-methylbenzene]-)	12083-24-8	CO	7.09 ± 0.1	EI	3788
$\text{C}_{10}\text{H}_{10}\text{O}_2\text{Cr}^+$	$(\text{C}_6\text{H}_4(\text{CH}_3)_2)(\text{CO})_3\text{Cr}$ (Chromium, tricarbonyl[(1,2,3,4,5,6- η)-1,2-dimethylbenzene]-)	12129-29-2	CO	7.00 ± 0.1	EI	3788
$\text{C}_{11}\text{H}_{12}\text{O}_2\text{Cr}^+$	$(\text{C}_6\text{H}_3(\text{CH}_3)_3)(\text{CO})_3\text{Cr}$ (Chromium, tricarbonyl[(1,2,3,4,5,6- η)-1,3,5-trimethylbenzene]-)	12129-67-8	CO	6.69 ± 0.1	EI	3788
$\text{C}_{14}\text{H}_{18}\text{O}_2\text{Cr}^+$	$(\text{C}_6(\text{CH}_3)_6)(\text{CO})_3\text{Cr}$ (Chromium, tricarbonyl[(1,2,3,4,5,6- η)-hexamethylbenzene]-)	12088-11-8	CO	6.45 ± 0.1	EI	3788
$\text{C}_9\text{H}_6\text{O}_3\text{Cr}^+$	$\text{C}_6\text{H}_6(\text{CO})_3\text{Cr}$ (Chromium, (η^6 -benzene)tricarbonyl-)	12082-08-5	**	7.42 ± 0.03 (V)	PE	4447
			**	6.74 ± 0.1	EI	3788
			**	7.0 ± 0.2	EI	3786
			**	7.28	CTS	4029
$\text{C}_9\text{H}_8\text{O}_3\text{Cr}^+$	$(\text{C}_6\text{H}_5\text{CH}_2\text{OH})(\text{CO})_3\text{Cr}$ (Chromium, [(1,2,3,4,5,6- η)-benzenemethanol]tricarbonyl-)	12116-45-9	CO	7.32 ± 0.1	EI	3788
	$(\text{C}_6\text{H}_5\text{OCH}_3)(\text{CO})_3\text{Cr}$ (Chromium, tricarbonyl[(1,2,3,4,5,6- η)-methoxybenzene]-)	12116-44-8	CO	6.95 ± 0.1	EI	3788
	$(\text{C}_6\text{H}_5\text{COOCH}_3)(\text{CO})_3\text{Cr}$ (Chromium, tricarbonyl[(1,2,3,4,5,6- η)-methylbenzoate]-)	12125-87-0	2CO	7.7	EI	5448
			2CO	8.27 ± 0.1	EI	3788
$\text{C}_{10}\text{H}_8\text{O}_3\text{Cr}^+$	$\text{C}_7\text{H}_6(\text{CO})_3\text{Cr}$ (Chromium, tricarbonyl[1,2,3,4,5,6- η]-1,3,5-cycloheptatriene]-)	12125-72-3	**	7.18 (V)	PE	5206
			**	7.30 ± 0.05 (V)	PE	4724
			**	6.9 ± 0.2	EI	5210
	$(\text{C}_6\text{H}_5\text{CH}_3)(\text{CO})_3\text{Cr}$ (Chromium, tricarbonyl[(1,2,3,4,5,6- η)-methylbenzene]-)	12083-24-8	**	6.6 ± 0.2	EI	5210
			**	6.69 ± 0.1	EI	3788
			**	7.29	CTS	4029
$\text{C}_{11}\text{H}_{10}\text{O}_3\text{Cr}^+$	$(\text{C}_6\text{H}_4(\text{CH}_3)_2)(\text{CO})_3\text{Cr}$ (Chromium, tricarbonyl[(1,2,3,4,5,6- η)-1,2-dimethylbenzene]-)	12129-29-2	**	6.70 ± 0.1	EI	3788
			**	7.29	CTS	4029
$\text{C}_{12}\text{H}_{12}\text{O}_3\text{Cr}^+$	$(\text{C}_6\text{H}_3(\text{CH}_3)_3)(\text{CO})_3\text{Cr}$ (Chromium, tricarbonyl[(1,2,3,4,5,6- η)-1,3,5-trimethylbenzene]-)	12129-67-8	**	7.20 ± 0.05 (V)	PE	4724
			**	7.20 (V)	PE	5286
			**	7.20 (V)	PE	5367
			**	6.60 ± 0.1	EI	3788
			**	7.29	CTS	4029

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{15}\text{H}_{18}\text{O}_3\text{Cr}^+$	$(\text{C}_6(\text{CH}_3)_6)(\text{CO})_3\text{Cr}$ (Chromium, tricarbonyl[(1,2,3,4,5,6- η)-hexamethylbenzene]-)	12088-11-8	**	7.00 (V)	PE	5286
			**	6.35 ± 0.1	EI	3788
$\text{C}_{10}\text{H}_8\text{O}_4\text{Cr}^+$	$(\text{C}_6\text{H}_5\text{CH}_2\text{OH})(\text{CO})_3\text{Cr}$ (Chromium, [(1,2,3,4,5,6- η)-benzenemethanol]tricarbonyl-)	12116-45-9	**	6.92 ± 0.1	EI	3788
	$(\text{C}_6\text{H}_5\text{OCH}_3)(\text{CO})_3\text{Cr}$ (Chromium, tricarbonyl[(1,2,3,4,5,6- η)-methoxybenzene]-)	12116-44-8	**	6.75 ± 0.1	EI	3788
	$(\text{C}_6\text{H}_5\text{COOCH}_3)(\text{CO})_3\text{Cr}$ (Chromium, tricarbonyl[(1,2,3,4,5,6- η)-methylbenzoate]-)	12125-87-0	CO	7.32	CTS	4029
				7.60 ± 0.1	EI	3788
$\text{C}_{11}\text{H}_8\text{O}_4\text{Cr}^+$	$(\text{C}_7\text{H}_9)(\text{CO})_4\text{Cr}$ (Chromium, [(2,3,5,6- η)-bicyclo[2.2.1]hepta-2,5-diene]tetracarbonyl-)	12146-36-0	**	7.28 (V)	PE	5367
$\text{C}_{11}\text{H}_8\text{O}_5\text{Cr}^+$	$(\text{C}_6\text{H}_5\text{COOCH}_3)(\text{CO})_3\text{Cr}$ (Chromium, tricarbonyl[(1,2,3,4,5,6- η)-methylbenzoate]-)	12125-87-0	**	7.02 ± 0.1	EI	3788
			**	7.1	EI	5448
$\text{C}_8\text{H}_6\text{O}_6\text{Cr}^+$	$\text{CH}_3\text{C}(\text{OCH}_3)(\text{CO})_3\text{Cr}$	20540-69-6	**	7.47 (V)	PE	4692
			**	7.46 ± 0.1	EI	3582
$\text{C}_{13}\text{H}_8\text{O}_6\text{Cr}^+$	$\text{C}_6\text{H}_5\text{C}(\text{OCH}_3)(\text{CO})_3\text{Cr}$ (Chromium, pentacarbonyl(methoxyphenylmethylene)-(OC-6-21)-)	27436-93-7	**	7.39 (V)	PE	4692
			**	7.26 ± 0.1	EI	3582
$\text{C}_{14}\text{H}_{10}\text{O}_6\text{Cr}^+$	$(\text{C}_6\text{H}_4(\text{CH}_3)\text{COCH}_3)(\text{CO})_3\text{Cr}$ (Chromium, pentacarbonyl(methoxy(4-methylphenyl)methylene)-(OC-6-21)-)	29160-36-9	**	7.13 ± 0.1	EI	3582
$\text{C}_{15}\text{H}_{21}\text{O}_6\text{Cr}^+$	$(\text{CH}_3\text{COCHCOCH}_3)_3\text{Cr}$ (Chromium, tris(2,4-pentanedionato-O,O')-(OC-6-11)-)	21679-31-2	**	7.46 ± 0.07 (V)	PE	3682
$\text{C}_{11}\text{H}_6\text{O}_7\text{Cr}^+$	$\text{C}_4\text{H}_3\text{OC}(\text{OCH}_3)(\text{CO})_3\text{Cr}$ (Chromium, pentacarbonyl(2-furanylmethoxymethylene)-(OC-6-21)-)	34741-93-0	**	7.37 (V)	PE	4692
$\text{C}_{14}\text{H}_{10}\text{O}_7\text{Cr}^+$	$(\text{C}_6\text{H}_4(\text{OCH}_3)\text{COCH}_3)(\text{CO})_3\text{Cr}$ (Chromium, pentacarbonyl(α -dimethoxybenzylidene)-)	27436-99-3	**	7.05 ± 0.1	EI	3582
$\text{C}_8\text{H}_{12}\text{O}_8\text{Cr}_2^+$	$\text{Cr}_2(\text{O}_2\text{CCH}_3)_4$	15020-15-2	**	8.65 ± 0.05 (V)	PE	4986
$\text{C}_{12}\text{H}_{20}\text{O}_8\text{Cr}_2^+$	$\text{Cr}_2(\text{O}_2\text{CC}_2\text{H}_5)_4$	XXXXX-XX-X	**	8.104 ± 0.05 (V)	PE	4986
$\text{C}_5\text{H}_5\text{NOCr}^+$	$(\text{C}_5\text{H}_5)(\text{CO})_2(\text{NO})\text{Cr}$ (Chromium, dicarbonyl(η^5 -2,4-cyclopentadien-1-yl)nitrosyl-)	36312-04-6	2CO	10.53 ± 0.1	EI	5348
$\text{C}_7\text{H}_7\text{NOCr}^+$	$(\text{C}_6\text{H}_5\text{NH}_2)(\text{CO})_3\text{Cr}$ (Chromium, (η^6 -benzenamine)tricarbonyl-)	12108-11-1	2CO	7.84 ± 0.1	EI	3788

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₆H₅NO₂Cr⁺	(C ₆ H ₅)(CO) ₂ (NO)Cr (Chromium,dicarbonyl(η^5 -2,4-cyclopentadien-1-yl)nitrosyl-)	36312-04-6	CO	9.37±0.1	EI	5348
C₈H₇NO₂Cr⁺	(C ₆ H ₅ NH ₂)(CO) ₃ Cr (Chromium, (η^1 -benzenamine)tricarbonyl-)	12108-11-1	CO	6.75±0.1	EI	3788
C₇H₅NO₃Cr⁺	(C ₆ H ₅)(NO)(CO) ₂ Cr (Chromium, dicarbonyl(η^5 -2,4-cyclopentadien-1-yl)nitrosyl-)	36312-04-6	**	7.80	EI	3579
			**	8.51±0.1	EI	5348
C₉H₇NO₃Cr⁺	(C ₆ H ₅ NH ₂)(CO) ₃ Cr (Chromium, (η^1 -benzenamine)tricarbonyl-)	12108-11-1	**	6.52±0.1	EI	3788
C₁₁H₁₁NO₃Cr⁺	C ₆ H ₅ N(CH ₃) ₂ Cr(CO) ₃ (Chromium, tricarbonyl(<i>N,N</i> -dimethylbenzenamine)-)	12109-10-3	**	7.38	CTS	4029
C₅H₃NO₅Cr⁺	(CO) ₅ NH ₃ Cr	15228-27-0	**	7.56 (V)	PE	4252
			**	7.56 (V)	PE	5540
C₇H₃NO₅Cr⁺	(CO) ₅ CNCH ₃ Cr	33726-04-4	**	7.61 (V)	PE	4252
C₇H₅NO₅Cr⁺	CH ₃ C(NH ₂)(CO) ₅ Cr	22852-50-2	**	7.45 (V)	PE	4692
C₈H₉NO₅Cr⁺	(CO) ₅ N(CH ₃) ₃ Cr	15228-26-9	**	7.45 (V)	PE	4252
C₉H₉NO₅Cr⁺	CH ₃ C(N(CH ₃) ₂)(CO) ₅ Cr	22852-52-4	**	7.12 (V)	PE	4692
C₁₀H₅NO₅Cr⁺	(C ₅ H ₅ N)(CO) ₅ Cr (Chromium,pentacarbonyl(pyridine)-(OC-6-22)-)	14740-77-3	**	7.30 (V)	PE	5566
C₁₀H₁₁NO₅Cr⁺	(C ₅ H ₁₀ NH)(CO) ₅ Cr (Chromium,pentacarbonyl(piperidine)-(OC-6-22))	15710-39-1	**	7.39 (V)	PE	5540
C₁₁H₇NO₅Cr⁺	(CH ₃ C ₅ H ₄ N)(CO) ₅ Cr (Chromium,pentacarbonyl(4-methylpyridine)-(OC-6-22)-)	64914-26-7	**	7.22 (V)	PE	5566
C₁₂H₇NO₅Cr⁺	C ₆ H ₅ C(NH ₂)(CO) ₅ Cr (Chromium, (aminophenylmethylene)pentacarbonyl-(OC-6-21)-)	32370-44-8	**	7.25 (V)	PE	4692
C₁₄H₁₁NO₅Cr⁺	C ₆ H ₅ C(N(CH ₃) ₂)(CO) ₅ Cr (Chromium, pentacarbonyl[(dimethylamino)phenylmethylene]-(OC-6-21)-)	30971-68-7	**	7.02 (V)	PE	4692
C₁₄H₁₃NO₅Cr⁺	(<i>tert</i> -C ₄ H ₉ C ₅ H ₄ N)(CO) ₅ Cr (Chromium,pentacarbonyl[4-(1,1-dimethylethyl)pyridine]-(OC-6-22)-)	64914-25-6	**	7.17 (V)	PE	5566

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_8\text{H}_4\text{N}_2\text{O}_5\text{Cr}^+$	$(\text{C}_3\text{H}_4\text{N}_2)(\text{CO})_5\text{Cr}$ (Chromium, pentacarbonyl(1H-pyrazole-N ²)-(OC-6-22)-)	71127-65-6	**	7.40 (V)	PE	5213
$\text{C}_{12}\text{H}_{14}\text{N}_2\text{O}_5\text{Cr}^+$	$(\text{C}_3\text{H}_4\text{N}_2(\text{C}_2\text{H}_5)_2)(\text{CO})_5\text{Cr}$	XXXXXX-XX-X	**	7.12 (V)	PE	5601
$\text{C}_8\text{H}_3\text{NO}_6\text{Cr}^+$	$(\text{C}_3\text{H}_3\text{NO})(\text{CO})_5\text{Cr}$ (Chromium, pentacarbonyl(isoxazole-N ²)-(OC-6-22)-)	71127-67-8	**	7.42 (V)	PE	5213
$\text{C}_{10}\text{H}_5\text{NO}_6\text{Cr}^+$	$\text{C}_4\text{H}_3\text{OC}(\text{NH}_2)(\text{CO})_5\text{Cr}$ (Chromium, (amino-2-furanylmethylene)pentacarbonyl-(OC-6-21)-)	29130-96-9	**	7.22 (V)	PE	4692
$\text{C}_{11}\text{H}_7\text{NO}_6\text{Cr}^+$	$(\text{CH}_3\text{OC}_5\text{H}_4\text{N})(\text{CO})_5\text{Cr}$ (Chromium, pentacarbonyl(4-methoxypyridine-N ¹)-(OC-6-22)-)	64914-33-6	**	7.18 (V)	PE	5566
$\text{C}_{12}\text{H}_7\text{NO}_6\text{Cr}^+$	$(\text{CH}_3\text{COC}_5\text{H}_4\text{N})(\text{CO})_5\text{Cr}$ (Chromium, pentacarbonyl[1-(4-pyridinyl)ethanone-N]-(OC-6-22)-)	64914-29-0	**	7.5 (V)	PI	5566
$\text{C}_{24}\text{H}_{24}\text{N}_4\text{O}_4\text{Cr}_2^+$	$(\text{C}_5\text{H}_3\text{N}(\text{O})\text{CH}_3)_4\text{Cr}_2$ (Chromium, tetrakis[μ-(6-methyl-2(1H)-pyridinato-N ¹ :O ²)]di-(Cr-Cr), stereoisomer)	67634-82-6	**	6.8 (V)	PE	5191
FCr^+						
	CrF	13943-42-5	**	9.3 ± 0.4	EI	5440
	CrF ₂	10049-10-2	F	14.7 ± 0.5	EI	5440
F_2Cr^+						
	CrF ₂	10049-10-2	**	10.6 ± 0.3	EI	5440
	CrF ₃	7788-97-8	F	14.8 ± 0.5	EI	5440
F_3Cr^+						
	CrF ₃	7788-97-8	**	12.5 ± 0.3	EI	5440
$\text{C}_{13}\text{H}_7\text{O}_6\text{FCr}^+$	$(\text{C}_6\text{H}_4\text{FCOCH}_3)(\text{CO})_5\text{Cr}$ (Chromium, pentacarbonyl[(4-fluorophenyl)methoxymethylene]-, (OC-6-21)-)	27436-94-8	**	7.32 ± 0.1	EI	3582
$\text{C}_{14}\text{H}_7\text{O}_6\text{F}_3\text{Cr}^+$	$(\text{C}_6\text{H}_4(\text{CF}_3)\text{COCH}_3)(\text{CO})_5\text{Cr}$ (Chromium, pentacarbonyl[α-methoxy-o-(trifluoromethyl)benzylidene]-)	32011-10-2	**	7.34 ± 0.1	EI	3582
	$(\text{C}_6\text{H}_4(\text{CF}_3)\text{COCH}_3)(\text{CO})_5\text{Cr}$ (Chromium, pentacarbonyl[methoxy[4-(trifluoromethyl)phenyl]methylene]-, (OC-6-21)-)	27637-27-0	**	7.42 ± 0.1	EI	3582
$\text{C}_{15}\text{H}_{12}\text{O}_6\text{F}_9\text{Cr}^+$	$(\text{CF}_3\text{COCHCOCH}_3)_3\text{Cr}$ (Chromium, tris(1,1,1-trifluoro-2,4-pentanedionato-O,O')	14592-89-3	**	8.58 ± 0.07 (V)	PE	3682
$\text{C}_{15}\text{H}_3\text{O}_6\text{F}_{18}\text{Cr}^+$	$(\text{CF}_3\text{COCHCOCF}_3)_3\text{Cr}$ (Chromium, tris(1,1,1,5,5-hexafluoro-2,4-pentanedionato-O,O')-	14592-80-4	**	9.53 (V)	PE	3681
			**	9.57 ± 0.07 (V)	PE	3682
$\text{C}_{16}\text{H}_{44}\text{Si}_4\text{Cr}^+$	$((\text{CH}_3)_3\text{SiCH}_2)_4\text{Cr}$	35394-18-4	**	7.26 ± 0.1 (V)	PE	3830

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{15}\text{H}_{24}\text{O}_6\text{Si}_2\text{Cr}^+$ $\text{C}_{15}\text{H}_{24}\text{O}_6\text{Si}_2\text{Cr}$		XXXXX-XX-X **		7.57 (V)	PE	5601
$\text{C}_{12}\text{H}_{27}\text{PCr}^+$ $((n\text{-C}_4\text{H}_9)_3\text{P})(\text{CO})_5\text{Cr}$		18497-59-1	5CO	11.05	EI	5564
$\text{C}_{18}\text{H}_{15}\text{PCr}^+$ $((\text{C}_6\text{H}_5)_3\text{P})(\text{CO})_5\text{Cr}$ (Chromium,pentacarbonyl(triphenylphosphine)-(OC-6-22))		14917-12-5	5CO	10.9	EI	5564
$\text{C}_6\text{H}_{18}\text{N}_3\text{PCr}^+$ $((\text{CH}_3)_2\text{N})_3\text{P})(\text{CO})_5\text{Cr}$ $((\text{CH}_3)_2\text{N})_3\text{P}_2(\text{CO})_4\text{Cr}$		XXXXX-XX-X 19976-85-3	5CO	12.5 ± 0.05 11.0 ± 0.05	EI EI	3952 3952
$\text{C}_{13}\text{H}_{27}\text{OPCr}^+$ $((n\text{-C}_4\text{H}_9)_3\text{P})(\text{CO})_5\text{Cr}$		18497-59-1	4CO	9.65	EI	5564
$\text{C}_{19}\text{H}_{15}\text{OPCr}^+$ $((\text{C}_6\text{H}_5)_3\text{P})(\text{CO})_5\text{Cr}$ (Chromium,pentacarbonyl(triphenylphosphine)-(OC-6-22))		14917-12-5	4CO	9.7	EI	5564
$\text{C}_{14}\text{H}_{27}\text{O}_2\text{PCr}^+$ $((n\text{-C}_4\text{H}_9)_3\text{P})(\text{CO})_5\text{Cr}$		18497-59-1	3CO	8.85	EI	5564
$\text{C}_{20}\text{H}_{15}\text{O}_2\text{PCr}^+$ $((\text{C}_6\text{H}_5)_3\text{P})(\text{CO})_5\text{Cr}$ (Chromium,pentacarbonyl(triphenylphosphine)-(OC-6-22))		14917-12-5	3CO	9.3	EI	5564
$\text{C}_3\text{H}_9\text{O}_3\text{PCr}^+$ $(\text{P}(\text{OCH}_3)_3)(\text{CO})_5\text{Cr}$		18461-34-2	5CO	11.5	EI	5564
$\text{C}_6\text{H}_{15}\text{O}_3\text{PCr}^+$ $(\text{P}(\text{OC}_2\text{H}_5)_3)(\text{CO})_5\text{Cr}$		18461-32-0	5CO	11.8	EI	5564
$\text{C}_{15}\text{H}_{27}\text{O}_3\text{PCr}^+$ $((n\text{-C}_4\text{H}_9)_3\text{P})(\text{CO})_5\text{Cr}$		18497-59-1	2CO	8.55	EI	5564
$\text{C}_{21}\text{H}_{15}\text{O}_3\text{PCr}^+$ $((\text{C}_6\text{H}_5)_3\text{P})(\text{CO})_5\text{Cr}$ (Chromium,pentacarbonyl(triphenylphosphine)-(OC-6-22))		14917-12-5	2CO	9.1	EI	5564
$\text{C}_4\text{H}_9\text{O}_4\text{PCr}^+$ $(\text{P}(\text{OCH}_3)_3)(\text{CO})_5\text{Cr}$		18461-34-2	4CO	10.1	EI	5564
$\text{C}_7\text{H}_{15}\text{O}_4\text{PCr}^+$ $(\text{P}(\text{OC}_2\text{H}_5)_3)(\text{CO})_5\text{Cr}$		18461-32-0	4CO	10.0	EI	5564
$\text{C}_{16}\text{H}_{27}\text{O}_4\text{PCr}^+$ $((n\text{-C}_4\text{H}_9)_3\text{P})(\text{CO})_5\text{Cr}$		18497-59-1	CO	8.25	EI	5564
$\text{C}_{22}\text{H}_{15}\text{O}_4\text{PCr}^+$ $((\text{C}_6\text{H}_5)_3\text{P})(\text{CO})_5\text{Cr}$ (Chromium,pentacarbonyl(triphenylphosphine)-(OC-6-22))		14917-12-5	CO	8.5	EI	5564
$\text{C}_5\text{H}_3\text{O}_5\text{PCr}^+$ $(\text{CO})_5\text{PH}_3\text{Cr}$		18116-53-5	**	7.90 (V)	PE	4252
$\text{C}_5\text{H}_9\text{O}_5\text{PCr}^+$ $(\text{P}(\text{OCH}_3)_3)(\text{CO})_5\text{Cr}$		18461-34-2	3CO	9.4	EI	5564
$\text{C}_8\text{H}_9\text{O}_5\text{PCr}^+$ $(\text{CO})_5\text{P}(\text{CH}_3)_3\text{Cr}$		26555-09-9	**	7.58 (V)	PE	4252

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_8\text{H}_9\text{O}_5\text{PCr}^+$	$(\text{CO})_5\text{P}(\text{CH}_3)_3\text{Cr}$	26555-09-9	**	7.6	PE	5602
$\text{C}_{8}\text{H}_{15}\text{O}_5\text{PCr}^+$	$(\text{P}(\text{OC}_2\text{H}_5)_3)(\text{CO})_5\text{Cr}$	18461-32-0	3CO	9.2	EI	5564
$\text{C}_{11}\text{H}_{15}\text{O}_5\text{PCr}^+$	$((\text{C}_2\text{H}_5)_3\text{P})(\text{CO})_5\text{Cr}$	21321-30-2	**	7.6	PE	5602
$\text{C}_{11}\text{H}_{17}\text{O}_5\text{PCr}^+$	$\text{C}_{13}\text{H}_{17}\text{O}_7\text{PCr}$ (Chromium,dicarbonyl[(1,2,3,4,5,6- η)-methylbenzoate](trimethyl phosphite-P)-)	53248-14-9	2CO	7.2	EI	5448
$\text{C}_{23}\text{H}_{15}\text{O}_5\text{PCr}^+$	$(\text{C}_6\text{H}_5)_3\text{P}(\text{CO})_5\text{Cr}$ (Chromium, pentacarbonyl (triphenylphosphine)-(OC-6-22)-)	14917-12-5	**	7.30 (V)	PE	5139
			**	7.40 \pm 0.05	EI	4600
$\text{C}_{23}\text{H}_{33}\text{O}_5\text{PCr}^+$	$(\text{C}_6\text{H}_{11})_3\text{P}(\text{CO})_5\text{Cr}$ (Chromium, pentacarbonyl (tricyclohexylphosphine)-(OC-6-22)-)	15603-93-7	**	7.24 (V)	PE	5139
$\text{C}_{26}\text{H}_{23}\text{O}_5\text{PCr}^+$	$\text{C}_{28}\text{H}_{21}\text{O}_7\text{PCr}$ (Chromium,dicarbonyl[(1,2,3,4,5,6- η)-methyl benzoate](triphenyl phosphite-P)-)	63928-77-8	2CO	7.2	EI	5448
$\text{C}_6\text{H}_6\text{O}_6\text{PCr}^+$	$(\text{P}(\text{OCH}_3)_3)(\text{CO})_5\text{Cr}$	18461-34-2	$\text{OCH}_3 + \text{CO}$	11.9	EI	5564
$\text{C}_6\text{H}_9\text{O}_6\text{PCr}^+$	$(\text{P}(\text{OCH}_3)_3)(\text{CO})_5\text{Cr}$	18461-34-2	2CO	8.7	EI	5564
$\text{C}_9\text{H}_{15}\text{O}_6\text{PCr}^+$	$(\text{P}(\text{OC}_2\text{H}_5)_3)(\text{CO})_5\text{Cr}$	18461-32-0	2CO	8.8	EI	5564
$\text{C}_7\text{H}_6\text{O}_7\text{PCr}^+$	$(\text{P}(\text{OCH}_3)_3)(\text{CO})_5\text{Cr}$	18461-34-2	OCH_3	10.8	EI	5564
$\text{C}_7\text{H}_9\text{O}_7\text{PCr}^+$	$(\text{P}(\text{OCH}_3)_3)(\text{CO})_5\text{Cr}$	18461-34-2	CO	8.25	EI	5564
$\text{C}_9\text{H}_{10}\text{O}_7\text{PCr}^+$	$(\text{P}(\text{OC}_2\text{H}_5)_3)(\text{CO})_5\text{Cr}$	18461-32-0	OC_2H_5	11.1	EI	5564
$\text{C}_{10}\text{H}_{15}\text{O}_7\text{PCr}^+$	$(\text{P}(\text{OC}_2\text{H}_5)_3)(\text{CO})_5\text{Cr}$	18461-32-0	CO	8.3	EI	5564
$\text{C}_{13}\text{H}_{17}\text{O}_7\text{PCr}^+$	$\text{C}_{13}\text{H}_{17}\text{O}_7\text{PCr}$ (Chromium,dicarbonyl[(1,2,3,4,5,6- η)-methylbenzoate](trimethyl phosphite-P)-)	53248-14-9	**	6.6	EI	5448
$\text{C}_{28}\text{H}_{23}\text{O}_7\text{PCr}^+$	$\text{C}_{28}\text{H}_{23}\text{O}_7\text{PCr}$ (Chromium,dicarbonyl[(1,2,3,4,5,6- η)-methyl benzoate](triphenyl phosphite-P)-)	63928-77-8	**	6.8	EI	5448
$\text{C}_8\text{H}_9\text{O}_8\text{PCr}^+$	$((\text{CH}_3\text{O})_3\text{P})(\text{CO})_5\text{Cr}$	18461-34-2	**	8.0	PE	5602

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₁H₁₅O₈PCr⁺ ((C ₂ H ₅ O) ₃ P)(CO) ₅ Cr		18461-32-0	**	7.9	PE	5602
C₁₄H₂₁O₈PCr⁺ (<i>iso</i> -C ₃ H ₇ O) ₃ P(CO) ₅ Cr		XXXXX-XX-X	**	7.61 (V)	PE	5139
C₂₃H₁₅O₈PCr⁺ (C ₆ H ₅ O) ₃ P(CO) ₅ Cr (Chromium, pentacarbonyl (triphenyl phosphite-P)-(OC-6-22)-)		18461-39-7	**	7.67 (V)	PE	5139
C₇H₁₈N₃OPCr⁺ (((CH ₃) ₂ N) ₃ P)(CO) ₅ Cr		XXXXX-XX-X	4CO	9.8±0.05	EI	3952
C₉H₁₈N₃O₃PCr⁺ (((CH ₃) ₂ N) ₃ P)(CO) ₅ Cr		XXXXX-XX-X	2CO	8.6±0.05	EI	3952
C₁₀H₁₈N₃O₄PCr⁺ (((CH ₃) ₂ N) ₃ P)(CO) ₅ Cr		XXXXX-XX-X	CO	7.6±0.05	EI	3952
C₁₁H₁₈N₃O₅PCr⁺ (((CH ₃) ₂ N) ₃ P)(CO) ₅ Cr		15137-66-3	** **	6.6±0.05 7.6	EI PE	3952 5602
C₁₆H₃₀N₂O₃P₂Cr⁺ C ₁₆ H ₃₀ N ₂ O ₃ P ₂ Cr		XXXXX-XX-X	**	6.70 (V)	PE	5601
C₁₅H₃₆N₆O₃P₂Cr⁺ (((CH ₃) ₂ N) ₃ P) ₂ (CO) ₄ Cr		19976-85-3	CO	9.5±0.05	EI	3952
C₁₆H₃₆N₆O₄P₂Cr⁺ (((CH ₃) ₂ N) ₃ P) ₂ (CO) ₄ Cr		19976-85-3	**	6.5±0.05	EI	3952
F₁₈P₆Cr⁺ (PF ₃) ₆ Cr		26117-61-3	** **	9.0 9.29 (V)	PE PE	4021 4456
C₃H₉N₃F₁₂P₆Cr⁺ (CH ₃ N(PF ₂) ₂) ₃ Cr		63404-40-0	**	7.70 (V)	PE	5376
C₅O₅F₃PCr⁺ (PF ₃)(CO) ₅ Cr		18461-42-2	** ** **	8.56 (V) 8.7 8.70	PE PE EI	5539 5602 5453
C₄O₄F₆P₂Cr⁺ (PF ₃) ₂ (CO) ₄ Cr		31616-42-9	**	8.85	EI	5453
C₃O₃F₉P₃Cr⁺ (PF ₃) ₃ (CO) ₃ Cr		31616-43-0	**	8.90	EI	5453
CSCr⁺ CS(CO) ₅ Cr		50358-90-2	5CO	13.68±0.04	EI	5291
C₅H₅NSCr⁺ (C ₅ H ₅)(CO) ₂ (NS)Cr (Chromium,dicarbonyl(η^5 -2,4-cyclopentadien-1-yl)thionitrosyl-)		66539-91-1	2CO	9.07±0.1	EI	5348
C₂OSCr⁺ CS(CO) ₅ Cr		50358-90-2	4CO	12.12±0.04	EI	5291
C₃O₂SCr⁺ CS(CO) ₅ Cr		50358-90-2	3CO	11.12±0.04	EI	5291

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_4\text{O}_3\text{SCr}^+$	$\text{CS}(\text{CO})_5\text{Cr}$	50358-90-2	2CO	10.22 ± 0.04	EI	5291
$\text{C}_5\text{O}_4\text{SCr}^+$	$\text{CS}(\text{CO})_5\text{Cr}$	50358-90-2	CO	9.39 ± 0.04	EI	5291
$\text{C}_6\text{O}_5\text{SCr}^+$	$(\text{CS})(\text{CO})_5\text{Cr}$	50358-90-2	** ** **	8.16 (V) 8.16 (V) 8.31 ± 0.03	PE PE EI	5333 5518 5291
$\text{C}_9\text{H}_8\text{O}_2\text{SCr}^+$	$(\text{C}_6\text{H}_5\text{COOCH}_3)(\text{CS})(\text{CO})_2\text{Cr}$ (Chromium,(carbonothioyl)dicarbonyl[(1,2,3,4,5,6- η)-methylbenzoate]-)	52140-27-9	2CO	9.2	EI	5448
$\text{C}_{11}\text{H}_8\text{O}_4\text{SCr}^+$	$(\text{C}_6\text{H}_5\text{COOCH}_3)(\text{CS})(\text{CO})_2\text{Cr}$ (Chromium,(carbonothioyl)dicarbonyl[(1,2,3,4,5,6- η)-methylbenzoate]-)	52140-27-9	**	7.8	EI	5448
$\text{C}_8\text{H}_6\text{O}_5\text{SCr}^+$	$\text{CH}_3\text{C}(\text{SCH}_3)(\text{CO})_5\text{Cr}$	35797-92-3	**	7.35 (V)	PE	4692
$\text{C}_8\text{H}_8\text{O}_5\text{SCr}^+$	$(\text{S}(\text{CH}_2)_4)(\text{CO})_5\text{Cr}$ (Chromium,pentacarbonyl(tetrahydrothiophene)-)	15038-40-1	** **	7.45 7.45 ± 0.05	EI EI	5292 3498
$\text{C}_7\text{H}_6\text{O}_6\text{SCr}^+$	$(\text{SO}(\text{CH}_3)_2)(\text{CO})_5\text{Cr}$	36083-80-4	** **	7.64 7.64 ± 0.05	EI EI	5292 3498
$\text{C}_7\text{H}_4\text{O}_8\text{SCr}^+$	$(\text{SO}(\text{OCH}_2)_2)(\text{CO})_5\text{Cr}$	36252-44-5	** **	7.80 7.80 ± 0.05	EI EI	5292 3498
$\text{C}_6\text{H}_5\text{NOSCr}^+$	$(\text{C}_5\text{H}_5)(\text{CO})_2(\text{NS})\text{Cr}$ (Chromium,dicarbonyl(η^5 -2,4-cyclopentadien-1-yl)thionitrosyl-)	66539-91-1	CO	8.15 ± 0.1	EI	5348
$\text{C}_7\text{H}_5\text{NO}_2\text{SCr}^+$	$(\text{C}_5\text{H}_5)(\text{CO})_2(\text{NS})\text{Cr}$ (Chromium,dicarbonyl(η^5 -2,4-cyclopentadien-1-yl)thionitrosyl-)	66539-91-1	**	7.83 ± 0.1	EI	5348
$\text{C}_8\text{H}_3\text{NO}_5\text{SCr}^+$	$(\text{C}_3\text{H}_3\text{NS})(\text{CO})_5\text{Cr}$ (Chromium,pentacarbonyl(isothiazole-N ²)-(OC-6-22)-) $(\text{C}_3\text{H}_3\text{NS})(\text{CO})_5\text{Cr}$ (Chromium,pentacarbonyl(thiazole-N ³)-(OC-6-22)-)	39554-14-8 55293-31-7	** **	7.32 (V) 7.36 (V)	PE PE	5213 5213
$\text{C}_{27}\text{H}_{23}\text{O}_5\text{PSCr}^+$	$\text{C}_{28}\text{H}_{23}\text{O}_6\text{PSCr}$ (Chromium,(carbonothioyl)carbonyl[(1,2,3,4,5,6- η)-methylbenzoate](triphenyl phosphite-P)-)	57546-01-7	CO	8.8 ± 0.3	EI	5448
$\text{C}_{28}\text{H}_{23}\text{O}_6\text{PSCr}^+$	$\text{C}_{28}\text{H}_{23}\text{O}_6\text{PSCr}$ (Chromium,(carbonothioyl)carbonyl[(1,2,3,4,5,6- η)-methylbenzoate](triphenyl phosphite-P)-)	57546-01-7	**	7.4	EI	5448
$\text{C}_{12}\text{H}_{30}\text{O}_6\text{P}_3\text{S}_6\text{Cr}^+$	$((\text{C}_2\text{H}_5)_2\text{S}_2\text{PO}_2)_3\text{Cr}$	14177-95-8	**	7.71 (V)	PE	5203

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
Cl₂Cr⁺	CrCl ₂	10049-05-5	**	9.97 (V)	PE	5172
C₆H₅ClCr⁺	(C ₆ H ₅ Cl)(CO) ₃ Cr (Chromium, tricarbonyl(η^6 -chlorobenzene)-)	12082-03-0	3CO	10.10±0.1	EI	3788
O₂Cl₂Cr⁺	CrO ₂ Cl ₂	14977-61-8	** **	11.8 (V) 11.85±0.03 (V)	PE PE	4455 5148
C₇H₅OClCr⁺	(C ₆ H ₅ Cl)(CO) ₃ Cr (Chromium, tricarbonyl(η^6 -chlorobenzene)-)	12082-03-0	2CO	8.18±0.1	EI	3788
C₈H₅O₂ClCr⁺	(C ₆ H ₅ Cl)(CO) ₃ Cr (Chromium, tricarbonyl(η^6 -chlorobenzene)-)	12082-03-0	CO	7.45±0.1	EI	3788
C₉H₅O₃ClCr⁺	(C ₆ H ₅ Cl)(CO) ₃ Cr (Chromium, tricarbonyl(η^6 -chlorobenzene)-)	12082-03-0	**	7.00±0.1	EI	3788
C₁₃H₇O₆ClCr⁺	(C ₆ H ₅ ClCOCH ₃)(CO) ₃ Cr (Chromium, pentacarbonyl[(4-chlorophenyl)methoxymethylene]-, (OC-6-21)-)	29160-37-0	**	7.34±0.1	EI	3582
C₁₀H₄NO₅ClCr⁺	(ClC ₅ H ₄ N)(CO) ₃ Cr (Chromium, pentacarbonyl(4-chloropyridine)-(OC-6-22)-)	64914-28-9	**	7.42 (V)	PI	5566
C₅O₅PCl₃Cr⁺	(PCl ₃)(CO) ₃ Cr	18461-41-1	** **	8.32 (V) 8.26	PE EI	5539 5453
Mn⁺						
(³ S)	Mn	7439-96-5	**	7.434	S	5497
(³ S)			**	7.43	PE	4858
(³ S)			**	8.61	PE	4858
(³ D)			**	14.26	PE	4858
	C ₅ H ₅ (CO) ₃ Mn	12079-65-1	3CO + C ₅ H ₅	15.32±0.02	EI	4661
	(Manganese, tricarbonyl(η^5 -2,4-cyclopentadien-1-yl)-)					
	C ₅ H ₄ (CH ₃)(CO) ₃ Mn	12108-13-3	3CO + C ₆ H ₇	16.33±0.02	EI	4661
	(Manganese, tricarbonyl[1,2,3,4,5- η]-1-methyl-2,4-cyclopentadien-1-yl]-)					
	(CO) ₅ MnH	16972-33-1		17.3	EI	3814
	((CH ₃) ₃ Si)(CO) ₅ Mn	26500-16-3		21.7	EI	3814
	((CH ₃) ₃ Si)(CO) ₄ (PF ₃)Mn	33989-27-4		21.9	EI	3814
	C ₅ H ₅ (CO) ₂ CSMn	31741-76-1	2CO + CS + C ₅ H ₅	16.51±0.04	EI	4661
	(Manganese, (carbonothioyl)dicarbonyl(η^5 -2,4-cyclopentadien-1-yl)-)					
	C ₅ H ₄ (CH ₃)(CO) ₂ CSMn	49716-52-1	2CO + CS + C ₆ H ₇	16.22±0.02	EI	4661
	(Manganese, (carbonothioyl)dicarbonyl[(1,2,3,4,5- η)-1-methyl-2,4-cyclopentadien-1-yl]-)					
	(C ₅ H ₅) ₂ (CS) ₂ (NO) ₂ Mn ₂	64090-73-9		26.71±0.03	EI	5423
	(Manganese, bis[μ -(carbonothioyl)bis(η^5 -2,4-cyclopentadien-1-yl)]dinitrosyldi-(Mn-Mn)-)					
	(C ₅ H ₅)(CS)(NO)MnI	58450-74-1		16.57±0.03	EI	5561
	(Manganese, (carbonothioyl)(η^5 -2,4-cyclopentadien-1-yl)iodonitrosyl-)					
	(CH ₃ C ₅ H ₄)(CS)(NO)MnI	XXXXX-XX-X		15.57±0.07	EI	5561
	(Manganese, (carbonothioyl)][1,2,3,4,5- η]-1-methyl-2,4-cyclopentadien-1-yl]iodonitrosyl-)					

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
Mn₂⁺	(C ₅ H ₅) ₂ (CS) ₂ (NO) ₂ Mn ₂ (Manganese,bis[μ-(carbonothioyl)bis(η ⁵ -2,4-cyclopentadien-1-yl)-dinitrosyldi-(Mn-Mn)-])	64090-73-9		28.38±0.02	EI	5423
HMn⁺	(CO) ₅ MnH	16972-33-1	5CO	13.8	EI	3814
C₈H₅Mn⁺	C ₅ H ₅ (CO) ₃ Mn (Manganese, tricarbonyl(η ⁵ -2,4-cyclopentadien-1-yl)-) C ₅ H ₅ (CO) ₂ CSMn (Manganese, (carbonothioyl)dicarbonyl(η ⁵ -2,4-cyclopentadien-1-yl)-) (C ₅ H ₅) ₂ (CS) ₂ (NO) ₂ Mn ₂ (Manganese,bis[μ-(carbonothioyl)bis(η ⁵ -2,4-cyclopentadien-1-yl)-dinitrosyldi-(Mn-Mn)-]) (C ₅ H ₅)(CS)(NO)MnI (Manganese,(carbonothioyl)(η ⁵ -2,4-cyclopentadien-1-yl)iodonitrosyl-)	12079-65-1 31741-76-1 64090-73-9 58450-74-1	3CO 2CO + CS NO + CS + I	11.67±0.04 12.25±0.03 21.33±0.16 12.84±0.03	EI EI EI EI	4661 4661 5423 5561
C₆H₇Mn⁺	C ₅ H ₄ (CH ₃)(CO) ₃ Mn (Manganese, tricarbonyl[(1,2,3,4,5-η)-1-methyl-2,4-cyclopentadien-1-yl]-) (CH ₃ C ₆ H ₄)(CO) ₂ ((C ₆ H ₅) ₃ P)Mn (Manganese,dicarbonyl[(1,2,3,4,5-η)-1-methyl-2,4-cyclopentadien-1-yl] (triphenylphosphine)-) C ₅ H ₄ (CH ₃)(CO) ₂ CSMn (Manganese, (carbonothioyl)dicarbonyl[(1,2,3,4,5-η)-1-methyl- 2,4-cyclopentadien-1-yl]-) C ₂₆ H ₂₂ OPSMn (Manganese,(carbonothioyl)carbonyl[(1,2,3,4,5-η)-1-methyl-2,4- cyclopentadien-1-yl](triphenylphosphine)-) (CH ₃ C ₆ H ₄)(CO) ₂ ((C ₆ H ₅) ₃ As)Mn (Manganese,dicarbonyl[(1,2,3,4,5-η)-1-methyl-2,4- cyclopentadien-1-yl](triphenylarsine)-) C ₂₆ H ₂₂ OSMnAs (Manganese,(carbonothioyl)carbonyl[(1,2,3,4,5-η)-1- methyl-2,4-cyclopentadien-1-yl](triphenylarsine)-) C ₂₆ H ₂₂ O ₂ MnSb (Manganese,dicarbonyl[(1,2,3,4,5-η)-1-methyl-2,4-cyclopentadien- 1-yl](triphenylstibine)-) (CH ₃ C ₆ H ₄)(CO)(CS)((C ₆ H ₅) ₃ Sb)Mn (Manganese,(carbonothioyl)carbonyl[(1,2,3,4,5-η)-1-methyl- 2,4-cyclopentadien-1-yl](triphenylstibine)-) (CH ₃ C ₆ H ₄)(CS)(NO)MnI (Manganese,(carbonothioyl)[(1,2,3,4,5-η)-1-methyl- 2,4-cyclopentadien-1-yl]iodonitrosyl-)	12108-13-3 12100-95-7 49716-52-1 70279-43-5 XXXXX-XX-X XXXXX-XX-X XXXXX-XX-X XXXXX-XX-X XXXXX-XX-X XXXXX-XX-X XXXXX-XX-X	3CO 2CO + CS NO + CS + I	11.21±0.03 15.97±0.09 12.20±0.01 16.23±0.02 14.73±0.02 15.68±0.06 14.51±0.04 14.95±0.10 13.19±0.04	EI EI EI EI EI EI EI EI EI	4661 5576 4661 5576 5576 5576 5576 5576 5561
C₁₀H₁₀Mn⁺	(C ₅ H ₅) ₂ Mn (Manganocene)	1271-27-8	**	6.26 (V)	PE	5394
			**	6.26 (V)	PE	5507
			**	6.55	PE	3725
	(C ₅ H ₅) ₂ (CS) ₂ (NO) ₂ Mn ₂ (Manganese,bis[μ-(carbonothioyl)bis(η ⁵ -2,4-cyclopentadien-1-yl)-dinitrosyldi-(Mn-Mn)-])	64090-73-9		16.16±0.03	EI	5423
C₁₁H₁₁Mn⁺	(C ₅ H ₅)(C ₆ H ₆)Mn (Manganese, (η ⁶ -benzene)(η ⁵ -2,4-cyclopentadien-1-yl)-)	1271-43-8	**	6.36±0.1 (V)	PE	3686
C₁₂H₁₄Mn⁺	(C ₅ H ₄ CH ₃) ₂ Mn (Manganocene,1,1'-dimethyl-)	32985-17-4	**	6.01 (V)	PE	5507

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₂H₁₄Mn⁺	(C ₅ H ₄ CH ₃) ₂ Mn	32985-17-4	**	6.06 (V)	PE	5394
C₂₀H₃₀Mn⁺	(C ₅ (CH ₃) ₂) ₂ Mn (Manganocene, decamethyl-)	XXXXX-XX-X	**	5.33 (V)	PE	5394
C₄₄H₂₈N₄Mn⁺	C ₂₀ H ₈ N ₄ (C ₆ H ₅) ₄ Mn (Manganese, [5,10,15,20-tetraphenyl-21H,23H-porphinato(2-)N ²¹ ,N ²² ,N ²³ ,N ²⁴]-(SP-4-1)-)	31004-82-7	**	6.44 (V)	PE	4557
C₃₂H₁₆N₈Mn⁺	C ₃₂ H ₁₆ N ₈ Mn (Manganese, [29H,31H-phthalocyaninato(2')-N ²⁹ ,N ³⁰ ,N ³¹ ,N ³²]-(SP-4-1)-)	14325-24-7	**	7.26±0.10	EI	3829
COMn⁺	((CH ₃) ₃ Si)(CO) ₅ Mn	26500-16-3		17.9	EI	3814
C₂O₂Mn⁺	(CO) ₅ MnH	16972-33-1		13.7	EI	3814
C₃O₃Mn⁺	(CO) ₅ MnH	16972-33-1		13.2	EI	3814
C₄O₄Mn⁺	(CO) ₅ MnH	16972-33-1		11.2	EI	3814
C₁₀O₁₀Mn₂⁺	(CO) ₁₀ Mn ₂	10170-69-1	**	8.02 (V)	PE	4492
CHOMn⁺	(CO) ₅ MnH	16972-33-1	4CO	12.7	EI	3814
C₆H₅OMn⁺	C ₅ H ₅ (CO) ₃ Mn (Manganese, tricarbonyl(η^5 -2,4-cyclopentadien-1-yl)-)	12079-65-1	2CO	9.28±0.01	EI	4661
C₇H₇OMn⁺	C ₅ H ₄ (CH ₃)(CO) ₃ Mn (Manganese, tricarbonyl[1,2,3,4,5- η]-1-methyl-2,4-cyclopentadien-1-yl]-)	12108-13-3	2CO	9.01±0.03	EI	4661
C₂HO₂Mn⁺	(CO) ₅ MnH	16972-33-1	3CO	10.3	EI	3814
C₇H₅O₂Mn⁺	C ₅ H ₅ (CO) ₃ Mn (Manganese, tricarbonyl(η^5 -2,4-cyclopentadien-1-yl)-)	12079-65-1	CO	8.37±0.01	EI	4661
C₈H₇O₂Mn⁺	C ₅ H ₄ (CH ₃)(CO) ₃ Mn (Manganese, tricarbonyl[1,2,3,4,5- η]-1-methyl-2,4-cyclopentadien-1-yl]-)	12108-13-3	CO	8.13±0.01	EI	4661
C₃HO₃Mn⁺	(CO) ₅ MnH	16972-33-1	2CO	9.9	EI	3814
C₈H₅O₃Mn⁺	C ₅ H ₅ (CO) ₃ Mn (Manganese, tricarbonyl (η^5 -2,4-cyclopentadien-1-yl)-)	12079-65-1	**	8.05 (V)	PE	4570
			**	8.06±0.01	EI	4661
			**	8.12±0.1	EI	3578
			**	8.12	EI	5453

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₉H₇O₃Mn⁺						
	C ₆ H ₇ (CO) ₃ Mn (Manganese, tricarbonyl[1,2,3,4,5- η]-2,4-cyclohexadien-1-yl]-)	12108-14-4	**	8.06±0.05 (V)	PE	4501
	C ₅ H ₄ (CH ₃)(CO) ₃ Mn (Manganese, tricarbonyl[(1,2,3,4,5- η)-1-methyl-2,4-cyclopentadien-1-yl]-)	12108-13-3	**	~8.1	PE	4995
			**	7.86±0.01	EI	4661
C₁₀H₇O₃Mn⁺						
	C ₇ H ₇ (CO) ₃ Mn (Manganese, tricarbonyl[(1,2,3,4,5- η)-2,4,6-cycloheptatrien-1-yl]-)	53011-14-6	**	7.78±0.05 (V)	PE	4501
C₁₀H₉O₃Mn⁺						
	C ₇ H ₉ (CO) ₃ Mn (Manganese, tricarbonyl[1,2,3,4,5- η]-2,4-cycloheptadien-1-yl]-)	32798-86-0	**	7.86±0.05 (V)	PE	4501
C₄HO₄Mn⁺						
	(CO) ₅ MnH	16972-33-1	CO	8.7	EI	3814
C₅HO₅Mn⁺						
	(CO) ₅ MnH	16972-33-1	**	8.85 (V)	PE	4448
			**	8.85 (V)	PE	4456
			**	8.5±0.1	EI	3814
C₆H₃O₅Mn⁺						
	CH ₃ (CO) ₅ Mn	13601-24-6	**	8.65 (V)	PE	4110
C₁₅H₂₁O₆Mn⁺						
	(CH ₃ COCHCOCH ₃) ₃ Mn (Manganese, tris(2,4-pentanedionato-O,O')-, (OC-6-11)-)	14284-89-0	**	7.32±0.07 (V)	PE	3682
C₁₆H₁₄N₂O₂Mn⁺						
	C ₁₆ H ₁₄ N ₂ O ₂ Mn (Manganese, [[2,2'-[1,2-ethanediylbis(nitrilomethylidyne)] bis[phenolato]](2--N,N',O,O')-])	XXXXX-XX-X	**	7.77±0.08	EI	4668
FMn⁺						
	MnF	13569-25-0	**	8.51±0.2	EI	3623
	MnF ₂	7782-64-1		13.60±0.2	EI	3623
F₂Mn⁺						
	MnF ₂	7782-64-1	**	11.38±0.2	EI	3623
	MnF ₃	7783-53-1		14.79±0.2	EI	3623
F₃Mn⁺						
	MnF ₃	7783-53-1	**	12.57±0.2	EI	3623
	MnF ₄	15195-58-1		15.50±0.2	EI	3623
F₄Mn⁺						
	MnF ₄	15195-58-1	**	13.46±0.2	EI	3623
O₃FMn⁺						
	MnO ₃ F	15586-97-7	**	12.20±0.05 (V)	PE	4632
C₆O₅F₃Mn⁺						
	CF ₃ (CO) ₅ Mn	13601-14-4	**	9.17 (V)	PE	4110
C₁₅H₃O₆F₁₈Mn⁺						
	(CF ₃ COCHCOCF ₃) ₃ Mn (Manganese, tris(1,1,1,5,5-hexafluoro-2,4-pentanedionato-O,O')-, (OC-6-11)-)	14354-50-8	**	9.2 (V)	PE	3682
C₁₉H₃O₁₀F₁₈Mn⁺						
	(CF ₃ COCHCOCF ₃) ₃ (CO) ₄ Mn (Tris(1,1,1,5,5-hexafluoro-2,4-pentanedionato)manganese tetracarbonyl)	XXXXX-XX-X	**	8.11±0.07 (V)	PE	3682

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₃H₉SiMn⁺	((CH ₃) ₃ Si)(CO) ₅ Mn	26500-16-3		12.8	EI	3814
C₄H₉OSiMn⁺	((CH ₃) ₃ Si)(CO) ₅ Mn ((CH ₃) ₃ Si)(CO) ₄ (PF ₃)Mn	26500-16-3 33989-27-4	4CO	12.0	EI	3814
				12.7	EI	3814
C₅H₉O₂SiMn⁺	((CH ₃) ₃ Si)(CO) ₅ Mn ((CH ₃) ₃ Si)(CO) ₄ (PF ₃)Mn	26500-16-3 33989-27-4	3CO	10.8	EI	3814
				11.1	EI	3814
C₆H₉O₃SiMn⁺	((CH ₃) ₃ Si)(CO) ₅ Mn	26500-16-3	2CO	10.2	EI	3814
C₇H₉O₄SiMn⁺	((CH ₃) ₃ Si)(CO) ₅ Mn ((CH ₃) ₃ Si)(CO) ₄ (PF ₃)Mn	26500-16-3 33989-27-4	CO PF ₃	9.2 9.9	EI EI	3814 3814
C₅H₉O₅SiMn⁺	(SiH ₃)(CO) ₅ Mn	15770-61-3	**	8.99±0.02 (V)	PE	3827
C₈H₉O₅SiMn⁺	(Si(CH ₃) ₃)(CO) ₅ Mn ((CH ₃) ₃ Si)(CO) ₅ Mn	XXXXX-XX-X 26500-16-3 **	**	9.0±0.1 (V) 8.47 8.7±0.2	PE PE EI	3827 5321 3814
C₂₄H₂₂PMn⁺	(CH ₃ C ₅ H ₄)(CO) ₂ ((C ₆ H ₅) ₃ P)Mn (Manganese,dicarbonyl[(1,2,3,4,5-η)-1-methyl-2,4-cyclopentadien-1-yl] (triphenylphosphine)-) C ₂₆ H ₂₂ OPSMn (Manganese,(carbonothioyl)carbonyl[(1,2,3,4,5-η)-1-methyl-2,4-cyclopentadien-1-yl](triphenylphosphine)-)	12100-95-7 70279-43-5	2CO CO+CS	8.54±0.03 9.19±0.03	EI EI	5576 5576
C₂₅H₂₂OPMn⁺	(CH ₃ C ₅ H ₄)(CO) ₂ ((C ₆ H ₅) ₃ P)Mn (Manganese,dicarbonyl[(1,2,3,4,5-η)-1-methyl-2,4-cyclopentadien-1-yl] (triphenylphosphine)-)	12100-95-7	CO	8.95±0.02	EI	5576
C₇H₈O₂PMn⁺	(C ₅ H ₅)(PH ₃)(CO) ₂ Mn (Manganese,dicarbonyl(η ⁵ -2,4-cyclopentadien-1-yl)(phosphine)-)	12300-46-8	**	7.28	EI	5453
C₂₆H₂₂O₂PMn⁺	(CH ₃ C ₅ H ₄)(CO) ₂ ((C ₆ H ₅) ₃ P)Mn (Manganese,dicarbonyl[(1,2,3,4,5-η)-1-methyl-2,4-cyclopentadien-1-yl] (triphenylphosphine)-)	12100-95-7	**	6.55±0.03	EI	5576
C₇H₄O₃PMn⁺	C ₄ H ₄ P(CO) ₃ Mn (Phosphacymantrene)	XXXXX-XX-X	**	8.25 (V)	PE	4995
C₉H₈O₃PMn⁺	C ₄ H ₂ P(CH ₃) ₂ (CO) ₃ Mn (Phosphacymantrene, 3,4-dimethyl-)	XXXXX-XX-X	**	8.13 (V)	PE	4995
C₁₁H₁₂O₃PMn⁺	C ₁₁ H ₁₂ O ₃ PMn (Phosphacymantrene, 3,4-dimethyl-2-ethyl-)	XXXXX-XX-X	**	8. (V)	PE	4995
C₁₁H₁₀O₄PMn⁺	C ₁₁ H ₁₀ O ₄ PMn (Phosphacymantrene, 2-acetyl-3,4-dimethyl-)	XXXXX-XX-X	**	8.2 (V)	PE	4995

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
HF₁₅P₅Mn⁺	H(PF ₃) ₃ Mn	20558-69-4	**	9.47 (V)	PE	4456
C₇H₅O₂F₃PMn⁺	(C ₅ H ₅)(PF ₃) ₂ (CO) ₂ Mn (Manganese,dicarbonyl(η^5 -2,4-cyclopentadien-1-yl) (phosphorus trichloride)-)	12275-47-7	**	8.24	EI	5453
C₇H₉O₄F₃SiPMn⁺	((CH ₃) ₃ Si)(CO) ₄ PF ₃ Mn	33989-27-4	**	8.7±0.2	EI	3814
C₆H₉O₃F₆SiP₂Mn⁺	((CH ₃) ₃ Si)(CO) ₃ (PF ₃) ₂ Mn	36087-62-4	**	8.1±0.1	EI	3814
C₅H₉O₂F₉SiP₃Mn⁺	((CH ₃) ₃ Si)(CO) ₂ (PF ₃) ₃ Mn	36087-61-3	**	9.1±0.2	EI	3814
CSMn⁺	C ₅ H ₅ (CO) ₂ CSMn (Manganese, (carbonothioyl)dicarbonyl(η^5 -2,4-cyclopentadien-1-yl)-) C ₅ H ₅ (CH ₃)(CO) ₂ CSMn (Manganese, (carbonothioyl)dicarbonyl[(1,2,3,4,5- η)-1-methyl- 2,4-cyclopentadien-1-yl]-) (C ₅ H ₅)(CS)(NO)MnI (Manganese,(carbonothioyl)(η^5 -2,4-cyclopentadien-1-yl)iodonitrosyl-) (CH ₃ C ₅ H ₄)(CS)(NO)MnI (Manganese,(carbonothioyl)][1,2,3,4,5- η)-1-methyl- 2,4-cyclopentadien-1-yl]iodonitrosyl-)	31741-76-1 49716-52-1 58450-74-1 XXXXX-XX-X	2CO + C ₅ H ₅ 2CO + C ₆ H ₇ C ₅ H ₅ + NO + I XXXXX-XX-X	16.91±0.02 17.97±0.01 17.74±0.03 18.00±0.05	EI	4661 4661 5561 5561
C₂S₂Mn₂⁺	(C ₅ H ₅) ₂ (CS)(NO) ₂ Mn ₂ (Manganese,bis[μ -(carbonothioyl)bis(η^5 -2,4-cyclopentadien-1-yl) dinitrosyldi-(Mn-Mn)-])	64090-73-9		20.92±0.04	EI	5423
C₆H₅SMn⁺	C ₅ H ₅ (CO) ₂ CSMn (Manganese, (carbonothioyl)dicarbonyl(η^5 -2,4-cyclopentadien-1-yl)-) (C ₅ H ₅) ₂ (CS) ₂ (NO) ₂ Mn ₂ (Manganese,bis[μ -(carbonothioyl)bis(η^5 -2,4-cyclopentadien-1-yl) dinitrosyldi-(Mn-Mn)-]) (C ₅ H ₅)(CS)(NO)MnI (Manganese,(carbonothioyl)(η^5 -2,4-cyclopentadien-1-yl)iodonitrosyl-)	31741-76-1 64090-73-9 58450-74-1	2CO 13.00±0.02 NO + I	9.25±0.01 10.17±0.03	EI	4661 5423 5561
C₇H₇SMn⁺	C ₅ H ₅ (CH ₃)(CO) ₂ CSMn (Manganese, (carbonothioyl)dicarbonyl[(1,2,3,4,5- η)-1-methyl- 2,4-cyclopentadien-1-yl]-) C ₂₀ H ₂₂ OPSMn (Manganese,(carbonothioyl)carbonyl[(1,2,3,4,5- η)-1-methyl-2,4- cyclopentadien-1-yl](triphenylphosphine)-) C ₂₀ H ₂₂ OSMnAs (Manganese,(carbonothioyl)carbonyl[(1,2,3,4,5- η)-1- methyl-2,4-cyclopentadien-1-yl](triphenylarsine)-) (CH ₃ C ₅ H ₄)(CO)(CS)(C ₆ H ₅) ₂ Sb)Mn (Manganese,(carbonothioyl)carbonyl[(1,2,3,4,5- η)-1-methyl- 2,4-cyclopentadien-1-yl](triphenylstibine)-) (CH ₃ C ₅ H ₄)(CS)(NO)MnI (Manganese,(carbonothioyl)][1,2,3,4,5- η)-1-methyl- 2,4-cyclopentadien-1-yl]iodonitrosyl-)	49716-52-1 70279-43-5 XXXXX-XX-X XXXXX-XX-X XXXXX-XX-X XXXXX-XX-X	2CO CO + (C ₆ H ₅) ₃ P 11.78±0.30 10.78±0.08 NO + I	9.15±0.01 12.54±0.50 10.21±0.03	EI	4661 5576 5576 5576 5561
C₁₀H₁₅SMn⁺	C ₄ H ₈ SC ₅ H ₄ CH ₃ Mn(CO) ₂ (Dicarbonyl[(1,2,3,4,5- η)-1-methyl-2,4-cyclopentadien-1-yl] (tetrahydrothiophene)manganese)	12153-94-5	2CO	7.9±0.1	EI	3498

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{18}\text{H}_{17}\text{SMn}^+$	$(\text{C}_5\text{H}_5)_2\text{SC}_5\text{H}_4\text{CH}_3\text{Mn}(\text{CO})_2$ (Dicarbonyl((1,2,3,4,5)-1-methyl-2,4-cyclopentadien-1-yl) (1,1'-thiobis(benzene)-S)manganese)	36154-47-9	2CO	8.0±0.1	EI	3498
$\text{C}_7\text{H}_4\text{S}_2\text{Mn}^+$	$(\text{C}_5\text{H}_5)_2(\text{CS})_2(\text{NO})_2\text{Mn}_2$ (Manganese,bis[μ-(carbonothioyl)bis(η^5 -2,4-cyclopentadien-1-yl)dinitrosyldi-(Mn-Mn)-])	64090-73-9		15.61±0.16	EI	5423
$\text{C}_7\text{H}_5\text{S}_2\text{Mn}^+$	$(\text{C}_5\text{H}_5)_2(\text{CS})_2(\text{NO})_2\text{Mn}_2$ (Manganese,bis[μ-(carbonothioyl)bis(η^5 -2,4-cyclopentadien-1-yl)dinitrosyldi-(Mn-Mn)-])	64090-73-9		16.02±0.05	EI	5423
$\text{C}_6\text{H}_5\text{SMn}_2^+$	$(\text{C}_5\text{H}_5)_2(\text{CS})_2(\text{NO})_2\text{Mn}_2$ (Manganese,bis[μ-(carbonothioyl)bis(η^5 -2,4-cyclopentadien-1-yl)dinitrosyldi-(Mn-Mn)-])	64090-73-9		16.18±0.07	EI	5423
$\text{C}_{11}\text{H}_{10}\text{SMn}_2^+$	$(\text{C}_5\text{H}_5)_2(\text{CS})_2(\text{NO})_2\text{Mn}_2$ (Manganese,bis[μ-(carbonothioyl)bis(η^5 -2,4-cyclopentadien-1-yl)dinitrosyldi-(Mn-Mn)-])	64090-73-9	CS+2NO	12.64±0.02	EI	5423
$\text{C}_7\text{H}_4\text{S}_2\text{Mn}_2^+$	$(\text{C}_5\text{H}_5)_2(\text{CS})_2(\text{NO})_2\text{Mn}_2$ (Manganese,bis[μ-(carbonothioyl)bis(η^5 -2,4-cyclopentadien-1-yl)dinitrosyldi-(Mn-Mn)-])	64090-73-9		11.87±0.03	EI	5423
$\text{C}_7\text{H}_5\text{S}_2\text{Mn}_2^+$	$(\text{C}_5\text{H}_5)_2(\text{CS})_2(\text{NO})_2\text{Mn}_2$ (Manganese,bis[μ-(carbonothioyl)bis(η^5 -2,4-cyclopentadien-1-yl)dinitrosyldi-(Mn-Mn)-])	64090-73-9	$\text{C}_5\text{H}_5+2\text{NO}$	12.89±0.02	EI	5423
$\text{C}_{12}\text{H}_{10}\text{S}_2\text{Mn}_2^+$	$(\text{C}_5\text{H}_5)_2(\text{CS})_2(\text{NO})_2\text{Mn}_2$ (Manganese,bis[μ-(carbonothioyl)bis(η^5 -2,4-cyclopentadien-1-yl)dinitrosyldi-(Mn-Mn)-])	64090-73-9	2NO	8.99±0.02	EI	5423
$\text{C}_7\text{H}_5\text{OSMn}^+$	$\text{C}_5\text{H}_5(\text{CO})_2\text{CSMn}$ (Manganese,(carbonothioyl)dicarbonyl(η^5 -2,4-cyclopentadien-1-yl)-)	31741-76-1	CO	8.18±0.01	EI	4661
$\text{C}_8\text{H}_7\text{OSMn}^+$	$\text{C}_5\text{H}_4(\text{CH}_3)(\text{CO})_2\text{CSMn}$ (Manganese,(carbonothioyl)dicarbonyl((1,2,3,4,5- η)-1-methyl-2,4-cyclopentadien-1-yl)-)	49716-52-1	CO	7.95±0.02	EI	4661
$\text{C}_8\text{H}_{13}\text{OSMn}^+$	$\text{C}_5\text{H}_4\text{CH}_3\text{Mn}(\text{CO})_2\text{SO}(\text{CH}_3)_2$ (Dicarbonyl((1,2,3,4,5)-1-methyl-2,4-cyclopentadien-1-yl)sulfinylbis(methane)-S)manganese)	12153-02-5	2CO	7.9±0.1	EI	3498
$\text{C}_{10}\text{H}_{15}\text{OSMn}^+$	$\text{C}_5\text{H}_4\text{SOC}_5\text{H}_4\text{CH}_3\text{Mn}(\text{CO})_2$ (Dicarbonyl((1,2,3,4,5)-1-methyl-2,4-cyclopentadiene-1-yl)(tetrahydrothiophene 1-oxide-S)manganese)	12153-95-6	2CO	7.5±0.1	EI	3498
$\text{C}_{16}\text{H}_{17}\text{OSMn}^+$	$(\text{C}_5\text{H}_5)_2\text{SOC}_5\text{H}_4\text{CH}_3\text{Mn}(\text{CO})_2$ (Dicarbonyl((1,2,3,4,5- η)-1-methyl-2,4-cyclopentadien-1-yl)(1,1'-sulfinylbis(benzene)-S)manganese)	36154-49-1	2CO	7.8±0.1	EI	3498

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_8\text{H}_5\text{O}_2\text{SMn}^+$	$(\text{C}_5\text{H}_5)(\text{CO})_2(\text{CS})\text{Mn}$ (Manganese,(carbonothioyl)dicarbonyl(η^5 -2,4-cyclopentadien-1-yl)-)	31741-76-1	** **	7.81 (V) 7.78 ± 0.01	PE EI	5518 4661
$\text{C}_9\text{H}_7\text{O}_2\text{SMn}^+$	$\text{C}_5\text{H}_4(\text{CH}_3)(\text{CO})_2\text{CSMn}$ (Manganese, (carbonothioyl)dicarbonyl[(1,2,3,4,5- η)-1-methyl-2,4-cyclopentadien-1-yl]-)	49716-52-1	**	7.72 ± 0.02	EI	4661
$\text{C}_{12}\text{H}_{15}\text{O}_2\text{SMn}^+$	$\text{C}_4\text{H}_8\text{SC}_5\text{H}_4\text{CH}_3\text{Mn}(\text{CO})_2$ (Dicarbonyl(1,2,3,4,5- η)-1-methyl-2,4-cyclopentadien-1-yl) (tetrahydrothiophene)manganese)	12153-94-5	** **	6.45 ± 0.05 6.45	EI EI	3498 5292
$\text{C}_{20}\text{H}_{17}\text{O}_2\text{SMn}^+$	$(\text{C}_6\text{H}_5)_2\text{SC}_5\text{H}_4\text{CH}_3\text{Mn}(\text{CO})_2$ (Dicarbonyl((1,2,3,4,5- η)-1-methyl-2,4-cyclopentadien-1-yl) (1,1'-thiobis(benzene)-S)manganese)	36154-47-9	** **	6.27 ± 0.05 6.27	EI EI	3498 5292
$\text{C}_8\text{H}_{11}\text{O}_3\text{SMn}^+$	$\text{C}_2\text{H}_4\text{O}_2\text{SOC}_5\text{H}_4\text{CH}_3\text{Mn}(\text{CO})_2$ (Dicarbonyl(1,3,2-dioxathiolane 2-oxide-S)((1,2,3,4,5- η)-1-methyl-2,4-cyclopentadien-1-yl)manganese)	12152-97-5	2CO	7.75 ± 0.1	EI	3498
$\text{C}_{10}\text{H}_{13}\text{O}_3\text{SMn}^+$	$\text{C}_5\text{H}_8\text{CH}_3\text{Mn}(\text{CO})_2\text{SO}(\text{CH}_3)_2$ (Dicarbonyl((1,2,3,4,5- η)-1-methyl-2,4-cyclopentadien-1-yl) (sulfinylbis(methane)-S)manganese)	12153-02-5	** **	7.19 ± 0.05 7.19	EI EI	3498 5292
$\text{C}_{12}\text{H}_{15}\text{O}_3\text{SMn}^+$	$\text{C}_4\text{H}_8\text{SOC}_5\text{H}_4\text{CH}_3\text{Mn}(\text{CO})_2$ (Dicarbonyl((1,2,3,4,5- η)-1-methyl-2,4-cyclopentadiene-1-yl) (tetrahydrothiophene 1-oxide-S)manganese) hydrothiophene-1-oxide-S-)	12153-95-6	** **	6.79 ± 0.05 6.79	EI EI	3498 5292
$\text{C}_{20}\text{H}_{17}\text{O}_3\text{SMn}^+$	$(\text{C}_6\text{H}_5)_2\text{SOC}_5\text{H}_4\text{CH}_3\text{Mn}(\text{CO})_2$ (Dicarbonyl((1,2,3,4,5- η)-1-methyl-2,4-cyclopentadien-1-yl) (1,1'-sulfinylbis(benzene)-S)manganese)	36154-49-1	** **	6.76 ± 0.05 6.76	EI EI	3498 5292
$\text{C}_{10}\text{H}_{11}\text{O}_5\text{SMn}^+$	$\text{C}_2\text{H}_4\text{O}_2\text{SOC}_5\text{H}_4\text{CH}_3\text{Mn}(\text{CO})_2$ (Dicarbonyl(1,3,2-dioxathiolane 2-oxide-S)((1,2,3,4,5- η)-1-methyl-2,4-cyclopentadien-1-yl)manganese)	12152-97-5	** **	7.38 ± 0.05 7.38	EI EI	3498 5292
$\text{C}_6\text{H}_5\text{NOSMn}^+$	$(\text{C}_5\text{H}_5)_2(\text{CS})_2(\text{NO})_2\text{Mn}_2$ (Manganese,bis[μ -(carbonothioyl)bis(η^5 -2,4-cyclopentadien-1-yl)dinitrosylid-($Mn-Mn$)-])	64090-73-9		11.04 ± 0.03	EI	5423
	$(\text{C}_5\text{H}_5)(\text{CS})(\text{NO})\text{MnI}$ (Manganese,(carbonothioyl)(η^5 -2,4-cyclopentadien-1-yl)iodonitrosyl-)	58450-74-1	I	8.77 ± 0.04	EI	5561
$\text{C}_7\text{H}_7\text{NOSMn}^+$	$(\text{CH}_3\text{C}_5\text{H}_4)(\text{CS})(\text{NO})\text{MnI}$ (Manganese,(carbonothioyl)(1,2,3,4,5- η)-1-methyl-2,4-cyclopentadien-1-yl)iodonitrosyl-)	XXXXX-XX-X	I	8.68 ± 0.02	EI	5561

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₇H₅NOS₂Mn⁺	(C ₅ H ₅) ₂ (CS) ₂ (NO) ₂ Mn ₂ (Manganese,bis[μ-(carbonothioyl)bis(η ⁵ -2,4-cyclopentadien-1-yl)dinitrosyldi-(Mn-Mn)])	64090-73-9		9.03±0.04	EI	5423
C₇H₅NOS₂Mn⁺	(C ₅ H ₅) ₂ (CS) ₂ (NO) ₂ Mn ₂ (Manganese,bis[μ-(carbonothioyl)bis(η ⁵ -2,4-cyclopentadien-1-yl)dinitrosyldi-(Mn-Mn)])	64090-73-9	C ₅ H ₅ +NO	11.97±0.02	EI	5423
C₁₂H₁₀NOS₂Mn⁺	(C ₅ H ₅) ₂ (CS) ₂ (NO) ₂ Mn ₂ (Manganese,bis[μ-(carbonothioyl)bis(η ⁵ -2,4-cyclopentadien-1-yl)dinitrosyldi-(Mn-Mn)])	64090-73-9	NO	7.90±0.02	EI	5423
C₁₂H₁₀N₂O₂S₂Mn⁺	(C ₅ H ₅) ₂ (CS) ₂ (NO) ₂ Mn ₂ (Manganese,bis[μ-(carbonothioyl)bis(η ⁵ -2,4-cyclopentadien-1-yl)dinitrosyldi-(Mn-Mn)])	64090-73-9	**	6.77±0.02	EI	5582
C₂₅H₂₂PSMn⁺	C ₂₆ H ₂₂ OPSMn (Manganese,(carbonothioyl)carbonyl[(1,2,3,4,5-η)-1-methyl-2,4-cyclopentadien-1-yl](triphenylphosphine)-])	70279-43-5	CO	7.37±0.02	EI	5576
C₂₆H₂₂OPSMn⁺	C ₂₆ H ₂₂ OPSMn (Manganese,(carbonothioyl)carbonyl[(1,2,3,4,5-η)-1-methyl-2,4-cyclopentadien-1-yl](triphenylphosphine)-])	70279-43-5	**	6.58±0.02	EI	5576
Cl₂Mn⁺	MnCl ₂	7773-01-5	**	11.03 (V)	PE	5172
C₄₄H₂₈N₄ClMn⁺	C ₂₀ H ₈ N ₄ (C ₆ H ₅) ₄ MnCl (Manganese, chloro[5,10,15,20-tetraphenyl-21H,23H-porphinato(2-)N ²¹ ,N ²² ,N ²³]-(SP-5-12)-)	32195-55-4	**	5.95±0.2	OTH	4962
O₃ClMn⁺	MnO ₃ Cl	15605-27-3	**	11.98±0.05 (V)	PE	4632
C₅O₅ClMn⁺	(CO) ₅ MnCl	14100-30-2	** **	8.87±0.05 (V) 8.94 (V)	PE PE	4492 3866
C₅O₅SiCl₃Mn⁺	(CO) ₅ SiCl ₃ Mn	38194-30-8	**	9.36±0.05	PE	4492
C₇H₅O₂PCl₃Mn⁺	(C ₅ H ₅)(PCl ₃)(CO) ₂ Mn (Manganese,dicarbonyl(η ⁵ -2,4-cyclopentadien-1-yl)(phosphorus trichloride)-)	12275-46-6	**	8.12	EI	5453
Fe⁺	Fe	7439-89-6	** **	7.7±0.2 8.0±0.5	EI EI	4618 4436
	(C ₅ H ₅) ₂ Fe (Ferrocene)	102-54-5		12.0±1.5	EI	3793
			(C ₅ H ₅) ₂	14.00±0.25 14.10±0.15	EI EI	3628 4072
	(CO) ₅ Fe	13463-40-6	5CO	16.2±0.2	EI	4618

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
Fe⁺						
	((CH ₃) ₂ N) ₃ P(CO) ₃ Fe	19372-47-5		17.0±0.05	EI	3952
	(C ₅ H ₅) ₂ (CS)(CO) ₃ Fe ₂	67113-80-8		20.94±0.03	EI	5423
	(Iron,μ-carbonothioyl-μ-carbonyldicarbonylbis(η ⁵ -2,4-cyclopentadien-1-yl)di(Fe-Fe)-)					
	(C ₅ H ₅) ₂ (CS) ₂ (CO) ₂ Fe ₂	67225-86-9		20.21±0.03	EI	5423
	(Iron,bis[μ-(carbonothioyl)]dicarbonylbis(η ⁵ -2,4-cyclopentadien-1-yl)di(Fe-Fe)-)					
Fe²⁺	Fe ⁺	7439-89-6	**	16.188±0.001	S	5233
Fe₂⁺						
	(C ₅ H ₅) ₂ (CS)(CO) ₃ Fe ₂	67113-80-8		26.71±0.06	EI	5423
	(Iron,μ-carbonothioyl-μ-carbonyldicarbonylbis(η ⁵ -2,4-cyclopentadien-1-yl)di(Fe-Fe)-)					
	(C ₅ H ₅) ₂ (CS) ₂ (CO) ₂ Fe ₂	67225-86-9		20.89±0.03	EI	5423
	(Iron,bis[μ-(carbonothioyl)]dicarbonylbis(η ⁵ -2,4-cyclopentadien-1-yl)di(Fe-Fe)-)					
C₂Fe⁺	(CO) ₅ Fe	13463-40-6		29.9±0.5	EI	4736
C₅H₃Fe⁺	(C ₅ H ₅) ₂ Fe (Ferrocene)	102-54-5		17.75±0.2	EI	4072
				18.06±0.10	EI	3628
C₅H₅Fe⁺	(C ₅ H ₅) ₂ Fe (Ferrocene)	102-54-5		12.95±0.15	EI	4072
			C ₅ H ₅	13.9±0.2	EI	3793
				14.25±0.25	EI	3628
	(C ₅ H ₅) ₂ (CS)(CO) ₃ Fe ₂	67113-80-8		15.82±0.03	EI	5423
	(Iron,μ-carbonothioyl-μ-carbonyldicarbonylbis(η ⁵ -2,4-cyclopentadien-1-yl)di(Fe-Fe)-)					
	(C ₅ H ₅) ₂ (CS) ₂ (CO) ₂ Fe ₂	67225-86-9		15.32±0.03	EI	5423
	(Iron,bis[μ-(carbonothioyl)]dicarbonylbis(η ⁵ -2,4-cyclopentadien-1-yl)di(Fe-Fe)-)					
C₁₀H₁₀Fe⁺	(C ₂ H ₅) ₂ Fe (Ferrocene)	102-54-5	**	6.90 (V)	PE	4565
			**	6.78±0.05	PI	3729
			**	6.72	PE	3725
			**	6.86 (V)	PE	5394
			**	6.88 (V)	PE	3688
			**	6.88 (V)	PE	5507
			**	~7.0 (V)	PE	3527
			**	7.10 (V)	PE	4072
			**	6.75±0.25	EI	3628
			**	6.9±0.1	EI	3793
			**	6.90±0.1	EI	4072
	(C ₅ H ₅) ₂ (CS)(CO) ₃ Fe ₂	67113-80-8		9.03±0.03	EI	5423
	(Iron,μ-carbonothioyl-μ-carbonyldicarbonylbis(η ⁵ -2,4-cyclopentadien-1-yl)di(Fe-Fe)-)					
	(C ₅ H ₅) ₂ (CS) ₂ (CO) ₂ Fe ₂	67225-86-9		8.62±0.03	EI	5423
	(Iron,bis[μ-(carbonothioyl)]dicarbonylbis(η ⁵ -2,4-cyclopentadien-1-yl)di(Fe-Fe)-)					
C₁₂H₁₂Fe⁺	(C ₅ H ₅)(C ₅ H ₁ C ₂ H ₃)Fe (Ferrocene, ethenyl-)	1271-51-8	**	6.75±0.05	PI	3729

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₂H₁₄Fe⁺	(C ₅ H ₄ CH ₃) ₂ Fe (Ferrocene, 1,1'-dimethyl-)	1291-47-0	**	6.72 (V)	PE	3688
			**	6.72 (V)	PE	5507
	(C ₅ H ₅)(C ₅ H ₄ C ₂ H ₅)Fe (Ferrocene, ethyl-)	1273-89-8	**	6.70±0.05	PI	3729
C₂₀H₃₀Fe⁺	(C ₅ (CH ₃) ₅) ₂ Fe (Ferrocene, decamethyl-)	12126-50-0	**	5.88 (V)	PE	5394
C₂₀H₁₆Fe₂⁺	(C ₁₀ H ₈) ₂ Fe ₂ (1,1":1',1'''-Biferrocene)	11105-90-1	**	6.55 (V)	PE	5373
C₂₀H₁₈Fe₂⁺	(C ₁₀ H ₈)(C ₅ H ₅) ₂ Fe ₂ (1,1"-Biferrocene)	1287-38-3	**	6.6 (V)	PE	5373
C₃₆H₄₄N₄Fe⁺	((C ₂ H ₅) ₂ C ₄ NCH) ₄ Fe (Iron, [2,3,7,8,12,13,17,18-octaethyl-21H,23H-porphinato(2)-N ²¹ ,N ²² ,N ²³ ,N ²⁴]- (SP-4-1)-)	61085-06-1	**	6.06±0.03 (V)	PE	5476
C₄₄H₂₈N₄Fe⁺	C ₂₀ H ₈ N ₄ (C ₆ H ₅) ₄ Fe (Iron, [5,10,15,20-tetraphenyl-21H,23H-porphinato(2)-N ²¹ ,N ²² ,N ²³ ,N ²⁴]- (SP-4-1)-)	16591-56-3	**	6.50 (V)	PE	4557
C₃₂H₁₆N₈Fe⁺	C ₃₂ H ₁₆ N ₈ Fe (Iron, [29H,31H-phthalocyaninato(2)-N ²⁹ ,N ³⁰ ,N ³¹ ,N ³²]- (SP-4-1)-)	132-16-1	**	7.22±0.10	EI	3829
OFe⁺	FeO (CO) ₅ Fe	1345-25-1 13463-40-6	**	8.71±0.10 22.5±0.5	EI EI	4436 4736
O₂Fe⁺	FeO ₂	12411-15-3	**	9.5±0.5	EI	4436
COFe⁺	(CO) ₅ Fe	13463-40-6	4CO	14.0±0.2	EI	4618
C₂OF₂Fe⁺	(CO) ₅ Fe	13463-40-6		20.2±0.5	EI	4736
C₂O₂Fe⁺	(CO) ₅ Fe	13463-40-6	3CO	11.0±0.2	EI	4618
C₃O₂Fe⁺	(CO) ₅ Fe	13463-40-6		18.2±0.5	EI	4736
C₃O₃Fe⁺	(CO) ₅ Fe	13463-40-6	2CO	10.1±0.2	EI	4618
C₃O₃Fe⁺²	(CO) ₅ Fe	13463-40-6		24.0±0.5	EI	4736
C₄O₄Fe⁺	(CO) ₅ Fe	13463-40-6	CO	9.3±0.2	EI	4618
C₅O₅Fe⁺	(CO) ₅ Fe	13463-40-6	**	8.6 (V)	PE	4376

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₅O₅Fe⁺						
	(CO) ₅ Fe	13463-40-6	**	8.60 (V)	PE	4456
			**	8.4±0.2	EI	4618
C₉H₈O₂Fe⁺						
	C ₅ H ₅ (CO) ₂ (CH ₃)Fe (Iron, dicarbonyl(η^5 -2,4-cyclopentadien-1-yl)methyl-)	12080-06-7	**	7.65 (V)	PE	4565
			**	7.79 (V)	PE	4570
			**	7.91 (V)	PE	5358
C₁₀H₁₀O₂Fe⁺						
	(C ₅ H ₅)(C ₅ H ₅)(CO) ₂ Fe (Iron, dicarbonyl(η^5 -2,4-cyclopentadien-1-yl)-2-propenyl-)	38960-10-0	**	7.97 (V)	PE	5358
C₁₂H₁₀O₂Fe⁺						
	(C ₅ H ₅) ₂ (CO) ₂ Fe (Iron, dicarbonyl-2,4-cyclopentadiene-1-yl(η^5 -2,4-cyclopentadien-1-yl)-)	12247-96-0	**	7.58 (V)	PE	5358
C₇H₄O₃Fe⁺						
	C ₄ H ₄ (CO) ₃ Fe (Iron, tricarbonyl(η^4 -1,3-cyclobutadiene)-)	12078-17-0	**	7.65±0.02	PE	4412
			**	8.1 (V)	PE	4937
			**	8.15 (V)	PE	5005
C₇H₆O₃Fe⁺						
	(CH ₂ =CHCH=CH ₂)(CO) ₃ Fe	12078-32-9	**	8.16 (V)	PE	5551
			**	8.23 (V)	PE	5044
	(1,3-n-C ₄ H ₉)(CO) ₃ Fe (Iron, (η^3 -1,3-butadiene)tricarbonyl-)	52610-59-0	**	8.22 (V)	PE	5005
	(CH ₃) ₃ C(CO) ₃ Fe (Trimethylenemethane-iron tricarbonyl-)	XXXXXX-XX-X	**	8.63 (V)	PE	5005
C₈H₈O₃Fe⁺						
	C ₈ H ₈ O ₃ Fe (Iron, tricarbonyl[(1,2,3,4- η)-2-methyl-1,3-butadiene]-, stereoisomer)	32731-93-4	**	8.11 (V)	PE	5005
	C ₈ H ₈ O ₃ Fe (Iron, tricarbonyl[(1,2,3,4- η)-1,3-pentadiene]-,(E)-)	XXXXXX-XX-X	**	8.07 (V)	PE	5005
C₉H₈O₃Fe⁺						
	C ₆ H ₆ (CO) ₃ Fe (Iron, tricarbonyl[(1,2,3,4- η)-1,3-cyclohexadiene]-)	12152-72-6	**	7.96 (V)	PE	5005
			**	7.98 (V)	PE	5551
C₉H₁₀O₃Fe⁺						
	C ₉ H ₁₀ O ₃ Fe (Iron, tricarbonyl[(1,2,3,4- η)-2,3-dimethyl-1,3-butadiene]-)	31741-56-7	**	7.95 (V)	PE	5005
	C ₉ H ₁₀ O ₃ Fe (Iron, tricarbonyl[(1,2,3,4- η)-2-methyl-1,3-pentadiene-,(E)-])	XXXXXX-XX-X	**	7.94 (V)	PE	5005
C₁₀H₈O₃Fe⁺						
	C ₇ H ₆ (CO) ₃ Fe (Iron, [(2,3,5,6- η)-bicyclo 2.2.1]hepta-2,5-diene] tricarbonyl-)	12307-07-2	**	7.51 (V)	PE	5005
			**	7.51 (V)	PE	5367
	(C ₇ H ₆)(CO) ₃ Fe (Iron, tricarbonyl[(1,2,3,4- η)-1,3,5-cycloheptatriene]-)	36343-88-1	**	7.76 (V)	PE	5551
C₁₀H₁₀O₃Fe⁺						
	(C ₇ H ₁₀)(CO) ₃ Fe (Iron, tricarbonyl[(1,2,3,4- η)-1,3-cycloheptadiene]-)	40674-86-0	**	7.78 (V)	PE	5551

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{11}\text{H}_8\text{O}_3\text{Fe}^+$	$(\text{C}_8\text{H}_9)(\text{CO})_3\text{Fe}$ (Iron, tricarbonyl[(1,2,3,4- η)-1,3,5,7-cyclooctatetraene]-)	12093-05-9	**	7.84 (V)	PE	5551
$\text{C}_{11}\text{H}_{12}\text{O}_3\text{Fe}^+$	$(\text{C}_8\text{H}_{12})(\text{CO})_3\text{Fe}$ (Iron, tricarbonyl[(1,2,3,4- η)-1,3-cyclooctadiene]-)	33270-50-7	**	7.45 (V)	PE	5551
$\text{C}_4\text{H}_2\text{O}_4\text{Fe}^+$	$\text{Fe}(\text{CO})_3\text{H}_2$	12002-28-7	**	9.65	PE	4372
$\text{C}_6\text{H}_4\text{O}_4\text{Fe}^+$	$(\text{CH}_2=\text{CH}_2)(\text{CO})_3\text{Fe}$	32799-25-0	** **	8.38 (V) 8.4-8.6 (V)	PE PE	4946 4376
$\text{C}_{11}\text{H}_{12}\text{O}_4\text{Fe}^+$	$(\text{C}_3(\text{CH}_3)_4)(\text{CO})_4\text{Fe}$	12264-26-5	**	7.84 (V)	PE	5362
$\text{C}_7\text{H}_4\text{O}_5\text{Fe}^+$	$\text{CH}_2=\text{CHCHO}(\text{CO})_4\text{Fe}$	12287-43-3	** **	8.69 (V) 9.35 (V)	PE PE	4908 5559
$\text{C}_8\text{H}_6\text{O}_5\text{Fe}^+$	$\text{CH}_3\text{CH}=\text{CHCHO}(\text{CO})_4\text{Fe}$	70520-16-0	**	8.60 (V)	PE	4908
$\text{C}_{11}\text{H}_{12}\text{O}_5\text{Fe}^+$	$\text{C}_{11}\text{H}_{12}\text{O}_5\text{Fe}$ (Iron, tricarbonyl[(1,2,3,4- η)-(2,4-hexadienoic acid ethyl ester, (E,E)-)-])	XXXXX-XX-X	**	8.19 (V)	PE	5005
$\text{C}_7\text{H}_3\text{O}_6\text{Fe}^+$	$\text{CH}_2=\text{CHCOO}(\text{CO})_4\text{Fe}$	12287-44-4	**	8.66 (V)	PE	4908
$\text{C}_8\text{H}_6\text{O}_6\text{Fe}^+$	$\text{CH}_2=\text{CHCOOCH}_3(\text{CO})_4\text{Fe}$	12287-67-1	**	8.50 (V)	PE	4908
$\text{C}_{15}\text{H}_{21}\text{O}_6\text{Fe}^+$	$(\text{CH}_3\text{COCHCOCH}_3)_3\text{Fe}$ (Iron, tris(2,4-pentanedionato-O,O')-, (OC-6-11)-)	14024-18-1	**	8.10±0.07 (V)	PE	3682
$\text{C}_{33}\text{H}_{57}\text{O}_6\text{Fe}^+$	$((\text{CH}_3)_3\text{CCOCHCO}(\text{CH}_3)_3)_3\text{Fe}$ (Iron, tris(2,2,6,6-tetramethyl-3,5-heptanedionato-O,O')-)	14876-47-2	**	7.92±0.07 (V)	PE	3682
$\text{C}_{10}\text{H}_8\text{O}_8\text{Fe}^+$	$\text{CH}_3\text{OOCCH}=\text{CHCOOCH}_3(\text{CO})_4\text{Fe}$	33248-78-1	**	8.68 (V)	PE	4908
$\text{C}_{14}\text{H}_{10}\text{O}_4\text{Fe}_2^+$	$trans-((\text{C}_5\text{H}_5)(\text{CO})_2\text{Fe})_2$ (Iron, di- μ -carbonyldicarbonylbis(η^5 -2,4-cyclopentadien-1-yl) di-(Fe-Fe))	32757-46-3	**	6.95 (V)	PE	5317
$\text{C}_{21}\text{H}_{20}\text{O}_4\text{Fe}_4^+$	$(\text{C}_5\text{H}_5\text{COFe})_4$ (Iron, tetra- μ 3-carbonyltetrakis(η^5 -2,4-cyclopentadien-1-yl)tetra-tetrahedro)	12203-87-1	**	6.45 (V)	PE	4565
$\text{B}_4\text{C}_3\text{H}_8\text{O}_3\text{Fe}^+$	$\text{B}_4\text{H}_4(\text{CO})_3\text{Fe}$ (Iron, tricarbonyl[(1,2,3,4- η)-tetraborane(8)]-)	54748-47-9	**	8.6 (V)	PE	4937
$\text{B}_5\text{C}_3\text{H}_9\text{O}_3\text{Fe}^+$	$\text{B}_5\text{H}_6(\text{CO})_3\text{Fe}$ (Iron, tricarbonyl [nonahydriopentaborate (2-)]-)	61403-41-6	**	8.4 (V)	PE	4937

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
B₃C₅H₅O₃Fe⁺	C ₂ H ₅ B ₃ (CO) ₃ Fe (Iron, tricarbonyl [η^4 -pentahydrodicarbapentaborato(2-)]-)	53363-10-3	**	8.6 (V)	PE	4937
B₃C₅H₇O₃Fe⁺	C ₂ H ₇ B ₃ (CO) ₃ Fe (Iron, tricarbonyl [(1,2,3,4,5- η)-heptahydro-1,2-dicarbapentaborate (2-)]-)	36657-30-4	**	8.7	PE	4937
B₅C₅H₃O₅Fe⁺	B ₅ H ₃ (CO) ₅ Fe (Iron, tricarbonyl [(2,3,4,5- η)-dicarbonyltrihydropentaborato (2-)]-)	61525-93-7	**	8.0 (V)	PE	4937
C₂N₂O₄Fe⁺	(CO) ₂ (NO) ₂ Fe	13682-74-1	**	8.16±0.04	PE	5225
C₇H₃NO₄Fe⁺	(CH ₂ =CHCN)(CO) ₄ Fe	15602-77-4	**	8.90 (V)	PE	5559
C₉H₅NO₄Fe⁺	(C ₅ H ₅ N)(CO) ₄ Fe (Iron,tetracarbonyl(pyridine)-(TB-5-12)-)	53317-88-7	**	7.65 (V)	PE	5559
C₁₅H₁₂O₆F₉Fe⁺	(CF ₃ COCHCOCH ₃) ₃ Fe (Iron, tris(1,1,1-trifluoro-2,4-pentanedionato-O,O')-	14526-22-8	**	9.18±0.07 (V)	PE	3682
C₁₅H₃O₆F₁₈Fe⁺	(CF ₃ COCHCOCF ₃) ₃ Fe (Iron, tris(1,1,1,5,5-hexafluoro-2,4-pentanedionato-O,O'), (OC-6-11)-)	17786-67-3	**	10.13±0.07 (V)	PE	3682
C₁₃H₁₈SiFe⁺	(C ₅ H ₅)(C ₅ H ₄ Si(CH ₃) ₃)Fe (Ferrocene, (trimethylsilyl)-)	12215-68-8	**	9.5±0.10	EI	3729
C₆H₁₈N₃PF₆⁺	((CH ₃) ₂ N) ₃ P)(CO) ₄ Fe	19372-47-5	4CO	10.2±0.05	EI	3952
C₁₂H₃₆N₆P₂Fe⁺	((CH ₃) ₂ N) ₃ P ₂ (CO) ₄ Fe	19372-46-4	3CO	11.7±0.05	EI	3952
C₇H₉O₄PF₆⁺	(P(CH ₃) ₃)(CO) ₄ Fe (JC-Mean value of Jahn-Teller components)	18475-02-0	**	7.77 (V)	PE	5559
C₂₂H₁₅O₄PF₆⁺	(P(C ₆ H ₅) ₃)(CO) ₄ Fe (Iron,tetracarbonyl(triphenylphosphine)-)	14649-69-5	**	7.55 (V)	PE	5559
C₇H₁₈N₃OPF₆⁺	((CH ₃) ₂ N) ₃ P)(CO) ₄ Fe	19372-47-5	3CO	10.2±0.05	EI	3952
C₈H₁₈N₃O₂PF₆⁺	((CH ₃) ₂ N) ₃ P)(CO) ₄ Fe	19372-47-5	2CO	9.8±0.05	EI	3952
C₉H₁₈N₃O₃PF₆⁺	((CH ₃) ₂ N) ₃ P)(CO) ₄ Fe	19372-47-5	CO	9.4±0.05	EI	3952
C₁₀H₁₈N₃O₄PF₆⁺	((CH ₃) ₂ N) ₃ P)(CO) ₄ Fe	19372-47-5	**	9.0±0.05	EI	3952

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{13}\text{H}_{36}\text{N}_6\text{OP}_2\text{Fe}^+$	$((\text{CH}_3)_2\text{N})_3\text{P}_2(\text{CO})_3\text{Fe}$	19372-46-4	2CO	10.2 ± 0.05	EI	3952
$\text{C}_{11}\text{H}_{36}\text{N}_6\text{O}_2\text{P}_2\text{Fe}^+$	$((\text{CH}_3)_2\text{N})_3\text{P}_2(\text{CO})_3\text{Fe}$	19372-46-4	CO	9.7 ± 0.05	EI	3952
$\text{C}_{15}\text{H}_{36}\text{N}_6\text{O}_3\text{P}_2\text{Fe}^+$	$((\text{CH}_3)_2\text{N})_3\text{P}_2(\text{CO})_3\text{Fe}$	19372-46-4	**	7.7 ± 0.05	EI	3952
$\text{F}_{15}\text{P}_5\text{Fe}^+$	$(\text{PF}_3)_5\text{Fe}$	13815-34-4	** ** **	8.9 9.15 (V) 8.83	PE PE EI	4021 4456 5453
$\text{H}_2\text{F}_{12}\text{P}_4\text{Fe}^+$	$\text{FeH}_2(\text{PF}_3)_4$	24899-55-6	**	9.78 (V)	PE	4720
$\text{C}_4\text{O}_4\text{F}_3\text{PFe}^+$	$\text{PF}_3\text{Fe}(\text{CO})_4$	16388-47-9	**	8.75 (V)	PE	4753
$\text{C}_3\text{O}_3\text{F}_6\text{P}_2\text{Fe}^+$	$(\text{PF}_3)_2\text{Fe}(\text{CO})_3$	16454-87-8	** **	8.95 (V) 8.47	PE EI	4753 5453
$\text{C}_2\text{O}_2\text{F}_9\text{P}_3\text{Fe}^+$	$(\text{PF}_3)_3(\text{CO})_2\text{Fe}$	16388-46-8	**	8.61	EI	5453
$\text{COF}_{12}\text{P}_4\text{Fe}^+$	$(\text{PF}_3)_4\text{FeCO}$	16388-45-7	** **	9.18 (V) 8.62	PE EI	4753 5453
CSFe^+	$(\text{C}_5\text{H}_5)_2(\text{CS})(\text{CO})_3\text{Fe}_2$ (Iron, μ -carbonothioyl- μ -carbonyldicarbonylbis(η^5 -2,4-cyclopentadien-1-yl)di-(Fe-Fe)-)	67113-80-8		11.88 ± 0.03	EI	5423
	$(\text{C}_5\text{H}_5)_2(\text{CS})_2(\text{CO})_2\text{Fe}_2$ (Iron, bis[μ -(carbonothioyl)]dicarbonylbis(η^5 -2,4-cyclopentadien-1-yl)di-(Fe-Fe)-)	67225-86-9		12.02 ± 0.06	EI	5423
CSFe_2^+	$(\text{C}_5\text{H}_5)_2(\text{CS})(\text{CO})_3\text{Fe}_2$ (Iron, μ -carbonothioyl- μ -carbonyldicarbonylbis(η^5 -2,4-cyclopentadien-1-yl)di-(Fe-Fe)-)	67113-80-8		26.00 ± 0.03	EI	5423
	$(\text{C}_5\text{H}_5)_2(\text{CS})_2(\text{CO})_2\text{Fe}_2$ (Iron, bis[μ -(carbonothioyl)]dicarbonylbis(η^5 -2,4-cyclopentadien-1-yl)di-(Fe-Fe)-)	67225-86-9		20.00 ± 0.03	EI	5423
$\text{C}_2\text{S}_2\text{Fe}_2^+$	$(\text{C}_5\text{H}_5)_2(\text{CS})(\text{CO})_3\text{Fe}_2$ (Iron, μ -carbonothioyl- μ -carbonyldicarbonylbis(η^5 -2,4-cyclopentadien-1-yl)di-(Fe-Fe)-)	67113-80-8		26.00 ± 0.03	EI	5582
	$(\text{C}_5\text{H}_5)_2(\text{CS})_2(\text{CO})_2\text{Fe}_2$ (Iron, bis[μ -(carbonothioyl)]dicarbonylbis(η^5 -2,4-cyclopentadien-1-yl)di-(Fe-Fe)-)	67225-86-9		22.53 ± 0.03	EI	5423
$\text{C}_6\text{H}_5\text{SFe}^+$	$(\text{C}_5\text{H}_5)_2(\text{CS})(\text{CO})_3\text{Fe}_2$ (Iron, μ -carbonothioyl- μ -carbonyldicarbonylbis(η^5 -2,4-cyclopentadien-1-yl)di-(Fe-Fe)-)	67113-80-8		14.74 ± 0.04	EI	5423
	$(\text{C}_5\text{H}_5)_2(\text{CS})_2(\text{CO})_2\text{Fe}_2$ (Iron, bis[μ -(carbonothioyl)]dicarbonylbis(η^5 -2,4-cyclopentadien-1-yl)di-(Fe-Fe)-)	67225-86-9		14.74 ± 0.09	EI	5423

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_6\text{H}_5\text{SFe}_2^+$	$(\text{C}_5\text{H}_5)_2(\text{CS})(\text{CO})_3\text{Fe}_2$ (Iron, μ -carbonothioyl- μ -carbonyldicarbonylbis(η^5 -2,4-cyclo pentadien-1-yl)di-(Fe-Fe)-)	67113-80-8	$\text{C}_5\text{H}_5 + 3\text{CO}$	16.79 ± 0.02	EI	5423
	$(\text{C}_5\text{H}_5)_2(\text{CS})_2(\text{CO})_2\text{Fe}_2$ (Iron, bis[μ -(carbonothioyl)]dicarbonylbis(η^5 -2,4-cyclopentadien-1-yl) di-(Fe-Fe)-)	67225-86-9		17.19 ± 0.02	EI	5423
$\text{C}_{11}\text{H}_{10}\text{SFe}_2^+$	$(\text{C}_5\text{H}_5)_2(\text{CS})(\text{CO})_3\text{Fe}_2$ (Iron, μ -carbonothioyl- μ -carbonyldicarbonylbis(η^5 -2,4-cyclo pentadien-1-yl)di-(Fe-Fe)-)	67113-80-8	3CO	10.44 ± 0.04	EI	5423
	$(\text{C}_5\text{H}_5)_2(\text{CS})_2(\text{CO})_2\text{Fe}_2$ (Iron, bis[μ -(carbonothioyl)]dicarbonylbis(η^5 -2,4-cyclopentadien-1-yl) di-(Fe-Fe)-)	67225-86-9	$2\text{CO} + \text{CS}$	12.61 ± 0.04	EI	5423
$\text{C}_7\text{H}_5\text{S}_2\text{Fe}_2^+$	$(\text{C}_5\text{H}_5)_2(\text{CS})_2(\text{CO})_2\text{Fe}_2$ (Iron, bis[μ -(carbonothioyl)]dicarbonylbis(η^5 -2,4-cyclopentadien-1-yl) di-(Fe-Fe)-)	67225-86-9		13.23 ± 0.02	EI	5423
$\text{C}_{12}\text{H}_{10}\text{S}_2\text{Fe}_2^+$	$(\text{C}_5\text{H}_5)_2(\text{CS})_2(\text{CO})_3\text{Fe}_2$ (Iron, bis[μ -(carbonothioyl)]dicarbonylbis(η^5 -2,4-cyclopentadien-1-yl) di-(Fe-Fe)-)	67225-86-9	2CO	8.89 ± 0.03	EI	5423
$\text{C}_9\text{H}_{18}\text{N}_3\text{S}_6\text{Fe}^+$	$[\text{S}_2\text{CN}(\text{CH}_3)_2]_3\text{Fe}$	14484-64-1	**	7.72 (V)	PE	4710
$\text{C}_6\text{O}_6\text{S}_2\text{Fe}_2^+$	$(\text{CO})_6\text{Fe}_2\text{S}_2$	14243-23-3	**	7.9 (V)	PE	5536
$\text{C}_7\text{H}_5\text{OSFe}^+$	$(\text{C}_5\text{H}_5)_2(\text{CS})(\text{CO})_3\text{Fe}_2$ (Iron, μ -carbonothioyl- μ -carbonyldicarbonylbis(η^5 -2,4-cyclo pentadien-1-yl)di-(Fe-Fe)-)	67113-80-8		13.13 ± 0.03	EI	5423
	$(\text{C}_5\text{H}_5)_2(\text{CS})_2(\text{CO})_2\text{Fe}_2$ (Iron, bis[μ -(carbonothioyl)]dicarbonylbis(η^5 -2,4-cyclopentadien-1-yl) di-(Fe-Fe)-)	67225-86-9		13.83 ± 0.11	EI	5423
$\text{C}_{12}\text{H}_{10}\text{OSFe}_2^+$	$(\text{C}_5\text{H}_5)_2(\text{CS})(\text{CO})_3\text{Fe}_2$ (Iron, μ -carbonothioyl- μ -carbonyldicarbonylbis(η^5 -2,4-cyclo pentadien-1-yl)di-(Fe-Fe)-)	67113-80-8	2CO	8.58 ± 0.02	EI	5423
$\text{C}_{13}\text{H}_{10}\text{O}_2\text{SFe}_2^+$	$(\text{C}_5\text{H}_5)_2(\text{CS})(\text{CO})_3\text{Fe}_2$ (Iron, μ -carbonothioyl- μ -carbonyldicarbonylbis(η^5 -2,4-cyclo pentadien-1-yl)di-(Fe-Fe)-)	67113-80-8	CO	7.62 ± 0.02	EI	5423
$\text{C}_{11}\text{H}_{10}\text{O}_3\text{SFe}_2^+$	$(\text{C}_5\text{H}_5)_2(\text{CS})(\text{CO})_3\text{Fe}_2$ (Iron, μ -carbonothioyl- η -carbonyldicarbonylbis(η^5 -2,4-cyclo pentadien-1-yl)di-(Fe-Fe)-)	67113-80-8	**	6.46 ± 0.02	EI	5423
$\text{C}_{13}\text{H}_{10}\text{OS}_2\text{Fe}_2^+$	$(\text{C}_5\text{H}_5)_2(\text{CS})_2(\text{CO})_2\text{Fe}_2$ (Iron, bis[μ -(carbonothioyl)]dicarbonylbis(η^5 -2,4-cyclopentadien-1-yl) di-(Fe-Fe)-)	67225-86-9	CO	7.47 ± 0.02	EI	5423
$\text{C}_{11}\text{H}_{10}\text{O}_2\text{S}_2\text{Fe}_2^+$	$(\text{C}_5\text{H}_5)_2(\text{CS})_2(\text{CO})_2\text{Fe}_2$ (Iron, bis[μ -(carbonothioyl)]dicarbonylbis(η^5 -2,4-cyclopentadien-1-yl) di-(Fe-Fe)-)	67225-86-9	**	6.76 ± 0.04	EI	5423

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{12}\text{H}_{11}\text{O}_6\text{S}_2\text{Fe}_2^+$ $(\text{iso-C}_3\text{H}_7\text{S})_2(\text{CO})_6\text{Fe}_2$		26411-94-9	**	7.5 (V)	PE	5536
$\text{C}_8\text{H}_{12}\text{N}_2\text{O}_2\text{S}_4\text{Fe}^+$ $[\text{S}_2\text{CN}(\text{CH}_3)_2]_2(\text{CO})_2\text{Fe}$		36309-89-4	**	8.51 (V)	PE	4710
$\text{C}_{14}\text{H}_{20}\text{N}_2\text{O}_2\text{S}_4\text{Fe}^+$ $[\text{C}_5\text{H}_{11}\text{N}(\text{CS}_2)]_2(\text{CO})_2\text{Fe}$ (Iron, dicarbonylbis(1-piperidinecarbodithioato-S,S')-(OC-6-21)-)		35816-66-1	**	8.57 (V)	PE	4710
$\text{C}_{18}\text{H}_{16}\text{N}_2\text{O}_2\text{S}_4\text{Fe}^+$ $\text{C}_{18}\text{H}_{16}\text{N}_2\text{O}_2\text{S}_4\text{Fe}$ (Iron, dicarbonylbis(methylphenylcarbamodithioato-S,S')-(OC-6-21)-)		63796-70-3	**	7.77 (V)	PE	4710
$\text{C}_{18}\text{H}_{28}\text{N}_2\text{O}_2\text{S}_4\text{Fe}^+$ $\text{C}_{18}\text{H}_{28}\text{N}_2\text{O}_2\text{S}_4\text{Fe}$ (Iron, dicarbonylbis(2,6-dimethyl-1-piperidinecarbodithioato-S,S')-(OC-6-21)-)		63796-67-8	**	8.26 (V)	PE	4710
$\text{C}_{20}\text{H}_{20}\text{N}_2\text{O}_2\text{S}_4\text{Fe}^+$ $\text{C}_{20}\text{H}_{20}\text{N}_2\text{O}_2\text{S}_4\text{Fe}$ (Iron, dicarbonylbis(ethylphenylcarbamodithioato-S,S')-(OC-6-21)-)		63796-69-0	**	7.76 (V)	PE	4710
$\text{C}_{22}\text{H}_{24}\text{N}_2\text{O}_2\text{S}_4\text{Fe}^+$ $\text{C}_{22}\text{H}_{24}\text{N}_2\text{O}_2\text{S}_4\text{Fe}$ (Iron, dicarbonylbis[ethyl(phenylmethyl)carbamodithioato-S,S']-(OC-6-21)-)		63796-64-5	**	7.90 (V)	PE	4710
$\text{C}_{28}\text{H}_{20}\text{N}_2\text{O}_2\text{S}_4\text{Fe}^+$ $\text{C}_{28}\text{H}_{20}\text{N}_2\text{O}_2\text{S}_4\text{Fe}$ (Iron, dicarbonylbis(diphenylcarbamodithioato-S,S')-(OC-6-21)-)		63796-68-9	**	7.58 (V)	PE	4710
$\text{C}_{12}\text{H}_{16}\text{N}_2\text{O}_4\text{S}_4\text{Fe}^+$ $\text{C}_{12}\text{H}_{16}\text{N}_2\text{O}_4\text{S}_4\text{Fe}$ (Iron, dicarbonylbis(4-morpholinecarbodithioato-S,S')-(OC-6-21)-)		63796-66-7	**	8.64 (V)	PE	4710
$\text{C}_{10}\text{H}_9\text{NO}_6\text{SFe}_2^+$ $(\text{tert-C}_4\text{H}_9\text{NS})(\text{CO})_6\text{Fe}_2$		41812-87-7	**	7.8 (V)	PE	5536
Cl_2Fe^+ FeCl_2		7758-94-3	**	10.34 (V)	PE	5172
$\text{C}_{10}\text{H}_9\text{ClFe}^+$ $(\text{C}_5\text{H}_5)(\text{C}_5\text{H}_4\text{Cl})\text{Fe}$ (Ferrocene, chloro-)		1273-74-1	**	6.83 \pm 0.05	PI	3729
$\text{C}_{10}\text{H}_8\text{Cl}_2\text{Fe}^+$ $(\text{C}_5\text{H}_4\text{Cl})_2\text{Fe}$ (Ferrocene, 1,1'-dichloro-)		1293-67-0	**	7.03 (V)	PE	3688
$\text{C}_{44}\text{H}_{28}\text{N}_4\text{ClFe}^+$ $\text{C}_{20}\text{H}_8\text{N}_4(\text{C}_6\text{H}_5)_4\text{FeCl}$ (Iron, chloro[5,10,15,20-tetraphenyl-21H,23H-porphinato(2-)N ²¹ ,N ²² ,N ²³ ,N ²⁴]- (SP-5-12)-)		16456-81-8	**	6.09 \pm 0.2	OTH	4962
$\text{C}_7\text{H}_5\text{O}_2\text{ClFe}^+$ $\text{C}_5\text{H}_5(\text{CO})_2\text{FeCl}$ (Iron, dicarbonylchloro(η^5 -2,4-cyclopentadien-1-yl)-)		12107-04-9	**	8.00 (V)	PE	4565
			**	8.00 (V)	PE	4570
$\text{C}_6\text{H}_2\text{O}_4\text{Cl}_2\text{Fe}^+$ $\text{CCl}_2\text{CH}_2(\text{CO})_4\text{Fe}$ $\text{trans-C}_2\text{H}_2\text{Cl}_2(\text{CO})_4\text{Fe}$		52613-75-9	**	8.82 (V)	PE	4908
		52646-80-7	**	8.72 (V)	PE	4908

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
Co⁺						
	(C ₅ H ₅) ₂ Co (Cobaltocene)	1277-43-6		14.10±0.15	EI	4072
	Cl ₃ SiCo(CO) ₃ PF ₃	37769-28-1		18.9±0.5	EI	3653
	Cl ₃ SiCo(CO) ₂ (PF ₃) ₂	37769-29-2		18.9±0.4	EI	3653
C₃H₃Co⁺						
	(C ₅ H ₅) ₂ Co (Cobaltocene)	1277-43-6		17.50±0.2	EI	4072
C₅H₅Co⁺						
	(C ₅ H ₅) ₂ Co (Cobaltocene)	1277-43-6		13.20±0.2	EI	4072
				14.0±0.3	EI	3793
C₁₀H₁₀Co⁺						
	(C ₅ H ₅) ₂ Co (Cobaltocene)	1277-43-6	**	5.55 (V)	PE	5394
			**	5.56 (V)	PE	5507
			**	5.7±0.2	EI	3793
			**	5.95±0.1	EI	4072
C₁₂H₁₄Co⁺						
	(C ₅ H ₅ CH ₃) ₂ Co (Cobaltocenium, 1,1'-dimethyl-)	40759-60-2	**	5.37 (V)	PE	5507
C₂₀H₃₀Co⁺						
	(C ₅ (CH ₃) ₅) ₂ Co (Cobaltocene, decamethyl-)	XXXXX-XX-X	**	4.705 (V)	PE	5394
C₂H₂Co⁺						
	CH≡CH(CO) ₆ Co ₂ (Cobalt, hexacarbonyl(ethyne)di-)	12553-66-1	6CO	15.58±0.05	EI	4116
C₄H₆Co₂⁺						
	(CO) ₆ CH ₃ C≡CCH ₃ Co ₂ (Cobalt, (2-butyne)hexacarbonyldi-)	12282-08-5	6CO	15.60±0.05	EI	4116
BC₁₁H₁₃Co⁺						
	(C ₅ H ₅)(C ₅ H ₅ BCH ₃)Co (Cobalt, (η^5 -2,4-cyclopentadien-1-yl)[(1,2,3,4,5,6- η)-1-methylboratabenzene]-)	36534-25-5	**	6.56±0.1	EI	3545
B₂C₁₂H₁₆Co⁺						
	(C ₅ H ₅ BCH ₃) ₂ Co (Cobalt, bis[(1,2,3,4,5,6- η)-1-methylboratabenzene]-)	36534-27-7	**	7.15±0.1	EI	3545
BC₁₆H₁₅Co⁺						
	(C ₅ H ₅)(C ₅ H ₅ BC ₆ H ₅)Co (Cobalt, (η^5 -2,4-cyclopentadien-1-yl)[(1,2,3,4,5,6- η)-1-phenylboratabenzene]-)	36682-12-9	**	6.63±0.1	EI	3545
B₂C₂₂H₂₀Co⁺						
	(C ₅ H ₅ BC ₆ H ₅) ₂ Co (Cobalt, bis[(1,2,3,4,5,6- η)-1-phenylboratabenzene]-)	36534-31-3	**	7.25±0.1	EI	3545
C₁₆H₁₆N₄Co⁺						
	C ₁₆ H ₁₆ N ₄ Co (Cobalt, [N,N'-bis(2-aminophenyl)methylene] -1,2-ethanediaminato(2-)-N,N',N'',N''']-)	21177-97-9	**	6.98±0.10	EI	4668
C₃₆H₄₄N₄Co⁺						
	((C ₅ H ₅) ₂ C ₆ NCH) ₂ Co (Cobalt,[2,3,7,8,12,13,17,18-octaethyl-21H,23H-porphinato(2-) N ²¹ ,N ²² ,N ²³ ,N ²⁴]-(SP-4-1)-)	17632-19-8	**	6.09±0.03 (V)	PE	5476

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{44}\text{H}_{28}\text{N}_4\text{Co}^+$	$\text{C}_{21}\text{H}_8\text{N}_4(\text{C}_6\text{H}_5)_4\text{Co}$ (Cobalt, [5,10,15,20-tetraphenyl-21H,23H-porphinato(2-)-N ²¹ ,N ²² ,N ²³ ,N ²⁴]- (SP-4-1)-)	14172-90-8	**	6.12±0.2	OTH	4962
$\text{C}_{32}\text{H}_{16}\text{N}_8\text{Co}^+$	$\text{C}_{12}\text{H}_{10}\text{N}_8\text{Co}$ (Cobalt, [29H,31H-phthalocyaninato(2-)-N ²⁹ ,N ³⁰ ,N ³¹ ,N ³²]- (SP-4-1)-)	3317-67-7	**	7.46±0.10	EI	3829
COCo^+	$\text{Cl}_3\text{SiCo}(\text{CO})_3\text{PF}_3$ $\text{Cl}_3\text{SiCo}(\text{CO})_2(\text{PF}_3)_2$	37769-28-1 37769-29-2		16.7±0.3 16.9±0.4	EI EI	3653 3653
$\text{C}_2\text{O}_2\text{Co}^+$	$\text{Cl}_3\text{SiCo}(\text{CO})_3\text{PF}_3$ $\text{Cl}_3\text{SiCo}(\text{CO})_2(\text{PF}_3)_2$	37769-28-1 37769-29-2		15.5±0.4 15.5±0.3	EI EI	3653 3653
$\text{C}_4\text{HO}_4\text{Co}^+$	$(\text{CO})_4\text{CoH}$	16842-03-8	** **	8.90±0.02 (V) 8.90 (V)	PE PE	3827 4456
$\text{C}_{10}\text{H}_{14}\text{O}_4\text{Co}^+$	$(\text{CH}_3\text{C}(\text{O})=\text{CHCOCH}_3)_2\text{Co}$	14024-48-7	**	8.50 (V)	PE	5100
$\text{C}_{22}\text{H}_{40}\text{O}_4\text{Co}^+$	$((\text{CH}_3)_3\text{CCO})_2\text{CH}_2\text{Co}$	XXXXX-XX-X	**	7.92 (V)	PE	5568
$\text{C}_{15}\text{H}_{21}\text{O}_6\text{Co}^+$	$(\text{C}_5\text{H}_7\text{O}_2)_3\text{Co}$ (Cobalt, tris(2,4-pentanedionato-0,0')-(OC-6-11)-)	21679-46-9	** **	7.52 (V) 7.52±0.07 (V)	PE PE	4965 3682
$\text{C}_3\text{H}_2\text{OCo}^+$	$\text{CH}\equiv\text{CH}(\text{CO})_6\text{Co}_2$ (Cobalt, hexacarbonyl(ethyne)di-)	12553-66-1	5CO	14.11±0.05	EI	4116
$\text{C}_5\text{H}_6\text{OCo}^+$	$(\text{CO})_6\text{CH}_3\text{C}\equiv\text{CCH}_3\text{Co}_2$ (Cobalt, (2-butyne)hexacarbonyldi-)	12282-08-5	5CO	13.85±0.05	EI	4116
$\text{C}_4\text{H}_2\text{O}_2\text{Co}^+$	$\text{CH}\equiv\text{CH}(\text{CO})_6\text{Co}_2$ (Cobalt, hexacarbonyl(ethyne)di-)	12553-66-1	4CO	12.40±0.05	EI	4116
$\text{C}_6\text{H}_6\text{O}_2\text{Co}^+$	$(\text{CO})_6\text{CH}_3\text{C}\equiv\text{CCH}_3\text{Co}_2$ (Cobalt, (2-butyne)hexacarbonyldi-)	12282-08-5	4CO	12.36±0.05	EI	4116
$\text{C}_5\text{H}_2\text{O}_3\text{Co}^+$	$\text{CH}\equiv\text{CH}(\text{CO})_6\text{Co}_2$ (Cobalt, hexacarbonyl(ethyne)di-)	12553-66-1	3CO	10.96±0.05	EI	4116
$\text{C}_7\text{H}_6\text{O}_3\text{Co}^+$	$(\text{CO})_6\text{CH}_3\text{C}\equiv\text{CCH}_3\text{Co}_2$ (Cobalt, (2-butyne)hexacarbonyldi-)	12282-08-5	3CO	10.98±0.05	EI	4116
$\text{C}_6\text{H}_2\text{O}_4\text{Co}^+$	$\text{CH}\equiv\text{CH}(\text{CO})_6\text{Co}_2$ (Cobalt, hexacarbonyl(ethyne)di-)	12553-66-1	2CO	9.74±0.05	EI	4116

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_8\text{H}_6\text{O}_4\text{Co}_2^+$	$(\text{CO})_6\text{CH}_3\text{C}\equiv\text{CCH}_3\text{Co}_2$ (Cobalt, (2-butyne)hexacarbonyldi-)	12282-08-5	2CO	9.68 ± 0.05	EI	4116
$\text{C}_7\text{H}_2\text{O}_5\text{Co}_2^+$	$\text{CH}\equiv\text{CH}(\text{CO})_6\text{Co}_2$ (Cobalt, hexacarbonyl(ethyne)di-)	12553-66-1	CO	8.71 ± 0.05	EI	4116
$\text{C}_9\text{H}_6\text{O}_5\text{Co}_2^+$	$(\text{CO})_6\text{CH}_3\text{C}\equiv\text{CCH}_3\text{Co}_2$ (Cobalt, (2-butyne)hexacarbonyldi-)	12282-08-5	CO	8.62 ± 0.05	EI	4116
$\text{C}_8\text{H}_2\text{O}_6\text{Co}_2^+$	$\text{CH}\equiv\text{CH}(\text{CO})_6\text{Co}_2$ (Cobalt, hexacarbonyl(ethyne)di-)	12553-66-1	**	7.96 ± 0.05	EI	4116
$\text{C}_{10}\text{H}_6\text{O}_6\text{Co}_2^+$	$(\text{CO})_6\text{CH}_3\text{C}\equiv\text{CCH}_3\text{Co}_2$ (Cobalt, (2-butyne)hexacarbonyldi-)	12282-08-5	**	7.80 ± 0.05	EI	4116
$\text{C}_{15}\text{H}_8\text{O}_6\text{Co}_2^+$	$(\text{CO})_6\text{CH}_3\text{C}\equiv\text{CC}_6\text{H}_5\text{Co}_2$ (Cobalt, hexacarbonyl[μ -{(1,2- η ;1,2- η)-1-propynylbenzene}]di-, (Co-Co))	53556-74-4	**	7.85 ± 0.05	EI	4116
$\text{B}_2\text{C}_{12}\text{H}_{16}\text{O}_2\text{Co}^+$	$(\text{C}_5\text{H}_5\text{BOCH}_3)_2\text{Co}$ (Cobalt, bis[(1,2,3,4,5,6- η)-1-methoxyboratabenzene]-)	36534-20-0	**	7.02 ± 0.1	EI	3545
$\text{N}_3\text{O}_9\text{Co}^+$	$(\text{NO}_3)_3\text{Co}$	55866-74-5	**	10.79 ± 0.03 (V)	PE	4999
$\text{C}_3\text{NO}_4\text{Co}^+$	$(\text{CO})_3\text{NOCO}$	14096-82-3	**	8.26 ± 0.03	PE	5225
$\text{C}_{16}\text{H}_{14}\text{N}_2\text{O}_2\text{Co}^+$	$\text{C}_{16}\text{H}_{14}\text{N}_2\text{O}_2\text{Co}$ (Cobalt, [[2,2'-[1,2-ethanediylbis(nitrilomethylidyne)] bis[phenolato]](2-)N,N',O,O']-)	14167-18-1	**	7.52 ± 0.06	EI	4668
$\text{C}_{21}\text{H}_{24}\text{N}_2\text{O}_2\text{Co}^+$	$\text{C}_{21}\text{H}_{24}\text{N}_2\text{O}_2\text{Co}$ (Cobalt, [[2,2'-[1,7-heptanediylyl]bis(nitrilomethylidyne)] bis[phenolato]](2-)N,N',O,O']-, (T-4)-)	17084-78-5	**	7.78 ± 0.08	EI	4213
$\text{C}_{20}\text{H}_{23}\text{N}_3\text{O}_2\text{Co}^+$	$\text{C}_{20}\text{H}_{23}\text{N}_3\text{O}_2\text{Co}$ (Cobalt, [[2,2'-[iminobis(3,1-propanediyl)nitrilomethylidyne)] bis[phenolato]](2-)N,N',N'', O,O']-)	15306-22-6	**	7.31 ± 0.07	EI	4213
$\text{C}_{20}\text{H}_{22}\text{N}_2\text{O}_3\text{Co}^+$	$\text{C}_{20}\text{H}_{22}\text{N}_2\text{O}_3\text{Co}$ (Cobalt, [[2,2'-[oxybis(3,1-propanediyl)nitrilomethylidyne)] bis[phenolato]](2-)N ² ,N ² ',O, O']-(T-4)-)	52279-51-3	**	7.53 ± 0.10	EI	4213
$\text{C}_{15}\text{H}_{18}\text{N}_3\text{O}_{12}\text{Co}^+$	$(\text{C}_6\text{H}_6\text{O}_2\text{NO}_2)_3\text{Co}$ (Cobalt, tris(3-nitro-2,4-pentanedionato-O ²⁻ ,O ⁴⁻)-(OC-6-11)-)	15169-25-2	**	8.51 (V)	PE	4965
$\text{C}_4\text{F}_6\text{Co}_2^+$	$\text{CF}_3\text{C}\equiv\text{CCF}_3(\text{CO})_6\text{Co}_2$ (Cobalt, hexacarbonyl(1,1,1,4,4,4-hexafluoro-2-butyne)di-)	12557-89-0	6CO	15.72 ± 0.05	EI	4116

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_5\text{H}_5\text{Ni}^+$	$(\text{C}_5\text{H}_5)_2\text{Ni}$ (Nickelocene)	1271-28-9		12.6 ± 0.2	EI	3793
	$\text{C}_5\text{H}_5\text{NiNO}$ (Nickel, (η^5 -2,4-cyclopentadien-1-yl)nitrosyl-)	12071-73-7	C_5H_5	13.00 ± 0.25 10.5	EI EI	3628 4015
$\text{C}_6\text{H}_{10}\text{Ni}^+$	$(\text{C}_5\text{H}_5)_2\text{Ni}$	12077-85-9	**	7.76 (V)	PE	5281
			**	7.33 ± 0.04	PE	3711
			**	7.76 (V)	PE	4396
$\text{C}_8\text{H}_8\text{Ni}^+$	$(\text{C}_5\text{H}_5)_2\text{Ni}$ (Nickelocene)	1271-28-9	C_2H_2	12.6 ± 0.1	EI	3628
$\text{C}_8\text{H}_{14}\text{Ni}^+$	$(\text{CH}_3\text{CH}=\text{CHCH}_2)_2\text{Ni}$ (Nickel, bis(1,2,3- η)-(1-methyl-2-propenyl)-)	12145-63-0	**	7.53 (V)	PE	4396
	$(\text{CH}_2=\text{C}(\text{CH}_3)\text{CH}_2)_2\text{Ni}$ (Nickel, bis(1,2,3- η)-2-methyl-2-propenyl-)	12261-14-2	**	7.53 (V)	PE	4396
			**	7.53 (V)	PE	5281
$\text{C}_{10}\text{H}_{10}\text{Ni}^+$	$(\text{C}_5\text{H}_5)_2\text{Ni}$ (Nickelocene)	1271-28-9	**	6.2	PE	3725
			**	6.50 (V)	PE	5394
			**	6.51 (V)	PE	5507
			**	6.50 ± 0.25	EI	3628
			**	6.8 ± 0.1	EI	3793
$\text{C}_{10}\text{H}_{18}\text{Ni}^+$	$(\text{CH}_3\text{CH}=\text{CHCHCH}_3)_2\text{Ni}$ (Nickel, bis[(1,2,3- η)-2-pentenyl]-)	43062-19-7	**	7.22 (V)	PE	4396
$\text{C}_{12}\text{H}_{14}\text{Ni}^+$	$(\text{C}_5\text{H}_4\text{CH}_3)_2\text{Ni}$ (Nickelocene,1,1'-dimethyl-)	1293-95-4	**	6.36 (V)	PE	5507
$\text{C}_{20}\text{H}_{30}\text{Ni}^+$	$(\text{C}_5(\text{CH}_3)_5)_2\text{Ni}$ (Nickelocene,decamethyl-)	XXXXX-XX-X	**	5.82 (V)	PE	5394
$\text{C}_{16}\text{H}_{16}\text{N}_4\text{Ni}^+$	$\text{C}_{16}\text{H}_{16}\text{N}_4\text{Ni}$ (Nickel, [N,N'-bis[(2-aminophenyl)methylene] -1,2-ethanediaminato(2-)-N,N',N'',N'''])-	15738-33-7	**	6.84 ± 0.08	EI	4668
$\text{C}_{36}\text{H}_{44}\text{N}_4\text{Ni}^+$	$((\text{C}_2\text{H}_5)_2\text{C}_4\text{NCH})_4\text{Ni}$ (Nickel,[2,3,7,8,12,13,17,18-octaethyl-21H,23H-porphinato(2-)- $\text{N}^{21},\text{N}^{22},\text{N}^{23},\text{N}^{24}]-(\text{SP}-4-1)-)$	24803-99-4	**	6.38 ± 0.03 (V)	PE	5476
$\text{C}_{44}\text{H}_{28}\text{N}_4\text{Ni}^+$	$\text{C}_{20}\text{H}_8\text{N}_4(\text{C}_6\text{H}_5)_4\text{Ni}$ (Nickel, [5,10,15,20-tetraphenyl-21H,23H-porphinato(2-)- $\text{N}^{21},\text{N}^{22},\text{N}^{23},\text{N}^{24}]-(\text{SP}-4-1)-)$	14172-92-0	**	6.29 ± 0.2	OTH	4962
			**	6.44 (V)	PE	4557
$\text{C}_{32}\text{H}_{16}\text{N}_8\text{Ni}^+$	$\text{C}_{32}\text{H}_{16}\text{N}_8\text{Ni}$ (Nickel, [29H,31H-phthalocyaninato(2-)- $\text{N}^{29},\text{N}^{30},\text{N}^{31},\text{N}^{32}]-(\text{SP}-4-1)-)$	14055-02-8	**	7.45 ± 0.10	EI	3829

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
CONi⁺	(CO) ₄ Ni	13463-39-3	3CO	12.6±0.2	EI	4618
C₂O₂Ni⁺	(CO) ₄ Ni	13463-39-3	2CO	10.6±0.2	EI	4618
C₃O₃Ni⁺	(CO) ₄ Ni	13463-39-3	CO	9.5±0.2	EI	4618
C₄O₄Ni⁺	(CO) ₄ Ni	13463-39-3	** **	8.21±0.03 8.8±0.2	PE EI	5225 4618
C₁₀H₁₄O₄Ni⁺	(CH ₃ COCHCOCH ₃) ₂ Ni (Nickel, bis(2,4-pentanedionato-O,O')-(SP-4-1)-)	3264-82-2	** ** **	7.40 (V) 7.41 (V) 7.61 (V)	PE PE PE	5100 4571 4384
C₁₀H₁₆O₄Ni⁺	((CH ₃ CO) ₂ CH ₂) ₂ Ni	XXXXX-XX-X	**	7.35 (V)	PE	5568
C₂₂H₄₀O₄Ni⁺	((CH ₃) ₃ CCO) ₂ CH ₂) ₂ Ni	XXXXX-XX-X	**	7.40 (V)	PE	5568
C₅H₅NONi⁺	C ₅ H ₅ NONi (Nickel, (η^1 -2,4-cyclopentadien-1-yl)nitrosyl-)	12071-73-7	** **	8.29 8.5	PE EI	4234 4015
C₆H₇NONi⁺	C ₆ H ₄ (CH ₃)NONi (Nickel, [(1,2,3,4,5- η)-1-methyl-2,4-cyclopentadien-1-yl]nitrosyl-)	32714-42-4	**	8.09	PE	4234
C₁₂H₁₈N₂O₂Ni⁺	C ₁₂ H ₁₈ O ₂ N ₂ Ni (Nickel, [[4,4'-(1,2-ethanediyl)dinitro]]bis[2-pentanato]](2)-N,N',O,O'-)	13878-48-3	**	6.80 (V)	PE	3822
C₁₆H₁₄N₂O₂Ni⁺	C ₁₆ H ₁₄ N ₂ O ₂ Ni (Nickel, [[2,2'-[1,2-ethanediyl]bis(nitrilomethylidyne)] bis[phenolato]](2)-N,N',O,O'-)	14167-20-5	**	7.57±0.09	EI	4668
C₂₁H₂₄N₂O₂Ni⁺	C ₂₁ H ₂₄ N ₂ O ₂ Ni (Nickel, [[2,2'-[1,7-heptanediyl]bis(nitrilomethylidyne)] bis[phenolato]](2)-N,N',O,O'-)	52358-03-9	**	7.69±0.09	EI	4213
C₂₀H₂₃N₃O₂Ni⁺	C ₂₀ H ₂₃ N ₃ O ₂ Ni (Nickel, [[2,2'-[iminobis(3,1-propanediyl)nitrilomethylidyne]] bis[phenolato]](2)-N,N',N'', O,O'-)	15391-40-9	**	7.41±0.08	EI	4213
C₂₀H₂₂N₂O₃Ni⁺	C ₂₀ H ₂₂ N ₂ O ₃ Ni (Nickel, [[2,2'-[oxybis(3,1-propanediyl)nitrilomethylidyne]] bis[phenolato]](2)-N,N',O,O', O''-)	52279-52-4	**	7.61±0.06	EI	4213
C₁₀H₈O₄F₆Ni⁺	(CF ₃ COCHCOCH ₃) ₂ Ni (Nickel, bis(1,1,1-trifluoro-2,4-pentanedionato-O,O')-)	14324-83-5	**	8.25 (V)	PE	4571

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₀H₂O₄F₁₂Ni⁺	(CF ₃ COCHCOCF ₃) ₂ Ni (Nickel, bis(1,1,1,5,5-hexafluoro-2,4-pentanedionato-O,O')-(SP-4-1)-)	14949-69-0	**	9.35 (V)	PE	4384
F₁₂P₁Ni⁺	(PF ₃) ₂ Ni	13859-65-9	** ** **	8.77 8.82±0.06 8.81	PE PE EI	5453 4187 5453
C₁₀H₁₁S₁Ni⁺	(CH ₃ CSCHCSCH ₃) ₂ Ni (Nickel, bis(2,4-pentanedithionato-S,S')-)	10170-79-3	** **	6.90 (V) 6.92 (V)	PE PE	5100 4571
C₁₂H₁₈N₂S₂Ni⁺	(CH ₃ C(=S)CH ₂ C(CH ₃) ₂ NCH ₂) ₂ Ni	41258-78-0	**	6.55 (V)	PE	5446
C₈H₁₈N₁S₁Ni⁺	(N(CH ₃) ₂ N=C(S)SCH ₃) ₂ Ni	53809-97-5	**	7.33	PE	5285
C₁₀H₁₁O₂S₂Ni⁺	(CH ₃ CSCHCOCH ₃) ₂ Ni (Nickel, bis(4-thioxo-2-pentanonato-O,S')-)	14239-84-0	** **	6.99 (V) 7.00 (V)	PE PE	4571 5100
C₂₀H₂₂N₂O₂SNi⁺	C ₂₀ H ₂₂ N ₂ O ₂ SNi (Nickel, [[2,2'-[thiobis(3,1-propanediyl)nitrilomethylidyne]]bis[phenolato]](2-)N,N',O,O' ,S)-)	52279-55-7	**	7.44±0.07	EI	4213
C₁₀H₈F₆S₁Ni⁺	(CF ₃ CSCHCSCH ₃) ₂ Ni (Nickel, bis(1,1,1-trifluoro-2,4-pentanedithionato-S,S')-)	21609-15-4	**	7.65 (V)	PE	4571
C₁₀H₈O₂F₆S₂Ni⁺	(CF ₃ CSCHCOCH ₃) ₂ Ni (Nickel, bis(1,1,1-trifluoro-4-thioxo-2-pentanonato-O,S')-)	15744-66-8	**	7.80 (V)	PE	4571
C₈H₂₀O₄P₂S₁Ni⁺	Ni(S ₂ P(OC ₂ H ₅) ₂) ₂	16743-23-0	**	7.24±0.05	PE	4636
Cl₂Ni⁺	NiCl ₂	7718-54-9	**	11.23 (V)	PE	5172
Cu⁺	Cu	7440-50-8	** ** ** ** ** ** **	7.72634±0.00002 S 7.72 10.44 10.55 10.69 10.98 7.71±0.05	PE PE PE PE PE EI	4011 4858 4858 4858 4858 4858 3745
	(¹ S ₀)					
	(³ D ₃)					
	(³ D ₂)					
	(³ D ₁)					
	(¹ D ₂)					
	Cu ₃ Cl ₃	11093-65-5		17.9±0.5	EI	4236
	Cu ₃ Cl ₄	11093-67-7		14.0±0.5	EI	4236
Cu₂⁺	Cu ₂	12190-70-4	** **	7.8±0.4 7.8	EI EI	5296 3775
	Cu ₃ Cl ₃	11093-65-5		18.1±0.5	EI	4236
C₁₆H₁₆N₁Cu⁺	C ₁₆ H ₁₆ N ₁ Cu (Copper, [N,N'-bis[(2-aminophenyl)methylene]-1,2-ethanediaminato(2-)N,N',N'',N'''])-	21177-98-0	**	7.15±0.11	EI	4668

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{36}\text{H}_{44}\text{N}_4\text{Cu}^+$	$((\text{C}_2\text{H}_5)_2\text{C}_4\text{NCH})_4\text{Cu}$ (Copper[2,3,7,8,12,13,17,18-octaethyl-21H,23H-porphinato(2-)- $\text{N}^{21},\text{N}^{22},\text{N}^{23},\text{N}^{24}]-(\text{SP}-4-1)-$)	14409-63-3	**	6.31 ± 0.03 (V)	PE	5476
$\text{C}_{44}\text{H}_{28}\text{N}_4\text{Cu}^+$	$\text{C}_{20}\text{H}_8\text{N}_4(\text{C}_6\text{H}_5)_4\text{Cu}$ (Copper, [5,10,15,20-tetraphenyl-21H,23H-porphinato(2-)- $\text{N}^{21},\text{N}^{22},\text{N}^{23},\text{N}^{24}]-(\text{SP}-4-1)-$)	14172-91-9	** **	6.24 ± 0.2 6.49 (V)	OTH PE	4962 4557
$\text{C}_{32}\text{H}_{16}\text{N}_8\text{Cu}^+$	$\text{C}_{32}\text{H}_{16}\text{N}_8\text{Cu}$ (Copper, [29H,31H-phthalocyaninato(2-)- $\text{N}^{29},\text{N}^{30},\text{N}^{31},\text{N}^{32}]-(\text{SP}-4-1)-$)	147-14-8	**	7.37 ± 0.10	EI	3829
$\text{C}_{10}\text{H}_{14}\text{O}_4\text{Cu}^+$	$(\text{CH}_3\text{C}(\text{O})=\text{CHCOCH}_3)_2\text{Cu}$ $(\text{CH}_3\text{COCHCOCH}_3)_2\text{Cu}$ (Copper, bis(2,4-pentanedionato-O,O')-(SP-4-1)-)	46369-53-3 13395-16-9	** **	8.35 (V) 8.20 (V)	PE PE	5100 4384
$\text{N}_2\text{O}_6\text{Cu}^+$	$(\text{NO}_3)_2\text{Cu}$	XXXXX-XX-X	**	10.47 ± 0.04 (V)	PE	4999
$\text{C}_{12}\text{H}_{18}\text{N}_2\text{O}_2\text{Cu}^+$	$\text{C}_{12}\text{H}_{18}\text{O}_2\text{N}_2\text{Cu}$ (Copper, [[4,4'-(1,2-ethanediyl)dinitrilo]bis[2-pentanonato]](2-)- $\text{N},\text{N}',\text{O},\text{O}']-$)	14263-53-7	**	7.00 (V)	PE	3822
$\text{C}_{16}\text{H}_{14}\text{N}_2\text{O}_2\text{Cu}^+$	$\text{C}_{16}\text{H}_{14}\text{N}_2\text{O}_2\text{Cu}$ (Copper, [[2,2'-[1,2-ethanediylbis(nitrilomethylidyne)] bis[phenolato]](2-)- $\text{N},\text{N}',\text{O},\text{O}']-$)	14167-15-8	**	7.69 ± 0.09	EI	4668
$\text{C}_{21}\text{H}_{24}\text{N}_2\text{O}_2\text{Cu}^+$	$\text{C}_{21}\text{H}_{24}\text{N}_2\text{O}_2\text{Cu}$ (Copper, [[2,2'-[1,7-heptanediylibis(nitrilomethylidyne)]bis[phenolato]](2-)- $\text{N},\text{N}',\text{O},\text{O}']-$)	52279-50-2	**	7.81 ± 0.07	EI	4213
$\text{C}_{20}\text{H}_{23}\text{N}_3\text{O}_2\text{Cu}^+$	$\text{C}_{20}\text{H}_{23}\text{N}_3\text{O}_2\text{Cu}$ (Copper, [[2,2'-[iminobis(3,1-propanediyl)nitrilomethylidyne)]bis[phenolato]](2-)- $\text{N},\text{N}',\text{N}''$, $\text{O},\text{O}']-$)	15391-22-7	**	7.54 ± 0.08	EI	4213
$\text{C}_{20}\text{H}_{22}\text{N}_2\text{O}_3\text{Cu}^+$	$\text{C}_{20}\text{H}_{22}\text{N}_2\text{O}_3\text{Cu}$ (Copper, [[2,2'-[oxybis(3,1-propanediyl)nitrilomethylidyne)] bis[phenolato]](2-)- $\text{N}^2,\text{N}^2,\text{O}^1,\text{O}^1]-$, (SP-4-2)-)	52279-53-5	**	7.75 ± 0.05	EI	4213
$\text{C}_{10}\text{H}_8\text{O}_4\text{F}_6\text{Cu}^+$	$(\text{CF}_3\text{C}(\text{O})=\text{CHCOCH}_3)_2\text{Cu}$	14324-82-4	**	8.95 (V)	PE	5100
$\text{C}_{10}\text{H}_2\text{O}_4\text{F}_{12}\text{Cu}^+$	$(\text{CF}_3\text{C}(\text{O})=\text{CHCOCF}_3)_2\text{Cu}$ (Copper, bis(1,1,1,5,5-hexafluoro-2,4-pentane- dionato-O,O')-(SP-4-1)-)	14781-45-4	** **	10.20 (V) 9.92 (V)	PE PE	5100 4384
$\text{C}_{12}\text{H}_{18}\text{N}_2\text{S}_2\text{Cu}^+$	$(\text{CH}_3\text{C}(=\text{S})\text{CH}_2\text{C}(\text{CH}_3)\text{NCH})_2\text{Cu}$	41192-46-5	**	6.35 (V)	PE	5446
$\text{C}_{10}\text{H}_{14}\text{O}_2\text{S}_2\text{Cu}^+$	$(\text{CH}_3\text{C}(\text{O})=\text{CHCSCH}_3)_2\text{Cu}$	27821-98-3	**	7.65 (V)	PE	5100

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₂₀H₂₂N₂O₂SCu⁺	C ₂₀ H ₂₂ N ₂ O ₂ SCu (Copper, [[2,2'-thiobis(3,1-propanediyl)nitrilomethylidyne)]bis[phenolato]](2-)-N,N,O,O'-]-, (SP-4-2)-)	52358-04-0	**	7.78±0.06	EI	4213
ClCu⁺	CuCl Cu ₃ Cl ₃	7758-89-6 11093-65-5	** 2CuCl CuCl+Cu+Cl 2Cu+2Cl	10.7±0.3 16.0±0.05 20.0±0.5 23.3±0.5	EI EI EI EI	5634 4236 4236 4236
ClCu₂⁺	Cu ₃ Cl ₃	11093-65-5	CuCl ₂	12.0±0.5	EI	3455
Cl₂Cu₂⁺	Gu ₂ Cl ₂ Cu ₃ Cl ₃ Cu ₄ Cl ₄	12258-96-7 11093-65-5 11093-67-7	** CuCl Cu+Cl	9.6±0.03 13.5±0.5 16.7±0.5 14.0±0.5	EI EI EI EI	5634 4236 4236 3455
Cl₂Cu₃⁺	Cu ₃ Cl ₃	11093-65-5	Cl	12.8±0.3	EI	5330
Cl₃Cu₃⁺	Cu ₃ Cl ₃ Cu ₄ Cl ₄	11093-65-5 11093-67-7	** ** ** ** CuCl	9.52 (V) 9.6±0.5 9.9±0.5 10.0±0.3 10.4±1.0	PE EI EI EI EI	5297 4236 3455 5330 4236
Cl₃Cu₄⁺	Cu ₄ Cl ₄	11093-67-7	Cl	12.2±0.5 12.4±0.5	EI EI	4236 3455
Cl₄Cu₄⁺	Cu ₄ Cl ₄	11093-67-7	** **	9.6±0.5 9.9±0.5	EI EI	4236 3455
Cl₄Cu₅⁺	Cu ₅ Cl ₅	11093-68-8	Cl	10.5-1.0 10.6±0.5	EI EI	4236 3455
Cl₅Cu₅⁺	Cu ₅ Cl ₅	11093-68-8	** **	9.2±1.0 9.7±0.5	EI EI	4236 3455
Zn⁺	Zn	7440-66-6	** **	9.394 9.57±0.07	S EI	5450 3745
C₂H₆Zn⁺	(CH ₃) ₂ Zn	544-97-8	**	9.4 (V)	PE	5300
C₄H₁₀Zn⁺	(C ₂ H ₅) ₂ Zn	557-20-0	**	8.6 (V)	PE	5300
C₃₆H₄₄N₄Zn⁺	((C ₂ H ₅) ₂ C ₄ NCH) ₄ Zn (Zinc,[2,3,7,8,12,13,17,18-octaethyl-21H,23H-porphinato(2-)-N ²¹ ,N ²² ,N ²³ ,N ²⁴]-(SP-4-1)-)	17632-18-7	**	6.29±0.03 (V)	PE	5476

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{14}\text{H}_{28}\text{N}_4\text{Zn}^+$	$\text{C}_{20}\text{H}_{18}\text{N}_4(\text{C}_6\text{H}_5)_4\text{Zn}$ (Zinc, [5,10,15,20-tetraphenyl-21H,23H-porphinato(2-)-N ²¹ ,N ²² ,N ²³ ,N ²⁴]-(SP-4-1)-)	14074-80-7	**	6.03 ± 0.2	OTH	4962
			**	6.42 (V)	PE	4557
$\text{C}_{32}\text{H}_{16}\text{N}_8\text{Zn}^+$	$\text{C}_{32}\text{H}_{16}\text{N}_8\text{Zn}$ (Zinc, [29H,31H-pthalocyaninato(2-)-N ²⁹ ,N ³⁰ ,N ³¹ ,N ³²]-(SP-4-1)-)	14320-04-8	**	7.37 ± 0.10	EI	3829
$\text{C}_{10}\text{H}_{14}\text{O}_4\text{Zn}^+$	$(\text{CH}_3\text{COCHCOCH}_3)_2\text{Zn}$ (Zinc, bis(2,4-pentanedionato-O,O')-(T-4)-)	14024-63-6	**	8.46 (V)	PE	4384
$\text{C}_{22}\text{H}_{40}\text{O}_4\text{Zn}^+$	$((\text{CH}_3)_3\text{CCO})_2\text{CH}_2\text{Zn}$	XXXXXX-XX-X	**	8.15 (V)	PE	5568
F_2Zn^+	ZnF ₂	7783-49-5	**	13.91 ± 0.03	PE	5433
$\text{C}_{10}\text{H}_2\text{O}_4\text{F}_{12}\text{Zn}^+$	$(\text{CF}_3\text{COCHCOCF}_3)_2\text{Zn}$ (Zinc, bis(1,1,1,5,5,5-hexafluoro-2,4-pentanedionato-O,O')-(T-4)-)	14949-70-3	**	10.25 (V)	PE	4384
$\text{C}_{12}\text{H}_{36}\text{N}_2\text{Si}_4\text{Zn}^+$	$(\text{N}(\text{Si}(\text{CH}_3)_3)_2)_2\text{Zn}$	3999-27-7	**	8.50 ± 0.05 (V)	PE	4725
Cl_2Zn^+	$(^2\Pi_g)$ ZnCl ₂ $(^2\Pi_{1/2g})$ $(^2\Pi_{3/2g})$ $(^2\Pi_u)$ $(^2\Pi_d)$ $(^2\Pi_{1/2u})$ $(^2\Pi_{3/2u})$ $(^2\Sigma_u)$ $(^2\Sigma_d)$ $(^2\Sigma_u^-)$ $(^2\Sigma_d^-)$ $(^2\Sigma_g^-)$ $(^2D_{5/2})$ $(^2D_{3/2})$	7646-85-7	** ** ** ** ** ** ** ** ** ** ** ** ** ** ** **	11.7 (V) 11.85 (V) 11.85 (V) 11.87 ± 0.05 (V) 12.3 (V) 12.39 ± 0.05 (V) 12.41 (V) 12.41 (V) 13.0 (V) 13.07 ± 0.05 (V) 13.09 (V) 14.0 (V) 14.10 ± 0.05 (V) 14.13 (V) 19.23 (V) 19.51 (V)	PE PE PE PE PE PE PE PE PE PE PE PE PE PE PE PE	3963 4232 4232 3833 3963 3833 4232 4232 3963 3833 4232 3963 3963 3833 4232 4232
Ga^+	Ga	7440-55-3	** ** **	6.0 ± 0.3 6.1 ± 0.4 6.1	EI EI EI	5067 4111 3472
	(CH ₃) ₃ Ga	1445-79-0	C ₆ H ₆ +CH ₃	13.24 ± 0.03	EI	3474
	(CH ₂ =CH) ₃ Ga	1188-13-2	C ₆ H ₆ +C ₂ H ₃	11.17 ± 0.05	EI	3474
	Ga ₂ S	12259-25-5	S	9.2 ± 0.3	EI	5229
Ga_2^+	Ga ₂ S	12259-25-5	S	11.5 ± 0.5	EI	5229
CH_3Ga^+	(CH ₃) ₃ Ga	1445-79-0	2CH ₃	13.65 ± 0.07	EI	3474
$\text{C}_2\text{H}_3\text{Ga}^+$	(CH ₂ =CH) ₃ Ga	1188-13-2	C ₆ H ₆	10.95 ± 0.05	EI	3474

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_2\text{H}_4\text{Ga}^+$	$(\text{CH}_2=\text{CH})_3\text{Ga}$	1188-13-2	$\text{C}_2\text{H}_3 + \text{C}_2\text{H}_2$	11.85 ± 0.05	EI	3474
$\text{C}_2\text{H}_6\text{Ga}^+$	$(\text{CH}_3)_3\text{Ga}$	1445-79-0	CH_3	10.16 ± 0.03	EI	3474
$\text{C}_3\text{H}_9\text{Ga}^+$	$(\text{CH}_3)_3\text{Ga}$	1445-79-0	** **	9.76 (V) 9.87 ± 0.02	PE EI	4398 3474
$\text{C}_4\text{H}_6\text{Ga}^+$	$(\text{CH}_2=\text{CH})_3\text{Ga}$	1188-13-2	C_2H_3	11.04 ± 0.08	EI	3474
$\text{C}_6\text{H}_9\text{Ga}^+$	$(\text{CH}_2=\text{CH})_3\text{Ga}$	1188-13-2	**	10.81 ± 0.1	EI	3474
$\text{C}_{12}\text{H}_{10}\text{Ga}^+$	$(\text{C}_6\text{H}_5)_3\text{Ga}$ (Gallium, triphenyl-)	1088-02-4	C_6H_5	8.63	PI	4055
$\text{C}_{18}\text{H}_{15}\text{Ga}^+$	$(\text{C}_6\text{H}_5)_3\text{Ga}$ (Gallium, triphenyl-)	1088-02-4	**	8.46 ± 0.03	PI	4055
CNGa^+	GaCN	51750-59-5	**	9 ± 1	EI	4205
FGa^+	GaF	13966-78-4	**	10.7 ± 0.6	EI	3613
F_2Ga^+	GaF_3	7783-51-9		15.1 ± 0.5	EI	3613
F_5Ga_2^+	Ga_2F_6	38586-87-7		15.6 ± 0.5	EI	3613
$\text{C}_{15}\text{H}_3\text{O}_6\text{F}_{18}\text{Ga}^+$	$(\text{CF}_3\text{COCHCOCF}_3)_3\text{Ga}$ (Gallium, tris(1,1,1,5,5-hexafluoro-2,4-pentanedionato-O,O')-, (OC-6-11)-)	19648-92-1	**	10.19 ± 0.07 (V)	PE	3682
SGa^+	Ga_2S	12259-25-5	Ga	12 ± 0.5	EI	5229
SGa_2^+	Ga_2S	12259-25-5	**	7.7 ± 0.3	EI	5229
Cl_3Ga^+	GaCl_3	13450-90-3	** ** **	11.52 11.96 (V) 11.96 (V)	PE PE PE	4215 4398 4256
Cl_6Ga_2^+	$(\text{GaCl}_3)_2$	15654-66-7	** **	11.81 (V) 11.81 (V)	PE PE	4559 4256
Ge^+	Ge $(^2\text{P}_{1/2})$	7440-56-4	** ** ** **	7.899 8.1186 7.8 ± 0.5 8.0 ± 0.3	S S EI EI	5495 5495 4200 3610
	GeF_4	14929-46-5		29.4 ± 0.2	EI	5154
	GeS	12024-10-1		13.51 ± 0.03	PI	4936

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
Ge₂⁺	Ge ₂	12596-05-3	**	7.8	EI	3775
H₄Ge⁺	GeH ₄	7782-65-2	** **	11.34 12.0 (V)	PE PE	3716 3508
C₂H₈Ge⁺	(CH ₃) ₂ GeH ₂ C ₂ H ₅ GeH ₃	1449-64-5 1747-99-5	** **	10.74 (V) 10.4 (V)	PE PE	5261 4985
C₃H₉Ge⁺	(CH ₃) ₄ Ge (CH ₃) ₃ CGe(CH ₃) ₃ ((CH ₃) ₃ Ge) ₂ (CH ₃) ₃ SiGe(CH ₃) ₃ C ₆ H ₅ SGe(CH ₃) ₃ (Germane, trimethyl(phenylthio)-) (CH ₃) ₃ GeCl C ₅ H ₅ (CO) ₃ CrGe(CH ₃) ₃ (Tricarbonyl(η^5 -2,4-cyclopentadien-1-yl)(trimethylgermyl)chromium) C ₅ H ₅ (CO) ₃ MoGe(CH ₃) ₃ (Tricarbonyl(η^5 -2,4-cyclopentadien-1-yl)(trimethylgermyl)molybdenum) ((CH ₃) ₃ Ge)(CH ₃) ₃ Sn C ₅ H ₅ (CO) ₃ WGe(CH ₃) ₃ (Tricarbonyl(η^5 -2,4-cyclopentadien-1-yl)(trimethylgermyl)tungsten)	865-52-1 1184-91-4 993-52-2 31608-80-7 4848-62-8 1529-47-1 34962-34-0 33306-91-1 16393-89-8 33306-93-3	CH ₃ CH ₃ (CH ₃) ₃ C (CH ₃) ₃ Ge (CH ₃) ₃ Si Cl (CH ₃) ₃ Sn (CH ₃) ₃ Sn	10.05±0.14 10.07±0.07 9.91±0.22 9.96±0.16 9.99±0.14 9.83±0.1 11.75±0.04 9.06±0.1 9.63±0.14 10.01±0.18 9.84±0.1	EI EI EI EI EI EI EI EI EI EI EI EI EI EI EI EI EI	3548 4126 3548 3548 3548 4198 3939 3495 3495 3548 3495 3495
C₃H₁₆Ge⁺	(C ₂ H ₅) ₃ GeH	1188-14-3	**	9.6 (V)	PE	4985
C₄H₇Ge⁺	(CH ₃) ₃ GeC≡CH	2290-58-6	CH ₃	10.56±0.07	EI	4126
C₄H₁₂Ge⁺	(CH ₃) ₄ Ge (C ₂ H ₅) ₂ GeH ₂	865-52-1 1631-46-5	** **	9.33±0.04 9.38±0.1 9.29±0.14 9.56±0.06 9.8 (V)	PE PE EI EI PE	3880 3677 3548 4126 4985
C₅H₅Ge⁺	(CH ₃) ₂ Ge(C≡CH) ₂	28056-58-8	CH ₃	10.94±0.04	EI	4126
C₅H₈Ge⁺	C ₅ H ₅ (GeH ₃) (Germane, 2,4-cyclopentadien-1-yl-)	35682-28-1	**	8.5 (V)	PE	4373
C₅H₁₀Ge⁺	(CH ₃) ₃ GeC≡CH	2290-58-6	**	9.77±0.04	EI	4126
C₆H₃Ge⁺	CH ₃ Ge(C≡CH) ₃	28056-56-6	CH ₃	10.74±0.05	EI	4126
C₆H₈Ge⁺	(CH ₃) ₂ Ge(C≡CH) ₂	28056-58-8	**	10.57±0.07	EI	4126
C₆H₁₂Ge⁺	C ₄ H ₆ Ge(CH ₃) ₂ (Germacyclopent-3-ene,1,1-dimethyl-)	1731-10-8	**	9.0 (V)	PE	5550

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_6\text{H}_{11}\text{Ge}^+$	$\text{CH}_2=\text{CHCH}_2\text{Ge}(\text{CH}_3)_3$	762-66-3	**	8.85 (V)	PE	4172
$\text{C}_7\text{H}_{18}\text{Ge}^+$	$(\text{CH}_3)_3\text{CGe}(\text{CH}_3)_3$	1184-91-4	**	8.98 ± 0.12	EI	3548
$\text{C}_8\text{H}_7\text{Ge}^+$	$\text{Ge}(\text{C}\equiv\text{CH})_4$	4531-35-5	**	11.04 ± 0.05	EI	4126
$\text{C}_8\text{H}_{18}\text{Ge}^+$	$\text{CH}_2=\text{CHGe}(\text{C}_2\text{H}_5)_3$	6207-41-6	**	9.2 (V)	PE	3850
$\text{C}_8\text{H}_{20}\text{Ge}^+$	$(\text{C}_2\text{H}_5)_4\text{Ge}$	597-63-7	** **	9.3 (V) 9.4 (V)	PE PE	3850 4985
$\text{C}_9\text{H}_{14}\text{Ge}^+$	$\text{C}_6\text{H}_5(\text{CH}_3)_3\text{Ge}$ (Germane, trimethylphenyl-)	1626-00-2	** ** **	8.98 ± 0.05 9.00 (V) ~ 8.75	PE PE CTS	4589 4280 3922
$\text{C}_9\text{H}_{20}\text{Ge}^+$	$\text{CH}_2=\text{CHCH}_2\text{Ge}(\text{C}_2\text{H}_5)_3$	1793-90-4	**	8.8 (V)	PE	3850
$\text{C}_{10}\text{H}_{14}\text{Ge}^+$	$\text{C}_6\text{H}_5\text{Ge}(\text{CH}_3)_2$ (1 <i>H</i> -2-Benzogermole, 2,3-dihydro-2,2-dimethyl-)	27490-21-7	**	8.39	CTS	3546
$\text{C}_{10}\text{H}_{16}\text{Ge}^+$	$\text{C}_6\text{H}_5\text{CH}_2(\text{CH}_3)_3\text{Ge}$ (Germane, trimethyl(phenylmethyl)-)	2848-62-6	** ** ** ** **	8.25 (V) 8.36 ± 0.05 8.40 (V) 8.19 8.26	PE PE PE CTS CTS	4172 4589 4280 3922 3546
$\text{C}_{12}\text{H}_{18}\text{Ge}^+$	$\text{C}_6\text{H}_5\text{Ge}(\text{CH}_3)_3$ (Germane, 1-indanyltrimethyl-)	27490-24-0	**	8.02	CTS	3546
$\text{C}_{13}\text{H}_{15}\text{Ge}^+$	$\text{C}_{10}\text{H}_7\text{Ge}(\text{CH}_3)_3$ (Germane, trimethyl-1-naphthalenyl-)	XXXXX-XX-X	**	8.00	CTS	3922
$\text{C}_{13}\text{H}_{22}\text{Ge}^+$	$\text{C}_6\text{H}_5\text{CH}_2\text{Ge}(\text{C}_2\text{H}_5)_3$ (Germane, triethyl(phenylmethyl)-)	2945-41-7	**	8.1 (V)	PE	4172
$\text{C}_{14}\text{H}_{18}\text{Ge}^+$	$\text{C}_{10}\text{H}_7\text{CH}_2\text{Ge}(\text{CH}_3)_3$ (Germane, trimethyl(1-naphthalenylmethyl)-)	51220-35-0	**	7.78	CTS	3922
$\text{C}_{16}\text{H}_{16}\text{Ge}^+$	$(\text{C}_6\text{H}_5)_3\text{GeH}$ (Germane, triphenyl-)	2816-43-5	**	9.15 ± 0.05 (V)	PE	4620
$\text{C}_{20}\text{H}_{14}\text{Ge}^+$	$((\text{CH}_3)_3\text{CCH}_2)_4\text{Ge}$	50654-36-9	**	9.01 ± 0.1 (V)	PE	4242
$\text{C}_6\text{H}_{18}\text{Ge}_2^+$	$((\text{CH}_3)_3\text{Ge})_2$	993-52-2	**	8.18 ± 0.11	EI	3548

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
N₂Ge⁺	Ge ₂ N	53262-45-6	**	8.4±0.5	EI	4200
H₃N₃Ge⁺	GeH ₃ N ₃	21138-22-7	**	10.01±0.02 (V)	PE	3670
H₉N₂Ge⁺	(GeH ₃) ₃ N	22856-27-5	**	9.2±0.1 (V)	PE	3661
C₉H₁₁N₂Ge⁺	C ₆ H ₅ N=NGe(CH ₃) ₃ (Diazene, phenyl(trimethylgermyl)-)	34472-62-3	**	7.65±0.2 (V)	PE	4581
C₈H₂₄N₄Ge⁺	(N(CH ₃) ₂) ₄ Ge	7344-40-3	**	8.48 (V)	PE	4588
O²Ge⁺ (²Σ)	GeO	20619-16-3	**	11.25±0.01 (V)	PE	4760
			**	11.25±0.01 (V)	PE	4883
			**	11.25 (V)	PE	4967
			**	11.40±0.01 (V)	PE	4760
			**	15.17±0.01	PE	4760
			**	11.0±0.3	EI	3610
O₂Ge⁺	Ge ₂ O ₂	XXXXX-XX-X	**	10.76±0.02 (V)	PE	4760
H₆O²Ge⁺	(GeH ₃) ₂ O	14939-17-4	**	10.40 (V)	PE	3656
C₅H₁₂O²Ge⁺	(CH ₃) ₃ (COCH ₃)Ge	53520-45-9	**	8.5 (V)	PE	4139
CH₃N²NOGe⁺	GeH ₃ NCO	6928-42-3	**	10.76±0.02 (V)	PE	3670
C₈H₁₃NOGe⁺	C ₅ H ₄ N(O)Ge(CH ₃) ₃ (Pyridine, 4-(trimethylgermyl)-, 1-oxide)	28867-08-5	**	8.12 (V)	PE	4222
FGe⁺	GeF ₄	14929-46-5		23.4±0.4	EI	5154
F₂Ge⁺	GeF ₂	13940-63-1	**	12.9±0.3	EI	3570
	GeF ₄	14929-46-5		20.7±0.3	EI	5154
F₃Ge⁺	GeF ₄	14929-46-5	F	15.7±0.2	EI	5154
F₄Ge⁺	GeF ₄	7783-58-6	**	16.06±0.04 (V)	PE	3880
F₄Ge₂⁺	Ge ₂ F ₄	12332-08-0	**	13.1±0.3	EI	3570
H₃FGe⁺	GeH ₃ F	13537-30-9	**	12.3±0.1 (V)	PE	3510
H₂F₂Ge⁺	GeH ₂ F ₂	14986-65-3	**	13.0±0.1 (V)	PE	3510

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₂H₆F₂Ge⁺	(CH ₃) ₂ GeF ₂	811-70-1	**	10.45 (V)	PE	5261
OF₂Ge⁺	GeOF ₂	XXXXX-XX-X	**	12.3±0.3	EI	3570
C₆H₁₈SiGe⁺	(CH ₃) ₃ SiGe(CH ₃) ₃	31608-80-7	**	8.31±0.10	EI	3548
C₁₁H₃₈Si₄Ge⁺	(CH(Si(CH ₃) ₃) ₂) ₂ Ge	60111-69-5	**	7.75±0.05 (V)	PE	4725
NSiGe⁺	GeSiN	53262-44-5	**	8.6±0.5	EI	4200
C₁₁H₃₆N₂Si₂Ge⁺	C ₁₁ H ₃₆ N ₂ Si ₂ Ge (N(Si(CH ₃) ₃)(tert-C ₄ H ₉)) ₂ Ge	55147-81-4 XXXXX-XX-X	** **	7.24±0.05 (V) 7.26 (V)	PE PE	4725 4157
C₁₂H₃₆N₂Si₄Ge⁺	(N(Si(CH ₃) ₃) ₂) ₂ Ge	55290-25-0	** **	7.71±0.05 (V) 7.72 (V)	PE PE	4725 4157
H₅PGe⁺	GeH ₃ PH ₂	13573-06-3	**	9.7±0.1 (V)	PE	3661
H₉PGe₃⁺	(GeH ₃) ₃ P	15587-38-9	**	9.0±0.1 (V)	PE	3661
SGe⁺ (² Π) (² Π) (² Σ ⁺) (² Π) (² Π _{1/2,3/2}) (² Σ)	GeS	12025-32-0	** ** ** ** ** **	9.98±0.02 10.18±0.03 (V) 10.35±0.08 (V) 10.25–10.28 (V) 10.36 (V) 10.39 (V)	PI PI PI PE PE EI	4936 4936 4936 4550 4967 4550
H₄SGe⁺	GeH ₃ SH	21847-06-3	**	9.69 (V)	PE	3656
H₆SGe₂⁺	(GeH ₃) ₂ S	18852-54-5	**	9.25 (V)	PE	3656
C₄H₁₂SGe⁺	(CH ₃) ₃ SCH ₃ Ge	3860-84-2	**	8.50±0.05 (V)	PE	4153
C₈H₁₁SGe⁺	C ₆ H ₅ S(CH ₃) ₃ Ge (Germane, trimethyl(phenylthio)–)	4848-62-8	CH ₃	9.95±0.1	EI	4198
C₉H₁₄SGe⁺	C ₆ H ₅ S(CH ₃) ₃ Ge (Germane, trimethyl(phenylthio)–)	4848-62-8	** **	8.52±0.05 8.08±0.1	PE EI	4589 4198
C₁₀H₁₆SGe⁺	C ₆ H ₅ (SCH ₃)(CH ₃) ₃ Ge (Germane, trimethyl[4-(methylthio)phenyl]–)	59163-56-3	**	7.90±0.05 (V)	PE	4627
C₁₃H₁₁SGe⁺	C ₁₂ H ₉ SGe(CH ₃) ₂ (10H-Phenothiagermanin, 10,10-dimethyl–)	63447-23-4	CH ₃	8.7±0.1	EI	4664

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{11}\text{H}_{14}\text{SGe}^+$	$\text{C}_{12}\text{H}_9\text{SGe}(\text{CH}_3)_2$ (10H-Phenothiagermanin, 10,10-dimethyl-)	63447-23-4	**	8.0 ± 0.1	EI	4664
$\text{C}_6\text{H}_{18}\text{SGe}^+$	$(\text{CH}_3)_3\text{Ge}_2\text{S}$	6199-00-4	** **	8.40 ± 0.05 (V) 8.60 ± 0.1	PE EI	4153 4198
CH_3NSGe^+	GeH_3NCS	16475-45-9	**	9.14 ± 0.02 (V)	PE	3670
Cl_3Ge^+	GeCl_4 CH_3GeCl_3	10038-98-9 993-10-2	Cl CH_3	12.12 ± 0.04 12.22 ± 0.05	EI EI	3939 3939
Cl_4Ge^+	GeCl_4	10038-98-9	**	11.68 ± 0.05	EI	3939
H_3ClGe^+	GeH_3Cl	13637-65-5	** **	11.30 ± 0.02 (V) 11.34 ± 0.05 (V)	PE PE	3510 3502
$\text{H}_2\text{Cl}_2\text{Ge}^+$	GeH_2Cl_2	15230-48-5	**	11.42 ± 0.02 (V)	PE	3510
$\text{C}_2\text{H}_6\text{ClGe}^+$	$(\text{CH}_3)_3\text{GeCl}$ $(\text{CH}_3)_2\text{GeCl}_2$	1529-47-1 1529-48-2	CH_3 Cl	10.44 ± 0.04 11.56 ± 0.04	EI EI	3939 3939
$\text{C}_3\text{H}_9\text{ClGe}^+$	$(\text{CH}_3)_3\text{GeCl}$	1529-47-1	** **	10.35 (V) 9.62 ± 0.04	PE EI	4566 3939
$\text{C}_9\text{H}_{13}\text{ClGe}^+$	$\text{C}_6\text{H}_5(\text{CH}_3)_3\text{GeCl}$ (Germane, (4-chlorophenyl)trimethyl-)	56866-67-2	**	8.84 (V)	PE	4438
$\text{CH}_3\text{Cl}_2\text{Ge}^+$	$(\text{CH}_3)_2\text{GeCl}_2$ CH_3GeCl_3	1529-48-2 993-10-2	CH_3 Cl	11.08 ± 0.05 11.78 ± 0.05	EI EI	3939 3939
$\text{C}_2\text{H}_6\text{Cl}_2\text{Ge}^+$	$(\text{CH}_3)_2\text{GeCl}_2$	1529-48-2	** **	10.65 (V) 10.18 ± 0.05	PE EI	5261 3939
$\text{CH}_3\text{Cl}_3\text{Ge}^+$	CH_3GeCl_3	993-10-2	**	11.11 ± 0.04	EI	3939
$\text{C}_8\text{H}_{14}\text{CrGe}^+$	$\text{C}_5\text{H}_5(\text{CO})_3\text{CrGe}(\text{CH}_3)_3$ (Tricarbonyl(η^5 -2,4-cyclopentadien-1-yl)(trimethylgermyl)chromium)	34962-34-0	3CO	10.57 ± 0.24	EI	3495
$\text{C}_9\text{H}_{14}\text{OCrGe}^+$	$\text{C}_5\text{H}_5(\text{CO})_3\text{CrGe}(\text{CH}_3)_3$ (Tricarbonyl(η^5 -2,4-cyclopentadien-1-yl)(trimethylgermyl)chromium)	34962-34-0	2CO	9.53 ± 0.15	EI	3495
$\text{C}_{10}\text{H}_{14}\text{O}_2\text{CrGe}^+$	$\text{C}_5\text{H}_5(\text{CO})_3\text{CrGe}(\text{CH}_3)_3$ (Tricarbonyl(η^5 -2,4-cyclopentadien-1-yl)(trimethylgermyl)chromium)	34962-34-0	CO	9.13 ± 0.1	EI	3495
$\text{C}_{11}\text{H}_{14}\text{O}_3\text{CrGe}^+$	$\text{C}_5\text{H}_5(\text{CO})_3\text{CrGe}(\text{CH}_3)_3$ (Tricarbonyl(η^5 -2,4-cyclopentadien-1-yl)(trimethylgermyl)chromium)	34962-34-0	**	7.79 ± 0.1	EI	3495

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_5\text{H}_3\text{O}_3\text{MnGe}^+$ $(\text{GeH}_3)(\text{CO})_3\text{Mn}$		25069-08-3	**	8.90 ± 0.02 (V)	PE	3827
$\text{C}_4\text{H}_3\text{O}_4\text{CoGe}^+$ $(\text{GeH}_3)(\text{CO})_3\text{Co}$		28360-37-4	**	8.80 ± 0.02 (V)	PE	3827
CuGe^+ GeCu		12394-89-7	**	7.5	EI	3775
As^+ As_2 AsF_3 AsCl_3		7440-38-2 7784-35-2 7784-34-1	** 3F 3Cl	>10.0 27.0 ± 0.4 20.4 ± 0.4	EI	3475 5016 5016
As_2^+ As_2		23878-46-8	** **	10.1 ± 0.2 11.0 ± 0.5	S EI	3567 3555
As_4^+ As_4		12187-08-5	**	9.9 ± 0.2	EI	3555
H_3As^+ AsH_3		7784-42-1	** **	9.89 10.58 ± 0.05 (V)	PE PE	3719 5419
$\text{C}_2\text{H}_7\text{As}^+$ $(\text{CH}_3)_2\text{AsH}$		593-57-7	** **	8.58 9.14 (V)	PE PE	3589 4185
$\text{C}_3\text{H}_9\text{As}^+$ $(\text{CH}_3)_3\text{As}$		593-88-4	** **	8.65 (V) 8.65 (V)	PE PE	4226 5368
$\text{C}_4\text{H}_9\text{As}^+$ $(\text{CH}_2=\text{CH})(\text{CH}_3)_2\text{As}$		13652-14-7	**	8.68 (V)	PE	5122
$\text{C}_4\text{H}_{11}\text{As}^+$ $(\text{CH}_3)_3\text{As}=\text{CH}_2$		19415-86-2	**	6.72 (V)	PE	5368
$\text{C}_5\text{H}_5\text{As}^+$ $\text{C}_5\text{H}_5\text{As}$ (Arsenin)		289-31-6	**	8.8 (V)	PE	3832
$\text{C}_5\text{H}_{11}\text{As}^+$ $(\text{CH}_2=\text{CHCH}_2)(\text{CH}_3)_2\text{As}$		691-35-0	**	8.57 (V)	PE	5122
$\text{C}_6\text{H}_5\text{As}^+$ $(\text{C}_6\text{H}_5)_3\text{As}$ (Arsine, triphenyl-)		603-32-7		8.2 ± 0.1	PI	4325
$\text{C}_8\text{H}_{11}\text{As}^+$ $(\text{C}_6\text{H}_5)(\text{CH}_3)_2\text{As}$ (Arsine,dimethylphenyl-)		696-26-4	**	8.67 (V)	PE	5122
$\text{C}_9\text{H}_{13}\text{As}^+$ $(\text{C}_6\text{H}_5\text{CH}_2)(\text{CH}_3)_2\text{As}$ (Arsine,dimethyl(phenylmethyl))		36678-76-9	**	8.45 (V)	PE	5122
$\text{C}_{12}\text{H}_{10}\text{As}^+$ $(\text{C}_6\text{H}_5)_3\text{As}$ (Arsine, triphenyl-)		603-32-7		9.35 ± 0.1	PI	4325

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{12}\text{H}_{11}\text{As}^+$	$(\text{C}_6\text{H}_5)_2\text{AsH}$ (Arsine, diphenyl-)	829-83-4	**	7.87 ± 0.01	PE	4154
$\text{C}_{12}\text{H}_{13}\text{As}^+$	$\text{C}_6\text{H}_5\text{C}_6\text{H}_2\text{As}(\text{CH}_3)_2$ (1H-Arsole, 2,5-dimethyl-1-phenyl-)	20527-10-0	**	8.0 (V)	PE	4090
$\text{C}_{18}\text{H}_{13}\text{As}^+$	$(\text{C}_6\text{H}_5)_3\text{As}$ (Arsine, triphenyl-)	603-32-7	**	7.32 ± 0.05	PI	4325
			**	7.60 ± 0.01	PE	4154
			**	8.03 ± 0.05 (V)	PE	4368
			**	8.11 (V)	PE	5139
$\text{C}_{19}\text{H}_{13}\text{As}^+$	$\text{C}_{13}\text{H}_8\text{AsC}_6\text{H}_5$ (Acridarsine, 10-phenyl-)	28660-45-9	**	7.05 (V)	PE	5630
$\text{C}_4\text{H}_{12}\text{As}_2^+$	$((\text{CH}_3)_2\text{As})_2\text{-trans}$ $((\text{CH}_3)_2\text{As})_2\text{-gauche}$	471-35-2	**	7.91 (V)	PE	4185
			**	8.85 (V)	PE	4185
			**	8.85 (V)	PE	4185
$\text{B}_2\text{C}_6\text{H}_{18}\text{N}_3\text{As}^+$	$\text{N}_3\text{B}_2(\text{CH}_3)_3\text{As}(\text{CH}_3)_2$ (1,2,4,3,5-Triazadiborolidine, 4-(dimethylarsino)-1,2,3,5-tetramethyl-)	57877-84-6	**	7.5 (V)	PE	4526
O_6As_4^+	As_4O_6	12505-67-8	**	10.01 ± 0.05 (V)	PE	4639
			**	10.05 (V)	PE	4704
			**	10.05 (V)	PE	5343
CH_2OAs^+	$\text{As}(\text{OCH}_3)_3$	6596-95-8		13.88	EI	4339
CH_3OAs^+	$\text{As}(\text{OCH}_3)_3$	6596-95-8		10.63	EI	4339
CH_4OAs^+	$\text{As}(\text{OCH}_3)_3$	6596-95-8		13.48	EI	4339
$\text{C}_2\text{H}_4\text{OAs}^+$	$\text{As}(\text{OC}_2\text{H}_5)_3$	3141-12-6		12.30	EI	4339
$\text{C}_2\text{H}_5\text{OAs}^+$	$\text{As}(\text{OC}_2\text{H}_5)_3$	3141-12-6		10.80	EI	4339
$\text{C}_2\text{H}_6\text{OAs}^+$	$\text{As}(\text{OC}_2\text{H}_5)_3$	3141-12-6		12.30	EI	4339
$\text{C}_3\text{H}_9\text{OAs}^+$	$(\text{CH}_3)_3\text{As=O}$	4964-14-1	**	9.08 (V)	PE	5368
$\text{C}_2\text{H}_5\text{O}_2\text{As}^+$	$\text{As}(\text{OCH}_3)_3$	6596-95-8	CH_3OH	8.98	EI	4339
$\text{C}_2\text{H}_6\text{O}_2\text{As}^+$	$\text{As}(\text{OCH}_3)_3$	6596-95-8	CH_3O	10.03	EI	4339
$\text{C}_2\text{H}_7\text{O}_2\text{As}^+$	$\text{As}(\text{OCH}_3)_3$	6596-95-8	CH_2O	8.80	EI	4339

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁H₉O₂As⁺	As(OC ₂ H ₅) ₃	3141-12-6	C ₂ H ₅ OH	8.60	EI	4339
C₁H₁₀O₂As⁺	As(OC ₂ H ₅) ₃	3141-12-6	C ₂ H ₅ O	10.03	EI	4339
C₁H₁₁O₂As⁺	As(OC ₂ H ₅) ₃	3141-12-6	C ₂ H ₄ O	8.52	EI	4339
C₃H₉O₃As⁺	As(OCH ₃) ₃	6596-95-8	** **	9.73 (V) 7.93	PE EI	4705 4339
FAs⁺	AsF ₃ AsF ₅	7784-35-2 7784-36-3		19.6±0.1 23.8±0.2	EI EI	5016 5016
F₂As⁺	AsF ₃ AsF ₅	7784-35-2 7784-36-3	F ⁻	12.80±0.1 15.8±0.2	EI EI	5016 5016
F₃As⁺	AsF ₃ AsF ₅	7784-35-2 7784-36-3	** ** F ₂ ⁻	12.3±0.05 12.84±0.05 14.0±0.1	EI EI EI	5016 3578 5016
F₄As⁺	AsF ₅	7784-36-3	F ⁻	13.8±0.2	EI	5016
C₄F₁₂As₂⁺	((CF ₃) ₂ As) ₂ (Trans conformer)	360-56-5	**	10.39 (V)	PE	4185
CH₂F₃As⁺	H ₂ AsCF ₃	XXXXX-XX-X	**	14.0±0.05 (V)	PE	5419
C₆H₇F₆As⁺	cis-(CH ₃) ₂ AsC(CF ₃)=C(CF ₃)H trans-(CH ₃) ₂ AsC(CF ₃)=C(CF ₃)H	4648-64-0 4648-63-9	** **	8.61 8.71	PE PE	3589 3589
C₈H₁₁F₆As⁺	(C ₂ H ₅) ₂ AsC(CF ₃)=C(CF ₃)H	XXXXX-XX-X	**	8.44	PE	3589
H₉Si₃As⁺	(SiH ₃) ₃ As	15110-34-6	**	9.3±0.1 (V)	PE	3661
C₇H₁₉SiAs⁺	(CH ₃) ₃ As=CHSi(CH ₃) ₃	3607-04-3	**	6.56 (V)	PE	5368
C₁₀H₂₇Si₂As⁺	(CH ₃) ₃ As=C(Si(CH ₃) ₃) ₂	58972-45-5	**	6.66 (V)	PE	5368
H₂F₃SiAs⁺	F ₃ SiAsH ₂	53098-12-7	**	10.90±0.05 (V)	PE	5419
C₂H₆F₃SiAs⁺	F ₃ SiAs(CH ₃) ₂	60387-29-3	**	9.4±0.05 (V)	PE	5419
PAs⁺	AsP	12255-33-3	** **	10.5±0.6 11.2±0.5	EI EI	4120 3555

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
P₃As⁺	AsP ₃	12511-95-4	**	10.3±0.3	EI	3555
P₂As₂⁺	As ₂ P ₂	12512-03-7	**	10.3±0.3	EI	3555
PA₃⁺	As ₃ P	12512-11-7	**	10.0±0.3	EI	3555
S₃As₃⁺	As ₄ S ₄	12279-90-2		9.0±0.7	EI	3475
S₃As₄⁺	As ₄ S ₃	12512-13-9	**	9.01 (V)	PE	4704
S₄As₄⁺	As ₄ S ₄	12279-90-2	**	9.0±0.7	EI	3475
ClAs⁺	AsCl ₃	7784-34-1		17.7±0.2	EI	5016
Cl₂As⁺	AsCl ₃	7784-34-1	Cl ⁻	12.4±0.2	EI	5016
Cl₃As⁺	AsCl ₃	7784-34-1	**	10.55±0.025	PE	3626
			**	10.90 (V)	PE	5473
			**	10.57±0.03	EI	3626
			**	11.6±0.05	EI	5016
C₂H₆SiCl₃As⁺	Cl ₃ SiAs(CH ₃) ₂	XXXXX-XX-X	**	9.20±0.05 (V)	PE	5419
C₂₄H₂₂MnAs⁺	(CH ₃ C ₅ H ₄)(CO) ₂ ((C ₆ H ₅) ₃ As)Mn (Manganese,dicarbonyl)[(1,2,3,4,5- η)-1-methyl-2,4-cyclopentadien-1-yl](triphenylarsine)-)	XXXXX-XX-X	2CO	8.44±0.03	EI	5576
	C ₂₆ H ₂₂ OSMnAs ⁺ (Manganese,(carbonothioyl)carbonyl)[(1,2,3,4,5- η)-1-methyl-2,4-cyclopentadien-1-yl](triphenylarsine)-)	XXXXX-XX-X	CO+CS	9.01±0.02	EI	5576
C₂₅H₂₂OMnAs⁺	(CH ₃ C ₅ H ₄)(CO) ₂ ((C ₆ H ₅) ₃ As)Mn (Manganese,dicarbonyl)[(1,2,3,4,5- η)-1-methyl-2,4-cyclopentadien-1-yl](triphenylarsine)-)	XXXXX-XX-X	CO	8.53±0.04	EI	5576
C₂₆H₂₂O₂MnAs⁺	(CH ₃ C ₅ H ₄)(CO) ₂ ((C ₆ H ₅) ₃ As)Mn (Manganese,dicarbonyl)[(1,2,3,4,5- η)-1-methyl-2,4-cyclopentadien-1-yl](triphenylarsine)-)	XXXXX-XX-X	**	6.38±0.03	EI	5576
C₂₅H₂₂SMnAs⁺	C ₂₆ H ₂₂ OSMnAs (Manganese,(carbonothioyl)carbonyl)[(1,2,3,4,5- η)-1-methyl-2,4-cyclopentadien-1-yl](triphenylarsine)-)	XXXXX-XX-X	CO	7.57±0.02	EI	5576
C₂₆H₂₂OSMnAs⁺	C ₂₆ H ₂₂ OSMnAs (Manganese,(carbonothioyl)carbonyl)[(1,2,3,4,5- η)-1-methyl-2,4-cyclopentadien-1-yl](triphenylarsine)-)	XXXXX-XX-X	**	6.71±0.02	EI	5576

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{22}\text{H}_{15}\text{O}_1\text{FeAs}^+$	$\text{As}(\text{C}_6\text{H}_5)_3(\text{CO})_3\text{Fe}$ (Iron,tetracarbonyl(triphenylarsine)-)	14375-84-9	**	7.50 (V)	PE	5559
Se^+						
	H_2Se	7783-07-5		12.6 ± 0.1	EI	3633
	CSe_2	506-80-9		13.38 ± 0.02	PI	4936
	$(^2\Sigma_u^+)$		CSe	13.4543	PI	5000
Se_2^+						
	$(^2\Pi_{g,1/2}) \text{Se}_2$	12185-17-0	**	8.70 ± 0.05	PE	4662
	$(^2\Pi_{g,3/2})$		**	9.13 (V)	PE	4662
	$(^1\Pi_\mu)$		**	10.68 (V)	PE	4662
	$(^1\Sigma_g^-)$		**	11.27 (V)	PE	4662
			**	12.27 (V)	PE	4662
			**	12.81 (V)	PE	4662
	$(^2\Sigma_g^-)$		**	13.31 (V)	PE	4662
			**	14.00 (V)	PE	4662
	CSe_2	506-80-9		15.21 ± 0.02	PI	4936
Se_3^+						
	Se_3	12597-28-3	**	7.83 ± 0.02	PE	4662
Se_6^+						
	Se_6	12597-30-7	**	8.23 ± 0.05	PE	4662
HSe^+						
	SeH	13940-22-2	**	9.79	S	3742
	H_2Se	7783-07-5	H	13.6 ± 0.2	EI	4610
			H	13.8 ± 0.2	EI	3633
H_2Se^+						
	$(^2\text{B}_1) \text{H}_2\text{Se}$	7783-07-5	**	9.88	PE	3719
	$(^2\text{B}_1)$		**	9.93	PE	4073
	$(^2\text{A}_1)$		**	12.40	PE	3719
	$(^2\text{B}_2)$		**	14.11	PE	3719
	$(^2\text{A}_1)$		**	21.0 (V)	PE	3719
			**	10.00 ± 0.05	EI	4610
CSe^+						
	CSe	16674-18-3	**	10.8 ± 0.5	EI	4966
			**	10.943	OTH	5000
	CSe_2	506-80-9		14.37 ± 0.02	PI	4936
			Se	14.586	PI	5000
CSe_2^+						
	CSe_2	506-80-9	**	9.25	S	5098
	$(^2\Sigma_u^+)$		**	13.6336	S	5000
	$(^3\Sigma_g^+)$		**	15.899	S	5000
	$(^2\Pi_{g,3/2})$		**	9.258 ± 0.0002	PI	5000
	$(\Pi_{3/2})$		**	9.26 ± 0.01	PI	4936
	$(^2\Pi_{1/2})$		**	9.52 ± 0.01	PI	4936
	$(^2\Pi_{g,1/2})$		**	9.524	PI	5000
	$(^2\Sigma_g^+)$		**	13.63 ± 0.02	PI	4936
	$(^2\Sigma_g^+)$		**	15.89 ± 0.02	PI	4936
	$(^2\Pi_{3/2g})$		**	9.26	PE	4309
	$(^2\Pi_{3/2g})$		**	9.27 ± 0.01	PE	3965
	$(^2\Pi_{1/2g})$		**	9.52	PE	4309
	$(^2\Pi_{1/2g})$		**	9.54 ± 0.01	PE	3965
	$(^2\Pi_u)$		**	11.45	PE	4309
	$(^2\Pi_u)$		**	11.49 ± 0.01	PE	3965
	$(^2\Sigma_u)$		**	13.61	PE	4309

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
CSe₂⁺ $(^2\Sigma_u^+)$	CSe ₂	506-80-9	**	13.63±0.01	PE	3965
			**	15.87	PE	4309
			**	15.90±0.01	PE	3965
			**	9.4±0.5	EI	4966
C₂H₂Se⁺	CH ₂ =C=Se	61134-37-0	**	8.72 (V)	PE	4982
C₂H₅Se⁺	CH ₃ SeCH ₂ CH ₂ CH(NH ₂)COOH	1464-42-2		12.03±0.06	EI	3443
C₂H₆Se⁺	(CH ₃) ₂ Se	593-79-3	**	8.400±0.010	S	3970
			**	8.40 (V)	PE	3656
C₃H₇Se⁺	CH ₃ SeCH ₂ CH ₂ CH(NH ₂)COOH	1464-42-2	C ₂ H ₅ NO ₂	9.34±0.15	EI	3443
C₄H₈Se⁺	C ₄ H ₈ Se (Selenophene)	288-05-1	**	8.776	S	5456
			**	8.80 (V)	PE	3858
			**	8.92 (V)	PE	4626
			**	≤8.92 (V)	PE	3804
			**	9.01±0.05	EI	3482
			**	8.96	CTS	4382
C₅H₆Se⁺	C ₅ H ₆ Se (Selenophene, tetrahydro-)	3465-98-3	**	8.14 (V)	PE	4145
C₆H₄Se⁺	C ₅ H ₄ SeCH ₃ (Selenophene, 2-methyl-)	7559-42-4	**	8.40±0.05 (V)	PE	4626
			**	8.38±0.1	EI	3804
C₆H₄Se⁺	C ₅ H ₄ =C=Se (Methaneselone, 2,4-cyclopentadien-1-ylidene-)	72443-10-8	**	8.34 (V)	PE	4982
C₈H₆Se⁺	C ₈ H ₆ Se (Benzo[b]selenophene)	272-30-0	**	8.03±0.05	PE	4435
C₈H₁₀Se₂⁺	C ₆ H ₄ (SeCH ₃) ₂ (Benzene, 1,4-bis(methylseleno)-)	40400-26-8	**	7.95 (V)	PE	5403
C₆H₄Se₁⁺	(C ₃ H ₅ Se ₂) ₂ (1,3-Diselenole, 2-(1,3-diselenol-2-ylidene)-)	54489-01-9	**	7.21	EI	5622
C₁₀H₁₂Se₁⁺	C ₆ Se ₁ (CH ₃) ₃ (1,3-Diselenole, 2-(4,5-dimethyl-1,3-diselenol-2-ylidene)-4,5-dimethyl-)	55259-49-9	**	6.58 (V)	PE	4481
C₃H₆NSe⁺	CH ₃ SeCH ₂ CH ₂ CH(NH ₂)COOH	1464-42-2		10.33±0.07	EI	3443
C₄H₁₀NSe⁺	CH ₃ SeCH ₂ CH ₂ CH(NH ₂)COOH	1464-42-2	CO ₂ H	9.83±0.16	EI	3443

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_7\text{H}_9\text{NSe}^+$	$\text{C}_6\text{H}_4(\text{SeCH}_3)\text{NH}_2$ (Benzenamine,2-(methylseleno)-) $\text{C}_6\text{H}_4(\text{SeCH}_3)\text{NH}_2$ (Benzenamine,3-(methylseleno)-) $\text{C}_6\text{H}_4(\text{SeCH}_3)\text{NH}_2$ (Benzenamine,4-(methylseleno)-)	70086-67-8 70086-66-7 35065-62-4	** ** **	7.95 (V) 7.83 (V) 7.88 (V)	PE	5403 5403 5403
$\text{C}_2\text{H}_2\text{N}_2\text{Se}^+$	$\text{C}_2\text{H}_2\text{N}_2\text{Se}$ (1,2,3-Selenadiazole)	26223-16-5	**	9.69 (V)	PE	4982
$\text{C}_6\text{H}_4\text{N}_2\text{Se}^+$	$\text{C}_6\text{H}_4\text{N}_2\text{Se}$ (1,2,3-Benzoselenadiazole)	123-92-7	**	8.83 (V)	PE	4982
O_2Se^+	${}^2\text{A}_1$ SeO_2 ${}^2\text{A}_2 + {}^2\text{B}_2$ ${}^2\text{B}_2$ ${}^2\text{A}_1 + {}^2\text{B}_1$ ${}^2\text{A}_1$	7446-08-4	** ** ** ** **	11.76 (V) 12.18 (V) 14.56 (V) 14.95 (V) 19.90 (V)	PE	4817 4817 4817 4817 4817
COSe^+	${}^2\Pi_{3/2}$ COSe $(X^2\pi_{3/2,1/2})$ ${}^2\Pi_{1/2}$ ${}^2\Pi$ ${}^2\Sigma^+$	1603-84-5	** ** ** ** **	10.36 ± 0.01 10.37 10.57 ± 0.01 14.58 ± 0.01 15.75 ± 0.01	PE	3965 4383 3965 3965 3965
$\text{C}_5\text{H}_4\text{OSe}^+$	$\text{C}_4\text{H}_3\text{SeCHO}$ (2-Selenophenecarboxaldehyde)	25109-26-6	**	9.47 ± 0.05	EI	3482
$\text{C}_5\text{H}_6\text{OSe}^+$	$\text{C}_4\text{H}_3\text{Se}(\text{=O})(\text{CH}_3)$ (2(5H)Selenophenone, 5-methyl-)	26562-65-2	**	8.84 ± 0.05	EI	4666
$\text{C}_6\text{H}_6\text{OSe}^+$	$\text{C}_4\text{H}_3\text{SeCOCH}_3$ (Ethanone, 1-selenophene-2-yl-)	15429-03-5	**	9.30 ± 0.05	EI	3482
$\text{C}_6\text{H}_8\text{OSe}^+$	$\text{C}_4\text{H}_2\text{Se}(\text{=O})(\text{CH}_3)_2$ (3(2H)-Selenophenone, 2,5-dimethyl-)	57556-10-2	**	8.24 ± 0.05	EI	4673
$\text{C}_7\text{H}_{10}\text{OSe}^+$	$\text{C}_4\text{HSe}(\text{CH}_3)_2\text{OCH}_3$ (Selenophene, 3-methoxy-2,5-dimethyl-) $\text{C}_4\text{HS}(\text{=O})(\text{CH}_3)_3$ (2(3H)-Selenophenone, 3,3,5-trimethyl-) $\text{C}_4\text{HS}(\text{=O})(\text{CH}_3)_3$ (3(2H)-Selenophenone, 2,2,5-trimethyl-)	57556-13-5 57556-20-4 57556-11-3	** ** **	7.69 ± 0.05 7.98 ± 0.05 8.21 ± 0.05	EI	4673 4666 4673
$\text{C}_8\text{H}_{10}\text{OSe}^+$	$\text{C}_6\text{H}_4(\text{SeCH}_3)(\text{OCH}_3)$ (Benzene,1-methoxy-2-(methylseleno)-) $\text{C}_6\text{H}_4(\text{SeCH}_3)(\text{OCH}_3)$ (Benzene,1-methoxy-3-(methylseleno)-) $\text{C}_6\text{H}_4(\text{SeCH}_3)(\text{OCH}_3)$ (Benzene,1-methoxy-4-(methylseleno)-)	1657-75-6 2726-42-3 1694-07-1	** ** **	7.86 (V) 7.93 (V) 8.05 (V)	PE	5403 5403 5403

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{12}\text{H}_8\text{OSe}^+$	$\text{C}_{12}\text{H}_8\text{OSe}$ (Phenoxyaselenin)	262-22-6	**	7.74 ± 0.05 (V)	PE	4743
$\text{C}_5\text{H}_4\text{O}_2\text{Se}^+$	$\text{C}_4\text{H}_3\text{SeCOOH}$ (2-Selenophenecarboxylic acid)	22968-45-2	**	9.19 ± 0.05 (V)	PE	4626
			**	9.25 ± 0.1	EI	3804
$\text{C}_6\text{H}_6\text{O}_2\text{Se}^+$	$\text{C}_4\text{H}_3\text{SeCOOCH}_3$ (2-Selenophenecarboxylic acid methyl ester)	39697-33-1	**	9.05 ± 0.05 (V)	PE	4626
$\text{C}_4\text{H}_6\text{NOSe}^+$	$\text{CH}_3\text{SeCH}_2\text{CH}_2\text{CH}(\text{NH}_2)\text{COOH}$	1464-42-2	$\text{H}_2\text{O} + \text{CH}_3$	10.00 ± 0.05	EI	3443
$\text{C}_5\text{H}_9\text{NOSe}^+$	$\text{CH}_3\text{SeCH}_2\text{CH}_2\text{CH}(\text{NH}_2)\text{COOH}$	1464-42-2	H_2O	8.73 ± 0.10	EI	3443
$\text{C}_7\text{H}_9\text{NOSe}^+$	$\text{C}_4\text{H}_3\text{SeCON}(\text{CH}_3)_2$ (2-Selenophenecarboxamide, N,N-dimethyl-)	55685-51-3	**	8.85 ± 0.05 (V)	PE	4626
$\text{C}_4\text{H}_3\text{NO}_2\text{Se}^+$	$\text{C}_4\text{H}_3\text{SeNO}_2$ (Selenophene, 2-nitro-)	15429-04-6	**	9.64 ± 0.05 (V)	PE	4626
$\text{C}_4\text{H}_8\text{NO}_2\text{Se}^+$	$\text{CH}_3\text{SeCH}_2\text{CH}_2\text{CH}(\text{NH}_2)\text{COOH}$	1464-42-2	CH_3	9.35 ± 0.10	EI	3443
$\text{C}_5\text{H}_{11}\text{NO}_2\text{Se}^+$	$\text{CH}_3\text{SeCH}_2\text{CH}_2\text{CH}(\text{NH}_2)\text{COOH}$	1464-42-2	**	8.26 ± 0.03	EI	3443
F_2Se^+	SeF_2	14017-34-6	**	10.20 (V)	PE	5074
$\text{C}_6\text{H}_3\text{OF}_3\text{Se}^+$	$\text{C}_4\text{H}_3\text{SeCOCF}_3$ (Ethanone, 2,2,2-trifluoro-1-(selenophene-2-yl)-)	26149-08-6	**	9.64 ± 0.05	EI	3482
$\text{H}_6\text{Si}_2\text{Se}^+$	$(\text{SiH}_3)_2\text{Se}$	14939-45-8	**	9.18 (V)	PE	3656
PSe^+	SeP	12509-41-0	**	8.2	EI	4001
P_4Se_3^+	P_4Se_3	1314-86-9	**	8.71 (V)	PE	4704
$\text{C}_3\text{H}_6\text{O}_3\text{PSe}^+$	$(\text{CH}_3\text{O})_3\text{PSe}$	152-19-2	**	8.67 (V)	PE	4705
SSe^+	SSe	7446-34-6	**	9.2 ± 0.3	EI	4682
CSSe^+ $(^2\Pi_{3/2})$ $(^2\pi_{3/2,1/2})$ $(X^2\Pi_{1/2})$	SCSe	5951-19-9	**	9.58 ± 0.01	PE	3965
			**	9.58	PE	4383
			**	9.77 ± 0.01	PE	3965
$\text{C}_4\text{H}_1\text{SSe}^+$	$\text{C}_4\text{H}_1\text{SSe}$ (1,4-Thiaselenin)	290-82-4	**	8.1 ± 0.1 (V)	PE	4841

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_5\text{H}_6\text{SSe}^+$	$\text{C}_4\text{H}_2\text{Se}(\text{SH})\text{CH}_3$ (2-Selenophenethiol, 5-methyl-)	63359-60-4	**	8.17 ± 0.05	EI	4706
$\text{C}_6\text{H}_8\text{SSe}^+$	$\text{C}_4\text{H}_2\text{Se}(\text{CH}_3)\text{SCH}_3$ (Selenophene, 2-methyl-5-(methylthiol)-)	63359-62-6	**	7.84 ± 0.05	EI	4706
	$\text{C}_4\text{HSe}(\text{SH})(\text{CH}_3)_2$ (3-Selenophenethiol, 2,5-dimethyl-)	63359-61-5	**	7.90 ± 0.05	EI	4706
$\text{C}_7\text{H}_{10}\text{SSe}^+$	$\text{C}_4\text{HSe}(\text{CH}_3)_2\text{SCH}_3$ (Selenophene, 2,5-dimethyl-3-(methylthiol)-)	63394-81-0	**	7.73 ± 0.05	EI	4706
$\text{C}_8\text{H}_{10}\text{SSe}^+$	$\text{C}_6\text{H}_4(\text{SeCH}_3)(\text{SCH}_3)$ (Benzene, 1-(methylseleno)-4-(methylthio)-)	70086-65-6	**	7.90 (V)	PE	5403
$\text{C}_6\text{H}_4\text{S}_2\text{Se}_2^+$	$(\text{C}_3\text{H}_2\text{SSe})_2$ (1,3-Thiaselenole,2-(1,3-thiaselenol-2-ylidene)-)		**	7.06	CTS	5622
Cl_2Se^+	SeCl_2	14457-70-6	** **	9.50 ± 0.2 (V) 9.52 (V)	PE PE	5023 5074
Cl_2Se_2^+	Se_2Cl_2	10025-68-0	**	9.81 ± 0.2 (V)	PE	5023
$\text{C}_4\text{H}_3\text{ClSe}^+$	$\text{C}_4\text{H}_3\text{SeCl}$ (Selenophene, 2-chloro-)	1449-67-8	**	8.83 ± 0.05 (V)	PE	4626
			**	8.72	CTS	4382
$\text{C}_2\text{H}_6\text{PClSe}^+$	$(\text{CH}_3)_2\text{P}(\text{Se})\text{Cl}$	XXXXX-XX-X	**	8.64 (V)	PE	5523
$\text{CH}_3\text{PCl}_2\text{Se}^+$	$\text{CH}_3\text{P}(\text{Se})\text{Cl}_2$	2171-82-6	**	9.16 (V)	PE	5523
$\text{C}_6\text{O}_5\text{SeCr}^+$	$(\text{CO})_5(\text{CSe})\text{Cr}$	63356-87-6	**	8.03 (V)	PE	5333
MnSe^+	MnSe	1313-22-0	**	8.2 ± 0.5 8.2 ± 0.5	EI EI	4901 4966
GeSe^+	SeGe $(^2\Pi_{3/2}, ^2\Pi_{1/2})$ $(^2\Sigma)$ $(^2\Sigma)$ $(^2\Sigma)$	12065-10-0	**	9.8 (V)	PE	4967
			**	9.95 (V)	PE	4550
			**	10.20 (V)	PE	4550
			**	13.4 (V)	PE	4550
			**	14.9 (V)	PE	4550
			**	10.2 ± 0.5 (V)	EI	4550
$\text{H}_6\text{Ge}_2\text{Se}^+$	$(\text{GeH}_3)_2\text{Se}$	24254-18-0	**	8.84 (V)	PE	3656
Br^+	Br $(^4\text{P}_2)$ $(^4\text{P}_0)$	10097-32-2	**	11.81	S	5209
			**	11.81 ± 0.02	PE	5087
			**	11.81	PE	5214

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
Br⁺						
(³ P ₁)	Br	10097-32-2	**	12.20±0.02	PE	5087
(³ P ₀)			**	12.20	PE	5214
(³ P ₀)			**	12.28	PE	5214
(³ P ₀)			**	12.30±0.02	PE	5087
(¹ D ₂)			**	13.28	PE	5214
(¹ D ₂)			**	13.30±0.02	PE	5087
(¹ S ₀)			**	15.26	PE	5214
(¹ S ₀)			**	15.27±0.02	PE	5087
CH ₃ Br		74-83-9	CH ₃	15.8±0.5	EI	4533
CH ₂ Br ₂		74-95-3	CH ₂ Br	15.5±0.1	EI	3442
			CH ₂ Br	16.0	EI	3490
AsBr ₃		7784-33-0	AsBr ₂	15.0±0.2	EI	5016
Br₂⁺						
(² π _{3/2p})	Br ₂	7726-95-6	**	10.57 (V)	PE	4564
(² π _{1/2p})			**	10.92 (V)	PE	4564
			**	10.8±0.2	EI	4906
AsBr ₃		7784-33-0	AsBr	13.4±0.1	EI	5016
HBr⁺						
(² Σ ⁺)	HBr	10035-10-6	**	15.2964±0.0025	S	4343
			**	11.66±0.02	PI	5307
(² Π _{3/2})			**	11.645±0.005	PE	3839
(² Π _{1/2})			**	11.979±0.005	PE	3839
(² Σ ⁺)			**	15.288±0.005	PE	3839
CH ₃ Br		74-83-9	CH ₂	15.9±0.3	EI	4533
C ₂ H ₃ Br		593-60-2		18.2±0.1	PI	5079
(CH ₃) ₂ CB ₂ NO		7119-91-7		11.60	EI	4809
DBr⁺						
(² Π _{3/2})	DBr	13536-59-9	**	11.673±0.005	PE	3839
(² Π _{1/2})			**	12.002±0.005	PE	3839
(² Σ ⁺)			**	15.284±0.005	PE	3839
H₂Br⁺						
(HBr) ₂		XXXXXX-XX-X	Br	11.42±0.03	PI	5307
H₂Br₂⁺						
(HBr) ₂		XXXXXX-XX-X	**	10.83±0.05	PI	5307
LiBr⁺						
LiBr		7550-35-8	**	9.43±0.05 (V)	PE	4950
			**	10.0 (V)	PE	4307
Li₂Br₂⁺						
(LiBr) ₂		XXXXXX-XX-X	**	10.05±0.08 (V)	PE	4950
H₈B₅Br⁺						
B ₅ H ₉ Br (Pentaborane(9), 1-bromo-)		23753-67-5	**	9.71 (V)	PE	4519
B ₅ H ₉ Br (Pentaborane(9), 2-bromo-)		23753-64-2	**	10.04 (V)	PE	4519
CBr⁺						
CH ₃ Br		74-83-9	H+H ₂	18.8±0.3	EI	4533
C₄Br₂⁺						
CBr≡CC≡CBr		36333-41-2	**	9.20±0.02	PE	4162
CBr₃⁺						
CBr ₃		558-13-4	Br	10.47±0.02	PI	4308

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₇H₉Br⁺	C ₇ H ₉ Br (Bicyclo[2.2.1]hept-2-ene, 5-bromo-, <i>exo</i> -)	5889-54-3	**	9.2	EI	5633
C₈H₅Br⁺	C ₆ H ₅ C≡CBr (Benzene, (bromoethynyl)-)	932-87-6	**	8.65 (V)	PE	4334
	C ₆ H ₄ (Br)C≡CH (Benzene, 1-bromo-4-ethynyl-)	766-96-1	**	8.62 (V)	PE	4334
C₁₀H₁₅Br⁺	C ₁₀ H ₁₅ Br (Tricyclo[3.3.1.1 ^{3,7}]decane, 1-bromo-)	768-90-1	**	9.2	PE	3907
			**	9.30±0.06	PE	3886
	C ₁₀ H ₁₅ Br (Tricyclo[3.3.1.1 ^{3,7}]decane, 2-bromo-)*	7314-85-4	**	9.31±0.05	PE	3886
C₁₂H₉Br⁺	C ₆ H ₅ C ₆ H ₄ Br (1,1'-Biphenyl, 4-bromo-)	92-66-0	**	8.05±0.02	PE	3702
C₁₄H₉Br⁺	C ₁₄ H ₉ Br (Anthracene, 9-bromo-)	1564-64-3	**	7.48±0.03 (V)	PE	4887
CHBr₂⁺	CHBr ₃	75-25-2	**	10.70±0.02	PI	4640
CH₂Br₂⁺	CH ₂ Br ₂	74-95-3	**	10.52±0.05	PI	4640
C₂H₂Br₂⁺	CBr ₂ CH ₂	593-92-0	**	9.78±0.01	S	5123
			**	9.78 (V)	PE	4303
	cis-CHBrCHBr	590-11-4	**	9.63±0.01	S	5123
			**	9.32±0.02	PE	3659
			**	9.63 (V)	PE	4303
	trans-CHBr=CHBr	590-12-5	**	9.47±0.01	S	4653
			**	9.55±0.01	S	5123
			**	9.30±0.02	PE	3659
			**	9.55 (V)	PE	4303
			**	9.56 (V)	PE	3648
C₂H₄Br₂⁺	CH ₂ BrCH ₂ Br	106-93-4	**	10.37	PI	5501
			**	10.42	PE	5501
			**	10.57±0.02 (V)	PE	4367
	CH ₃ CHBr ₂	557-91-5	**	10.17	PI	5501
			**	10.17	PE	5501
C₃H₆Br₂⁺	Br(CH ₂) ₃ Br	109-64-8	**	10.26 (V)	PE	4482
C₄H₈Br₂⁺	Br(CH ₂) ₄ Br	110-52-1	**	10.27 (V)	PE	4482
	CH ₃ (CHBr) ₂ CH ₃ (erythro) (Butane, (R',R')-(±)-2,3-dibromo-)	598-71-0	**	10.12 (V)	PE	4482
	CH ₃ (CHBr) ₂ CH ₃ (threo) (Butane, (R',S')-2,3-dibromo-)	5780-13-2	**	10.16 (V)	PE	4482
C₅H₈Br₂⁺	C ₅ H ₈ Br ₂ (Cyclopentane, 1,2-dibromo-, <i>cis</i> -)	33547-17-0	**	10.02±0.02	PE	4003

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₅H₈Br₂⁺	C ₅ H ₈ Br ₂ (Cyclopentane, 1,2-dibromo-, <i>trans</i> -)	10230-26-9	** **	10.08±0.02 10.04 (V)	PE PE	4003 4482
C₅H₁₀Br₂⁺	Br(CH ₂) ₅ Br	111-24-0	**	10.23 (V)	PE	4482
C₆H₄Br₂⁺	C ₆ H ₄ Br ₂ (Benzene, 1,2-dibromo-)	583-53-9	** **	8.99±0.03 (V) 9.02 (V) 9.05±0.03 (V)	PE PE PE	4890 3873 4890
	C ₆ H ₄ Br ₂ (Benzene, 1,3-dibromo-)	108-36-1	** **	9.10 (V) 8.90±0.03 (V)	PE PE	3873 4890
	C ₆ H ₄ Br ₂ (Benzene, 1,4-dibromo-)	106-37-6	** ** ** **	8.909 (V) 8.91 (V)	PE PE	5257 3873
C₆H₁₀Br₂⁺	C ₆ H ₁₀ Br ₂ (Cyclohexane, 1,2-dibromo- <i>cis</i> -)	19246-38-9	**	9.94±0.02	PE	4003
	C ₆ H ₁₀ Br ₂ (Cyclohexane, 1,2-dibromo-, <i>trans</i> -)	7429-37-0	** **	10.02±0.02 10.06±0.01 (V)	PE PE	4003 5218
C₁₀H₆Br₂⁺	C ₁₀ H ₆ Br ₂ (Azulene, 1,3-dibromo-)	14658-95-8	**	7.40 (V)	PE	5397
C₁₂H₈Br₂⁺	(C ₆ H ₅ Br) ₂ (1,1'-Biphenyl, 2,2'-dibromo-)	13029-09-9	**	8.40±0.02	PE	3702
C₁₁H₈Br₂⁺	C ₁₁ H ₈ Br ₂ (Anthracene, 9,10-dibromo-)	523-27-3	**	7.58	PE	4364
CHBr₃⁺	CHBr ₃	75-25-2	** **	10.48±0.02 10.47 (V)	PI PE	4640 4146
C₆H₃Br₃⁺	C ₆ H ₃ Br ₃ (Benzene, 1,3,5-tribromo-)	626-39-1	** **	8.91 (V) 9.21±0.02	PE PE	3873 5305
Be C₅H₅Br⁺	(C ₅ H ₅)BeBr (Beryllium,bromo(η^5 -2,4-cyclopentadien-1-yl)-)	52140-35-9	**	9.52 (V)	PE	5384
BC₂H₆Br⁺	(CH ₃) ₂ BrB	5158-50-9	** **	10.35 (V) 10.25	PE PE	4398 5485
B₁C₂H₅Br⁺	C ₂ B ₁ H ₅ Br (1,6-Dicarbahexaborane(6),2-bromo-)	XXXXX-XX-X	**	9.43 (V)	PE	5553
BCH₃Br₂⁺	CH ₃ Br ₂ B	17933-16-3	**	10.61 (V)	PE	4398

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
BCH₃Br₂⁺	CH ₃ Br ₂ B	17933-16-3	**	10.60	PE	5485
B₁C₂H₁Br₂⁺	C ₂ B ₁ H ₁ Br ₂ (1,6-Dicarbahexaborane(6),2,4-dibromo-)	XXXXX-XX-X	**	9.17 (V)	PE	5553
N₃Br⁺	BrN ₃	13973-87-0	**	10.00±0.01	PE	5001
H₂NBr⁺	NH ₂ Br	14519-10-9	**	10.18±0.04 (V)	PE	4947
HNBr₂⁺	NHBr ₂	14519-03-0	**	10.1±0.2 (V)	PE	4948
C₃NBr⁺	CBr≡CCN	3114-46-3	**	10.71±0.02	PE	4765
CH₃NBr⁺	CH ₃ NHBr	10218-87-8	** **	9.67 (V) 9.12	PE PE	4775 5329
C₂H₂NBr⁺	CH ₂ BrCN	590-17-0	**	11.28 (V)	PE	4684
C₂H₆NBr⁺	(CH ₃) ₂ NBr	10218-90-3	** **	9.15 (V) 8.61	PE PE	5304 5329
C₅H₄NBr⁺	C ₅ H ₄ NBr (Pyridine, 2-bromo-) C ₅ H ₄ NBr (Pyridine, 3-bromo-) C ₅ H ₄ NBr (Pyridine, 4-bromo-)	109-04-6 626-55-1 1120-87-2	** ** **	9.7±0.1 9.75±0.1 9.95±0.1	EI EI EI	4302 4302 4302
C₅H₁₀NBr⁺	C ₅ H ₁₀ NBr (Piperidine, 1-bromo-)	60094-06-6	**	8.92±0.10 (V)	PE	5308
C₆H₆NBr⁺	C ₆ H ₅ BrNH ₂ (Benzeneamine, 2-bromo-) C ₆ H ₅ BrNHCOCH ₃ (Acetamide, N-(2-bromophenyl)-) C ₆ H ₅ BrNHCOCH ₃ (Acetamide, N-(4-bromophenyl)-)	615-36-1 614-76-6 103-88-8	** CH ₂ =C=O CH ₂ =C=O	8.45 11.17±0.03 10.56±0.03	EI EI EI	4834 3483 3483
C₇H₁₂NBr⁺	C ₇ H ₁₂ NBr (1-Azabicyclo[2.2.2]octane, 4-bromo-)	2181-19-3	**	8.46±0.015 (V)	PE	4286
C₈H₁₄NBr⁺	C ₈ H ₁₄ NBr (8-Azabicyclo[3.2.1]octane, 3-bromo-8-methyl- <i>exo</i> -)	2292-11-7	**	7.8±0.15	EI	5401
C₁₃H₁₆NBr⁺	C ₆ H ₅ BrC(=CH ₂)C ₅ H ₄ N (Pyridine, 2-[1-(2-bromophenyl)ethenyl]-) C ₆ H ₅ BrC(=CH ₂)C ₅ H ₄ N (Pyridine, 2-[1-(4-bromophenyl)ethenyl]-)	XXXXX-XX-X XXXXX-XX-X	** **	8.6 8.62	OTH EI	5570 5570

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₃H₁₀NBr⁺	C ₆ H ₄ (Br)CH=CHC ₅ H ₃ N (Pyridine, <i>trans</i> -3-[2-(4-bromophenyl)ethenyl]-)	5847-71-2	**	8.15±0.05 (V)	PE	4377
C₉H₁₀N₂Br⁺	C ₆ H ₄ (Br)N=CHN(CH ₃) ₂ (Methanimidamide, <i>N'</i> -(2-bromophenyl)- <i>N,N</i> -dimethyl-)	53746-69-3	H	8.7	EI	4337
C₉H₁₁N₂Br⁺	C ₆ H ₄ (Br)N=CHN(CH ₃) ₂ (Methanimidamide, <i>N'</i> -(2-bromophenyl)- <i>N,N</i> -dimethyl-)	53746-69-3	**	7.2	EI	4337
C₁₈H₁₇N₂Br⁺	C ₆ H ₄ (Br)C ₃ H ₃ (CN)C ₆ H ₄ N(CH ₃) ₂ (Cyclopropanecarbonitrile, 1-(<i>p</i> -bromophenyl)-2-(<i>p</i> -(dimethylamino)phenyl)-)	32589-49-4	**	7.10±0.05	EI	3575
C₂H₂N₃Br⁺	C ₂ H ₂ N ₃ Br (1H-1,2,4-Triazole,5-bromo-)	XXXXX-XX-X	**	9.9 (V)	PE	5228
C₃H₄N₃Br⁺	C ₂ HN ₃ Br(CH ₃) (1H-1,2,4-Triazole,3-bromo-1-methyl-)	56616-91-2	**	9.55 (V)	PE	5228
	C ₂ HN ₃ Br(CH ₃) (1H-1,2,4-Triazole,5-bromo-1-methyl-)	16681-72-4	**	9.6 (V)	PE	5228
	C ₂ HN ₃ Br(CH ₃) (1H-1,2,4-Triazole,5-bromo-3-methyl-)	XXXXX-XX-X	**	9.6 (V)	PE	5228
	C ₂ HN ₃ Br(CH ₃) (4H-1,2,4-Triazole,3-bromo-4-methyl-)	16681-73-5	**	9.7 (V)	PE	5228
C₄H₆N₃Br⁺	C ₂ N ₃ Br(CH ₃) ₂ (1H-1,2,4-Triazole,3-bromo-1,5-dimethyl-)	56616-93-4	**	9.3 (V)	PE	5228
	C ₂ N ₃ Br(CH ₃) ₂ (1H-1,2,4-Triazole,5-bromo-1,3-dimethyl-)	56616-96-7	**	9.4 (V)	PE	5228
	C ₂ N ₃ Br(CH ₃) ₂ (4H-1,2,4-Triazole,3-bromo-4,5-dimethyl-)	56616-84-3	**	9.25 (V)	PE	5228
CH₃NBr₂⁺	CH ₃ NBr ₂	10218-83-4	** **	9.68 (V) 9.15	PE PE	4775 5329
C₅H₁₃NBr₂⁺	(C ₂ H ₅) ₂ NCH ₃ Br ₂	59777-81-0	**	10.60 (V)	PE	4564
C₆H₅NBr₂⁺	C ₆ H ₃ Br ₂ NHOCH ₃ (Acetamide, <i>N</i> -(2,4-dibromophenyl)-)	23373-04-8	CH ₂ =C=O	10.24±0.03	EI	3480
	C ₆ H ₃ Br ₂ NHOCH ₃ (Acetamide, <i>N</i> -(2,6-dibromophenyl)-)	33098-80-5	CH ₂ =C=O	10.02±0.03	EI	3480
C₆H₁₅NBr₂⁺	(C ₂ H ₅) ₃ NBr ₂	56348-00-6	**	10.60 (V)	PE	4564
C₉H₂₁NBr₂⁺	(<i>n</i> -C ₃ H ₇) ₃ NBr ₂	59777-82-1	**	10.77 (V)	PE	4564
C₁₂H₂₇NBr₂⁺	(<i>n</i> -C ₄ H ₉) ₃ NBr ₂	59777-83-2	**	10.66 (V)	PE	4564
BC₅H₇NBr⁺	C ₅ H ₄ N(Br)·BH ₃ (Pyridine, 4-bromo-, compound with borane (1:1))	56898-53-4	**	9.71 (V)	PE	4536

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
BC₁H₁₂N₂Br⁺	((CH ₃) ₂ N) ₂ BBR	6990-27-8	** **	8.13 8.16 (V)	PE PE	3584 3704
BC₂H₆NBr₂⁺	(CH ₃) ₂ NBBR	7360-64-7	** **	9.55 (V) 9.60	PE PE	3704 3584
B₂C₁H₁₂N₂Br₂⁺	(BrCH ₃ BNCH ₃) ₂	73775-15-2	**	9.58 (V)	PE	5628
B₂C₃H₉N₃Br₂⁺	N ₃ B ₂ Br ₂ (CH ₃) ₃ (1,2,4,3,5-Triazadiborolidine, 3,5-dibromo-1,2,4-trimethyl-)	53246-10-9	**	8.14 (V)	PE	4526
OBr⁺ (X ³ Σ ⁻)	BrO(X ² Π _{3/2})	14380-62-2	**	10.29±0.01 (V)	PE	5222
COBr₂⁺	CBr ₂ O	593-95-3	**	11.0 (V)	PE	3726
C₂O₂Br₂⁺	(COBr) ₂	15219-34-8	**	10.49±0.1	PE	4696
C₂H₃OB⁺	CH ₃ CBrO	506-96-7	**	10.68±0.05 (V)	PE	4220
C₂H₅OB⁺	CH ₂ BrCH ₂ OH-gauche <i>trans</i> -CH ₂ BrCH ₂ OH	XXXXX-XX-X XXXXX-XX-X	** **	10.75 (V) 10.65 (V)	PE PE	5088 5088
C₃H₇OB⁺	CH ₂ BrCH ₂ OCH ₃ -gauche <i>trans</i> -CH ₂ BrCH ₂ OCH ₃	XXXXX-XX-X XXXXX-XX-X	** **	10.13 (V) 10.20 (V)	PE PE	5088 5088
C₄H₃OB⁺	C ₄ H ₃ OB ⁺ (Furan, 3-bromo-)	22037-28-1	**	9.14	CTS	4382
C₅H₉OB⁺	C ₅ H ₈ (Br)OH (Cyclopentanol, 2-bromo-, <i>cis</i> -) C ₅ H ₈ (Br)OH (Cyclopentanol, 2-bromo-, <i>trans</i> -)	28435-62-3 20377-79-1	** **	10.19±0.02 10.11±0.02	PE PE	4003 4003
C₆H₄OB⁺	C ₆ H ₄ BrOCH ₃ (Benzene, 1-bromo-3-methoxy-) C ₆ H ₄ BrOCH ₃ (Benzene, 1-bromo-4-methoxy-) C ₆ H ₄ BrNO ₂ (Benzene, 1-bromo-3-nitro-) C ₆ H ₄ BrNO ₂ (Benzene, 1-bromo-4-nitro-)	2398-37-0 104-92-7 585-79-5 586-78-7	CH ₃ CH ₃ NO NO	12.29±0.1 11.89±0.1 10.26±0.1 10.55±0.1	EI	3446 3446 3447 3447
C₆H₅OB⁺	C ₆ H ₅ (OH)Br (Phenol, 2-bromo-) C ₆ H ₅ BrOOCCH ₃ (Phenol, 2-bromo-, acetate) C ₆ H ₅ BrOOCCH ₃ (Phenol, 3-bromo-, acetate)	95-56-7 1829-37-4 35065-86-2	** CH ₂ =C=O CH ₂ =C=O	9.09±0.1 9.62±0.03 10.02±0.2	EI EI EI	3553 3483 3484

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_6\text{H}_5\text{OBr}^+$	$\text{C}_6\text{H}_5\text{BrOOCCH}_3$ (Phenol, 4-bromo-, acetate)	1927-95-3	$\text{CH}_2=\text{C=O}$	9.84 ± 0.03	EI	3483
			$\text{CH}_2=\text{C=O}$	10.08 ± 0.2	EI	3484
$\text{C}_7\text{H}_5\text{OBr}^+$	$\text{C}_6\text{H}_5(\text{Br})\text{COOH}$ (Benzoic acid, 3-bromo-)	585-76-2	OH	12.23 ± 0.2	EI	3973
		586-76-5	OH	12.34 ± 0.2	EI	3973
	$\text{C}_6\text{H}_5(\text{Br})\text{COOH}$ (Benzoic acid, 4-bromo-)					
$\text{C}_7\text{H}_7\text{OBr}^+$	$\text{C}_6\text{H}_5\text{BrOCH}_3$ (Benzene, 1-bromo-3-methoxy-)	2398-37-0	**	8.69 ± 0.1	EI	3446
		104-92-7	**	8.11	PE	4621
	$\text{C}_6\text{H}_5(\text{Br})\text{OCH}_3$ (Benzene, 1-bromo-4-methoxy-)		**	8.39 ± 0.1	EI	3446
	$\text{C}_6\text{H}_5\text{Br}(\text{COCH}_3)$ (Ethanone, 1-(4-bromophenyl)-)	99-90-1	**	9.0 ± 0.1	PE	4401
$\text{C}_8\text{H}_9\text{OBr}^+$	$\text{C}_6\text{H}_5\text{OCH}_2\text{CH}_2\text{Br}$ (Benzene, 2-bromoethoxy-)	589-10-6	**	8.42	EI	5083
			**	8.49 ± 0.05	EI	5484
	$\text{C}_6\text{H}_5\text{OC}_3\text{H}_7\text{Br}$ (Benzene,(3-bromopropoxy)-)	XXXXX-XX-X	**	8.56 ± 0.05	EI	5484
$\text{C}_{10}\text{H}_{11}\text{OBr}^+$	$\text{C}_6\text{H}_5\text{OC}_4\text{H}_9\text{Br}$ (Benzene,4-bromobutoxy-)	XXXXX-XX-X	**	8.54 ± 0.05	EI	5484
$\text{C}_{11}\text{H}_{16}\text{OBr}^+$	$\text{C}_6\text{H}_5\text{OC}_5\text{H}_{11}\text{Br}$ (Benzene,[5-bromopentyl]oxy)-)	XXXXX-XX-X	**	8.59 ± 0.05	EI	5484
$\text{C}_{12}\text{H}_{18}\text{OBr}^+$	$\text{C}_6\text{H}_5\text{OC}_6\text{H}_{13}\text{Br}$ (Benzene,[6-bromohexyl]oxy)-)	XXXXX-XX-X	**	8.60 ± 0.05	EI	5484
$\text{C}_2\text{H}_3\text{O}_2\text{Br}^+$	CH_2BrCOOH	79-08-3	**	11.0 (V)	PE	3874
$\text{C}_7\text{H}_5\text{O}_2\text{Br}^+$	$\text{C}_6\text{H}_5(\text{Br})\text{COOH}$ (Benzoic acid, 3-bromo-)	585-76-2	**	9.66 ± 0.2	EI	3973
		586-76-5	**	9.72 ± 0.2	EI	3973
	$\text{C}_6\text{H}_5(\text{Br})\text{COOCCH}_3$ (Cyclopentanol, 2-bromo-, acetate, <i>cis</i> -)	53093-41-7	**	10.00 ± 0.02	PE	4003
$\text{C}_7\text{H}_{11}\text{O}_2\text{Br}^+$	$\text{C}_5\text{H}_8(\text{Br})\text{OCOCH}_3$ (Cyclopentanol, 2-bromo-, acetate, <i>trans</i> -)	53093-42-8	**	10.07 ± 0.02	PE	4003
		1829-37-4	**	8.66 ± 0.03	EI	3483
	$\text{C}_6\text{H}_5\text{BrOOCCH}_3$ (Phenol, 3-bromo-, acetate)	35065-86-2	**	8.79 ± 0.2	EI	3484

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₈H₇O₂Br⁺						
	C ₆ H ₅ BrOOCCH ₃ (Phenol, 4-bromo-, acetate)	1927-95-3	**	8.42±0.03	EI	3483
			**	8.61±0.2	EI	3484
C₆H₅OB₂Br⁺						
	C ₆ H ₅ Br ₂ OOCCH ₃ (Phenol, 2,4-dibromo-, acetate)	36914-79-1	CH ₂ =C=O	9.45±0.03	EI	3480
	C ₆ H ₅ Br ₂ OOCCH ₃ (Phenol, 2,6-dibromo-, acetate)	28165-72-2	CH ₂ =C=O	9.74±0.03	EI	3480
C₈H₆O₂Br₂⁺						
	C ₆ H ₅ Br ₂ OOCCH ₃ (Phenol, 2,4-dibromo-, acetate)	36914-79-1	**	8.21±0.03	EI	3480
	C ₆ H ₅ Br ₂ OOCCH ₃ (Phenol, 2,6-dibromo-, acetate)	28165-72-2	**	8.42±0.03	EI	3480
NOBr⁺						
(² A'; ² A'')	NOBr	13444-87-6	**	10.17	PE	4404
(² A'; ² A'')			**	10.20±0.05	PE	4420
CNOBr⁺						
	BrNCO	3644-72-2	**	10.46±0.01	PE	5001
CNOBr₃⁺						
	CBr ₃ NO	XXXXX-XX-X		9.96±0.05 (V)	PE	5298
C₅H₄NOBr⁺						
	C ₅ H ₄ N(O)Br (Pyridine, 4-bromo-, 1-oxide)	14248-50-1	**	8.44 (V)	PE	4222
C₆H₁₂NOBr⁺						
	C ₆ H ₁₂ NOBr	52761-86-1	**	9.06±0.1 (V)	PE	4465
C₈H₇NOBr⁺						
	C ₆ H ₅ Br ₂ NHCOCH ₃ (Acetamide, N-(2,4-dibromophenyl)-)	23373-04-8		8.84±0.03	EI	3480
	C ₆ H ₅ Br ₂ NHCOCH ₃ (Acetamide, N-(2,6-dibromophenyl)-)	33098-80-5		8.88±0.03	EI	3480
C₈H₈NOBr⁺						
	C ₆ H ₅ BrNHCOCH ₃ (Acetamide, N-(2-bromophenyl)-)	614-76-6	**	8.50	EI	4834
			**	8.17±0.03	EI	3483
	C ₆ H ₅ BrNHCOCH ₃ (Acetamide, N-(4-bromophenyl)-)	103-88-8	**	8.17±0.03	EI	3483
C₁₂H₈NOBr⁺						
	C ₆ H ₅ BrCOC ₅ H ₄ N (Methanone, (2-bromophenyl)-2-pyridinyl-)	XXXXX-XX-X	**	8.93	EI	5459
C₄H₃N₂OB₂Br⁺						
	C ₄ H ₃ N ₂ Br(=O) (2(1H)-Pyrimidinone, 5-bromo-)	38353-06-9	**	9.47±0.05	EI	5159
C₅H₅N₂OB₂Br⁺						
	C ₄ H ₂ N ₂ BrOCH ₃ (Pyrimidine, 5-bromo-2-methoxy-)	14001-66-2	**	9.11±0.05	EI	5159
	C ₄ H ₂ N ₂ Br(=O)CH ₃ (2(1H)Pyrimidinone, 5-bromo-1-methyl-)	14248-01-2	**	8.78±0.05	EI	5159

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₇H₇N₂OBr⁺	C ₆ H ₅ BrNHCONH ₂ (Urea, (2-bromophenyl)-)	13114-90-4	**	8.45	EI	4834
C₄H₄NO₂Br⁺	C ₆ H ₅ N(Br)(=O) ₂ (2,5-Pyrrolidinedione, 1-bromo-)	128-08-5	**	10.12 (V)	PE	4742
			**	10.12 (V)	PE	4810
C₅H₈NO₂Br⁺	C ₃ H ₂ NO(=O)(Br)(CH ₃) ₂ (2-Oxazolidinone, 3-bromo-4,4-dimethyl-)	60491-95-4	**	9.45 (V)	PE	4742
C₆H₄NO₂Br⁺	C ₆ H ₅ BrNO ₂ (Benzene, 1-bromo-3-nitro-)	585-79-5	**	9.82±0.1	EI	3447
	C ₆ H ₅ BrNO ₂ (Benzene, 1-bromo-4-nitro-)	586-78-7	**	9.76±0.1	EI	3447
C₁₁H₁₆NO₂Br⁺	C ₁₁ H ₁₆ NO ₂ Br (Benzeneethanamine, 4-bromo-2,5-dimethoxy- α -methyl-(\pm)-)	64638-07-9	**	7.94±0.06	PE	4758
C₈H₇NOBr⁺	C ₆ H ₅ Br ₂ NHCOCH ₃ (Acetamide, N-(2,4-dibromophenyl)-)	23373-04-8	**	8.08±0.03	EI	3480
	C ₆ H ₅ Br ₂ NHCOCH ₃ (Acetamide, N-(2,6-dibromophenyl)-)	33098-80-5	**	8.32±0.03	EI	3480
FBr⁺	BrF	13863-59-7	**	11.77±0.01	PE	4755
			**	11.78±0.01	PE	3680
			**	12.09±0.01	PE	3680
			**	12.10±0.01	PE	4755
			**	15.92±0.01 (V)	PE	4755
F₃Br⁺	BrF ₃	7787-71-5	**	12.15±0.04	PE	3680
F₅Br⁺	BrF ₅	7789-30-2	**	13.172±0.005	PE	3655
CF₃Br⁺	CF ₃ Br	75-63-8	**	12.0 (V)	PE	3914
			**	12.08±0.05 (V)	PE	4727
			**	12.12±0.02 (V)	PE	4026
C₂F₃Br⁺	C ₂ F ₃ Br	598-73-2	**	9.67	PE	3589
			**	10.11 (V)	PE	4303
C₃F₃Br⁺	CF ₃ C≡CBr	819-01-2	**	10.81±0.02	PE	4765
C₆F₅Br⁺	C ₆ F ₅ Br (Benzene, bromopentafluoro-)	344-04-7	**	9.57 (V)	PE	5252
			**	9.67±0.02	PE	5305
CF₂Br₂⁺	CF ₂ Br ₂	75-61-6	**	11.18 (V)	PE	5470

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_2\text{F}_4\text{Br}_2^+$	$(\text{CF}_2\text{Br})_2$	124-73-2	**	11.44 ± 0.01 (V)	PE	4613
$\text{C}_6\text{F}_4\text{Br}_2^+$	$\text{C}_6\text{F}_4\text{Br}_2$ (Benzene, 1,2-dibromo-3,4,5,6-tetrafluoro-) $\text{C}_6\text{F}_4\text{Br}_2$ (Benzene, 1,3-dibromo-2,4,5,6-tetrafluoro-) $\text{C}_6\text{F}_4\text{Br}_2$ (Benzene, 1,4-dibromo-2,3,5,6-tetrafluoro-)	827-08-7 27516-63-8 344-03-6	**	9.50 ± 0.02 9.45 ± 0.02 9.42 ± 0.02	PE	5305 5305 5305
CFBr_3^+	CFBr_3	353-54-8	**	10.67 ± 0.01	PE	4365
$\text{C}_6\text{F}_3\text{Br}_3^+$	$\text{C}_6\text{F}_3\text{Br}_3$ (Benzene, 1,3,5-tribromo-2,4,6-trifluoro-)	XXXXX-XX-X	**	9.33 ± 0.02	PE	5305
$\text{C}_2\text{H}_2\text{FBr}^+$	<i>cis</i> -CHF=CHBr	2366-31-6	**	9.75 (V)	PE	4303
$\text{C}_2\text{H}_4\text{FBr}^+$	$\text{CH}_2\text{FCH}_2\text{Br}$	762-49-2	**	10.57 (V)	PE	4482
$\text{C}_3\text{H}_6\text{FBr}^+$	$\text{CH}_2\text{FCH}_2\text{CH}_2\text{Br}$ $\text{CH}_3\text{CHFCH}_2\text{Br}$	352-91-0 1871-72-3	**	10.38 (V) 10.44 (V)	PE PE	4482 4482
$\text{C}_4\text{H}_8\text{FBr}^+$	$\text{CH}_3\text{CH}_2\text{CHFCH}_2\text{Br}$ $(\text{CH}_3)_2\text{CFCH}_2\text{Br}$ $\text{CH}_3\text{CHFCHBrCH}_3$ (erythro) $\text{CH}_3\text{CHFCHBrCH}_3$ (threo)	1871-73-4 19869-78-4 57302-15-5 5780-13-2	**	10.32 (V) 10.28 (V) 10.91 (V) 10.21 (V)	PE PE PE PE	4482 4482 4482 4482
$\text{C}_5\text{H}_8\text{FBr}^+$	$\text{C}_5\text{H}_8\text{FBr}$ (Cyclopentane, 1-bromo-2-fluoro-, <i>cis</i> -) $\text{C}_5\text{H}_8\text{FBr}$ (Cyclopentane, 1-bromo-2-fluoro-, <i>trans</i> -)	51422-72-1 51422-73-2	**	10.10 ± 0.02 10.25 ± 0.02	PE PE	4003 4003
$\text{C}_6\text{H}_4\text{FBr}^+$	$\text{C}_6\text{H}_4\text{FBr}$ (Benzene, 1-bromo-2-fluoro-)	1072-85-1	**	9.11 ± 0.02	PE	5305
			**	9.14 (V)	PE	4567
	$\text{C}_6\text{H}_4\text{FBr}$ (Benzene, 1-bromo-3-fluoro-)	1073-06-9	**	9.11 ± 0.02	PE	5305
			**	9.25 (V)	PE	4567
	$\text{C}_6\text{H}_4\text{FBr}$ (Benzene, 1-bromo-4-fluoro-)	460-00-4	**	9.02 ± 0.02	PE	5305
			**	9.03 (V)	PE	4567
$\text{C}_6\text{H}_{10}\text{FBr}^+$	$\text{C}_6\text{H}_{10}\text{FBr}$ (Cyclohexane, 1-bromo-2-fluoro-, <i>cis</i> -)	51422-74-3	**	10.04 ± 0.02	PE	4003
			**	10.06 (V)	PE	4482
	$\text{C}_6\text{H}_{10}\text{FBr}$ (Cyclohexane, 1-bromo-2-fluoro-, <i>trans</i> -)	17170-96-6	**	10.18 ± 0.02	PE	4003
			**	10.05 (V)	PE	4482
$\text{C}_{12}\text{H}_8\text{FBr}^+$	$\text{C}_6\text{H}_5(\text{Br})\text{C}_6\text{H}_3\text{F}$ (1,1'-Biphenyl, 4-bromo-4'-fluoro-)	398-21-0	**	8.10 ± 0.02	PE	3702

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₆H₃F₂Br⁺						
	C ₆ H ₃ F ₂ Br (Benzene,1-bromo-2,4-difluoro-)	348-57-2	**	9.16±0.02	PE	5305
	C ₆ H ₃ F ₂ Br (Benzene,2-bromo-1,4-difluoro-)	399-94-0	**	9.18±0.02	PE	5305
	C ₆ H ₃ F ₂ Br (Benzene,4-bromo-1,2-difluoro-)	348-61-8	**	9.19±0.02	PE	5305
C₆H₂F₃Br⁺						
	C ₆ H ₂ F ₃ Br (Benzene,1-bromo-2,4,5-trifluoro-)	327-52-6	**	9.25±0.02	PE	5305
	C ₆ H ₂ F ₃ Br (Benzene,2-bromo-1,3,5-trifluoro-)	2367-76-2	**	9.34±0.02	PE	5305
C₇H₄F₃Br⁺						
	C ₆ H ₄ BrCF ₃ (Benzene, 1-bromo-2-trifluoromethyl-)	392-83-6	**	9.38 (V)	PE	4567
	C ₆ H ₄ BrCF ₃ (Benzene, 1-bromo-3-trifluoromethyl-)	401-78-5	**	9.36 (V)	PE	4567
	C ₆ H ₄ BrCF ₃ (Benzene, 1-bromo-4-trifluoromethyl-)	402-43-7	**	9.48 (V)	PE	4567
C₆HF₄Br⁺						
	C ₆ HF ₄ Br (Benzene,1-bromo-2,3,4,5-tetrafluoro-)	1074-91-5	**	9.50±0.02	PE	5305
	C ₆ HF ₄ Br (Benzene,2-bromo-1,3,4,5-tetrafluoro-)	1559-86-0	**	9.46±0.02	PE	5305
	C ₆ HF ₄ Br (Benzene,3-bromo-1,2,4,5-tetrafluoro-)	1559-88-2	**	9.45±0.02	PE	5305
C₆H₃FBr₂⁺						
	C ₆ H ₃ FBr ₂ (Benzene,2,4-dibromo-1-fluoro-)	XXXXX-XX-X	**	9.05±0.02	PE	5305
C₂H₂F₂Br₂⁺						
	CF ₂ BrCH ₂ Br	75-82-1	**	10.86±0.01 (V)	PE	4613
C₆H₂F₂Br₂⁺						
	C ₆ H ₂ F ₂ Br ₂ (Benzene,1,2-dibromo-4,5-difluoro-)	XXXXX-XX-X	**	9.13±0.02	PE	5305
	C ₆ H ₂ F ₂ Br ₂ (Benzene,1,4-dibromo-2,5-difluoro-)	XXXXX-XX-X	**	9.09±0.02	PE	5305
C₃H₂OF₃Br⁺						
	CH ₂ BrCOCF ₃	431-35-6	**	10.92±0.02 (V)	PE	4524
NaBr⁺						
(³ P _{3/2})	NaBr	7647-15-6	**	8.31±0.1	PE	4344
(³ P _{3/2})			**	8.31±0.1	PE	5035
(³ P _{1/2})			**	8.7 (V)	PE	4307
(³ P _{1/2})			**	9.45±0.04 (V)	PE	5035
AlBr⁺						
	AlBr	22359-97-3	**	9.3	PE	4860
AlBr₃⁺						
	AlBr ₃	7727-15-3	**	10.91 (V)	PE	4398
			**	10.91 (V)	PE	4256
Al₂Br₆⁺						
	(AlBr ₃) ₂	18898-34-5	**	10.97 (V)	PE	4559
			**	10.97 (V)	PE	4256

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_2\text{H}_6\text{AlBr}^+$	$(\text{CH}_3)_2\text{BrAl}$	3017-85-4	**	9.90 (V)	PE	4398
$\text{CH}_3\text{AlBr}_2^+$	$\text{CH}_3\text{Br}_2\text{Al}$	3017-75-2	**	10.65 (V)	PE	4398
$\text{C}_4\text{H}_{12}\text{Al}_2\text{Br}_2^+$	$((\text{CH}_3)_2\text{BrAl})_2$	15218-96-9	**	9.68 (V)	PE	4559
SiBr^+	SiBr	14791-57-2	** **	7.3 $9.0 \pm 1.$	S EI	3558 5166
SiBr_2^+	SiBr_2	14877-32-8	**	$12.0 \pm 1.$	EI	5166
SiBr_3^+	SiBr_3	13842-48-3	**	$12.5 \pm 1.$	EI	5166
SiBr_4^+	SiBr_4	7789-66-4	**	$14.0 \pm 1.$	EI	5166
H_3SiBr^+	SiH_3Br	13465-73-1	** ** ** **	10.90 (V) 11.03 10.96 ± 0.02 (V) 11.03 ± 0.05 (V)	PE S PE PE	3511 4697 3510 3502
$\text{H}_2\text{SiBr}_2^+$	SiH_2Br_2	13768-94-0	**	10.92 ± 0.02 (V)	PE	3510
$\text{C}_3\text{H}_9\text{SiBr}^+$	$(\text{CH}_3)_3\text{SiBr}$	2857-97-8	**	10.23 (V)	PE	4566
$\text{C}_5\text{H}_9\text{SiBr}^+$	$(\text{CH}_3)_3\text{SiC}\equiv\text{CBr}$	18243-59-9	**	9.4 ± 0.1	PE	4002
$\text{C}_9\text{H}_{13}\text{SiBr}^+$	$\text{BrC}_6\text{H}_4\text{Si}(\text{CH}_3)_3$ (Silane,(4-bromophenyl)trimethyl-)	6999-03-7	**	8.67 (V)	PE	5380
$\text{C}_8\text{H}_{15}\text{Si}_2\text{Br}_2^+$	$\text{C}_8\text{H}_{15}\text{Si}_2\text{Br}_2$	65411-95-2	**	8.83 (V)	PE	4715
$\text{C}_9\text{H}_{13}\text{OSiBr}^+$	$\text{BrC}_6\text{H}_4\text{Si}(\text{CH}_3)_2\text{OCH}_3$ (Silane,(3-bromophenyl)methoxydimethyl-) $\text{BrC}_6\text{H}_4\text{Si}(\text{CH}_3)_2\text{OCH}_3$ (Silane,(4-bromophenyl)methoxydimethyl-)	62244-46-6 17021-92-0	** **	9.22 9.16	EI EI	5421 5421
F_3SiBr^+	SiF_3Br	14049-39-9	**	12.46 ± 0.02 (V)	PE	4026
$\text{C}_7\text{H}_7\text{FSiBr}^+$	$\text{BrC}_6\text{H}_4\text{Si}(\text{CH}_3)_2\text{F}$ (Silane,(3-bromophenyl)fluorodimethyl-) $\text{BrC}_6\text{H}_4\text{Si}(\text{CH}_3)_2\text{F}$ (Silane,(4-bromophenyl)fluorodimethyl-)	62244-54-6 62244-53-5	CH_3 CH_3	11.11 10.91	EI EI	5366 5366
$\text{C}_8\text{H}_{10}\text{FSiBr}^+$	$\text{BrC}_6\text{H}_4\text{Si}(\text{CH}_3)_2\text{F}$ (Silane,(3-bromophenyl)fluorodimethyl-)	62244-54-6	**	9.11	EI	5421

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_8\text{H}_{10}\text{FSiBr}^+$	$\text{BrC}_6\text{H}_5\text{Si}(\text{CH}_3)_2\text{F}$ (Silane,(4-bromophenyl)fluorodimethyl-)	62244-53-5	**	9.01	EI	5421
PBr^+	PBr_3	7789-60-8		14.2 ± 0.2	EI	3556
PBr_2^+	PBr_3	7789-60-8	Br	11.2 ± 0.1	EI	3556
PBr_3^+	PBr_3	7789-60-8	** ** ** ** ** **	9.96 (V) 9.99 (V) 10.0 (V) 10.00 ± 0.03 (V) 10.00 (V) 10.1 ± 0.1	PE PE PE PE PE EI	4023 5539 5190 3669 4146 3556
OPBr_3^+	POBr_3	7789-59-5	** ** **	10.75 ± 0.02 10.99 (V) 11.03 ± 0.03 (V)	PE PE PE	3835 4023 3669
$\text{CH}_3\text{O}_2\text{PBr}_2^+$	$\text{PBr}_2\text{O}(\text{OCH}_3)$	63560-73-6	**	9.97 (V)	PE	4699
F_2PBr^+	PF_2Br	15597-40-7	**	11.08 ± 0.1 (V)	PE	3662
SBr_2^+	SBr_2	14312-20-0	** **	9.33 ± 0.05 (V) 9.4 (V)	PE PE	5031 5466
S_2Br_2^+	S_2Br_2	13172-31-1	**	9.5	PE	4188
$\text{H}_8\text{B}_9\text{SBr}^+$	$\text{SB}_9\text{H}_8(\text{Br})$ (1-Thiadecaborane(9),10-bromo-) $\text{SB}_9\text{H}_8(\text{Br})$ (1-Thiadecaborane(9),6-bromo-)	58568-92-6 58575-43-2	** **	9.52 (V) 9.51 (V)	PE PE	5324 5324
C_1SBr_1^+	C_1SBr_1 (Thiophene, tetrabromo-)	3958-03-0	**	8.53 (V)	PE	4690
$\text{C}_6\text{S}_2\text{Br}_1^+$	$\text{C}_6\text{S}_2\text{Br}_1^+$ (Thieno[2,3- <i>b</i>]thiophene,2,3,4,5-tetrabromo-)	53255-86-0	**	8.39 (V)	PE	5478
$\text{C}_1\text{H}_3\text{SBr}^+$	$\text{C}_1\text{H}_3\text{SBr}$ (Thiophene, 2-bromo-)	1003-09-4	** ** ** ** ** **	8.60 (V) 8.664 ± 0.005 8.664 8.82 ± 0.05 (V) 8.93 ± 0.05 8.80	PE PE PE PE EI CTS	4690 3911 3645 4626 3482 3787
	$\text{C}_1\text{H}_3\text{SBr}$ (Thiophene, 3-bromo-)	872-31-1	** **	8.812 ± 0.005 8.812 8.97 (V)	PE PE	3911 3645 4690

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_4\text{H}_3\text{SBr}^+$	$\text{C}_4\text{H}_3\text{SBr}$	872-31-1	** **	9.02 ± 0.05 8.87	EI CTS	3482 4382
$\text{C}_7\text{H}_7\text{SBr}^+$	$\text{C}_6\text{H}_5(\text{Br})\text{SCH}_3$ (Benzene, 1-bromo-4-(methylthio)-)	104-95-0	**	8.17 ± 0.05 (V)	PE	4627
$\text{C}_6\text{H}_3\text{S}_2\text{Br}^+$	$\text{C}_6\text{H}_3\text{S}_2\text{Br}$ (Thieno[2,3- <i>b</i>]thiophene,2-bromo-) $\text{C}_6\text{H}_3\text{S}_2\text{Br}$ (Thieno[2,3- <i>b</i>]thiophene,3-bromo-)	25121-81-7 25121-84-0	** **	8.35 (V) 8.43 (V)	PE PE	5478 5478
$\text{C}_4\text{H}_2\text{SBr}_2^+$	$\text{C}_4\text{H}_3\text{SBr}_2$ (Thiophene, 2,5-dibromo-) $\text{C}_4\text{H}_3\text{SBr}_2$ (Thiophene, 3,4-dibromo-)	3141-27-3 3141-26-2	** **	8.49 (V) 8.94 (V)	PE PE	4690 4690
$\text{C}_6\text{H}_2\text{S}_2\text{Br}_2^+$	$\text{C}_6\text{H}_3\text{S}_2\text{Br}_2$ (Thieno[2,3- <i>b</i>]thiophene,2,5-dibromo-) $\text{C}_6\text{H}_3\text{S}_2\text{Br}_2$ (Thieno[2,3- <i>b</i>]thiophene,3,4-dibromo-)	25121-86-2 53255-78-0	** **	8.19 (V) 8.30 (V)	PE PE	5478 5478
$\text{C}_8\text{H}_4\text{S}_3\text{Br}_2^+$	$(\text{C}_4\text{H}_2\text{S}(\text{Br}))_2\text{S}$ (Thiophene,2,2'-thiobis[3-bromo-]) $(\text{C}_4\text{H}_2\text{S}(\text{Br}))_2\text{S}$ (Thiophene,2,2'-thiobis[4-bromo-]) $(\text{C}_4\text{H}_2\text{S}(\text{Br}))_2\text{S}$ (Thiophene,3,3'-thiobis[2-bromo-]) $(\text{C}_4\text{H}_2\text{S}(\text{Br}))_2\text{S}$ (Thiophene,3,3'-thiobis[4-bromo-])	28504-80-5 65828-00-4 65827-99-8 28504-81-6	** ** ** **	8.50 (V) 8.60 (V) 8.10 (V) 8.20 (V)	PE PE PE PE	5356 5356 5356 5356
$\text{C}_6\text{HS}_2\text{Br}_3^+$	$\text{C}_6\text{HS}_2\text{Br}_3$ (Thieno[2,3- <i>b</i>]thiophene,2,3,4-tribromo-) $\text{C}_6\text{HS}_2\text{Br}_3$ (Thieno[2,3- <i>b</i>]thiophene,2,3,5-tribromo-)	53255-84-8 53255-85-9	** **	8.35 (V) 8.28 (V)	PE PE	5478 5478
$\text{BC}_{12}\text{H}_{18}\text{SBr}^+$	$\text{C}_6\text{H}_5(\text{Br})\text{SB}(\text{n-C}_3\text{H}_7)_2$ (Borinic acid, dipropylthio-4-bromophenyl ester)	64503-49-7	**	8.67 ± 0.05 (V)	PE	4848
$\text{C}_8\text{H}_6\text{NSBr}^+$	$\text{C}_7\text{H}_5\text{NS}(\text{Br})\text{CH}_3$ (Benzothiazole, 6-bromo-2-methyl-)	5304-21-2	**	8.55 (V)	PE	4437
$\text{C}_8\text{H}_8\text{NSBr}^+$	$\text{C}_7\text{H}_5\text{BrNHCSCH}_3$ (Ethanethioamide, N-(2-bromophenyl)-)	62635-46-5	**	8.05	EI	4834
$\text{C}_7\text{H}_7\text{N}_2\text{SBr}^+$	$\text{C}_6\text{H}_5\text{BrNHCSNH}_2$ (Thiourea, (2-bromophenyl)-)	5391-30-0	**	8.10	EI	4834
OSBr^+	SOBr_2	507-16-4	** ** **	10.54 (V) 10.54 (V) 10.63 (V)	PE PE PE	3646 4295 3705

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
OSBr₃⁺	SOBr ₃		XXXXX-XX-X **	9.41±0.02	PE	3835
C₁₂H₈O₂SBBr₂⁺	(C ₆ H ₅ Br) ₂ SO ₂ (Benzene, 1,1'-sulfonylbis[4-bromo-])	2050-48-8	**	8.84±0.05	PI	5040
PSBr₃⁺	PSBr ₃	3931-89-3	** **	9.82 (V) 9.89±0.03 (V)	PE PE	4023 3669
C₂H₆PSBr⁺	(CH ₃) ₂ P(S)Br	6839-93-6	**	8.18 (V)	PE	5523
CH₃PSBr₂⁺	CH ₃ P(S)Br ₂	5827-24-7	**	9.53 (V)	PE	5523
F₂PSBr⁺	F ₂ P(S)Br	13706-09-7	**	10.58 (V)	PE	5523
FPSBr₂⁺	FP(S)Br ₂	13706-10-0	**	10.23 (V)	PE	5523
CCl₂Br₂⁺	CBr ₂ Cl ₂	594-18-3	**	10.67±0.02 (V)	PE	4880
C₂H₅ClBr⁺	CH ₂ BrCH ₂ Cl <i>trans</i> -CH ₂ BrCH ₂ Cl CH ₂ BrCH ₂ Cl-gauche CH ₃ CHClBr	107-04-0 593-96-4	** ** ** ** ** ** **	10.55 10.52 10.65±0.01 (V) 10.67±0.1 (V) 10.52±0.1 (V) 10.42 10.37	PI PE PE PE PE PI PE	5501 5501 4613 4751 4751 5501 5501
C₅H₈ClBr⁺	C ₅ H ₈ ClBr (Cyclopentane, 1-bromo-2-chloro-, <i>cis</i> -) C ₅ H ₈ ClBr (Cyclopentane, 1-bromo-2-chloro-, <i>trans</i> -)	37722-39-7 14376-82-0	** **	10.13±0.02 10.23±0.02	PE PE	4003 4003
C₆H₁₀ClBr⁺	C ₆ H ₁₀ ClBr (Cyclohexane, 1-bromo-2-chloro-, <i>cis</i> -) C ₆ H ₁₀ ClBr (Cyclohexane, 1-bromo-2-chloro-, <i>trans</i> -)	51422-75-4 13898-96-9	** **	10.03±0.02 10.13±0.02	PE PE	4003 4003
CNOCl₂Br⁺	CCl ₂ BrNO		XXXXX-XX-X	10.22±0.05 (V)	PE	5298
CNOClBr₂⁺	CClBr ₂ NO		XXXXX-XX-X	10.02±0.05 (V)	PE	5298
CF₂ClBr⁺	CF ₂ BrCl	353-59-3	**	11.83 (V)	PE	5470
PClBr⁺	PClBr ₂	13550-32-8	Br	11.3±0.1	EI	3556
PCl₂Br⁺	PCl ₂ Br	13536-48-6	**	10.4±0.1	EI	3556

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
PClBr₂⁺	PClBr ₂	13550-32-8	**	10.2±0.1	EI	3556
KBr⁺ (² P _{3/2}) (² P _{3/2}) (² P _{1/2})	KBr	7758-02-3	**	7.85±0.1	PE	4344
			**	7.85±0.1	PE	5035
			**	8.1 (V)	PE	4307
			**	8.82±0.04 (V)	PE	5035
CaBr⁺	CaBr	10024-43-8	**	5.6	PE	4860
TiBr₁⁺	TiBr ₁	7789-68-6	**	10.55 (V)	PE	5148
			**	10.59 (V)	PE	4694
C₁₀H₁₀TiBr₂⁺ (η-C ₅ H ₅) ₂ TiBr ₂ (Titanium dibromobis(η ⁵ -2,4-cyclopentadien-1-yl)-)		1293-73-8	**	8.8±0.1 (V)	PE	4987
C₁₀H₄NO₅CrBr⁺ (BrC ₅ H ₄ N)(CO) ₅ Cr (Chromium,(4-bromopyridine)pentacarbonyl-(OC-6-22)-)		64914-27-8	**	7.37 (V)	PE	5566
C₅O₅PCrBr₃⁺ (PBr ₃)(CO) ₅ Cr		22466-06-4	**	8.32 (V)	PE	5539
C₅O₅MnBr⁺ (CO) ₅ MnBr		14516-54-2	**	8.83±0.05 (V)	PE	4492
			**	8.86 (V)	PE	3866
C₆H₃NO₄MnBr⁺ cis-(CO) ₄ (CCH ₃)MnBr		37474-14-9	**	8.26 (V)	PE	3866
C₇H₅O₂PMnBr₃⁺ (C ₅ H ₅)(PBr ₃)(CO) ₂ Mn (Manganese,dicarbonyl(η ⁵ -2,4-cyclopentadien-1-yl)- (phosphorus tribromide)-)		XXXXX-XX-X	**	8.01	EI	5453
C₄O₄FeBr₂⁺ (CO) ₄ FeBr ₂		18475-84-8	**	8.68 (V)	PE	4431
C₇H₅O₂FeBr⁺ C ₅ H ₅ (CO) ₂ FeBr (Iron, bromodicarbonyl (η ⁵ -2,4-cyclopentadien-1-yl)-)		12078-20-5	**	7.93 (V)	PE	4570
			**	7.95 (V)	PE	4565
C₆H₂O₄FeBr₂⁺ trans-C ₂ H ₂ Br ₂ (CO) ₄ Fe		52646-68-1	**	8.74 (V)	PE	4908
C₁₅H₁₈O₆CoBr₃⁺ (C ₅ H ₆ O ₂ Br) ₃ Co (Cobalt, tris(3-bromo-2,4-pentanedionato-0,0')-(OC-6-11)-)		15218-44-7	**	7.58 (V)	PE	4965
Cu₃Br₃⁺ Cu ₃ Br ₃		37190-22-0	**	9.7	EI	3954
			**	9.50±0.02 (V)	PE	4839
Cu₄Br₄⁺ Cu ₄ Br ₄		XXXXX-XX-X		10.4	EI	3954
Cu₄Br₄⁺ Cu ₄ Br ₄		XXXXX-XX-X	**	9.2	EI	3954

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
ZnBr₂⁺						
(³ P _{3/2})	ZnBr ₂	7699-45-8	**	10.89±0.05 (V)	PE	3833
(³ P _{1/2})			**	11.22±0.05 (V)	PE	3833
(³ P ₀)			**	11.40±0.05 (V)	PE	3833
(³ S ₁)			**	12.28±0.05 (V)	PE	3833
(³ S ₀)			**	13.55±0.05 (V)	PE	3833
(³ P _{3/2})			**	10.8 (V)	PE	3963
(³ P _{1/2})			**	10.90 (V)	PE	4232
(³ P ₀)			**	11.1 (V)	PE	3963
(³ P _{1/2})			**	11.2 (V)	PE	3963
(³ P _{1/2})			**	11.285 (V)	PE	4232
(³ P _{1/2})			**	11.4 (V)	PE	3963
(³ P _{3/2})			**	11.46 (V)	PE	4232
(³ P _{1/2})			**	11.625 (V)	PE	4232
(³ S ₀)			**	12.3 (V)	PE	3963
(³ S ₁)			**	12.33 (V)	PE	4232
(³ S ₀)			**	13.0 (V)	PE	3963
(³ S ₁)			**	13.55 (V)	PE	4232
(² D _{5/2})			**	18.89 (V)	PE	4232
(² D _{3/2})			**	19.19 (V)	PE	4232
GaBr₃⁺						
GaBr ₃		13450-88-9	**	10.40	PE	4215
			**	10.94 (V)	PE	4398
			**	10.94 (V)	PE	4256
H₃GeBr⁺						
GeH ₃ Br		13569-43-2	**	10.72±0.05 (V)	PE	3502
H₂GeBr₂⁺						
GeH ₂ Br ₂		13769-36-3	**	10.69±0.02 (V)	PE	3510
C₁₈H₁₅GeBr⁺						
(C ₆ H ₅) ₃ GeBr (Germane, bromotriphenyl-)		3005-32-1	**	9.17±0.05 (V)	PE	4620
AsBr⁺						
AsBr ₃		7784-33-0		12.5±0.2	EI	5016
AsBr₂⁺						
AsBr ₃		7784-33-0	Br ⁻	8.4±0.2	EI	5016
AsBr₃⁺						
AsBr ₃		7784-33-0	**	10.19 (V)	PE	5473
			**	10.21±0.04 (V)	PE	4635
			**	8.7±0.05	EI	5016
SeBr₂⁺						
SeBr ₂		22987-45-7	**	9.07 (V)	PE	5074
			**	9.17±0.05 (V)	PE	5031
Kr⁺						
(³ P _{3/2})	Kr	7439-90-9	**	13.9997±0.00001	S	5162
(³ P _{3/2})			**	14.0010±0.0012	S	3881
(³ P _{1/2})			**	14.6655±0.00002	S	5162
(³ P _{3/2})			**	13.992±0.002	PE	3525
(³ P _{1/2})			**	14.661±0.002	PE	3525
(³ P _{3/2})			**	13.974±0.004	PEN	3541
KrF₂⁺						
KrF ₂		13773-81-4	F+F ⁻	11.517	PI	4998
Kr⁺²						
Kr		7439-90-9	**	38.4±0.2	EI	4503
Kr₂⁺						
Kr ₂		12596-40-6	**	12.86±0.015	PI	4923
(1/2)u			**	13.76±0.02 (V)	PE	4885

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
Kr₂⁺						
(3/2)g	Kr ₂	12596-40-6	**	13.90±0.015 (V)	PE	4885
(1/2)u			**	14.57±0.015 (V)	PE	4885
			**	13.45±0.3	EI	5350
FKr⁺	KrF ₂	13773-81-4	F	~13.38	PI	4998
F₂Kr⁺						
(³ P ₀)	KrF ₂	13773-81-4	**	13.06–13.16	PE	3642
(³ P _{1/2,u})			**	13.34 (V)	PE	3501
(³ P _{1/2,d})			**	13.47 (V)	PE	3501
(³ S ₁)			**	13.75	PE	3642
(³ S ₀)			**	13.90 (V)	PE	3501
(³ P ₂)			**	14.0	PE	3642
(³ P ₁)			**	14.37 (V)	PE	3501
(³ P ₀)			**	16.25	PE	3642
(³ P _{3/2})			**	16.92 (V)	PE	3501
(³ S ₁)			**	17.7 (V)	PE	3501
(³ S ₀)			**	17.7 (V)	PE	3642
(³ P _{1/2})			**	23.0 (V)	PE	3501
ArKr⁺	KrAr	51184-77-1	**	13.425±0.02	PI	4926
			**	14.0±0.2	EI	5350
Rb⁺						
Rb		7440-17-7	**	4.18	PE	4642
RbOH		1310-82-3	OH	~10	EI	3461
RbCl		7791-11-9	Cl	8.695±0.03	PI	3536
(² P _{3/2})			Cl ⁻	21.17±0.04 (V)	PE	5035
(² P _{1/2})				22.00±0.04 (V)	PE	5035
RbBr		7789-39-1	Br	8.12±0.03	PI	3536
(² P _{3/2})			Br ⁻	21.10±0.04 (s)	PE	5035
(² P _{1/2})				21.77±0.04 (s)	PE	5035
RbI		7790-29-6	I	7.53±0.03	PI	3536
(² P _{3/2})			I ⁻	21.33±0.04 (V)	PE	5035
(² P _{1/2})				22.21±0.04 (V)	PE	5035
			I	9.4±0.4	EI	5239
Rb²⁺	Rb ⁺	22537-38-8	**	27.285±0.003	S	3924
			**	27.2898±0.0001	S	5180
NO₃Rb⁺	RbNO ₃	XXXXX-XX-X	**	8.89±0.03 (V)	PE	5354
O₃PRb⁺	RbPO ₃	XXXXX-XX-X	**	9.70±0.04 (V)	PE	4840
ClRb⁺						
RbCl		7791-11-9	**	8.50±0.03	PI	3536
(² P _{3/2})			**	8.26±0.1	PE	4344
(² P _{1/2})			**	8.26±0.1	PE	5035
			**	8.7 (V)	PE	4307
Cl₂Rb₂⁺	(RbCl) ₂	12265-61-1	**	9.30 (V)	PE	5035
			**	9.30 (V)	PE	4344
AlCl₄Rb⁺	RbAlCl ₄	17992-02-8	**	10.39±0.05 (V)	PE	5238

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
BrRb⁺	RbBr	7789-39-1	** ** ** ** **	7.935±0.03 7.75±0.1 7.75±0.1 8.0 (V) 8.62±0.04 (V)	PI PE PE PE PE	3536 4344 5035 4307 5035
(² P _{3/2})						
(² P _{3/2})						
(² P _{1/2})						
BrRb₂⁺	Rb ₂ Br ₂	12409-58-4	Br	8.485±0.05	PI	3536
Sr⁺	Sr	7440-24-6	** ** **	5.5 5.5±0.3 ~5.7	PE EI EI	4860 5067 3486
Sr⁺²	Sr	7440-24-6	**	16	EI	3486
OSr⁺	SrO	1314-11-0	**	6.5±1	EI	4881
ClSr⁺	SrCl	14989-33-4	**	5.10±0.06	EI	3526
Cl₂Sr⁺	SrCl ₂	10476-85-4	**	10.2 (V)	PE	4761
BrSr⁺	SrBr	14519-13-2	**	5.5	PE	4860
Y⁺	Y	7440-65-5	** ** ** ** ** **	6.35±0.10 6.4±0.5 6.45±0.15 6.5±0.5 6.6±0.6 6.9±0.1	EI EI EI EI EI EI	5342 5349 4114 4528 4902 4147
YO		12036-00-9		~13	EI	4147
C₂Y⁺	YC ₂	12071-35-1	**	6.0±1.0	EI	5349
OY⁺	YO	12036-00-9	** **	5.85±0.15 6.0±0.1	EI EI	4114 4147
SY⁺	YS	12210-79-6	** ** **	6.0 6.5±0.5 7.2±0.6	EI EI EI	4001 4528 4902
Zr⁺	Zr	7440-67-7	** ** **	5.8±0.2 6.4±0.1 6.48±0.07	EI EI EI	4483 4114 5342
H₁₆B₁Zr⁺	Zr(BH ₄) ₄	12370-59-1	**	11.6±0.1 (V)	PE	4825
C₂Zr⁺	ZrC ₂	12340-54-4	**	7.5±0.5	EI	4112

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{12}\text{H}_{12}\text{Zr}^+$	$(\text{C}_7\text{H}_7)(\text{C}_5\text{H}_5)\text{Zr}$ (Zirconium, $(\eta^7\text{-cycloheptatrienylum})(\eta^5\text{-2,4-cyclopentadien-1-yl})$)	54006-95-0	**	6.94 ± 0.05 (V)	PE	4428
$\text{C}_{20}\text{H}_{14}\text{Zr}^+$	$((\text{CH}_3)_3\text{CCH}_2)_4\text{Zr}$	38010-72-9	**	8.33 ± 0.1 (V)	PE	4242
$\text{C}_8\text{H}_{21}\text{N}_4\text{Zr}^+$	$(\text{N}(\text{CH}_3)_2)_4\text{Zr}$	XXXXX-XX-X	**	7.23 (V)	PE	4588
$\text{C}_{16}\text{H}_{10}\text{N}_4\text{Zr}^+$	$(\text{N}(\text{C}_2\text{H}_5)_2)_4\text{Zr}$	XXXXX-XX-X	**	6.76 (V)	PE	4588
OZr^+	ZrO	12036-01-0	** **	5.8 ± 0.2 6.2 ± 0.1	EI EI	4483 4114
O_2Zr^+	ZrO ₂	1314-23-4	** **	9.4 ± 0.2 9.55 ± 0.1	EI EI	4483 4114
$\text{C}_{20}\text{H}_{28}\text{O}_6\text{Zr}^+$	$((\text{CH}_3\text{CO})_2\text{CH})_4\text{Zr}$	17501-44-9	**	7.95 (V)	PE	5338
$\text{C}_{16}\text{H}_{14}\text{Si}_2\text{Zr}^+$	$((\text{CH}_3)_3\text{SiCH}_2)_4\text{Zr}$	32665-18-2	**	8.64 ± 0.1 (V)	PE	4242
ClZr^+	ZrCl ₄	10026-11-6		21.9	EI	3783
Cl_2Zr^+	ZrCl ₄	10026-11-6		16.8	EI	3783
Cl_3Zr^+	ZrCl ₄	10026-11-6		12.3	EI	3783
Cl_4Zr^+	ZrCl ₄	10026-11-6	** **	11.94 (V) 10.6	PE EI	4694 3783
$\text{C}_{10}\text{H}_{10}\text{Cl}_2\text{Zr}^+$	$(\eta\text{-C}_5\text{H}_5)_2\text{ZrCl}_2$ (Zirconium, dichlorobis($\eta^5\text{-2,4-cyclopentadien-1-yl}$))	1291-32-3	** **	8.6 ± 0.1 (V) 8.60 ± 0.05 (V)	PE PE	4987 4375
$\text{C}_{20}\text{H}_{30}\text{Cl}_2\text{Zr}^+$	$(\text{C}_5(\text{CH}_3)_2)_2\text{ZrCl}_2$ (Zirconium, dichlorobis[(1,2,3,4,5)- η]-1-(1-ethylpropyl)- 2,4-cyclopentadien-1-yl])	58628-41-4	**	7.55 (V)	PE	5560
Br_4Zr^+	ZrBr ₄ (JC-Mean value of Jahn-Teller components)	13777-25-8	**	10.86 (V)	PE	4694
$\text{C}_{10}\text{H}_{10}\text{Br}_2\text{Zr}^+$	$(\eta\text{-C}_5\text{H}_5)_2\text{ZrBr}_2$ (Zirconium dibromobis($\eta^5\text{-2,4-cyclopentadien-1-yl}$))	1294-67-3	**	8.9 ± 0.1 (V)	PE	4987
Nb^+	Nb	7440-03-1	** **	6.61 ± 0.05 10.1 ± 1.0	EI EI	5342 4900

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₂H₁₂Nb⁺	(C ₅ H ₅)(C ₅ H ₅)Nb (Niobium, (η^5 -cycloheptatrienylum)(η^3 -2,4-cyclopentadien-1-yl)-)	54360-38-2	**	5.98±0.05 (V)	PE	4428
C₁₃H₁₅Nb⁺	(C ₅ H ₅) ₂ (CH ₂ CH=CH ₂)Nb (Niobium, bis(η^5 -2,4-cyclopentadien-1-yl)(η^3 -2-propenyl)-)	39413-65-5	**	5.7±0.1 (V)	PE	4425
C₁₀H₃₀N₅Nb⁺	(N(CH ₃) ₂) ₂ Nb	XXXXX-XX-X	**	6.77 (V)	PE	5036
ClNb⁺	NbCl ₅	10026-12-7		24.2	EI	3783
Cl₂Nb⁺	NbCl ₅	10026-12-7		19.5	EI	3783
Cl₃Nb⁺	NbCl ₅	10026-12-7		14.6	EI	3783
Cl₄Nb⁺	NbCl ₅	10026-12-7		10.7	EI	3783
Cl₅Nb⁺	NbCl ₅	10026-12-7	**	10.97 (s)	PE	4764
C₁₂H₁₄Cl₂Nb⁺	(η -CH ₃ C ₅ H ₅) ₂ NbCl ₂ (Niobium, dichlorobis(η^5 -2,4-cyclopentadien-1-yl)-)	12793-14-5	**	6.4±0.1 (V)	PE	4987
Mo⁺	Mo	7439-98-7	**	7.10	S	4864
			**	7.0±0.3	EI	4864
			**	7.22±0.06	EI	5342
			**	10.5±1.0	EI	4900
	(CO) ₆ Mo	13939-06-5	6CO	18.24±0.06	EI	5291
	((CH ₃) ₂ N) ₂ P(CO) ₅ Mo	14971-43-8		18.4±0.05	EI	3952
	((CH ₃) ₂ N) ₂ P ₂ (CO) ₅ Mo	27342-90-1		15.3±0.05	EI	3952
	CS(CO) ₅ Mo	50358-91-3	5CO+CS	19.12±0.30	EI	5291
Mo₂⁺	Mo ₂	10241-05-1		23.1	EI	3783
C₁₀H₁₂Mo⁺	(C ₅ H ₅) ₂ H ₂ Mo (Molybdenum, bis(η^5 -2,4-cyclopentadien-1-yl)dihydro-)	1291-40-3	**	6.4±0.1 (V)	PE	4425
C₁₂H₁₂Mo⁺	(C ₆ H ₅) ₂ Mo (Molybdenum, bis(η^6 -benzene)-)	12129-68-9	**	5.52±0.05 (V)	PE	4132
	(C ₅ H ₅)(C ₅ H ₅)Mo (Molybdenum, (η^5 -cycloheptatrienylum)(η^3 -2,4-cyclopentadien-1-yl)-)	12301-35-8	**	5.87±0.05 (V)	PE	4428
C₁₂H₁₁Mo⁺	(C ₅ H ₅) ₂ (η -CH ₂ =CH ₂)Mo (Molybdenum, bis(η^5 -2,4-cyclopentadien-1-yl)(η^2 -ethene)-)	37343-05-8	**	6.0±0.1 (V)	PE	4425
C₁₂H₁₆Mo⁺	(C ₅ H ₅) ₂ (CH ₃) ₂ Mo (Molybdenum, bis(η^5 -2,4-cyclopentadien-1-yl)dimethyl-)	39333-52-3	**	6.1±0.1 (V)	PE	4425

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₁H₁₆Mo⁺	(C ₆ H ₅ CH ₃) ₂ Mo (Molybdenum, bis[(1,2,3,4,5,6- η)-methylbenzene]–)	12131-22-5	**	5.32±0.05 (V)	PE	4132
C₁₈H₂₄Mo⁺	(C ₆ H ₃ (CH ₃) ₃) ₂ Mo (Molybdenum, bis[(1,2,3,4,5,6- η)-1,3,5-trimethylbenzene]–)	12131-50-9	**	5.13±0.05 (V)	PE	4132
C₈H₂₄N₄Mo⁺	(N(CH ₃) ₂) ₄ Mo	XXXXX-XX-X	**	5.30 (V)	PE	5036
C₁₆H₄₀N₄Mo⁺	(N(C ₂ H ₅) ₂) ₄ Mo	XXXXX-XX-X	**	5.3 (V)	PE	5036
C₁₂H₃₆N₆Mo₂⁺	(N(CH ₃) ₂) ₆ Mo ₂	51956-20-8	**	6.74 (V)	PE	5565
COMo⁺	(CO) ₆ Mo CS(CO) ₅ Mo	13939-06-5 50358-91-3	5CO 4CO+CS	16.52±0.03 17.54±0.30	EI EI	5291 5291
C₂O₂Mo⁺	(CO) ₆ Mo CS(CO) ₅ Mo	13939-06-5 50358-91-3	4CO 3CO+CS	14.86±0.02 15.82±0.30	EI EI	5291 5291
C₃O₃Mo⁺	(CO) ₆ Mo CS(CO) ₅ Mo	13939-06-5 50358-91-3	3CO 2CO+CS	13.29±0.02 14.05±0.20	EI EI	5291 5291
C₄O₄Mo⁺	(CO) ₆ Mo CS(CO) ₅ Mo	13939-06-5 50358-91-3	2CO CO+CS	11.61±0.02 12.39±0.20	EI EI	5291 5291
C₅O₅Mo⁺	(CO) ₆ Mo CS(CO) ₅ Mo	13939-06-5 50358-91-3	CO CS	10.02±0.02 10.96±0.20	EI EI	5291 5291
C₆O₆Mo⁺	(CO) ₆ Mo	13939-06-5	** ** ** **	8.50±0.02 (V) 8.50 (V) 8.46±0.01 8.50±0.05	PE PE EI EI	3979 4456 5291 4600
C₁₁H₁₀OMo⁺	(C ₅ H ₅) ₂ COMo (Molybdenum, carbonylbis(η^5 -2,4-cyclopentadien-1-yl)–)	12701-85-8	**	5.9±0.1 (V)	PE	4425
C₁₀H₈O₃Mo⁺	C ₇ H ₈ (CO) ₃ Mo (Molybdenum,tricarbonyl[(1,2,3,4,5,6,- η)-1,3,5-cycloheptatriene]–)	12125-77-8 **		7.44 (V) 7.46±0.05 (V)	PE PE	5206 4724
C₁₂H₁₂O₃Mo⁺	(C ₆ H ₅ (CH ₃) ₃)(CO) ₃ Mo (Molybdenum, tricarbonyl[(1,2,3,4,5,6- η)-1,3,5-trimethylbenzene]–)	12089-15-5 **		7.35±0.05 (V) 7.37 (V)	PE PE	4724 5367
C₁₁H₈O₄Mo⁺	(C ₇ H ₈)(CO) ₄ Mo (Molybdenum,[$(2,3,5,6-\eta)$ -bicyclo[2.2.1]hepta-2,5-diene]tetracarbonyl–)	XXXXX-XX-X	**	7.48 (V)	PE	5367

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_4\text{H}_10\text{O}_8\text{Mo}_2^+$	$\text{Mo}_2(\text{O}_2\text{CH})_4$	51329-49-8	** **	7.60 ± 0.05 (V) 7.5 (V)	PE PE	4986 4426
$\text{C}_8\text{H}_{12}\text{O}_8\text{Mo}_2^+$	$\text{Mo}_2(\text{O}_2\text{CCH}_3)_4$	14221-06-8	** **	6.92 ± 0.05 (V) 6.8 (V)	PE PE	4986 4426
$\text{C}_{20}\text{H}_{36}\text{O}_8\text{Mo}_2^+$	$\text{Mo}_2(\text{O}_2\text{CC}(\text{CH}_3)_3)_4$ $\text{Mo}_2(\text{COOC}(\text{CH}_3)_3)_4$ (Molybdenum, tetrakis[μ -(2,2-dimethylpropanoato-0:0')di-(Mo-Mo)])	XXXXX-XX-X	** **	6.75 ± 0.05 (V) 6.7 (V)	PE PE	4986 4426
$\text{C}_{12}\text{H}_{14}\text{N}_2\text{O}_5\text{Mo}^+$	$(\text{C}_3\text{H}_4\text{N}_2(\text{C}_2\text{H}_5)_2)(\text{CO})_5\text{Mo}$	XXXXX-XX-X	**	6.90 (V)	PE	5601
$\text{C}_{21}\text{H}_{24}\text{N}_1\text{O}_4\text{Mo}_2^+$	$(\text{C}_5\text{H}_3\text{N}(\text{O})(\text{CH}_3)_4\text{Mo}_2$ (Molybdenum,tetrakis[μ -(6-methyl-2(1H)-pyridinonato-N ¹ :O ²)di-(Mo-Mo),stereoisomer])	67634-80-4	**	5.89 (V)	PE	5191
$\text{C}_{16}\text{H}_{20}\text{N}_2\text{O}_6\text{Mo}_2^+$	$\text{C}_{16}\text{H}_{20}\text{N}_2\text{O}_6\text{Mo}_2$	XXXXX-XX-X	**	6.24 ± 0.04	PE	5596
FMo^+	MoF	60388-18-3	**	8.0 ± 0.3	EI	4864
F_2Mo^+	MoF ₂ MoF ₃ MoF ₄	20205-60-1 20193-58-2 XXXXX-XX-X	** F	9.00 ± 0.15 14.3 ± 1.0 19.0 ± 1.0	EI EI EI	4864 5424 5424
F_3Mo^+	MoF ₃ MoF ₄	20193-58-2 23412-45-5	** ** F	9.88 ± 0.10 10.2 ± 0.5 14.01 ± 0.5	EI EI EI	4864 5424 5424
F_4Mo^+	MoF ₄	23412-45-5	** **	9.74 ± 0.2 10.11 ± 0.10	EI EI	5424 4864
F_5Mo^+	MoF ₅ MoF ₆	13819-84-6 7783-77-9	** ** F	10.60 ± 0.10 10.81 ± 0.2 15.2 ± 0.2	EI EI EI	4864 5424 4864
F_6Mo^+	MoF ₆	7783-77-9	**	14.5 ± 0.1	PE	4989
OF_3Mo^+	MoOF ₃	22529-29-9	**	11.0 ± 0.5	EI	5434
$\text{C}_8\text{F}_{12}\text{O}_8\text{Mo}_2^+$	$\text{Mo}_2(\text{O}_2\text{CCF}_3)_4$	36608-07-8	**	8.67 ± 0.05 (V)	PE	4986
$\text{O}_4\text{Na}_2\text{Mo}^+$	Na ₂ MoO ₄	XXXXX-XX-X	**	7.2	EI	4578
$\text{C}_{15}\text{H}_{24}\text{O}_6\text{Si}_2\text{Mo}^+$	$\text{C}_{15}\text{H}_{24}\text{O}_6\text{Si}_2\text{Mo}$	XXXXX-XX-X	**	7.27 (V)	PE	5601

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₆H₁₈N₃PMo⁺						
	((CH ₃) ₂ N) ₃ P)(CO) ₅ Mo	14971-43-8	5CO	10.3±0.05	EI	3952
	((CH ₃) ₂ N) ₃ P) ₂ (CO) ₄ Mo	27342-90-1		16.1±0.05	EI	3952
C₁₂H₃₆N₆P₂Mo⁺						
	((CH ₃) ₂ N) ₃ P) ₂ (CO) ₄ Mo	27342-90-1	4CO	14.8±0.05	EI	3952
C₈H₉O₃PMo⁺						
	((CH ₃) ₃ P)(CO) ₅ Mo	16917-96-7	**	7.7	PE	5602
C₁₁H₁₅O₅PMo⁺						
	((C ₂ H ₅) ₃ P)(CO) ₅ Mo	19217-79-9	**	7.7	PE	5602
C₂₃H₁₅O₅PMo⁺						
	(C ₆ H ₅) ₃ (CO) ₅ PMo (Molybdenum, pentacarbonyl(triphenylphosphine)-)	14971-42-7	**	7.70±0.05	EI	4600
C₂₃H₃₃O₅PMo⁺						
	(C ₆ H ₁₁) ₃ P(CO) ₅ Mo (Molybdenum, pentacarbonyl (tricyclohexylphosphine)-(OC-6-22)-)	15603-94-8	**	7.44 (V)	PE	5139
C₈H₉O₈PMo⁺						
	((CH ₃ O) ₃ P)(CO) ₅ Mo	15631-20-6	**	8.1	PE	5602
C₁₁H₁₅O₈PMo⁺						
	((C ₂ H ₅ O) ₃ P)(CO) ₅ Mo	15603-75-5	**	8.0	PE	5602
C₁₀H₃₀O₄P₂Mo⁺						
	C ₁₀ H ₃₀ O ₄ P ₂ Mo (Molybdenum, tetracarbonylbis(triphenylphosphine)-(OC-6-12)-)	16244-53-4	**	7.60±0.05	EI	4600
C₇H₁₈N₃OPMo⁺						
	((CH ₃) ₂ N) ₃ P)(CO) ₅ Mo	14971-43-8	4CO	12.1±0.05	EI	3952
C₈H₁₈N₃O₂PMo⁺						
	((CH ₃) ₂ N) ₃ P)(CO) ₅ Mo	14971-43-8	3CO	9.9±0.05	EI	3952
C₉H₁₈N₃O₃PMo⁺						
	((CH ₃) ₂ N) ₃ P)(CO) ₅ Mo	14971-43-8	2CO	9.6±0.05	EI	3952
C₁₀H₁₈N₃O₄PMo⁺						
	((CH ₃) ₂ N) ₃ P)(CO) ₅ Mo	14971-43-8	CO	7.8±0.05	EI	3952
C₁₁H₁₈N₃O₅PMo⁺						
	((CH ₃) ₂ N) ₃ P)(CO) ₅ Mo	14971-43-8	**	7.8	PE	5602
			**	5.7±0.05	EI	3952
C₁₃H₃₆N₆OP₂Mo⁺						
	((CH ₃) ₂ N) ₃ P) ₂ (CO) ₄ Mo	27342-90-1	3CO	14.0±0.05	EI	3952
C₁₄H₃₆N₆O₂P₂Mo⁺						
	((CH ₃) ₂ N) ₃ P) ₂ (CO) ₄ Mo	27342-90-1	2CO	11.2±0.05	EI	3952
C₁₅H₃₆N₆O₃P₂Mo⁺						
	((CH ₃) ₂ N) ₃ P) ₂ (CO) ₄ Mo	27342-90-1	CO	11.1±0.05	EI	3952
C₁₆H₃₆N₆O₄P₂Mo⁺						
	((CH ₃) ₂ N) ₃ P) ₂ (CO) ₄ Mo	27342-90-1	**	6.8±0.05	EI	3952
F₁₈P₆Mo⁺						
	(PF ₃) ₆ Mo	15339-46-5	**	9.17 (V)	PE	4456

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or ~ appearance potential (eV)	Method	Ref.
C₃H₉N₃F₁₂P₆Mo⁺	(CH ₃ N(PF ₃) ₂) ₃ Mo	63353-75-3	**	7.93 (V)	PE	5376
C₅O₅F₃PMo⁺	(PF ₃)(CO) ₅ Mo	15322-05-1	** **	8.55 (V) 8.8	PE PE	5539 5602
CSMo⁺	CS(CO) ₅ Mo	50358-91-3	5CO	16.07±0.09	EI	5291
C₂OSMo⁺	CS(CO) ₅ Mo	50358-91-3	4CO	14.46±0.05	EI	5291
C₃O₂SMo⁺	CS(CO) ₅ Mo	50358-91-3	3CO	12.39±0.09	EI	5291
C₄O₃SMo⁺	CS(CO) ₅ Mo	50358-91-3	2CO	11.02±0.05	EI	5291
C₅O₄SMo⁺	CS(CO) ₅ Mo	50358-91-3	CO	9.36±0.05	EI	5291
C₆O₅SMo⁺	CS(CO) ₅ Mo	50358-91-3	**	8.18±0.02	EI	5291
ClMo⁺	MoCl ₅	10241-05-1		20.3	EI	3783
Cl₂Mo⁺	MoCl ₅	10241-05-1		17.1	EI	3783
Cl₃Mo⁺	MoCl ₅	10241-05-1		12.9	EI	3783
Cl₄Mo⁺	MoCl ₅	10241-05-1		10.1	EI	3783
Cl₅Mo⁺	MoCl ₅	10241-05-1	** **	9.27 (V) 9.2	PE EI	4764 3783
C₁₂H₁₄Cl₂Mo⁺	(η-CH ₃ C ₅ H ₅) ₂ MoCl ₂ (Molybdenum,dichlorobis[(1,2,3,4,5-η)-1-methyl-2,4-cyclopentadien-1-yl]-)	63374-10-7	**	6.8±0.1 (V)	PE	4987
O₂Cl₂Mo⁺	MoO ₂ Cl ₂	XXXXX-XX-X	**	11.93±0.02	PE	5148
C₅O₅PCl₃Mo⁺	(PCl ₃)(CO) ₅ Mo	19212-18-1	**	8.36 (V)	PE	5539
C₈H₁₂O₈CrMo⁺	CrMo(O ₂ CCH ₃) ₄	XXXXX-XX-X	**	7.06±0.05 (V)	PE	4986
C₂₁H₂₁N₁O₁CrMo⁺	(C ₅ H ₃ N(O)CH ₃) ₂ MoCr (Molybdenum,(chromium)tetrakis[μ-(6-methyl-2(1H)-pyridinonato-N ¹ :O ²)]-(Cr-Mo))	72070-57-6	**	6.0 (V)	PE	5191
C₂₃H₁₅O₃AsMo⁺	(C ₆ H ₅) ₂ (CO) ₅ AsMo (Molybdenum, pentacarbonyl(triphenylarsine)-(OC-6-22-))	19212-22-7	**	7.80±0.05	EI	4600

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₂H₁₁Br₂Mo⁺	(η -CH ₃ C ₅ H ₄) ₂ MoBr ₂ (Molybdenum,dibromobis[(1,2,3,4,5- η)-1-methyl-2,4-cyclopentadien-1-yl]-)	63984-91-8	**	6.9±0.1 (V)	PE	4987
C₅O₅PBr₃Mo⁺	(PBr ₃)(CO) ₅ Mo	22466-07-5	**	8.33 (V)	PE	5539
Ru⁺	Ru (C ₅ H ₅) ₂ Ru (Ruthenocene)	7440-18-8 1287-13-4	** (C ₅ H ₅) ₂	7.16±0.07 16.50±0.25	EI EI	5342 3628
C₃H₃Ru⁺	(C ₅ H ₅) ₂ Ru (Ruthenocene)	1287-13-4		19.6±0.2	EI	3628
C₅H₅Ru⁺	(C ₅ H ₅) ₂ Ru (Ruthenocene)	1287-13-4	C ₅ H ₅	14.75±0.25	EI	3628
C₈H₈Ru⁺	(C ₅ H ₅) ₂ Ru (Ruthenocene)	1287-13-4	C ₂ H ₂	14.6±0.2	EI	3628
C₁₀H₁₀Ru⁺	(C ₅ H ₅) ₂ Ru (Ruthenocene)	1287-13-4	**	7.45 (V)	PE	3688
			**	7.45 (V)	PE	5394
			**	7.50±0.25	EI	3628
C₁₂H₁₄Ru⁺	(C ₅ H ₅ CH ₃) ₂ Ru (Ruthenocene, 1,1'-dimethyl-)	33292-37-4	**	7.25 (V)	PE	3688
O₄Ru⁺	RuO ₄	20427-56-9	** ** ** **	12.09 12.15 12.15 ±0.02 (V) 12.16	PE PE PE PE	3836 4166 5148 3838
C₁₂O₁₂Ru₃⁺	(CO) ₁₂ Ru ₃ (Ruthenium, dodecacarbonyltri-)	15243-33-1	** **	7.7±0.2 (V) 7.91 (V)	PE PE	4882 5189
C₉H₈O₃Ru⁺	(C ₆ H ₈)(CO) ₃ Ru (Ruthenium,tricarbonyl[(1,2,3,4- η)-1,3-cyclohexadiene]-)	12108-25-7	**	8.01 (V)	PE	5551
C₁₀H₁₀O₃Ru⁺	(C ₇ H ₁₀)(CO) ₃ Ru (Ruthenium,tricarbonyl[(1,2,3,4- η)-1,3-cycloheptadiene]-)	41550-67-8	**	7.96 (V)	PE	5551
C₁₅H₃O₆F₁₈Ru⁺	(CF ₃ COCHCOCF ₃) ₃ Ru (Ruthenium, tris(1,1,1,5,5-hexafluoropentanedionato-O,O')-, (OC-6-11)-)	16827-63-7	**	8.85±0.07 (V)	PE	3682
F₁₅P₅Ru⁺	Ru(CO) ₅	19702-30-8	**	9.17 (V)	PE	4456
Rh⁺	Rh	7440-16-6	**	7.1±0.6	EI	4909

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
Rh⁺	Rh	7440-16-6	**	7.42±0.08	EI	5342
Rh₂⁺	Rh ₂	12596-98-4	**	7.1±1.0	EI	4206
CRh⁺	RhC	12127-42-3	** ** ** ** **	7.2±0.5 9.2±1.0 8.1±0.6 8.6±0.4 8.6±0.4	EI EI EI EI EI	4909 5349 3978 4206 5635
C₂Rh⁺	RhC ₂	37306-47-1	**	8.1±0.4	EI	5635
C₇H₇O₁Rh⁺	(CH ₃ COCHCOCH ₃)Rh(CO) ₂ (Dicarbonyl(2,4-pentanedionato)rhodium)	14874-82-9	**	8.6±0.1	EI	3497
C₁₂H₉O₁Rh⁺	(CH ₃ COCHCOC ₆ H ₅)Rh(CO) ₂ (Dicarbonyl(1-phenyl-1,3-butanedionato)rhodium)	24151-55-1	**	8.4±0.1	EI	3497
C₁₇H₁₁O₁Rh⁺	(C ₆ H ₅ COCHCOC ₆ H ₅)Rh(CO) ₂ (Dicarbonyl(1,3-diphenyl-1,3-propanedionato)rhodium)	24151-56-2	**	8.4±0.1	EI	3497
C₁₅H₂₁O₆Rh⁺	(CH ₃ COCHCOCH ₃) ₃ Rh (Tris(2,4-pentanedionato)rhodium)	14284-92-5	** **	7.34±0.01 7.75±0.05	EI EI	3496 3497
C₁₅H₂₀NO₈Rh⁺	((CH ₃ CO) ₂ CH) ₂ Rh(NO ₂) ₂ (OCCH ₃) ₂ (OC-6-22-(3-Nitro-2,4-pentanedionato-O ² ,O')bis(2,4-pentanedionato-O,O')rhodium)	36530-11-7	**	7.65±0.02	EI	3496
C₁₅H₁₉N₂O₁₀Rh⁺	((CH ₃ CO) ₂ CNO ₂) ₂ Rh(CH(OCCH ₃) ₂) (OC-6-21-Bis(3-nitro-2,4-pentanedionato-O ² ,O')(2,4-pentanedionato-O,O')rhodium)	36530-12-8	**	7.97±0.03	EI	3496
C₁₅H₁₈N₃O₁₂Rh⁺	(CH ₃ COC(NO ₂)COCH ₃) ₃ Rh (OC-6-11-Tris(3-nitro-2,4-pentanedionato-O ² ,O')rhodium)	36530-13-9	**	8.39±0.04	EI	3496
C₂₁H₃₆N₁O₄Rh₂⁺	(C ₅ H ₅ N(CH ₃)OH) ₄ Rh ₂ (Rhenium,tetrakis(6-methyl-2-pyridinol)-)	XXXXX-XX-X	**	6.49±0.02	PE	5579
C₇H₁O₁F₃Rh⁺	(CH ₃ COCHCOCF ₃)Rh(CO) ₂ (Dicarbonyl(1,1,1-trifluoro-2,4-pentanedionato)rhodium)	18517-13-0	**	8.85±0.05	EI	3497
C₇HO₁F₆Rh⁺	(CF ₃ COCHCOCF ₃)Rh(CO) ₂ (Dicarbonyl(1,1,1,5,5-hexafluoro-2,4-pentanedionato)rhodium)	18517-12-9	**	9.2±0.1	EI	3497
P₂Rh⁺	RhP ₂	11092-25-4	**	7.7±1.0	EI	4532
HF₁₂P₁Rh⁺	H(PF ₃) ₂ Rh	16842-03-8	**	9.70 (V)	PE	4456

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
HF₁₂P₄Rh⁺	(PF ₃) ₄ RhH	16949-48-7	**	9.7	PE	4021
C₁₂H₃₀O₆P₃S₆Rh⁺	((C ₂ H ₅) ₂ S ₂ PO ₂) ₃ Rh	33991-54-7	**	7.70 (V)	PE	5203
C₁O₁Cl₂Rh₂⁺	((CO) ₂ RhCl) ₂ (CO) ₁ Rh ₂ Cl ₂	14404-25-2 14523-22-9	**	8.89±0.03 (V) 9.01 (V)	PE PE	5255 5327
F₁₂P₄Cl₂Rh₂⁺	(PF ₃) ₄ Rh ₂ Cl ₂	14876-98-3		9.0 (V)	PE	5327
ScRh⁺	RhSc	12166-12-0	**	8.0±1.0	EI	5349
TiRh⁺	RhTi	12600-90-7	**	8.2±1.0	EI	4206
Ti₂Rh⁺	RhTi ₂	12067-05-9	**	7.9±1.0	EI	4206
YRh⁺	RhY	XXXXX-XX-X	**	7.2±1.0	EI	5349
Pd⁺	Pd	7440-05-3	** **	8.0±0.4 8.35±0.05	EI EI	3597 5342
C₆H₁₀Pd⁺	(C ₃ H ₅) ₂ Pd	12240-87-8	** **	7.56 (V) 7.24±0.03	PE PE	5281 3711
C₈H₁₁Pd⁺	(CH ₂ C(CH ₃)CH ₂) ₂ Pd	41348-25-8	**	7.33 (V)	PE	5281
C₃₆H₄₁N₁Pd⁺	((C ₂ H ₅) ₂ C ₁ NCH) ₁ Pd (Palladium,[2,3,7,8,12,13,17,18-octaethyl-21H,23H-porphinato(2-)–N ²¹ ,N ²² ,N ²³ ,N ²⁴]–(SP-4-1)–)	24804-00-0	**	6.37±0.03 (V)	PE	5476
C₁₀H₁₆O₁Pd⁺	((CH ₃ CO) ₂ CH ₂) ₂ Pd	XXXXX-XX-X	**	7.79 (V)	PE	5568
C₂₂H₄₀O₁Pd⁺	((CH ₃ COO) ₂ CH ₂) ₂ Pd	XXXXX-XX-X	**	7.67 (V)	PE	5568
C₁₂H₁₈N₂O₂Pd⁺	C ₁₂ H ₁₈ O ₂ N ₂ Pd (Palladium, [[4,4'-(1,2-ethanediyl)dinitrilo)bis[2-pentanone]](2 ⁻)-N,N',O,O']–(SP-4-2)–)	38337-62-1	**	6.88 (V)	PE	3822
F₁₂P₄Pd⁺	Pd(PF ₃) ₄	13815-33-3	**	9.9±0.1 (V)	PE	4187
C₁₂H₁₈N₂S₂Pd⁺	(CH ₃ C(=S)CH ₂ C(CH ₃)NCH) ₂ Pd	41391-03-1	**	6.70 (V)	PE	5446
C₈H₂₀O₁P₂S₁Pd⁺	Pd(S ₂ P(OC ₂ H ₅) ₂) ₂	21312-72-1	**	7.90±0.05	PE	4636
Ag⁺ (² P ⁰)	Ag	7440-22-4	**	7.576	S	5494

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
Ag^+						
($^1\text{S}_0$)	Ag	7440-22-4	**	7.57	PE	4858
($^3\text{D}_3$)			**	12.43	PE	4858
($^3\text{D}_2$)			**	12.62	PE	4858
($^3\text{D}_1$)			**	12.80	PE	4858
($^1\text{D}_2$)			**	13.28	PE	4858
			**	7.5 ± 0.3	EI	4865
			**	7.51 ± 0.07	EI	3574
			**	7.6	EI	3472
			**	7.62 ± 0.07	EI	5342
			**	7.8 ± 0.2	EI	3609
AgCl		7783-90-6		11.1 ± 0.3	EI	3622
Ag ₂ Cl ₃		12444-97-2		14.5	EI	3622
AgBr		7785-23-1	Br	11.3 ± 0.1	EI	4313
Ag_2^+						
	Ag ₂	12187-06-3	**	6.4 ± 0.7	EI	3440
			**	7.35 ± 0.05	EI	3574
			**	7.4 ± 0.8	EI	3597
			**	8.0 ± 1.0	EI	3609
	Ag ₃ Cl ₃	12444-97-2		18.0 ± 0.5	EI	3622
Ag_3^+						
	Ag ₃ Cl ₃	12444-97-2		18.4 ± 0.5	EI	3605
FAg^+						
	AgF	7775-41-9	**	11.0 ± 0.3	EI	4865
NaAg^+						
	NaAg	38782-42-2	**	7.0 ± 1.5	EI	4919
AlAg^+						
	AgAl	12379-67-8	**	7.8 ± 0.5	EI	3796
$\text{O}_2\text{P}\text{Ag}^+$						
	AgPO ₂	XXXXX-XX-X	**	9.3	EI	4098
ClAg^+						
($^2\Pi_{3/2}$)	AgCl	7783-90-6	**	10.08 (V)	PE	5297
($^2\Pi_{1/2}$)			**	10.14 (V)	PE	5297
($^2\Sigma_{1/2}$)			**	10.62 (V)	PE	5297
($^2\Sigma^+$)			**	11.03 ± 0.1 (V)	PE	4778
($E_{5/2}$)			**	13.50 (V)	PE	5297
($E_{3/2}$)			**	13.68 (V)	PE	5297
($E_{1/2}$)			**	13.80 (V)	PE	5297
($E_{3/2}$)			**	14.15 (V)	PE	5297
($E_{1/2}$)			**	14.26 (V)	PE	5297
			**	10.8 ± 0.4	EI	3622
			**	11.3 ± 0.5	EI	3605
	Ag ₃ Cl ₃	12444-97-2		14.2	EI	3622
ClAg_2^+						
	Ag ₃ Cl ₃	12444-97-2		12.9	EI	3622
Cl_2Ag_2^+						
	Ag ₂ Cl ₂	XXXXX-XX-X	**	10.3 ± 0.5	EI	3605
ClAg_3^+						
	Ag ₃ Cl ₃	12444-97-2		14.9 ± 0.5	EI	3605
Cl_2Ag_3^+						
	Ag ₃ Cl ₃	12444-97-2		11.1 ± 0.3	EI	3622

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
Cl₂Ag₃⁺	Ag ₃ Cl ₃	12444-97-2		11.1±0.5	EI	3605
Cl₃Ag₃⁺	Ag ₃ Cl ₃	12444-97-2	** ** ** ** (AgCl) ₃	11.44±0.04 (V) 10.0±0.5 10.4±0.3 10.4±0.3 67244-69-3 **	PE EI EI EI PE	5297 3605 3622 5330 4839
Cl₃Ag₁⁺	Ag ₁ Cl ₁	XXXXXX-XX-X		10.9±0.5	EI	3605
Cl₁Ag₁⁺	Ag ₁ Cl ₁	XXXXXX-XX-X	**	9.6±1.0	EI	3605
Cl₂Cu₂Ag⁺	AgCu ₂ Cl ₃	XXXXXX-XX-X Cl		12.0±0.4	EI	5330
Cl₃Cu₂Ag⁺	AgCu ₂ Cl ₃	XXXXXX-XX-X	**	10.3±0.3	EI	5330
Cl₂CuAg₂⁺	Ag ₂ CuCl ₃	XXXXXX-XX-X Cl		11.4±0.3	EI	5330
Cl₃CuAg₂⁺	Ag ₂ CuCl ₃	XXXXXX-XX-X	**	10.1±0.3	EI	5330
BrAg⁺	AgBr	7785-23-1	** ** ** ** ** ** ** ** ** ** ** **	9.59 (V) 9.85 (V) 10.47 (V) 11.15±0.1 (V) 13.27 (V) 13.39 (V) 13.485 (V) 13.98 (V) 14.15 (V) 9.1±0.1 9.5±0.3	PE PE PE PE PE PE PE PE PE EI EI	5297 5297 5297 4778 5297 5297 5297 5297 5297 4313 3467
BrAg₂⁺	Ag ₃ Br ₃	11078-33-4		11.9±0.6	EI	4313
Br₂Ag₃⁺	Ag ₃ Br ₂ Ag ₃ Br ₃	11078-32-3 11078-33-4	** Br	10.0±0.2 9.8±0.2	EI EI	3467 4313
Br₃Ag₃⁺	Ag ₃ Br ₃	11078-33-4	** ** ** ** **	9.60 (V) 11.46±0.04 (V) 9.6±0.3 9.6±0.3 9.8±0.2	PE PE EI EI EI	4981 5297 5330 4313 3467
Cl₂BrAg₃⁺	Ag ₃ Cl ₂ Br	XXXXXX-XX-X	**	10.35±0.2	EI	5330
ClBr₂Ag₃⁺	AgClBr ₂	XXXXXX-XX-X	**	9.8±0.3	EI	5330

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
Cd⁺						
(³ S _{1/2})	Cd	7440-43-9	**	8.993	S	5450
(² P _{1/2})			**	8.99	PEN	3537
(² P _{3/2})			**	14.5	PEN	3537
(² D _{5/2})			**	14.9	PEN	3537
(² D _{3/2})			**	17.6	PEN	3537
(² D _{5/2})			**	18.4	PEN	3537
(² D _{3/2})			**	20.2	PEN	3537
			**	9.07±0.07	EI	3745
C₂H₆Cd⁺						
(CH ₃) ₂ Cd		506-82-1	**	8.8 (V)	PE	5300
			**	17.349 (V)	PE	4822
C₄H₁₀Cd⁺						
(C ₂ H ₅) ₂ Cd		592-02-9	**	8.2 (V)	PE	5300
C₆H₁₄Cd⁺						
(n-C ₃ H ₇) ₂ Cd		5905-48-6	**	8.2 (V)	PE	5300
F₂Cd⁺						
CdF ₂		7790-79-6	**	13.18±0.04	PE	5433
C₈H₂₂Si₂Cd⁺						
((CH ₃) ₃ SiCH ₂) ₂ Cd		XXXXXX-XX-X	**	8.8 (V)	PE	5300
Cl₂Cd⁺						
(³ P ₁)	CdCl ₂	10108-64-2	**	11.3 (V)	PE	3963
(² P _{3/2g})			**	11.42 (V)	PE	4232
(² P _{1/2g})			**	11.42 (V)	PE	4232
(² P ₁)			**	11.44±0.05 (V)	PE	3833
(² P ₃)			**	11.8 (V)	PE	3963
(² P _{1/2u})			**	11.92 (V)	PE	4232
(² P _{3/2u})			**	11.92 (V)	PE	4232
(² P ₅)			**	11.93±0.05 (V)	PE	3833
(² S ₁)			**	12.4 (V)	PE	3963
(² S ₃)			**	12.46 (V)	PE	4232
(² S ₅)			**	12.53±0.05 (V)	PE	3833
(² S ₇)			**	13.1 (V)	PE	3963
(² S ₉)			**	13.12±0.05 (V)	PE	3833
(² S ₁₁)			**	13.29 (V)	PE	4232
(² D _{5/2})			**	19.55 (V)	PE	4232
(² D _{3/2})			**	20.27 (V)	PE	4232
Br₂Cd⁺						
(³ P _{3/2g})	CdBr ₂	7789-42-6	**	10.3 (V)	PE	3963
(³ P _{1/2g})			**	10.58±0.05 (V)	PE	3833
(³ P _{3/2u})			**	10.59 (V)	PE	4232
(³ P _{1/2u})			**	10.6 (V)	PE	3963
(² P _{1/2g})			**	10.7 (V)	PE	3963
(² P _{1/2u})			**	10.8 (V)	PE	3963
(² P _{1/2p})			**	10.94±0.05 (V)	PE	3833
(² P _{1/2p})			**	10.965 (V)	PE	4232
(² P ₁)			**	11.15±0.05 (V)	PE	3833
(² P _{1/2u})			**	11.31 (V)	PE	4232
(² S ₁)			**	11.7 (V)	PE	3963
(² S ₃)			**	11.85±0.05 (V)	PE	3833
(² S ₅)			**	11.85 (V)	PE	4232
(² S ₇)			**	12.4 (V)	PE	3963
(² S ₉)			**	12.78±0.05 (V)	PE	3833
(² S ₁₁)			**	12.84 (V)	PE	4232
(² D _{5/2})			**	19.31 (V)	PE	4232

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
Br₂Cd⁺ (² D _{3/2}) CdBr ₂		7789-42-6	**	19.95 (V)	PE	4232
In⁺						
(¹ S ₀)	In	7440-74-6	**	5.78±0.03	PE	5052
(³ P ₀)			**	11.01±0.06	PE	5052
(³ P ₁)			**	11.15±0.04	PE	5052
(¹ P ₁)			**	13.56±0.05	PE	5052
(³ P ₂)			**	24.27±0.03	PE	5052
(³ F ₂)			**	24.88±0.04	PE	5052
(³ P ₀)			**	24.96±0.04	PE	5052
(³ D ₃)			**	25.06±0.03	PE	5052
(¹ P ₁)			**	25.32	PE	5052
(³ D ₁)			**	25.91±0.04	PE	5052
	InBO ₂	XXXXX-XX-X	BO ₂	10.65±0.11	EI	5587
C₅H₅In⁺						
C ₅ H ₅ In (Indium, (η^5 -2,4-cyclopentadien-1-yl)-)		34822-89-4	**	8.28 (V)	PE	4777
			**	8.3±0.1 (V)	PE	4853
OIn₂⁺						
In ₂ O		12030-22-7	**	8.3±0.3	EI	3491
BO₂In⁺						
InBO ₂		XXXXX-XX-X	**	9.65±0.08	EI	5587
C₁₂H₃₀O₆P₃S₆In⁺ ((C ₂ H ₅) ₂ S ₂ PO ₂) _n In		21602-84-6	**	8.3 (V)	PE	5203
ClIn⁺						
(² Σ)	InCl	13465-10-6	**	9.51	PE	3640
(² $\Sigma_{1/2}$)			**	9.71 (V)	PE	4713
(² Π)			**	10.17	PE	3640
(² $\Pi_{3/2}$ + ² $\Pi_{1/2}$)			**	10.85 (V)	PE	4713
(² Σ)			**	12.82	PE	3640
(² $\Sigma_{1/2}$)			**	13.11 (V)	PE	4713
(² D _{5/2})			**	25.30 (V)	PE	5035
(² D _{5/2})			**	25.31 (V)	PE	4713
(² D _{5/2})			**	25.46 (V)	PE	4713
(² D _{3/2})			**	26.20 (V)	PE	5035
(² D _{3/2})			**	26.24 (V)	PE	4713
(² D _{3/2})			**	26.46 (V)	PE	4713
Cl₃In⁺						
InCl ₃		10025-82-8	**	~11.4 (V)	PE	4398
			**	11.45	PE	4215
BrIn⁺						
(² Π)	InBr	14280-53-6	**	6.62	PE	3640
(² Σ)			**	9.09	PE	3640
(² $\Sigma_{1/2}$)			**	9.35 (V)	PE	4713
(² $\Pi_{3/2}$)			**	9.90 (V)	PE	4713
(² $\Pi_{1/2}$)			**	10.13 (V)	PE	4713
(² Σ)			**	12.38	PE	3640
(² $\Sigma_{1/2}$)			**	12.78 (V)	PE	4713
(² D _{5/2})			**	25.19 (V)	PE	4713
(² D _{5/2})			**	25.26 (V)	PE	5035
(² D _{5/2})			**	25.30 (V)	PE	4713
(² D _{3/2})			**	26.07 (V)	PE	4713
(² D _{3/2})			**	26.14 (V)	PE	5035

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
BrIn⁺ $(^3D_{3/2})$ $(^2\Sigma_{1/2})$	InBr	14280-53-6	** **	26.18 (V) 26.40 (V)	PE PE	4713 4713
Br₃In⁺	InBr ₃	13465-09-3	** **	10.3 (V)? 10.32	PE PE	4398 4215
Sn⁺ $(^3P_{1/2})$ $(^3P_{3/2})$	Sn	7440-31-5	** ** **	7.344 7.871 7.28±0.07	S S EI	5496 5496 3745
H₄Sn⁺	SnH ₄	2406-52-2	**	10.75	PE	3716
C₃H₉Sn⁺						
(CH ₃) ₃ Sn	594-27-4	CH ₃	9.58±0.19	EI	3548	
(tert-C ₄ H ₉)(CH ₃) ₃ Sn	3531-47-3	(CH ₃) ₃ C	9.32±0.16	EI	3548	
((CH ₃) ₃ Sn) ₂	661-69-8	(CH ₃) ₃ Sn	9.51±0.22	EI	3548	
((CH ₃) ₃ Si)(CH ₃) ₃ Sn	16393-88-7	(CH ₃) ₃ Si	9.80±0.24	EI	3548	
C ₆ H ₅ SSn(CH ₃) ₃ (Stannane, trimethyl(phenylthio)-)	1007-27-8		9.42±0.1	EI	4198	
C ₅ H ₅ (CO) ₂ CrSn(CH ₃) ₃ (Tricarbonyl(η^5 -2,4-cyclopentadien-1-yl)(trimethylstannylyl)chromium)	31854-87-2		9.09±0.1	EI	3495	
((CH ₃) ₃ Sn)(CO) ₅ Mn	14126-94-4		8.85±0.13	EI	5321	
((CH ₃) ₃ Sn)(CO) ₅ Co	13964-90-4		9.06±0.15	EI	5321	
((CH ₃) ₃ Ge)(CH ₃) ₃ Sn	16393-89-8	(CH ₃) ₃ Ge	9.85±0.22	EI	3548	
(C ₅ H ₅) ₂ Sn(CH ₃) ₃ MoH (Molybdenum,bis(η^5 -2,4-cyclopentadien-1-yl)hydro(trimethylstannylyl)-)	51159-64-9		9.19±0.15	EI	5321	
C ₅ H ₅ (CO) ₂ MoSn(CH ₃) ₃ (Tricarbonyl(η^5 -2,4-cyclopentadien-1-yl)(trimethylstannylyl)molybdenum)	12214-92-5		9.85±0.1	EI	3495	
C ₁₉ H ₂₀ O ₁ SnMo (Molybdenum,bis(η^5 -2,4-cyclopentadien-1-yl)[3-methoxy-1-(methoxycarbonyl)-3-oxo-1-propenyl](trimethylstannylyl)-)	51231-85-7		9.44±0.13	EI	5321	
(C ₅ H ₅) ₂ Sn(CH ₃) ₃ MoCl (Molybdenum,chlorobis(η^5 -2,4-cyclopentadien-1-yl)(trimethylstannylyl)-)	51231-83-5		9.30±0.14	EI	5321	
(C ₅ H ₅) ₂ Sn(CH ₃) ₃ MoBr (Molybdenum,bromobis(η^5 -2,4-cyclopentadien-1-yl)(trimethylstannylyl)-)	51231-84-6		9.36±0.12	EI	5321	
(C ₅ H ₅) ₂ Sn(CH ₃) ₃ MoI (Molybdenum,bis(η^5 -2,4-cyclopentadien-1-yl)iodo(trimethylstannylyl)-)	51249-26-4		9.42±0.15	EI	5321	
(C ₅ H ₅) ₂ Sn(CH ₃) ₃ TaH ₂ (Tantalum,bis(η^5 -2,4-cyclopentadien-1-yl)dihydro(trimethylstannylyl)-)	51192-04-2		9.46±0.11	EI	5321	
(C ₅ H ₅) ₂ Sn(CH ₃) ₃ WH (Tungsten,bis(η^5 -2,4-cyclopentadien-1-yl)hydro(trimethylstannylyl)-)	51192-18-8		9.73±0.12	EI	5321	
C ₅ H ₅ (CO) ₂ WSn(CH ₃) ₃ (Tricarbonyl(η^5 -2,4-cyclopentadien-1-yl)(trimethylstannylyl)tungsten)	12093-29-7		10.05±0.15	EI	3495	
((CH ₃) ₃ Sn)(CO) ₅ Re	15219-90-6		9.59±0.13	EI	5321	
C₃H₁₀Sn⁺						
(CH ₃) ₃ SnH	1631-73-8	**	9.9 (V)	PE	4985	
C₄H₇Sn⁺						
(CH ₃) ₃ SnC≡CH	1112-00-1	CH ₃	9.84±0.08	EI	4126	
C₅H₁₂Sn⁺						
(CH ₃) ₃ Sn	594-27-4	** ** ** ** ** **	8.85±0.1 8.93±0.04 9.7 (V) 9.75 (V) 9.75 (V) 8.76±0.12	PE PE PE PE PE EI	3677 3880 5571 4457 4241 3548	

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₅H₁₂Sn⁺	CH ₂ =CHSn(CH ₃) ₃	754-06-3	**	9.7 (V)	PE	4457
C₅H₁₁Sn⁺	C ₂ H ₅ Sn(CH ₃) ₃	3531-44-0	** **	9.1 (V) 9.1 (V)	PE PE	4457 5571
C₆H₁₁Sn⁺	CH ₂ =CHCH ₂ Sn(CH ₃) ₃	762-73-2	** **	8.50 (V) 8.70 (V)	PE PE	4172 4241
C₆H₁₆Sn⁺	(C ₂ H ₅) ₃ SnH (CH ₃) ₃ (n-C ₃ H ₇)Sn	997-50-2 3531-45-1	** ** **	9.1 (V) 9.1 (V) 8.9 (V)	PE PE PE	4985 5571 4457
	(CH ₃) ₂ (C ₂ H ₅) ₂ Sn <i>iso</i> -C ₃ H ₇ Sn(CH ₃) ₃	4282-05-7 3531-46-2	** ** **	9.01 (V) 8.77 (V) 8.9 (V)	PE PE PE	5571 4457 5571
C₇H₁₆Sn⁺	(CH ₃) ₃ SnCH ₂ CH ₂ CH=CH ₂ C ₃ H ₅ CH ₂ Sn(CH ₃) ₃ (Stannane, (cyclopropylmethyl)trimethyl-)	17314-38-4 51675-53-7	** **	9.71 (V) 8.85 (V)	PE PE	4241 4241
C₇H₁₈Sn⁺	(CH ₃) ₃ (C ₃ H ₇)Sn (C ₂ H ₅) ₃ (CH ₃)Sn (CH ₃) ₃ (<i>iso</i> -C ₃ H ₇)Sn (CH ₃) ₃ (<i>tert</i> -C ₄ H ₉)Sn	1527-99-7 2097-60-1 1118-10-1 3531-47-3	** ** ** ** ** ** **	9.0 (V) 9.52 (V) 8.95 (V) 9.05 (V) 9.33 (V) 8.50 (V) 8.65 (V) 8.34±0.11	PE PE PE PE PE PE PE EI	5571 4241 5571 5571 4241 5571 4457 3548
C₈H₁₈Sn⁺	(CH ₃) ₃ Sn(CH ₂) ₃ CH=CH ₂ C ₅ H ₅ Sn(CH ₃) ₃ (Stannane, cyclopentyltrimethyl-)	34232-11-6 15095-84-8	** **	9.72 (V) 8.72 (V)	PE PE	4241 4457
C₈H₂₀Sn⁺	(C ₂ H ₅) ₄ Sn (CH ₃) ₂ (C ₂ H ₅) ₂ Sn (CH ₃) ₂ (<i>iso</i> -C ₃ H ₇) ₂ Sn	597-64-8 56535-52-5 XXXXX-XX-X	** ** ** **	8.87 (V) 8.93 (V) 9.0 (V) 8.8 (V) 8.56 (V)	PE PE PE PE PE	4457 5571 4985 5571 5571
C₉H₁₁Sn⁺	C ₆ H ₅ (CH ₃) ₃ Sn (Stannane, trimethylphenyl-)	934-56-5	** ** **	8.83±0.05 8.94 (V) ~8.75	PE PE CTS	4589 4280 3922
C₉H₂₀Sn⁺	C ₆ H ₁₁ Sn(CH ₃) ₃ (Stannane, cyclohexyltrimethyl-)	3531-48-4	**	8.57 (V)	PE	4457
C₉H₂₂Sn⁺	(<i>iso</i> -C ₃ H ₇) ₃ SnH	759-23-9	**	8.6 (V)	PE	4985
C₁₀H₁₀Sn⁺	(C ₅ H ₅) ₂ Sn (Stannocene)	1294-75-3	**	7.75±0.05 (V)	PE	4853

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{10}\text{H}_{16}\text{Sn}^+$	$\text{C}_6\text{H}_5\text{CH}_2(\text{CH}_3)_3\text{Sn}$ (Stannane, trimethyl(phenylmethyl)-) $\text{C}_6\text{H}_5\text{CH}_2(\text{CH}_3)_3\text{Sn}$ (Stannane, trimethyl(phenylmethyl)-)	4314-94-7 4314-94-7	** ** ** **	8.08 ± 0.05 8.1 (V) 8.21 (V) 7.91	PE PE PE CTS	4589 4172 4280 3922
$\text{C}_{10}\text{H}_{18}\text{Sn}^+$	$\text{C}_7\text{H}_9\text{Sn}(\text{CH}_3)_3$ (Stannane, bicyclo[2.2.1]hept-2-en-2-yltrimethyl-)	38573-92-1	**	8.45 (V)	PE	4457
$\text{C}_{10}\text{H}_{24}\text{Sn}^+$	$(\text{CH}_3)_2(\text{C}_4\text{H}_9)_2\text{Sn}$ $(\text{CH}_3)_2(\text{tert-C}_4\text{H}_9)_2\text{Sn}$	1528-00-3 35569-11-0	** **	8.8 (V) 8.22 (V)	PE PE	5571 5571
$\text{C}_{12}\text{H}_{16}\text{Sn}^+$	$(\text{C}_6\text{H}_5)(\text{CH}_3)_3\text{Sn}$ (Stannane, 1 <i>H</i> -inden-1-yltrimethyl-)	23022-40-4	**	7.29 ± 0.01	EI	3805
$\text{C}_{12}\text{H}_{18}\text{Sn}^+$	$(\text{C}_6\text{H}_9)(\text{CH}_3)_3\text{Sn}$ (Stannane, (2,3-dihydro-1 <i>H</i> -inden-1-yl)trimethyl-)	41273-55-6	**	7.29 ± 0.01	EI	3805
$\text{C}_{12}\text{H}_{28}\text{Sn}^+$	$(\text{C}_3\text{H}_7)_4\text{Sn}$ $(n\text{-C}_4\text{H}_9)_4\text{SnH}$ $(iso\text{-C}_3\text{H}_7)_4\text{Sn}$	2176-98-9 688-73-3 2949-42-0	** ** **	8.82 (V) 8.8 (V) 8.46 (V)	PE PE PE	5571 4985 5571
$\text{C}_{13}\text{H}_{16}\text{Sn}^+$	$\text{C}_{10}\text{H}_7\text{Sn}(\text{CH}_3)_3$ (Stannane, trimethyl-1-naphthalenyl-)	944-85-4	**	7.99	CTS	3922
$\text{C}_{13}\text{H}_{22}\text{Sn}^+$	$\text{C}_6\text{H}_5\text{CH}_2\text{Sn}(\text{C}_2\text{H}_5)_3$ (Stannane, triethyl(phenylmethyl)-)	18629-74-8	**	7.9 (V)	PE	4172
$\text{C}_{14}\text{H}_{13}\text{Sn}^+$	$\text{C}_{13}\text{H}_{10}\text{Sn}(\text{CH}_3)_2$ (Dibenzo[<i>b,e</i>]stannin, 5,10-dihydro-5,5-dimethyl-)	23708-66-9	CH_3	9.0	EI	4228
$\text{C}_{14}\text{H}_{18}\text{Sn}^+$	$\text{C}_{10}\text{H}_7\text{CH}_2(\text{CH}_3)_3\text{Sn}$ (Stannane, trimethyl(1-naphthalenylmethyl)-)	51220-36-1	**	~7.6	CTS	3922
$\text{C}_{14}\text{H}_{30}\text{Sn}^+$	$(\text{CH}_2=\text{CH})(n\text{-C}_4\text{H}_9)_3\text{Sn}$	7486-35-3	**	8.6 (V)	PE	3850
$\text{C}_{15}\text{H}_{16}\text{Sn}^+$	$\text{C}_{13}\text{H}_{10}\text{Sn}(\text{CH}_3)_2$ (Dibenzo[<i>b,e</i>]stannin, 5,10-dihydro-5,5-dimethyl-)	23708-66-9	**	<8.6	EI	4228
$\text{C}_{15}\text{H}_{32}\text{Sn}^+$	$(\text{CH}_2=\text{CHCH}_3)(n\text{-C}_4\text{H}_9)_3\text{Sn}$	24850-33-7	**	8.4 (V)	PE	3850
$\text{C}_{16}\text{H}_{36}\text{Sn}^+$	$(\text{C}_4\text{H}_9)_4\text{Sn}$ $(sec\text{-C}_4\text{H}_9)_4\text{Sn}$ $(iso\text{-C}_4\text{H}_9)_4\text{Sn}$	1461-25-2 6031-41-0 3531-43-9	** ** **	8.76 (V) 8.7 (V) 8.45 (V) 8.68 (V)	PE PE PE PE	5571 3850 5571 5571

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₈H₁₅Sn⁺						
	((C ₆ H ₅) ₃ Sn)(CO) ₅ Mn (Manganese, pentacarbonyl(triphenylstannyly)-(OC-6-22)-)	14405-84-6		8.38±0.15	EI	5321
				9.00±0.24	EI	4204
	(C ₆ H ₅) ₃ SnFe(CO) ₅ C ₅ H ₅ (Iron, dicarbonyl(η^5 -2,4-cyclopentadien-1-yl)(triphenylstannyl)-)	12132-09-1		9.16±0.21	EI	4204
	((C ₆ H ₅) ₃ Sn)(CO) ₅ Re (Rhenium, pentacarbonyl(triphenylstannyly)-(OC-6-22)-)	15614-21-8		9.16±0.11	EI	5321
C₁₈H₁₆Sn⁺						
	(C ₆ H ₅) ₃ SnH (Stannane, triphenyl-)	892-20-6	**	9.11±0.05 (V)	PE	4620
C₁₉H₃₁Sn⁺						
	C ₆ H ₅ CH ₂ Sn(C ₆ H ₅) ₃ (Stannane, tributyl(phenylmethyl)-)	28493-54-1	**	7.9 (V)	PE	4172
C₂₀H₁₄Sn⁺						
	((CH ₃) ₃ CCH ₂) ₂ Sn	13356-21-3	**	8.58±0.1 (V)	PE	4242
			**	8.67 (V)	PE	5571
C₂₄H₂₀Sn⁺						
	(C ₆ H ₅) ₃ Sn (Stannane, tetraphenyl-)	595-90-4	**	8.34±0.03	PI	4055
C₆H₁₈Sn₂⁺						
	((CH ₃) ₃ Sn) ₂	661-69-8	**	8.02±0.15	EI	3548
C₈H₂₂Sn₂⁺						
	((CH ₃) ₃ Sn) ₂ CHCH ₃ (CH ₃) ₃ Sn(CH ₂) ₂ Sn(CH ₃) ₃	XXXXX-XX-X	**	8.25 (V)	PE	4457
		56580-70-2	**	8.06 (V)	PE	4457
C₉H₂₁Sn₂⁺						
	(CH ₃) ₃ Sn(CH ₂) ₃ Sn(CH ₃) ₃	35434-81-2	**	9.46 (V)	PE	4457
C₁₃H₂₉Sn₂⁺						
	C ₇ H ₁₀ (Sn(CH ₃) ₃) ₂ (Stannane, bicyclo[2.2.1]heptane-2,3-diylibis(trimethyl-, (2- <i>endo</i> , 3- <i>exo</i>)-)	56580-71-3	**	8.0 (V)	PE	4457
C₈H₂₁N₁Sn⁺						
	(N(CH ₃) ₂) ₄ Sn	1066-77-9	**	7.67 (V)	PE	4588
B₂C₇H₂₁N₃Sn⁺						
	N ₃ B ₂ (CH ₃) ₄ (CH ₃) ₃ Sn (1,2,4,3,5-Triazadiborolidine, 1,2,3,5-tetramethyl-4-(trimethylstannyl)-)	53246-13-2	**	7.28 (V)	PE	4526
	N ₃ B ₂ (CH ₃) ₄ (CH ₃) ₃ Sn (1,2,4,3,5-Triazadiborolidine, 1,3,4,5-tetramethyl-2-(trimethylstannyl)-)	53246-19-8	**	7.27 (V)	PE	4526
OSn⁺						
	SnO	21651-19-4	**	9.5±1	EI	3819
C₁₃H₁₁OSn⁺						
	C ₁₂ H ₈ OSn(CH ₃) ₂ (10 <i>H</i> -Phenoxyastannin, 10,10-dimethyl-)	1802-94-4	CH ₃	9.4	EI	4228
			CH ₃	9.40	EI	4228
			CH ₃	8.5±0.1	EI	4664
	C ₂₁ H ₁₆ O ₂ Sn ₂ (CH ₃) ₄ (10 <i>H</i> ,20 <i>H</i> -Tetrabenz[<i>b,e,h,k</i>][1,7,4,10]dioxadistannacyclododecin, 10,10,20,20-tetramethyl-)	51452-88-1		11.05	EI	4228
C₁₁H₁₁OSn⁺						
	C ₁₂ H ₈ OSn(CH ₃) ₂ (10 <i>H</i> -Phenoxyastannin, 10,10-dimethyl-)	1802-94-4	**	8.1	EI	4228

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{11}\text{H}_{11}\text{OSn}^+$	$\text{C}_{12}\text{H}_n\text{OSn}(\text{CH}_3)_2$	1802-94-4	**	8.0 ± 0.1	EI	4664
$\text{C}_{12}\text{H}_{20}\text{O}_1\text{Sn}^+$	$\text{C}_{12}\text{H}_{20}\text{O}_1\text{Sn}$ (Tin,dimethylbis(2,4-pentanedionato-O,O')-(OC-6-21)-)	40866-48-6	**	8.35 (V)	PE	5103
$\text{C}_8\text{H}_{13}\text{NOSn}^+$	$\text{C}_5\text{H}_1\text{N}(\text{O})\text{Sn}(\text{CH}_3)_3$ (Pyridine, 4-(trimethylstannyly)-, 1-oxide)	28867-09-6	**	8.04 (V)	PE	4222
F_2Sn^+	SnF_2	7783-47-3	**	8.0 ± 0.1	PE	5054
$\text{C}_6\text{H}_{14}\text{SiSn}^+$	$((\text{CH}_3)_3\text{Si})(\text{CH}_3)_3\text{Sn}$	16393-88-7	**	8.18 ± 0.14	EI	3548
$\text{C}_{11}\text{H}_{38}\text{Si}_1\text{Sn}^+$	$(\text{CH}(\text{Si}(\text{CH}_3)_3)_2)_2\text{Sn}$	41823-72-7	**	7.42 ± 0.05 (V)	PE	4725
$\text{C}_{16}\text{H}_{44}\text{Si}_1\text{Sn}^+$	$((\text{CH}_3)_3\text{SiCH}_2)_3\text{Sn}$	18547-12-1	**	8.71 ± 0.1 (V)	PE	3830
$\text{C}_{11}\text{H}_{36}\text{N}_2\text{Si}_2\text{Sn}^+$	$\text{C}_{11}\text{H}_{36}\text{N}_2\text{Si}_2\text{Sn}$ $(\text{N}(\text{Si}(\text{CH}_3)_3)(\text{tert}-\text{C}_1\text{H}_9))_2\text{Sn}$	55147-80-3 XXXXX-XX-X	** **	7.26 ± 0.05 (V) 7.25 (V)	PE PE	4725 4157
$\text{C}_{12}\text{H}_{36}\text{N}_2\text{Si}_1\text{Sn}^+$	$(\text{N}(\text{Si}(\text{CH}_3)_3)_2)_2\text{Sn}$	55147-78-9	** **	7.75 ± 0.05 (V) 7.75 ± 0.05 (V)	PE PE	4725 4725
SSn^+ $(^2\Pi_{1/2,3/2})$ $(^2\Sigma, ^2\Pi)$	SnS	1314-95-0	** ** **	9.42 (V) 10.20 (V) 9.7 ± 0.5 (V)	PE PE EI	4967 4550 4550
$\text{C}_4\text{H}_{12}\text{SSn}^+$	$(\text{CH}_3)_3\text{SCH}_3\text{Sn}$	993-46-4	**	8.37 ± 0.05 (V)	PE	4153
$\text{C}_8\text{H}_{11}\text{SSn}^+$	$\text{C}_6\text{H}_5\text{SSn}(\text{CH}_3)_3$ (Stannane, trimethyl(phenylthio)-)	1007-27-8	CH_3	9.22 ± 0.1	EI	4198
$\text{C}_9\text{H}_{11}\text{SSn}^+$	$\text{C}_6\text{H}_5\text{S}(\text{CH}_3)_3\text{Sn}$ (Stannane, trimethyl(phenylthio)-)	1007-27-8	** **	8.40 ± 0.05 (V) 7.87 ± 0.1	PE EI	4589 4198
$\text{C}_{10}\text{H}_{16}\text{SSn}^+$	$\text{C}_6\text{H}_5(\text{SCH}_3)\text{Sn}(\text{CH}_3)_3$ (Stannane, trimethyl[4-(methylthio)phenyl]-) $\text{C}_6\text{H}_5\text{SCH}_2\text{Sn}(\text{CH}_3)_3$ (Stannane, trimethyl[(phenylthio)methyl]-)	17113-79-0 59163-59-6	** **	7.87 ± 0.05 (V) 7.74 ± 0.05 (V)	PE PE	4627 4627
$\text{C}_{11}\text{H}_{18}\text{SSn}^+$	$\text{C}_{11}\text{H}_{18}\text{SSn}$ (Stannane, trimethyl[[4-(methylthio)phenyl]methyl]-)	59163-58-5	**	7.70 ± 0.05 (V)	PE	4627
$\text{C}_{13}\text{H}_{11}\text{SSn}^+$	$\text{C}_{12}\text{H}_n\text{SSn}(\text{CH}_3)_2$ (10 <i>H</i> -Phenothiastannin, 10,10-dimethyl-)	42371-86-8	CH_3 CH_3	9.4 8.9 ± 0.1	EI EI	4228 4664

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₁H₁₁SSn⁺	C ₁₂ H ₈ SSn(CH ₃) ₂ (10 <i>H</i> -Phenothiastannin, 10,10-dimethyl-)	42371-86-8	** **	8.1 8.0±0.1	EI EI	4228 4664
C₄H₁₀S₂Sn⁺	C ₂ H ₅ S ₂ Sn(CH ₃) ₂ (1,3,2-Dithiastannolane,2,2-dimethyl-)	1072-55-5	**	8.70 (V)	PE	5369
C₅H₁₂S₂Sn⁺	C ₂ H ₅ S ₂ Sn(CH ₃) ₃ (1,3,2-Dithiastannolane,2,2,4-trimethyl-)	61235-66-3	**	8.15 (V)	PE	5369
C₇H₁₆S₂Sn⁺	C ₂ H ₅ S ₂ Sn(CH ₃)(C ₂ H ₅) ₂ (1,3,2-Dithiastannolane,2,2-diethyl-4-methyl-)	69032-03-7	**	7.98 (V)	PE	5369
C₁₁H₂₁S₂Sn⁺	C ₂ H ₅ S ₂ Sn(CH ₃)(C ₄ H ₉) ₂ (1,3,2-Dithiastannolane,2,2-dibutyl-4-methyl-)	61235-67-4	**	7.65 (V)	PE	5369
C₄H₈S₄Sn⁺	C ₂ H ₅ S ₂ SnS ₂ C ₂ H ₄ (1,4,6,9-Tetrathia-5-stannaspiro[4.4]nonane)	176-56-7	**	8.77 (V)	PE	5369
C₆H₁₂S₄Sn⁺	C ₂ H ₅ S ₂ (CH ₃)SnS ₂ C ₂ H ₃ (CH ₃) (1,4,6,9-Tetrathia-5-stannaspiro[4.4]nonane,2,7-dimethyl-)	7191-35-7	**	8.70 (V)	PE	5369
C₆H₁₈SSn₂⁺	((CH ₃) ₃ Sn) ₂ S	1070-91-3	** **	8.22±0.05 (V) 9.2±0.1	PE EI	4153 4198
C₆H₁₅NS₂Sn⁺	(CH ₃) ₃ (S ₂ CN(CH ₃) ₂)Sn	33726-89-5	**	7.86 (V)	PE	5569
C₁₁H₂₅NS₂Sn⁺	(C ₂ H ₅) ₃ (S ₂ CN(C ₂ H ₅) ₂)Sn	XXXXXX-XX-X	**	7.46 (V)	PE	5569
C₁₀H₂₄N₂S₄Sn⁺	((S ₂ CN(CH ₃) ₂)(CH ₃) ₂ Sn) ₂	XXXXXX-XX-X	**	7.70 (V)	PE	5569
C₂₂H₄₈N₂S₄Sn⁺	((S ₂ CN(CH ₃) ₂)(C ₄ H ₉) ₂ Sn) ₂	XXXXXX-XX-X	**	8.01 (V)	PE	5569
C₁₃H₁₁O₂SSn⁺	C ₁₂ H ₈ SSn(O) ₂ (CH ₃) ₂ (10 <i>H</i> -Phenothiastannin, 10,10-dimethyl- 5,5-dioxide)	17068-20-1	CH ₃	9.6	EI	4228
C₁₄H₁₁O₂SSn⁺	C ₁₂ H ₈ SSn(O) ₂ (CH ₃) ₂ (10 <i>H</i> -Phenothiastannin, 10,10-dimethyl- 5,5-dioxide)	17068-20-1	**	<9.3	EI	4228
C₅H₁₅PS₂Sn⁺	(CH ₃) ₃ (S ₂ P(CH ₃) ₂)Sn	XXXXXX-XX-X	**	8.60 (V)	PE	5569
C₁₂H₃₂P₂S₄Sn⁺	((CH ₃) ₂ (S ₂ P(C ₂ H ₅) ₂)Sn) ₂	XXXXXX-XX-X	**	8.34 (V)	PE	5569
C₂₄H₅₆P₂S₄Sn⁺	((C ₄ H ₉) ₂ (S ₂ P(C ₂ H ₅) ₂)Sn) ₂	XXXXXX-XX-X	**	8.35 (V)	PE	5569

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
Cl₂Sn⁺						
(² B ₁)	SnCl ₂	7772-99-8	**	7.30 (V)	PE	4837
(² A ₁)			**	10.31±0.05 (V)	PE	4826
(² B ₂)			**	10.37±0.05 (V)	PE	4725
(² A ₂)			**	11.0 (V)	PE	4725
(² B ₂)			**	11.33±0.05 (V)	PE	4725
(² A ₁)			**	12.12±0.05 (V)	PE	4725
(² A ₁)			**	12.77±0.05 (V)	PE	4725
(² D _{3/2})			**	15.90±0.05 (V)	PE	4725
(² D _{3/2})			**	33.48 (V)	PE	5035
			**	34.53 (V)	PE	5035
C₂H₆ClSn⁺						
	((CH ₃) ₂ SnCl)(CO) ₅ Mn	17501-04-1		9.74±0.12	EI	5321
C₃H₉ClSn⁺						
	(CH ₃) ₃ SnCl	1066-45-1	**	9.90	PE	5168
			**	10.16 (V)	PE	4566
C₉H₁₃ClSn⁺						
	C ₆ H ₅ (CH ₃) ₃ SnCl (Stannane, (4-chlorophenyl)trimethyl-)	14064-15-4	**	8.95 (V)	PE	4438
C₁₈H₁₅ClSn⁺						
	(C ₆ H ₅) ₃ SnCl (Stannane, chlorotriphenyl-)	639-58-7	**	9.29±0.05 (V)	PE	4620
C₂H₆Cl₂Sn⁺						
	(CH ₃) ₂ SnCl ₂	753-73-1	**	10.43	PE	5168
C₁₀H₁₁O₂Cl₂Sn⁺						
	C ₁₀ H ₁₁ O ₂ SnCl ₂ (Tin,dichlorobis(2,4-pentanedionato-O,O')-)	16919-46-3	**	9.10 (V)	PE	5103
C₈H₉O₅MnSn⁺						
	Mn(CO) ₅ Sn(CH ₃) ₃	14126-94-4	**	8.63±0.05	PE	4492
			**	8.24±0.11	EI	5321
C₂₃H₁₅O₅MnSn⁺						
	((C ₆ H ₅) ₃ Sn)(CO) ₅ Mn (Manganese,pentacarbonyl(triphenylstannyly)-(OC-6-22)-)	14405-84-6	**	7.94±0.11	EI	5321
C₇H₆O₅ClMnSn⁺						
	((CH ₃) ₂ SnCl)(CO) ₅ Mn	17501-04-1	**	8.21±0.12	EI	5321
C₇H₉O₄CoSn⁺						
	((CH ₃) ₃ Sn)(CO) ₄ Co	13964-90-4	**	8.25	PE	5321
CuSn⁺						
	CuSn	12054-11-4	**	7.2±1.0	EI	5061
Cu₂Sn⁺						
	Cu ₂ Sn	52935-15-6	**	7.7±1.0	EI	5061
C₆H₁₈GeSn⁺						
	(CH ₃) ₃ GeSn(CH ₃) ₃	16393-89-8	**	8.20±0.10	EI	3548
SeSn⁺ (X ² P _{3/2})	SnSe	1315-06-6	**	9.0 (V)	PE	4967
Br₂Sn⁺						
	SnBr ₂	10031-24-0	**	6.84 (V)	PE	4837

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
Br₂Sn⁺						
(² B ₁)	SnBr ₂	10031-24-0	**	9.85±0.05 (V)	PE	4826
(² A ₁)			**	9.87±0.05 (V)	PE	4725
(² Br ₂)			**	10.2 (V)	PE	4725
(² A ₂)			**	10.65±0.05 (V)	PE	4725
(² B ₁)			**	11.35±0.05 (V)	PE	4725
(² A ₁)			**	12.05±0.05 (V)	PE	4725
(² A ₁)			**	15.24±0.05 (V)	PE	4725
(² D _{5/2})			**	33.15 (V)	PE	5035
(² D _{3/2})			**	34.21 (V)	PE	5035
C₃H₉BrSn⁺						
	(CH ₃) ₃ SnBr	1066-44-0	**	9.60 (V)	PE	4566
ClBrSn⁺						
	SnBrCl	13595-90-9	**	10.3±0.3	EI	3800
ClBr₃Sn⁺						
	SnBr ₃ Cl	14779-73-8	**	11.1±0.3	EI	3800
C₁₃H₂₀MoSn⁺						
	(C ₅ H ₅) ₂ (Sn(CH ₃) ₃)MoH	51159-64-9	**	6.48±0.11	EI	5321
	(Molybdenum,bis(η^5 -2,4-cyclopentadien-1-yl)hydro(trimethylstannyl)-)					
C₁₉H₂₆O₄MoSn⁺						
	C ₁₉ H ₂₆ O ₄ SnMo	51231-85-7	**	6.80±0.13	EI	5321
	(Molybdenum,bis(η^5 -2,4-cyclopentadien-1-yl)[3-methoxy-1-(methoxycarbonyl)-3-oxo-1-propenyl]trimethylstannyl)-)					
C₁₃H₁₉ClMoSn⁺						
	(C ₅ H ₅) ₂ (Sn(CH ₃) ₃)MoCl	51231-83-5	**	6.55±0.12	EI	5321
	(Molybdenum,chlorobis(η^5 -2,4-cyclopentadien-1-yl)trimethylstannyl)-)					
C₁₃H₁₉BrMoSn⁺						
	(C ₅ H ₅) ₂ (Sn(CH ₃) ₃)MoBr	51231-84-6	**	6.60±0.13	EI	5321
	(Molybdenum,bromobis(η^5 -2,4-cyclopentadien-1-yl)trimethylstannyl)-)					
Sb⁺						
	Sb	7440-36-0	**	8.3±0.4	EI	4111
			**	8.68±0.06	EI	3956
	Sb ₂	32679-33-7	Sb	11.5±0.5	EI	4111
Sb₂⁺						
	Sb ₂	32679-33-7	**	9.3±0.2	S	3567
			**	8.64±0.06	EI	3956
			**	8.9±0.3	EI	3961
			**	9.5±0.5	EI	3555
Sb₃⁺						
	Sb ₃	37267-70-2	**	7.50±0.13	EI	3956
			**	9.0±0.2	EI	3961
	Sb ₁	12597-17-0	Sb	10.8±0.3	EI	3961
Sb₁⁺						
	Sb ₁	12597-17-0	**	7.70±0.06	EI	3956
			**	8.4±0.3	EI	3961
			**	9.1±0.3	EI	3555
H₃Sb⁺						
	SbH ₃	7803-52-3	**	9.51	PE	3719

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_3\text{H}_9\text{Sb}^+$	$(\text{CH}_3)_3\text{Sb}$	594-10-5	**	8.48 (V)	PE	4226
$\text{C}_5\text{H}_5\text{Sb}^+$	$\text{C}_5\text{H}_5\text{Sb}$ (Antimonin)	289-75-8	**	8.3 (V)	PE	3832
$\text{C}_6\text{H}_5\text{Sb}^+$	$(\text{C}_6\text{H}_5)_3\text{Sb}$ (Stibine, triphenyl-)	603-36-1		8.7 ± 0.1	PI	4325
$\text{C}_{12}\text{H}_{10}\text{Sb}^+$	$(\text{C}_6\text{H}_5)_3\text{Sb}$ (Stibine, triphenyl-)	603-36-1		9.0 ± 0.1	PI	4325
$\text{C}_{18}\text{H}_{15}\text{Sb}^+$	$(\text{C}_6\text{H}_5)_3\text{Sb}$ (Stibine, triphenyl-)	603-36-1	**	7.26 ± 0.05	PI	4325
			**	7.80 ± 0.01	PE	4154
			**	8.08 ± 0.05 (V)	PE	4368
O_6Sb^+	Sb_4O_6	72926-13-7	**	9.31 (V)	PE	5343
F_3Sb^+	SbF_3	7783-56-4	**	12.61 ± 0.1	EI	3578
PSb^+	SbP	25889-81-0	**	9.9 ± 0.3	EI	3596
Cl_3Sb^+	SbCl_3	10025-91-9	**	10.70 (V)	PE	5473
			**	10.73	PE	4146
$\text{C}_{24}\text{H}_{22}\text{MnSb}^+$	$\text{C}_{26}\text{H}_{22}\text{O}_2\text{MnSb}$ (Manganese,dicarbonyl)[(1,2,3,4,5- η)-1-methyl-2,4-cyclopentadien-1-yl](triphenylstibine)-)	XXXXX-XX-X	2CO	8.38 ± 0.03	EI	5576
	$\text{C}_{26}\text{H}_{22}\text{OSMnSb}$ (Manganese,(carbonothioyl)carbonyl)[(1,2,3,4,5- η)-1-methyl-2,4-cyclopentadien-1-yl](triphenylstibine)-)	XXXXX-XX-X	CO+CS	8.83 ± 0.03	EI	5576
$\text{C}_{25}\text{H}_{22}\text{OMnSb}^+$	$(\text{CH}_3\text{C}_5\text{H}_4)(\text{CO})_2((\text{C}_6\text{H}_5)_3\text{Sb})\text{Mn}$ (Manganese,dicarbonyl)[(1,2,3,4,5- η)-1-methyl-2,4-cyclopentadien-1-yl](triphenylstibine)-)	XXXXX-XX-X	CO	8.46 ± 0.04	EI	5576
$\text{C}_{26}\text{H}_{22}\text{O}_2\text{MnSb}^+$	$(\text{CH}_3\text{C}_5\text{H}_4)(\text{CO})_2((\text{C}_6\text{H}_5)_3\text{Sb})\text{Mn}$ (Manganese,dicarbonyl)[(1,2,3,4,5- η)-1-methyl-2,4-cyclopentadien-1-yl](triphenylstibine)-)	XXXXX-XX-X	**	6.37 ± 0.03	EI	5576
$\text{C}_{25}\text{H}_{22}\text{SMnSb}^+$	$\text{C}_{26}\text{H}_{22}\text{OSMnSb}$ (Manganese,(carbonothioyl)carbonyl)[(1,2,3,4,5- η)-1-methyl-2,4-cyclopentadien-1-yl](triphenylstibine)-)	XXXXX-XX-X	CO	7.30 ± 0.04	EI	5576
$\text{C}_{26}\text{H}_{22}\text{OSMnSb}^+$	$\text{C}_{26}\text{H}_{22}\text{OSMnSb}$ (Manganese,(carbonothioyl)carbonyl)[(1,2,3,4,5- η)-1-methyl-2,4-cyclopentadien-1-yl](triphenylstibine)-)	XXXXX-XX-X	**	6.61 ± 0.03	EI	5576

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
SbGa⁺	SbGa	12064-03-8	**	7.6±1.0	EI	4111
Br₃Sb⁺	SbBr ₃	7789-61-9	** **	9.77 (V) 10.07 (V)	PE PE	4146 5473
C₂₃H₁₅O₅MoSb⁺	(C ₆ H ₅) ₃ (CO) ₅ SbMo (Molybdenum, pentacarbonyl(triphenylstibine)-(OC-6-22)-)	19212-21-6	**	7.90±0.05	EI	4600
Te₂⁺	(³ P _{1/2}) Te ₂ (³ P) (³ P _{1/2}) (³ P _{3/2}) (¹ P ₁) (¹ P ₀) (¹ S ₀) (³ S ₁)	10028-16-7	** ** ** ** ** ** ** ** ** ** ** ** ** ** ** **	8.05 8.22 (V) 8.30 (V) 8.77 (V) 9.42 (V) 9.44 (V) 10.10 (V) 11.02 (V) 11.58 (V) 11.87 (V) 12.42 (V)	PE PE PE PE PE PE PE PE PE PE PE PE PE PE PE	5475 4643 4662 4662 4662 4643 4662 4662 4662 4662 4662 4662 4662 4662 4662
Te₃⁺	Te ₃	50645-41-5	**	9.3	EI	5294
Te₄⁺	Te ₄	12597-49-8	**	9.5	EI	5294
Te₅⁺	Te ₅	50645-42-6	**	7.4	EI	5294
Te₆⁺	Te ₆	XXXXX-XX-X	**	7.2	EI	5294
HTe⁺	TeH H ₂ Te	13940-36-8 7783-09-7	** H	9.09 11.9±0.2	S EI	3742 4610
H₂Te⁺	H ₂ Te	7783-09-7	** ** ** ** **	9.14 11.63 13.04 18.6 (V) 9.2±0.1	PE PE PE PE EI	3719 3719 3719 3719 4610
C₂H₆Te⁺	(CH ₃) ₂ Te	593-80-6	** **	7.926±0.010 7.89 (V)	S PE	3970 3656
C₄H₈Te⁺	C ₄ H ₈ Te (Tellurophene)	288-08-4	** ** ** ** **	8.27 8.40±0.03 8.40±0.05 (V) 8.60±0.1 8.32	PE PE PE EI CTS	3858 3804 4626 3804 4382
C₄H₈Te⁺	C ₄ H ₈ Te (Tellurophene, tetrahydro-)	3465-99-4	**	7.73 (V)	PE	4145

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_5\text{H}_6\text{Te}^+$	$\text{C}_5\text{H}_5\text{TeCH}_3$ (Tellurophene, 2-methyl-)	35246-25-4	**	8.20 ± 0.05 (V)	PE	4626
			**	8.25 ± 0.1	EI	3804
			**	8.22	CTS	4382
$\text{C}_7\text{H}_8\text{Te}^+$	$\text{C}_6\text{H}_5\text{TeCH}_3$ (Benzene,(methyltelluro)-)	872-89-9	**	7.6 (V)	PE	5520
$\text{C}_8\text{H}_6\text{Te}^+$	$\text{C}_8\text{H}_6\text{Te}$ (Benzo[b]tellurophene)	272-35-5	**	7.76 ± 0.05	PE	4435
OTe^+						
$(^2\Pi_{1/2})$	TeO	13451-17-7	**	8.72 (V)	PE	4643
$(^2\Pi_{3/2})$			**	9.32 (V)	PE	4643
$(^1\Pi)$			**	10.80 (V)	PE	4643
$(^2\Pi + ^1\Sigma)$			**	11.17 (V)	PE	4643
$(^2\Sigma)$			**	12.00 (V)	PE	4643
$(^2\Pi?)$			**	12.7 (V)	PE	4643
$(^2\Pi)$			**	13.49 (V)	PE	4643
O_2Te^+						
$(^2\text{A}_1 + ^3\text{A}_2 + ^2\text{B}_2)$	TeO ₂	59863-17-1	**	11.17 (V)	PE	4643
			**	12.7 (V)	PE	4643
			**	13.49 (V)	PE	4643
$\text{C}_5\text{H}_4\text{OTe}^+$						
	$\text{C}_5\text{H}_5\text{TeCHO}$ (2-Tellurophenecarboxaldehyde)	35273-64-4	**	8.88 ± 0.1	EI	3804
$\text{C}_6\text{H}_6\text{OTe}^+$						
	$\text{C}_5\text{H}_5\text{TeCOCH}_3$ (Ethanone, 1-tellurophene-2-yl-)	35273-65-5	**	8.60 ± 0.1	EI	3804
$\text{C}_{12}\text{H}_8\text{OTe}^+$						
	$\text{C}_{12}\text{H}_8\text{OTe}$ (Phenoxatellurin)	262-24-8	**	7.61 ± 0.05 (V)	PE	4743
$\text{C}_5\text{H}_4\text{O}_2\text{Te}^+$						
	$\text{C}_5\text{H}_5\text{TeCOOH}$ (2-Tellurophenecarboxylic acid)	35246-22-1	**	8.62 ± 0.05 (V)	PE	4626
			**	8.80 ± 0.1	EI	3804
$\text{C}_6\text{H}_6\text{O}_2\text{Te}^+$						
	$\text{C}_5\text{H}_5\text{TeCOOCH}_3$ (2-Tellurophenecarboxylic acid methyl ester)	35246-23-2	**	8.51 ± 0.05 (V)	PE	4626
			**	8.64 ± 0.1	EI	3804
$\text{C}_7\text{H}_9\text{N}^+\text{OTe}^+$						
	$\text{C}_5\text{H}_5\text{TeCON}(\text{CH}_3)_2$ (2-Tellurophenecarboxamide, N,N-dimethyl-)	55685-52-4	**	8.39 ± 0.05 (V)	PE	4626
$\text{Si}_2\text{H}_6\text{Te}^+$						
	$(\text{SiH}_3)_2\text{Te}$	19415-73-7	**	8.63 (V)	PE	3656
PTe^+						
	TeP	51890-39-2	**	7.8 ± 1.0	EI	4001
$\text{C}_4\text{H}_1\text{STe}^+$						
	$\text{C}_4\text{H}_1\text{STe}$ (1,4-Thiatellurin)	3092-46-4	**	7.9 ± 0.1 (V)	PE	4841

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₅H₆STE⁺	C ₅ H ₃ TeSCH ₃ (Tellurophene, 2-(methylthio)-)	51299-95-7	**	8.15±0.1	EI	3804
C₅H₃ClTe⁺	C ₅ H ₃ TeCl (Tellurophene, 2-chloro-)	59163-66-5	**	8.68±0.05 (V)	PE	4626
GeTe⁺ (X ² P _{3/2})	GeTe	12025-39-7	**	9.1 (V)	PE	4967
H₆Ge₂Te⁺	(GeH ₃) ₂ Te	24312-07-0	**	8.34 (V)	PE	3656
SeTe⁺	SeTe	12067-42-4	**	8.8±0.3	EI	4682
C₄H₃BrTe⁺	C ₄ H ₃ TeBr (Tellurophene, 2-bromo-)	59163-67-6	**	8.59±0.05 (V)	PE	4626
SnTe⁺ (² P _{3/2})	SnTe	12040-02-7	**	8.61 (V)	PE	4967
(² P _{3/2})			**	8.65 (V)	PE	4550
(² P _{1/2})			**	8.95 (V)	PE	4550
(² P ₁)			**	9.39 (V)	PE	4550
I⁺						
(³ P ₂)	I(² P _{3/2})	14362-44-8	**	10.43±0.02	PE	5087
(³ P ₂)			**	10.45	PE	5178
(³ P ₀)			**	11.23±0.02	PE	5087
(³ P ₀)			**	11.25	PE	5178
(³ P ₁)			**	11.30±0.02	PE	5087
(³ P ₁)			**	11.33	PE	5178
(¹ D ₂)			**	12.13±0.02	PE	5087
(¹ D ₂)			**	12.15	PE	5178
			**	10.5	EI	5177
I ₂		7553-56-2	I	13.0	EI	5177
HI		10034-85-2		13.49±0.13	PI	4991
CH ₂ I ₂		75-11-6	CH ₂ I	13.2±0.1	EI	3442
			CH ₂ I	13.8	EI	3490
AgI		7783-96-2	Ag	11.1	EI	4313
I₂⁺						
(² P _{3/2})	I ₂	7553-56-2	**	9.311±0.002	PE	3870
(² P _{1/2})			**	9.953±0.002	PE	3870
			**	9.5	EI	5177
Ag ₃ I ₃		37375-12-5		10.2	EI	4313
WO ₂ I ₂		14447-89-3		15.0±0.8	EI	3451
I₂⁺						
I ₂		7553-56-2	**	25.5±0.4	EI	4052
			**	25.5±0.4	EI	4311
HI⁺						
(² P _{3/2})	HI	10034-85-2	**	10.386±0.001	S	4991
(² P _{1/2})			**	11.0495±0.001	S	4991
DI⁺						
(² P _{3/2})	DI	14104-45-1	**	10.387	S	4991
(² P _{1/2})			**	11.0505±0.001	S	4991

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
LiI⁺	Lil	10377-51-2	**	8.44±0.03 (V)	PE	4950
Li₂I₂⁺	(Lil) ₂	37279-36-0	**	9.23±0.06 (V)	PE	4950
H₈B₃I⁺	B ₅ H ₈ I (Pentaborane(9), 1-iodo-)	30624-33-0	**	9.06 (V)	PE	4519
	B ₅ H ₈ I (Pentaborane(9), 2-iodo-)	20199-87-5	**	9.30 (V)	PE	4519
C₄I₂⁺	Cl≡CC≡Cl	53214-97-4	**	8.76±0.02	PE	4162
CH₂I⁺	CH ₂ I ₂	75-11-6	**	10.55±0.02	PI	4640
CH₃I⁺	CH ₃ I	74-88-4	** ** ** ** ** ** ** ** ** ** **	9.538 9.538 9.533±0.01 9.52 9.53 (V) 9.54 9.9 (V) 10.14 9.48±0.03	S S PI PE PE PE PE PE EI	3748 5245 4640 3532 5249 4194 4193 3532 3626
C₂HI⁺	CH≡Cl	14545-08-5	**	9.7397	S	3751
C₂H₃I⁺	C ₂ H ₃ I	593-66-8	** ** ** ** **	9.296 9.32 (V) 9.33 9.35 (V) 9.32	S PE PE PE PE	5145 4194 3863 4310 4542
C₂H₅I⁺	C ₂ H ₅ I	75-03-6	** ** ** ** ** ** ** ** **	9.346 9.33 (V) 9.34 (V) 9.34 (V) 9.35 9.35 9.45±0.02 (V) 9.6 (V)	S PE PE PE PE PE PE PE	3748 5249 4076 5088 3532 4194 3987 4193
C₃H₃I⁺	CH ₃ C≡Cl	624-66-8	**	9.18±0.02 9.20	PE EI	4765 5282
C₃H₅I⁺	CH ₂ CHCH ₂ I	556-56-9	** ** ** ** ** **	9.298 9.25 (V) 9.30 9.30 (V) 9.32 (V) 9.37	S PE PE PE PE PE	5145 4260 4091 3863 4194 5145

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₃H₇I⁺						
	<i>n</i> -C ₃ H ₇ I	107-08-4	**	9.269	S	3748
			**	9.26	PI	5069
			**	9.25	PE	3532
			**	9.26	PE	4194
			**	9.27	PE	4076
			**	9.5 (V)	PE	4193
	<i>iso</i> -C ₃ H ₇ I	75-30-9	**	9.175	S	5145
			**	9.18	PI	5069
			**	9.18	PE	4194
			**	9.18	PE	5145
			**	9.19	PE	3532
			**	9.4 (V)	PE	4193
			**	9.2±<0.1	EI	3735
C₄HI⁺						
	CH≡CC≡Cl	6088-91-1	**	9.24±0.02	PE	4162
C₄H₉I⁺						
	CH ₂ CHC≡Cl	40589-39-7	**	8.94±0.02	PE	4374
C₄H₉I⁺						
	<i>n</i> -C ₄ H ₉ I	542-69-8	**	9.229	S	3748
			**	9.23	PE	3532
			**	9.23	PE	4194
			**	9.24	PE	4076
			**	9.5 (V)	PE	4193
	<i>sec</i> -C ₄ H ₉ I	513-48-4	**	9.4 (V)	PE	4193
	<i>iso</i> -C ₄ H ₉ I	513-38-2	**	9.202	S	5145
			**	9.20	PE	4194
			**	9.20	PE	5145
			**	9.4 (V)	PE	4193
			**	9.04	PE	5145
			**	9.04	PE	4194
			**	9.08	PE	3532
			**	9.4 (V)	PE	4193
C₅H₉I⁺						
	CH ₃ C≡CC≡Cl	40201-91-0	**	8.82±0.02	PE	4162
C₅H₉I⁺						
	C ₅ H ₉ I (Cyclopentane, iodo-)	1556-18-9	**	9.076	S	5145
			**	9.07	PE	4194
			**	9.07	PE	5145
C₅H₁₁I⁺						
	CH ₂ I-CH ₂ CH(CH ₃) ₂	541-28-6	**	9.192	S	5145
			**	9.20	PE	4194
			**	9.20	PE	5145
	<i>n</i> -C ₅ H ₁₁ I	628-17-1	**	9.201	S	3748
			**	9.20	PE	4194
			**	9.22	PE	3532
			**	9.4 (V)	PE	4193
C₆H₅I⁺						
	C ₆ H ₅ I (Benzene, iodo-)	591-50-4	**	8.67	PE	4194
			**	8.67 (V)	PE	5125
			**	8.70	PE	4621
			**	8.801 (V)	PE	5257
			**	9.05	EI	4834

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_6\text{H}_{11}\text{I}^+$	$\text{C}_6\text{H}_{11}\text{I}$ (Cyclohexane, iodo-)	626-62-0	** ** **	9.003 8.91 8.91	S PE PE	5145 4194 5145
$\text{C}_6\text{H}_{13}\text{I}^+$	$n-\text{C}_6\text{H}_{13}\text{I}$	638-45-9	** **	9.179 9.20	S PE	3748 1494
$\text{C}_7\text{H}_7\text{I}^+$	$\text{C}_6\text{H}_5\text{CH}_2\text{I}$ (Benzene, (iodomethyl)-) $\text{C}_6\text{H}_5\text{ICH}_3$ (Benzene, 1-iodo-2-methyl-) $\text{C}_6\text{H}_5\text{ICH}_3$ (Benzene, 1-iodo-3-methyl-) $\text{C}_6\text{H}_5\text{ICH}_3$ (Benzene, 1-iodo-4-methyl-)	620-05-3 615-37-2 625-95-6 624-31-7	** ** ** **	8.91 (V) 8.53 ± 0.1 8.55 ± 0.1 8.38 8.60 ± 0.1	PE EI EI PE EI	3992 3777 3777 4621 3777
$\text{C}_8\text{H}_5\text{I}^+$	$\text{C}_6\text{H}_5\text{C}\equiv\text{Cl}$ (Benzene, (iodoethynyl)-)	932-88-7	**	8.55 (V)	PE	4334
$\text{C}_{11}\text{H}_9\text{I}^+$	$\text{C}_{11}\text{H}_9\text{I}$ (1,4-Methanonaphthalene, 1,4-dihydro-5-iodo-) $\text{C}_{11}\text{H}_9\text{I}$ (1,4-Methanonaphthalene, 1,4-dihydro-6-iodo-)	63608-69-5 63509-78-4	** **	8.52 ± 0.05 (V) 8.29 ± 0.05 (V)	PE PE	5019 5019
$\text{C}_{12}\text{H}_9\text{I}^+$	$\text{C}_6\text{H}_5\text{C}_6\text{H}_4\text{I}$ (1,1'-Biphenyl, 2-iodo-)	2113-51-1	**	8.20 ± 0.02	PE	3702
CHI_2^+	CHI_3	75-47-8	**	9.77 ± 0.02	PI	4640
CH_2I_2^+	CH_2I_2	75-11-6	**	9.46 ± 0.02	PI	4640
$\text{C}_2\text{H}_2\text{I}_2^+$	<i>cis</i> - $\text{CHI}=\text{CHI}$ <i>trans</i> - $\text{CHI}=\text{CHI}$	590-26-1 590-27-2	** ** **	8.94 (V) 8.92 (V) 8.92 (V)	PE PE PE	4310 4310 3648
$\text{C}_2\text{H}_4\text{I}_2^+$	$\text{CH}_2\text{I}\text{CH}_2\text{I}$	624-73-7	**	9.50 ± 0.02 (V)	PE	4367
$\text{C}_6\text{H}_4\text{I}_2^+$	$\text{C}_6\text{H}_4\text{I}_2$ (Benzene, 1,4-diido-)	624-38-4	**	8.60 (V)	PE	5257
CHI_3^+	CHI_3	75-47-8	** **	9.25 ± 0.02 9.21	PI PE	4640 5198
$\text{BC}_2\text{H}_6\text{I}^+$	$(\text{CH}_3)_2\text{IB}$	17933-09-4	**	9.48 (V)	PE	4398
$\text{B}_4\text{C}_2\text{H}_5\text{I}^+$	$\text{C}_2\text{B}_4\text{H}_5\text{I}$ (1,6-Dicarbahexaborane(6),2-iodo-)	38744-24-0	**	9.16 (V)	PE	5553

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
B₁C₂H₁I₂⁺	C ₂ B ₁ H ₁ I ₂ (1,6-Dicarbahexaborane(6),2,4-diiodo-)	XXXXXX-XX-X **		8.86 (V)	PE	5553
C₃NI⁺	Cl≡CCN	2003-32-9	**	10.18±0.02	PE	4765
C₆H₆NI⁺	C ₆ H ₅ NH ₂ (Benzeneamine, 2-iodo-) C ₆ H ₅ (I)NH ₂ (Benzeneamine, 4-iodo-) C ₆ H ₅ NHCOCH ₃ (Acetamide, N-(2-iodophenyl)-) C ₆ H ₅ NHCOCH ₃ (Acetamide, N-(4-iodophenyl)-)	615-43-0 540-37-4 19591-17-4 622-50-4	** ** CH ₂ =C=O CH ₂ =C=O	8.35 7.51 10.48±0.03 9.72±0.03	EI PE EI EI	4834 4621 3483 3483
C₇H₄NI⁺	C ₆ H ₅ (I)CN (Benzonitrile, 4-iodo-)	3058-39-7	**	9.13	PE	4621
C₇H₁₂NI⁺	C ₇ H ₁₂ NI (1-Azabicyclo[2.2.2]octane, 4-iodo-)	27701-90-2	**	8.35±0.015 (V)	PE	4286
C₁₃H₁₀NI⁺	C ₆ H ₅ IC(=CH ₂)C ₅ H ₄ N (Pyridine,2-[1-(2-iodophenyl)ethenyl]-)	XXXXXX-XX-X **		8.3	OTH	5570
C₉H₁₀N₂I⁺	C ₆ H ₅ (I)N=CHN(CH ₃) ₂ (Methanimidamide, N-(2-iodophenyl)-N,N-dimethyl-)	53666-10-7	H	8.7	EI	4337
C₉H₁₁N₂I⁺	C ₆ H ₅ (I)N=CHN(CH ₃) ₂ (Methanimidamide, N-(2-iodophenyl)-N,N-dimethyl-)	53666-10-7	**	7.3	EI	4337
C₂₅H₂₅N₂I⁺	C ₂₅ H ₂₅ N ₂ I (Quinolinium, 1-ethyl-2-[3-(1-ethyl-2(1H)-quinolinylidene)-1-propenyl]-, iodide)	605-91-4	**	7.25	PI	3586
C₂₉H₃₅N₂I⁺	C ₂₉ H ₃₅ N ₂ I (Quinolinium, 1-(3-methylbutyl)-4-[[1-(3-methylbutyl)-4(1H)-quinolinylidene]methyl]-, iodide)	523-42-2	**	7.35	PI	3586
BC₄H₁₂N₂I⁺	((CH ₃) ₂ N) ₂ BI	7318-71-0	**	8.11 (V)	PE	3704
BC₂H₆NI₂⁺	(CH ₃) ₂ NBI ₂	7318-72-1	**	8.95 (V)	PE	3704
C₂H₅OI⁺	CH ₃ ICH ₂ OH CH ₃ ICH ₂ OH-gauche <i>trans</i> -CH ₂ ICH ₂ OH	624-76-0 XXXXXX-XX-X ** XXXXXX-XX-X **	** 9.66±0.07 (V) 9.73 (V) 9.60 (V)	PE PE PE	3987 5088 5088	
C₃H₇OI⁺	CH ₃ ICH ₂ OCH ₃ CH ₃ ICH ₂ OCH ₃ -gauche <i>trans</i> -CH ₂ ICH ₂ OCH ₃	4296-15-5 XXXXXX-XX-X ** XXXXXX-XX-X **	** 9.43±0.04 (V) 9.43 (V) 9.40 (V)	PE PE PE	3987 5088 5088	

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₆H₅OI⁺						
	C ₆ H ₅ (I)OH (Phenol, 4-iodo-)	540-38-5	**	8.06	PE	4621
	C ₆ H ₅ IOOCCH ₃ (Phenol, 2-iodo-, acetate)	32865-61-5	CH ₂ =C=O	9.72±0.03	EI	3483
	C ₆ H ₅ IOOCCH ₃ (Phenol, 4-iodo-, acetate)	33527-94-5	CH ₂ =C=O	9.38±0.03	EI	3483
C₇H₇OI⁺						
	C ₆ H ₅ IOCH ₃ (Benzene, 1-iodo-4-methoxy-)	696-62-8	**	7.97	PE	4621
C₂H₃O₂I⁺						
	CH ₂ ICOOH	64-69-7	**	11.03 (V)	PE	3874
C₈H₇O₂I⁺						
	C ₆ H ₅ (I)COOCH ₃ (Benzoic acid, 4-iodo-, methyl ester)	619-44-3	**	8.73	PE	4621
	C ₆ H ₅ IOOCCH ₃ (Phenol, 2-iodo-, acetate)	32865-61-5	**	8.25±0.03	EI	3483
	C ₆ H ₅ IOOCCH ₃ (Phenol, 4-iodo-, acetate)	33527-94-5	**	8.20±0.03	EI	3483
C₆H₅OI₂⁺						
	C ₆ H ₅ I ₂ OOCCH ₃ (Phenol, 2,4-diiodo-, acetate)	36914-80-4	CH ₂ =C=O	8.94±0.03	EI	3480
	C ₆ H ₅ I ₂ OOCCH ₃ (Phenol, 2,6-diiodo-, acetate)	28165-73-3	CH ₂ =C=O	9.18±0.03	EI	3480
C₈H₆O₂I₂⁺						
	C ₆ H ₅ I ₂ OOCCH ₃ (Phenol, 2,4-diiodo-, acetate)	36914-80-4	**	7.90±0.03	EI	3480
	C ₆ H ₅ I ₂ OOCCH ₃ (Phenol, 2,6-diiodo-, acetate)	28165-73-3	**	8.07±0.03	EI	3480
CNOI⁺	INCO	3607-48-5	**	9.89±0.01	PE	5001
C₈H₈NOI⁺						
	C ₆ H ₅ INHCOCH ₃ (Acetamide, N-(2-iodophenyl)-)	19591-17-4	**	8.45	EI	4834
			**	7.98±0.03	EI	3483
	C ₆ H ₅ INHCOCH ₃ (Acetamide, N-(4-iodophenyl)-)	622-50-4	**	7.87±0.03	EI	3483
C₁₂H₈NOI⁺						
	C ₆ H ₅ ICOC ₅ H ₄ N (Methanone, (2-iodophenyl)-2-pyridinyl-)	XXXXX-XX-X	**	8.76	EI	5459
C₇H₇N₂OI⁺						
	C ₆ H ₅ INHCONH ₂ (Urea, (2-iodophenyl)-)	13114-93-7	**	8.30	EI	4834
C₆H₅NO₂I⁺						
	C ₆ H ₅ (I)NO ₂ (Benzene, 1-iodo-4-nitro-)	636-98-6	**	9.24	PE	4621
FI⁺						
(² P _{3/2})	IF	13873-84-2	**	10.54±0.01	PE	4755
(² P _{1/2})			**	11.24±0.01	PE	4755
(² S)			**	15.22±0.01 (V)	PE	4755
(² Π)			**	15.94±0.01 (V)	PE	4755

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
F_3I^+	IF_3	7783-66-6	**	12.943 ± 0.005	PE	3655
CF_3I^+	CF_3I	2314-97-8	**	10.45 ± 0.05 (V)	PE	4727
$\text{C}_2\text{F}_3\text{I}^+$	$\text{CF}_3\text{C}\equiv\text{Cl}$	39130-85-3	**	10.17 ± 0.02	PE	4765
$\text{C}_2\text{F}_5\text{I}^+$	$\text{C}_2\text{F}_5\text{I}$	354-64-3	**	10.66 ± 0.1	EI	4862
$\text{C}_6\text{F}_5\text{I}^+$	$\text{C}_6\text{F}_5\text{I}$ (Benzene, pentafluoroiodo-)	827-15-6	**	9.54 (V)	PE	5252
$\text{C}_2\text{F}_4\text{I}_2^+$	$(\text{CF}_2\text{I})_2$	354-65-4	**	10.11 ± 0.01 (V)	PE	4613
$\text{C}_2\text{H}_2\text{F}_3\text{I}^+$	$\text{CF}_3\text{CH}_2\text{I}$	353-83-3	**	9.998	S	5145
NaI^+	NaI	7681-82-5	** ** ** ** ** **	7.64 ± 0.02 7.60 ± 0.1 7.60 ± 0.1 8.0 (V) 9.21 ± 0.04 (V) 9.21 ± 0.04 (V)	PI PE PE PE PE PE	3536 4344 5035 4307 4344 5035
MgI_2^+	MgI_2	10377-58-9	** **	9.57 ± 0.03 10.5 (V)	PI PE	3536 4761
AlI^+	AlI	29977-41-1	**	9.3 ± 0.3	EI	5067
AlI_3^+	AlI_3	7784-23-8	** **	9.66 (V) 9.66 (V)	PE PE	4398 4256
$\text{C}_2\text{H}_6\text{AlI}^+$	$(\text{CH}_3)_2\text{AlI}$	2938-72-9	**	9.48 (V)	PE	4398
$\text{CH}_3\text{AlI}_2^+$	$\text{CH}_3\text{I}_2\text{Al}$	2938-46-7	**	9.73 (V)	PE	4398
$\text{C}_4\text{H}_{12}\text{Al}_2\text{I}_2^+$	$((\text{CH}_3)_2\text{AlI})_2$	59585-02-3	**	9.38 (V)	PE	4559
H_3SiI^+	SiH_3I	13598-42-0	** **	9.78 ± 0.02 (V) 10.05 ± 0.05 (V)	PE PE	3510 3502
H_2SiI_2^+	SiH_2I_2	13760-02-6	**	9.69 ± 0.02 (V)	PE	3510
$\text{C}_5\text{H}_9\text{SiI}^+$	$(\text{CH}_3)_3\text{SiC}\equiv\text{Cl}$	18163-47-8	**	9.1 ± 0.1	PE	4002
PI_3^+	PI_3	13455-01-1	**	9.15 (V)	PE	4023

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
F₂PI⁺	PF ₂ I	13819-11-9	** **	10.1±0.1 (V) 9.6±0.1	PE EI	3662 4305
C₁SI₄⁺	C ₁ SI ₄ (Thiophene, tetraiodo-)	19259-11-1	**	8.27 (V)	PE	4690
C₄H₃SI⁺	C ₄ H ₃ SI (Thiophene, 3-iodo-)	10486-61-0	**	8.46 (V)	PE	4690
C₄H₃SI⁺	C ₄ H ₃ SI (Thiophene, 2-iodo-)	3437-95-4	** **	8.46 (V) 8.52±0.05 (V)	PE PE	4690 4626
C₆H₃S₂I⁺	C ₆ H ₃ S ₂ I (Thieno[2,3- <i>b</i>]thiophene, 2-iodo-) C ₆ H ₃ S ₂ I (Thieno[2,3- <i>b</i>]thiophene, 3-iodo-)	53020-10-3 53020-11-4	** **	8.18 (V) 8.24 (V)	PE PE	5478 5478
C₄H₂SI₂⁺	C ₄ H ₂ SI ₂ (Thiophene, 2,5-diiodo-)	625-88-7	** ** **	8.28 (V) 8.32 8.35	PE EI CTS	4690 3787 3787
	C ₄ H ₂ SI ₂ (Thiophene, 3,4-diiodo-)	19259-08-6	**	8.45 (V)	PE	4690
C₆H₅NSI⁺	C ₆ H ₅ INHCSCH ₃ (Ethanethioamide, N-(2-iodophenyl)-)	39184-84-4	**	8.10	EI	4834
C₆H₅N₂SI⁺	C ₆ H ₅ INHCSNH ₂ (Thiourea, (2-iodophenyl)-)	62635-52-3	**	8.15	EI	4834
ClI⁺ (² P _{3/2})	ICl	7790-99-0	** **	10.088±0.01 10.662±0.01	S S	4027 4027
KI⁺ (² P _{3/2})	KI	7681-11-0	** ** **	7.21±0.1 7.21±0.1 7.4 (V)	PE PE PE	4344 5035 4307
	(² P _{1/2})		**	8.66±0.04 (V)	PE	5035
CaI⁺	CaI	15923-87-2	**	6.1±0.3	EI	5067
CaI₂⁺	CaI ₂	10102-68-8	**	10.1 (V)	PE	4761
TiI₄⁺	TiI ₄ (JC-Mean value of Jahn-Teller components)	7720-83-4	**	9.27 (V)	PE	4694
C₁₀H₁₀I₂Ti⁺	(η-C ₅ H ₅) ₂ TiI ₂ (Titanium,bis(η-2,4-cyclopentadien-1-yl)diiodo-)	12152-92-0	**	8.0±0.1 (V)	PE	4987

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
MnI⁺						
	(C ₅ H ₅)(CS)(NO)MnI (Manganese,(carbonothioyl)(η ⁵ -2,4-cyclopentadien-1-yl)iodonitrosyl-)	58450-74-1		16.15±0.04	EI	5561
	(CH ₃ C ₅ H ₄)(CS)(NO)MnI (Manganese,(carbonothioyl)[(1,2,3,4,5-η)-1-methyl-2,4-cyclopentadien-1-yl]iodonitrosyl-)	XXXXX-XX-X		17.11±0.03	EI	5561
C₅H₅MnI⁺						
	(C ₅ H ₅)(CS)(NO)MnI (Manganese,(carbonothioyl)(η ⁵ -2,4-cyclopentadien-1-yl)iodonitrosyl-)	58450-74-1	NO + CS	10.92±0.03	EI	5561
C₆H₇MnI⁺						
	(CH ₃ C ₅ H ₄)(CS)(NO)MnI (Manganese,(carbonothioyl)[(1,2,3,4,5-η)-1-methyl-2,4-cyclopentadien-1-yl]iodonitrosyl-)	XXXXX-XX-X	NO + CS	10.93±0.02	EI	5561
C₅O₅MnI⁺						
	(CO) ₅ MnI	14879-42-6	**	8.40±0.05 (V)	PE	4492
			**	8.44-8.74 (V)	PE	3866
CSMnI⁺						
	(C ₅ H ₅)(CS)(NO)MnI (Manganese,(carbonothioyl)(η ⁵ -2,4-cyclopentadien-1-yl)iodonitrosyl-)	58450-74-1	NO + C ₅ H ₅	13.97±0.03	EI	5561
	(CH ₃ C ₅ H ₄)(CS)(NO)MnI (Manganese,(carbonothioyl)[(1,2,3,4,5-η)-1-methyl-2,4-cyclopentadien-1-yl]iodonitrosyl-)	XXXXX-XX-X		14.91±0.04	EI	5561
C₆H₅SMnI⁺						
	(C ₅ H ₅)(CS)(NO)MnI (Manganese,(carbonothioyl)(η ⁵ -2,4-cyclopentadien-1-yl)iodonitrosyl-)	58450-74-1	NO	8.81±0.02	EI	5561
C₇H₇SMnI⁺						
	(CH ₃ C ₅ H ₄)(CS)(NO)MnI (Manganese,(carbonothioyl)[(1,2,3,4,5-η)-1-methyl-2,4-cyclopentadien-1-yl]iodonitrosyl-)	XXXXX-XX-X	NO	8.90±0.02	EI	5561
C₆H₅NOSMnI⁺						
	(C ₅ H ₅)(CS)(NO)MnI (Manganese,(carbonothioyl)(η ⁵ -2,4-cyclopentadien-1-yl)iodonitrosyl-)	58450-74-1	**	7.45±0.02	EI	5561
C₇H₇NOSMnI⁺						
	(CH ₃ C ₅ H ₄)(CS)(NO)MnI (Manganese,(carbonothioyl)[(1,2,3,4,5-η)-1-methyl-2,4-cyclopentadien-1-yl]iodonitrosyl-)	XXXXX-XX-X	**	7.35±0.02	EI	5561
C₄O₄FeI₂⁺						
	(CO) ₄ FeI ₂	14911-55-8	**	8.76 (V)	PE	4431
C₇H₅O₂FeI⁺						
	C ₅ H ₅ (CO) ₂ FeI (Iron, dicarbonyl (η ⁵ -2,4-cyclopentadien-1-yl)iodo-)	12078-28-3	**	7.77 (V)	PE	4570
			**	7.81 (V)	PE	4565
Cu₃I₃⁺						
	(CuI) ₃	67244-68-2	**	8.99±0.02 (V)	PE	4839
ZnI₂⁺						
(² Π _{3/2g})	ZnI ₂	10139-47-6	**	9.73±0.05 (V)	PE	3833
(² Π _{3/2g})			**	9.7 (V)	PE	3963
(² Π _{3/2g})			**	9.76 (V)	PE	4232
(² Π _{3/2g})			**	10.2 (V)	PE	3963
(² Π _{1/2g} , ² Π ₁)			**	10.32±0.05 (V)	PE	3833
(² Π _{1/2g})			**	10.32 (V)	PE	4232
(² Π _{1/2g})			**	10.35 (V)	PE	3963

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
ZnI₂⁺						
(³ P _{3/2})	ZnI ₂	10139-47-6	**	10.40 (V)	PE	4232
(³ P _{1/2})			**	10.5 (V)	PE	3963
(³ P _{1/2})			**	10.575 (V)	PE	4232
(³ S _{1/2})			**	11.4 (V)	PE	3963
(³ S _{1/2})			**	11.45±0.05 (V)	PE	3833
(³ S _{1/2})			**	10.32±0.05 (V)	PE	3833
(³ S _{1/2})			**	11.53 (V)	PE	4232
(³ S _{1/2})			**	12.4 (V)	PE	3963
(³ S _{1/2})			**	12.74±0.05 (V)	PE	3833
(³ S _{1/2})			**	12.80 (V)	PE	4232
(² D _{5/2})			**	18.40 (V)	PE	4232
(² D _{3/2})			**	18.71 (V)	PE	4232
GaI⁺						
GaI		15605-68-2	**	9.0±0.3	EI	5067
GaI₃⁺						
GaI ₃		13450-91-4	**	9.40	PE	4215
			**	9.51 (V)	PE	4398
			**	9.51 (V)	PE	4256
GeI₄⁺						
GeI ₄		13450-95-8	**	9.42	PE	5148
H₃GeI⁺						
GeH ₃ I		13573-02-9	**	9.59±0.02 (V)	PE	3510
			**	9.84±0.05 (V)	PE	3502
H₂GeI₂⁺						
GeH ₂ I ₂		14694-31-6	**	12.6±0.1 (V)	PE	3510
AsI₃⁺						
AsI ₃		7784-45-4	**	9.00±0.04 (V)	PE	4635
			**	9.11 (V)	PE	5473
BrI⁺						
(³ P _{3/2})	IBr	7789-33-5	**	9.790±0.004	PE	3870
(³ P _{1/2})			**	10.386±0.004	PE	3870
C₆H₅BrI⁺						
C ₆ H ₅ (I)Br (Benzene, 1-bromo-4-iodo-)		589-87-7	**	8.52	PE	4621
RbI⁺						
RbI		7790-29-6	**	7.308±0.03	PI	3536
(³ P _{3/2})			**	7.12±0.1	PE	4344
(³ P _{3/2})			**	7.12±0.1	PE	5035
(³ P _{1/2})			**	7.3 (V)	PE	4307
(³ P _{1/2})			**	8.48±0.04 (V)	PE	5035
				6.6±0.4	EI	5239
Rb₂I⁺						
Rb ₂ I ₂		12532-37-5	I	7.674	PI	3536
			I	7.2±0.4	EI	5239
SrI⁺						
SrI		14696-99-2	**	5.5±0.3	EI	5067
SrI ₂		10476-86-5	**	9.5±0.3	EI	5067
SrI₂⁺						
SrI ₂		10476-86-5	**	10.0 (V)	PE	4761

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
OF₄Xe⁺	XeOF ₄	13774-85-1	**	>12.0	PE	3943
ArXe⁺	XeAr	58206-67-0	**	11.985±0.017	PI	4926
KrXe⁺	XeKr	12521-42-5	** **	11.757±0.017 12.2±0.2	PI EI	4926 5350
Cs⁺	Cs	7440-46-2	** **	3.89 3.89	PE EI	4642 4352
	CsOH	21351-79-1	OH	~10	EI	3461
	CsNO ₃	XXXXX-XX-X		10.50±0.5	EI	4100
(² P _{3/2})	CsCl	7647-17-8	Cl ⁻	17.46±0.04 (V)	PE	5035
(² P _{1/2})				18.86±0.04 (V)	PE	5035
(² P _{3/2})	CsBr	7787-69-1	Br ⁻	17.52±0.04 (V)	PE	5035
(² P _{1/2})				18.53±0.04 (V)	PE	5035
(² P _{3/2})	CsI	7789-17-5	I ⁻	17.60±0.04 (V)	PE	5035
(² P _{1/2})				19.31±0.04 (V)	PE	5035
Cs²⁺	Cs ⁺	18459-37-5	**	23.14±0.02	S	5179
Cs₂⁺	Cs ₂	12184-83-7	**	3.60–3.71	PI	3772
Cs₂O⁺	Cs ₂ MoO ₄	XXXXX-XX-X	MoO ₃	~12.	EI	4578
NO₃Cs⁺	CsNO ₃	XXXXX-XX-X	**	8.78±0.06 (V)	PE	5354
Cs₂NO₃⁺	(CsNO ₃) ₂	XXXXX-XX-X		14.1±1.0	EI	4100
FCs⁺	CsF	13400-13-0	** ** ** ** **	8.80±0.10 9.0±0.2 9.68±0.05 (V) 9.7 (V) 10.22±0.05 (V)	PE PE PE PE PE	3958 4606 4353 4307 4353
(² Π)						
(² Σ)						
F₄AlCs⁺	CsAlF ₄	39211-00-2	**	13.12±0.05 (V)	PE	5238
O₃PCs⁺	CsPO ₃	XXXXX-XX-X	**	9.41±0.04 (V)	PE	4840
ClCs⁺	CsCl	7647-17-8	** ** ** ** ** ** ** ** ** **	7.84±0.05 7.9±0.2 8.32±0.1 8.32±0.1 8.5 (V) 8.7±0.1 (V) 8.83±0.05 (V) 8.9±0.1 (V) 9.48±0.05 (V)	PE PE PE PE PE PE PE PE PE	3958 4606 4344 5035 4307 4353 4266 4353 4353
(² P _{3/2})						
(² P _{3/2})						
(² Π _{3/2})						
(² Π _{1/2})						
(² Σ)						

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
Cl₂Cs₂⁺	(CsCl) ₂	12258-95-6	** **	9.15 (V) 9.15 (V)	PE PE	5035 4344
AlCl₃Cs⁺	CsAlCl ₃	17992-03-9	**	10.50±0.05 (V)	PE	5238
BrCs⁺	CsBr	7787-69-1	** ** ** ** ** ** ** ** ** ** ** ** ** ** ** ** ** **	7.3±0.2 7.46±0.05 7.74±0.1 7.74±0.1 8.0 (V) 8.47±0.5 (V) 8.57±0.04 (V) 8.88±0.05 (V) 9.21±0.05 (V)	PE PE PE PE PE PE PE PE PE PE PE PE PE PE PE PE PE	4606 3958 4344 5035 4307 4353 5035 4353 4353
O₃MoCs₂⁺	Cs ₂ MoO ₄		XXXXX-XX-X O	~12.	EI	4578
O₄MoCs₂⁺	Cs ₂ MoO ₄		XXXXX-XX-X **	7.	EI	4578
ICs⁺	CsI	7789-17-5	** ** ** ** ** ** ** ** ** ** ** ** ** ** ** ** ** **	6.5±0.2 7.10±0.05 7.10±0.1 7.10±0.1 7.2 (V) 7.46±0.05 (V) 8.00±0.10 8.12±0.05 (V) 8.40±0.04 (V) 8.46±0.05 (V)	PE PE PE PE PE PE PE PE PE PE PE PE PE PE PE PE PE	4606 3958 4344 5035 4307 4353 3958 4353 5035 4353
Ba⁺	Ba	7440-39-3	** ** ** ** ** ** ** ** ** ** BaO	5.1±0.2 5.0 5.22±0.03 5.0±0.3 5.15±0.1 5.17±0.08 ~5.2 10.95±0.18	EI PE PE EI EI EI EI EI	4458 4860 4381 5067 4114 5342 3486 3821
Ba⁺²	Ba	7440-39-3	**	12	EI	3486
OBa⁺	BaO	1304-28-5	** ** ** **	6.5±0.2 6.85±0.1 6.97±0.12 7.±1	EI EI EI EI	4458 5275 3821 4506
BO₂Ba⁺	BaBO ₂	54597-36-3	**	10.8±0.2	EI	5585
ClBa⁺	BaCl	14832-99-6	**	5.0	PE	4860

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
Cl_2Ba^+	BaCl_2	10361-37-2	**	10.0 (V)	PE	4761
BrBa^+	BaBr	14832-97-4	**	5.0	PE	4860
BaI^+	BaI	12524-20-8	**	5.0 ± 0.3	EI	5067
	BaI_2	13718-50-8	**	9.0 ± 0.3	EI	5067
I_2Ba^+	BaI_2	13718-50-8	**	9.7 (V)	PE	4761
La^+	La	7439-91-0	**	5.45 ± 0.2	EI	4114
			**	5.5 ± 0.7	EI	5303
			**	5.51 ± 0.09	EI	5342
			**	5.6 ± 0.1	EI	4560
			**	6.9 ± 1.2	EI	3978
	LaF_3	13709-38-1		26	EI	3456
				26.9	EI	3466
ClLa^+	LaC_2	12071-15-7	C	14.9 ± 0.5	EI	3457
C_2La^+	LaC_2	12071-15-7	**	5.4 ± 0.3	EI	3457
C_3La^+	LaC_3	12602-63-0	**	6.8 ± 0.5	EI	3457
C_4La^+	LaC_4	12603-31-5	**	4.7 ± 0.5	EI	3457
$\text{C}_5\text{H}_5\text{La}^+$	$(\text{C}_5\text{H}_5)_3\text{La}$ (Lanthanum,tris[(1,2,3,4,5- η)-2,4-cyclopentadien-1-yl]-)	1272-23-7	$2\text{C}_5\text{H}_5$	17.3 ± 0.3	EI	5490
$\text{C}_8\text{H}_8\text{La}^+$	$(iso-\text{C}_5\text{H}_7\text{C}_5\text{H}_1)_3\text{La}$ (Lanthanum,tris[(1,2,3,4,5- η)-1-(1-methylethyl)-2,4-cyclopentadien-1-yl]-)	68959-87-5		22.5 ± 0.3	EI	5490
$\text{C}_{10}\text{H}_{10}\text{La}^+$	$(\text{C}_5\text{H}_5)_3\text{La}$ (Lanthanum,tris[(1,2,3,4,5- η)-2,4-cyclopentadien-1-yl]-)	1272-23-7	C_5H_5	10.2 ± 0.3	EI	5490
$\text{C}_{15}\text{H}_{15}\text{La}^+$	$(\text{C}_5\text{H}_5)_3\text{La}$ (Lanthanum,tris[(1,2,3,4,5- η)-2,4-cyclopentadien-1-yl]-)	1272-23-7	**	7.9 ± 0.3	EI	5490
$\text{C}_{16}\text{H}_{22}\text{La}^+$	$(iso-\text{C}_5\text{H}_7\text{C}_5\text{H}_1)_3\text{La}$ (Lanthanum,tris[(1,2,3,4,5- η)-1-(1-methylethyl)-2,4-cyclopentadien-1-yl]-)	68959-87-5	$\text{C}_5\text{H}_7\text{C}_5\text{H}_1$	13.8 ± 0.3	EI	5490
$\text{C}_{21}\text{H}_{33}\text{La}^+$	$(iso-\text{C}_5\text{H}_7\text{C}_5\text{H}_1)_3\text{La}$ (Lanthanum,tris[(1,2,3,4,5- η)-1-(1-methylethyl)-2,4-cyclopentadien-1-yl]-)	68959-87-5	**	8.0 ± 0.3	EI	5490
OLa^+	LaO	12031-20-8	**	4.90 ± 0.1	EI	4560
			**	4.95 ± 0.1	EI	4114
			**	5.2	EI	4119

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
FLa⁺						
	LaF ₃	13709-38-1		16 18.5	EI EI	3456 3466
F₂La⁺						
	LaF ₃	13709-38-1		9 11.8	EI EI	3456 3466
F₅La₂⁺						
	(LaF ₃) ₂	12592-31-3		12.4	EI	3466
RhLa⁺						
	LaRh	12142-68-6	**	7.7±1.0	EI	3978
Ce⁺						
	Ce	7440-45-1	** ** ** ** ** ** ** ** ** ** CeO CeF ₃ CeL ₃	5.5387±0.0004 5.5387±0.0004 5.537±0.0004 5.44±0.1 5.6±0.5 5.7±0.3 5.9±0.4 5.9±0.6 6.0±0.5 6.0±0.5 ~13.5 25.2 16.75±0.15 3I	S S PI EI EI EI EI EI EI EI EI EI	5056 5186 5056 4624 3969 3597 3468 3621 3473 3986 4061 3607 4607 3820
Ce₂⁺						
	Ce ₂	12595-88-9	**	5.9±0.4	EI	3471
C₂Ce⁺						
	C ₂ Ce	12012-32-7	** **	5.6±0.4 6.2±0.5	EI EI	3597 3969
NCe⁺						
	CeN	25764-08-3	**	5.8±0.6	EI	3469
O Ce⁺						
	CeO	12014-74-3	** ** ** **	4.90±0.1 5.2±0.2 5.3±0.5 6.0±0.5	EI EI EI EI	4624 4061 3986 3473
	CeO ₂	1306-38-3		~11	EI	4061
O₂Ce⁺						
	CeO ₂	1306-38-3	** **	9.7±0.5 10.3±0.2	EI EI	3986 4061
O₂Ce₂⁺						
	(CeO) ₂	12258-89-8	**	8±1	EI	3986
F Ce⁺						
	CeF ₃	7758-88-5		17.2	EI	3607
F₂Ce⁺						
	CeF ₃	7758-88-5		13.5	EI	3607
F₃Ce⁺						
	CeF ₃	7758-88-5	**	11.4	EI	3607

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
F₃Ce⁺	Ce ₂ F ₆	37346-47-7		13.1	EI	3607
CSiCe⁺	CSiCe	51257-45-5	**	~9	EI	3969
SCe⁺	CeS	12014-82-3	**	6.0±0.6	EI	3621
S₂Ce⁺	CeS ₂	12133-58-3	**	13.5±1	EI	3621
CRuCe⁺	RuCeC	70378-92-6	**	6.5±1	EI	5331
C₂RuCe⁺	RuCeC ₂	XXXXX-XX-X	**	7.5±0.8	EI	5331
RhCe⁺	CeRh	12157-69-6	**	6.8±1.0	EI	4209
CRhCe⁺	RhCeC	70378-91-5	**	6.±1	EI	5331
C₂RhCe⁺	RhCeC ₂	53262-56-9	**	7.6±0.8	EI	5331
PdCe⁺	CePd	12292-14-7	**	6.2±0.5	EI	3597
ICe⁺	CeI ₃	7790-87-6	I, 2I	13.15±0.15 13.6±0.5	EI EI	4607 3820
ICe⁺²	CeI ₃	7790-87-6		28±1	EI	3820
I₂Ce⁺	CeI ₃	7790-87-6	I I	9.55±0.1 9.7±0.5	EI EI	4607 3820
I₃Ce⁺	CeI ₃	7790-87-6	** **	9.05±0.1 9.6±0.5	EI EI	4607 3820
Pr⁺	Pr	7440-10-0	** ** **	5.464±0.006 5.464 5.37±0.1	PI PI EI	5056 5186 4624
	PrI ₃	13813-23-5	3I	17.0±0.2	EI	3820
C₅H₈Pr⁺	(C ₅ H ₅) ₃ Pr (Praseodymium,tris[(1,2,3,4,5- η)-2,4-cyclopentadien-1-yl]-)	11077-59-1	2C ₅ H ₅	17.0±0.4	EI	5490
C₈H₈Pr⁺	(iso-C ₅ H ₅ C ₅ H ₅) ₃ Pr (Praseodymium,tris[(1,2,3,4,5- η)-1-(1-methylethyl)-2,4-cyclopentadien-1-yl]-)	69021-86-9		22.1±0.3	EI	5490
C₁₀H₁₀Pr⁺	(C ₅ H ₅) ₃ Pr (Praseodymium,tris[(1,2,3,4,5- η)-2,4-cyclopentadien-1-yl]-)	11077-59-1	C ₅ H ₅	10.0±0.2	EI	5490

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₅H₁₅Pr⁺	(C ₅ H ₅) ₃ Pr (Praseodymium,tris[(1,2,3,4,5- η)-2,4-cyclopentadien-1-yl]-)	11077-59-1	**	8.2±0.2	EI	5490
C₁₆H₂₂Pr⁺	(iso-C ₃ H ₇ C ₅ H ₅) ₃ Pr (Praseodymium,tris[(1,2,3,4,5- η)-1-(1-methylethyl)-2,4-cyclopentadien-1-yl]-)	69021-86-9	C ₃ H ₇ C ₅ H ₅	12.4±0.3	EI	5490
C₂₁H₃₃Pr⁺	(iso-C ₃ H ₇ C ₅ H ₅) ₃ Pr (Praseodymium,tris[(1,2,3,4,5- η)-1-(1-methylethyl)-2,4-cyclopentadien-1-yl]-)	69021-86-9	**	8.2±0.3	EI	5490
CNPr⁺	PrCN	57137-34-5	**	5.5±0.5	EI	4505
OPr⁺	PrO	12035-81-3	**	4.90±0.1	EI	4624
IPr⁺	PrI ₃	13813-23-5	2I	12.9±0.2	EI	3820
I₂Pr⁺	PrI ₃	13813-23-5	I	10.0±0.2	EI	3820
I₃Pr⁺	PrI ₃	13813-23-5	**	9.2±0.2	EI	3820
Nd⁺						
	Nd	7440-00-8	**	5.5250±0.0006	S	5056
			**	5.5250±0.0006	S	5186
			**	5.523±0.003	PI	5056
			**	5.49±0.1	EI	4624
			**	6.5	EI	4030
	NdCl ₃	10024-93-8	3Cl?	20.9±1.0	EI	3802
	NdB ₃	13536-80-6		16.9±0.7	EI	3976
	NdI ₃	13813-24-6	3I	15.9±0.2	EI	3820
C₅H₅Nd⁺	(C ₅ H ₅) ₃ Nd (Neodymium,tris[(1,2,3,4,5- η)-2,4-cyclopentadien-1-yl]-)	1273-98-9	2C ₅ H ₅	16.8±0.2	EI	5490
C₈H₈Nd⁺	(iso-C ₃ H ₇ C ₅ H ₅) ₃ Nd (Neodymium,tris[(1,2,3,4,5- η)-1-(1-methylethyl)-2,4-cyclopentadien-1-yl]-)	69021-85-8		18.9±0.3	EI	5490
C₁₀H₁₀Nd⁺	(C ₅ H ₅) ₃ Nd (Neodymium,tris[(1,2,3,4,5- η)-2,4-cyclopentadien-1-yl]-)	1273-98-9	C ₅ H ₅	9.8±0.2	EI	5490
C₁₅H₁₅Nd⁺	(C ₅ H ₅) ₃ Nd (Neodymium,tris[(1,2,3,4,5- η)-2,4-cyclopentadien-1-yl]-)	1273-98-9	**	8.0±0.2	EI	5490
C₁₆H₂₂Nd⁺	(iso-C ₃ H ₇ C ₅ H ₅) ₃ Nd (Neodymium,tris[(1,2,3,4,5- η)-1-(1-methylethyl)-2,4-cyclopentadien-1-yl]-)	69021-85-8	C ₃ H ₇ C ₅ H ₅	10.8±0.3	EI	5490
C₂₁H₃₃Nd⁺	(iso-C ₃ H ₇ C ₅ H ₅) ₃ Nd (Neodymium,tris[(1,2,3,4,5- η)-1-(1-methylethyl)-2,4-cyclopentadien-1-yl]-)	69021-85-8	**	7.9±0.3	EI	5490

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
ONd⁺	NdO	12035-20-0	**	4.97±0.1	EI	4624
ClNd⁺	NdCl ₃	10024-93-8	2Cl?	17.3±1.0	EI	3802
Cl₂Nd⁺	NdCl ₃	10024-93-8	Cl?	11.9±0.3	EI	3802
Cl₃Nd⁺	NdCl ₃	10024-93-8	**	<11.4	EI	3802
Br₂Nd⁺	NdBr ₃	13536-80-6		10.5±0.7	EI	3976
INd⁺	NdI ₃	13813-24-6	2I	13.6±0.5	EI	3820
I₂Nd⁺	NdI ₃	13813-24-6	I	9.3±0.5	EI	3820
I₃Nd⁺	NdI ₃	13813-24-6	**	9.2±0.5	EI	3820
Pm⁺	Pm	7440-12-2	**	5.582±0.01	OTH	5056
			**	5.582±0.010	OTH	5186
Sm⁺	Sm	7440-19-9	**	5.6437±0.0006	S	5186
			**	5.6437±0.001	S	5056
			**	5.639±0.003	PI	5056
			**	5.5	EI	4872
			**	5.58±0.1	EI	4624
	SmI ₂	32248-43-4		12.5	EI	3820
			2I	13.1±0.2	EI	4122
OSm⁺	SmO	12035-88-0	**	5.5	EI	4872
			**	5.55±0.1	EI	4624
ISm⁺	SmI ₂	32248-43-4		9.2	EI	3820
			I	9.8±0.2	EI	4122
I₂Sm⁺	SmI ₂	32248-43-4	**	8.7	EI	3820
			**	9.0±0.2	EI	4122
Eu⁺	Eu	7440-53-1	**	5.6704±0.0003	S	5056
			**	5.6704±0.0003	S	5186
			**	5.67045±0.00002	S	5511
			**	5.666±0.003	PI	5056
			**	5.5	EI	4872
			**	5.6±0.5	EI	3611
			**	5.68±0.1	EI	4624
			**	5.9±0.2	EI	3459
			**	6.1±0.5	EI	4869
	EuI ₂	22015-35-6		12.45±0.2	EI	3612

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
Eu⁺²	Eu ⁺	15065-79-9	**	11.241±0.006	S	4210
Eu₂⁺	Eu ₂	12596-00-8	**	6.3±1.0	EI	4012
C₂Eu⁺	EuC ₂	12127-44-5	**	6.6±0.7	EI	3611
CNEu⁺	EuCN	50647-38-6	**	5.5±1.5	EI	3798
OEu⁺	EuO	12020-60-9	** ** ** **	6.2 6.3±0.2 6.3±0.8 6.48±0.1	EI EI EI EI	4872 5468 4869 4624
OEu₂⁺	Eu ₂ O	62462-47-9	**	6.1±0.9	EI	4869
O₂Eu₂⁺	Eu ₂ O ₂	62462-48-0	**	7.4±1.0	EI	4869
SEu⁺	EuS	12020-65-4	** **	6.8±0.3 6.8±0.3	EI EI	4486 4874
S₂Eu⁺	EuS ₂	55957-42-1	** **	7.2±0.5 7.2±0.5	EI EI	4486 4874
SEu₂⁺	Eu ₂ S	62462-49-1	** **	6.7±0.5 6.7±0.5	EI EI	4486 4874
S₂Eu₂⁺	Eu ₂ S ₂	62462-51-5	** **	6.6±0.5 6.6±0.5	EI EI	4486 4874
AgEu⁺	EuAg	12249-50-2	**	6.1±0.5	EI	4012
IEu⁺	EuI ₂	22015-35-6		9.90±0.2	EI	3612
I₂Eu⁺	EuI ₂	22015-35-6	**	8.85±0.2	EI	3612
Gd⁺	Gd	7440-54-2	** ** ** ** **	6.1502±0.0006 6.1502±0.0006 6.1±0.6 6.24±0.1 6.3±0.6	S S EI EI EI	5056 5186 4902 4624 4869
	GdCl ₃	10138-52-0	3Cl?	20.9±1.0	EI	3802
	GdI ₃	13572-98-0	3I	17.0±0.2	EI	3820
OGd⁺	GdO	12024-77-0	** **	5.75±0.1 6.5±0.8	EI EI	4624 4869

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
O₂Gd⁺	GdO ₂	53789-25-6	**	9.5±1.0	EI	4869
OGd₂⁺	Gd ₂ O	62462-54-8	**	6.5±1.0	EI	4869
O₂Gd₂⁺	Gd ₂ O ₂	62462-55-9	**	8.2±1.0	EI	4869
SGd⁺	GdS	12134-74-6	**	6.9±0.6	EI	4902
ClGd⁺	GdCl ₃	10138-52-0	2Cl?	16.5±1.0	EI	3802
Cl₂Gd⁺	GdCl ₃	10138-52-0	Cl?	11.9±0.3	EI	3802
NaCl₃Gd⁺	NaGdCl ₃	XXXXX-XX-X		10.1±0.5	EI	3802
IGd⁺	GdI ₃	13572-98-0	2I	13.5±0.2	EI	3820
I₂Gd⁺	GdI ₃	13572-98-0	I	10.1±0.2	EI	3820
I₃Gd⁺	GdI ₃	13572-98-0	**	9.2±0.2	EI	3820
Tb⁺	Tb	7440-27-9	**	5.8639±0.0006	S	5056
			**	5.8639±0.0006	S	5186
			**	5.84±0.1	EI	4624
	TbI ₃	13813-40-6	3I	17.6±0.2	EI	3820
OTb⁺	TbO	12035-91-5	**	5.62±0.1	EI	4624
			**	6.1±0.7	EI	4869
OTb₂⁺	Tb ₂ O	62462-71-9	**	6.6±0.8	EI	4869
O₂Tb₂⁺	Tb ₂ O ₂	62462-78-6	**	6.0±0.8	EI	4869
CuTb⁺	TbCu	12019-22-6	**	5.3±0.3	EI	5296
ITb⁺	TbI ₃	13813-40-6	2I	13.7±0.2	EI	3820
I₂Tb⁺	TbI ₃	13813-40-6	I	10.5±0.2	EI	3820
I₃Tb⁺	TbI ₃	13813-40-6	**	9.5±0.2	EI	3820
Dy⁺	Dy	7429-91-6	**	5.9390±0.0006	S	5056
			**	5.9390±0.0006	S	5186
			**	5.936±0.003	PI	5056
			**	5.90±0.1	EI	4624

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
Dy ⁺	DyI ₃	15474-63-2	3I	16.4±0.2	EI	3820
ODy ⁺	DyO	12175-28-9	**	6.08±0.1	EI	4624
CuDy ⁺	DyCu	12018-73-4	**	5.4±0.4	EI	5296
IDy ⁺	DyI ₃	15474-63-2	2I	13.1±0.2	EI	3820
I ₂ Dy ⁺	DyI ₃	15474-63-2	I	10.5±0.2	EI	3820
I ₃ Dy ⁺	DyI ₃	15474-63-2	**	9.6±0.2	EI	3820
Ho ⁺	Ho	7440-60-0	** ** ** ** ** ** **	6.0216±0.0006 6.0216±0.0006 6.017±0.003 5.8±0.2 5.99±0.1 6.1±0.6	S S PI EI EI EI	5056 5186 5056 3440 4624 4869
	HoI ₃	13813-41-7	3I	16.7±0.2	EI	3820
Ho ₂ ⁺	Ho ₂	12596-28-0	**	6.0±1.0	EI	3440
OHo ⁺	HoO	12281-10-6	** **	6.17±0.1 6.2±0.7	EI EI	4624 4869
OHo ₂ ⁺	Ho ₂ O	62462-59-3	**	6.2±0.7	EI	4869
O ₂ Ho ₂ ⁺	Ho ₂ O ₂	62462-60-6	**	7.5±0.1	EI	4869
CuHo ⁺	HoCu	12018-93-8	**	5.3±0.3	EI	5296
AgHo ⁺	HoAg	12002-74-3	**	5.7±0.6	EI	3440
IHo ⁺	HoI ₃	13813-41-7	2I	13.2±0.2	EI	3820
I ₂ Ho ⁺	HoI ₃	13813-41-7	I	10.4±0.2	EI	3820
I ₃ Ho ⁺	HoI ₃	13813-41-7	**	9.2±0.2	EI	3820
Er ⁺	Er	7440-52-0	** ** ** **	6.1077±0.0006 6.1077±0.0010 6.104±0.003 5.93±0.1	S S PI EI	5056 5186 5056 4624
	ErI ₃	13813-42-8	3I	16.2±0.2	EI	3820

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
OEr⁺	ErO	12280-61-4	**	6.30±0.1	EI	4624
IEr⁺	ErI ₃	13813-42-8	2I	13.3±0.2	EI	3820
I₂Er⁺	ErI ₃	13813-42-8	I	10.2±0.2	EI	3820
I₃Er⁺	ErI ₃	13813-42-8	**	9.0±0.2	EI	3820
Tm⁺	Tm	7440-30-4	**	5.7	EI	3460
			**	6.11±0.1	EI	4624
	TmBr ₃	14456-51-0		17.5±0.7	EI	3976
	TmI ₃	13813-43-9	3I	16.1±0.2	EI	4122
OTm⁺	TmO	12281-29-7	**	6.44±0.1	EI	4624
Br₂Tm⁺	TmBr ₃	14456-51-0		11.1±0.7	EI	3976
Br₃Tm⁺	TmBr ₃	14456-51-0	**	9.6±0.7	EI	3976
ITm⁺	TmI ₃	13813-43-9	2I	12.4±0.2	EI	4122
I₂Tm⁺	TmI ₃	13813-43-9	I	10.5±0.2	EI	4122
I₃Tm⁺	TmI ₃	13813-43-9	**	9.2±0.2	EI	4122
Yb⁺	Yb	7440-64-4	**	6.21±0.1	EI	4624
			**	6.3±0.3	EI	4105
	YbCl ₂	13874-77-6		15.05±0.26	EI	3614
	YbBr ₃ ?	13759-89-2		14.7±0.7	EI	3976
Yb⁺²	Yb ⁺	20205-78-1	**	12.184±0.006	S	3974
Yb₂⁺	Yb ₂	12771-79-8	**	4-5	EI	4105
OYb⁺	YbO	25578-79-4	**	6.55±0.1	EI	4624
ClYb⁺	YbCl ₂	13874-77-6		10.70±0.21	EI	3614
Cl₂Yb⁺	YbCl ₂	13874-77-6	**	9.73±0.21	EI	3614
BrYb⁺	YbBr ₃ ?	25502-05-0		10.0±0.7	EI	3976
Br₂Yb⁺	YbBr ₃ ?	13759-89-2		10.0±0.7	EI	3976

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
Lu⁺						
	Lu	7439-94-3	** ** ** **	5.425889±0.00001 S 5.2±0.5 5.28±0.1 5.3±0.3	EI EI EI EI	4060 4869 4624 3618
C₂Lu⁺	LuC ₂	12175-89-2	**	7.8±1	EI	3618
C₄Lu⁺	LuC ₄	37215-84-2	**	11.1±1	EI	3618
OLu⁺	LuO	12032-02-9	** **	6.79±0.1 7.8±0.6	EI EI	4624 4869
OLu₂⁺	Lu ₂ O	12339-78-5	**	6.5±0.7	EI	4869
Hf⁺	Hf	7440-58-6	** **	6.65±0.1 6.65±0.10	EI EI	4114 5342
Hf₁₆B₄Hf⁺	Hf(BH ₄) ₄ Hf(BH ₃) ₄	37274-93-4 53608-70-1	** **	11.6±0.1 (V) 11.6±0.1 (V)	PE PE	4825 4888
C₂₀H₁₄Hf⁺	((CH ₃) ₃ CCH ₂) ₄ Hf	50654-35-8	**	8.51±0.1 (V)	PE	4242
NHf⁺	HfN	25817-87-2	**	<10	EI	4207
C₈H₂₁N₄Hf⁺	(N(CH ₃) ₂) ₄ Hf	XXXXXX-XX-X	**	7.50 (V)	PE	4588
C₁₆H₁₀N₄Hf⁺	(N(C ₂ H ₅) ₂) ₄ Hf	XXXXXX-XX-X	**	7.15 (V)	PE	4588
OHf⁺	HfO	12029-22-0	**	7.55±0.1	EI	4114
O₂Hf⁺	HfO ₂	12055-23-1	**	9.35±0.2	EI	4114
C₁₆H₁₄Si₄Hf⁺	((CH ₃) ₃ SiCH ₂) ₄ Hf	40334-04-1	**	8.58±0.1 (V)	PE	4242
Cl₄Hf⁺	HfCl ₄	13499-05-3	**	12.03 (V)	PE	4694
C₁₀H₁₀Cl₂Hf⁺	(η-C ₅ H ₅) ₂ HfCl ₂ (Hafnium,dichlorobis(η ⁵ -2,4-cyclopentadien-1-yl)-)	12116-66-4	** **	8.9±0.1 (V) 8.87±0.05 (V)	PE PE	4987 4375
Br₄Hf⁺	HfBr ₄ (JC-Mean value of Jahn-Teller components)	13777-22-5	**	11.06 (V)	PE	4694
I₄Hf⁺	HfI ₄ (JC-Mean value of Jahn-Teller components)	13777-23-6	**	9.53 (V)	PE	4694

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
Ta ⁺	Ta	7440-25-7	**	7.31±0.09	EI	5342
C ₅ H ₁₅ Ta ⁺	(CH ₃) ₅ Ta	53378-72-6	**	8.83±0.02 (V)	PE	4733
C ₁₀ H ₁₃ Ta ⁺	(C ₅ H ₅) ₂ H ₃ Ta (Tantalum, bis(η^5 -2,4-cyclopentadien-1-yl)dihydro-)	54474-28-1	**	8.1±0.1 (V)	PE	4425
C ₁₀ H ₃₀ N ₅ Ta ⁺	(N(CH ₃) ₂) ₅ Ta	XXXXXX-XX-X	**	6.89 (V)	PE	5036
OTa ⁺	TaO	12035-90-4	** **	7.5±0.5 7.92±0.1	EI EI	4678 4624
O ₂ Ta ⁺	TaO ₂	12036-14-5	**	8.5±0.5	EI	4678
Cl ₂ Ta ⁺	TaCl ₅	7721-01-9		20.3	EI	3783
Cl ₃ Ta ⁺	TaCl ₅	7721-01-9		15.2	EI	3783
Cl ₄ Ta ⁺	TaCl ₅	7721-01-9		10.9	EI	3783
Cl ₅ Ta ⁺	TaCl ₅	7721-01-9	**	11.08 (s)	PE	4764
C ₁₀ H ₁₀ Cl ₂ Ta ⁺	(η -C ₅ H ₅) ₂ TaCl ₂ (Tantalum,dichlorobis(η^5 -2,4-cyclopentadien-1-yl)-)	54039-37-1	**	6.4±0.1 (V)	PE	4987
C ₁₀ H ₁₀ Br ₂ Ta ⁺	(η -C ₅ H ₅) ₂ TaBr ₂ (Tantalum,dibromobis(η^5 -2,4-cyclopentadien-1-yl)-)	69005-97-6	**	6.4±0.1 (V)	PE	4987
C ₁₃ H ₂₁ SnTa ⁺	(C ₅ H ₅) ₂ (Sn(CH ₃) ₃)TaH ₂ (Tantalum,bis(η^5 -2,4-cyclopentadien-1-yl)dihydro(trimethylstannyl)-)	51192-04-2	**	6.77±0.12	EI	5321
W ⁺	W (CO) ₆ W CS(CO) ₅ W WBr ₄ WBr ₅	7440-33-7 14040-11-0 50358-92-4 14055-81-3 13470-11-6	** 6CO 5CO+CS Br 5Br	7.49±0.08 21.01±0.05 21.97±0.13 23.1±0.3 25.0±0.3	EI EI EI EI EI	5342 5291 5291 4906 4906
C ₃ H ₃ W ⁺	C ₅ H ₅ (CO) ₃ W (Tungsten, tricarbonyl(η^5 -2,4-cyclopentadien-1-yl)-)	12079-77-5	3CO+C ₂ H ₂	20.1±1.0	EI	4598
C ₅ H ₅ W ⁺	C ₅ H ₅ (CO) ₃ W (Tungsten, tricarbonyl(η^5 -2,4-cyclopentadien-1-yl)-)	12079-77-5	3CO	14.5±0.5	EI	4598
C ₆ H ₁₈ W ⁺	(CH ₃) ₆ W	36133-73-0	**	8.59±0.02 (V)	PE	4733

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{10}\text{H}_{12}\text{W}^+$	$(\text{C}_5\text{H}_5)_2\text{H}_2\text{W}$ (Tungsten, bis(η^5 -2,4-cyclopentadien-1-yl)dihydro-)	1271-33-6	**	6.4 ± 0.1 (V)	PE	4425
			**	6.35 ± 0.2	OTH	5278
$\text{C}_{12}\text{H}_{11}\text{W}^+$	$(\text{C}_5\text{H}_5)_2(\eta-\text{CH}_2=\text{CH}_2)\text{W}$ (Tungsten, bis(η^5 -2,4-cyclopentadien-1-yl)(η^2 -ethene)-)	37343-06-9	**	6.0 ± 0.1 (V)	PE	4425
$\text{C}_{12}\text{H}_{16}\text{W}^+$	$(\text{C}_5\text{H}_5)_2(\text{CH}_3)_2\text{W}$ (Tungsten, bis(η^5 -2,4-cyclopentadien-1-yl)dimethyl-)	39333-53-4	**	6.0 ± 0.1 (V)	PE	4425
$\text{C}_{13}\text{H}_{16}\text{W}^+$	$(\text{C}_5\text{H}_5)_2(\eta-\text{CH}_2=\text{CHCH}_3)\text{W}$ (Tungsten, bis(η^5 -2,4-cyclopentadien-1-yl)[1,2- η]-1-propene)-)	37343-23-0	**	5.9 ± 0.1 (V)	PE	4425
$\text{C}_6\text{H}_6\text{W}_2^+$	$[\text{C}_5\text{H}_5(\text{CO})_3\text{W}]_2$ (Tungsten, hexacarbonylbis (η^5 -2,4-cyclopentadien-1-yl)di-)	12566-66-4		29.0 ± 1.0	EI	4598
$\text{C}_8\text{H}_8\text{W}_2^+$	$[\text{C}_5\text{H}_5(\text{CO})_3\text{W}]_2$ (Tungsten, hexacarbonylbis (η^5 -2,4-cyclopentadien-1-yl)di-)	12566-66-4		25.0 ± 1.0	EI	4598
$\text{C}_{10}\text{H}_{10}\text{W}_2^+$	$[\text{C}_5\text{H}_5(\text{CO})_3\text{W}]_2$ (Tungsten, hexacarbonylbis (η^5 -2,4-cyclopentadien-1-yl)di-)	12566-66-4	6CO	15.5 ± 0.5	EI	4598
$\text{C}_{12}\text{H}_{36}\text{N}_6\text{W}^+$	$(\text{N}(\text{CH}_3)_2)_6\text{W}$	54935-70-5	**	6.73 (V)	PE	4588
O_2W^+	WO_2	12036-22-5	**	9.6 ± 0.3	EI	4556
O_3W_2^+	W_2O_6	XXXXX-XX-X		$35. \pm 1$	EI	4131
O_4W_2^+	W_2O_6	XXXXX-XX-X		17.1 ± 0.2	EI	4131
O_5W_2^+	W_2O_6	XXXXX-XX-X O		15.3 ± 0.2	EI	4131
O_6W_2^+	W_2O_6	XXXXX-XX-X	**	12.2 ± 0.2	EI	4131
O_8W_3^+	W_3O_9	XXXXX-XX-X O		14.6 ± 0.2	EI	4131
O_9W_3^+	W_3O_9	XXXXX-XX-X	**	12.0 ± 0.2	EI	4131
$\text{O}_{11}\text{W}_4^+$	W_4O_{12}	XXXXX-XX-X O		13.9 ± 0.2	EI	4131
$\text{W}_4\text{O}_{12}^+$	W_4O_{12}	XXXXX-XX-X	**	12.0 ± 0.2	EI	4131
BO_4W^+	$\text{W}(\text{BO}_3)_4\text{O}$	56644-98-5	**	10.9 ± 0.3	EI	4556

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
B₂O₆W⁺	B ₂ O ₃ ·WO ₃	XXXXX-XX-X **		12.3±0.3	EI	4556
BO₇W₂⁺	BO·W ₂ O ₆	XXXXX-XX-X **		12.1±0.3	EI	4556
BO₁₀W₃⁺	BO·W ₃ O ₉	XXXXX-XX-X **		12.5±0.3	EI	4556
B₂O₁₂W₃⁺	B ₂ O ₃ ·W ₃ O ₉	XXXXX-XX-X **		12.4±0.3	EI	4556
BO₁₃W₄⁺	BO·W ₄ O ₁₂	XXXXX-XX-X **		13.1±0.3	EI	4556
COW⁺	(CO) ₆ W CS(CO) ₅ W	14040-11-0 50358-92-4	5CO 4CO+CS	18.36±0.06 19.48±0.21	EI EI	5291 5291
C₂O₂W⁺	(CO) ₆ W CS(CO) ₅ W	14040-11-0 50358-92-4	4CO 3CO+CS	16.29±0.04 17.21±0.27	EI EI	5291 5291
C₃O₃W⁺	(CO) ₆ W CS(CO) ₅ W	14040-11-0 50358-92-4	3CO 2CO+CS	14.06±0.02 14.86±0.11	EI EI	5291 5291
C₄O₄W⁺	(CO) ₆ W CS(CO) ₅ W	14040-11-0 50358-92-4	2CO CO+CS	12.22±0.03 13.12±0.11	EI EI	5291 5291
C₅O₅W⁺	(CO) ₆ W CS(CO) ₅ W	14040-11-0 50358-92-4	CO CS	10.30±0.03 11.46±0.14	EI EI	5291 5291
C₆O₆W⁺	(CO) ₆ W	14040-11-0	** ** **	8.30±0.02 (V) 8.56 (V) 8.60±0.02	PE PE EI	3979 4456 5291
C₆H₅OW⁺	C ₅ H ₅ (CO) ₃ W (Tungsten, tricarbonyl(η^5 -2,4-cyclopentadien-1-yl)-)	12079-77-5	2CO	13.2±1.0	EI	4598
C₇H₅O₂W⁺	C ₅ H ₅ (CO) ₃ W (Tungsten, tricarbonyl(η^5 -2,4-cyclopentadien-1-yl)-)	12079-77-5	CO	12.3±0.2	EI	4598
C₈H₅O₃W⁺	C ₅ H ₅ (CO) ₃ W (Tungsten, tricarbonyl(η^5 -2,4-cyclopentadien-1-yl)-) [C ₅ H ₅ (CO) ₃ W] ₂ (Tungsten, hexacarbonylbis η^5 -2,4-cyclopentadien-1-yl)di-	12079-77-5 12566-66-4	** C ₅ H ₅ (CO) ₃ W	7.66±0.05 10.05±0.2	EI	4598 4598
C₁₀H₈O₃W⁺	C ₅ H ₅ (CO) ₃ W (Tungsten, tricarbonyl[(1,2,3,4,5,6- η)-1,3,5-cycloheptatriene]-) **	12128-81-3 **		7.32 (V) 7.55±0.05 (V)	PE PE	5206 4724
C₁₂H₁₂O₃W⁺	(C ₆ H ₅ (CH ₃) ₃)(CO) ₃ W (Tungsten, tricarbonyl[(1,2,3,4,5,6- η)-1,3,5-trimethylbenzene]-)	12129-69-0	**	7.20±0.05 (V)	PE	4724

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{11}\text{H}_{10}\text{OW}_2^+$	$[\text{C}_5\text{H}_5(\text{CO})_3\text{W}]_2$ (Tungsten, hexacarbonylbis (η^5 -2,4-cyclopentadien-1-yl)di-)	12566-66-4	5CO	13.85 ± 0.10	EI	4598
$\text{C}_{12}\text{H}_{10}\text{O}_2\text{W}_2^+$	$[\text{C}_5\text{H}_5(\text{CO})_3\text{W}]_2$ (Tungsten, hexacarbonylbis (η^5 -2,4-cyclopentadien-1-yl)di-)	12566-66-4	4CO	12.89 ± 0.10	EI	4598
$\text{C}_{13}\text{H}_{10}\text{O}_3\text{W}_2^+$	$[\text{C}_5\text{H}_5(\text{CO})_3\text{W}]_2$ (Tungsten, hexacarbonylbis (η^5 -2,4-cyclopentadien-1-yl)di-)	12566-66-4	3CO	11.00 ± 0.20	EI	4598
$\text{C}_{14}\text{H}_{10}\text{O}_4\text{W}_2^+$	$[\text{C}_5\text{H}_5(\text{CO})_3\text{W}]_2$ (Tungsten, hexacarbonylbis (η^5 -2,4-cyclopentadien-1-yl)di-)	12566-66-4	2CO	8.61 ± 0.05	EI	4598
$\text{C}_{15}\text{H}_{10}\text{O}_5\text{W}_2^+$	$[\text{C}_5\text{H}_5(\text{CO})_3\text{W}]_2$ (Tungsten, hexacarbonylbis (η^5 -2,4-cyclopentadien-1-yl)di-)	12566-66-4	CO	7.70 ± 0.05	EI	4598
$\text{C}_{16}\text{H}_{10}\text{O}_6\text{W}_2^+$	$[\text{C}_5\text{H}_5(\text{CO})_3\text{W}]_2$ (Tungsten, hexacarbonylbis (η^5 -2,4-cyclopentadien-1-yl)di-)	12566-66-4	**	6.65 ± 0.05	EI	4598
$\text{C}_5\text{H}_3\text{NO}_5\text{W}^+$	$(\text{CO})_5\text{NH}_3\text{W}$	15133-64-9	**	7.54 (V)	PE	4252
$\text{C}_7\text{H}_7\text{NO}_5\text{W}^+$	$(\text{CO})_5\text{NH}(\text{CH}_3)_2\text{W}$	15228-31-6	**	7.41 (V)	PE	4252
$\text{C}_8\text{H}_9\text{NO}_5\text{W}^+$	$(\text{CO})_5\text{N}(\text{CH}_3)_3\text{W}$	15228-32-7	**	7.41 (V)	PE	4252
$\text{C}_{10}\text{H}_5\text{NO}_5\text{W}^+$	$\text{C}_5\text{H}_5\text{NW}(\text{CO})_5$ (OC-6-22)-Pentacarbonyl(pyridine)tungsten)	14586-49-3	**	7.53 ± 0.05	EI	3498
			**	7.53	EI	5292
$\text{C}_{10}\text{H}_{11}\text{NO}_5\text{W}^+$	$(\text{C}_5\text{H}_{10}\text{NH})(\text{CO})_5\text{W}$ (Tungsten,pentacarbonyl(piperidine)-(OC-6-22))	31082-68-5	**	7.35 (V)	PE	5540
$\text{C}_{11}\text{H}_7\text{NO}_5\text{W}^+$	$\text{C}_5\text{H}_4\text{N}(\text{CH}_3)\text{W}(\text{CO})_5$ (Pentacarbonyl(4-methylpyridine)tungsten)	17000-14-5	**	7.46 ± 0.05	EI	3498
			**	7.46	EI	5292
$\text{C}_{12}\text{H}_9\text{NO}_5\text{W}^+$	$\text{C}_5\text{H}_3\text{N}(\text{CH}_3)_2\text{W}(\text{CO})_5$ ((OC-6-22)-Pentacarbonyl(2,6-dimethylpyridine)tungsten)	36252-39-8	**	7.43 ± 0.05	EI	3498
			**	7.43	EI	5292
$\text{C}_{11}\text{H}_4\text{N}_2\text{O}_5\text{W}^+$	$\text{C}_5\text{H}_4\text{N}(\text{CN})\text{W}(\text{CO})_5$ ((OC-6-22)-Pentacarbonyl(2-pyridinecarbonitrile- N^1)tungsten)	36252-42-3	**	7.65 ± 0.05	EI	3498
			**	7.65	EI	5292
$\text{C}_{12}\text{H}_{11}\text{N}_2\text{O}_5\text{W}^+$	$(\text{C}_5\text{H}_4\text{N}_2(\text{C}_2\text{H}_5)_2)(\text{CO})_5\text{W}$	XXXXX-XX-X	**	7.02 (V)	PE	5601
$\text{C}_{21}\text{H}_{21}\text{N}_1\text{O}_1\text{W}_2^+$	$(\text{C}_5\text{H}_3\text{N}(\text{O})\text{CH}_3)\text{W}_2$ (Tungsten,tetrakis[μ -(6-methyl-2(1H)-pyridinonato- N^1 : O^2)]di-(W-W) stereoisomer)	67634-84-8	**	5.3 (V)	PE	5191

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
FW⁺	WF	51621-16-0	**	8.5±1	EI	4580
F₂W⁺	WF ₂	33963-15-4	**	9.0±0.3	EI	4580
F₃W⁺	WF ₃ WF ₆	51621-17-1 7783-82-6	**	9.0±0.2 24.0±0.5	EI EI	4580 4580
F₄W⁺	WF ₄ WF ₆	13766-47-7 7783-82-6	** 2F	9.89±0.10 19.5±0.3	EI EI	4580 4580
F₅W⁺	WF ₅ WF ₆	19357-83-6 7783-82-6	** ** F	14.9±0.1 10.03±0.10 15.24±0.10	PE EI EI	4989 4580 4580
C₁₅H₂₁O₆Si₂W⁺	C ₁₅ H ₂₁ O ₆ Si ₂ W	XXXXX-XX-X	**	7.55 (V)	PE	5601
C₁₂H₃₆N₆P₂W⁺	((CH ₃) ₂ N) ₃ P) ₂ (CO) ₄ W	19976-86-4	4CO	10.7±0.05	EI	3952
C₂₀H₁₅O₂PW⁺	((C ₆ H ₅) ₃ P)(CO) ₅ W (Tungsten,pentacarbonyl(triphenylphosphine)-(OC-6-22)-)	15444-65-2	3CO	9.5	EI	5564
C₉H₉O₃PW⁺	(P(OCH ₃) ₃)(CO) ₅ W	23306-42-5	5CO	13.1	EI	5564
C₂₁H₁₅O₃PW⁺	((C ₆ H ₅) ₃ P)(CO) ₅ W (Tungsten,pentacarbonyl(triphenylphosphine)-(OC-6-22)-)	15444-65-2	2CO	9.1	EI	5564
C₇H₉O₄PW⁺	(P(OCH ₃) ₃)(CO) ₅ W	23306-42-5	4CO	12.3	EI	5564
C₇H₁₅O₄PW⁺	(P(OC ₂ H ₅) ₃)(CO) ₅ W	23306-43-6	5CO	12.2	EI	5564
C₁₆H₂₇O₄PW⁺	((n-C ₄ H ₉) ₃ P)(CO) ₅ W	17000-19-0	CO	9.4	EI	5564
C₂₂H₁₅O₄PW⁺	((C ₆ H ₅) ₃ P)(CO) ₅ W (Tungsten,pentacarbonyl(triphenylphosphine)-(OC-6-22)-)	15444-65-2	CO	8.5	EI	5564
C₅H₉O₅PW⁺	(P(OCH ₃) ₃)(CO) ₅ W	23306-42-5	3CO	11.1	EI	5564
C₈H₉O₅PW⁺	((CH ₃) ₃ P)(CO) ₅ W	26555-11-3	**	7.9	PE	5602
C₈H₁₅O₅PW⁺	(P(OC ₂ H ₅) ₃)(CO) ₅ W	23306-43-6	3CO	11.3	EI	5564
C₁₁H₁₅O₅PW⁺	((C ₂ H ₅) ₃ P)(CO) ₅ W	21321-31-3	**	7.8	PE	5602

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{23}\text{H}_{15}\text{O}_5\text{PW}^+$	$(\text{C}_6\text{H}_5)_3\text{P}(\text{CO})_5\text{W}$ (Tungsten, pentacarbonyl (triphenylphosphine)-(OC-6-22)-)	15444-65-2 **		7.36 (V)	PE	5139
			**	7.80±0.05	EI	4600
$\text{C}_{23}\text{H}_{33}\text{O}_5\text{PW}^+$	$(\text{C}_6\text{H}_{11})_3\text{P}(\text{CO})_5\text{W}$ (Tungsten, pentacarbonyl (tricyclohexylphosphine)-(OC-6-22)-)	18474-91-4 **		7.29 (V)	PE	5139
$\text{C}_6\text{H}_6\text{O}_6\text{PW}^+$	$(\text{P}(\text{OCH}_3)_3)(\text{CO})_5\text{W}$	23306-42-5	$\text{OCH}_3 + \text{CO}$	13.0	EI	5564
$\text{C}_6\text{H}_9\text{O}_6\text{PW}^+$	$(\text{P}(\text{OCH}_3)_3)(\text{CO})_5\text{W}$	23306-42-5	2CO	9.8	EI	5564
$\text{C}_8\text{H}_{10}\text{O}_6\text{PW}^+$	$(\text{P}(\text{OC}_2\text{H}_5)_3)(\text{CO})_5\text{W}$	23306-43-6	$\text{OC}_2\text{H}_5 + \text{CO}$	12.5	EI	5564
$\text{C}_9\text{H}_{15}\text{O}_6\text{PW}^+$	$(\text{P}(\text{OC}_2\text{H}_5)_3)(\text{CO})_5\text{W}$	23306-43-6	2CO	10.5	EI	5564
$\text{C}_7\text{H}_6\text{O}_7\text{PW}^+$	$(\text{P}(\text{OCH}_3)_3)(\text{CO})_5\text{W}$	23306-42-5	OCH_3	11.2	EI	5564
$\text{C}_7\text{H}_9\text{O}_7\text{PW}^+$	$(\text{P}(\text{OCH}_3)_3)(\text{CO})_5\text{W}$	23306-42-5	CO	9.0	EI	5564
$\text{C}_9\text{H}_{10}\text{O}_7\text{PW}^+$	$(\text{P}(\text{OC}_2\text{H}_5)_3)(\text{CO})_5\text{W}$	23306-43-6	OC_2H_5	11.5	EI	5564
$\text{C}_{10}\text{H}_{15}\text{O}_7\text{PW}^+$	$(\text{P}(\text{OC}_2\text{H}_5)_3)(\text{CO})_5\text{W}$	23306-43-6	CO	9.4	EI	5564
$\text{C}_8\text{H}_9\text{O}_8\text{PW}^+$	$((\text{CH}_3\text{O})_3\text{P})(\text{CO})_5\text{W}$	23306-42-5 **		8.2	PE	5602
$\text{C}_{11}\text{H}_{15}\text{O}_8\text{PW}^+$	$((\text{C}_2\text{H}_5\text{O})_3\text{P})(\text{CO})_5\text{W}$	23306-43-6 **		8.1	PE	5602
$\text{C}_{11}\text{H}_{21}\text{O}_8\text{PW}^+$	$(\text{i}-\text{C}_3\text{H}_7\text{O})_3\text{P}(\text{CO})_5\text{W}$	XXXXX-XX-X **		7.82 (V)	PE	5139
$\text{C}_{23}\text{H}_{15}\text{O}_8\text{PW}^+$	$(\text{C}_6\text{H}_5)_3\text{P}(\text{CO})_5\text{W}$ (Tungsten, pentacarbonyl (triphenyl phosphite-P)-(OC-6-22)-)	23306-41-4 **		7.90 (V)	PE	5139
$\text{C}_{16}\text{H}_{30}\text{O}_4\text{P}_2\text{W}^+$	$\text{C}_{16}\text{H}_{30}\text{O}_4\text{P}_2\text{W}$ (Tungsten, tetracarbonylbis(triphenylphosphine)-(OC-6-12)-)	16743-03-6 **		7.50±0.05	EI	4600
$\text{C}_{11}\text{H}_{18}\text{N}_3\text{O}_5\text{PW}^+$	$((\text{CH}_3)_2\text{N})_3\text{P}(\text{CO})_5\text{W}$	19976-82-0 **		7.9	PE	5602
$\text{C}_{11}\text{H}_{36}\text{N}_6\text{O}_2\text{P}_2\text{W}^+$	$((\text{CH}_3)_2\text{N})_3\text{P}_2(\text{CO})_4\text{W}$	19976-86-4	2CO	12.2±0.05	EI	3952
$\text{C}_{15}\text{H}_{36}\text{N}_6\text{O}_3\text{P}_2\text{W}^+$	$((\text{CH}_3)_2\text{N})_3\text{P}_2(\text{CO})_4\text{W}$	19976-86-4	CO	10.3±0.05	EI	3952
$\text{C}_{16}\text{H}_{36}\text{N}_6\text{O}_4\text{P}_2\text{W}^+$	$((\text{CH}_3)_2\text{N})_3\text{P}_2(\text{CO})_4\text{W}$	19976-86-4 **		5.5±0.05	EI	3952

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
F₁₈P₆W⁺	(PF ₃) ₆ W	13815-35-5	**	9.30 (V)	PE	4456
C₃H₉N₃F₁₂P₆W⁺	(CH ₃ N(PF ₂) ₂) ₃ W	63371-85-7	**	7.70 (V)	PE	5376
C₅O₅F₃PW⁺	(PF ₃)(CO) ₅ W	18461-47-7	** **	8.68 (V) 8.9	PE PE	5539 5602
CSW⁺	CS(CO) ₅ W	50358-92-4	5CO	18.07±0.04	EI	5291
C₂OSW⁺	CS(CO) ₅ W	50358-92-4	4CO	15.83±0.04	EI	5291
C₃O₂SW⁺	CS(CO) ₅ W	50358-92-4	3CO	13.46±0.04	EI	5291
C₄O₃SW⁺	CS(CO) ₅ W	50358-92-4	2CO	11.61±0.04	EI	5291
C₅O₄SW⁺	CS(CO) ₅ W	50358-92-4	CO	9.74±0.04	EI	5291
C₆O₅SW⁺	CS(CO) ₅ W	50358-92-4	** **	8.08 (V) 8.22±0.01	PE EI	5518 5291
F₂SW⁺	WSF ₂	41831-78-1	**	9.5±0.3	EI	4580
F₃SW⁺	WSF ₃ ⁺	41831-79-2	**	9.0±0.3	EI	4580
F₄SW⁺	WSF ₄	XXXXX-XX-X	**	12.0±0.3	EI	4580
F₂S₂W⁺	WS ₂ F ₂	41831-81-6	**	10.0±0.3	EI	4580
ClW⁺	WCl ₆	13283-01-7		22.9	EI	3783
Cl₂W⁺	WCl ₆	13283-01-7		19.4	EI	3783
Cl₃W⁺	WCl ₆	13283-01-7		15.4	EI	3783
Cl₄W⁺	WCl ₆	13283-01-7		11.4	EI	3783
Cl₅W⁺	WCl ₅ WCl ₆	13470-14-9 13283-01-7	**	8.84 (V) 10.9	PE EI	4764 3783
Cl₆W⁺	WCl ₆	13283-01-7	**	9.5	EI	3783
C₅O₅PCl₃W⁺	(PCl ₃)(CO) ₅ W	21223-85-8	**	8.39 (V)	PE	5539

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
O₅VW⁺	VW ₂ O ₈		XXXXX-XX-X	11.7±0.3	EI	4131
O₁₀V₃W⁺	V ₃ WO ₁₀		XXXXX-XX-X **	11.5±0.3	EI	4131
O₈VW₂⁺	VW ₂ O ₈		XXXXX-XX-X **	10.4±0.2	EI	4131
O₉V₂W₂⁺	V ₂ W ₂ O ₁₀		XXXXX-XX-X	12.2±0.2	EI	4131
O₁₀V₂W₂⁺	V ₂ W ₂ O ₁₀		XXXXX-XX-X **	11.9±0.2	EI	4131
O₁₃V₃W₂⁺	V ₃ W ₂ O ₁₃		XXXXX-XX-X **	11.1±0.2	EI	4131
O₁₁VW₃⁺	VW ₃ O ₁₁		XXXXX-XX-X **	10.7±0.2	EI	4131
O₁₃V₂W₁₃⁺	V ₂ W ₁₃ O ₁₁		XXXXX-XX-X O	12.3±0.4	EI	4131
C₂₃H₁₅O₅AsW⁺	(C ₆ H ₅) ₃ As(CO) ₅ W (Tungsten, pentacarbonyl (triphenylarsine)-(OC-6-22)-)	29743-02-0	**	7.37 (V)	PE	5139
BrW⁺	WBr ₁ WBr ₅ WOBr ₁	14055-81-3 13470-11-6 13520-77-9	3Br 4Br —	19.4±0.3 20.9±0.3 26.1±0.5	EI EI EI	4906 4906 4906
Br₂W⁺	WBr ₁ WBr ₅ WOBr ₁	14055-81-3 13470-11-6 13520-77-9	2Br 3Br —	15.1±0.3 16.6±0.2 20.9±0.4 21.4±0.5	EI EI EI EI	4906 4906 4906 3450
Br₃W⁺	WBr ₁ WBr ₅ WOBr ₁	14055-81-3 13470-11-6 13520-77-9	Br 2Br —	11.2±0.2 13.4±0.2 17.9±0.4 18.1±0.5	EI EI EI EI	4906 4906 4906 3450
Br₄W⁺	WBr ₁ WBr ₅	14055-81-3 13470-11-6	** Br	8.2±0.2 10.0±0.2	EI EI	4906 4906
Br₅W⁺	WBr ₅	13470-11-6	**	8.3±0.2	EI	4906
Br₃W₂⁺	W ₂ Br ₆	56729-72-7	3Br	19.5±0.3	EI	4906
Br₄W₂⁺	W ₂ Br ₆	56729-72-7	2Br	15.2±0.3	EI	4906
Br₅W₂⁺	W ₂ Br ₆	56729-72-7	Br	11.0±0.2	EI	4906

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
Br₆W⁺	W ₂ Br ₆	56729-72-7	**	9.0±0.2	EI	4906
OBrW⁺	WO ₂ Br ₂ WOBr ₃	13520-75-7 13520-77-9		20.0±0.8 18.3±0.5 18.1±0.8	EI EI EI	3450 4906 3450
O₂BrW⁺	WO ₂ Br ₂	13520-75-7		13.0±0.4	EI	3450
OBr₂W⁺	WOBr ₃	13520-77-9		14.5±0.2 14.4±0.5	EI EI	4906 3450
O₂Br₂W⁺	WO ₂ Br ₂	13520-75-7	**	11.4±0.2	EI	3450
OBr₃W⁺	WOBr ₃	13520-77-9		10.5±0.2 10.3±0.2	EI EI	4906 3450
OBr₄W⁺	WOBr ₃	13520-77-9	**	10.3±0.3	EI	3450
C₂₁H₂₁N₁O₁MoW⁺	(C ₅ H ₃ N(O)CH ₃) ₂ WMo (Tungsten, tetrakis[μ-(6-methyl-2(1H)-pyridinonato-N ¹ ,O ²)] (molybdenum)-(Mo-W))	67577-06-4	**	5.60 (V)	PE	5191
C₁₃H₂₀SnW⁺	(C ₅ H ₃) ₂ Sn(CH ₃) ₃ WH (Tungsten, bis(η ⁵ -2,4-cyclopentadien-1-yl)hydro(trimethylstannyl)-)	51192-18-8	**	6.18±0.11	EI	5321
C₂₃H₁₅O₅SbW⁺	(C ₆ H ₅) ₃ (CO) ₅ SbW (Tungsten, pentacarbonyl(triphenylstibine)-(OC-6-22)-)	29743-03-1	**	7.90±0.05	EI	4600
O₂IW⁺	WO ₂ I ₂	14447-89-3		12.5±0.5	EI	3451
O₂I₂W⁺	WO ₂ I ₂	14447-89-3	**	10.4±0.4	EI	3451
Re⁺	Re	7440-15-5	**	7.76±0.03	EI	5342
C₆H₁₈Re⁺	(CH ₃) ₆ Re	56090-02-9	**	7.89±0.03 (V)	PE	4733
C₁₀H₁₁Re⁺	(C ₅ H ₃) ₂ HRe (Rhenium, bis(η ⁵ -2,4-cyclopentadien-1-yl)hydro-)	1271-32-5	**	6.4±0.1 (V)	PE	4425
ORe⁺	ReO ₃	1314-28-9		~18	EI	4016
O₂Re⁺	ReO ₃ Re ₂ O ₇	1314-28-9 1314-68-7		14.4±1.0 21.9±1.0	EI EI	4016 4016

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
O_3Re^+	ReO ₃	1314-28-9	**	12.1±0.3	EI	4245
			**	12.5±0.4	EI	4016
	Re ₂ O ₇	1314-68-7		16.2±0.5	EI	4016
$O_5Re_2^+$	Re ₂ O ₇	1314-68-7		17.5±0.2	EI	4016
$O_6Re_2^+$	Re ₂ O ₇	1314-68-7		16.2±0.5	EI	4016
$O_7Re_2^+$	Re ₂ O ₇	1314-68-7	**	12.7±0.2	EI	4016
$C_{10}O_{10}Re_2^+$	(CO) ₁₀ Re ₂	14285-68-8	**	8.07 (V)	PE	4492
			**	8.86 (V)	PE	4448
$C_4H_{12}ORe^+$	(CH ₃) ₄ ORe	53022-70-1	**	8.86±0.05 (V)	PE	4733
$C_8H_5O_3Re^+$	C ₅ H ₅ (CO) ₃ Re (Rhenium, tricarbonyl (η^5 -2,4-cyclopentadien-1-yl)-)	12079-73-1	**	8.13 (V)	PE	4570
$C_5HO_5Re^+$	(CO) ₅ ReH	16457-30-0	**	8.86±0.02 (V)	PE	3827
			**	8.89±0.08	PE	4492
			**	8.94 (V)	PE	4448
$C_6H_3O_5Re^+$	(CO) ₅ CH ₃ Re	14524-92-6	**	8.71±0.05 (V)	PE	4492
			**	8.72 (V)	PE	4448
$C_{12}H_3O_{12}Re_3^+$	(CO) ₁₂ Re ₃ H ₃	XXXXX-XX-X	**	8.45 (V)	PE	5547
	(CO) ₁₂ Re ₃ H ₃	73463-62-4	**	8.45 (V)	PE	5357
	(Rhenium,dodecacarbonyltri- μ -hydrotri- <i>triangulo</i>)					
F_6Re^+	ReF ₆	10049-17-9	**	7.99	S	3565
			**	11.1±0.1	PE	4989
F_7Re^+	ReF ₇	17029-21-9	**	14.1±0.1	PE	4989
O_3FRe^+	ReO ₃ F	42246-24-2	**	12.37±0.1 (V)	PE	4989
OF_5Re^+	ReOF ₅	23377-53-9	**	13.2±0.1	PE	4989
$C_7O_6F_3Re^+$	COCF ₃ (CO) ₅ Re	55615-47-9	**	8.80 (V)	PE	4448
O_4NaRe^+	NaReO ₄	XXXXX-XX-X	**	10.62±0.03 (V)	PE	4806
$C_5H_3O_5SiRe^+$	(SiH ₃)(CO) ₅ Re	40628-33-9	**	8.9±0.1 (V)	PE	3827

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{16}\text{H}_{11}\text{OSi}_1\text{Re}^+$	$((\text{CH}_3)_3\text{SiCH}_2)_1\text{ORe}$	56519-47-2	**	8.00 ± 0.1 (V)	PE	4733
Cl_5Re^+	ReCl_3 (JC-Mean value of Jahn-Teller components)	13596-35-5	**	9.50 (V)	PE	4764
Cl_9Re_3^+	Re_3Cl_9 (Rhenium, tri- μ -chlorohexachlorotri- <i>triangulo</i>)	14973-59-2	**	9.15 ± 0.05 (V)	PE	5024
$\text{C}_5\text{O}_5\text{ClRe}^+$	$(\text{CO})_5\text{ReCl}$	14099-01-5	** ** **	8.80 (V) 9.02 (V) 9.06 (V)	PE PE PE	4448 4167 4492
O_1KRe^+	KReO_1	XXXXX-XX-X	**	9.98 ± 0.05 (V)	PE	4806
$\text{C}_5\text{H}_9\text{O}_5\text{GeRe}^+$	$(\text{GeH}_3)(\text{CO})_5\text{Re}$	30012-26-1	**	8.9 ± 0.1 (V)	PE	3827
Br_9Re^+	Re_3Br_9 (Rhenium, tri- μ -bromohexabromotri- <i>triangulo</i>)	33517-16-7	**	8.72 ± 0.10 (V)	PE	5024
$\text{C}_5\text{O}_5\text{BrRe}^+$	$(\text{CO})_5\text{ReBr}$	14220-21-4	** ** **	8.80 (V) 8.83 (V) 8.86 (V)	PE PE PE	4448 4492 4167
O_1RbRe^+	RbReO_1	XXXXX-XX-X	**	10.03 ± 0.06 (V)	PE	4806
$\text{C}_8\text{H}_9\text{O}_5\text{SnRe}^+$	$((\text{CH}_3)_3\text{Sn})(\text{CO})_5\text{Re}$	15219-90-6	**	8.30 ± 0.10	EI	5321
$\text{C}_{23}\text{H}_{15}\text{O}_5\text{SnRe}^+$	$((\text{C}_6\text{H}_5)_3\text{Sn})(\text{CO})_5\text{Re}$ (Rhenium, pentacarbonyl(triphenylstannyl)-(OC-6-22)-)	15614-21-8	**	7.98 ± 0.09	EI	5321
O_1IRe^+	ReO_3I	39327-80-5	**	10.9 ± 0.5	EI	4013
$\text{C}_5\text{O}_5\text{IRe}^+$	$(\text{CO})_5\text{ReI}$	13821-00-6	** ** **	8.32 (V) 8.36 (V) 8.50 (V)	PE PE PE	4448 4492 4167
O_1CsRe^+	CsReO_1	XXXXX-XX-X	**	9.83 ± 0.03 (V)	PE	4806
O_1BaRe^+	$\text{Ba}(\text{ReO}_4)_2?$	XXXXX-XX-X		13.4 ± 0.5	EI	4108
Os^+	Os	7440-04-2	**	8.15 ± 0.09	EI	5342
$\text{C}_{12}\text{H}_{11}\text{Os}^+$	$(\text{C}_5\text{H}_1\text{CH}_3)_2\text{Os}$ (Osmocene, 1,1'-dimethyl-)	40672-07-9	**	6.93 (V)	PE	3688
O_1Os^+	OsO ₄	20816-12-0	** **	12.320 12.35 ± 0.02 (V)	PE PE	3836 5148

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
O₁Os⁺	OsO ₄	20816-12-0	** **	12.35 12.39	PE PE	4166 3838
C₁₂O₁₂Os₃⁺	(CO) ₁₂ Os ₃	15696-40-9	** ** **	7.83 (V) 7.83±0.2 (V) 7.83 (V)	PE PE PE	5547 4882 5357
C₁₈O₁₈Os₆⁺	(CO) ₁₈ Os ₆ (Osmium, octadecacarbonylhexa-)	37216-50-5	**	7.50±0.2 (V)	PE	4882
Ir⁺	Ir	7439-88-5	** **	8.8±0.7 8.87±0.05	EI EI	5303 5342
C₇H₇O₁Ir⁺	(CH ₃ COCHCOCH ₃)Ir(CO) ₂ (Dicarbonyl(2,4-pentanedionato)iridium)	14023-80-4	**	8.6±0.1	EI	3497
C₇HO₁F₆Ir⁺	(CF ₃ COCHCOOCF ₃)Ir(CO) ₂ (Dicarbonyl(1,1,1,5,5,5-hexafluoro-2,4-pentanedionato)iridium)	14049-69-5	**	8.85±0.05	EI	3497
HF₁₂P₁Ir⁺	H(PF ₃) ₄ Ir	22372-64-1	**	9.82 (V)	PE	4456
LaIr⁺	LaIr	53095-72-0	**	6.0±1.0	EI	5303
CeIr⁺	IrCe	53239-19-3	**	6.0±1.0	EI	4209
Pt⁺	Pt	7440-06-4	**	8.82±0.04	EI	5342
C₆H₁₀Pt⁺	(C ₆ H ₅) ₂ Pt	12240-88-9	**	7.91 (V)	PE	5281
C₈H₁₁Pt⁺	(CH ₂ C(CH ₃)CH ₂) ₂ Pt	33010-07-0	**	7.65 (V)	PE	5281
C₁₀H₁₆O₁Pt⁺	((CH ₃ CO) ₂ CH ₂)Pt	XXXXXX-XX-X	**	7.60 (V)	PE	5568
C₈H₂₁P₂Pt⁺	C ₈ H ₂₁ P ₂ Pt	51351-75-8	**	7.68 (V)	PE	4739
C₁₈H₂₆P₂Pt⁺	C ₁₈ H ₂₆ P ₂ Pt (Platinum, bis(dimethylphenylphosphine)dimethyl-(SP-4-2)-)	24917-48-4	**	7.43 (V)	PE	4739
F₁₂P₁Pt⁺	Pt(PF ₃) ₄	19529-53-4	**	8.89±0.03	PE	4187
C₈H₂₀O₁P₂S₁Pt⁺	Pt(S ₂ P(OC ₂ H ₅) ₂) ₂	37583-01-0	**	7.60±0.05	PE	4636
C₇H₂₁P₂ClPt⁺	C ₇ H ₂₁ P ₂ ClPt	36512-52-4	**	7.76 (V)	PE	4739

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₇H₂₅P₂ClPt⁺	C ₁₇ H ₂₅ P ₂ ClPt (Platinum, chlorobis(dimethylphenylphosphine)methyl-(SP-4-3)-)	24833-58-7	**	7.54 (V)	PE	4739
C₆H₁₈P₂Cl₂Pt⁺	C ₆ H ₁₈ P ₂ Cl ₂ Pt	21545-76-6	**	7.86 (V)	PE	4739
TiPt⁺	PtTi	12038-31-2	**	10.1±1.0	EI	5150
C₁₇H₂₅P₂BrPt⁺	C ₁₇ H ₂₅ P ₂ BrPt (Platinum, bromobis(dimethylphenylphosphine)methyl-(SP-4-3)-)	24833-62-3	**	7.43 (V)	PE	4739
C₇H₂₁P₂IPt⁺	C ₇ H ₂₁ P ₂ IPt	68146-10-1	**	7.33 (V)	PE	4739
C₁₇H₂₅P₂IPt⁺	C ₁₇ H ₂₅ P ₂ IPt (Platinum, bis(dimethylphenylphosphine)iodomethyl-(SP-4-3)-)	24882-77-7	**	7.12 (V)	PE	4739
C₆H₁₈P₂I₂Pt⁺	trans-((CH ₃) ₃ P) ₂ I ₂ Pt	15703-03-4	**	7.49 (V)	PE	4739
C₁₆H₂₂P₂I₂Pt⁺	C ₁₆ H ₂₂ P ₂ I ₂ Pt (Platinum, bis(dimethylphenylphosphine)diido-(SP-4-1)-)	41119-53-3	**	7.39 (V)	PE	4739
CePt⁺	PtCe	12157-68-5	**	6.4±1.0	EI	4209
Au⁺	Au	7440-57-5	**	9.23	S	5500
	(² P ₁)		**	9.22	PE	4858
	(¹ S ₀)		**	11.08	PE	4858
	(³ D ₃)		**	11.41	PE	4858
	(³ D ₂)		**	12.66	PE	4858
	(³ D ₁)		**	12.89	PE	4858
	(¹ D ₂)		**	8.5±0.8	EI	3978
			**	9.0±0.5	EI	3473
			**	9.21±0.05	EI	3745
	NaAu	XXXXX-XX-X	Na	7.8	EI	4578
Au₂⁺	Au ₂	12187-09-6	**	9.5±0.3	EI	4014
			**	8.7±1.0	EI	5391
			**	9.5±0.3	EI	4005
			**	9.7±0.4	EI	3468
BAu⁺	AuB	12408-81-0	**	8.7±0.5	EI	3468
BOAu⁺	AuBO	12588-90-8	**	9.7±0.2	EI	3473
NaAu⁺	NaAu	61115-29-5	**	6.2	EI	4578
			**	8.5±1.5	EI	4919
AlAu⁺	AuAl	12250-38-3	**	7.6±0.3	EI	4014
			**	7.6±0.3	EI	4005

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
AlAu⁺						
	AuAl	12250-38-3	** **	7.8±0.3 9.0±1.0	EI EI	3440 3796
Al₂Au⁺						
	AuAl ₂	12004-03-4	**	6.2±1.0	EI	3966
AlAu₂⁺						
	Au ₂ Al	12250-39-4	**	7.7±1.0	EI	3966
C₄H₁₂PAu⁺						
	((CH ₃) ₃ P)(CH ₃)Au	32407-79-7	**	8.27 (V)	PE	4739
C₆H₁₈PAu⁺						
	((CH ₃) ₃ P)(CH ₃) ₃ Au	33012-33-8	**	7.80 (V)	PE	4739
C₁₁H₂₀PAu⁺						
	C ₁₁ H ₂₀ PAu (Gold, (dimethylphenylphosphine)trimethyl-(SP-4-2)-)	54854-73-8	**	7.69 (V)	PE	4739
C₁₆H₂₂PAu⁺						
	C ₁₆ H ₂₂ PAu (Gold, trimethyl(methyldiphenylphosphine)-(SP-4-2)-)	52170-97-5	**	7.64 (V)	PE	4739
GeAu⁺						
	AuGe	12256-41-6	**	7.7	EI	3775
CsAu⁺						
	CsAu	12256-37-0	**	6.6±0.3	EI	5153
LaAu⁺						
	LaAu	12429-32-2	**	5.8±1.0	EI	5303
CeAu⁺						
	AuCe	12408-82-1	**	6.0±0.3	EI	3468
AuEu⁺						
	EuAu	56214-25-6	**	5.6±1.0	EI	4529
Au₂Eu⁺						
	EuAu ₂	51198-56-2	**	5.9±1.0	EI	4529
HoAu⁺						
	AuHo	12044-80-3	**	6.2±0.5	EI	3440
Hg⁺						
(² S _{1/2})	Hg	7439-97-6	** ** ** ** ** ** **	10.4 14.8 10.487±0.005 14.907±0.015 16.787±0.015 18.050±0.050 10.47±0.05	PE PE PEN PEN PEN PEN EI	3672 3672 3541 3541 3541 3541 3745
Hg₂⁺						
	Hg ₂	12596-25-7	**	9.40±0.08	EI	5428
Hg₃⁺						
	Hg ₃	11062-37-6	**	8.90±0.08	EI	5428
Hg₄⁺						
	Hg ₄	XXXXX-XX-X	**	8.65±0.08	EI	5428

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
Hg ₅ ⁺	Hg ₅	XXXXX-XX-X **		8.60±0.08	EI	5428
Hg ₆ ⁺	Hg ₆	XXXXX-XX-X **		8.50±0.08	EI	5428
Hg ₇ ⁺	Hg ₇	XXXXX-XX-X **		8.35±0.08	EI	5428
Hg ₈ ⁺	Hg ₈	XXXXX-XX-X **		8.28±0.08	EI	5428
Hg ₉ ⁺	Hg ₉	XXXXX-XX-X **		8.25±0.08	EI	5428
Hg ₁₀ ⁺	Hg ₁₀	XXXXX-XX-X **		8.25±0.08	EI	5428
Hg ₁₁ ⁺	Hg ₁₁	XXXXX-XX-X **		8.22±0.08	EI	5428
Hg ₁₂ ⁺	Hg ₁₂	XXXXX-XX-X **		8.12±0.08	EI	5428
C ₁₂ H ₁₀ Hg ⁺	(C ₆ H ₅) ₂ Hg (Mercury, diphenyl-)	587-85-9	**	8.30±0.03	PI	4055
C ₂ H ₆ Hg ⁺	(CH ₃) ₂ Hg	593-74-8	** **	9.3 (V) 9.33 (V)	PE PE	5300 4574
C ₃ H ₈ Hg ⁺	(CH ₃)(C ₂ H ₅)Hg	29138-86-1	**	8.84 (V)	PE	4574
C ₄ H ₁₀ Hg ⁺	(C ₂ H ₅) ₂ Hg	627-44-1	** **	8.45 (V) 8.9 (V)	PE PE	4574 5300
	(CH ₃)(iso-C ₃ H ₇)Hg	29138-88-3	**	8.48 (V)	PE	4574
C ₅ H ₁₂ Hg ⁺	(C ₂ H ₅)(iso-C ₃ H ₇)Hg (CH ₃)(iso-C ₄ H ₉)Hg (CH ₃)(tert-C ₄ H ₉)Hg	59049-79-5 59643-44-6 59049-78-4	** ** **	8.18 (V) 8.75 (V) 8.31 (V)	PE PE PE	4574 4574 4574
C ₆ H ₁₁ Hg ⁺	(n-C ₃ H ₇) ₂ Hg (iso-C ₃ H ₇) ₂ Hg (C ₂ H ₅)(tert-C ₄ H ₉)Hg	628-85-3 1071-39-2 59049-80-8	** ** **	8.29 (V) 8.03 (V) 8.06 (V)	PE PE PE	4574 4574 4574
C ₇ H ₁₆ Hg ⁺	C ₇ H ₁₆ Hg	59049-81-9	**	7.73 (V)	PE	4574
C ₈ H ₁₈ Hg ⁺	(n-C ₁ H ₁₁) ₂ Hg (iso-C ₁ H ₁₁) ₂ Hg (tert-C ₁ H ₁₁) ₂ Hg	629-35-6 24470-76-6 23587-90-8	** ** **	8.35 (V) 8.30 (V) 7.57 (V)	PE PE PE	4574 4574 4574
C ₉ H ₂₀ Hg ⁺	(iso-C ₁ H ₁₁)(neo-C ₅ H ₁₁)Hg	59643-45-7	**	8.33 (V)	PE	4574

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₀H₁₀Hg⁺	(C ₅ H ₅) ₂ Hg (Mercurocene)	12083-67-9	**	8.4±0.1 (V)	PE	4853
C₁₀H₂₂Hg⁺	(<i>neo</i> -C ₅ H ₁₁) ₂ Hg	10284-49-8	**	8.30 (V)	PE	4574
C₈H₆O₂Hg⁺	(C ₄ H ₃ O) ₂ Hg (Mercury,di-2-furanyl-)	28752-79-6	**	8.39 (V)	PE	5323
	(C ₄ H ₃ O) ₂ Hg (Mercury,di-3-furanyl-)	28752-80-9	**	8.70 (V)	PE	5323
CN₃F₃Hg⁺	CF ₃ N ₃ Hg	51353-52-7	**	9.87 (V)	PE	4512
C₂NOF₃Hg⁺	CF ₃ NCOHg	51353-51-6	**	10.83 (V)	PE	4512
CNO₃F₃Hg⁺	CF ₃ ONO ₂ Hg	461-40-5	**	11.07 (V)	PE	4512
C₁₄H₃₈Si₄Hg⁺	(CH(Si(CH ₃) ₃) ₂) ₂ Hg	13294-24-1	**	8.12±0.05 (V)	PE	4725
C₁₂H₃₆N₂Si₄Hg⁺	(N(Si(CH ₃) ₃) ₂) ₂ Hg	4104-81-8	**	8.33±0.05 (V)	PE	4725
C₈H₆S₂Hg⁺	(C ₄ H ₃ S) ₂ Hg (Mercury,di-2-thienyl-)	5980-89-2	**	8.47 (V)	PE	5323
	(C ₄ H ₃ S) ₂ Hg (Mercury,di-3-thienyl-)	28752-81-0	**	8.72 (V)	PE	5323
C₂F₆S₂Hg⁺	(SCF ₃) ₂ Hg	XXXXX-XX-X	**	10.2 (V)	PE	4512
Cl₂Hg⁺	HgCl ₂	7487-94-7	**	11.5 (V)	PE	3963
C₃H₅ClHg⁺	CH ₂ =CHCH ₂ HgCl	14155-77-2	**	9.35 (V)	PE	3859
C₇H₇ClHg⁺	C ₆ H ₅ CH ₂ HgCl (Mercury, chloro(phenylmethyl)-)	2117-39-7	**	8.65 (V)	PE	4490
C₄H₃OClHg⁺	C ₄ H ₃ O(HgCl) (Mercury,chloro-2-furanyl-)	5857-37-4	**	8.96 (V)	PE	5323
	C ₄ H ₃ O(HgCl) (Mercury,chloro-3-furanyl-)	5857-38-5	**	9.10 (V)	PE	5323
C₅H₅OClHg⁺	C ₄ H ₃ O(CH ₂ HgCl) (Mercury,chloro(3-furanylmethyl)-)	73057-78-0	**	8.80 (V)	PE	5323
C₄H₃SClHg⁺	C ₄ H ₃ S(HgCl) (Mercury,chloro-2-thienyl-)	5857-39-6	**	9.05 (V)	PE	5323
	C ₄ H ₃ S(HgCl) (Mercury,chloro-3-thienyl-)	73057-79-1	**	9.23 (V)	PE	5323

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₅H₅SClHg⁺	C ₅ H ₅ S(CH ₂ HgCl) (Mercury, chloro(3-thienylmethyl)-)	73057-80-4	**	8.79 (V)	PE	5323
CF₃IHg⁺	CF ₃ HgI	421-11-4	**	9.89 (V)	PE	4512
Tl⁺	TlBO ₂	XXXXX-XX-X	BO ₂	10.43±0.07	EI	4096
Tl₂⁺	Tl ₂ O	1314-12-1		11.97±0.09	EI	4096
C₅H₅Tl⁺	C ₅ H ₅ Tl (Thallium, (η^5 -2,4-cyclopentadien-1-yl)-)	34822-90-7	** **	7.96 (V) 8.12±0.05 (V)	PE PE	4777 4853
OTl⁺	TlBO ₂	XXXXX-XX-X		10.68±0.11	EI	4096
OTl₂⁺	Tl ₂ O	1314-12-1	**	8.02±0.10	EI	4096
BO₂Tl⁺	TlBO ₂	XXXXX-XX-X	** **	10.2±0.05 (V) 9.92±0.11	PE EI	4871 4096
BO₂Tl₂⁺	(TlBO ₂) ₂	XXXXX-XX-X		9.17±0.10	EI	4096
NO₃Tl⁺	TlNO ₃	XXXXX-XX-X	**	9.9±0.05 (V)	PE	4871
FTl⁺ ($^2\Sigma$)	TlF	7789-27-7	**	10.80±0.02 (V)	PE	4552
FTl⁺ ($^2\Pi$)			**	11.90±0.02 (V)	PE	4552
FTl⁺ ($^2\Sigma$)			**	14.20±0.02 (V)	PE	4552
FTl⁺ ($^2\Sigma$)			**	10.52	PE	3971
FTl⁺ ($^2\Pi$)			**	11.15	PE	3971
FTl⁺ ($^2\Sigma$)			**	14.05	PE	3971
FTl₂⁺	(TlF) ₂	31970-97-5		9.97±0.02	PI	3971
F₂Tl₂⁺	(TlF) ₂	31970-97-5	** ** **	9.71±0.02 9.62 9.96±0.02 (V)	PI PE PE	3971 3971 4552
O₄STl₂⁺	Tl ₂ SO ₄	XXXXX-XX-X	**	9.8±0.05 (V)	PE	4871
CITl⁺ ($^2\Sigma$)	TlCl	7791-12-0	** **	9.894 (V) 9.91 (V)	PE PE	3913 4826
CITl⁺ ($^2\Sigma_{1/2}$)			**	9.92 (V)	PE	4713
CITl⁺ ($^2\Pi$)			**	9.925 (V)	PE	3536
CITl⁺ ($^2\Pi_{3/2} + ^2\Pi_{1/2}$)			**	10.38 (V)	PE	4713
CITl⁺ ($^2\Pi$)			**	10.384 (V)	PE	3913
CITl⁺ ($^2D_{5/2}$)			**	11.04 (V)	PE	4713
CITl⁺ ($^2D_{3/2}$)			**	11.95 (V)	PE	4713

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
CITI⁺						
(² P _{3/2})	TlCl	7791-12-0	**	10.38 (V)	PE	4713
(² D _{3/2})			**	13.17 (V)	PE	4713
(² D _{5/2})			**	13.41 (V)	PE	4713
(² D _{5/2})			**	13.68 (V)	PE	4713
(² Σ)			**	13.79	PE	3913
(² Σ _{1/2})			**	13.89 (V)	PE	4713
(² D _{3/2})			**	15.86 (V)	PE	4713
(² Π _{1/2})			**	18.55 (V)	PE	4713
(² Π _{3/2})			**	20.23 (V)	PE	4713
(² D _{5/2})			**	20.97 (V)	PE	4713
(² D _{5/2})			**	21.16 (V)	PE	5035
(² D _{5/2})			**	21.24 (V)	PE	4713
(² D _{5/2})			**	21.41 (V)	PE	4713
(² D _{3/2})			**	23.30 (V)	PE	4713
(² D _{3/2})			**	23.32 (V)	PE	5035
(² D _{3/2})			**	23.42 (V)	PE	4713
AsTl⁺						
TlAs		12006-09-6	**	9±1	EI	3947
BrTl⁺						
(² Σ _{1/2})	TlBr	7789-40-4	**	9.50 (V)	PE	4713
(² Π)			**	9.832 (V)	PE	3913
(² Π _{3/2} + ² Π _{1/2})			**	9.85 (V)	PE	4713
(² Σ)			**	13.57	PE	3913
(² Σ _{1/2})			**	13.69 (V)	PE	4713
(² Π _{1/2} ?)			**	17.78 (V)	PE	4713
(² Π _{3/2})			**	20.60 (V)	PE	4713
(² D _{5/2})			**	20.86 (V)	PE	4713
(² D _{5/2})			**	21.04 (V)	PE	5035
(² D _{5/2})			**	21.13 (V)	PE	4713
(² D _{3/2})			**	23.11 (V)	PE	4713
(² D _{3/2})			**	23.23 (V)	PE	5035
(² D _{3/2})			**	23.25 (V)	PE	4713
TI⁺						
TlI		7790-30-9	**	8.47±0.02	PI	3536
(² Σ _{1/2} , ² Π _{3/2})			**	8.47±0.02	PE	3913
(² Σ _{1/2} + ² Π _{3/2})			**	8.89 (V)	PE	4713
(² Π)			**	8.93 (V)	PE	3676
(² Π _{1/2})			**	9.39	PE	3913
(² Σ)			**	9.73 (V)	PE	4713
(² Σ _{1/2})			**	13.0	PE	3913
(² Σ _{1/2})			**	13.10 (V)	PE	4713
(² Π _{1/2} ?)			**	13.47 (V)	PE	4713
(² D _{5/2})			**	18.07 (V)	PE	4713
(² D _{5/2})			**	20.59 (V)	PE	4713
(² D _{5/2})			**	20.75 (V)	PE	5035
(² D _{5/2})			**	20.78 (V)	PE	4713
(² D _{3/2})			**	22.87 (V)	PE	4713
(² D _{3/2})			**	23.04 (V)	PE	5035
(² D _{3/2})			**	23.05	PE	4713
O₁ReTl⁺						
TlReO ₁		XXXXX-XX-X	**	10.6±0.05 (V)	PE	4871
Pb⁺						
(² P _{1/2})	Pb	7439-92-1	**	7.417	S	5449
(² P _{3/2})			**	9.163	S	5449
(² P _{1/2})			**	7.42±0.01	PE	5534
(² P _{3/2})			**	9.16±0.01	PE	5534
(¹ P _{1/2})			**	14.59±0.01	PE	5534

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
Pb⁺						
(¹ P _{3/2})	Pb	7439-92-1	**	15.61±0.01	PE	5534
(² D _{5/2})			**	15.97±0.01	PE	5534
(² D _{3/2})			**	16.06±0.01	PE	5534
(¹ P _{1/2})			**	16.57±0.01	PE	5534
(³ P _{1/2})			**	18.35±0.01	PE	5534
LiPb⁺	PbLi	12372-50-8	**	6.4±0.5	EI	5426
C₃H₉Pb⁺						
(CH ₃) ₄ Pb		75-74-1	CH ₃	8.77±0.16	EI	3548
(tert-C ₄ H ₉)(CH ₃) ₃ Pb		32997-03-8	(CH ₃) ₃ C	8.67±0.21	EI	3548
((CH ₃) ₃ Pb) ₂		6713-83-3	(CH ₃) ₃ Pb	9.02±0.14	EI	3548
C ₆ H ₅ SPb(CH ₃) ₃	(Plumbane, trimethyl(phenylthio)-)	40560-63-2		8.37±0.1	EI	4198
C₄H₁₂Pb⁺						
(CH ₃) ₄ Pb		75-74-1	**	8.50±0.04	PE	3880
			**	8.83±0.1	PE	3677
			**	8.26±0.17	EI	3548
C₇H₁₈Pb⁺	(tert-C ₄ H ₉)(CH ₃) ₃ Pb	32997-03-8	**	7.99±0.13	EI	3548
C₉H₁₄Pb⁺						
C ₆ H ₅ (CH ₃) ₃ Pb	(Plumbane, trimethylphenyl-)	19040-53-0	**	~8.82	PE	4589
C₁₀H₁₀Pb⁺						
(C ₅ H ₅) ₂ Pb	(Plumbocene)	1294-74-2	**	7.53±0.05 (V)	PE	4853
C₁₀H₁₆Pb⁺						
C ₆ H ₅ CH ₂ (CH ₃) ₃ Pb	(Plumbane, trimethyl(phenylmethyl)-)	54338-54-4	**	7.87±0.05	PE	4589
C₆H₁₈Pb₂⁺	((CH ₃) ₃ Pb) ₂	6713-83-3	**	7.41±0.10	EI	3548
C₁₄H₂₈N₄Pb⁺						
C ₂₀ H ₈ N ₄ (C ₆ H ₅) ₄ Pb	(Lead, [5,10,15,20-tetraphenyl-21H,23H-porphinato(2-)—N ²¹ ,N ²² ,N ²³ ,N ²⁴]—(SP-4-1)-)	14784-17-9	**	5.99±0.2	OTH	4962
OPb⁺						
PbO		1317-36-8	**	9.08±0.10	EI	5163
O₂Pb⁺						
PbO ₂		1309-60-0	**	8.87±0.10	EI	5163
C₁₄H₃₈Si₄Pb⁺						
(CH(Si(CH ₃) ₃) ₂) ₂ Pb		41823-73-8	**	7.25±0.05 (V)	PE	4725
C₁₆H₄₄Si₄Pb⁺						
((CH ₃) ₂ SiCH ₂) ₄ Pb		18547-13-2	**	8.14±0.1 (V)	PE	3830
C₁₄H₃₆N₂Si₂Pb⁺						
(N(Si(CH ₃) ₃)(tert-C ₄ H ₉)) ₂ Pb		55147-79-0	**	7.26±0.05 (V)	PE	4725
			**	7.18 (V)	PE	4157

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
$\text{C}_{12}\text{H}_{36}\text{N}_2\text{Si}_4\text{Pb}^+$	$(\text{N}(\text{Si}(\text{CH}_3)_3)_2)_2\text{Pb}$	55147-59-6	** **	7.92 (V) 7.92 ± 0.05 (V)	PE PE	4157 4725
$\text{C}_4\text{H}_{12}\text{SPb}^+$	$(\text{CH}_3)_3\text{SCH}_3\text{Pb}$	14326-59-1	**	8.13 ± 0.05 (V)	PE	4153
$\text{C}_8\text{H}_{11}\text{SPb}^+$	$\text{C}_6\text{H}_5\text{SPb}(\text{CH}_3)_3$ (Plumbane, trimethyl(phenylthio)-)	40560-63-2	CH_3	8.11 ± 0.1	EI	4198
$\text{C}_9\text{H}_{11}\text{SPb}^+$	$\text{C}_6\text{H}_5\text{S}(\text{CH}_3)_3\text{Pb}$ (Plumbane, trimethyl(phenylthio)-)	40560-63-2	** **	8.15 ± 0.05 7.75 ± 0.1	PE EI	4589 4198
$\text{C}_{10}\text{H}_{16}\text{SPb}^+$	$\text{C}_6\text{H}_5(\text{SCH}_3)(\text{CH}_3)_3\text{Pb}$ (Plumbane, trimethyl[4-(methylthio)phenyl]-)	59163-57-4	**	<8.02 (V)	PE	4627
$\text{C}_6\text{H}_{18}\text{SPb}_2^+$	$((\text{CH}_3)_3\text{Pb})_2\text{S}$	14511-33-2	**	7.78 ± 0.05 (V)	PE	4153
Cl_2Pb^+	PbCl_2 $(^2\text{D}_{5/2})$ $(^2\text{D}_{3/2})$	7758-95-4	** ** **	10.11 (V) 27.34 (V) 29.92 (V)	PE PE PE	3650 5035 5035
$\text{C}_3\text{H}_9\text{ClPb}^+$	$(\text{CH}_3)_3\text{PbCl}$	1520-78-1	**	9.70 (V)	PE	4566
Br_2Pb^+	PbBr_2 $(^2\text{B}_1)$ $(^2\text{D}_{5/2})$ $(^2\text{D}_{3/2})$	10031-22-8	** ** **	9.81 ± 0.05 (V) 27.02 (V) 29.58 (V)	PE PE PE	4826 5035 5035
$\text{C}_3\text{H}_9\text{BrPb}^+$	$(\text{CH}_3)_3\text{PbBr}$	6148-48-7	**	9.30 (V)	PE	4566
TePb^+	TePb $(^2\Pi_{3/2})$ $(^2\Pi_{1/2})$ $(^2\Sigma)$	1314-91-6	** ** **	8.04 (V) 8.34 (V) 9.01 (V)	PE PE PE	4550 4550 4550
I_2Pb^+	PbI_2 $(^2\text{D}_{5/2})$ $(^2\text{D}_{3/2})$	10101-63-0	** ** **	8.86 ± 0.03 26.48 (V) 29.20 (V)	PI PE PE	3536 5035 5035
Bi^+	Bi Bi_2	7440-69-9 12187-12-1	**	7.2 ± 0.5 9.6 ± 0.5	EI EI	4128 4128
Bi_2^+	Bi_2	12187-12-1	** **	7.3 ± 0.5 7.6 ± 0.5	EI EI	4120 4128
Bi_3^+	Bi_3	XXXXX-XX-X	**	8.8 ± 0.5	EI	4128
Bi_4^+	Bi_4	12595-65-2	**	7.3 ± 0.5	EI	4128

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
LiBi⁺	BiLi	12048-27-0	**	6.0±0.5	EI	5426
C₅H₅Bi⁺	C ₅ H ₅ Bi (Bismin)	289-52-1	**	7.9	PE	4416
C₆H₅Bi⁺	(C ₆ H ₅) ₃ Bi (Bismuthine, triphenyl-)	603-33-8		7.75±0.1	PI	4325
C₁₂H₁₀Bi⁺	(C ₆ H ₅) ₃ Bi (Bismuthine, triphenyl-)	603-33-8		7.9±0.1	PI	4325
C₁₈H₁₅Bi⁺	(C ₆ H ₅) ₃ Bi (Bismuthine, triphenyl-)	603-33-8	**	7.45±0.05	PI	4325
F₃Bi⁺	BiF ₃	7787-61-3	**	~12	EI	3551
F₄Bi⁺	BiF ₅	7787-62-4		14.5-15	EI	3551
GaBi⁺	GaBi	12010-43-4	**	7±1	EI	3608
TlBi⁺	BiTl	26257-16-9	**	7.5±0.4	EI	3949
Ac⁺	Ac	7440-34-8	**	5.17±0.12	OTH	3875
Th⁺	Th	7440-29-1	**	6.11±0.02	PE	5052
(¹ S ₀)			**	12.22±0.07	PE	5052
(³ P ₀)			**	12.56±0.06	PE	5052
(³ P ₁)			**	13.75±0.04	PE	5052
(³ P ₂)			**	15.49±0.03	PE	5052
(¹ P ₁)			**	5.9±0.15	EI	3962
			**	6.0±0.1	EI	4114
			**	6.2±0.2	EI	4123
			**	6.8	EI	4119
			**	6.9±0.5	EI	4909
			**	6.9±0.5	EI	5306
			**	7.0±0.5	EI	4208
			**	6.08±0.12	OTH	3875
			**	7.4±0.3	OTH	5149
ThO		12035-93-7	O	15.9±0.1	EI	4123
				16	EI	4208
CTh⁺	CTh	12012-16-7	**	7.9±1.0	EI	5306
			**	8.0±1.0	EI	4112
C₂Th⁺	C ₂ Th	12071-31-7	**	6.4±0.5	EI	5306
			**	6.5±0.3	EI	4112
C₃Th⁺	C ₃ Th	XXXXX-XX-X	**	8.4±1.0	EI	5306

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₃Th⁺	C ₃ Th		XXXXX-XX-X **	9.2±1.0	EI	4112
C₄Th⁺	C ₄ Th	52931-63-2	** **	9.0±0.5 10.0±1.0	EI EI	5306 4112
C₁₆H₁₆Th⁺	(C ₄ H ₈) ₂ Th (Thorium, bis(η^8 -1,3,5,7-cyclooctatetraene)-)	12702-09-9	** **	6.75 (V) 6.79 (V)	PE PE	4562 4612
OTh⁺	ThO	12035-93-7	** ** ** **	>6.0±0.1 6.1±0.1 6.1±0.15 6.1	EI EI EI EI	4208 4114 3962 4119
O₂Th⁺	ThO ₂	1314-20-1	** ** **	8.7±0.15 8±1 8.7±0.15	EI EI EI	4114 4208 3962
C₂₀H₂₈O₈Th⁺	((CH ₃ CO) ₂ CH) ₄ Th	17499-48-8	**	7.85 (V)	PE	5338
Cl₄Th⁺	ThCl ₄	10026-08-1	**	12.7±0.3	EI	3795
C₁₅H₁₅ClTh⁺	(C ₅ H ₅) ₃ ThCl (Thorium, chlorotris(η^5 -2,4-cyclopentadien-1-yl)-)	1284-82-8	**	7.85 (V)	PE	4585
C₁₈H₂₁ClTh⁺	(C ₅ H ₄ CH ₃) ₃ ThCl (Thorium, chlorotris(1,2,3,4,5- η)-1-methyl-2,4-cyclopentadien-1-yl]-)	62156-90-5	**	7.75 (V)	PE	4585
RuTh⁺	RuTh	52014-55-8	** **	6.4±0.5 7.1±1.0	EI EI	4909 4130
PtTh⁺	ThPt	12038-30-1	**	8±2	EI	3968
Pa⁺	Pa	7440-13-3	**	5.89±0.12	OTH	3875
U⁺	U	7440-61-1	** ** ** ** ** ** ** ** **	6.22±0.5 6.0±0.5 6.0±0.5 6.1±0.1 6.1±0.3 6.8±1.5 ~6±0.5 6.05±0.07 6.3±0.3	S EI EI EI EI EI EI OTH OTH	3566 4909 5169 4114 3557 3595 3448 3875 5149
U⁺²	U ⁺	15721-70-7	**	10.6±1	S	3566

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
H₁₆B₄U⁺	U(BH ₄) ₄	12523-77-2	** ** **	9.58±0.1 (V) 9.59± (V) 9.0±0.5	PE PE EI	4825 4888 5375
CU⁺	UC	12070-09-6	**	7.8±1.0	EI	5169
C₂U⁺	UC ₂	12071-33-9	**	6.4±0.5	EI	5169
C₃U⁺	UC ₃	XXXXX-XX-X	**	8.1±1.0	EI	5169
C₄U⁺	UC ₄	XXXXX-XX-X	**	8.7±0.5	EI	5169
C₁₆H₁₆U⁺	(C ₄ H ₈) ₂ U (Uranium, bis(η^5 -1,3,5,7-cyclooctatetraene)-)	11079-26-8	** **	6.15 (V) 6.20 (V)	PE PE	4562 4612
BC₁₈H₂₅U⁺	(C ₅ H ₅ CH ₃) ₃ UBH ₄ (Uranium, tris[(1,2,3,4,5- η)-1-methyl-2,4-cyclopentadien-1-yl]-tetrahydroborate(1-))	62156-96-1	**	6.35 (V)	PE	4585
OU⁺	UO	12035-97-1	** ** ** **	4.3±1.5 5.6±0.1 5.7±0.4 ~6±0.5	EI EI EI EI	3595 4114 3557 3448
O₂U⁺	UO ₂	1344-57-6	** ** **	4.5±1.5 5.4±0.1 5.5±0.4	EI EI EI	3595 4114 3557
O₃U⁺	UO ₃	1344-58-7	** ** **	9.5±1.5 10.6±0.1 11.1±0.4	EI EI EI	3595 4114 3557
C₁₀H₁₄O₆U⁺	((CH ₃ CO) ₂ CH) ₂ UO ₂	18039-69-5	**	8.40 (V)	PE	5338
C₂₀H₂₈O₈U⁺	((CH ₃ CO) ₂ CH) ₄ U	17923-26-1	**	6.65 (V)	PE	5338
F₄U⁺	UF ₄	10049-14-6	** ** **	9.51 9.96±0.1 10.0±0.3	PE EI EI	5371 4865 4865
	UF ₆	7783-81-5	2F	17.35±0.1	EI	4865
F₅U⁺	UF ₅	13775-07-0	** **	11.28±0.1 11.5±0.3	EI EI	4865 4865
	UF ₆	7783-81-5	F	14.24±0.10	EI	4865
F₆U⁺	UF ₆	7783-81-5	**	14.00±0.10	EI	4865

Table of Ion Energetics Measurements—Continued

Ion (state)	Reactant	CAS Registry Number	Other products	Ionization or appearance potential (eV)	Method	Ref.
C₁₀H₂O₆F₁₂U⁺	((CF ₃ CO) ₂ CH) ₂ UO ₂	67316-66-9	**	10.05 (V)	PE	5338
C₂₀H₁₆O₈F₁₂U⁺	(CF ₃ COCHCOCH ₃) ₄ U	32627-13-7	**	7.83 (V)	PE	5338
OSU⁺	UOS	22201-28-1	**	~8±0.5	EI	3448
Cl₁U⁺	UCl ₁	10026-10-5	** **	9.18 11.0±0.3	PE EI	5371 3795
C₁₅H₁₅ClU⁺	(C ₅ H ₅) ₃ UCl (Uranium, chlorotris(η ⁵ -2,4-cyclopentadien-1-yl)-)	1284-81-7	**	6.90 (V)	PE	4585
C₁₈H₂₁ClU⁺	(C ₅ H ₁ CH ₃) ₃ UCl (Uranium, chlorotris[(1,2,3,4,5-η)-1-methyl-2,4-cyclopentadien-1-yl]-)	59834-82-1	**	7.10 (V)	PE	4585
C₁₈H₂₁BrU⁺	(C ₅ H ₁ CH ₃) ₃ UBr (Uranium, bromotris[(1,2,3,4,5-η)-1-methyl-2,4-cyclopentadien-1-yl]-)	62050-82-2	**	6.95 (V)	PE	4585
RuU⁺	RuU	12316-41-5	**	6.1±0.5	EI	4909
Np⁺	Np	7439-99-8	** ** **	6.1±0.1 6.20±0.12 6.2657±0.0005	EI OTH S	4560 3875 5165
ONp⁺	NpO	XXXXX-XX-X	**	5.7±0.1	EI	4560
Pu⁺	Pu	7440-07-5	**	6.06±0.02	OTH	3875
Am⁺	Am	7440-35-9	**	5.993±0.010	OTH	3875
Cm⁺	Cm	7440-51-9	**	6.09±0.02	OTH	3875
Bk⁺	Bk	7440-40-6	**	6.30±0.09	OTH	3875
Cf⁺	Cf	7440-71-3	**	6.41±0.10	OTH	3875
Es⁺	Es	7429-92-7	**	6.52±0.10	OTH	3875
Fm⁺	Fm	7440-72-4	**	6.64±0.11	OTH	3875
Md⁺	Md	7440-11-1	**	6.74±0.12	OTH	3875
No⁺	No	10028-14-5	**	6.84±0.12	OTH	3875

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- [3440] Cocke, D. L., and Gingerich, K. A. Mass spectrometric determination of the dissociation energies of the molecules Ho_2 , HoAg , and HoAu , *J. Phys. Chem.* **75**, 3264 (1971).
- [3441] Baylis, A. B., Pressley, G. A., Jr., and Stafford, F. E. Mass spectrometric investigation of the pyrolysis of boranes. IV. Diborane, *J. Am. Chem. Soc.* **88**, 2428 (1966).
- [3442] DeCorpo, J. J., Bafus, D. A., and Franklin, J. L. Enthalpies of formation of the monohalomethyl radicals from mass spectrometric studies of the dihalomethanes, *J. Chem. Thermodyn.* **3**, 125 (1971).
- [3443] Bentley, T. W., Johnstone, R. A. W., and Mellon, F. A. Aspects of mass spectra of organic compounds. Part IX. Evidence against charge localization in the fragmentation of methionine and selenomethionine, *J. Chem. Soc. (B)*, 1800 (1971).
- [3444] Borossay, J., Csákvári, B., and Szepes, L. Determination of bond energies of organic silicon compounds on the basis of appearance potentials, *Intern. J. Mass Spectrom. Ion Phys.* **7**, 47 (1971).
- [3445] Daly, N. R. Higher autoionization processes in argon and xenon, *Proc. Phys. Soc.* **85**, 897 (1965).
- [3446] Brown, P. Kinetic studies in mass spectrometry. VIII. Competing $[\text{M}-\text{CH}_3]$ and $[\text{M}-\text{CH}_2\text{O}]$ reactions in substituted anisoles. Approximate activation energies from ionization and appearance potentials, *Org. Mass Spectrom.* **4**, 519 (1970).
- [3447] Brown, P. Kinetic studies in mass spectrometry. IX. Competing $[\text{M}-\text{NO}_2]$ and $[\text{M}-\text{NO}]$ reactions in substituted nitrobenzenes. Approximate activation energies from ionization and appearance potentials, *Org. Mass Spectrom.* **4**, 533 (1970).
- [3448] Cater, E. D., Rauh, E. G., and Thorn, R. J. Thermochemistry of UOS; evaporation of US-UO₂ mixtures; on the attainment of equilibrium in Knudsen cells, *J. Chem. Phys.* **49**, 5244 (1968).
- [3449] Edwards, J. G., Franzen, H. F., and Gilles, P. W. High-temperature mass spectrometry, vaporization, and thermodynamics of titanium monosulfide, *J. Chem. Phys.* **54**, 545 (1971).
- [3450] Gupta, S. K. A thermodynamic investigation of the tungsten-oxygen-bromine system, *J. Phys. Chem.* **75**, 112 (1971).
- [3451] Gupta, S. K. Thermal stabilities of tungsten oxyiodides, *J. Phys. Chem.* **73**, 4086 (1969).
- [3452] Daly, N. R., and Powell, R. E. Electron collisions in nitrogen, *Proc. Phys. Soc.* **89**, 273 (1966).
- [3453] Cantone, B., Emma, V., and Grasso, F. Fine structure near the ionization threshold of Kr, O₂, NO by electron impact, *Advan. Mass Spectrom.* **4**, 599 (1968).
- [3454] Dougherty, R. C., Bertorello, H. E., and Martínez de Bertorello, M. Mass spectra and thermochemistry of methyl phenanthrenes. A contribution to the analogy between mass spectral and thermal fragmentation reactions, *Org. Mass Spectrom.* **5**, 1321 (1971).
- [3455] Guido, M., Balducci, G., Gigli, G., and Spoliti, M. Mass spectrometric study of the vaporization of cuprous chloride and the dissociation energy of Cu_2Cl_3 , Cu_3Cl_5 , and Cu_4Cl_7 , *J. Chem. Phys.* **55**, 4566 (1971).
- [3456] Smagina, E. I., and Kutsev, V. S. A mass-spectrometric study of the mechanism of the decomposition of lanthanum oxide fluoride, *Zh. Fiz. Khim.* **45**, 46 (1971) [Engl. transl.: *Rus. J. Phys. Chem.* **45**, 24 (1971)].
- [3457] Stearns, C. A., and Kohl, F. J. Vaporization thermodynamics of the lanthanum carbon system. Mass spectrometric determination of the dissociation energy of LaC_2 , LaC_3 , and LaC_5 , *J. Chem. Phys.* **54**, 5180 (1971).
- [3458] Smoes, S., Myers, C. E., and Drowart, J. Determination of the atomization energies of CP, C_2P , CP_2 and C_2P_2 by high temperature Knudsen cell mass spectrometry, *Chem. Phys. Letters* **8**, 10 (1971).
- [3459] Hariharan, A. V., and Eick, H. A. Vaporization thermodynamics of europium(II) sulfide, *High Temp. Sci.* **3**, 123 (1971).
- [3460] Seiver, R. L., and Eick, H. A. Vapor pressure measurements in the samarium dicarbide-carbon and thulium dicarbide-carbon systems, *High Temp. Sci.* **3**, 292 (1971).
- [3461] Schoonmaker, R. C., and Porter, R. F. Mass spectrometric study of alkali hydroxide vapors, *J. Chem. Phys.* **31**, 830 (1959).
- [3462] Srivastava, R. D., and Farber, M. Mass spectrometric determination of the heats of formation of AlOCl(g) and AlOF(g) , *J. Phys. Chem.* **75**, 1760 (1971).
- [3463] Farber, M., Srivastava, R. D., and Uy, O. M. Mass spectrometric determination of the heat of formation of the AlO_2 molecule, *J. Chem. Phys.* **55**, 4142 (1971).
- [3464] Porter, R. F., and Schoonmaker, R. C. Mass spectrometric study of the vaporization of LiF , NaF , and LiF-NaF mixtures, *J. Chem. Phys.* **29**, 1070 (1958).
- [3465] Srivastava, R. D., and Farber, M. Thermodynamic properties of the B-Cl-F system from mass spectrometer investigations, *Trans. Faraday Soc.* **67**, 2298 (1971).
- [3466] Skinner, H. B., and Searcy, A. W. Demonstration of the existence of La_2F_6 gas and determination of its stability, *J. Phys. Chem.* **75**, 108 (1971).
- [3467] Visnapuu, A., and Marek, B. C. Properties of silver bromide vapors, *J. Less-Common Metals* **25**, 89 (1971).
- [3468] Gingerich, K. A. Gaseous metal borides. III. The dissociation energy and heat of formation of gold monoboride, *J. Chem. Phys.* **54**, 2646 (1971).
- [3469] Gingerich, K. A. Gaseous metal nitrides. IV. The dissociation energy of cerium mononitride, *J. Chem. Phys.* **54**, 3720 (1971).
- [3470] Kohl, F. J., and Stearns, C. A. Mass spectrometric determination of the dissociation energy of ScC_2 and ScC_3 , *J. Chem. Phys.* **54**, 1414 (1971).
- [3471] Gingerich, K. A., and Finkbeiner, H. C. Dissociation energy of diatomic cerium and predicted stability of gaseous intermetallic cerium compounds, *J. Chem. Phys.* **54**, 2621 (1971).
- [3472] Gingerich, K. A., and Piacente, V. Gaseous phosphorus compounds. IV. Thermodynamic study of gallium monophosphide with a mass spectrometer and dissociation energy of aluminum diphosphide, *J. Chem. Phys.* **54**, 2498 (1971).
- [3473] Gingerich, K. A., and Pupp, C. Mass spectrometric determination of the heats of formation and atomization of gaseous AuBO , *J. Chem. Phys.* **54**, 3713 (1971).
- [3474] Glockling, F., and Strafford, R. G. Electron impact studies on some group III metal alkyls, *J. Chem. Soc. (A)*, 1761 (1971).
- [3475] Munir, Z. A., Street, G. B., and Winters, H. F. Mass-spectrometric and vapor pressure studies on the sublimation of realgar (As_4S_4), *J. Chem. Phys.* **55**, 4520 (1971).
- [3476] Hedaya, E., Kent, M. E., McNeil, D. W., Lossing, F. P., and McAllister, T. The thermal rearrangement of phenylnitrene to cyanocyclopentadiene, *Tetrahedron Letters* **30**, 3415 (1968).
- [3477] Harris, M. M., Loudon, A. G., and Mazengo, R. Z. Ring expansion reactions in aromatic systems. A study of steric strain in some *n,n'*-dimethyl-1,1'-binaphthyls, *Org. Mass Spectrom.* **5**, 1123 (1971).
- [3478] Simmie, J. M., and Tschiukow-Roux, E. Mass spectrum, appearance potentials and bond dissociation energies of

- 1,1,1-trifluoroethane, Intern. J. Mass Spectrom. Ion Phys. **7**, 41 (1971).
- [3479] Williams, D. H., Cooks, R. G., and Howe, I. Studies in mass spectrometry. XXXI. A comparison of reaction rates in common ions generated via fragmentation and direct ionization, J. Am. Chem. Soc. **90**, 6759 (1968).
- [3480] Benezra, S. A., and Bursey, M. M. *ortho*-Effects in mass spectra. Alteration of the molecular-ion energy distribution in disubstituted acetanilides, Z. Naturforsch. **27a**, 670 (1972).
- [3481] Pihlaja, K., and Jalonen, J. Appearance potentials determined by the electron-impact method as an analytical aid in the evaluation of conformational energies and clarification of ring conformation-I: Appearance potentials of the [M-R]⁺ ions formed in the primary fragmentation of stereoisomeric 1,3-dioxans. A direct route to conformational energies, Org. Mass Spectrom. **5**, 1363 (1971).
- [3482] Linda, P., Marino, G., and Pignataro, S. A comparison of sensitivities to substituent effects of five-membered heteroaromatic rings in gas phase ionization, J. Chem. Soc. (B), 1585 (1971).
- [3483] Benezra, S. A., and Bursey, M. M. *ortho*-Effects on ordering factors in mass spectral rearrangements. Loss of keten from halogenated phenyl acetates and acetanilides, J. Chem. Soc. (B), 1515 (1971).
- [3484] Gamble, A. A., Gilbert, J. R., and Tillett, J. G. Substituent effects on the mass spectra of substituted phenyl acetates, Org. Mass Spectrom. **5**, 1093 (1971).
- [3485] Johnstone, R. A. W., Mellon, F. A., and Ward, S. D. On-line computer methods used in conjunction with the measurement of ionization and appearance potentials, Advan. Mass Spectrom. **5**, 334 (1971).
- [3486] Okudaira, S. Multiple ionization of Ca, Sr and Ba by electron impact, J. Phys. Soc. Japan **29**, 409 (1970).
- [3487] Haney, M. A., and Franklin, J. L. Heats of formation of H₃O⁺, H₂S⁺, and NH₃⁺ by electron impact, J. Chem. Phys. **50**, 2028 (1969).
- [3488] Hickling, R. D., and Jennings, K. R. Kinetic shifts and metastable transitions, Org. Mass Spectrom. **3**, 1499 (1970).
- [3489] Redhead, P. A. Multiple ionization in carbon monoxide by successive electron impacts, Can. J. Phys. **47**, 2449 (1969).
- [3490] Franklin, J. L., and Haney, M. A. Energy distribution in ionic decomposition processes, Recent Developments in Mass Spectroscopy, ed. K. Ogata and T. Hayakawa (Baltimore University Park Press, Baltimore, 1970) p. 909.
- [3491] Makowiecki, D. M., Lynch, D. A., and Carlson, K. D. Infrared spectra of the aluminum family suboxides, J. Phys. Chem. **75**, 1963 (1971).
- [3492] Brion, C. E., Farmer, J. S. H., Pincock, R. E., and Stewart, W. B. Mass spectra of some geometric isomers at 1216 Å and 584 Å: The photoionization of isomeric tricyclo [3.2.1.0^{2,1}] octanes and related compounds, Org. Mass Spectrom. **4**, 587 (1970).
- [3493] Meisels, G. G., and Giessner, B. G. Threshold behavior and the determination of appearance potentials from second differential ionization efficiencies, Intern. J. Mass Spectrom. Ion Phys. **7**, 489 (1971).
- [3494] Hvistendahl, G., and Undheim, K. Ionization potentials of stable free radicals, Chemica Scripta **1**, 123 (1971).
- [3495] Cardin, D. J., Keppie, S. A., Lappert, M. F., Litzow, M. R., and Spalding, T. R. Binuclear organometallic compounds. Part III. Metal-metal bond dissociation energies, Raman, and infrared spectra for the series (π -C₅H₅)(CO)₂M¹M²Me₃: (M¹=Cr, Mo, or W; M²=Ge or Sn), J. Chem. Soc. (A), 2262 (1971).
- [3496] Bursey, M. M., and Rogerson, P. F. The electronimpact ionization potentials of successively substituted acetylacetones of rhodium(III), Inorg. Chem. **10**, 1313 (1971).
- [3497] Bonati, F., Distefano, G., Innorta, G., Minghetti, G., and Pignataro, S. Ionization energies of rhodium and iridium β -diketonates: on the nature of the last occupied orbital, Z. Anorg. Allg. Chem. **386**, 107 (1971).
- [3498] Distefano, G., Foffani, A., Innorta, G., and Pignataro, S. Mass spectrometric study of transition metal complexes with ligands having nitrogen or sulphur as donor atom, Advan. Mass Spectrom. **5**, 696 (1971).
- [3499] Cornford, A. B., Frost, D. C., Herring, F. G., and McDowell, C. A. The photoelectron spectrum of the free radical chlorine dioxide, Chem. Phys. Letters **10**, 345 (1971).
- [3500] Berkowitz, J. Experimental potential energy curves for X²H and ³S⁺ states of HF⁺, Chem. Phys. Letters **11**, 21 (1971).
- [3501] Brundle, C. R., and Jones, G. R. The molecular orbital energy levels and bonding in krypton difluoride, Chem. Commun., 1198 (1971).
- [3502] Cradock, S., and Ebsworth, E. A. V. Photo-electron spectra of silyl and germyl halides and (p→d) π -bonding, Chem. Commun., 57 (1971).
- [3503] Green, M. C., Lappert, M. F., Pedley, J. B., Schmidt, W., and Wilkins, B. T. Photoelectron spectra and energy level trends in Me_nSiCl_{4-n} and related series, J. Organometal. Chem. **31**, C55 (1971).
- [3504] Bock, H., and Ensslin, W. Bond-bond interaction in polysilanes, Angew. Chem. Intern. Ed. **10**, 404 (1971).
- [3505] Haselbach, E., Heilbronner, E., Mannschreck, A., and Seitz, W. Lone pair interaction in 3,3-dimethylazirine, Angew. Chem. Intern. Ed. **9**, 902 (1970).
- [3506] Lloyd, D. R., and Lyraugh, N. Photoelectron spectra of the symmetric trimethylborazines, Chem. Commun. **3**, 125 (1971).
- [3507] Anderson, C. P., Mamantov, G., Bull, W. E., Grimm, F. A., Carver, J. C., and Carlson, T. A. Photoelectron spectrum of chlorine monofluoride, Chem. Phys. Letters **12**, 137 (1971).
- [3508] Cradock, S. The photoelectron spectra of GeH₄ and GeF₄, Chem. Phys. Letters **10**, 291 (1971).
- [3509] Bischof, P., Heilbronner, E., Prinzbach, H., and Martin, H. D. A photoelectron-spectroscopic investigation of the homoconjugative interaction between π - and *Walsh*-orbitals in *endo*- and *exo*-cyclopropano-norbornene, Helv. Chim. Acta **54**, 1072 (1971).
- [3510] Cradock, S., and Whiteford, R. A. Photo-electron spectra of the mono and dihalo silanes and germanes, Trans. Faraday Soc. **67**, 3425 (1971).
- [3511] Frost, D. C., Herring, F. G., Katrib, A., McLean, R. A. N., Drake, J. E., and Westwood, N. P. C. Photoelectron spectra and bonding in some halosilanes, Can. J. Chem. **49**, 4033 (1971).
- [3512] Cetinkaya, B., King, G. H., Krishnamurthy, S. S., Lappert, M. F., and Pedley, J. B. Photoelectron spectra of electron-rich olefins and an isostructural boron compound; olefins of exceptionally low first ionisation potential, Chem. Commun., 1370 (1971).
- [3513] Gleiter, R., Heilbronner, E., and Hornung, V. Lone pair interaction in pyridazine, pyrimidine, and pyrazine, Angew. Chem. Intern. Ed. **9**, 901 (1970).
- [3514] Frost, D. C., Herring, F. G., Katrib, A., McLean, R. A. N., Drake, J. E., and Westwood, N. P. C. (p→d) π Bonding in halosilanes: evidence from photoelectron spectroscopy, Chem. Phys. Letters **10**, 347 (1971).
- [3515] Frost, D. C., Katrib, A., McDowell, C. A., and McLean, R. A. N. The ²A₁ band in the photoelectron spectrum of hydrogen sulphide, Intern. J. Mass Spectrom. Ion Phys. **7**, 485 (1971).
- [3516] Edqvist, O., Åsbrink, L., and Lindholm, E. On the photoelectron spectrum of NO, Z. Naturforsch. **26a**, 1407 (1971).
- [3517] Chadwick, D., Frost, D. C., and Weiler, L. The photoelectron

- spectra of cyclic ketones, *Tetrahedron Letters* **47**, 4543 (1971).
- [3518] Cowan, D. O., Gleiter, R., Glemser, O., Heilbronner, E., and Schäuble, J. The photoelectron spectrum of thiacyl fluoride (NSF), *Helv. Chim. Acta* **54**, 1559 (1971).
- [3520] Baker, A. D., Brundle, C. R., and Turner, D. W. The interpretation of photoelectron spectra especially those of benzene and water, *Intern. J. Mass Spectrom. Ion Phys.* **1**, 443 (1968).
- [3521] Appell, J., and Kubach, C. On the formation of energetic protons by electron impact on methane, *Chem. Phys. Letters* **11**, 486 (1971).
- [3523] Potapov, V. K., and Sorokin, V. V. Photoionization and ion-molecule reactions in quinones and alcohols, *Khim. Vys. Energ.* **5**, 487 (1971) [Engl. transl.: High Energy Chem. **5**, 435 (1971)].
- [3524] Nicholson, A. J. C. Determination of bond dissociation energies from photoionization efficiency curves, *Recent Developments in Mass Spectrometry*, ed. K. Ogata and T. Hayakawa (University Park Press, Baltimore, 1970) p. 745.
- [3525] Spohr, R., Guyon, P. M., Chupka, W. A., and Berkowitz, J. Threshold photoelectron detector for use in the vacuum ultraviolet, *Rev. Sci. Instr.* **42**, 1872 (1971).
- [3526] Loginov, M. V., and Mittsev, M. A. Thermal dissociation of SrCl_2 molecules at a tungsten surface, *Zh. Tekh. Fiz.* **41**, 709 (1971) [Engl. transl.: Sov. Phys.-Tech. Phys. **16**, 557 (1971)].
- [3527] Evans, S., Orchard, A. F., and Turner, D. W. A simple, medium resolution helium(I) photoelectron spectrometer, *Intern. J. Mass Spectrom. Ion Phys.* **7**, 261 (1971).
- [3528] Carlson, T. A., and Anderson, C. P. Angular distribution of the photoelectron spectrum for benzene, *Chem. Phys. Letters* **10**, 561 (1971).
- [3529] Rabalais, J. W., Bergmark, T., Werme, L. O., Karlsson, L., and Siegbahn, K. The Jahn-Teller effect in the electron spectrum of methane, *Phys. Scr.* **3**, 13 (1971).
- [3530] Åsbrink, L., and Rabalais, J. W. Comments on the high resolution photoelectron spectrum of H_2O and D_2O , *Chem. Phys. Letters* **12**, 182 (1971).
- [3531] Åsbrink, L. The photoelectron spectrum of H_2 , *Chem. Phys. Letters* **7**, 549 (1970).
- [3532] Brogli, F., and Heilbronner, E. The competition between spin orbit coupling and conjugation in alkyl halides and its repercussion on their photoelectron spectra, *Helv. Chim. Acta* **54**, 1423 (1971).
- [3533] Frost, D. C., and Sandhu, J. S. Ionization potentials of ethylene and some methyl-substituted ethylenes as determined by photoelectron spectroscopy, *Indian J. Chem.* **9**, 1105 (1971).
- [3534] Jonathan, N., Morris, A., Ross, K. J., and Smith, D. J. High resolution vacuum ultraviolet photoelectron spectra of transient species: $\text{O}_2(^1\Delta_g)$ and previously unobserved states of O_2^+ , *J. Chem. Phys.* **54**, 4954 (1971).
- [3535] Holmes, J. L., and McGillivray, D. The mass spectra of isomeric hydrocarbons—I: Norbornene and nortricyclene; The mechanisms and energetics of their fragmentations, *Org. Mass Spectrom.* **5**, 1349 (1971).
- [3536] Berkowitz, J. Photoionization mass spectrometry and photoelectron spectroscopy of high temperature vapors, *Advan. High Temp. Chem.* **3**, 123 (1971).
- [3537] Čermák, V. Penning ionization electron spectroscopy. III. Ionization of cadmium, *Coll. Czech. Chem. Com.* **36**, 948 (1971).
- [3538] Matsumoto, A., Taniguchi, S., and Hayakawa, T. Studies of dissociation of hydrogen and n-butane metastable ions by a pulsed ion source, *Recent Developments in Mass Spectrometry*, ed. K. Ogata and T. Hayakawa (University Park Press, Baltimore, 1970) p. 820.
- [3539] Syrvatka, B. G., Bel'ferman, A. L., Gil'burd, M. M., and Moin, F. B. Determination of the dissociation energy of the double bond in some fluorochlorosubstituted ethylenes and their ions by electron bombardment, *Zh. Org. Khim.* **7**, 9 (1971) [Engl. transl.: *J. Org. Chem. USSR* **7**, 8 (1971)].
- [3540] Murphy, Jr., C. B., and Enrione, R. E. Mass spectrometric determination of bond dissociation energies in $\text{BF}_3\text{-OEt}_2$, *Chem. Comm.*, 1622 (1971).
- [3541] Hotop, H., and Niehaus, A. Reactions of excited atoms and molecules with atoms and molecules. II. Energy analysis of Penning electrons, *Z. Physik* **228**, 68 (1969).
- [3543] Foster, R. Ionization potentials of electron donors, *Nature* **183**, 1253 (1959).
- [3544] Gross, M. L., and Wilkins, C. L. Computer-assisted ion cyclotron resonance appearance potential measurements for C_5H_{10} isomers, *Anal. Chem.* **43**, 1624 (1971).
- [3545] Herberich, G. E., Greiss, G., Heil, H. F., and Müller, J. Paramagnetic borabenzene cobalt complexes, *Chem. Comm.*, 1328 (1971).
- [3546] Pitt, C. G. Hyperconjugation: an alternative to the concept of the $p_{\pi}-d_{\pi}$ bond in Group IV chemistry, *J. Organometal. Chem.* **23**, C35 (1970).
- [3547] Murphy, C. B., and Enrione, R. E. Bond dissociation energies of the 1- and 2- isomers of pentaborane derivatives by electron impact and by extended Hückel calculations, *Intern. J. Mass Spectrom. Ion Phys.* **7**, 327 (1971).
- [3548] Lappert, M. F., Pedley, J. B., Simpson, J., and Spalding, T. R. Bonding studies of compounds of boron and the Group IV elements. VI. Mass spectrometric studies on compounds Me_2M and $\text{Me}_3\text{M-M'Me}_3$ (M and $\text{M}' = \text{C, Si, Ge, Sn, and Pb}$): thermochemical data, *J. Organometal. Chem.* **29**, 195 (1971).
- [3549] Gaidis, J. M., Briggs, P. R., and Shannon, T. W. Mass spectra of disilanes. Phenyl-silicon interaction and silicon-silicon bond strength, *J. Phys. Chem.* **75**, 974 (1971).
- [3550] Majer, J. R., Olavesen, C., and Robb, J. C. Wavelength effect in the photolysis of halogenated ketones, *J. Chem. Soc. (B)*, 48 (1971).
- [3551] Lawless, E. W. Mass spectrometric evidence of dimers in bismuth pentafluoride and antimony pentafluoride, *Inorg. Chem.* **10**, 2084 (1971).
- [3552] Grützmacher, H. F., and Hübner, J. Bildung und Struktur von $\text{C}_6\text{H}_4\text{O}$ bei der Pyrolyse von Salicylsäureestern und verwandten Verbindungen, *Liebigs Ann. Chem.* **748**, 154 (1971).
- [3553] Grützmacher, H. F., and Hübner, J. Massenspektrometrie instabiler Moleküle. VII. Thermische Bildung von Fulven-6-on durch zweifache Ringverengung von 3-Bromtropolon, *Tetrahedron Letters* **19**, 1455 (1971).
- [3554] Warneck, P. Photoionisation von Methanol und Formaldehyd, *Z. Naturforsch.* **26a**, 2047 (1971).
- [3555] Gutbier, H. Massenspektrometrische Untersuchungen der Verdampfungsvorgänge bei einigen Verbindungen mit Zinkblende-Gitter im Temperaturbereich um 1000°K, *Z. Naturforsch.* **16a**, 268 (1961).
- [3556] Varmuza, K., and Krenmayr, P. Massenspektrometrische Untersuchungen einfacher und gemischter Phosphortrihalogenide, *Monatsh. Chem.* **102**, 1037 (1971).
- [3557] Pattoret, A., Drowart, J., and Smoes, S. Etudes thermodynamiques par spectrométrie de masse sur le système uranium-oxygène, *Thermodyn. Nucl. Mater., Proc. Symp.*, Vienna, 1967, 613 (1968).
- [3558] Kuznetsova, L. A., Kuzmenko, N. E., and Kuzyakov, Yu. Ya. Emission spectrum of the SiBr^+ molecule, *Opt. Spektrosk.* **24**, 812 (1968) [Engl. transl.: *Opt. Spectry.* **24**, 434 (1968)].
- [3559] Smith, D. R., and Raymonda, J. W. Rydberg states in fluorinated benzenes; hexa-, penta-, and mono-fluorobenzene, *Chem. Phys. Letters* **12**, 269 (1971).

- [3560] Verma, R. D., Dixit, M. N., Jois, S. S., Nagaraj, S., and Singhal, S. R. Emission spectrum of the PO molecule. Part II. $^3\Sigma$ - $^2\Sigma$ transitions, Can. J. Phys. **49**, 3180 (1971).
- [3561] Worley, R. E., and Jenkins, F. A. A new Rydberg series in N₂, Phys. Rev. **54**, 305 (1938).
- [3564] Rao, T. V. R., and Lakshman, S. V. J. The true potential energy curves and Franck-Condon factors of SiH and SiH⁺ molecules, Physica, **56**, 322 (1971).
- [3565] McDiarmid, R. Higher electronic states of ReF_n, J. Mol. Spectry. **39**, 332 (1971).
- [3566] Radziszewski, L. J., Jr., Steinhaus, D. W., and Cowan, R. D. Present status of the analysis of U_i and U_{II} as derived from measurements of optical spectra, J. Opt. Soc. Am. **60**, 1556 (1970).
- [3567] Donovan, R. J., and Strachan, P. Vacuum U.-V. spectra of transient molecules and radicals, Trans. Faraday Soc. **67**, 3407 (1971).
- [3569] Gleiter, R., Hornung, V., Lindberg, B., Höglberg, S., and Lozac'h, N. The He-584 Å and X-ray photoelectron spectra of thiathiophthenes, Chem. Phys. Letters **11**, 401 (1971).
- [3570] Adams, G. P., Margrave, J. L., Steiger, R. P., and Wilson, P. W. The enthalpy of sublimation of germanium difluoride and the thermodynamics of sublimation of the Group IVa difluorides, J. Chem. Thermodyn. **3**, 297 (1971).
- [3571] Zaretskii, V. I., Sadovskaya, V. L., Wulfson, N. S., Sizov, V. F., and Merimnon, V. G. Mass spectrometry of steroid systems-XXI. Appearance and ionization potentials for the stereoisomers of the D-homoestrane series, Org. Mass Spectrom. **5**, 1179 (1971).
- [3573] Weissler, G. L., Ogawa, M., and Judge, D. L. Absorption of O₂, CO₂ and CS₂; fluorescence from CS₂; and photoionization of atomic carbon, J. Physique Supp. **32**, C4-154 (1971).
- [3574] Cabaud, B., Uzan, R., and Nounou, P. Étude des processus d'ionisation à haute température des vapeurs métalliques par couplage d'une cellule de Knudsen et d'une source Fox. I. Interprétation des processus d'ionisation de Ag et influence de la température sur les courbes d'efficacité d'ionisation, Intern. J. Mass Spectrom. Ion Phys. **6**, 89 (1971).
- [3575] Lageot, C. Potentiel d'ionisation, courbes d'efficacité d'ionisation différentielle, localisation de la charge de 9 cyclopropanes, Org. Mass Spectrom. **5**, 845 (1971).
- [3576] Gleiter, R., Heilbronner, E., and de Meijere, A. Die konjugative Wechselwirkung zwischen π -und Walsh-Orbitalen: das Photoelektron-Spektrum des Homofulvens, Helv. Chim. Acta **54**, 1029 (1971).
- [3577] Briegleb, G., Czekalla, J., and Reuss, G. Mesomeriemomente und Elektronenüberführungsbänder von Elektronen-Donator-Akzeptor-Komplexen des Chloranils und Tetracyanäthylen mit aromatischen Kohlenwasserstoffen, Z. Physik. Chem. **30**, 333 (1961).
- [3578] Müller, J., and Fenderl, K. Reaktionen des π -Cyclopentadienyl-mangantricarbonyl-Kations mit einfachen Fluorverbindungen in der Gasphase, Chem. Ber. **104**, 2207 (1971).
- [3579] Müller, J., and Fenderl, K. Sekundär-Ionen in den Massenspektren von Organochrom-Komplexen, Chem. Ber. **104**, 2199 (1971).
- [3580] Sucrow, W., Bethke, H., and Chondromatidis, G. Thermolyse von 1,2,4,5-Tetramethyl-hexahydro- 1,2,4,5-tetrazinen im Massenspektrometer, Tetrahedron Letters **19**, 1481 (1971).
- [3581] Lageot, C. Potentiel d'ionisation, potentiel d'apparition et courbes d'ionisation différentielle pour les 1-2 dimethylcyclohexanes *cis* et *trans*, Org. Mass Spectrom. **5**, 839 (1971).
- [3582] Fischer, E. O., Kreiter, C. G., Kollmeier, H. J., Müller, J., and Fischer, R. D. Übergangsmetall-Carben-Komplexe XXVII. Ringsubstituierte (Methoxyphenylcarben)-pentacarbonylchrom(0)-Komplexe, J. Organometal. Chem. **28**, 237 (1971).
- [3583] Grützmacher, H. -F., and Hübner, J. Massenspektrometrie instabiler Moleküle. III: Nachweis und Untersuchungen zur Stabilität chlorsubstituierter Dehydrobenzole in der Gasphase, Org. Mass Spectrom. **2**, 649 (1969).
- [3584] Bock, H., and Fuss, W. Ionisierungsgenergien und Geometrien von Aminoboranen, Chem. Ber. **104**, 1687 (1971).
- [3585] Praet, M. -Th. Ionisation et dissociation du 1-methylcyclopentene, du methylenecyclopentane et de quelques isomères par impact d'électrons et de photons, Org. Mass Spectrom. **4**, 65 (1970).
- [3586] Terenin, A. Charge transfer in organic solids, induced by light, Proc. Chem. Soc., 321 (1961).
- [3587] Salmona, G., Ferré, Y., and Vincent, E. J. Études expérimentales et théoriques de potentiels d'ionisation dérivés de la série de l'isothiazole, C. R. Acad. Sci., Ser. C **273**, 863 (1971).
- [3588] Bonnier, J. -M., Gelus, M., and Nounou, P. Contribution à l'étude de l'effet inductif et de l'effet d'hyperconjugaison dans quelques méthylaromatiques, J. Chim. Phys. **10**, 1191 (1965).
- [3589] Cullen, W. R., Frost, D. C., and Leeder, W. R. The ultraviolet and photoelectron spectra of some unsaturated fluorocarbon derivatives, J. Fluorine Chem. **1**, 227 (1971/72).
- [3590] Audier, H. E., Bouchoux, G., and Fetizon, M. Ionisation et fragmentation en spectrométrie de masse. II. Influence du substituant sur des fragmentations compétitives en série aromatique, Bull. Soc. Chim. Fr. **3**, 858 (1971).
- [3594] Balducci, G., De Maria, G., Guido, M.; and Piacente, V. Dissociation energy of TiO and TiO₂ gaseous molecules, J. Chem. Phys. **56**, 3422 (1972).
- [3595] Blackburn, P. E., and Danielson, P. M. Electron impact relative ionization cross sections and fragmentation of U, UO, UO₂, and UO₃, J. Chem. Phys. **56**, 6156 (1972).
- [3596] Kordis, J., and Gingerich, K. A. Gaseous phosphorus compounds. VIII. Thermodynamic study of antimony monophosphide with a mass spectrometer, J. Phys. Chem. **76**, 2336 (1972).
- [3597] Cocke, D. L., and Gingerich, K. A. Mass spectrometric determination of the bond dissociation energies of the molecules CePd and CeC₂, J. Phys. Chem. **76**, 2332 (1972).
- [3598] Conde-Caprace, G., and Collin, J. E. Ionization and dissociation of cyclic ethers and thioethers by electron-impact. A comparison between 1,3-dioxolane, 1,3-dithiolane and 1,3-oxathiolane, Org. Mass Spectrom. **6**, 415 (1972).
- [3605] Wagner, L. C., and Grimley, R. T. A study of ionization processes by the angular distribution technique. The AgCl system, J. Phys. Chem. **76**, 2819 (1972).
- [3606] Uy, O. M., Srivastava, R. D., and Farber, M. Mass spectrometric determination of the heats of formation of the gaseous molecules AlOF₂ and AlF₂, High Temp. Sci. **4**, 227 (1972).
- [3607] Roberts, J. A., Jr., and Searcy, A. W. The stabilities of Ce₂F₆(g) and La₂F₆(g), High Temp. Sci. **4**, 411 (1972).
- [3608] Piacente, V., and Desideri, A. Mass spectrometric determination of the dissociation energy of GaBi molecule, J. Chem. Phys. **57**, 2213 (1972).
- [3609] Piacente, V., and Gingerich, K. A. Thermodynamic study of the molecule NaAg with a mass spectrometer, High Temp. Sci. **4**, 312 (1972).
- [3610] Hildenbrand, D. L. The gaseous equilibrium Ge + SiO = GeO + Si and the dissociation energy of SiO, High Temp. Sci. **4**, 244 (1972).
- [3611] Balducci, G., De Maria, G., and Guido, M. Mass spectrometric

- determination of the dissociation energy of $\text{EuC}_2(\text{g})$, *J. Chem. Phys.* **56**, 1431 (1972).
- [3612] Hariharan, A. V., and Eick, H. A. Vaporization thermodynamics of EuI_3 , *High Temp. Sci.* **4**, 379 (1972).
- [3613] Feather, D. H., Büchler, A., and Searcy, A. W. The vapor pressures of gallium trifluoride monomer and dimer, *High Temp. Sci.* **4**, 290 (1972).
- [3614] Hariharan, A. V., Fishel, N. A., and Eick, H. A. Vaporization thermodynamics of YbCl_3 , *High Temp. Sci.* **4**, 405 (1972).
- [3615] Muenow, D. W., and Margrave, J. L. Mass spectrometric observations of gaseous phosphorus sulfides and oxysulfides, *J. Inorg. Nucl. Chem.* **34**, 89 (1972).
- [3616] Hildenbrand, D. L. Thermochemistry of the molecules CS and CS^+ , *Chem. Phys. Letters* **15**, 379 (1972).
- [3617] Farber, M., Srivastava, R. D., and Uy, O. M. Mass spectrometric determination of the thermodynamic properties of the vapour species from alumina, *J. Chem. Soc. Faraday Trans. I.* **68**, 249 (1972).
- [3618] Guido, M., Balducci, G., and De Maria, G. Thermodynamics of rare-earth–carbon systems. IV. The lutetium–carbon system, *J. Chem. Phys.* **57**, 1475 (1972).
- [3619] Gingerich, K. A. Gaseous phosphorus compounds. VII. The dissociation energy and heat of formation of boron monophosphide, *J. Chem. Phys.* **56**, 4239 (1972).
- [3620] Farber, M., Uy, O. M., and Srivastava, R. D. Effusion–mass spectrometric determination of the heats of formation of the gaseous molecules V_1O_{10} , V_1O_9 , VO_2 , and VO , *J. Chem. Phys.* **56**, 5312 (1972).
- [3621] Gingerich, K. A., Pupp, C., and Campbell, B. E. Mass spectrometric determination of the heats of atomization of the molecules Ce_2S , CeS_2 , Ce_2S_2 , and Ce_3S_3 , *High Temp. Sci.* **4**, 236 (1972).
- [3622] Gräber, P., and Weil, K. G. Mass spectrometric investigations of silver halides I: mass spectrum, appearance potentials, and fragmentation scheme of silver chloride, *Ber. Bunsenges. Phys. Chem.* **76**, 410 (1972).
- [3623] Ehlert, T. C., and Hsia, M. Mass spectrometric and thermochemical studies of the manganese fluorides, *J. Fluorine Chem.* **2**, 33 (1972–73).
- [3624] Tajima, S., Shimizu, Y., and Tsuchiya, T. The effect of the shield voltage on appearance potential measurements using a mass spectrometer, *Bull. Chem. Soc. Japan* **45**, 931 (1972).
- [3625] Crowe, A., Preston, J. A., and McConkey, J. W. Ionization of argon by electron impact, *J. Chem. Phys.* **57**, 1620 (1972).
- [3626] Johnstone, R. A. W., and Mellon, F. A. Electron–impact ionization and appearance potentials, *J. Chem. Soc. Faraday Trans. II* **68**, 1209 (1972).
- [3627] Hvistendahl, G., and Undheim, K. Mass spectrometry of ‘onium compounds. Part XIV. Methiodides of methyl pyridylacetates, *J. Chem. Soc., Perkin Trans. II* **14**, 2030 (1972).
- [3628] Flesch, G. D., Junk, G. A., and Svec, H. J. Ionization efficiency data and fragmentation mechanisms for ferrocene, nickelocene, and ruthenocene, *J. Chem. Soc. Dalton Trans.*, 1102 (1972).
- [3629] Lightner, D. A., Majeti, S., Nicoletti, R., and Thommen, E. Benzyl vs. tropylum ions in the electron impact induced decomposition of *n*-butylbenzenes, *Intra–Sci. Chem. Rep.* **6**, 113 (1972).
- [3630] Hvistendahl, G., and Undheim, K. Mass spectrometry of ‘onium compounds. IX: on the evaporation of anilinium oxides. Ionization potential measurements, *Org. Mass Spectrom.* **6**, 217 (1972).
- [3631] Benezra, S. A., and Bursey, M. M. Hydrogen bonding in mass spectral activated complexes. A correction, *J. Chem. Soc., Perkin Trans. II* **1537** (1972).
- [3632] Foster, M. S., and Beauchamp, J. L. Gas–phase ion chemistry of azomethane by ion cyclotron resonance spectroscopy, *J. Am. Chem. Soc.* **94**, 2425 (1972).
- [3633] Dixon, D. A., Holtz, D., and Beauchamp, J. L. Acidity, basicity, and gas-phase ion chemistry of hydrogen selenide by ion cyclotron resonance spectroscopy, *Inorg. Chem.* **11**, 960 (1972).
- [3634] Hertzberg, M., White, G., Olfky, R. S., and Saalfeld, F. E. Bisdifluorominoalkanes: the mass spectral decomposition of isomeric propanes, *J. Phys. Chem.* **76**, 60 (1972).
- [3635] Gronneberg, T., and Undheim, K. Mass spectrometry of onium compounds. X; on the evaporation of pyridinium-3-oxides—ionization potential measurements, *Org. Mass Spectrom.* **6**, 225 (1972).
- [3636] Gronneberg, T., and Undheim, K. Mass spectrometry of onium compounds. XI: ionization potentials of hydroxy and mercapto pyridines, *Org. Mass Spectrom.* **6**, 823 (1972).
- [3637] Brundle, C. R., Robin, M. B., and Kuebler, N. A. Perfluoro effect in photoelectron spectroscopy. II. Aromatic molecules, *J. Am. Chem. Soc.* **94**, 1466 (1972).
- [3638] Brogli, F., Heilbronner, E., and Kobayashi, T. Photoelectron spectra of azabenzenes and azanaphthalenes: II. A reinvestigation of azanaphthalenes by high-resolution photoelectron spectroscopy, *Helv. Chim. Acta* **55**, 274 (1972).
- [3639] Åsbrink, L., Fridh, C., Jonsson, B. Ö., and Lindholm, E. Rydberg series in small molecules. XVII. Photoelectron, UV, mass and electron impact spectra of pyridazine, *Intern. J. Mass Spectrom. Ion Phys.* **8**, 229 (1972).
- [3640] Berkowitz, J., and Dehmer, J. L. Photoelectron spectroscopy of high-temperature vapors. II. Chemical bonding in the Group III monohalides, *J. Chem. Phys.* **57**, 3194 (1972).
- [3641] Basset, P. J., and Lloyd, D. R. Photoelectron spectra of halides. Part III. Trifluorides and oxide trifluorides of nitrogen and phosphorus, and phosphorus oxide trichloride, *J. Chem. Soc. Dalton Trans.* 248 (1972).
- [3642] Brundle, C. R., and Jones, G. R. Electronic structure of KrF_2 , studied by photoelectron spectroscopy, *J. Chem. Soc. Faraday Trans. II* **68**, 959 (1972).
- [3643] Brundle, C. R., Kuebler, N. A., Robin, M. B., and Basch, H. Ionization potentials of the tetraphosphorus molecule, *Inorg. Chem.* **11**, 20 (1972).
- [3644] Boschi, R., Murrell, J. N., and Schmidt, W. Photoelectron spectra of polycyclic aromatic hydrocarbons, *Faraday Discuss. Chem.* **54**, 116 (1972).
- [3645] Bergmark, T., Rabalais, J. W., Werme, L. O., Karlsson, L., and Siegbahn, K. High-resolution electron spectra of methane, thiophene, 2-bromothiophene, and 3-bromothiophene, *Electron Spectroscopy*, ed. D. A. Shirley (North-Holland Pub. Co., Amsterdam, 1972).
- [3646] Bock, H., and Solouki, B. The “sulfoxide bond”, *Angew. Chem. Intern. Ed.* **11**, 436 (1972).
- [3647] Boekelheide, V., and Schmidt, W. A photoelectron spectroscopic study of a classically conjugated but orbitally unconjugated tris-bridged cyclophane: [2.2.2](1,3,5)cyclophane-1,9,17-triene, *Chem. Phys. Lett.* **17**, 410 (1972).
- [3648] Bock, H., and Wittel, K. Photoelectron spectra and molecular properties of *trans*-dihalogenoethylenes: substituent effects spin-orbit coupling, *J. Chem. Soc. Chem. Commun.* 602 (1972).
- [3649] Brundle, C. R., Robin, M. B., Kuebler, N. A., and Basch, H. Perfluoro effect in photoelectron spectroscopy. I. Nonaromatic molecules, *J. Am. Chem. Soc.* **94**, 1451 (1972).
- [3650] Berkowitz, J. Photoelectron spectroscopic studies with a cylindrical-mirror analyzer, in: *Electron Spectroscopy*, ed. D. A. Shirley (North-Holland Pub. Co., Amsterdam, 1972).
- [3651] Åsbrink, L., Fridh, C., Jonsson, B. Ö., and Lindholm, E. Rydberg series in small molecules. XVI. Photoelectron,

- UV, mass and electron impact spectra of pyrimidine, *Intern. J. Mass Spectrom. Ion Phys.* **8**, 215 (1972).
- [3652] Paine, R. T., Sodeck, G., and Stafford, F. E. Molecular beam mass spectra and pyrolyses of fluorophosphine-triborane(7) complexes. Formation and mass spectrum of triborane(7), *Inorg. Chem.* **11**, 2593 (1972).
- [3653] Saalfeld, F. E., McDowell, M. V., MacDiarmid, A. G., and Highsmith, R. E. Nature of the bonding between silicon and the cobalt tetracarbonyl group in silylcobalt tetracarbonyl. III. Mass spectral studies of trichlorosilyl trifluorophosphine cobalt carbonyl derivatives, *Intern. J. Mass Spectrom. Ion Phys.* **9**, 197 (1972).
- [3654] Carmichael, P. J., Gowenlock, B. G., and Johnson, C. A. F. Carbon-nitrogen bond dissociation energy values in C-nitrosocompounds, *Intern. J. Chem. Kinet.* **4**, 339 (1972).
- [3655] DeKock, R. L., Higginson, B. R., and Lloyd, D. R. Photoelectron spectra of halides. Part 6.—The spectra of SF₅Cl, BrF₅ and IF₅, *Faraday Discuss. Chem. Soc.* **54**, 84 (1972).
- [3656] Cradock, S., and Whiteford, R. A. Photoelectron spectra of the methyl, silyl and germyl derivatives of the group VI elements, *J. Chem. Soc. Faraday Trans. II* **68**, 281 (1972).
- [3657] Dewar, M. J. S., and Goodman, D. W. Photoelectron spectra of molecules. Part 5.—Polycyclic aromatic hydrocarbons, *J. Chem. Soc. Faraday Trans. II* **68**, 1784 (1972).
- [3658] Chizhov, Yu. V., Kleimenov, V. I., Medynskii, G. S., and Vilesov, F. I. Photoelectron spectroscopy study of benzene, *Opt. Spektrosk.* **33**, 661 (1972) [Engl. transl.: *Opt. Spectry. (USSR)* **33**, 365 (1972)].
- [3659] Chadwick, D., Frost, D. C., Katrib, A., McDowell, C. A., and McLean, R. A. N. Photoelectron spectra of some bromoethylenes and 2-bromopropene, *Can. J. Chem.* **50**, 2642 (1972).
- [3660] Cowan, D.O., Gleiter, R., Glemser, O.; and Heilbronner, E. The photoelectron spectra of NSCl, NSF and NSF₂, *Helv. Chim. Acta* **55**, 2418 (1972).
- [3661] Cradock, S., Ebsworth, E. A. V., Savage, W. J., and Whiteford, R. A. Photoelectron spectra of some methyl, silyl and germyl amines, phosphines and arsines, *J. Chem. Soc. Faraday Trans. II* **68**, 1972 (1972).
- [3662] Cradock, S., and Rankin, D. W. H. Photoelectron spectra of PF₃H and some substituted difluorophosphines, *J. Chem. Soc. Faraday Trans. II* **68**, 940 (1972).
- [3663] Cradock, S., and Savage, W. The photoelectron spectrum and electronic structure of hexamethyl tungsten, *Inorg. Nucl. Chem. Lett.* **8**, 753 (1972).
- [3664] Collin, J. E., Delwiche, J., and Natalis, P. Autoionization observed by photoelectron spectrometry at different wavelengths, *Electron Spectroscopy*, ed. D. A. Shirley (North-Holland Publishing Co., Amsterdam, 1972) p. 401.
- [3665] Dixon, R. N.; Duxbury, G.; Fleming, G. R., and Hugo, J. M. V. The photoelectron spectrum of thiaetyl fluoride, *Chem. Phys. Lett.* **14**, 60 (1972).
- [3666] DeKock, R. L., Lloyd, D. R., Breeze, A., Collins, G. A. D., Cruickshank, D. W. J.; and Lempka, H. J. Photoelectron spectroscopy and ab initio LCAO MO SCF calculations on thiaetyl fluoride, *Chem. Phys. Lett.* **14**, 52 (1972).
- [3667] Chadwick, D. Photoelectron spectra of phosgene and thiophosgene, *Can. J. Chem.* **50**, 737 (1972).
- [3668] Clark, P. A., and Brogli, F., and Heilbronner, E. The π -orbital energies of the acenes, *Helv. Chim. Acta* **55**, 1415 (1972).
- [3669] Cox, P. A., Evans, S., Orchard, A. F., Richardson, N. V., and Roberts, P. J. Simple quantitative molecular orbital methods used in connection with photoelectron spectroscopy, *Faraday Discuss. Chem. Soc.* **54**, 26 (1972).
- [3670] Cradock, S., Ebsworth, E. A. V., and Murdoch, J. D. Photoelectron spectra of some Group 4 pseudohalides and related compounds, *J. Chem. Soc. Faraday Trans. II* **68**, 86 (1972).
- [3671] Cornford, A. B., Frost, D. C., Herring, F. G., and McDowell, C. A. Photoelectron spectra of some free radicals, *Faraday Discuss. Chem. Soc.* **54**, 56 (1972).
- [3672] Blake, A. J. Photoionization study of mercury by photoelectron spectroscopy, *Proc. Roy. Soc. (London)* **325**, 555 (1971).
- [3673] Bock, H., and Fuss, W. Arguments concerning the orbital sequence in borazin, *Angew. Chem. Intern. Ed.* **10**, 182 (1971).
- [3674] Plotnikov, V. F., Bogolyubov, G. M., Maretina, I. A., and Petrov, A. A. Organic derivatives of elements of Groups V and VI. V. Mass spectra of 1-buten-3-ynylamines, *Zh. Org. Khim.* **5**, 1157 (1969) [Engl. transl.: *J. Org. Chem. USSR* **5**, 1137 (1969)].
- [3675] DeKock, R. L., Lloyd, D. R., Hillier, I. H., and Saunders, V. R. Experimental and theoretical study of the electronic structures of sulphuryl fluoride and perchloryl fluoride, *Proc. Roy. Soc. (London)* **328**, 401 (1972).
- [3676] Evans, S. General discussion, *Faraday Discuss. Chem. Soc.* **54**, 143 (1972).
- [3677] Evans, S., Green, J. C., Joachim, P. J., Orchard, A. F., Turner, D. W., and Maier, J. P. Electronic structures of the Group IV_b tetramethyls by helium(I) photoelectron spectroscopy, *J. Chem. Soc. Faraday Trans. II* **68**, 905 (1972).
- [3678] Frost, D. C., Herring, F. G., Katrib, A., McDowell, C. A., and McLean, R. A. N. Photoelectron spectra of CH₃SH, (CH₃)₂S, C₆H₅SH, and C₆H₅CH₂SH; the bonding between sulfur and carbon, *J. Phys. Chem.* **76**, 1030 (1972).
- [3679] Gleiter, R., Heilbronner, E., and Hornung, V. Photoelectron spectra of azabenzenes and azanaphthalenes: I. Pyridine, diazines *s*-triazine and *s*-tetrazine, *Helv. Chim. Acta* **55**, 255 (1972).
- [3680] DeKock, R. L., Higginson, B. R., Lloyd, D. R., Breeze, A., Cruickshank, D. W. J., and Armstrong, D. R. Photoelectron spectra of halides. V. Experimental and theoretical study of the electronic structures of ClF, ClF₃, BrF and BrF₅, *Mol. Phys.* **24**, 1059 (1972).
- [3681] Evans, S., Hamnett, A., and Orchard, A. F. The relative orbital energies of metal and ligand electrons in some tris(hexafluoroacetylacetoneato) transition-metal complexes, *J. Coord. Chem.* **2**, 57 (1972).
- [3682] Evans, S., Hamnett, A., Orchard, A. F., and Lloyd, D. R. Study of the metal-oxygen bond in simple tris-chelate complexes by He(I) photoelectron spectroscopy, *Faraday Discuss. Chem. Soc.* **54**, 227 (1972).
- [3683] Evans, S., Joachim, P. J., Orchard, A. F., and Turner, D. W. A study of the orbital electronic structure of the P₄ molecule by photoelectron spectroscopy, *Intern. J. Mass Spectrom. Ion Phys.* **9**, 41 (1972).
- [3684] Eland, J. H. D. Photoelectron spectra and ionization potentials of aromatic hydrocarbons, *Intern. J. Mass Spectrom. Ion Phys.* **9**, 214 (1972).
- [3685] Heilbronner, E., Hornung, V., Pinkerton, F. H., and Thamess, S. F. 31. Photoelectron spectra of azabenzenes and azanaphthalenes: III. The orbital sequence in methyl- and trimethylsilyl-substituted pyridines, *Helv. Chim. Acta* **55**, 289 (1972).
- [3686] Evans, S., Green, J. C., and Jackson, S. E. He(I) photoelectron spectra of some π -arene complexes, *J. Chem. Soc. Faraday Trans. II* **68**, 249 (1972).
- [3687] Heilbronner, E., and Martin, H. D. The π -orbital sequence in norbornadiene and related hydrocarbons, *Helv. Chim. Acta* **55**, 1490 (1972).
- [3688] Evans, S., Green, M. L. H., Jewitt, B., Orchard, A. F., and Pygall, C. F. Electronic structure of metal complexes containing π -cyclopentadienyl and related ligands, *J.*

- Chem. Soc. Faraday Trans. II* **68**, 1847 (1972).
- [3690] Frost, D. C., Lee, S. T., and McDowell, C. A. The high resolution photoelectron spectrum of CS, *Chem. Phys. Letters* **17**, 153 (1972).
- [3691] Jonathan, N., Morris, A., Okuda, M.; Ross, K. J., and Smith, D. J. Photoelectron spectroscopy of transient species. The CS molecule, *Faraday Discuss. Chem. Soc.* **54**, 48 (1972).
- [3692] McDowell, C. A. General Discussion, *Faraday Discuss. Chem. Soc.* **54**, 68 (1972).
- [3693] Herring, F. G. General Discussion, *Faraday Discuss. Chem. Soc.* **54**, 68 (1972).
- [3694] Chadwick, D., Cornford, A. B., Frost, D. C., Herring, F. G., Katrib, A., McDowell, C. A., and McLean, R. A. N. Photoelectron spectra of some dihalocompounds, in *Electron Spectroscopy*, ed. D. A. Shirley (North-Holland Publishing Co., Amsterdam, 1972) p. 453.
- [3695] Potts, A. W., Glenn, K. G., and Price, W. C. General Discussion, *Faraday Discuss. Chem. Soc.* **54**, 65 (1972).
- [3696] King, G. H., Kroto, H. W., and Suffolk, R. J. The photoelectron spectrum of a short-lived species in the decomposition products of CS₂, *Chem. Phys. Letters* **13**, 457 (1972).
- [3697] Kroto, H. W., and Suffolk, R. J. The photoelectron spectrum of an unstable species in the pyrolysis products of dimethyldisulphide, *Chem. Phys. Letters* **15**, 545 (1972).
- [3698] Morris, A. General Discussion, *Faraday Discuss. Chem. Soc.* **54**, 64 (1972).
- [3699] Lloyd, D. R., and Lyraugh, N. Photoelectron studies of boron compounds. Part 3. Complexes of borane with Lewis bases, *J. Chem. Soc. Faraday Trans. II* **68**, 947 (1972).
- [3700] Potts, A. W., Glenn, K. G., and Price, W. C. General Discussion, *Faraday Discuss. Chem. Soc.* **54**, 64 (1972).
- [3701] Jonathan, N., Morris, A., Okuda, M., and Smith, D. J. Electron spectroscopy of transient species, *Electron Spectroscopy*, ed. D. A. Shirley (North-Holland Pub. Co., Amsterdam, 1972) p. 345.
- [3702] Maier, J. P., and Turner, D. W. Steric inhibition of resonance studied by molecular photoelectron spectroscopy. Part I. Biphenyls, *Faraday Discuss. Chem. Soc.* **54**, 149 (1972).
- [3703] Maier, J. P., and Turner, D. W. Photoelectron spectroscopy and the geometry of the phosphine and phosphorous trifluoride molecular ions, *J. Chem. Soc. Faraday Trans. II* **68**, 711 (1972).
- [3704] King, G. H., Krishnamurthy, S. S., Lappert, M. F., and Pedley, J. B. Bonding studies of compounds of boron and the Group 4 elements. Part 9. Photoelectron spectra and bonding studies of halogeno-, dimethylamino-, and methyl-boranes, BX₃ and BX₂Y, *Faraday Discuss. Chem. Soc.* **54**, 70 (1972).
- [3705] Mines, G. W., Thomas, R. K., and Thompson, H. Photoelectron spectra of compounds containing thionyl and sulphuryl groups, *Proc. Roy. Soc. (London)* **A329**, 275 (1972).
- [3707] King, G. H., Murrell, J. N., and Suffolk, R. J. The vacuum-ultraviolet photoelectron spectra of fluoropyridines, *J. Chem. Soc. Dalton Trans.*, 564 (1972).
- [3708] Kroto, H. W., and Suffolk, R. J. The photoelectron spectrum of F₃CS and fluorine substitution shifts, *Chem. Phys. Letters* **17**, 213 (1972).
- [3709] Lyraugh, N., Lloyd, D. R., Guest, M. F., Hall, M. B., and Hillier, I. H. Photoelectron studies of boron compounds. Part 4. Experimental and theoretical studies of diboron tetrachloride and diboron tetrafluoride, *J. Chem. Soc. Faraday Trans. II* **68**, (1972).
- [3710] Murrell, J. N., and Schmidt, W. Photoelectron spectroscopic correlation of the molecular orbitals of methane, ethane, propane, isobutane and neopentane, *J. Chem. Soc. Faraday Trans. II* **68**, 1709 (1972).
- [3711] Lloyd, D. R., and Lyraugh, N. Photoelectron spectra of the bis-(π -allyl) complexes of nickel and palladium, *Electron Spectroscopy*, ed. D. A. Shirley (North-Holland Pub. Co., Amsterdam, 1972) p. 445.
- [3712] Morishima, I., Yoshikawa, K., Yonezawa, T., and Matsumoto, H. Photoelectron spectral studies of organic free radicals. The nitroxide radical, *Chem. Phys. Letters* **16**, 336 (1972).
- [3713] Klessinger, M. Ionization potentials of substituted benzenes, *Angew. Chem. Intern. Ed.* **11**, 525 (1972).
- [3714] Okuda, M. General Discussion, *Faraday Discuss. Chem. Soc.* **54**, 140 (1972).
- [3716] Potts, A. W., and Price, W. C. The photoelectron spectra of methane, silane, germane and stannane, *Proc. Roy. Soc. (London)* **A165**, (1972).
- [3717] Jonathan, N. General Discussion, *Faraday Discuss. Chem. Soc.* **54**, 64 (1972).
- [3718] Sweigart, D. A., and Turner, D. W. Lone pair orbitals and their interactions studied by photoelectron spectroscopy. I. Carboxylic acids and their derivatives, *J. Am. Chem. Soc.* **94**, 5592 (1972).
- [3719] Potts, A. W., and Price, W. C. Photoelectron spectra and valence shell orbital structures of groups V and VI hydrides, *Proc. Roy. Soc. (London)* **A326**, 181 (1972).
- [3720] Fridh, C., Åsbrink, L., Jonsson, B. O., and Lindholm, E. Rydberg series in small molecules. XIV. Photoelectron, UV, mass and electron impact spectra of s-triazine, *Intern. J. Mass Spectrom. Ion Phys.* **8**, 85 (1972).
- [3721] Rabalais, J. W. Photoelectron spectroscopic investigation of the electronic structure of nitromethane and nitrobenzene, *J. Chem. Phys.* **57**, 960 (1972).
- [3722] Van Den Ham, D. M. W., and Van Der Meer, D. The photoelectron spectra of the diazanaphthalenes, *Chem. Phys. Letters* **12**, 447 (1972).
- [3723] Van Den Ham, D. M. W., and Van Der Meer, D. Perfluoro effect in the photoelectron spectra of quinoline and isoquinoline, *Chem. Phys. Letters* **15**, 549 (1972).
- [3724] Palmer, M. H., and Findlay, R. H. Ab initio molecular orbital calculations, the electronic structure and electron spectrum of norbornadiene, *Chem. Phys. Letters* **15**, 416 (1972).
- [3725] Rabalais, J. W., Werme, L. O., Bergmark, T., Karlsson, L., Hussain, M., and Siegbahn, K. Electron spectroscopy of open-shell systems: spectra of Ni(C₅H₅)₂, Fe(C₅H₅)₂, Mn(C₅H₅)₂, and Cr(C₅H₅)₂, *J. Chem. Phys.* **57**, 1185 (1972).
- [3726] Thomas, R. K., Thompson, H. Photoelectron spectra of carbonyl halides and related compounds, *Proc. Roy. Soc. (London)* **A327**, 13 (1972).
- [3727] Robin, M. B., Brundle, C. R., Kuebler, N. A., Ellison, G. B., and Wiberg, K. B. Photoelectron spectra of small rings. IV. The unsaturated three-membered rings, *J. Chem. Phys.* **57**, 1758 (1972).
- [3728] Rabalais, J. W., Bergmark, T., Werme, L. O.; Karlsson, L., Hussain, M., and Siegbahn, K. The high-resolution electron spectrum of carbon suboxide, *Electron Spectroscopy*, ed. D. A. Shirley (North-Holland Pub. Co., Amsterdam, 1972), p. 425.
- [3729] Adamchuk, V. K., Dmitriev, A. B., Prudnikova, G. V., and Sorokin, L. S. Photoionization of low-volatility molecules in a geiger counter, *Opt. Spektros.*, **33**, 358 (1972) [Eng. transl. *Opt. Spectrosc. USSR* **33**, 191 (1972)].
- [3730] Daisey, J. M., and Sonnessa, A. J. A study of the thermodynamic and spectral properties of molecular complexes of iodine with several aminopyridines, *J. Phys. Chem.* **76**, 1895 (1972).
- [3731] DiLonardo, G., Galloni, G., Trombetti, A., and Zauli, C. Electronic spectrum of thiophen and some deuterated thiophens, *J. Chem. Soc. Faraday Trans. II* **68**, 2009 (1972).
- [3732] Lossing, F. P. Free radicals by mass spectrometry. XLIV.

- [3752] Sergeev, Yu. L., Akopyan, M. E., and Vilesov, F. I. Photoionization of the phenyl radical, Opt. Spektrosk. **32**, 230 (1972) [Engl. transl.: Opt. Spectry. (USSR) **32**, 121 (1972)].
- [3753] Stebbings, W. L., and Taylor, J. W. Photoionization mass spectrometry. II. Contrasting fragmentation of toluene by photons and by electrons, Intern. J. Mass Spectrom. Ion Phys. **9**, 471 (1972).
- [3754] Kaufman, V., and Minnhausen, L. Accurate ground-term combinations in NeI, J. Opt. Soc. Am. **62**, 92 (1972).
- [3755] Narayan, B. Spectra and ionization potential of cyanoacetylene, Proc. Indian Acad. Sci. A **75**, 92 (1972).
- [3756] Radzimski, L. J., and Kaufman, V. New wavelengths and energy levels in the spectrum of singly ionized chlorine (Cl II), J. Opt. Soc. Am. **62**, 1371 (1972).
- [3757] Raymonda, J. W. Rydberg states in cyclic alkanes, J. Chem. Phys. **56**, 3912 (1972).
- [3758] Pitt, C. G., Carey, R. N., and Toren, E. C. Nature of the electronic interactions in aryl-substituted polysilanes, J. Am. Chem. Soc. **94**, 3806 (1972).
- [3759] Nishida, S., Moritani, I., and Teraji, T. Ionization potentials of cyclopropylethylenes, J. Chem. Soc. Chem. Commun., 1114 (1972).
- [3760] Ogawa, M., and Ogawa, S. Absorption spectrum of CO in the Hopfield helium continuum region, 600–1020 Å, J. Mol. Spectry. **41**, 393 (1972).
- [3761] Narayana, B., and Price, W. C. Ionization of the σ^2 s orbital of NO and configuration interaction effects on the spin-split states arising from inner orbital ionization in paramagnetic molecules, J. Phys. B **5**, 1784 (1972).
- [3762] Ackermann, F., Lefebvre-Brion, H., and Roche, A. L. Calculated Rydberg states of the PO molecule, Can. J. Phys. **50**, 692 (1972).
- [3763] Takezawa, S., and Tanaka, Y. Absorption spectrum of HD in the vacuum-uv region. Rydberg states and ionization energy, J. Chem. Phys. **56**, 6125 (1972).
- [3764] Gilbert, R., Sauvageau, P., and Sandorfy, C. Far-UV and photoelectron spectra of 1,3,5-trifluorobenzene, Chem. Phys. Letters **17**, 465 (1972).
- [3765] Vilesov, F. I. The photoionization of vapors of compounds whose molecules contain carbonyl groups, Dokl. Akad. Nauk SSSR **132**, 1332 (1960) [Engl. transl.: Dokl. Phys. Chem. **132**, 521 (1960)].
- [3766] Akopyan, M. E., and Loginov, Yu. V. Mass-spectrometric study of the photoionization of free α -aminoacids, Khim. Vys. Energ. **1**, 97 (1967) [Engl. transl.: High Energy Chem. **1**, 83 (1967)].
- [3767] Polyakova, A. A., Zimina, K. I., Petrov, A. A., and Khmel'nikskii, R. A. Mass spectra and the structure of vinylacetylenes, Dokl. Akad. Nauk SSSR **127**, 386 (1959) [Engl. transl.: Dokl. Phys. Chem. **127**, 597 (1959)].
- [3768] Velasco, R. Espectro ultravioleta de la molécula Li₂, An. R. Soc. Esp. Fis. Quim. **175**, (1960).
- [3769] Gil'burd, M. M., Syrvatka, B. G., Shevchuk, V. U., Bel'ferman, A. L., and Moin, F. B. Mass spectrometric study of fluorine-containing compounds. I. Comparative study of methylacetylene and difluoromethylacetylene, Khim. Vys. Energ. **1**, 411 (1967) [Engl. transl.: High Energy Chem. **1**, 359 (1967)].
- [3770] Herzberg, G., and Jungen, Ch. Rydberg series and ionization potential of the H₂ molecule, J. Mol. Spectry. **41**, 425 (1972).
- [3772] Marr, G. V., and Wherrett, S. R. The ionization of caesium vapour by the method of space charge amplification, J. Phys. B **5**, 1735 (1972).
- [3773] Scheps, R., Florida, D., and Rice, S. A. Comments on the Rydberg spectrum of pyrazine, J. Mol. Spectry. **44**, 1 (1972).
- [3774] Iverson, A. A., and Russell, B. R. A medium resolution study of allene in the vacuum ultraviolet. I. Spectra and a
- [3733] Sweigart, D. A., and Turner, D. W. Lone pair orbitals and their interactions studied by photoelectron spectroscopy. II. Equivalent orbitals in saturated oxygen and sulfur heterocycles, J. Am. Chem. Soc. **94**, 5599 (1972).
- [3734] Thomas, R. K. Photoelectron spectroscopy of hydrogen-bonded systems: spectra of monomers, dimers and mixed complexes of carboxylic acids, Proc. Roy. Soc. (London) **A331**, 249 (1972).
- [3735] Gross, M. L. Ion cyclotron resonance spectrometry. A means of evaluating 'kinetic shifts', Org. Mass Spectrom. **6**, 827 (1972).
- [3736] Salmona, Y. F., and Vincent, E.-J. Études expérimentales et théoriques de potentiels d'ionisation de dérivés de la série de l'isothiazole, C. R. Acad. Sci., Ser. C. **273**, 863 (1971).
- [3737] Lageot, C. Étude des états excités de l'ion HCN⁺, J. Chim. Phys. Phys.-Chim.. Biol. **68**, 214 (1972).
- [3738] Wanckez, K.-P., Lebert, K.-H., and Hartmann, H. Untersuchung der Ion-Molekül-Reaktionen des Thiothionylfluorids mit Hilfe der Ionen-Cyclotronresonanz-Spektrometrie, Z. Naturforsch. **27a**, 155 (1972).
- [3739] Mason, D. C., Kuppermann, A., and Mintz, D. M. Angular distribution of electrons from the photoionization of ethylene, in Electron Spectroscopy, ed. D. A. Shirley (North Holland, Amsterdam, 1972) p. 269.
- [3740] Fridh, C., Åsbrink, L., Jonsson, B. Ö., and Lindholm, E. Rydberg series in small molecules XVIII. Photoelectron, UV, mass, and electron impact spectra of *s*-tetrazine, Intern. J. Mass Spectrom. Ion Phys. **9**, 485 (1972).
- [3741] Haselbach, E., Heilbronner, E., Musso, H., and Schmelzer, A. Notiz über die Photoelektronen-Spektren des Nortricyclens und des Triasterans, Helv. Chim. Acta **55**, 302 (1972).
- [3742] Donovan, R. J., Little, D. J., and Konstantatos, J. Vacuum ultraviolet spectra of transient molecules and radicals, J. Chem. Soc. Faraday Trans. II **68**, 1812 (1972).
- [3743] Gole, J. L., and Margrave, J. L. The vacuum ultraviolet spectrum of molecular fluorine, J. Mol. Spectry. **43**, 65 (1972).
- [3745] Cabaud, B., Hoareau, A., Nounou, P., and Uzan, R. Étude des processus d'ionisation à haute température des vapeurs métalliques par couplage d'une cellule de Knudsen et d'une source fox. II. Interprétation des processus d'autoionisation des vapeurs métalliques monoatomiques et influence de la température sur les courbes d'efficacité d'ionisation, Int. J. Mass Spectrom. Ion Phys. **8**, 181 (1972).
- [3746] Wittel, K., Hass, A., and Bock, H. Photoelektronenspektren und Moleküleigenschaften, XVI. Thiocarbonylhalogenide-Orbitale und Ladungen, Chem. Ber. **105**, 3865 (1972).
- [3747] Barz, P., and Fritz, H. P. Untersuchungen an biochemisch Ligandensystemen, V. Komplexchemische und physikalisch-chemische Untersuchungen am 1,2-Dimethylhydrazin, Z. Naturforsch. **27b**, 1131 (1972).
- [3748] Boschi, R. A., and Salahub, D. R. The far ultra-violet spectra of some 1-iodoalkanes, Mol. Phys. **24**, 289 (1972).
- [3749] Doucet, J., Sauvageau, P., and Sandorfy, C. The vacuum ultraviolet spectrum of tetrahydrofuran, Chem. Phys. Letters **17**, 316 (1972).
- [3750] Fridh, C., Åsbrink, L., Jonsson, B. Ö., and Lindholm, E. Rydberg series in small molecules. XV. Photoelectron, UV, mass and electron impact spectra of pyrazine, Intern. J. Mass Spectrom. Ion Phys. **8**, 101 (1972).
- [3751] Salahub, D. R., and Boschi, R. A. The far ultraviolet spectrum of iodoacetylene, Chem. Phys. Letters **16**, 320 (1972).

- preliminary ionization potential, Spectrochim. Acta **28A**, 447 (1972).
- [3775] Neckel, A., and Sodeck, G. Bestimmung der Dissoziationsenergien der gasförmigen Moleküle CuGe, AgGe und AuGe, Monatsh. Chem. **103**, 367 (1972).
- [3776] Scott, J. D., and Russell, B. R. Vacuum-ultraviolet spectral studies of several chlorofluoroethylenes, J. Am. Chem. Soc. **94**, 2634 (1972).
- [3777] Tajima, S.; and Tsuchiya, T. The effects of the repeller voltage and the shield voltage on appearance potential measurements by electron impact, Shitsurgo Bunseki **20**, 117 (1972).
- [3778] Bock, H., and Stafast, H. Photoelektronenspektren und Moleküleigenschaften, IX. Die π -Systeme der cis- und trans-Dicyan-äthylene, Chem. Ber. **105**, 1158 (1972).
- [3779] Ferreira, M. A. A., and Costa, M. L. Impacto electrónico no oxi-sulfureto de carbono: potenciais de aparecimento de iões positivos, calores de formação e energias de dissociacão, Rev. Port. Quim., **14**, 21 (1972).
- [3780] Brogli, F., Heilbronner, E., and Ipaktschi, J. Die Wechselwirkung zwischen Walsh- und π -Orbitalen im 7-Cyclopropyliden-norbornadien, Helv. Chem. Acta **55**, 2447 (1972).
- [3781] Bock, H., Wagner, G., and Kroner, J. Photoelektronenspektren und Moleküleigenschaften, XIV. Die Delokalisation des Schwefel-Elektronenpaars in CH₃S-substituierten Aromaten, Chem. Ber. **105**, 3850 (1972).
- [3782] Schmidbaur, H., and Vornberger, W. Die Organosiliciumchemie der Phosphor-Ylide, XVI. Si-Si-Struktureinheiten als Carbanion-Substituenten in Yilden, Chem. Ber. **105**, 3173 (1972).
- [3783] Preiss, H. Massenspektrometrische Untersuchungen an einigen Halogeniden der 5. Haupt- und Nebengruppe, Z. Anorg. Allg. Chem. **389**, 280 (1972).
- [3784] Bentley, T. W., Johnstone, R. A. W., and McMaster, B. N. Appearance potentials of metastable and normal ions and the kinetic shift, J. Chem. Soc. Chem. Commun., 510 (1973).
- [3785] Foner, S. N., and Hudson, R. L. Mass spectrometric studies of tetrafluorohydrazine and the difluoroamino radical, J. Chem. Phys. **58**, 581 (1973).
- [3786] Gaivoronskii, P. E., Larin, N. V., Sirokin, N. I., Artemov, A. N., and Shushunov, N. V. Study of arenechromium tricarbonyl complexes by mass spectrometry, Izv. Akad. Nauk SSSR, Ser. Khim. **11**, 2618 (1973) [Engl. transl.: Bull. Acad. Sci. USSR, Div. Chem. Sci. **22**, 2557 (1973)].
- [3787] Aloisi, G. G., and Pignataro, S. Molecular complexes of substituted thiophens with σ and π acceptors, J. Chem. Soc. Faraday Trans. I **69**, 534 (1973).
- [3788] Gilbert, J. R., Leach, W. P., and Miller, J. R. Ionisation and appearance potential measurements in arene chromium tricarbonyls, J. Organometal. Chem. **49**, 219 (1973).
- [3789] Hvistendahl, G., Undheim, K., and Györösi, P. Mass spectrometry of tropylium halides, Org. Mass Spectrom. **7**, 903 (1973).
- [3790] Gross, M. L., and Aerni, R. J. The unusual loss of hydrogen from ionized 1,5-hexadiyne, J. Am. Chem. Soc. **95**, 7875 (1973).
- [3791] Flesch, G. D., and Svec, H. J. Fragmentation reactions in the mass spectrometer for C₂-C₅ alkanes, J. Chem. Soc. Faraday Trans. II **69**, 1187 (1973).
- [3792] Benoit, F. The benzoyl cation: The participation of isolated electronic excited states in the dissociation of molecular ions of the form [C₆H₅COX]⁺, Org. Mass Spectrom. **7**, 1407 (1973).
- [3793] Begun, G. M., and Compton, R. N. Electron impact ionization studies of ferrocene, cobaltocene, nickelocene, and magnesocene, J. Chem. Phys. **58**, 2271 (1973).
- [3794] Branton, G. R., and Pua, C. K. N. Low energy electron impact ionization and fragmentation—cyclobutanone, Can. J. Chem. **51**, 624 (1973).
- [3795] Ackermann, R. J., and Rauh, E. G. The preparation and characterization of the metastable monoxides of thorium and uranium, J. Inorg. Nucl. Chem. **35**, 3787 (1973).
- [3796] Cuthill, A. M., Fabian, D. J., and Shu-Shou-Shen, S. Bond dissociation energies of the metallic vapor species aluminum–silver and aluminum–gold measured by Knudsen-cell mass spectrometry, J. Phys. Chem. **77**, 2008 (1973).
- [3797] Crowe, A., and McConkey, J. W. Dissociative ionization by electron impact II. N⁺ and N⁺⁺ from N₂, J. Phys. B (Proc. Phys. Soc.) **6**, 2108 (1973).
- [3798] Cocke, D. L., Gingerich, K. A., and Kordis, J. Mass spectrometric observation of gaseous EuCN and the determination of its atomization energy, J. Chem. Soc. Chem. Commun. 561 (1973).
- [3799] Crowe, A., and McConkey, J. W. Dissociative ionization by electron impact. I. Protons from H₂, J. Phys. B (Proc. Phys. Soc.) **6**, 2088 (1973).
- [3800] Ciach, S., Knowles, D. J., Nicholson, A. J. C., and Swingler, D. L. Vaporization of tin(II) halides. I. Stannous chloride and stannous bromide, Inorg. Chem. **12**, 1443 (1973).
- [3801] Farber, M., and Srivastava, R. D. Effusion-mass spectrometric study of thermodynamic properties of vanadium and vanadium nitride, J. Chem. Soc. Faraday Trans. I **69**, 390 (1973).
- [3802] Ciach, S., Nicholson, A. J. C., Swingler, D. L., and Thistlethwaite, P. J. Mass spectrometric study of the vapor phase over neodymium chloride and gadolinium chloride, Inorg. Chem. **12**, 2072 (1973).
- [3803] Jalonen, J., Pasanen, P., and Pihlaja, K. Ionisation and appearance potentials in the evaluation of nonbonded interactions. IV: Conformational effects in methyl-substituted 1,3-oxathianes, Org. Mass Spectrom. **7**, 949 (1973).
- [3804] Distefano, G., Pignataro, S., Innorta, G., Fringuelli, F., Marino, G., and Taticchi, A. Ionization energies of selenophen, tellurophen and some of their derivatives, Chem. Phys. Letters **22**, 132 (1973).
- [3805] Rakita, P. E., Hoffman, M. K., Andrews, M. N., and Bursey, M. M. σ - π Conjugation in group IVA compounds of indene and indane, J. Organometal. Chem. **49**, 213 (1973).
- [3806] Pignataro, S., Mancini, V., Innorta, G., and Distefano, G. Ionization energies and ring orbital interaction in diarylmethanes and diarylethylenes, Z. Naturforsch. **27**, 534 (1972).
- [3807] Innorta, G., Torroni, S., Pignataro, S., and Mancini, V. The activation energy as guiding factor in the fragmentation of substituted diphenylmethanes, Org. Mass Spectrom. **7**, 1399 (1973).
- [3808] Sen Sharma, D. K., and Franklin, J. L. Heats of formation of free radicals by mass spectrometry, J. Am. Chem. Soc. **95**, 6562 (1973).
- [3809] Smolinsky, G., and Vasile, M. J. Mass spectra of vinyltrimethylsilane and vinyltri(methyl-d₃)silane, Org. Mass Spectrom. **7**, 1069 (1973).
- [3810] Muenow, D. W. Mass spectrometric evidence for the gaseous silicon oxide nitride molecule and its heat of atomization, J. Phys. Chem. **77**, 970 (1973).
- [3811] Morrison, J. D., and Traeger, J. C. Ionization and dissociation by electron impact. II. NH₃ and PH₃, Intern. J. Mass Spectrom. Ion Phys. **11**, 277 (1973).
- [3812] Momigny, J., Mathieu, G., Wankenne, H., and Ferreira, M. A. A. Collision- and non-collision-induced predissociation in the appearance of S⁺ and CS⁺ ions from CS₂ under electron impact, Chem. Phys. Letters **21**, 606 (1973).
- [3813] Morrison, J. D., and Traeger, J. C. Ionization and dissociation by electron impact. III. CH₄ and SiH₄, Intern. J. Mass Spectrom. Ion Phys. **11**, 289 (1973).

- [3814] Saalfeld, F. E., McDowell, M. V., DeCorpo, J. J., Berry, A. D., and MacDiarmid, A. G. Mass spectral studies of some manganese carbonyls, *Inorg. Chem.* **12**, 48 (1973).
- [3815] Glockling, F., Morrison, R. J., and Wilson, J. W. Diphenylberyllium: electron impact and calorimetric studies, *J. Chem. Soc. Dalton Trans.* 94 (1973).
- [3816] Hildenbrand, D. L. Dissociation energies of the molecules AlO and Al₂O, *Chem. Phys. Letters* **20**, 127 (1973).
- [3817] Henion, J. D., and Kingston, D. G. I. Mass spectrometry of organic compounds. VII. Energetics of substituent isomerization in diphenyl sulfide and diphenyl ether, *J. Am. Chem. Soc.* **95**, 8358 (1973).
- [3818] Hildenbrand, D. L. Mass spectrometric studies of some gaseous sulfur fluorides, *J. Phys. Chem.* **77**, 897 (1973).
- [3819] Dewart, J., Myers, C. E., Szwarc, R., Vander Auwera-Mahieu, A., and Uy, O. M. Determination by the mass spectrometric Knudsen cell method of the atomization energies of the molecules PO and PO₂, *J. Chem. Soc. Faraday Trans. II* **68**, 1749 (1972).
- [3820] Hirayama, C., and Castle, P. M. Mass spectra of rare earth triiodides, *J. Phys. Chem.* **77**, 3110 (1973).
- [3821] Panchenkov, I. G., Gusarov, A. V., and Gorokhov, L. N. Dissociation energy of the barium oxide molecule, *Zh. Fiz. Khim.* **47**, 101 (1973) [Engl. transl. *Russ. J. Phys. Chem.* **47**, 55 (1973)].
- [3822] Condorelli, G., Fragalà, I., Centineo, G., and Tondello, E. The electronic structure and photoelectron spectra of Ni^{II}, Cu^{II} and Pd^{II} complexes with N,N'-ethylene-bis(acetylacetoneiminato) dianion, *Inorg. Chim. Acta* **7**, 725 (1973).
- [3823] Scheppelle, S. E., Mitchum, R. K., Kinneberg, K. F., Meissels, G. G., and Emmel, R. H. Internal energy distributions and the fragmentation of gaseous organic ions. Dissociation of ions produced by electron impact on 4-methylbenzil, *J. Am. Chem. Soc.* **95**, 5105 (1973).
- [3824] Cowling, S. A., Johnstone, R. A. W., Gorman, A. A., and Smith, P. G. Photoelectron spectrum of 5-methylenenorborn-2-ene and through-space interactions (homobutadiene conjugation), *J. Chem. Soc. Chem. Commun.* 627 (1973).
- [3825] Cowley, A. H., Dewar, M. J. S., Goodman, D. W., Schweiger, J. R. Stereochemical dependence of lone pair interactions in the photoelectron spectra of nitrogen-phosphorus compounds, *J. Am. Chem. Soc.* **95**, 6506 (1973).
- [3826] Bain, A. D., and Frost, D. C. Studies of the carbonyl group in some five-membered ring compounds by photoelectron spectroscopy, *Can. J. Chem.* **51**, 1245 (1973).
- [3827] Cradock, S., Ebsworth, E. A. V., and Robertson, A. Photoelectron spectra of some silyl and germyl transition-metal carbonyls and related species, *J. Chem. Soc. Dalton Trans.* 22 (1973).
- [3828] Boyd, R. J., Bünzli, J. C., Snyder, J. P., and Heyman, M. L. Photoelectron spectra of 2,3-diazabicyclo[2.2n]alk-2-enes ($n = 1, 2, 3, 4$), *J. Am. Chem. Soc.* **95**, 6478 (1973).
- [3829] Eley, D. D., Hazeldine, D. J., and Palmer, T. F. Mass spectra, ionisation potentials and related properties of metal-free and transition metal phthalocyanines, *J. Chem. Soc. Faraday Trans. II* **69**, 1808 (1973).
- [3830] Evans, S., Green, J. C., and Jackson, S. E. He-(I) photoelectron spectra of some metal complexes containing the ligands trimethylsilylmethyl and neopentyl, *J. Chem. Soc. Faraday Trans. II* **69**, 191 (1973).
- [3831] Berkowitz, J., Dehmer, J. L., and Appelman, E. H. Photoelectron spectrum of hypofluorous acid, HOF, *Chem. Phys. Letters* **19**, 334 (1973).
- [3832] Batich, C., Heilbronner, E., Hornung, V., Ashe, A. J., Clark, D. T., Cobley, U. T., Kilcast, D., and Scanlan, I. Photoelectron spectra of phosphabenzenes, arsabenzenes, and stibabenzenes, *J. Am. Chem. Soc.* **95**, 928 (1973).
- [3833] Cocksey, B. G., Eland, J. H. D., and Danby, C. J. Photoelectron spectra of the zinc and cadmium halides, *J. Chem. Soc. Faraday Trans. II* **69**, 1558 (1973).
- [3834] Dromey, R. G., Morrison, J. D., and Peel, J. B. Time-averaged and deconvoluted photoelectron spectrum of the first band of O₂, *Chem. Phys. Letters* **23**, 30 (1973).
- [3835] Bünzli, J. C., Frost, D. C., and McDowell, C. A. Photoelectron spectra of phosphoryl and thiophosphoryl chlorides and bromides, *J. Electron Spectrosc. Relat. Phenom.* **1**, 481 (1972/73).
- [3836] Foster, S., Felps, S., Cusachs, L. C., and McGlynn, S. P. Photoelectron spectra of osmium and ruthenium tetroxides, *J. Am. Chem. Soc.* **95**, 5521 (1973).
- [3837] Frost, D. C., Herring, F. G., Katrib, A., and McDowell, C. A. The photoelectron spectrum of ethylene sulphide, *Chem. Phys. Letters* **20**, 401 (1973).
- [3838] Diemann, E., and Müller, A. The He(I) photoelectron spectra of OsO₄ and RuO₄, *Chem. Phys. Letters* **19**, 538 (1973).
- [3839] Delwiche, J., Natalis, P., Momigny, J., and Collin, J. E. On the photoelectron spectra of HBr and DBr, *J. Electron Spectrosc. Relat. Phenom.* **1**, 219 (1972/73).
- [3840] Frost, D. C., Lee, S. T., and McDowell, C. A. The photoelectron spectrum of HCP and comments on the first photoelectron band of HCN, *Chem. Phys. Letters* **23**, 472 (1973).
- [3841] Frost, D. C., Lee, S. T., and McDowell, C. A. The He(I) photoelectron spectrum of S₂O, *Chem. Phys. Letters* **22**, 243 (1973).
- [3842] Bünzli, J. C., Frost, D. C., and Weiler, L. Photoelectron spectrum of 7-thiabicyclo[2.2.1]heptane, *J. Am. Chem. Soc.* **95**, 7880 (1973).
- [3843] Bain, A. D., Bünzli, J. C., Frost, D. C., and Weiler, L. Photoelectron spectra of cyclic ethers, *J. Am. Chem. Soc.* **95**, 291 (1973).
- [3844] Bock, H., Mollère, P., Becker, G., and Fritz, G. Photoelectron spectra and molecular properties. XX. Dimethyl ether, methoxysilane, and disiloxane, *J. Organometal. Chem.* **61**, 113 (1973).
- [3845] Cooks, R. G., Bertrand, M., Beynon, J. H., Rennekamp, M. E., and Setser, D. W. Energy partitioning data as an ion structure probe. Substituted anisoles, *J. Am. Chem. Soc.* **95**, 1732 (1973).
- [3846] Boschi, R., and Schmidt, W. The photoelectron spectrum and structure of sulfur in the gas phase at 140°C, *Inorg. Nucl. Chem. Lett.* **9**, 643 (1973).
- [3847] Beez, M., Bieri, G., Bock, H., and Heilbronner, E. The ionization potentials of butadiene, hexatriene, and their methyl derivatives: evidence for through space interaction between double bond π -orbitals and non-bonded pseudo- π orbitals of methyl groups?, *Helv. Chim. Acta* **56**, 1028 (1973).
- [3848] Houk, K. N., Davis, L. P., Newkome, G. R., Duke, Jr., R. E., and Nauman, R. V. Photoelectron spectroscopy of cyclic β -diketones and their enolone tautomers, *J. Am. Chem. Soc.* **95**, 8364 (1973).
- [3849] Heilbronner, E., Gleiter, R., Hoshi, T., and Meijere, A. The interaction of Walsh-orbitals in diademane and related hydrocarbons, *Helv. Chim. Acta* **56**, 1594 (1973).
- [3850] Schweig, A., Weidner, U., and Manuel, G. Theory and application of photoelectron spectroscopy. XVI. Photoelectron spectroscopy and molecular conformations: Ge-C and Sn-C hyperconjugation and the conformation of allylgermanes and -stannanes, *J. Organometal. Chem.* **54**, 145 (1973).
- [3851] Miller, L. L., Koch, V. R., Koenig, T., Tuttle, M. Photoelectron spectroscopy and the anodic fragmentation of adamantane derivatives, *J. Am. Chem. Soc.* **95**, 5075 (1973).
- [3852] Johnstone, R. A. W., and Mellon, F. A. Photoelectron spectroscopy of sulphur-containing heteroaromatics and molecular orbital calculations, *J. Chem. Soc. Faraday*

- Trans. II **69**, 1155 (1973).
- [3853] Haselbach, E., and Eberbach, W. The photolysis of tricyclo[4.2.1.0^{2,5}] nonadiene: support of a Dougherty Type N' mechanism from photoelectron spectroscopical studies, *Helv. Chim. Acta* **56**, 1944 (1973).
- [3854] Maier, J. P., and Turner, D. W. Steric inhibition of resonance studied by molecular photoelectron spectroscopy. Part 2. Phenylethylenes, *J. Chem. Soc. Faraday Trans. II* **69**, 196 (1973).
- [3855] Schmidt, W., Wilkins, B. T., Fritz, G., and Huber, R. Energy level trends in 1,3,5,7-tetrasilaadamantanes ("carborundanes") and related molecules from photoelectron spectroscopy, *J. Organometal. Chem.* **59**, 109 (1973).
- [3856] Khalil, O. S., Meeks, J. L., and McGlynn, S. P. Electronic spectroscopy of highly polar aromatics. VII. Photoelectron spectra of nitroanilines, *J. Am. Chem. Soc.* **95**, 5876 (1973).
- [3857] Koch, E. E., Otto, A., and Radler, K. The absorption spectrum of the anthracene molecule in the vacuum ultraviolet, *Chem. Phys. Letters* **21**, 501 (1973).
- [3858] Schäfer, W., Schweig, A., Gronowitz, S., Taticchi, A., and Fringuelli Reversal in the sequence of two highest occupied molecular orbitals in the series thiophen, selenophen, and tellurophen, *J. Chem. Soc. Chem. Commun.*, 541 (1973).
- [3859] Schmidt, H., Schweig, A., and Manuel, G. Preliminary communication. Theory and application of photoelectron spectroscopy. XXX. Hg-C hyperconjugation, *J. Organometal. Chem.* **55**, C1 (1973).
- [3860] Haselbach, E., Lanyiova, Z., and Rossi, M. On the correlation between ionization potentials and excitation energies, part III: pyrazine, *Helv. Chim. Acta* **56**, 2889 (1973).
- [3861] Schweig, A., and Thiel, W. Photoionization cross sections: He I and He II photoelectron spectra of saturated three-membered rings, *Chem. Phys. Letters* **21**, 541 (1973).
- [3862] Osafune, K., Katsumata, S., and Kimura, K. Photoelectron spectroscopic study of hydrazine, *Chem. Phys. Letters* **19**, 369 (1973).
- [3863] Mines, G. W., and Thompson, H. W. Photoelectron spectra of vinyl and allyl halides, *Spectrochim. Acta* **29A**, 1377 (1973).
- [3864] Katrib, A., and Rabalais, J. W. Electronic interaction between the vinyl group and its substituents, *J. Phys. Chem.* **77**, 2358 (1973).
- [3865] Lloyd, D. R., Roberts, P. J. The assignment of the photoelectron spectrum of sulphur dioxide, *Mol. Phys.* **26**, 225 (1973).
- [3866] Lichtenberger, D. L., Sarapu, A. C., and Fenske, R. F. Photoelectron spectra and electronic structure of pentacarbonylmanganese halides, *Inorg. Chem.* **12**, 702 (1973).
- [3867] Mollère, P., Bock, H., Becker, G., Fritz, G. Photoelectron spectra and molecular properties. XXI. Dimethyl sulfide, methyl silyl sulfide, and disilyl sulfide, *J. Organometal. Chem.* **61**, 127 (1973).
- [3868] McLean, R. A. N. The bonding of a silicon atom with a phenyl ring: the photoelectron spectrum of phenylsilane, *Can. J. Chem.* **51**, 2089 (1973).
- [3869] Jones, R. W., Koski, W. S. Photoelectron spectrum of pentaborane, *J. Chem. Phys.* **59**, 1228 (1973).
- [3870] Higginson, B. R., Lloyd, D. R., and Roberts, P. J. Variable temperature photoelectron spectroscopy. The adiabatic ionization potential of the iodine molecule, *Chem. Phys. Letters* **19**, 480 (1973).
- [3871] Fehlner, T. P., and Turner, D. W. Photoelectron spectrum of HBS, *J. Am. Chem. Soc.* **95**, 7175 (1973).
- [3872] Goodman, D. W., Dewar, M. J. R., Schweiger, J. R., and Cowley, A. H. The photoelectron spectrum of phosphorus pentafluoride, *Chem. Phys. Letters* **21**, 474 (1973).
- [3873] Streets, D. G., and Ceasar, G. P. Inductive and mesomeric effects on the π orbitals of halobenzenes, *Mol. Phys.* **26**, 1037 (1973).
- [3874] Watanabe, I., Yokoyama, Y., and Ikeda, S. Lone pair ionization potentials of carboxylic acids determined by He(I) photoelectron spectroscopy, *Bull. Chem. Soc. Japan* **46**, 1959 (1973).
- [3875] Sugar, J. Ionization energies of the neutral actinides, *J. Chem. Phys.* **59**, 788 (1973).
- [3876] Evans, K., Scheps, R., Rice, S. A., and Heller, D. Primary photochemical and photophysical processes in chloro- and bromo-acetylene, *J. Chem. Soc. Faraday Trans. II* **69**, 856 (1973).
- [3877] Aihara, J., and Inokuchi, H. Ionization potentials of anthracene, *Chem. Letters*, 421 (1973).
- [3878] Collins, R. J., Husain, D., and Donovan, R. J. Kinetic and spectroscopic studies of $O_2(a^1\Delta_g)$ by time-resolved absorption spectroscopy in the vacuum ultra-violet, *J. Chem. Soc. Faraday Trans. II* **69**, 145 (1973).
- [3879] Chadwick, D., Frost, D. C., Herring, F. G., Katrib, A., McDowell, C. A., and McLean, R. A. N. Photoelectron spectra of sulfonyl and thionyl halides, *Can. J. Chem.* **51**, 1893 (1973).
- [3880] Jonas, A. E., Schweitzer, G. K., Grimm, F. A., and Carlson, T. A. The photoelectron spectra of the tetrafluoro and tetramethyl compounds of the group IV elements, *J. Electron Spectrosc. Relat. Phenom.* **1**, 29 (1972/73).
- [3881] Chaghtai, M. S. Z., and Hassan, V. The ionization potential and the $4s4p^6n l$ levels of ^{86}Kr I, *J. Phys. B* **6**, 433 (1973).
- [3882] Basco, N., and Morse, R. D. Analysis of the Rydberg transitions in ethylene sulphide, *Chem. Phys. Letters* **20**, 404 (1973).
- [3883] Watanabe, I., Yokoyama, Y., and Ikeda, S. Vibrational structures in the photoelectron spectrum of formic acid, *Chem. Phys. Letters* **19**, 406 (1973).
- [3884] Zverev, V. V., Vovna, V. I., Èl'man, M. S., Kitaev, Yu. P., and Vilesov, F. I. Photoelectron spectra and structure of hydrazones, *Dokl. Akad. Nauk SSSR* **213**, 1319 (1973) [Engl. Transl.: *Dokl. Phys. Chem.* **213**, 945 (1973)].
- [3885] Gompper, R., Holsboer, F., Schmidt, W., and Seybold, G. Rapid double bond shift in a donor acceptor substituted cyclobutadiene. Evidence from 584-Å and X-ray photoelectron spectroscopy, *J. Am. Chem. Soc.* **95**, 8479 (1973).
- [3886] Worley, S. D., Mateescu, G. D., McFarland, C. W., Fort, R. C., Jr., and Sheley, C. F. Photoelectron spectra and MINDO-SCF-MO calculations for adamantine and some of its derivatives, *J. Am. Chem. Soc.* **95**, 7580 (1973).
- [3887] Nelsen, S. F., Buschek, J. M., and Hintz, P. J. Photoelectron spectra of hydrazines. II. Conformations of hexahydropyridazines, *J. Am. Chem. Soc.* **95**, 2013 (1973).
- [3888] Haselbach, E., Mannschreck, A., and Seitz, W. Lone pair electronic structure, conformation and oxidation behaviour of diaziridines, *Helv. Chim. Acta* **56**, 1614 (1973).
- [3889] Nelsen, S. F., and Buschek, J. M. Photoelectron spectra of hydrazines. I. Dependence of the lone pair-lone pair splitting on dihedral angle for tetraalkylhydrazines, *J. Am. Chem. Soc.* **95**, 2011 (1973).
- [3890] Maier, J. P., and Turner, D. W. Steric inhibition of resonance studied by molecular photoelectron spectroscopy, *J. Chem. Soc. Faraday Trans. II* **69**, 521 (1973).
- [3891] Gronneberg, T., and Undheim, K. Mass spectrometry of onium compounds - XV. ionization potentials of amino pyridines, *Tetrahedron Letters*, 3193 (1973).
- [3892] Sustmann, R.; and Schubert, R. Photoelektronenspektroskopische Bestimmung von Substituenten-Effekten. I. Substituierte Butadiene,

- Tetrahedron Letters **27**, 2739 (1972).
- [3906] Appell, J., and Durup, J. The formation of protons by impact of low energy electrons on water molecules, Inten. J. Mass Spectrom. Ion Phys. **10**, 247 (1972/73).
- [3907] Mateescu, G. D., and Workey, S. D. Electron spectroscopy. II. Photoelectron spectra of adamantane and 1-bromoadamantane, Tetrahedron Letters **52**, 5285 (1972).
- [3908] Weidner, U., Schweig, A. Nature of the "silicon β -effect" in allyltrimethylsilane, Angew. Chem. Intern. Ed. **11**, 146 (1972).
- [3909] Hildenbrand, D. L. Thermochemistry of the molecular species LiO, LiO⁺, and Li₂O⁺, J. Chem. Phys. **57**, 4556 (1972).
- [3910] Klebe, K. J., Houte, J. J. v., and Thuijl, J. v. Loss of HCN and H from the molecular ion of imidazole, Org. Mass Spectrom **6**, 1363 (1972).
- [3911] Rabalais, J. W., Werme, L. O., Bergmark, T., Karlsson, L., and Siegbahn, K. The high resolution electron spectra of thiophene, 2-bromothiophene and 3-bromothiophene, Intern. J. Mass Spectrom. Ion Phys. **9**, 185 (1972).
- [3913] Berkowitz, J. Photoelectron spectroscopy of high-temperature vapors. I. TiCl, TiBr, and TiI, J. Chem. Phys. **56**, 2766 (1972).
- [3914] Doucet, J., Sauvageau, P., and Sandorfy, C. Vacuum ultraviolet and photoelectron spectra of fluoro-chloro derivatives of methane, J. Chem. Phys. **58**, 3708 (1973).
- [3915] Van Der Helm, D., Christian, S. D., and Lin, L.-N. Charge transfer complexes of purines and pyrimidines. 9-Cyclohexyladenine-iodine in organic solvents and in the solid state, J. Am. Chem. Soc. **95**, 2410 (1973).
- [3916] McLafferty, F. W., Bente, P. F., III., Kornfeld, R., Tsai, S.-C., and Howe, I. Collisional activation spectra of organic ions, J. Am. Chem. Soc. **95**, 2120 (1973).
- [3918] Fedorova, M. S., Denisov, Yu. V., and Potapov, V. K. Mass-spectrometric study of the photoionisation processes of tricyclo[5.2.1.0^{2,6}] decane and its alkyl derivatives, Zh. Fiz. Khim. **47**, 2667 (1973) [Engl. transl.: Russ. J. Phys. Chem. **47**, 1498 (1973)].
- [3919] Stockbauer, R. Threshold electron-photoion coincidence mass spectrometric study of CH₄, CD₄, C₂H₆, and C₂D₆, J. Chem. Phys. **58**, 3800 (1973).
- [3920] Berkowitz, J., Dehmer, P. M., and Chupka, W. A. Photoionization mass spectrometry of F₂O, J. Chem. Phys. **59**, 925 (1973).
- [3921] Dibeler, V. H., and Walker, J. A. Photoionization of acetylene near threshold, Intern. J. Mass Spectrom. Ion Phys. **11**, 49 (1973).
- [3922] Pitt, C. G. Hyperconjugation and its role in group IV chemistry, J. Organometal. Chem. **61**, 49 (1973).
- [3923] Minnagen, L. Spectrum and the energy levels of neutral argon, Ar I, J. Opt. Soc. Am. **63**, 1185 (1973).
- [3924] Reader, J., and Epstein, G. Zeeman effect and revised analysis of singly ionized rubidium (Rb II), J. Opt. Soc. Am. **63**, 1153 (1973).
- [3925] McCulloh, K. E. Photoionization of carbon dioxide, J. Chem. Phys. **59**, 4250 (1973).
- [3927] Killgoar, P. C., Jr., Leroi, G. E., Chupka, W. A., and Berkowitz, J. Photoionization study of NO₂. I. The ionization potential, J. Chem. Phys. **59**, 1370 (1973).
- [3928] Berkowitz, J., and Wahl, A. C. The dissociation energy of fluorine, Advan. Fluorine Chem. **7**, 147 (1973).
- [3929] Okabe, H., and Dibeler, V. H. Photon impact studies of C₂HCN and CH₃CN in the vacuum ultraviolet; heats of formation of C₂H and CH₃CN, J. Chem. Phys. **59**, 2430 (1973).
- [3930] Reinke, D., Kraessig, R., and Baumgärtel, H. Photoreactions of small organic molecules, Z. Naturforsch. **28a**, 1021 (1973).
- [3931] Dibeler, V. H., Walker, J. A., and McCulloh, K. E. Observations on hot bands in the molecular and dissociative photoionization of acetylene and the heat of formation of the ethynyl ion, J. Chem. Phys. **59**, 2264 (1973).
- [3932] Berkowitz, J., Appelman, E. H., and Chupka, W. A. Photoionization of HOF with mass analysis, J. Chem. Phys. **58**, 1950 (1973).
- [3933] Hoffmann, R. W., Schüttler, R., Schäfer, W., and Schweig, A. Methylen-norbornadien, ein Bicycloheptafulven, Angew. Chem. **84**, 533 (1972).
- [3934] Oehling, H., Schäfer, W., and Schweig, A. Sequence of highest occupied molecular orbitals in the phosphorin system, Angew. Chem. Intern. Ed. **10**, 656 (1971).
- [3935] Lloyd, D. R. Calibration of a He(I) photoelectron spectrometer, J. Phys. E **3**, 629 (1972).
- [3936] Cowan, D. O., Gleiter, R., Hashmall, J. A., Heilbronner, E., and Hornung, V. Interaction between the orbitals of lone pair electrons in dicarbonyl compounds, Angew. Chem. Intern. Ed. **10**, 401 (1971).
- [3937] Sustmann, R., and Trill, H. Photoelektronenspektroskopische Bestimmung von Substituenten-Effekten. II. α,β -ungesättigte Carbonester, Tetrahedron Letters **42**, 4271 (1972).
- [3938] Rabalais, J. W., and Colton, R. J. Electronic interaction between the phenyl group and its unsaturated substituents, J. Electron Spectrosc. Relat. Phenom. **1**, 83 (1972/73).
- [3939] Tamás, J., Czira, G., Mal'tsev, A. K., and Nefedov, O. M. Electron impact studies on some organochlorogermaines: mass spectra and bond dissociation energies, J. Organometal. Chem. **40**, 311 (1972).
- [3940] Weidner, U., and Schweig, A. Theory and application of photoelectron spectroscopy. V. The nature of bonding in vinyl-and allylsilanes: the effects of $\sigma-\pi$ (hyperconjugation) and $p_{\pi}-d_{\pi}$ conjugation in these compounds, J. Organometal. Chem. **39**, 261 (1972).
- [3941] Robin, M. B., and Kuebler, N. A. Excited electronic states of the simple alcohols, J. Electron Spectrosc. Relat. Phenom. **1**, 13 (1972/73).
- [3942] Golob, L., Jonathan, N., Morris, A., Okuda, M., and Ross, K. J. The first ionization potential of the methyl radical as determined by photoelectron spectroscopy, J. Electron Spectrosc. Relat. Phenom. **1**, 506 (1972/73).
- [3943] Brundelle, C. R., and Jones, G. R. The high resolution photoelectron spectra and the electronic structure of XeOF₄, J. Electron Spectrosc. Relat. Phenom. **1**, 403 (1972/73).
- [3945] Dôong, P., and Bizot, M. Dissociation uni- et bi-moléculaire des ions NO⁺, Intern. J. Mass Spectrom. Ion Phys. **10**, 227 (1972/73).
- [3946] Pitt, C. G., and Bock, H. $\sigma-\pi$ Mixing in phenylpentamethyldisilane, J. Chem. Soc. Chem. Commun., 28 (1972).
- [3947] Piacente, V., and Malaspina, L. Dissociation energy of the TIAs molecule, J. Chem. Phys. **56**, 1780 (1972).
- [3948] Boekelheide, V., Murrell, J. N., and Schmidt, W., The photoelectron spectrum of *trans*-15,16-dimethyl-dihydropyrene, Tetrahedron Letters **7**, 575 (1972).
- [3949] De Maria, G., Malaspina, L., and Piacente, V. Dissociation energy of the gaseous TiBi molecule, J. Chem. Phys. **56**, 1978 (1972).
- [3950] Mollère, P., Bock, H., Becker, G., and Fritz, G. Photoelectron spectra and molecular properties. XV. The effects of α - and β -silyl substituents on π -systems, J. Organometal. Chem. **46**, 89 (1972).
- [3951] Boschi, R., and Schmidt, W. Photoelectron spectra of polycyclic aromatic hydrocarbons. Pyrene and coronene, Tetrahedron Letters **25**, 2577 (1972).
- [3952] Saalfeld, F. E., DeCorpo, J. J., and McDowell, M. V. The mass spectra of some metal carbonyl complexes of tris(dimethylamino)phosphine, J. Organometal. Chem. **44**,

- 333 (1972).
- [3953] Boschi, R., Schmidt, W., and Gfeller, J.-C. The electronic structure of 1,6-methano-cyclodecapentaene, *Tetrahedron Letters*, 4107 (1972).
- [3954] Schaaf, D. W., and Gregory, N. W. Mass spectrometric study of the vaporization of cuprous bromide, *J. Phys. Chem.* **76**, 3271 (1972).
- [3955] Debies, T. P., and Rabalais, J. W. Photoelectron spectra of substituted benzenes. II. Seven valence electron substituents, *J. Electron Spectrosc. Relat. Phenom.* **1**, 355 (1972/73).
- [3956] Cabaud, B., Hoareau, A., Nounou, P., and Uzan, R. High temperature mass spectrometric study of polyatomic antimony species by electron impact. Direct evidence for the existence of Sb₃ molecules, *Intern. J. Mass Spectrom. Ion Phys.* **11**, 157 (1973).
- [3957] Masclet, P., Grosjean, D., and Mouvier, G. Alkene ionization potentials. Part I. Quantitative determination of alkyl group structural effects, *J. Electron Spectrosc. Relat. Phenom.* **2**, 225 (1973).
- [3958] Berkowitz, J., Dehmer, J. L., and Walker, T. E. H. PES of hightemperature vapors. IV. The cesium halides. Effect of spin-orbit interaction on the photoelectron and mass spectra of the alkali halides, *J. Chem. Phys.* **59**, 3645 (1973).
- [3959] Van Den Ham, D. M. W., and Van Der Meer, D. Photoelectron spectra of some fluorine substituted diazanaphthalenes, *J. Electron Spectrosc. Relat. Phenom.* **2**, 247 (1973).
- [3960] Berkowitz, J., Dehmer, J. L., Shimada, K., and Szwarc, M. Photoelectron spectroscopic studies of (α -naphthyl)-(CH₂)_n(α -naphthyl) vapor: open chain or cyclic conformation?, *J. Elec. Spectrosc. Relat. Phenom.* **2**, 211 (1973).
- [3961] Kordis, J., and Gingerich, K. A. Mass spectroscopic investigation of the equilibrium dissociation of gaseous Sb₂, Sb₃, Sb₄, SbP, SbP₃, and P₂, *J. Chem. Phys.* **58**, 5141 (1973).
- [3962] Ackermann, R. J., and Rauh, E. G. High temperature properties of the thorium-oxygen system: a revision of the thermodynamic properties of ThO(g) and ThO₂(g), *High Temp. Sci.* **5**, 463 (1973).
- [3963] Boggess, G. W., Allen, J. D., Jr., and Schweitzer, G. K. The photoelectron spectra of gaseous zinc(II) and cadmium(II) chlorides, bromides, and iodides, *J. Electron Spectrosc. Relat. Phenom.* **2**, 467 (1973).
- [3964] Kobayashi, T., Yokota, K., and Nagakura, S. Photoelectron spectra of styrenes, *J. Electron Spectrosc. Relat. Phenom.* **3**, 449 (1973).
- [3965] Frost, D. C., Lee, S. T., and McDowell, C. A. Photoelectron spectra of OCSe, SCSe, and CSe₂, *J. Chem. Phys.* **59**, 5484 (1973).
- [3966] Gingerich, K. A., Cocke, D. L., Finkbeiner, H. C., and Chang, C.-A. High temperature Knudsen cell mass spectrometric determination of the heats of atomization of AlAu₂ and Al₂Au, *Chem. Phys. Letters* **18**, 102 (1973).
- [3967] Morrison, J. D., and Traeger, J. C. Ionization and dissociation by electron impact. I. H₂O and H₂S, *Intern. J. Mass Spectrom. Ion Phys.* **11**, 77 (1973).
- [3968] Gingerich, K. A. Mass spectrometric evidence for the very high stability of gaseous ThIr and ThPt and method of calculating dissociation energies of diatomic intermetallic compounds with multiple bonds, *Chem. Phys. Letters* **23**, 270 (1973).
- [3969] Guido, M., Gigli, G. Mass spectrometric study of the CeSiC molecules, *J. Chem. Phys.* **59**, 3437 (1973).
- [3970] Scott, J. D., Causley, G. C., and Russell, B. R. Vacuum ultraviolet absorption spectra of dimethylsulfide, dimethylselenide, and dimethyltelluride, *J. Chem. Phys.* **59**, 6577 (1973).
- [3971] Dehmer, J. L., Berkowitz, J., and Cusachs, L. C. Photoelectron spectroscopy of high-temperature vapors. III. Monomer and dimer spectra of thallous fluoride, *J. Chem. Phys.* **58**, 5681 (1973).
- [3973] Benoit, F. Substituent effects in mass spectrometry. III. Substituent effects in the dissociation of the molecular ions of *para* and *meta* substituted benzoic acids, *Org. Mass Spectrom.* **7**, 295 (1973).
- [3974] Kaufman, V., and Sugar, J. One-electron spectrum of singly ionized ytterbium (Yb II), *J. Opt. Soc. Am.* **63**, 1168 (1973).
- [3975] Gardner, J. L., and Samson, J. A. R. 304. Å photoelectron spectra of CO, N₂, O₂ and CO₂, *J. Electron Spectrosc. Relat. Phenom.* **2**, 259 (1973).
- [3976] Work, D. E., and Eick, H. A. An investigation of the incongruent sublimation of some lanthanide (III) oxobromides, *High Temp. Sci.* **5**, 313 (1973).
- [3977] Fjeldstad, P. E., and Undheim, K. Mass spectrometry of onium compounds. XXX. Ionisation potential in structural assignment of some gaseous molecules, *Org. Mass Spectrom.* **7**, 639 (1973).
- [3978] Cocke, D. L., Gingerich, K. A., and Kordis, J. Determination of the high bond dissociation energy of the molecule LaRh, *High Temp. Sci.* **5**, 474 (1973).
- [3979] Higginson, B. R., Lloyd, D. R., Burroughs, P., Gibson, D. M., and Orchard, A. F. Photoelectron studies of metal carbonyls. Part 2. The valence region photoelectron spectra of the Group VIA hexacarbonyls, *J. Chem. Soc. Faraday Trans. II* 1659 (1973).
- [3980] Mollere, P. D. The photoelectron spectrum of oxetane: non-degenerate Walsh orbitals in a four-membered heterocycle, *Tetrahedron Letters*, 2791 (1973).
- [3981] Gleiter, R., Schmidt, E., Cowan, D. O., and Ferraris, J. P. The electronic structure of tetrathiofulvalene, *J. Electron Spectrosc. Relat. Phenom.* **2**, 207 (1973).
- [3982] Kroto, H. W., Suffolk, R. J., and Westwood, N. P. C. The photoelectron spectrum of thiaborine, HBS, *Chem. Phys. Letters* **22**, 495 (1973).
- [3983] Katayama, D. H., Huffman, R. E., and O'Bryan, C. L. Absorption and photoionization cross sections for H₂O and D₂O in the vacuum ultraviolet, *J. Chem. Phys.* **59**, 4309 (1973).
- [3984] Yamazaki, T., Katsumata, S., and Kimura, K. Photoelectron spectra and orbital assignments by sum rule consideration: ethyl and *n*-propyl fluorides, *J. Electron Spectrosc. Relat. Phenom.* **2**, 335 (1973).
- [3985] Thompson, K. R. Mass spectrometric determination of the atomization energies of AlSiO(g) and Al₂O(g), *High Temp. Sci.* **5**, 62 (1973).
- [3986] Piacente, V., Bardi, G., Malaspina, L., and Desideri, A. Dissociation energy of CeO₂ and Ce₂O₂ molecules, *J. Chem. Phys.* **59**, 31 (1973).
- [3987] Leavell, S., Steichen, J., and Franklin, J. L. Photoelectron spectra of intramolecularly hydrogen bonded compounds, *J. Chem. Phys.* **59**, 4343 (1973).
- [3988] Cowling, S. A., and Johnstone, R. A. W. Photoelectron spectroscopy: the effects of steric inhibition to resonance in anilines, *J. Electron Spectrosc. Relat. Phenom.* **2**, 161 (1973).
- [3989] Santoro, E. The fragmentation of some alkyl thiophosphate esters by electron-impact, *Org. Mass Spectrom.* **7**, 589 (1973).
- [3990] Boschi, R., Schmidt, W., Suffolk, R. J., Wilkins, B. T., Lempka, H. J., and Ridyard, J. N. A. Complete valence shell electronic structure of adamantane from He I and He II photoelectron spectroscopy, *J. Electron Spectrosc. Relat. Phenom.* **2**, 377 (1973).
- [3991] Goldstein, M. J., Natowsky, S., Heilbronner, E., and Hornung, V. Near cancellation of through space and through bond interaction in bicyclo[3.2.2]nona-6,8-diene, *Helv. Chim.*

- Acta **56**, 294 (1973).
- [3992] Schmidt, H., and Schweig, A. C-Hal Hyperkonjugation, Tetrahedron Letters, 981 (1973).
- [3993] Müller, C., and Schweig, A. Konjugation in Sulfonen, Tetrahedron **29**, 3973 (1973).
- [3994] Schweig, A., Weidner, U., Berger, J. G., and Grahn, W. Spirokonjugation, Tetrahedron Letters, 557 (1973).
- [3995] Schmidt, H., and Schweig, A. Ausschluss transanularer Wechselwirkung in 2,5-Dihydrothiophen, Tetrahedron Letters, 1437 (1973).
- [3996] Mathar, W., Bohlmann, F., and Schwarz, H. Massenspektrometrische Untersuchung von Amiden. V. Über den Einfluss der N-Donatorstärke auf die Aktivierungs-Energie der Methyl-Abspaltung aus Crotonsäureamiden, Tetrahedron Letters, 4583 (1973).
- [3997] Bruckmann, P., and Klessinger, M. Photoelektronenspektren organischer Verbindungen. III. Photoelektronenspektren acetylenesubstituierter kleiner Ringe, J. Electron Spectrosc. Relat. Phenom. **2**, 341 (1973).
- [3998] Eland, J. H. D. Predissociation of N_2O^+ and COS^+ ions studied by photoelectronphotoion coincidence spectroscopy, Intern. J. Mass Spectrom. Ion Phys. **12**, 389 (1973).
- [3999] Batich, C., Bischof, P., and Heilbronner, E. The photoelectron spectra of cyclooctatetraene and its hydrogenated derivatives, J. Electron Spectrosc. Relat. Phenom. **1**, 333 (1972/73).
- [4000] Schmidt, W. Photoelectron spectra of diamondoid molecules, adamantane, silamantane and urotropine, Tetrahedron **29**, 2129 (1973).
- [4001] Drowth, J., Myers, C. E., Szwarc, R., Vander Auwera-Mahieu, A., and Uy, O. M. The dissociation energies of the molecules PS, PSe, and PTe, High Temp. Sci. **5**, 482 (1973).
- [4002] Bieri, G., Brogli, F., Heilbronner, E., and Kloster-Jensen, E. A photoelectron spectroscopic investigation of the electronic structure of trimethylsilylhaloacetylenes, J. Electron Spectrosc. Relat. Phenom. **1**, 67 (1972/73).
- [4003] Botter, R., Menes, F., Gounelle, Y., Pechine, J. M., and Solgadi, D. The ionization potentials of geometrical isomers: the *cis* and *trans* 2-substituted cyclopentyl and cyclohexyl bromides, Intern. J. Mass Spectrom. Ion Phys. **12**, 188 (1973).
- [4004] Bünzli, J. C., Frost, D. C., Weiler, L. The photoelectron spectrum of triquinacene, Tetrahedron Letters, 1159 (1973).
- [4005] Stearns, C. A., and Kohl, F. J. Mass spectrometric determination of the dissociation energies of gaseous Al_2 , $AlSi$, and $AlSiO$, High Temp. Sci. **5**, 113 (1973).
- [4006] Gleiter, R., Heilbronner, E., Paquette, L. A., Thompson, G. L., and Wingard, R. E., Jr. Photoelectron spectra of polyunsaturated [4,4,2]propellanes, Tetrahedron **29**, 565 (1973).
- [4008] Ceasar, G. P., Green, J., Paquette, L. A., and Wingard, R. E., Jr. Orbital interaction in 2a,8b-dihydrocyclop[cd]azulene, Tetrahedron Letters, 1721 (1973).
- [4009] Cradock, S., Findlay, R. H., and Palmer, M. H. The molecular energy levels of the azoles: a study by photoelectron spectroscopy and ab initio molecular orbital calculations, Tetrahedron **29**, 2173 (1973).
- [4010] Bünzli, J. C., Burak, A. J., and Frost, D. C. Through-space interaction in non-conjugated acyclic dienes studied by photoelectron spectroscopy, Tetrahedron **29**, 3735 (1973).
- [4011] Tondello, G. Absorption spectrum of Cu I in the vacuum ultraviolet, J. Opt. Soc. Am. **63**, 346 (1973).
- [4012] Kordis, J., and Gingerich, K. A. Dissociation energies and heats of formation of the gaseous Eu_2 and $EuAg$ molecules, J. Phys. Chem. **77**, 700 (1973).
- [4013] Rabeneck, H., Rinke, K., and Schäfer, H. ReO_3Jg Bildung, Massenspektrum, Ionisierungsenergie und Bildungsenthalpie, Z. Anorg. Allg. Chem. **397**, 112 (1973).
- [4014] Stearns, C. A., and Kohl, F. J. Mass spectrometric determination of the dissociation energies of AlC_2 , Al_2C_2 , and $AlAuC_2$, J. Phys. Chem. **77**, 136 (1973).
- [4015] Müller, J., and Goll, W. Ion-Molekül-Reaktionen von (Cyclopentadienyl)-nitrosoNickel mit σ - und π -Donatoren in der Gasphase, Chem. Ber. **106**, 1129 (1973).
- [4016] Skinner, H. B., and Searcy, A. W. Mass spectrometric studies of gaseous oxides of rhenium, J. Phys. Chem. **77**, 1573 (1973).
- [4017] Clark, P. A., Gleiter, R., and Heilbronner, E. Photoelectron spectra of planar sulfur heterocycles, Tetrahedron **29**, 3085 (1973).
- [4018] Fortin, C. J., and Rousseau, Y. Spectrométrie de masse des cyclohexanones *gem*-diphénylées. II. Chaleurs de formation et structures possibles des principaux ions fragmentaires, Can. J. Chem. **51**, 3457 (1973).
- [4019] Brogli, F., Crandall, J. K., Heilbronner, E., Kloster-Jensen, E., and Sojka, S. A. The photoelectron spectra of methyl-substituted allenes and of tetramethyl-bisallenyl, J. Electron Spectrosc. Relat. Phenom. **2**, 455 (1973).
- [4020] Tanaka, K., and Tanaka, I. Photoelectron spectra from some autoionizing states of O_2 near the ionization threshold, J. Chem. Phys. **59**, 5042 (1973).
- [4021] Nixon, J. F. Photoelectron spectra and bonding in metal-trifluorophosphine complexes, J. Chem. Soc. Dalton Trans. **21**, 2226 (1973).
- [4022] Katrib, A., Debies, T. P., Colton, R. J., Lee, T. H., and Rabalais, J. W. The use of differential photoionization cross sections as a function of excitation energy in assigning photoelectron spectra, Chem. Phys. Letters **22**, 196 (1973).
- [4023] Berkosky, J. L., Ellison, F. O., Lee, T. H., and Rabalais, J. W. Model for calculating spin-orbit interactions with applications to photoelectron spectroscopy, J. Chem. Phys. **59**, 5342 (1973).
- [4024] Kröner, J., Strack, W., Holsboer, F., and Kosbahn, W. Zur Elektronenstruktur der Thiokumulene, Z. Naturforsch. **28b**, 188 (1973).
- [4025] Akopyan, M. E., Sergeev, Yu. L., and Vilesov, F. I. Photoionization in vapors of aliphatic sulfides. I. Methylmercaptan, dimethyl and diethyl sulfides, Khim. Vys. Energ. **4**, 305 (1970) [Engl. transl.: High Energy Chem. **4**, 265 (1970)].
- [4026] Cradock, S., Ebsworth, E. A. V., and Whiteford, R. A. Photoelectron spectra of some simple fluorosilanes, J. Chem. Soc. Dalton Trans. **22**, 2401 (1973).
- [4027] Venkateswarlu, P. The vacuum ultraviolet spectrum of ICl , Can. J. Phys. **53**, 812 (1975).
- [4028] Potapov, V. K., and Iskakov, L. I. Electronic structure and photoionization of aromatic amines, Khim. Vys. Energ. **5**, 264 (1971) [Engl. transl.: High Energy Chem. **5**, 237 (1971)].
- [4029] Kobayashi, H., Kobayashi, M., and Kaizu, Y. Molecular complexes of arenetricarbonylchromium, Bull. Chem. Soc. Japan **46**, 3109 (1973).
- [4030] Antonova, N. L., Kutsev, V. S. Mass-spectrometric investigation of the thermal dissociation of neodymium dicarbide, Zh. Fiz. Khim. **47**, 2446 (1973) [Engl. transl.: Russ. J. Phys. Chem. **47**, 1385 (1973)].
- [4031] Iskakov, L. I., Potapov, V. K. Photoionization and decomposition of benzaldehyde, acetophenone, and benzophenone, Khim. Vys. Energ. **5**, 265 (1971) [Engl. transl.: High Energy Chem. **5**, 238 (1971)].
- [4032] Ogata, H., Onizuka, H., Nihei, Y., Kamada, H. The photoelectron spectra of alcohols, mercaptans and amines, Bull. Chem. Soc. Japan **46**, 3036 (1973).

- [4033] Hoshino, H., Tajima, S., and Tsuchiya, T. The effect of the temperature on the mass spectra of aliphatic primary alcohols and 1-alkenes. I. Bull. Chem. Soc. Japan **46**, 3043 (1973).
- [4034] Askani, R., Gleiter, R., Heilbronner, E., Hornung, V., and Musso, H. The orbital sequence in semibullvalene, barbaralene and dihydrobullvalene, Tetrahedron Letters, 4461 (1971).
- [4035] Pozharskii, A. F., Kashparov, I. S., Holls, P. J., and Zaletov, V. G. Heterocyclic pleiadene analogs. VI. Electronic properties of perimidine, Khim. Geterotsikl. Soedin. **4**, 543 (1971) [Engl. transl.: Chem. Heterocycl. Compd. **4**, 507 (1971)].
- [4036] Shen, K.-W., and Kuebler, N. A. Synthesis, reactions, and photoelectron spectrum of 8,11-dimethylenepentacyclo [5.4.0^{2,6}.0^{3,10}.0^{5,9}] tridecane, Tetrahedron Letters, 2145 (1973).
- [4037] Bischof, P., Haselbach, E., and Heilbronner, E. Photoelectron spectrum of cyclobutane, Angew. Chem. Intern. Ed. **9**, 953 (1970).
- [4038] Heilbronner, E., and Muszkat, K. A. On the relative importance of through-space vs. through-bond interaction between the lone pairs in 1,4-diazabicyclo[2.2.2]octane, J. Am. Chem. Soc. **92**, 3818 (1970).
- [4039] Brogli, F., and Heilbronner, E. The photoelectron spectra of benzenoid hydrocarbons C₁₈H₁₂, Angew. Chem. Intern. Ed. **11**, 538 (1972).
- [4040] Brogli, F., Eberbach, W., Haselbach, E., Heilbronner, E., Hornung, V., and Lemal, D. M. 199. Die Photoelektronen-Spektren des Tricyclo[4.2.1.0^{2,5}]nonadiens und seines 3,4-Diaza-Analogons. Ein Beitrag zur Kenntnis der Wechselwirkung zwischen den einsamen Elektronenpaaren der cis-konfigurierten Azogruppe, Helv. Chim. Acta **56**, 1933 (1973).
- [4041] Schwarz, H., Bohlmann, F., and Russ, B. Elektronenstossinduzierte Fragmentierung von Polymethylbenzaldehyden. II. Mechanismus der Methyl-Abspaltung aus dem Molekül-Ion von tri- und Pentamethylbenzaldehyden, Org. Mass Spectrom. **7**, 1001 (1973).
- [4043] Zverev, V. V., Vovna, V. I., El'man, M. S., Kitaev, Yu. P., and Vilessov, F. I. Photoelectronic spectra and electronic and three-dimensional structures of acyclic azines, Dokl. Akad. Nauk. SSSR **213**, 1117 (1973) [Engl. transl.: Dokl. Phys. Chem. **213**, 1100 (1973)].
- [4044] Schwarz, H., and Bohlmann, F. Elektronenstossinduzierte Fragmentierung von Acetylenverbindungen. VI. Struktur und Bildungsenthalpie der Ionen [C₁₁H₉]⁺ und [C₉H₇]⁺, Org. Mass Spectrom. **7**, 395 (1973).
- [4045] Gleiter, R., Heilbronner, E., Hekman, M., and Martin, H.-D. π-Orbital-Wechselwirkungen "through space" and "through bond" in tricyclo[4.2.0.0^{2,5}]octadienen, Chem. Ber. **106**, 28 (1973).
- [4046] Schwarz, H., and Bohlmann, F. Massenspektrometrische Untersuchung von Amiden. I. Energetische Betrachtungen zur elektronenstossinduzierten Fragmentierung von Piperidin- und Piperidein-Amiden, Org. Mass Spectrom. **7**, 1197 (1973).
- [4047] Heilbronner, E., and Martin, H.-D. Über die Orbitalsequenz in aliphatischen Diazoverbindungen, Chem. Ber. **106**, 3376 (1973).
- [4048] Brogli, F., Heilbronner, E., Hornung, V., and Kloster-Jensen, E. 230. Die Photoelektronen-Spektren methylsubstituierter Acetylene, Helv. Chim. Acta **56**, 2171 (1973).
- [4049] Batich, C., Heilbronner, E., and Semmelhack, M. F. 225. Bemerkung zur Gleichheit der Aufspaltungen ΔI (zwischen den ersten beiden π-Ionisationspotentialen) und ΔE (zwischen den entsprechenden π*→Π Übergangsenergien) des Spiro[4,4] nonatetraens, Helv. Chim. Acta **56**, 2110 (1973).
- [4050] Schmidt, W., and Wilkins, B. T. Das "Equivalent Orbital" (EO)-Verfahren zur Interpretation von Photoelektronen(PE)-Spektren: Neopentan, Angew. Chem. **84**, 168 (1972).
- [4051] Schwarz, H., Bohlmann, F., and Vorlaender, W. Elektronenstossinduzierte Fragmentierung von Polymethylbenzaldehyden. III. Bildung und Zerfall des Formyltropylium-Ions aus Dimethylbenzaldehyd, Org. Mass Spectrom. **7**, 1005 (1973).
- [4052] Paulus, J.-M., and Abbé, J.-C. Potentiel d'apparition de I₂²⁺ a partir de I₂, J. Chim. Phys. **70**, 690 (1973).
- [4053] Schweig, A., Schäfer, W., and Dimroth, K. Unusual sequence of the two highest occupied π-molecular orbitals in the phosphorin system, Angew. Chem. Intern. Ed. **11**, 631 (1972).
- [4054] Uy, O. M., Srivastava, R. D., and Farber, M. Mass spectrometric determination of the heats of formation of gaseous BO₂ and BOF₂, High Temp. Sci. **3**, 462 (1971).
- [4055] Rodionov, A. N., Potapov, V. K., and Rogozhin, K. L. Photoionization of certain aromatic heteroorganic compounds, Khim. Vys. Energ. **7**, 278 (1973) [Engl. transl.: High Energy Chem. **7**, 249 (1973)].
- [4056] Ikuta, S., Yoshihara, K., Shiokawa, T., Jinno, M., Yokoyama, Y., and Ikeda, S. Photoelectron spectroscopy of cyclohexane, cyclopentane, and some related compounds, Chem. Letters, 1237 (1973).
- [4057] Iskakov, L. I., and Potapov, V. K. Photoionization and decomposition of benzaldehyde, acetophenone, and benzophenone, Khim. Vys. Energ. **5**, 265 (1971) [Engl. transl.: High Energy Chem. **5**, 238 (1971)].
- [4058] Potapov, V. K., and Bazhenov, B. A. The photoionization of pyrrole, furan, and thiophene, Khim. Vys. Energ. **4**, 553 (1970) [Engl. transl.: High Energy Chem. **4**, 505 (1970)].
- [4060] Camus, P., and Tomkins, F. S. Absorption-line series in Lu I, J. Phys. (Paris) **33**, 197 (1972).
- [4061] Ackermann, R. J., and Rauth, E. G. A high-temperature study of the stoichiometry, phase behavior, vaporization characteristics, and thermodynamic properties of the cerium + oxygen system, J. Chem. Thermodyn. **3**, 609 (1971).
- [4062] Schwarz, H., Praefcke, K., and Martens, J. Organische Schwefelverbindungen. III. Elektronenstossinduzierte Untersuchungen von Arylestern der Monothioind Dithiophthalsäure und der isomeren 3,3-substituierten Phthalide, Tetrahedron **29**, 2877 (1973).
- [4063] Brogli, F., Giovannini, E., Heilbronner, E., and Schurter, R. Die Photoelektronen-Spektren der Benzocycloalkene, Chem. Ber. **106**, 961 (1973).
- [4065] Kröner, J., Nölle, D., and Nöth, H. Photoelektronenspektroskopische Untersuchungen an Bor-Verbindungen. I. Orbitalreihenfolgen und Ladungsdichten in Methylthioind Methoxyborane, Z. Naturforsch. Teil B **28**, 416 (1973).
- [4066] Schäfer, W., Schweig, A., Märkl, G., and Heier, K.-H. Zur Elektronenstruktur der λ₃- und λ₅-Phosphanaphthaline—ungewöhnlich grosse MO Destabilisierungen, Tetrahedron Letters 3743 (1973).
- [4067] Stafast, H., Bock, H. Photoelectron spectra and molecular properties. XVII. Hyperconjugation in dicyano methane and 2,2-dicyano propane, Z. Naturforsch. **28b**, 746 (1973).
- [4068] Katsumata, S., Iwai, T., and Kimura, K. Photoelectron spectra and sum rule consideration. Higher alkyl amines and alcohols, Bull. Chem. Soc. Japan **46**, 3391 (1973).
- [4069] Parr, G. R., and Taylor, J. W. A photoionization mass spectrometer utilizing a high intensity molecular beam sampling system and synchrotron radiation, Rev. Sci.

- Instrum. **44**, 1578 (1973).
[4070] Syrvatka, B. G., Gil'burd, M. M., Bel'ferman, A. L. Mass spectrometric study of chlorofluoro-substituted ethylenes, Zh. Org. Khim. **8**, 1553 (1972) [Engl. transl.: J. Org. Chem. USSR **8**, 1587 (1972)].
[4071] Ivko, A. A. Use of mass spectroscopy and isotope labelling for determining the structure of ions and molecules, Org. Katal., **20** (1970).
[4072] Puttemans, J.-P., and Hanson, A. Etude énergétique du ferrocène et du cobaltoène par impact électronique. Enthalpie de formation du radical cyclopentadiényle, Ing. Chim. (Brussels) **53**, 17 (1971).
[4073] Natalis, P. Contribution à la spectroscopie photoélectronique. Effets de l'autoionisation dans les spectres photoélectroniques de molécules diatomiques et triatomiques, Acad. R. Belg. Mem. Cl. Sci. Collect. 8°-2° Ser. T **41/1**, (1973).
[4074] Fortin, C. J., Forest, M., Vaziri, C., Gravel, D., and Rousseau, Y. Spectrométrie de masse des cyclohexanones *gem*-diphénylées. I. Localisation de la charge positive, Can. J. Chem. **51**, 3445 (1973).
[4075] Rosenstock, H. M., Larkins, J. T., and Walker, J. A. Interpretation of photoionization thresholds: quasiequilibrium theory and the fragmentation of benzene, Intern. J. Mass Spectrom. Ion Phys. **11**, 309 (1973).
[4076] Kimura, K., Katsumata, S., Achiba, Y., Matsumoto, H., and Nagakura, S. Photoelectron spectra and orbital structures of higher alkyl chlorides, bromides, and iodides, Bull. Chem. Soc. Japan **46**, 373 (1973).
[4077] Cundy, C. S., Lappert, M. F., Pedley, J. B., Schmidt, W., and Wilkins, B. T. Bonding studies of compounds of boron and the Group IV elements. XI. Photoelectron spectra of strained cyclic organosilicon compounds, J. Organometal. Chem. **51**, 99 (1973).
[4078] Sergeev, Yu. L., Akopyan, M. E., Vilesov, F. I., and Chizhov, Yu. V. Photoionization processes in gaseous cyclohexane, and chloro- and bromocyclohexane, Khim. Vys. Energ. **7**, 418 (1973) [Engl. transl.: High Energy Chem. **7**, 369 (1973)].
[4079] Poltorakov, A. P., Pirnazarova, F. N., But, P. G., Piruzyan, L. A., Chibrikov, V. M., Vikhlyayev, Yu. I., and Ul'yanova, O. V. Ionization potentials of phenothiazine derivatives and their correlation with the pharmacological effect, Izv. Akad. Nauk SSSR, Ser. Khim. 2106 (1973) [Engl. transl.: Bull. Acad. Sci. USSR, Div. Chem. Sci. **22**, 2050 (1973)].
[4080] Mines, G. W., Thomas, R. K., and Thompson, H. The photoelectron spectra of thiocarbonyl fluoride and thiocarbonyl chloride, Proc. Roy. Soc. Lond. A **333**, 171 (1973).
[4081] Schweig, A., Weidner, U., Hellwinkel, D., and Krapp, W. Spiroconjugation, Angew. Chem. Intern. Ed. **12**, 310 (1973).
[4082] Kobayashi, T., and Nagakura, S. Photoelectron spectra of tetrahydropyran, 1,3-dioxane, and 1,4-dioxane, Bull. Chem. Soc. Japan **46**, 1558 (1973).
[4083] Schweig, A., Weidner, U., Hill, R. K., and Cullison, D. A. A quantitative account of spiroconjugation, J. Am. Chem. Soc. **95**, 5426 (1973).
[4084] Robin, M. B., Taylor, G. N., Kuebler, N. A., and Bach, R. D. Planarity of the carbon skeleton in various alkylated olefins, J. Org. Chem. **38**, 1049 (1973).
[4085] Rademacher, P. Photoelectron spectra and conformation of hydrazine derivatives, Angew. Chem. Intern. Ed. **12**, 408 (1973).
[4086] Vovna, V. I., Lopatin, S. N., Pettsold, R., Vilesov, F. I., and Akopyan, M. E. Photoelectron spectra of thiophosphorylchloride and some of its aminosubstituted derivatives, Opt. i Spektrosk. **34**, 868 (1973) [Engl. transl.: Opt. Spectry. **34**, 501 (1973)].
[4087] Ogata, H., Onizuka, H., Nihei, Y., and Kamada, H. On the first bands of the photoelectron spectra of amines, alcohols, and mercaptans, Chem. Letters 895 (1972).
[4088] Boschi, R., and Schmidt, W. Transannular π - π interaction in cyclophanes, Angew. Chem. Intern. Ed. **12**, 402 (1973).
[4089] Johnstone, R. A. W., and Mellon, F. A. Effects of induction and resonance in the calculation of ionization potentials of substituted benzenes by perturbation molecular orbital theory, J. Chem. Soc. Faraday Trans. II **69**, 36 (1973).
[4090] Schäfer, W., Schweig, A., Märkl, G., Hauptmann, H., and Mathey, F. Direct proof of the non-aromaticity of phospholes and arsoles, Angew. Chem. Intern. Ed. **12**, 145 (1973).
[4091] Schmidt, H., and Schweig, A. Semiquantitative proof of hyperconjugation, Angew. Chem. Intern. Ed. **12**, 307 (1973).
[4092] Bock, H., Solouki, B., Rosmus, P., and Steudel, R. Photoelectron spectra and molecular properties: SSO and OSO, Angew. Chem. Intern. Ed. **12**, 933 (1973).
[4094] Reetz, M. T., Hoffmann, R. W., Schäfer, W., and Schweig, A. Methylenecyclo [4.2.1]nona-2,4,7-triene, Angew. Chem. Intern. Ed. **12**, 81 (1973).
[4095] Samson, J. A. R., and Gardner, J. L. Fluorescence excitation and photoelectron spectra of CO₂ induced by vacuum ultraviolet radiation between 185 and 716 angstroms, J. Geophys. Res. **78**, 3663 (1973).
[4096] Bagarat'yan, N. V., Il'in, M. K., and Nikitin, O. T. Mass-spectrometric study of thallium metaborate, Teplosif. Vysokikh Temperatur **11**, 995 (1973) [Engl. transl.: High Temp. (USSR) **11**, 888 (1973)].
[4097] Benito, I., Seidl, H., and Bock, H. Efectos electronicos y estericos de sustituyentes alquilicos y silicicos sobre el sistema electronico π del estireno, Rev. Fac. Cienc. Univ. Oviedo **14**, 95 (1973).
[4098] Smoes, S., and Drowart, J. Atomization energies of phosphorus oxides, Faraday Symp. Chem. Soc. **139** (1973).
[4099] Tan, H.-S., and Lampe, F. W. The reaction of ethyl radicals with nitric oxide. Nitrosoethane and triethylhydroxylamine formation, J. Phys. Chem. **76**, 3303 (1972).
[4100] Ames, L. L., Wang, J. L.-F., and Margrave, J. L. The vaporization of cesium nitrate, Inorg. Nucl. Chem. Letters **9**, 1243 (1973).
[4102] Smoes, S., Depière, D., and Drowart, J. The atomization energies of the gaseous molecules, Rev. Int. Hautes Temp. Refract. **9**, 171 (1972).
[4103] Wu, H. Y., and Wahlbeck, P. G. Vapor pressures of TiO(g) in equilibrium with Ti₂O₃(s) and Ti₃O₅(s, β); dissociation energy of TiO(g), J. Chem. Phys. **56**, 4534 (1972).
[4104] Schäfer, W., and Schweig, A. Evidence against the significance of C-S hyperconjugation in determining the conformation of allyl methyl sulphide, J. Chem. Soc. Chem. Commun., 824 (1972).
[4105] Guido, M., and Balducci, G. Dissociation energy of Yb₂, J. Chem. Phys. **57**, 5611 (1972).
[4106] Kobayashi, T., and Nagakura, S. Photoelectron spectra of anilines, Chem. Letters, 1013 (1972).
[4107] Kobayashi, T., and Nagakura, S. Photoelectron spectra of nitro-compounds, Chem. Letters, 903 (1972).
[4108] Semenov, G. A., Nikolaev, E. N., and Opendak, I. G. Mass-spectrometric investigation of the vaporisation of barium and magnesium per-rhenates, Zh. Neorg. Khim. **17**, 1819 (1972) [Engl. transl.: Russ. J. Inorg. Chem. **17**, 943 (1972)].
[4110] Lichtenberger, D. I., and Fenski, R. F. Assignment of the photoelectron spectra of Mn(CO)₅CH₃ and Mn(CO)₅CF₃, Inorg. Chem. **13**, 486 (1974).
[4111] Piacente, V., and Balducci, G. The dissociation energy of the

- molecule GaSb, *High Temp. Sci.* **6**, 254 (1974).
- [4112] Kohl, F. J., and Stearns, C. A. Vaporization and dissociation energies of the molecular carbides of titanium, zirconium, hafnium, and thorium, *High Temp. Sci.* **6**, 284 (1974).
- [4113] Farber, M., and Srivastava, R. D. Dissociation energies of BeF and BeCl and the heat of formation of BeClF, *J. Chem. Soc. Faraday Trans. I* **70**, 1581 (1974).
- [4114] Rauh, E. G., and Ackermann, R. J. First ionization potentials of some refractory oxide vapors, *J. Chem. Phys.* **60**, 1396 (1974).
- [4115] Hoffman, M. K. Hidden rearrangements in the mass spectral decomposition of cycloheptatriene, *Z. Naturforsch.* **29a**, 1077 (1974).
- [4116] Gambino, O., Vaglio, G. A., Ferrari, R. P., and Valle, M. Ionization and appearance potentials of cobalt carbonyl complexes, *J. Organometal. Chem.* **76**, 89 (1974).
- [4117] Undheim, K., El-Gendy, M. A. F., and Hurum, T. Mass spectrometry of onium compounds. XXVI. Ionisation potentials in structure analysis of gaseous aminopyridine 1-oxides, *Acta Chem. Scand., Ser. B* **28**, 743 (1974).
- [4118] Maeda, K., Suzuki, I. H., and Koyama, Y. Ionization efficiency curves of ethylene by electron impact, *Intern. J. Mass Spectrom. Ion Phys.* **14**, 273 (1974).
- [4119] Neubert, A., and Zmbov, K. F. Mass spectrometric determination of the dissociation energy of the ThO molecule, *High Temp. Sci.* **6**, 303 (1974).
- [4120] Gingerich, K. A., Cocke, D. L., and Kordis, J. Gaseous phosphorus compounds. X. Mass spectrometric determination of the dissociation energies of arsenic and bismuth monophosphides, *J. Phys. Chem.* **78**, 603 (1974).
- [4121] Dube, G., and Chvalovsky, V. Electron impact fragmentation of silyl-substituted phenyldimethylsilanes, *Coll. Czech. Chem. Commun.* **39**, 2641 (1974).
- [4122] Hirayama, C., Castle, P. M., Liebermann, R. W., Zollweg, R. J., and Camp, F. E. Vapor pressure of samarium diiodide and mass spectra of vapors over samarium diiodide and thulium triiodide, *Inorg. Chem.* **13**, 2804 (1974).
- [4123] Hildenbrand, D. L., and Murad, E. Ionization potential of thorium, *J. Chem. Phys.* **61**, 5466 (1974).
- [4124] Beauchamp, J. L., Caserio, M. C., and McMahon, T. B. Ion-molecule reactions of *tert*-butyl alcohol by ion cyclotron resonance spectroscopy, *J. Am. Chem. Soc.* **96**, 6243 (1974).
- [4125] Dube, G., and Chvalovsky, V. Electron impact fragmentation of substituted phenyldimethylsilanes, *Coll. Czech. Chem. Commun.* **39**, 2621 (1974).
- [4126] MacLean, D. I., and Sacher, R. E. A study of some spectroscopic properties of Group IVA acetylides, *J. Organometal. Chem.* **74**, 197 (1974).
- [4127] Price, S. J. W., and Sapiano, H. J. C₆F₅X bond dissociation energies: determination from appearance potential measurements and correlation with thermochemical data, *Can. J. Chem.* **52**, 4109 (1974).
- [4128] Wagner, L. C., and Grimley, R. T. A mass spectrometric study of the bismuth vapor system by the angular distribution technique, *Chem. Phys. Letters* **29**, 594 (1974).
- [4129] Crowe, A., and McConkey, J. W. Dissociative ionization by electron impact. III. O⁺, CO⁺ and C⁺ from CO₂, *J. Phys. B* **7**, 349 (1974).
- [4130] Gingerich, K. A. Thermodynamic evidence for quadruple bond formation in the molecule ThRu and possible maximum bond energy between ligand-free metal atoms, *Chem. Phys. Letters* **25**, 523 (1974).
- [4131] Bennett, S. L., Lin, S.-S., and Gilles, P. W. High-temperature vaporization of ternary systems. I. Mass spectrometry of oxygen-rich vanadium-tungsten-oxygen species, *J. Phys. Chem.* **78**, 266 (1974).
- [4132] Evans, S., Green, J. C., Jackson, S. E., and Higginson, B. He(I) photoelectron spectra of some transition-metal sandwich complexes, *J. Chem. Soc. Dalton Trans.*, 304 (1974).
- [4133] Yoshikawa, K., Hashimoto, M., and Morishima, I. Photoelectron spectroscopic study of cyclic amines. The relation between ionization potentials, basicities, and s character of the nitrogen lone pair electrons, *J. Am. Chem. Soc.* **96**, 288 (1974).
- [4134] Nelsen, S. F., and Buschek, J. M. Photoelectron spectra of hydrazines. V. Pyrazolidine and hexahydropyridazine derivatives, *J. Am. Chem. Soc.* **96**, 6987 (1974).
- [4135] Schmidt, H., Schweig, A., Trost, B. M., Neubold, H. B., and Scudder, P. H. Influence of geometry on cyclopropyl participation in the thermolysis of azo compounds. A photoelectron spectroscopic rationalization, *J. Am. Chem. Soc.* **96**, 622 (1974).
- [4136] Schmidt, H., Schweig, A., Anastassiou, A. G., and Yamamoto, H. Photoelectron spectroscopic evidence for bicycloconjugation in 9-azabicyclo[4.2.1]nona-2,4,7-triene, *J. Chem. Soc. Chem. Commun.*, 218 (1974).
- [4137] Nelsen, S. F., and Buschek, J. M. Photoelectron spectra of hydrazines. III. Evidence for similar lone pair-lone pair dihedral angles for acyclic hydrazines, *J. Am. Chem. Soc.* **96**, 2392 (1974).
- [4138] Westwood, N. P. C. The photoelectron spectrum of silicon difluoride, *Chem. Phys. Letters* **25**, 558 (1974).
- [4139] Ramsey, B. G., Brock, A., Bassindale, A. R., and Bock, H. $\sigma\pi^*$, a reassignment of the long wavelength UV transition in acyl-silanes and -germanes by photoelectron spectroscopy, *J. Organometal. Chem.* **74**, C41 (1974).
- [4140] Bünzli, J. C., Frost, D. C., and Weiler, L. Photoelectron spectrum of tropone. Inductive effect of carbonyl group, *J. Am. Chem. Soc.* **96**, 1952 (1974).
- [4141] Nelsen, S. F., and Buschek, J. M. Photoelectron spectra of some cyclic di- and polyamines. Lone pair-lone pair interaction in 1,3- and 1,4-diamines, *J. Am. Chem. Soc.* **96**, 7930 (1974).
- [4142] Martin, H.-D., Heller, C., Haselbach, E., and Lanyova, Z. Quadracyclanes. Part I: photoelectron spectra and electronic structure, *Helv. Chim. Acta* **57**, 465 (1974).
- [4143] Maier, J. P. Photoelectron spectroscopy of *peri*- amino naphthalenes, *Helv. Chim. Acta* **57**, 994 (1974).
- [4144] Raymonda, J. W., Edwards, L. O., and Russell, B. R. Vacuum ultraviolet absorption spectra of some chloroalkanes, *J. Am. Chem. Soc.* **96**, 1708 (1974).
- [4145] Pignataro, S., and Distefano, G. $n-\sigma$ mixing in pentatomic heterocyclic compounds of sixth group by photoelectron spectroscopy, *Chem. Phys. Letters* **26**, 356 (1974).
- [4146] Nicholson, D. G., and Rademacher, P. Photoelectron spectra and electronic structures of antimony(III) halides, *Acta Chem. Scand. Ser. A* **28**, 1136 (1974).
- [4147] Liu, M. B., and Wahlbeck, P. G. Knudsen effusion and mass spectrometric studies of the vaporization of Y₂O₃(s). Dissociation energy of YO(g), *High Temp. Sci.* **6**, 179 (1974).
- [4148] Mines, G. W., and Thomas, R. K. The photoelectron spectrum of sulphur trioxide: Jahn-Teller distortion in SO₃⁺, *Proc. Roy. Soc. (London) A* **336**, 355 (1974).
- [4149] Solouki, B., Rosmus, P., and Bock, H. Photoelectron spectra and molecular properties. SCl₂-ionization potentials and ab initio SCF calculations, *Chem. Phys. Letters* **26**, 20 (1974).
- [4150] Staley, R. H., and Beauchamp, J. L. Basicities and ion-molecule reactions of the methylphosphines in the gas phase by ion cyclotron resonance spectroscopy, *J. Am. Chem. Soc.* **96**, 6252 (1974).
- [4151] Distefano, G., Ricci, A., Colonna, F. P., Pietropaolo, D., and Pignataro, S. Bonding between sulfur and the elements of Group IV studied by UV photoelectron spectroscopy, *J. Organometal. Chem.* **78**, 93 (1974).
- [4152] Debies, T. P., and Rabalais, J. W. Photoelectron spectra of

- substituted benzenes. III. Bonding with Group V substituents, Inorg. Chem. **13**, 308 (1974).
- [4155] Berlinsky, A. J., Carolan, J. F., and Weiler, L. Photoelectron spectrum and electronic structure of tetrathiofulvalene (TTF), Can. J. Chem. **52**, 3373 (1974).
- [4156] Nelsen, S. F., and Buschek, J. M. Photoelectron spectra of hydrazines. IV. Empirical estimation of lone pair-lone pair dihedral angles and prediction of lone pair ionization potentials for some cyclic and bicyclic hydrazines, J. Am. Chem. Soc. **96**, 6982 (1974).
- [4157] Harris, D. H., and Lappert, M. F. Monomeric, volatile bivalent amides of Group IV_b elements, $M(NR^1_2)_2$ and $M(NR^1R^2)_2$ ($M=Ge, Sn$ or Pb ; $R^1=Me_3Si$, $R^2=Me_3C$), J. Chem. Soc. Chem. Commun., 895 (1974).
- [4158] Heilbronner, E., and Maier, J. P. Consequences of substitution in the photoelectron spectra of [2,2] paracyclophanes: separation of 'through-space' and 'through-bond' interactions as a consequence of fluorosubstitution, Helv. Chim. Acta **57**, 151 (1974).
- [4159] Haink, H. J., Adams, J. E., and Huber, J. R. The electronic structure of aromatic amines: photoelectron spectroscopy of diphenylamine, iminobiphenyl, acridan and carbazole, Ber. Bunsenges. **78**, 436 (1974).
- [4160] Ensslin, W., Bock, H., and Becker, G. Photoelectron spectra and molecular properties. XXX. π interactions in H_3Si - and H_3C -substituted acetylenes, J. Am. Chem. Soc. **96**, 2757 (1974).
- [4161] Bischof, P., Gleiter, R., and Hofmann, P. Reversal in the sequence of the two highest occupied molecular orbitals in the series pyrazine, 2,6-dimethyl-pyrazine, and tetramethylpyrazine, J. Chem. Soc. Chem. Commun., 767 (1974).
- [4162] Heilbronner, E., Hornung, V., Maier, J. P., and Kloster-Jensen, E. The photoelectron spectra of halodiacetylenes, dihalodiacetylenes, and halomethylidiacetylenes, J. Am. Chem. Soc. **96**, 4252 (1974).
- [4163] Sasaki, T., Eguchi, S., Kiriyama, T., Sakito, Y., and Kato, H. Photoelectron spectroscopic evidence concerning ground-state through- σ -bond interaction in the 1,3-diazadamantan-6-one system, J. Chem. Soc. Chem. Commun., 725 (1974).
- [4165] Freiser, B. S., and Beauchamp, J. L. Gas phase ion chemistry and photochemistry of ions generated from perfluoropropylene. Photodissociation of the perfluoroallyl cation, J. Am. Chem. Soc. **96**, 6260 (1974).
- [4166] Evans, S., Hamnett, A., and Orchard, A. F. Concerning the ultraviolet photoelectron spectra of osmium and ruthenium tetroxide, J. Am. Chem. Soc. **96**, 6221 (1974).
- [4167] Ceasar, G. P., Milazzo, P., Cihonski, J. L., and Levenson, R. A. Photoelectron spectra of the rhenium pentacarbonyl halides, Inorg. Chem. **13**, 3035 (1974).
- [4168] Osafune, K., and Kimura, K. Photoelectron spectroscopic study of hydrogen peroxide, Chem. Phys. Letters **25**, 47 (1974).
- [4169] Brundle, C. R. He (I) and He (II) photoelectron spectra of ozone, Chem. Phys. Letters **26**, 25 (1974).
- [4170] Frost, D. C., Lee, S. T., and McDowell, C. A. High resolution photoelectron spectroscopy of ozone, Chem. Phys. Letters **24**, 149 (1974).
- [4171] Hino, S., and Inokuchi, H. Photoelectron spectrum (He I) of 9-methylanthracene, Chem. Letters, 363 (1974).
- [4172] Schweig, A., Weidner, U., and Manuel, G. Theory and application of photoelectron spectroscopy. XLIV. A quantitative account of hyperconjugation in allyl and benzyl compounds of elements of Group IVB, J. Organometal. Chem. **67**, C4 (1974).
- [4173] Fedorova, M. S., Potapov, V. K., Denisov, Yu. V., Sorokin, V. V., and Evlasheva, T. I. A mass-spectrometric study of the photoionisation of certain cyclic hydrocarbons, Zh. Fiz. Khim. **48**, 1828 (1974) [Engl. transl.: Russ. J. Phys. Chem. **48**, 1078 (1974)].
- [4174] Gorodyskii, V. A., Pozdnyakov, V. P., Siretskii, Yu. G., Fadeeva, I. I., and Kozlov, L. P. The structures and properties of complexes of tetranitromethane and pyromellitic dianhydride with hydrocarbons of the benzene-tetracene series. II. Spectroscopic and thermodynamic characteristics, Zh. Fiz. Khim. **48**, 2190 (1974) [Engl. transl.: Russ. J. Phys. Chem. **48**, 1298 (1974)].
- [4175] Radziemski, L. J., Jr., and Kaufman, V. Wavelengths, energy levels, and analysis of the second spectrum of chlorine (Cl II), J. Opt. Soc. Am. **64**, 366 (1974).
- [4176] Sasanuma, M.; Morioka, Y.; Ishiguro, E., and Nakamura, M. Rydberg series in the NO molecule near 600 Å, J. Chem. Phys. **60**, 327 (1974).
- [4177] Warneck, P. Heat of formation of the HCO radical, Z. Naturforsch. **29a**, 350 (1974).
- [4178] Grönneberg, T., Hurum, T., and Undheim, K. Ionisation potentials in structure analysis of gaseous hydroxypyridine 1-oxides, Acta Chem. Scand., Ser. B **28**, 986 (1974).
- [4179] Cradock, S., Findlay, R. H., and Palmer, M. H. Bonding in methyl-and silyl-cyclopentadiene compounds: a study by photoelectron spectroscopy and *ab initio* molecular-orbital calculations, J. Chem. Soc., Dalton Trans., 1650 (1974).
- [4180] Bieri, G., Heilbronner, E., Kloster-Jensen, E., Schmelzer, A., and Wirz, J. Electronic states of 1,5-cyclooctadiyne radical cation and of related systems: the electronic structure of *cis*-bent carbon-carbon triple bonds, Helv. Chim. Acta **57**, 1265 (1974).
- [4181] Starzewski, K. A. O., Dieck, H. t., and Bock, H. Photoelectron spectra and molecular properties. XXIII. Photoelectron spectra of silicon-substituted ylidic systems, J. Organometal. Chem. **65**, 311 (1974).
- [4182] Brown, C. M., Tilford, S. G., and Ginter, M. L. Absorption spectrum of B I in the 1350–1900-Å region, J. Opt. Soc. Am. **64**, 877 (1974).
- [4183] Colin, R., DeGreef, D., Goethals, P., and Verhaegen, G. The ionization potential of the BeH molecule, Chem. Phys. Letters **25**, 70 (1974).
- [4184] Dorko, E. A., Scheps, R., and Rice, S. A. Comments on the ultraviolet spectrum and photophysical properties of trimethylenecyclopropane, J. Phys. Chem. **78**, 568 (1974).
- [4185] Cowley, A. H., Dewar, M. J. S., Goodman, D. W., and Padolina, M. C. Detection of rotational isomerism in diphosphines and diarsines by photoelectron spectroscopy, J. Am. Chem. Soc. **96**, 2648 (1974).
- [4186] Golob, L., Jonathan, N., Morris, A., Okuda, M., Ross, K. J., and Smith, D. J. Vacuum ultraviolet photoelectron spectroscopy of transient species: the $SO(^3\Sigma^-)$ radical, Ber. Bunsenges. **78**, 201 (1974).
- [4187] Bassett, P. J., Higginson, B. R., Lloyd, D. R., Lynaugh, N., and Roberts, P. J. Helium-I photoelectron spectra of tetrakis(trifluorophosphine)-nickel(O), -palladium(O), and -platinum(O), J. Chem. Soc., Dalton. Trans., 2316 (1974).
- [4188] Colton, R. J., and Rabalais, J. W. Photoelectron and electronic absorption spectra of SCl_2 , S_2Cl_2 , S_2Br_2 and $(CH_3)_2S_2$, J. Electron Spectrosc. Relat. Phenom. **3**, 345 (1974).
- [4189] Batich, C., Heilbronner, E., Rommel, E., Semmelhack, M. F., and Foos, J. S. Equivalence of the energy gaps $\Delta I(1,2)$ and $\Delta E(1,2)$ between corresponding bands in the photoelectron (I) and electronic absorption (E) spectra of spiro[4.4] nonatetraene. An amusing consequence of spiroconjugation, J. Am. Chem. Soc. **96**, 7662 (1974).
- [4191] Cowley, A. H., Dewar, M. J. S., Goodman, D. W., and Padolina, M. C. A Photoelectron spectroscopic study of polyphosphines. The question of $p\pi:d\pi$ bonding, J. Am. Chem. Soc. **96**, 3666 (1974).

- [4192] Akopyan, M. E., and Loginov, Yu. V. Photoelectron spectra of trimethylamine derivatives, Opt. Spektrosk. **37**, 442 (1974) [Engl. transl.: Opt. Spectrosc. **37**, 250 (1974)].
- [4193] Dromey, R. G., and Peel, J. B. Photoelectron spectroscopic correlation of the molecular orbitals of the alkanes and alkyl iodides, J. Mol. Struct. **23**, 53 (1974).
- [4194] Boschi, R. A. A., and Salahub, D. R. The high resolution photoelectron spectra of some iodoalkanes, iodoacycloalkanes, iodoalkenes, and fluoroiodohydrocarbons, Can. J. Chem. **52**, 1217 (1974).
- [4195] Schweig, A., Vermeer, H., and Weidner, U. A photoelectron spectroscopic study of keto-enol tautomerism in acetylacetones - a new application of photoelectron spectroscopy, Chem. Phys. Letters **26**, 229 (1974).
- [4196] Boschi, R., Clar, E., and Schmidt, W. Photoelectron spectra of polynuclear aromatics. III. The effect of nonplanarity in sterically overcrowded aromatic hydrocarbons, J. Chem. Phys. **60**, 4406 (1974).
- [4197] Suzuki, I. H., and Maeda, K. Ionization efficiency curves and fragmentations of deuterated ethylenes by electron impact, Intern. J. Mass Spectrom. Ion Phys. **15**, 281 (1974).
- [4198] Distefano, G., Ricci, A., Danieli, R., Foffani, A., Innorta, G., and Torroni, S. Mass Spectrometric evidence for $p_{\pi}-d_{\pi}$ bonding between sulfur and the Group IVB elements, J. Organometal. Chem. **65**, 205 (1974).
- [4199] Loudon, A. G., and Mazengo, R. Z. Steric strain and electron-impact. The behaviour of some n,n' -dimethyl-1,1-binaphthyls, some n,n' -dimethylbiphenyls and model compounds, Org. Mass Spectrom. **8**, 179 (1974).
- [4200] Muenow, D. W. Mass spectrometric determination of the heats of formation and atomization energies of the molecules Ge_2N and GeSiN , J. Chem. Phys. **60**, 3382 (1974).
- [4201] Tsuji, K., Saito, M., and Tani, T. Ionization potentials of phenylenediamines and steric effect in the ortho isomer, Denki Kagaku Oyobi Kogyo Butsuri Kagaku **41**, 688 (1973).
- [4202] Owzarski, T. P., and Franzen, H. F. High temperature mass spectrometry, vaporization, and thermodynamics of vanadium monosulfide, J. Chem. Phys. **60**, 1113 (1974).
- [4203] Holmes, J. L. The mass spectra of isomeric hydrocarbons - II: The C_5H_8 isomers, spiropentane, cyclopentene, 1,3-pentadiene and isoprene; the mechanisms and energetics of their fragmentations, Org. Mass Spectrom. **8**, 247 (1974).
- [4204] Clark, H. C., and Rake, A. T. Mass spectrometry of systems containing a Group IVB - transition metal bond. I. The phenyl-and pentafluorophenyl-silicon, -germanium and -tin derivatives of pentacarbonylmanganese, J. Organometal. Chem. **82**, 159 (1974).
- [4205] Guido, M., and Gigli, G. Mass spectrometric study of the gaseous gallium monocyanide, J. Chem. Phys. **60**, 721 (1974).
- [4206] Cocke, D. L., and Gingerich, K. A. Thermodynamic investigation of the gaseous molecules TiRh , Rh_2 , and Ti,Rh by mass spectrometry, J. Chem. Phys. **60**, 1958 (1974).
- [4207] Kohl, F. J., and Stearns, C. A. Identification and dissociation energy of gaseous hafnium mononitride, J. Phys. Chem. **78**, 273 (1974).
- [4208] Hildenbrand, D. L., and Murad, E. Mass spectrometric studies of gaseous ThO and ThO_2 , J. Chem. Phys. **61**, 1232 (1974).
- [4209] Gingerich, K. A. Mass spectrometric evidence for the very high stability of diatomic cerium compounds with some platinum metals and predicted dissociation energies of selected diatomic intermetallic compounds with multiple bonds, J. Chem. Soc. Faraday Trans. II **70**, 471 (1974).
- [4210] Martin, W. C., Hagan, L., Reader, J., and Sugar, J. Ground levels and ionization potentials for lanthanide and actinide atoms and ions, J. Phys. Chem. Ref. Data **3**, 771 (1974).
- [4211] Eaton, D. F., and Traylor, T. G. Distortional stabilization in phenyl participations, J. Am. Chem. Soc. **96**, 7109 (1974).
- [4212] Kroto, H. W., Landsberg, B. M., Suffolk, R. J., and Vodden, A. The photoelectron and microwave spectra of the unstable species thioacetaldehyde, CH_3CHS , and thioacetone, $(\text{CH}_3)_2\text{CS}$, Chem. Phys. Letters **29**, 265 (1974).
- [4213] Taylor, L. T., and Dillard, J. G. Mass spectrometric study of polydentate Schiff base coordination complexes. II. Cobalt(II), nickel(II), and copper(II) complexes of N,N' -bis(salicylidene)heptanediamine, N,N' -bis(salicylidene)-3,3'-bis(aminopropyl)amine, N,N' -bis(salicylidene)-3,3'-bis(aminopropyl) ether, and N,N' -bis(salicylidene)-3,3'-bis(aminopropyl) sulfide, Inorg. Chem. **13**, 2620 (1974).
- [4214] Nelsen, S. F., and Buschek, J. M. Charge delocalization in saturated systems. The radical cation of 1,3,6,8-tetraazatricyclo[4.4.1.1^{3,8}]dodecane, J. Am. Chem. Soc. **96**, 6424 (1974).
- [4215] Dehmer, J. L., Berkowitz, J., Cusachs, L. C., and Aldrich, H. S. Photoelectron spectroscopy of high temperature vapors. V. HeI spectra of GaX_3 and InX_3 ($X = \text{Cl}, \text{Br}, \text{I}$), J. Chem. Phys. **61**, 594 (1974).
- [4216] Kaving, B., and Lindgren, B. Ultraviolet absorption spectrum of the CaH molecule. II. The structure of the d-complex around 2850 Å, Physica Scripta **10**, 81 (1974).
- [4217] Worrell, C., Verhoeven, J. W., and Speckamp, W. N. Through-bond interaction in 1-aza-adamantane derivatives, Tetrahedron **30**, 3525 (1974).
- [4218] Baker, A. D., Brisk, M., and Gellender, M. Photoelectron spectra and dihedral angles of disulfides, J. Electron Spectrosc. Relat. Phenom. **3**, 227 (1974).
- [4219] Yokoyama, Y., Jinno, M., Watanabe, I., and Ikeda, S. Identification of accidentally degenerate bands in UV photoelectron spectra of ethylene carbonate and propylene carbonate, J. Electron Spectrosc. Relat. Phenom. **5**, 1095 (1974).
- [4220] Chadwick, D., and Katrib, A. Photoelectron spectra of acetaldehyde and acetyl halides, J. Electron Spectrosc. Relat. Phenom. **3**, 39 (1974).
- [4221] Debies, T. P., and Rabalais, J. W. Electronic structure of amino acids and ureas, J. Electron Spectrosc. Relat. Phenom. **3**, 315 (1974).
- [4222] Weiner, M. A., and Lattman, M. Photoelectron spectra of 4-substituted pyridine N-oxides, Tetrahedron Letters, 1709 (1974).
- [4223] McLafferty, F. W., and Winkler, J. Gaseous tropylum, benzyl, tolyl, and norbornadienyl cations, J. Am. Chem. Soc. **96**, 5182 (1974).
- [4224] Tam, W.-C., Yee, D., and Brion, C. E. Photoelectron spectra of some aldehydes and ketones, J. Electron Spectrosc. Relat. Phenom. **4**, 77 (1974).
- [4225] White, R. M., Carlson, T. A., and Spears, D. P. Angular distribution of the photoelectron spectra for ethylene, propylene, butene and butadiene, J. Electron Spectrosc. Relat. Phenom. **3**, 59 (1974).
- [4226] Elbel, S., Bergmann, H., Enßlin, W. Photoelectron spectra of the trimethyl compounds of the Group V elements, J. Chem. Soc. Faraday Trans. II **70**, 555 (1974).
- [4228] Terlouw, J. K., Heerma, W., Frintrop, P. C. M., Dijkstra, G., and Meinema, H. A. Electron-impact induced fragmentation of some heterocyclic-tin compounds, J. Organometal. Chem. **64**, 205 (1974).
- [4229] Barrow, R. F., Kopp, I., and Malmberg, C. The electronic spectrum of gaseous AlF , Physica Scripta, **10**, 86 (1974).
- [4230] Dyke, J. M., Golob, L., Jonathan, N., Morris, A., Okuda, M., and Smith, D. J. Vacuum ultraviolet photoelectron spectroscopy of transient species. Part 3. The $\text{SO}^{(\Sigma)}$

- radical, J. Chem. Soc. Faraday Trans. II **70**, 1818 (1974).
- [4231] Koenig, T., Tuttle, M., and Wieleske, R. A. The He(I) photoelectron spectra of xylenes and metacyclophanes. A reassignment of the lowest ionic state of [2.2] metacyclophane, Tetrahedron Letters, 2537 (1974).
- [4232] Berkowitz, J. PES of high-temperature vapors. VI. 304 Å and 584 Å PES of Zn, Cd, and Hg halides and the electronegativity scale, J. Chem. Phys. **61**, 407 (1974).
- [4233] Kelder, J., Cefontain, H., Higginson, B. R., and Lloyd, D. R. Photoelectron and ultraviolet absorption spectra of cyclopropyl conjugated 1,2-diketones, Tetrahedron Letters, 739 (1974).
- [4234] Evans, S., Guest, M. F., Hillier, I. H., and Orchard, A. F. Theoretical and experimental study of the low energy ionic states of π -cyclopentadienyl-nickel nitrosyl, J. Chem. Soc., Faraday Trans. II **70**, 417 (1974).
- [4235] Gavin, R. M., Jr., and Rice, S. A. Spectroscopic properties of polyenes. II. The vacuum ultraviolet spectra of *cis*- and *trans*-1,3,5-hexatriene, J. Chem. Phys. **60**, 3231 (1974).
- [4236] Wagner, L. C., Robert, P., Grindstaff, Q., and Grimley, R. T. A mass spectrometric study of the fragmentation of the cuprous chloride vapor system, Intern. J. Mass Spectrom. Ion Phys. **15**, 255 (1974).
- [4237] Eidelberg, M. The spectrum and term system of helium-like boron, B IV, J. Phys. B **7**, 1476 (1974).
- [4238] McDiarmid, R. Assignments of Rydberg and valence transitions in the electronic absorption spectrum of dimethyl sulfide, J. Chem. Phys. **61**, 274 (1974).
- [4239] Dyke, J. M., Golob, L., Jonathan, N., Morris, A., and Okuda, M. Vacuum ultraviolet photoelectron spectroscopy of transient species. Part 4. Difluoromethylene and ozone, J. Chem. Soc., Faraday Trans. II **70**, 1828 (1974).
- [4240] Kobayashi, T., and Nagakura, S. Photoelectron spectra of aminopyridines and cyanopyridines, J. Electron Spectrosc. Relat. Phenom. **4**, 207 (1974).
- [4241] Brown, R. S., Eaton, D. F., Hosomi, A., Traylor, T. G., and Wright, J. M. Photoelectron spectra of cyclopropylcarbinyltrimethyltin and allyltrimethyltin. A comparison of σ - σ and σ - π conjugation, J. Organometal. Chem. **66**, 249 (1974).
- [4242] Lappert, M. F., Pedley, J. B., and Sharp, G. Bonding studies of transition metal complexes. I. He(I) photoelectron spectra of d^0 silylmethyl and neopentyl derivatives of Group IVA metals, J. Organometal. Chem. **66**, 271 (1974).
- [4243] Fuss, W., and Bock, H. Photoelectron spectra and molecular properties. XXXVI. $(\text{H}_3\text{C})_2\text{B}$, $(\text{H}_3\text{C})_2\text{BF}$, $(\text{H}_3\text{C})_2\text{BN}(\text{CH}_3)_2$, and $(\text{H}_3\text{C})_2\text{CC}(\text{CH}_3)_2$: The use of ionization potentials in assigning UV spectra, J. Chem. Phys. **61**, 1613 (1974).
- [4244] Rosmus, P., Dacre, P. D., Solouki, B., and Bock, H. Hartree-Fock calculations and photoelectron spectra of SSO and NSF, Theoret. Chim. Acta **35**, 129 (1974).
- [4245] Maleev, A. N., Semenov, G. A., and Kholodov, A. I. Evaporator with gas flooding for high-temperature thermodynamic investigations on a mass spectrometer, Zavod. Lab. **40**, 987 (1974) [Engl. transl.: Ind. Lab. (USSR) **40**, 1179 (1974)].
- [4246] Planckaert, A. A., Doucet, J., and Sandorfy, C. Comparative study of the vacuum ultraviolet absorption and photoelectron spectra of some simple ethers and thioethers, J. Chem. Phys. **60**, 4846 (1974).
- [4247] Eland, J. H. D. Predissociation of triatomic ions studied by photoelectron-photoion coincidence spectroscopy, Advan. Mass Spectrom. **6**, 917 (1974).
- [4248] Lee, T. H., and Rabalais, J. W. Vibrational transition probabilities in photoelectron spectra, J. Chem. Phys. **61**, 2747 (1974).
- [4249] Asmus, P., and Klessinger, M. Photoelectron spectra of organic compounds. VI. Exocyclic methylene compounds, Tetrahedron **30**, 2477 (1974).
- [4250] Aarons, L. J., Connor, J. A., Hillier, I. H., Schwarz, M., and Lloyd, D. R. Electronic structure of diazocyclopentadiene. A study using low and high energy photoelectron spectroscopy and *ab initio* molecular orbital calculations, J. Chem. Soc., Faraday Trans. II **70**, 1106 (1974).
- [4251] Batisch, C., and Adam, W. The photoelectron spectra of alkylperoxides, Tetrahedron Letters, 1467 (1974).
- [4252] Higginson, B. R., Lloyd, D. R., Connor, J. A., and Hillier, I. H. Photoelectron studies of metal carbonyls. Part 4. Mono-substituted complexes of chromium and tungsten carbonyls, J. Chem. Soc., Faraday Trans. II **70**, 1418 (1974).
- [4253] Guimon, C., Pfister-Guillouzo, G., Arbelot, M., and Chanon, M. Electronic structure of sulphur compounds. VII. Photoelectron spectra of thiocarbonyl heterocycles, Tetrahedron **30**, 3831 (1974).
- [4254] Connors, R. E., Roebber, J. L., and Weiss, K. Vacuum ultraviolet spectroscopy of cyanogen and cyanoacetylenes, J. Chem. Phys. **60**, 5011 (1974).
- [4255] Koenig, T., and Longmaid, H. Photoelectron spectra of 1,4-dihydropyridine and *N*-methyl-1,4-dihydropyridine, J. Org. Chem. **39**, 560 (1974).
- [4256] Lappert, M. F., Pedley, J. B., Sharp, G. J., and Westwood, N. P. C. Bonding studies of compounds of boron and the Group III and IV elements. XII. Variable temperature He I photoelectron spectra of Group III halides, $2\text{MX}_3 \rightleftharpoons \text{M}_2\text{X}_4$ ($\text{M} = \text{Al}$ or Ga , $\text{X} = \text{Cl}$, Br , or I), J. Electron Spectrosc. Relat. Phenom. **3**, 237 (1974).
- [4258] Bodor, N., Chen, B. H., and Worley, S. D. Photoelectron spectra and SCF MO calculations for the dimers of cyclobutadiene, J. Electron Spectrosc. Relat. Phenom. **4**, 65 (1974).
- [4259] Bischof, P., Gleiter, R., Kukla, M. J., and Paquette, L. A. The photoelectron spectra of tricyclo[3.3.0.0^{2,6}]octene and tricyclo[3.3.0.0^{2,6}]octane, J. Electron Spectrosc. Relat. Phenom. **4**, 177 (1974).
- [4260] Worrell, C. W. The photoelectron and absorption spectra of allyl halides, J. Electron Spectrosc. Relat. Phenom. **3**, 359 (1974).
- [4261] Cowley, A. H., Dewar, M. J. S., Gilje, J. W., Goodman, D. W., and Schweiger, J. R. Relationship between the photoelectron spectra and torsional barriers of aminophosphines, J. Chem. Soc. Chem. Commun., 340 (1974).
- [4262] Schäfer, W., Schweig, A., Bickelhaupt, F., and Vermeer, H. The electronic structure of a λ_3 -phosphaphhenanthrene; a direct indication of the similarity of electronic effects produced by sp^2 hybridized carbon and phosphorus atoms, Recl. Trav. Chim. Pays-Bas **93**, 1 (1974).
- [4263] Batisch, C., Heilbronner, E., and Vogel, E. The ionization energies of bridged [14]annulenes and of dicyclohepta[cd, gh]pentalene, Helv. Chim. Acta **57**, 2288 (1974).
- [4264] Van Deurzen, C. H. H., Conway, J. G., and Davis, S. P. Spectrum and energy levels of quadruple-ionized vanadium (V v), J. Opt. Soc. Am. **64**, 498 (1974).
- [4265] Roberts, P. Ph.D. Thesis (Univ. Birmingham, 1974).
- [4266] Akopyan, M. E., and Timoshenko, M. M. A photoelectron spectrometer for investigating vapors of nonvolatile substances, Prib. Tekh. Eksp., 164 (1974) [Engl. transl.: Instrum. Exp. Tech., 185 (1974)].
- [4267] Clary, D. C., Lewis, A. A., Morland, D., Murrell, J. N., and Heilbronner, E. Ionization potentials of cycloalkenes, J. Chem. Soc., Faraday Trans. II **70**, 1889 (1974).
- [4268] Bischof, P., Gleiter, R., de Meijere, A., and Meyer, L.-U. The conjugative interaction between π -orbitals and cyclobutane-orbitals in spiro[3.4]octa-5,7-diene and spiro[3.4]octene-5, Helv. Chim. Acta **57**, 1519 (1974).
- [4269] Almemark, M., Backvall, J. E., Moberg, C., Åkermark, B., Åsbrink, L., and Roos, B. *Ab initio* calculations and

- assignment of photoelectron spectra of maleic and succinic anhydride, *Tetrahedron* **30**, 2503 (1974).
- [4270] Harshbarger, W. R., Kuebler, N. A., and Robin, M. B. Electronic structure and spectra of small rings. V. Photoelectron and electron impact spectra of cyclopropenone, *J. Chem. Phys.* **60**, 345 (1974).
- [4271] Gilbert, R., and Sandorfy, C. Rydberg transitions in the ultraviolet spectra of difluorobenzenes, *Chem. Phys. Letters* **27**, 457 (1974).
- [4272] Aloisi, G. G., Santini, S., and Sorriso, S. Molecular complexes of substituted diphenyl sulphides with π acceptors. Charge transfer spectra and ionization potentials of the donors, *J. Chem. Soc., Faraday Trans. I* **70**, 1908 (1974).
- [4274] Bock, H., and Seidl, H. "d-Orbital effects" in silicon-substituted π -electron systems. XI. Syntheses and properties of the isomeric bis(trimethylsilyl)-1,3-butadienes, *J. Am. Chem. Soc.* **90**, 5694 (1968).
- [4275] Maier, J. P., and Muller, J.-F. Ionisation energies of pyridine N-oxides determined by photoelectron spectroscopy, *J. Chem. Soc., Faraday Trans. II* **70**, 1991 (1974).
- [4276] Wagner, G., and Bock, H. Photoelektronenspektren und Moleküleigenschaften, XXVI. Die Delokalisation von Schwefel-Elektronenpaaren in Alkylsulfiden und -disulfiden, *Chem. Ber.* **107**, 68 (1974).
- [4277] Rademacher, P. Photoelectron spectra and conformation of cyclic N,N'-dimethylhydrazines, *Tetrahedron Letters*, 83 (1974).
- [4278] Bodor, N., Kaminski, J. J., Worley, S. D., Colton, R. J., Lee, T. H., and Rabalais, J. W. Photoelectron spectra, hydrolytic stability, and antimicrobial activity of N-chlorinated piperidines, *J. Pharm. Sci.* **63**, 1387 (1974).
- [4279] Vilesov, F. I., Lopatin, S. N., Vovna, V. I., Paetzold, R., and Niendorf, K. Untersuchungen an Molekülkomplexen. 3. Mitteilung: Zusammenhang zwischen der Ionisierungsenergie von $R_1R_2R_3PS$ -Molekülen und der Stabilität ihrer Molekülkomplexe mit Jod, *Z. Phys. Chem. (Leipzig)* **255**, 661 (1974).
- [4280] Bischof, P. K., Dewar, M. J. S., Goodman, D. W., and Jones, T. B. Photoelectron spectra of molecules. VI. Hyperconjugation versus $p_{\pi}-d_{\pi}$ bonding in group IVb compounds, *J. Organometal. Chem.* **82**, 89 (1974).
- [4281] Martin, H.-D., Kagabu, S., and Schwesinger, R. Photoelektronenspektroskopische Untersuchung der intramolekularen π, π -Wechselwirkungen in endo-Tricyclo[4.2.1.0^{5,6}]nonadienen. Homokonjugation mit Cyclobutanen, *Chem. Ber.* **107**, 3130 (1974).
- [4284] Harada, Y., Ohno, K., Seki, K., and Inokuchi, H. Photoelectron spectrum of dispiro[2.2.2.2]deca-4,9-diene. Conjugation of Walsh orbitals of cyclopropane rings with orbitals of diene, *Chem. Letters*, 1081 (1974).
- [4285] Henrich, G., Gunkel, E., and Klessinger, M. Photoelektronenspektren organischer Verbindungen. 4. Photoelektronenspektren ungesättigter Carbonylverbindungen, *J. Mol. Struct.* **21**, 231 (1974).
- [4286] Bieri, G., and Heilbronner, E. Der Einfluss von Substituenten in Stellung 4 auf das n-Ionisations- potential des Chinuclidins, *Helv. Chim. Acta* **57**, 546 (1974).
- [4287] Hartmann, O.-R., Lebert, K.-H., and Chun, H.-U. Elektronenstoss- und Ionenstossuntersuchungen an Schwefeldichlorid und Dischwefeldichlorid, *Z. Phys. Chem. (Frankfurt am Main)* **92**, 311 (1974).
- [4288] Samson, J. A. R., and Petrosky, V. E. Continuum ionization transition probabilities of atomic oxygen, *Phys. Rev. A* **9**, 2449 (1974).
- [4289] Salisbury, K. Quenching of the fluorescence of styrenes by ground state oxygen, *J. Photochem.* **2**, 401 (1973/1974).
- [4290] Schmidt, H., and Schweig, A. Notiz zur transanularen n- π -Wechselwirkung in 2,5-Dihydrofuran, *Chem. Ber.* **107**, 725 (1974).
- [4291] Bock, H., Wagner, G., Wittel, K., Sauer, J., and Seebach, D. Photoelektronenspektren und Moleküleigenschaften. XXXII. n/ π -Konjugation in heterosubstituierten Äthylenen, *Chem. Ber.* **107**, 1869 (1974).
- [4292] Höhne, G., Marschner, F., and Praefcke, K. Organische Schwefelverbindungen. X. Photoelektronen- und UV-spektroskopische Untersuchung eines nichtkonjugierten 1,3-Oxathiol-Spiroketons, *Z. Naturforsch.* **29b**, 546 (1974).
- [4293] Schäfer, W., Schweig, A., Maier, G., and Sayrac, T. Inductive effect of a carbonyl group. The electronic structure of [5]annulenones, *J. Am. Chem. Soc.* **96**, 279 (1974).
- [4294] Stafast, H., and Bock, H. Photoelektronenspektren und Moleküleigenschaften, XXXIV. Cyanamid, *Chem. Ber.* **107**, 1882 (1974).
- [4295] Bock, H., and Solouki, B. Photoelektronenspektren und Moleküleigenschaften, XXXV. Sulfoxide X_2SO – Beispiele für den Nutzen von Korrelations – diagrammen bei der Diskussion von Substituenteneffekten und von geometrischen Störungen, *Chem. Ber.* **107**, 2299 (1974).
- [4296] Bougeard, D., Schrader, B., Bleckmann, P., and Plessner, T. Ramanpekroskopie und Molekülstruktur. VII. Infrarot-, Raman- und Photoelektronenspektrum des Hexamethyl-Dewarbenzols; Normalkoordinatenberechnung und Untersuchung der π -Elektronen-Wechselwirkung, *Justus Liebigs Ann. Chem.* **156**, 137 (1974).
- [4297] Marschner, F., Juds, H., and Goetz, H. Zur Homokonjugation im Hexamethyl-Dewar-Benzol, *Tetrahedron Letters*, 3983 (1973).
- [4298] Kroner, J., Nöth, H., and Niedenzu, K. Photoelektronenspektroskopische Untersuchungen an Bor-Verbindungen. III. π -Elektronendelokalisation in 1,3,2-Diazaborolinen, *J. Organometal. Chem.* **71**, 165 (1974).
- [4299] Kroner, J., Nölle, D., Nöth, H., and Winterstein, W. Photoelektronenspektroskopische Untersuchungen an Bor-Verbindungen. IV. Ionisierungsenergien und Geometrie von Tetraazadiborinen, *Z. Naturforsch.* **29b**, 476 (1974).
- [4300] Köppel, C., Schwarz, H., and Bohlmann, F. Energetische Betrachtungen zur intramolekularen Solvatisierung von Ionen bei der elektronenstossinduzierten Fragmentierung von α,ω -Bis-(Trimethylsilyl)-äthern, *Org. Mass Spectrom.* **9**, 567 (1974).
- [4301] Martin, H.-D., and Schwesinger, R. Notiz über eine beobachtete through-bond-Wechselwirkung über vier σ -Bindungen, *Chem. Ber.* **107**, 3143 (1974).
- [4302] Stefanović, D., and Grützmacher, H. F. The ionisation potential of some substituted pyridines, *Org. Mass Spectrom.* **9**, 1052 (1974).
- [4303] Wittel, K., and Bock, H. Photoelektronenspektren und Moleküleigenschaften. XXVII. Chlor- und Bromäthylene – Beispiele für Änderungen in π - und σ -Systemen, *Chem. Ber.* **107**, 317 (1974).
- [4304] Lauer, G., Müller, C., Schulte, K.-W., Schweig, A., and Krebs, A. Struktur und Spin-Multiplicität eines [4]Annulen-Systems, *Angew. Chem.* **86**, 597 (1974).
- [4305] Dean, C. R. S., Finch, A., Gardner, P. J., and Payling, D. W. Appearance and ionization potentials of ions produced by electron-impact on some phosphorus-fluorine compounds: the phosphorus-phosphorus bond dissociation energy in diphosphorus tetrafluoride, *J. Chem. Soc., Faraday Trans. I* **70**, 1921 (1974).
- [4306] Knowles, D. J., and Nicholson, A. J. C. Ionization energies of formic and acetic acid monomers, *J. Chem. Phys.* **60**, 1180 (1974).
- [4307] Goodman, T. D., Allen, J. D., Jr., Cusachs, L. C., and Schweitzer, G. K. The photoelectron spectra of gaseous alkali halides, *J. Electron Spectrosc. Relat. Phenom.* **3**, 289 (1974).

- [4308] Werner, A. S., Tsai, B. P., and Baer, T. Photoionization study of the ionization potentials and fragmentation paths of the chlorinated methanes and carbon tetrabromide, *J. Chem. Phys.* **60**, 3650 (1974).
- [4309] Cradock, S., and Duncan, W. The photoelectron spectrum of CSe_2 , *Mol. Phys.* **27**, 837 (1974).
- [4310] Wittel, K., Bock, H., and Manne, R. Photoelectron spectra of iodo ethylenes. A simple method to incorporate spin orbit coupling in molecular orbital models, *Tetrahedron* **30**, 651 (1974).
- [4311] Paulus, J.-M., and Abbé, J.-C. Potentiel d'apparition de I_2^{2+} à partir de I_2 , *J. Chim. Phys.*, 690 (1973).
- [4312] Cantú, A. M., Jannitti, E., and Tondello, G. Spectra of N VI, O VII, F VII, and F VIII in grazing- incidence region, *J. Opt. Soc. Am.* **64**, 699 (1974).
- [4313] Bernauer, O., and Weil, K. G. Massenspektrometrische Untersuchungen an Silberhalogeniden. III: Silberbromid und Silberjodid, *Ber. Bunsenges.* **72**, 1339 (1974).
- [4316] Thorstad, O., and Undheim, K. Mass spectrometry of onium compounds. XXIV. Ionisation potential in structure analysis of pyridodiazooxides, *Chem. Scr.* **6**, 222 (1974).
- [4317] Thorstad, O., Undheim, K., and Hvistendahl, G. Mass spectrometry of onium compounds. XXIII. Ionisation potentials in the structural assignment of $[\text{M}-\text{N}_2]$ ions from diazo-oxides, *Org. Mass Spectrom.* **9**, 548 (1974).
- [4318] Locht, R., and Schopman, J. The dissociative ionization in oxygen, *Intern. J. Mass Spectrom. Ion Phys.* **15**, 361 (1974).
- [4319] Puttemans, J. P. Ionisation de cycloalcanes (C_5 à C_{12}) en spectroscopie photoélectronique et par impact électronique, *Ing. Chim. (Brussels)* **56**, 64 (1974).
- [4320] Gordon, S. M., Krige, G. J., and Reid, N. W. Isotope effects in the unimolecular decomposition of ethylene by low-energy electron impact, *Intern. J. Mass Spectrom. Ion Phys.* **14**, 109 (1974).
- [4321] Sauvageau, P., Doucet, J., Gilbert, R., and Sandorfy, C. Vacuum ultraviolet and photoelectron spectra of fluoroethanes, *J. Chem. Phys.* **61**, 391 (1974).
- [4322] Fehlner, T. P., and Turner, D. W. The photoelectron spectrum of SiF_2 , *Inorg. Chem.* **13**, 754 (1974).
- [4323] Guimon, C., Gonbeau, D., Pfister-Guillouzo, G., Åsbrink, L., and Sandström, J. Electronic structure of sulphur compounds. VI. Photoelectron spectra of some simple thiocarbonyl compounds, *J. Electron Spectrosc. Relat. Phenom.* **4**, 49 (1974).
- [4324] Müller, C., Schweig, A., and Mock, W. L. Through-conjugation through the sulfone group in 2,5-di-*tert*-butylthiophene 1,1-dioxide, *J. Am. Chem. Soc.* **96**, 280 (1974).
- [4325] Potapov, V. K., Rodionov, A. N., Evlasheva, T. I., and Rogozhin, K. L. Photoionization of triphenyl derivatives of elements in group VB of the periodic table, *Khim. Vys. Energ.* **8**, 559 (1974) [Engl. Transl.: High Energy Chem. (USSR) **8**, 486 (1974)].
- [4326] Müller, C., Schweig, A., Anastassiou, A. G., and Wetzel, J. C. Hetero homoconjugation versus heterobicycloconjugation in 9-thiabicyclo[4.2.1]nona- 2,4,7,-triene, *Tetrahedron* **30**, 4089 (1974).
- [4327] Dewar, P. S., Ernstbrunner, E., Gilmore, J. R., Godfrey, M., and Mellor, J. M. Conformational analysis of alkyl aryl ethers and alkyl aryl sulphides by photoelectron spectroscopy, *Tetrahedron* **30**, 2455 (1974).
- [4328] Pykhtina, E. V., Cherednichenko, L. V., Kardash, I. E., Evlasheva, T. I., Sorokin, V. V., Potapov, V. K., and Pravednikov, A. N. Ionization potentials of amines and energies of charge transfer bands in the absorption spectra of complexes of 7,7,8,8-tetracyanoquinodimethan, *Khim. Vys. Energ.* **8**, 307 (1974) [Engl. transl.: High Energy Chem. (USSR) **8**, 257 (1974)].
- [4329] Harada, Y., Seki, K., Suzuki, A., and Inokuchi, H. Photoelectron spectrum of vinylcyclopropane, *Chem. Letters.*, 893 (1973).
- [4330] Suffolk, R. J. The photoelectron spectra of the perfluorodiazines, *J. Electron Spectrosc. Relat. Phenom.* **3**, 53 (1974).
- [4331] Tajima, S., and Tsuchiya, T. Energetics consideration of C_nH_n^+ ions produced from various precursors by electron impact, *Bull. Chem. Soc. Japan* **46**, 3291 (1973).
- [4332] Wagner, G., Bock, H., Budenz, R., and Seel, F. Photoelektronenspektren und Moleküleigenschaften. XIX. FSSF and SSF₂, *Chem. Ber.* **106**, 1285 (1973).
- [4333] Goetz, H., Marschner, F., and Juds, H. Zur $n-\pi$ Wechselwirkung im Benzalanilin, *Tetrahedron* **30**, 1133 (1974).
- [4334] Griebel, R., Hohlneicher, G., and Dörr, F. A photoelectro-spectroscopic study of benzonitrile, ethynylbenzene and some of its substituted derivatives, *J. Electron Spectrosc. Relat. Phenom.* **4**, 185 (1974).
- [4335] Heller, R., Varmuza, K., and Krenmayr, P. Massenspektrometrische Untersuchung des Substituenteneffektes bei einfach substituierten Benzophenonen, *Monatsh. Chem.* **105**, 787 (1974).
- [4336] Köppel, C., Schwarz, H., and Bohlmann, F. Elektronenstossinduzierte Fragmentierung von Acetylenverbindungen. VIII. Struktur der stabilen und instabilen Ionen $[\text{C}_n\text{H}_m]^+$ aus isomeren C_nH_{10} -Kohlenwasserstoffen, *Org. Mass Spectrom.* **8**, 25 (1974).
- [4337] Kuschel, H., and Grützmacher, H.-F. Zum Mechanismus massenspektrometrischer Fragmentierungsreaktionen. XIII. Einfluss der Dissoziationsenergie auf intramolekulare aromatische Substitutionen in den Molekül-Ionen von *N,N*-Dimethyl-*N'*-phenylformamidinen, *Org. Mass Spectrom.* **9**, 408 (1974).
- [4338] Martin, H.-D., Heller, C., and Werp, J. Bishomofulvenkonjugation. Photoelektronenspektren und Elektronenstruktur homologer Quadricyclane, *Chem. Ber.* **107**, 1393 (1974).
- [4339] Preiss, H. Die Massenspektren einiger Arsen- und Arsenigsäureester. Kinetische Untersuchung ihres Fragmentierungsverhaltens, *Z. Anorg. Allg. Chem.* **404**, 175 (1974).
- [4340] Marschner, F., and Goetz, H. Korrelation zwischen Photoelektronen- und Elektronen-Spektren. III. Eine Methode zur Deutung der PE- und UV-Spektren vom Toluol, *Tetrahedron* **30**, 3451 (1974).
- [4341] Marschner, F., and Goetz, H. Korrelation zwischen Photoelektronen- und Elektronen-Spektren. II. Untersuchung aromatischer π -Systeme mit modifizierten PPP-SCF-Cl-Parametern, *Tetrahedron* **30**, 3159 (1974).
- [4342] Foster, M. S., Williamson, A. D., and Beauchamp, J. L. Photoionization mass spectrometry of trans-azomethane, *Intern. J. Mass Spectrom. Ion Phys.* **15**, 429 (1974).
- [4343] Terwilliger, D. T., and Smith, A. L. Analysis of autoionizing Rydberg states in the vacuum ultraviolet absorption spectrum of HBr and DBr, *J. Mol. Spectrosc.* **50**, 30 (1974).
- [4344] Potts, A. W., Williams, T. A., and Price, W. C. Photoelectron spectra and electronic structure of diatomic alkali halides, *Proc. Roy. Soc. (London) A* **341**, 147 (1974).
- [4345] Bock, H., Wittel, K., and Haas, A. Photoelektronenspektren und Moleküleigenschaften. XLI. Vergleich von F_3CS^- , Cl- und F-Substituenten in Thiocarbonyl-Verbindungen, *Z. Anorg. Allg. Chem.* **408**, 107 (1974).
- [4346] Bohlmann, F., Herrmann, R., Mathar, W., and Schwarz, H. Massenspektrometrische Untersuchung von Amiden. VII. Intramolekulare Wechselwirkungen bifunktioneller Gruppen bei der elektronenstoßinduzierten Fragmentierung von Piperidinen und Piperideiden, *Chem. Ber.* **107**, 1081 (1974).
- [4347] Bruckmann, P., and Klessinger, M. Photoelektronenspektren

- organischer Verbindungen. V. Wechselwirkung kleiner Ringe mit π -Systemen, *Chem. Ber.* **107**, 1108 (1974).
- [4348] Chinone, A., and Ohta, M. Charge-transfer complex of sydnone, *Bull. Chem. Soc. Japan* **47**, 1032 (1974).
- [4349] Parr, G. R., and Taylor, J. W. Photoionization mass spectrometry. IV. Carbon dioxide, *Intern. J. Mass Spectrom. Ion Phys.* **14**, 467 (1974).
- [4350] Kräfig, R., Reinke, D., and Baumgärtel, H. Photoreaktionen kleiner organischer Moleküle II. Die Photoionenspektren der Isomeren Propylen-Cyclopropan und Acetaldehyd-Äthylenoxyd, *Ber. Bunsenges.* **78**, 425 (1974).
- [4351] Bergmark, T., Karlsson, L., Jadrny, R., Mattsson, L., Albridge, R. G., and Siegbahn, K. Isotopic effects in the electron spectra of $H_2^{16}O$, $H_2^{18}O$, and $D_2^{16}O$, *J. Electron Spectrosc. Relat. Phenom.* **4**, 85 (1974).
- [4352] Nygaard, K. J., and Hahn, Y. B. Ion fluorescene in cesium, *Physica* **75**, 333 (1974).
- [4353] Timoshenko, M. M., and Akopyan, M. E. Photoelectron spectra of cesium halides, *Khim. Vys. Energ.* **8**, 211 (1974) [Engl. transl.: High Energy Chem. (USSR) **8**, 175 (1974)].
- [4355] Dehmer, P. M., Berkowitz, J., and Chupka, W. A. Photoionization of atomic nitrogen, *J. Chem. Phys.* **60**, 2676 (1974).
- [4356] Coppens, P., Smets, J., Fishel, M. G., and Drowart, J. Mass spectrometric study of the photoionization of nitrous oxide in the wavelength interval 1000–600 Å, *Intern. J. Mass Spectrom. Ion Phys.* **14**, 57 (1974).
- [4357] Houk, K. N., George, J. K., and Duke, Jr., R. E. A frontier molecular orbital treatment of fulvene cycloadditions. Molecular orbital calculations and photoelectron spectra of substituted fulvenes, *Tetrahedron* **30**, 523 (1974).
- [4358] Krenmayr, P., Heller, R., and Varmuza, K. Massenspektrometrische Untersuchungen an Benzophenon und substituierten Benzophenonen. I. Ermittlung thermodynamischer Größen, *Organic Mass Spectrom.* **9**, 998 (1974).
- [4359] Kuschel, H., and Grützmacher, H.-F. Zum Mechanismus massenspektrometrischer Fragmentierungsreaktionen. XI. Einfluß von Substituenten auf die Bildung cyclischer Fragment-Ionen in den Massenspektren von *N,N*-Dimethyl-*N'*-2-chlorphenylformamidinen und 2-Chlorformaniliden, *Org. Mass Spectrom.* **9**, 395 (1974).
- [4360] Leduc, G., and Rousseau, Y. Spectrometrie de masse d'imines aliphatiques, *Can. J. Chem.* **52**, 1648 (1974).
- [4361] Schäfer, W., Schweig, A., Maier, G., Sayrac, T., and Crandall, J. K. Electronic structure of cyclopropenone, *Tetrahedron Letters*, 1213 (1974).
- [4362] Schmidt, H., Schweig, A., and Krebs, A. Splitting of the degenerate acetylenic π MOs; a probe for ring strain, *Tetrahedron Letters*, 1471 (1974).
- [4363] Schäfer, W., Schmidt, H., Schweig, A., Hoffmann, R. W., and Kurz, H. Evidence for strong nonbonded n/π interaction in bicyclo[4.2.1]nona-2,4,7-trien-9-one, *Tetrahedron Letters*, 1953 (1974).
- [4364] Streets, D. G., and Williams, T. A. Photoelectron spectroscopy of 9,10-dihaloanthracenes, *J. Electron Spectrosc. Relat. Phenom.* **3**, 71 (1974).
- [4365] Chau, F. T., and McDowell, C. A. Photoelectron spectra of fluorotribromomethane and fluorotrichloromethane, *J. Electron Spectrosc. Relat. Phenom.* **6**, 357 (1975).
- [4366] Doucet, J., Sauvageau, P., and Sandorfy, C. Photoelectron and far-ultraviolet absorption spectra of chlorofluoro derivatives of ethane, *J. Chem. Phys.* **62**, 355 (1975).
- [4367] Chau, F. T., and McDowell, C. A. Photoelectron spectra of 1,2 dichloro-, 1,2 dibromo- and 1,2 diiodo-ethane, *J. Electron Spectrosc. Relat. Phenom.* **6**, 365 (1975).
- [4368] Distefano, G., Pignataro, S., Szepes, L., and Borossay, J. Photoelectron spectroscopy study of the triphenyl derivatives of the group V elements, *J. Organometal. Chem.* **102**, 313 (1975).
- [4369] Buttrill, S. E. Jr., Williamson, A. D., and LeBreton, P. Photoionization measurement of the heat of formation of allyl cations, *J. Chem. Phys.* **62**, 1586 (1975).
- [4370] J. M., Golob, L., Jonathan, N., and Morris, A. Vacuum ultraviolet photoelectron spectroscopy of transient species, *J. Chem. Soc. Faraday Trans. II* **71**, 1026 (1975).
- [4371] Cowley, A. H., Dewar, M. J. S., and Goodman, D. W. Molecular photoelectron spectroscopic studies of some trifluoromethyl-substituted phosphines and chlorophosphines, *J. Am. Chem. Soc.* **97**, 3653 (1975).
- [4372] Guest, M. F., Higginson, B. R., Lloyd, D. R., and Hillier, I. H. Interpretation of the valence photoelectron spectra of $Mn(CO)_5H$, $Mn(CO)_5CH_3$, and $Fe(CO)_5H_2$, *J. Chem. Soc. Faraday Trans. II* **71**, 902 (1975).
- [4373] Cradock, S., Ebsworth, E. A. V., Moretto, H., and Rankin, D. W. H. Photoelectron spectra and fluxional behaviour in some σ -cyclopentadienes, *J. Chem. Soc. Dalton Trans.* 390, (1975).
- [4374] Brogli, F., Heilbronner, E., Wirz, J., Kloster-Jensen, E., Bergman, R. G., Vollhardt, K. P. C., and Ashe III, A. J. The consequences of σ and π conjugative interactions in mono-, di- and triacetylenes. A photoelectron spectroscopic investigation, *Helv. Chim. Acta* **58**, 2620 (1975).
- [4375] Condorelli, G., Fragalà, I., Centineo, A., and Tondello, E. The electronic structure and photoelectron spectra of dichlorodi- π -cyclopentadienyl-titanium(IV), -zirconium(IV) and-hafnium(IV), *J. of Organomet. Chem.* **87**, 311 (1975).
- [4376] Baerends, E. J., Oudshoorn, Ch., and Oskam, A. Photoelectron spectra and Xα calculations of iron pentacarbonyl and ethyleneiron tetracarbonyl, *J. Electron Spectrosc. Relat. Phenom.* **6**, 259 (1975).
- [4377] Distefano, G., Mazzucato, U., Modelli, A., Pignataro, S., and Orlandi, G. Photoelectron (He I) spectroscopic study of styrylpyridines, *J. Chem. Soc. Faraday Trans. II* **71**, 1583 (1975).
- [4378] Arnold, D. E. J., and Rankin, D. W. H. Preparation and properties of bis(difluorophosphino)- and tris(difluorophosphino)-amine, *J. Chem. Soc. Dalton Trans.* 889 (1975).
- [4379] Bergmann, H., and Bock, H. Photoelectron spectra and molecular properties, XLVI nitroso compounds – electron-rich molecules, *Z. Naturforsch.* **30b**, 629 (1975).
- [4380] Astrup, E. E., Bock, H., Wittel, K., and Heimbach, P. Photoelectron spectra and molecular properties. LIII. Methyl-substituent effects on the hexatriene π system, *Acta Chem. Scand. Ser. A* **29**, 827 (1975).
- [4381] Brehm, B., and Höfler, K. The 21.22-eV photoelectron spectrum of barium, *Intern. J. Mass Spectrom. Ion Phys.* **17**, 371 (1975).
- [4382] Aloisi, G., Santini, S., and Savelli, G. Molecular complexes of heteroaromatic five membered ring compounds with tetracyanoethylene. Charge transfer spectra, equilibrium constants and ionization potentials of the donors, *J. Chem. Soc. Faraday Trans. I* **70**, 2045 (1975).
- [4383] Cradock, S., and Duncan, W. Photoelectron spectra of OCSe and SCSe, *J. Chem. Soc. Faraday Trans. II* **6**, 1262 (1975).
- [4384] Brittain, H. G., and Disch, R. L. The He(I) photoelectron spectra of some bivalent transition metal β -diketonate complexes*, *J. Electron Spectrosc. Relat. Phenom.* **7**, 475 (1975).
- [4385] Asmus, P., Klessinger, M., Meyer, L.-U., and deMeijere, A. Conjugative interaction between cyclopropyl walsh orbitals and π -orbitals in dispiro[2.2.2.2]deca-4,9-diene, *Tetrahedron Letters* **6**, 381 (1975).

- [4386] Bernardi, F., Colonna, F. P., Dembech, P., Distefano, G., and Vivarelli, P. Through space and through bond interactions in [2,2] metacyclo-2,6-pyridinophane studied by ultraviolet photoelectron spectroscopy, *Chem. Phys. Letters* **36**, 539 (1975).
- [4387] Bischof, P., Gleiter, R., and Hofmann, P. 229. Photoelectron spectra of vicinal tricarbonyls, *Helv. Chim. Acta* **58**, 2130 (1975).
- [4388] DeKock, R. L., and Lloyd, D. R. The He¹photoelectron spectrum of sulphur trioxide, *J. Chem. Soc. Dalton Trans.* 526 (1973).
- [4389] Bernardi, F., Distefano, G., Mangini, A., Pignataro, S., and Spunta, G. Photoelectron spectra of substituted anisoles and thioanisoles, *J. Electron Spectrosc. Relat. Phenom.* **7**, 457 (1975).
- [4390] Bally, T., and Haselbach, E. The photoelectron spectrum of triisopropylidenecyclopropane ('Hexamethyl-3-radialene'), *Helv. Chim. Acta* **58**, 321 (1975).
- [4391] Allan, M., Heilbronner, E., and Kloster-Jensen, E. A photoelectron-spectroscopic investigation of benzologue tropones, *J. Electron Spectrosc. Relat. Phenom.* **6**, 181 (1975).
- [4392] Bak, B., Jansen, P., and Stafast, H. Cyanogen azide: ionisation potentials and AB initio SCF MO calculation, *Chem. Phys. Letters* **35**, 247 (1975).
- [4393] Anthony, M. T., Green, M. L. H., and Young, D. Preparation of zerovalent di(η -arene)titanium compounds using titanium vapour, *J. Chem. Soc. Dalton Trans.* 1419 (1975).
- [4394] Bieri, G., Heilbronner, E., Goldstein, M. J., Leight, R. S., and Lipton, M. S. The photoelectron spectrum of dewar benzene, *Tetrahedron Letters* **8**, 581 (1975).
- [4395] Brown, R. S. A photoelectron investigation of acyl silanes: The photoelectron spectra of trimethylsilyl phenyl ketone and phenyl *tert*-butyl ketone, *Can. J. Chem.* **53**, 2446 (1975).
- [4396] Batich, C. Photoelectron spectroscopy of bis(π -allyl) nickel and its methyl substituted derivatives: support for the near validity of Koopmans' Theorem, *J. Am. Chem. Soc.* **98**, 7585 (1976).
- [4397] Bischof, P., Gleiter, R., Hopf, H., and Lenich, F. T. Photoelectron spectra of open chain C_nH_n isomers, *J. Am. Chem. Soc.* **97**, 5467 (1975).
- [4398] Barker, G. K., Lappert, M. F., Pedley, J. B., Sharp, G. J., and Westwood, N. P. C. Bonding studies of boron and the group 3-5 elements. Part XV. He(I) photoelectron spectra of monomeric group 3 trihalide, trimethyl, and mixed halogenomethyl species, *J. Chem. Soc. Dalton Trans.* 1765 (1975).
- [4399] Brown, R. S. Photoelectron studies on intramolecularly hydrogen-bonded systems. I. The photoelectron spectra of *cis*-and *trans*-2-amino-cyclopentanol, and *cis*-and *trans*-2-(N,N-dimethylamino) cyclopentanol, *Can. J. Chem.* **54**, 642 (1976).
- [4400] Bischof, P., Gleiter, R., and Müller, E. The electronic structure of benzvalene, *Tetrahedron* **32**, 2769 (1976).
- [4401] Egdell, R., Green, J. C., and Rao, C. N. R. Photoelectron spectra of substituted benzenes, *Chem. Phys. Letters* **33**, 600 (1975).
- [4402] Guimon, M. F., Guimon, C., Metras, F., and Pfister-Guillouzo, G. Application of photoelectron spectroscopy to conformational analysis of S-tetrathianes, *J. Am. Chem. Soc.* **98**, 2078 (1976).
- [4403] Gonbeau, D., Guimon, C., Deschamps, J., and Pfister-Guillouzo, G. Electronic structure of sulphur compounds IX. Photoelectron spectra of various 1,2-dithiole-3-thiones, *J. Electron Spectrosc. Relat. Phenom.* **6**, 99 (1975).
- [4404] Frost, D. C., Lee, S. T., McDowell, C. A., and Westwood, N. P. C. Photoelectron spectroscopic studies of some nitrosyl and nitryl halides and nitric acid, *J. Electron Spectrosc. Relat. Phenom.* **7**, 331 (1975).
- [4405] Guimon, C., Guimon, M.-F., and Pfister-Guillouzo, G. Application of photoelectron spectroscopy to conformational analysis of two 1,3-dithia compounds, *Tetrahedron Letters* **17**, 1413 (1975).
- [4406] Gleiter, R., Gygax, R., and Reid, D. H. The He 584 Å photoelectron spectra of analogs of thiathiophethene, *Helv. Chim. Acta* **58**, 1591 (1975).
- [4407] Guimon, C., and Pfister-Guillouzo, G. Electronic structure of sulfur compounds XVI. Photoelectron spectra of ethylene trithiocarbonate and ethylene dithiocarbonate, *J. Electron Spectrosc. Relat. Phenom.* **7**, 191 (1975).
- [4408] Frost, D. C., Lee, S. T., McDowell, C. A., and Westwood, N. P. C. The photoelectron spectrum of diazene (diimine), *Chem. Phys. Letters* **30**, 26 (1975).
- [4409] Cradock, S., Ebsworth, E. A. V., Meikle, G. D., and Rankin, D. W. H. Preparation, properties, and molecular structure of silylsulphinyllamine, *J. Chem. Soc. Dalton Trans.* 805 (1975).
- [4410] Guimon, M.-F., Guimon, C., and Pfister-Guillouzo, G. Application of photoelectron spectroscopy to conformational analysis of 1,2,4-trithiolanes, *Tetrahedron Letters* **7**, 441 (1975).
- [4411] Richardson, N. V., and Weinberger, P. The electronic structure of the S₈ molecule, *J. Electron Spectrosc. Relat. Phenom.* **6**, 109 (1975).
- [4412] Hall, M. B., Hillier, I. H., Connor, J. A., Guest, M. F., and Lloyd, D. R. The electronic structure of transition metal complexes containing organic ligands III. Low energy photoelectron spectra and *ab initio* SCF MO calculations of iron tricarbonyl cyclobutadiene, *Mol. Phys.* **30**, 839 (1975).
- [4413] Cradock, S., Ebsworth, E. A. V., and Muiry, I. B. Photoelectron spectra and bonding of (N-B)-2,8,9-trioxa-5-aza-1-bora-bicyclo[3.3.3]undecane (boratran) and some 2,8,9-trioxa-5-aza-1-sila-bicyclo[3.3.3]undecanes (silatrans), *J. Chem. Soc. Dalton Trans.* 25 (1975).
- [4414] Delwiche, J. P., and Praet, M. Th. Spectre de photoélectrons He(I) de l'hexafluorobicyclo-[2.2.0]-hexadiène-2,5, *J. Electron Spectrosc. Relat. Phenom.* **7**, 317 (1975).
- [4415] Banna, M. S., and Shirley, D. A. Molecular photoelectron spectroscopy at 132.3 eV: N₂, CO, C₂H₄ and O₂, *J. Electron Spectrosc. Relat. Phenom.* **8**, 255 (1976).
- [4416] Bastide, J., Heilbronner, E., Maier, J. P., and Ashe III, A. J. The photoelectron spectrum of bismabenzene, *Tetrahedron Letters* **6**, 411 (1976).
- [4417] Bock, H., Haselbach, E., Maier, E., and Stafast, H. The photoelectron spectrum of tetracyanomethane, *Helv. Chim. Acta* **59**, 1035 (1976).
- [4418] Baker, A. D., Brisk, M. A., Venanzi, T. J., Kwon, Y. S., and Sadka, S. Spiroconjugation involving sulfur 3p atomic orbitals, *Tetrahedron Letters* **38**, 3415 (1976).
- [4419] Alder, R. W., Goode, N. C., King, T. J., Mellor, J. M., and Miller, B. W. A 1,5-Diazabicyclo[3.3.3] undecane derivative with almost planar bridgehead nitrogens, *J. Chem. Soc. Chem. Commun.* 173, (1976).
- [4420] Alderdice, D. S., and Dixon, R. N. Photoelectron spectra of nitrosyl halides, *J. Chem. Soc. Faraday Trans. II* **73**, 245 (1977).
- [4421] Bally, T., Haselbach, E., Lanyiova, S., Marschner, F., and Rossi, M. Concerning the conformation of isolated benzylideneaniline, *Helv. Chim. Acta* **59**, 486 (1976).
- [4422] Abbas, M. I., Dyke, J. M., and Morris, A. Photoelectron spectrum of nitrosyl chloride, *J. Chem. Soc. Faraday Trans. II* **72**, 814 (1976).
- [4423] Schmidt, H., Schweig, A., Mathey, F., and Müller, G. Molecular conformation and hyperconjugation. P-C-hyperconjugation and the conformation of vinyl-, allyl-,

- phenyl-and benzylphosphines, *Tetrahedron*, **31**, 1287 (1975).
- [4424] Hardin, A. H., and Sandorfy, C. Photoelectron and vacuum ultraviolet spectra of a series of fluoroethers, *J. of Fluorine Chem.* **5**, 435 (1975).
- [4425] Green, J. C., Jackson, S. E., and Higginson, B. Photoelectron studies of some bent bis(η -cyclopentadienyl)metal complexes. Part I. Some eighteen-electron systems with hydride, alkyl, olefin, allyl, and carbonyl ligands, *J. Chem. Soc. Dalton Trans.* **403** (1975).
- [4426] Green, J. C., and Hayes, A. J. Ionization energies of an Mo-Mo quadruple bond; a He(I) photoelectron study of some molybdenum-dycarboxylate dimers, *Chem. Phys. Letters* **31**, 306 (1975).
- [4427] Flamini, A., Semprini, E., and Condorelli, G. Electronic structure of donor groups by photoelectron spectroscopy: thioderivatives of the carboxylate group, *Chem. Phys. Letters* **32**, 365 (1975).
- [4428] Groenenboom, C. J., de Liefde Meijer, H. J., Jellinek, F., and Oskam, A. UV photoelectron spectra of (cyclopentadienyl)-(cycloheptatrienyl)- zirconium-, niobium and-molybdenum, *J. Organometal. Chem.* **97**, 73 (1975).
- [4429] Houk, K. N., Chang, Y.-M., and Engel, P. S. Photoelectron spectroscopy of azo compounds, *J. Am. Chem. Soc.* **97**, 1824 (1975).
- [4430] Hush, N. S., Cheung, A. S., and Hilton, P. R. Binding energies of π -and "lone-pair"-levels in mono- and diaza-phenanthrenes and anthracenes: an He(I) photoelectron spectroscopic study, *J. Electron Spectrosc. Relat. Phenom.* **7**, 385 (1975).
- [4431] Levenson, R. A., Cihonski, J. L., Milazzo, P., and Ceasar, G. P. Photoelectron spectra of $\text{Fe}(\text{CO})_2\text{X}_2$ ($\text{X}=\text{Br}, \text{I}$). Transition metal analogs of the methylene halides, *Inorg. Chem.* **14**, 2578 (1975).
- [4432] Kröner, J., Wiberg, N., and Bayer, H. Photoelektronenspektrum von tetrazen, *Angew. Chem.* **87**, 203 (1975).
- [4433] Asmus, P., and Klessinger, M. Photoelektronenspektren organischer Verbindungen, IX. Sterisch fixierte Bicyclopropyle, *Liebigs. Ann. Chem.* **12**, 2169 (1975).
- [4434] Klessinger, M., Asmus, P., and Kraatz, U. Photoelektronenspektren organischer Verbindungen-VII, *Tetrahedron* **31**, 517 (1975).
- [4435] Müller, J.-F. 287. Spectres photoélectroniques He(I) et He(II) du benzo[*b*]sélénophène et du benzo[*b*]tellurophène, *Helv. Chim. Acta* **58**, 2646 (1975).
- [4436] Hildenbrand, D. L. Thermochemistry of molecular FeO , FeO^+ and FeO_2 , *Chem. Phys. Letters* **34**, 352 (1975).
- [4437] Salmona, G., Faure, R., and Vincent, E.-J. Structure électronique en série benzothiazolique: Spectres photoélectroniques et calculs C.N.D.O., *C. R. Acad. Sci. Paris* **280**, 605 (1975).
- [4438] Limouzin, Y., and Maire, J. C. Spectres photoélectroniques des composés organométalliques, *J. Organometal. Chem.* **92**, 169 (1975).
- [4439] Guimon, C., Pfister-Guillouzo, G., and Arbelot, M. Spectres photoélectroniques d'hétérocycles carbonyles et thiocarbonyles, *Tetrahedron* **31**, 2769 (1975).
- [4440] Hush, N. S., and Cheung, A. S. Ionization potentials and donor properties of nucleic acid bases and related compounds, *Chem. Phys. Letters* **34**, 11 (1975).
- [4441] Fehlner, T. P. Photoelectron spectroscopy of *closocarbonanes*. Observation of exo-polyhedral molecular orbitals, *Inorg. Chem.* **14**, 934 (1975).
- [4442] Guest, M. F., Hillier, I. H., Higginson, B. R., and Lloyd, D. R. The electronic structure of trans- π -ition metal complexes containing organic ligands II. Low energy photoelectron spectra and *ab initio* SCF MO calculations of dibenzene chromium and benzene chromium tricarbonyl, *Mol. Phys.* **29**, 113 (1975).
- [4443] Hall, M. B. The use of spin-orbit coupling in the interpretation of photoelectron spectra. I. Application to substituted rhodium pentacarbonyls, *J. Am. Chem. Soc.* **97**, 2057 (1975).
- [4444] Boyd, R. J., Bünzli, J.-C. G., and Snyder, J. P. Photoelectron spectra of diazabasketene, diazadeltacyclene, and related polycyclic *cis*- azoalkanes, *J. Am. Chem. Soc.* **98**, 2398 (1976).
- [4445] Brown, R. S. Application of photoelectron spectroscopy to intramolecularly hydrogen-bonded systems. Part III. The photoelectron spectra of *cis* and *trans* 2-substituted cyclopentanols and *cis* and *trans* 2-substituted cyclohexanols, *Can. J. Chem.* **54**, 1929 (1976).
- [4446] Kobayashi, T. Photoelectron spectra of *N,N*-dimethyl-nitrosamine and related compounds, *Z. Phys. Chem. N. Folge* **97**, 269 (1975).
- [4447] Colonna, F. P., Distefano, G., Pignataro, S.; Pitacco, G.; and Valentin, E. Ionization energies of some amines and enamines and an estimation of their relative basicity in gaseous phase, *J. Chem. Soc., Faraday Trans. II* **71**, 1572 (1975).
- [4448] Bieri, G., Heilbronner, E., Kobayashi, T., Schmelzer, A., Goldstein, M. J., Leight, R. S., and Lipton, M. S. 286. *Dewar* benzene and some of its derivatives. A photoelectron spectroscopic analysis, *Helv. Chim. Acta* **59**, 2657 (1976).
- [4449] Lloyd, D. R., Lynaugh, N., Roberts, P. J., and Guest, M. F. Photoelectron studies of boron compounds, Part 5. - Higher boron hydrides B_3H_{10} , B_5H_9 , and $\text{B}_{10}\text{H}_{11}$, *J. Chem. Soc. Faraday Trans. II* **71**, 1382 (1975).
- [4450] Lee, T. H., and Rabalaïs, J. W. Photoelectron spectrum and ground state electronic structure of chromyl chloride vapor, *Chem. Phys. Letters* **34**, 135 (1975).
- [4451] Head, R. A., Nixon, J. F., Sharp, G. J., and Clark, R. J. Photoelectron spectroscopic study of metal trifluorophosphine and hydridotrifluorophosphine complexes, *J. Chem. Soc. Dalton Trans.* 2054 (1975).
- [4452] Hosomi, A., and Taylor, T. G. Studies of interactions of adjacent carbon-metal σ bonds by photoelectron spectroscopy, *J. Am. Chem. Soc.* **97**, 3682 (1975).
- [4453] Hilpert, K., Naoumidis, A., and Wolff, G. Mass spectrometric study of the evaporation of BaAl_2O_4 , *High Temp. Sci.* **7**, 1 (1975).
- [4454] Mollere, P. D., Houk, K. N., Bomse, D. S., and Morton, T. H. Photoelectron spectra of sterically congested alkenes and dienes, *J. Am. Chem. Soc.* **98**, 4732 (1976).
- [4455] Allan, M., Heilbronner, E., Kloster-Jensen, E., and Maier, J. P. The π -states of tetraacetylene radical cation, *Chem. Phys. Letters* **41**, 228 (1976).
- [4456] Spanget-Larsen, J., Gleiter, R., and Hünig, S. The electronic structure of dibenzotetrathiafulvalene, *Chem. Phys. Letters* **37**, 29 (1976).
- [4457] Meeks, J. L., and McGlynn, S. P. Photoelectron spectra of carbonyls. Oxamide, parabanic acid, and their *N*-methyl derivatives, *J. Am. Chem. Soc.* **97**, 5079 (1975).
- [4458] Kobayashi, T. Photoelectron spectra of p-benzoquinones, *J. Electron Spectrosc. Relat. Phenom.* **7**, 349 (1975).
- [4459] Kobayashi, T., Yokota, K., and Nagakura, S. Photoelectron spectra of the *cis*- and *trans*-isomers of some ethylene derivatives, *Bull. Chem. Soc. Japan* **48**, 412 (1975).
- [4460] Egdell, R., Green, J. C., Rao, C. N. R., Gowenlock, B. G., and Pfab, J. He(I) photoelectron studies of *C*-nitroso compounds, *J. Chem. Soc. Faraday Trans. II*, 988 (1975).
- [4461] Utsunomiya, C., Kobayashi, T., and Nagakura, S. Photoelectron spectra of substituted naphthalenes, *Bull. Chem. Soc. Japan* **48**, 1852 (1975).
- [4462] Rao, C. N. R. Lone-pair ionization bands of chromophores in the photoelectron spectra of organic molecules, *Indian J. Chem.* **13**, 950 (1975).
- [4463] Neijzen, B. J. M., Schmitz, R. F., Klumpp, G. W., and

- DeLange, C. A. $n-\pi$ Interactions in homoallylic methyl ethers. A photoelectron spectroscopic study, *Tetrahedron* **31**, 873 (1975).
- [4469] Mines, G. W., and Thompson, H. W. The photoelectron spectra of amides, thioamides, ureas and thioureas, *Spectrochim. Acta* **31A**, 137 (1975).
- [4470] Maier, J. P., Muller, J.-F., and Kubota, T. Ionisation energies and the electronic structures of the N-oxides of diazenes, *Helv. Chim. Acta* **58**, 1634 (1975).
- [4471] Meeks, J. L., Arnett, J. F., Larson, D., and McGlynn, S. P. Photoelectron spectroscopy of carbonyls. Ionization assignments, *Chem. Phys. Letters* **30**, 190 (1975).
- [4472] Kobayashi, T., and Nagakura, S. Angular distribution for the photoelectron spectra of benzene and hexafluorobenzene, *J. Electron Spectrosc. Relat. Phenom.* **7**, 187 (1975).
- [4473] Kobayashi, T., and Nagakura, S. Photoelectron spectra of nitrophenols and nitroanisoles, *J. Electron Spectrosc. Relat. Phenom.* **6**, 421 (1975).
- [4474] Lappert, M. F., Pedley, J. B., Wilkins, B. T., Stelzer, O., and Unger, E. Bonding studies of compounds of boron and the group 3–5 elements. Part XIII. He(I) photoelectron spectra of phosphines R_nPX_{3-n} (R =Me or Bu'; X=H, Cl, or F; $n=1-3$), $(Me_2N)PCl_{3-n}$ ($n=1-3$), and $(R_2N)PF_2$ (R =Me or Et), *J. Chem. Soc. Dalton Trans.* 1207 (1975).
- [4475] Kobayashi, T., Yokota, K., and Nagakura, S. Photoelectron spectra of trans-azobenzene and benzalaniline, *J. Electron Spectrosc. Relat. Phenom.* **6**, 167 (1975).
- [4476] Rosmus, P., Stafast, H., and Bock, H. Sulphur dicyanide: Ionisation potentials and Hartree-Fock calculations, *Chem. Phys. Letters* **34**, 275 (1975).
- [4477] Lloyd, D. R., Roberts, P. J., and Hillier, I. H. Electronic structure of nitric acid studied by photoelectron spectroscopy and molecular orbital calculation, *J. Chem. Soc. Faraday Trans. II* **71**, 496 (1975).
- [4478] Hino, S., Seki, K., and Inokuchi, H. Photoelectron spectra of *p*-terphenyl in gaseous and solid states, *Chem. Phys. Letters* **36**, 335 (1975).
- [4479] Schweig, A., and Thon, N. Measurement of relative conformational stabilities by variable temperature photoelectron spectroscopy. A study of rotational isomerism in thioanisole, *Chem. Phys. Letters* **38**, 482 (1976).
- [4480] Aue, D. H., Webb, H. M., and Bowers, M. T. Quantitative proton affinities, ionization potentials, and hydrogen affinities of alkylamines, *J. Am. Chem. Soc.* **98**, 311 (1976).
- [4481] Gleiter, R., Kobayashi, M., Spanget-Larsen, J., Ferraris, J. P., Bloch, A. N., Bechgaard, K., and Cowan, D. O. Photoelectron and electronic absorption spectra of tetrathiafulvalene and related compounds, *Ber. Bunsenges.* **79**, 1218 (1975).
- [4482] Gounelle, Y., Menard, C., Pechine, J. M., Solgadi, D., Menes, F., and Botter, R. Conformational effects on ionization potentials; Photoelectron spectra of dibromo- and bromofluoro- alkyl compounds, *J. Electron Spectrosc. Relat. Phenom.* **7**, 247 (1975).
- [4483] Murad, E., and Hildenbrand, D. L. Thermochemical properties of gaseous ZrO and ZrO_2 , *J. Chem. Phys.* **63**, 1133 (1975).
- [4484] Peel, J. B., and Willett, G. D. Photoelectron spectroscopic studies of the higher alcohols, *Aust. J. Chem.* **28**, 2357 (1975).
- [4485] Alderdice, D. S., and Dixon, R. N. Photoelectron spectrum of sulphur trioxide, *J. Chem. Soc. Faraday Trans II* **72**, 372 (1976).
- [4486] Smoes, S., Drowart, J., and Welter, J. M. Thermodynamic study of the vaporization of europium monosulfide by Knudsen-cell mass spectrometry Atomization energies of $EuS(g)$, $Eu_2S(g)$, $EuS_2(g)$, $Eu_2O(g)$, $Eu_2O_2(g)$, $Eu_2OS(g)$, and $Eu_2S_2(g)$, *J. Chem. Thermodyn.* **9**, 275 (1977).
- [4487] Meeks, J. L., Arnett, J. F., Larson, D. B., and McGlynn, S. P. Photoelectron spectroscopy of carbonyls. Urea, oxamide, oxalic acid and oxamic acid, *J. Am. Chem. Soc.* **97**, 3905 (1975).
- [4488] Obenland, S., and Schmidt, W. Photoelectron spectra of polynuclear aromatics. IV. The helicenes, *J. Am. Chem. Soc.* **97**, 6633 (1975).
- [4489] Peel, J. B., and Willett, G. D. Photoelectron spectrum of methylenimine by spectrum stripping, *J. Chem. Soc., Faraday Trans. II* **71**, 1799 (1975).
- [4490] Schmidt, H., Schweig, A., and Manuel, G. Vapourphase conformation of benzylmercuric chloride, *J. Chem. Soc. Chem. Commun.* 667, (1975).
- [4491] Samson, J. A. R., and Gardner, J. L. On the ionization potential of molecular oxygen, *Can. J. Phys.* **53**, 1948 (1975).
- [4492] Higginson, B. R., Lloyd, D. R., Evans, S., and Orchard, A. F. Photoelectron studies of metal carbonyls. Part 5. - Substituted group VIIA carbonyls, *J. Chem. Soc. Faraday Trans. II* **71**, 1913 (1975).
- [4493] Runge, W., Kosbahn, W., and Kroner, J. The molecular structure of allenes and ketenes II [1] Photoelectron spectra, absorption spectra, and CNDO/S-calculations of phenyl and methyl substituted allenes, *Ber. Bunsenges.* **79**, 371 (1975).
- [4494] Stockbauer, R., and Inghram M. G. Vibrational structure in the ground state of ethylene and ethylene-d₄ molecular ions investigated by threshold photoelectron spectroscopy, *J. Electron Spectrosc. Relat. Phenom.* **7**, 492 (1975).
- [4495] Kobayashi, T., and Nagakura, S. Photoelectron spectra of phenyl isocyanates and phenyl isothiocyanate, *J. Electron Spectrosc. Relat. Phenom.* **7**, 488 (1975).
- [4496] Wittel, K. The photoelectron spectrum of formylfluoride, *J. Electron Spectrosc. Relat. Phenom.* **8**, 245 (1976).
- [4497] Aue, D. H., Webb, H. M., and Bowers, M. T. Photoelectron spectrum and gas-phase basicity of manxine. Evidence for a planar bridgehead nitrogen, *J. Am. Chem. Soc.* **97**, 4136 (1975).
- [4498] Wu, M., and Fehlner, T. P. Photoelectron spectroscopy of unstable species. The PN molecule, *Chem. Phys. Letters* **36**, 114 (1975).
- [4499] Zverev, V. V., Vovna, V. I., Kitaev, Yu. P., and Vilessov, F. I. Photoelectron spectrum and structure of formaldehyde azine, *J. Struct. Chem.* **16**, 947 (1975).
- [4500] Lee, T. H., Colton, R. J., White, M. G., and Rabalais, J. W. Electronic structure of hydrazoic acid and the azide ion from x-ray and ultra- violet electron spectroscopy, *J. Am. Chem. Soc.* **97**, 4845 (1975).
- [4501] Whitesides, T. H., Lichtenberger, D. L., and Budnik, R. A. Bonding in ring whizzers. I. Photoelectron spectra and molecular orbital calculations for $(\eta^5-C_5H_5)Mn(CO)_3$, $(\eta^5-C_6H_5)Mn(CO)_3$, and $(\eta^5-C_7H_7)Mn(CO)_3$, *Inorg. Chem.* **14**, 68 (1975).
- [4503] Johnson, L. P., and Morrison, J. D. Double ionization to low-lying states of the doubly- charged rare gases, *Intern. J. Mass Spectrom. Ion Phys.* **18**, 355 (1975).
- [4504] Arnold, D. E. J., Ebsworth, E. A. V., and Rankin, D. W. H. Preparation and properties of difluoro-(difluorophosphinoamino)borane, *J. Chem. Soc. Dalton*, 823 (1976).
- [4505] Guido, M., and Gigli, G. Mass spectrometric study of the gaseous PrCN molecule, *High Temp. Sci.* **7**, 122 (1975).
- [4506] Farber, M., and Srivastava, R. D. The dissociation energy of barium oxide, *High Temp. Sci.* **7**, 74 (1975).
- [4507] Compernolle, F. Mass spectrum and heat of formation of isocyanic acid. Production of $[HCO]^+$ from discrete electronic state of molecular ion, *Org. Mass Spectrom.* **10**, 289 (1975).
- [4508] Müller, C., Schweig, A., and Vermeer, H. Thiirene dioxides. Electronic structure, *J. Am. Chem. Soc.* **97**, 982 (1975).

- [4509] Brown, R. S. Application of photoelectron spectroscopy to intramolecularly hydrogen-bonded systems. Part II. On the $n-\pi^*$ blue shift of carbonyl-containing molecules, *Can. J. Chem.* **54**, 3203 (1976).
- [4510] Koenig, T., Wielesek, R., Snell, W., and Balle, T. Helium(I) photoelectron spectrum of *p*-quinodimethane, *J. Am. Chem. Soc.* **97**, 3225 (1975).
- [4511] Brown, R. S. Application of photoelectron spectroscopy to intramolecularly hydrogen-bonded systems. Part IV. π -systems; the photoelectron spectra of *syn*- and *anti*-7-norbornenol, *Can. J. Chem.* **54**, 3206 (1976).
- [4512] Wittel, K., Bock, H., Haas, A., and Pfleger, K. H. Photoelectron spectra and molecular properties XLVII. F₃C-substituted mercury compounds, *J. Electron Spectrosc. Relat. Phenom.* **7**, 365 (1975).
- [4513] Kimura, K., Katsumata, S., Yamazaki, T., and Wakabayashi, H. UV photoelectron spectra and sum rule consideration; Out-of-plane orbitals of unsaturated compounds with planar-skeleton structure, *J. Electron Spectrosc. Relat. Phenom.* **6**, 41 (1975).
- [4514] Kimura, K., Katsumata, S., and Osafune, K. Photoelectron spectroscopic study of skew compounds. Methylhydrazine and unsymmetrical dimethylhydrazine, *Bull. Chem. Soc. Japan* **48**, 2736 (1975).
- [4515] Schäfer, W., Schweig, A., Vermeer, H., Bickel-haupt, F., and De Graaf, H. On the nature of the "free electron pair" on phosphorus in aromatic phosphorus compounds: the photoelectron spectrum of 2-phosphanaphthalene, *J. Electron Spectrosc. Relat. Phenom.* **6**, 91 (1975).
- [4516] Lloyd, D. R., Roberts, P. J., Hillier, I. H., and Shenton, I. C. On the photoelectron spectrum of sulphur trioxide, *Mol. Phys.* **31**, 1549 (1976).
- [4517] Bertoti, I., Cradock, S., Ebsworth, E. A. V., and Whiteford, R. A. Photoelectron spectra and transannular interactions in 1-silacyclopent-3-enes, *J. Chem. Soc. Dalton* 937 (1976).
- [4518] Gingerich, K. A., and Miller, F. Thermodynamic study of gaseous sodium-phosphorus-oxygen ternary molecules by high temperature mass spectrometry, *J. Chem. Phys.* **63**, 1211 (1975).
- [4519] Ulman, J. A., and Fehlner, T. P. Substituent effects in cluster species. Photoelectron spectra of 1-, 2-, and μ -substituted pentaborane(9), *J. Am. Chem. Soc.* **98**, 1119 (1976).
- [4520] McGlynn, S. P., and Meeks, J. L. Photoelectron spectra of carbonyls: Acetaldehyde, acetamide, biacetyl, pyruvic acid, methyl pyruvate and pyruvamide, *J. Electron Spectrosc. Relat. Phenom.* **6**, 269 (1975).
- [4521] Vovna, V. I., Vilesov, F. I., and Lopatin, S. N. Photoelectron spectra of hydrazine and some alkyl derivatives, *Opt. Spectrosc.* **38**, 143 (1975).
- [4522] Rothgery, E. F., McGee, H. A. Jr., and Pusatcioglu, S. Aminodifluoroborane, *Inorg. Chem.* **14**, 2236 (1975).
- [4523] Van Den Ham, D. M. W., Beerlage, M., Van Der Meer, D., and Feil, D. Photoelectron spectra of fluorine substituted diazanaphthalenes: "even cases," *J. Electron Spectrosc. Relat. Phenom.* **7**, 33 (1975).
- [4524] Young, V. Y., and Cheng, K. L. The photoelectron spectra of halogen substituted acetones, *J. Chem. Phys.* **65**, 3187 (1976).
- [4525] Fridh, C., and Åsbrink, L. Photoelectron and electron impact spectrum of HCN, *J. Electron Spectrosc. Relat. Phenom.* **7**, 119 (1975).
- [4526] Kröner, J., Nölle, D., Nöth, H., and Winterstein, W. Photoelektronenspektroskopische Untersuchungen an Bor-Verbindungen, II Fünfgliedrige Hetero-cyclen des Bors, *Chem. Ber.* **108**, 3807 (1975).
- [4527] Aue, D. H., Webb, H. M., and Bowers, M. T. Proton affinities, ionization potentials, and hydrogen affinities of nitrogen and oxygen bases. Hybridization effects, *J. Am. Chem. Soc.* **97**, 4137 (1975).
- [4528] Steiger, R. A., and Cater, E. D. Vaporization, thermodynamics, and dissociation energy of yttrium monosulfide, *High Temp. Sci.* **7**, 204 (1975).
- [4529] Cocke, D. L., Gingerich, K. A., and Kordis, J. Atomization energies of gaseous EuAu, EuAu₂, EuRh, RhAu, Eu₂, and EuAg, and predicted stability of selected diatomic europium compounds, *High Temp. Sci.* **7**, 61 (1975).
- [4530] Van Veen, E. H., and Plantenga, F. L. Threshold electron-impact excitation spectrum of pyridine, *Chem. Phys. Letters* **30**, 28 (1975).
- [4531] Heilbronner, E., Brogli, F., and Vogel, E. Photoelectron spectroscopic assignment of symmetry to the ground state and first excited state of the 1,4-cyclohexadiene radical cation, *J. Electron Spectrosc. Relat. Phenom.* **9**, 227 (1976).
- [4532] Cocke, D. L., Gingerich, K. A., and Kordis, J. Gaseous phosphorus compounds XI. Thermodynamic investigation of the gaseous molecule RhP₂, *High Temp. Sci.* **7**, 20 (1975).
- [4533] Kaposi, O., Riedel, M., and Sánchez, G. R. Mass-spectrometric study of electron-impact and heterogeneous pyrolytic decomposition of methyl bromide, *Acta Chim. Acad. Sci. Hung.* **85**, 361 (1975).
- [4534] Selim, E. T. M. Electron impact study of benzene, *Egypt. J. Phys.* **7**, 91, (1976).
- [4535] Mouvier, G., and Hernández, R. Ionisation and appearance potentials of alkylketones, *Org. Mass Spectrom.* **10**, 958 (1975).
- [4536] Weiner, M. A., and Lattman, M. Ultraviolet photoelectron spectra of 4-substituted pyridineboranes, *Inorg. Nucl. Chem. Lett.* **11**, 723 (1975).
- [4537] Wieczorek, J. S., Koenig, T., and Balle, T. The He(I) photoelectron spectra of amine n-oxides, *J. Electron Spectrosc. Relat. Phenom.* **6**, 215 (1975).
- [4538] Van Hoorn, M. D. He(I) ionisation potentials and MO calculations of butenyne and the monomethyl-substituted butenyne, *J. Electron Spectrosc. Relat. Phenom.* **6**, 65 (1975).
- [4539] Radwan, T. N., and Turner, D. W. Molecular photoelectron spectroscopy. Part V. Ozone, *J. Chem. Soc. (A)*, 85 (1966).
- [4540] Allan, M., Heilbronner, E., and Kaupp, G. 203. The photoelectron spectrum of dibenzo-*p*-quinodimethane, *Helv. Chim. Acta* **59**, 1949 (1976).
- [4541] Haselbach, E., and Rossi, M. 33. Electronic structure, molecular conformation and reactivity of benzonorboradiene systems, *Helv. Chim. Acta* **59**, 278 (1976).
- [4542] Salahub, D. R., and Boschi, R. A. A pot-pourri of UV and PE spectra of iodides, Chemical spectroscopy and photochemistry in the vacuum-ultraviolet 191 (1974).
- [4543] Harland, P. W., Rankin, D. W. H., and Thynne, J. C. J. Ionisation by electron impact of phosphorus trifluoride and difluorocyanophosphine, *Int. J. Mass Spectrom. Ion Phys.* **13**, 395 (1974).
- [4544] Hildenbrand, D. L. Dissociation energy and ionization potential of the molecule CF, *Chem. Phys. Letters* **32**, 523 (1975).
- [4545] Lossing, F. P., and Traeger, J. C. Stabilization in cyclopentadienyl, cyclopentenyl, and cyclopentyl cations, *J. Am. Chem. Soc.* **97**, 1579 (1975).
- [4546] Flesch, G. D., and Svec, H. J. Thermochemistry of vanadium oxytrichloride and vanadium oxytrifluoride by mass spectrometry, *Inorg. Chem.* **14**, 1817 (1975).
- [4547] Katsumata, S., and Kimura, K. Photoelectron spectra and sum rule consideration. Effect of chlorine substitution on ionization energies for chloroethanes, chloroacetaldehydes and chloroacetyl chlorides, *J. Electron Spectrosc. Relat. Phenom.* **6**, 309 (1975).

- [4548] Green, M. M., Bafus, D., and Franklin, J. L. Short communication; Combined deuterium labeling and appearance potential measurements to uncover competing reaction mechanisms in the electron- impact-induced loss of water from cyclohexanol, *Org. Mass Spectrom.* **10**, 679 (1975).
- [4549] Wittel, K., Astrup, E. E., Bock, H., Graeffe, G., and Juslén, H. Photoelectron spectra and molecular properties, XLVIII Carbonates and thiocarbonates, *Z. Naturforsch.* **30b**, 862 (1975).
- [4550] Wu, M., and Fehlner, T. P. Valence level photoelectron spectra of some heavy group 4–6 diatomic molecules, *J. Am. Chem. Soc.* **98**, 7578 (1976).
- [4551] Maier, J. P., Muller,J.-F., Kubota, T., and Yamakawa, M. 183. Ionisation energies and the electronic structures of the N-oxides of azanaphthalenes and azaanthracenes, *Helv. Chim. Acta* **58**, 1641 (1975).
- [4552] Streets, D. G., and Berkowitz, J. The structure of Tl_2F_7 from photoelectron spectroscopy, *Chem. Phys. Letters* **38**, 475 (1976).
- [4553] Ravishankara, A. R., and Hanrahan, R. J. An electron impact investigation of 1,1,2,2-tetrafluorocyclobutane, *J. Phys. Chem.* **79**, 876 (1975).
- [4554] Hildenbrand, D. L. Vertical ionization potential of the CF_2 radical, *Chem. Phys. Letters* **30**, 32 (1975).
- [4555] Guimon, C., Arbelot, M., and Pfister-Guillouzo, G. Structure électronique de dérivés sulfures—VIII. Spectres photoélectroniques et électroniques d'hétérocycles thiocarbonylés benzo-substitués, *Spectrochim. Acta* **31A**, 985 (1975).
- [4556] Gusarov, A. V., Gotkis, I. S., and Gorokhov, L. N. Mass-spectrometric study of the evaporation products of the B_2O_3 – WO_3 system; Heat of formation of BWO_3 (gas), *High Temp. (USSR)* **13**, 324 (1975).
- [4557] Khandelwal, S. C., and Roebber, J. L. The photoelectron spectra of tetraphenylporphine and some metallotetraphenylporphyrins, *Chem. Phys. Letters* **34**, 355 (1975).
- [4558] Bock, H., Ensslin, W., Fehér, F., and Freund, R. Photoelectron spectra and molecular properties. LI. Ionization potentials of silanes Si_nH_{2n+2} , *J. Am. Chem. Soc.* **98**, 668 (1976).
- [4559] Lappert, M. F., Pedley, J. B., Sharp, G. J., and Guest, M. F. Bonding studies of compounds of boron and elements of groups 3–5, *J. Chem. Soc. Faraday Trans. II* **72**, 539 (1976).
- [4560] Rauh, E. G., and Ackermann, R. J. The first ionization potentials of neptunium and neptunium monoxide, *J. Chem. Phys.* **62**, 1584 (1975).
- [4561] Ames, D. L., and Turner, D. W. Photoelectron spectroscopic studies of dinitrogen tetroxide and dinitrogen pentoxide, *Proc. R. Soc. London Ser. A* **348**, 175 (1976).
- [4562] Fragala, I., Condorelli, G., Zanella, P., and Tondello, E. Photoelectron spectroscopy of actinide organometallic compounds I. Bis (cyclooctatetraene)actinide(IV) complexes, *J. Organometal. Chem.* **122**, 357 (1976).
- [4564] Utsunomiya, C., Kobayashi, T., and Nagakura, S. Photoelectron spectra of electron donor-acceptor complexes between bromine and alkylamines, *Chem. Phys. Letters* **39**, 245 (1976).
- [4565] Symon, D. A., and Waddington, T. C. Valence-band photoelectron spectra of some dicarbonyl(η -cyclopentadienyl)(ligand)iron compounds and tetrakis [carbonyl(η -cyclopentadienyl)iron(1)], *J. Chem. Soc. Dalton Trans.* 2140 (1975).
- [4566] Flaminii, A., Semprini, E., Stefani, F., Sorriso, S., and Cardaci, G. He(I) photoelectron spectra and semiempirical molecular-orbital calculations on methylmetal halides of group 4A elements, *J. Chem. Soc. Dalton Trans.* 731 (1976).
- [4567] Gounelle, Y., Jullien, J., Solgadi, D., Botter, R., and Menes, F. No. 157. – Effets de l'isomérisation sur les potentiels d'ionisation: Spectres des photoélectrons de dihalogénobenzènes, *J. Chim. Phys.* **10**, 1094 (1975).
- [4568] Ihle, H. R., and Wu, C. H. Mass spectrometric determination of the ionization potential and dissociation energy of LiD, *J. Chem. Phys.* **63**, 1605 (1975).
- [4569] Batich, C., Heilbronner, E., Quinn, C. B., and Wiseman, J. R. The electronic structure of vinyl ethers and sulfides with interrupted conjugation examined by photoelectron spectroscopy, *Helv. Chim. Acta* **59**, 512 (1976).
- [4570] Lichtenberger, D. L., and Fenske, R. F. The helium(I)photoelectron spectra and electronic structure of (η^5 -Cyclopentadienyl) d^n metal carbonyls, *J. Am. Chem. Soc.* **98**, 50 (1976).
- [4571] Cauletti, C., and Furlani, C. He(I) photoelectron spectra of bis(β -diketonate)nickel(II) complexes and their mono- and di-thio analogues, *J. Electron Spectrosc. Relat. Phenom.* **6**, 465 (1975).
- [4572] Bischof, P., Gleiter, R., Hafner, K., Kobayashi, M., and Spanget-Larsen, J. Polarized absorption and photoelectron spectra of aceheptylene, 3,5-dimethylaceheptylene and 3,5,8,10-tetramethylaceheptylene, *Ber. Bunsenges.* **80**, 532 (1976).
- [4573] Behan, J. M., Dean, F. M., and Johnstone, R. A. W. Photoelectron spectra of cyclic aromatic ethers. The question of the Mills-Nixon effect, *Tetrahedron* **32**, 167 (1976).
- [4574] Fehlner, T. P., Ulman, J., Nugent, W. A., and Kochi, J. K. Effect of alkyl substituents on the first ionization potential and on $5d^{10}$ ionization in dialkylmercury compounds, *Inorg. Chem.* **15**, 2544 (1976).
- [4575] Carlier, P., Dubois, J. E., Masclet, P., and Mouvier, G. Spectres de photoélectrons des alcynes, *J. Electron Spectrosc. Relat. Phenom.* **7**, 55 (1975).
- [4576] Battiste, D. R., Davis, L. P., and Nauman, R. V. Photoelectron spectroscopy and quantum chemical analysis of some *N*-nitrosamines, *J. Am. Chem. Soc.* **97**, 5071 (1975).
- [4577] Brown, R. S. A photoelectron investigation of the peroxide bond, *Can. J. Chem.* **53**, 3439 (1975).
- [4578] Johnson, I. Mass spectrometric study of the vaporization of cesium and sodium molybdates, *J. Phys. Chem.* **79**, 722 (1975).
- [4579] Starzewski, K. A. O., and Bock, H. Photoelectron spectra and molecular properties. 58. 1,2 Phosphorus ylides: Gas phase ionization potentials and charge distribution, *J. Am. Chem. Soc.* **98**, 8486 (1976).
- [4580] Hildenbrand, D. L. Thermochemistry of the gaseous tungsten fluorides, *J. Chem. Phys.* **62**, 3074 (1975).
- [4581] Bock, H., Wittel, K., Veith, M., and Wiberg, N. Photoelectron spectra and molecular properties. L. $^{1-3}$ On the blue color of bis(trimethylsilyl)diimine, *J. Am. Chem. Soc.* **98**, 109 (1976).
- [4582] Brown, C. M., Tilford, S. G., Tousey, R., and Ginter, M. L. Absorption spectrum of Si I between 1500 and 1900 Å, *J. Opt. Soc. Am.* **64**, 1665 (1974).
- [4583] Brown, C. M., Tilford, S. G., and Ginter, M. L. Absorption spectrum of Ca I in the 1580–2090 Å region, *J. Opt. Soc. Am.* **63**, 1454 (1973).
- [4584] Elbel, S., Dieck, H.t., Becker, G., and Ensslin, W. Photoelectron spectra of group 5 compounds. II. Conformational analysis of diphosphine (P_2H_4), *Inorg. Chem.* **15**, 1235 (1976).
- [4585] Fragala, I., Ciliberto, E., Fischer, R. D., Sienel, G. R., and Zanella, P. Photoelectron spectroscopy of f -element organometallic compounds II. Tricyclopentadienyl derivatives of uranium (IV) and thorium (IV). *J. Organometal. Chem.* **120**, C9 (1976).
- [4586] Dolby, L. J., Hanson, G., and Koenig, T. The He I

- photoelectron spectra of *N*-methylisoindole and *N*-methylindole, *J. Org. Chem.* **41**, 3537 (1976).
- [4587] Frost, D. C., Lee, S. T., McDowell, C. A., and Westwood, N. P. C. The photoelectron spectra of diazene, diazene-d₂, and *trans*-methyldiazene, *J. Chem. Phys.* **64**, 4719 (1976).
- [4588] Gibbins, S. G., Lappert, M. F., Pedley, J. B., and Sharp, G. J. Bonding studies of transition-metal complexes. Part II. Helium-I photoelectron spectra of homoleptic d⁰, d¹, and d¹⁰tetrakis(dialkylamides) of transition and group 4B metals and tungsten hexakis(dimethylamide), *J. Chem. Soc. Dalton* 72 (1975).
- [4589] Distefano, G., Pignataro, S., Ricci, A., Colonna, F. P., and Pietropaolo, D. Interactions of π orbitals with the group IV elements studied by ionization energy measurements, *Ann. Chim.* **64**, 153 (1974).
- [4590] Albini, A., and Mark, F. Photoelectron spectra of phenazine N-oxide and some of its derivatives, *J. Chem. Soc., Faraday Trans. II* **72**, 463 (1976).
- [4591] Lossing, F. P., and Traeger, J. C. Free radicals by mass spectrometry XLVI. Heats of formation of C₅H₇ and C₅H₉ radicals and cations, *Intern. J. Mass Spectrom. Ion Phys.* **19**, 9 (1975).
- [4592] Evlasheva, T. I., Potapov, V. K., and Tulupov, V. A. The structure of the ammonium chloride molecule, *Russ. J. Phys. Chem.* **49**, 738 (1975).
- [4593] Dougherty, D., Bloomfield, J. J., Newkome, G. R., Arnett, J. F., and McGlynn, S. P. Photoelectron spectra of carbonyls. Propellenes and propellanones, *J. Phys. Chem.* **80**, 2212 (1976).
- [4595] Bastide, J., and Maier, J. P. Electronic states of the radical cations of the 1,3-dipoles: HCNO, CH₂N₂, and N₃H, studied by photoelectron spectroscopy, *Chem. Phys.* **12**, 177 (1976).
- [4596] Dyke, J., Jonathan, N., Lee, E., and Morris, A. Vacuum ultraviolet photoelectron spectroscopy of transient species, *J. Chem. Soc. Faraday Trans. II* **72**, 1385 (1976).
- [4597] Bulgin, D. K., Dyke, J. M., and Morris, A. Hel photoelectron spectrum of the P₂(X¹ Σ_{g}^+) molecule, *J. Chem. Soc. Faraday Trans. II* **72**, 2225 (1976).
- [4598] Krause, J. R., and Bidinosti, D. R. Mass spectrometric studies of the ionization and thermal decomposition of tungsten π -cyclopentadienyl tricarbonyl dimer, *Can. J. Chem.* **53**, 628 (1975).
- [4599] Dougherty, D., Wittel, K., Meeks, J., and McGlynn, S. P. Photoelectron spectroscopy of carbonyls. Ureas, uracils, and thymine, *J. Am. Chem. Soc.* **98**:13, 3815 (1976).
- [4600] Paetzold, R., and Abd-el-Mottaleb, S. Correlative studies of some spectroscopic and bonding parameters in octahedrally coordinated metal carbonyl complexes, *J. Mol. Struct.* **24**, 357 (1975).
- [4602] Dixon, R. N., Duxbury, G., Rabalais, J. W., and Åsbrink, L. Ro-vibronic structure in the photoelectron spectra of H₂O, D₂O and HDO, *Mol. Phys.* **31**, 423 (1976).
- [4603] Phillips, G. R., Russell, M. E., and Solka, B. H. The structure of the [C₂H₅O]⁺ ion in the mass spectrum of diethyl ether, *Org. Mass. Spectrom.* **10**, 819 (1975).
- [4604] DeKock, R. L., Shehfeh, M. A., Lloyd, D. R., and Roberts, P. J. Ultraviolet photoelectron spectra of thiaryl chloride, *J. Chem. Soc. Faraday Trans. II* **72**, 807 (1976).
- [4606] Smith, J. A., and Pong, W. Ultraviolet photoelectron spectra of cesium halides, *Phys. Rev. B* **12**, 5931 (1975).
- [4607] Chantry, P. J. Positive ion appearance potentials measured in Cel₃, *J. Chem. Phys.* **65**, 4421 (1976).
- [4608] Dewar, M. J. S., Fonken, G. J., Jones, T. B., and Minter, D. E. Photoelectron spectra of molecules. Part VII. Cyclopropylallenes, *J. Chem. Soc. Perkin II*, 764 (1976).
- [4609] Koenig, T., Snell, W., and Chang, J. C. The He(I) photoelectron spectra of benzyl and α -cyanoisopropyl radicals, *Tetrahedron Letters*, **50**, 4569 (1976).
- [4610] Balkis, T., Gaines, A. F., Özgen, G., Özgen, I. T., and Flowers, M. C. Ionization of hydrogen sulphide, selenide and telluride by electron impact, *J. Chem. Soc. Faraday Trans. II* **72**, 524 (1976).
- [4611] Güsten, H., Klasinc, L., Tóth, T., and Knop, J. V. Photoelectron spectroscopy of heterocycles. 5H-dibenzo(a,d) cycloheptene analogs, *J. Electron Spectrosc. Relat. Phenom.* **8**, 417 (1976).
- [4612] Clark, J. P., and Green, J. C. The He-I photoelectron spectra of uranocene and thorocene, *J. Organometal. Chem.* **112**, C14 (1976).
- [4613] Chau, F. T., and McDowell, C. A. Photoelectron spectra of 1,2-dibromo-1,1-difluoroethane, 1,2-bromochloroethane, and 1,2-dichloro-, 1,2-dibromo-, and 1,2-diiodotetrafluoroethane, *J. Phys. Chem.* **80**, 2923 (1976).
- [4614] Koenig, T., Balle, T., and Snell, W. Helium(I) photoelectron spectra of organic radicals, *J. Am. Chem. Soc.* **97**, 662, (1975).
- [4615] Gardner, J. L., and Samson, J. A. R. Photoion and photoelectron spectroscopy of CO and N₂, *J. Chem. Phys.* **62**, 1447 (1975).
- [4616] Koenig, T., Smith, M., and Snell, W. The He(I) photoelectron spectrum of cyclopentadienone, *J. Am. Chem. Soc.* **99**, 6663 (1977).
- [4617] Derrick, P. J., Holmes, J. L., and Morgan, R. P. Kinetics and mechanisms of the loss of water from the cyclohexanol radical ion at times from 50 picoseconds to 10 microseconds following field ionization, *J. Am. Chem. Soc.* **97**, 4936 (1975).
- [4618] Clements, P. J., and Sale, F. R. A mass spectrometric study of nickel tetracarbonyl, iron pentacarbonyl and binary mixtures of these compounds, *Metall. Trans.* **7B**, 171 (1976).
- [4619] Brown, R. S. The influence of remote substituents upon ionization potential. Part I. The effect of two allylic oxygens upon the ionization of the π -bond, *Can. J. Chem.* **54**, 805 (1976).
- [4620] Distefano, G., Pignataro, S., Szepes, L., and Borossay, J. Photoelectron spectroscopy study of the triphenyl derivatives of the group IV elements, *J. Organometal. Chem.* **104**, 173 (1976).
- [4621] Behan, J. M., Johnstone, R. A. W., and Bentley, T. W. An evaluation of empirical methods for calculating the ionization potentials of substituted benzenes, *Organic Mass Spectrom.* **11**, 207 (1976).
- [4622] Arnold, D. E. J., and Rankin, D. W. H. Preparation and properties of diaminodifluorophosphorane, *J. Chem. Soc. Dalton* 1130 (1976).
- [4623] Debies, T. P., and Rabalais, J. W. Calculated photoionization cross-sections and angular distributions for the isoelectronic series Ne, HF, H₂O, NH₃, and CH₄, *J. Am. Chem. Soc.* **97**, 487 (1975).
- [4624] Ackermann, R. J., Rauh, E. G., and Thorn, R. J. The thermodynamics of ionization of gaseous oxides; the first ionization potentials of the lanthanide metals and monoxides, *J. Chem. Phys.* **65**, 1027 (1976).
- [4625] Bradshaw, D. I., Moyes, R. B., and Wells, P. B. Mass spectra of some deuterium-labelled methylsilanes and the analysis of mixtures, *Can. J. Chem.* **54**, 599 (1976).
- [4626] Fringuelli, F., Marino, G., Taticchi, A., Distefano, G., Colonna, F. P., and Pignataro, S. Photoelectron spectra of the α -substituted derivatives of furan, thiophen, selenophen, and tellurophen. A comparative study of the molecular orbital energies, *J. Chem. Soc. Perkin Trans. II*, 276 (1976).
- [4627] Colonna, F. P., Danieli, R., Distefano, G., and Ricci, A. The electronic effect of XMe₃ and CH₂XMe₃ (X=Si, Ge, Sn, or Pb) substituents in organometallic sulphides. A kinetic and photoelectron spectroscopic study, *J. Chem. Soc.*

- Perkin Trans. II, 306 (1976).
- [4628] Györosi, P., Hvistendahl, G., and Undheim, K. Mass spectrometry of some triphenylcyclopropenium salts. Competition between dimerisation and adduct formation, Org. Mass Spectrom. **10**, 744 (1975).
- [4629] Gardner, J. L., and Samson, J. A. R. Photoion and photoelectron spectroscopy of oxygen, J. Chem. Phys. **62**, 4460 (1975).
- [4630] Eland, J. H. D., Frey, R., Schulte, H., and Brehm, B. New results on the fragmentation of the benzene ion, Intern. J. Mass Spectrom. Ion Phys. **21**, 209 (1976).
- [4631] Yamazaki, T., and Kimura, K. He I photoelectron spectrum of dinitrogen tetraoxide (N_2O_4), Chem. Phys. Letters **43**, 502 (1976).
- [4632] Diemann, E., Varetti, E. L.; and Müller, A. The He(I) photoelectron spectra of the substituted permanganates MnO_3F and MnO_3Cl , Chem. Phys. Letters **51**, 460 (1977).
- [4633] Delwiche, J. P.; Praet, M.-Th.; Caprace, G.; Hubin-Franksin, M.-J.; Natalis, P.; and Collin, J. E. The photoelectron spectra of perfluoro-2- butyne and perfluorohexa-2,4-diyne. The perfluorination effect, J. Electron Spectrosc. Relat. Phenom. **12**, 395 (1977).
- [4634] Dyke, J., Jonathan, N., Lee, E., Morris, A., and Winter, M. Vacuum ultraviolet photoelectron spectroscopy of transient species: Part 8, the t-butyl radical, Physica Scripta **16**, 197 (1977).
- [4635] Peel, J. B., and Willett, G. D. The photoelectron spectra of arsenic tribromide and arsenic triiodide, J. Electron Spectrosc. Relat. Phenom. **9**, 175 (1976).
- [4636] Maier, J. P., and Swegart, D. A. Ultraviolet photoelectron spectra of nickel, palladium, and platinum diethyl dithiophosphate complexes, Inorg. Chem. **15**, 1989 (1976).
- [4637] Gleiter, R., Spanget-Larsen, J., Thulstrup, E. W., Murata, I., Nakasuiji, K., and Jutz, C. 155. The electronic structure of azuleno[1,2,3-cd] phenalene and azuleno[5,6,7-cd]phenalene, a comparison, Helv. Chim. Acta **59**, 1459 (1976).
- [4638] Müller, C., Schäfer, W., Schweig, A., Thon, N., and Vermeer, H. Detection of rotational isomers by variable temperature photoelectron spectroscopy. A new technique in the realm of molecular conformational analysis, J. Am. Chem. Soc. **98**, 5440 (1976).
- [4639] Diemann, E. The orbital electronic structure of the As_4O_6 molecule by photoelectron spectroscopy, Inorg. Chim. Acta **24**, L27 (1977).
- [4640] Tsai, B. P., Baer, T., Werner, A. S., and Lin, S. F. A photoelectron-photoion coincidence study of the ionization and fragment appearance potentials of bromo- and iodomethanes, J. Phys. Chem. **79**, 570 (1975).
- [4641] Klasinc, L. Application of photoelectron spectroscopy to biologically active molecules and their constituent parts, J. Electron Spectrosc. Relat. Phenom. **8**, 161 (1976).
- [4642] Williams, T. A., and Potts, A. W. Complexities in the He I photoelectron spectra of alkali metal vapours, J. Electron Spectrosc. Relat. Phenom. **8**, 331 (1976).
- [4643] Potts, A. W., and Williams, T. A. The He I photoelectron spectrum of TeO , Chem. Phys. Letters **42**, 550 (1976).
- [4644] Peng, S., Padva, A., and LeBreton, P. R. Ultraviolet photoelectron studies of biological purines: The valence electronic structure of adenine, Proc. Nat. Acad. Sci. U.S.A. **73**, 2966 (1976).
- [4645] Pullen, B. P., and Stockdale, J. A. D. Dissociative ionization of SF_6 by electron impact, Intern. J. Mass Spectrom. Ion Phys. **19**, 35 (1976).
- [4646] Popkie, H. E., Koski, W. S., and Kaufman, J. J. Ab-Initio LCAO-MO-SCF calculations of morphine and nalorphine and measurement of their photoelectron spectra, J. Am. Chem. Soc. **98**, 1342 (1976).
- [4647] White, M. G., Colton, R. J., Lee, T. H., and Rabalais, J. W. Electronic structure of N,N-dimethylnitramine and N,N-dimethylnitrosamine from X-ray and uv electron spectroscopy, Chem. Phys. **8**, 391 (1975).
- [4648] McGlynn, S. P., and Meeks, J. L. Photoelectron spectra of carbonyls, carbonates, oxalates and esterification effects, J. Electron Spectrosc. Relat. Phenom. **8**, 85 (1976).
- [4649] Young, V. Y., and Cheng, K. L. The ultraviolet photoelectron spectra of aliphatic and aromatic isonitriles, J. Electron Spectrosc. Relat. Phenom. **9**, 317 (1976).
- [4650] Dargelos, A., Sandorfy, C. The photoelectron and far-ultraviolet absorption spectra of simple oximes, J. Chem. Phys. **67**, 3011 (1977).
- [4651] Domelsmith, L. N., Houk, K. N., Timberlake, J. W., and Szilagyi, S. The photoelectron spectrum of tetramethyldiazetine: The elucidation of ring size effects on azo group ionization potentials, Chem. Phys. Letters **48**, 471 (1977).
- [4652] Wirz, J. 173. Electronic structure and photophysical properties of planar conjugated hydrocarbons with 4n-membered rings. I. Photoelectron spectra of 1,5,9-tridehydro[12]-annulene and related compounds, Helv. Chim. Acta **59**, 1647 (1976).
- [4653] Wittel, K., Felps, W. S., Klasinc, L., and McGlynn, S. P. Molecular Rydberg transitions. VI. trans-di-bromoethylene. The relation between vacuum ultraviolet and photoelectron spectroscopy, J. Chem. Phys. **65**, 3698 (1976).
- [4654] Domelsmith, L. N., and Houk, K. N. Photoelectron spectra of cyclopentanone and cyclohexanone enamines, Tetrahedron Letters **23**, 1981 (1977).
- [4655] Guyon, P.-M., Spohr, R., Chupka, W. A., and Berkowitz, J. Threshold photoelectron spectra of HF, DF, and F_2 , J. Chem. Phys. **65**, 1650 (1976).
- [4656] Van Veen, E. H. Triplet $\pi \rightarrow \pi^*$ transitions in thiophene, furan and pyrrole by low-energy electron-impact spectroscopy, Chem. Phys. Letters **41**, 535 (1976).
- [4657] Dyke, J. M., Morris, A., and Trickl, I. R. Characterization of the ground ionic state of the NS molecule using photoelectron spectroscopy, J. Chem. Soc. Faraday Trans. II, **73**, 147 (1977).
- [4658] Van Veen, E. H., and Plantenga, F. L. Low-energy electron-impact excitation spectra of acetylene, Chem. Phys. Letters **38**, 493 (1976).
- [4659] Gleiter, R., Kobayashi, M., and Kuthan, J. The n-orbital sequence in 1,3 diazadamantane, Tetrahedron **32**, 2775 (1976).
- [4660] Compernolle, F., and DeSchryver, F. Nonconcerted cycloreversion by electron impact of substituted 2,4-azetidinediones. Relations between the mass spectral and photochemical reactions, J. Am. Chem. Soc. **97**, 3909 (1975).
- [4661] Efraty, A., Huang, M. H. A., and Weston, C. A. Mass spectra of organometallic compounds. V. Electron-impact study of the cyclopentadienylmanganese thiocarbonyl and carbonyl complexes $RC_5H_4Mn(CO)_2CX$ [$R=H, CH_3$; $X=S, O$], Inorg. Chem. **14**, 2796 (1975).
- [4662] Streets, D. G., and Berkowitz, J. Photoelectron spectroscopy of Se_2 and Te_2 , J. Electron Spectrosc. Relat. Phenom. **9**, 269 (1976).
- [4663] Shevchenko, V. E., Iljin, M. K., Nikitin, O. T., and Sidorov, L. N. Mass-spectrometric study of mixed dimers M_2BO_2F , Intern. J. Mass Spectrom. Ion Phys. **21**, 279 (1976).
- [4664] Innorta, G., Torroni, S., Distefano, G., Pietropaolo, D., and Ricci, A. Mass spectra and energetic data for the formation of the $[M-CH_3]^+$ ion of 10,10-dimethylphenylthio-silyl, -germyl and -stannyl derivatives, and of 9,9-dimethylthioxantene, Org. Mass Spectrom. **12**, 766 (1977).
- [4665] Pfeffer, H.-U., Klessinger, M., Erker, G., and Roth, W. R.

- Photoelektronenspektren organischer Verbindungen, VIII
7,8-Dimethylenbicyclo [2.2.2]octa-2,5-dien und verwandte Verbindungen, Chem. Ber. **108**, 2923 (1975).
- [4666] Thorstad, O., Undheim, K., Cederlund, B., and Hörnfeldt, A.-B. Ionisation potentials in tautomeric analysis of 2-hydroxy derivatives of thiophenes, selenophenes, and furans, Acta Chem. Scand. **B29**, 647 (1975).
- [4667] Domelsmith, L. N., Munchausen, L. L., and Houk, K. N. Photoelectron spectra of psychotropic drugs. 2. Phenothiazine and related tranquilizers, J. Am. Chem. Soc. **99**, 6506 (1977).
- [4668] Gilbert, W. C., Taylor, L. T., and Dillard, J. G. Mass spectrometric study of polydentate Schiff base coordination compounds. I. Cobalt(II), nickel(II), and copper(II) complexes of salen and oaben, J. Am. Chem. Soc. **95**, 2477 (1973).
- [4669] Wiberg, K. B., Ellison, G. B., Wendoloski, J. J., Brundle, C. R., and Kuebler, N. A. Electronic states of organic molecules. 3. Photoelectron spectra of cycloalkenes and methylenecycloalkanes, J. Am. Chem. Soc. **98**, 7179 (1976).
- [4670] Dehmer, P. M., and Dehmer, J. L. Photoelectron spectrum of the Xe₂ van der Waals molecule, J. Chem. Phys. **67**, 1774 (1977).
- [4671] Van Veen, E. H. Low-energy electron-impact spectroscopy on ethylene, Chem. Phys. Letters **41**, 540 (1976).
- [4672] Domelsmith, L. N., Munchausen, L. L., and Houk, K. N. Photoelectron spectra of psychotropic drugs. 1. Phenethylamines, tryptamines, and LSD, J. Am. Chem. Soc. **99**, 4311 (1977).
- [4673] Thorstad, O., Undheim, K., Lantz, R., and Hörnfeldt, A.-B. Ionisation potentials in tautomeric analysis of 3-hydroxy derivatives of thiophenes, selenophenes, and furans, Acta Chem. Scand. **B29**, 652 (1975).
- [4674] Bastide, J., Maier, J. P., and Kubota, T. Ionisation energies and electronic structures of the phenyl 1,3-dipoles, J. Electron Spectrosc. Relat. Phenom. **9**, 307 (1976).
- [4675] Vilesov, F. I., and Lopatin, S. N. Photoelectron spectrometer, Zh. Tekh. Fiz. **42**, 176 (1972).
- [4676] Bieri, G. Cyanogen fluoride: A photoelectron-spectroscopic investigation, Chem. Phys. Letters **46**, 107 (1977).
- [4677] Thorstad, O., Undheim, K., and El-Gendy, M. A. F. Ionisation potentials in structure analysis of isomeric nitrones, oxaziranes, O-ether oximes and acid amides, Org. Mass Spectrom. **10**, 1155 (1975).
- [4678] Smoes, S., Drowart, J., and Myers, C. E. Determination of the atomization energies of the molecules TaO(g) and TaO₂(g) by the mass-spectrometric Knudsen-cell method, J. Chem. Thermodyn. **8**, 225 (1976).
- [4679] Staley, R. H., Kleckner, J. E., and Beauchamp, J. L. Relationship between orbital ionization energies and molecular properties. Proton affinities and photoelectron spectra of nitriles, J. Am. Chem. Soc. **98**, 2081 (1976).
- [4680] Solouki, B., Rosmus, P., and Bock, H. Unstable intermediates. 4. Thioformaldehyde, J. Am. Chem. Soc. **98**, 6054 (1976).
- [4681] Bieri, G., Heilbronner, E., Stadelmann, J.-P., Vogt, J., and von Niessen, W. Electronic states of difluoroacetylene, difluorodiacetylene, and perfluoropentadiyne-1,3 radical cations. A photoelectron spectroscopic investigation, J. Am. Chem. Soc. **99**, 6832 (1977).
- [4682] Drowart, J., and Smoes, S. Determination by the mass spectrometric Knudsen cell method and discussion of the dissociation energies of the molecules Se₂(g), SSe(g) and SeTe(g), J. Chem. Soc. Faraday Trans. II **73**, 1755 (1977).
- [4683] Block, T. F., Biernbaum, M., and West, R. Cyclic polysilanes XII. Photoelectron spectra and bonding in 1,2,3,4-tetra-t-butyltetramethylcyclotetrasilane and related t-butylsilicon compounds, J. Organometal. Chem. **131**, 199 (1977).
- [4684] Botter, R., Gounelle, Y., Henry, Y., Jullien, J., Menes, F., and Solgadi, D. Photoelectron spectra of halogeno-3-propynes (XCH₂=CH) and halogeno-methylcyanides (XCH₂C≡N) and comments on the attribution of fluorine IP, J. Electron Spectrosc. Relat. Phenom. **10**, 393 (1977).
- [4685] Bulgin, D. K., Dyke, J. M., and Morris, A. Vacuum ultraviolet photoelectron spectrum of the PN(X¹Σ⁺) molecule, J. Chem. Soc. Faraday II **73**, 983 (1977).
- [4686] Bieri, G., Dill, J. D., Heilbronner, E., and Maier, J. P. 68. The electronic states of the pentatetraene radical cation, Helv. Chim. Acta **60**, 629 (1977).
- [4687] Paule, R. C. Mass spectrometric studies of Al₂O₃ vaporization processes, High Temp. Sci. **8**, 257 (1976).
- [4688] Schmidt, H., Schweig, A., Anastassiou, A. G., and Wetzel, J. C. The dominant role of hyperconjugation in the 9-oxabicyclo[4.2.1] nona-2,4,7-triene series, Tetrahedron **32**, 2239 (1976).
- [4689] Rothgery, E. F., Holt, R. J., and McGee, H. A., Jr. Cryochemical synthesis and molecular energetics of cyclopropanone and some related compounds, J. Am. Chem. Soc. **97**, 4971 (1975).
- [4690] Božić, Z., Humski, K., Cvitaš, T., and Klasinc, L. Photoelectron spectra of bromo- and iodo-thiophens, J. Chem. Soc. Perkin Trans. II, 1413 (1977).
- [4691] Bünzli, J.-C. G., Olsen, H., and Snyder, J. P. Photoelectron spectra of bicyclic azo N-oxides and azo N,N'-dioxides, J. Org. Chem. **42**, 614 (1977).
- [4692] Block, T. F., and Fenske, R. F. A photoelectron spectroscopic study of some pentacarbonylchromium carbene complexes, J. Am. Chem. Soc. **99**, 4321 (1977).
- [4693] Bussières, N., and Marmet, P. Ionization and dissociative ionization of CO₂ by electron impact, Can. J. Phys. **55**, 1889 (1977).
- [4694] Egdell, R. G., and Orchard, A. F. Photoelectron spectra of the group IVA halides, J. Chem. Soc. Faraday Trans. II **74**, 485 (1978).
- [4695] Ashmore, F. S., and Burgess, A. R. Photoelectron spectra of the unbranched C₅-C₇ alkenes, aldehydes and ketones, J. Chem. Soc. Faraday Trans. II **74**, 734 (1978).
- [4696] Frost, D. C., McDowell, C. A., Pouzard, G., and Westwood, N. P. C. The photoelectron spectra of the oxalyl halides (COX)₂, [X=F, Cl and Br], J. Electron Spectrosc. Relat. Phenom. **10**, 273 (1977).
- [4697] Causley, G. C., Clark, J. B., and Russell, B. R. The vacuum ultraviolet spectrum of bromosilane, Chem. Phys. Letters **38**, 602 (1976).
- [4698] Bock, H., Solouki, B., Bert, G., and Rosmus, P. Unstable Intermediates. 5. ¹Thioketene, J. Am. Chem. Soc. **99**, 1663 (1977).
- [4699] Furlani, C., and Andreucci, M. V. Valence-shell photoionization spectra of some methyl dihalogenophosphates, J. Chem. Soc. Dalton Trans. **7**, 673 (1977).
- [4700] Frost, D. C., McDowell, C. A., and Westwood, N. P. C. The photoelectron spectrum of dinitrogen tetroxide, J. Electron Spectrosc. Relat. Phenom. **10**, 293 (1977).
- [4701] Clar, E., and Schmidt, W. Correlations between photoelectron and phosphorescence spectra of polycyclic hydrocarbons, Tetrahedron **32**, 2563 (1976).
- [4702] Bieri, G., Burger, F., Heilbronner, E., and Maier, J. P. 223. Valence ionization energies of hydrocarbons, Helv. Chim. Acta **60**, 2213 (1977).
- [4703] Brown, R. S., and Marcinko, R. W. Application of photoelectron spectroscopy to intramolecular hydrogen bonding. 6. The relative importance of electrostatic and covalent contributions to the H bond of H-bonded alcohols containing a conjugated olefin as the electron donor, J. Am. Chem. Soc. **99**, 6500 (1977).
- [4704] Cannington, P. H., and Whitfield, H. J. Photoelectron spectra

- of $\text{As}_3\text{S}_3\text{P}_2\text{S}_3\text{P}_2\text{Se}_3$, and As_3O_6 , J. Electron Spectrosc. Relat. Phenom. **10**, 35 (1977).
- [4705] Cowley, A. H., Lattman, M., Montag, R. A., and Verkade, J. G. The coordination behavior of acyclic phosphites; a UV photoelectron spectroscopic study, Inorg. Chim. Acta **25**, L151 (1977).
- [4706] Cederlund, B., Lantz, R., Hörfeldt, A.-B., Thorstad, O., and Undheim, K. Preparation of some thiophene-, selenophene-and furan thiols. Ionisation potentials in tautomer analysis, Acta Chem. Scand. Ser. B **31**, 198 (1977).
- [4707] Gleiter, R., Kobayashi, M., Neunhoeffer, H., and Spanget-Larsen, J. Photoelectron spectra of 1,2,4-triazine and some methyl derivatives, Chem. Phys. Letters **46**, 231 (1977).
- [4708] Galasso, V., Colonna, F. P., and Distefano, G. Photoelectron spectra of 1,2-indandione, 1,3-indandione and heterocyclic analogues, J. Electron Spectrosc. Relat. Phenom. **10**, 227 (1977).
- [4709] Gan, T. H., Peel, J. B., and Willett, G. D. Reinterpretation of the photoelectron spectrum of dinitrogen tetroxide, J. Chem. Soc. Faraday Trans. II **73**, 1459 (1977).
- [4710] Cauletti, C., Duffy, N. V., and Furlani, C. L. UV Photoelectron spectra of some substituted iron dithiocarbamates, Inorg. Chim. Acta **23**, 181 (1977).
- [4711] Cook, M. J., El-Abbad, S., Katritzky, A. R., Guimon, C., and Pfister-Guillouzo, G. Photoelectron spectra of hydroxy- and mercapto-pyridines and models of fixed structure, J. Chem. Soc. Perkin II, 1652 (1977).
- [4712] Clar, E., and Schmidt, W. Correlations between photoelectron and ultraviolet absorption spectra of polycyclic hydrocarbons. The perylene, coronene and bisanthene series, Tetrahedron, **33**, 2093 (1977).
- [4713] Egdell, R. G., and Orchard, A. F. He(II) Photoelectron spectra of indium(I) and thallium(I) halides, J. Chem. Soc. Faraday Trans. II **74**, 1179 (1978).
- [4714] Reeher, J. R., Flesch, G. D., and Svec, H. J. The mass spectra and ionization potentials of the neutral fragments produced during the electron bombardment of aromatic compounds, Org. Mass. Spectrom. **11**, 154 (1976).
- [4715] Ensslin, W., Schmidtko, H.-H., and Kühn, Th. Photoelectron spectra and electronic structure of symmetrically *trans*-substituted disilyethylenes, Inorg. Chim. Acta **24**, 159 (1977).
- [4716] Harman, P. J., Kent, J. E., Gan, T. H., Peel, J. B., and Willett, G. D. The photoelectron spectrum of benzvalene, J. Am. Chem. Soc. **99**, 943 (1977).
- [4718] Frost, D. C., LeGeyt, M. R., Paddock, N. L., and Westwood, N. P. C. Helium I photoelectron spectrum of disulphur dinitride, J. Chem. Soc. Chem. Commun. 217 (1977).
- [4719] Houk, K. N., Caramella, P., Munchausen, L. L., Chang, Y.-M., Battaglia, A., Sims, J., and Kaufman, D. C. Photoelectron spectra of nitrones and nitrile oxides, J. Electron Spectrosc. Relat. Phenom. **10**, 441 (1977).
- [4720] Head, R. A., Nixon, J. F., and Clark, R. J. UV photoelectron spectra of first-row transition metal hydridocarbonyl and hydridotrifluorophosphine complexes, J. Organometal. Chem. **135**, 209 (1977).
- [4721] Gan, T. H., Peel, J. B., and Willett, G. D. A comparison of the photoelectron spectra of the trimethyl- and triallyl-orthoformates, Chem. Phys. Letters **51**, 464 (1977).
- [4722] Houle, F. A., and Beauchamp, J. L. Detection and investigation of allyl and benzyl radicals by photoelectron spectroscopy, J. Am. Chem. Soc. **100**, 3290 (1978).
- [4723] Gleiter, R., Bischof, P., Volz, W. E., and Paquette, L. A. Conjugative interaction between II and cyclobutane orbitals. The synthesis and electronic structure of bicyclo[4.1.1]octa-2,4-diene, J. Am. Chem. Soc. **99**, 8 (1977).
- [4724] Gower, M., Kane-Maguire, L. A. P., Maier, J. P., and Sweigart, D. A. Ultraviolet photoelectron spectra of cyclohepta-1,3,5-triene and mesitylene tricarbonyl complexes of the group 6A metals, J. Chem. Soc. Dalton **316** (1977).
- [4725] Harris, D. H., Lappert, M. F., Pedley, J. B., and Sharp, G. J. Bonding studies of compounds of group 3-5 elements. Part XVIII. He(I) photoelectron spectra of bivalent homoleptic alkyls and amides, especially of group 4 elements, and of tin(II) chloride and bromide, J. Chem. Soc. Dalton II, 945 (1976).
- [4726] Bischof, P., Eaton, P. E., Gleiter, R., Heilbronner, E., Jones, T. B., Musso, H., Schmelzer, A., and Stober, R. 44. The electronic structure of cubane (C_8H_8) as revealed by photoelectron spectroscopy, Helv. Chim. Acta **61**, 547 (1978).
- [4727] Cvitaš, T., Güsten, H., Klasinc, L., Novadj, I., and Vančík, H. Photoelectron spectra of bromo- and iodotrifluoromethane, Z. Naturforsch. **33a**, 1528 (1978).
- [4728] Bally, T., Buser, U., and Haselbach, E. I. Tetrakis (methylidene)cyclobutane('4]radialene'): Electronic states of the molecular ion, Helv. Chim. Acta **61**, 38 (1978).
- [4729] Holmes, J. L., Terlouw, J. K., and Lossing, F. P. The thermochemistry of $\text{C}_2\text{H}_5\text{O}^+$ ions, J. Phys. Chem. **80**, 2860 (1976).
- [4730] Gan, T. H., Peel, J. B., and Willett, G. D. Comparison of He I and He II photoelectron spectra of phosphoryl chloride, Chem. Phys. Letters **48**, 483 (1977).
- [4731] Heilbronner, E., Jones, T. B., and Maier, J. P. 170. The ionization energies of di-*n*-alkyl diacetylenes, Helv. Chim. Acta **60**, 1697 (1977).
- [4732] Gan, T. H., Peel, J. B., and Willett, G. D. Photoelectron spectra of the *gauche* and *trans* conformers of 1,2-dichloroethane, J. Chem. Soc. Faraday Trans. II **73**, 965 (1977).
- [4733] Green, J. C., Lloyd, D. R., Galyer, L., Mertis, K., and Wilkinson, G. Photoelectron spectra of some transition metal alkyls and oxoalkyls, J. Chem. Soc. Dalton Trans. **10**, 1403 (1978).
- [4734] Basso-Bert, M., Cassoux, P., Crasnier, F., Gervais, D., Labarre, J.-F., and DeLoth, P. Molecular orbitals and photoelectron spectra of some titanium(IV) organometallic compounds, J. Organometal. Chem. **136**, 201 (1977).
- [4735] Bewick, A., Edwards, C. J., Jones, S. R., and Mellor, J. M. The electrochemical difunctionalisation of saturated hydrocarbons, Tetrahedron Letters 631 (1976).
- [4736] Conard, B. R., and Sridhar, R. Appearance potentials of ion fragments of iron pentacarbonyl, Can. J. Chem. **56**, 2607 (1978).
- [4737] Colbourne, D., Frost, D. C., McDowell, C. A., and Westwood, N. P. C. The photoelectron spectra of the chloramines NH_2Cl , NHCl_2 , NCl_3 and the methyl chloramines CH_3NHCl , CH_3NCl_2 , and $(\text{CH}_3)_2\text{NCl}$, J. Chem. Phys. **69**, 1078 (1978).
- [4738] Basch, H., Bieri, G., Heilbronner, E., and Jones, T. B. 2. The photoelectron spectrum of tetrafluorobutatriene, Helv. Chim. Acta **61**, 46 (1978).
- [4739] Behan, J., Johnstone, R. A. W., and Puddephatt, R. J. Photoelectron spectra and reactivity of methyl (tertiary phosphine)-platinum and -gold complexes, J. Chem. Soc. Chem. Commun. 444 (1978).
- [4740] Bloch, M., Brogli, F., Heilbronner, E., Jones, T. B., Prinzbach, H., and Schweikert, O. 138. Photoelectron spectra of unsaturated oxides. I. 1,4-dioxin and related systems, Helv. Chim. Acta **61**, 1388 (1978).
- [4741] Nagy-Felsobuki, E., and Peel, J. B. Photoelectron spectra of methylchloramine and methyldichloramine, J. Chem. Soc. Faraday Trans. II **74**, 1927 (1978).
- [4742] Gerson, S. H., Worley, S. D., Bodor, N., Kaminski, J. J., and Flechtner, T. W. The photoelectron spectra of some

- heterocyclic compounds which contain N, O, Cl, and Br, *J. Electron Spectrosc. Relat. Phenom.* **13**, 421 (1978).
- [4743] Colonna, F. P., Distefano, G., Galasso, V., Irgolic, K. J., King, C. E., and Pappalardo, G. C. The conformation, uv-absorption spectra and photoelectron spectra of phenoxyachalcogenins, *J. Organometal. Chem.* **146**, 235 (1978).
- [4744] Eck, V., Schweig, A., and Vermeer, H. The ultraviolet photoelectron spectrum of o-benzoquinone methide, *Tetrahedron Letters* **27**, 2433 (1978).
- [4745] Ehlert, T. C. Mass spectrometric investigations of the oxides of potassium, *High Temp. Sci.* **9**, 237 (1977).
- [4746] Frost, D. C., Kroto, H. W., McDowell, C. A., and Westwood, N. P. C. The HeI photoelectron spectra of the isoelectronic molecules, cyanogen azide, NCN_3 , and cyanogen isocyanate, NCNCO , *J. Electron Spectrosc. Relat. Phenom.* **11**, 147 (1977).
- [4747] McAluff, E. J., and Houk, K. N. Photoelectron spectra of substituted oxiranes and thiiranes. Substituent effects on ionization potentials involving σ orbitals, *Can. J. Chem.* **55**, 318 (1977).
- [4748] Kröner, J., Kosbahn, W., and Runge, W. The molecular structure of allenes and ketenes, IX [1] π and σ interactions in allenes: Photoelectron spectra, electronic absorption spectra, and quantumchemical calculations, *Ber. Bunsenges.* **81**, 826 (1977).
- [4749] Dehmer, P. M., and Dehmer, J. L. Photoelectron spectrum of Xe_2 and potential energy curves for Xe_2^+ , *J. Chem. Phys.* **68**, 3462 (1978).
- [4750] Cavell, R. G., and Allison, D. A. Photoelectron spectra of acetylene with $\text{He}^1\text{He}^{11}\text{Zr}$ M, and MgK_{α} radiation sources, *J. Chem. Phys.* **69**, 159 (1978).
- [4751] Carnovale, F., Gan, T. H., and Peel, J. B. Photoelectron spectra of the *gauche* and *trans* conformers of 1,2-bromochloroethane, *J. Electron Spectrosc. Relat. Phenom.* **16**, 87 (1979).
- [4752] Berkowitz, J., and Eland, J. H. D. Photoionization of N_2O : Mechanisms of photoionization and ion dissociation, *J. Chem. Phys.* **67**, 2740 (1977).
- [4753] Head, R. A., Nixon, J. F., Westwood, N. P. C., and Clark, R. J. He(I) photoelectron spectra of mixed carbonyl trifluorophosphine complexes of zero-valent iron, *J. Organometal. Chem.* **145**, 75 (1978).
- [4754] Lauer, G., Schäfer, W., and Schweig, A. Functional subunits in the nucleic acid bases uracil and thymine, *Tetrahedron Letters* **45**, 3939 (1975).
- [4755] Colbourn, E. A., Dyke, J. M., Fayad, N. K., and Morris, A. The He(I) photoelectron spectra of BrF and IF , *J. Electron Spectrosc. Relat. Phenom.* **14**, 443 (1978).
- [4756] Kobayashi, M., Gleiter, R., Coffen, D. L., Bock, H., Schulz, W., and Stein, U. Spiroconjugation in orthothiocarbonates, *Tetrahedron* **33**, 433 (1977).
- [4757] Ajello, J. M., Huntress, W. T., Jr., and Rayermann, P. A photoionization mass spectrometer study of CFCl_3 , CF_2Cl_2 and CF_3Cl , *J. Chem. Phys.* **64**, 4746 (1976).
- [4758] Domelsmith, L. N., and Houk, K. N. Photoelectron spectroscopic studies of hallucinogens: The use of ionization potentials in QSAR, *NIDA Res. Monogr.* **22**, 423 (1978).
- [4759] Baldwin, M. A., Loudon, A. G., Webb, K. S., and Cardnell, P. C. Charge location and fragmentation under electron impact, *Org. Mass Spectrom.* **12**, 279 (1977).
- [4760] Colbourn, E. A., Dyke, J. M., Fackerell, A., Morris, A., and Trickle, I. R. Vacuum ultraviolet photoelectron spectrum of the $\text{GeO}(\text{X}\Sigma^+)$ molecule, *J. Chem. Soc. Faraday Trans. II* **74**, 2278 (1978).
- [4761] Hammer, C. A., Allen, J. D. Jr., Cusachs, L. C., and Schweitzer, G. K. The high-temperature photoelectron spectra of alkaline-earth chlorides and iodides, *J. Electron Spectrosc. Relat. Phenom.* **13**, 149 (1978).
- [4762] Berkowitz, J., Eland, J. H. D., and Appelman, E. H. Photoionization mass spectrometry and heat of formation of S_2O , *J. Chem. Phys.* **66**, 2183 (1977).
- [4763] Colbourne, D., Frost, D. C., McDowell, C. A., and Westwood, N. P. C. The photoelectron spectra of the isoelectronic molecules hypochlorous acid HOCl and chloramine NH_2Cl , *J. Chem. Phys.* **68**, 3574 (1978).
- [4764] Egglest, R. G., and Orchard, A. F. Photoelectron spectra of some transition metal pentachlorides, *J. Electron Spectrosc. Relat. Phenom.* **14**, 277 (1978).
- [4765] Bieri, G., Heilbronner, E., Hornung, V., Kloster-Jensen, E., Maier, J. P., Thommen, F., and von Niessen, W. Electronic states of substituted haloacetylene and cyanoacetylene radical cations, *Chem. Phys.* **36**, 1 (1979).
- [4766] Hemmersbach, P., Klessinger, M., and Bruckmann, P. Electronic structure of *exo*-dimethylenecycloalkanes, *J. Am. Chem. Soc.* **100**, 6344 (1978).
- [4767] Klessinger, M., and Gunkel, E. The electronic structure of polyenes and unsaturated carbonyl compounds, *Tetrahedron* **34**, 3591 (1978).
- [4768] Kimura, K., and Katsumata, S. Photoelectron spectroscopic study of hydroxylamine (NH_2OH) and *o*-methylhydroxylamine (NH_2OCH_3), *J. Chem. Phys.* **67**, 1225 (1977).
- [4769] Nagata, S., Yamabe, T., and Fukui, K. A study of the electronic spectra of thioacetic acid and its ethyl ester, *J. Phys. Chem.* **79**, 2335 (1975).
- [4770] MacNeil, K. A. G., and Dixon, R. N. High-resolution photoelectron spectroscopy of methanol and its deuterated derivatives: Internal rotation in the ground ionic state, *J. Electron Spectrosc. Relat. Phenom.* **11**, 315 (1977).
- [4771] Koenig, T., and Southworth, S. The He I photoelectron spectrum of 3,7-dimethyl-*p*-quinodimethane. A non-Koopmans theorem effect, *J. Am. Chem. Soc.* **99**, 2807 (1977).
- [4772] Gerson, S. H., Worley, S. D., Bodor, N., and Kaminski, J. J. Electronic structures of some antimicrobial *N*-chloramines. Possible existence of intramolecular hydrogen bonding and its effect on germicidal efficiency, *J. Med. Chem.* **21**, 686 (1978).
- [4773] Katsumata, S., and Lloyd, D. R. The photoelectron spectra of the OH and OD radicals, *Chem. Phys. Letters* **45**, 519 (1977).
- [4774] Carnovale, F., Livett, M. K., and Peel, J. B. The photoelectron spectrum of the dimethyl etherhydrogen chloride complex, *J. Am. Chem. Soc.* **102**, 569 (1980).
- [4775] Carnovale, F., Nagy-Felsobuki, E., Peel, J. B., and Willet, G. D. The photoelectron spectra of *N*-bromo-methylamine and *N,N*-dibromo-methylamine, *J. Electron Spectrosc. Relat. Phenom.* **14**, 163 (1978).
- [4776] Frost, D. C., MacDonald, B., McDowell, C. A., and Westwood, N. P. C. Pyrolysis of trimethyl hexahydro-s-triazines. The photoelectron spectra of *N*-methylmethylenimine, $\text{CH}_2=\text{NCH}_3$, and *C*-methyl-methylenimine, $\text{CH}_3\text{CH}=\text{NH}$, *J. Electron Spectrosc. Relat. Phenom.* **14**, 379 (1978).
- [4777] Egglest, R. G., Fragala, I., and Orchard, A. F. UV photoelectron spectra of the indium(I) and thallium(I) cyclopentadienides, *J. Electron Spectrosc. Relat. Phenom.* **14**, 467 (1978).
- [4778] Vonbach, P. S., Saltzburg, H., and Caesar, G. P. The photoelectron spectra of gaseous silver halides, *J. Electron Spectrosc. Relat. Phenom.* **8**, 359 (1976).
- [4779] Müller, C., Schweig, A., Thiel, W., Grahn, W., Bergman, R. G., and Vollhardt, K. P. C. Photoelectron spectra of 2,5-dehydrotropyldiene, 3,6-dehydrooxepin, and fulvenallene, *J. Am. Chem. Soc.* **101**, 5579 (1979).
- [4780] Nelsen, S. F., Peacock, V. E., Weisman, G. R., Landis, M. E., and Spencer, J. A. Conformations of fourmembered ring

- hydrazines and hydrazine radical cations, J. Am. Chem. Soc. **100**, 2806 (1978).
- [4781] Houk, K. N., Strozier, R. W., Santiago, C., Gandour, R. W., and Vollhardt, K. P. C. Electronic structure and photoelectron spectrum of 1,5,9-cyclododecatriyne, J. Am. Chem. Soc. **101**, 5183 (1979).
- [4782] Sandman, D. J., Ceasar, G. P., Nielsen, P., Epstein, A. J., and Holmes, T. J. Electronic structure of the π donor naphthalene 1,8-disulfide, J. Am. Chem. Soc. **100**, 202 (1978).
- [4803] Overman, L. E., Taylor, G. F., Houk, K. N., and Domelsmith, L. N. Diels-Alder reactions between *trans*-1-N-acylamino-1,3-dienes and methyl acrylate. A correlation between diene photoelectron ionization potentials and reactivity, stereoselectivity, and regioselectivity, J. Am. Chem. Soc. **100**, 3182 (1978).
- [4804] Koenig, T., Wielesek, R., Miller, L. L., and So, Y.-H. Correlation of electrochemical reactivity and photoelectron spectra of aromatic ketones, J. Am. Chem. Soc. **99**, 7061 (1977).
- [4805] Novak, I., Klasinc, L., and Knop, J. V. Photoelectron spectroscopy of heterocycles. Dipyridylethylenes, Z. Naturforsch **32a**, 886 (1977).
- [4806] Vick, D. O., Woodley, D. G., Bloor, J. E., Allen, J. D., Jr., Mui, T. C., and Schweitzer, G. K. The photoelectron spectra of gaseous alkali perrhenates, J. Electron Spectrosc. Relat. Phenom. **13**, 247 (1978).
- [4807] Stockbauer, R., and McCulloh, K. E., and Parr, A. C. The ionization potential of allene, Intern. J. Mass Spectrom. Ion Phys. **31**, 187 (1979).
- [4808] Schang, P., Gleiter, R., and Rieker, A. The He(I) photoelectron spectrum of cyclobutene-1,2-dione and o-benzoquinone, Ber. Bunsenges. Phys. Chem. **82**, 629 (1978).
- [4809] Day, J. S., Gowenlock, B. G., Johnson, C. A. F., McInally, I. D., and Pfab, J. Appearance potential studies of some geminal substituted C-nitroso- compounds, J. Chem. Soc. Perkin Trans. II **10**, 1110, (1978).
- [4810] Worley, S. D., Gerson, S. H., Bodor, N., Kaminski, J. J., and Flechtner, T. W. On the structure of N-chlorosuccinimide and N-bromosuccinimide. A photoelectron spectroscopic study, J. Chem. Phys. **68**, 1313 (1978).
- [4811] Ulman, J. A., Andersen, E. L., and Fehlner, T. P. Characterization of ferraboranes by ultraviolet photoelectron spectroscopy, J. Am. Chem. Soc. **100**, 456 (1978).
- [4812] Palmer, M. H., Leaver, D., Nisbet, J. D., Millar, R. W., and Eggell, R. The electronic structure of some heterocycles with bridgehead nitrogen: photoelectron spectra and ab initio molecular orbital calculations, J. Mol. Struct. **42**, 85 (1977).
- [4813] Streets, D. G., and Berkowitz, J. Electronic structures of fulvalene and octachlorofulvalene, Chem. Phys. **23**, 79 (1977).
- [4814] Vocelle, D., Dargelos, A., Pottier, R., and Sandorfy, C. Photoelectron and far-ultraviolet absorption spectra of nonaromatic azomethine compounds, J. Chem. Phys. **66**, 2860 (1977).
- [4815] Prins, I., Verhoeven, J. W., DeBoer, Th. J., and Worrell, C. Conformational studies on arylcyclopropanes-I, Tetrahedron **33**, 127 (1977).
- [4816] Heilbronner, E., Jones, T. B., Kloster-Jensen, E., and Maier, J. P. 191. Electronic states of di-*t*- butylpolyacetylene radical cations, Helv. Chim. Acta **61**, 2040 (1978).
- [4817] Potts, A. W. The valence electronic structure of SeO_2 studied by UPS and its relation to that of other group VIB dioxides, J. Electron Spectrosc. Relat. Phenom. **11**, 157 (1977).
- [4818] Peel, J. B., and Willett, G. D. Photoelectron spectroscopic studies of the propyl- and allyl- substituted amines, Aust. J. Chem. **30**, 2571 (1977).
- [4819] Domelsmith, L. N., and Houk, K. N. Photoelectron spectra of cyclopentanone and cyclohexanone enamines, Tetrahedron Letters **23**, 1981 (1977).
- [4820] Koenig, T., and Chang, J. C. Helium(I) photoelectron spectrum of tropyl radical, J. Am. Chem. Soc. **100**, 2240 (1978).
- [4821] Bastide, J., Hall, D., Heilbronner, E., and Maier, J. P. He(Ia) photoelectron spectra of some higher aromatic perfluoro compounds, J. Electron Spectrosc. Relat. Phenom. **16**, 205 (1979).
- [4822] Bancroft, G. M., Coatsworth, L. L., Creber, D. K., and Tse, J. High Resolution gas phase photoelectron spectra of core *d* levels using He II radiation, Physica Scripta **16**, 217 (1977).
- [4823] Hammer, C. A., Allen, J. D., Jr., Cusachs, L. C., and Schweitzer, G. K. The high-temperature photoelectron spectra of alkaline-earth chlorides and iodides, J. Electron Spectrosc. Relat. Phenom. **13**, 149 (1978).
- [4824] Reingold, I. D., Schmidt, W., and Boekelheide, V. Syntheses, properties, and photoelectron spectra of substituted and layered [2.2](2,6)pyridinoparacyclophanes, J. Am. Chem. Soc. **101**, 2121 (1979).
- [4825] Downs, A. J., Eggell, R. G., Orchard, A. F., and Thomas, P. D. P. Photoelectron spectra of metal tetrahydroborates, J. Chem. Soc. Dalton Trans. **12**, 1755 (1978).
- [4826] Evans, S., and Orchard, A. F. The vapour-phase ultra-violet photoelectron spectroscopy of metal halides at elevated temperatures, J. Electron Spectrosc. Relat. Phenom. **6**, 207 (1975).
- [4827] Solouki, B., Bock, H., and Appel, R. Photoelektronenspektren und Moleküleigenschaften, XLV Schwefelsäure-Derivate X_2SY_2 : Alkyl-, Vinyl- und Arylsulfone, Alkylsulfonimide und Sulfurylhalogenede, Chem. Ber. **108**, 897 (1975).
- [4828] Heilbronner, E., Hoshi, T., von Rosenberg, J. L., and Hafner, K. Alkyl-induced, natural hypsochromic shifts of the $^2\text{A} \leftarrow ^2\text{X}$ and $^2\text{B} \leftarrow ^2\text{X}$ transitions of azulene and naphthalene radical cations, Nouveau Journal de Chimie, **1**, 105 (1976).
- [4829] Delwiche, J., Praet, M.-Th., Caprace, G., Franskin-Hubin, M.-J., Natalis, P., and Collin, J. E. The He(I) photoelectron spectra of methyl-substituted 1,2,4-pentatrienes (vinylallene), J. Electron Spectrosc. Relat. Phenom. **16**, 35 (1979).
- [4830] Morishima, I., Yoshikawa, K., Hashimoto, M., and Bekki, K. Homoallylic interaction between the nitrogen lone pair and the nonadjacent π bond in cyclic and bicyclic amines. I. Photoelectron spectroscopic study, J. Am. Chem. Soc. **97**, 4283 (1975).
- [4831] Benoit, F. M., Harrison, A. G., and Lossing, F. P. Hydrogen migrations in mass spectrometry III- Energetics of formation of $[\text{R}'\text{CO}_2\text{H}_2]^+$ in the mass spectra of $\text{R}'\text{CO}_2\text{R}$, Organic Mass Spectrom. **12**, 78 (1977).
- [4832] Christoph, G. G., Muthard, J. L., Paquette, L. A., Böhm, M. C., and Gleiter, R. Quantitative assessment of pp- σ overlap in a topologically convex triene. Electronic and crystal structure analysis of C_{16} -hexaquinacene, J. Am. Chem. Soc. **100**, 7782 (1978).
- [4833] Domelsmith, L. N., Houk, K. N., Piedrahita, C., and Dolbier, W. J., Jr. The photoelectron spectrum of 1,1-difluoroallene. On π electron donation and withdrawal by fluorine, J. Am. Chem. Soc. **100**, 6908 (1978).
- [4834] Baldwin, M. A., Loudon, A. G., MacColl, A., and Webb, K. S. The nature and fragmentation pathways of the molecular ions of some arylureas, arylthioureas, acetanilides, thioacetanilides and related compounds, Org. Mass Spectrom. **11**, 1181 (1976).
- [4835] Ku, A. Y., Paquette, L. A., Rozeboom, M. D., and Houk, K. N. Polar effects on di- π -methane rearrangements.

- Regiospecificity in the triplet-sensitized photoisomerizations of 2-cyanobenzonorbornadienes carrying methoxy aryl substituents, *J. Am. Chem. Soc.* **101**, 5981 (1979).
- [4836] Kroto, H. W., Nixon, J. F., Simmons, N. P. C., and Westwood, N. P. C. FC≡P, C-fluorophosphaethyne: Preparation and detection by photoelectron and microwave spectroscopy, *J. Am. Chem. Soc.* **100**, 446 (1978).
- [4837] Poole, R. T., Nicholson, J. A., Jenkin, J. G., Leckey, R.C.G., Peel, J. B., and Liesegang, J. Electronic structure of the valence bands of SnCl_2 and SnBr_2 studied by ultraviolet photoelectron spectroscopy, *J. Electron Spectrosc. Relat. Phenom.* **15**, 91 (1979).
- [4838] Müller, C., Schweig, A., Cava, M. P., and Lakshminathan, M. V. Thieno[3,4-c]thiophenes. Electronic structure, *J. Am. Chem. Soc.* **98**, 7187 (1976).
- [4839] Potts, A. W., and Lyus, M. L. The photoelectron spectrum and valence shell structure of $(\text{CuX})_3$ and $(\text{AgCl})_3$, *J. Electron Spectrosc. Relat. Phenom.* **13**, 305 (1978).
- [4840] Bowling, R. A., Allen, J. D., Jr., and Schweitzer, G. K. The He(I) photoelectron spectra of gaseous alkali metaphosphates, *J. Electron Spectrosc. Relat. Phenom.* **17**, 25 (1979).
- [4841] Colonna, F. P., Distefano, G., and Galasso, V. Photoelectron spectra of 1,4-dithiin and related compounds, *J. Electron Spectrosc. Relat. Phenom.* **18**, 75 (1980).
- [4842] Brown, R. S., and Marcinko, R. W. Influence of substituents upon ionization potential. Dependence of the π -ionization energy on the orientation of an allylic hydroxyl or methoxyl substituent, *J. Am. Chem. Soc.* **100**, 5721 (1978).
- [4843] Kibel, M. H., and Nyberg, G. L. Angular distribution valence photoelectron spectra of nitric oxide, *J. Electron Spectrosc. Relat. Phenom.* **17**, 1 (1979).
- [4844] McAllduff, E. J., and Bunbury, D. L. Photoelectron spectra of some aromatic mono-and di-ketones, *J. Electron Spectrosc. Relat. Phenom.* **17**, 81 (1979).
- [4845] Campbell, M. J., Liesegang, J., Riley, J. D., Leckey, R. C. G., Jenkin, J. G., and Poole, R. T. The electronic structure of the valence bands of solid NH_3 and H_2O studied by ultraviolet photoelectron spectroscopy, *J. Electron Spectrosc. Relat. Phenom.* **15**, 83 (1979).
- [4846] Jones, T. B., and Maier, J. P. Study of the radical cation of all *trans*-1,3,5,7-octatetraene by its emission, $\text{A}^2\text{A}_{\text{u}} \rightarrow \text{X}^2\text{B}_{\text{g}}$, and by photoelectron spectroscopy, *Intern. J. Mass Spectrom. Ion Phys.* **31**, 287 (1979).
- [4847] Andreucci, M. V., Bitchev, P., Carusi, P., and Furlani, A. Valence shell photoionization spectra of some substituted hydroxy-acetylenes. A tentative correlation with their cyclotrimerization reactions, *J. Electron Spectrosc. Relat. Phenom.* **16**, 25 (1979).
- [4848] Bernardi, F., Danieli, R., Distefano, G., Modelli, A., and Ricci, A. Photoelectron spectra of substituted dialkylphenylthioboranes, *J. Organometal. Chem.* **136**, 161 (1977).
- [4849] Bartetzko, R., Gleiter, R., Muthard, J. L., and Paquette, L. A. Long range electronic transmission in conformationally rigid α -diketones, *J. Am. Chem. Soc.* **100**, 5589 (1978).
- [4850] Benoit, F. M., and Harrison, A. G. Predictive value of proton affinity. Ionization energy correlations involving oxygenated molecules, *J. Am. Chem. Soc.* **99**, 3980 (1977).
- [4851] Dougherty, D., Blankespoor, R. L., and McGlynn, S. P. Photoelectron spectroscopy of carbonyls. Bicyclo[2.2.1]-heptane-2,3-diones and bicyclo[2.2.1] hept-5-ene-2,3-diones. *J. Electron Spectrosc. Relat. Phenom.* **16**, 245 (1979).
- [4852] Clar, E., and Schmidt, W. Correlations between photoelectron and ultraviolet absorption spectra of polycyclic hydrocarbons. The terrylene and peropyrene series, *Tetrahedron* **34**, 3219 (1978).
- [4853] Cradock, S., and Duncan, W. Photoelectron spectra of cyclopentadienyl derivatives of mercury, thallium, indium, tin and lead, *J. Chem. Soc. Faraday Trans. II* **74**, 194 (1978).
- [4854] Distefano, G., Jones, D., Colonna, F. P., Bigotto, A., Galasso, V., Pappalardo, G. C., and Scarlata, G. Evidence from the ultraviolet photoelectron and x-ray photoelectron spectra of phthalimide, quinolinimide, and their *N*-methyl derivatives regarding the prevailing tautomeric form of quinolinimide, *J. Chem. Soc. Perkin II*, 441 (1978).
- [4855] Domelsmith, L. N., Houk, K. N., Degenhardt, C. R., and Paquette, L. A. Photoelectron spectra and orbital interactions in methyleneoertriquinacenes, *J. Am. Chem. Soc.* **100**, 100 (1978).
- [4856] Müller, C., Schweig, A., and Vermeer, H. Photoelectron spectra of 1,2-diphenylcyclopropene, 2,3-diphenylcyclopropene, 2,3-diphenylthiirene 1-oxide, and *cis*-stilbene. An experimental verification of conjugative and inductive interactions, *J. Am. Chem. Soc.* **100**, 8056 (1978).
- [4857] Kirby, C., Kroto, H. W., and Westwood, N. P. C. Detection of chlorothioborine, ClB=S , a new unstable triatomic molecule, by photoelectron and microwave spectroscopy, *J. Am. Chem. Soc.* **100**, 3766 (1978).
- [4858] Dyke, J. M., Fayad, N. K., Morris, A., and Trickle, I. R. Gas-phase He I photoelectron spectra of some transition metals: Cu, Ag, Au, Cr and Mn, *J. Phys. B.* **12**, 2985 (1979).
- [4859] Houk, K. N., and Munchausen, L. L. Ionization potentials, electron affinities, and reactivities of cyanoalkenes and related electron-deficient alkenes. A frontier molecular orbital treatment of cyanoalkene reactivities in cycloaddition, electrophilic, nucleophilic, and radical reactions, *J. Am. Chem. Soc.* **98**, 937 (1976).
- [4860] Hildenbrand, D. L. Dissociation energies of CaBr , SrBr , BaBr , and BaCl from mass spectrometric studies of gaseous equilibria, *J. Chem. Phys.* **66**, 3526 (1977).
- [4861] Gleiter, R., Schang, P., and Seitz, G. Photoelectron spectra of cyclobutene-1,2-dione derivatives, *Chem. Phys. Letters* **55**, 144 (1978).
- [4862] Hsieh, T., and Hanrahan, R. J. An electron impact investigation of pentafluoroethyl iodide, *Int. J. Mass Spectrom. Ion Phys.* **23**, 201 (1977).
- [4863] Hansen, P. E., and Undheim, K. Mass spectrometry of onium compounds. XXIX. Ionisation potential in structure analysis of valence isomers, *Acta Chem. Scand.* **B29**, 221 (1975).
- [4864] Hildenbrand, D. L. Thermochemical studies of the gaseous lower-valent fluorides of molybdenum, *J. Chem. Phys.* **65**, 614 (1976).
- [4865] Hildenbrand, D. L. Thermochemistry of gaseous UF_5 and UF_4 , *J. Chem. Phys.* **66**, 4788 (1977).
- [4866] Domelsmith, L. N., Mollere, P. D., Houk, K. N., Hahn, R. C., and Johnson, R. P. Photoelectron and charge transfer spectra of benzobicycloalkenes. Relationships between through-space interactions and reactivity, *J. Am. Chem. Soc.* **100**, 2959 (1978).
- [4867] Utsunomiya, C., Kobayashi, T., and Nagakura, S. Photoelectron angular distribution measurements for some pyridines, *Bull. Chem. Soc. Jpn.* **51**, 3482 (1978).
- [4868] Corderman, R. R., LeBreton, P. R., Buttrill, S. E., Jr., Williamson, A. D., and Beauchamp, J. L. Photoionization and ion cyclotron resonance studies of the ion chemistry of ethylene oxide, *J. Chem. Phys.* **65**, 4929 (1976).
- [4869] Kordis, J.; and Gingerich, K. A. Mass spectrometric observations of some polyatomic gaseous rare earth oxides and their atomization energies, *J. Chem. Phys.* **66**, 483

- (1977).
- [4870] Smith, R. D., Wyatt, J. R., DeCorpo, J. J., Saalfeld, F. E., Moran, M. J., and MacDiarmid, A. G. Identification of the vapor phase species of (SN), *Chem. Phys. Letters* **41**, 362 (1976).
- [4871] Schweitzer, G. K., McMurtrie, A. C., Allen, J. D., Jr., Cusachs, L. C., Vick, D. O., and Finkelstein, G. The photoelectron spectra of some gaseous thallium (I) oxyanion salts, *J. Electron Spectrosc. Relat. Phenom.* **10**, 155 (1977).
- [4872] Hildenbrand, D. L. Dissociation energy of samarium monoxide and its relation to that of europium monoxide, *Chem. Phys. Letters* **48**, 340 (1977).
- [4873] Simmons, L. L., Lowden, L. F., and Ehlert, T. C. A mass spectrometric study of K_2CO_3 and K_2O , *J. Phys. Chem.* **81**, 706 (1977).
- [4874] Smoes, S., Drowart, J., and Welter, J. M. Thermodynamic study of the vaporization of europium sulfide by the mass spectrometric Knudsen cell method, *Adv. Mass Spectrom.* **7A**, 622 (1978).
- [4875] Simmons, L. L., Lowden, L. F., and Ehlert, T. C. A mass spectrometric study of potassium cyanide, *J. Phys. Chem.* **81**, 709 (1977).
- [4876] Suzuki, I. H., and Maeda, K. Ionization efficiency curves of acetylene by electron impact, *Mass Spectroscopy* **25**, 223 (1977).
- [4877] Sahini, V. E., Constantin, V., and Serban, I. Determination of ionization potentials using a MI-1305 mass spectrometer, *Rev. Roum. Chim.* **23**, 479 (1978).
- [4878] Loudon, A. G., and Webb, K. S. The nature of the $[C_2H_6N]^+$ and $[CH_4N]^+$ ions formed by electron impact on methylated formamides, acetamides, ureas, thioureas and hexamethylphosphoramide, *Org. Mass Spectrom.* **12**, 283 (1977).
- [4879] Foner, S. N., and Hudson, R. L. Mass spectrometry of excited state molecules: Observation of highly vibrationally excited HF by ionization potential measurement, *J. Chem. Phys.* **68**, 2987 (1978).
- [4880] Bunzli, J. C., Frost, D. C., Herring, F. G., and McDowell, C. A. Assignment of the doublet states arising from ionization of chlorine lone-pairs in molecules possessing C_2 symmetry, *J. Electron Spectrosc. Relat. Phenom.* **9**, 289 (1976).
- [4881] Farber, M., and Srivastava, R. D. The dissociation energies of calcium oxide and strontium oxide, *High Temp. Sci.* **8**, 73 (1976).
- [4882] Green, J. C., Seddon, E. A., and Mingos, D. M. P. U.V. photoelectron spectral studies on the metal carbonyl cluster compounds $Os_n(CO)_{12}$, $Ru_3(CO)_{12}$, and $Os_n(CO)_{18}$, *J. Chem. Soc. Chem. Commun.* 94 (1979).
- [4883] Colbourn, E. A., Dyke, J. M., Fackerell, A., Morris, A., and Trickle, I. A. Vacuum ultraviolet photoelectron spectrum of the $GeO(X^1\Sigma^+)$ molecule, *J. Chem. Soc., Faraday Trans. II*, 2278 (1978).
- [4884] Kobayashi, T. A simple general tendency in photoelectron angular distributions of some monosubstituted benzenes, *Phys. Letters* **69A**, 105 (1978).
- [4885] Dehmer, P. M., and Dehmer, J. L. Photoelectron spectra of Ar_2 and Kr_2 and dissociation energies of the rare gas dimer ions, *J. Chem. Phys.* **69**, 125 (1978).
- [4886] Eland, J. H. D., and Berkowitz, J. Formation and predissociation of $CO_2^+(C^2\Sigma_g^+)$, *J. Chem. Phys.* **67**, 2782 (1977).
- [4887] Klasinc, L., Kovač, B., Schoof, S., and Güsten, H. Photoelectron spectroscopy of 9-substituted anthracenes, *Croat. Chem. Acta* **51**, 307 (1978).
- [4888] Downs, A. J., Egdell, R. G., Orchard, A. F., and Thomas, P. D. P. Photoelectron spectra of metal tetrahydroborates, *J. Chem. Soc. Dalton Trans.* 1755 (1978).
- [4889] Bigotto, A., Galasso, V., Colonna, F. P., Distefano, G., Pappalardo, G. C., and Scarlata, G. Electronic structure and photophysical properties of quinolinic anhydride, quinolinic thioanhydride, and *N*-methylquinolinimide, *J. Chem. Soc. Perkins II* **11**, 1194 (1978).
- [4890] Cvitaš, T., and Klasinc, L. Photoelectron spectra of bromobenzenes, *Croat. Chem. Acta* **50**, 291 (1977).
- [4891] Palmer, M. H., Moyes, W., Spiers, M., and Ridyard, J. N. A. The electronic structure of substituted benzenes; ab initio calculations and photoelectron spectra for phenol, the methyl- and fluoro-derivatives, and the dihydroxybenzenes, *J. Mol. Struct.* **52**, 293 (1979).
- [4892] Palmer, M. H., Moyes, W., Spiers, M., and Ridyard, J. N. A. The electronic structure of substituted benzenes; ab initio calculations and photoelectron spectra for nitrobenzene, the nitrotoluenes, dinitrobenzenes and fluoronitrobenzenes, *J. Mol. Struct.* **55**, 243 (1979).
- [4893] Palmer, M. H., Moyes, W., Spiers, M., and Ridyard, J. N. A. The electronic structure of substituted benzenes; a study of aniline, the toluidines, phenylenediamines and fluoroanilines by photoelectron spectroscopy and ab initio calculations, *J. Mol. Struct.* **53**, 235 (1979).
- [4894] Farber, M., and Srivastava, R. D. Mass spectrometric determination of the heats of formation of the silane fluorides, *Chem. Phys. Letters* **51**, 307 (1977).
- [4895] Lossing, F. P., and MacColl, A. Free radicals by mass spectrometry, XLVII. Ionization potentials and ionic heats of formation of C_5-C_7 alkyl radicals, *Can. J. Chem.* **54**, 990 (1976).
- [4896] Willis, C., Lossing, F. P., and Back, R. A. The heat of formation of N_2H_2 and the proton affinity of N_2 , *Can. J. Chem.* **54**, 1 (1976).
- [4897] Hubin-Franskin, M.-J., Locht, R., and Katihabwa, J. Dissociative ionization of carbon disulphide in the gas phase. Heat of formation of the CS radical, *Chem. Phys. Letters* **37**, 488 (1976).
- [4898] Houle, F. A., and Beauchamp, J. L. Detection and investigation of allyl and benzyl radicals by photoelectron spectroscopy, *J. Am. Chem. Soc.* **100**, 3290 (1978).
- [4899] Houle, F. A., and Beauchamp, J. L. Photoelectron spectroscopy of methyl, ethyl, isopropyl, and *tert*-butyl radicals. Implications for the thermochemistry and structures of the radicals and their corresponding carbonium ions, *J. Am. Chem. Soc.* **101**, 4067 (1979).
- [4900] Gupta, S. K., and Gingerich, K. A. Mass spectrometric observation and dissociation energy of the gaseous molecule $MoNb$, *J. Chem. Phys.* **69**, 4318 (1978).
- [4901] Smoes, S., Pattje, W. R., and Drowart, J. Thermodynamic study of the vaporization of manganese metal and manganese selenide by Knudsen-cell mass spectrometry, *High Temp. Sci.* **10**, 109 (1978).
- [4902] Fries, J. A., and Cater, E. D. Vaporization, thermodynamics, and dissociation energy of gadolinium monosulfide: Systematics of vaporization of the rare earth monosulfides, *J. Chem. Phys.* **68**, 3978 (1978).
- [4903] Foner, S. N., and Hudson, R. L. Determination of the proton affinity of N_2 from ionization data on *trans*-diimide, *J. Chem. Phys.* **68**, 3169 (1978).
- [4904] Foner, S. N., and Hudson, R. L. On the heat of formation of diimide, *J. Chem. Phys.* **68**, 3162 (1978).
- [4905] Hubin-Franskin, M. J., Huard, D., and Marmet, P. On the heat of formation of CS from CS_2 and OCS, *Int. J. Mass Spectrom. Ion Phys.* **27**, 263 (1978).
- [4906] Kaposi, O., Popović, A., and Marsel, J. Mass spectrometric studies of tungsten bromides and oxybromides, *J. Inorg. Nucl. Chem.* **39**, 1809 (1977).
- [4907] Murphy, M. K., and Beauchamp, J. L. Photoionization mass spectrometry of the fluoromethylsilanes $(CH_3)_nF_{4-n}Si(n=1-4)$, *J. Am. Chem. Soc.* **99**, 2085 (1977).
- [4908] Van Dam, H., and Oskam, A. $He(I)$ and $He(II)$ photoelectron spectra of iron tetracarbonyl-olefin complexes, *J. Electron*

- Spectrosc. Relat. Phenom. **16**, 307 (1979).
- [4909] Gingerich, K. A., and Gupta, K. A. Dissociation energies of the molecules RhTh and RhU from high temperature mass spectrometry and predicted thermodynamic stabilities of selected diatomic actinide-platinum metal intermetallic molecules, J. Chem. Phys. **69**, 505 (1978).
- [4910] Batten, C. F., Taylor, J. A., and Meisels, G. G. Photoionization processes at threshold. I. Threshold photoelectron and photoionization spectra of CO₂, J. Chem. Phys. **65**, 3316 (1976).
- [4911] Suzuki, I. H., and Maeda, K. Ionization efficiency curves of ethane by electron impact, Intern. J. Mass Spectrom. Ion Phys. **24**, 147 (1977).
- [4912] Zmbov, K. F., Wu, C. H., and Ihle, H. R. A mass spectrometric study of heteronuclear diatomic alkali metal molecules. Dissociation energies and ionization potentials of NaLi, KLi, and NaK, J. Chem. Phys. **67**, 4603 (1977).
- [4913] Schmidt, W. Photoelectron spectra of polynuclear aromatics. V. Correlations with ultraviolet absorption spectra in the catacondensed series, J. Chem. Phys. **66**, 828 (1977).
- [4914] Herrmann, A., Leutwyler, S., Schumacher, E., and Wöste, L. 38. On metal-atom clusters IV. Photoionization thresholds and multiphoton ionization spectra of alkali-metal molecules, Helv. Chim. Acta **61**, 453 (1978).
- [4915] Lossing, F. P. Heats of formation of some isomeric [C_nH_{2n+1}O]⁺ ions. Substitutional effects on ion stability, J. Am. Chem. Soc. **99**, 7526 (1977).
- [4916] Jadrny, R., Karlsson, L., Mattsson, L., and Siegbahn, K. Valence electron spectra of the chlorofluoromethanes CF₃Cl, CF₂Cl₂ and CFCI₃, Physica Scripta **16**, 235 (1977).
- [4917] Sasanuma, M., Ishiguro, E., Hayaisha, T., Masuko, H., Morioka, Y., Nakajima, T., and Nakamura, M. Photoionisation of SF₆ in the XUV region, J. Phys. B. **12**, 4057 (1979).
- [4918] McAllduff, E. J., Lynch, B. M., and Houk, K. N. Photoelectron spectra of substituted benzamides, Can. J. Chem. **56**, 495 (1978).
- [4919] Piacente, V., and Gingerich, K. A. The mass spectrometric determination of the dissociation energies of the molecules NaAu and NaAg, High Temp. Sci. **9**, 189 (1977).
- [4920] Tal'roze, V. L., Butkovskaya, N. I., Larichev, M. N., Leipunskii, I. O., Morozov, I. I., Dodonov, A. F., Kudrov, B. V., Zelenov, V. V., and Raznikov, V. V. Advances in the mass spectrometry of free radicals, Advan. Mass Spectrom. **7A**, 693 (1978).
- [4921] Sullivan, S. A., and Beauchamp, J. L. Positive and negative ion chemistry of sulfuryl halides, Intern. J. Mass Spectrom. Ion Phys. **28**, 69 (1978).
- [4922] Gordon, S. M., Harvey, G. A., Jackson, J. R., Tresling, J. D., and Van Niekerk, J. M. Computer-assisted retarding potential difference system for ionization efficiency measurements, Intern. J. Mass Spectrom. Ion Phys. **23**, 259 (1977).
- [4923] Ng, C. Y., Trevor, D. J., Mahan, B. H., and Lee, Y. T. Photoionization studies of the Kr₂ and Ar₂ van der Waals molecules, J. Chem. Phys. **66**, 446 (1977).
- [4924] Smets, J., Coppens, P., and Drowart, J. Photoionization with mass spectrometric analysis of the tetraphosphorus molecule, Chem. Phys. **20**, 243 (1977).
- [4925] Kuck, D., and Grützmacher, H. F. Hydrogen rearrangement in molecular ions of alkyl benzenes: appearance potentials and substituent effects on the formation of [C_nH_n]⁺ ions, Org. Mass Spectrom. **13**, 81 (1978).
- [4926] Ng, C. Y., Tiedemann, P. W., Mahan, B. H., and Lee, Y. T. Photoionization studies of the diatomic heteronuclear rare gas molecules XeKr, XeAr, and KrAr, J. Chem. Phys. **66**, 5737 (1977).
- [4927] Shudo, K., Kobayashi, T., and Utsunomiya, C. Photoelectron spectral studies on the interaction of three-membered rings with aryl groups, Tetrahedron **33**, 1721 (1977).
- [4928] McLoughlin, R. G., Morrison, J. D., and Traeger, J. C. A photoionization study of the [C_nH_n]⁺ ion formed from some monosubstituted alkyl benzenes, Org. Mass Spectrom. **13**, 483 (1978).
- [4929] Paine, A. J., and Werstiuk, N. H. SCF molecular orbitals and the photoelectron spectrum of 5,5-dimethyl-Δ¹-1,3,4-oxadiazolin-2-one, Can. J. Chem. **56**, 1319 (1978).
- [4930] Ng, C. Y., Trevor, D. J., Mahan, B. H., and Lee, Y. T. Photoionization study of the Xe₂ van der Waals molecule, J. Chem. Phys. **65**, 4327 (1976).
- [4931] Leng, F. J., and Nyberg, G. L. Angular-distribution He(I)/Ne(I) photoelectron spectra of allene, J. Chem. Soc., Faraday Trans. II **73**, 1719 (1977).
- [4932] Jones, G. G., and Taylor, J. W. A photoionization study of carbon dioxide dimers in a supersonic molecular beam, J. Chem. Phys. **68**, 1768 (1978).
- [4933] Van Der Greef, J., Molenaar-Langeveld, T. A., and Nibbering, N. M. M. The elimination of HCN (or HNC) from the molecular ions of some isomeric C_nH_nN compounds, studied by field ionization kinetic and kinetic energy release measurements, Int. J. Mass Spectrom. Ion Phys. **29**, 11 (1979).
- [4934] Rettig, W., and Wirz, J. 111. Electronic structure and photophysical properties of isoindole and its benzo[f]-and dibenzo[e,g]-derivatives, Helv. Chim. Acta **59**, 1054 (1976).
- [4935] Drowart, J., Smets, J., Reynaert, J. C., and Coppens, P. Mass spectrometric study of the photo-ionization of inorganic gases and vapours, Adv. Mass Spectrom. **7A**, 647 (1978).
- [4936] Ulman, J. A., Andersen, E. L., Fehlner, T. P. Characterization of ferraboranes by ultraviolet photoelectron spectroscopy, J. Am. Chem. Soc. **100**, 456 (1978).
- [4937] Santiago, C., Houk, K. N., DeCicco, G. J., Scott, L. T. The photoelectron and ultraviolet spectra of octamethylcyclododeca-1,3,7,9-tetrayne: a weakly antiaromatic molecule, J. Am. Chem. Soc. **100**, 692 (1978).
- [4938] Krause, D. A., Taylor, J. W., and Fenske, R. F. An analysis of the effects of alkyl substituents on the ionization potentials of n-alkenes, J. Am. Chem. Soc. **100**, 718 (1978).
- [4939] Gerson, S. H., Worley, S. D., Bodor, N., Kaminski, J. J., and Flechtnner, T. W. The photoelectron spectra of some heterocyclic compounds which contain N, O, Cl and Br, J. Electron Spectrosc. Relat. Phenom. **13**, 421 (1978).
- [4940] Gerson, S. H., Worley, S. D., Bodor, N., and Kaminski, J. J. Electronic structures of some antimicrobial N-chloramines. Possible existence of intramolecular hydrogen bonding and its effect on germicidal efficiency, J. Med. Chem. **21**, 686 (1978).
- [4941] Cowley, A. H., Dewar, M. J. S., Lattman, M., Mills, J. L., McKee, M. An ultraviolet photoelectron spectroscopic-molecular orbital study of some cyclopolyphosphines, J. Am. Chem. Soc. **100**, 3349 (1978).
- [4942] Carlier, J., and Botter, R. Photoelectron spectra of ethylene and of the six deuterated derivatives, J. Electron Spectros. Relat. Phenom. **17**, 91 (1979).
- [4943] Bulgin, D. K., Dyke, J. M., Jonathan, N., and Morris A. Vacuum ultraviolet photoelectron spectroscopy of transient species, J. Chem. Soc. Faraday Trans. II **75**, 456 (1979).
- [4944] Koenig, T., Imre, D., and Hoobler, J. A. He(I) photoelectron spectrum of benzocyclobutadiene, J. Am. Chem. Soc. **101**, 6446 (1979).
- [4945] Van Dam, H., and Oskam, A. A remeasurement of the UV photoelectron spectrum of Fe(CO)₅ethylene, J. Electron Spectrosc. Relat. Phenom. **17**, 357 (1979).
- [4946] Nagy-Felsobuki, E., Peel, J. B., and Willett, G. D. The

- photoelectron spectrum of bromamine, *J. Electron Spectrosc. Relat. Phenom.* **13**, 17 (1978).
- [4948] Nagy-Felsobuki, E., and Peel, J. B. The photoelectron spectra of unstable intermediates: Dibromamine, *J. Electron Spectrosc. Relat. Phenom.* **15**, 61 (1979).
- [4949] Ulman, J. A., and Fehlner, T. P. Ultraviolet photoelectron spectroscopy of boranes and carboranes. Five-, six-, and seven-atom frameworks, *J. Am. Chem. Soc.* **100**, 449 (1978).
- [4950] Potts, A. W., and Lee, E. P. F. Photoelectron spectra and electronic structure of lithium halide monomers and dimers, *J. Chem. Soc. Faraday Trans. II* **75**, 941 (1979).
- [4951] Carnovale, F., Gan, T. H., and Peel, J. B. Semi-empirical calculations and the assignment of valence photoelectron spectra of large molecules: phenalen-9-amino-1-imine, *J. Electron Spectrosc. Relat. Phenom.* **15**, 173 (1979).
- [4952] Santiago, C., Gandour, R. W., Houk, K. N., Nutakul, W., Cravey, W. E., and Thummel, R. P. Photoelectron and ultraviolet spectra of small-ring fused aromatic molecules as probes of aromatic ring distortions, *J. Am. Chem. Soc.* **100**, 3730 (1978).
- [4954] McAllduff, E. J., Caramella, P., and Houk, K. N. Photoelectron spectra of 3-substituted cyclopentenes. Correlations between ionization potentials and cycloaddition regioselectivity, *J. Am. Chem. Soc.* **100**, 105 (1978).
- [4955] Gassman, P. G., and Yamaguchi, R. Electrochemical oxidation of strained hydrocarbons, *J. Am. Chem. Soc.* **101**, 1308 (1979).
- [4956] Ramsey, B. G., and O'Neill, S. J. A comparison of modified CNDO/2 calculations and the photoelectron, NMR, and UV spectroscopic properties of boronsubstituted monophenylboranes, *J. Organometal. Chem.* **141**, 257 (1977).
- [4957] Berkowitz, J. Photoionization of CH₃OH, CD₃OH, and CH₃OD: Dissociative ionization mechanisms and ionic structures, *J. Chem. Phys.* **69**, 3044 (1978).
- [4958] Hille, E., and Märk, T. D. Cross section for single and double ionization of carbon monoxide by electron impact from threshold up to 180 eV, *J. Chem. Phys.* **69**, 4600 (1978).
- [4959] Akopyan, M. E., and Villem, Ya. Ya. Ion-molecule reactions in the photoionization of formic and acetic acid vapors, *High Energy Chem.* **10**, 24 (1976).
- [4960] Gross, M. L., Chiu, E., Pokorny, D., and DeRoos, F. L. Regiospecificity for water elimination. A mass spectral study of 1-tetralol and 2-tetralol, *Org. Mass Spectrom.* **12**, 55 (1977).
- [4961] Praet, M.-Th., Hubin-Franskin, M. J., Delwiche, J. P., and Schoos, R. Ionization and dissociation of C₆F₆ isomers under electron impact, *Organic Mass Spectrom.* **12**, 297 (1977).
- [4962] Nakato, Y., Abe, K., and Tsubomura, H. Experimental determination of ionization potentials of tetraphenylporphine and metallo-tetraphenylporphines, *Chem. Phys. Letters*, **39**, 358 (1976).
- [4963] Gleiter, R., Haider, R., Conia, J.-M., Barnier, J.-P., de Meijere, A., and Weber, W. Interaction of Walsh orbitals in rotanes. Photoelectron spectroscopic investigation, *J. Chem. Soc. Chem. Commun.* 130 (1979).
- [4964] Spanget-Larsen, J., Gleiter, R., Detty, M. R., and Paguette, L. A. Interaction of Walsh orbitals in trishomocycloheptatrienes and related hydrocarbons, *J. Am. Chem. Soc.* **100**, 3005 (1978).
- [4965] Brittain, H. G., Horozoglu, G., and Baker, A. D. The He(I) photoelectron spectra of some γ -substituted Co(III) acetylacetone complexes, *J. Electron Spectrosc. Relat. Phenom.* **16**, 107 (1979).
- [4966] Smoes, S., and Drowart, J. Determination of the atomization energies of the molecules CSe(g) and CSe₂(g) by the mass spectrometric Knudsen cell method, *J. Chem. Soc.* Faraday Trans. II **73**, 1746 (1977).
- [4967] White, M. G., Rosenberg, R. A., Lee, S. T., and Shirley, D. A. The He(I) photoelectron spectroscopy of heavy group IV-VI diatomics, *J. Electron Spectrosc. Relat. Phenom.* **17**, 323 (1979).
- [4968] Worley, S. D., Taylor, K. G., Venugopalan, B., and Clark, M. S., Jr. Photoelectron spectra, reactions and structures of some acyclic 2-azadiene systems, *Tetrahedron* **34**, 833 (1978).
- [4969] Neijzen, B. J. M., and DeLange, C. A. Photoelectron spectroscopy of mono- and dicyanobenzenes and their perfluoro derivatives, *J. Electron Spectrosc. Relat. Phenom.* **14**, 187 (1978).
- [4970] Banna, M. S., and Shirley, D. A. Molecular photoelectron spectroscopy at 132.3 eV. The second-row hydrides, *J. Chem. Phys.* **63**, 4759 (1975).
- [4971] Holmes, J. L., and Osborne, A. D. Energy partitioning in the metastable fragmentation [C₃H₇]⁺ → [C₃H₅]⁺ + H₂, *Org. Mass Spectrom.* **13**, 133 (1978).
- [4972] Roberge, R., Sandorf, C., Matthews, J. I., and Strausz, O.P. The far ultraviolet and HeI photoelectron spectra of alkyl and fluorine substituted silane derivatives, *J. Chem. Phys.* **69**, 5105 (1978).
- [4979] Hubin-Franskin, M.-J., Delwiche, J., Natalis, P., Caprace, G., and Roy, D. On the photoelectron spectrum of CS₂, *J. Elec. Spectrosc. Rel. Phenom.* **18**, 295 (1980).
- [4980] Brown, R. S., and Marcinko, R. W. Photoelectron spectra of the ozonides of ethylene, cyclopentene, and cyclohexene. Experimental evidence for the magnitude of the "pure" inductive effect of an ether oxygen on ionization energy, *J. Am. Chem. Soc.* **100**, 5584 (1978).
- [4981] MacNaughton, R. M., Allen, J. D., Jr., and Schweitzer, G. K. The He(I) photoelectron spectra of gaseous (CuCl)_n, (CuBr)_n, (AgCl)_n, (AgBr)_n, and (AgI)_n, *J. Electron Spectrosc. Relat. Phenom.* **18**, 363 (1980).
- [4982] Schulz, R., and Schweig, A. 6-Fulveneselone, *Angew. Chem. Int. Ed.* **19**, 69 (1980).
- [4983] Ajò, D., Granozzi, G., Ciliberto, E., and Fragalà, I. Investigation of the electronic structure of 2-(acetylamino)prop-2-enoic acid (N-acetyldehydroalanine) by He¹ and He^{II} photoelectron spectroscopy, *J. Chem. Soc. Perkins Trans. 2* 483, (1980).
- [4984] Bock, H., Boggs, J. E., Kleemann, G., Lentz, D., Oberhammer, H., Peters E. M., Seppelt, K., Simon, A., and Solouki, B. Structure and reactions of methylenesulfur tetrafluoride, *Angew. Chem. Int. Ed.* **18**, 944 (1979).
- [4985] Beltram, G., Fehlner, T. P., Mochida, K., and Kochi, J. K. UV photoelectron spectra of group IV alkyl hydrides, *J. Electron Spectrosc. Relat. Phenom.* **18**, 153 (1980).
- [4986] Coleman, A. W., Green, J. C., Hayes, A. J., Seddon, E. A., Lloyd, D. R., and Niwa, Y. A comparison of the electronic structure of some group 6a dimetal tetracarboxylates using photoelectron spectroscopy, *J. Chem. Soc. Dalton Trans.* **75**, 1057 (1979).
- [4987] Cauletti, C., Clark, J. P., Green, J. C., Jackson, S. E., Fragala, I. L., Ciliberto, E., and Coleman, A. W. Photoelectron spectra of bis-cyclopentadienyl metal dihalides, *J. Electron Spectrosc. Relat. Phenom.* **18**, 61 (1980).
- [4988] Ebsworth, E. A. V., Rankin, D. W. H., and Wright, J. G. Preparation and chemical and spectroscopic properties of (disilylamo)-difluorophosphine and bis(difluorophosphino)silylamine, *J. Chem. Soc. Dalton Trans.* **6**, 1065 (1979).
- [4989] Vovna, V. I., Dudin, A. S., Avkhutskii, L. M., Lopatin, S. N., and Ippolitov, E. G. Photoelectron spectra and electronic spectra of volatile rhenium fluorides [and oxide fluorides], *Russ. J. Inorg. Chem.* **24**, 1135 (1979).
- [4990] Aue, D. H., Webb, H. M., Davidson, W. R., Vidal, M., Bowers, M. T., Goldwhite, H., Vertal, L. E., Douglas, J.

- E., Kollman, P. A., and Kenyon, G. L. Proton affinities and photoelectron spectra of three-membered-ring heterocycles, *J. Am. Chem. Soc.* **102**, 5151 (1980).
- [4991] Eland, J. H. D., and Berkowitz, J. Photoionization mass spectrometry of HI and DI at high resolution, *J. Chem. Phys.* **67**, 5034 (1977).
- [4992] Eweg, J. K., Müller, F., van Dam, H., Terpstra, A., and Oskam, A. He(I) and He(II) photoelectron spectra of alloxazines and isoalloxazines, *J. Am. Chem. Soc.* **102**, 51 (1980).
- [4993] Dannacher, J., Schmelzer, A., Stadelmann, J.-P., and Vogt, J. A photoelectron-photoion coincidence study of vinylfluoride, *Intern. J. Mass Spectrom. Ion Phys.* **31**, 175 (1979).
- [4994] Frey, R., Gotchev, B., Peatrman, W. B., Pollak, H., and Schlag, E. W. Photoionization resonance study of the X(Σ^+), A(Σ^+), B(Σ^+) and C(Σ^+) states of CS₂⁺ and COS⁺ *Intern. J. Mass Spectrom. Ion Phys.* **26**, 137 (1978).
- [4995] Guimon, C., Pfister-Guillouzo, G., and Mathey, F. Electronic structure of phosphacymanrene by photoelectron spectroscopy (He I, He II) and E.H.T. calculations, *Nouveau J. de Chimie* **3**, 725 (1979).
- [4996] Pesterev, V. I., Gabdrakipov, V. Z., Artyukhin, V. I., and Agashkin, O. V. The ionisation and excitation of the conformers of piperidine and its alkyl derivatives, *Russ. J. Phys. Chem.* **53**, 845 (1979).
- [4997] Batten, C. F., Taylor, J. A., Tsai, B. P., and Meisels, G. G. Photoionization processes at threshold. II. Threshold photoelectron, photoionization, and coincidence ion-threshold photoelectron spectra of BF₃, *J. Chem. Phys.* **69**, 2547 (1978).
- [4998] Berkowitz, J., and Holloway, J. H. Photoionization mass spectrometric study of KrF₂, *J. Chem. Soc. Faraday Trans. II* **74**, 2077 (1978).
- [4999] Garner, C. D., Hawksworth, R. W., Hillier, I. H., MacDowell, A. A., and Guest, M. F. Electronic structure of the transition-metal nitrates Ti(NO₃)₄, VO(NO₃)₃, Co(NO₃)₃, and Cu(NO₃)₂. Studied by low-energy photoelectron spectroscopy and ab initio molecular orbital and scattered wave-X α calculations, *J. Am. Chem. Soc.* **102**, 4325 (1980).
- [5000] Radler, K., and Berkowitz, J. Photoionization mass spectrometric study of CS₂, *J. Chem. Phys.* **66**, 2176 (1977).
- [5001] Frost, D. C., MacDonald, C. B., McDowell, C. A., and Westwood, N. P. C. The HeI photoelectron spectra of the halogen azides, XN₃(X=Cl and Br) and the halogen isocyanates, XNCO(X=Cl, Br and I), *Chem. Phys.* **47**, 111 (1980).
- [5002] Colonna, F. P., Distefano, G., Guerra, M., and Jones, D. Photoelectron (He(I), He(II) and X-ray) spectroscopy of γ -pyrone and its related sulphur derivatives: valence and core ionization energies and shake-up satellites, *J. Electron Spectrosc. Relat. Phenom.* **18**, 309 (1980).
- [5003] Williamson, A. D., LeBreton, P. R., and Beauchamp, J. L. Photoionization mass spectrometry of 2-fluoropropane and 2,2-difluoropropane. A novel determination of the proton affinity of vinyl fluoride and 1,1-difluoroethylene, *J. Am. Chem. Soc.* **98**, 2705 (1976).
- [5004] Weiss, M. J., Berkowitz, J., and Appelman, E. H. Photoionization of ozone: Formation of O₄⁺ and O₅⁺, *J. Chem. Phys.* **66**, 2049 (1977).
- [5005] Worley, S. D., Webb, T. R., Gibson, D. H., and Ong, T.-S. The photoelectron spectra of some iron tricarbonyl complexes of 4 π -electron donor ligands, *J. Electron Spectrosc. Relat. Phenom.* **18**, 189 (1980).
- [5006] Huebner, R. H., Celotta, R. J., Mielczarek, S. R., and Kuyatt, C. E. Electron energy loss spectroscopy of acetone vapor, *J. Chem. Phys.* **59**, 5434 (1973).
- [5007] Appell, J., Durup, J., Fehsenfeld, F. C., and Fournier, P. Double charge transfer spectroscopy of diatomic molecules, *J. Phys. B* **6**, 197 (1973).
- [5008] Dyke, J. M., Jonathan, N. B. H., Morris, A., and Winter, M. J. The first ionization potential of the formyl radical, HCO(X²A), studied using photoelectron spectroscopy, *Mol. Phys.* **39**, 629 (1980).
- [5009] Parr, A. C., Jason, A. J., Stockbauer, R., and McCulloch, K. E. Photoionization and threshold photoelectron-photoion coincidence study of propyne from onset to 20 eV, *Intern. J. Mass Spectrom. Ion Phys.* **30**, 319 (1979).
- [5010] Worley, S. D., Webb, T. R., Gibson, D. H., and Ong, T. -S. On the electronic structures of cyclobutadiene and trimethylenemethane, *J. Organometal. Chem.* **168**, C16 (1979).
- [5011] Dunlavy, S. J., Dyke, J. M., Jonathan, N., and Morris, A. Vacuum ultraviolet photoelectron spectroscopy of transient species. Part 11. The NH₂(X²B₁) radical, *Mol. Phys.* **39**, 1121 (1980).
- [5012] Bock, H., Kaim, W., and Rohwer, H. E. Radical ions XI*. One-electron oxidation of alkylsilyl benzenes in the gas phase and in solution, *J. Organometal. Chem.* **135**, C14 (1977).
- [5013] Szepes, L., Distefano, G., and Pignataro, S. Steric inhibition of resonance in acetanilides by UV photoelectron spectroscopy, *Ann. Chim.* **64**, 159 (1974).
- [5014] Parr, A. C., Jason, A. J., and Stockbauer, R. Photoionization and threshold photoelectronphotoion coincidence study of cyclopropene from onset to 20 eV. *Int. J. Mass Spectrom. Ion Phys.* **33**, 243 (1980).
- [5015] Ng, C. Y., Trevor, D. J., Tiedemann, P. W., Ceyer, S. T., Kronebusch, P. L., Mahan, B. H., and Lee, Y. T. Photoionization of dimeric polyatomic molecules: proton affinities of H₂O and HF, *J. Chem. Phys.* **67**, 4235 (1977).
- [5016] Pabst, R. E., Sharpe, M. C., Margrave, J. L., and Franklin, J. L. An electron impact study of the appearance energies of positive ions from AsF₃, AsCl₃, AsBr₃ and AsF₅, *Int. J. Mass Spectrom. Ion Phys.* **33**, 187 (1980).
- [5017] Sell, J. A., and Kuppermann, A. Variable angle photoelectron spectroscopy of the fluoroethylenes, *J. Chem. Phys.* **71**, 4703 (1979).
- [5018] Wood, K. V., and Taylor, J. W. A photoionization mass spectrometric study of autoionization in ethylene and trans-2-butene, *Int. J. Mass Spectrom. Ion Phys.* **30**, 307 (1979).
- [5019] Santiago, C., McAlduff, E. J., Houk, K. N., Snow, R. A., and Paquette, L. A. Photoelectron spectra of ortho- and meta-substituted benzonorbornadienes. Relationships to regioselectivities in triplet di- π -methane rearrangements, *J. Am. Chem. Soc.* **100**, 6149 (1978).
- [5020] Gleiter, R., Hofmann, P., Schang, P., and Sieber, A. The orbital sequence in cyclic 1,3-diketones, *Tetrahedron* **36**, 655 (1980).
- [5021] Zverev, V. V., Villem, Y. Y., Ermolaeva, L. V., and Lisin, A. F. The photoelectron spectra and electronic structure of unsaturated phosphoryl compounds, *Doklady Akademii Nauk SSSR*, **246**, 1368 (1978).
- [5022] Freiser, B. S. Electron impact ionization of argon ions by trapped ion cyclotron resonance spectroscopy, *Int. J. Mass Spectrom. Ion Phys.* **33**, 263 (1980).
- [5023] Nagy-Felsobuki, E., and Peel, J. B. Photoelectron spectra of selenium dichloride and diselenium dichloride, *J. Chem. Soc. Faraday II*, **76**, 148 (1980).
- [5024] Bursten, B. E., Cotton, F. A., Green, J. C., Seddon, E. A., and Stanley, G. G. Electronic structures and photoelectron spectra of the metal atom cluster species Re_nCl_n, Re_nBr_n, and [Re₃Cl₁₂]³⁻, *J. Am. Chem. Soc.* **102**, 955 (1980).
- [5025] Mead, P. T., Donchi, K. F., Traeger, J. C., Christie, J. R., and Derrick, P. J. Secondary hydrogen isotope effect in the

- unimolecular decomposition of 2-methylpropane radical cations, J. Am. Chem. Soc. **102**, 3364 (1980).
- [5026] Neijzen, B. J. M., and DeLange, C. A. Photoelectron spectroscopy of some thiocyanates, isocyanates and isothiocyanates, J. Electron Spectrosc. Relat. Phenom. **18**, 179 (1980).
- [5027] Hubin-Franskin, M.-J., Marmet, P., and Huard, D. Excitation and ionization of OCS and CS₂ by electron impact, Int. J. Mass Spectrom. Ion Phys. **33**, 311 (1980).
- [5028] Eland, J. H. D., Berkowitz, J., Schulte, H., and Frey, R. Rates of unimolecular pyridine ion decay and the heat of formation of C₆H₅⁺, Int. J. Mass Spectrom. Ion Phys. **28**, 297 (1978).
- [5029] Lee, L. C., Judge, D. L., and Ogawa, M. CS₂⁺(B²Σ_u⁺, A²Π_u → X²Π_g) fluorescence from photoionization excitation of CS₂ vapor, Can. J. Phys. **53**, 1861 (1975).
- [5030] Colbourne, D., Frost, D. C., McDowell, C. A., and Westwood, N. P. C. Dichloroketen; gas phase preparation and characterisation by photoelectron spectroscopy, J. Chem. Soc. Chem. Comm. 250, (1980).
- [5031] Nagy-Felsobuki, E., and Peel, J. B. Photoelectron spectra of sulfur dibromide and selenium dibromide, Chem. Phys. **45**, 189 (1980).
- [5032] Zverev, V. V., Villem, Ya. Ya., Bel'skii, V. E., and Kitaev, Yu. P. The photoelectronic spectra of phosphoryl compounds, Izv. Akad. Nauk SSSR, Ser. Khim. **1**, 84 (1979).
- [5033] Westwood, N. P. C., Kroto, H. W., Nixon, J. F., and Simmons, N. P. C. Formation of 1-phosphapropyne, CH₃C≡P, by pyrolysis of dichloro(ethyl)phosphine: a He(I) photoelectron spectroscopic study, J. Chem. Soc. Dalton Trans. **9**, 1405 (1979).
- [5034] Brogli, F., Heilbronner, E., Kloster-Jensen, E., Schmelzer, A., Manocha, A. S., Pople, J. A., and Radom, L. The photoelectron spectrum of butatriene, Chem. Phys. **4**, 107 (1974).
- [5035] Potts, A. W., and Price, W. C. Photoelectron studies of ionic materials using molecular beam techniques, Physica Scripta **16**, 191 (1977).
- [5036] Chisholm, M. H., Cowley, A. H., and Lattman, M. A UV photoelectron spectroscopic investigation of the bonding in some tri-, tetra-, and penta- coordinated dialkylamino compounds of chromium, molybdenum, niobium, and tantalum, J. Am. Chem. Soc. **102**, 46 (1980).
- [5037] Walker, T. E. H., Dehmer, P. M., and Berkowitz, J. Rotational band shapes in photoelectron spectroscopy: HF and DF, J. Chem. Phys. **59**, 4292 (1973).
- [5038] Jaudon, P., and Tabet, J.-C. Rearrangement of molecular ions following electron impact. IV-Origin of the [M-28]⁺ ions in the mass spectra of α-methyl-2-decalone, Org. Mass Spectrom. **15**, 65 (1980).
- [5039] Holmes, J. L., and Lossing, F. P. Gas-phase heats of formation of keto and enol ions of carbonyl compounds, J. Am. Chem. Soc. **102**, 1591 (1980).
- [5040] Evlasheva, T. I., Puchkova, V. V., Potapov, V. K., and Gur'yanova, E. N. Ionisation potentials and electron-donating properties of sulphones, Russ. J. Phys. Chem. **49**, 453 (1975).
- [5041] Johnson, K. M., Powis, I., and Danby, C. J. The fragmentation of COCl₂⁺ and COF₂⁺ ions studied by the photoelectron-photoion coincidence technique, Int. J. Mass Spectrom. Ion Phys. **32**, 1 (1979).
- [5042] Hodges, R. V., Houle, F. A., Beauchamp, J. L., Montag, R. A., and Verkade, J. G. Effects of molecular structure on basicity. Gas phase proton affinities of cyclic phosphites, J. Am. Chem. Soc. **102**, 932 (1980).
- [5043] Kovač, B., and Klasic, L. Photoelectron spectroscopy of adamantane and some adamantanones, Croat. Chem. Acta **51**, 55 (1978).
- [5044] Connor, J. A., Derrick, L. M. R., Hall, M. B., Hillier, I. H., Guest, M. F., Higginson, B. R., and Lloyd, D. R. The electronic structure of transition metal complexes containing organic ligands 1. Low and high energy photoelectron spectra and ab initio SCF MO calculations of iron tricarbonyl butadiene, Mol. Phys. **28**, 1193 (1974).
- [5045] Parker, D. H., and El-Sayed, M. A. Determination of excited state lifetimes and ionization potentials by dual beam visible lasers, Chem. Phys. **42**, 379 (1979).
- [5046] Lefavre, D., and Marmet, P. Electroionization of D₂O and H₂O and study of fragments H⁺ and OH⁺, Can. J. Phys. **56**, 1549 (1978).
- [5048] Larzilliere, M., and Damany, N. Rydberg series of carbon disulfide converging to A²Π_u states of CS₂⁺, Can. J. Phys. **56**, 1150 (1978).
- [5049] Lagerqvist, A., and Renhorn, I. The spectrum of silicon monoxide (oxygen-18) in the vacuum ultraviolet region, J. Mol. Spectrosc. **19**, 157 (1974).
- [5050] Parr, A. C., Jason, A. J., and Stockbauer, R. Photoionization and threshold photoelectron-photoion coincidence study of allene from onset to 20eV, Intern. J. Mass Spectrom. Ion Phys. **26**, 23 (1978).
- [5051] Locht, R., Schopman, J., Wankenne, H., and Momigny, J. The dissociative ionization of nitrogen, Chem. Phys. **7**, 393 (1975).
- [5052] Lee, E. P. F., and Potts, A. W. The VUV photoelectron spectra of atomic In and Tl, J. Electron Spectrosc. Relat. Phenom. **19**, 65 (1980).
- [5053] Leupin, W., and Wirz, J. 161. Cyclooct-1-en-5-yne. Preparation, spectroscopic characteristics and chemical reactivity, Helv. Chim. Acta **61**, 1663 (1978).
- [5054] Poole, R. T., Orders, P. J., Jenkin, J. G., Leckey, R. C. G., and Liesegang, J. Electronic structure of the valence bands of SnF₂ studied by ultraviolet photoelectron spectroscopy, Chem. Phys. Letters **54**, 220 (1978).
- [5055] Potts, A. W., and Williams, T. A. The observation of "forbidden" transitions in He II photoelectron spectra, J. Electron Spectrosc. Relat. Phenom. **3**, 3 (1974).
- [5056] Paisner, J. A., Solarz, R. W., Worden, E. F., and Conway, J. G. IV. Highly excited states, ionization, and high intensity interactions, Springer Series in Optical Sciences, Laser Spectroscopy III, **7**, 160 (1977).
- [5058] Broer, W. J., Weringa, W. D., and Nieuwpoort, W. C. Rearrangements and fragmentations of [C₂H₅S]⁺ ions, Org. Mass Spectrom. **14**, 543 (1979).
- [5059] Helal, A. I., and Zahran, N. F. Kinetic shift in some para-substituted acetophenones, Org. Mass Spectrom. **13**, 549 (1978).
- [5060] Karlsson, L., Mattsson, L., Jadrny, R., Bergmark, T., and Siegbahn, K. Vibrational and vibronic structure in the valence electron spectrum of H₂S, Physica Scripta **13**, 229 (1976).
- [5061] Kingcade, J. E., Dufner, D. C., Gupta, S. K., and Gingerich, K. A. A thermodynamic study of the gaseous molecules CuSn₂ and Cu₂Sn, High Temp. Sci. **10**, 213 (1978).
- [5063] Kimura, K., and Osafune, K. Sum rule consideration on valence orbital ionization energies in methyl amines, Mol. Phys. **29**, 1073 (1975).
- [5064] Kronebusch, P. L., and Berkowitz, J. Photodissociative ionization in the 21–41 eV region: O₂, N₂, CO, NO, CO₂, H₂O, NH₃ and CH₄, Int. J. Mass Spectrom. Ion Phys. **22**, 283 (1976).
- [5066] Powis, I., and Danby, C. J. The unimolecular fragmentation of energy-selected acetone ions, Int. J. Mass Spectrom. Ion Phys. **32**, 27 (1979).
- [5067] Kleinschmidt, P. D., and Hildenbrand, D. L. Dissociation energies of Cal, Sri, and Bal from high temperature mass spectrometry, J. Chem. Phys. **68**, 2819 (1978).
- [5068] Kimura, K., and Osafune, K. Photoelectron spectroscopic study of skew compounds. III. N,N'-dimethylhydrazine,

- dimethyl peroxide, and dimethyl disulfide, Bull. Chem. Soc. Japan **48**, 2421 (1975).
- [5069] Traeger, J. C. Photoionization mass spectrometry of the propyl halides, Int. J. Mass Spectrom. Ion Phys. **32**, 309 (1980).
- [5070] Holmes, J. L., and Lossing, F. P. Keto and enol forms of methyl acetate molecular ions, their stability and interconvertibility prior to fragmentation in the gas phase, Org. Mass Spectrom. **14**, 512 (1979).
- [5071] Drury-Lessard, C. R., and Moule, D. C. The higher Rydberg states of formaldehyde, Chem. Phys. Letters **47**, 300 (1977).
- [5072] Holmes, J. L., Rye, R. T. B., and Terlouw, J. K. On the loss of ethylene from $[C_3H_5O]^+$ ions of structure CH_3CH_2CHOH , Org. Mass Spectrom. **14**, 606 (1979).
- [5073] De Leeuw, D. M., Mooyman, R., and De Lange, C. A. He(I) photoelectron spectroscopy of transient species: The SF₂ molecule, Chem. Phys. **34**, 287 (1978).
- [5074] De Leeuw, D. M., Mooyman, R., and De Lange, C. A. He(I) photoelectron spectroscopy of transient species: The SeX₂ molecules (X=F, Cl and Br), Chem. Phys. **38**, 21 (1979).
- [5079] Lohr, W., Jochims, H. W., and Baumgartel, H. Photoreaktionen kleiner organischer Moleküle IV Absorptionsspektren, Photoionen- und Resonanzphotoelektronenspektren von Vinylbromid, Ber. Bunsenges. **79**, 901 (1975).
- [5080] Baldwin, M. A. Appearance energies and the kinetic shift. Loss of HCN from the benzonitrile molecular ion, Org. Mass. Spectrom. **14**, 601 (1979).
- [5081] Kaufman, V., and Hagan, L. Spectrum and energy levels of singly ionized aluminum (Al II), J. Opt. Soc. Am. **69**, 232 (1979).
- [5082] Lauer, G., Schäfer, W., and Schweig, A. Assignment of the four lowest ionized states of *p*-benzoquinone and the question of "lone pair splitting" in this system, Chem. Phys. Letters **33**, 312 (1975).
- [5083] Russell, D. H., Gross, M. L., Van der Greef, J., and Nibbering, N. M. M. The chemistry of C_nH_nO radical cations: A study of rearrangement reactions of halogen substituted ethyl phenyl ethers, Org. Mass Spectrom. **14**, 474 (1979).
- [5084] Bieri, G., and Åsbrink, L. 30.4-nm He(II) photoelectron spectra of organic molecules, J. Electron Spectrosc. Relat. Phenom. **20**, 149 (1980).
- [5085] Loudet, M., Grimaud, M., Metras, F., and PfisterGuillouzo, G. Interactions intramoléculaires en série cyclohexanique partie II. Spectres photoélectroniques de chloro-2 cyclohexanones, J. Mol. Struct. **35**, 213 (1976).
- [5086] Holmes, J. L., Terlouw, J. K., Vijhuizen, P. C., and A'Campo, C. Metastable ion studies XII—Molecular and fragment ion structures for isomeric C_nH_nO₂ acids, Org. Mass Spectrom. **14**, 204 (1979).
- [5087] De Leeuw, D. M., Mooyman, R., and De Lange, C. A. He(I) photoelectron spectroscopy of halogen atoms, Chem. Phys. Letters **54**, 231 (1978).
- [5088] Hoppilliard, Y., and Solgadi, D. Conformational analysis of 2-haloethanols and 2-methoxyethyl- halides in a photoelectron-spectrometer, Tetrahedron **36**, 377 (1980).
- [5089] Weidner, U., and Schweig, A. Evidence for conjugation through a saturated silicon atom, Angew. Chem. Int. Ed. **11**, 536 (1972).
- [5090] Dougherty, D., Brint, P., and McGlynn, S. P. Photoelectron spectroscopy of carbonyls. Lone-pair interactions in α -, β -, γ -, and δ -dicarbonyls, J. Am. Chem. Soc. **100**, 5597 (1978).
- [5091] Nelsen, S. F., Kessel, C. R., and Brien, D. J. Bredt's rule kinetically stabilized nitrogen- centered radical cations and radicals in the 9-azabicyclo[3.3.1]nonyl system, J. Am. Chem. Soc. **102**, 702 (1980).
- [5092] Ramsey, B. G. Substituent effects on imidazole basicity and photoelectron spectroscopy determined ionization energies, J. Org. Chem. **44**, 2093 (1979).
- [5093] Dougherty, D., Younathan, E. S., Voll, R., Abdulnur, S., and McGlynn, S. P. Photoelectron spectroscopy of some biological molecules, J. Electron Spectrosc. Relat. Phenom. **13**, 379 (1978).
- [5094] Heilbronner, E., Jones, T. B., Krebs, A., Maier, G., Malsch, K. D., Pocklington, J., and Schmelzer, A. A photoelectron spectroscopic investigation of tetra-*tert*-butyltetrahedrane and of tetra-*tert*- butylcyclobutadiene, J. Am. Chem. Soc. **102**, 564 (1980).
- [5095] Bigotto, A., Galasso, V., Distefano, G., and Modelli, A. Photoelectron and electronic spectra of acenaphthenequinone, naphthalic anhydride, and naphthalimide, J. Chem. Soc. Perkin II, 1502 (1979).
- [5096] de Jong, A. P., and van Dam, H. Ultraviolet photoelectron spectroscopy of cyclic amidines. I. Electronic structure of some α -adrenergic benzylimidazolines, J. Med. Chem. **22**, 1290 (1979).
- [5097] McAlduff, E. J. Photoelectron spectra of substituted acetophenones. Correlations with reactivity, Can. J. Chem. **58**, 622 (1980).
- [5098] Greening, F. R., and King, G. W. Rydberg states of carbon diselenide, J. Mol. Spectrosc. **61**, 459 (1976).
- [5099] Houk, K. N., Bimanand, A., Mukherjee, D., Sims, J., Chang, Y.-M., Kaufman, D. C., and Domelsmith, L. N. Nitrone ionization potentials and cycloaddition regioselectivities, Heterocycles **7**, 293 (1977).
- [5100] Cauletti, C., Furlani, C., and Storto, G. Coordinative bond and *d*-shell ionisations in the UV photoelectron spectra of bis(β -diketonato) cobalt (II) and copper (II) complexes, and their thioanalogue, J. Electron Spectrosc. Relat. Phenom. **18**, 329 (1980).
- [5101] Gürtler, P., Saile, V., and Koch, E. E. Rydberg series in the absorption spectra of H₂O and D₂O in the vacuum ultraviolet, Chem. Phys. Letters **51**, 386 (1977).
- [5102] Bock, H., Kaim, W., Kira, M., Osawa, H., and Sakurai, H. Radical ions XXVIII. Tris(trimethylsilylmethyl) aminium, $^+N(CH_3Si(CH_3)_3)_3$: A stable fluxional aminium radical cation, J. Organometal. Chem. **164**, 295 (1979).
- [5103] Fragalà, I., Ciliberto, E., Finocchiaro, P., and Recca, A. He(I) and He(II) excited photoelectron spectra and electronic structure of 'pseudooctahedral' dichloro- and dimethyl-bis(pentane- 2,4-dionato) tin(IV), J. Chem. Soc. Dalton II, 240 (1979).
- [5104] Kajitani, M., Sugimori, A., Sato, N., Seki, K., Inokuchi, H., and Harada, Y. Ultraviolet photoelectron spectra of crown ethers, Bull. Chem. Soc. Japan **52**, 2199 (1979).
- [5105] Thomas, R. K., and Thompson, H. The photoelectron spectra of allene, deuterioallenes and tetrafluoroallene, Proc. R. Soc. London Ser. A, **339**, 29 (1974).
- [5106] Stockbauer, R., and Rosenstock, H. M. Kinetic shift in methane and allene ion fragmentation, Intern. J. Mass Spectrom. Ion Phys. **27**, 185 (1978).
- [5107] Solouki, B., Rosmus, P., Bock, H., and Maier, G. Short-path pyrolysis: Silabenzene, Angew. Chem. Int. Ed. **19**, 51 (1980).
- [5108] Gleiter, R., Böhm, M. C., Haaland, A., Johansen, R., and Lusztyk, J. Beryllocene, (C₅H₅)₂Be. The He (I) photoelectron spectrum and ab initio molecular orbital calculations, J. Organometal. Chem. **170**, 285 (1979).
- [5119] Martin, H.-D., and Pföhler, P. Pentacyclo[6.4.0.0^{2,5}.0^{8,10}.0^{1,9}]dodeca-6,11-diene, an *o,o':op'*-dimer of benzene, Angew. Chem. Int. Ed. **17**, 847 (1978).
- [5120] Traeger, J. C., and McLoughlin, R. G. Threshold photoionization and dissociation of toluene and cycloheptatriene, J. Am. Chem. Soc. **99**, 7351 (1977).
- [5121] Tajima, S., Azami, T., and Tsuchiya, T. An investigation of the decomposition of the common intermediate ions produced by electron impact, Org. Mass. Spectrom. **12**, 24 (1977).

- [5122] Schmidt, H., Schweig, A., and Vermeer, H. On the conformation of unsaturated arsines, *J. Mol. Struct.* **37**, 93 (1977).
- [5123] Schander, J., and Russell, B. R. Vacuum ultraviolet spectra of bromoethylene and dibromoethylenes, *J. Am. Chem. Soc.* **98**, 6900 (1976).
- [5124] Hudson, B. S., Ridyard, J. N. A., and Diamond, J. Polyene spectroscopy. Photoelectron spectra of the diphenylpolyenes, *J. Am. Chem. Soc.* **98**, 1126 (1976).
- [5125] Sell, J. A., and Kupperman, A. Angular distributions in the photoelectron spectra of benzene and its monohalogenated derivatives, *Chem. Phys.* **33**, 367 (1978).
- [5126] Smyth, K. C., Schiavone, J. A., and Freund, R. S. Dissociative excitation of CO by electron impact: Translational spectroscopy of long-lived high-Rydberg fragment atoms, *J. Chem. Phys.* **60**, 1358 (1974).
- [5127] Samson, J. A. R., Kemeny, P. C., and Haddad, G. N. Double ionization of CO₂ by photon impact, *Chem. Phys. Letters* **51**, 75 (1977).
- [5128] Suzuki, I. H., and Maeda, K. Behavior of hydrogen atoms in the fragmentation of CH₃CD₃, *Can. J. Chem.* **55**, 3124 (1977).
- [5129] Suzuki, I. H., and Maeda, K. Ionization efficiency curves of acetylene by mono-energetic electron impact, *Adv. Mass Spectrom.* **7**, 182 (1978).
- [5130] Stockbauer, R., and Inghram, M. G. Threshold photoelectron-photoion coincidence mass spectrometric study of ethylene and ethylene-d₄, *J. Chem. Phys.* **62**, 4862 (1975).
- [5131] Schulz, R., and Schweig, A. Existence of 1,2,3-benzoxadiazole in the gas phase, *Angew. Chem. Int. Ed.* **18**, 692 (1979).
- [5132] Frey, R., Gotchev, B., Kalman, O. F., Peatman, W. B., Pollak, H., and Schlag, E. W. Photoionization resonance spectra of CO₂⁺ and threshold electron-ion coincidence measurements of the fragmentation of CO₂⁺, *Chem. Phys.* **21**, 89 (1977).
- [5133] Nelsen, S. F., Hollinsed, W. C., Grezzo, L. A., and Parmelee, W. P. Conformational effects in 2,3-cycloalkyl-2,3-diazabicyclic tetraalkylhydrazines, *J. Am. Chem. Soc.* **101**, 7347 (1979).
- [5134] Fragala, I., Millefiori, S., and Recca, A. Gas phase ultraviolet photoelectron spectra of aromatic azomethine compounds, *J. Chem. Res.* **1**, 28 (1980).
- [5135] Golovin, A. V., Akopyan, M. E., Vilesov, F. I., and Sergeev, Y. L. Ion-electron coincidence study of the photoionization of formic and acetic acids, *Khim. Vys. Energ.* **13**, 200 (1979).
- [5136] Ghosh, S. N., and Verma, R. D. Rydberg states of the PO molecule, *J. Mol. Spectrosc.* **72**, 200 (1978).
- [5137] Frost, D. C., Lee, S. T., McDowell, C. A., and Westwood, N. P. C. The photoelectron spectrum of diazene (diimine), *Chem. Phys. Letters* **30**, 26 (1975).
- [5138] Maier, J. P., and Marthaler, O. Emission spectra of the radical cations of 1,3-dichlorobenzene, 1,4-dichlorobenzene and 1,3,5-trichlorobenzene in the gas phase, *Chem. Phys.* **32**, 419 (1978).
- [5139] Daamen, H., Oskam, A., and Stufkens, D. J. U.V.-photoelectron (He I and He II) studies of M(CO)₅PR₃ (M = Cr, W and R = C₆H₁₁, C₆H₅, O-i-C₃H₇, OC₆H₅) and W(CO)₅As(C₆H₅)₃, *Inorg. Chim. Acta* **38**, 71 (1980).
- [5140] Takezawa, S., and Tanaka, Y. The absorption spectrum of D₂ in the vacuum-uv region, Rydberg bands, np_σ¹Σ_u⁺ ← X¹Σ_g⁺ and np_π¹Π_u[←] X¹Σ_g⁺ with n=4–6, and the ionization energy, *J. Mol. Spectrosc.* **54**, 379 (1975).
- [5141] Cooks, R. G., Ast, T., and Beynon, J. H. A new method for the determination of double and triple ionization potentials of organic ions, *Intern. J. Mass. Spectrom. Ion Phys.* **11**, 490 (1973).
- [5142] Jonathan, N., Morris, A., Okuda, M., Ross, K. J., and Smith, D. J. Vacuum ultraviolet photoelectron spectroscopy of transient species, *J. Chem. Soc. Faraday Trans. II*, 70, 1810 (1974).
- [5143] Mathur, B. P., Rothe, E. W., Reck, G. P., and Lightman, A. J. Two-photon ionization of Li₂: isotopic separation and determination of IP(Li₂) and D₂(Li₂⁺), *Chem. Phys. Letters* **56**, 336 (1978).
- [5144] Miescher, E. High resolution absorption spectrum of nitric oxide (NO) in the region of the first ionization limit, *Can. J. Phys.* **54**, 2074 (1976).
- [5145] Boschi, R. A., and Salahub, D. R. The far ultraviolet spectra of some branched chain iodo-alkanes, iodo-cycloalkanes, fluoro-iodo-alkanes and iodoalkenes, *Mol. Phys.* **24**, 735 (1972).
- [5146] McCulloch, K. E. Energetics and mechanisms of fragment ion formation in the photoionization of normal and deuterated water and ammonia, *Int. J. Mass Spectrom. Ion Phys.* **21**, 333 (1976).
- [5147] Appell, J., Durup, J., Fehsenfeld, F. C., and Fournier, P. Doubly ionized states of some polyatomic molecules studied by double charge transfer spectroscopy, *J. Phys. B*, **7**, 406 (1974).
- [5148] Burroughs, P., Evans, S., Hamnett, A., Orchard, A. F., and Richardson, N. V. He-I photoelectron spectra of some d⁰ transition metal compounds, *J. Chem. Soc. Faraday Trans. II* **70**, 1895 (1974).
- [5149] Avni, R., and Klein, F. S. The first ionization potential of uranium and thorium measured in a d.c. arc plasma, *Spectrochim. Acta* **25B**, 331 (1973).
- [5150] Gupta, S. K., Pelino, M., and Gingerich, K. A. Dissociation energy of the gaseous molecule PtTi by high-temperature Knudsen effusion mass spectrometry, *J. Phys. Chem.* **83**, 2335 (1979).
- [5151] Cvitaš, T., and Klasinc, L. High resolution photoelectron spectrum of hydrazoic acid, *J. Chem. Soc. Faraday Trans. II*, **72**, 1240 (1976).
- [5152] Busse, B., and Weil, K. G. Existence and bond energy of the cesium auride molecule, *Angew. Chem. Int. Ed.* **18**, 629 (1979).
- [5153] Harland, P. W., Cradock, S., and Thynne, J. C. J. Positive- and negative-ion formation due to the electron bombardment of germanium tetrafluoride, *Int. J. Mass Spectrom. Ion Phys.* **10**, 169 (1972/73).
- [5154] Gacek, M., Thorstad, O., Ongstad, L., and Undheim, K. Ionisation potentials in tautomer analysis of 2-hydroxypyrimidines, *Chem. Scripta* **13**, 99 (1978).
- [5155] Watanabe, I., Yokoyama, Y., and Ikeda, S. Vibrational structures in the He(I) photoelectron spectra of carboxylic acids, *Bull. Chem. Soc. Japan* **47**, 627 (1974).
- [5156] Yoshino, K., and Tanaka, Y. Absorption spectrum of krypton in the vacuum UV region, *J. Opt. Soc. Am.* **69**, 159 (1979).
- [5157] Zmbov, K. F., and Miletic, M. Mass spectrometric determination of the dissociation energy of PbO₂ and ionization potentials of PbO and PbO₂ molecules, *Advan. Mass Spectrom.* **7A**, 573 (1978).
- [5158] Wu, C. H. Thermochemical properties of gaseous Li₂ and Li₃, *J. Chem. Phys.* **65**, 3181 (1976).
- [5159] Worden, E. F., and Conway, J. G. Laser spectroscopy of neptunium; first ionization potential, lifetimes and new high-lying energy levels of Np I, *J. Opt. Soc. Am.* **69**, 733 (1979).
- [5160] Farber, M., and Srivastava, R. D. Mass spectrometric determination of the heats of formation of the silicon bromides SiBr(g), SiBr₂(g), and SiBr₃(g). *High Temp. Sci.* **12**, 21 (1980).
- [5161] Fock, J.-H., Gürler, P., and Koch, E. E. Molecular Rydberg transitions in carbon monoxide: term value/ionization energy correlation of BF, CO and N₂, *Chem. Phys.* **47**, 87 (1980).

- [5168] Fragala, I., Ciliberto, E., Egde^{II}, R. G., and Granozzi, G. He(I) and He(II) photoelectron spectra of methyltin chlorides, *J. Chem. Soc. Dalton*, **145** (1980).
- [5169] Gupta, S. K., and Gingerich, K. A. Observation and atomization energies of the gaseous uranium carbides, UC, UC₂, UC₃, UC₄, and UC₅, by high temperature mass spectrometry, *J. Chem. Phys.* **71**, 3072 (1979).
- [5170] Hitchcock, A. P., Brion, C. E., and Van der Wiel, M. J. Absolute oscillator strengths for valence-shell ionic photofragmentation of N₂O and CO₂(8–75 eV), *Chem. Phys.* **45**, 461 (1980).
- [5171] Ho, P., and Burns, R. P. A mass spectrometric study of the AlO₂ molecule, *High Temp. Sci.* **12**, 31 (1980).
- [5172] Lee, E. P. F., Potts, A. W., Doran, M., Hillier, I. H., Delaney, J. J., Hawksworth, R. W., and Guest, M. F. Photoelectron spectra and electronic structure of the transition metal dichlorides, MCl₂(M = Cr, Mn, Fe, Co, Ni), *J. Chem. Soc. Faraday II*, **76**, 506 (1980).
- [5173] Momigny, J., Wankenne, H., and Krier, C. Correlation diagram approach to the dissociative ionization mechanisms of methanol, *Intern. J. Mass Spectrom. Ion Phys.* **35**, 151 (1980).
- [5174] Nishimura, T., Niwa, Y., Tsuchiya, T., and Nozoye, H. Ionic dissociation of methanol studied by photoelectron-photoion coincidence spectroscopy, *J. Chem. Phys.* **72**, 2222 (1980).
- [5175] Powis, I. The dissociation of state-selected CF₃X⁺ molecular ions, *Mol. Phys.* **39**, 311 (1980).
- [5176] Stephan, K., Helm, H., Kim, Y. B., Seykora, G., Ramler, J., Grössl, M., Märk, E., and Märk, T. D. Single and double ionization of nitrogen dioxide by electron impact from threshold up to 180 eV, *J. Chem. Phys.* **73**, 303 (1980).
- [5177] Pittermann, U., and Weil, K. G. Massenspektrometrische Untersuchungen an Silberhalogeniden V: Verdampfung von Silberiodid, *Ber. Bunsenges. Phys. Chem.* **84**, 542 (1980).
- [5178] Imre, D., and Koenig, T. The He(I) photoelectron spectrum of atomic iodine by photodissociation of molecular iodine, *Chem. Phys. Letters* **73**, 62 (1980).
- [5179] Reader, J. Energy levels of singly ionized cesium (Cs II), *Phys. Rev. A* **13**, 507 (1976).
- [5180] Reader, J. Spectrum and energy levels of singly ionized rubidium (Rb II), *J. Opt. Soc. Am.* **65**, 286 (1975).
- [5181] Rosenstock, H. M., Stockbauer, R., and Parr, A. C. Kinetic shift in chlorobenzene ion fragmentation and the heat of formation of the phenyl ion, *J. Chem. Phys.* **71**, 3708 (1979).
- [5182] Nielsen, U., and Schwarz, W. H. E. VUV spectra of the xenon fluorides, *Chem. Phys.* **13**, 195 (1976).
- [5183] Causley, G. C., and Russell, B. R. Vacuum ultraviolet absorption spectra of dichlorosilane, dichloromethylsilane and dichlorodimethylsilane, *J. Electron Spectrosc. Relat. Phenom.* **8**, 71 (1976).
- [5184] Maier, J. P., Marthaler, O., and Mohraz, M. Emission spectra of the cations of some fluorosubstituted phenols in the gaseous phase, *J. Electron Spectrosc. Relat. Phenom.* **19**, 11 (1980).
- [5185] Brown, R. S., Influence of remote substituents on ionization potential. Part II. Enamines, *Can. J. Chem.* **54**, 1521 (1976).
- [5186] Worden, E. F., Solarz, R. W., Paisner, J. A., and Conway, J. G. First ionization potentials of lanthanides by laser spectroscopy, *J. Opt. Soc. Am.* **68**, 52 (1978).
- [5187] Leutwyler, S., Herrmann, A., Wöste, L., and Schumacher, E. Isotope selective two-step photoionization study of K₂ in a supersonic molecular beam, *Chem. Phys.* **48**, 253 (1980).
- [5188] Ihle, H. R., Wu, C. H., Miletic, M., and Zmbov, K. F. Mass spectrometric studies of gas species in the systems Si–Cl and Si–Li, *Adv. Mass Spectrom.* **7A**, 670 (1978).
- [5189] Ajò, D., Granozzi, G., Tondello, E., and Fraga^{II}, I. Nature of the metal–metal bond in *triangulo-Ru₃(CO)₁₂* from UV photoelectron spectroscopy and quantum mechanical calculations, *Inorg. Chim. Acta* **37**, 191 (1979).
- [5190] Betteridge, D., Thompson, M., Baker, A. D., and Kemp, N. R. Photoelectron spectra of phosphorus halides, alkyl phosphites and phosphates, organophosphorus pesticides, and related compounds, *Anal. Chem.* **44**, 2005 (1972).
- [5191] Bursten, B. E., Cotton, F. A., Cowley, A. H., Hanson, B. E., Lattman, M., and Stanley, G. S. Strong metal-to-metal quadruple bonds in a series of five isostructural compounds as indicated by photoelectron spectroscopy, *J. Am. Chem. Soc.* **101**, 6244 (1979).
- [5192] Spanget-Larsen, J., Gleiter, R., De Meijere, A., and Binger, P. Linear combination of Walsh orbitals in tris- σ -homobenzenes, *Tetrahedron* **35**, 1385 (1979).
- [5193] Carnovale, F., Gan, T. H., Peel, J. B., and Franz, K.-D. Photoelectron spectroscopic studies of some alkoxy phenalenones, *Tetrahedron* **35**, 129 (1979).
- [5194] Böhml, M. C., and Gleiter, R. Electronic structure and reactivity of propellanes, *Tetrahedron* **35**, 675 (1979).
- [5195] Moseley, J. T., Saxon, R. P., Huber, B. A., Cosby, P. C., Abouaf, R., and Tadjeddine, M. Photofragment spectroscopy and potential curves of Ar₂⁺, *J. Chem. Phys.* **67**, 1659 (1977).
- [5196] Jochims, H. W., Lohr, W., and Baumgärtel, H. Photoreactions of small organic molecules V. Absorption-, photoion- and resonancephotoelectronspectra of CF₃Cl, CF₂Cl₂, CFCl₃ in the energy range 10–25 eV, *Ber. Bunsenges.* **80**, 130 (1976).
- [5197] Mattsson, L., Karlsson, L., Jadrny, R., and Siegbahn, K. Valence electron spectrum of C₆H₆ excited by linearly polarized HeI radiation, *Phys. Scripta* **16**, 221 (1977).
- [5198] Manne, R., Wittel, K., and Mohanty, B. S. Spin-orbit interaction in molecular photoelectron spectra An intermediate coupling approach, *Mol. Phys.* **29**, 485 (1975).
- [5199] McDiarmid, R. On the ultraviolet spectrum of *trans*-1,3-butadiene, *J. Chem. Phys.* **64**, 514 (1976).
- [5200] Harris, D., McKinnon, S., and Boyd, R. K. The origins of the base peak in the electron impact spectrum of limonene, *Org. Mass Spectrom.* **14**, 265 (1979).
- [5201] Willett, G. D., and Baer, T. Thermochemistry and dissociation dynamics of state-selected C_nH_nX ions. 3. C₄H₃N⁺ *J. Am. Chem. Soc.* **102**, 6774 (1980).
- [5202] Rücker, C., Lang, D., Sauer, J., Friege, H., and Sustmann, R. Reaktivität substituierter 1,3-Butadiene in Diels-Alder-Reaktionen, *Chem. Ber.* **113**, 1663 (1980).
- [5203] Ciliberto, E., Costanzo, L. L., Fraga^{II}, I., and Granozzi, G. Ultraviolet photoelectron spectra of ‘octahedral’ (diethyldithiophosphato) metal(III) complexes, *Inorg. Chim. Acta* **44**, L25 (1980).
- [5204] Bock, H., and Brähler, U. G. Oxidation und Reduktion methylthio-substituierter Naphthaline – ein Vergleich von Molekülezuständen, *Chem. Ber.* **112**, 3081 (1979).
- [5205] Locht, R., Olivier, J. L., and Momigny, J. Dissociative autoionization as a mechanism for the proton formation from methane and methane-d₄ by low energy electron impact, *Chem. Phys.* **43**, 425 (1979).
- [5206] Fraga^{II}, I., Ciliberto, E., Granozzi, G., and Deganello, G. He-I and He-II excited photoelectron spectra of cycloheptatrienetricarbonyl complexes of group VIA metals, *J. Organometal. Chem.* **182**, 511 (1979).
- [5207] Solouki, B., Bock, H., and Appel, R. Sequence of orbitals in sulfones and sulfodiimides, *Angew. Chem. Int. Ed.* **11**, 927 (1972).
- [5208] Jonathan, N., Morris, A., Okuda, M., Smith, D. J., and Ross, K. J. Photoelectron spectroscopy of transient species: The CS molecule, *Chem. Phys. Letters* **13**, 334 (1972).
- [5209] Čermák, V. Electron spectroscopy of autoionizing states of

- oxygen, chlorine and bromine atoms, *J. Electron Spectrosc. Relat. Phenom.* **6**, 135 (1975).
- [5210] Davis, R., Ojo, I. A., and Webb, M. L. Mass spectrometry of transition-metal π -complexes, V.-Fragmentation and structure of chromium-coordinated $[\text{C}_7\text{H}_{11}]^+$ and $[\text{C}_8\text{H}_{10}]^+$ ions, *Org. Mass Spectrom.* **13**, 547 (1978).
- [5211] Hall, D., Maier, J. P., and Rosmus, P. Electronic states of ketene radical cation, *Chem. Phys.* **24**, 373 (1977).
- [5212] Coughlin, D. J., Brown, R. S., and Salomon, R. G. The prostaglandin endoperoxide nucleus and related bicyclic peroxides. Synthetic and spectroscopic studies, *J. Am. Chem. Soc.* **101**, 1533 (1979).
- [5213] Daamen, H., Oskam, A., Stufkens, D. J., and Waaijers, H. W. Bonding properties of group VIB metal pentacarbonyl azole complexes studied by electronic absorption, photoelectron, ^{13}C NMR and vibrational spectroscopy, *Inorg. Chim. Acta* **34**, 253 (1979).
- [5214] Kimura, K., Yamazaki, T., and Achiba, Y. He I (584 Å) photoelectron spectra and photoionization cross sections of atomic chlorine and bromine, *Chem. Phys. Letters* **58**, 104 (1978).
- [5215] Katritzky, A. R., Baker, V. J., Brito-Palma, F. M. S., Patel, R. C., Pfister-Guillouzo, G., and Guimon, C. Conformational analysis of saturated heterocycles. Part 94. Applications of photoelectron spectroscopy to molecular properties. Part 3. Photoelectron spectroscopic studies of some 1-oxa-3,4-diaza-, 1-thia-3,4-diaza-, 1,2,4-triaza-, and 1,2,4,5-tetraaza-cyclohexanes, *J. Chem. Soc. Perkin II* **91**, (1980).
- [5216] Bock, H., Bowling, R. A., Solouki, B., Barton, T. J., and Burns, G. T. Analysis and optimization of gas phase reactions. 13. ^{13}Si Latoluene, *J. Am. Chem. Soc.* **102**, 429 (1980).
- [5217] Fragalà, I., Ciliberto, E., and Thomas, J. L. He(I) and He(II) excited photoelectron spectra of cyclopentadienyldicarbonyltitanium(II), *J. Organometal. Chem.* **175**, C25 (1979).
- [5218] Chau, F. T., and McDowell, C. A. The HeI photoelectron spectrum of *trans*-1,2-dibromocyclohexane, *J. Mol. Struct.* **34**, 93 (1976).
- [5219] Syrvatka, B. G., and Gil'burd, M. M. Mass-spectrometric study of trifluorotriazosmethane, *Russ. J. Phys. Chem.* **47**, 1215 (1973).
- [5220] Dunlavy, S. J., Dyke, J. M., and Morris, A. The first ionization potential of the $\text{BrO}(\text{X}^2\text{H}_2)$ radical obtained using photoelectron spectroscopy, *Chem. Phys. Letters* **53**, 382 (1978).
- [5221] Colin, R., and De Greef, D. The absorption spectrum of the BeH and BeD molecules in the vacuum ultraviolet, *Can. J. Phys.* **53**, 2142 (1975).
- [5222] Guimon, C., Pfister-Guillouzo, G., and Arbelot, M. Spectres photoélectroniques de la dithiole-1,3-thione-2 et de son dérivé benzosubstitué, *J. Mol. Struct.* **30**, 339 (1976).
- [5223] Hillier, I. H., Guest, M. F., Higginson, B. R., and Lloyd, D. R. *Ab initio* calculations of transition metal complexes V. The electronic structure and He(I) photoelectron spectra of $\text{Fe}(\text{CO})_2(\text{NO})_2$, $\text{Co}(\text{CO})_3\text{NO}$ and $\text{Ni}(\text{CO})_4$, *Mol. Phys.* **27**, 215 (1974).
- [5224] Kováčik, V., Mihálov, V., and Kováč, P. Identification of methyl (methyl *O*-acetyl-*O*-methyl hexopyranosid)uronates by mass spectrometry, *Carbohydr. Res.* **54**, 23 (1977).
- [5225] Guimon, C., Pfister-Guillouzo, G., Bernardini, A., and Viallefond, P. A photoelectron study (HeI, HeII) of the tautomeric equilibrium of chloro- and bromo-1,2,4-triazoles, *Tetrahedron* **36**, 1071 (1980).
- [5226] Piacente, V., Bardi, G., di Paolo, V., and Ferro, D. The vapour pressure over Ga_2S_2 and Ga_2Se_2 , *J. Chem. Thermodyn.* **8**, 391 (1976).
- [5227] Kuck, D., and Grützmacher, H.-Fr. The activation energy of the skeletal isomerization in the radical cations of toluene and cycloheptatriene by mass spectrometry of their 2-phenylethyl derivatives, *Org. Mass Spectrom.* **14**, 86 (1979).
- [5228] Shaw, R. W., Jr., and Thomas, T. D. Auger electron spectrum and ionization potentials of the HF molecule, *Phys. Rev. A* **11**, 1491 (1975).
- [5229] Sell, J. A., and Kupperman, A. Angular distributions in the photoelectron spectroscopy of SF_6 , *Chem. Phys.* **33**, 379 (1978).
- [5230] Johansson, S. The spectrum and term system of Fe II, *Physica Scripta* **18**, 217 (1978).
- [5231] Paguette, L. A., Ku, A. Y., Santiago, C., Rozenboom, M. D., and Houk, K. N. Control of regioselectivity in the di- π -methane rearrangement. Tripletsensitized photoisomerization of benzonorbornadienes carrying cyano substituents in the aryl and vinyl segments, *J. Am. Chem. Soc.* **101**, 5972 (1979).
- [5232] Lassiter, T. W., Allen, J. D., Jr., and Schweitzer, G. K. The photoelectron spectroscopic characterization of vapors above heated alkali tetrafluoroaluminates, alkali tetrachloroaluminates, and ammonium tetrachloroaluminate, *J. Electron Spectrosc. Relat. Phenom.* **19**, 321 (1980).
- [5233] Biefeld, R. M. The vaporization thermodynamics of rubidium iodide as determined by mass-loss Knudsen effusion and mass spectrometry, *J. Chem. Thermodyn.* **10**, 907 (1978).
- [5234] Mark, T. D., and Hille, E. Cross section for single and double ionization of carbon dioxide by electron impact threshold up to 180 eV, *J. Chem. Phys.* **69**, 2492 (1978).
- [5235] Stadelmann, J. P., and Vogt, J. A photoelectronphotoion coincidence study of *cis*- and *trans*- difluoroethene, *Intern. J. Mass Spectrom. Ion Phys.* **35**, 83 (1980).
- [5236] Miletic, M., Ereš, D., Veljković, M., and Zmbov, K. F. Mass spectrometric study of the ionization and fragmentation of carbon disulphide by mono- energetic electron impact, *Intern. J. Mass Spectrom. Ion Phys.* **35**, 231 (1980).
- [5237] Ajò, D., Ciliberto, E., Fragalà, I., and Granozzi, G. Lone-pair interactions in the photoelectron spectra of dicarboxylic acids: Malonic acid and its α -alkyl derivatives, *J. Mol. Struct.* **62**, 189 (1980).
- [5238] Selim, E. T. M. Ionization and dissociation of propylene by electron impact, *Indian J. Pure Appl. Phys.* **18**, 31 (1980).
- [5239] Hochmann, P., Templet, P. H., Wang, H.-t., and McGlynn, S. P. Molecular Rydberg transitions. I. Low- energy Rydberg transitions in methyl halides, *J. Chem. Phys.* **62**, 2588 (1975).
- [5240] Domelsmith, L. N., Houk, K. N., Piedrahita, C., and Dolbier, W. J., Jr. The photoelectron spectrum of 1,1-difluoroallene. On π electron donation and withdrawal by fluorine, *J. Am. Chem. Soc.* **100**, 6908 (1978).
- [5241] Palenius, H. P., Huffman, R. E., Larrabee, J. C., and Tanaka, Y. The absorption spectrum of fluorine F I observed with the helium continuum, *J. Opt. Soc. Am.* **68**, 1564 (1978).
- [5242] Wiberg, N., Fischer, G., and Bachhuber, H. Diazlen und andere Distickstoffhydride: Bildungswärmen, Dissoziationsenergien, Auftrittspotentiale, Protonenaffinitäten, *Z. Naturforsch.* **34b**, 1385 (1979).
- [5243] Utsunomiya, C., Kobayashi, T., and Nagakura, S. Photoelectron angular distribution measurements for some aliphatic alcohols, amines, and halides, *Bull. Chem. Soc. Japan* **53**, 1216 (1980).
- [5244] Fragalà, I., Ciliberto, E., Granozzi, G., and Deganello, G. He-I and He-II excited photoelectron spectra of cycloheptatrienetricarbonyl complexes of group VIA metals, *J. Organometal. Chem.* **182**, 511 (1979).
- [5245] Carnovale, F., Gan, T. H., and Peel, J. B. Photoelectron spectroscopic studies of the monomers and dimers of

- acetic and trifluoracetic acids, J. Electron Spectrosc. Relat. Phenom. **20**, 53 (1970).
- [5252] Trudell, B. C., and Price, S. J. W. The ultraviolet photoelectron spectra of C_6F_5X compounds, $X=(F,Cl,Br,I,H,CH_3)$, Can. J. Chem. **57**, 2256 (1979).
- [5253] Colbourne, D., Frost, D. C., McDowell, C. A., and Westwood, N. P. C. The vacuum ultraviolet photoelectron spectrum of difluoramine, Chem. Phys. Letters **72**, 247 (1980).
- [5254] Wu, C. H. Binding energies of LiH_2 and LiH_2^+ and the ionization potential of LiH_2 , J. Chem. Phys. **71**, 783 (1979).
- [5255] Dickson, R. S., Carnovale, F., and Peel, J. B. A photoelectron spectroscopic study of $[Rh(CO)_2Cl]_2$, J. Organometal. Chem. **179**, 115 (1979).
- [5256] Potts, A. W., and Fattahallah, G. H. High-resolution ultraviolet photoelectron spectroscopy of CO_2 , COS and CS_2 , J. Phys. B: Atom. Molec. Phys. **13**, 2545 (1980).
- [5257] Potts, A. W., Lyus, M. L., Lee, E. P. F., and Fattahallah, G. H. High resolution ultraviolet photoelectron spectra of C_6H_5X and $p-C_6H_4X_2$ where $X = Cl, Br$ or I , J. Chem. Soc. Faraday II, **76**, 556 (1980).
- [5258] Klasinc, L., Novak, I., Scholz, M., and Kluge, G. Photoelektronenspektren substituierter Pyridine und Benzole und ihre Interpretation durch die CNDO/SWW-Methode, Croat. Chem. Acta **51**, 43 (1978).
- [5259] Palmer, M. H., Moyes, W., and Spiers, M. The electronic structure of substituted benzenes: Ab initio calculations and photoelectron spectra for benzonitrile, the tolunitriles, fluorobenzonitriles, dicyanobenzenes and ethynylbenzene, J. Mol. Struct. **62**, 165 (1980).
- [5260] Arnold, D. R., and Wong, P. C. The oxidation potentials of *cis*- and *trans*-1,2-diphenylcyclopropane and *cis*- and *trans*-2,3-diphenyloxirane, Can. J. Chem. **57**, 2098 (1979).
- [5261] Drake, J. E., Glavinčevski, B. M., and Gorzelska, K. The photoelectron spectra of dimethylgermane, difluoro- and dichlorodimethyl germane, Can. J. Chem. **57**, 2278 (1979).
- [5262] Nomoto, K., Achiba, Y., and Kimura, K. HeI and HeII photoelectron study of N_2O_4 , Bull. Chem. Soc. Japan **52**, 1614 (1979).
- [5263] Holmes, J. L., and Lossing, F. P. Thermochemistry and unimolecular reactions of ionized acetic acid and its enol in the gas phase, J. Am. Chem. Soc. **102**, 3732 (1980).
- [5264] Morgan, R. P., Derrick, P. J., and Loudon, A. G. Kinetics and mechanisms of the decompositions of the molecular ions of pentanal and its monomethyl- substituted homologues in the picosecond to microsecond time interval following field ionization, J. Chem. Soc. Perkin II, 306 (1980).
- [5265] Scharf, H.-D., Plum, H., Fleischhauer, J., and Schleker, W. Zur Diels-Alder-Reaktivität *s-cis*- fixierter 1,3-Diene, Chem. Ber. **112**, 862 (1979).
- [5266] Appell, J., and Horsley, J. A. Electronic states of doubly ionized ammonia, J. Chem. Phys. **60**, 3445 (1974).
- [5267] Holmes, J. L., Yuan, D., and Rye, R. T. B. Metastable ion studies, VII-Loss of water from the molecular ion of cyclopentanol, Org. Mass Spectrom. **12**, 254 (1977).
- [5268] Holmes, J. L., Weese, G. M., Blair, A. S., and Terlouw, J. K. Metastable ion studies IX-Thermochemistry and ion structures among fragmenting $[C_4H_8]^+$ ions, an electron impact and field ionization investigation, Org. Mass Spectrom. **12**, 424 (1977).
- [5269] Domcke, W., Cederbaum, L. S., Schirmer, J., Von Niessen, W., and Maier, J. P. Breakdown of the molecular orbital picture of ionization for inner valence electrons: experimental and theoretical study of H_2S and PH_3 , J. Electron Spectrosc. Relat. Phenom. **14**, 59 (1978).
- [5270] Jochims, H. W., Lohr, W., and Baumgärtel, H. Photoionization mass spectrometry studies of deuterated acetaldehydes CH_3CDO and CD_3CHO , Chem. Phys. Letters **54**, 594 (1978).
- [5271] Schäfer, W., Schweig, A., Dimroth, K., and Kanter, H. Nature of bonding in λ^3 -phosphorins, J. Am. Chem. Soc. **98**, 4410 (1976).
- [5272] Kobayashi, T., and Nagakura, S. Photoelectron spectra of substituted benzenes, Bull. Chem. Soc. Japan **47**, 2563 (1974).
- [5273] Ogata, H., Kitayama, J., Koto, M., Kojima, S., Nihei, Y., and Kamada, H. Vacuum ultraviolet absorption and photoelectron spectra of aliphatic ketones, Bull. Chem. Soc. Japan **47**, 958 (1974).
- [5274] Williamson, A. D., Compton, R. N., and Eland, J. H. D. Accurate photoionization thresholds by multiphoton ionization: Pyrrole, J. Chem. Phys. **70**, 590 (1979).
- [5275] Rauh, E. G., and Ackermann, R. J. Erratum: First ionization potentials of some refractory oxide vapors, J. Chem. Phys. **64**, 1862 (1976).
- [5276] Potzinger, P., Ritter, A., and Krause, J. Massenspektrometrische Bestimmung von Bindungsenergien in siliciumorganischen Verbindungen, Z. Naturforsch. **30a**, 347 (1975).
- [5277] Nakato, Y., Ozaki, M., and Tsubomura, H. Ionization energies and Rydberg states of tetraaminoethylenes, Bull. Chem. Soc. Japan **45**, 1299 (1972).
- [5278] Nakato, Y., Chiyoda, T., and Tsubomura, H. Experimental determination of ionization potentials of organic amines, β -carotene and chlorophyll a, Bull. Chem. Soc. Japan **47**, 3001 (1974).
- [5279] Orlov, V. M., Varshavsky, Y. M., and Miroshnikov, A. I. Photoionisation mass spectra of volatile derivatives of short peptides and appearance potentials of their characteristic ions, Org. Mass Spectrom. **9**, 811 (1974).
- [5280] Nelsen, S. F., Peacock, V., and Weisman, G. R. Single-electron oxidation equilibria of tetraalkylhydrazines. Comparison of solution E° values and vapor-phase ionization potentials, J. Am. Chem. Soc. **98**, 5269 (1976).
- [5281] Böhm, M. C., Gleiter, R., and Batich, C. D. 106. The photoelectron spectra of $Ni(Pd,Pt-Diallyl)$, Helv. Chim. Acta **63**, 990 (1980).
- [5282] Holmes, J. L., and Lossing, F. P. The reactivity of $[C_3H_3^+]$ ions; a thermochemical study, Can. J. Chem. **57**, 249 (1979).
- [5283] Butler, J. J., and Baer, T. Thermochemistry and dissociation dynamics of state-selected C_4H_4X ions. 1. Thiophene, J. Am. Chem. Soc. **102**, 6764 (1980).
- [5284] Wolkoff, P., and Holmes, J. L. Fragmentations of alkane molecular ions, J. Am. Chem. Soc. **100**, 7346 (1978).
- [5285] Cauletti, C., Tarli, F., Monaci, A., Bonapasta, A. A., and Bossa, M. Ultraviolet photoelectron spectra of some methyl esters of dithiocarbazic acids and of $[Ni\{N(CH_3)_2-N=C(SCH_3)\}_2]$ and comparison with quantum-mechanical calculations, J. Chem. Soc. Dalton 1087 (1980).
- [5286] Connor, J. A., Martinho-Simoes, J. A., Skinner, H. A., and Zafarani-Moattar, M. T. Thermochemistry of *bis*-Arene- and Arenetricarbonyl-chromium compounds containing hexamethylbenzene, 1,3,5-trimethylbenzene and naphthalene, J. Organometal. Chem. **179**, 331 (1979).
- [5287] Gusel'nikov, L. E., and Nametkin, N. S. 1,1-dimethyl-1-silaethylene. Heat of formation, ionization potential and the energy of the silicon-carbon π -bond, J. Organometal. Chem. **169**, 155 (1979).
- [5288] Rademacher, P., and Freckmann, B. Photoelektronenspektren und Konformations-verhalten von Hydroxylamin und Methylhydroxylaminen, J. Electron Spectrosc. Relat. Phenom. **19**, 251 (1980).
- [5289] Willett, G. D., and Baer, T. Thermochemistry and dissociation dynamics of state-selected C_4H_4X ions. 2. Furan and 3-butyn-2-one, J. Am. Chem. Soc. **102**, 6769 (1980).
- [5290] Tang, S.-Y., McGowan, J. C., Singh, M., Galatsis, P., Ellis, B.

- E., Boyd, R. K., and Brown, S. A. Mass spectrometry of some furanocoumarins, *Can. J. Chem.* **57**, 1995 (1979).
- [5291] Michels, G. D., Flesch, G. D., and Svec, H. J. Comparative mass spectrometry of the group 6B hexacarbonyls and pentacarbonyl thiocarbonyls, *Inorg. Chem.* **19**, 479 (1980).
- [5292] Distefano, G., Eoffani, A., Innorta, G., and Pignataro, S. Electron impact ionization potentials of some manganese, chromium and tungsten organometallic derivatives, *Int. J. Mass Spectrom. Ion Phys.* **7**, 383 (1971).
- [5293] McLoughlin, R. G., Morrison, J. D., and Traeger, J. C. Photoionization of the C-1 - C-4 monosubstituted alkyl benzenes: Thermochemistry of $[C_6H_5]^+$ and $[C_6H_4]^+$ formation, *Org. Mass Spectrom.* **14**, 104 (1979).
- [5294] Neubert, A. Mass spectrometric determination of the atomization energies of Te_n ($n=3-7$) molecules, *High Temp. Sci.* **10**, 261 (1978).
- [5295] Noodeman, L., Westwood, N. P. C., and Mitchell, K. A. R. Ionization energies and electronic structure of $N_3P_3Cl_6$ as determined by UV photoelectron spectroscopy and the Xα scattered wave method, *Chem. Phys. Letters* **58**, 252 (1978).
- [5296] Hilpert, K. Mass spectrometric determination of the dissociation energies of CuTb(g), CuDy(g), and CuHo(g), *Ber. Bunsenges. Phys. Chem.* **83**, 161 (1979).
- [5297] Berkowitz, J., Batson, C. H., and Goodman, G. L. Photoelectron spectroscopy of AgCl, AgBr, and AgI vapors, *J. Chem. Phys.* **72**, 5829 (1980).
- [5298] Ernsting, N. P., Pfab, J.; Green, J. C., and Römel, J. Photoelectron spectra of nitrosomethane, t- nitrosobutane and some perhalogenonitrosomethanes, *J. Chem. Soc. Faraday II* **76**, 844 (1980).
- [5299] Trott, W. M., Blais, N. C., and Walters, E. A. Photoionization of carbon disulfide monomers and dimers in a supersonic molecular beam, *J. Chem. Phys.* **71**, 1692 (1979).
- [5300] Creber, D. K., and Bancroft, G. M. Photoelectron studies of dialkyl group 2B compounds: Ligand field splittings and intensity variations with photon energy, *Inorg. Chem.* **19**, 643 (1980).
- [5301] Rademacher, P., and Freckmann, B. Photoelectron spectra and conformations of tetrahydro-1,2- oxazines and isoxazolines, *Tetrahedron Letters* 841 (1978).
- [5303] Haque, R., Pelino, M., and Gingerich, K. A. Investigation of the thermodynamic stability of the molecule LaIr(g) by high temperature mass spectrometry, *J. Chem. Phys.* **71**, 2929 (1979).
- [5304] Carnovale, F., Gan, T.-H., and Peel, J. B. The photoelectron spectra of the *N*-chloro and *N*-bromo derivatives of dimethylamine, *Aust. J. Chem.* **32**, 719 (1979).
- [5305] Mohraz, M., Maier, J. P., and Heilbronner, E. He(Iα) and He(IIα) photoelectron spectra of fluorinated chloro-and bromo-benzenes, *J. Electron Spectrosc. Relat. Phenom.* **19**, 429 (1980).
- [5306] Gupta, S. K., and Gingerich, K. A. A thermodynamic study of the gaseous thorium carbides, ThC , ThC_2 , ThC_3 , ThC_4 , ThC_5 , and ThC_6 , *J. Chem. Phys.* **72**, 2795 (1980).
- [5307] Tiedemann, P. W., Anderson, S. L., Ceyer, S. T., Hirooka, T., Ng, C. Y., Mahan, B. H., and Lee, Y. T. Proton affinities of hydrogen halides determined by the molecular beam photoionization method, *J. Chem. Phys.* **71**, 605 (1979).
- [5308] Gan, T.-H., and Peel, J. B. Photoelectron spectroscopic studies of piperidine and its *N*-halo derivatives, *Aust. J. Chem.* **32**, 475 (1979).
- [5309] Guimon, C., Pfister-Guillouzo, G., Begtrup, M. Photoelectron spectra of some pyrazolthiones, 1,2,3-triazolthiones, and 4-(1,2,3-triazolio) sulfides. Evidence of an abnormal effect of methylation, *J. Am. Chem. Soc.* **100**, 1275 (1978).
- [5310] Friege, H., and Klessinger, M. Elektronenstruktur von Alkyl-aryl-und Alkyl-vinyl-ethern, *Chem. Ber.* **112**, 1614 (1979).
- [5311] Potzinger, P., Stracke, H.-U., Küpper, W., and Gollnick, K. Ionisierungs- und Auftrittspotentialmessungen an Dialkylsulfoxiden, *Z. Naturforsch.* **30a**, 340 (1975).
- [5313] Bieri, G., Schmelzer, A., Asbrink, L., and Jonsson, M. Fluorine and the fluoroderivatives of acetylene and diacetylene studied by 30.4 nm He(II) photoelectron spectroscopy, *Chem. Phys.* **49**, 213 (1980).
- [5314] Martin, H.-D., Heller, C., Mayer, B., and Beckhaus, H.-D. Synthese bicyclischer, nicht-konjugierter Polyene. Stereochemie und transannulare Wechselwirkungen, *Chem. Ber.* **113**, 2589 (1980).
- [5315] Mohraz, M., Jian-qi, W., Heilbronner, E., Vogel, P., and Pilet, O. 57. Radical cation states of 2,3,5,6,7,8-hexamethylidenebicyclo[2.2.2]octane, *Helv. Chim. Acta* **63**, 568 (1980).
- [5316] Broer, W. J., and Weringa, W. D. Potential energy profiles for the unimolecular reactions of $[C_3H_7S]^+$ ions, *Org. Mass Spectrom.* **15**, 229 (1980).
- [5317] Granozzi, G., Tondello, E., Bénard, M., and Fraga, I. Electronic structure of *trans*- $[(\eta^5-C_5H_5Fe(CO)_2)_2]$ by He(I) and He(II) photoelectron spectroscopy and ab initio calculations, *J. Organometal. Chem.* **194**, 83 (1980).
- [5318] Kondo, T., Tanimoto, M., Matsumoto, M., Nomoto, K., Achiba, Y., and Kimura, K. Cyclic peroxides: Dihedral angle around the peroxide bond by microwave and photoelectron spectroscopic studies, *Tetrahedron Letters* **21**, 1649 (1980).
- [5319] Bock, H., and Kaim, W. Reduktion R_3SiO -substituierter Benzol-Derivate, *Z. Anorg. Allg. Chem.* **459**, 103 (1979).
- [5320] Millefiori, S., Millefiori, A., Pignataro, S., Distefano, G., and Colonna, F. P. Gas phase UPS investigation of *trans*-azobenzenes, *Z. Naturforsch.* **34a**, 1496 (1979).
- [5321] Harris, D. H., and Spalding, T. R. Electron impact study of compounds with main group IV element to transition metal bonds, *Inorg. Chim. Acta* **39**, 187 (1980).
- [5322] Rademacher, P., Bass, V.-M., Wildemann, M., and Weger, H. Photoelektronenspektren und Konformation von Hydrazobenzolen, *Chem. Ber.* **110**, 1939 (1977).
- [5323] Colonna, F. P., Distefano, G., Guerra, M., Jones, D., and Modelli, A. Furyl- and thienyl-mercury derivatives studied by means of ultraviolet photoelectron spectroscopy. Evidence for the participation in bonding of the vacant $6p\pi$ orbitals of mercury in bis-2-furyl- and bis-2-thienylmercury, *J. Chem. Soc. Dalton*, 2037 (1979).
- [5324] Fehlner, T. P., Wu, M., Meneghelli, B. J., and Rudolph, R. W. Ultraviolet photoelectron spectroscopy of thiaboranes, *Inorg. Chem.* **19**, 49 (1980).
- [5325] Martin, H.-D., and Heller, C. Synthese homokonjugierter Polyene: 2,4-Dimethylenbicyclo[3.2.0]oct-6-en und 2,5-Dimethylenbicyclo[4.2.0]non-7-en, *Monatsh. Chem.* **110**, 1271 (1979).
- [5326] Modelli, A., Innorta, G., and Torroni, S. He(I) UPS spectra of some α -diazoketones, *J. Electron Spectrosc. Relat. Phenom.* **18**, 359 (1980).
- [5327] Nixon, J. F., Suffolk, R. J., Taylor, M. J., Norman, J. G., Jr., Hoskins, D. E., and Gmur, D. J. Photoelectron and electronic spectra of $Rh_2Cl_2(CO)_4$ and $Rh_2Cl_2(PF_3)_4$. Assignments from SCF-Xα-SW calculations, *Inorg. Chem.* **19**, 810 (1980).
- [5328] Zverev, V. V., Villem, Y. Y., Islamov, R. G., and Kitaev, Y. P. Photoelectron spectrum and electron structure of vinylphosphonic dichloride $Cl_2P(O)CH=CH_2$, *Zh. Obshch. Khim.* **49**, 1737 (1979).
- [5329] Colbourne, D., Frost, D. C., McDowell, C. A., and Westwood, N. P. C. The photoelectron spectra of the methylbromamines and unsubstituted bromamines, *Can. J. Chem.* **57**, 1279 (1979).
- [5330] Bernauer, O., Busse, B., and Weil, K. G. Massenspek-

- trometrische Untersuchungen an Kupfer- und Silberhalogeniden IV: Die Systeme $\text{Ag}_x\text{Cu}_{1-x}\text{Cl}$ ($0 \leq x \leq 1$) und $\text{AgCl}_x\text{Br}_{1-x}$ ($0 \leq x \leq 1$), Ber. Bunsenges. Phys. Chem. **83**, 603 (1979).
- [5331] Gingerich, K. A., and Cocke, D. L. The atomization energies of ternary cerium–platinum metal monocarbides, Inorg. Chim. Acta **33**, L107 (1979).
- [5332] Kaim, W., and Bock, H. Radical ions XVII. Radical ions of tetrakis(trimethylsilyl)butatriene, J. Organometal. Chem. **164**, 281 (1979).
- [5333] English, A. M., Plowman, K. R., Butler, I. S., Diemann, E., and Müller, A. He(I) photoelectron spectra of pentacarbonyl(selenocarbonyl)chromium(0) and related complexes, Inorg. Chim. Acta **32**, 113 (1979).
- [5334] Wu, C. H., and Ihle, H. R. The existence of the molecule Cl_3 and its ionization potential, Chem. Phys. Letters **61**, 54 (1979).
- [5335] Martin, H.-D., Heller, C., Haider, R., Hoffmann, R. W., Becherer, J., and Kurz, H. R. Vertikale oder nichtvertikale Stabilisierung im Pentacyclo-[4.3.0.0^{2,1}.0^{3,8}.0^{5,7}]non-9-yl-Kation? Das 9-Methylenpentacyclononan-Radikalkation, Chem. Ber. **110**, 3010 (1977).
- [5336] Weschke, W., Timpe, H.-J., and Just, G. Untersuchungen an EDA-Komplexen von Benzylarylethern, J. Prakt. Chem. **321**, 605 (1979).
- [5337] Mathur, B. P., Abbey, L. E., Burgess, E. M., and Moran, T. F. Doubly charged ion mass spectra, Org. Mass Spectrom. **15**, 312 (1980).
- [5338] Fragalà, I., Condorelli, G., Tondello, A., and Cassol, A. Photoelectron spectroscopy of f-element coordination compounds. 1. β -diketonate complexes of uranium(IV), thorium(IV), zirconium(IV), and dioxouranium(VI), Inorg. Chem. **17**, 3175 (1978).
- [5339] Schmidt, H., Schweig, A., Thiel, W., and Jones, M., Jr. Photoelektronenspektren und MNDO-Rechnungen für [n]Paracyclophane, Chem. Ber. **111**, 1958 (1978).
- [5340] Sahini, V. E., Constantin, V., Serban, I., and Vlădescu, C. Electron impact-induced fragmentation of dibenzo/b,e/oxepin-11-one and dibenzo/b,e/ thiepin-11-one, Rev. Roum. Chim. **23**, 163 (1978).
- [5341] Gotthardt, H., Reiter, F., Gleiter, R., and Bartetzko, R. Thieno[3,4-c]isothiazol. Synthese und Eigenschaften eines neuen nichtklassisch kondensierten Thiophens, Chem. Ber. **112**, 260 (1979).
- [5342] Rauh, E. G., and Ackermann, R. J. The first ionization potentials of the transition metals, J. Chem. Phys. **70**, 1004 (1979).
- [5343] Egdell, R. G., Palmer, M. H., and Findlay, R. H. Electronic structure of the group 5 oxides: Photoelectron spectra and ab initio molecular orbital calculations, Inorg. Chem. **19**, 1314 (1980).
- [5344] Asmus, P., and Klessinger, M. Wechselwirkung der Walsh-Orbitale in Bicyclopropyl, Angew. Chem. **88**, 343 (1976).
- [5345] McLoughlin, R. G., and Traeger, J. C. Heat of formation for *tert*-butyl cation in the gas phase, J. Am. Chem. Soc. **101**, 5791 (1979).
- [5346] Locht, R., and Momigny, J. Dissociative ionization by low energy electron impact. Energy distribution and appearance energy of doubly ionized fragments from N_2 and O_2 , Chem. Phys. Letters **66**, 574 (1979).
- [5347] Kröner, J., and Strack, W. Geometry and n-ionization energies of alkyl-substituted triketones, Angew. Chem. Int. Ed. **11**, 220 (1972).
- [5348] Greenhough, T. J., Kolthammer, B. W. S., Legzdins, P., and Trotter, J. Organometallic nitrosyl chemistry. 10. Synthesis, X-ray structural characterization, and properties of dicarbonyl (η^5 -cyclopentadienyl)(thionitrosyl)chromium, Inorg. Chem. **18**, 3548 (1979).
- [5349] Haque, R., and Gingerich, K. A. Investigation of the thermodynamic stabilities of the gaseous molecules RhScC , RhScC_2 and RhYC_2 by Knudsen effusion mass spectrometry, J. Chem. Soc. Faraday Trans. II, 985 (1979).
- [5350] Helm, H., Stephan, K., and Märk, T. D. Electronimpact ionization of Ar_2 , ArKr , Kr_2 , KrXe , and Xe_2 , Phys. Rev. A **19**, 2154 (1979).
- [5351] Jochims, H. W., Lohr, W., and Baumgärtel, H. Photoreactions of small organic molecules V. Absorption-, photoion- and resonancephotoelectrospectra of CF_3Cl , CF_2Cl_2 , CFCl_3 in the energy range 10–25eV, Ber. Bunsenges. **80**, 130 (1976).
- [5352] Reinke, D., Baumgärtel, H., Cvitaš, T., Klasinc, L., and Güsten, H. Vergleich der Photoelektronenspektren und Photoionenspektren von Vinylfluorid, Ber. Bunsenges. **78**, 1145 (1974).
- [5353] Rademacher, P., and Koopman, H. Cyclische und bicyclische Hydrazine, Chem. Ber. **108**, 1557 (1975).
- [5354] Bowling, R. A., Sherrod, R. E., Bloor, J. E., Allen, J. D., Jr., and Schweitzer, G. K. He I photoelectron spectra of gaseous alkali nitrates, Inorg. Chem. **17**, 3418 (1978).
- [5355] Findlay, R. H., Palmer, M. H., Downs, A. J., Egdell, R. G., and Evans, R. Electronic structure of the sulfur nitrides. Ab initio calculations and photoelectron spectra, Inorg. Chem. **19**, 1307 (1980).
- [5356] Meunier, P., and Pfister-Guilhouzo, G. Etude conformationnelle par spectroscopie photoélectronique de sulfures de bithiényle, Can. J. Chem. **55**, 3901 (1977).
- [5357] Green, J. C., Mingos, D. M. P., and Seddon, E. A. UV photoelectron spectral studies of carbonylhydrido-clusters and the development of a topological bonding model, J. Organometal. Chem. **185**, C20 (1980).
- [5358] Fabian, B. D., Fehlner, T. P., Hwang, L.-S.J., and Labinger, J. A. Allylic interactions in organometallics: Probing electronic structure in ($\eta^5\text{-C}_5\text{H}_5$) $\text{Fe}(\text{CO})_2\text{R}$, $\text{R} = \text{CH}_3$, $\eta^1\text{-C}_3\text{H}_5$, $\eta^1\text{-C}_5\text{H}_5$, J. Organometal. Chem. **191**, 409 (1980).
- [5359] de Meijere, A. Dispiro[2.0.2.4]deca-7,9-dien und Vergleichsverbindungen: Darstellung, UV-, NMR- und Photoelektronenspektroskopische Untersuchungen, Chem. Ber. **107**, 1684 (1974).
- [5360] Masclet, P., and Mouvier, G. Étude par spectrométrie photoélectronique d'aldéhydes et de cétones éthyléniques conjugués, J. Electron Spectrosc. Relat. Phenom. **14**, 77 (1978).
- [5361] Hemmersbach, P., and Klessinger, M. Der Einfluss von Wechselwirkungen durch den Raum und über Bindungen auf die Walsh-Orbitale von Spirocyclopropylverbindungen, Tetrahedron **36**, 1337 (1980).
- [5362] Hill, W. E., Ward, C. H., Webb, T. R., and Worley, S. D. Electronic structures of (alkene)iron tetracarbonyl compounds, Inorg. Chem. **18**, 2029 (1979).
- [5363] Frost, D. C., Kirby, G., Lau, W. M., MacDonald, C. A., McDowell, C. A., and Westwood, N. P. C. Thiocyanogen (SCN_2). Preparation, ultraviolet photoelectron spectrum and structure, Chem. Phys. Letters **69**, 1 (1980).
- [5364] Akiyama, I., Li, K. C., LeBreton, P. R., Fu, P. P., and Harvey, R. G. Ultraviolet photoelectron studies of polycyclic aromatic hydrocarbons. The ground-state electronic structure of aryloxiranes and metabolites of benzo[a]pyrene, J. Phys. Chem. **83**, 2997 (1979).
- [5365] Reineke, W., and Strein, K. Erzeugung eines methylenhaltigen Molekularstrahls durch Pyrolyse von Keten, Ber. Bunsenges. Phys. Chem. **80**, 343 (1976).
- [5366] Dube, G., and Gey, E. Das Verhalten disubstituierter Benzole $\text{RC}_6\text{H}_4\text{Si}(\text{CH}_3)_2\text{F}$ bei Elektronenstob, Org. Mass Spectrom. **14**, 17 (1979).
- [5367] Worley, S. D., and Webb, T. R. The electronic structure of transition-metal carbonyl complexes of norbornadiene and mesitylene, J. Organometal. Chem. **192**, 139 (1980).

- [5368] Starzewski, K.-H. A. O., Richter, W., and Schmidbaur, H. Photoelektronenspektren und Struktur von Arsenyleniden. P versus As: Ein Beitrag zur Problematik der ylidischen Bindung, *Chem. Ber.* **109**, 473 (1976).
- [5369] Delmas, M. A., and Maire, J. C. Conformation du cycle dithia-1,3 stanna-2 cyclopentane: etude par RMN ¹H et par spectroscopie photoelectronique, *J. Organometal. Chem.* **161**, 13 (1978).
- [5370] Dannacher, J. Photoelectron-photoion-coincidence measurements on 2,4-headiyne, *Chem. Phys.* **29**, 339 (1978).
- [5371] Dyke, J. M., Fayad, N. K., Morris, A., Trickle, I. R., and Allen, G. C. A study of the electronic structure of the actinide tetrahalides UF₆, ThF₄, UCl₄, and ThCl₄ using vacuum ultraviolet photoelectron spectroscopy and SCF-Xα scattered wave calculations, *J. Chem. Phys.* **72**, 3822 (1980).
- [5372] Martin, H.-D., Kunze, M., and Beckhaus, H.-D., Walsh, R. and Gleiter, R. Cis,trans-1,5-Cyclooctadien Thermolyse, Spannungsergie und transannulare Wechselwirkungen, *Tetrahedron Letters* 3069 (1979).
- [5373] Böhm, M. C., Gleiter, R., Delgado-Pena, F., and Cowan, D. O. Photoelectron spectra of biferrocylene and biferrocene, *Inorg. Chem.* **19**, 1081 (1980).
- [5374] Schwarz, H., Wesdemiotis, C., Hess, B., and Levsen, K. Massenspektrometrische Untersuchung organischer Stickstoffverbindungen, *Org. Mass Spectrom.* **10**, 595 (1975).
- [5375] Armentrout, P. B., and Beauchamp, J. L. Properties and reactions of uranium(IV) tetrahydroborate by ion cyclotron mass spectrometry, *Inorg. Chem.* **18**, 1349 (1979).
- [5376] Andreoccetti, M. V., Cauletti, C., Furlani, C., and King, R. B. UV photoelectron study of mononuclear metal complexes of methylaminobis(difluorophosphine), *Inorg. Chem.* **18**, 954 (1979).
- [5377] Kunz, H., Lindig, M., Bicker, R., and Bock, H. Intramolekulare Wasserstoffbrücken in sterisch fixierten Amino-alkoholen, *Chem. Ber.* **111**, 2282 (1978).
- [5378] Schäfer, W., and Schweig, A. Zur Konjugation in aromatischen Aminen und Phosphanen, *Angew. Chem.* **84**, 898 (1972).
- [5379] Sümmermann, W., and Deffner, U. Die elektrochemische Oxidation aliphatischer Nitroxyl-Radikale, *Tetrahedron* **31**, 593 (1975).
- [5380] Limouzin, Y., and Maire, J. C. Spectres photoélectroniques des composés organométalliques, *J. Organometal. Chem.* **105**, 179 (1976).
- [5381] Rademacher, P. Acyclische Hydrazine, *Chem. Ber.* **108**, 1548 (1975).
- [5382] Kain, W., and Bock, H. R₂P- und R₂N-substituierte Benzole: Die Ladungsverteilung in ihren Kationen, Anionen und Trianionen, *Chem. Ber.* **111**, 3843 (1978).
- [5383] Nomoto, K., Achiba, Y., and Kimura, K. HeII(304 Å) photoelectron spectrum of N₂O₄, *Chem. Phys. Letters* **63**, 277 (1979).
- [5384] Böhm, M. C., Gleiter, R., Morgan, G. L., Lusztyk, J., and Starowieyski, K. B. Electronic structure of metallocenes V. Photoelectron spectra of cyclopentadienyl-beryllium compounds, *J. Organometal. Chem.* **194**, 257 (1980).
- [5386] Solouki, B., Rosmus, P., and Bock, H. Photoelektronenspektrum von HN=S=O, *Angew. Chem.* **88**, 381 (1976).
- [5387] Flitsch, W., Peeters, H., Schulten, W., and Rademacher, P. Photoelektronenspektren und Konformation von 1,1'-bipyrrrolyl, *Tetrahedron* **34**, 2301 (1978).
- [5388] Weiss, M. J., Hsieh, T.-C., and Meisels, G. G. Fragmentation of SO₂⁺ prepared in state selected vibrational levels, *J. Chem. Phys.* **71**, 567 (1979).
- [5389] Guimon, C., Pfister-Guillouzo, G., Manuel, G., and Mazerolles, P. Structure électronique de silacyclopentenes-2 et -3, *J. Organometal. Chem.* **149**, 149 (1978).
- [5390] Dehmlow, E. V., Dehmlow, S. S., and Marschner, F. Chemische und photoelektronenspektroskopische Eigenschaften von cyclopropylsubstituierten Cyclopropenonen, *Chem. Ber.* **110**, 154 (1977).
- [5391] Gupta, S. K., and Gingerich, K. A. Knudsen effusion mass spectrometric determination of the dissociation energy of niobium, Nb₂(g), and the heat of sublimation of solid niobium, *J. Chem. Phys.* **70**, 5350 (1979).
- [5392] Böhm, M. C., and Gleiter, R. Das He(I)-Photoelektronenspektrum von 1,2,6,7-Cyclodecatetraen. Hinweis auf eine starke transannulare Wechselwirkung, *Chem. Ber.* **111**, 3516 (1978).
- [5393] Wu, C. H., Ihle, H. R., and Zmbov, K. Atomization energies of the molecules LiSiO(g) and Si₂O₂(g) by mass spectrometric gaseous equilibria, *J. Chem. Soc. Faraday II* **76**, 447 (1980).
- [5394] Cauletti, C., Green, J. C., Kelly, M. R., Powell, P., and Van Tilborg, J. Photoelectron spectra of metallocenes, *J. Electron Spectrosc. Relat. Phenom.* **19**, 327 (1980).
- [5395] Jorgensen, F. S., and Snyder, J. P. Search for a *trans*-disulfide: Structural analysis of di-*tert*adamantyl disulfide by photoelectron spectroscopy, derivation of $\sigma_{(t\text{-Ad})}$, and molecular mechanics calculations for related bulky disulfides, *J. Org. Chem.* **45**, 1015 (1980).
- [5396] Kováč, B., Klasinc, L., Stanovník, B., and Tišler, M. Photoelectron spectroscopy of heterocycles. Azaindenes and azaindolizines (1), *J. Heterocycl. Chem.* **17**, 689 (1980).
- [5397] Dougherty, D., Lewis, J., Nauman, R. V., and McGlynn, S. P. Photoelectron spectroscopy of azulenes, *J. Electron Spectrosc. Relat. Phenom.* **19**, 21 (1980).
- [5398] Rankin, D. W. H., and Wright, J. G. Preparation and properties of difluorophosphino(tetrafluorophosphoranyl)amine, *J. Chem. Soc. Dalton*, 1070 (1979).
- [5399] Schenck, H., Oertel, H., and Baumgärtel, H. Photoreactions of small organic molecules VII. Photoionization studies on the ion-pair formation of the fluorochloromethanes CF₂Cl₂, CF₃Cl, and CFCI₃, *Ber. Bunsenges. Phys. Chem.* **83**, 683 (1979).
- [5400] Audier, H.-E., Fetizon, M., Henry, Y., and Prange, T. Mécanismes de fragmentation de l'oxazole, *Org. Mass Spectrom.* **11**, 1047 (1976).
- [5401] Grützmacher, H.-F., and Lange, G. Untersuchungen zur synchronen oder zweistufigen Fragmentierung der Molekül-Ionen von 3-substituierten Tropanen, *Chem. Ber.* **111**, 1962 (1978).
- [5403] Chmutova, G. A., Vtyurina, N. N., Komina, T. V., Gazizov, I. G., and Bock, H. Molecular-orbital characteristics of anisole isologs X-C₆H₄-ECH₃ (E = O, S, Se) containing donor substituents X, *Zh. Obshch. Khim.* **49**, 192 (1979).
- [5404] Dannacher, J., Heilbronner, E., Stadelmann, J.-P., and Vogt, J. 225. Fragmentation of energy selected butadiyne- and 1,3-pentadiyne radical cations, *Helv. Chim. Acta* **62**, 2186 (1979).
- [5405] Mellink, W. A., and Janssen, M. J. Photoelectron spectra of aromatic sulphides and sulphones, *J. Chem. Res.(S)*, 422 (1978).
- [5406] Nakagaki, R., Kobayashi, T., and Nagakura, S. Conformations of acetanilide and related compounds studied by ultraviolet photoelectron spectroscopy, *Bull. Chem. Soc. Jpn.* **53**, 901 (1980).
- [5407] Carlier, P., Mouvier, G., Mesnard, D., and Miginiac, L. Etude par spectrométrie de photoélectrons de la structure électronique de enynes conjuguées, *J. Electron Spectrosc. Relat. Phenom.* **16**, 147 (1979).
- [5408] Sell, J. A., Mintz, D. M., and Kupperman, A. Photoelectron

- angular distributions of carbon- carbon pi electrons in ethylene, benzene, and their fluorinated derivatives, *Chem. Phys. Letters* **58**, 601 (1978).
- [5409] Carlier, P., and Mouvier, G. Etude par spectrometrie de photoelectrons de la structure electronique des phenyl-alkynes conjugues, *J. Electron Spectrosc. Relat. Phenom.* **16**, 169 (1979).
- [5410] Guimon, C., Pfister-Guillouzo, G., and Arbelot, M. Spectres photoélectroniques de la dithiole-1,3-thione-2 et de son dérivé benzosubstitué, *J. Mol. Struct.* **30**, 339 (1976).
- [5411] Bieri, G., Burger, F., Heilbronner, E., and Maier, J. P. 223. Valence ionization energies of hydrocarbons, *Helv. Chim. Acta* **60**, 2213 (1977).
- [5412] Trott, W. M., Blais, N. C., and Walters, E. A. Molecular beam photoionization study of acetone and acetone-d₆, *J. Chem. Phys.* **69**, 3150 (1978).
- [5413] Zaretskii, Z. V. I.: Oren, D., and Kelner, L. Automatic method for the measurement of the electron impact ionization and appearance potentials, *Appl. Spectrosc.* **30**, 366 (1976).
- [5414] Sahini, V. E., Constantin, V., Serban, I., and Vladescu, C. Electron-impact-induced fragmentation of some dibenz[b, e] thiophenes, *Rev. Roum. Chim.* **23**, 315 (1978).
- [5415] Guimon, M. F., Guimon, C., Metras, F., and Pfister-Guillouzo, G. Analyse conformationnelle de trithiolannes-1,2,4, *Can. J. Chem.* **54**, 146 (1976).
- [5416] Laerum, T., and Undheim, K. N-Quaternary compounds. Part 52. Photochemically induced valence bond isomerism and rearrangement to pyridinones of thiazolo[3,2-a]-pyridinium-8-olate derivatives, *J. Chem. Soc. Perkin Trans. I*, 1150 (1978).
- [5417] Goetz, H., Marschner, F., Juds, H., and Pohle, H. Korrelation zwischen pK_a-Werten und vertikalen Ionisationspotentialen I, bei p-X-Phenyl-dicyclohexyl-phosphinen, *Phosphorus* **6**, 137 (1976).
- [5418] Demuth, R. Photoelektronenspektren von einigen Trihalogensilylphosphanen und -arsanen X₃SiER₂ (X=F, Cl; E=N, P, As; R=H, CH₃), *Z. Naturforsch.* **32b**, 1252 (1977).
- [5419] Brown, R. S., Marcinko, R. W., and Tse, A. Application of photoelectron spectroscopy to substituent effects. Conformational analysis of some flexible allylic ethers and alcohols, *Can. J. Chem.* **57**, 1890 (1979).
- [5420] Gey, E., and Dube, G. Die Ionisierungsenergien von substituierten Phenyldimethylmethoxysilanen und Phenyldimethylfluorsilanen, *Int. J. Mass Spectrom. Ion Phys.* **22**, 103 (1976).
- [5421] Meunier, P., and Pfister-Guillouzo, G. Etude conformati- onnelle par spectroscopie photoélectronique de dihydro "épinnes", *Can. J. Chem.* **55**, 2867 (1977).
- [5422] Efraty, A., Liebman, D., Huang, M. H. A., Weston, C. A., and Angelici, R. J. Mass spectra of organo- metallic compounds. 7. Electron-impact study of some cyclopentadienylmetal thiocarbonyl-bridged dimers, *Inorg. Chem.* **17**, 2831 (1978).
- [5423] Alikhanyan, A. S., Steblevskii, A. V., Malkerova, I. P., Pervov, V. S., Butskii, V. D., and Gorgoraki, V. I. Mass spectrometric study of the sublimation of molybdenum trifluoride, *Russ. J. Inorg. Chem.* **23**, 814 (1978).
- [5424] Dyke, J. M., Jonathan, N., Mills, J. D., and Morris, A. Vacuum ultraviolet photoelectron spectroscopy of transient species Part 12. The FO(X² Π) radical, *Mol. Phys.* **40**, 1177 (1980).
- [5425] Neubert, A., Ihle, H. R., and Gingerich, K. A. Thermodynamic study of the molecules BiLi and PbLi by Knudsen effusion mass spectrometry, *J. Chem. Phys.* **73**, 1406 (1980).
- [5426] Schäfer, W., and Schweig, A. C-S Hyperconjugation, *Tetrahedron Letters*, 5205 (1972).
- [5427] Cabaud, B., Hoareau, A., and Melinon, P. Time-of-flight spectroscopy of a supersonic beam of mercury. Intensities and appearance potentials of Hg_n aggregates, *J. Phys. D.* **13**, 1831 (1980).
- [5428] Bohlmann, F., Köppel, C., Müller, B., Schwarz, H., and Weyerstahl, P. Massenspektrometrische Untersuchung isomerer Kohlenwasserstoffe: Struktur und Bildungsenthalpie stabiler (C₁₃H₁₁)⁺-Ionen, *Tetrahedron* **30**, 1011 (1974).
- [5429] Cooper, C. D., Williamson, A. D., Miller, J. C., and Compton, R. N. Resonantly enhanced multiphoton ionization of pyrrole, N-methyl pyrrole, and furan, *J. Chem. Phys.* **73**, 1527 (1980).
- [5430] Bally, T., and Haselbach, E. 65. Tris(methylidene)-cyclopropane("[3]radialene"). Part 2. Electronic states of the molecular cation and revised uv-absorption spectrum of the parent neutral, *Helv. Chim. Acta* **61**, 754 (1978).
- [5431] Allan, M., Dannacher, J., and Maier, J. P. Radiative and fragmentation decay of the cations of trans- and cis 1,3,5-hexatriene and of all trans-1,3,5-heptatriene in the A(Π^+) states, studied by emission and photoelectron-photoion coincidence spectroscopy, *J. Chem. Phys.* **73**, 3114 (1980).
- [5432] Lee, E. P. F., Law, D., and Potts, A. W. Photoelectron spectra and valence shell electronic structure of zinc and cadmium difluoride, *J. Chem. Soc. Faraday II* **76**, 1314 (1980).
- [5433] Alikhanyan, A. S., Steblevskii, A. V., Pervov, V. S., Butskii, V. D., and Gorgoraki, V. I. Mass-spectrometric study of molybdenum oxide fluorides, *Russ. J. Inorg. Chem.* **23**, 1412 (1978).
- [5434] Coppens, P., Reynaert, J. C., and Drowart, J. Mass spectrometric study of the photoionization of carbon disulphide in the wavelength interval 125-60nm, *J. Chem. Soc. Faraday II* **75**, 292 (1979).
- [5435] Jongsma, C., Vermeer, H., Bickelhaupt, F., Schäfer, W., and Schweig, A. 10-methyl-9-phosphaanthracene, *Tetrahedron* **31**, 2931 (1975).
- [5436] Matyuk, V. M., Potapov, V. K., and Prokhoda, A. L. Photoexcitation and photoionisation of nitro- derivatives of benzene and toluene, *Russ. J. Phys. Chem.* **53**, 538 (1979).
- [5437] Weiner, M. A., Lattman, M., and Grim, S. O. Ultraviolet photoelectron spectra of some substituted triarylphosphines, *J. Org. Chem.* **40**, 1292 (1975).
- [5438] Ono, Y., Linn, S. H., Prest, H. F., Gress, M. E., and Ng, C. Y. Molecular beam photoionization study of carbon disulfide, carbon disulfide dimer and clusters, *J. Chem. Phys.* **73**, 2523 (1980).
- [5439] Malkerova, I. P., Alikhanyan, A. S., Pervov, V. S., Tripol'skaya, T. A., Gorgoraki, V. I., and Malyusov, V. A. High-temperature investigations of chromium lower fluorides, *Russ. J. Inorg. Chem.* **24**, 1775 (1979).
- [5440] Bischof, P., Gleiter, R., Taylor, R. T., Browne, A. R., and Paquette, L. A. Electronic structure of tricyclo[4.1.0.0^{2,7}]hept-3-enes. Correlation with the regioselectivity of electrophilic attack, *J. Org. Chem.* **43**, 2391 (1978).
- [5441] Starzewski, K. A. O., and Dieck, H. t. Electronic structure and reactivity. 8. Iminophosphoranessimple ylide analogues? An investigation of quantitative and phenomenological differences, *Inorg. Chem.* **18**, 3307 (1979).
- [5442] Allen, C. W., and Green, J. C. Organophosphazenes. 12. He I photoelectron spectra of selected phenyl- and *l*p-(dimethylamino)phenyl]fluorocyclotriphosphazenes, *Inorg. Chem.* **19**, 1719 (1980).
- [5443] Müller, C., Schweig, A., and Vermeer, H. Methode zur Berechnung induktiver und konjugativer Effekte -

- Andwendung auf Tropon, Angew. Chem. **86**, 275 (1974).
- [5445] Holmes, J. L., Terlouw, J. K., and Burgers, P. C. $[C_6H_5O]^+$ ions; Reacting and non-reacting configurations, Org. Mass Spectrom. **15**, 140 (1980).
- [5446] Cauletti, C., and Sima, J. The electronic structure of some N,N'-ethylenebis(thioacetylacetoneiminato) complexes studied by uv photoelectron spectroscopy, J. Electron Spectrosc. Relat. Phenom. **19**, 1 (1980).
- [5447] Bischof, P., Böhm, M., Gleiter, R., Snow, R. A., Doecke, C. W., and Paquette, L. A. Evaluation of through-space interaction in 9-substituted pentacyclononane derivatives, J. Org. Chem. **43**, 2387 (1978).
- [5448] Simonneaux, G., Jaouen, G., Dabard, R., and Guenot, P. Contribution à l'étude des métaux asymétriques IV. Etude des complexes areniques du chrome(0) par spectroscopie de masse: modes de liaison du métal avec divers ligands à deux électrons: phosphines, phosphites, carbonyle et thiocarbonyle, J. Organometal. Chem. **132**, 231 (1977).
- [5449] Brown, C. M., Tilford, S. G., and Ginter, M. L. Absorption spectrum of Pb I between 1350 and 2041 Å, J. Opt. Soc. Am. **67**, 1240 (1977).
- [5450] Brown, C. M., Tilford, S. G., and Ginter, M. L. Absorption spectra of Zn I and Cd I in the 1300–1750 Å region, J. Opt. Soc. Am. **65**, 1404 (1975).
- [5451] Zaikin, V. G., Musaev, I. A., and Kurashova, E. K. Ionization and appearance potentials in organic chemistry, Org. Mass Spectrom. **13**, 685 (1978).
- [5452] Zaikin, V. G., and Wulfson, N. S. Dissociation energies of axial and equatorial carbon–carbon bonds in substituted decahydroquinolols, Tetrahedron Letters, 2935 (1978).
- [5453] Müller, J., Fenderl, K., and Mertschenk, B. Die Donor-Akzeptor-Eigenschaften des Liganden Trifluorphosphin in Übergangsmetallkomplexen, Chem. Ber. **104**, 700 (1971).
- [5454] Rosenstock, H. M., McCulloh, K. E., and Lossing, F. P. On the mechanisms of C_6H_6 ionization and fragmentation, Intern. J. Mass Spectrom. Ion Phys. **25**, 327 (1977).
- [5455] McLean, W., Murray, P. T., Baer, T., and Jarnagin, R. C. Dissociative photoionization of *t*-butyl-lithium, J. Chem. Phys. **69**, 2715 (1978).
- [5456] Bavia, M., Zauli, C., and Fusina, L. Rydberg states in selenophene, Mol. Phys. **30**, 1289 (1975).
- [5457] Bieri, G., and Jonsson, B.-Ö. HNC⁺ radical cation studied by charge-exchange mass spectrometry, Chem. Phys. Letters **56**, 446 (1978).
- [5458] Vogt, J., Williamson, A. D., and Beauchamp, J. L. Properties and reactions of ketene in the gas phase by ion cyclotron resonance spectroscopy and photoionization mass spectrometry. Proton affinity, site specificity of protonation, and heat of formation of ketene, J. Am. Chem. Soc. **100**, 3478 (1978).
- [5459] Schubert, R., and Grützmacher, H.-F. Kinetic energy release and position of transition state during the intramolecular substitution of ionized 2-benzoyl pyridines, Org. Mass Spectrom. **15**, 122 (1980).
- [5460] Vajda, J. H., and Harrison, A. G. Proton affinities of some olefinic carbonyl compounds and heats of formation of $C_nH_{2n-1}O^+$ ions, Intern. J. Mass Spectrom. Ion Phys. **30**, 293 (1979).
- [5461] Maier, J. P., Marthaler, O., and Mohraz, M. Decay of some substituted benzene cations in the B states in the gaseous phase, J. Chim. Phys. **77**, 661 (1980).
- [5462] Baldwin, M. A., Loudon, A. G., Dunmur, R. E., Schmutzler, R., and Gregor, I. K. Charge location and fragmentation under electron impact, Org. Mass Spectrom. **12**, 275 (1977).
- [5463] Kovač, B., Heilbronner, E., Prinzbach, H., and Weidmann, K. 288. The photoelectron spectra of D_{2d} and C_{2v} hydrocarbons containing two norbornadiene or quadricyclane groups, Helv. Chim. Acta **62**, 2841 (1979).
- [5465] Bell, S., Ng, T. L., and Walsh, A. D. Vacuum ultraviolet spectra of formic and acetic acids, J. Chem. Soc. Faraday Trans. II **71**, 393 (1975).
- [5466] De Leeuw, D. M., Mooyman, R., and De Lange, C. A. He(I) photoelectron spectroscopy of transient species: The SBr₂ molecule, Chem. Phys. Letters **61**, 191 (1979).
- [5467] Solka, B. H., and Russell, M. E. Energetics of formation of some structural isomers of gaseous $C_2H_5O^+$ and $C_2H_6N^+$ ions, J. Phys. Chem. **78**, 1268 (1974).
- [5468] Murad, E., and Hildenbrand, D. L. Thermochemical properties of gaseous EuO, J. Chem. Phys. **65**, 3250 (1976).
- [5469] Peel, J. B., and Willett, G. D. Photoelectron spectra of the allyl amines, Chem. Phys. Letters **38**, 479 (1976).
- [5470] Doucet, J., Gilbert, R., Sauvageau, P., and Sandorfy, C. Photoelectron and far-ultraviolet spectra of CF_3Br , CF_2BrCl , and CF_3Cl , J. Chem. Phys. **62**, 366 (1975).
- [5471] Hildenbrand, D. L. Mass spectrometric studies of the thermochemistry of gaseous TiO and TiO₂, Chem. Phys. Letters **44**, 281 (1976).
- [5472] Padva, A., LeBreton, P. R., Dinerstein, R. J., and Ridyard, J. N. A. UV photoelectron studies of biological pyrimidines: The electronic structure of uracil, Biochem. Biophys. Res. Commun. **60**, 1262 (1974).
- [5473] Lee, T. H., and Rabalais, J. W. Model for spin-orbit interactions with inclusion of *d* electrons: Applications to photoelectron spectroscopy, J. Chem. Phys. **60**, 1172 (1974).
- [5474] Streets, D. G., Hall, W. E., and Ceasar, G. P. Mesomeric mixing in the π energy levels of aminobenzenes studied by photoelectron spectroscopy, Chem. Phys. Letters **17**, 90 (1972).
- [5475] Berkowitz, J. PES of high temperature vapors. VII. S_2 and Te_2 , J. Chem. Phys. **62**, 4074 (1975).
- [5476] Kitagawa, S., Morishima, I., Yonezawa, T., and Sato, N. Photoelectron spectroscopic study on metallo-octaethylporphyrins, Inorg. Chem. **18**, 1345 (1979).
- [5477] Worley, S. D., Hargis, J. H., Chang, L., Mattson, G. A., and Jennings, W. B. A study of the electronic structure of 2-(dialkyl-amino)-1,3-dimethyl-1,3,2-diazaphospholanes and related molecules, Inorg. Chem. **18**, 3581 (1979).
- [5478] Coustale, M., Guimon, C., Arriau, J., and Pfister-Guillouzo, G. Spectres photoélectroniques de divers thiéno[2,3-*b*]thiophènes, J. Heterocyclic Chem. **13**, 231 (1976).
- [5479] Rabalais, J. W., Debies, T. P., Berkosky, J. L., Huang, J.-T. J., and Ellison, F. O. Calculated photoionization cross sections and relative experimental photoionization intensities for a selection of small molecules, J. Chem. Phys. **61**, 516 (1974).
- [5480] Bischof, P., Gleiter, R., Dürr, H., Ruge, B., and Herbst, P. Das Photoelektronenspektrum von 1,2-Diäthylspiro[2.4]hepta-1,4,6-trien. Beispiel für eine π - σ -Wechselwirkung, Chem. Ber. **109**, 1412 (1976).
- [5481] Barracough, P., Bilgic, S., Pedley, J. B., Rogers, A. J., and Young, D. W. Conjugative and homo-conjugative effects in 2-heterobicyclo[3.2.1]octa-3, 6-dienes, Tetrahedron **35**, 99 (1979).
- [5482] Chesnavich, W. J., Su, T., and Bowers, M. T. Reactions of vibrationally excited ions. A theoretical and experimental analysis of the reaction $(C_6H_5^+) + NH_3 \rightarrow NH_2^+ + C_6H_4$, J. Am. Chem. Soc. **100**, 4362 (1978).
- [5483] Wolkoff, P., Holmes, J. L., and Lossing, F. P. On the formation of cyclopentenium cations from all $C_6H_{10}^+$ molecular ion structures, Can. J. Chem. **58**, 251 (1980).
- [5484] Schwarz, H., Petersen, R. D., and Van De Sande, C. C. Entropische und energetische Effekte bei der Bildung cyclischer Oxonium-, Sulfonium- und Bromonium-Ionen aus ω -substituierten Phenoxyalkanen, Org. Mass Spectrom. **12**, 391 (1977).

- [5485] Berger, H.-O., Kroner, J., and Nöth, H. Die BorHalogen-Bindung in Methylhalogenboranen: Photo-elektronenspektren und ab initio-Rechnungen, *Chem. Ber.* **109**, 2266 (1976).
- [5486] Akaba, R., Tokumaru, K., Kobayashi, T., and Utsunomiya, C. Electronic structures and conformations of *N*-benzylideneanilines. II. Photoelectron spectral study, *Bull. Chem. Soc. Japan* **53**, 2002 (1980).
- [5487] Maquestiau, A., Van Haverbeke, Y., Flamming, R., and Cooks, R. G. Fragmentation d'azoles sous l'impact électronique, *Org. Mass Spectrom.* **10**, 946 (1975).
- [5488] Holzmann, G., Rothkopf, H. W., Müller, R., and Wöhrl, D. Massenspektren heteroaromatischer Nitrile, *Org. Mass Spectrom.* **10**, 97 (1975).
- [5489] Rademacher, P., Breier, H., and Poppek, R. Photo-elektronenspektren und Konformationen bi- und tricyclischer Hexahydro-1,2,4,5-tetrazine, *Chem. Ber.* **112**, 853 (1979).
- [5490] Gaivoronskii, P. E., Gavrilchuk, E. M., Chernyaev, N. P., and Zverev, Y. B. Mass spectrometric study of tris(isopropylcyclopentadienyl) π - complexes of lanthanum, praseodymium, and neodymium, *Russ. J. Inorg. Chem.* **23**, 1742 (1978).
- [5491] Loutfy, R. O., Still, I. W. J., Thompson, M., and Leong, T. S. Correlation of the photoelectron and electronic spectra of thiochromones and thiochromanones with their electrochemical data, *Can. J. Chem.* **57**, 638 (1979).
- [5492] Lin, J., Yu, C., Peng, S., Akiyama, I., Li, K., Lee, L. K., LeBreton, P. R. Ultraviolet photoelectron studies of the ground-state electronic structure and gas-phase tautomerism of purine and adenine, *J. Am. Chem. Soc.* **102**, 4627 (1980).
- [5493] Grützmacher, H.-F., and Schubert, R. Substituent effects in the mass spectra of benzoyl heteroarenes, *Org. Mass Spectrom.* **14**, 567 (1979).
- [5494] Brown, C. M., and Ginter, M. L. Absorption spectrum of Ag I between 1540 and 1850 Å, *J. Opt. Soc. Am.* **67**, 1323 (1977).
- [5495] Brown, C. M., Tilford, S. G., and Ginter, M. L. Absorption spectrum of Ge I between 1500 and 1900 Å, *J. Opt. Soc. Am.* **67**, 584, 1977.
- [5496] Brown, C. M., Tilford, S. G., and Ginter, M. L. Absorption spectrum of Sn I between 1580 and 2040 Å, *J. Opt. Soc. Am.* **67**, 607 (1977).
- [5497] Brown, C. M., and Ginter, M. L. Absorption spectrum of Mn I between 1305 and 2040 Å, *J. Opt. Soc. Am.* **68**, 1541 (1978).
- [5498] Behan, J. M., Johnstone, R. A. W., Worman, J. J., and Fehlner, T. P. Photoelectron spectroscopy (HeI) of the dithione, bisimine and monoimine of tetramethyl-1,3-cyclobutanedione, *J. Mol. Struct.* **40**, 151 (1977).
- [5500] Brown, C. M., and Ginter, M. L. Absorption spectrum of Au I between 1300 and 1900 Å, *J. Opt. Soc. Am.* **68**, 243 (1978).
- [5501] Berman, D. W., Anicich, V., and Beauchamp, J. L. Stabilities of isomeric halonium ions $C_2H_4X^+$ ($X = Cl, Br$) by photoionization mass spectrometry and ion cyclotron resonance spectroscopy. General considerations of the relative stabilities of cyclic and acyclic isomeric onium ions, *J. Am. Chem. Soc.* **101**, 1239 (1979).
- [5502] Ellingsen, P., and Hvistendahl, G. Mass spectrometry of the simple *N*-vinylpyridinium salts, *Org. Mass Spectrom.* **15**, 18 (1980).
- [5503] Finney, C. D., and Harrison, A. G. A third-derivative method for determining electron-impact onset potentials, *Intern. J. Mass Spectrom. Ion Phys.* **9**, 221 (1972).
- [5504] Bock, H., Kaim, W., Nöth, H., and Semkow, A. Radical ions. 36. Structural changes accompanying the one-electron oxidation of hydrazine and its silyl derivatives, *J. Am. Chem. Soc.* **102**, 4421 (1980).
- [5505] Kotov, B. V., and Potapov, V. K. Ionization potentials of strong organic electron acceptors, *Khim. Vys. Energ.* **6**, 375 (1972).
- [5506] Karlsson, L., Mattson, L., Jadny, R., Albridge, R. G., Pinchas, S., Bergmark, T., and Siegbahn, K. Isotopic and vibronic coupling effects in the valence electron spectra of $H_2^{16}O$, $H_2^{18}O$, and $D_2^{16}O$, *J. Chem. Phys.* **62**, 4745 (1975).
- [5507] Evans, S., Green, M. L. H., Jewitt, B., King, G. H., and Orchard, A. F. Electronic structures of metal complexes containing the π -cyclopentadienyl and related ligands, *J. Chem. Soc. Faraday II*, **70**, 356 (1974).
- [5508] Isakov, L. I., and Potapov, V. K. Dissociative photoionization of *n*-alkylamines, *Khim. Vys. Energ.* **5**, 534 (1971).
- [5509] Berkowitz, J., Batson, C. H., and Goodman, G. L. Photoionization of lithium chloride vapors: The structure and stability of alkali halide molecules and ions, *J. Chim. Phys.* **77**, 631 (1980).
- [5510] Vovna, V. I., and Vilessov, F. I. Photoelectron spectra and the structure of molecular orbitals of methyl amines, *Opt. Spectrosc.* **36**, 251 (1974).
- [5511] Smith, G., and Tomkins, F. S. Autoionization resonances in the Eu I absorption spectrum and a new determination of the ionization potential, *Proc. Roy. Soc. Lond.* **A342**, 149 (1975).
- [5512] Potapov, V. K., and Sorokin, V. V. Kinetic energies of products of dissociative photoionization of molecules. I. Aliphatic ketones and alcohols, *Khim. Vys. Energ.* **6**, 387 (1972).
- [5513] Selim, E. T. M., and El-Kholy, S. B. Mass spectrometric ionization and dissociation of methane, *Indian J. Pure Appl. Phys.* **13**, 233 (1975).
- [5514] Vovna, V. I., Lopatin, S. N., Pettsold, R., Vilessov, F. I., and Akopyan, M. E. Photoelectron spectra of a number of substitution products of thiophosphoryl chloride, *Opt. Spectrosc.* **36**, 99 (1974).
- [5515] Akopyan, M. E., Vilessov, F. I., and Lopatin, S. N. Photoionization of benzyl chloride, *Khim. Vys. Energ.* **6**, 110 (1972).
- [5516] Hodges, R. V., McDonnell, T. J., and Beauchamp, J. L. Properties and reactions of trimethyl phosphite, trimethyl phosphate, triethyl phosphate, and trimethyl phosphorothionate by ion cyclotron resonance spectroscopy, *J. Am. Chem. Soc.* **102**, 1327 (1980).
- [5517] Arnett, J. F., Newkome, G., Mattice, W. L., and McGlynn, S. P. Excited electronic states of the α -dicarbonyls, *J. Am. Chem. Soc.* **96**, 4385 (1974).
- [5518] Lichtenberger, D. L., and Fenske, R. F. Electronic structure of transition metal thiocarbonyl complexes, *Inorg. Chem.* **15**, 2015 (1976).
- [5519] Hernandez, R., Masclet, P., and Mouvier, G. Spectroscopic de photoelectrons d'aldehydes et de cétones aliphatiques, *J. Electron Spectrosc. Relat. Phenom.* **10**, 333 (1977).
- [5520] Tschmutowa, G., and Bock, H. Photoelektron-Spektren und Moleküleigenschaften, LXIII Ionisierungssenergien von Phenyl-Methyl-Tellurid und Vergleich der Effekte von Sauerstoff-, Schwefel-, Selen- und Tellur-Substituenten auf das Benzol- π -System, *Z. Naturforsch.* **31b**, 1611 (1976).
- [5521] Maier, J. P., Marthaler, O., and Mohraz, M. Decay of some substituted benzene cations in the B states in the gaseous phase, *J. Chim. Phys.* **77**, 661 (1980).
- [5522] Güsten, H., Klasinc, L., and Ruščić, B. Photoelectron spectroscopy of heterocycles. Indene analogs, *Z. Naturforsch.* **31a**, 1051 (1976).
- [5523] Elbel, S., and Dieck, H. t. Photoelectron spectra of group 5 compounds. Part V. Phosphorus halides, $R_2P(X)Y$ and $RP(X)Y_2R = Me$ or F ; $X = O, S$, or Se ; $Y = Cl$ or Br , *J. Chem. Soc. Dalton* **1762** (1976).
- [5525] Åsbrink, L., Von Niessen, W., and Bieri, G. 30.4-nm He(II)

- photoelectron spectra of organic molecules, *J. Electron Spectrosc. Relat. Phenom.* **21**, 93 (1980).
- [5526] Weiner, M. A., and Lattman, M. Ultraviolet photoelectron spectra of some $\text{Cr}(\text{CO})_5\text{L}$ complexes containing organosulfide and organophosphine ligands, *Inorg. Chem.* **17**, 1084 (1978).
- [5527] Ramsey, B. G., and Walker, F. A. A linear relationship between substituted pyridine lone pair vertical ionization potentials and pK_a , *J. Am. Chem. Soc.* **96**, 3314 (1974).
- [5528] Al-Khafaji, J. A., and Shanshal, M. The photoelectron spectra of cyclopropyl ketones, *Z. Naturforsch.* **32a**, 109 (1977).
- [5529] Elbel, S., and Dieck, H. t. Photoelectron spectra of group 5 compounds. Part IV. A study of the E-X bond in R_3EX compounds ($\text{R}=\text{Me}$ or F ; $\text{E}=\text{N}$ or P ; $\text{X}=\text{O}$ or S), *J. Chem. Soc. Dalton* 1757 (1976).
- [5530] Van den Ham, D. M. W., Van der Meer, D., and Feil, D. Photoelectron spectra of fluorine-substituted diazenes, *J. Electron Spectrosc. Relat. Phenom.* **3**, 479 (1974).
- [5531] Trofimov, B. A., Mel'der, U. K., Pikver, R. I., and Vyalykh, E. P. Ionization potentials of unsaturated sulfides and the participation of neighboring multiple bonds of heteroatoms in the stabilization of the radical cation, *Teor. Eksp. Khim.* **11**, 129 (1975).
- [5532] Remane, H., Graefe, J., and Herzschnuh, R. Ionisationspotentiale von cis- und trans- Cycloalkenen, *Z. Chem.* **12**, 194 (1972).
- [5533] Süzer, S., Banna, M. S., and Shirley, D. A. Relativistic and correlation effects in the 21.2-eV photoemission spectrum of atomic lead, *J. Chem. Phys.* **63**, 3473 (1975).
- [5535] Bock, H., and Kaim, W. Radical ions. 37. Ionization and one-electron oxidation of electron-rich silylalkyl olefins, *J. Am. Chem. Soc.* **102**, 4429 (1980).
- [5536] Van Dam, H., Louwen, J. N., Oskam, A., Doran, M., and Hillier, I. H. The electronic structure of dinuclear transition-metal complexes containing metal-metal interactions, *J. Electron Spectrosc. Relat. Phenom.* **21**, 57 (1980).
- [5537] Heilbronner, E., and Jones, T. B. Spectral differences between "isospectral" molecules, *J. Am. Chem. Soc.* **100**, 6506 (1978).
- [5538] Kobayashi, T. A new rule for photoelectron angular distributions of molecules, *Phys. Letters* **69A**, 31 (1978).
- [5539] Daamen, H., Boxhoorn, G., and Oskam, A. U. V. photoelectron (He I and He II) studies of $\text{M}(\text{CO})_5\text{PX}_3$ ($\text{M}=\text{Cr}, \text{Mo}$, W and $\text{X}=\text{F}, \text{Cl}, \text{Br}$) *Inorg. Chim. Acta* **28**, 263 (1978).
- [5540] Daamen, H., and Oskam, A. Bonding properties of some monosubstituted chromium and tungsten hexacarbonyls $\text{M}(\text{CO})_5\text{L}$ ($\text{L}=\text{amine}$, substituted pyridine, azine), *Inorg. Chim. Acta* **26**, 81 (1978).
- [5541] Kirby, C., and Kroto, H. W. Microwave and photoelectron study of *cis*- and *trans*-isocyanato ethene, $\text{CH}_2=\text{CHNCO}$ (vinyl isocyanate), *J. Mol. Spectrosc.* **70**, 216 (1978).
- [5543] Akopyan, M. E., Vilesov, F. I., and Loginov, Y. V. Photoionization of amines and the monomolecular decomposition of excited molecular ions, *Khim. Vys. Energ.* **9**, 327 (1975).
- [5544] Livett, M. K., Nagy-Felsobuki, E., Peel, J. B., and Willett, G. D. Photoelectron spectra of chloramine and dichloramine, *Inorg. Chem.* **17**, 1608 (1978).
- [5545] de Jong, A. P., and Van Dam, H. Ultraviolet photoelectron spectroscopy of cyclic amidines. 2. Electronic structure of clonidine and some related 2-(phenylimino)imidazolidines with α -adrenergic activity, *J. Med. Chem.* **23**, 889 (1980).
- [5546] Barlos, K., and Nöth, H. Beiträge zur Chemie des Bors, CVIII[1] Synthese und Konformation von N,N'-Bis(boryl)-N,N'-dimethylhydrazinen, *Z. Naturforsch* **35b**, 125 (1980).
- [5547] Mingos, D. M. P. Theoretical and structural studies on organometallic cluster molecules, *Pure Appl. Chem.* **52**, 705 (1980).
- [5548] Dreckschmidt, R., Kessel, H., and Marschner, F. Korrelation zwischen Photoelektronen- und Elektronen-Spektren VI, *Tetrahedron* **33**, 101 (1977).
- [5549] Meeks, J. L., and McGlynn, S. P. Photoelectron spectra of carbonyls. Oxalyl chloride, ethyl oxalyl chloride, ethyl oxamate and N,N-dimethyl ethyl oxamate, *Spectrosc. Lett.* **8**, 439 (1975).
- [5550] Schweig, A., Weidner, U., and Manuel, G. Photoelektron-Spektroskopie und transannulare Wechselwirkungen - Zur Frage transannularer d- π -Wechselwirkung in 1-Sila-und 1-Germa-3- cyclopentenen, *Angew. Chem.* **84**, 899 (1972).
- [5551] Green, J. C., Powell, P., and Van Tilborg, J. He(I) photoelectron spectra of tricarbonyl-iron and -ruthenium complexes of cyclic dienes, and of tricarbonylcycloheptatriene-and tricarbonylcyclo-octatetraene-iron, *J. Chem. Soc. Dalton* 1974, (1976).
- [5552] Potapov, V. K., Kardash, I. E., Sorokin, V. V., Sokolov, S. A., and Evlasheva, T. I. Photoionization of heteroaromatic compounds, *Khim. Vys. Energ.* **6**, 392 (1972).
- [5553] Beltram, G. A., and Fehlner, T. P. Substituent effects in cluster species. 2. Photoelectron spectra of 2- and 2,4-substituted 1,6-dicarba- *clos*-hexaborane(6), *J. Am. Chem. Soc.* **101**, 6237 (1979).
- [5554] Syrvatka, B. G., Gil'burd, M. M., and Bel'ferman, A. L. Ion-dissociative processes of some halogen-containing butadienes and particles structure, *Zh. Org. Khim.* **9**, 1117 (1973).
- [5555] Verkin, B. I., Sukodub, L. F., and Yanson, I. K. Ionization potentials of nitrogenous bases of nucleic acids, *Dokl. Akad. Nauk SSSR* **228**, 1452 (1976).
- [5556] Rang, S., Paldoia, P., and Talvari, A. Ionization potentials of unsaturated hydrocarbons. 2. Mono- substituted cyclopentenes and cyclohexenes, *Eesti NSV Teaduste Akad. Toimetised* 354 (1974).
- [5557] Ponomarev, D. A., Takhimova, V. V., Akopyan, M. E., and Sergeyev, Y. L. Effect of alkyl groups on the stability of cations in the absence of solvation effects, *Zh. Org. Khim.* **10**, 403 (1974).
- [5558] Dougherty, D., and McGlynn, S. P. Photoelectron spectroscopy of carbonyls. 1,4-Benzozquinones, *J. Am. Chem. Soc.* **99**, 3234 (1977).
- [5559] Flamini, A., Semprini, E., Stefani, F., Cardaci, G., Bellachoma, G., and Andreucci, M. He(I) photoelectron spectra of tetracarbonyliron complexes of group 5 ligands and of olefinic ligands, *J. Chem. Soc. Dalton* 695 (1978).
- [5560] Fragala, I., Marks, T. J., Fagan, P. J., and Manriquez, J. M. He(I)-and He(II)- excited photoelectron spectra of bis(pentamethylcyclopentadienyl)zirconium dichloride, *J. Electron Spectrosc. Relat. Phenom.* **20**, 249 (1980).
- [5561] Efraty, A., and Huang, M. H. A. Mass spectra of organometallic compounds. 8. Electron-impact study of the cyclopentadienylmanganese derivatives $\text{RC}_5\text{H}_5\text{Mn}(\text{CS})(\text{NO})\text{I}$ ($\text{R}=\text{H}, \text{CH}_3$), *Inorg. Chem.* **19**, 2296 (1980).
- [5562] Borden, W. T., Young, S. D., Frost, D. C., Westwood, N. P. C., and Jorgensen, W. L. Photoelectron spectra of the 1,2,5,6-tetramethyl-3,4,7,8-tetramethylene derivatives of tricyclo[3.3.0.0^{2,6}]octane and tricyclo[4.2.0.0^{2,5}]octane, *J. Org. Chem.* **44**, 737 (1979).
- [5563] Gleiter, R., Schang, P., Adam, W., Eggeleit, H. J., Erden, I., and Bloodworth, A. J. Photoelectron spectra of bicyclic peroxides, *J. Electron Spectrosc. Relat. Phenom.* **19**, 223 (1980).
- [5564] Torroni, S., Innorta, G., Foffani, A., and Distefano, G. Interpretation of the mass spectra of substituted chromium and tungsten carbonyls by means of

- appearance potential measurements, *J. Organometall. Chem.* **65**, 209 (1974).
- [5565] Bursten, B. E., Cotton, F. A., Green, J. C., Seddon, E. A., and Stanley, G. G. Molecular orbital and spectroscopic studies of triple bonds between transition-metal atoms. 1. The d^3-d^3 Mo_2L_6 compounds ($\text{L}=\text{OR}, \text{NR}_2, \text{CH}_2\text{R}$), *J. Am. Chem. Soc.* **102**, 4579 (1980).
- [5566] Weiner, M. A., Gin, A., and Lattman, M. Ultraviolet photoelectron spectra of 4-substituted pyridine-pentacarbonylchromium complexes, *Inorg. Chim. Acta* **24**, 235 (1977).
- [5567] Anderson III, G. M., Kollman, P. A., Domelsmith, L. N., and Houk, K. N. Methoxy group nonplanarity in *o*-dimethoxybenzenes. Simple predictive models for conformations and rotational barriers in alkoxyaromatics, *J. Am. Chem. Soc.* **101**, 2344 (1979).
- [5568] Fragalà, I., Costanzo, L. L., Ciliberto, E., Condorelli, G., and D'Arrigo, C. He-I and He-II excited photoelectron spectra of tetracoordinated complexes of transition metal ions with β -diketonate ligands, *Inorg. Chim. Acta* **40**, 15 (1980).
- [5569] Cauletti, C., Nicotra, G., and Piancastelli, M. N. The electronic structure of some alkyl-tin derivatives with sulphur-containing ligands studied by UV photoelectron spectroscopy, *J. Organomet. Chem.* **190**, 147 (1980).
- [5570] Schubert, R., and Grützmacher, H.-F. Kinetic energy release and position of transition state during intramolecular aromatic substitution in ionized 1-phenyl-1-(2-pyridyl)ethylenes, *J. Am. Chem. Soc.* **102**, 5323 (1980).
- [5571] Kochi, J. K. The role of electron transfer and charge transfer in organometallic chemistry, *Pure Appl. Chem.* **52**, 571 (1980).
- [5572] Bock, H., and Wagner, G. "Einsame" Elektronenpaare in organischen Sulfiden und Disulfiden, *Angew. Chem.* **84**, 119 (1972).
- [5574] Bock, H., Kaim, W., and Rohwer, H. E. Die hyperkonjugative Stabilisierung von *p*-Xylol-Radikalkationen durch $(\text{H}_3\text{C})_3\text{Si}-$ Substituenten, *Chem. Ber.* **111**, 3573 (1978).
- [5575] Kovač, B., Allan, M., Heilbronner, E., Maier, J. P., Gleiter, R., Haenel, M. W., Keehn, P. M., and Reiss, J. A. He(Iα) photoelectron spectra of [2.2] cyclophanes, *J. Elect. Spectrosc. Relat. Phenom.* **19**, 167 (1980).
- [5576] Efraty, A., Liebman, D., Huang, M. H. A., and Weston, C. A. Mass spectra of organometallic compounds. 6. Energetics of fragmentations of manganese-containing ions with intact ligands in the series $\text{CH}_3\text{C}_5\text{H}_7\text{Mn}(\text{CO})_2\text{L}$ and $\text{CH}_3\text{C}_5\text{H}_7\text{Mn}(\text{CO})(\text{CS})\text{L}$ [$\text{L}=\text{Ph}_3\text{As}$ and Ph_3Sb], *Inorg. Chim. Acta* **39**, 105 (1980).
- [5577] Palmer, M. H., Simpson, I., and Platenkamp, R. J. The electronic structure of flavin derivatives, *J. Mol. Struct.* **66**, 243 (1980).
- [5578] Spanget-Larsen, J., Gleiter, R., Klein, G., Doecke, C. W., and Paquette, L. A. Orbital interactions in "hypostrophene" and its hydro- and homo-derivatives, A photoelectron spectroscopic investigation, *Chem. Ber.* **113**, 2120 (1980).
- [5579] Berry, M., Garner, C. D., Hillier, I. H., MacDowell, A. A., and Clegg, W. Crystal structure and U. V. photoelectron spectra of tetrakis-(6-methyl-2-oxopyridinato)dirhodium, *J. Chem. Soc. Chem. Commun.* 494 (1980).
- [5581] Nelsen, S. F., Kessel, C. R., Brien, D. J., and Weinhold, F. 9-(9-Borabicyclo[3.3.1]nonyl)-9-azabicyclo[3.3.1]nonane radical cation: A failure of Bredt's Rule kinetic stabilization, *J. Org. Chem.* **45**, 2116 (1980).
- [5582] Efraty, A., Liebman, D., Huang, M. H. A., Weston, C. A., and Angelici, R. J. Mass spectra of organometallic compounds. 7. Electron-impact study of some cyclopentadienylmetal thiocarbonyl-bridged dimers, *Inorg. Chem.* **17**, 2831 (1978).
- [5583] Rang, S., and Martinson, E. Ionization potentials of unsaturated hydrocarbons. 3. *n*-Alkynes C_nC_{11} , *Eesti NSV Teaduste Akadeemia Toimetised* **27**. Köide Keemia Geol. 44 (1978).
- [5584] Rang, S., Martinson, E., and Müürisepp, M. Ionization potentials of unsaturated hydrocarbons. 1. *n*-Alkenes, *Eesti NSV Teaduste Akadeemia Toimetised* **23**. Köide Keemia Geol. 352 (1974).
- [5585] Il'in, M. K., Makarov, A. V., and Nikitin, O. T. A study of evaporating products of barium metaborate, *Vestn. Mosk. Univ., Ser. II: Khim.* **15**, 436 (1974).
- [5586] Shikhmamedbekova, A. Z., Aslanov, F. A., Gadzhiev, M. M., Gulamova, T. E., and Akhmedova, F. N. Mass spectrometric study of methylene cycloalkenes, *Dokl. Akad. Nauk Azerb. SSR* **26**, 34 (1970).
- [5587] Makarov, A. V., Nikitin, O. T., and Chervonny, A. D. The mass-spectrometric study of evaporation of indium metaborate, *Vestn. Mosk. Univ., Ser. II: Khim.* **15**, 193 (1974).
- [5588] Semenov, G. A., Volkov, A. D., and Franktseva, K. E. Mass-spectrometric study of sodium carbonate vaporization, *Tr. Leningrad. Tekhnol. Inst. Tsellyul.-Bum. Prom.* **30**, 153 (1973).
- [5589] Zverev, V. V., Vovna, V. I., Elman, M. S., Kitaev, Y. N., and Vilesov, F. I. Photoelectron spectra and spatial structure of acyclic azines, *Dokl. Akad. Nauk USSR* **213**, 1117 (1973).
- [5590] Armen, G. H., Braunstein, C., Weinstein, M. I., and Baker, A. D. Reaction of azoxyanisole with oxalyl chloride: Use of photoelectron spectroscopy in seeking new reactions, *Tetrahedron Letters* 4197 (1979).
- [5591] Kokars, V., Kampars, V., and Neilands, O. Characteristics of π -electron structure of betainelike active methylene group derivatives. VII. Electron absorption spectra and electron-donor properties of the series of β -diketone onium betaines, *Latv. Psr. Zinat. Akad. Vestis Kim. Ser.* **6**, 734 (1975).
- [5592] Kampars, V., and Neilands, O. Characteristics of π -electron structure of betainelike active methylene group derivatives. VI. Character of long-wave absorption bands and charge transfer complexes of 1,3-indanodine onium betaines, *Latv. Psr. Zinat. Akad. Vestis Kim. Ser.* **6**, 727 (1975).
- [5593] Kreicberga, Y. N., Kampars, V., and Neilands, O. Y. Onium derivatives of tetraarylcylopentadienes, *Zh. Org. Khim.* **11**, 1508 (1975).
- [5594] Yu, C., Peng, S., Akiyama, I., Lin, J., and LeBreton, P. R. Ultraviolet photoelectron studies of biological pyrimidines. The valence electronic structure of cytosine, *J. Am. Chem. Soc.* **100**, 2303 (1978).
- [5595] Carnovale, F., Gan, T. H., Peel, J. B., and Franz, K.-D. The photoelectron spectra of some 1,9-disubstituted phenalenes, *J. Chem. Soc. Perkin II*, 957 (1980).
- [5596] Garner, C. D., Hillier, I. H., Knight, M. J., Mac Dowell, A. A., Walton, I. B., and Guest, M. F. Electronic structure and assignment of the ultraviolet photoelectron spectra of 6-methyl-2-oxo-pyridine complexes of dichromium(II) and dimolybdenum(II), *J. Chem. Soc. Faraday II* **76**, 885 (1980).
- [5597] Gleiter, R., Haider, R., Murata, I., and Pagni, R. M. Photoelectron spectra of three isomeric $(\text{C}_{11}\text{H}_{10})$ naphtho[1,8]-fused hydrocarbons, *J. Chem. Res. (S)*, 72 (1979).
- [5598] Zaikin, V. G., and Wulfson, N. S. Ionization and appearance potentials in organic chemistry, *Org. Mass Spectrom.* **13**, 680 (1978).
- [5599] Paquette, L. A., Bellamy, F., Böhm, M. C., and Gleiter, R. Electronic control of stereoselectivity. 6. Directionality of singlet oxygen addition to 1,4-dimethoxynaphthalenes

- laterally fused to bridged bicyclic systems, *J. Org. Chem.* **45**, 4913 (1980).
- [5600] Kovač, B., Mohraz, M., Heilbronner, E., Boekelheide, V., and Hopf, H. Photoelectron spectra of the cyclophanes, *J. Am. Chem. Soc.* **102**, 4314 (1980).
- [5601] Lappert, M. F., McCabe, R. W., MacQuitty, J. J., Pye, P. L., and Riley, P. I. Paramagnetic carbenemetal complexes. Part 1. Cationic chromium(I) complexes and the chemistry of their chromium(O) precursors and of related molybdenum(O) and tungsten(O) complexes, especially with bulky carbene ligands $C(OR')R[R=CH(SiMe₃)₂ or CH₂SiMe₃]$, *J. Chem. Soc. Dalton* 90 (1980).
- [5602] Yarbrough II, L. W., and Hall, M. B. Photoelectron spectra of substituted chromium, molybdenum, and tungsten pentacarbonyls. Relative π -acceptor and σ -donor properties of various phosphorus ligands, *Inorg. Chem.* **17**, 2269 (1978).
- [5603] Ratkovskii, I. A., and Butilin, B. A. Appearance potential of ions in the mass spectrum of LaPO₄, *Vest. Akad. Nauk* **5**, 115 (1973).
- [5604] Bischof, P., Gleiter, R., Dach, R., Enders, D., and Seebach, D. Zur Wechselwirkung einsamer Elektronenpaare in Δ^2 Tetrazenen. *Photoelektronenspektroskopische Untersuchungen*, *Tetrahedron* **31**, 1415 (1975).
- [5605] Bruckmann, P., and Klessinger, M. Photoelektronenspektren ungesättigter Systeme mit Cyclopropan- und Cyclobutanringen, *Angew. Chem.* **84**, 543 (1972).
- [5606] Bischof, P., Bosse, D., Gleiter, R., Kukla, M. J., de Meijere, A., and Paquette, L. A. Zur Frage der Homokonjugation in Triquinacen und 2a,4a,8a,8b-Tetrahydrocyclopent[cd]azulen. *Photoelektronenspektroskopische Untersuchungen*, *Chem. Ber.* **108**, 1218 (1975).
- [5607] Bruckmann, P., and Klessinger, M. Konjugative Wechselwirkungen des Cyclobutans, *Chem. Ber.* **111**, 944 (1978).
- [5608] Bouchoux, G. Ionisation et fragmentation en spectrométrie de masse VIII. Energies d'activation de deux fragmentations compétitives, dissociation de l'acétate de phényle et de l'acetanilide sous impact électronique, *Int. J. Mass Spectrom. Ion Phys.* **26**, 379 (1978).
- [5609] Bock, H., and Fuss, W. Notiz zu Ionisierungsenergien und Geometrie von Aminoboranen, *Chem. Ber.* **109**, 799 (1976).
- [5610] Bock, H., Hirabayashi, T., Mohmand, S., and Solouki, B. Instabile Zwischenprodukte in der Gasphase: Der thermische Zerfall von Carbonsäurechloriden RCOCl, *Angew. Chem.* **89**, 106 (1977).
- [5611] Bouchoux, G. Ionisation et fragmentation en spectrométrie de masse, *Org. Mass Spectrom.* **13**, 184 (1978).
- [5612] Bock, H., and Brähler, G. Oxidation und Reduktion methylthio-substituierter π -Systeme und die Elektronenverteilung in ihren Radikalionen, *Angew. Chem.* **89**, 893 (1977).
- [5613] Bischof, P., Gleiter, R., Hafner, K., Knauer, K. H., Spanget-Larsen, J., and Süß, H. U. Das Photoelektronenspektrum des 1,3,5-Tri-*tert*-butylpentals. Hinweis auf Bindungsalternanz im Grundzustand, *Chem. Ber.* **111**, 932 (1978).
- [5614] Klasinc, L., Trinajstić, N., and Knop, J. V. Application of photoelectron spectroscopy to biologically active molecules and their constituent parts. VIII. Thalidomide, *Int. J. Quantum Chem.* **7**, 403 (1980).
- [5615] Stockbauer, R., and Inghram, M. G. The fragmentation of propane and deuteropropane molecular ions, *J. Chem. Phys.* **65**, 4081 (1976).
- [5616] Williamson, A. D., and Beauchamp, J. L. Ion molecule reactions in vinyl fluoride by photoionization. Effects of vibrational excitation on major reaction pathways, *J. Chem. Phys.* **65**, 3196 (1976).
- [5617] Smyth, K. C., Schiavone, J. A., and Freund, R. S. Dissociative excitation of N₂ by electron impact: Translational spectroscopy of long-lived high- Rydberg fragment atoms, *J. Chem. Phys.* **59**, 5225 (1973).
- [5618] Schäfer, W., Schweig, A., and Mathey, F. Phospholes. Electronic structure, *J. Am. Chem. Soc.* **98**, 407 (1976).
- [5619] Ruščić, B., Kovač, B., Klasinc, L., and Güsten, H. Photoelectron spectroscopy of heterocycles. Fluorene analogues, *Z. Naturforsch.* **33a**, 1006 (1978).
- [5620] Dudin, A. V., Gorokhov, L. N., and Baluev, A. V. A study of the electron-impact ionization of chlorine trifluoride and its decomposition products by mass spectrometry, *Izvest. Akad. Nauk SSSR, Ser. Khim.* 2408 (1979).
- [5621] Andrews, G. D., Baldwin, J. E., and Gilbert, K. E. Photoelectron spectrum of bicyclo[2.1.0]pent-2-ene: Electronic destabilization of a homo[4n] annulene, *J. Org. Chem.* **45**, 1523 (1980).
- [5622] Engler, E. M., Kaufman, F. B., Green, D. C., Klots, C. E., and Compton, R. N. Ionization potentials and donor properties of selenium analogs of tetrathiafulvalene, *J. Am. Chem. Soc.* **97**, 2921 (1975).
- [5623] Hamada, Y., Hirakawa, A. Y., Tsuboi, M., and Ogata, H. Interaction between lone pair electrons on the nitrogen atoms in 1,5-diazabicyclo[3.2.1]octane, *Bull. Chem. Soc. Japan* **46**, 2244 (1973).
- [5624] Vovna, V. I., Lopatin, S. N., Pettsol'd, R., and Vilesov, F. I. Photoelectron spectra and electronic structure of some phosphoryl compounds, *Khim. Vys. Energ.* **9**, 9 (1975).
- [5625] Pasto, D. J., Fehlner, T. P., Schwartz, M. E., and Baney, H. F. On the orbital interactions of three-membered rings with π systems. Electronic structure of alkenylenecyclopropanes, *J. Am. Chem. Soc.* **98**, 530 (1976).
- [5626] Botter, R., and Carlier, J. Spectre de photoélectrons et calcul des facteurs de Franck-Condon pour H₂O, D₂O, HDO, *J. Electron Spectrosc. Relat. Phenom.* **12**, 55 (1977).
- [5627] Zverev, V. V., Vilesov, F. I., Vovna, V. I., Lopatin, S. N., and Kitaev, Y. P. Photoelectron spectra, *Izv. Akad. Nauk. SSSR, Ser. Khim.* **1975**, 1051 (1975).
- [5628] Barlos, K., and Nöth, H. Beiträge zur Chemie des Bors, CVIII Synthese und Konformation von N,N'-Bis(boryl)-N,N'-dimethylhydrazinen, *Z. Naturforsch.* **35b**, 125 (1980).
- [5629] Bock, H., and Kaim, W. Einelektronen-Oxidationen (H₃C)₃SiCH₂-substituierter benzole in der gasphase und in Lösung, *Chem. Ber.* **111**, 3552 (1978).
- [5630] Schäfer, W., Schweig, A., Bickelhaupt, F., and Vermeer, H. Photoelectron spectroscopy and conjugationdirect proof of the unusual sequence of the two highest occupied π -molecular orbitals in the phosphorin (phosphabenzenes) and the arsenin (arsabenzene) system, *Angew. Chem. Intern. Ed.* **11**, 924 (1972).
- [5631] Turk, J., and Shapiro, R. H. Formation of benzoyl ions: a complicated cleavage reaction, *Org. Mass Spectrom.* **6**, 189 (1972).
- [5632] Bock, H., and Wagner, G. "Electron lone pairs" in organic sulfides and disulfides, *Angew. Chem. Intern. Ed.* **11**, 150 (1972).
- [5633] Tomer, K. B., Turk, J., and Shapiro, R. H. Anchimeric assistance in electron-impact reactions: homoallylic systems, *Org. Mass. Spectrom.* **6**, 235 (1972).
- [5634] Guido, M., Gigli, G., and Balducci, G. Dissociation energy of CuCl and Cu₂Cl₂ gaseous molecules, *J. Chem. Phys.* **57**, 3731 (1972).
- [5635] Cocke, D. L., and Gingerich, K. A. Determination of the heats of atomization of the molecules RhC₂, RhC, and TiC₂ by high temperature mass spectrometry, *J. Chem. Phys.* **57**, 3654 (1972).

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